



## *N*-*tert*-Butanesulfinyl imines in the asymmetric synthesis of nitrogen-containing heterocycles

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### Review

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## Abstract

The synthesis of nitrogen-containing heterocycles, including natural alkaloids and other compounds presenting different types of biological activities have proved to be successful employing chiral sulfinyl imines derived from *tert*-butanesulfinamide. These imines are versatile chiral auxiliaries and have been extensively used as electrophiles in a wide range of reactions. The electron-withdrawing sulfinyl group facilitates the nucleophilic addition of organometallic compounds to the iminic carbon with high diastereoisomeric excess and the free amines obtained after an easy removal of the *tert*-butanesulfinyl group can be transformed into enantioenriched nitrogen-containing heterocycles. The goal of this review is to highlight enantioselective syntheses of heterocycles involving the use of chiral *N*-*tert*-butanesulfinyl imines as reaction intermediates, including the synthesis of several natural products. The synthesis of nitrogen-containing heterocycles in which the nitrogen atom is not provided by the chiral imine will not be considered in this review. The sections are organized according to the size of the heterocycles. The present work will comprehensively cover the most pertinent contributions to this research area from 2012 to 2020. We regret in advance that some contributions are excluded in order to maintain a concise format.

## Introduction

Chiral imines derived from *tert*-butanesulfinamide have been extensively used as electrophiles in a wide range of reactions. The presence of the chiral electron-withdrawing sulfinyl group facilitates the nucleophilic addition of organometallic compounds to the iminic carbon [1-3]. The ready availability of both enantiomers of *tert*-butanesulfinamide in large-scale processes, the easy deprotection of the amine under mild acidic conditions, and a practical procedure for recycling the chiral auxiliary [4,5] have contributed to the widespread use of these imines as precursors of chiral compounds with a nitrogen atom bonded to a stereogenic center. The amine derivatives, resulting after removal of the *tert*-butanesulfinyl group, can be transformed into enantioenriched nitrogen-containing heterocycles [6,7] including natural alkaloids [8-11] and other compounds that show different types of biological activities [12,13]. The way to achieve these transformations is by intramolecular cyclizations, involving the free primary amine, and appropriate reactive positions (those positions bearing a leaving group) in the electrophile or in the carbonyl component of the starting imine (Scheme 1).

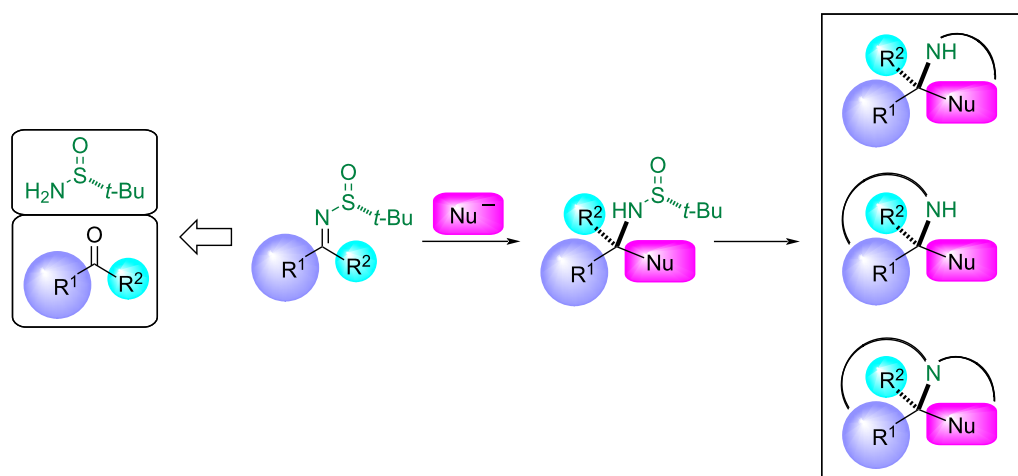
## Synthesis of *tert*-butane *N*-sulfinyl imines

The first method developed for the synthesis of enantiomerically pure *N*-*tert*-butanesulfonylamine **1** was reported by Ellman and co-workers [14,15]. In 1999, they described the synthesis of imines from the condensation reaction of aldehydes or ketones with *tert*-butanesulfinamides. In this work, the condensation with aldehydes was carried using Ti(OEt)<sub>4</sub> in tetrahydrofuran (THF), or CuSO<sub>4</sub> in dichloromethane at room temperature. The combination of MgSO<sub>4</sub> in the presence of a catalytic amount of pyridinium *p*-toluenesulfonate (PPTS) also worked well to

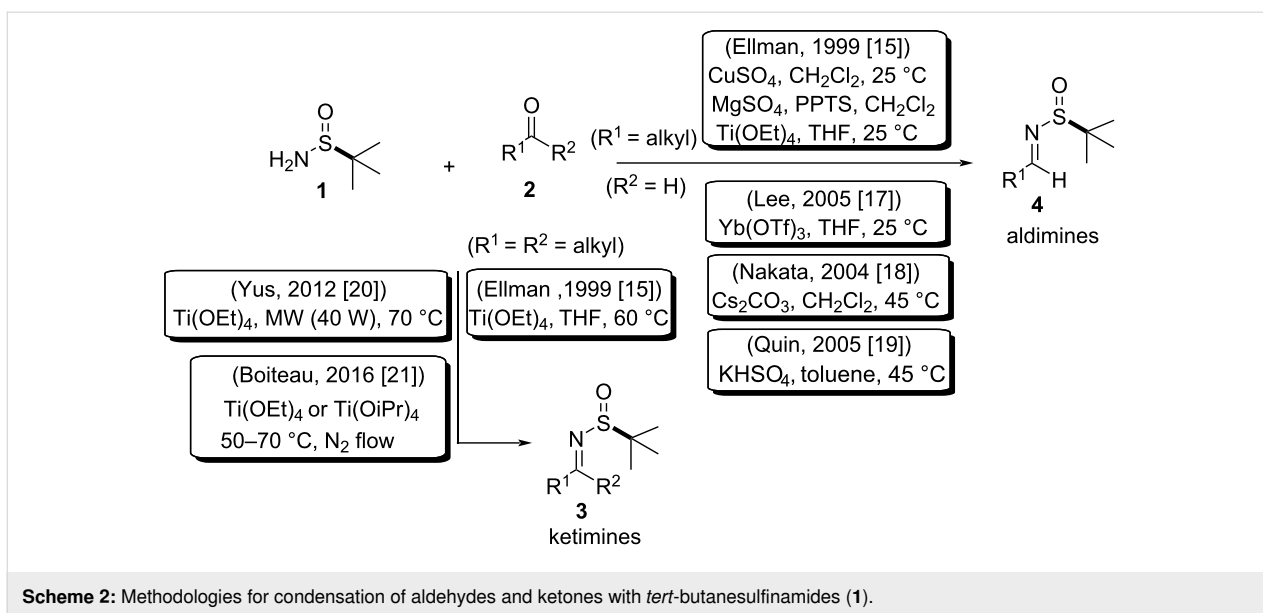
perform these condensations [1,16]. For the formation of aldimines, other methodologies are described in the literature using condensation reagents such as Yb(OTf)<sub>3</sub> [17], Cs<sub>2</sub>CO<sub>3</sub> [18] and KHSO<sub>4</sub> [19]. However, for the synthesis of ketimines, Ti(OEt)<sub>4</sub> was the only effective reagent when performing the reaction at 60 °C in THF [16]. Ketimines were also synthesized with Ti(OEt)<sub>4</sub>, under microwave irradiation in a solvent-free system [20]. In hindered ketones, Ti(OiPr)<sub>4</sub> or Ti(OEt)<sub>4</sub> using vacuum or under a nitrogen flow were effective to *tert*-butanesulfinyl ketimine condensation (Scheme 2) [21].

## Mechanism of addition of nucleophiles to *N*-sulfinyl imines

The *p*-toluenesulfinamide **5** was first described by Davis and co-workers in a racemic form [22], and subsequently, the compound was prepared and isolated as a single enantiomer [23,24], becoming an important tool in the asymmetric synthesis of aziridines [25,26],  $\alpha$ -amino acids [27,28],  $\beta$ -amino acids [23,29] and branched  $\alpha$ -amines [30,31]. The Darzens-type asymmetric synthesis of *N*-(*p*-toluenesulfinyl)aziridine 2-carboxylate esters (**7** and **8**) was described through the addition of lithium  $\alpha$ -bromo enolates to enantiopure *p*-toluenesulfinamide **5**. *cis*-aziridine **7a** was formed as the major diastereoisomer in 89% yield and the *trans*-isomer in 8% yield in a one-step procedure using lithium enolates of methyl bromoacetate **6a** and sulfinyl imine **5**. Lithium enolates of methyl  $\alpha$ -bromopropionate gave *trans*-aziridine in 50% yield under the same conditions. The transition state is proposed with a six-membered chair-like transition containing a four-membered metallocycle. In *cis*-aziridine the enolate of methyl  $\alpha$ -bromoacetate has *E*-geometry and the *trans*-aziridine **8a** has *Z*-geometry [32,33]. In the transition



**Scheme 1:** General strategy for the enantioselective synthesis of *N*-containing heterocycles from *N*-*tert*-butanesulfinyl imines.

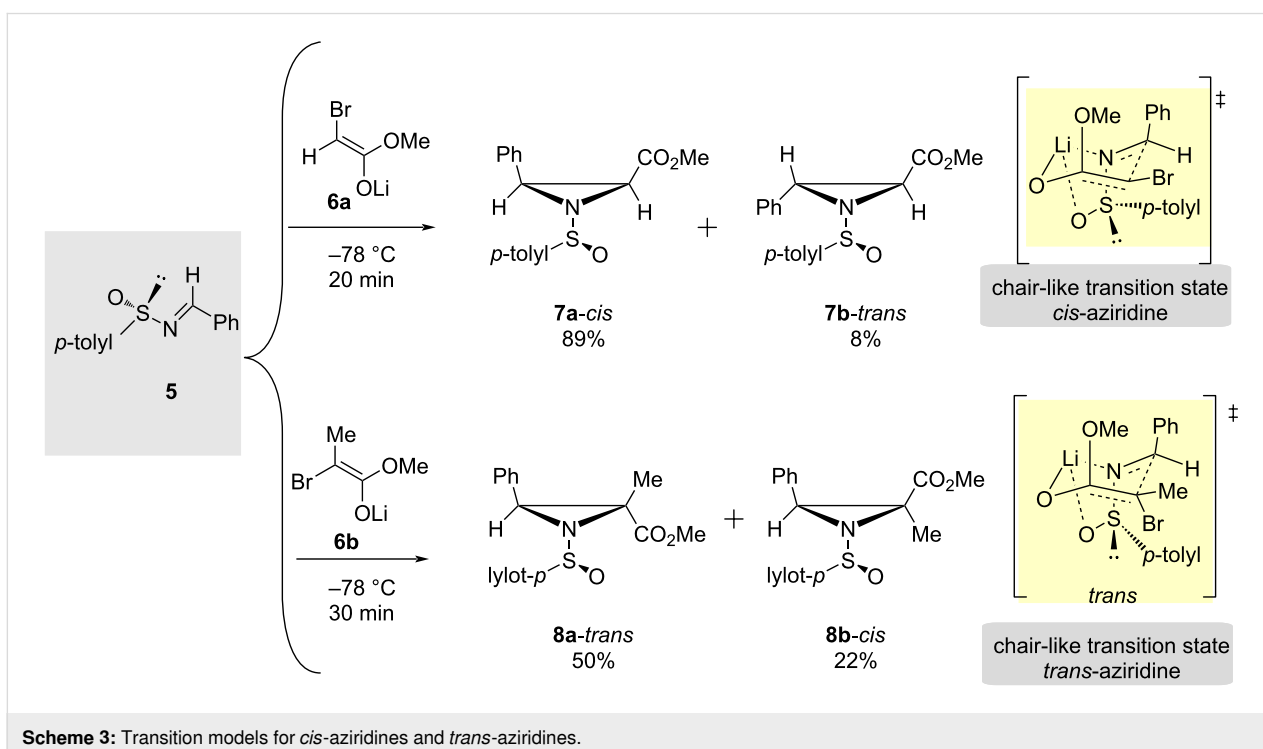


**Scheme 2:** Methodologies for condensation of aldehydes and ketones with *tert*-butanesulfinamides (1).

state, the metal cation of the enolate is being coordinated with both nitrogen and oxygen atoms of the sulfinimine [25,26] (Scheme 3).

In 1999, Ellman and co-workers described the reduction of sulfinyl imines using sodium borohydride (NaBH<sub>4</sub>) [34] or *L*-selectride [35]. Davis–Ellman transition state models were proposed to rationalize organometallic additions to *N*-sulfinyl imines. The mechanism for obtaining these two stereoisomers

was elucidated in the work published by Andersen and co-workers [36]. The origin of the reversal of the diastereofacial selectivity on the change of reducing agents is based on the operating transition states [37,38]. A cyclic transition state is proposed in the reaction with sodium borohydride. In this transition state, the oxygen of the sulfinyl group interacts with the boron atom, facilitating the release of the hydride, directing the attack to the *Si*-face of the imine with (*R*,*E*) configuration. When the reduction is performed with *L*-selectride, with the



**Scheme 3:** Transition models for *cis*-aziridines and *trans*-aziridines.

poorly coordinating metal hydride, an open transition state operates due to the bulkiness of the isobutyl groups bonded to the boron atom. In this case, the attack of the hydride takes place to the less hindered *Re*-face of the imine (Scheme 4) [1,36].

The nucleophilic addition reactions to *N-tert*-butanesulfinyl imines were also described by Ellman and co-workers who reported the addition of allylmagnesium bromide to ketimines. The employment of Grignard reagents showed greater diastereoselectivity than reactions using organolithium and organocerium compounds. In some examples, the use of organolithium is feasible through the use of aluminum-derived additives [15]. In this study, the influence of solvents on diastereoselectivity was also observed. They found that the reactions performed in noncoordinating solvents, such as toluene and dichloromethane, took place with high diastereoselectivity. However, solvents such as ether and THF had a negative impact on stereoselectivity [15]. On the other hand, recent studies developed by Sirvent and Foubelo demonstrated the influence of the solvents in both the yield and diastereoselectivity in these reactions. They found that working in THF led to higher yields and poorer diastereoselectivities than when the reactions were performed in less coordinating solvents, such as diethyl ether and toluene [39].

Based on a broader analysis related to the effects of the solvent, metal and additives in 1,2-addition reactions to *N-tert*-butanesulfinyl imines of organometallic compounds, different transition models have been proposed to explain the stereochemical outcomes. The cyclic model justified by the Zimmermann–Traxler transition state [40–42] is the typical mechanism

operating in reactions involving Grignard reagents in noncoordinating solvents, such as toluene and dichloromethane, while an acyclic model [43] is common in organolithium compounds in solvents such as THF. In the cyclic model, the bulky *tert*-butyl group occupies an equatorial position due to steric hindrance [1,14,44,45] (Scheme 5).

