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OPEN 2-benzothiazoleacetonitrile based two-photon fluorescent probe for hydrazine and its bio-imaging and environmental applications

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A novel turn-on two-photon fluorescent probe NS-N₂H₄ was developed with the 2-benzothiazoleacetonitrile as a new recognition site for the detection of hydrazine (N_2H_4). The twophoton probe exhibited favorable properties including high selectivity, low cytotoxicity and almost 16-fold fluorescence enhancement in the presence of N_2H_4 in solution. The probe could be used to image hydrazine in the living cells. Notably, we also used the two-photon fluorescent probe to image hydrazine in the tissue imaging for the first time. Furthermore, by the way of probe-loaded TLC plate, we further monitored vapor of hydrazine. Therefore, the novel two-photon probe is expected to be employed to detect N₂H₄ in biosamples and environmental pollution and the new recognition site will be widely applied to construct fluorescent probes for the detection of N_2H_4 .

Hydrazine (N_2H_4) has been widely employed in space system as rocket propellant due to its special chemical properties including flammability and explosion¹. According to its basic and reductive properties, hydrazine has been used as catalyst, corrosion inhibitor, and reducing agent in pharmaceutical, agricultural, and applied chemical industries²⁻⁴. However, it is also regarded as an important industrial pollutant to humans and animals with high toxicity, which could cause the lungs, livers, and kidneys cancerous⁵. Thus, the concentration of N_2H_4 must be controlled as low as 10 ppb⁶. Therefore, it is highly significant to develop powerful means for the tracking and detection of N₂H₄ in living systems with high sensitivity and good selectivity.

There are some analytical methods for the detection of N_2H_4 , which were exploited in the previous work, such as including chromatography-mass spectrometry, titrimetry and electro-chemical methods^{7,8}. However, sophisticated instrumentation and highly personal operating techniques must be needed in these processes, which are complex and time-consuming. In the past few decades, organic fluorescent probes, which were regarded as the most powerful monitoring tools, have become an important tool used in biological studies with excellent merits including high sensitivity, good selectivity and real-time detection9-12

Very recently, a number of fluorescent probes for monitoring N_2H_4 in living biosystem have been reported¹³⁻³⁴, most of which were reported by deprotection of the leaving group for the detection of hydrazine¹³⁻²⁷. Also, only few examples were developed by the cleavage of carbon-carbon double bond²⁸⁻³⁰. Besides, some fluorescent probes were used for the detection of N₂H₄ by the way of open ring, closed ring and effect of ESIPT after reacting with $N_2H_4^{31-34}$. Hence, it is very crucial to develop a new recognition site for the detection of N_2H_4 . Furthermore, all the previous probes were excited by one-photon wavelengths leading to photobleaching of fluorescent dyes and damage to living cells and tissues. Although the two-photon confocal microscopes are relatively not common, there are significant merits of two-photon microscopy (TPM) with long excitation wavelengths such as three-dimensional detection of living tissues, depressed the photodamage to biological samples, increased penetration ability of tissue and reduced fluorescent interference of background. Therefore, it is very important and necessary to construct two-photon fluorescent probe, which could be suitable for imaging N2H4 specifically in living cells and tissues.

In this report, we have constructed a novel two-photon fluorescent probe $NS-N_2H_4$ for the detection of N_2H_4 with 2-benzothiazoleacetonitrile as a new recognition site. (Fig. 1) The novel turn-on fluorescent probe $NS-N_2H_4$ was designed for the recognition of N₂H₄ with good selectivity over other analytes. Besides, the cell imaging and

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Figure 1. The structure of $NS-N_2H_4$ and the proposed sensoring mechanism for N_2H_4 .



Figure 2. Synthesis of the two-photon fluorescent probe NS-N₂H₄.



Figure 3. Fluorescence spectra of NS-N₂H₄ (10 μ M) in pH 7.4 PBS/DMSO (v/v=2/1) in the absence or presence of N₂H₄.

the first tissue imaging confirmed that the probe $NS-N_2H_4$ can be used to monitor the level of N_2H_4 in living biosystem. Furthermore, the probe $NS-N_2H_4$ could monitor vapor of hydrazine by the way of probe-loaded TLC plate. Therefore, the two-photon probe is expected to be employed to detect N_2H_4 in biosamples and environmental pollution.

