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Met125 is essential for maintaining the structural integrity of calmodulin's C-terminal domain

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We have used NMR and circular dichroism spectroscopy to investigate the structural and dynamic effects of oxidation on calmodulin (CaM), using peroxide and the Met to Gln oximimetic mutations. CaM is a Ca²⁺-sensitive regulatory protein that interacts with numerous targets. Due to its high methionine content, CaM is highly susceptible to oxidation by reactive oxygen species under conditions of cell stress and age-related muscle degeneration. CaM oxidation alters regulation of a host of CaM's protein targets, emphasizing the importance of understanding the mechanism of CaM oxidation in muscle degeneration and overall physiology. It has been shown that the M125Q CaM mutant can mimic the functional effects of methionine oxidation on CaM's regulation of the calcium release channel, ryanodine receptor (RyR). We report here that the M125Q mutation causes a localized unfolding of the C-terminal lobe of CaM, preventing the formation of a hydrophobic cluster of residues near the EF-hand Ca²⁺ binding sites. NMR analysis of CaM oxidation by peroxide offers further insights into the susceptibility of CaM's Met residues to oxidation and the resulting structural effects. These results further resolve oxidation-driven structural perturbation of CaM, with implications for RyR regulation and the decay of muscle function in aging.

Calmodulin (CaM) is a 16.7 kDa Ca²⁺-binding protein that converts intracellular [Ca²⁺] into cellular processes by targeting numerous target proteins, including voltage- and ligand-gated ion channels, kinases, phosphatases, transcription factors, and metabolic enzymes^{1–3}. Indeed, several missense mutations in CaM have been associated with severe cardiac arrhythmia⁴. CaM's Ca²⁺ sensitivity stems from the cooperatively paired EF-hand motifs in the N- and C-terminal domains, which are tethered by a flexible linker that facilitates CaM's binding to a variety of cellular targets^{2,5}. For most targets, this binding interaction depends on the calcium occupancy state, either Ca²⁺-free (apo) or Ca²⁺-bound (holo). Each domain contains two EF-hand motifs, allowing CaM to bind a total of four Ca²⁺ ions^{5,6}. The N-domain contains EF-hands I and II, with a ~ tenfold lower Ca²⁺ affinity (~ 10 μM) relative to the C-domain EF-hands III and IV^{7,8}. Thus changes in cytoplasmic [Ca²⁺] shift the distribution of apo-CaM and holo-CaM; this distribution shift can be disrupted by oxidative stress as a byproduct of CaM's high methionine (Met) content^{9,10}. Indeed, with the initiator Met residue excluded, eight of the nine Met residues play key roles in stabilizing CaM's Ca²⁺-binding pockets and participate in the structural changes and hydrophobic packing that enables CaM to bind and regulate a multitude of cellular targets (Fig. 1)¹⁰.

In addition to the hydrophobic environment around the Met residues, the EF-hand motifs contain paired aromatic residues in helices 1 and 4 that shift from nearly antiparallel orientation in the apo state to nearly perpendicular in the holo state^{11–16}. The stacking of these aromatic residues and stabilization of hydrophobic interactions throughout CaM's domains play key roles in the cooperativity of Ca²⁺ binding¹³. CaM's structural transition from the apo to holo state exposes hydrophobic clefts formed around CaM's Met residues (Fig. 1)^{13,16–18}. In each of the two domains, the hydrophobic cleft contains a cluster of four Met residues that adopt multiple conformations, facilitating CaM's conformational flexibility in binding to different target proteins^{19,20}. The N-domain hydrophobic core includes M37, M52, M72, and M73, while the C-domain contains M110, M125, M145, and M146. Notably, the CaM initiator Met, M1, is typically co-translationally cleaved in *E. coli*, which has been the case for previous studies^{21,22}. For this reason, the CaM literature has an inconsistent history of excluding the initiator Met in the residue number assignment. For this study, we included the initiator Met in the residue number assignment. The remaining Met residue, M77, is more solvent-exposed and resides in the linker region between the two terminals. The role of these Met residues, as well as the rest of the hydrophobic cores, in mediating CaM's

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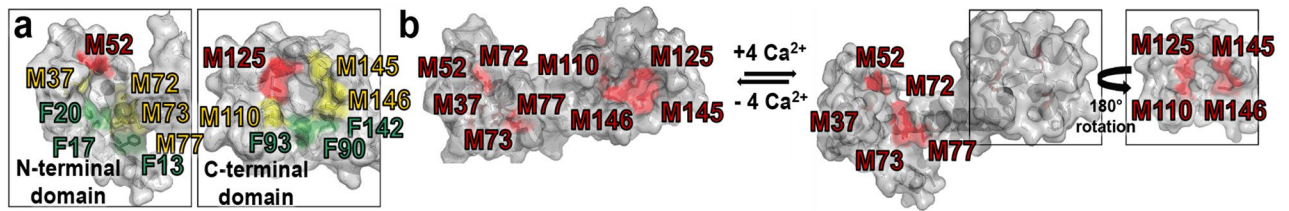


Figure 1. Hydrophobic interactions stabilize CaM's EF hand domains. (a) N- and C-terminal domains of Ca²⁺-bound (holo) CaM (PDB 1CLL) with Met residues shown in yellow, Phe residues shown in green and the mutation sites 52 and 125 shown in red. (b) Ca²⁺ binding alters the solvent exposure of CaM's Met residues (red; PDB 1DMO³⁰ and 1CLL³¹). For simplicity, residue assignment includes initiator Met. The image was generated using PyMOL Molecular Graphics System, Version 1.8³² (<https://pymol.org/2/>).

interaction with target peptides has been established through previous structural studies^{18,23–25}. Selective Met mutations in CaM to selenomethionines or other non-natural amino acids significantly alter CaM's binding affinity for several targets^{23,26–28}. In particular, site-specific Met to Leu mutations have been exploited to identify the role of individual Met residues in binding target peptides²⁹.

Functionally, Met oxidation has been shown to alter CaM's ability to regulate its target proteins^{29,33,34}. Excluding the initiator Met, all nine Met residues in CaM are susceptible to oxidation both *in vivo* and *in vitro*^{35–38}. Met oxidation is a reversible biological process that plays an important role in a variety of signaling and regulatory pathways. Reactive oxygen species (ROS), which generate the bulk of this oxidation, are produced as a result of normal metabolism. However, under conditions of oxidative stress, proteins become excessively oxidized, generating changes in cellular function^{36,39,40}. While some forms of oxidized CaM are selectively degraded by the 20S proteasome^{38,41}, CaM oxidation can also be reversed by methionine sulfoxide reductase (Msr), and CaM's oxidation level plays a role in cellular signaling³⁵.

Spectroscopic studies have shown that oxidation disrupts CaM's α -helical secondary structure^{37,42}. Since site-specific oxidation of CaM's Met residues is difficult to achieve experimentally, several groups have studied site-specific mutations that mimic the effects of oxidation. Particular attention has been devoted to the M125Q-CaM mutant since M125 is crucial for recognition and regulation of the ryanodine receptor (RyR)^{21,43–45}. A spin-labeling EPR study showed that H₂O₂ mediated oxidation induces a large shift in the conformational equilibrium of apo-CaM, and that M125Q presents a similar structural conversion, though with a lower population shifting to the oxidized conformation²¹. Thus, M125Q appears to partially mimic the fully oxidized state. This structural result complements functional studies of RyR, which have shown that the M125Q mutation partially mimics the oxidation-driven abolishment of the CaM-mediated effect on ryanodine receptor activity^{29,43}. This is consistent with X-ray crystallography showing contacts formed between M125 in CaM and W3620 within a peptide that spans RyR1 residues 3614–3643 (which form CaM binding domain 2⁴⁴)⁴⁵. Here, we have analyzed the structural and dynamic features of M125Q-CaM using circular dichroism (CD) and high-resolution NMR spectroscopy. For domain comparisons, we have also compared these effects with the equivalent mutation in the N lobe, M52Q-CaM, and with the effects of global oxidation of CaM with H₂O₂. In our investigation of H₂O₂-mediated oxidation of CaM, we have also resolved the time course for site-specific oxidation in the absence of free Ca²⁺. These results provide structural insight into the functional consequences of site-directed oxidation in CaM, with particular relevance to dysregulation of RyR.