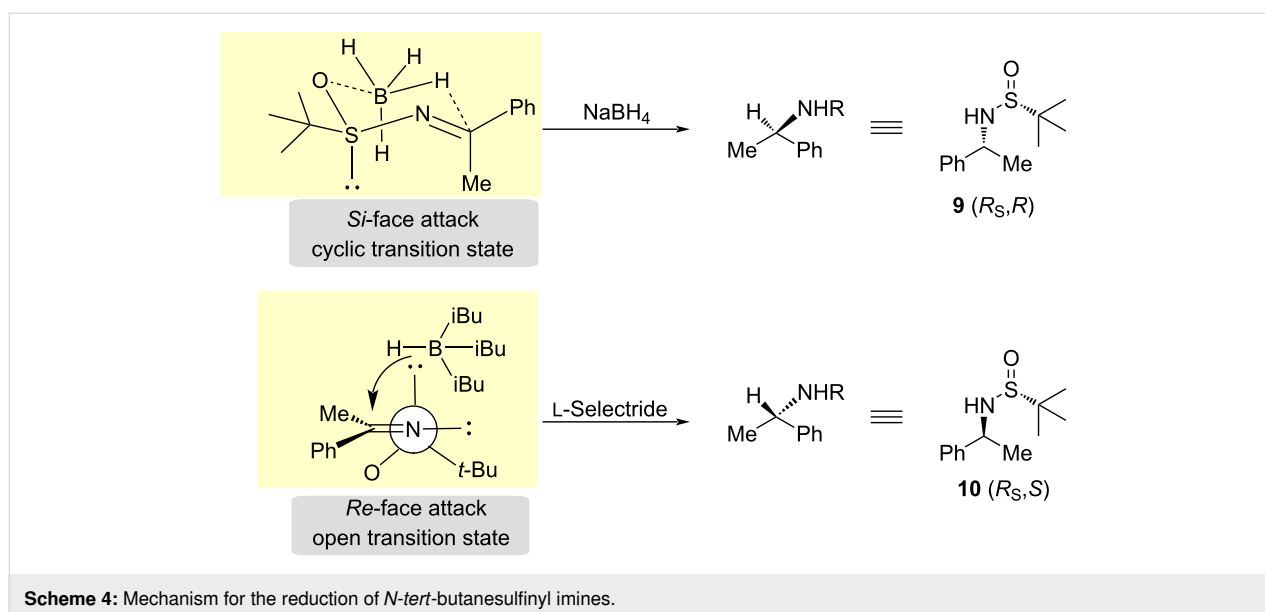
Other contributions to the nucleophilic addition reactions to *N-tert*-butanesulfinyl imines were made by Yus and co-workers employing organozincates [46–49] and indium [50].

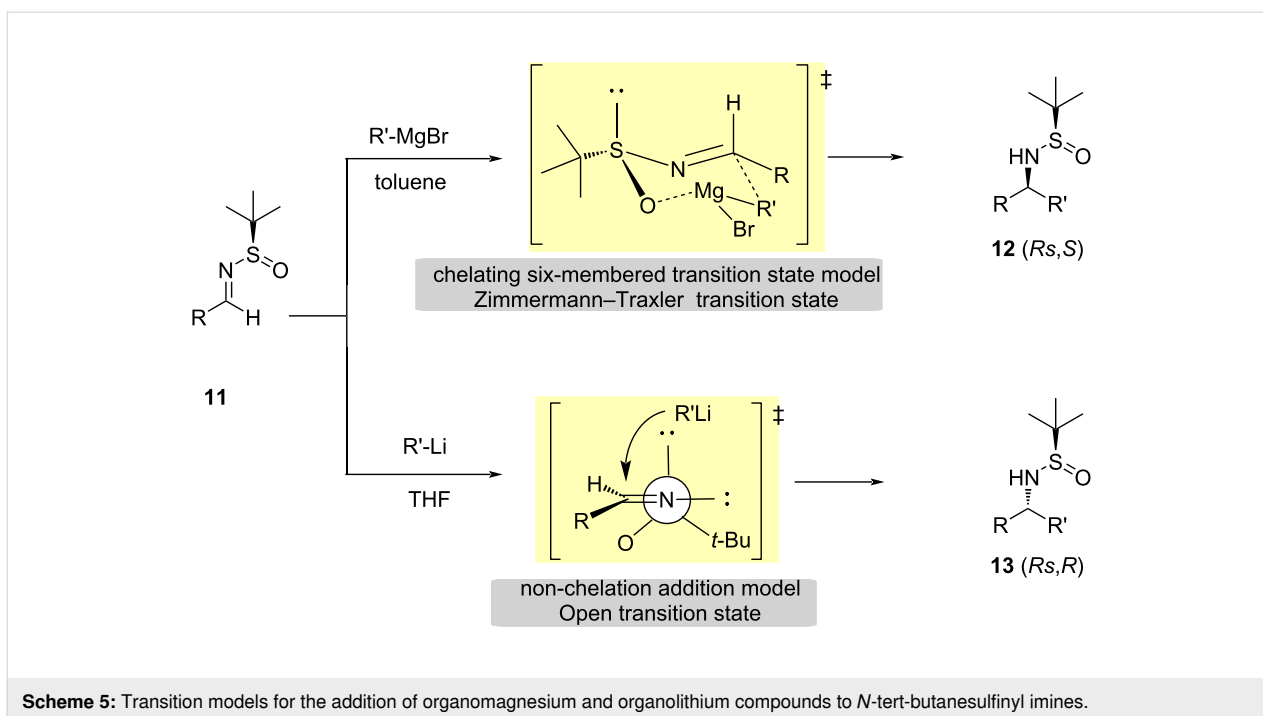
After this brief about the synthesis of enantiomerically pure *N-tert*-butanesulfonamide and applications in some nucleophilic additions, the next sections will describe the synthesis of several alkaloids according to the size of the heterocycles. We regret in advance that some contributions are excluded in order to maintain a concise format.

## Review

### Asymmetric synthesis of aziridines

Saturated nitrogen-containing three-membered heterocycles have attracted increasing interest in recent years because compounds with this structural motif display quite diverse pharmacological activities. Chiral aziridines [51] also play an important role in asymmetric synthesis because they can act both as ligands [52–55] and as chiral auxiliaries [56]. The most widely used synthetic methods to form the aziridine ring [57–61] include intramolecular cyclizations in amines bearing potential leaving groups. Stereoselective syntheses of aziridines have been successfully carried out by combining a nucleophilic addition to *N-tert*-butanesulfinyl  $\alpha$ -chloroimines, and an intramolecular cyclization, the chlorine atom being finally displaced.

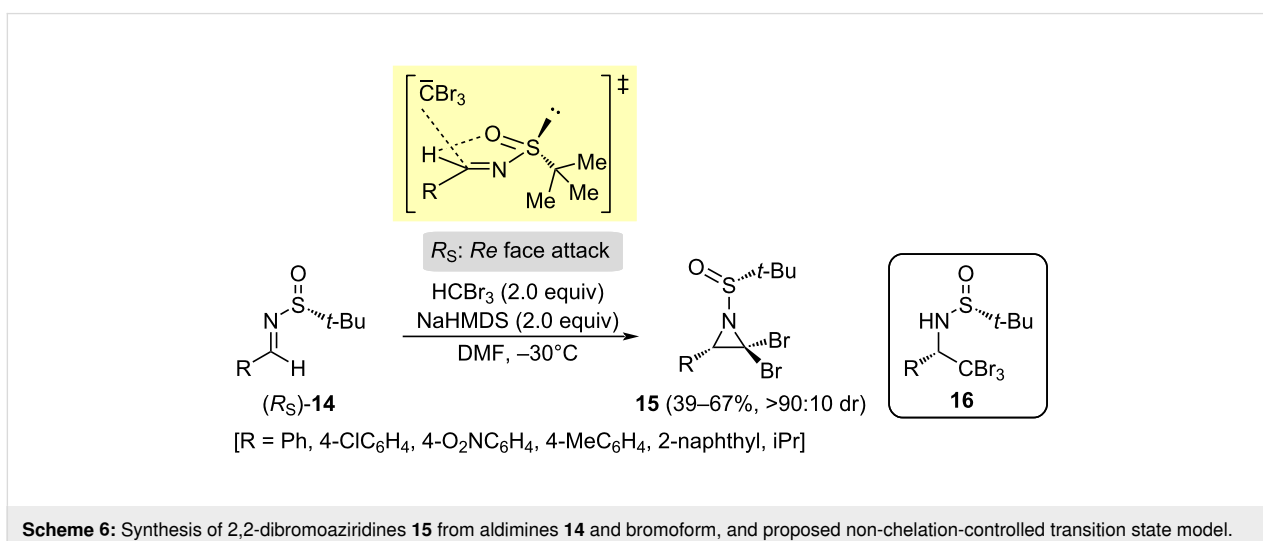




Chiral *N*-tert-butanesulfinyl aldimines and ketimines have also been used successfully to form aziridines through aza-Darzens and Corey–Chaykovsky reactions [62,63].

The first asymmetric synthesis of 2,2-dibromoaziridines **15** was achieved by performing the nucleophilic addition of the anion resulting from the deprotonation of bromoform with sodium hexamethyldisilazide (NaHMDS) to chiral *N*-tert-butanesulfinyl aldimines (*RS*)-**14**, at low temperature, and using DMF as solvent. After addition, a subsequent intramolecular cyclization involving the resulting amide and the vicinal carbon with bromine atoms took place. By contrary, when the reaction was

carried out in THF, the elimination process was suppressed, leading exclusively to enantiomerically pure  $\alpha$ -tribromomethylamines **16**. The structure and configuration of aziridines **15** were determined unambiguously by single crystal X-ray analysis [64]. In order to explain the experimental results, a nonchelation controlled transition state was proposed. Based on computational studies, it is known that a kind of *s-cis* arrangement of the sulfinyl group is the most stable conformation of the imine, due to the contribution of the hydrogen bonding of the oxygen and the iminic hydrogen. In this scenario, the tribromomethyl anion attacked the less hindered *Re* face of the imine with (*RS*) configuration (Scheme 6).

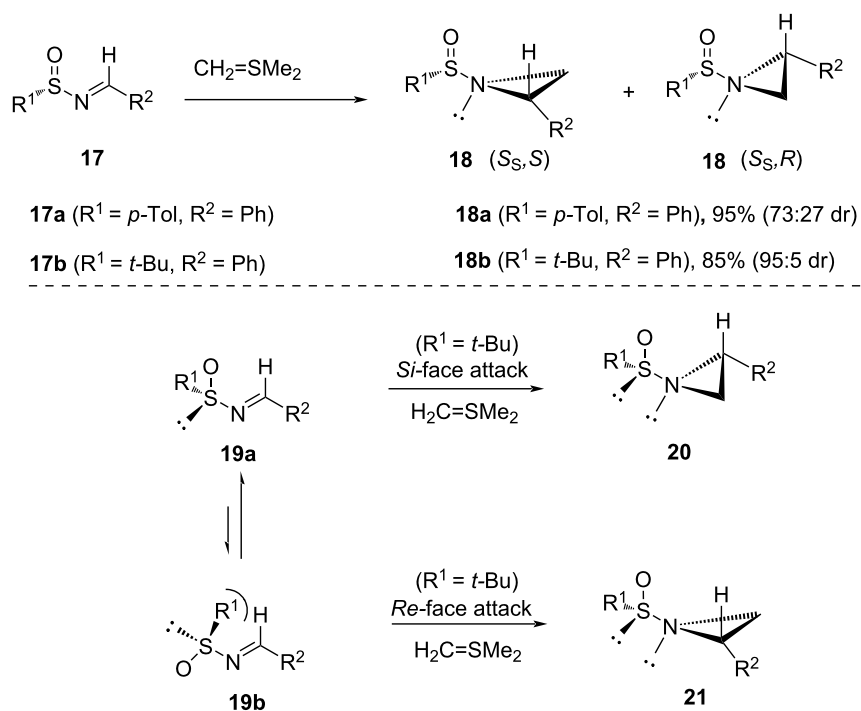


The applicability of sulfinyl imines was also shown by Garcia Ruano and co-workers. (*S<sub>S</sub>*)-*tert*-Butanesulfinyl imine **17b** provided better diastereoselectivity to obtain aziridines **18b** than (*S<sub>S</sub>*)-*tert*-butanesulfinyl imine **17a** to obtain aziridines **18a** (Scheme 7) [65].

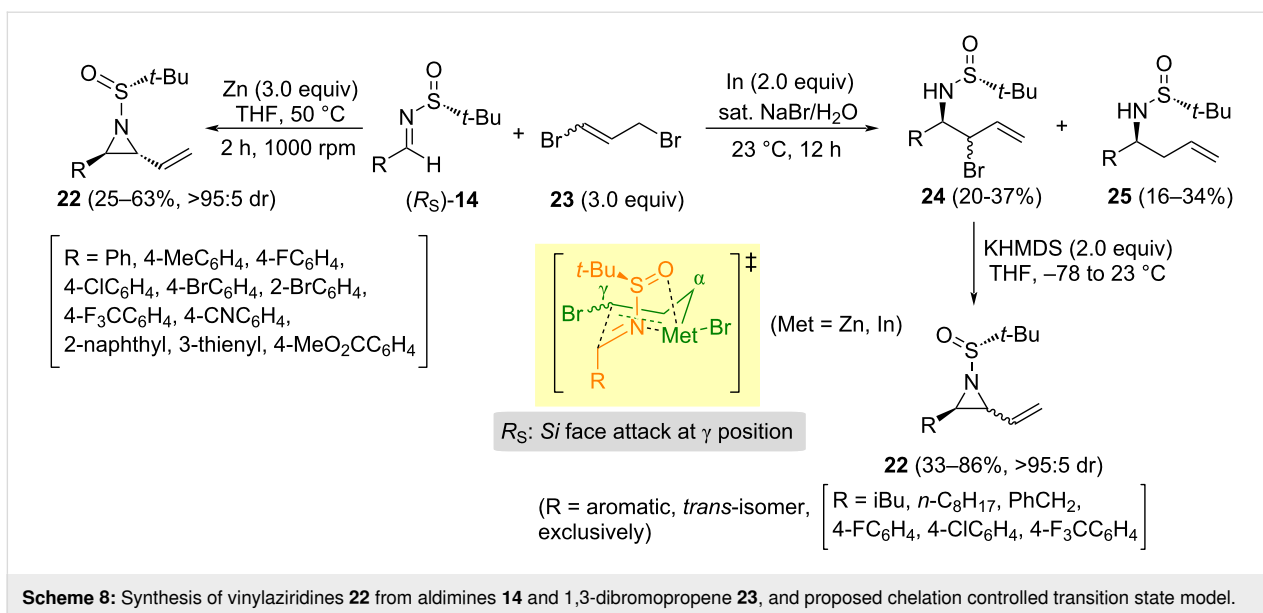
Allylation of *N*-*tert*-butanesulfinyl imines **14** with allylic bromides in the presence of zinc or indium metals is a well-known reaction [66,67]. It is possible to control and predict the stereochemistry of the addition to get the corresponding homoallylamine derivative with a high level of stereocontrol. The reaction of chiral imine **14** with an excess of 1,3-dibromopropene (**23**) in THF at 50 °C for 2 h, and a 1000 rpm rotation speed, led to *trans*-vinylaziridines **22**, in moderate yields and high diastereoselectivities. Those were the reaction conditions that Sun and co-workers found to be optimal for the formation of the vinylaziridines [68]. However, when the allylation was performed in the presence of indium metal, in a saturated aqueous solution of sodium bromide, a mixture of the bromoallylation and allylation products **25** and **26**, respectively, were obtained [69]. The addition of the allyl unit to the imines proceeded with total facial diastereoselectivity, producing also preferably diastereoisomers with *anti* relative configuration. According to the mechanism depicted on Scheme 8, the allyl unit reacted at  $\gamma$ -position, taking place the addition to the *Si* face of the imines with *R<sub>S</sub>* configuration. The bromoallylated product **24** was obtained as a mixture of *anti/syn* diastereoisomers. Treatment of com-

pounds **24** with potassium hexamethyldisilazide provided vinylaziridines **22** through an intramolecular cyclization step. This intramolecular nucleophilic substitution is a stereospecific process. In the case of aromatic compounds **24**, *trans*-vinylaziridines were the only reaction products, meanwhile for aliphatic derivatives **24** (R = alkyl), *trans*- and *cis*-aziridines **22** were isolated in practically the same ratio as the *anti/syn* ratio of their precursors **24**. This shows that the cyclization reaction is stereospecific (Scheme 8). Comparing both methodologies, the indium-mediated bromoallylation seemed to be superior, since aliphatic aldimines **14** were compatible with this approach.

