Supplemental Information

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Conformational Conversion during Amyloid Formation at Atomic Resolution

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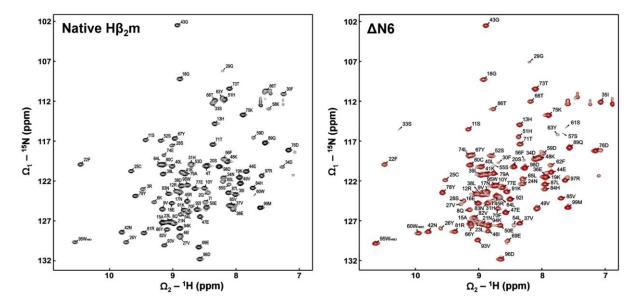


Figure S1, related to Figure 2. $^1H^{-15}N$ HSQC Spectra of Native H β_2m and $\Delta N6$ (each 80 μM) Obtained at pH 7.5, 25 $^{\circ}C$

Missing assignments for native H β_2 m (residues 1, 57, 60-62, 88) and for Δ N6 (residues 7, 53, 58, 60, 88) suggest these resonances are either weak in intensity due to line broadening or have chemical shifts that are degenerate with other resonances.

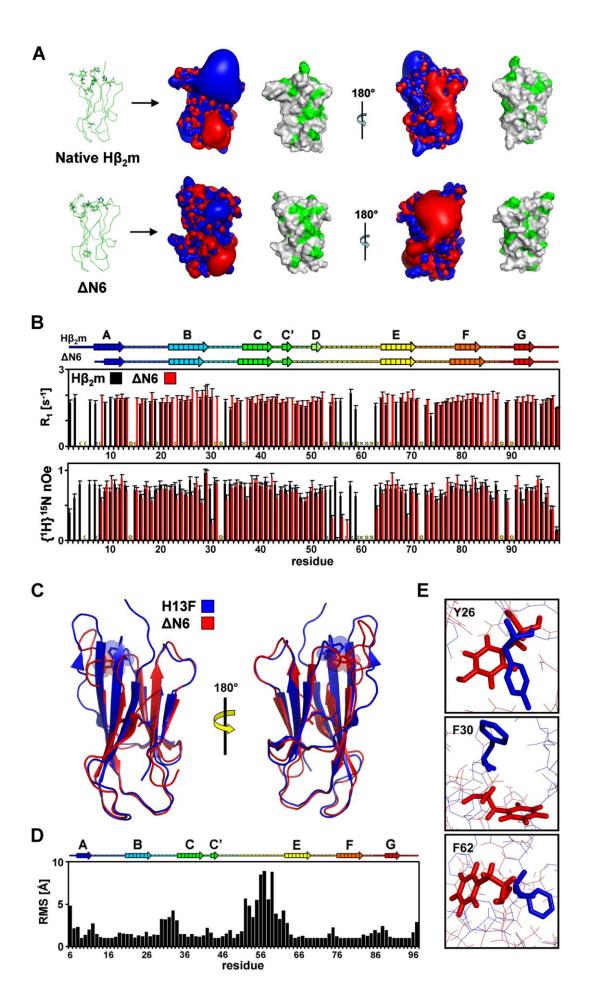


Figure S2, related to Figure 3. Comparison of the Structures and Dynamics of Native $H\beta_2 m$, $\Delta N6$, and H13F

- (A) Representation of the hydrophobic (green) and electrostatic (blue/red) surfaces of $H\beta_2m$ and $\Delta N6$ (lowest energy structures) using APBS (Baker et al., 2001) and Pymol (DeLano, 2002). Highlighted residues are Phe30, Pro32, Phe62, Leu64, Phe70 and His84 (sticks) that show some of the largest movements of sidechain orientation in the two structures (see also Figure 3C).
- (B) 15 N longitudinal relaxation (R₁=1/T₁) and { 1 H} 15 N nOe relaxation measurements of 500 μ M H β_2 m (black) and 500 μ M Δ N6 (red) at pH 7.5, 25 °C. Circles highlight residues for which no transverse relaxation rate could be determined due to resonance overlap, line broadening or the residue being a proline. Black crosses mark missing assignments. Rainbow coloured ribbons above indicate the secondary structure content of H β_2 m and Δ N6 deduced from the final set of 30 lowest energy structures using DSSPcont (Carter et al., 2003). The error was estimated using duplicates.
- (C) Cartoon overlay showing one monomer of the hexameric β_2 m taken from the crystal structure of H13F (3CIQ, in blue) (Calabrese et al., 2008) and the lowest energy structure of Δ N6 (red). Pro32 (blue and red sticks, spheres, respectively) and the disulfide bond (Cys25-Cys80, sticks) are highlighted.
- (D) RMS $C\alpha$ [Å] of the overlay of the structures shown in (C). The rainbow coloured ribbon indicates the secondary structure elements of $\Delta N6$ deduced from the final set of 30 lowest energy structures using DSSPcont (Carter et al., 2003).
- (E) Overlay of the structures of H13F and Δ N6 showing residues whose sidechains differ most significantly (> 2 Å) in orientation in the two structures shown in (C).

