Review

## Development of a Miniature Mass Spectrometer and an Automated Detector for Sampling Explosive Materials

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The development of a robust ionization source using the counter-flow APCI, miniature mass spectrometer, and an automated sampling system for detecting explosives are described. These development efforts using mass spectrometry were made in order to improve the efficiencies of on-site detection in areas such as security, environmental, and industrial applications. A development team, including the author, has struggled for nearly 20 years to enhance the robustness and reduce the size of mass spectrometers to meet the requirements needed for on-site applications. This article focuses on the recent results related to the detection of explosive materials where automated particle sampling using a cyclone concentrator permitted the inspection time to be successfully reduced to 3 s.



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# EXPLOSIVE DETECTION USING MASS SPECTROMETRY

The threats of terrorism and criminal bombing are becoming ever more serious problems for all countries. To find hidden explosives, various types of detectors for explosives have been developed.<sup>1-3)</sup> They are classified into two categories: bulk detection and trace detection. Bulk detection includes X-ray imaging, nuclear quadrupole resonance (NQR), and neutron techniques that are used to determine the shape and size of suspicious objects in luggage, letters, packages, etc. However, bulk detection is not capable of easily identifying what the suspicious objects are. Trace detection, on the other hand, including chemical-analysis methods such as chemical luminescence,4<sup>)</sup> ion mobility spectroscopy,<sup>5-9)</sup> and mass spectrometry,<sup>10-13)</sup> has been applied to the detection of trace contaminants that are present on a passenger's body, clothes, and luggage. Trace detection permits specific chemicals to be identified, and its selectivity is higher than that of bulk detection. However, it cannot determine the actual amount of chemicals in a suspicious item such as luggage. As mentioned above, the characteristics of bulk detection and trace detection are different. To improve

security at important facilities, therefore, the combined use of bulk detection and trace detection is recommended. At current security checkpoints at most airports, not all carryon luggage is tested by trace detector, although all luggage is examined by X-ray imaging. One of the reasons for this is that the use of a trace detector is not so popular, because its throughput is not sufficiently rapid to permit the method to be applied for all passengers and luggage. Typical test procedures for a trace detector are as follows: a security guard at a security checkpoint wipes the surface of the luggage with a wiping-sheet to collect any contaminants that are adhered to the luggage, inserts the wiping-sheet into the heating unit of the trace detector, and waits to see if the alarm of the trace detector is activated or not. The total analysis time from wiping to detection is typically 30s per test, when applied to human screening. To shorten this inspection time, we evaluated the use of mass spectrometry because its high separation and high sensitivity might possibility lead to much higher throughput. In the mass spectrometry studies, both vapor or particle detection was used for volatile explosives or less-volatile explosives, respectively. The use of a mass spectrometer for the detection of vapor derived from explosives using various ambient ionization sources such as atmospheric pressure chemical ionization (APCI)<sup>11,13)</sup> and

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Mass Spectrometry Society of Japan bestowed the MSSJ AWARD for Technical Development 2016 on the author. This is an invited review of the achievement. Abbreviations: APCI, atmospheric pressure chemical ionization; AREX, axially resonant excitation; CID, collision induced dissociation; ECD, electron capture dissociation; LP-DBDI, low-pressure dielectric barrier discharge ionization; NQR, nuclear quadrupole resonance; TATP, triacetone triperoxide; TNT, 2,4,6-trinitrotoluene; RDX, research department explosive, 1,3,5-trinitroperhydro-1,3,5-triazine; SESI, secondary electrospray ionization

secondary electrospray ionization (SESI)<sup>12)</sup> ionized the vapor molecules derived from an explosive has been reported in previous studies. Ewing *et al.*<sup>13)</sup> reported on the detection of very low concentrations of vapor (below 25 ppq) produced by an explosive vapor using a laboratory based triplequadrupole mass spectrometer, but a much higher sensitivity would be needed to achieve the real-time detection of less volatile explosives such as like RDX, and PETN. Many researchers who are involved in the detection of explosive are still seeking a method that is faster and more accurate in this field.

