

Article

Anti-Inflammatory and Antioxidant Components from *Hygroryza aristata*

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Abstract: Twenty-six known compounds and two new compounds, including a new lignan, $(7S^*, 8R^*, 7^*, 8^*, 8^*, 5^*)$ -icariol A₂-9-*O*- β -xylopyranoside (1), and a new indole alkaloid, hygarine (2), were isolated from the extracts of *Hygroryza aristata* (Gramineae). The structures of all compounds were elucidated on the basis of NMR spectral analysis. The compounds (-)-epigallocatechin-3-*O*-gallate (4) and (-)-epicatechin-3-*O*-gallate (5) possess free radical scavenging activities and compound 1 could inhibit superoxide anion generation and elastase release by fMLP/CB-induced human neutrophils with IC₅₀ values of 19.33 ± 0.86 and 24.14 ± 1.59 µM, respectively.

Keywords: Hygroryza aristata; Gramineae; hygarine; anti-inflammatory; antioxidant

1. Introduction

Hygroryza aristata (Gramineae) is a stoloniferous perennial plant widely distributed in tropical Asia [1]. This plant grows in paddy fields and ponds at 400–800 meters above sea level, often forming floating clusters in lakes and slow-flowing rivers, and is a forage favored by cattle. Nevertheless, to our knowledge the chemical and biological activities of this plant have not been reported previously. In the search for biologically active constituents from natural sources, we determined that the methanolic extract of this plant was able to scavenge DPPH free radicals. In the present study two new compounds, $(7S^*, 8R^*, 7'R^*, 8'S^*)$ -icariol A₂-9-*O*- β -xylopyranoside (1) and hygarine (2), were isolated together with 26 known compounds from the extracts of the *H. aristata* (Figure 1). Several of these compounds were evaluated for their antioxidant and anti-inflammatory activities.

OCH₃ OCH₃ HO OH H₃CO OCH₃ но 1 2 OH HO соон HO. ОН || 0 OH ÓН

Figure 1. Structures of compounds 1–5.

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2. Results and Discussion

Compound 1 was obtained as a colorless syrup. Its IR spectrum exhibited hydroxyl and aromatic group absorption bands at 3,375 cm⁻¹ and 1,517 cm⁻¹, respectively. The UV spectrum revealed absorption maxima at 237 and 279 nm. The ¹H-NMR data (Table 1) showed two typical 1,3,4,5-tetrasubstituted phenyl ring protons at δ 6.75 and 6.79 (each 2H, *s*), one anomeric proton at δ 4.23 (1H, *d*, *J* = 7.6 Hz) and four methoxyl groups at δ 3.87 (12H, s). Four methines at δ 2.34, 2.46, 5.01, 5.10 and two methylenes at δ 3.62, 3.67, 3.74, 4.01 were observed in the ¹H-NMR and HMQC spectra. The ¹³C signals at δ 66.0, 70.2, 73.9, 77.0, 104.3 and the coupling constant (*J* = 7.6 Hz) of anomeric proton suggested that 1 contained a β -D-xylose moiety.

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4 :R=OH

5 :R=H

Position	δc	δ _H
1	133.2	
2, 6	103.9	6.75 (2H, <i>s</i>)
3, 5	148.3	
4	135.2	
7	83.6	5.10 (1H, d, J = 8.4 Hz)
8	50.9	2.34 (1H, <i>m</i>)
9a	68.6	3.62 (1H, <i>m</i>)
9b		4.01 (1H, <i>dd</i> , <i>J</i> = 10.0, 4.8 Hz)
1'	133.2	
2', 6'	104.0	6.79 (2H, <i>s</i>)
3', 5'	148.4	
4'	135.3	
7'	83.3	5.01 (1H, d, J = 8.4 Hz)
8'	53.5	2.46 (1H, <i>m</i>)
9'a	60.1	3.67 (1H, <i>m</i>)
9'b		3.74 (1H, <i>dd</i> , <i>J</i> = 11.6, 4.4 Hz)
3, 5, 3', 5'-OCH ₃	55.9	3.87 (12H, <i>s</i>)
Xyl-1	104.3	4.23 (1H, d, J = 7.6 Hz)
Xyl-2	73.9	3.20 (1H, <i>m</i>)
Xyl-3	77.0	3.31 (1H, <i>m</i>)
Xyl-4	70.2	3.48 (1H, <i>m</i>)
Xyl-5	66.0	3.20 (1H, <i>m</i>)
		3.84 (1H, dd, J = 6.4, 2.4 Hz)

Table 1. ¹H- and ¹³C-NMR data for compound **1** (400 and 100 MHz in methanol- d_4).