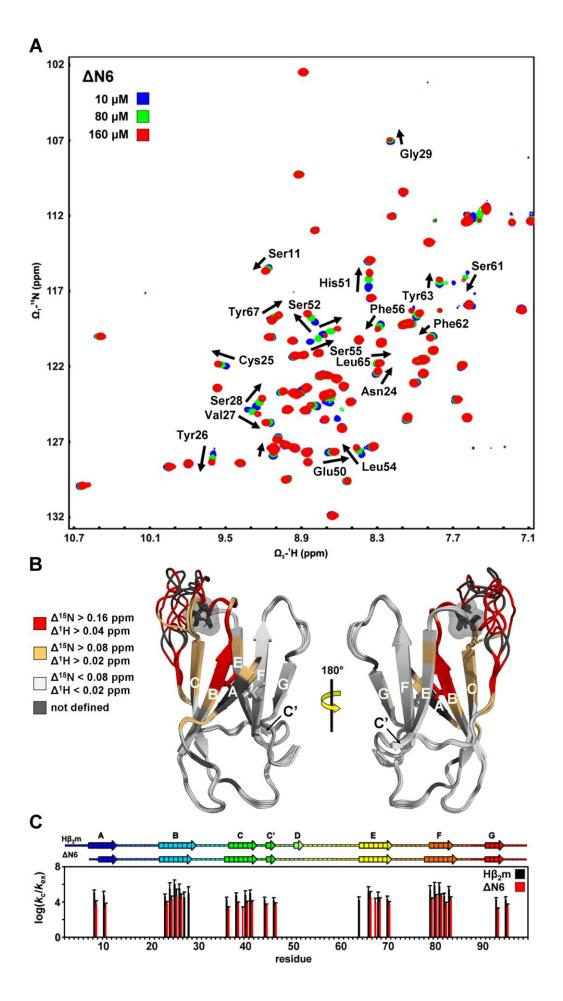
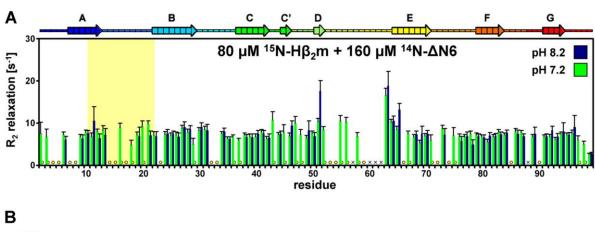


Figure S3, related to Figure 4. Hydrogen Exchange Kinetics of Native $H\beta_2 m$ and the Concentration Dependence of the 1H - ^{15}N Spectrum of $\Delta N6$

- (A) ^{1}H - ^{15}N HSQC spectra of 10 μM (blue), 80 μM (green) and 160 μM (red) $\Delta\text{N}6$ at pH 7.2, 25 $^{\circ}\text{C}$. Resonances that shift significantly with protein concentration are labelled and highlighted with an arrow (pH 8.2 \Rightarrow pH 6.2).
- (B) Cartoon representation of the five lowest energy structures of $\Delta N6$. Highlighted in red and light orange are the residues that correspond to resonances that shift significantly (^{1}H / $^{15}N > 0.04$ / 0.16 ppm or ^{1}H / $^{15}N > 0.02$ / 0.08 ppm, respectively) shown in (A). Residues coloured in black are not assigned or are proline residues or, for residues in the A-strand, resonate in a crowded region in the spectrum and hence their chemical shift *versus* concentration could not be determined. Residues coloured in light grey do not show significant chemical shift (^{1}H / ^{15}N < 0.02 / 0.08 ppm) under the conditions employed.
- (C) Hydrogen exchange kinetics of 80 μ M H β_2 m or Δ N6 measured at pH 7.2, 37 °C or pH 7.2, 25 °C, respectively. The intrinsic rates of HX (k_c) were calculated as described (Bai et al., 1993). Measured rates are denoted k_{ex} . The β -strands in native H β_2 m and Δ N6 are shown above in rainbow colour. The rate of HX of Δ N6 is too fast to measure by NMR at 37 °C. The error was estimated from the noise level of the experiment.



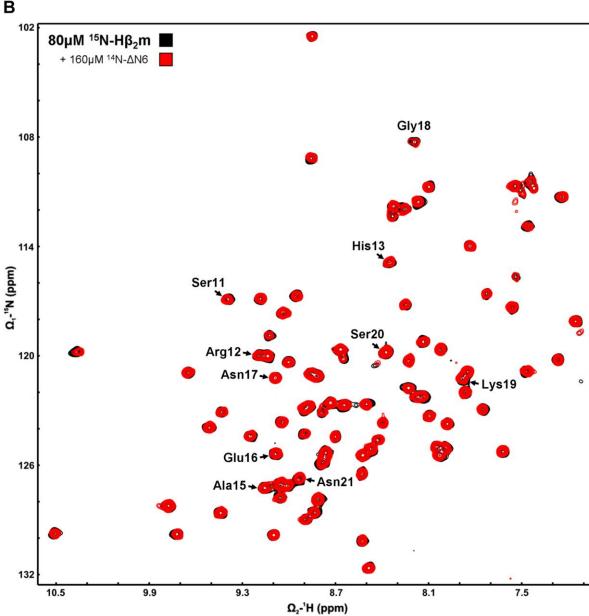


Figure S4, related to Figure 6. Interactions between Native $H\beta_2 m$ and $\Delta N6$ Are pH Dependent

