RESEARCH ARTICLE



Exotic properties of a voltage-gated proton channel from the snail *Helisoma trivolvis*

Sarah Thomas¹, Vladimir V. Cherny², Deri Morgan², Liana R. Artinian³, Vincent Rehder³, Susan M.E. Smith¹, and Thomas E. DeCoursey² D

Voltage-gated proton channels, H_V 1, were first reported in *Helix aspersa* snail neurons. These H⁺ channels open very rapidly, two to three orders of magnitude faster than mammalian H_v1. Here we identify an H_v1 gene in the snail *Helisoma* trivolvis and verify protein level expression by Western blotting of H. trivolvis brain lysate. Expressed in mammalian cells, HtH_v1 currents in most respects resemble those described in other snails, including rapid activation, 476 times faster than hHv1 (human) at pHo 7, between 50 and 90 mV. In contrast to most Hv1, activation of HtHv1 is exponential, suggesting first-order kinetics. However, the large gating charge of ~5.5 e_0 suggests that HtH_v1 functions as a dimer, evidently with highly cooperative gating. HtHv1 opening is exquisitely sensitive to pHo, whereas closing is nearly independent of pHo. Zn²⁺ and Cd²⁺ inhibit HtH_v1 currents in the micromolar range, slowing activation, shifting the proton conductance-voltage $(g_{H^-}V)$ relationship to more positive potentials, and lowering the maximum conductance. This is consistent with HtH_v1 possessing three of the four amino acids that coordinate Zn²⁺ in mammalian H_v1. All known H_v1 exhibit ΔpH-dependent gating that results in a 40-mV shift of the g_{H} -V relationship for a unit change in either pH₀ or pH₁. This property is crucial for all the functions of H_V1 in many species and numerous human cells. The HtH_V1 channel exhibits normal or supernormal pH_o dependence, but weak pH_i dependence. Under favorable conditions, this might result in the HtH_v1 channel conducting inward currents and perhaps mediating a proton action potential. The anomalous ΔpH -dependent gating of HtH_v1 channels suggests a structural basis for this important property, which is further explored in this issue (Cherny et al. 2018. J. Gen. Physiol. https://doi.org/10.1085/jgp.201711968).

Introduction

Voltage-gated proton channels, H_v1, remain relative newcomers to the ion channel family. Although the idea of a depolarization-activated proton-selective ion channel was proposed in 1972 by J. Woodland Hastings and colleagues (Fogel and Hastings, 1972), the first voltage-clamp study that established the existence of this channel type occurred a decade later in the snail Helix aspersa (Thomas and Meech, 1982). An H_v1 gene was not identified until 2006 (Ramsey et al., 2006; Sasaki et al., 2006). Strong interest in this channel has arisen for two main reasons. First, its structure, with just four transmembrane helices, closely resembles the voltage-sensing domain of other voltage-gated ion channels, making it a unique model for voltage-gating mechanisms. By combining voltage sensing, gating, and conduction into a single module, H_v1 uniquely provides a direct readout of its gating state. Second, exceedingly diverse functions have been identified for H_v1 in many species and in many human tissues (DeCoursey, 2013).

The first systematic voltage-clamp characterization of voltage-gated proton currents was in *Lymnaea stagnalis* snail

neurons (Byerly et al., 1984). When mammalian proton currents were identified a decade later (DeCoursey, 1991), the most obvious difference was that H_v1 in snails activated two to three orders of magnitude faster. Here, we investigate the properties of the Helisoma trivolvis snail H_V1 gene product. We searched a transcriptome of *H. trivolvis* and found a putative HtH_v1; we then cloned the gene from a cDNA pool constructed from H. trivolvis brain tissue. We find many similarities to native proton currents studied in situ in other snail species, including rapid gating kinetics and other significant differences from mammalian Hv1. HtHv1 currents differ from mammalian H_v1 in having exponential (vs. sigmoid) activation, similarity of τ_{act} and τ_{tail} at overlapping voltages, and maximal time constants near the midpoint of the proton conductance-voltage $(g_{\rm H}-V)$ relationship, all features suggestive of simple first-order gating kinetics expected of a monomeric protein. However, the existence of an extensive coiled-coil motif in the C terminus together with steep voltage dependence suggests "cooperative" gating of a dimeric protein. Potent

¹Department of Molecular and Cellular Biology, Kennesaw State University, Kennesaw, GA; ²Department of Physiology & Biophysics, Rush University, Chicago, IL; ³Department of Biology, Georgia State University, Atlanta, GA.

Correspondence to T.E. DeCoursey: tdecours@rush.edu.

© 2018 Thomas et al. This article is distributed under the terms of an Attribution–Noncommercial–Share Alike–No Mirror Sites license for the first six months after the publication date (see http://www.rupress.org/terms/). After six months it is available under a Creative Commons License (Attribution–Noncommercial–Share Alike 4.0 International license, as described at https://creativecommons.org/licenses/by-nc-sa/4.0/).

&JGP

inhibition of HtH_v1 by Zn²⁺ and Cd²⁺ is explained by conservation of three of four members of the Zn²⁺-binding site (Takeshita et al., 2014). The most remarkable property of the HtH_v1 channel is that its sensitivity to pH_i is anomalously weak. The voltage-gating mechanism of all H_v1 identified to date is unique in being nearly equally responsive to pH_o and pH_i, such that a one-unit change in either shifts the $g_{\rm H}$ -V relationship by 40 mV. This "rule of forty" (DeCoursey, 2013) has the biologically crucial effect of ensuring that H_v1 channels open only when there is an outward electrochemical gradient for H⁺. In other words, H_V1 channels open only when doing so will result in acid extrusion from cells. Extensive mutation of hH_v1 has failed to produce any significant violation of the rule of forty (Ramsey et al., 2010; DeCoursey, 2016). In this issue, Cherny et al. identify a single amino acid difference between HtH_V1 and hH_V1 that appears to be responsible for the anomalous ΔpH dependence of the snail channel.

Materials and methods

Snail tissue

H. trivolvis, a pulmonate snail (order: Basommatophora; family: Planorbidae) from an albino stock maintained and continuously bred in aquaria at Georgia State University, was used for experiments. Snails were originally caught in the wild and introduced as an experimental model animal by S.B. Kater (Kater, 1974).

Gene cloning, mutagenesis, antibody synthesis, and Western blotting

Basic Local Alignment Search Tool searches of a transcriptome from *H. trivolvis* (unpublished data) yielded a hit that matched the criteria for an H_V1 sequence (Smith et al., 2011). Brains were dissected from H. trivolvis (Cohan et al., 2003), RNA was extracted from brain tissue using the RNeasy kit (Qiagen), and a cDNA pool was constructed using the SuperScript III kit (Life Technologies) according to the manufacturer's instructions. Primers designed against the transcriptome hit were used to clone the putative HtH_{V1} coding sequence; the sequence was confirmed by commercial sequencing (SourceBio Science). This coding sequence was subcloned into eukaryotic expression vector pCA-IRES-eGFP. Site-directed mutagenesis of HtH_v1 was performed and sequence verified commercially (Genewiz). Antibody was raised in rabbit to a synthetic peptide (RSPSDHGEGFEEPLC) based on the predicted HtH_v1 epitope and affinity purified (Genscript) with a final concentration of 0.904 mg/ml. Total lysate was prepared from *H. trivolvis* brains that had been stored whole in Qiagen RLT buffer at -80°C for 12 mo. Brains were thawed and triturated briefly on ice; the lysate was cleared by centrifugation at 10,000 × g for 5 min. Proteins from *H. trivolvis* brain lysate were separated by SDS-PAGE, Western blotted, and probed with anti-HtH_v1 antibody (diluted 1:10,000 in blocking buffer) either alone or preincubated with 1,000-fold molar excess of synthetic peptide corresponding to the epitope.

Electrophysiology

HEK-293 cells were grown to \sim 80% confluence in 35-mm culture dishes. HEK-293 cells were transfected with 0.4–0.5 µg cDNA using Lipofectamine 2000 (Invitrogen) or polyethylenimine (Sigma).

Plasmids that did not include GFP were cotransfected with GFP. After 24 h at 37°C in 5% CO₂, cells were trypsinized and replated onto glass coverslips at low density for patch-clamp recording. We selected green cells under fluorescence for recording. Because HEK-293 cells often have small endogenous H_v1 currents (Musset et al., 2011), cells that exhibited small currents suspected to be native were exposed to 1 μ M Zn²⁺, which has generally weaker effects on HtH_v1 (20% slowing of τ_{act} , ~5 mV shift of the g_{H} -Vrelationship, and a 24% decrease in $g_{H,max}$ in three to four cells at pH_o 7) than on hH_v1 (more than a twofold slowing of τ_{act} , ~20 mV shift of the g_{H} -Vrelationship; Musset et al., 2010b). Cells determined on this basis to exhibit native currents were excluded from the study.