HMBC correlations between H-1', H-2', H-6'/C-7' and H-1, H-2, H-6/C-7 indicated that the 1,3,4,5- tetrasubstituted phenyl rings were connected to C-7' and C-7, respectively (Figure 2). The ${}^{3}J$ -correlation signals between H-9 (δ H 3.62, 4.01) and C-1" (δ C 104.3) indicated the presence of a xylose unit attached at C-9.





The planar structure of **1** is similar to that of icariol A₂-9-O- β -xylopyranoside [2,3]. In a NOESY experiment, H-8/ H-2, 6; H-8'/H-2', 6' and H-8/H-8' NOE correlations were observed, revealing that

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relative configuration of **1** as shown in Figure 2. Thus, the structure of compound **1** was determined to be $(7S^*, 8R^*, 7^*R^*, 8^*S^*)$ -icariol A₂-9-*O*- β -xylopyranoside.

Compound **2** was obtained as a brown syrup with an elemental composition of $C_{19}H_{18}N_2O_3$, as determined by its HRFAB-MS ($[M+1]^+ m/z$ 323.3734). The IR spectrum of **2** displayed absorption bands from a hydroxyl group (3,404 cm⁻¹) and an amide group (1,643 cm⁻¹). In the UV spectrum of **2**, absorption maxima were observed at 224 (sh), 288, and 312 (sh) nm, revealing that the compound has an indole skeleton [4].

Position	δ _C	δ _H
2	123.5	7.09 (1H, d, J = 2.4 Hz)
3	112.1	
4	112.0	7.19 (1H, d, J = 8.4 Hz)
5	111.9	6.70 (1H, <i>dd</i> , <i>J</i> = 8.4, 2.0 Hz)
6	151.1	
7	103.0	7.03 (1H, $d, J = 2.0$ Hz)
8	132.0	
9	128.8	
1'	26.1	2.89 (2H, $t, J = 7.6$ Hz)
2'	40.2	3.59 (2H, <i>dd</i> , <i>J</i> = 13.6, 7.6 Hz)
1"	166.1	
2"	119.4	7.48 (1H, d , J = 16.0 Hz)
3"	139.6	6.49 (1H, d, J = 16.0 Hz)
4"	127.3	
5", 9"	129.6	7.42 (2H, d, J = 8.4 Hz)
6", 8"	116.1	6.85 (2H, d, J = 8.4 Hz)
7"	159.3	
1-NH		7.74 (1H, br. <i>s</i>)
3'-NH		7.30 (1H, br.s)
6-OH		8.86 (1H, br.s)
7" - OH		9.75 (1H, br.s)

Table 2. ¹H- and ¹³C-NMR data for compound **2** (400 and 100 MHz in acetone- d_6).

In the ¹H-NMR data of **2**, shown in Table 2, the three mutually coupled aromatic protons at δ 6.70 (1H, *dd*, *J* = 8.4, 2.0 Hz), 7.03 (1H, *d*, *J* = 2.0 Hz) and 7.19 (1H, *d*, *J* = 8.4 Hz) were assigned to H-5, H-7 and H-4 in the indole skeleton, respectively. The signal at δ 7.09 (1H, *d*, *J* = 2.4 Hz) was contributed by H-2 in the indole skeleton. In the aromatic region of the ¹H-NMR spectrum, the signals at δ 6.85 (2H, *d*, *J* = 8.4 Hz) and 7.42 (2H, *d*, *J* = 8.4Hz) indicated the presence of a 1,4-disubstituted benzene ring, and *trans*-olefinic protons appeared at δ 6.49 (1H, *d*, *J* = 16.0 Hz) and 7.48 (1H, *d*, *J* = 16.0 Hz). Therefore, the combined data for compound **2** indicated the presence of a partial coumaric acid. Two methylene proton signals at δ 2.89 (2H, *t*, *J* = 7.6 Hz) and 3.59 (2H, *dd*, *J* = 13.6, 7.6 Hz) correlated with the carbon signal at δ 26.1 (C-1') and 40.2 (C-2') in the HSQC experiment. The COSY spectrum of **2** showed two methylenes coupled with NH ($\delta_{\rm H}$ 7.30). Thus, compound **2** contains a partial NH-CH₂-CH₂ structure. The HMBC spectrum exhibited correlations between