Results and Discussion

Design and synthesis. The platform of 6-hydroxy-2-naphthaldehyde was chosen as the fluorescence reporting group due to two-photon properties and easy modification. By engineering a new recognition moiety of 2-benzothiazoleacetonitrile to the fluorescent platform, we designed and synthesized the two-photon probe NS-N₂H₄, which was outlined in Fig. 2 by condensation reaction in one step easily with good yield. The structure of target compound NS-N₂H₄ was fully characterized by ¹H NMR, ¹³C NMR and HRMS.

Fluorescent properties of NS-N₂H₄. With the two-photon probe NS-N₂H₄ in hand, its optical properties were measured in the absence or presence of N₂H₄ including absorption (Fig. S1) and fluorescence spectroscopy. The probe NS-N₂H₄ showed almost no fluorescence with excitation at 360 nm (Fig. 3). In presence of N₂H₄, the probe NS-N₂H₄ exhibited strong emission at 448 nm in PBS-DMSO (v/v = 2/1, pH = 7.4) at ambient temperature. That is to say, PBS-DMSO (v/v = 2/1, pH = 7.4) was regarded as the suitable solvent for the fluorescence experiment. With the time extended, the fluorescence intensity was increased gradually (Fig. 3). Notably, a large fluorescence enhancement (up to 16-fold) was shown. The two-photon probe showed relatively high sensitivity in presence of N₂H₄. In addition, the probe NS-N₂H₄ is also stable under irradiation depicted in the Fig. S2.

Mechanism. To get insight into the proposed sensing process, we studied the reaction of $NS-N_2H_4$ with N_2H_4 by mass spectrometry. When the probe $NS-N_2H_4$ (20 μ M) was treated with N_2H_4 (400 μ M) in pH 7.4 PBS/DMSO (v/v = 2/1) at room temperature, a significant peak at m/z 187.0875 corresponding to the product $NS-N_2H_4$ -adduct appears in the ESI-MS spectrum (Fig. S4), in good agreement with the proposed sensing mechanism (Fig. 1).

Effect of pH. We then decided to examine the effect of pH on the fluorescence response of $NS-N_2H_4$ to N_2H_4 . As shown in Fig. 4, the emissions intensities of $NS-N_2H_4$ are quite low and do not change significantly over wide



Figure 4. The emission intensity changes of $NS-N_2H4$ (10 μ M) upon addition of $N2H_4$ (20 equiv) at different pH PBS buffer, containing 33.3% DMSO as a cosolvent.



Figure 5. Fluorescence spectra of NS-N2H4 (10 μ M) in pH 7.4 PBS/DMSO (v/v=2/1) in the absence or presence of N₂H₄ (20 equiv).

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ranges of pH from 2.0–9.5, indicating that the free probe was stable in the wide pH range. Upon treated **NS**-N₂H₄ with N₂H₄, we found that the pH value of solution has a great influence on the probe **NS**-N₂H₄ response to N₂H₄. With the increase of pH from 7.0 to 9.5, an enhancement trend is observed in **NS**-N₂H₄ fluorescence intensity of response to N₂H₄, which covers well the physiological pH range of mitochondria (about pH 7.99), indicating that the free probe is suitable for detecting N₂H₄ in living cells and tissues.

Response rate and selectivity. The time courses of the fluorescence intensities of $NS-N_2H_4$ (10 μ M) in the presence of N_2H_4 (20 equiv) in pH 7.4 PBS/DMSO (v/v = 2/1) was measured in Fig. 5. Notably, a gradual increase in fluorescence intensity was observed in the presence of N_2H_4 in 240 min at ambient temperature. However, the fluorescence intensity increased rapidly at 37 °C (Fig. S3). The fact is that the probe $NS-N_2H_4$ could be fit for the detection of N_2H_4 in real time. The high selectivity to the target molecule over other potentially competing molecules is another important property for a bioimaging probe with potential application in the biosystem. Therefore, we had performed some research on the selectivity of the free probe $NS-N_2H_4$ to investigate the selectivity. As shown in Fig. 6, When other analytes such as Al^{3+} , Ca^{2+} , Co^{2+} , Cu^+ , Cu^{2+} , Mg^{2+} , Zn^{2+} , SO_3^{2-} , Cys, Cl^- were treated with $NS-N_2H_4$, the fluorescence intensity was almost unchanged compared with a strong fluorescent response when treated with N_2H_4 . These results suggest that the probe $NS-N_2H_4$ is highly selective for N_2H_4 over other tested species.