- (A) 15 N transverse relaxation measurements (R_2 =1/ T_2) of 80 μ M 15 N-H β_2 m in the presence of 160 μ M 14 N- Δ N6 at pH 8.2 or pH 7.2 (blue and green, respectively), 37 $^{\circ}$ C. Circles highlight residues for which no transverse relaxation rate could be determined due to resonance overlap, line broadening or the corresponding residue being a proline. Black crosses mark missing assignments. The rainbow coloured ribbon above indicates the secondary structure content of native H β_2 m deduced from the final set of 30 lowest energy structures using DSSPcont (Carter et al., 2003). The error was estimated using duplicates.
- (B) ^{1}H - ^{15}N HSQC spectra of 80 μM ^{15}N -H $\beta_{2}\text{m}$ alone (black) or presence of 160 μM ^{14}N - $\Delta\text{N}6$ at pH 6.2 (red), 37 °C. Residues of H $\beta_{2}\text{m}$ within the AB-loop (residues 11-21) that show enhanced transverse relaxation rates under those conditions (see Figure 6A) are highlighted.

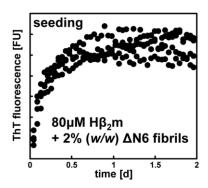


Figure S5, related to Figure 5. Cross-Seeding of Native H β_2 m Using Δ N6 Fibrillar Seeds ThT fluorescence of 80 μ M H β_2 m in 25 mM sodium phosphate buffer at pH 7.2, 37 °C seeded quiescently by using 2 % (w/w) Δ N6 fibrillar seeds grown *de novo* in 25 mM sodium phosphate buffer pH 7.2, 37 °C. Note: no fibrils form when 80 μ M H β_2 m is incubated without seed (Figure 5A).

Table S1, related to Figure 2. Assignment of cis- and trans-X-proline residues in native $H\beta_2 m$ and $\Delta N6$

	Native Hβ₂m			ΔN6		
	¹³ Cβ [ppm]	¹³ Cγ [ppm]	$\Delta^{13}C\beta^{13}C\gamma$ [ppm]	¹³ Cβ [ppm]	¹³ Cγ [ppm]	Δ ¹³ C β ¹³ C γ [ppm]
Pro5	32.14 ± 0.14	26.99 ± 0.05	5.15 ± 0.19	-	=	-
Pro14	31.48 ± 0.08	28.21 ± 0.14	3.27 ± 0.22	31.44 ± 0.09	28.20 ± 0.09	3.24 ± 0.18
Pro32	35.08 ± 0.11	25.10 ± 0.13	9.98 ± 0.24	31.51 ± 0.12	27.17 ± 0.13	4.34 ± 0.25
Pro72	32.15 ± 0.07	26.82 ± 0.16	5.33 ± 0.23	32.15 ± 0.09	26.86 ± 0.08	5.29 ± 0.17
Pro90	31.97 ± 0.16	27.71 ± 0.14	4.26 ± 0.30	32.41 ± 0.09	27.17 ± 0.13	5.24 ± 0.22

¹³Cβ and ¹³Cγ chemical shifts of proline residues and their difference Δ^{13} Cβ¹³Cγ in native Hβ₂m and Δ N6 are shown. Values of Δ^{13} Cβ¹³Cγ < 4.80 ppm correspond to a 100 % probability that the proline is *trans*. Values > 9.15 ppm correspond to a 100 % probability that the proline is *cis*. Values between > 4.80 ppm and < 9.15 ppm indicate mixed probabilities (Schubert et al., 2002).