The group of Stockman reported the synthesis of 2,2',3-substituted aziridines **27** from *N*-*tert*-butanesulfinyl imines **14** and  $\alpha$ -bromoesters **26** by applying an aza-Darzens methodology [70]. The reactions were performed in THF at -78 °C, using lithium hexamethyldisilazide as base. Aziridines with relative *trans*-configuration were obtained in good yields and excellent stereoselectivities with methyl  $\alpha$ -bromo- $\alpha$ -phenylacetate (**26**, R<sup>2</sup> = Ph). Lower yields, and poorer diastereoselectivities were observed with less bulky methyl 2-bromo-2-butenate (R<sup>2</sup> = CH<sub>3</sub>CH=), leading to an almost complete loss of *cis/trans* selectivity by the reaction with aliphatic aldimines **14**. The absolute configuration of the reaction products was unambiguously determined after X-ray crystallographic analysis of some of the reaction products. In order to rationalize the observed stereochemical outcome, a six-membered cyclic transition state



**Scheme 7:** Diastereoselective synthesis of aziridines from *tert*-butanesulfinyl imines.

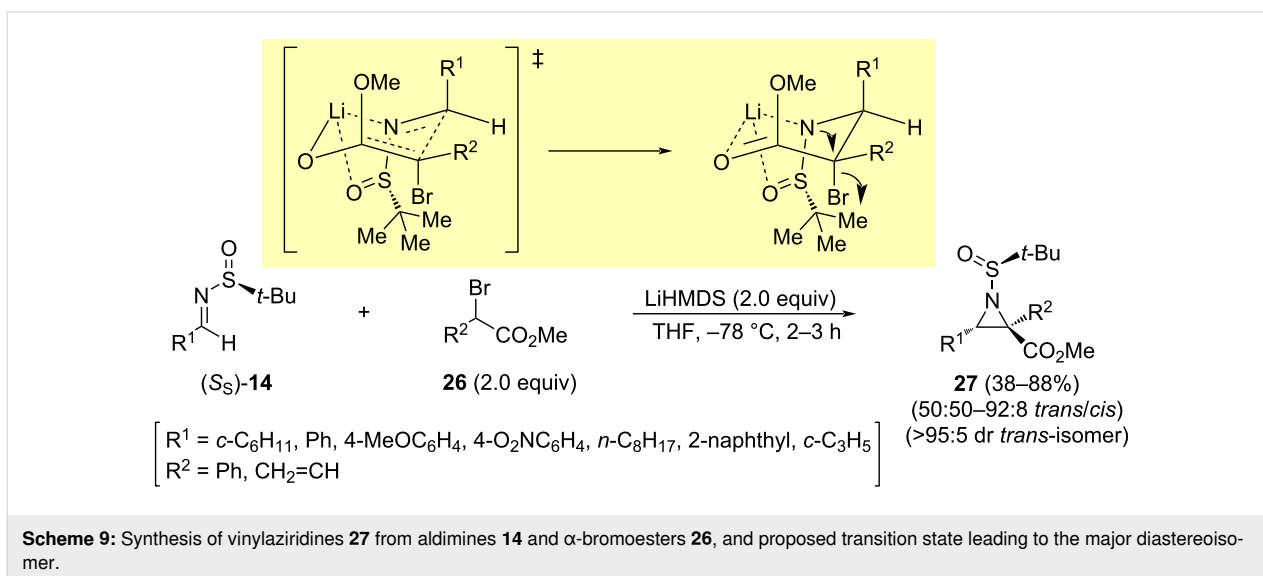


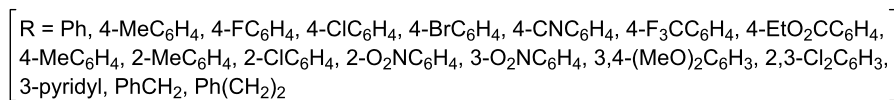
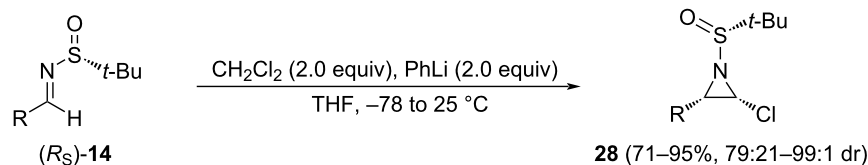
has been proposed. The addition to imine **14** with  $S_S$  configuration takes place on the less hindered *Re* face, on the other hand, the *cis/trans* selectivity observed being a consequence of the *E* stereochemistry of both the imine **14** and the enolate derived from bromoester **26** (Scheme 9).

A two-step protocol carried out in a single synthetic operation was developed by Chen and Zhang to synthesize 3-substituted 2-chloroaziridines with relative *cis* configuration [71]. The reaction of chiral imines **14** in dichloromethane in the presence of 2 equivalents of phenyllithium at  $-78$  °C to room temperature produced the expected 2-chloroaziridines with excellent yields and diastereoselectivities. Under the previously commented optimized reaction conditions, dichloromethylithium is first

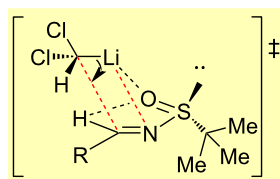
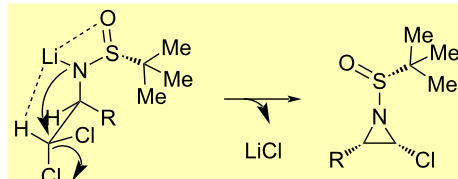
generated, taking place a diastereoselective nucleophilic dichloromethylation of the imine. The addition took place almost exclusively to the *Re* face of the imine ( $R_S$ )-**14**, which is the less sterically hindered in the most stable *s-cis* conformation (see Scheme 2). The second intramolecular *N*-alkylation step produced the 2-chloroaziridines **28** with relative *cis* configuration. The absolute proposed configurations were confirmed by X-ray crystallographic analysis (Scheme 10).

An interesting asymmetric vinylogous aza-Darzens reaction was employed to access *cis*-vinylaziridines **30** and **31**. The group of Njardarson found that the reaction of different aromatic and aliphatic chiral imines ( $S_S$ )-**14** with the dienolate resulting from the deprotonation of bromomethyl butenolide **29** in THF at





step 1: nucleophilic dichloromethylation

step 2: *N*-alkylation $R_S$ : *Re* face attack

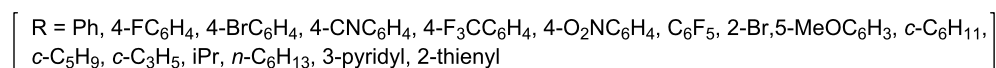
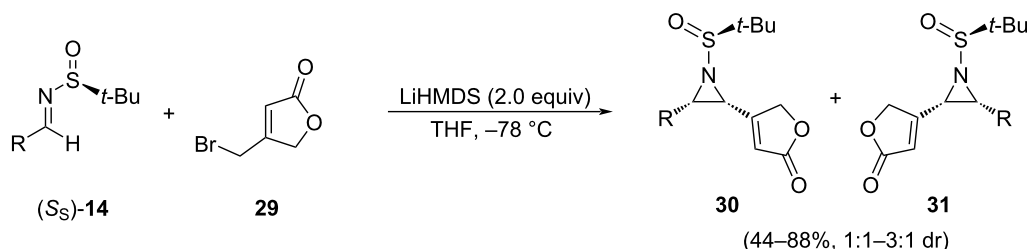
intramolecular nucleophilic substitution

**Scheme 10:** Synthesis of 2-chloroaziridines **28** from aldimines **14** and dichloromethane, and proposed transition state model for the nucleophilic addition and for the elimination step.

–78 °C led to a mixture of diastereomeric *cis*-vinylaziridines **30** and **31** with good yields in most cases. Lithium hexamethyldisilazide was the base of choice to perform the deprotonation, and it must be added very slowly to the reaction mixture in order to suppress self-dimerization of the butenolide [72]. The structures as well as the absolute and relative stereochemistry of reaction products **30** and **31** were also unambiguously determined following a single-crystal X-ray analysis (Scheme 11).

The stereoselective synthesis of diastereomeric 2-chloro-2-arylaziridines **36** and **32** was successfully accomplished through a three-component cascade coupling reaction of silyldichloro-methanes **33**, aryl nitriles **34** and chiral *N*-*tert*-butanesulfonyl aldimines ( $R_S$ )-**14**. The process reported by Lu, Xu and co-workers started with the deprotonation of silyl com-

pounds **33** with LDA at –78 °C, leading to the corresponding silyldichloromethyl lithium derivative, which reacted with aryl nitrile **34**. After nucleophilic addition and [1,3]-aza-Brook rearrangement, *N*-silyllithiumamide **35** was formed. This strongly nucleophilic species could be trapped by the chiral imine ( $R_S$ )-**14**, producing 2-chloro-2-arylaziridines via an aza-Darzens reaction [73]. Importantly, the structure of the final aziridine is determined by the silyl group, and the order of the addition of HMPA and imine **14** in the multicomponent coupling. When the bulky TBS group was used, and HMPA was added to the reaction mixture before the imine **14**, aziridines **36** were formed. The addition of lithium metaloenamine took place through an open transition state to the *Re* face of the imine ( $R_S$ )-**14**, followed by an intramolecular nucleophilic substitution to form the aziridines ring. On the contrary, aziridines **32**



**Scheme 11:** Synthesis of *cis*-vinylaziridines **30** and **31** from aldimines **14** and bromomethylbutenolide **29**.

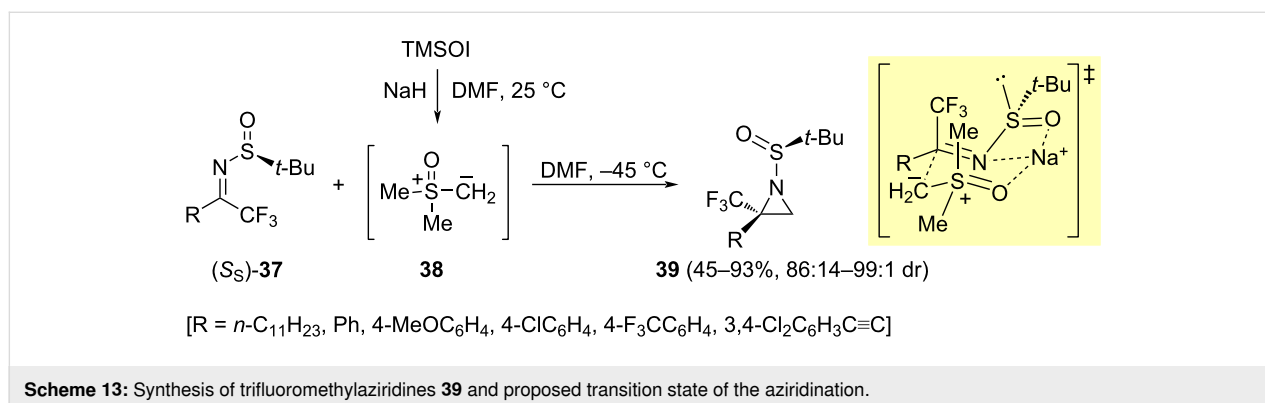
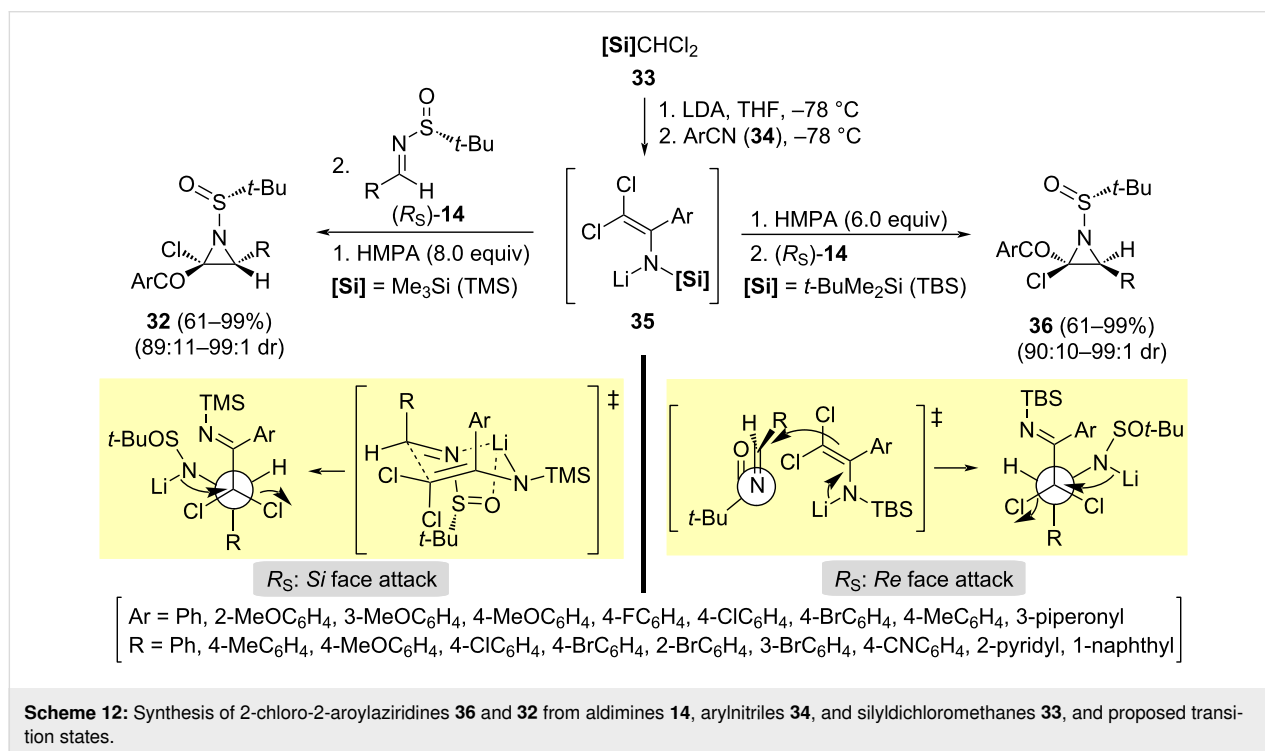


were obtained starting from dichloromethyltrimethylsilane (**33**, [Si] = TMS), and adding the chiral imine before HMPA to the reaction mixture. The nucleophilic addition of metaloenamine occurred through a cyclic transition state to the *Si* face of the imine ( $R_S$ )-**14**. In this way, both *cis*-aziridines diastereoisomers **36** and **32** were formed from the same chiral imine **14** and aryl nitriles **34** (Scheme 12).

