



Artificial bioconjugates with naturally occurring linkages: the use of phosphodiester

Takao Shoji¹, Hiroki Fukutomi¹, Yohei Okada² and Kazuhiro Chiba^{*1}

Full Research Paper

Open Access

Address:

¹Department of Applied Biological Science, Tokyo University of Agriculture and Technology, 3-5-8 Saiwai-cho, Fuchu, Tokyo 183-8509, Japan and ²Department of Chemical Engineering, Tokyo University of Agriculture and Technology, 2-24-16 Naka-cho, Koganei, Tokyo 184-8588, Japan

Email:

Kazuhiro Chiba^{*} - chiba@cc.tuat.ac.jp

^{*} Corresponding author

Keywords:

alkyl chain soluble supports; artificial bioconjugates; naturally occurring linkages; 5'-phosphitylation; phosphodiester bonds

Beilstein J. Org. Chem. **2018**, *14*, 1946–1955.

doi:10.3762/bjoc.14.169

Received: 18 March 2018

Accepted: 06 July 2018

Published: 27 July 2018

Associate Editor: K. N. Allen

© 2018 Shoji et al.; licensee Beilstein-Institut.

License and terms: see end of document.

Abstract

Artificial orthogonal bond formations such as the alkyne–azide cycloaddition have enabled selective bioconjugations under mild conditions, yet naturally occurring linkages between native functional groups would be more straightforward to elaborate bioconjugates. Herein, we describe the use of a phosphodiester bond as a versatile option to access various bioconjugates. An opposite activation strategy, involving 5'-phosphitylation of the supported oligonucleotides, has allowed several biomolecules that possess an unactivated alcohol to be directly conjugated. It should be noted that there is no need to pre-install artificial functional groups and undesired and unpredictable perturbations possibly caused by bioconjugation can be minimized.

Introduction

Peptides and oligonucleotides are of exceptional importance since they are promising pharmaceutical candidates for the treatment of a range of diseases that are beyond traditional small molecule drugs [1-7]. Due to their iterative structures, chemical syntheses can technically be divided into two parts; deprotection and coupling reactions, enabling simple repeated procedures for their production, and bioactivities can poten-

tially be tuned by alteration of their sequences. Nowadays, not only canonical amino acids and/or nucleosides but also artificial building blocks with various functions are available, expanding chemical libraries of peptides and oligonucleotides significantly [8-15]. Furthermore, thus designed and synthesized peptides and oligonucleotides can be conjugated with each other to integrate their bioactivities and/or functions, which has been

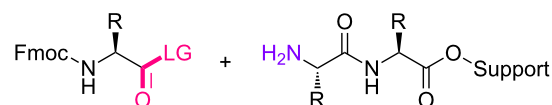
one of the central topics in the field of chemical biology [16–26]. Since peptide synthesis and oligonucleotide synthesis require different chemistries, such conjugations are typically carried out in the latter stages of the synthesis, otherwise subsequent steps are complicated.

Synthetic chemists, armed with various elegant artificial orthogonal bond formations, including alkyne–azide cycloaddition [27–34], thiol–ene ligation [35–38], Staudinger ligation [39,40], inverse-electron-demand Diels–Alder reaction [41–44], and hydrazone/oxime formation [45–48], have developed selective conjugation reactions under mild conditions. Although these bond-forming reactions have proven to be truly powerful approaches and will remain as first options to create novel bioconjugates, artificial functional groups such as alkynes or azides must be pre-installed into the respective biomolecules. Thus generated linkages are also artificial, possibly causing significant perturbation from their native forms, which generally affect bioactivities negatively. Arguably, naturally occurring linkages that can be formed between native functional groups would be safe alternatives, and are known as natural conjugates such as nucleopeptides and nucleolipids [49].

We have been developing alkyl chain soluble support (ACSS)-assisted liquid-phase methods, specifically for peptide and oligonucleotide syntheses [50–61]. In both cases, the supported reactants and products are soluble in less-polar solvents, allowing their chemical syntheses even in submolar concentrations. The supported products are readily separated as precipitates by the addition of polar solvents, and washing the precipitates with polar solvents simultaneously rinses away excess amino acids or nucleosides and coupling reagents. We have demonstrated multistep syntheses of up to 28-mers for peptides and 21-mers for oligonucleotides without column purification. In all cases, the C-terminal-activated amino acids or 3'-terminal-activated nucleosides are coupled to the N- or 5'-terminus of the supported reactants via amide or phosphodiester linkages (Figure 1). Such couplings could also be possible in the opposite activating manner. Namely, the activation of the N- or 5'-terminus of the supported reactants would be unique alternatives that allow the use of unactivated amino acids or nucleosides. In peptide synthesis, activation of the N-terminus is rather rare, except for some recent encouraging examples [62–68]; however, this is not the case for oligonucleotide synthesis [69–73]. The activation of the 5'-primary alcohol is expected to be even more effective than that of the 3'-secondary alcohols (Figure 2), and the 5'-primary alcohol could then be coupled to other nucleosides but also various alcohols via a naturally occurring phosphodiester linkage. Described herein is a simple and straightforward access to artificial bioconjugates with naturally occurring linkages.