Micropipettes were pulled using a Flaming Brown automatic pipette puller (Sutter Instruments) from Custom 8520 Patch Glass (equivalent to Corning 7052 glass; Harvard Apparatus), coated with Sylgard 184 (Dow Corning Corp.), and heat polished to a tip resistance range of typically 3–10 M Ω with highly buffered TMA⁺ pipette solutions. Electrical contact with the pipette solution was achieved by a thin sintered Ag-AgCl pellet (In Vivo Metric Systems) attached to a Teflon-encased silver wire, or simply a chlorided silver wire. A reference electrode made from a Ag-AgCl pellet was connected to the bath through an agar bridge made with Ringer's solution. The current signal from the patch clamp (EPC-9 from HEKA Instruments or Axopatch 200B from Axon Instruments) was recorded and analyzed using Pulse and PulseFit software (HEKA Instruments), or P-CLAMP software supplemented by Sigmaplot (SPSS). Seals were formed with Ringer's solution (in mM: 160 NaCl, 4.5 KCl, 2 CaCl₂, 1 MgCl₂, 5 HEPES, pH 7.4) in the bath, and the potential was zeroed after the pipette was in contact with the cell. Current records are displayed without correction for liquid junction potentials.

Whole-cell or excised inside-out patch configurations of the patch-clamp technique were performed. Bath and pipette solutions were used interchangeably. They contained (in mM) 2 MgCl₂, 1 EGTA, 80–100 buffer, and 75–120 TMA⁺ CH₃SO₃⁻, adjusted to bring the osmolality to ~300 mOsm, and were titrated using TMAOH. Buffers with pK_a near the desired pH were used: homo-PIPES for pH 4.5–5.0, Mes for pH 5.5–6.0, Bis-Tris for pH 6.5, *N*,*N*-bis(2-hydroxyethyl)-2-aminoethanesulfonic acid for pH 7.0, HEPES for pH 7.5, Tricine for pH 8.0, and *N*-cy-clohexyl-2-aminoethanesulfonic acid for pH 9.0. Experiments were done at room temperature (~20–25°C). Current records are shown without leak correction.

Reversal potentials (V_{rev}) in most cases were determined from the direction and amplitude of tail current relaxation over a range of voltages after a prepulse that activated the proton conductance, g_H . When the g_H was activated negative to V_{rev} , the latter could be determined directly from families of currents. Currents were fitted with a single exponential to obtain the activation time constant (τ_{act}), and the fitted curve was extrapolated to infinite time to obtain the steady-state current amplitude (I_H), from which the g_H was calculated as $g_H = I_H/(V - V_{rev})$. Thus we assume that the time-dependent component reflects H⁺ current, and time-independent current represents leak. Because of the strong voltage dependence of activation kinetics, we frequently applied longer pulses near threshold voltages and shorter pulses for large depolarizations to resolve kinetics and avoid proton

Thomas et al.

depletion associated with large H⁺ flux. The voltage at which $g_{\rm H}$ was 10% of $g_{\rm H,max}$ ($V_{\rm gH,max/10}$) was determined after defining $g_{\rm H,max}$ as the largest $g_{\rm H}$ measured.

Results

HtH_v1 is a voltage-gated proton-selective channel

The gene coding for a putative voltage-gated proton channel was identified based on criteria established previously, namely the presence of four transmembrane helices homologous to S1-S4 of voltage sensor domains with an Asp in the middle of the S1 transmembrane helix and the RxWRxxR motif in S4 (Musset et al., 2011; Smith et al., 2011). We cloned the putative HtH_v1 gene from a cDNA pool of brain tissue, verifying that this gene is expressed at the RNA level. Protein level expression was verified by Western blotting of *H. trivolvis* brain lysate probed with a commercially raised antibody to a synthetic peptide based on a HtH_v1 epitope (Fig. 1, inset). The single protein detected ran at ~50 kD, somewhat larger than the predicted size of 40 kD. Glycosylation at the five putative N-glycosylation sites in the S1-S2 linker could account for this discrepancy, given that N-linked oligosaccharides range from 1,884 to 2,851 D (Imperiali and O'Connor, 1999). Excess synthetic peptide abolished the binding of antibody to brain lysate, establishing the specificity of the antibody. The antibody did not significantly bind to two proteins that do not contain the epitope: human glutathione S-transferase and luciferin binding protein from Lingulodinium polyedrum.

The HtH_V1 channel protein (Fig. 1) is substantially larger than the human hH_V1, with 360 amino acids (hH_V1 has 273). Much of this excess resides in the S1–S2 extracellular linker with 73 residues (vs. eight in hH_V1), which contains five potential N-glycosylation sites (vs. 0 in hH_V1). Focusing on the transmembrane regions, HtH_V1 has charged amino acids nearly identical to those of hH_V1. One exception is at the outer end of the S1 helix, where hH_V1 has basic Lys¹²⁵ but snail HtH_V1 has acidic Glu¹²⁰. There is extensive predicted coiled-coil in the C terminus: 36 residues (positions 289–324) with 90% stringency, and 28 residues (294– 321) with 99% stringency according to MARCOIL (Delorenzi and Speed, 2002). H_V1 in several species have been shown to exist as dimers, largely because of coiled-coil interactions in the C terminus (Koch et al., 2008; Lee et al., 2008; Tombola et al., 2008).

The HtH_V1 gene was transfected into HEK-293 cells. Under voltage clamp, transfected cells displayed depolarization-activated currents. The selectivity of these currents was established by measuring the reversal potential, $V_{\rm rev}$, over a range of pH_o and pH_i values (Fig. 2). The measured values of $V_{\rm rev}$ are close to the Nernst potential for H⁺, $E_{\rm H}$, shown as a dashed line. Clearly, the HtH_V1 channel is highly proton selective over the pH range studied.

HtH_v1 gating is rapid with unusual voltage dependence

A family of currents generated by HtH_V1 at symmetrical pH 6.0 is illustrated in Fig. 3 A. The currents activate rapidly with depolarization, and activation becomes much faster at higher voltages. Although H_V1 currents in all species activate more rapidly at more positive voltages, the τ_{act} of HtH_V1 currents (solid and open red squares in Fig. 3 C) exhibits noticeably steeper voltage

dependence. The maximum slope of the τ_{act} -V relationship in seven cells at pH_o 6 was 13.0 ± 3.4 mV/*e*-fold change in τ_{act} (mean ± SD). In several mammalian H_v1, τ_{act} changes *e*-fold in 40–72 mV (DeCoursey, 2003). Channel closing in HtH_v1 was also steeply voltage dependent (Fig. 3 B and blue diamonds in Fig. 3 C), with τ_{tail} changing *e*-fold in 14.2 ± 1.9 mV in six cells. In mammalian cells, the slope is typically much flatter, 26–44 mV/*e*-fold change in τ_{tail} (DeCoursey, 2003).

A remarkable feature of HtH_Vl is that at intermediate voltages where the measurements overlap, the time constants of H⁺ current turn-on (τ_{act}) and deactivation (τ_{tail}) essentially superimpose (Fig. 3 C). This behavior is suggestive of simple first-order kinetics, such as a two-state system:

in which α is the rate of channel opening and β is the rate of channel closing, and the time constant τ is $(\alpha + \beta)^{-1}$ (Hodgkin and Huxley, 1952). Another feature suggestive of first-order kinetics is evident in the $g_{\rm H}$ -V relationship from this cell (Fig. 3 D). The voltage at which the $g_{\rm H}$ is half-maximal is ~40 mV, where the time constants are maximal (Fig. 3 C). However, the limiting slope of the $g_{\rm H}$ -V relationship in Fig. 3 D, i.e., the slope of the most negative values obtained, indicates a gating charge of ~6 e_0 . The mean gating charge in 18 limiting slope measurements was 5.5 ± 0.9 e_0 (mean ± SD). Because the range of $g_{\rm H}$ values resolved did not exceed three orders of magnitude, these gating charge estimates should be considered lower limits. In most species, cooperative gating of the dimeric H_V1 channel doubles the gating charge from 2–3 to 4–6 e_0 (Gonzalez et al., 2010, 2013; Fujiwara et al., 2012).

Mean gating kinetics determined at symmetrical pH 7.0 is shown in Fig. 4 A. As was also seen at pH 6.0 (Fig. 3 C), at voltages where τ_{act} and τ_{tail} overlap, they have similar values, suggestive of first-order gating kinetics. In the first description of proton currents in snail neurons, the activation time to half-peak current was 25 ms or less at pH_o 7.4 (Byerly et al., 1984). With this in mind, the activation kinetics of HtH_v1 is quite similar to that reported in neurons from *L. stagnalis*. When proton currents were first identified in mammalian species, they were found to be radically slower (DeCoursey, 1991; Bernheim et al., 1993; DeCoursey and Cherny, 1993; Demaurex et al., 1993; Kapus et al., 1993). The activation kinetics of HtH_v1 is two to three orders of magnitude faster than that of hH_v1 (Fig. 4 B), averaging 476 times faster between 50 and 90 mV at pH_o 7.