Results

M125Q mutation causes unfolding in CaM's C-terminal domain. We first analyzed the effects of the M125Q mutation on CaM's secondary structure using CD spectroscopy. Upon introduction of the M125Q mutation, the dichroic profiles show a decrease in the molar ellipticity at 222 nm, indicating a loss of α -helical secondary structure (Supplementary Fig. 1a). Deconvolution of the CD spectra indicates a decrease in both the regular helix content (Helix1), and the total helical content for M52Q-CaM and M125Q-CaM relative to WT-CaM (Supplementary Fig. 1b). To identify the domains affected by the mutation and the cause of this shift in helical content, we analyzed the amide backbone fingerprint of U-¹⁵N WT-CaM and M125Q-CaM with NMR spectroscopy. In conditions with free Ca²⁺ buffered by EGTA, to promote apo-CaM state, the [¹H-¹⁵N] HSQC spectra for WT-CaM and M125Q-CaM show that the M125-specific oximimetic mutation causes chemical shift perturbations in the protein fingerprint (Supplementary Fig. 2a). More specifically, significant chemical shift changes were localized to the C-terminal domain residues of CaM; while the resonances for the dynamic linker and the N-terminal lobe remain largely unperturbed (Fig. 2). In fact, the resonances assigned to CaM's N-domain and linker region consistently overlay between the two spectra, while a few residues in the C-domain of M125Q move toward the 8–9 ppm region of the ¹H dimension relative to WT-CaM, suggesting a shift toward a disordered conformation for those residues (Supplementary Fig. 2a). In addition, several resonances of the C-domain are visibly broadened (Supplementary Fig. 2a). This combination of line broadening and collapse to the 8–9 ppm range prevented complete assignment of M125Q-CaM's C-domain, particularly for residues immediately surrounding the mutation site (Figs. 2, 3). These chemical shift perturbations suggest that upon modification of M125, the C-domain undergoes localized structural destabilization and increased molecular motions, whereas the N-domain remains essentially unperturbed.

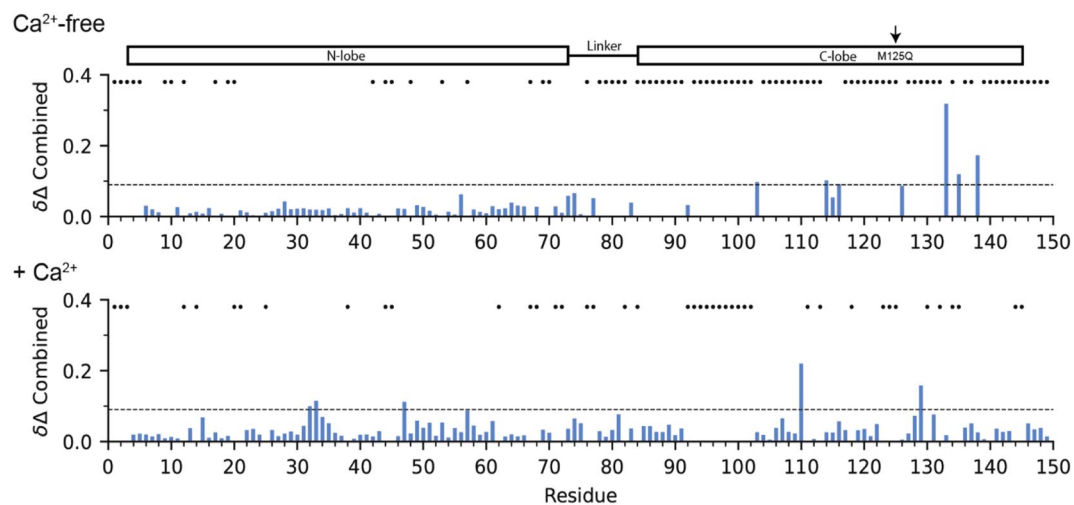


Figure 2. M125Q mutation alters CaM's C-terminal domain. Combined $^1\text{H}/^{15}\text{N}$ chemical shift perturbations for M125Q-CaM in the absence and presence of Ca^{2+} , relative to WT-CaM based on $[^1\text{H}, ^{15}\text{N}]$ HSQC spectra. The dashed line represents two standard deviations (0.09) within the data. Asterisks indicate missing peaks or unassigned residues. NMR Spectra were analyzed using NMRFAM-Sparky⁴⁶.

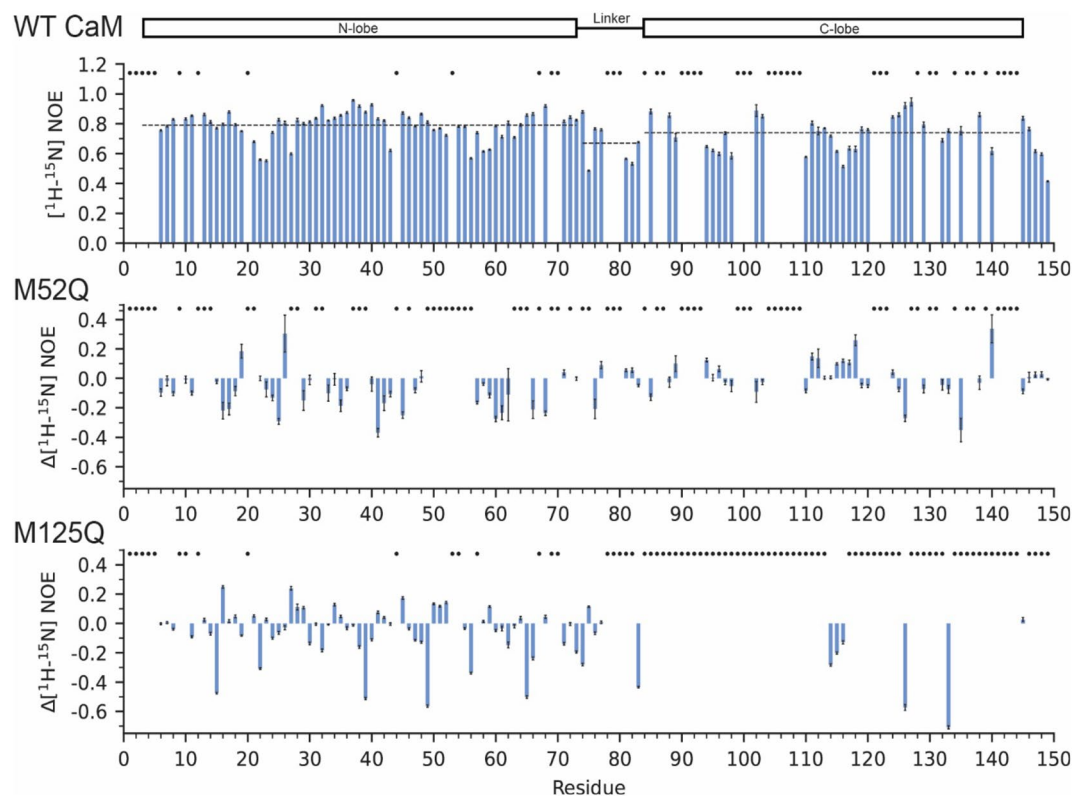


Figure 3. $[^1\text{H}, ^{15}\text{N}]$ Backbone NOE values in the absence of Ca^{2+} . $[^1\text{H}, ^{15}\text{N}]$ backbone heteronuclear NOE values for WT-CaM in the absence of Ca^{2+} (top). Relative differences in NOEs for Ca^{2+} -free M52Q-CaM, and M125Q-CaM are shown in the lower panels. NOE values for each residue were determined using the ratio between the peak intensities of the saturated and unsaturated spectra. Error bars represent the uncertainty in each NOE value, determined relative to the baseline noise level in the saturated and unsaturated spectra. Dashed lines indicate the average NOE of each domain. NMR Spectra were analyzed using NMRFAM-Sparky⁴⁶.