Application in vapor gas detection. Encouraged by the above excellent properties of the probe $NS-N_2H_4$, we evaluated its potential utility for the detection of hydrazine in real samples. At the beginning, TLC plates were soaked in the solution of $NS-N_2H_4$ (0.1 mM, in DMSO). After dried, the $NS-N_2H_4$ probe-loaded TLC plates were used to detect gaseous hydrazine, which can further discriminate hydrazine aqueous solutions with different concentrations (Fig. 7). When exposed to the vapor of hydrazine for 10 min, distinctive fluorescence color changes of $NS-N_2H_4$ -loaded TLC plates were observed (Fig. 7b-f), which were highly dependent on the concentration of hydrazine in aqueous solution and easy to be distinguished by the naked eyes. However, no visible change was



Figure 6. The fluorescence intensity of probe NS-N₂H₄ (10 μ M) in the presence of various analytes (10 equiv) in PBS buffer (pH 7.4 PBS/DMSO (v/v=2/1)). 1: none; 2: Al³⁺; 3: Ca²⁺; 4: Co²⁺; 5: Cu⁺; 6: Cu²⁺; 7: Mg²⁺; 8: Zn²⁺; 9: S²⁻; 10: SO₃²⁻; 11: Cys; 12: Cl⁻; 13: N₂H₄.



Figure 7. Photographs of TLC plates, soaked in the solution of NS-N₂H₄ followed by addition of different amounts of hydrazine. (a): water; (b): 10% N₂H₄; (c) 30% N₂H₄; (d) 40% N₂H₄; (e) 50% N₂H₄; (f) 80% N₂H₄.

observed by applying blank solvent (distilled H_2O , Fig. 7a). Therefore, these results demonstrate that the probe **NS-N₂H₄** is suitable for the instant visualization of trace amounts of hydrazine in environmental samples

Bioimaging in living cells. The above measurements indicate that the two-photon fluorescent probe has good properties including sensing appropriately at physiological pH, a very large turn-on signal, in particular a new recognition site, good selectivity. Thus, the probe $NS-N_2H_4$ seems to be fit for the detection of N_2H_4 in real biosamples. We evaluated $NS-N_2H_4$ imaging assays in live cells, and fluorescence imaging experiments were carried out in living cells (HeLa cells) on confocal laser scanning microscopy.

The cytotoxicity of **NS-N**₂**H**₄ was examined toward Hela cells by a MTT assay (see Supplementary Fig. S5). The results have proved to be that the probe **NS-N**₂**H**₄ at the low concentrations has no marked cytotoxicity to the cells after a long period (>90% HeLa cells survived after 24 h with **NS-N**₂**H**₄ (30.0 μ M) incubation). Therefore, the probe **NS-N**₂**H**₄ is suitable for imaging N₂H₄ in living cells due to the low cytotoxicity.

After established the excellent sensing performance and the low cytotoxicity of the probe $NS-N_2H_4$, we examine whether the probe could be functional in living cells. The utility of $NS-N_2H_4$ for fluorescence imaging of N_2H_4 in living cells was investigated (Fig. 8). When HeLa cells were incubated with $NS-N_2H_4$ for 30 min, no detectable fluorescence was observed. However, when the cells were pre-treated with $NS-N_2H_4$ for 30 min and then incubated with N_2H_4 (10 equiv) solution for another 30 min, the strong fluorescence was shown in the blue channel (Fig. 8e) at the same test conditions, confirming that the probe possess good membrane permeability and could image N_2H_4 in cellular environment.