H-1'/C-2, C-3 and C-9, H-2'/C-1" and C-1', H-2"/C-1", C-3", and C-4" indocating the three aforementioned structures to be connected as shown in Figure 3. In addition, the presence of a carbonyl group in the coumaric acid moiety connected to NH with an amide bond was further confirmed by the fragment at m/z 147 in the FAB-MS spectrum. According to previous reports and the above NMR analysis, compound **2** was deduced to have the structure shown and it was named hygarine.

Figure 3. Significant correlations observed in HMBC of compound 2.



Twenty-six known compounds, including *N-p*-coumarylanthranilic acid (3) [5], (-)-epigallocatechin-3-*O*-gallate (4) [6], (-)-epicatechin-3-*O*-gallate (5) [7], L-tyrosine (6) [8], quercetin (7) [9], rutin (8) [9], kaempferol (9) [10], adenine (10) [11], adenosine (11) [11], inosine (12) [11], uracil (13) [12], uridine (14) [13], coumaric acid (15) [14], methyl coumarate (16) [15], ferulic acid (17) [16], *n*-hexacosyl ferulate (18) [17], vanillic acid (19) [18], *p*-hydroxybenzoic acid (20) [19], 2,6-dimethoxyquinone (21) [20], physcion (22) [21], cycloeucalenol (23) [22], squalane (24) [23], a β -sitosterol and stigmasterol mixture (25) [15], eicosanoic acid 2,3-dihydroxypropyl ester (26) [24], butcosanoic acid 2,3-dihydroxypropyl ester (27) [24] and sucrose (28) [25], were isolated and identified by comprehensive studies of their physical and spectral data.

Figure 4. Effect of $(7S^*, 8R^*, 7^*R^*, 8^*S^*)$ -icariol A₂-9-*O*- β -xylopyranoside (1) and *N*-*p*-coumarylanthranilic acid (3) isolated from *H. aristata* on generation of superoxide anion (left) and elastase release (right) in fMLP/CB-activated human neutrophils.



All data are expressed as mean \pm S.E.M. (n = 3). *P < 0.05, ***P < 0.001 compared with the control value.

Among all these compounds, several were evaluated for their antioxidant and anti-inflammatory activities. The inhibitory activity of $(7S^*, 8R^*, 7^*R^*, 8^*S^*)$ -icariol A₂-9-*O*- β -xylopyranoside (1) and *N*-*p*-coumarylanthranilic acid (3) against the inflammatory response in human neutrophils was

investigated. Compound 1 showed moderate anti-inflammatory activities and further, it inhibited superoxide anion generation and elastase release by fMLP/CB-induced human neutrophils, with IC₅₀ values of 19.33 ± 0.86 and $24.14 \pm 1.59 \mu$ M, respectively (Figure 4). In addition, the *n*-butanol soluble layer of the *H. aristata* methanolic extract was found to be a powerful scavenger of DPPH free radicals (Figure 5). In further bioactivity-guided investigation of the butanol layer, (-)-epigallocatechin-3-*O*-gallate (4) (2.92 g, 0.31% of plant material), and (-)-epicatechin-3-*O*-gallate (5) (0.55 g, 0.06%) were obtained; these compounds were the major components of *H. aristata*. Their radical scavenging activities were greater by approximately 3.8 and 3.1-fold, respectively, than that of α -tocopherol (Figure 6). These results suggest that *H. aristata* could be a rich natural source of catechins due to its high content of these substances.