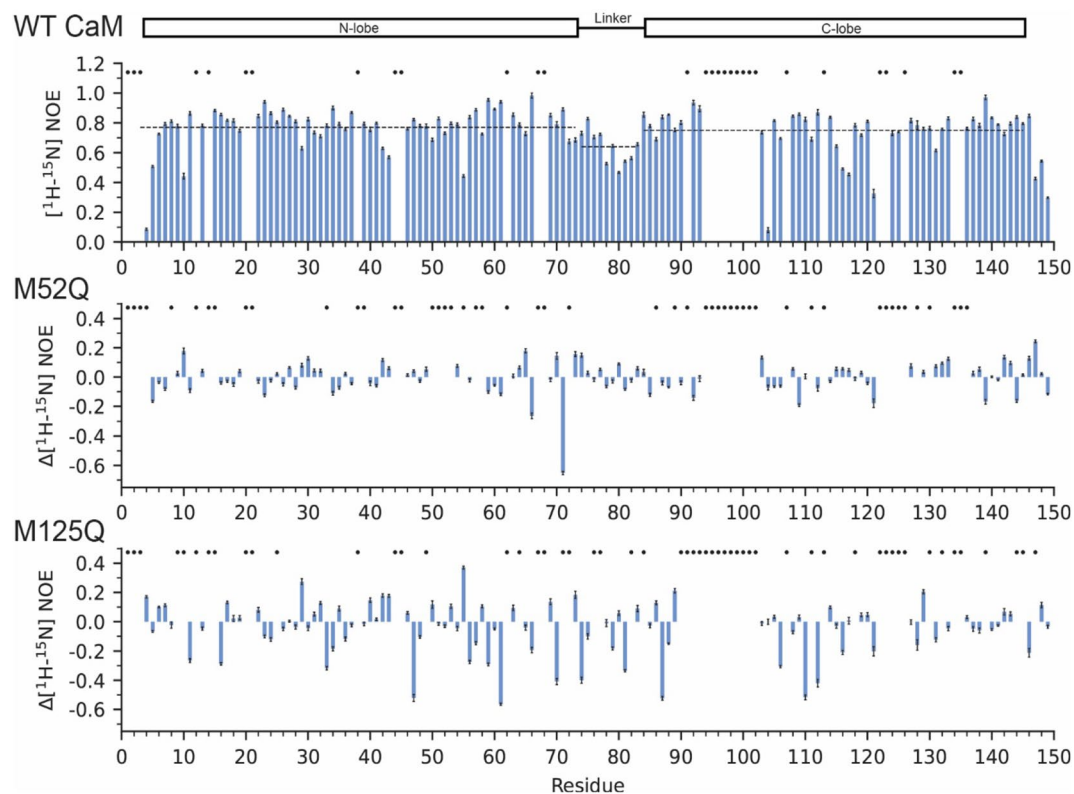


Figure 4. $[^1\text{H}, ^{15}\text{N}]$ Backbone NOE values in the presence of saturating Ca^{2+} . $[^1\text{H}, ^{15}\text{N}]$ backbone heteronuclear NOE values for WT CaM in the presence of 6 mM CaCl_2 (top). Relative differences in NOEs for M52Q-CaM, and M125Q-CaM are shown in the lower panels. NOE values for each residue were determined by taking the ratio between the peak intensities of the saturated and unsaturated spectra. Error bars represent the uncertainty in each NOE value, determined relative to the baseline noise level in the saturated and unsaturated spectra. Dashed lines indicate the average NOE of each domain. NMR Spectra were analyzed using NMRFAM-Sparky⁴⁶.

To test whether Ca^{2+} binding would restore the native folding of the C-terminal domain, we carried out experiments with saturating Ca^{2+} added, and followed the amide fingerprint using $[^1\text{H}, ^{15}\text{N}]$ HSQC experiments (Supplementary Fig. 2b). In the NMR spectra, WT-CaM has a very distinct and global response to Ca^{2+} binding, reflecting an overall stabilization of the secondary structure as described previously¹⁴. In the presence of saturating $[\text{Ca}^{2+}]$, the residues in WT-CaM progressively move toward a fully Ca^{2+} -saturated form that exhibits sharp, well-dispersed peaks (Supplementary Fig. 2b). Furthermore, a few amides in the N-domain of M125Q-CaM show subtle chemical shift perturbations, and several residues exhibit increased line broadening relative to Ca^{2+} -bound WT-CaM (Supplementary Fig. 2b). The profile corresponding to several C-domain residues in M125Q-CaM differed from that of WT-CaM (Fig. 2), suggesting that the M125Q mutation disrupts the conformational equilibrium adopted by the C-domain in the presence of Ca^{2+} . Overall, the chemical shift analysis supports the hypothesis that the M125Q mutation causes destabilization of the C-terminal globular domain, with disruption of the hydrophobic packing around the EF hand motifs that is not rescued by the addition of Ca^{2+} .

To further test this hypothesis, we carried out $[^1\text{H}-^{15}\text{N}]$ heteronuclear NOE experiments to monitor CaM's backbone dynamics in the ps-ns time scale⁴⁷. As expected from previous studies, WT-CaM exhibits backbone NOE values between 0.2 and 0.9 (Figs. 3, 4). Accordingly, CaM's linker region and C-domain exhibit slightly lower NOEs (Fig. 3), indicating that these regions undergo faster structural dynamics than the N-terminal domain. In Ca^{2+} -buffered conditions, large reductions in NOE values are observed across the entire protein, with the largest changes attributed to the few residues of the C-domain that could be assigned (Figs. 3, 4). This behavior is typical of small unfolded peptides or intrinsically disordered domains, and indicates widespread conformational and structural instability as a result of the M125Q mutation, consistent with prior EPR studies²¹. With saturating $[\text{Ca}^{2+}]$, the localized changes are also observed in the C-domain's amide fingerprint (i.e. peak doubling and broadening) for M125Q-CaM, as are the widespread reductions in NOE values, particularly in the helices preceding the Ca^{2+} binding sites and hydrophobic core (Fig. 4, Supplementary Fig. 3 and Supplementary Fig. 4). Furthermore, expanded regions of residues G114, G133 and G135 after Ca^{2+} titrations show peak intensity at their apo positions, indicating two distinct structural states at saturating Ca^{2+} and partial impairment of Ca^{2+} -binding to the C-domain (Supplementary Fig. 3). Attempts were made to increase the concentration of free Ca^{2+} (> 4.5 mM) in the presence of EGTA, but this caused significant reductions in spectral quality (data not shown). In contrast, the resonances associated with the N-domain undergo the Ca^{2+} -driven structural transitions typical to WT-CaM, reaching the likely full-holo state at $[\text{Ca}^{2+}]$ at concentrations lower than 4.5 mM (Supplementary Fig. 3). These

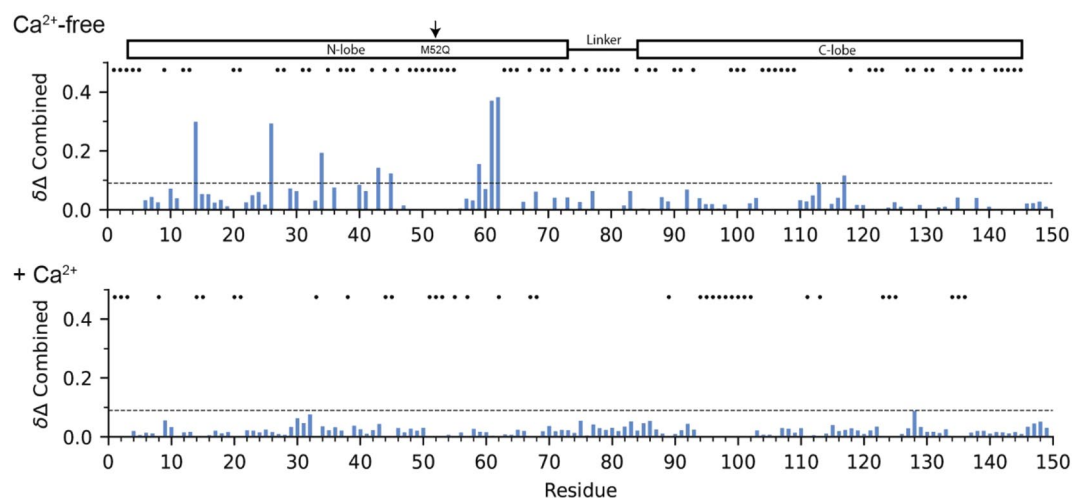


Figure 5. M52Q mutation perturbs CaM's N-terminal domain in the absence of Ca²⁺. Combined ¹H/¹⁵N chemical shift perturbations for M52Q-CaM in the absence and presence of Ca²⁺, relative to WT-CaM based on [¹H, ¹⁵N] HSQC spectra. The dashed line represents two standard deviations (0.09) within the data. Asterisks indicate missing peaks or unassigned residues. NMR Spectra were analyzed using NMRFAM-Sparky⁴⁶.

patterns support the hypothesis that the M125Q mutation induces localized structural changes to the C-domain and global dynamic changes in both the Ca²⁺-free and bound forms.