Bioimaging in living tissues. Encouraged by the above ideal results of the probe in the blue channel for monitoring N_2H_4 and the advantages of two-photon fluorescence microscopy, we decided to further operate experiment for the detection of N_2H_4 in living tissues by two-photon fluorescence microscopy. The living tissues slices of the fresh rat liver were prepared with thickness at 400 µm, which were measured by two-photon fluorescence microscopy. At the beginning, tissue slices incubated with only the probe **NS-N_2H_4** (20.0 µM) for 30 min at 37 °C in PBS exhibit no fluorescence at the emission window of 0–75 nm (Fig. S6). When tissue slices were incubated with **NS-N_2H_4** (20.0 µM) for 30 min, and then treated with N_2H_4 (20 equiv) for another 30 min, significant fluorescence emerged up to 75 µm depth of living tissues by the way of two-photon fluorescence microscopy, which has exhibited its two-photon fluorescence properties (Fig. 9).

Conclusions

In conclusion, we have developed a turn-on two-photon fluorescent probe with the 2-benzothiazoleacetonitrile as a new recognition site for the detection of hydrazine N_2H_4 . Desirable properties including good selectivity and low cytotoxicity are emerged. The probe **NS-N₂H₄** could be used to image hydrazine in living cells as well



Figure 8. Brightfield and fluorescence images of HeLa cells stained with the probe $NS-N_2H_4$. (a) Brightfield image of HeLa cells costained only with $NS-N_2H_4$; (b) Fluorescence images of (a) from blue channel; (c) overlay of (a and b); (d) Brightfield image of HeLa cells costained with $NS-N_2H_4$ and treated with N_2H_4 ; (e) Fluorescence images of (d) from blue channel; (f) overlay of the brightfield image (d) and blue channels (e).

as in living tissues for the first time. At last, the novel probe was applied to monitor vapor of hydrazine by the way of probe-loaded TLC plate. We expect that the novel probe $NS-N_2H_4$ could be helpful for investigation and detection of N_2H_4 in living organisms and environmental pollution and many other fluorescent probes would be developed with this new recognition site in the future.

Methods

Fluorometric analysis. Without other noted, all the tests were operated according to the following procedure. A stock solution (1.0 mM) of $NS-N_2H_4$ was prepared in DMSO. In a 10 mL tube the test solution of compounds $NS-N_2H_4$ was prepared by placing 0.09 mL of stock solution, 3 mL of DMSO, 6 mL of 0.1 M PBS buffer and an appropriate volume of N_2H_4 sample solution. After adjusting the final volume to 10 mL with 0.1 M PBS buffer, standing at room temperature 3 min, 3 mL portion of it was transferred to a 1 cm quartz cell to measure absorbance or fluorescence. All fluorescence measurements were conducted at room temperature on a Hitachi F4600 Fluorescence Spectrophotometer. The slight pH variations of the solutions were achieved by adding the minimum volumes of NaOH (0.1 M) or HCl (0.2 M).

Vapor gas detection. TLC plates were soaked in the solution of $NS-N_2H_4$ (0.1 mM, in DMSO). After dried, the $NS-N_2H_4$ probe-loaded TLC plates were placed to cover a flask containing different concentration of N_2H_4 for 10 min at room temperature before observation.

Cytotoxicity assay. The living cells line were treated in DMEM (Dulbecco's Modified Eagle Medium) supplied with fetal bovine serum (10%, FBS), penicillin (100 U/mL) and streptomycin (100 μ g/mL) under the atmosphere of CO₂ (5%) and air (95%) at 37 °C. The HeLa cells were then seeded into 96-well plates, and 0, 1, 5, 10, 20, 30 μ M (final concentration) of the probe **NS-N₂H₄** (99.9% DMEM and 0.1% DMSO) were added respectively. Subsequently, the cells were cultured at 37 °C in an atmosphere of CO₂ (5%) and air (95%) for 24 hours. Then the HeLa cells were washed with PBS buffer, and DMEM medium (100 μ L) was added. Next, MTT (10 μ L, 5 mg/mL) was injected to every well and incubated for 4 h. Violet formazan was treated with sodium dodecyl sulfate solution (100 μ L) in the H₂O-DMF mixture. Absorbance of the solution was measured at 570 nm by the way of a microplate reader. The cell viability was determined by assuming 100% cell viability for cells without **NS-N₂H₄**.