Phe30 Native Hβ₂m 27 lle35 HG → Phe30 HZ Phe30 HZ Phe30 HB1 Phe30 HA Arg3 HB1 3.2 ± 0.5 Leu64 HB1 → Phe30 HZ 3.5 ± 0.5 Phe30 HB1 = lle35 HD 3.2 ± 0.6 Phe30 HB1 = lle35 HD 3.2 ± 0.5 Leu64 HD 2.5 ± 0.5 Leu64 HD 2 Pro32 HB2 → Leu64 HD 2 Pro32 HB2 → Leu64 HD 2 Pro32 HA 2.8 ± 0.4 Leu64 HD 2 Pro32 HB 2.8 ± 0.4 Leu64 HD 2 Pro32 HB 2.8 ± 0.4 leu64 HD 2 Pro32 HB 2.8 ± 0.6 Phe30 HB2 + lle35 HD 3.2 ± 0.7 lle35 HD 3 Phe30 HB2 HB 2.4 ± 0.5 lle35 HD 3 Val82 HG 2.4 ± 0.5 lle35 HD 3 Val82 HB 2.4 ± 0.5 lle35 HD 3 Leu64 HD 3.3 ± 0.6 lle35 HD 3 Leu64 HD 3 S.5 ± 0.6 lle35 HD 3 Leu64 HD 3 S.5 ± 0.6 lle35 HD 3 Leu64 HD 3 S.5 ± 0.6 lle35 HD 3 Leu64 H	residue	construct	Total $(i \pm > 4)$ nOes	nOe	distance [Å]
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$ \frac{ \text{Ile35 HD} \rightarrow \text{Val82 HB} 2.4 \pm 0.5 }{ \text{AN6} } = 40 $					
$ \frac{\Delta N6}{1825 \text{ HD}} \rightarrow \text{Val27 HG} \\ 1825 \text{ HD}} \rightarrow \text{Val82 HG} \\ 1825 \text{ HD}} \rightarrow \text{Val82 HG} \\ 2.0 \pm 0.4 \\ 1825 \text{ HD}} \rightarrow \text{Val82 HG} \\ 2.0 \pm 0.5 \\ 1825 \text{ HD}} \rightarrow \text{Val82 HG} \\ 2.8 \pm 0.5 \\ 1825 \text{ HG}} \rightarrow \text{Val82 HG} \\ 2.8 \pm 0.5 \\ 1825 \text{ HG}} \rightarrow \text{Val82 HG} \\ 2.8 \pm 0.5 \\ 1825 \text{ HD}} \rightarrow \text{Ser33 HB2}} \rightarrow \text{Phe62 HD} \\ 2.8 \pm 0.5 \\ 2.8 \pm 0.6 \\ 2.8 \pm 0.7 $					
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		4316	40		
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		Δ N6	40		
Phe62 Native Hβ₂m S Phe62 HD Ph					
Phe62 Native Hβ ₂ m 6 Ser33 HB2 → Phe62 HD 4.4 ± 0.6 Phe62 HD → Leu54 HD 5.3 ± 0.8 Phe62 HD → Leu54 HD 5.8 ± 0.6 Phe56 HB2 → Phe62 HD 5.8 ± 0.7 Phe56 HB1 → Pro32 HG2 4.9 ± 0.7 Phe62 HZ → Pro32 HG2 5.1 ± 0.7 Phe62 HZ → Pro32 HG3 5.6 ± 0.9 Phe62 HZ → Pro32 HG1 6.8 ± 0.8 Pro32 HA → Phe62 HE 7.4 ± 1.0 Phe62 HZ → Pro32 HG1 6.8 ± 0.8 Pro32 HA → Phe62 HE 7.4 ± 1.0 Phe62 HZ → Pro32 HG1 1.7 ± 0.3 Leu64 HD → Asp34 HA 3.1 ± 0.4 Leu64 HD → Phe30 HZ 3.5 ± 0.5 Ile35 HN → Leu64 HD 3.5 ± 0.5 Leu64 HD → Phe30 HZ 3.5 ± 0.5 Pro32 HB2 → Leu64 HD 2.1 ± 0.5 Pro32 HB2 → Leu64 HD 2.5 ± 0.5 Leu64 HD → Pro32 HA 2.8 ± 0.4 Ile35 HD → Leu64 HD 3.2 ± 0.6 Val27 HB → Leu64 HD 3.2 ± 0.6 Val27 HB → Leu64 HD 3.3 ± 0.5 Phe70 HN → Phe70 HN 3.0 ± 0.4 Asp21 HB2 → Phe70 HD 3.5 ± 0.4 Asp21 HB2 → Phe70 HD 3.6 ± 0.6 Phe70 HN → Phe22 HA 3.6 ± 0.5 Asp21 HB2 → Phe70 HZ 3.7 ± 0.5 Asp21 HB1 → Phe70 HZ 4.1 ± 0.8					
Phe62 Native Hβ₂m 6 Ser33 HB2 → Phe62 HD \rightarrow Leu54 HD \rightarrow Ser3 8.8 ± 0.6 Phe62 HD → Ser33 HB1 4.4 ± 0.6 Phe62 HD → Leu54 HD \rightarrow Ser33 HB1 5.8 ± 0.6 Phe62 HD → Ser33 HB1 5.8 ± 0.7 Ser30 HB1 5.8 ± 0.7 Phe62 HD → Ser33 HB1 5.8 ± 0.7 Phe56 HB2 → Phe62 HD \rightarrow Ser30 HB1 5.8 ± 0.7 Phe56 HB1 → Phe62 HD \rightarrow Ser30 HG2 4.9 ± 0.7 Phe62 HE \rightarrow Pro32 HG2 4.9 ± 0.7 Phe62 HE \rightarrow Pro32 HG2 5.1 ± 0.7 Phe62 HE \rightarrow Pro32 HG1 5.6 ± 0.9 Phe62 HZ \rightarrow Pro32 HB1 6.8 ± 0.8 Pro32 HB1 6.8 ± 0.8 Pro32 HB1 7.4 ± 1.0 Leu64 Native Hβ₂m 18 Ile35 HG \rightarrow Leu64 HD 1.7 ± 0.3 Algorithm 1.7 ± 0.3 Algorithm 2.3 ± 0.5 Algorithm 2.2 ± 0.5					
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Phe62	Native $H\beta_2 m$	6	Ser33 HB2 \rightarrow Phe62 HD	
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$				Phe62 HD → Leu54 HD	
				Phe62 HD → Ser33 HB1	5.8 ± 0.6
$ \frac{\Delta N6}{Phe62} = \frac{8}{Phe62} = \frac{Pro32}{Pro32} = \frac{4.9 \pm 0.7}{S.1 \pm 0.7} $ $ \frac{Phe62}{Phe62} = \frac{Pro32}{Pro32} = \frac{1}{S.1 \pm 0.7} $ $ \frac{Phe62}{Phe62} = \frac{Pro32}{Pro32} = \frac{1}{S.1 \pm 0.7} $ $ \frac{Phe62}{Phe62} = \frac{Pro32}{Pro32} = \frac{1}{S.1 \pm 0.7} $ $ \frac{Phe62}{Phe62} = \frac{Pro32}{Pro32} = \frac{1}{S.1 \pm 0.7} $ $ \frac{Phe62}{Pro32} = \frac{1}{A} \Rightarrow \frac{1}{Phe62} = \frac{1}{A} =$				Phe56 HB2 → Phe62 HD	5.8 ± 0.7