(a) **general** peptide synthesis

activated amino acid + **unactivated** supported N-terminus

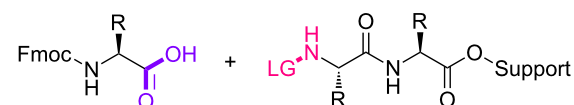


Fmoc: 9-fluorenylmethyloxycarbonyl

LG: leaving group

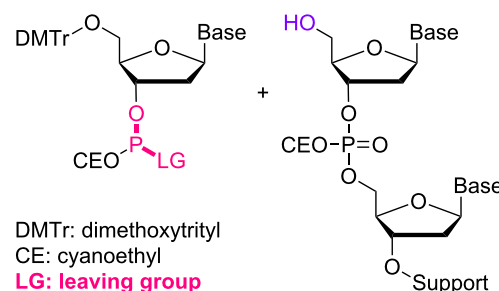
(b) **opposite** peptide synthesis (limited examples)

unactivated amino acid + **activated** supported N-terminus



(c) **general** oligonucleotide synthesis

activated nucleoside + **unactivated** supported 5'-terminus



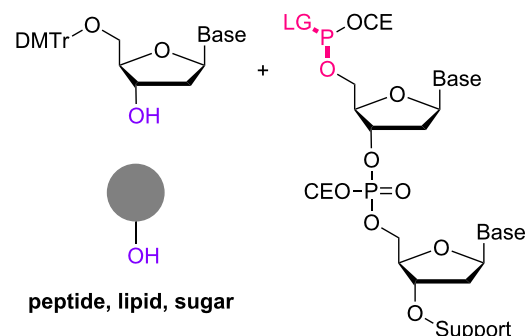
DMTr: dimethoxytrityl

CE: cyanoethyl

LG: leaving group

(d) **opposite** oligonucleotide synthesis (**this work**)

unactivated nucleoside + **activated** supported 5'-terminus



peptide, lipid, sugar

Figure 1: Schematic illustration of possible support-assisted methods.

Results and Discussion

The present work began with the optimization of the reaction conditions for the activation of the 5'-terminus, more specifically, 5'-phosphitylation [74]. The supported trinucleotide **1** was prepared from the support in 83% yield over 8 steps (Scheme S1, Supporting Information File 1) and used as a model in combination with 5-(benzylmercapto)-1*H*-tetrazole

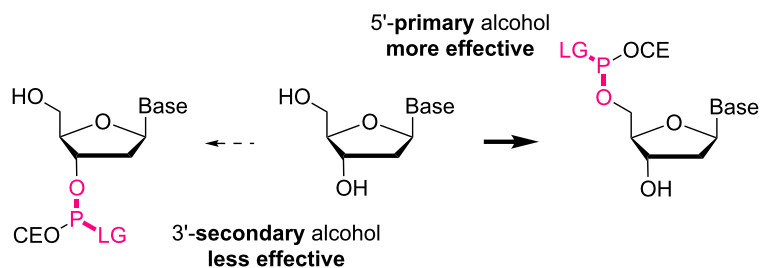


Figure 2: Expected reactivity of 5'- and 3'-terminus for the activation.

(BMT) as an activator (Table 1). The reaction was found to be rather sensitive to the concentration of the starting material (Table 1, entries 1–3). The best result was obtained at 25 mM concentration in dichloromethane (CH_2Cl_2). Although tetrahydrofuran (THF) is one of the typical reaction solvents for ACSS-assisted liquid-phase synthesis, this was not the case for the 5'-phosphitylation (Table 1, entry 4). When the activator was switched to tetrazole, the yield was slightly decreased

(Table 1, entry 5), while dicyanoimidazole (DCI) was proven to be an inefficient option for the reaction (Table 1, entry 6). It should be noted that the 5'-activated supported trinucleotide **2** was stable throughout the work-up procedure routinely used for the ACSS-assisted liquid-phase method. Namely, the 5'-activated supported trinucleotide **2** was readily separated as a precipitate by the addition of acetonitrile, and washing the precipitate with acetonitrile simultaneously rinses away excess

Table 1: Optimization of the conditions for the 5'-phosphitylation.

entry ^a	activator ^b	solvent	concentration (mM)	yield (%) ^{c,d}
1	BMT	CH_2Cl_2	5	61 (24)
2	BMT	CH_2Cl_2	25	91 (0)
3	BMT	CH_2Cl_2	100	0 (81)
4	BMT	THF	25	0 (86)
5	tetrazole	CH_2Cl_2	25	83 (11)
6	DCI	CH_2Cl_2	25	40 (46)

^aReactions were carried out using 3.0 equiv of 2-cyanoethyl-*N,N,N',N'*-tetraisopropylphosphorodiamidite in the presence of 3 Å MS at rt for 30 min. ^b3.0 equiv was used. ^cYields were determined by ³¹P NMR analysis. ^dRecovered starting material was reported in parenthesis.

reagents to afford the pure form, which could be used for further reactions without column purification.