HtH_v1 is sensitive to inhibition by external Zn²⁺ and Cd²⁺

The polyvalent metal cations Zn^{2+} and Cd^{2+} were among the first H_Vl inhibitors identified (Thomas and Meech, 1982; Mahaut-Smith, 1989b). Zn^{2+} in particular has been used widely on H_Vl identified in new species and remains the most potent inhibitor (Cherny and DeCoursey, 1999). Fig. 5 illustrates the effects of 100 μ M Zn^{2+} or Cd^{2+} on HtH_Vl currents. Three main effects are evident: the current amplitude is reduced, the current activates more slowly (scaled currents in Fig. 5 D), and the $g_{H^-}V$ relationship is shifted positively along the voltage axis. These three parameters are interrelated in that a positive shift of the



Figure 1. **The HtH_v1 proton channel sequence.** Topology of the HtH_v1 channel, with transmembrane regions defined by alignment with those determined for hH_v1 by electron paramagnetic resonance (Li et al., 2015). Amino acids in transmembrane regions are color coded as follows: red, acids; blue, bases; brown, amines; purple, aromatics; green, hydroxyls; and orange stars, putative (extracellular) glycosylation sites. Noteworthy are D107 in S1, which presumably confers H⁺ selectivity, and the RxWRxxR motif in S4, both of which are conserved universally in all known H_v1. Sequence in black is the epitope used to generate antibody. Inset shows Western blots of *H. trivolvis* brain, confirming the presence of HtH_v1 protein. Drawn with TOPO2 (http://www.sacs.ucsf.edu/TOPO2/).

 $g_{\rm H}$ -V relationship will in itself decrease the current and slow $\tau_{\rm act}$ at any given voltage. The mean changes in these three parameters produced by 10 or 100 μ M of the two metals are summarized in Fig. 5 E.

These three effects of polyvalent metal cations have been observed for H_Vl from many species. As in rat H_Vl (Cherny and DeCoursey, 1999), Zn^{2+} is more potent than Cd^{2+} in HtH_Vl . Focusing on the three main effects, HtH_Vl was more sensitive, similar to, or less sensitive than human H_Vl (hH_Vl). The reduction of $g_{H,max}$ is glaringly obvious for HtH_Vl , whereas in mammalian H_Vl this effect is small and difficult to detect because of the interrelatedness of the three effects (Cherny and DeCoursey, 1999).

Zn²⁺ at 10 μ M slows τ_{act} by four- to fivefold in both hH_V1 (Musset et al., 2010b) and HtH_V1 (Fig. 5). In contrast, the shift of the g_{H} -V relationship by Zn²⁺ is far more profound in human hH_V1, with a 20-mV shift produced by 1 μ M Zn²⁺ (Musset et al., 2010b) compared with a 12-mV shift by 10 μ M Zn²⁺ in HtH_V1 (Fig. 5).

Unique ΔpH dependence of HtH_v1 gating

Families of proton currents generated by the *H. trivolvis* proton channel gene product, HtH_{vl} , in a cell studied at four pH_o values with pH_i 6 are illustrated in Fig. 6 (A–D). The currents activate with depolarization, and activation becomes much faster at higher voltages. Both voltage dependence and kinetics were

\$JGP



Figure 2. **The HtH_v1 proton channel is highly proton selective.** Reversal potentials (V_{rev}) were measured over a range of pH_i (5.0–8.0) and pH_o (4.8–8.0) values, as described in Materials and methods, in eight cells and one inside-out patch. Data from each cell or patch are connected by lines. The dashed gray line shows identity between V_{rev} and the Nernst potential for H⁺, E_{H} , i.e., the expectation for perfect selectivity.

exquisitely sensitive to pH_o . At higher pH_o , the proton conductance, g_H , turned on at more negative voltages and turned on much more rapidly (note the different time bases). Fig. 6 (E-H) shows deactivation kinetics at each pH_o during tail current measurements in this cell. Channel closing becomes much more rapid at more negative voltages.

In Fig. 6 I, time constants of H⁺ current turn-on (activation, τ_{act}) and turn-off (deactivation, τ_{tail}) from the same cell are plotted. Several intriguing features emerge. Unlike H_vl in other species, at intermediate voltages where τ_{act} and τ_{tail} overlapped, they were of similar magnitude (as was seen in Figs. 3 and 4). Because this behavior suggests first-order kinetics (Scheme 1), the data in Fig. 6 I were analyzed in this way, and the expressions for each rate constant are given on the graph in the following form:

$$\alpha(V) = \alpha e^{V/k_{\alpha}} \tag{1}$$

and

$$\beta(V) = \beta e^{-V/k_{\beta}}.$$
 (2)

The voltage dependence of τ_{act} is steep (small k_{α}) and appears to become steeper at lower pH_o. The voltage dependence of channel closing, τ_{tail} , is also steep (small k_{β}) but appears to be independent of pH_o. To a first approximation, β is independent of pH_o, whereas α is markedly influenced by pH_o. Mean values for the rate constants are plotted in Fig. 7. Confirming the impression from Fig. 6 I (where k_{α} was 11.4, 18, 26, and 30 mV at pH_o 5, 6, 7, and 8), k_{α} increased with pH_o, and k_{β} was pH independent. But by far the strongest effect of pH is that increasing pH_o massively increases the opening rate constant α , which increased more than an order of magnitude per unit increase in pH_o. Stated differently, protonation at the external face of the HtH_V1 channel strongly inhibits channel opening. More subtly, it is evident that for any given pH_o, α is higher and β is lower at pH_i 6 than at pH_i 7; hence, lower pH_i both promotes opening and slows closing.

The $g_{\rm H}$ -V relationships from the cell in Fig. 6 (A–I) are plotted in Fig. 6 J. Like other H_V1, HtH_V1 exhibits robust pH_o-dependent shifts with increasing pH_0 shifting the $g_{\rm H}$ -V relationship negatively. The shifts for $pH_0 5 \rightarrow 6$ and $6 \rightarrow 7$ are closer to 50 than 40 mV, indicating that HtH_v1 exceeds the rule of forty for changes in pH_o . To reconstruct $g_{\rm H}$ - V relationships using the simple first-order assumption (Scheme 1 and Eqs. 1 and 2), which predicts that $P_{open} = \alpha/(\alpha + \beta)$, the solid curves in Fig. 6 J were drawn from the rate constant equations in Fig. 6 I scaled by $g_{H,max}$. Their limiting slope at negative voltages is shallower than observed. Squaring the P_{open}-V relationships, as in the classic Hodgkin-Huxley n^2 approach (Hodgkin and Huxley, 1952), produces the steeper dashed curves, which better approximate the data. Without pushing the model too far, we conclude that it is probable that, like several other H_v1 (Gonzalez et al., 2010; Musset et al., 2010b; Tombola et al., 2010; Fujiwara et al., 2012), HtH_v1 functions as a dimer in which both protomers must activate before either one conducts.

The effects of changes in pH_i were explored in inside-out patches, as illustrated in Fig. 8. The resolution was limited somewhat by the typically small current amplitude combined with



Figure 3. Strong voltage dependence of HtHv1 gating kinetics. (A) A family of proton currents at pH_{o} 6 and pH_{i} 6 in a HEK-293 cell transfected with HtH_V1 . Pulses were applied in 10-mV increments up to 100 mV from a holding potential, $V_{\text{hold}} = -40 \text{ mV}$. (B) Tail currents in the same cell elicited by a prepulse to 80 mV, in 10-mV increments from -60 to 40 mV. (C) Gating kinetics in the same cell. The time constant of channel opening, τ_{act} (red squares) was obtained from single exponential fits to rising currents. Tail current (deactivation, channel closing) time constants, τ_{tail} were also from single exponential fits. Open squares and dashed lines show τ_{act} from a second family. (D) The proton conductance was calculated from the extrapolated single exponential fits of currents and the measured V_{rev} .







Figure 4. Activation and deactivation kinetics of HtH_v1 at symmetrical pH 7.0 (pH_o 7, pH_i 7) overlap. (A) The time constant of channel opening, τ_{actv} was obtained from single exponential fits to currents during activation. Tail current (deactivation, channel closing) time constants, τ_{tail} , were also from single exponential fits. Error bars represent mean ± SEM for 7–17 cells for τ_{act} and 3–7 cells for τ_{tail} . (B) Comparison of channel opening kinetics in HtH_v1 and hH_v1. Activation time constants (τ_{act}) in WT HtH_v1 (triangles, from A) and WT hH_v1 (diamonds, from Cherny et al., 2015) at pH_o 7, pH_i 7. Error bars represent mean ± SEM.

rapid activation kinetics at some pH. Activation kinetics could be resolved at low but not at high pH_i. For example, at pH_i 8 (Fig. 8 C), inward current is clearly activated, but the kinetics is ambiguous. Nevertheless, it is evident in Fig. 8 D that activation kinetics depends only weakly on pH_i, in stark contrast to the strong dependence seen for pH_o (Figs. 6 and 7), and in contrast to mammalian H_v1, in which lowering pH_i speeds activation fivefold/unit (DeCoursey and Cherny, 1995; Villalba-Galea, 2014). Deactivation kinetics was poorly resolved in most patches. The most surprising feature (Fig. 8 E) is that the heretofore universal rule of forty governing Δ pH-dependent gating is violated by HtH_v1. Changing pH_i shifts the $g_{H^-}V$ relationship of HtH_v1 by just 20 mV/unit or less. The aberrant behavior of HtH_v1 provides clues to the mechanism of Δ pH-dependent gating.