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				Phe56 HB1 → Phe62 HD	5.8 ± 0.7
Phe62 HE → Pro32 HG1		ΔΝ6	8	Phe62 HE → Pro32 HG2	4.9 ± 0.7
Phe62 HZ → Pro32 HB1				Phe62 HZ → Pro32 HG2	5.1 ± 0.7
Leu64 Native Hβ₂m 18 Ile35 HG → Leu64 HD Leu64 HD Asp34 HA Asn21 HB2 → Phe30 HZ 3.3 ± 0.5 Ile35 HN → Leu64 HD Asp34 HA Leu64 HB1 → Phe30 HZ 3.5 ± 0.6 Leu64 HD → Phe30 HZ 3.5 ± 0.5 ΔN6 46 Ile35 HD → Leu64 HD Leu64 HD Leu64 HD Asp32 HA Leu64 HD Pro32 HB2 → Leu64 HD Asp32 HA Leu64 HD Asp32 HA Leu64 HD Pro32 HA Leu64 HD Pro32 HA Leu64 HD Pro32 HA Leu64 HD Asp32 HA Leu64 HD Asp32 HA Leu64 HD Asp32 HA Leu64 HD Asp32 HA Leu64 HD Leu64 HD Asp32 HA Asp32 HB2 → Phe70 HD Asp32 HA Leu64 HD Asp32 HA Asp32 HB2 → Phe70 HZ Asp32 HA Asp32				Phe62 HE → Pro32 HG1	5.6 ± 0.9
Leu64 Native Hβ₂m 18 Ile35 HG → Leu64 HD 1.7 ± 0.3 1.2 0.4 2.3 ± 0.4 2.3 ± 0.5 1.2 0.4 2.3 ± 0.5 1.2 0.4 2.3 ± 0.5 1.2				Phe62 HZ → Pro32 HB1	6.8 ± 0.8
Leu64 HD → Asp34 HA Leu64 HB1 → Phe30 HZ 3.3 ± 0.5 Ile35 HN → Leu64 HD 3.5 ± 0.6 Leu64 HD → Phe30 HZ 3.5 ± 0.5 ΔN6 46 Ile35 HD → Leu64 HD 2.1 ± 0.5 Pro32 HB2 → Leu64 HD 2.5 ± 0.5 Leu64 HD → Pro32 HA Ile35 HD → Leu64 HD 3.2 ± 0.6 Val27 HB → Leu64 HD 3.3 ± 0.5 Phe70 Native Hβ ₂ m 27 Asn21 HN → Phe70 HN 3.0 ± 0.4 Asn21 HB2 → Phe70 HD 3.5 ± 0.4 Ser11 HB2 → Phe70 HZ 3.6 ± 0.6 Leu23 HD → Phe70 HZ 3.7 ± 0.5 ΔN6 36 Phe70 HN → Phe22 HA 3.6 ± 0.6 Ser11 HB1 → Phe70 HZ 3.6 ± 0.5 Ser11 HB1 → Phe70 HZ 4.1 ± 0.8				Pro32 HA → Phe62 HE	7.4 ± 1.0
Leu64 HD → Asp34 HA Leu64 HB1 → Phe30 HZ 3.3 ± 0.5 Ile35 HN → Leu64 HD 3.5 ± 0.6 Leu64 HD → Phe30 HZ 3.5 ± 0.5 ΔN6 46 Ile35 HD → Leu64 HD 2.1 ± 0.5 Pro32 HB2 → Leu64 HD 2.5 ± 0.5 Leu64 HD → Pro32 HA Ile35 HD → Leu64 HD 3.2 ± 0.6 Val27 HB → Leu64 HD 3.3 ± 0.5 Phe70 Native Hβ ₂ m 27 Asn21 HN → Phe70 HN 3.0 ± 0.4 Asn21 HB2 → Phe70 HD 3.5 ± 0.4 Ser11 HB2 → Phe70 HZ 3.6 ± 0.6 Leu23 HD → Phe70 HZ 3.7 ± 0.5 ΔN6 36 Phe70 HN → Phe22 HA 3.6 ± 0.6 Ser11 HB1 → Phe70 HZ 3.6 ± 0.5 Ser11 HB1 → Phe70 HZ 4.1 ± 0.8	Leu64	Native Hß, m	18		
Leu64 HB1 → Phe30 HZ Ile35 HN → Leu64 HD 3.5 ± 0.6 Leu64 HD → Phe30 HZ 3.5 ± 0.5 ΔN6 46 Ile35 HD → Leu64 HD 2.1 ± 0.5 Pro32 HB2 → Leu64 HD 2.5 ± 0.5 Leu64 HD → Pro32 HA Ile35 HD → Leu64 HD 3.2 ± 0.6 Val27 HB → Leu64 HD 3.3 ± 0.5 Phe70 Native Hβ ₂ m 27 Asn21 HN → Phe70 HN 3.0 ± 0.4 Phe70 HN → Phe22 HA Asn21 HB2 → Phe70 HD 3.5 ± 0.4 Ser11 HB2 → Phe70 HZ 3.6 ± 0.6 Leu23 HD → Phe70 HZ 3.7 ± 0.5 ΔN6 36 Phe70 HN → Phe22 HA 3.6 ± 0.6 Ser11 HB1 → Phe70 HZ 3.6 ± 0.5 Ser11 HB1 → Phe70 HZ					
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Leu64 HD → Pro32 HA 2.8 ± 0.4 Ile35 HD → Leu64 HD 3.2 ± 0.6 Val27 HB → Leu64 HD 3.3 ± 0.5 Phe70 Native Hβ₂m 27 Asn21 HN → Phe70 HN 3.0 ± 0.4 Phe70 HN → Phe22 HA 3.4 ± 0.4 Asn21 HB2 → Phe70 HD 3.5 ± 0.4 Ser11 HB2 → Phe70 HZ 3.6 ± 0.6 Leu23 HD → Phe70 HZ 3.7 ± 0.5 ΔN6 36 Phe70 HN → Phe22 HA 3.6 ± 0.5 Ser11 HB1 → Phe70 HZ 4.1 ± 0.8		<u> </u>	10		
Phe70 Native Hβ ₂ m 27 Asn21 HN → Phe70 HN 3.2 ± 0.6					
Phe70 Native Hβ ₂ m 27 Asn21 HN → Phe70 HN 3.0 ± 0.4 Phe70 HN → Phe70 HN 3.0 ± 0.4 Phe70 HN → Phe22 HA 3.4 ± 0.4 Asn21 HB2 → Phe70 HD 3.5 ± 0.4 Ser11 HB2 → Phe70 HZ 3.6 ± 0.6 Leu23 HD → Phe70 HZ 3.7 ± 0.5 ΔN6 36 Phe70 HN → Phe22 HA 3.6 ± 0.5 Ser11 HB1 → Phe70 HZ 4.1 ± 0.8					
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	Phe/0	rvauve $\pi p_2 m$	<i>∠1</i>		
Ser11 HB2 → Phe70 HZ 3.6 ± 0.6 Leu23 HD → Phe70 HZ 3.7 ± 0.5 ΔN6 36 Phe70 HN → Phe22 HA 3.6 ± 0.5 Ser11 HB1 → Phe70 HZ 4.1 ± 0.8					
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ΔN6 Phe70 HN → Phe22 HA Ser11 HB1 → Phe70 HZ 3.6 ± 0.5 4.1 ± 0.8					
Ser11 HB1 \rightarrow Phe70 HZ 4.1 \pm 0.8		1276			
		Δ N6	36		
1 01 TD 1 TO TTE 10 0 T					
				Asn21 HN \rightarrow Phe70 HE	4.3 ± 0.5
Leu23 HN \rightarrow Phe70 HE 4.2 \pm 0.5				Leu23 HN → Phe70 HE	4.2 ± 0.5