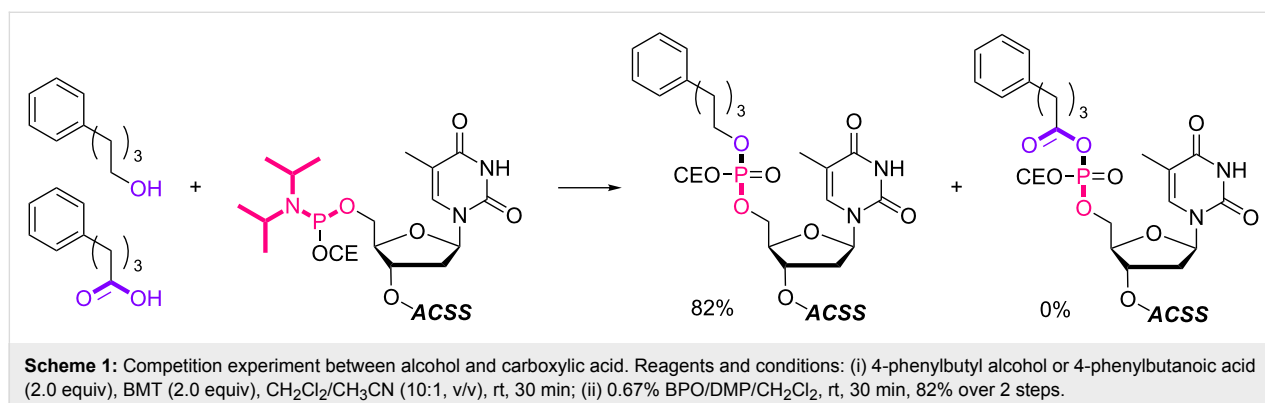
With the optimized 5'-phosphitylation conditions in hand, we then investigated conjugation using the 5'-activated supported trinucleotide **2** as a model. As expected, unactivated nucleosides could be coupled to the activated 5'-terminus without difficulty (Scheme S2, Supporting Information File 1). Furthermore, to our delight, the conjugation was compatible with carboxylic acids that are potential nucleophiles for the activated 5'-terminus and can also induce side reactions of the phosphoramidite (Scheme 1). This compatibility is of immense versatility since synthesized peptide fragments, which are cleaved from their supports thus freeing the C-terminus, could directly be conjugated to the 5'-terminus of oligonucleotides via tyrosine, serine, or threonine side chains. In order to demonstrate such a versatile conjugation, tripeptide **3** was prepared from the support in 94% yield over 6 steps (Scheme S3, Supporting Information File 1) and used as a model in combination with the 5'-activated supported trinucleotide **2** (Scheme 2 and Figure 3). To our satisfaction, the conjugation took place smoothly to afford the desired bioconjugation product **4** and basic deprotection gave the native form **5** with a naturally occurring linkage in 70% yield over 5 steps. The impurities can reasonably be assigned as the hydrolyzed and/or oxidized products of the 5'-activated supported trinucleotide **2** [75].

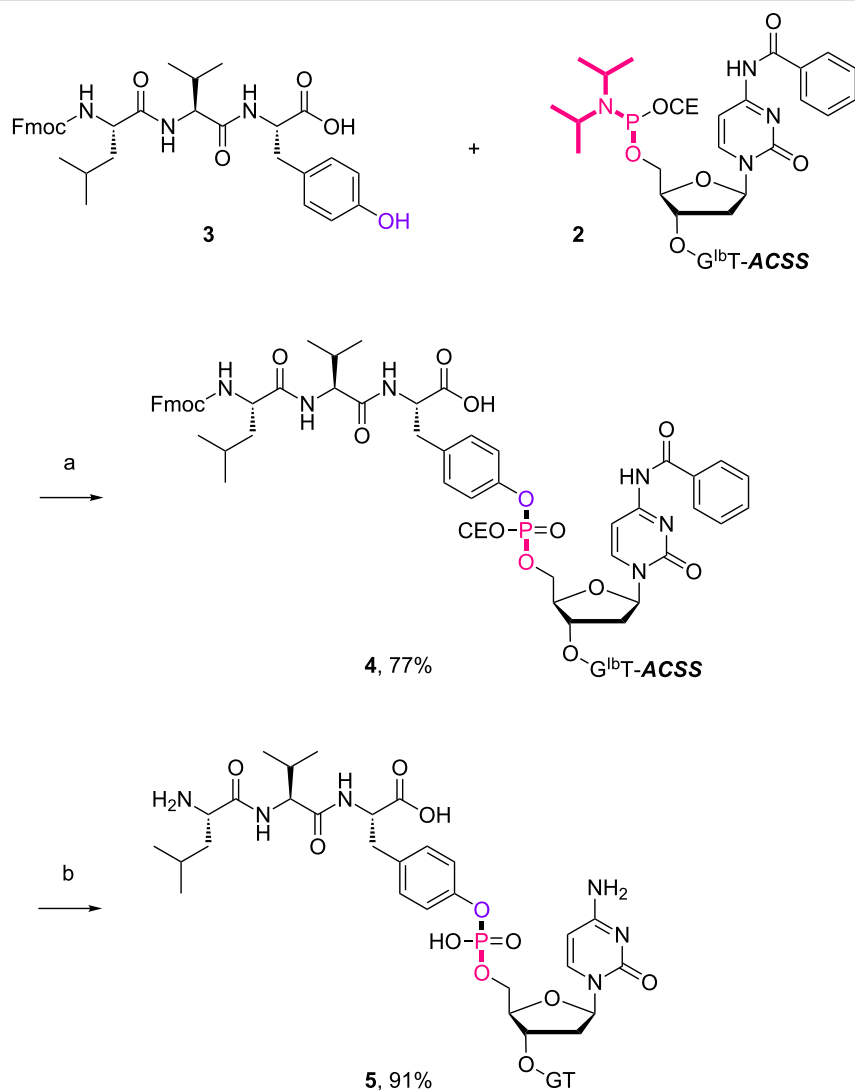
In order to investigate the generality of the current conjugation approach, we then turned our attention to the use of relatively longer peptides, sugars, and lipids as biomolecules (Table 2) [76,77]. When pentapeptide **6** was used instead (see Scheme S4 in Supporting Information File 1 for preparation of pentapeptide **6**), however, the conjugation was not successful at all (Table 2, entry 1). No desired conjugate was obtained under the same reaction conditions that were effective for the conjugation with tripeptide **3**. In this case, no other functional group that could inhibit the conjugation existed in the pentapeptide **6**, and therefore, it could be rationalized that the physical properties,

such as solubility, were the issue. This hypothesis was confirmed by the successful conjugation with supported pentapeptide **7**, affording the desired bioconjugate **8** in 67% yield over 2 steps (Table 2, entry 2). It is well-known that peptides with specific sequences can cause aggregation and thus severely inhibit further reactions; however, our support could possibly address such technical problems. Gratifyingly, although further optimization of the reaction conditions is needed for the deprotection reactions (see Schemes S5 and S6 in Supporting Information File 1 for side reactions), both the lipid **9** and the protected sugar **10** were also effectively conjugated to the activated 5'-terminus under the same reaction conditions to give the desired bioconjugates **11** and **12**, respectively (Table 2, entries 3 and 4). We finally examined whether relatively longer supported oligonucleotides were compatible with the methodology disclosed here. Therefore, supported decanucleotide **13** was prepared from the support in 54% yield over 27 steps (Scheme S7, Supporting Information File 1). The results were better than expected, our optimized conditions for the 5'-phosphitylation, conjugation, and deprotection were all even more effective with the supported decanucleotide **13** than with the supported trinucleotide **1** and the desired bioconjugate **16** was obtained in 84% yield over 6 steps (Scheme 3, Scheme 4 and Figure 4).