M52Q-CaM exhibits divergent behavior to M125Q-CaM. To determine whether a N-domain mutation equivalent to M125Q destabilizes the N-domain, we engineered the M52Q-CaM mutant. M52Q-CaM has been previously studied in the context of its binding to RyR under reducing conditions⁴⁸. M52 is situated in the N-domain and belongs to the hydrophobic cluster of residues in a location equivalent to M125 in the C-domain. While CD analysis indicates that M52Q-CaM exhibits a loss of α -helical content relative to WT-CaM, this loss is not as severe as in M125Q-CaM (Supplementary Fig. 1). In the absence of Ca²⁺, M52Q-CaM's chemical shift perturbations are isolated to residues in close structural proximity to M52, and these perturbations are fully rescued by the addition of Ca²⁺ (Fig. 5 and Supplementary Fig. 5). In addition, M52Q-CaM reaches Ca²⁺ saturation at lower concentrations of Ca²⁺ than M125Q-CaM (Supplementary Fig. 3b). [¹H, ¹⁵N] heteronuclear NOE experiments on M52Q-CaM confirmed these observations with lower backbone NOE values for M52Q-CaM's N-domain residues relative to that of WT-CaM (average of 0.69 compared to 0.79 for WT), which was only in the absence, not presence, of Ca²⁺ (Figs. 3, 4). The analysis of M52Q-CaM suggests that, despite the sequence similarity and conserved structural elements between CaM's N- and C-domains, selective modification of the N-domain has a lower impact on structural responses to Ca²⁺, relative to modification of the C-domain. This agrees with our previous report that indirectly demonstrated this using [³H]-ryanodine binding to RyR²⁹.

M125Q-CaM exhibits hallmarks of partially oxidized WT-CaM. It has been reported that both M125Q-CaM and oxidized WT CaM (oxCaM; oxidized with H₂O₂) show a greatly reduced affinity for RyR relative to WT-CaM^{29,43}. To resolve the structural basis for the functional effects of M125Q-CaM, M52Q-CaM, and oxCaM, we used CD and NMR to obtain the structural fingerprint of these CaM mutants relative to WT-CaM. By tracking the molar ellipticity at 222 nm, we show that addition of H₂O₂ induces a loss of α -helical secondary structure of WT-CaM in the absence of Ca²⁺ (Fig. 6 and Supplementary Fig. 6a). This is in agreement with previous studies showing that H₂O₂ causes a significant loss of α -helical secondary structure in the absence of Ca²⁺^{37,49,50}. Using NMR, we carried out parallel studies following the time dependence of oxidation on the chemical shifts in the amide fingerprint for WT-CaM in the absence and presence of Ca²⁺ (Fig. 6).

In the first three hours after the addition of H₂O₂, spectra of WT-CaM in the absence of Ca²⁺ show the greatest changes in chemical shift and peak intensity in all of the resonances corresponding to the C-domain Met residues (Fig. 6b and Supplementary Fig. 6), which is consistent with previous *in vivo* and *in vitro* assays that report the sensitivity of these sites to oxidation in the absence of Ca²⁺^{37,51}. In the N-domain, the response varied between the different Met residues. Specifically, M52 and M73 exhibited oxidation susceptibility similar to the C-domain Met residues, while M37 and M72 proved least sensitive to oxidative insult, and did not become fully oxidized until 14 and 11 h of H₂O₂ exposure, respectively (Fig. 6b and Supplementary Fig. 6). Comparison of the oxidation profiles and NMR spectra for M52 and M125 revealed that although in the first three hours of H₂O₂ exposure both residues exhibited a similar susceptibility to oxidation, M125 is slightly more susceptible and became fully oxidized following five hours of H₂O₂ exposure compared to eight hours for M52 (Fig. 6c and Supplementary Fig. 6). Interestingly, the solvent-exposed M77 exhibited a similar susceptibility to oxidation as M145 and M146 during the first three hours of H₂O₂, but took 14 h to become completely oxidized (Fig. 6c, Supplementary Fig. 6). During the course of H₂O₂ exposure in the absence of Ca²⁺, most Met residues exhibit gradual changes in their chemical shifts, indicating fast exchange between oxidized states of WT-CaM, probably reflecting gradual changes

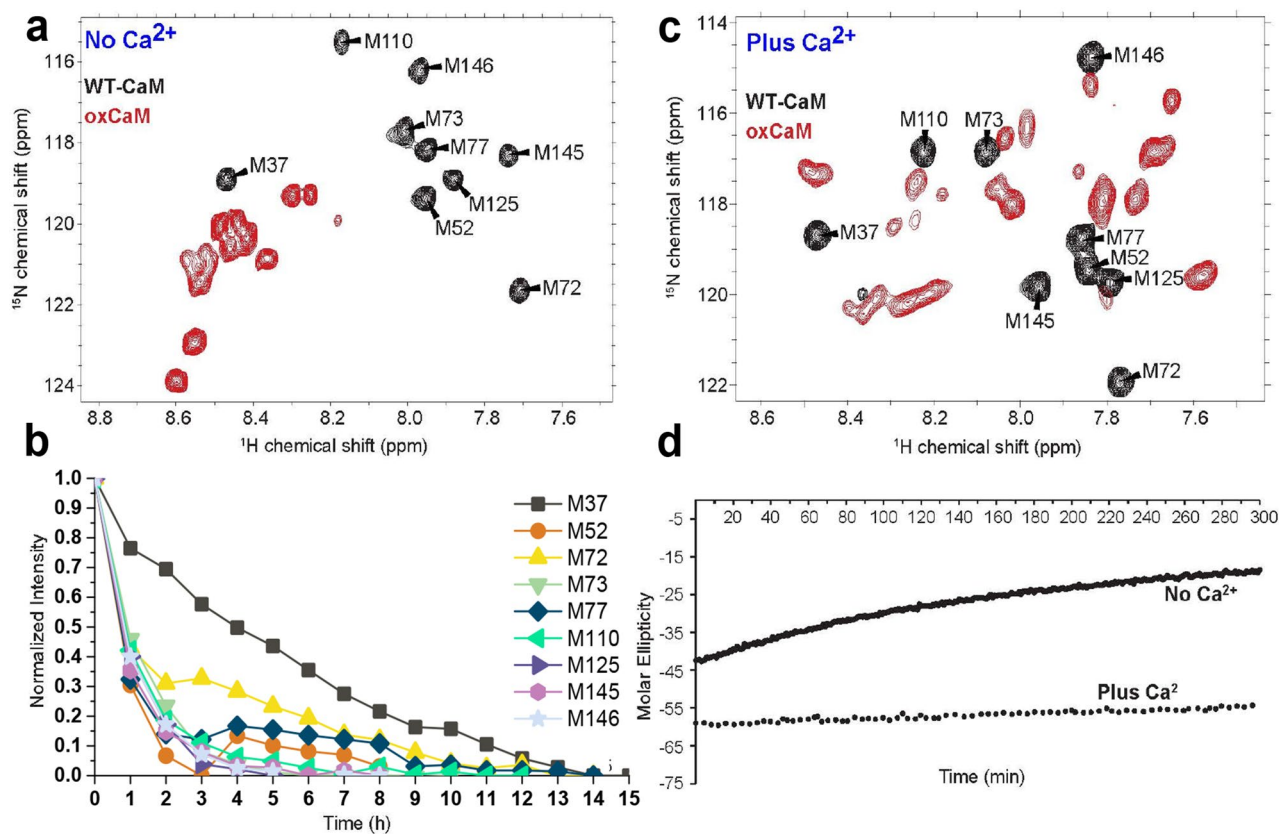


Figure 6. Ca²⁺ protects against structural degradation from oxidation. (a) In the absence of Ca²⁺, [¹H, ¹⁵N] HSQC overlay of apo ¹³C/¹⁵N Met labeled WT-CaM before and after exposure to 50 mM H₂O₂. (b) Change in normalized [¹H, ¹⁵N] HSQC peak intensity for each Met residue in apo CaM-WT illustrating the increased susceptibility of C-terminal Met residues to oxidation. The complete oxidation of CaM corresponds to the plateau of the intensity of the resonances and was confirmed by MALDI. (c) In the presence of saturating Ca²⁺, [¹H, ¹⁵N] HSQC overlay of Ca²⁺-bound ¹³C/¹⁵N Met labeled WT-CaM before and after exposure to 50 mM H₂O₂. All NMR spectra were acquired on a Bruker 900 MHz spectrometer at 25 °C. (d) CD spectra tracking the loss of α-helical secondary structure for WT-CaM in the absence and presence of Ca²⁺, following exposure to 50 mM H₂O₂ for 5 h. CD spectra were acquired at 25 °C and tracked the α-helical signal at 222 nm. Spectral images were generated using NMRFAM-Sparky⁴⁶, and graphs in panels (b) and (d) were generated using Origin 2015 (<https://www.originlab.com/>) and Excel 2016 (<https://office.microsoft.com/excel>), respectively.

in structure induced by the oxidation of nearby Met residues (Supplementary Fig. 6D). This is in contrast with the behavior of Met residues in the presence of Ca²⁺ (Supplementary Fig. 7), where residues showed the presence of multiple conformational states under slow exchange. The full amide fingerprints of oxCaM in the absence and presence of Ca²⁺ also show significant differences in the dispersion of the residues. With complete oxidation of CaM in the absence of Ca²⁺, the ¹H chemical shifts in both the N- and C-domain residues of WT-CaM collapse toward the 8–9 ppm range, indicating that CaM's structure is largely denatured (Fig. 6a and Supplementary Fig. 6), while the resonances in Ca²⁺-saturated oxCaM remain well-dispersed, indicating a structured protein (Fig. 6c and Supplementary Fig. 7).