Fig. 9 summarizes the ΔpH dependence of HtH_{v1} . For a variety of reasons discussed elsewhere (Cherny et al., 2015), we





Figure 5. **HtHv1 currents are moderately sensitive to inhibition by divalent metal cations.** (**A**) A family of HtHv1 currents at pH_o 7, pH_i 7, in 10-mV increments up to 80 mV. (**B and C**) Families of currents in the same cell at the same voltages in the presence of 100 μ M Cd²⁺ or 100 μ M Zn²⁺, respectively. Calibrations apply to A-C. (**D**) The currents at 80 mV from the three families in A-C are superimposed after scaling to the same maximum current and time base. (**E**) Three main effects of Zn²⁺ and Cd²⁺ are summarized: *I*_H (relative to 1) is the fractional current remaining at 60 mV in the presence of metal; τ slowing (also relative to 1) is the slowing of activation time constants (the ratio τ_{act} [metal/control] at 80 mV); and *g*_H-*V* shift is the shift of *V*(*g*_{H,max}/10). Mean ± SEM is plotted for four cells at 100 μ M metal and five cells at 100 μ M, all studied at pH_o 7. In some cells, a second kinetic component appeared at high metal concentrations. In these cases, we used the faster of the two time constants for this comparison. *, Zn²⁺ effect is significantly greater than Cd²⁺ effect (P < 0.05 by Student's *t* test).

have adopted $V(g_{\rm H,max}/10)$, the voltage at which the $g_{\rm H}$ is 10% of its maximal value, as a parameter to define the position of the $g_{\rm H}$ -V relationship. We find this preferable to other parameters that have been used for this purpose, such as the midpoint of a Boltzmann curve (which frequently does not fit the data well or is ill determined) or the threshold voltage at which current is first detectable (which is arbitrary, depends on the signal-tonoise ratio, and is particularly difficult to resolve when it occurs near $V_{\rm rev}$, as frequently occurs in $\rm HtH_{v1}$). It is evident in Fig. 9 that when $\rm pH_o < 7$, changes in $\rm pH_o$ shift $V(g_{\rm H,max}/10)$ by more than 40 mV/unit (for reference, this slope is shown as a dashed green line in Fig. 9). H_VI in two other species (coccolithophore EhH_VI and insect NpH_VI) also exhibit shifts with pH_o greater than 40 mV/unit (Cherny et al., 2015; Chaves et al., 2016). At pH_o higher than 7, the shift decreases, which may reflect saturation of the response caused by the ambient pH approaching the p K_a of a critical titratable group. Saturation of Δ pH dependence has been observed previously in hH_vI at pH > 8 (Cherny et al., 2015).

The most striking result in Fig. 9 is the data for changes in pH_i (dark red diamonds), which reveal that the position of the







Thomas et al. A snail proton channel with anomalous properties

\$JGP



Figure 7. Rate constants extracted from fits of time-constant data at various pH values (pH_o/pH_i). Time constants were fitted to Eqs. 1 and 2, as illustrated in Fig. 6 I. Mean ± SEM plotted for *n* = 1–4.

 $g_{\rm H^-}V$ relationship depends only weakly on pH_i. There is no clear indication of saturation, although the slope appears to increase at larger ΔpH (i.e., lower pH_i). This is qualitatively like the whole-cell pH_o response, which is steepest at low pH_o and saturates at high pH_o. Over the entire ΔpH range, the mean slope is only 15.3 mV/unit change in pH_i. HtH_v1 is the first H_v1 in which such weak ΔpH dependence has been identified.

Discussion

The rapid kinetics of HtH_v1 resembles that of other snail proton channels but differs from mammalian H_v1

The snail H_V1 , HtH_V1 , exhibits all of the major features of H_V1 in all species studied thus far. It is highly proton selective and it is voltage gated, opening with depolarization, and opening more rapidly at more positive voltages. Furthermore, its voltage dependence is strongly modulated by pH, such that increasing pH_o or decreasing pH_i shifts the $g_{H^-}V$ relationship negatively, in what has been called ΔpH -dependent gating (Cherny et al., 1995). Beyond these qualitative similarities, however, HtH_V1 differs markedly from H_V1 in humans and other mammalian species. The main differences include very rapid activation kinetics, steeply voltage-dependent activation kinetics, activation in a more negative voltage range, exponential rather than sigmoid activation, and distinctly aberrant ΔpH dependence. These properties are discussed below.

The first voltage-gated proton channels to be characterized by voltage clamp were in neurons from the snails *L. stagnalis* (Byerly et al., 1984), *H. aspersa* (Thomas and Meech, 1982; Mahaut-Smith, 1989b), and *Helix pomatia* (Doroshenko et al., 1986). All activated

rapidly, with time constants, τ_{act} , of a few milliseconds. When mammalian proton currents were identified, the most obvious difference was much slower activation, with τ_{act} in the range of seconds (DeCoursey, 1991; Bernheim et al., 1993; Demaurex et al., 1993; Kapus et al., 1993) or even minutes (DeCoursey and Cherny, 1993). A more subtle difference was that mammalian $H_{v}l$ activate with a distinct delay, whereas snail $H_{v}l$ activate exponentially. We show here that the HtHvl channel shares both properties with other snail H_vl . Byerly et al. (1984) reported half-times for activation of less than 25 ms for proton currents in *L. stagnalis* neurons at pH_o 7.4, as observed here at pH_o 7 (Fig. 4 A).

Paradoxically, some aspects of gating in HtH_v1 suggest a simple first-order transition between closed and open states

Activation and deactivation time constants in HtHvl are of similar magnitude at voltages where they overlap. This property is typical of a simple first-order system (Scheme 1). In mammalian H_v1 (DeCoursey, 1991; Cherny et al., 1995, 2001; DeCoursey and Cherny, 1996, 1997; Cherny and DeCoursey, 1999; Schilling et al., 2002), activation tends to be slower than deactivation. This asymmetrical behavior is typical of cooperatively gated multimeric channels (Hodgkin and Huxley, 1952; Hille, 2001), because all subunits must activate before the channel conducts, whereas only one subunit needs to deactivate to close the pore. In rat H_v1, deactivation was rapid, pH_o independent, and weakly voltage dependent at large negative voltages (Cherny et al., 1995). However, near the threshold voltage for $g_{\rm H}$ activation, a second slower component of τ_{tail} appeared that was pH_o dependent and of comparable magnitude to τ_{act} . Also suggestive of a first-order system in HtH_VI is that τ_{act} and τ_{tail} were slowest at the midpoint of the











Figure 9. **Anomalous** ΔpH dependence of gating in HtH_v1. The position of the g_{H} -V relationship was defined in terms of $V(g_{H,max}/10)$, the voltage at which the g_H was 10% of $g_{H,max}$, the maximal value measured. In whole-cell measurements, pH_o was varied, with mean ± SEM data for pH_i 6 or 7 plotted separately. The slope of these data is well above 40 mV/unit (indicated as a dashed green line) and appears to begin to saturate above pH_o 7 ($\Delta pH > 0$ for pH_i 7 and $\Delta pH > 1$ for pH_i 6). In contrast, when pH_i was varied using inside-out patches, all with pH_o 7, there was very little shift of the g_{H} -V relationship. The linear regression slope of the pH_o 7 data is 15.3 mV/unit change in pH_i. Numbers of cells for increasing ΔpH at pH_i 6 = 3, 7, 5, 4; pH_i 7 = 3, 8, 11, 5, 3; and pH_o 7 = 3, 5, 4, 5, 6.

 $g_{\rm H}$ -V relationship (Figs. 3 and 6). Finally, activation kinetics was well described by a single exponential and could not be fitted reasonably with a higher-order function.

There is general agreement that the H_{V1} dimer in several species gates "cooperatively," but it is less clear what this word means; in drug binding, cooperativity can be produced by quite different mechanisms (Colquhoun, 1973). One sense is that, like the Hodgkin-Huxley model, multiple subunits must move before the channel can conduct. Another sense is that, like oxygen binding to the four hemes of hemoglobin, the movement of one H_v1 protomer promotes the movement of the other. The sigmoid activation kinetics of H_v1 in several species (Gonzalez et al., 2010; Musset et al., 2010b; Tombola et al., 2010; Fujiwara et al., 2012) appears to reflect that both protomers must undergo a conformational change before either can conduct (first sense; Gonzalez et al., 2010) or highly cooperative gating (second sense; Tombola et al., 2010). When H_v1 is forced to exist as a monomer, by splicing it with the N terminus of Ciona intestinalis voltage-sensing phosphatase or by truncating the C terminus, the current turns on exponentially and five to seven times faster than with the WT dimeric protein (Koch et al., 2008; Musset et al., 2010a,b; Tombola et al., 2010; Fujiwara et al., 2012). The dimerization of H_v1 in several species appears strongly dependent on coiled-coil interactions in the C terminus (Koch et al., 2008; Lee et al., 2008; Tombola et al., 2008; Li et al., 2010). HtH_v1 has extensive predicted coiled-coil in its C-terminal region. This complicates the interpretation for HtH_V1, because the exponential activation and the apparently first-order kinetics suggest monomeric behavior.