Conclusion

In conclusion, we have demonstrated that the phosphodiester bond can be an effective linkage not only to construct oligonucleotides but also to conjugate them to various biomolecules, including peptides, sugars, and lipids. The development of a method to activate the supported 5'-terminus, affording useful stable phosphoramidites that are compatible with ACSS-assisted liquid-phase synthesis, has enabled direct conjugation using unactivated alcohols. It should be noted that the thus obtained bioconjugates were all artificial, but constructed via naturally occurring linkages between native functional groups. The approach presented herein should be promising to design and synthesize novel bioconjugates without undesired and unpredictable perturbations possibly derived from artificial linkages.





Scheme 2: Conjugation between the 5'-activated supported trinucleotide **2** and the tripeptide **3**. Reagents and conditions: (a) (i) tripeptide (2.0 equiv), BMT (2.0 equiv), CH₂Cl₂/CH₃CN/DMF (20:2:1, v/v/v), rt, 30 min; (ii) 0.67% BPO/DMP/CH₂Cl₂, rt, 30 min, 77% over 2 steps. (b) (i) 5% DCA/CH₂Cl₂, rt, 5 min; (ii) NH₃ aq/EtOH (3:1, v/v), 70 °C, 3 h; (iii) TEA·3HF/DMF, rt, 24 h, 91% over 3 steps.

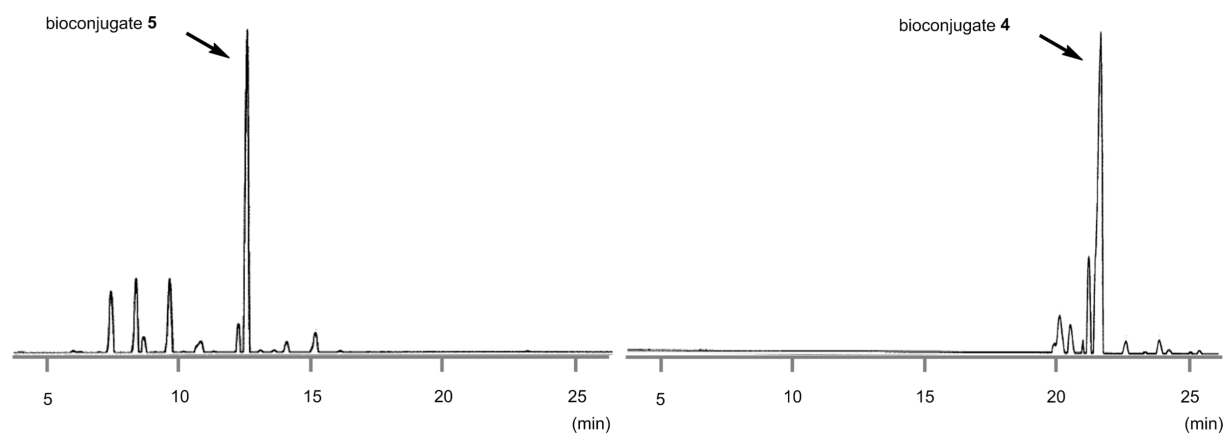


Figure 3: HPLC spectra of the crude protected bioconjugate **4** and the crude deprotected bioconjugate **5**.

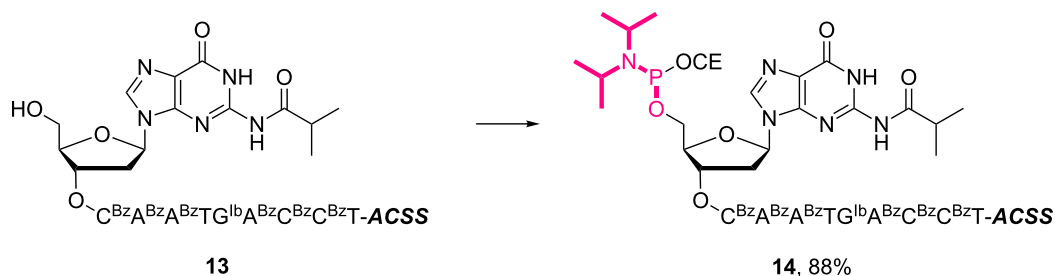
Table 2: Conjugation between the 5'-activated supported trinucleotide **2** and several biomolecules. Reagents and conditions: (a) (i) biomolecule (2.0 equiv), BMT (2.0 equiv), CH₂Cl₂/CH₃CN/DMF (20:2:1, v/v/v), rt, 30 min; (ii) 0.67% BPO/DMP/CH₂Cl₂, rt, 30 min. (b) (i) 5% DCA/CH₂Cl₂, rt, 5 min; (ii) NH₃ aq/EtOH (3:1, v/v), 70 °C, 3 h; (iii) TEA 3HF/DMF, rt, 24 h.

entry	biomolecule	conjugation yield (%) ^a	bioconjugate	deprotection yield (%) ^b
1	 6	n.d.	 8	–
2	 7	70	 8	67
3 ^c	 9	68	 11	28