## **ROBUST IONIZATION SOURCE**

Before starting the development of systems for detecting explosives, we focussed on the development of a highly robust, miniaturized version of a mass spectrometer. A technology for on-site applications would only be acceptable if they met the user's requirements for usability, size, cost, and related issues.

Although it was not specifically designed for detecting explosives but was originally designed for an environmental applications, i.e., monitoring dioxin precursors, we developed the counter-flow introduction APCI14,15) depicted in Fig. 1. The direction of the high-speed air opposite the electric field can avoid overlapping the secondary ionization (ion-molecular reaction) area and the NO<sub>X</sub> abundant area. Because ions are first generated by Corona discharge they do not perturb the  $NO_x$  abundant area, ions that are generated first have more chances to react with target molecules like explosives. This less-overlapping configuration improved the ionization efficiency for negative ions by about 3 times. As another advantage of this configuration is that it extended the maintenance frequency to typically every 6 months; our conventional APCI ion source required irregular maintenance between a few hours and a week under this condition. We conclude that the high speed air flow prevents the deposition of dust around the tip of the needle electrode, which makes the Corona discharge much more stable. The robust ionization source has been a key technology and is deployed in our mass spectrometers for on-site applications.



Fig. 1. Schematic of counter flow introduction APCI. The direction of the air flow opposite to the electric field can avoid overlapping the ionization area and the  $NO_X$  flowing area.

## LINEAR ION TRAP DEVELOPMENT FOR MIN-IATURISED MASS SPECTROMETER

Miniaturization of mass spectrometer was made for onsite applications. We aimed to develop a miniature mass spectrometer that meets, not only size requirements but also has sufficient sensitivity and selectivity for detecting explosives for purposes of trace detection. Since an ion trap has intrinsic characteristics such as small dimension, and lessstrict vacuum conditions, we developed several types of linear ion trap mass spectrometers, such as an axially resonant excitation (AREX) linear ion trap mass spectrometer,<sup>16,17)</sup> and a wire type linear ion trap.<sup>18)</sup>

The AREX linear ion trap depicted in Fig. 2, inserted in a set of vane lenses between each quadrupole rod, produces ab electrostatic potential that is approximately harmonic along the central axis of the quadrupole field. After ions with a specific m/z are resonantly oscillated in the axial direction, they are mass selectively ejected in the axial direction. At a high scan rate of 11 Th/ms, the AREX achieved a high ejection efficiency of more than 60%, which is more than three times higher than a conventional mass selective axial ejection method from a linear trap using a fringing field. In contrast, CID excitation in a conventional linear ion trap is in a pseudo-harmonic radio frequency potential that destabilizes fragment ions whose m/z is less than 1/4 that of the precursor ions. ECD was achieved in an axial magnetic field of about 0.2 Tesla superimposed along the AREX linear ion trap mass spectrometer. We observed that sequential ECD/CID inside the trap improved the sequence coverage for a peptide.

In another linear ion trap, the wire type linear ion trap shown in Fig. 3, a set consisting of a trap wire lens and an extraction wire lens positioned orthogonally to each other



Fig. 2. Schematic of AREX LIT. A set of vane lenses are inserted between each quadrupole rod to produce a DC potential that is approximately harmonic along the central axis of the quadrupole field. Since ions with specific *m/z* can are resonantly oscillated in the axial harmonic potential, an MS/MS analysis, which consists of accumulation, isolation, collision induced dissociation (CID), electron capture dissociation (ECD), and mass selective ejection can be performed [Reproduced from ref. 17 with Copyright permission of MSSJ].

was placed between the quadrupole rods. The trap wire lens confines the ions inside the trap, and the extraction wire axially extracts ions from the trap. The ions are resonantly excited in the direction perpendicular to the trap wire lens by applying a supplemental alternating current to the excitation lenses. Excited ions with a large motion pass over the trap wire lens, while unexcited ions remain trapped inside. Ions that have passed over the trap wire lens are then extracted by the extraction wire lens. The characteristics of mass-selective ejection with a direct current extraction field were investigated by both simulations and experiments. A mass resolving power of  $m/\Delta m=1300$  was achieved at a scan rate of 500 Th/s. The dependence of ejection efficiency on trap wire lens bias was measured, and an ejection efficiency of 20% at a scan rate of 500 Th/s was achieved by optimizing the DC bias on the trap wire lens.