One explanation might be that HtH_v1 exists in the membrane as a dimer because of the coiled-coil region, but the coupling between C terminus and S4 segment is dysfunctional, as can be achieved experimentally by introducing a flexible linker between S4 and the C terminus (Fujiwara et al., 2012). However, this appears unlikely to be the case, because the apparent gating charge of HtH_V1 is nearly 6 e_0 (5.5 ± 0.9, mean ± SD in a sample of 18 $g_{\rm H}$ -V curves), based on the limiting slope of the $g_{\rm H}$ -V relationship. Monomeric H_v1 typically exhibit gating charge roughly half that of the dimer, 2–3 versus 4–6 e_0 . When the coupling between the C terminus and the S4 helix was disrupted by a flexible linker, the gating charge was halved (Fujiwara et al., 2012). Given that HtH_v1 has charged amino acids in its transmembrane regions similar to those of other H_v1, we assume that its gating charge has analogous origins. The $g_{\rm H}$ -V relationships in Fig. 6 J also are compatible with Hodgkin-Huxley-type gating. One possibility is that a concerted rate-limiting step in opening occurs late, presumably after the conformational changes in each monomer (Gonzalez et al., 2010; Musset et al., 2010b; Villalba-Galea, 2014). The voltage-dependent movement of monomers may be so rapid in HtH_v1 that the concerted opening step becomes rate limiting. Another speculative explanation for its exponential activation is that HtH_v1 enjoys tighter coupling between protomers than H_v1 in other species; in essence, both S4 helices move together.

The gating of $HtH_{V}1$ depends steeply on voltage

Perceptibly different from mammalian $H_V l$, the activation kinetics of snail $H_V l$, $HtH_V l$, is more steeply voltage dependent, giving



a family of currents a distinctive gestalt (Fig. 3 A). In HtH_V1, τ_{act} decreased *e*-fold in 13.8 mV, in contrast to several mammalian H_V1, where τ_{act} changes *e*-fold in 40–72 mV (DeCoursey, 2003). In addition, τ_{tail} increased *e*-fold in 14.0 mV in HtH_V1, compared with a slope typically 26–44 mV/*e*-fold change in τ_{tail} in mammalian cells (DeCoursey, 2003). The steeply voltage-dependent gating kinetics of HtH_V1 is strikingly reminiscent of voltage-gated K⁺ channel behavior (Cahalan et al., 1985).

Gating kinetics in $HtH_{\nu}1$ is strongly dependent on pH_{\circ}

Byerly et al. (1984) noted that activation kinetics in snail LsH_{V1} slowed at lower pH_o more than could be accounted for by the shift of the $g_{\rm H}$ -V relationship. This is clearly true of HtH_V1 as well. The τ_{act} -V relationship shifts positively with lower pH_o (Fig. 6 I), but its maximum increases by roughly an order of magnitude per unit decrease in pH_o. In stark contrast, in rat H_v1, the τ_{act} -V relationship mainly shifted along the voltage axis with changes in pH_o, with little change in kinetics. However, the τ_{act} -V relationship in rat was strongly affected by pH_i, slowing fivefold per unit increase in pH_i (DeCoursey and Cherny, 1995). In one study of human hH_vl, gating was described by three exponentials with activation generally faster at lower pH_i and deactivation faster at higher pH_i (Villalba-Galea, 2014). Qualitatively similar results were reported in mouse macrophages, but the largest change was a two- to threefold slowing of τ_{act} for a 1.5-unit increase in pH_i or a 2.1-unit decrease in pH_o (Kapus et al., 1993). An insect H_v1, NpH_v1, however, exhibited nearly as strong pH_o dependence of kinetics as found here in HtH_v1 (Chaves et al., 2016). In contrast to the strong dependence of activation kinetics on pH_i in rat (DeCoursey and Cherny, 1995), in HtH_v1, τ_{act} was in a similar range at all pH_i values from 5 to 8. These differences in gating kinetics among species may make it challenging to produce a single universal model that describes the voltage and pH dependence of gating in all species.

Metal binding site in $HtH_{V}1$

HtH_V1 was moderately sensitive to inhibition by Zn²⁺, the classic (Thomas and Meech, 1982; Mahaut-Smith, 1989a) and still most potent (Cherny and DeCoursey, 1999) H_V1 inhibitor. Somewhat weaker effects were observed for Cd²⁺ (Fig. 5). The principal effects of Zn²⁺ on mammalian H_V1 are a slowing of activation, a positive shift of the $g_{\rm H}$ -V relationship, and possibly a reduction of the maximum H⁺ conductance, $g_{\rm H,max}$ (Cherny and DeCoursey, 1999). These effects are also observed in HtH_V1, but the decrease in $g_{\rm H,max}$ is much more obvious, whereas the shift of the $g_{\rm H}$ -V relationship is substantially weaker in HtH_V1 than in hH_V1. Thus, the $g_{\rm H}$ -V relationship in human hH_V1 is shifted more by 1 μM Zn²⁺ (Musset et al., 2010b) than HtH_V1 is shifted by 10 μM Zn²⁺ (Fig. 5).

In mammalian $H_V l$, Zn^{2+} binds mainly to two His: His¹⁴⁰ and His¹⁹³ in $hH_V l$ (Ramsey et al., 2006; Musset et al., 2010b). Surprisingly, when $mH_V l$ was crystallized, it contained a Zn^{2+} atom, coordinated by the corresponding two His with contributions from two acids, Glu^{115} and Asp^{119} , given in Table 1 (Takeshita et al., 2014). Mutation of both acids simultaneously decreases Zn^{2+} affinity of $mH_V l$, but neutralizing either alone does not (Takeshita et al., 2014). As indicated in Table 1, three of these

four corresponding residues are conserved in HtHv1: Glu¹¹⁴ Glu¹¹⁸ (conservatively replacing Asp), and His²⁰¹, with Val²⁵⁴ replacing the second His. Consistent with the partial conservation of the mammalian Zn²⁺ site, Zn²⁺ was generally less potent in HtH_v1 but still quite effective. The main difference in the presumed Zn²⁺ binding residues in HtH_v1 is the lack of His¹⁹³. One might therefore speculate that His¹⁹³ in human hH_v1 is important in Zn^{2+} shifting the g_{H} -V relationship positively. Evidently, when the binding site includes His¹⁹³ located in the external S2–S3 linker, Zn²⁺ binding biases the membrane potential more effectively. Coordination by four amino acids is more typical of a structural Zn²⁺ binding site, whereas catalytic Zn²⁺ binding sites usually have three amino acids and one water as a ligand (Auld, 2001). Of interest is a study showing that the metal transport site of ZnT transporters is selective for Zn²⁺ over Cd²⁺ when the four ligands are 2 His + 2 acids, but cannot discriminate the two metals with 1 His + 3 acids (Hoch et al., 2012). The HtH_{v1} channel has 1 His + 2 acids and is moderately selective for Zn²⁺ over Cd²⁺. As shown in Table 1, the NpH_v1, CiH_v1, CpH_v1, SpH_v1, and DrH_v1 channels share a 1 His + 3 acids scheme and are much less sensitive to Zn^{2+} than mammalian H_V1 (Cd²⁺ was not tested) and generally less sensitive than HtH_v1. Intriguingly, the D145H mutation in NpH_v1 results in 2 His + 2 acids, which markedly increases its Zn²⁺ sensitivity (Chaves et al., 2018). Empirically, Table 1 indicates that the configurations of H_v1 for Zn²⁺ binding to H_v1 in order of decreasing efficacy are: 2 His + 2 acids > 1 His + 2 acids > 1 His + 3 acids. It appears that the 1 His + 3 acids motif is somewhat less favorable for Zn²⁺ binding than 1 His + 2 acids as found in HtH_v1, which seems paradoxical, because the 1 His + 3 acids motif has four ligands instead of three, possibly plus water. Perhaps geometrical factors can be more important than the number of ligands.