In the presence of saturating Ca²⁺, the structural stability of hydrophobic interactions in the EF hand motif appears to confer a protective effect in preserving CaM's secondary structure. While all nine of CaM's Met residues (excluding the initiator Met) are still susceptible to oxidation in the presence of Ca²⁺, instead of adopting the fully denatured structure observed in the absence of Ca²⁺, oxCaM retains its α-helical structure and shifts to a different conformation (Fig. 6c, d, and Supplementary Fig. 7). In saturating [Ca²⁺], all Met residues are highly susceptible to oxidation upon exposure to H₂O₂ (Supplementary Fig. 7). The ability of Ca²⁺ binding to rescue α-helical structure in oxCaM is demonstrated in Fig. 7.

Although M52Q-CaM exhibits strong similarities to WT-CaM, particularly in the presence of Ca²⁺, M125Q-CaM exhibits hallmarks of oxCaM in both the presence and absence of Ca²⁺. In particular, the collapse of residues into the 8–9 ppm range indicates a similar less-ordered state. In addition, comparison of the amide fingerprints for Ca²⁺-bound oxCaM and Ca²⁺-bound M125Q-CaM show several similarities in both chemical shift and line broadening patterns (Supplementary Fig. 8).

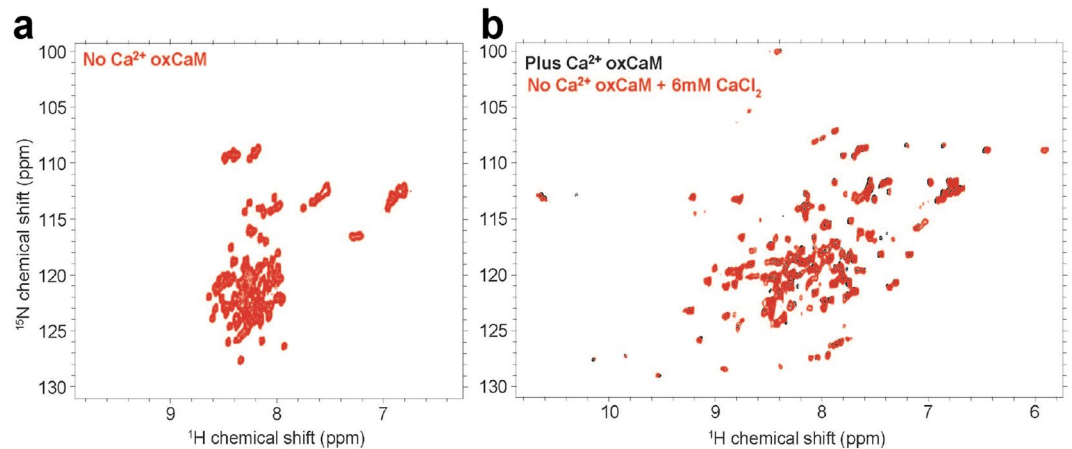


Figure 7. Ca^{2+} rescues secondary structure in fully oxidized CaM. **(a)** $[\text{}^1\text{H}, \text{}^{15}\text{N}]$ HSQC of fully oxidized WT-CaM in the absence of Ca^{2+} . **(b)** $[\text{}^1\text{H}, \text{}^{15}\text{N}]$ HSQC overlay of fully oxidized WT-CaM in the presence of Ca^{2+} , and initial absence of Ca^{2+} during oxidation with the addition of 6 mM CaCl_2 after full oxidation. Spectra were acquired on a Bruker 900 MHz spectrometer at 25 °C. Spectral images were generated using NMRFAM-Sparky⁴⁶.

Discussion

This study provides direct evidence that the C-domain of CaM is a sensor for oxidative processes. In particular, the M125Q mutation, located in the hydrophobic cluster of the C-terminal lobe, partially mimics the effect of oxidative insults on CaM. This agrees with the previously established impact of the M125Q mutation on CaM-binding and modulation of targets, particularly smooth muscle myosin light chain kinase, CaM-dependent protein kinase IIa, CaM-dependent protein kinase IV, and RyR⁵². Moreover, it has been shown that the M125Q mutation weakens the interactions with RyR in a Ca^{2+} -dependent manner, which is consistent with the altered response to Ca^{2+} observed in this study^{29,43}. Recent EPR experiments focusing on the conformational equilibrium between closed-to-open conformations of WT-CaM and M125Q-CaM showed that the populations of these states are significantly affected by this single mutation²¹. That study also showed that more dramatic changes are observed when CaM is completely oxidized by treatment with 50 mM H_2O_2 for 24 h. In the present study, NMR provides direct, detailed structural insight into the effects of this mutation, which induces partial melting in the C-domain that disrupts EF-hands III and IV and the high-affinity binding of Ca^{2+} . The changes in CaM's structure, target regulation, and Ca^{2+} binding due to the M125Q mutation are probably the result of disruptions in the hydrophobic interactions of CaM's C-domain and the resulting conformational transition of the C-domain toward a partially folded state. Remarkably, almost identical behavior is observed for the C-domain of CaM upon exposure (> 2 h) to 50 mM H_2O_2 . In the presence of saturating $[\text{Ca}^{2+}]$, M125 is surrounded by hydrophobic residues that contribute to the hydrophobic surface that is involved in CaM's binding to several target proteins (Fig. 1). The mutation of Met to Gln introduces a bulkier amino acid sidechain, preventing the formation of the hydrophobic core that holds together the C-terminal lobe. While M145 and M146 likely cause small changes in the structure upon oxidation, the oxidation of M125 to Met sulfoxide, as we mimic with the M125Q mutation, would very likely destabilize the tertiary interactions within the hydrophobic cluster formed by L106, M110, Met125, V122, and L117. Indeed, M145 and M146 oxidation is unlikely to have a causative role in the observed structural disorder, given that we previously demonstrated using molecular dynamics studies of a CaM fragment (V137-T147) that selective oxidation of M145 and M146 stabilizes inter-residue interactions via an interaction with Y139⁵³. In contrast to our previous study, here we investigate the full length CaM protein, and demonstrate the modification of the Met thioether to a sulfoxide within the hydrophobic cluster of the C-terminal domain causes a significant destabilization of the tertiary structure that melts into a partially folded state. In accord with M145 and M146 having a minimal role in this destabilization, Anbanandam et al.⁴² found, using NMR, that a CaM mutant with selectively oxidized M145 and M146 residues sustained only nominal disruption to secondary structure.

Since the C-domain is central for the binding to RyR and a variety of kinases, partial unfolding of this domain provides a compelling explanation for the loss of affinity of CaM for these targets as well as the similarity between M125Q-CaM and oxCaM. We conclude that the main role of M125 is to maintain the structural integrity of CaM's C-domain, providing a rationale for why oxidation at Met residues, such as M125, causes such large changes in CaM's structure and disruptions in the regulation of CaM's cellular targets. This new insight for M125 is further supported by the comparison of M125Q-CaM to partially oxidized forms of WT-CaM, suggesting that oxidation of CaM perturbs its structure such that it can no longer bind its targets and is tagged for selective degradation by the 20S proteasome^{38,41}.

In conclusion, we have established a direct correlation between the unfolding of the C-terminal domain, caused either by the M125Q mutation or by oxidation, and the functional effects of these modifications toward CaM targets such as RyR⁴³ and smooth muscle myosin light chain kinase, CaM-dependent protein kinase IIa,

and CaM-dependent protein kinase IV⁵². This study emphasizes the importance of the C-terminal domain of CaM in target recognition and binding.