Chiral sulfinyl imines have been also used in the stereoselective synthesis of aziridines. The reaction of *N*-*tert*-butanesulfinyl trifluoromethyl ketimines ( $S_S$ )-**37** with dimethylsulfoxonium methylide **38** gave trifluoromethylated aziridines **39** in moderate to excellent yields (45–93%), and good diastereoselectivities (86:14 to >99:1 dr). The absolute configuration of compounds **39** was determined by X-ray crystallographic analy-

sis, and it was found that the configuration of the newly generated stereocenter was *R*. Huang and co-workers proposed a cyclic transition state determined by the coordination of the nitrogen of the imine, and the oxygen atoms of the sulfoxonium and sulfinyl units to the sodium cation, which is present in the reaction medium [74]. The trifluoromethyl group occupies an equatorial position to avoid electrostatic repulsion with the lone pair of electrons of the sulfinyl group. The nucleophilic attack took place to the *Si* face of the ketimine **37** (Scheme 13).

Recently, Yang and co-workers described the diastereoselective synthesis of aziridines **42** in one-step using the Cu(I)/*L*-proline complex as a catalyst and *N*-*tert*-butanesulfinylamide in an aminotrifluoromethylation reaction of alkenes. All the aziridines **42** were obtained with high diastereoselectivity



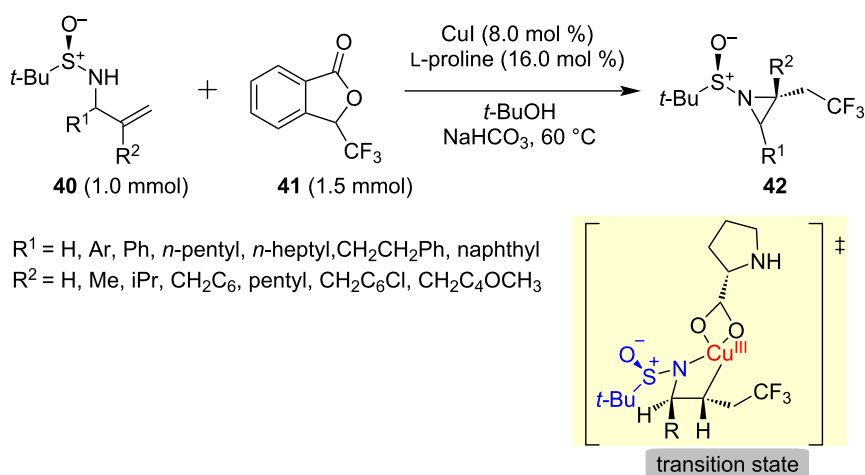
(dr > 25:1) and good yields (56–98%) from allylic sulfonamides **40** and Togni's reagent (**41**). The reaction mechanism is proposed based on DFT calculations. In this study, they observed an intramolecularly intermediate Cu(III) species, and the sulfinamide acts as a direction group and nucleophile [75] (Scheme 14).

### Asymmetric synthesis of azetidines

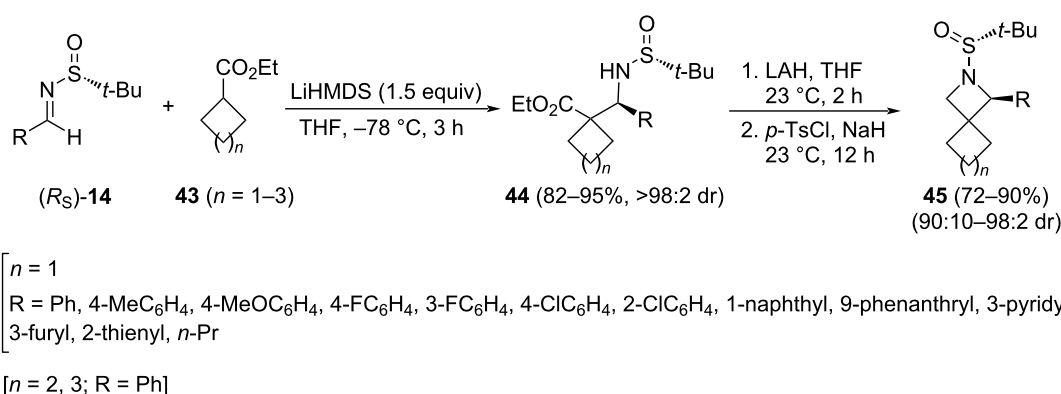
Azetidine [76,77] has attracted less attention than aziridines, pyrrolidines and piperidines, among small and medium size aza-heterocycles, because there are no general methods for their preparation. However, four-membered nitrogen-containing heterocycles have recently found applicability in pharmacy as highly biological active compounds. Among this group of nitrogenated heterocycles,  $\beta$ -lactams (azetid-2-ones) have reached especial attention [78-80], being easily accessible from  $\beta$ -aminoesters.

A stereoselective synthesis of 1-substituted 2-azaspiro[3.3]heptanes **45** ( $n = 1$ ) was reported by the group of Reddy [81] starting from ethyl cyclobutanecarboxylate **43** and chiral *N*-tert-butanesulfinyl aldimines ( $R_S$ )-**14**. In this three-step procedure, a highly diastereoselective addition of the ethyl cyclobutanecarboxylate anion occurred first, followed by the reduction of the ester group, and an intramolecular nucleophilic substitution of the tosylate of the resulting primary alcohol (Scheme 15). This methodology was applicable to the synthesis of 1-phenyl-2-azaspiro[3.4]octane (**45**,  $n = 2$ , R = Ph) and 1-phenyl-2-azaspiro[3.5]nonane (**45**,  $n = 3$ , R = Ph). The structure and absolute stereochemistry of these compounds were assigned based on the single-crystal X-ray diffraction analysis of azaspiroheptane **45** with a 9-fenanthryl substituent at 1-position.

The same three-step procedure was applied by Reddy and co-workers to synthesize 1-substituted 2,6-diazaspiro[3.3]hep-



**Scheme 14:** Synthesis of aziridines **42** and proposed state transition.



**Scheme 15:** Synthesis of 1-substituted 2-azaspiro[3.3]heptanes, 1-phenyl-2-azaspiro[3.4]octane and 1-phenyl-2-azaspiro[3.5]nonane **45** from chiral imines **14**, and ethyl cycloalkancarboxylates **43**.

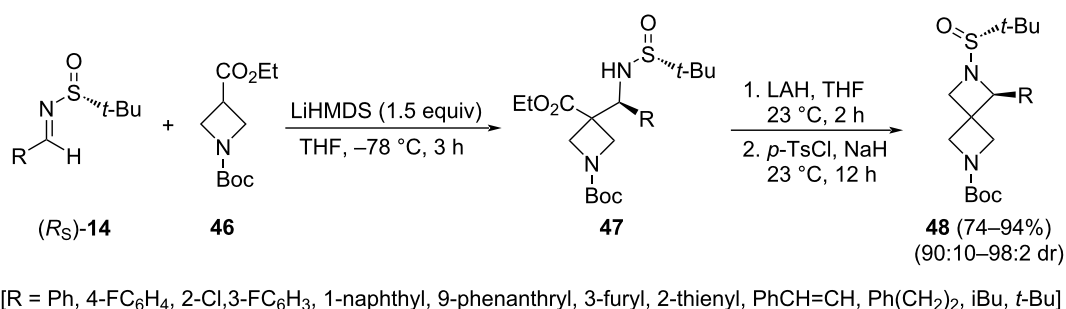
anes **48**, starting in this case from 1-Boc-azetidine-3-carboxylate **46**, instead of ethyl cyclobutanecarboxylate **43** [82]. This structural motif was found to have similar physicochemical properties as 2-substituted piperazines, which are key intermediates in drug discovery. The applied protocol was found to be practical for the asymmetric synthesis of a variety of aromatic, heteroaromatic, and aliphatic 1-substituted 2,6-diazaspiro[3.3]heptanes **48**, with overall yields ranging from 74 to 94% (Scheme 16).

### Asymmetric synthesis of $\beta$ -lactams

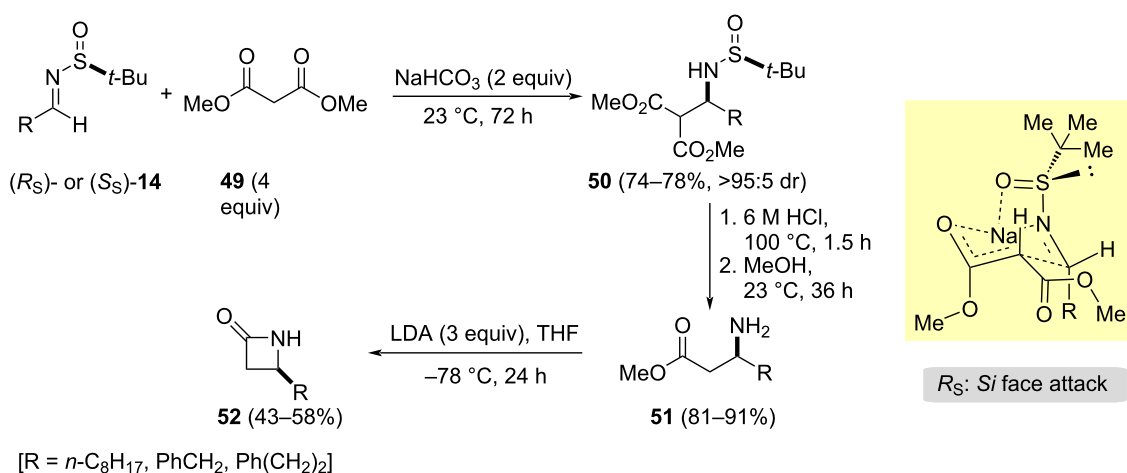
$\beta$ -Lactam antibiotics are important drug class of antibacterial agents [83] and in the literature, several strategies are reported using the sulfinyl group [84–88]. In this context, an enantioselective synthesis of 4-substituted azetidin-2-ones **52** was also accomplished starting from chiral imines **14** and dimethyl malonate (**49**). A diastereoselective coupling of these components under solvent-free conditions was carried out, using sodium carbonate as base promoter. The resulting dimethyl 2-(1-aminoalkyl)malonates **50** were obtained in moderate to good

yields as single diastereoisomers in all cases except for aromatic aldimines. Compounds **50** could be easily transformed successively into  $\beta$ -amino esters **51**, and the corresponding  $\beta$ -lactams **52** with high optical purity (Scheme 17) [89]. The absolute configurations of compounds **52** were obtained by the comparison of the signs of specific rotation of **52** with  $R = \text{Ph}(\text{CH}_2)_2$ , with that of known (*R*)-4-(2-phenylethyl)azetidin-2-one. A 6/4-fused bicyclic transition state model was proposed to rationalize the stereochemical outcome, in which the sodium metal is chelated by the oxygen and the nitrogen atoms of sulfinyl imine (Scheme 17), occurring the nucleophilic attack to the *Si* face of the imines with  $R_S$  configuration.

Su and Xu reported the stereoselective synthesis of spiro  $\beta$ -lactam **57** from chiral ( $R_S$ )-*N*-*tert*-butanesulfinyl isatin ketimine **53** ( $R^1 = \text{H}$ ), with a bulky trityl protecting group bonded to the nitrogen indolic atom ( $\text{Tr} = \text{triphenylmethyl}$ ), and ethyl bromoacetate. The Zn/Cu-mediated Reformatsky-type reaction furnished enantiomerically pure compound **54** after column chromatographic purification. Selective desulfinylation



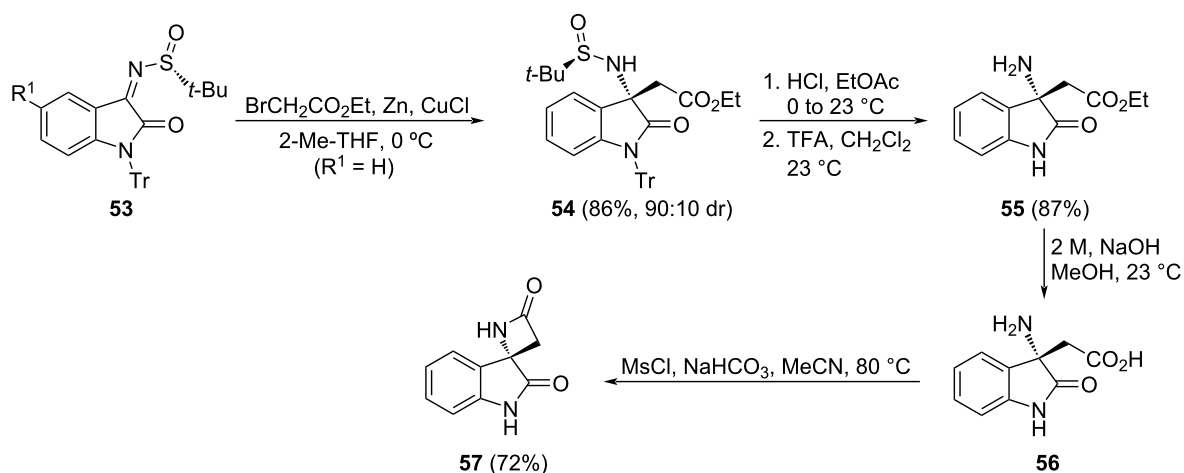
**Scheme 16:** Synthesis of 1-substituted 2,6-diazaspiro[3.3]heptanes **48** from chiral imines **14** and 1-Boc-azetidine-3-carboxylate (**46**).