The g_{H} -V relationship of HtH_v1 depends more on pH_o and less on pH_i than H_v1 in other species

A unique property of H_v1 is that its voltage-dependent gating is strongly modulated by pH in a manner called Δ pH dependence (Cherny et al., 1995). The $g_{\rm H}$ -V relationship is shifted equally by increasing pH_0 or decreasing pH_i , by -40 mV/unit change, thus responding to the pH gradient (Δ pH) rather than to the absolute pH (Cherny et al., 1995). The practical consequence is that H_{v1} opens only when the electrochemical gradient for H⁺ is outward, such that when the channel opens it will always extrude acid from the cell (Doroshenko et al., 1986; DeCoursey and Cherny, 1994). To a rough approximation, all H_v 1 appear to shift by 40 mV/unit at all pH values (DeCoursey, 2003). Until recently, the rare exceptions to this rule of forty were ignored as anomalies, perhaps reflecting difficulties of the measurements, in particular with control over pH (DeCoursey and Cherny, 1997). However, measurements explicitly addressing this point revealed that the ΔpH -dependent gating of hH_v1, kH_v1, and EhH_v1 does indeed deviate by saturating at high pH, namely above pH_o 8 or pH_i 8 (Cherny et al., 2015). Byerly et al. (1984) reported little shift between pH_o 7.4 and 8.4 in *L. stagnalis*, and this observation is consistent with the saturation at high pH_o observed here for HtH_{V1} (Fig. 9). The slope in HtH_{V1} begins to decrease above pH_{o} 7 (Fig. 9), suggesting that saturation begins at lower pH_0 than in

Table 1.	Putative Zn ²⁺ -binding residues in H _v 1 f	rom several species
----------	---	---------------------

Species	H _v 1 name	Acid ₁	Acid ₂	His ₁	His₂	Zn ²⁺ potency	Reference
						μM^{a}	
Mus musculus	mH _v 1	Glu ¹¹⁵	Asp ¹¹⁹	His ¹³⁶	His ¹⁸⁹	1	Takeshita et al., 2014
Homo sapiens	hH_V1	Glu ¹¹⁹	Asp ¹²³	His ¹⁴⁰	His ¹⁹³	1	Musset et al., 2010b
Rattus norvegicus	RnH_V1	Glu ¹¹⁵	Asp ¹¹⁹	His ¹³⁸	His ¹⁹¹	1	Cherny and DeCoursey, 1999
Helisoma trivolvis	HtH _v 1	Glu ¹¹⁴	Glu ¹¹⁸	His ²⁰¹	Val ²⁵⁴	10	This study
Nicoletia phytophila	NpH _V 1	Glu ⁷³	Asp ⁷⁷	His ⁹²	Asp ¹⁴⁵	10	Chaves et al., 2018
Ciona intestinalis	CiH _V 1	Glu ¹⁶⁷	Asp ¹⁷¹	His ¹⁸⁸	Glu ²⁴³	10	Qiu et al., 2016
Coccolithus pelagicus	CpHv1	Glu ⁸⁰	Asp ⁸⁴	His ¹⁰⁵	Glu ¹⁵⁸	30	Taylor et al., 2011
Strongylocentrotus purpuratus	SpH _V 1	Glu ⁸⁰	Asp ⁸⁴	His ¹¹⁴	Glu ¹⁶⁷	100	Sakata et al., 2016
Danio rerio	DrH_V1	Glu ⁸⁴	Asp ⁸⁸	His ¹⁰⁵	Asp ¹⁵⁸	100	Ratanayotha et al., 2017
Lingulodinium polyedrum ^b	LpHv1	Ser ⁵²	Glu ⁵⁶	Ala ¹⁶¹	Thr ²¹⁰ - Asn ²¹¹	~100	Rodriguez et al., 2017
Emiliania huxleyi ^c	EhH _V 1	Glu ¹¹⁷	Asp ¹²¹	Thr ²¹⁰	Gly ²⁹⁸	~500	Taylor et al., 2011
Karlodinium veneficum	kH _V 1	Gly ⁵⁸	Glu ⁶²	Glu ¹⁰⁹	Gly ¹⁶²	Weak	DeCoursey, 2012

Acidic amino acids, red; His, aqua; and neutral residues, gray.

^aZn²⁺ potency is defined very approximately as the concentration required to shift the g_H-V relationship by 20 mV or reduce current during a test pulse by 50%. The latter is highly arbitrary and depends strongly on the test voltage used (DeCoursey et al., 2001), but in some cases is the only information available. The four amino acids implicated in Zn²⁺ binding were identified in the crystal structure of mH_v1, which actually included a bound Zn²⁺ atom, confirmed by mutation of each individually (Takeshita et al., 2014). The importance of the two His in hH_v1 have been confirmed in mutation studies (Ramsey et al., 2006; Musset et al., 2010b); all other amino acids in this table are simply located at analogous positions as assessed by multiple alignment. ^bGiven the moderate Zn²⁺ sensitivity of LpH_v1, which lacks three of four coordinating groups, it is quite possible that Zn²⁺ binds at a different location. ^cEhH_v1 has two alternatively located His in the S1–S2 linker that bind Zn²⁺ (Taylor et al., 2011).

hH_v1. Saturation of Δ pH-dependent gating suggests that pH is approaching the effective p K_a of one or more titratable groups that sense pH_o. Given this interpretation, the effective p K_a is roughly 1 unit lower in HtH_v1 than in hH_v1.

Another deviation from the rule of forty is that the $g_{\rm H}$ -V relationship in HtH_V1 shifted ~60 mV/unit change in pH_o between pH_o 5 and 7 (Fig. 9), well above the classic value of 40 mV/unit change in pH_o (Cherny et al., 1995). This unusual property is shared by several disparate species, including other snails. In *H. pomatia*, the shift was 63 mV from pH_o 7.5 to 6.6 (Doroshenko et al., 1986). In *L. stagnalis*, the shift was 46 mV from pH_o 7.4 to 6.4 (Byerly et al., 1984). Changes in pH_o in a coccolithophore EhH_v1 produced shifts of ~50 mV/unit (Cherny et al., 2015). An insect H_v1 (NpH_v1) shifts 54 mV/unit change in pH_o (Chaves et al., 2016).

More dramatically, changes in pH_i produced much smaller shifts of the $g_{\rm H^-}V$ relationship in HtH_V1 than the 40 mV in mammalian species (Cherny et al., 1995). The mean shift in HtH_V1 between pH_i 5 and 9 was only 15.3 mV/unit (Fig. 9). This is in remarkable agreement with the 15 mV/unit reported in the snail LsH_V1 between pH_i 5.9 and 7.3 (Byerly et al., 1984). Meech (2012) recently emphasized the stronger effects of pH_o over pH_i after reanalyzing old data. However, in another snail, *H. pomatia*, HpH_V1 apparently shifted normally, roughly 30–50 mV/unit change in pH_i (Doroshenko et al., 1986), so on this point it is not possible to generalize about molluscan H_v1.

The ΔpH dependence of mammalian H_VI results in only outward H^+ currents under most circumstances, which is crucial to many if not all of its functions (DeCoursey, 2003). One

striking consequence of the anomalous ΔpH dependence of HtH_v1 and perhaps of other snail H_v1 is that inward currents are readily observed at certain ΔpH . Even at symmetrical pH, there are often inward currents. More conspicuously, because of the weak dependence on pH_i , an inward pH gradient (ΔpH < 0) produces inward currents over an extensive voltage range (e.g., Fig. 8 C). Inward currents would affect neuronal excitability by providing a depolarizing current. They at first appear incompatible with an early proposal that proton currents in snail neurons function to extrude protons that enter via Ca²⁺/ H⁺ exchange after each Ca²⁺-mediated action potential (Ahmed and Connor, 1980; Thomas and Meech, 1982; Byerly et al., 1984), but under normal conditions of an outward H⁺ gradient, inward currents would likely not be activated. Nevertheless, the possibility arises that H_v1 might mediate action potentials in molluscan neurons under certain conditions, although Ca²⁺ channels are thought to be primarily responsible (Hagiwara and Byerly, 1981). H_vl appears to mediate action potentials in bioluminescent dinoflagellates (Fogel and Hastings, 1972; Smith et al., 2011; Rodriguez et al., 2017).

As *L. stagnalis* and *H. trivolvis* snails live in similar habitats, we expect that their proton channels should function similarly. This view is supported by the similarity of the LsH_V1 sequence to that of HtH_V1, especially in the S2/S3 region that we have identified for its importance in pH_i sensing in the accompanying paper (Cherny et al., 2018). In that paper, we identify a single amino acid difference between hH_V1 and HtH_V1 that appears largely responsible for the difference in pH_i sensing.



Acknowledgments

We thank Kristie Bishop (Albany State University) for technical assistance with Western blotting experiments. We gratefully acknowledge heroic efforts by David Colquhoun to disambiguate our fuzzy thoughts on cooperativity.

This work was supported by the National Institutes of Health (grants GM121462 to T.E. DeCoursey and GM102336 to T.E. DeCoursey and S.M.E. Smith) and the National Science Foundation (grant MCB-1242985 to T.E. DeCoursey and S.M.E. Smith and Neuroscience Cluster award 0843173 to V. Rehder). Ms. Bishop was a participant in the Kennesaw State Research Experience for Undergraduates, Chemistry and Biochemistry Summer Undergraduate Research Experience (National Science Foundation CHE-1560329).