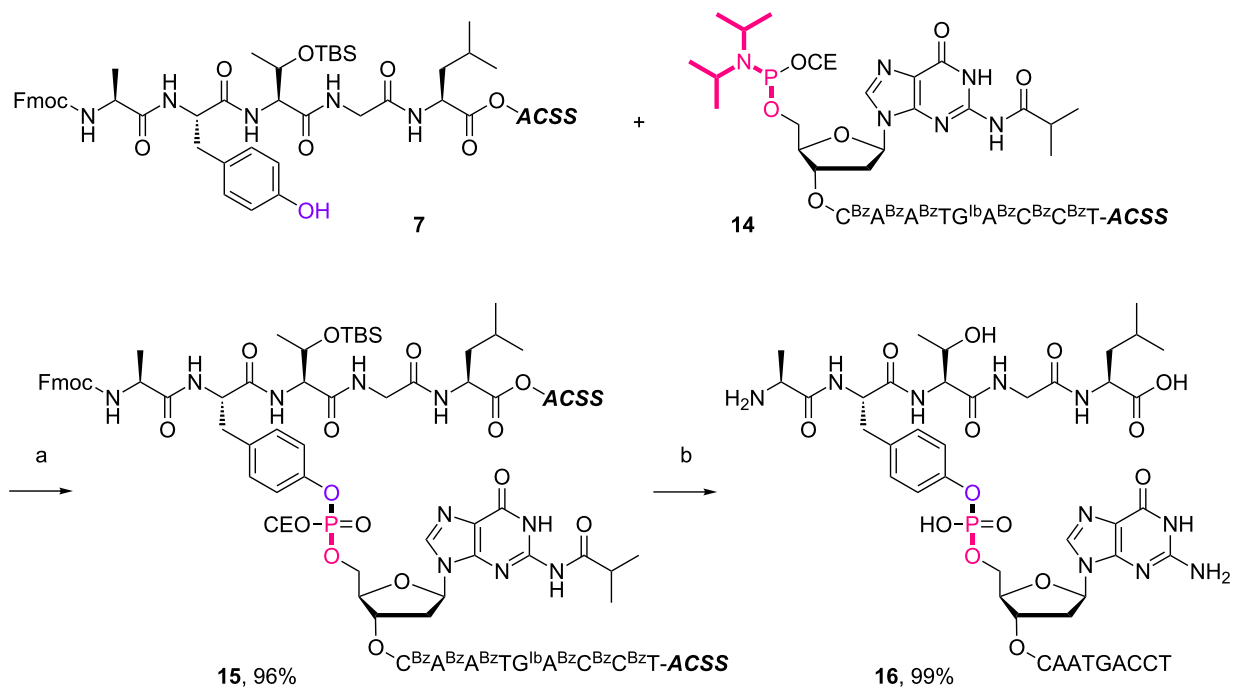
Table 2: Conjugation between the 5'-activated supported trinucleotide **2** and several biomolecules. Reagents and conditions: (a) (i) biomolecule (2.0 equiv), BMT (2.0 equiv), CH₂Cl₂/CH₃CN/DMF (20:2:1, v/v/v), rt, 30 min; (ii) 0.67% BPO/DMP/CH₂Cl₂, rt, 30 min. (b) (i) 5% DCA/CH₂Cl₂, rt, 5 min; (ii) NH₃ aq/EtOH (3:1, v/v), 70 °C, 3 h; (iii) TEA 3HF/DMF, rt, 24 h. (continued)



^aYields were determined by ³¹P NMR and HPLC analyses. ^bYields were determined by HPLC analysis. ^cCH₂Cl₂/CH₃CN (10:1, v/v) was used as a solvent.



Scheme 3: 5'-Phosphitylation of supported decanucleotide **13**. Reagents and conditions: 2-cyanoethyl-*N,N,N',N'*-tetraisopropylphosphorodiamidite (3.0 equiv), BMT (2.0 equiv), CH₂Cl₂/CH₃CN (10:1, v/v), rt, 30 min, 88%.



Scheme 4: Conjugation between 5'-activated supported decanucleotide **14** and supported pentapeptide **7**. Reagents and conditions: (a) (i) pentapeptide (2.0 equiv), BMT (2.0 equiv), CH₂Cl₂/CH₃CN/DMF (20:2:1, v/v/v), rt, 30 min; (ii) 0.67% BPO/DMP/CH₂Cl₂, rt, 30 min, 96% over 2 steps. (b) (i) 5% DCA/CH₂Cl₂, rt, 5 min; (ii) NH₃ aq/EtOH (3:1, v/v), 70 °C, 3 h; (iii) TEA 3HF/DMF, rt, 24 h, 99% over 3 steps.