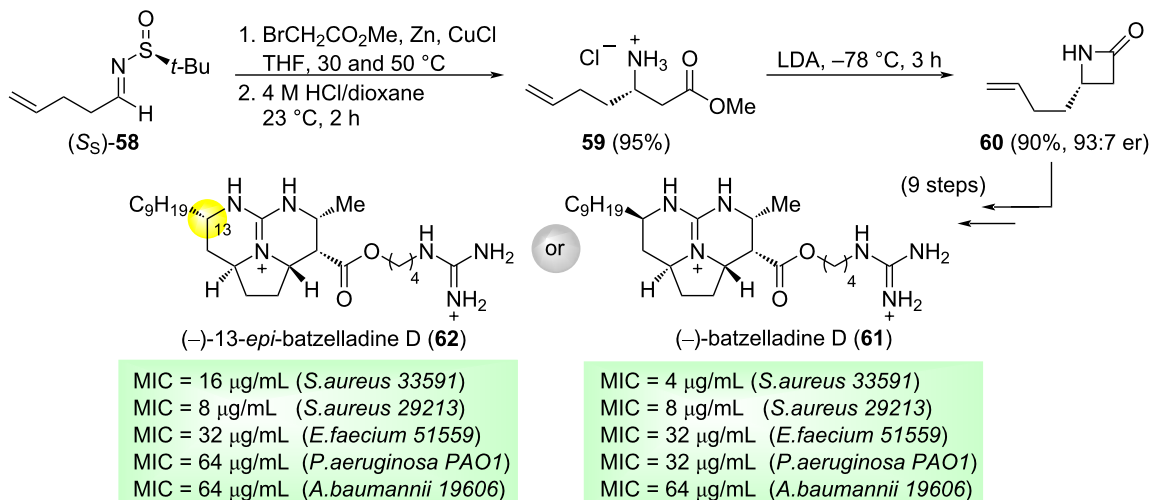
**Scheme 17:** Synthesis of  $\beta$ -lactams **52** from chiral imines **14** and dimethyl malonate (**49**).

of **54** was carried out by using 1.0 M HCl in EtOAc, and further removal of the Tr group by the employment of TFA in dichloromethane afforded the 2-oxindolinyl amino ester derivative **55**, the key intermediate for the synthesis of (–)-AG-041R, a gastrin/cholecystokinin-B receptor antagonist. In addition, amino ester **55** was transformed into the fully unprotected amino acid **56** under basic conditions in MeOH, and after that, further treatment with MsCl and NaHCO<sub>3</sub> in MeCN at 80 °C led to spiro-β-lactam derivative **57** in 72% combined yield (Scheme 18) [88].

In 2020, Pierce and co-workers developed a method for the synthesis of guanidinium alkaloid batzelladine D in enantiomeric and racemic form, along with a series of stereochemical analogues. The batzelladines are a family of polycyclic guanidinium alkaloids that were isolated in the mid-1990s from the Caribbean sponge *bataella* sp. From a biological point of view, the batzelladines have received attention due to their reported activity as inhibitors of HIV gp120-human CD4 binding. Chiral *N*-*tert*-butanesulfinyl aldimine (*S*<sub>S</sub>)-**58** was used as a precursor in the synthesis of (–)-batzelladine D **61** and (–)-13-*epi*-batzelladine D **62**. The reaction of (*S*<sub>S</sub>)-**58** with methyl bromoacetate in the presence on Zn and CuCl in THF, left, after removal of the sulfinyl group under acidic conditions, to β-amino ester ammonium chloride **59** in high yield. This compound was transformed into β-lactam **60** in 90% yield by treatment with LDA in THF at –78 °C [90]. Compound **60** was converted after 9 steps in target batzelladines D **61** and **62** (Scheme 19). The authors explored also the antimicrobial activity of these compounds against a series of pathogens with starting promising results.



**Scheme 18:** Synthesis of spiro-β-lactam **57** from chiral (*R*<sub>S</sub>)-*N*-*tert*-butanesulfinyl isatin ketimine **53** and ethyl bromoacetate.



**Scheme 19:** Synthesis of β-lactam **60**, a precursor of (–)-batzelladine D (**61**) and (–)-13-*epi*-batzelladine D (**62**) from chiral (*S*<sub>S</sub>)-*N*-*tert*-butanesulfinyl imine **58**, and antimicrobial evaluation of MIC values for these compounds.

## Asymmetric synthesis of pyrrolidines

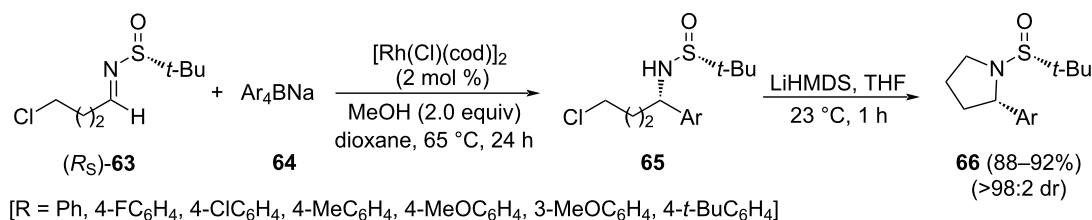
The pyrrolidine ring is more represented within natural products than the 3- and 4-membered nitrogen-containing heterocycles. This molecular array is also found in drugs and other biologically active molecules. For this reason, there are numerous examples of synthetic methodologies for these compounds in the literature. In most cases, the pyrrolidine ring is formed from an amine with a hydrocarbon chain that also carries a functional group at the appropriate distance that allows the cyclization process to take place. In the case of substituted pyrrolidines, the stereoselective synthesis is especially interesting, highlighting the 1,3-dipolar cycloaddition reactions with azomethine ylides as an example of transformation that take place with great stereocontrol, and allow the synthesis of polyfunctionalized pyrrolidines in a single reaction step [91,92]. On the other hand, natural amino acids proline and hydroxyproline are functionalized pyrrolidines that have found great application in organic synthesis as chiral organocatalysts in stereoselective processes [93,94].

### Cyclizations involving a position in the starting chiral imine

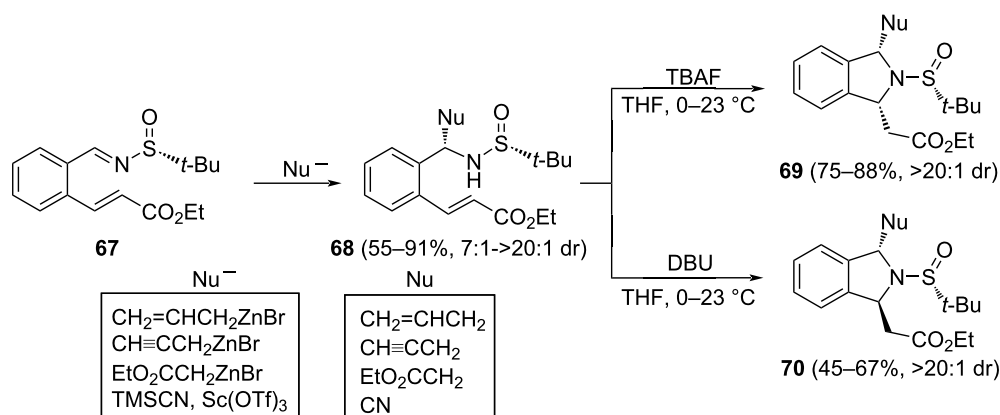
Arylation of chiral sulfinyl imines with sodium tetraarylboronates **64** was found to proceed with high diastereoselectivity under rhodium catalysis. Reddy and co-workers applied this

methodology to the synthesis of 2-substituted pyrrolidines **66** [95]. The arylation of chlorinated imine (*R<sub>S</sub>*)-**63** was performed with 2 mol % of an air-stable rhodium catalyst in dioxane, in the presence of 2 equivalents of MeOH, at 65 °C, leading to compounds **65** with high diastereomeric ratio. Crude amides **65** were converted into the corresponding pyrrolidines **66** in high yield by stirring at room temperature for 1 h in presence of 2.0 equivalents of LiHMDS (Scheme 20). It is important to highlight that cyclization occurred without epimerization with such as strong base.

Isindolines with substituents at 1 and 3 positions were synthesized from an aromatic *N-tert*-butanesulfinyl imine **67**, bearing a Michael acceptor in the *ortho*-position. Fustero, Barrio and co-workers found that combining an asymmetric nucleophilic addition to the chiral imine, with an intramolecular conjugate aza-Michael reaction, the expected 1,3-disubstituted isindolines were produced with high diastereoselectivity [96]. Importantly, depending on the base involved in the intramolecular aza-Michael reaction, it was possible to reach either *cis*- or *trans*-isindolines, **69** and **70**, respectively, from the same precursor **68**. The authors proposed that the thermodynamically more stable *cis*-isomer **69** is formed when TBAF was used. Meanwhile, working under kinetic conditions (DBU as base), *trans*-isomer **70** was obtained (Scheme 21).



**Scheme 20:** Rhodium-catalyzed asymmetric synthesis of 3-substituted pyrrolidines **66** from chiral imine (*R<sub>S</sub>*)-**63** and sodium tetraarylboronates **64**.



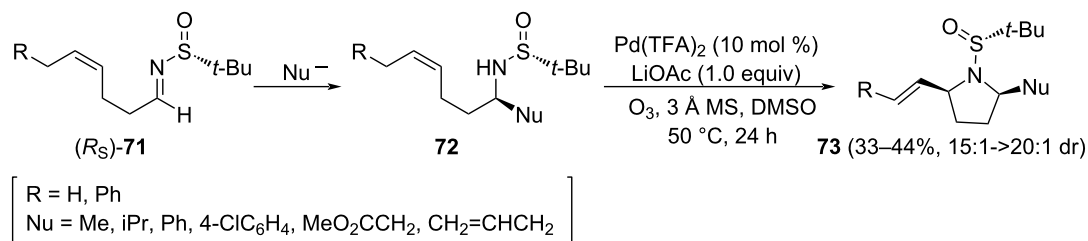
**Scheme 21:** Asymmetric synthesis of 1,3-disubstituted isindolines **69** and **70** from chiral imine **67**.

The group of Stahl provided an elegant approach to the synthesis of 2,5-disubstituted pyrrolidines **73**, from alkenyl sulfonamides **72** [97]. These substrates were prepared from chiral *N*-*tert*-butanesulfinyl imine (*R<sub>S</sub>*)-**71**. Nucleophilic additions to this imine took place with high diastereoselectivity to the *Si* face of the iminic carbon. After that, the combination of Pd(TFA)<sub>2</sub>, lithium acetate and DMSO as solvent led to optimal results in the oxidative cyclization process to produce pyrrolidines **73** as a single diastereoisomer, with relative *cis*-configuration (Scheme 22). This has been the first reported nucleophilic attack of the sulfinyl group to a  $\pi$ -allylpalladium intermediate.

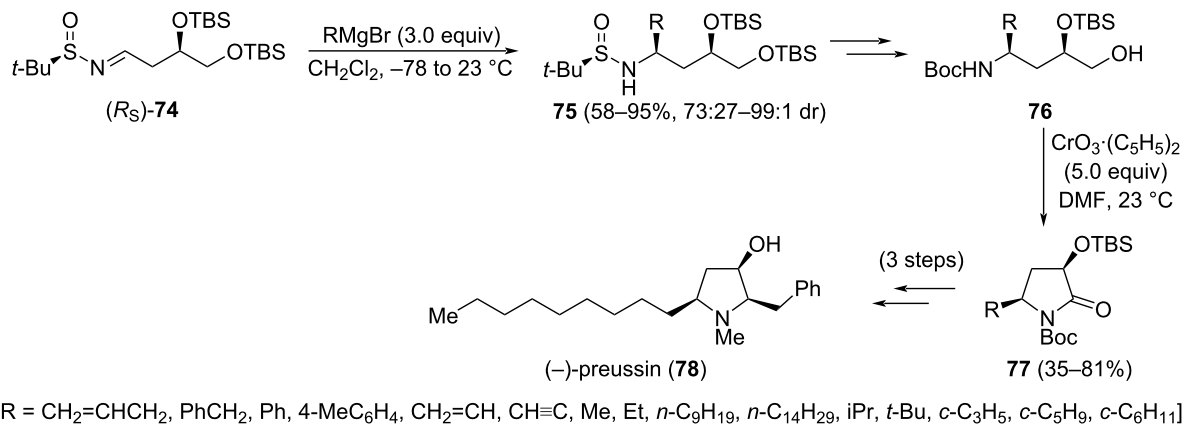
The asymmetric synthesis of 3-hydroxy-5-substituted pyrrolidin-2-ones **77**, with relative *cis*-configuration, was reported by the group of Huang and Wei [98]. A diastereoselective addition of Grignard reagents to chiral aldimine (*R<sub>S</sub>*)-**74**, and an intramolecular oxidative cyclization of aminoalcohols derivatives **76**, are key steps of this approach. Both diastereoisomers of aldimines **74** (*R<sub>S</sub>* and *S<sub>S</sub>*) were prepared from *D*-malic acid and the corresponding enantiomer of *tert*-butanesulfinamide. Importantly, the choice of the solvent was crucial for obtaining high diastereoselectivities in the Grignard addition step, in which dichloromethane was performing better than THF. On the other

hand, diastereoselectivities for addition products **75** were higher working with (*R<sub>S</sub>*)-**74** than its (*S<sub>S</sub>*) diastereoisomer, indicating a mismatch between the chiral auxiliary and the stereocenter in this substrate. Concerning the oxidative cyclization reaction, pyridinium dichromate (PDC) provided low yields of expected lactam **77**. Many oxidants were checked for this transformation to take place, and the Sarett reagent [CrO<sub>3</sub>·(C<sub>5</sub>H<sub>5</sub>)<sub>2</sub>] in DMF was the best to produce lactams **77**. The synthetic interest of these functionalized lactams was demonstrated in the synthesis of alkaloid (–)-preussin (**78**) from the appropriate precursor **77** in three steps (Scheme 23).

Similar strategies were employed to obtain similar heterocycles [99]. Other regio- and stereoisomeric 2-pyrrolidones **80** were also prepared by a stereoselective tandem Barbier process of **79** with alkyl and aryl bromide [100]. The process was performed in this case under Barbier reaction conditions. Namely, the formation of the nucleophile (organomagnesium reagent) is carried out in the presence of the electrophile (chiral imine **79**). Surprisingly, both diastereomeric aldimines **79** (*R<sub>S</sub>* and *S<sub>S</sub>*) gave similar results concerning the stereochemical outcome, suggesting that the chiral sulfonamide moiety was not involved in the stereocontrol during this tandem Barbier addition process. After addition of the organomagnesium reagent to the imine **79**, cycli-



**Scheme 22:** Asymmetric synthesis of *cis*-2,5-disubstituted pyrrolidines **73** from chiral imine (*R<sub>S</sub>*)-**71**.