			Phe78 HD → Phe70 HB1	4.5 ± 0.5
His84	Native Hβ ₂ m	21	His84 HD → Ile35 HG	2.4 ± 0.4
			Ile35 HA → His84 HD	3.0 ± 0.4
			Ile35 HB → His84 HD	3.2 ± 0.4
			His84 HN → Pro90 HA	3.3 ± 0.3
			Ile35 HG → His84 HD	3.3 ± 0.4
	ΔΝ6	18	Phe30 HB1 → His84 HD	3.2 ± 0.6
			Ile35 HA → His84 HA	3.3 ± 0.5
			His84 HD → Ile35 HB	3.7 ± 0.7
			Ile35 HG → His84 HB2	4.0 ± 0.5
			Ile35 HD → His84 HD	4.0 ± 0.6
Trp95	Native Hβ ₂ m	37	Ala15 HB2 → Trp95 HD	2.8 ± 0.4
			Val9 HG → Trp95 HB1	2.5 ± 0.4
			Tyr10 HA → Trp95 HD	2.8 ± 0.4
			Val9 HG → Trp95 HA	2.9 ± 0.3
			Leu23 HD → Trp95 HD	2.9 ± 0.5
	ΔΝ6	68	Val9 HG → Trp95 HA	2.8 ± 0.4
			Val9 HG → Trp95 HB1	2.8 ± 0.4
			Trp95 HN → Val9 HG	2.9 ± 0.3
			Leu23 HD → Trp95 HB1	3.2 ± 0.5
			Trp95 HE → Tyr10 HA	3.2 ± 0.4