Figure 5. Effects of different crude extracts in scavenging DPPH radicals.



Reduction of DPPH was measured spectophotometrically at 517 nm. All data are expressed as the mean \pm S.E.M. of three separate experiments. *P < 0.05, **P < 0.01, ***P < 0.001 compared with the control value. HAM: *H. aristata* methanolic crude extract, HAC: *H. aristata* CHCl₃ layer, HAB: *H. aristata* n-BuOH layer, HAW: *H. aristata* water layer, Vit E: α -tocopherol.

Figure 6. DPPH radical scavenging effects of compounds 1, 4 and 5 isolated from the *n*-butanol layer.



Reduction of DPPH was measured spectophotometrically at 517 nm. All data are expressed as the mean \pm S.E.M. of three separate experiments. ***P < 0.001 compared with the control value.

3. Experimental

3.1. General

Melting points were measured on a Yanagimoto MP-S3 micro melting point apparatus and are uncorrected. The UV spectra were recorded on a Hitachi U-3010 spectrophotometer in MeOH solution. The IR spectra were recorded on a Jasco IR Report-100 spectrophotometer as KBr discs. The ¹H- and ¹³C-NMR spectra were recorded on Bruker Avance-400 spectrometer. Chemical shifts are shown in δ values with tetramethylsilane as internal reference. The mass spectra were performed in the EI or FAB (matrix: glycerol) mode on a VG 70-250 S spectrometer. Specific rotations were determined on a Jasco P-1010 polarimeter.

3.2. Plant Material

Hygroryza aristata was collected and authenticated by Prof. C. S. Kouh at I-Lang, Taiwan. A voucher specimen (CGU-HA-1) was deposited in the herbarium of Chang Gung University, Taoyuan, Taiwan.

3.3. Extraction and Isolation

Dry *H. aristata* (948 g) was extracted with MeOH (6 L × 6) under reflux for 8 hours and concentrated to give a brown syrup (122.0 g). The syrup was suspended in H₂O and partitioned successively with CHCl₃ and *n*-BuOH. The CHCl₃ extract (17.0 g) was subjected to column chromatography over silica gel and eluted with a CHCl₃ and MeOH step gradients to afford eight fractions. Repeated column chromatography of the first fraction, over silica gel with *n*-hexane and EtOAc mixtures yielded squalane (24, 2.3 mg). The second and third fractions were treated as the first fraction to obtain *n*-hexacosyl ferulate (18, 7.2 mg), *p*-hydroxybenzoic acid (20, 5.0 mg), 2,6-dimethoxyquinone (21, 2.2 mg) and physcion (22, 1.5 mg), separately. The fourth fraction was repeatedly chromatographed over silica gel with CHCl₃ and MeOH (18:1) to afford cycloeucalenol (23, 7.1 mg), *β*-sitosterol & stigmasterol mixture (25, 100.0 mg), eicosanoic acid 2,3-dihydroxypropyl ester (26, 5.2 mg) and butcosanoic acid 2,3-dihydroxypropyl ester (27, 12.1 mg), successively. The fifth fraction was repeatedly column chromatographed over silica gel with CHCl₃-MeOH gradients to give coumaric acid (15, 11.5 mg) and methyl coumarate (16, 5.3 mg). The sixth fraction was purified by recrystallisation to afford vanillic acid (19, 2.5 mg).

The *n*-BuOH layer (18.4 g) was applied on Diaion HP-20 gel and eluted with gradients of H_2O and MeOH to give eight fractions. The second fraction was subjected to column chromatography over silica gel and eluted with a CHCl₃ and MeOH step gradient to afford L-tyrosine (**6**, 10.0 mg), inosine (**12**, 11.0 mg) and uridine (**14**, 20.0 mg), successively. The third fraction was chromatographed on silica gel column and eluted with CHCl₃ and MeOH mixtures to give adenine (**10**, 15.6 mg) and uracil (**13**, 5.6 mg). The fourth fraction was filtered to give adenosine (**11**, 20.3 mg).The fifth fraction was