For producing portable mass spectrometers used in the illicit drug detection applications, we developed a mass spectrometer with discontinuous sample gas introduction.<sup>19,20)</sup> The study was intended to produce a mass spectrometer that was sufficiently compact that it would be portable and be used in an onsite analysis. The development of a small mass spectrometer required the use of a compact pump whose displacement was small, thus decreasing the sensitivity of that spectrometer. To obtain a high sensitivity with a small mass spectrometer, we integrated some novel techniques: a highly sensitive ionization source and the efficient extraction of sample vapor. The low-pressure dielectric barrier discharge ionization (LP-DBDI) source made it possible to increase the conductance between the source and the mass analyzer, compared with ambient ionization sources, thus enhancing the efficiency of ion transfer from the ionization source to the mass analyzer. We also developed a vacuumed headspace and probe heating methods for efficiently transporting the sample vapor to the ionization source. The sensitivity was also further enhanced by using a discontinuous sample gas introduction technique. A prototype portable mass spectrometer using those novel techniques was found to be sufficiently sensitive to permit the detection of 0.1 ppm methamphetamine, 1 ppm amphetamine, 1 ppm 3,4-methylenedioxymethamphetamine, and 10 ppm cocaine in a liquid. These size reduced mass spectrometers could make on-site mass spectrometry possible.



Fig. 3. Schematic of the wire type LIT. A set consisting of a trap wire lens and an extraction wire lens positioned orthogonally with respect to each other was placed between quadrupole rods. The trap wire lens confines the ions inside the trap, and the extraction wire axially extracts ions from the trap when excitation electrodes resonantly oscillate the ions inside.

## **EXPLOSIVE DETECTION SYSTEM**

#### Vapor explosive sampling portal

We started the development of a high-throughput explosive detection system using these component devices in 2007. We first developed a high-throughput portal system for detecting triacetone triperoxide (TATP) vapor emitted from passengers and luggage, as shown in Fig. 4.<sup>21,22)</sup> The portal system consists of a push-pull air sampler, a counter flow introduction APCI ion source, and a wire-type linear ion trap mass spectrometer. TATP signals were clearly obtained within 2s after the subject passed through the portal system. By optimizing the shape of the nozzle that controls the air flow of the vapor sampler, TATP vapor could be detected at a high throughput, i.e., 1200 persons/hour. The false-positive rate of the detection portal system for TATP was evaluated by a field test performed at a train station. A multimarker logic to determine was adopted, and no falsepositive alarms were obtained for over 3000 passengers during the field test. Moreover, acetone, which is an inflammable liquid, could be detected from the passengers during the field test. We therefore concluded from these results that this detection portal system would be useful for detecting dangerous chemicals that have high vapor pressure (such as TATP and inflammable liquids). The detection limits of the vapor portal were between the sub-ppb and ppb level, which is not sufficient to permit explosive vapors such as TNT, and RDX to be detected.

### Automated particle sampler

In 2010, we started another development of a highthroughput explosive detection system for detecting more comprehensive explosives including less volatile explosives such as TNT, and RDX. This detector consists of an automated particle sampler for collecting contaminants in explosives and an ion-trap mass spectrometer for analyzing them. The automated sampler, which uses a cyclone concentrator and mass spectrometer in this manner enables the rapid detection of explosives' particles that had adhered to the surface of a synthetic leather sheet within a few seconds.<sup>23)</sup> Figure 5 shows the experimental setup of the pro-



Fig. 4. Photograph of the vapor sampler. Three sampling ports are aligned vertically in the sampling-port unit. To avoid the adsorption of various chemicals onto the surface of the sampling ports, heating the sampling ports is advisable. Vapors sampled by the sampling ports are introduced into the ion source of the mass spectrometer *via* a metal tube. The temperature of the metal tube was maintained at 100°C by a heater. Flow rate of the sample gas introduced into the ion-source unit was set at 2L/min [Reproduced from ref. 22 with Copyright permission of IEEE].