**Scheme 23:** Asymmetric synthesis of 3-hydroxy-5-substituted pyrrolidin-2-ones **77** from chiral imine (*R<sub>S</sub>*)-**74**.

zation involving the magnesium amide and the ester occurred without the need of an extra cyclization step to give, after *N*-Boc protection, 4-hydroxy-5-substituted pyrrolidin-2-ones **90**, with relative *trans*-configuration (Scheme 24).

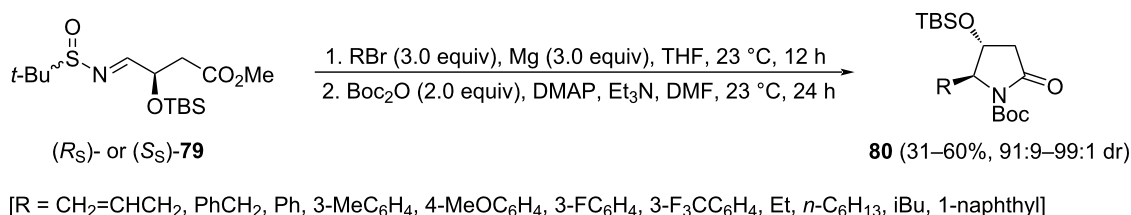
### Cyclizations involving a position in the attacking nucleophile

The reaction of ethyl 4-bromocrotonate (**81**) with LDA at  $-78\text{ }^{\circ}\text{C}$  and subsequent addition of chiral imines **14** afforded 3-pyrrolines **82** with high diastereoselectivity. Chogii and Njardarson proposed that after deprotonation of **81**, the resulting dienolate reacted at  $\alpha$ -position with the chiral imine **14**. The addition was highly diastereoselective, being the configuration of the newly created stereogenic center dependent on the configuration of the sulfur atom of the starting imine **14**. After nucleophilic addition, and subsequent elimination, 3-pyrrolines **82** were formed as single diastereomers [101]. The whole process could be considered a [3 + 2] annulation, and aziridines were not observed as competing reaction products (see above Scheme 11). In addition, hindered imines, ethers, sulfonates, heteroaryl substituents, and conjugated imines were all well tolerated (Scheme 25).

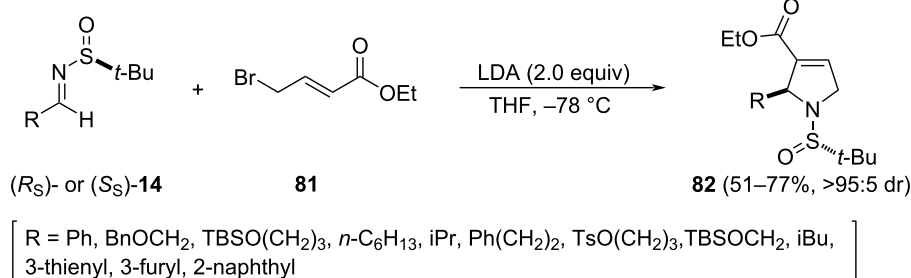
A  $\text{SmI}_2$ -mediated coupling of allenolate **83** with chiral ( $R_S$ )-**14** provided  $\gamma$ -amino ester derivatives **84** in good yields and moderate diastereomeric ratios. Huang and Py found that the better yields and diastereoselectivities were found working in THF as solvent in the presence of *t*-BuOH and LiBr as additives [102].

The isopropyl-substituted derivative **84** was easily converted into the corresponding methylene lactam **85**, upon removal of the sulfinyl unit under acidic conditions. Finally, ozonolysis of **85** yielded tetramic acid **86** in 60% yield (Scheme 26). The configuration of the newly generated chiral center in compounds **84** was assigned from the sign of the optical rotation of enantio-enriched tetramic acid **86**, which was previously characterized. Based on this, it can be stated that addition of the allenolate **83** takes place mainly to the *Re* face of imines with ( $R_S$ )-configuration.

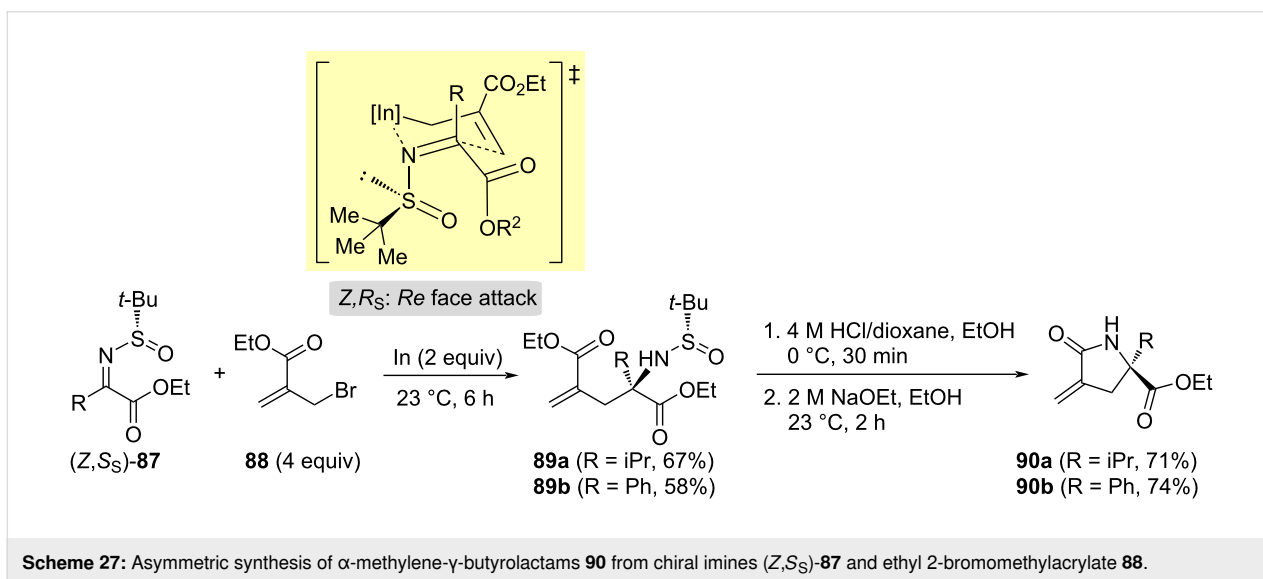
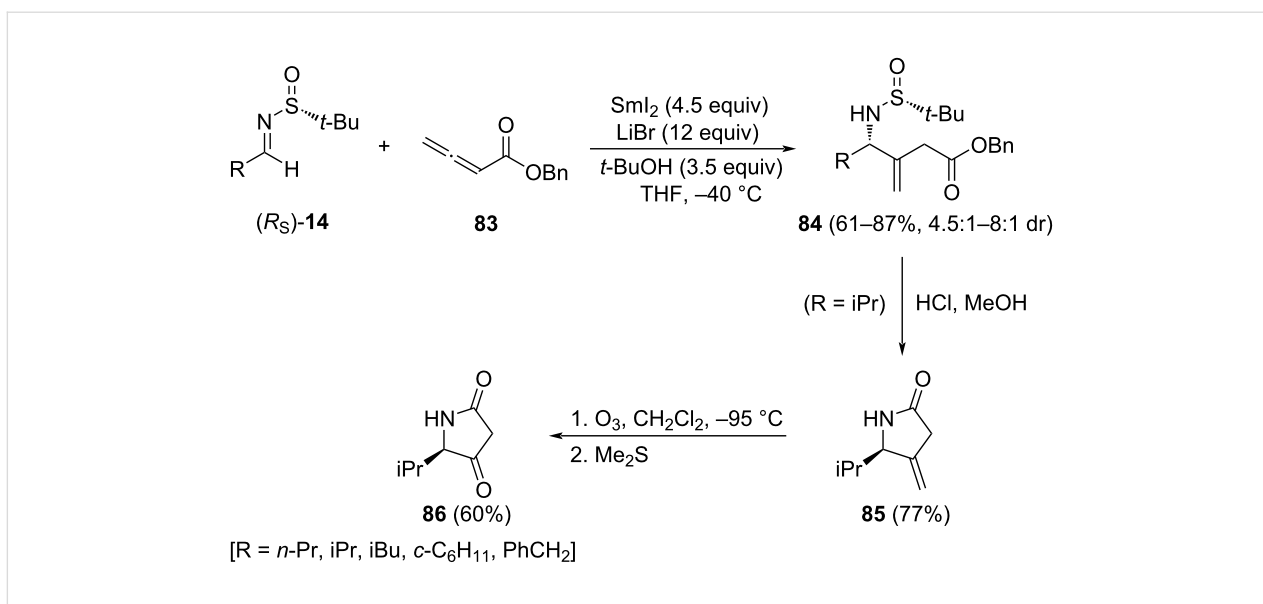
Excellent diastereoselectivities were also achieved in the indium-mediated allylation of chiral *tert*-butanesulfinyl glyoxylate imine derivatives **87** with ethyl 2-bromomethylacrylate (**88**). Working at room temperature without any additional solvent provided the highest yields in these coupling reactions, amino diesters **89** being isolated as single diastereoisomers (Scheme 27). Removal of the sulfinyl group under acidic conditions, and further treatment of the resulting ammonium salts with sodium ethoxide, yielded  $\alpha$ -methylene- $\gamma$ -butyrolactams **90**, in a one-pot, two-step process [103]. A six-membered chair-like transition state model with the indium coordinated to the nitrogen atom of the imine, and the sulfinyl and R groups located at axial positions, in a kind of *s-cis* conformation, was proposed to rationalize the stereochemical outcome. By considering this working model, the nucleophilic attack took place to the *Re* face of imines with (*Z,S*<sub>S</sub>)-configuration (Scheme 27).



**Scheme 24:** Asymmetric synthesis of 4-hydroxy-5-substituted pyrrolidin-2-ones **80** from chiral imines **79**.



**Scheme 25:** Asymmetric synthesis of 3-pyrrolines **82** from chiral imines **14** and ethyl 4-bromocrotonate (**81**).

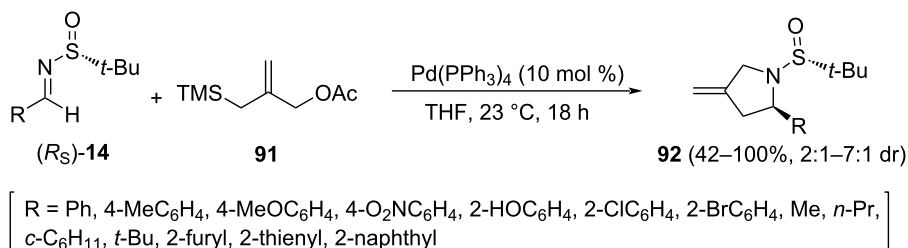


The cycloaddition of chiral sulfinyl imines ( $R_S$ )-**14** with 2-(trimethylsilylmethyl)allyl acetate (**91**) could also be promoted by Pd(0) to give methylenepyrrolidines **92**. The group of Stockman demonstrated that Pd(PPh<sub>3</sub>)<sub>4</sub> was the best source of Pd(0) and that the reaction worked well in different solvents, with dry THF giving the best diastereoselectivities and good yields at room temperature [104]. The configuration of the major diastereoisomer was assigned by X-ray crystallographic analysis. From this, authors rationalized the stereochemical outcome of the cyclization considering that the stereoinduction is derived from the dipole–dipole repulsion of the sulfinyl imine, which places the *tert*-butyl group on the *Si* face, and thus the cycloaddition occurs from the less sterically hindered *Re*

face. The cyclization process worked also in *tert*-butanesulfinyl ketimines, but yields and diastereoselectivities were significantly lower (Scheme 28).

Recently, a series of alkaloids like dibenzoazaspirodecanes **97** have been synthesized by addition of 2-bromobenzylmagnesium bromide (**94**) to chiral *N-tert*-butanesulfinyl imines **93**. These reactions proceeded with high levels of diastereocontrol. The resulting sulfonamide derivatives **95** were transformed into the target spiro compound **97** by performing successive desulfinylation and intramolecular palladium-catalyzed *N*-arylation. To rationalize the stereochemical course of the addition, DFT calculations were performed and they predicted correctly



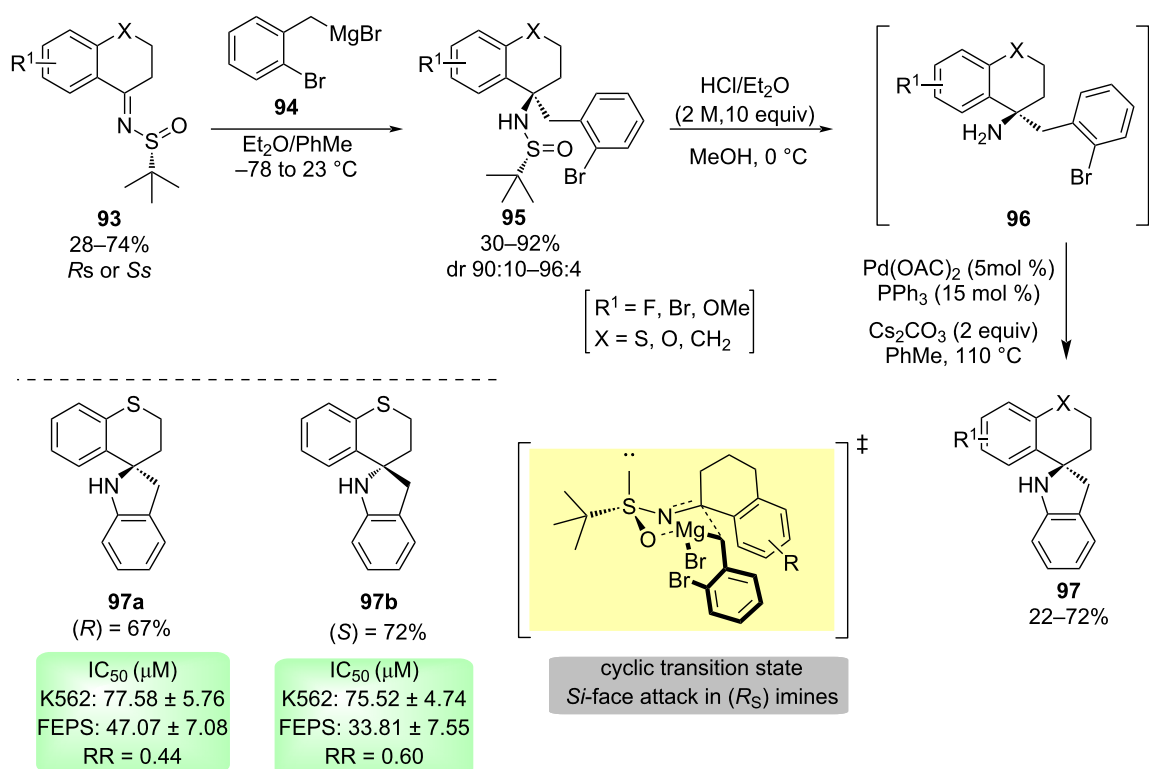


**Scheme 28:** Asymmetric synthesis of methylenepyrrolidines **92** from chiral imines  $(R_S)\text{-14}$  and 2-(trimethylsilylmethyl)allyl acetate **91**.

the observed experimental results considering a six-membered ring cyclic transition state. The addition took place to the *Si*-face of the imines with  $(R_S)$ -configuration. Compounds **97** were also evaluated in a preliminary study in leukemia strains (Scheme 29) [105].