repeatedly column chromatographed over silica gel with CHCl₃-MeOH gradients to give quercetin (5, 3.5 mg), kaempferol (7, 10.0 mg), (-)-epigallocatechin-3-*O*-gallate (4, 2,919.3 mg) and (-)-epicatechin-3-*O*-gallate (5, 552.0 mg). The seventh fraction was also rechromatographed over silica gel and eluted with EtOAc to give ($7S^*$, $8R^*$, 7^*R^* , 8^*S^*)-icariol A₂-9-*O*- β -xylopyranoside (1, 15.7 mg) and rutin (6, 3.2 mg). The eighth fraction was repeatedly column chromatographed over silica gel with CHCl₃-MeOH gradients to give hygarine (2, 3.0 mg), *N*-*p*-coumarylanthranilic acid (3, 8.6 mg) and ferulic acid (17, 2.8 mg), separately.

The H₂O layer (82.1 g) was applied on Diaion HP-20 gel and eluted with gradients of H₂O and MeOH to give four fractions. The first fraction was chromatographed on Sephadex LH-20 column and eluted with gradients of H₂O and MeOH to afford sucrose (**28**, 20.5 mg). The second fraction was repeatedly column chromatographed over Sephadex LH-20 with H₂O:MeOH gradients to give uridine (**14**, 20.0 mg).

3.4. DPPH Free-Radical Scavenging Activity

The scavenging activity of the DPPH radical was assayed by the modified method of Shimada *et al.* [26]. The absorbance was measured at 517 nm. Lower absorbance of the reaction mixture indicates higher free-radical scavenging activity. The DPPH radicalscavengingactivity (%) was calculated by the following equation:

scavenging activity (%) = $(1 - A_{\text{sample}}/A_{\text{control}}) \times 100\%$

3.5. Preparation of Human Neutrophils

Human neutrophils from venous blood of healthy, adult volunteers (20–30 years old) were isolated with a standard method of Dextran sedimentation prior to centrifugation in Ficoll Hypaque gradient and hypotonic lysis of erythrocytes. Purified neutrophils that contained > 98% viable cells, as determined by trypan blue exclusion, were resuspended in HBSS buffer at pH 7.4, and kept at 4 °C before use.

3.6. Measurement of Superoxide Anion (O_2^{\bullet}) Generation

The measurement of the generation of O_2^{-} was based on the superoxide dismutase (SOD)-inhibitable reduction of ferricytochrome *c* [27]. In brief, after supplementing with ferricytochrome *c* (0.5 mg/mL), neutrophils (6×10^5 /mL) were equilibrated at 37 °C for 2 min and incubated with either control or different concentrations of tested compounds for 5 min. Cells were activated by fMLP (0.1 μ M) or PMA (5 nM) for 10 min. When fMLP was used as stimulant, cytochalasin B (CB, 1 μ g/mL) was incubated for 3 min before peptide activation. The changes in absorbance with the reduction of ferricytochrome *c* at 550 nm were continuously monitored in a double-beam, six-cell positioner spectrophotometer with constant stirring. Calculation is based on the difference of the reactions with and without SOD (100 U/mL) divided by the extinction coefficient for the reduction of ferricytochrome *c* ($\epsilon = 21.1/mM/10$ mm).

(7*S**,8*R**,7'*R**,8'*S**)-*Icariol A*₂-9-*O*-β-xylopyranoside (1). Colorless syrup. Formula: C₂₇H₃₆O₁₃. [α]_D +0.63° (*c* 0.35, MeOH). UV λ_{max} nm (logε) : 237 (2.89), 279 (2.70). IRv_{max} cm⁻¹: 3,375, 1,613, 1,517, 1,113. FAB-MS *m/z* (*rel. int.%*): 591 ([M+Na]⁺, 8), 175 (7), 161 (9), 153 (6), 149 (20), 133 (9). ¹H- and ¹³C-NMR: see Table 1.

Hygarine (2). Brown syrup. UV λ_{max} nm (log ϵ): 224 (sh) (3.52), 288 (3.27), 312 (sh) (3.22). IR ν_{max} cm⁻¹: 3,404, 1,643, 1,048. FAB-MS *m/z* (*rel. int.%*): 323 ([M+1]⁺, 12), 147 (30), 55 (100), 94 (30). HRFAB-MS: Anal. Calcd. for C₁₉H₁₈N₂O₃ 323.3726, found 323.3734. ¹H- and ¹³C-NMR: see Table 2.