HeLa Cells culture. HeLa cells were grown in modified Eagle's medium (MEM) replenished with 10% FBS with the atmosphere of 5% CO₂ and 95% air at 37 °C for 24 h. The HeLa cells were washed with PBS when used. HeLa cells treated with NS-N₂H₄ (20.0 μ M) for 30 min, then with N₂H₄ (200.0 μ M) for 30 min at 37 °C. The ideal fluorescence images were acquired with a Nikon A1MP confocal microscopy with the equipment of a CCD camera.



Figure 9. Two-photon fluorescence images of a fresh mouse liver slice pretreated with NS-N₂H₄ (20μ M) and then with N₂H₄ (20 equiv) in PBS buffer at the depths of approximately 0–75 μ m. Excitation at 800 nm with fs pulse.

Tissue imaging. The Kunning mice were purchased from Shandong University Laboratory Animal Center (Jinan, China). All procedures for this study were approved by the Animal Ethical Experimentation Committee of Shandong University according to the requirements of the National Act on the use of experimental animals (China). The fresh mouse liver slices were obtained from the liver of 14-day-old mouse. The living liver slices were gained with 400 micron thickness using a vibrating-blade microtome in 25 mM PBS (pH 7.4). The living liver slices were pre-treated with NS-N₂H₄ (20 μ M) for 30 min. The slices were washed three times by PBS buffer and imaged. To obtain the two-photon fluorescence images of the tissues incubated with both the probe and anlysis sample (N₂H₄), the slices were pre-treated with NS-N₂H₄ (20 μ M) for 30 min at 37 °C, the slices were washed three times by PBS buffer and imaged. The two-photon fluorescence emission was collected at between 420 and 495 nm upon excitation at 800 nm with a femtosecond laser.

Synthesis of the probe NS-N₂H₄. A mixture of 6-hydroxy-2-naphthaldehyde (0.5 mmol, 100.0 mg, 1.0 equiv) and benzothiazole-2-acetonitrile (0.55 mmol, 58.9 mg, 1.1 equiv) were dissolved in EtOH (5.0 mL). The piperidine (0.55 mmol, 46.8 mg, 1.1 equiv) was added under N₂. After stirred at room temperature for 8 h, the reaction mixture was adjusted to distilled water (2.0 mL), and then extracted with ethyl acetate. The organic layer was washed with saturated sodium chloride, dried over Na₂SO₄, filtered, and concentrated under vacuum, and the product was obtained by silica column chromatography to give the probe NS-N₂H₄ in the yield of 83%. ¹H NMR (400 MHz, DMSO-*d*₆) δ 10.4 (s, 1H), 8.49 (d, *J* = 12.4 Hz, 2H), 8.22–8.19 (m, 2H), 8.09 (d, *J* = 8.0 Hz, 1H), 7.90

(dd, J = 14.4, 7.6 Hz, 2H), 7.60 (t, J = 8.0 Hz, 1H), 7.52 (t, J = 7.2 Hz, 1H), 7.23–7.16 (m, 2H). ¹³C NMR (101 MHz, DMSO- d_6) δ 164.6, 159.3, 153.9, 149.3, 137.6, 135.2, 134.7, 132.2, 128.1, 128.0, 127.9, 127.7, 125.8, 123.9, 123.4, 120.9, 117.6, 110.2, 104.0; HRMS (ESI) m/z calcd for C₂₀H₁₃ON₂ S⁺ (M + H)⁺: 329.0743; found 329.0744.