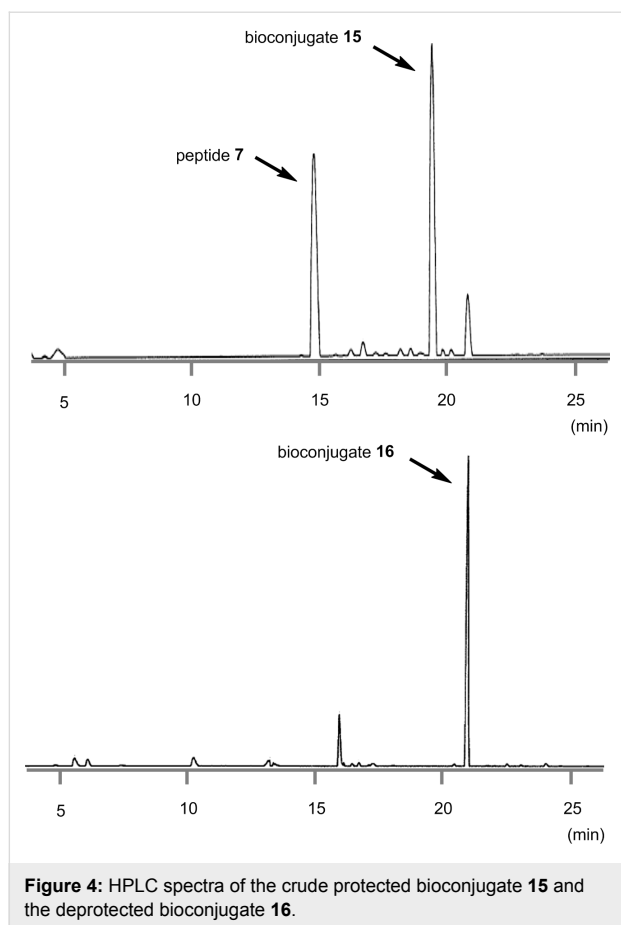


Figure 4: HPLC spectra of the crude protected bioconjugate **15** and the deprotected bioconjugate **16**.

Supporting Information

Supporting Information File 1

Additional schemes and figures, general remarks, synthesis and characterization data, including copies of ^1H and ^{13}C NMR.

[<https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-14-169-S1.pdf>]

Acknowledgements

This work was partially supported by JSPS KAKENHI Grant Numbers 15H04494, 17K19222 (to K. C.), 16H06193, and 17K19221 (to Y. O.).

ORCID® IDs

Yohei Okada - <https://orcid.org/0000-0002-4353-1595>

References

- Passioura, T.; Katoh, T.; Goto, Y.; Suga, H. *Annu. Rev. Biochem.* **2014**, *83*, 727–752. doi:10.1146/annurev-biochem-060713-035456
- Giordanetto, F.; Kihlberg, J. *J. Med. Chem.* **2014**, *57*, 278–295. doi:10.1021/jm400887j
- Bhat, A.; Roberts, L. R.; Dwyer, J. J. *Eur. J. Med. Chem.* **2015**, *94*, 471–479. doi:10.1016/j.ejmech.2014.07.083
- Craik, D. J.; Fairlie, D. P.; Liras, S.; Price, D. *Chem. Biol. Drug Des.* **2013**, *81*, 136–147. doi:10.1111/cbdd.12055
- Wittrup, A.; Lieberman, J. *Nat. Rev. Genet.* **2015**, *16*, 543–552. doi:10.1038/nrg3978
- Mansoori, B.; Shotorbani, S. S.; Baradaran, B. *Adv. Pharm. Bull.* **2014**, *4*, 313–321. doi:10.5681/apb.2014.046
- Khorkova, O.; Hsiao, J.; Wahlestedt, C. *Pharm. Pat. Anal.* **2013**, *2*, 215–229. doi:10.4155/ppa.13.4
- Tao, K.; Levin, A.; Adler-Abramovich, L.; Gazit, E. *Chem. Soc. Rev.* **2016**, *45*, 3935–5679. doi:10.1039/C5CS00889A
- Avan, I.; Hall, C. D.; Katritzky, A. R. *Chem. Soc. Rev.* **2014**, *43*, 3575–3594. doi:10.1039/c3cs60384a
- Wojciechowska, F.; Leumann, C. J. *Chem. Soc. Rev.* **2011**, *40*, 5669–5679. doi:10.1039/c1cs15027h
- Bandy, T. J.; Brewer, A.; Burns, J. A.; Marth, G.; Nguyen, T.; Stulz, E. *Chem. Soc. Rev.* **2011**, *40*, 138–148. doi:10.1039/B820255A
- Lebreton, J.; Escudier, J.-M.; Arzel, L.; Len, C. *Chem. Rev.* **2010**, *110*, 3371–3418. doi:10.1021/cr800465j
- Sinkeldam, R. W.; Greco, N. J.; Tor, Y. *Chem. Rev.* **2010**, *110*, 2579–2619. doi:10.1021/cr900301e
- Prakash, T. P. *Chem. Biodiversity* **2011**, *8*, 1616–1641. doi:10.1002/cbdv.201100081
- Leumann, C. J. *Bioorg. Med. Chem.* **2002**, *10*, 841–854. doi:10.1016/S0968-0896(01)00348-0
- Albada, B.; Metzler-Nolte, N. *Chem. Rev.* **2016**, *116*, 11797–11839. doi:10.1021/acs.chemrev.6b00166
- Liu, S. *Bioconjugate Chem.* **2015**, *26*, 1413–1438. doi:10.1021/acs.bioconjchem.5b00327
- Liu, H.; Irvine, D. J. *Bioconjugate Chem.* **2015**, *26*, 791–801. doi:10.1021/acs.bioconjchem.5b00103
- Kobayashi, H.; Turkbey, B.; Watanabe, R.; Choyke, P. L. *Bioconjugate Chem.* **2014**, *25*, 2093–2100. doi:10.1021/bc500481x
- Melnyk, O.; Ollivier, N.; Besret, S.; Melnyk, P. *Bioconjugate Chem.* **2014**, *25*, 629–639. doi:10.1021/bc500052r
- El-Mahdi, O.; Melnyk, O. *Bioconjugate Chem.* **2013**, *24*, 735–765. doi:10.1021/bc300516f
- Koniev, O.; Wagner, A. *Chem. Soc. Rev.* **2015**, *44*, 5495–5551. doi:10.1039/C5CS00048C
- Singh, Y.; Murata, P.; Defrancq, E. *Chem. Soc. Rev.* **2010**, *39*, 2054–2070. doi:10.1039/b911431a
- Canalle, L. A.; Löwik, D. W. P. M.; van Hest, J. C. M. *Chem. Soc. Rev.* **2010**, *39*, 329–353. doi:10.1039/B807871H
- Gunnou, S. B.; Madder, A. *Org. Biomol. Chem.* **2016**, *14*, 8002–8013. doi:10.1039/C6OB00808A
- Schilling, C. I.; Jung, N.; Biskup, M.; Schepers, U.; Bräse, S. *Chem. Soc. Rev.* **2011**, *40*, 4840–4871. doi:10.1039/c0cs00123f
- Johansson, J. R.; Beke-Somfai, T.; Stalsmeden, A. S.; Kann, N. *Chem. Rev.* **2016**, *116*, 14726–14768. doi:10.1021/acs.chemrev.6b00466
- Haldón, E.; Nicasio, M. C.; Pérez, P. J. *Org. Biomol. Chem.* **2015**, *13*, 9528–9550. doi:10.1039/C5OB01457C
- Paredes, E.; Das, S. R. *ChemBioChem* **2011**, *12*, 125–131. doi:10.1002/cbic.201000466
- El-Sagheer, A. H.; Brown, T. *Chem. Soc. Rev.* **2010**, *39*, 1388–1405. doi:10.1039/b901971p