Li and Xu reported a method for the enantioselective synthesis of  $\beta,\gamma$ -unsaturated  $\alpha$ -amino acids **100**, by a Lewis acid-promoted diastereoselective Petasis reaction of vinylboronic acids **98**,  $(R)$ -*N*-*tert*-butanesulfinamide and glyoxylic acid (**99**). They found that the best results were obtained working with  $\text{InBr}_3$  as Lewis acid, in dichloromethane at room temperature [106]. Under these reaction conditions, sulfinyl imine is formed first along with the boronate by interaction of the correspond-

ing vinylboronic acid with the carboxylic group of the imino acid intermediate. The transfer of the vinyl unit to the electrophilic iminic carbon took place in a quite rigid system, with chelation of the Lewis acid with the nitrogen of the imine and carboxylate oxygen, forming a five-membered ring. The migration of the vinyl group occurred to the *Re* face of the imine, which is less shielded than the *Si* face, because of the influence of the bulky *tert*-butyl in the most stable conformation of the sulfinyl imine (Scheme 30). The authors also demonstrated the synthetic utility of compounds **100**. Their reaction with thionyl chloride in methanol produced removal of the sulfinyl group and formation of the corresponding methyl ester, to give compounds **101**. Subsequent, reductive amination with 3-phenyl-2-propynal led to reaction intermediates **102**, which under



**Scheme 29:** Synthesis of dibenzoazaspirodecanes from cyclic *N*-*tert*-butanesulfinyl imines.

typical Pauson–Khand reaction conditions gave cyclopenta[*c*]proline derivatives **103** in moderate yields, with high diastereoselectivities (Scheme 30).

### Asymmetric synthesis of piperidines

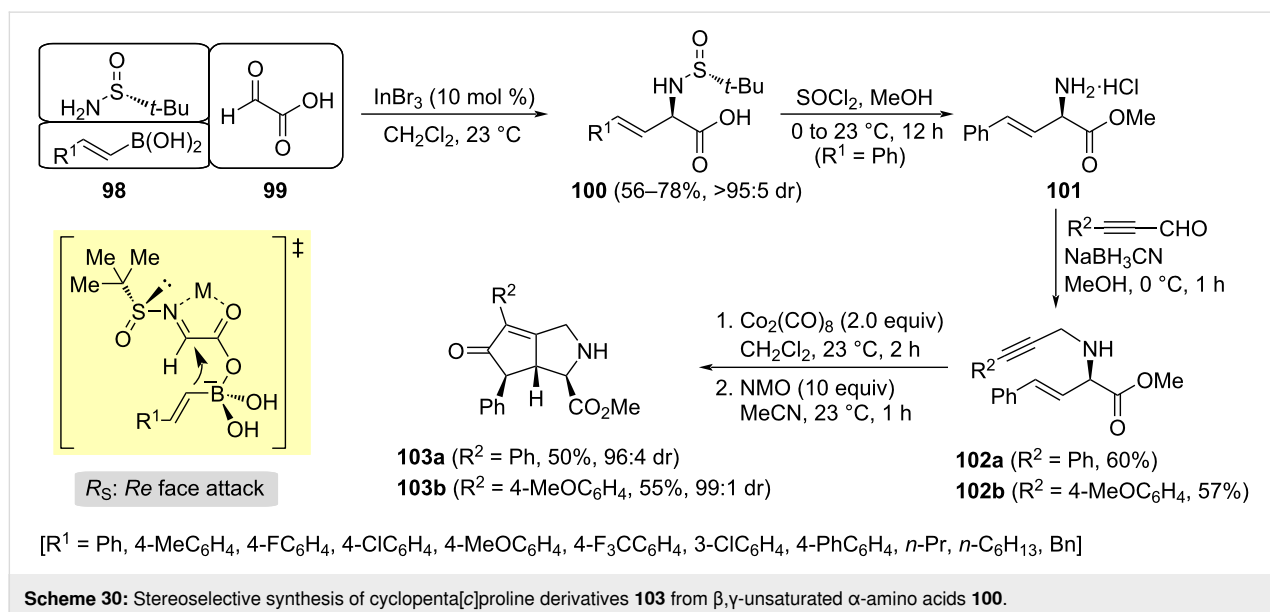
The six-membered nitrogen-containing rings are the most common heterocycles among natural products and also synthetic pharmaceutical drugs [107,108]. For this reason, the piperidine unit has attracted great attention among organic chemists [109]. Due to that, a large number of classical methodologies have been used for their synthesis, in which the key step is the generation of the six-membered ring, including the aldol reaction, the reductive amination, Mannich reaction, ring closing metathesis, Diels–Alder reaction with imines as dienophiles, aza-Prins cyclization, and intramolecular Michael reaction, among others [110–115]. Despite the considerable effort made to date in this field, the development of new methodologies that allow accessing these heterocycles in a stereoselective way, and taking into account environmental considerations as one of the most important points, continue to be of great interest [109,116–118].

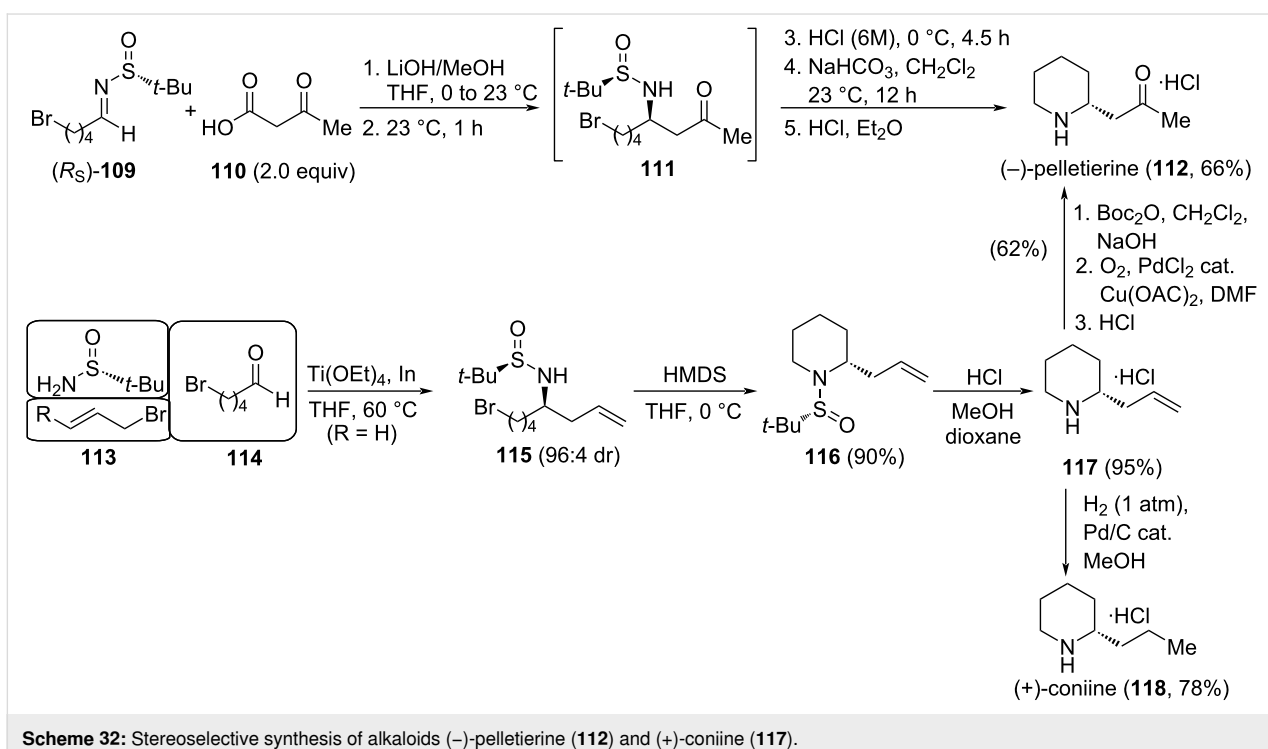
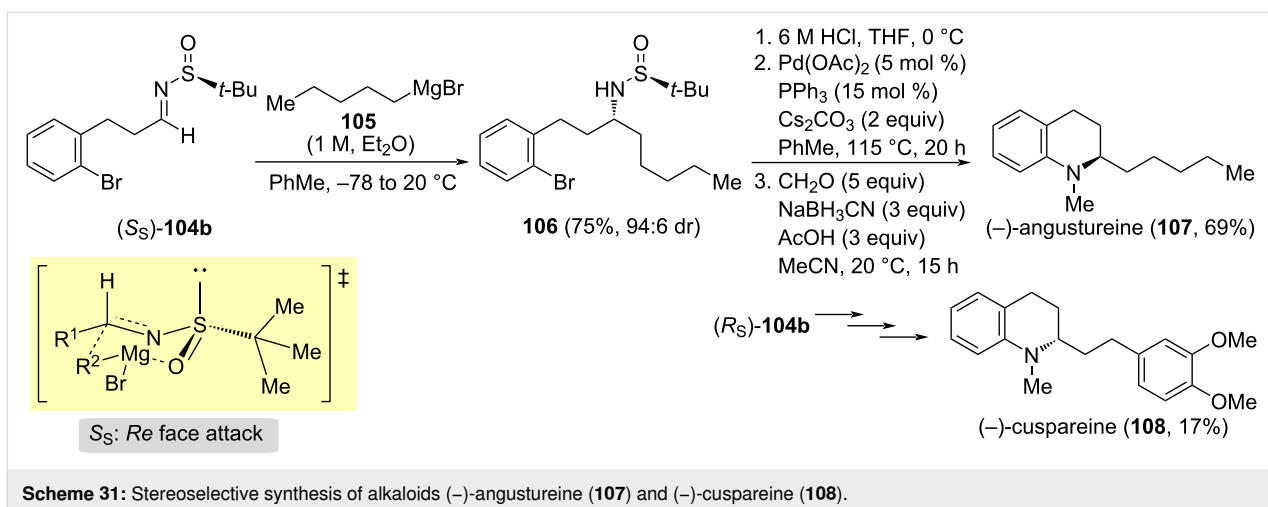
### Cyclizations involving a position in the starting chiral imine

Enantiomeric *N*-*tert*-butanesulfinyl imines **104b** derived from 3-(2-bromophenyl)propanal have been used as reaction intermediates in the synthesis of tetrahydroquinoline alkaloids (–)-angustureine (**107**) and (–)-cuspareine (**108**) reported by Sirvent et al. [119]. The diastereoselective addition of a Grignard reagent was a key step in this methodology. The addition proceeded with high diastereoselectivity in toluene, and the attack of the Grignard reagent occurred on the *Re* face of the

imine with *S* configuration at the sulfur atom, through a chelated transition state. The reaction of chiral aldimine (*S<sub>S</sub>*)-**104b** with pentylmagnesium bromide gave compound **106** in 75% yield. Further successive *N*-desulfinylation, intramolecular palladium-catalyzed *N*-arylation, and final *N*-methylation led to (–)-angustureine (**107**) in high overall yield (Scheme 31). The same methodology was applied to the synthesis of (–)-cuspareine (**108**), starting in this case from enantiomeric imine (*R<sub>S</sub>*)-**104b**, and using 2-(3,4-dimethoxyphenyl)ethylmagnesium bromide as Grignard reagent.

A straightforward synthesis of the alkaloid (–)-pelletierine (**112**) was accomplished by the diastereoselective coupling of 3-oxobutanoic acid (**110**) and the *N*-*tert*-butanesulfinyl imine (*R<sub>S</sub>*)-**109** derived from 5-bromopentanal (**114**). The base-promoted decarboxylative-Mannich coupling of these reagents led to β-amino ketone derivative **111**, which was not isolated. After removal of the sulfinyl group under acidic conditions, and intramolecular *N*-alkylation upon treatment with sodium bicarbonate, (–)-pelletierine (**112**) was formed, and easily isolated as its hydrochloride derivative (Scheme 32) [120]. Compound **112** is a key intermediate in the biomimetic synthesis of natural alkaloids. Interestingly, amino allylation of 5-bromopentanal (**114**) with (*R*)-*tert*-butanesulfinamide and allyl bromide (**113**, R = H) in the presence of indium metal gave homoallylamine derivative **115**. In this transformation, imine (*R<sub>S</sub>*)-**109** is a reaction intermediate that was not isolated. Treatment of **115** with potassium hexamethyldisilazide (KHMDs) led to the sulfinyl piperidine derivative **116**, and final deprotection under acidic conditions produced enantioenriched 2-allylpiperidine (**117**) as its hydrochloride (Scheme 32) [121]. Compound **117** has been also an advanced intermediate in the synthesis of alkaloids. For



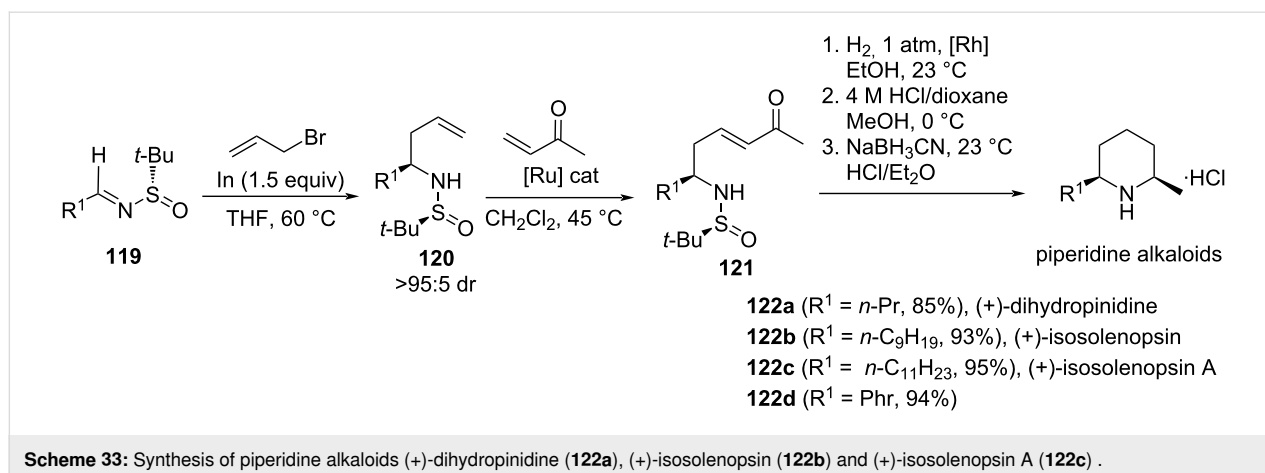


instance, hydrogenation of **117** yielded (+)-coniine (**118**), the major alkaloid extracted from poison hemlock and responsible for its toxicity, as its hydrochloride, and *N*-Boc protected derivative of **117** submitted to Wacker oxidation led to (-)-pelletierine (**112**) alkaloid.