Methods

Expression and purification of CaM. WT-CaM, M52Q-CaM, and M125Q-CaM were expressed using BL21(DE3) *E. coli* in LB and M9 media and purified using Phenyl-Sepharose CL-4B resin (Sigma Aldrich)⁴³. NMR samples were uniformly labeled using ¹⁵NH₄Cl (Sigma Aldrich) and ¹³C-D-glucose (Cambridge Isotope Laboratories Inc). Selectively ¹³C/¹⁵N Met labeled samples of WT-CaM for NMR were prepared using established protocols^{54–56}. Briefly, CaM was expressed in M9 media containing 0.2 mg/mL of uniformly ¹³C/¹⁵N labeled Met (Sigma Aldrich), 0.9 mg/mL of unlabeled Lys, Thr, and Ile, and 0.6 mg/mL of unlabeled Ala, Arg, Asn, Asp, Cys, Glu, Gln, Gly, His, Phe, Pro, Ser, Trp, Tyr, and Val. NMR samples were prepared by buffer exchange using Amicon Centrifugal Filter Units with a MWCO of 3 kDa with a final concentration of 0.4 mM CaM determined by absorbance at 280 nm. NMR buffers contained 20 mM imidazole, 100 mM KCl, 1 mM NaN₃, and either 1.5 mM EGTA or 6 mM CaCl₂ at pH 6.5 to promote the apo- and holo-CaM structural states, respectively.

NMR spectroscopy. NMR spectra were obtained on Varian 600 MHz, Bruker 850 MHz, or Bruker 900 MHz spectrometers (Minnesota NMR Center). Backbone assignments of WT-CaM were determined using triple resonance HNCACB⁵⁷ and CBCA(CO)NH⁵⁸ experiments in combination with the PINE server developed by the NMR Facility at Madison^{59–61}. Assignments obtained using the PINE server were compared to manual assignments and checked using PINE-SPARKY⁶⁰. TOCSY^{62,63} and NOESY-HSQC spectra were used to resolve ambiguities in assignments. NMR data was processed and analyzed using NMRPipe⁶⁴, Sparky⁶⁰, and NMRView. The resonance assignments correspond well with the previously published data by Urbauer et al.⁵⁹.

NMR Ca²⁺ titrations were performed by adding 409.6 μM, 1.6 mM, 2.5 mM, and 6 mM CaCl₂ to 0.4 mM samples of WT-CaM, M52Q-CaM and M125Q-CaM in 1.5 mM EGTA, for a final free [Ca²⁺] of 2.3 M, 157 M, 1 mM, and 4.5 mM (as determined by MaxChelator⁶⁵), respectively. Structural changes in CaM's response to Ca²⁺ were tracked using chemical shift perturbations in [¹H-¹⁵N] HSQC and backbone dynamics was measured using [¹H-¹⁵N] heteronuclear NOE experiments⁴⁷.

WT-CaM oxidation was induced by addition of 50 mM (final) H₂O₂. A time course of [¹H-¹⁵N] HSQC spectra were collected consecutively for 14–21 h and the chemical shift perturbations were tracked using NMRView's titration analysis software.

Chemical shift perturbations of M52Q-CaM and M125Q-CaM relative to WT-CaM were normalized to the spectral dispersions of the ¹H and ¹⁵N dimensions using the following equation⁶⁶:

$$\Delta\delta_{combined} = \sqrt{(\delta_{1H,WT} - \delta_{1H,Mutant})^2 + \left(\frac{\delta_{15N,WT} - \delta_{15N,Mutant}}{5}\right)^2}, \quad (1)$$

where $\Delta\delta_{combined}$ is the difference in chemical shift relative to WT-CaM. [¹H, ¹⁵N] heteronuclear NOE values for WT-CaM, M52Q-CaM, and M125Q-CaM were determined by taking the ratio between peak intensities in the proton saturated (I_{sat}) and unsaturated (I_{unsat}) spectra as previously described^{47,66}:

$$[{}^1H, {}^{15}N]NOE = \frac{I_{sat}}{I_{unsat}} \quad (2)$$

The uncertainty in the NOE measurement (σ_{HN-NOE}) was determined for each residue^{47,66}:

$$\sigma_{HN-NOE} = [{}^1H, {}^{15}N]NOE \sqrt{\left(\frac{\sigma_{I_{sat}}}{I_{sat}}\right)^2 + \left(\frac{\sigma_{I_{unsat}}}{I_{unsat}}\right)^2}, \quad (3)$$

where $\sigma_{I_{sat}}$ and $\sigma_{I_{unsat}}$ represent the baseline noise levels in the proton saturated and unsaturated spectra, respectively.

Circular dichroism spectroscopy. CD spectra of WT-CaM, M52Q-CaM, and M125Q-CaM were recorded from 280 to 190 nm using a JASCO J-815 spectrophotometer (University of Minnesota Biophysical Technology Center). Spectra were recorded at 25 °C in a buffer containing 2 mM HEPES, 50 mM NaCl, and 1 mM DTT at pH 7.4 in a quartz cuvette with a path length of 1 mm⁶⁷. Samples for CD were prepared in the presence of 2 mM EDTA or 6 mM CaCl₂. Baseline-subtracted spectra were recorded at 20 nm/min, signal-averaged six times, and reported using the molar ellipticity. Spectra were deconvoluted using the CDSSTR algorithm^{68–70} and SP175 reference database⁷¹ via the DichroWeb server^{72,73}. The time-dependent oxidation of WT-CaM was obtained by adding a final H₂O₂ concentration of 50 mM. Following the addition of H₂O₂, the α-helical signal at 222 nm was monitored for five hours.

Data availability

The authors declare that all data supporting the findings of this study are available within the article and its supplementary information file.