References

- Serov, A. & Kwak, C. Direct hydrazine fuel cells: A review. *Appl. Catal. B: Environ.* 98, 1–9, doi:10.1016/j.apcatb.2010.05.005 (2010).
 Rosca, V. & Koper, M. T. M. Electrocatalytic oxidation of hydrazine on platinum electrodes in alkaline solutions. *Electrochim. Acta.*
- 53, 5199–5205, doi:10.1016/j.electacta.2008.02.054 (2008).
 3. Kean, T., Miller, J. H., Skellern, G. G. & Snodin, D. Acceptance criteria for levels of hydrazine in substances for pharmaceutical use and analytical methods for its determination. *Pharmeur. Sci. Notes.* 2, 23–33 (2006).
- Khaled, K. F. Experimental and theoretical study for corrosion inhibition of mild steel in hydrochloric acid solution by some new hydrazine carbodithioic acid derivatives. *Appl. Surf. Sci.* 252, 4120–4128, doi:10.1016/j.apsusc.2005.06.016 (2006).
- Garrod, S. *et al.* Integrated metabonomic analysis of the multiorgan effects of hydrazine toxicity in the rat. *Chem. Res. Toxicol.* 18, 115–122, doi:10.1021/tx0498915 (2005).
- 6. Occupational safety and health guideline for hydrazine potential human carcinogen. US Department of Health and Human Services (1988).
- 7. McAdam, K. *et al.* Analysis of hydrazine in smokeless tobacco products by gas chromatography-mass spectrometry. *Chem. Cent. J.* 9, 13–26, doi:10.1186/s13065-015-0089-0 (2015).
- Karimi-Maleh, H., Moazampour, M., Ensafi, A. A., Mallakpour, S. & Hatami, M. An electrochemical nanocomposite modified carbon paste electrode as a sensor for simultaneous determination of hydrazine and phenol in water and wastewater samples. *Environ. Sci. Pollut. Res. Int* 21, 5879–5888, doi:10.1007/s11356-014-2529-0 (2014).
- 9. Lakowicz, J. R. Principles of Fluorescence Spectroscopy, Springer: New York, NY, (2006).
- Tang, Y. et al. Development of fluorescent probes based on protection-deprotection of the key functional groups for biological imaging. Chem. Soc. Rev. 44, 5003–5015, doi:10.1039/c5cs00103j (2015).
- 11. Zhou, X., Lee, S., Xu, Z. & Yoon, J. Recent progress on the development of chemosensors for gases. *Chem. Rev.* **115**, 7944–8000, doi:10.1021/cr500567r (2015).
- Li, X., Gao, X., Shi, W. & Ma, H. Design strategies for water-soluble small molecular chromogenic and fluorogenic probes. *Chem. Rev.* 114, 590–659, doi:10.1021/cr300508p (2014).
- Chen, W. et al. A novel fluorescent probe for sensitive detection and imaging of hydrazine in living cells. Talanta. 162, 225–231, doi:10.1016/j.talanta.2016.10.026 (2017).
- 14. Ma, J. et al. Probing hydrazine with a near-infrared fluorescent chemodosimeter. Dyes Pigm. 138, 39-46, doi:10.1016/j. dyepig.2016.11.026 (2017).
- Mahapatraa, A. K., Karmakara, P., Mannaa, S., Maitia, K. & Mandal, D. Benzthiazole-derived chromogenic fluorogenic and ratiometric probes for detection of hydrazine in environmental samples and living cells. J. Photochem. Photobiol., A. 334, 1–12, doi:10.1016/j.jphotochem.2016.10.032 (2017).
- Goswami, S., Paul, S. & Manna, A. Fast and ratiometric "naked" eye detection of hydrazine for both solid and vapour phase sensing. New J. Chem. 39, 2300–2305, doi:10.1039/C4NJ02220C (2015).
- Jin, X. et al. A flavone-based ESIPT fluorescent sensor for detection of N₂H₄ in aqueous solution and gas state and its imaging in living cells. Sens. Actuators, B 216, 141–149, doi:10.1016/j.snb.2015.03.088 (2015).
- Sun, Y., Zhao, D., Fan, S. & Duan, L. A 4-hydroxynaphthalimide-derived ratiometric fluorescent probe for hydrazine and its *in vivo* applications. Sens. Actuators, B. 208, 512–517, doi:10.1016/j.snb.2014.11.057 (2015).