The authors declare no competing financial interests.

Author contributions: Conceptualization: S.M.E. Smith and T.E. DeCoursey; data curation: S.M.E. Smith and T.E. DeCoursey; formal analysis: V.V. Cherny, D. Morgan, L.R. Artinian, V. Rehder, and T.E. DeCoursey; funding acquisition: V. Rehder, S.M.E. Smith, and T.E. DeCoursey; investigation: S. Thomas, V.V. Cherny, and D. Morgan; project administration: S.M.E. Smith and T.E. DeCoursey; resources: S. Thomas, L.R. Artinian, V. Rehder, and S.M.E. Smith; visualization: S.M.E. Smith and T.E. DeCoursey; writing (original draft): T.E. DeCoursey; writing (review and editing): V.V. Cherny and S.M.E. Smith.

Richard W. Aldrich served as editor.

Submitted: 7 December 2017 Accepted: 27 March 2018

References

- Ahmed, Z., and J.A. Connor. 1980. Intracellular pH changes induced by calcium influx during electrical activity in molluscan neurons. J. Gen. Physiol. 75:403–426. https://doi.org/10.1085/jgp.75.4.403
- Auld, D.S. 2001. Zinc coordination sphere in biochemical zinc sites. Biometals. 14:271-313. https://doi.org/10.1023/A:1012976615056
- Bernheim, L., R.M. Krause, A. Baroffio, M. Hamann, A. Kaelin, and C.R. Bader. 1993. A voltage-dependent proton current in cultured human skeletal muscle myotubes. J. Physiol. 470:313–333. https://doi.org/10.1113/ jphysiol.1993.sp019860
- Byerly, L., R. Meech, and W. Moody Jr. 1984. Rapidly activating hydrogen ion currents in perfused neurones of the snail, *Lymnaea stagnalis. J. Physiol.* 351:199–216. https://doi.org/10.1113/jphysiol.1984.sp015241
- Cahalan, M.D., K.G. Chandy, T.E. DeCoursey, and S. Gupta. 1985. A voltage-gated potassium channel in human T lymphocytes. J. Physiol. 358:197-237. https://doi.org/10.1113/jphysiol.1985.sp015548
- Chaves, G., C. Derst, A. Franzen, Y. Mashimo, R. Machida, and B. Musset. 2016. Identification of an H_V 1 voltage-gated proton channel in insects. *FEBS J.* 283:1453–1464. https://doi.org/10.1111/febs.13680
- Chaves, G., S. Bungert-Plümke, A. Franzen, and B. Musset. 2018. Zinc inhibition of an insect voltage-gated proton channel. *Biophys. J.* 114:492a. https://doi.org/10.1016/j.bpj.2017.11.2697
- Cherny, V.V., and T.E. DeCoursey. 1999. pH-dependent inhibition of voltage-gated H⁺ currents in rat alveolar epithelial cells by Zn²⁺ and other divalent cations. *J. Gen. Physiol.* 114:819–838. https://doi.org/10.1085/jgp .114.6.819
- Cherny, V.V., V.S. Markin, and T.E. DeCoursey. 1995. The voltage-activated hydrogen ion conductance in rat alveolar epithelial cells is determined by the pH gradient. J. Gen. Physiol. 105:861–896. https://doi.org/10.1085/ jgp.105.6.861
- Cherny, V.V., L.L. Thomas, and T.E. DeCoursey. 2001. Voltage-gated proton currents in human basophils. Биол. мембраны. 18:458–465.

- Cherny, V.V., D. Morgan, B. Musset, G. Chaves, S.M.E. Smith, and T.E. DeCoursey. 2015. Tryptophan 207 is crucial to the unique properties of the human voltage-gated proton channel, hH_v1. *J. Gen. Physiol.* 146:343–356.
- Cherny, V.V., D. Morgan, S. Thomas, S.M.E. Smith, and T.E. DeCoursey. 2018. Histidine¹⁶⁸ is crucial for Δ pH-dependent gating of the human voltage-gated proton channel, hH_v1. *J. Gen. Physiol.* https://doi.org/10.1085/jgp.201711968
- Cohan, C.S., J.L. Karnes, and F.Q. Zhou. 2003. Culturing neurons from the snail Helisoma. Methods Cell Biol. 71:157–170. https://doi.org/10.1016/S0091 -679X(03)01009-4
- Colquhoun, D. 1973. The relation between classical and cooperative models for drug action. *In* Drug Receptors. H.P. Rang, editor. Palmgrave, London. 149–182. https://doi.org/10.1007/978-1-349-00910-7_11.
- DeCoursey, T.E. 1991. Hydrogen ion currents in rat alveolar epithelial cells. *Biophys. J.* 60:1243–1253. https://doi.org/10.1016/S0006-3495(91)82158-0
- DeCoursey, T.E. 2003. Voltage-gated proton channels and other proton transfer pathways. *Physiol. Rev.* 83:475–579. https://doi.org/10.1152/physrev .00028.2002
- DeCoursey, T.E. 2012. Voltage-gated proton channels. Compr. Physiol. 2:1355– 1385. https://doi.org/10.1002/cphy.c100071
- DeCoursey, T.E. 2013. Voltage-gated proton channels: molecular biology, physiology, and pathophysiology of the H_V family. *Physiol. Rev.* 93:599–652. https://doi.org/10.1152/physrev.00011.2012
- DeCoursey, T.E. 2016. The intimate and controversial relationship between voltage-gated proton channels and the phagocyte NADPH oxidase. *Immunol. Rev.* 273:194–218. https://doi.org/10.1111/imr.12437
- DeCoursey, T.E., and V.V. Cherny. 1993. Potential, pH, and arachidonate gate hydrogen ion currents in human neutrophils. *Biophys. J.* 65:1590–1598. https://doi.org/10.1016/S0006-3495(93)81198-6
- DeCoursey, T.E., and V.V. Cherny. 1994. Voltage-activated hydrogen ion currents. J. Membr. Biol. 141:203–223. https://doi.org/10.1007/BF00235130
- DeCoursey, T.E., and V.V. Cherny. 1995. Voltage-activated proton currents in membrane patches of rat alveolar epithelial cells. J. Physiol. 489:299–307. https://doi.org/10.1113/jphysiol.1995.sp021051
- DeCoursey, T.E., and V.V. Cherny. 1996. Voltage-activated proton currents in human THP-1 monocytes. J. Membr. Biol. 152:131–140. https://doi.org/10.1007/s002329900092
- DeCoursey, T.E., and V.V. Cherny. 1997. Deuterium isotope effects on permeation and gating of proton channels in rat alveolar epithelium. *J. Gen. Physiol*. 109:415–434. https://doi.org/10.1085/jgp.109.4.415
- DeCoursey, T.E., V.V. Cherny, A.G. DeCoursey, W. Xu, and L.L. Thomas. 2001. Interactions between NADPH oxidase-related proton and electron currents in human eosinophils. J. Physiol. 535:767–781. https://doi.org/10 .1111/j.1469-7793.2001.00767.x
- Delorenzi, M., and T. Speed. 2002. An HMM model for coiled-coil domains and a comparison with PSSM-based predictions. *Bioinformatics*. 18:617– 625. https://doi.org/10.1093/bioinformatics/18.4.617
- Demaurex, N., S. Grinstein, M. Jaconi, W. Schlegel, D.P. Lew, and K.H. Krause. 1993. Proton currents in human granulocytes: regulation by membrane potential and intracellular pH. J. Physiol. 466:329–344.
- Doroshenko, P.A., P.G. Kostyuk, and A.E. Martynyuk. 1986. Transmembrane outward hydrogen current in intracellularly perfused neurones of the snail *Helix pomatia. Gen. Physiol. Biophys.* 5:337-350.
- Fogel, M., and J.W. Hastings. 1972. Bioluminescence: mechanism and mode of control of scintillon activity. Proc. Natl. Acad. Sci. USA. 69:690–693. https://doi.org/10.1073/pnas.69.3.690
- Fujiwara, Y., T. Kurokawa, K. Takeshita, M. Kobayashi, Y. Okochi, A. Nakagawa, and Y. Okamura. 2012. The cytoplasmic coiled-coil mediates cooperative gating temperature sensitivity in the voltage-gated H⁺ channel Hv1. Nat. Commun. 3:816. https://doi.org/10.1038/ncomms1823
- Gonzalez, C., H.P. Koch, B.M. Drum, and H.P. Larsson. 2010. Strong cooperativity between subunits in voltage-gated proton channels. Nat. Struct. Mol. Biol. 17:51–56. https://doi.org/10.1038/nsmb.1739
- Gonzalez, C., S. Rebolledo, M.E. Perez, and H.P. Larsson. 2013. Molecular mechanism of voltage sensing in voltage-gated proton channels. J. Gen. Physiol. 141:275–285. https://doi.org/10.1085/jgp.201210857
- Hagiwara, S., and L. Byerly. 1981. Calcium channel. Annu. Rev. Neurosci. 4:69– 125. https://doi.org/10.1146/annurev.ne.04.030181.000441
- Hille, B. 2001. Ion Channels of Excitable Membranes. Third edition. Sinauer Associates, Sunderland, MA. 814 pp.
- Hoch, E., W. Lin, J. Chai, M. Hershfinkel, D. Fu, and I. Sekler. 2012. Histidine pairing at the metal transport site of mammalian ZnT transporters controls Zn²⁺ over Cd²⁺ selectivity. *Proc. Natl. Acad. Sci. USA*. 109:7202–7207. https://doi.org/10.1073/pnas.1200362109