4. Conclusions

Two new compounds, $(7S^*, 8R^*, 7'R^*, 8'S^*)$ -icariol A₂-9-*O*- β -xylopyranoside (1) and hygarine (2), together with twenty-six known compounds, were isolated from the extracts of *Hygroryza aristata*. (-)-Epigallocatechin-3-*O*-gallate (4) and (-)-epicatechin-3-*O*-gallate (5) possess free radical scavenging activities. (7*S**, 8*R**, 7'*R**, 8'*S**)-Icariol A₂-9-*O*- β -xylopyranoside (1) could inhibit superoxide anion generation and elastase release by fMLP/CB-induced human neutrophils with IC₅₀ values of 19.33 ± 0.86 and 24.14 ± 1.59 µM, respectively.

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References

- 1. Hsu, C.C.; Kuoh, C.S.; Lin, W.C.; Liu, H.Y. *Flora of Taiwan*, 2nd ed.; Epoch Publishing Co., Ltd.: Taipei, Taiwan, 2000; Volume 5, pp. 320-321.
- 2. Takara, K.; Matsui, D.; Wada, K.; Ichiba, T.; Chinen, I.; Nakasone, Y. New phenolic compounds from Kokuto, non-centrifuged cane sugar. *Biosci. Biotechnol. Biochem.* **2003**, *67*, 376-379.
- 3. Su, D.G.; Tang, W.Z.; Hu, Y.C.; Liu, Y.B.; Yu, S.S.; Ma, S.G.; Qu, J.; Yu, D.Q. Lignan glycosides from *Neoalsomitra integrifoliola*. J. Nat. Prod. **2008**, 71, 784-788.
- 4. Ghosal, S.; Srivastava, R.S. β-Phenethylamine, tetrahydroisoquinoline and indole alkaloids of *Desmodium tiliaefolium. Phytochemistry* **1973**, *12*, 193-197.
- Ponchet, M.; Jean, F.B.; Hauteville, M.; Riccl, P. Dianthramides (*N*-benzoyl and *N*-paracoumarylanthranilic acid derivativies) from elicited tissues of *Dianthus caryophyllus*. *Phytochemistry* 1988, 27, 2725-2730.
- 6. Goetz, G.; Fkyerat, A.; Metais, N. Resistance factors to grey mould in grape berries: Identification of some phenolics inhibitors of *Botrytis cinerea* stilbene oxidase. *Phytochemistry* **1999**, *52*, 759-767.
- 7. Fan, W.; Tezuka, Y.; Kadota, S. Proly endopeptidase inhibitory activity of fourteen kampo formulas and inhibitory constituents of Tokaku-joki-to. *Chem. Pharm. Bull.* **2000**, *48*, 1055-1061.