Table S2, related to Figure 3. Five long-range (i \pm > 4) nOes of native H β_2 m and Δ N6 ranked for closest distance (pH 7.5, 25 °C) in the ensembles of structures calculated The residues listed show the most intense long-range nOes for both constructs that define the arrangement of residues highlighted in Figure 3C. Note that a full annotation of chemical shifts and peak lists is deposited in BMRB with accession numbers 17165 and 17166 for H β_2 m and Δ N6, respectively.

Supplemental Experimental Procedures

Assembly of amyloid fibrils. For all experiments a protein aliquot stored in 10 mM sodium phosphate buffer, pH 7.2 at -80 °C was thawed on ice and centrifuged (10 min, 10,000 g, 10 °C). For fibril growth assays samples (100 µl) were prepared in double sealed (UC-500 sealing film, Axygen) 96-well plates (Corning Incorporated, Costar) and contained 0.8-500 μM protein, 81-89.5 mM NaCl, 10 μM ThT and 0.02 % (w/v) sodium azide in 10 mM sodium phosphate buffer, pH 6.2-8.2. All reaction mixtures had an ionic strength of 100 mM and were sterile filtered (0.2 µm, Sarstedt). For seeded reactions, each sample contained additionally 10 % (w/w) Δ N6 seeds formed at pH 7.2, which were collected by centrifugation (5,000 g, 10 min) and resuspended in H₂O (3 times) prior to use. De novo fibril growth was carried out by incubating the plate at 37 °C with agitation (200 rpm) for several weeks. Approximately every 48 hours fibril growth was assessed by measuring the fluorescence of ThT (excitation 440 nm, emission 480 nm) using a Fluorostar Optima, BMG Labtech plate reader at 37 °C (40 readings per well, gain 1200). Seeded fibril growth was carried out by incubating the plate at 37 °C quiescently for a week. Every hour fibril growth was assessed by measuring the fluorescence of ThT (excitation 440 nm, emission 480 nm) using a Fluorostar Optima, BMG Labtech plate reader at 37 °C (40 readings per well, gain 1200). The soluble fraction obtained after centrifugation (14,000 g, 10 min) was analysed by SDS-PAGE.

Negative-stain EM. Carbon coated copper grids were prepared by the application of a thin layer of formvar with an overlay of thin carbon. Samples were centrifuged (14,000 g, 10 min) and the pellets resuspended in deionised water and then applied to the grid in a drop-wise fashion. The grid was then carefully dried with filter paper before it was negatively stained

by the addition of 18 μ l of 2 % (w/v) uranyl acetate. Micrographs were recorded on a Philips CM10 electron microscope at moderate dose ($\sim 100 \text{ e Å}^{-1}$).

NMR spectroscopy and structure determination. Sequential assignments were obtained from analysis of HNCA, HNCO, HN(CO)CA, CBCA(CO)NH, HNHA and 1H-15N NOESY-HSQC (Vuister and Bax, 1993; Kay et al., 1994; Muhandiram and Kay, 1994; Zhang et al., 1994; Zhang et al., 1997). Spectra were processed using NMRPipe (Delaglio et al., 1995) and analysed in CCPN analysis (Vranken et al., 2005). Aliphatic sidechain resonances were assigned on the basis of H(C)CH-TOCSY, (H)CCH-TOCSY and (H)CCH-COSY (Bottomley et al., 1999). Aromatic specific sidechain assignments were made using ¹H-¹³C CT HSQC, HB(CBCGCD)HD and HB(CBCGCDCE)HE spectra (Yamazaki et al., 1993) and short Hβ-Hδ nOes. For the measurement of residual dipolar couplings ¹H-¹⁵N J-modulated HSQC spectra (Tjandra et al., 1996) were acquired in the presence and absence of 7 mg ml⁻¹ or 15 mg ml⁻¹ pf1 bacteriophage for Hβ₂m and ΔN6, respectively. NOe distance restraints were derived from 120 ms three-dimensional ¹H-¹⁵N NOESYHSQC, three-dimensional ¹H-¹³C NOESY-HSQC (Muhandiram et al., 1993; Smallcombe et al., 1995) and threedimensional aromatic ¹³C filtered NOESY spectra. Torsion angles phi and psi were predicted from ¹Hα, ¹³Cα, ¹³Cβ, C' and backbone ¹⁵N chemical shifts using TALOS (Cornillescu et al., 1999). NOes were assigned and structures calculated in a two stage process. In the initial stage structures were calculated using the Marvin/PASD simulated annealing protocol (Kuszewski et al., 2004) from X-PLOR-NIH v2.17 (Schwieters et al., 2006) with all measured nOe peaks and TALOS restraints. In the second stage the 50 structures with the lowest energy were then transferred into the first round of an ARIA 2.1 calculation (Nilges et al., 1997) along with high probability assignments. The RDC alignment magnitude and rhombicity were calculated as an average from the initial 50 structures from the first stage