Fig. 5. Schematic of the automated particle sampler combined with a mass spectrometer. Cyclone dimensions: diameter of body: 32 mm, diameter of outlet: 14.5 mm, height of the body: 88 mm, width of inlet: 9 mm, height of inlet: 11 mm [Reproduced from ref. 23 with Copyright permission of Wiley].

totype sampler, which consists of an air nozzle (inner diameter: 1.5 mm), evacuation port, cyclone particle concentrator, vaporizer, and a homemade mass spectrometer. A wire type linear ion trap mass spectrometer with counter-flow introduction atmospheric pressure chemical ionization (APCI) records the mass spectrum at a rate of about 4 spectra/s. Explosive particles are wiped off with a sampling probe, which is an aluminium rod with a diameter of 2mm. This wiping operation was made to cover only the tip of probe with particles from the sample reservoir. Explosive particles are deposited onto a synthetic leather sheet with the sample probe. First, the explosive particles on a sheet are detached by an air jet pulse with a duration of 100 ms. Particles detached by the air jet pulse are collected from the evacuation port and then introduced into the cyclone particle concentrator. The mechanism and structure of the concentrator has been described in the literature,<sup>24)</sup> and our dimensions are given in the caption for Fig. 5. The pumping speed from the evacuation port into the concentrator is about 80 L/min. Particles accumulate near the bottom of the concentrator as the result of centrifugal wind and gravity forces. The vaporizer, which is placed at the bottom of the concentrator, has a transfer line between the concentrator and the mass spectrometer. Because the vaporizer is constantly heated at about 200°C, the explosives particles that arrive here are converted into vapor. Explosive vapor is then introduced into the mass spectrometer at a flow rate of about 1 L/min. To prevent particles from reaching the spectrometer, a metal filter with a filtration accuracy of  $1 \,\mu m$  (Nippon Seisen Co., Ltd., Osaka, Japan) is set inside the vaporizer. Explosive particles wiped off with the sampling probe were deposited onto and directly injected into the vaporizer after removing the cyclone particle concentrator (number of trials, n=20). By comparing the calibration curve and the signal intensity at the direct injections, the amount of TNT contained in a single deposit of particles was estimated to be about 47 ng (C.V. of 24%). The weight of deposited silica (C18) particles with the sample probe was calculated to be about 47  $\mu$ g by calculating the amount of TNT of 47 ng and mass fraction of 0.1%. In the following, we calculate the amounts of explosive particles from the calculated deposit particle weight of 47  $\mu$ g

and each mass fraction. We investigated the dependencies versus several parameters such as the nozzle gas pressure, angles of gas injections onto the object, distance between the nozzle and objects, pumping speed into the cyclone using the experimental setup. An example for determining the optimum pressure inside the jet nozzle, and the pumping speed are explained in the following. Figures 6(A) and (B) show surface images on the synthetic leather sheet before and after an air jet pulse at a nozzle pressure of 0.05 MPa is introduced. These images were taken with a fluorescence microscope after silica (C18) particles were prepared by doping with the fluorescent dye fluorescein. From the images, we calculated a detachment efficiency of 86.4%. Figure 7 shows changes in detachment efficiency against nozzle pressure. The detachment efficiency increased with the pressure increment and was saturated at more than 80% over 0.05 MPa. Since a higher nozzle pressure might spread the detached particles with a strong air flow and degrade the subsequent collection efficiency into the evacuation port, a nozzle pressure of 0.05 MPa was found to be optimum for this configuration. The detection efficiency reached a maximum value of about 20% when optimal conditions were employed. The cyclone particle concentrator enriches the explosive particles by about 80 times compared to the absence of a concentration process. Figures 8(A) and (B) show mass spectra when explosive particles were directly deposited into the metal filter inside the vaporizer using the setup shown in Fig. 5. In the negative ion mode, strong peaks originating from the explosives were observed. Molecular ions of  $M^-$  with a peak of m/z 227 were dominantly observed in the TNT mass spectrum of Fig. 8(A). Unlike TNT, adduct ions of  $[M+NO_2]^-$  with a peak of m/z 268 were dominant in the RDX mass spectrum shown in Fig. 8(B). These spectra show that TNT and RDX particles can be vaporized with minimal fragmentation at a temperature of 200°C. We used the intensities of the dominant peaks at m/z 227 and m/z 268 for the TNT and RDX concentration index. Figures 9(A) and (B) show the signal time response after the explosive particles are exposed to an air jet pulse. An air jet pulse was injected at a timing of 1.0 s, depicted as arrows in the figures. The signal for TNT rose in about 1s and returned to the