31. Kolb, H. C.; Finn, M. G.; Sharpless, K. B. *Angew. Chem., Int. Ed.* **2001**, *40*, 2004–2021. doi:10.1002/1521-3773(20010601)40:11<2004::AID-ANIE2004>3.0.CO;2-5
32. Rostovtsev, V. V.; Green, L. G.; Fokin, V. V.; Sharpless, K. B. *Angew. Chem., Int. Ed.* **2002**, *41*, 2596–2599. doi:10.1002/1521-3773(20020715)41:14<2596::AID-ANIE2596>3.0.CO;2-4
33. Tornøe, C. W.; Christensen, C.; Meldal, M. *J. Org. Chem.* **2002**, *67*, 3057–3064. doi:10.1021/jo011148j
34. Agard, N. J.; Prescher, J. A.; Bertozzi, C. R. *J. Am. Chem. Soc.* **2004**, *126*, 15046–15047. doi:10.1021/ja044996f
35. Dondoni, A.; Marra, A. *Chem. Soc. Rev.* **2012**, *41*, 573–586. doi:10.1039/C1CS15157F
36. Hoyle, C. E.; Bowman, C. N. *Angew. Chem., Int. Ed.* **2010**, *49*, 1540–1573. doi:10.1002/anie.200903924
37. Meyer, A.; Vasseur, J.-J.; Morvan, F. *Eur. J. Org. Chem.* **2013**, 465–473. doi:10.1002/ejoc.201201311
38. Killups, K. L.; Campos, L. M.; Hawker, C. J. *J. Am. Chem. Soc.* **2008**, *130*, 5062–5064. doi:10.1021/ja8006325
39. Weisbrod, S. H.; Baccaro, A.; Marx, A. *Methods Mol. Biol. (N. Y., NY, U. S.)* **2011**, *751*, 195–207. doi:10.1007/978-1-61779-151-2_12
40. Saxon, E.; Bertozzi, C. R. *Science* **2000**, *287*, 2007–2010. doi:10.1126/science.287.5460.2007
41. Knall, A.-C.; Slugovc, C. *Chem. Soc. Rev.* **2013**, *42*, 5131–5142. doi:10.1039/c3cs60049a
42. Bußkamp, H.; Batroff, E.; Niederwieser, A.; Abdel-Rahman, O. S.; Winter, R. F.; Wittmann, V.; Marx, A. *Chem. Commun.* **2014**, *50*, 10827–10829. doi:10.1039/C4CC04332D
43. Schoch, J.; Staudt, M.; Samanta, A.; Wiessler, M.; Jäschke, A. *Bioconjugate Chem.* **2012**, *23*, 1382–1386. doi:10.1021/bc300181n
44. Blackman, M. L.; Roysen, M.; Fox, J. M. *J. Am. Chem. Soc.* **2008**, *130*, 13518–13519. doi:10.1021/ja8053805
45. Collins, J.; Xiao, Z.; Müllner, M.; Connal, L. A. *Polym. Chem.* **2016**, *7*, 3812–3826. doi:10.1039/C6PY00635C
46. Meyer, A.; Spinelli, N.; Dumy, P.; Vasseur, J. J.; Morvan, F.; Defranco, E. *J. Org. Chem.* **2010**, *75*, 3927–3930. doi:10.1021/jo100599m
47. Dirksen, A.; Hackeng, T. M.; Dawson, P. E. *Angew. Chem., Int. Ed.* **2006**, *45*, 7581–7584. doi:10.1002/anie.200602877
48. Zatsepin, T. S.; Stetsenko, D. A.; Arzumanov, A. A.; Romanova, E. A.; Gait, M. J.; Oretskaya, T. S. *Bioconjugate Chem.* **2002**, *13*, 822–830. doi:10.1021/bc020016+
49. Rosemeyer, H. *Chem. Biodiversity* **2005**, *2*, 977–1063. doi:10.1002/cbdv.200590082
50. Ogami, K.; Okada, Y.; Chiba, K. *Chem. Lett.* **2018**, *47*, 138–140. doi:10.1246/cl.170971
51. Wakamatsu, H.; Okada, Y.; Sugai, M.; Hussaini, S. R.; Chiba, K. *Asian J. Org. Chem.* **2017**, *6*, 1584–1588. doi:10.1002/ajoc.201700401
52. Okada, Y.; Asama, H.; Wakamatsu, H.; Chiba, K.; Kamiya, H. *Eur. J. Org. Chem.* **2017**, 5961–5965. doi:10.1002/ejoc.201700697
53. Matsuno, Y.; Shoji, T.; Kim, S.; Chiba, K. *Org. Lett.* **2016**, *18*, 800–803. doi:10.1021/acs.orglett.6b00077
54. Okada, Y.; Wakamatsu, H.; Sugai, M.; Kauppinen, E. I.; Chiba, K. *Org. Lett.* **2015**, *17*, 4264–4267. doi:10.1021/acs.orglett.5b02057
55. Okada, Y.; Hosoya, S.; Suzuki, H.; Chiba, K. *Org. Lett.* **2014**, *16*, 6448–6451. doi:10.1021/ol5032798
56. Shoji, T.; Kim, S.; Chiba, K. *Chem. Lett.* **2014**, *43*, 1251–1253. doi:10.1246/cl.140355