with the DC utility from NMRPipe (Delaglio et al., 1995) and RDCs were used as variable angle restraints (VEAN) and SANI restraints (Clore et al., 1998) with a force constant of 0.1. The final structures were refined in a water box using standard ARIA parameters, the length of the two slow cooling stages in the ARIA protocol were extended by a factor of four as described (Fossi et al., 2005). Network anchoring was switched on throughout the calculation (Linge et al., 2004). During refinement the nOe distance restraint network was corrected for spin diffusion. The adjustment is based on the calculation of a theoretical intensity matrix from the set of structures produced each iteration. The theoretical intensity values were then used to calibrate the experimental volumes, and to correct the distance target. The calibrated volumes could then be used to estimate the error. All ARIA calculations were carried out using CNS 1.1 (Brunger et al., 1998). The final structure ensemble of native H β_2 m and Δ N6 was based on a total of 2065 or 2565 experimental nOe restraints, 128 or 118 dihedral angle restraints and 75 or 76 ¹H-¹⁵N residual coupling restraints, respectively (Table 1, Table S2 and Figure 3). Structures were validated with WHAT-CHECK (Hooft et al., 1996) and PROCHECKNMR (Laskowski et al., 1996). The molecular structure figures were generated using MolMol (Koradi et al., 1996) and Pymol (DeLano, 2002).

SOFAST-HMQC NMR experiments were carried out as described (Schanda and Brutscher, 2005). For refolding experiments: H β_2 m was denaturated in 8 M urea at 37 °C prior to refolding *via* 10-fold dilution with 25 mM sodium phosphate buffer pH 7.5 at 25 °C and the sample was then placed immediately into the NMR spectrometer. The first spectrum was recorded approximately 2 min after refolding was initiated. The length of each experiment was between 30 s and 5 min, d1 delays were between 0.500 and 0.550 seconds and the number of scans collected was between 2 and 8. Chemical shift referencing of all samples

was carried out using DSS as an internal standard for ¹H. ¹⁵N and ¹³C were indirectly referenced.

Secondary structure assignment: Using the continuous secondary structure assignment protocol DSSPcont (Carter et al., 2003), the secondary structure of native H β_2 m and Δ N6 was calculated based on 30 lowest energy solution structures. The output suggests a likelihood for each residue to be in one of eight secondary structure states: 3₁₀-helix, pi-helix, helix-turn, extended beta sheet, beta bridge, bend, and other/loop. Each residue in native $H\beta_2$ m and $\Delta N6$ was individually assigned to the structure with the highest likelihood (usually > 50%) over all 30 members of the ensemble calculated. Accordingly, Hβ₂m contains eight β-strands (residues 6-11 (A), 21-28 (B), 36-41 (C), 44-45 (C'), 50-51 (D), 64-70 (E), 79-83 (F), 91-94 (G)). ΔN6 contains seven β-strands (residues 8-11 (A), 21-27 (B), 35-41 (C), 44-45 (C'), 64-70 (E), 78-84 (F), 91-94 (G)), while residues 50 and 51 that form β-strand D in native $H\beta_2$ m adopted variable conformations in the 30 lowest energy structures calculated for ΔN6 with only 32 % and 0 % of molecules adopting β-strand conformation for these residues, respectively. By contrast in native Hβ₂m residues 50 and 51 establish both 53 % βstrand structure. β-strand C' in native Hβ₂m has a likelihood of 100 % whereas the likelihood in Δ N6 is reduced to 74 %. The 3₁₀-helix for residues 32-34 probably stabilises the *trans*isomer of X-Pro32 and has a likelihood of 100 % in Δ N6 whereas the likelihood in native $H\beta_2 m$ is 0 %.

¹⁵N NMR relaxation experiments. Backbone ¹⁵N transverse relaxation (R_2 =1/ T_2), ¹⁵N longitudinal relaxation (R_1 =1/ T_1), { 1 H} 15 N nOe relaxation measurements were carried out as described (Farrow et al., 1994). Duplicate measurements and spectral noise levels were used to obtain an estimate of the error. The R_2 relaxation measurements of all constructs were

performed at 500 MHz using a series of 10 experiments with relaxation delays ranging from 16.512 ms to 165.12 ms. The R_1 relaxation measurements of all constructs were performed at 500 MHz using a series of 11 experiments with mixing times ranging from 0 s to 1.28 s. For ${}^{1}H{}^{15}N$ nOe relaxation time experiments amide protons were pre-saturated with 120 ° pulses for 3.5 s prior to the experiment. All relaxation measurements were performed using 80-500 μ M protein in 81-89.5 mM NaCl (giving a total ionic strength of 100 mM), 10 % (ν/ν) D₂O and 0.02 % (ν/ν) sodium azide in 10 mM sodium phosphate buffer, pH 6.2-8.2.

Hydrogen Exchange NMR. All samples were adjusted to pH 7.2 or 6.2 in 10 mM sodium phosphate buffer and freeze-dried prior to dissolving them in 85 or 89.5 mM NaCl (to a total ionic strength of 100 mM), 100 % (v/v) D₂O and 0.02 % (w/v) sodium azide. The dead-time of the experiment was approximately 5-10 min and data acquisition (5-15 min) was carried out using SOFAST-HMQC NMR methods (Schanda et al., 2005). The data obtained were fitted (Origin, Originlab[©]) to a single exponential and the error was estimated from the noise level of the experiment.

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