- Houck, D.R.; Hanners, J.L.; Unkefer, C.J. Biosynthesis of pyrroloquinoline quinine.1. Identification of biosynthetic precursors using ¹³C labeling and NMR spectroscopy. *J. Am. Chem. Soc.* 1988, *110*, 6920-6921.
- 9. Liu, J.X.; Di, D.L.; Shi, Y.P. Diversity of chemical constituents from *Saxifraga montana* H. J. *Chin. Chem. Soc.* **2008**, *55*, 863-870.
- Slimestad, R.; Andersen, M.; Francis, G.W.; Marston, A.; Hostettmann, K. Syringetin 3-O-(6"-acetyl)-β-glucopyranoside and other flavonols from needles of Norway spruce, *Picea abies*. *Phytochemistry* **1995**, *40*, 1537-1542.
- 11. Wu, T.S.; Leu, Y.L.; Chan, Y.Y. Aristolochic acids as a defensive substance for the Aristolochiaceous plant-feeding swallowtail butterfly, *Pachliopta aristolochiae interpositus*. J. Chin. Chem. Soc. 2000, 47, 221-226.
- 12. Kitajima, J.; Tanaka, T.; Ida, Y. Water-soluble constituents of fennel. IX. glucosides and nucleosides. *Chem. Pharm. Bull.* **1999**, *47*, 988-992.
- Sako, M.; Kawada, H. A new and efficient synthetic method for ¹⁵N₃-labeled cytosine nucleoside: dimroth rearrangement of cytidine N₃-oxides. *J. Org. Chem.* **2004**, *69*, 8148-8150.
- 14. Lin, Y.L.; Wang, W.Y.; Kuo, Y.H.; Chen, C.F. Nonsteroidal constituents from *Solanum incanum* L. *J. Chin. Chem. Soc.* **2000**, *47*, 247-251.
- 15. Leu, Y.L.; Chan, Y.Y.; Hsu, M.Y.; Chen, I.S.; Wu, T.S. The constituents of the stem and roots of *Aristolochia foveolata. J. Chin. Chem. Soc.* **1998**, *45*, 539-541.
- 16. Kosuge, K.; Mitsunaga, K.; Koike, K.; Ohmoto, T. Studies on the constituents of *Ailanthus integrifolia*. *Chem. Pharm. Bull.* **1994**, *29*, 1669-1671.
- 17. Shen, C.C.; Syu, W.J.; Li, S.Y.; Lin, C.H.; Lee, G.H.; Sun, C.M. Antimicrobial activities of naphthazarins from *Arnebia euchroma. J. Nat. Prod.* **2002**, *65*, 1857-1862.
- 18. Chen, C.Y.; Chang, F.R.; Teng, C.M.; Wu, Y.C. Cheritamine, a new *N*-fatty acyl tryptamine and other constituents from the stems of *Annona cherimola*. J. Chin. Chem. Soc. **1999**, *46*, 77-86.
- 19. Wu, T.S.; Yeh, J.H; Wu, P.L. The heartwood constituents of *Tetradium glabrifolium*. *Phytochemistry* **1995**, *40*, 121-124.
- Lin, W.Y.; Kuo, Y.H.; Chang, Y.L.; Teng, C.M.; Wang, E.C.; Tsutomu, I.; Chen, I.S. Anti-platelet aggregation and chemical constituents from the rhizome of *Gynura japonica*. *Planta Med.* 2003, 69, 757-764.
- 21. Wang, E.C.; Shih, M.H.; Liu, M.C.; Chen, M.T.; Lee, G.H. Studies on constituents of *Saururus chinensis*. *Heterocycles* **1996**, *43*, 969-975.
- 22. Lee, C.K.; Chang, M.H. The chemical constituents from the heartwood of *Eucalyptus citriodora*. *J. Chin. Chem. Soc.* **2000**, *47*, 555-560.
- 23. Wu, T.S.; Chang, F.C.; Wu, P.L.; Kaoh, C.S.; Chen, I.S. Constituents of *Tetradium glabrifolium*. J. Chin. Chem. Soc. **1995**, 42, 929-934.
- 24. Susumu, K.; Toshiro, I.; Ken, Y.; Sakae, Y.; Michio, T. (+)-α-Viniferin an antiflammatory compound from *Caragana chamlagu* root. *Chem. Pharm. Bull.* **1990**, *38*, 432-435.
- 25. Wolters, A.M.; Jayawickrama, D.A.; Larive, C.K.; Sweedler, J.V. Capillary isotachophoresis/NMR: Extension to trace impurity analysis and improved instrumental coupling. *Anal. Chem.* **2002**, *74*, 2306-2313.

- 26. Shimada, K.; Fujikawa, K.; Yahara, K.; Nakamura, T. Antioxidative properties of xanthan on the autoxidation of soybean oil in cyclodextrin emulsion. *J. Agric. Food Chem.* **1992**, *40*, 945-948.
- 27. Hwang, T.L.; Su, Y.C.; Chang, H.L.; Leu, Y.L.; Chung, P.J.; Kuo, L.M.; Chang, Y.J. Suppression of superoxide anion and elastase release by C₁₈ unsaturated fatty acids in human neutrophils. *J. Lipid Res.* **2009**, *50*, 1395-1408.

Sample Availability: Samples of the compounds 1, 3-28 are available from the authors.

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