Fig. 6. Typical example of surface images (A) before and (B) after a pulse gas injection. The duration of gas injection was 0.1 s. Nozzle pressure: 0.03 MPa [Reproduced from ref. 23 with Copyright permission of Wiley].



Fig. 7. Detachment efficiencies versus injected gas pressure (N=7 for each pressure). Detachment conditions such as distance, angle, and gas pressure were determined through these experiments [Reproduced from ref. 23 with Copyright permission of Wiley].





Fig. 8. Mass spectra of particles of explosive material obtained with (A) 0.1% mass fraction TNT (corresponding to about 47 ng of TNT) and (B) 0.4% mass fraction RDX (corresponding to about 188 ng of RDX) [Reproduced from ref. 23 with Copyright permission of Wiley].



Fig. 9. Mass chromatogram of explosive particles. (A) Signal intensity for m/z 227 obtained with a 0.1% mass fraction of TNT (corresponding to about 47 ng of TNT). (B) Signal intensity with m/z 268 obtained with 0.4% mass fraction RDX (corresponding to about 188 ng of RDX). Arrows around 1.0s indicates timings of air jet injection. Pumping speed through the cyclone particle concentrator was set at 80 L/min [Reproduced from ref. 23 with Copyright permission of Wiley].

former level in less than 10 s after the air jet pulse. The time necessary for the transportation of the particles from the sample sheet to the mesh, and the transportation time of the vapor from the mesh to the mass spectrometer are estimated to be about 50, and 100 ms, respectively. In comparison, the signal of RDX increased in 2 s and then returned to the former level in about 15 s. The time response for RDX was slower than that for TNT, which can be explained by the difference of their volatility. The vapor pressure of RDX is much lower than that of TNT,<sup>3)</sup> which means RDX is vaporized more slowly than TNT with heating. If we take 1 s of data after signal-rising as its explosive's concentration, detection is accomplished in 2 s for TNT and in 3 s for RDX after the air jet pulse.

# Installation of the automated particles sampler into a commercial boarding gate

As the next step towards practical security applications of the high-throughput explosive detection system, we installed the automated particle sampler in a commercial boarding gate and baggage screener to check all passengers and baggage boarding an aircraft for traces explosives.<sup>25)</sup> We designed these prototypes using several parameters such as the nozzle gas pressure, angles of gas injection onto the object, and distance between the nozzle and objects, which we obtained in the sample testing described above. Photographs of boarding gate type are shown in Figs. 10(A) and (B). The prototype is based on a commercial boarding gate used at airports (Nippon Signal, Co., Ltd., Japan). As shown in Fig.



Fig. 10. Photographs of a boarding gate fitted with the built-in explosives detector: (A) outlook of the boarding gate, and (B) enlarged area around the electronic ticket reader [Reproduced from ref. 25 with Copyright permission of IEEE].

10(A), the boarding gate consists of a main unit, a sub unit, and a footboard. To install the particle sampler and the mass spectrometer in the main unit, the width of the side cover of the main unit is 15 cm wider than that of the original cover. The controllers of the particle sampler and the mass spectrometer are installed in the sub unit. Cables connecting the main and sub units are covered by the footboard. An air compressor (not shown in the figure) is needed to supply compressed air to the boarding gate. However, the air compressor does not always need to be located nearby the boarding gate because compressed gas may be supplied via a long tube. An enlargement of the electronic-ticket (e-ticket) reader is shown in Fig. 10(B). When a passenger places his/her IC-card or e-ticket on the reader, an air jet is emitted from the nozzle. Particles that had adhered to the e-ticket and/or the passenger's hand are blown by the jet and collected by the inlet. The main challenge to installing the detectors into a gate system is the simultaneous detection of various explosives. We carried out the simultaneous detection of TNT and TATP as an example where the switching of the ionization polarity to detect nitro-compounds (which have strong electron affinity) and peroxides (which have a strong proton affinity) was confirmed. To change the polarity of the ion source and the mass spectrometer, the interval between the negative and positive mass analysis was set to 0.3 s. The simulants of TNT and TATP particles that had adhered to the surface of the plastic card, and the plastic card was tested by the boarding gate set up. The results for detecting TNT and TATP are shown in Fig. 11. The data acquisition frequency in the case of the polarity switching analysis was expanded from every 0.25s in the case of only the negative mode to every 0.9 s, but the signals for TNT and TATP were clearly obtained, as shown in Fig. 11. The findings reported herein