- Hodgkin, A.L., and A.F. Huxley. 1952. A quantitative description of membrane current and its application to conduction and excitation in nerve. J. Physiol. 117:500–544. https://doi.org/10.1113/jphysiol.1952 .sp004764
- Imperiali, B., and S.E. O'Connor. 1999. Effect of N-linked glycosylation on glycopeptide and glycoprotein structure. Curr. Opin. Chem. Biol. 3:643–649. https://doi.org/10.1016/S1367-5931(99)00021-6
- Kapus, A., R. Romanek, A.Y. Qu, O.D. Rotstein, and S. Grinstein. 1993. A pH-sensitive and voltage-dependent proton conductance in the plasma membrane of macrophages. J. Gen. Physiol. 102:729–760. https://doi.org/ 10.1085/jgp.102.4.729
- Kater, S.B. 1974. Feeding in *Helisoma trivolvis*: The morphological and physiological bases of a fixed action pattern. *Am. Zool.* 14:1017–1036. https:// doi.org/10.1093/icb/14.3.1017
- Koch, H.P., T. Kurokawa, Y. Okochi, M. Sasaki, Y. Okamura, and H.P. Larsson. 2008. Multimeric nature of voltage-gated proton channels. Proc. Natl. Acad. Sci. USA. 105:9111–9116. https://doi.org/10.1073/pnas .0801553105
- Lee, S.Y., J.A. Letts, and R. Mackinnon. 2008. Dimeric subunit stoichiometry of the human voltage-dependent proton channel Hv1. Proc. Natl. Acad. Sci. USA. 105:7692–7695. https://doi.org/10.1073/pnas.0803277105
- Li, Q., R. Shen, J.S. Treger, S.S. Wanderling, W. Milewski, K. Siwowska, F. Bezanilla, and E. Perozo. 2015. Resting state of the human proton channel dimer in a lipid bilayer. Proc. Natl. Acad. Sci. USA. 112:E5926–E5935. https://doi.org/10.1073/pnas.1515043112
- Li, S.J., Q. Zhao, Q. Zhou, H. Unno, Y. Zhai, and F. Sun. 2010. The role and structure of the carboxyl-terminal domain of the human voltage-gated proton channel Hv1. J. Biol. Chem. 285:12047–12054. https://doi.org/10 .1074/jbc.M109.040360
- Mahaut-Smith, M.P. 1989a. The effect of zinc on calcium and hydrogen ion currents in intact snail neurones. J. Exp. Biol. 145:455–464.
- Mahaut-Smith, M.P. 1989b. Separation of hydrogen ion currents in intact molluscan neurones. J. Exp. Biol. 145:439–454.
- Meech, R. 2012. A contribution to the history of the proton channel. Wiley Interdiscip. Rev. Membr. Transp. Signal. 1:533–557. https://doi.org/10 .1002/wmts.59
- Musset, B., S.M.E. Smith, S. Rajan, V.V. Cherny, D. Morgan, and T.E. DeCoursey. 2010a. Oligomerization of the voltage-gated proton channel. *Channels* (Austin). 4:260–265. https://doi.org/10.4161/chan.4.4.12789
- Musset, B., S.M.E. Smith, S. Rajan, VV. Cherny, S. Sujai, D. Morgan, and T.E. DeCoursey. 2010b. Zinc inhibition of monomeric and dimeric proton channels suggests cooperative gating. J. Physiol. 588:1435–1449. https:// doi.org/10.1113/jphysiol.2010.188318
- Musset, B., S.M.E. Smith, S. Rajan, D. Morgan, VV. Cherny, and T.E. DeCoursey. 2011. Aspartate112 is the selectivity filter of the human voltage-gated proton channel. *Nature*. 480:273–277. https://doi.org/10.1038/ nature10557
- Qiu, F., A. Chamberlin, B.M. Watkins, A. Ionescu, M.E. Perez, R. Barro-Soria, C. González, S.Y. Noskov, and H.P. Larsson. 2016. Molecular mechanism of Zn²⁺ inhibition of a voltage-gated proton channel. *Proc. Natl. Acad. Sci.* USA. 113:E5962–E5971. https://doi.org/10.1073/pnas.1604082113

- Ramsey, I.S., M.M. Moran, J.A. Chong, and D.E. Clapham. 2006. A voltage-gated proton-selective channel lacking the pore domain. *Nature*. 440:1213–1216. https://doi.org/10.1038/nature04700
- Ramsey, I.S., Y. Mokrab, I. Carvacho, Z.A. Sands, M.S.P. Sansom, and D.E. Clapham. 2010. An aqueous H⁺ permeation pathway in the voltage-gated proton channel Hv1. *Nat. Struct. Mol. Biol.* 17:869–875. https://doi.org/10 .1038/nsmb.1826
- Ratanayotha, A., T. Kawai, S.I. Higashijima, and Y. Okamura. 2017. Molecular and functional characterization of the voltage-gated proton channel in zebrafish neutrophils. *Physiol. Rep.* 5:e13345. https://doi.org/10.14814/ phy2.13345
- Rodriguez, J.D., S. Haq, T. Bachvaroff, K.F. Nowak, S.J. Nowak, D. Morgan, V.V. Cherny, M.M. Sapp, S. Bernstein, A. Bolt, et al. 2017. Identification of a vacuolar proton channel that triggers the bioluminescent flash in dinoflagellates. *PLoS One.* 12:e0171594. https://doi.org/10.1371/journal.pone .0171594
- Sakata, S., N. Miyawaki, T.J. McCormack, H. Arima, A. Kawanabe, N. Özkucur, T. Kurokawa, Y. Jinno, Y. Fujiwara, and Y. Okamura. 2016. Comparison between mouse and sea urchin orthologs of voltage-gated proton channel suggests role of S3 segment in activation gating. *Biochim. Biophys. Acta*. 1858:2972–2983. https://doi.org/10.1016/j.bbamem.2016.09.008
- Sasaki, M., M. Takagi, and Y. Okamura. 2006. A voltage sensor-domain protein is a voltage-gated proton channel. *Science*. 312:589–592. https://doi .org/10.1126/science.1122352
- Schilling, T., A. Gratopp, T.E. DeCoursey, and C. Eder. 2002. Voltage-activated proton currents in human lymphocytes. J. Physiol. 545:93–105. https:// doi.org/10.1113/jphysiol.2002.028878
- Smith, S.M.E., D. Morgan, B. Musset, V.V. Cherny, A.R. Place, J.W. Hastings, and T.E. DeCoursey. 2011. Voltage-gated proton channel in a dinoflagellate. Proc. Natl. Acad. Sci. USA. 108:18162–18167. https://doi.org/10.1073/pnas .1115405108
- Takeshita, K., S. Sakata, E. Yamashita, Y. Fujiwara, A. Kawanabe, T. Kurokawa, Y. Okochi, M. Matsuda, H. Narita, Y. Okamura, and A. Nakagawa. 2014. X-ray crystal structure of voltage-gated proton channel. *Nat. Struct. Mol. Biol.* 21:352–357. https://doi.org/10.1038/nsmb.2783
- Taylor, A.R., A. Chrachri, G. Wheeler, H. Goddard, and C. Brownlee. 2011. A voltage-gated H⁺ channel underlying pH homeostasis in calcifying coccolithophores. *PLoS Biol.* 9:e1001085. https://doi.org/10.1371/journal .pbio.1001085
- Thomas, R.C., and R.W. Meech. 1982. Hydrogen ion currents and intracellular pH in depolarized voltage-clamped snail neurones. *Nature*. 299:826–828. https://doi.org/10.1038/299826a0
- Tombola, F., M.H. Ulbrich, and E.Y. Isacoff. 2008. The voltage-gated proton channel Hv1 has two pores, each controlled by one voltage sensor. *Neuron.* 58:546–556. https://doi.org/10.1016/j.neuron.2008.03.026
- Tombola, F., M.H. Ulbrich, S.C. Kohout, and E.Y. Isacoff. 2010. The opening of the two pores of the Hv1 voltage-gated proton channel is tuned by cooperativity. Nat. Struct. Mol. Biol. 17:44–50. https://doi.org/10.1038/ nsmb.1738
- Villalba-Galea, C.A. 2014. Hv1 proton channel opening is preceded by a voltage-independent transition. *Biophys. J.* 107:1564–1572. https://doi.org/ 10.1016/j.bpj.2014.08.017