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References

- Urrutia, J. *et al.* The crossroad of ion channels and calmodulin in disease. *Int. J. Mol. Sci.* <https://doi.org/10.3390/ijms20020400> (2019).
- Tidow, H. & Nissen, P. Structural diversity of calmodulin binding to its target sites. *FEBS J.* **280**, 5551–5565. <https://doi.org/10.1111/febs.12296> (2013).
- Chin, D. & Means, A. R. Calmodulin: a prototypical calcium sensor. *Trends Cell Biol.* **10**, 322–328. [https://doi.org/10.1016/s0962-8924\(00\)01800-6](https://doi.org/10.1016/s0962-8924(00)01800-6) (2000).
- Jensen, H. H., Brohus, M., Nyegaard, M. & Overgaard, M. T. Human calmodulin mutations. *Front. Mol. Neurosci.* **11**, 396. <https://doi.org/10.3389/fnmol.2018.00396> (2018).
- Grabarek, Z. Structure of a trapped intermediate of calmodulin: calcium regulation of EF-hand proteins from a new perspective. *J. Mol. Biol.* **346**, 1351–1366. <https://doi.org/10.1016/j.jmb.2005.01.004> (2005).
- Bhattacharya, S., Bunick, C. G. & Chazin, W. J. Target selectivity in EF-hand calcium binding proteins. *Biochem. Biophys. Acta.* **1742**, 69–79. <https://doi.org/10.1016/j.bbamcr.2004.09.002> (2004).
- Valeyev, N. V., Bates, D. G., Heslop-Harrison, P., Postlethwaite, I. & Kotov, N. V. Elucidating the mechanisms of cooperative calcium-calmodulin interactions: a structural systems biology approach. *BMC Syst. Biol.* **2**, 48. <https://doi.org/10.1186/1752-0509-2-48> (2008).
- VanScyoc, W. S. & Shea, M. A. Phenylalanine fluorescence studies of calcium binding to N-domain fragments of *Paramecium* calmodulin mutants show increased calcium affinity correlates with increased disorder. *Protein Sci.* **10**, 1758–1768. <https://doi.org/10.1110/ps.11601> (2001).
- Oda, T. *et al.* Oxidation of ryanodine receptor (RyR) and calmodulin enhance Ca release and pathologically alter RyR structure and calmodulin affinity. *J. Mol. Cell. Cardiol.* **85**, 240–248. <https://doi.org/10.1016/j.yjmcc.2015.06.009> (2015).
- Xiong, Y. *et al.* High-affinity and cooperative binding of oxidized calmodulin by methionine sulfoxide reductase. *Biochemistry* **45**, 14642–14654. <https://doi.org/10.1021/bi0612465> (2006).
- Babu, Y. S., Bugg, C. E. & Cook, W. J. Structure of calmodulin refined at 2.2 Å resolution. *J. Mol. Biol.* **204**, 191–204. [https://doi.org/10.1016/0022-2836\(88\)90608-0](https://doi.org/10.1016/0022-2836(88)90608-0) (1988).
- Babu, Y. S. *et al.* Three-dimensional structure of calmodulin. *Nature* **315**, 37–40. <https://doi.org/10.1038/315037a0> (1985).
- Lakowski, T. M., Lee, G. M., Okon, M., Reid, R. E. & McIntosh, L. P. Calcium-induced folding of a fragment of calmodulin composed of EF-hands 2 and 3. *Protein Sci.* **16**, 1119–1132. <https://doi.org/10.1110/ps.072777107> (2007).
- Kuboniwa, H. *et al.* Solution structure of calcium-free calmodulin. *Nat. Struct. Biol.* **2**, 768–776. <https://doi.org/10.1038/nsb0995-768> (1995).
- Schumacher, A. M., Schavocky, J. P., Velentza, A. V., Mirzoeva, S. & Watterson, D. M. A calmodulin-regulated protein kinase linked to neuron survival is a substrate for the calmodulin-regulated death-associated protein kinase. *Biochemistry* **43**, 8116–8124. <https://doi.org/10.1021/bi049589v> (2004).
- Ferrer-Acosta, Y., Rodriguez Cruz, E. N., Vaquer Adel, C. & Vega, I. E. Functional and structural analysis of the conserved EFhd2 protein. *Protein Pept. Lett.* **20**, 573–583. <https://doi.org/10.2174/0929866511320050011> (2013).
- Yun, C. H. *et al.* Structure of potato calmodulin PCM6: the first report of the three-dimensional structure of a plant calmodulin. *Acta Crystallogr. D Biol. Crystallogr.* **60**, 1214–1219. <https://doi.org/10.1107/S0907444904009771> (2004).
- O'Neil, K. T. & DeGrado, W. F. A predicted structure of calmodulin suggests an electrostatic basis for its function. *Proc. Natl. Acad. Sci. U. S. A.* **82**, 4954–4958. <https://doi.org/10.1073/pnas.82.15.4954> (1985).
- Ikura, M. *et al.* Solution structure of a calmodulin-target peptide complex by multidimensional NMR. *Science* **256**, 632–638. <https://doi.org/10.1126/science.1585175> (1992).
- Kurokawa, H. *et al.* Target-induced conformational adaptation of calmodulin revealed by the crystal structure of a complex with nematode Ca(2+)/calmodulin-dependent kinase kinase peptide. *J. Mol. Biol.* **312**, 59–68. <https://doi.org/10.1006/jmbi.2001.4822> (2001).
- McCarthy, M. R. *et al.* Impact of methionine oxidation on calmodulin structural dynamics. *Biochem. Biophys. Res. Commun.* **456**, 567–572. <https://doi.org/10.1016/j.bbrc.2014.11.091> (2015).
- Rebbeck, R. T. *et al.* S100A1 Protein does not compete with calmodulin for ryanodine receptor binding but structurally alters the ryanodine receptor. Calmodulin complex. *J. Biol. Chem.* **291**, 15896–15907. <https://doi.org/10.1074/jbc.M115.713107> (2016).
- Yamniuk, A. P., Ishida, H., Lippert, D. & Vogel, H. J. Thermodynamic effects of noncoded and coded methionine substitutions in calmodulin. *Biophys. J.* **96**, 1495–1507. <https://doi.org/10.1016/j.bpj.2008.10.060> (2009).
- Strynadka, N. C. & James, M. N. Model for the interaction of amphiphilic helices with troponin C and calmodulin. *Proteins* **7**, 234–248. <https://doi.org/10.1002/prot.340070305> (1990).
- Ikura, M. Calcium binding and conformational response in EF-hand proteins. *Trends Biochem. Sci.* **21**, 14–17 (1996).
- Zhang, M. & Vogel, H. J. Two-dimensional NMR studies of selenomethionyl calmodulin. *J. Mol. Biol.* **239**, 545–554. <https://doi.org/10.1006/jmbi.1994.1393> (1994).
- Yuan, T., Weljie, A. M. & Vogel, H. J. Tryptophan fluorescence quenching by methionine and selenomethionine residues of calmodulin: orientation of peptide and protein binding. *Biochemistry* **37**, 3187–3195. <https://doi.org/10.1021/bi9716579> (1998).
- Weljie, A. M. & Vogel, H. J. Tryptophan fluorescence of calmodulin binding domain peptides interacting with calmodulin containing unnatural methionine analogues. *Protein Eng.* **13**, 59–66. <https://doi.org/10.1093/protein/13.1.59> (2000).
- Balog, E. M., Norton, L. E., Thomas, D. D. & Fruen, B. R. Role of calmodulin methionine residues in mediating productive association with cardiac ryanodine receptors. *Am. J. Physiol. Heart Circ. Physiol.* **290**, H794–799. <https://doi.org/10.1152/ajpheart.00706.2005> (2006).
- Zhang, M., Tanaka, T. & Ikura, M. Calcium-induced conformational transition revealed by the solution structure of apo calmodulin. *Nat. Struct. Biol.* **2**, 758–767. <https://doi.org/10.1038/nsb0995-758> (1995).
- Chattopadhyaya, R., Meador, W. E., Means, A. R. & Quijoch, F. A. Calmodulin structure refined at 1.7 Å resolution. *J. Mol. Biol.* **228**, 1177–1192. [https://doi.org/10.1016/0022-2836\(92\)90324-d](https://doi.org/10.1016/0022-2836(92)90324-d) (1992).
- Schrodinger, LLC. *The PyMOL Molecular Graphics System, Version 1.8* (2015).
- Boschek, C. B., Jones, T. E., Smallwood, H. S., Squier, T. C. & Bigelow, D. J. Loss of the calmodulin-dependent inhibition of the RyR1 calcium release channel upon oxidation of methionines in calmodulin. *Biochemistry* **47**, 131–142. <https://doi.org/10.1021/bi701352w> (2008).
- Sun, H. *et al.* Repair of oxidized calmodulin by methionine sulfoxide reductase restores ability to activate the plasma membrane Ca-ATPase. *Biochemistry* **38**, 105–112. <https://doi.org/10.1021/bi981295k> (1999).
- Squier, T. C. Oxidative stress and protein aggregation during biological aging. *Exp. Gerontol.* **36**, 1539–1550. [https://doi.org/10.1016/s0531-5565\(01\)00139-5](https://doi.org/10.1016/s0531-5565(01)00139-5) (2001).
- Stadtman, E. R., Moskovitz, J. & Levine, R. L. Oxidation of methionine residues of proteins: biological consequences. *Antioxid. Redox Signal.* **5**, 577–582. <https://doi.org/10.1089/152308603770310239> (2003).
- Gao, J. *et al.* Loss of conformational stability in calmodulin upon methionine oxidation. *Biophys. J.* **74**, 1115–1134. [https://doi.org/10.1016/S0006-3495\(98\)77830-0](https://doi.org/10.1016/S0006-3495(98)77830-0) (1998).
- Balog, E. M., Lockamy, E. L., Thomas, D. D. & Ferrington, D. A. Site-specific methionine oxidation initiates calmodulin degradation by the 20S proteasome. *Biochemistry* **48**, 3005–3016. <https://doi.org/10.1021/bi802117k> (2009).