Fig. 12. Photograph of the baggage screener [Reproduced from ref. 25 with Copyright permission of IEEE].

show that the high-throughput detection for both positive and negative ionized explosives could be performed within 3 s after a passenger places his/her IC-card or e-ticket on the reader. For baggage screening, on the other hand, the traceexplosives detector was combined with a conventional X-ray baggage screener shown in Fig. 12. When a bag is put on the conveyor belt of the automated particle sampler, compressed air jets are emitted from nozzles. The trace analysis is finished, while an X-ray image of the bag is being obtained by the X-ray baggage screener. In both security equipment cases, the simulated TNT, RDX, and TATP particles that adhered to the detection targets were successfully detected. Accordingly, the developed security equipment will be useful tools for improving airport security in the near future.

### SUMMARY

This article summarizes the development of an MS-based system for detecting explosives. The instrument includes a counter-flow introduction APCI, a miniature mass spectrometer, and an automated sampling explosive detector system combined with these component devices. These developments have been ongoing for nearly 20 years in order to realize on-site mass spectrometry. Counter flow introduction APCI enhances robustness, ion trap development reduces the size of the mass spectrometry, permitting it to meet the requirements needed for on-site application. The detection of explosives at the level of the boarding gate and baggage screener were demonstrated, where the automated particle sampling using cyclone concentrator successfully reduced the inspection time and labour-intensive sampling process.

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## REFERENCES

- J. Yinon, S. Zitrin. Modern Methods and Applications in Analysis of Explosives, John Wiley and Sons Ltd., New York, 1993.
- 2) H. Schubert, A. Kuzentsov. Detection and Disposal of Improvised Explosives, Springer, Dordrecht, the Netherlands, 2006.
- M. Marshall, J. C. Oxley (Eds.). Aspects of Explosives Detection. Elsevier, Amsterdam, 2009.
- D. T. McQuade, A. E. Pullen, T. M. Swager. Conjugated polymerbased chemical sensors. *Chem. Rev.* 100: 2537–2574, 2000.
- G. A. Eiceman. Ion-mobility spectrometry as a fast monitor of chemical composition. *TrAC Trends Analyt. Chem.* 21: 259–275, 2002.
- 6) G. A. Eiceman, E. V. Krylov, N. S. Krylova, E. G. Nazarov, R. A. Miller. Separation of ions from explosives in differential mobility spectrometry by vapor-modified drift gas. *Anal. Chem.* 76: 4937–4944, 2004.
- L. Fricano, M. Goledzinowski, R. Jackson, F. Kuja, L. May, S. Nacson. An automatic trace detection system for the detection of explosives' vapors and particles in luggage. *Int. J. Ion Mobil. Spectrom.* 4: 22, 2001.
- C. D. Corrigan, L. V. Haley, D. P. Menagh. U.S.Patent 4,987,767, Explosive Detection Screening System. 1991.
- K. L. Linker, F. J. Conrad, C. A. Custer, C. L. Rhykerd. U.S.Patent 5,854,431, Particle Preconcentrator. 1998.
- J. Yinon. Field detection and monitoring of explosives. TrAC Trends Analyt. Chem. 21: 292–301, 2002.
- 11) Y. Takada, H. Nagano, M. Suga, Y. Hashimoto, M. Yamada, M. Sakairi, K. Kusumoto, T. Ota, J. Nakamura. Detection of military explosives by atmospheric pressure chemical ionization mass spectrometry with counter-flow introduction. *Propellants Explos.*

Pyrotech. 27: 224–228, 2002.