- 19. Yu., S. *et al.* A ratiometric two-photon fluorescent probe for hydrazine and its applications. *Sens. Actuators, B.* **220**, 1338–1345, doi:10.1016/j.snb.2015.07.051 (2015).
- Zhang, J. et al. Naked-eye and near-infrared fluorescence probe for hydrazine and its applications in *in vitro* and *in vivo* bioimaging. Anal. Chem. 87, 9101–9107, doi:10.1021/acs.analchem.5b02527 (2015).
- Zhou, J. *et al.* An ESIPT-based fluorescent probe for sensitive detection of hydrazine in aqueous solution. Org. Biomol. Chem. 13, 5344–5348, doi:10.1039/c5ob00209e (2015).
- 22. Cui., L. et al. Unique tri-output optical probe for specific and ultrasensitive detection of hydrazine. Anal. Chem. 86, 4611–4617, doi:10.1021/ac5007552 (2014).
- Goswami, S. et al. A reaction based colorimetric as well as fluorescence "turn on" probe for the rapid detection of hydrazine. RSC Adv. 4, 14210–14214, doi:10.1039/c3ra46663a (2014).
- 24. Liu, B. et al. Fluorescence monitor of hydrazine in vivo by selective deprotection of flavonoid. Sens. Actuators, B. 202, 194–200, doi:10.1016/j.snb.2014.05.010 (2014).
- Qian, Y., Lin, J., Han, L., Lin, L. & Zhu, H. A resorufin-based colorimetric and fluorescent probe for live-cell monitoring of hydrazine. *Biosens. Bioelectron.* 58, 282–286, doi:10.1016/j.bios.2014.02.059 (2014).
- Qu, D.-Y., Chen, J.-L. & Di, B. A fluorescence "switch-on" approach to detect hydrazine in aqueous solution at neutral pH. Anal. Methods. 6, 4705–4709, doi:10.1039/c4ay00533c (2014).
- Raju, M. V. R., Prakash, E. C., Chang, H.-C. & Lin, H.-C. A facile ratiometric fluorescent chemodosimeter for hydrazine based on Ing-Manske hydrazinolysis and its applications in living cells. *Dyes Pigm.* 103, 9–20, doi:10.1016/j.dyepig.2013.11.015 (2014).
- Li, Z. et al. A colorimetric and ratiometric fluorescent probe for hydrazine and itsapplication in living cells with low dark toxicity. Sens. Actuators, B. 241, 665–671, doi:10.1016/j.snb.2016.10.141 (2017).
- Imam Reja, S. I. et al. A charge transfer based ratiometric fluorescent probe for detection of hydrazine in aqueous medium and living cells. Sens. Actuators, B. 222, 923–929, doi:10.1016/j.snb.2015.08.078 (2016).
- Sun, M., Guo, J., Yang, Q., Xiao, N. & Li, Y. A new fluorescent and colorimetric sensor for hydrazine and its application in biological systems. J. Mater. Chem. B. 2, 1846–1851, doi:10.1039/c3tb21753a (2014).
- Dai, X., Wang, Z.-Y., Du, Z.-F., Miao, J.-Y. & Zhao, B.-X. A simple but effective near-infrared ratiometric fluorescent probe forhydrazine and its application in bioimaging. *Sens. Actuators, B.* 232, 369–374, doi:10.1016/j.snb.2016.03.159 (2016).
- Nandi, S. et al. Hydrazine selective dual signaling chemodosimetric probe in physiological conditions and its application in live cells. Anal. Chim. Acta. 893, 84–90, doi:10.1016/j.aca.2015.08.041 (2015).
- Goswami, S., Das, S., Aich, K., Sarkar, D. & Mondal, T. K. A coumarin based chemodosimetric probe for ratiometric detection of hydrazine. *Tetrahedron Lett.* 55, 2695–2699, doi:10.1016/j.tetlet.2014.03.041 (2014).
- 34. Xiao, L. *et al.* A fluorescent probe for hydrazine and its *in vivo* applications. *RSC Adv.* 4, 41807–41811, doi:10.1039/C4RA08101C (2014).

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Author Contributions

W. Lin and J.-Y. Wang conceived the idea and directed the work. J.-Y. Wang and Z.-R. Liu designed the experiments and performed the organic synthesis and spectral measurements. J.-Y. Wang and M. Ren performed the bioimaging and environmental experiments. All authors contributed to data analysis, manuscript writing and participated in research discussions.

Additional Information

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