39. Vogt, W. Oxidation of methionyl residues in proteins: tools, targets, and reversal. *Free Radic. Biol. Med.* **18**, 93–105. [https://doi.org/10.1016/0891-5849\(94\)00158-g](https://doi.org/10.1016/0891-5849(94)00158-g) (1995).
40. Stadtman, E. R., Van Remmen, H., Richardson, A., Wehr, N. B. & Levine, R. L. Methionine oxidation and aging. *Biochem. Biophys. Acta.* **1703**, 135–140. <https://doi.org/10.1016/j.bbapap.2004.08.010> (2005).
41. Ferrington, D. A. *et al.* Selective degradation of oxidized calmodulin by the 20 S proteasome. *J. Biol. Chem.* **276**, 937–943. <https://doi.org/10.1074/jbc.M005356200> (2001).
42. Anbanandam, A. *et al.* Mediating molecular recognition by methionine oxidation: conformational switching by oxidation of methionine in the carboxyl-terminal domain of calmodulin. *Biochemistry* **44**, 9486–9496. <https://doi.org/10.1021/bi0504963> (2005).
43. Balog, E. M. *et al.* Calmodulin oxidation and methionine to glutamine substitutions reveal methionine residues critical for functional interaction with ryanodine receptor-1. *J. Biol. Chem.* **278**, 15615–15621. <https://doi.org/10.1074/jbc.M209180200> (2003).
44. Lau, K., Chan, M. M. & Van Petegem, F. Lobe-specific calmodulin binding to different ryanodine receptor isoforms. *Biochemistry* **53**, 932–946. <https://doi.org/10.1021/bi401502x> (2014).
45. Maximciuc, A. A., Putkey, J. A., Shamoo, Y. & Mackenzie, K. R. Complex of calmodulin with a ryanodine receptor target reveals a novel, flexible binding mode. *Structure* **14**, 1547–1556. <https://doi.org/10.1016/j.str.2006.08.011> (2006).
46. Lee, W., Tonelli, M. & Markley, J. L. NMRFAM-SPARKY: enhanced software for biomolecular NMR spectroscopy. *Bioinformatics* **31**, 1325–1327. <https://doi.org/10.1093/bioinformatics/btu830> (2015).
47. Farrow, N. A. *et al.* Backbone dynamics of a free and phosphopeptide-complexed Src homology 2 domain studied by 15N NMR relaxation. *Biochemistry* **33**, 5984–6003. <https://doi.org/10.1021/bi00185a040> (1994).
48. Tikunova, S. B., Rall, J. A. & Davis, J. P. Effect of hydrophobic residue substitutions with glutamine on Ca(2+) binding and exchange with the N-domain of troponin C. *Biochemistry* **41**, 6697–6705. <https://doi.org/10.1021/bi011763h> (2002).
49. Jones, E. M., Squier, T. C. & Sacksteder, C. A. An altered mode of calcium coordination in methionine-oxidized calmodulin. *Biophys. J.* **95**, 5268–5280. <https://doi.org/10.1529/biophysj.108.139634> (2008).
50. Jas, G. S. & Kuczera, K. Free-energy simulations of the oxidation of c-terminal methionines in calmodulin. *Proteins* **48**, 257–268. <https://doi.org/10.1002/prot.10133> (2002).
51. Lim, J. C., Kim, G. & Levine, R. L. Stereospecific oxidation of calmodulin by methionine sulfoxide reductase A. *Free Radic. Biol. Med.* **61**, 257–264. <https://doi.org/10.1016/j.freeradbiomed.2013.04.004> (2013).
52. Chin, D. & Means, A. R. Methionine to glutamine substitutions in the C-terminal domain of calmodulin impair the activation of three protein kinases. *J. Biol. Chem.* **271**, 30465–30471. <https://doi.org/10.1074/jbc.271.48.30465> (1996).
53. Lewis, A. K. *et al.* Oxidation increases the strength of the methionine-aromatic interaction. *Nat. Chem. Biol.* **12**, 860–866. <https://doi.org/10.1038/nchembio.2159> (2016).
54. Lian, L. Y. & Middleton, D. A. Labelling approaches for protein structural studies by solution-state and solid-state NMR. *Prog. Nucl. Mag. Res. Sp.* **39**, 171–190. [https://doi.org/10.1016/S0079-6565\(01\)00034-6](https://doi.org/10.1016/S0079-6565(01)00034-6) (2001).
55. Foster, M. P., McElroy, C. A. & Amero, C. D. Solution NMR of large molecules and assemblies. *Biochemistry* **46**, 331–340. <https://doi.org/10.1021/bi0621314> (2007).
56. Verardi, R., Traaseth, N. J., Masterson, L. R., Vostrikov, V. V. & Veglia, G. Isotope labeling for solution and solid-state NMR spectroscopy of membrane proteins. *Adv. Exp. Med. Biol.* **992**, 35–62. https://doi.org/10.1007/978-94-007-4954-2_3 (2012).
57. Wittekind, M. & Mueller, L. Hncacb, a high-sensitivity 3d Nmr experiment to correlate amide-proton and nitrogen resonances with the alpha-carbon and beta-carbon resonances in proteins. *J. Magn. Reson. Ser. B* **101**, 201–205. <https://doi.org/10.1006/jmr.1993.1033> (1993).
58. Grzesiek, S. & Bax, A. Correlating backbone amide and side-chain resonances in larger proteins by multiple relayed triple resonance Nmr. *J. Am. Chem. Soc.* **114**, 6291–6293. <https://doi.org/10.1021/ja00042a003> (1992).
59. Urbauer, J. L., Short, J. H., Dow, L. K. & Wand, A. J. Structural analysis of a novel interaction by calmodulin—high-affinity binding of a peptide in the absence of calcium. *Biochemistry* **34**, 8099–8109. <https://doi.org/10.1021/bi00025a016> (1995).
60. Lee, W., Westler, W. M., Bahrami, A., Eghbalnia, H. R. & Markley, J. L. PINE-SPARKY: graphical interface for evaluating automated probabilistic peak assignments in protein NMR spectroscopy. *Bioinformatics* **25**, 2085–2087. <https://doi.org/10.1093/bioinformatics/btp345> (2009).
61. Wang, Y. & Jardetzky, O. Probability-based protein secondary structure identification using combined NMR chemical-shift data. *Protein Sci.* **11**, 852–861. <https://doi.org/10.1110/ps.3180102> (2002).
62. Braunschweiler, L. & Ernst, R. R. Coherence transfer by isotropic mixing—application to proton correlation spectroscopy. *J. Magn. Reson.* **53**, 521–528. [https://doi.org/10.1016/0022-2364\(83\)90226-3](https://doi.org/10.1016/0022-2364(83)90226-3) (1983).
63. Bax, A. & Davis, D. G. Mlev-17-based two-dimensional homonuclear magnetization transfer spectroscopy. *J. Magn. Reson.* **65**, 355–360. [https://doi.org/10.1016/0022-2364\(85\)90018-6](https://doi.org/10.1016/0022-2364(85)90018-6) (1985).
64. Delaglio, F. *et al.* Nmrpipe—a multidimensional spectral processing system based on unix pipes. *J. Biomol. NMR* **6**, 277–293. <https://doi.org/10.1007/Bf00197809> (1995).
65. Bers, D. M., Patton, C. W. & Nuccitelli, R. A practical guide to the preparation of Ca²⁺ buffers. *Method Cell Biol.* **99**, 1–26. [https://doi.org/10.1016/S0091-679x\(10\)99001-8](https://doi.org/10.1016/S0091-679x(10)99001-8) (2010).
66. Gustavsson, M. *et al.* Lipid-mediated folding/unfolding of Phospholamban as a regulatory mechanism for the sarcoplasmic reticulum Ca²⁺-ATPase. *J. Mol. Biol.* **408**, 755–765. <https://doi.org/10.1016/j.jmb.2011.03.015> (2011).
67. Greenfield, N. J. Using circular dichroism spectra to estimate protein secondary structure. *Nat. Protoc.* **1**, 2876–2890. <https://doi.org/10.1038/nprot.2006.202> (2006).
68. Compton, L. A. & Johnson, W. C. Analysis of protein circular-dichroism spectra for secondary structure using a simple matrix multiplication. *Anal. Biochem.* **155**, 155–167. [https://doi.org/10.1016/0003-2697\(86\)90241-1](https://doi.org/10.1016/0003-2697(86)90241-1) (1986).
69. Manavalan, P. & Johnson, W. C. Variable selection method improves the prediction of protein secondary structure from circular-dichroism spectra. *Anal. Biochem.* **167**, 76–85. [https://doi.org/10.1016/0003-2697\(87\)90135-7](https://doi.org/10.1016/0003-2697(87)90135-7) (1987).
70. Sreerama, N. & Woody, R. W. Estimation of protein secondary structure from circular dichroism spectra: comparison of CONTIN, SELCON, and CDSSTR methods with an expanded reference set. *Anal. Biochem.* **287**, 252–260. <https://doi.org/10.1006/abio.2000.4880> (2000).
71. Lees, J. G., Miles, A. J., Wien, F. & Wallace, B. A. A reference database for circular dichroism spectroscopy covering fold and secondary structure space. *Bioinformatics* **22**, 1955–1962. <https://doi.org/10.1093/bioinformatics/btl327> (2006).
72. Whitmore, L. & Wallace, B. A. DICHROWEB, an online server for protein secondary structure analyses from circular dichroism spectroscopic data. *Nucleic Acids Res.* **32**, W668–673. <https://doi.org/10.1093/nar/gkh371> (2004).
73. Whitmore, L. & Wallace, B. A. Protein secondary structure analyses from circular dichroism spectroscopy: methods and reference databases. *Biopolymers* **89**, 392–400. <https://doi.org/10.1002/bip.20853> (2008).

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Author contributions

S.E.D.N. purified labeled protein for NMR, acquired NMR and CD data. S.E.D.N. and G.V. analyzed NMR data. S.E.D.N. and D.K.W. analyzed CD data. S.E.D.N., D.K.W., R.T.R., R.L.C., G.V., and D.D.T. wrote the paper.

Competing interests

DDT and RLC hold equity in, and serve as an executive officers for Photonic Pharma LLC. These relationships have been reviewed and managed by the University of Minnesota. Photonic Pharma had no role in this study. None of the other authors (SEDN, DKW, RTR, GV) have any potential competing interests.

Additional information

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