- 12) P. Martínez-Lozano, J. Rus, G. Fernández de la Mora, M. Hernández, J. Fernández de la Mora. Secondary electrospray ionization (SESI) of ambient vapors for explosive detection at concentrations below parts per trillion. J. Am. Soc. Mass Spectrom. 20: 287–294, 2009.
- R. G. Ewing, D. A. Atkinson, B. H. Clowers. Direct real-time detection of RDX vapors under ambient conditions. *Anal. Chem.* 85: 389–397, 2013.
- M. Sakairi, Y. Hashimoto, M. Yamada, M. Suga, K. Kojima. US Patent 6686592, Mass spectrometer, mass spectrometry, and monitoring system, 1999.
- 15) K. Kojima, M. Sakairi, Y. Takada, J. Nakamura. Vapor detection of TNT and RDX using atmospheric pressure chemical ionization mass spectrometry with counter-flow introduction (CFI). J. Mass Spectrom. Soc. Jpn. 48: 360–362, 2000.
- 16) Y. Hashimoto, H. Hasegawa, T. Baba, I. Waki. Mass selective ejection by axial resonant excitation from a linear ion trap. J. Am. Soc. Mass Spectrom. 17: 685–690, 2006.
- 17) Y. Hashimoto, H. Hasegawa, M. Sugiyama, H. Satake, T. Baba, I. Waki. Tandem mass spectrometry using an axially resonant excitation linear ion trap. *J. Mass Spectrom. Soc. Jpn.* 55: 339–342, 2007.
- M. Sugiyama, H. Hasegawa, Y. Hashimoto. Mass-selective axial ejection from a linear ion trap with a direct current extraction field. *Rapid Commun. Mass Spectrom.* 23: 2917–2922, 2009.
- 19) S. Kumano, M. Sugiyama, M. Yamada, K. Nishimura, H. Hasegawa, H. Morokuma, H. Inoue, Y. Hashimoto. Development of a portable mass spectrometer characterized by discontinuous sample gas introduction, a low-pressure dielectric barrier discharge ionization source, and a vacuumed headspace technique. *Anal. Chem.* 85: 5033–5039, 2013.
- 20) S. Kumano, M. Sugiyama, H. Hasegawa, H. Morokuma, H. Inoue, Y. Hashimoto. Probe heating method for solid sample analysis by portable mass spectrometer. *Mass Spectrom.* (Tokyo) 4: A0038, 2015.
- 21) Y. Takada, H. Nagano, Y. Suzuki, M. Sugiyama, E. Nakajima, Y. Hashimoto, M. Sakairi. High-throughput walkthrough detection portal for counter terrorism: Detection of triacetone triperoxide (TATP) vapor by atmospheric-pressure chemical ionization ion trap mass spectrometry. *Rapid Commun. Mass Spectrom.* 25: 2448–2452, 2011.
- 22) Y. Takada, Y. Suzuki, H. Nagano, M. Sugiyama, E. Nakajima, M. Sugaya, Y. Hashimoto, M. Sakairi. High-throughput walkthrough detection portal as a measure for counter terrorism: Design of a Vapor sampler for detecting triacetone triperoxide vapor by atmospheric-pressure chemical-ionization ion-trap mass spectrometry. *IEEE Sens. J.* 12: 1673–1680, 2012.
- 23) Y. Hashimoto, H. Nagano, Y. Takada, H. Kashima, M. Sugaya, K. Terada, M. Sakairi. Real-time explosives particle detection using a cyclone particle concentrator. *Rapid Commun. Mass Spectrom.* 28: 1376, 2014.
- M. Rhodes (Ed.). Introduction to Particle Technology, 2nd ed. John Wiley and Sons, Chichester, 2008.
- 25) Y. Takada, H. Nagano, Y. Kawaguchi, H. Kashima, M. Sugaya, K. Terada, Y. Hashimoto, M. Sakairi. Automated trace-explosives detection for passenger and baggage screening. *IEEE Sens. J.* 16: 1119–1129, 2016.