57. Kitada, S.; Fujita, S.; Okada, Y.; Kim, S.; Chiba, K. *Tetrahedron* **2013**, *69*, 2555–2559. doi:10.1016/j.tet.2013.01.068
58. Okada, Y.; Suzuki, H.; Nakae, T.; Fujita, S.; Abe, H.; Nagano, K.; Yamada, T.; Ebata, N.; Kim, S.; Chiba, K. *J. Org. Chem.* **2013**, *78*, 320–327. doi:10.1021/jo302127d
59. Fujita, Y.; Fujita, S.; Okada, Y.; Chiba, K. *Org. Lett.* **2013**, *15*, 1155–1157. doi:10.1021/ol4003477
60. Kitada, S.; Takahashi, M.; Yamaguchi, Y.; Okada, Y.; Chiba, K. *Org. Lett.* **2012**, *14*, 5960–5963. doi:10.1021/ol302863r
61. Tana, G.; Kitada, S.; Fujita, S.; Okada, Y.; Kim, S.; Chiba, K. *Chem. Commun.* **2010**, *46*, 8219–8221. doi:10.1039/c0cc03090b
62. Zhu, Y.-P.; Sergeyev, S.; Franck, P.; Orru, R. V. A.; Maes, B. U. W. *Org. Lett.* **2016**, *18*, 4602–4605. doi:10.1021/acs.orglett.6b02247
63. Wu, X.; Stockdill, J. L.; Park, P. K.; Danishefsky, S. J. *J. Am. Chem. Soc.* **2012**, *134*, 2378–2384. doi:10.1021/ja2103372
64. Li, X.; Yuan, Y.; Kan, C.; Danishefsky, S. J. *J. Am. Chem. Soc.* **2008**, *130*, 13225–13227. doi:10.1021/ja804709s
65. Suppo, J.-S.; Subra, G.; Berges, M.; de Figueiredo, R. M.; Campagne, J.-M. *Angew. Chem., Int. Ed.* **2014**, *53*, 5389–5393. doi:10.1002/anie.201402147
66. Grzyb, J. A.; Batey, R. A. *Tetrahedron Lett.* **2003**, *44*, 7485–7488. doi:10.1016/j.tetlet.2003.08.026
67. Kosal, A. D.; Wilson, E. E.; Ashfeld, B. L. *Angew. Chem., Int. Ed.* **2012**, *51*, 12036–12040. doi:10.1002/anie.201206533
68. Hegarty, A. F.; McCormack, M. T.; Ferguson, G.; Roberts, P. J. *J. Am. Chem. Soc.* **1977**, *99*, 2015–2016. doi:10.1021/ja00448a075
69. van der Heden van Noort, G. J.; van Delft, P.; Meeuwenoord, N. J.; Overkleef, H. S.; van der Marel, G. A.; Filippov, D. V. *Chem. Commun.* **2012**, *48*, 8093–8095. doi:10.1039/c2cc33477a
70. van der Heden van Noort, G. J.; Overkleef, H. S.; van der Marel, G. A.; Filippov, D. V. *J. Org. Chem.* **2010**, *75*, 5733–5736. doi:10.1021/jo100757t
71. Kriek, N. M. A. J.; Meeuwenoord, N. J.; van den Elst, H.; Heus, H. A.; van der Marel, G. A.; Filippov, D. V. *Org. Biomol. Chem.* **2006**, *4*, 3576–3586. doi:10.1039/b608544j
72. Schmidt, K. S.; Filippov, D. V.; Meeuwenoord, N. J.; van der Marel, G. A.; van Boom, J. H.; Lippert, B.; Reedijk, J. *Angew. Chem., Int. Ed.* **2000**, *39*, 375–377. doi:10.1002/(SICI)1521-3773(20000117)39:2<375::AID-ANIE375>3.0.CO;2-M
73. Zhou, X.; Remaud, G.; Chattopadhyaya, J. *Tetrahedron* **1988**, *44*, 6471–6489. doi:10.1016/S0040-4020(01)89837-9
74. Wei, X. *Tetrahedron* **2013**, *69*, 3615–3637. doi:10.1016/j.tet.2013.03.001
75. Kungurtsev, V.; Laakkonen, J.; Molina, A. G.; Virta, P. *Eur. J. Org. Chem.* **2013**, 6687–6693. doi:10.1002/ejoc.201300864
76. Venkatesan, N.; Kim, B. H. *Chem. Rev.* **2006**, *106*, 3712–3761. doi:10.1021/cr0502448
77. Spinelli, N.; Defranco, E.; Morvan, F. *Chem. Soc. Rev.* **2013**, *42*, 4557–4573. doi:10.1039/C2CS35406C

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (<http://creativecommons.org/licenses/by/4.0>). Please note that the reuse, redistribution and reproduction in particular requires that the authors and source are credited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (<https://www.beilstein-journals.org/bjoc>)

The definitive version of this article is the electronic one which can be found at:
[doi:10.3762/bjoc.14.169](https://doi.org/10.3762/bjoc.14.169)