Yus and co-workers developed the synthesis of the piperidine alkaloids (+)-dihydropinidine (**122a**), (+)-isosolenopsin (**122b**), (+)-isosolenopsin A (**122c**) and (2*R*,6*R*)-6-methylpipercolic acid hydrochloride by oxidation of the aromatic ring of (2*R*,6*R*)-2-methyl-6-phenylpiperidine (**122d**). The diastereoselective allylation of (*S*<sub>S</sub>)-*N*-*tert*-butanesulfinyl imines **119** mediated by

indium metal under Barbier's reaction connections (formation of the allylindium intermediate in the presence of the imine electrophile) is the key step in these syntheses (*Re*-face attack). The natural products were obtained after four additional steps: cross-metathesis of allylated compounds **120** with methyl vinyl ketone, reduction of conjugated C=C double bond, removal of the sulfinyl group under acidic conditions, and final stereoselective reduction of the imine formed by intramolecular cyclization (Scheme 33) [122].

The group of Prasad reported the diastereoselective synthesis of β-amino ketone derivatives from *N*-*tert*-butanesulfinyl imines

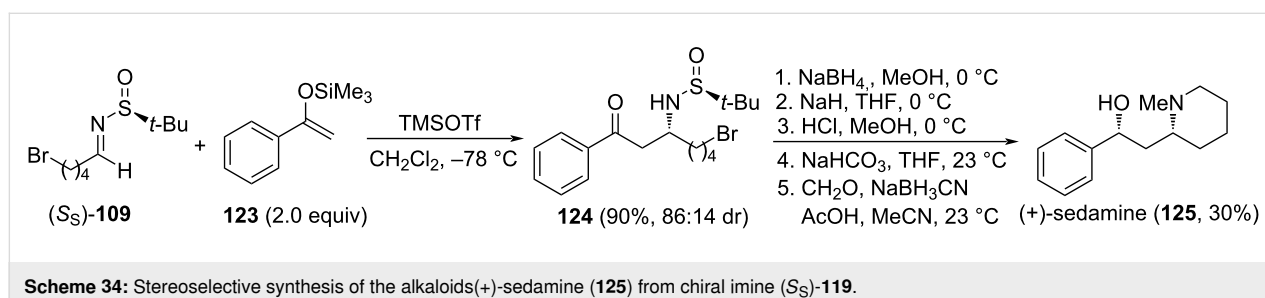


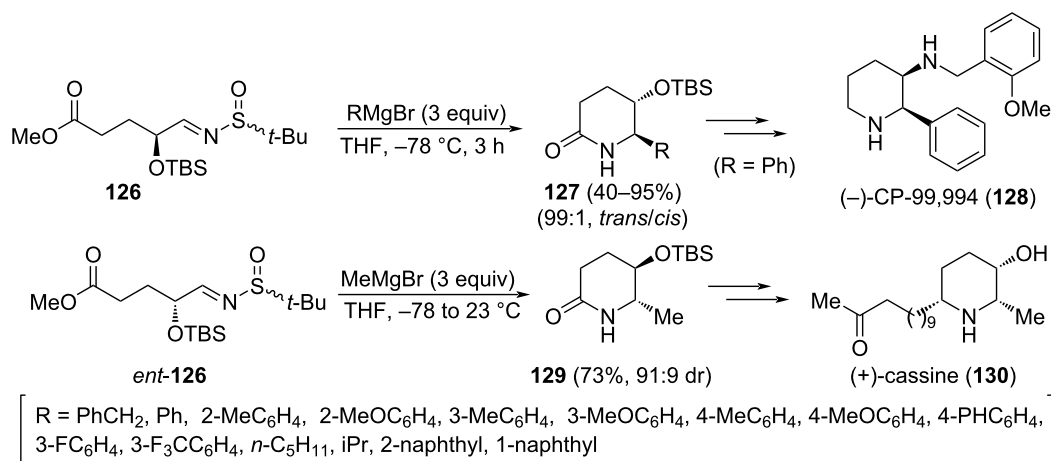
and silyl enol ethers of aryl methyl ketone [123]. The synthetic interest in  $\beta$ -amino ketones was exemplified in the synthesis of alkaloid (+)-sedamine (**125**), which has been shown to display memory-enhancing properties and was also effective for the treatment of cognitive disorders. The reaction of the *N-tert*-butanesulfinyl imine ( $S_S$ )-**119** with trimethylsilyl enol ether derived from acetophenone **123** in the presence of TMSOTf at low temperature, produced  $\beta$ -amino ketone derivative **124** in high yield and diastereoselectivity (Scheme 34). A reduction of **124** gave a mixture of diastereomeric alcohols, and the one with (*R*)-configuration at the benzylic position was isolated in 54% yield. Further treatment of the alcohol with NaH furnished a cyclized product, which after desulfination and *N*-methylation led to expected (+)-sedamine (**125**) in 30% overall yield from ketone derivative **124**.

The stereoselective synthesis of *trans*-5-hydroxy-6-substituted-2-piperidinones was also reported by the group of Wei, taking advantage of the addition of Grignard reagents to *N-tert*-butanesulfinyl  $\alpha$ -alkoxy aldimines **126** [124]. In this one-pot approach, a successive nucleophilic addition–cyclization–desulfinylation took place, leading directly to piperidinones **127**. The reactions were performed in THF at  $-78$  °C for 3 hours. Yields ranged from moderate to excellent with aliphatic and aromatic organomagnesium compounds. Based on X-ray crystallographic analyses, the relative configurations of the products **127** were

unambiguously assigned as *trans*-form. The stereocontrol was governed by the stereogenic center bearing the OTBS group at  $\alpha$ -position of the imine, showing no influence on it the configuration of the sulfur atom of the sulfinyl unit. This methodology was applied to the asymmetric synthesis of (–)-CP-99,994 (**128**), the enantiomer of a promising clinical agent which displays a variety of biological activities, including neurogenic inflammation, pain transmission, and regulation of the immune response. Starting from  $\alpha$ -alkoxy aldimines *ent*-**126**, the utility of this methodology was also demonstrated in the synthesis of alkaloid (+)-cassine (**130**), isolated from the leaves and twigs of *Cassia excelsa*, displaying antimicrobial activity. Methylmagnesium bromide was the Grignard reagent in this synthesis, with a 18% overall yield after seven steps from aldimine *ent*-**126** (Scheme 35) [125].

The reaction of chiral  $\alpha$ -siloxy imine ( $S_S$ )-**126** with enolates derived from methyl ketones **131** was also investigated. The enolate was formed with LDA at  $-78$  °C and reacted at the same temperature with imine ( $S_S$ )-**126** for 2.5 hours. The addition proceeded with high diastereoselectivity, followed by cyclization. Final acid treatment produced the removal of the sulfinyl group leading to *trans*-5-hydroxy-6-substituted ethanone-2-piperidinones **132** in moderate to high yields, as a single diastereoisomer [126]. The diastereoselectivity of the addition was controlled in this case by both the  $\alpha$ -siloxy group and the chiral





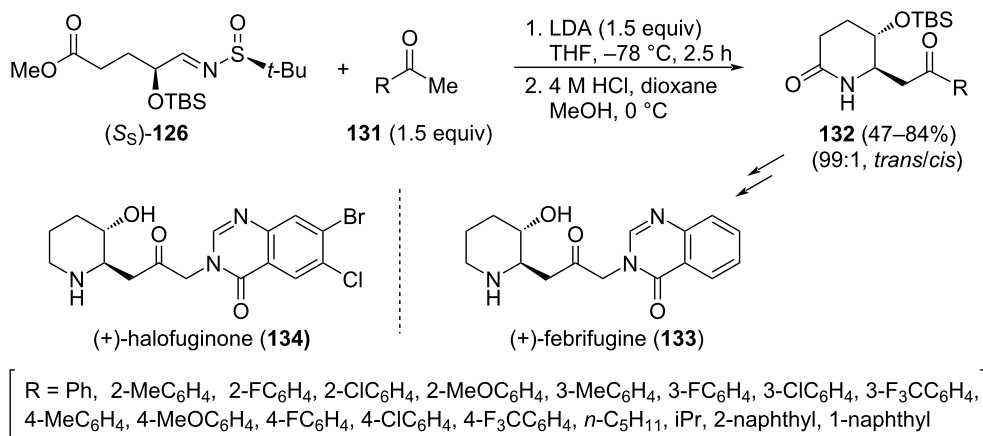
**Scheme 35:** Stereoselective synthesis of *trans*-5-hydroxy-6-substituted-2-piperidinones **127** and **129** from chiral imines **126**.

sulfinamide moiety. Interestingly, the utility of this approach was also demonstrated by the synthesis of (+)-febrifugine (**133**), a natural product isolated from Chinese medicinal plants *Dichroa febrifuga* Lour., and (+)-halofuginone (**134**), which is a pharmaceutical candidate developed from febrifugine for the treatment of human scleroderma (Scheme 36).

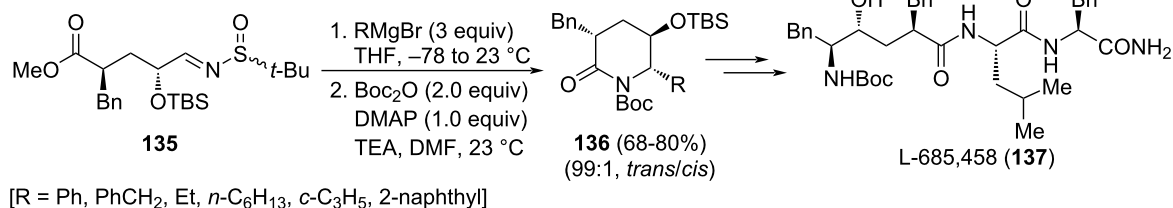
The reaction of chiral imine **135** with Grignard reagents in THF took also place with high diastereoselectivity. Starting imine **135** with two well-defined stereogenic centers at the hydrocarbon backbone were prepared as a mixture of (*R<sub>S</sub>*)- and (*S<sub>S</sub>*)-diastereoisomers from D-glutamic acid. After nucleophilic addition to the imine, a successive cyclization–desulfinylation occurred to give the corresponding piperidinone. Final reaction with di-*tert*-butyl dicarbonate led to functionalized 2-piperidinones **136** [127]. These compounds are interesting reaction intermediates because they can be transformed by conventional

reactions into, for instance, compound L-685,458 (**137**), an inhibitor of  $\gamma$ -secretase, with potential interest for the treatment of Alzheimer's disease and other neurological disorders (Scheme 37).

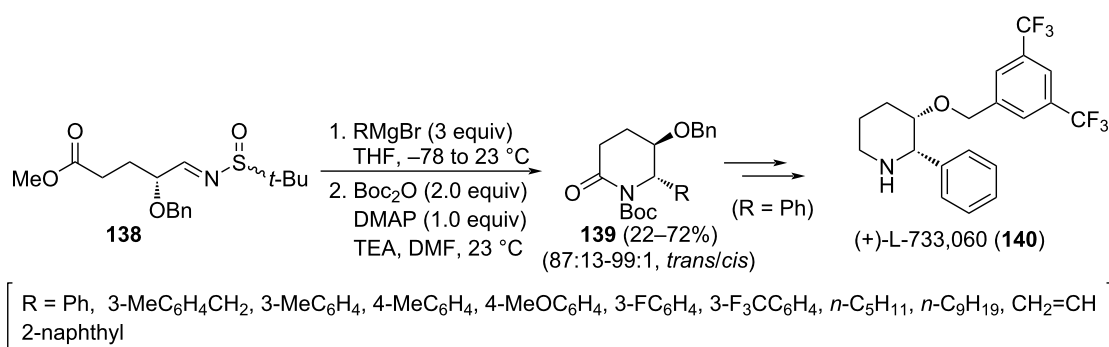
The diastereoselective synthesis of *trans*-5-hydroxy-6-substituted 2-piperidinones **139** was also achieved from *O*-benzyl protected aldimine **138** following the previously commented tandem Grignard reagent addition, subsequent cyclization–desulfinylation, and final *N*-Boc protection. The stereochemical pathway is controlled exclusively again by the configuration of the stereogenic center bearing the benzyloxy group [128]. Interestingly, chiral  $\delta$ -lactams **139** are synthetic intermediates that can be transformed into compound (+)-L-733,060 (**140**), a potent neurokinin substance P receptor antagonist. This compound displays a wide variety of biological activities, including inhibition of neurogenic inflammation, blocking



**Scheme 36:** Stereoselective synthesis of *trans*-5-hydroxy-6-substituted ethanone-2-piperidinones **132** from chiral imine (*S<sub>S</sub>*)-**126**.



**Scheme 37:** Stereoselective synthesis of *trans*-3-benzyl-5-hydroxy-6-substituted-2-piperidinones **136** from chiral imines **135**.



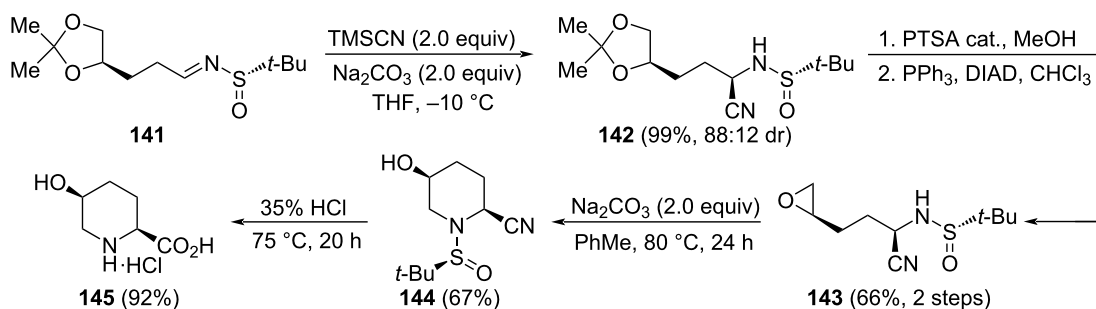
**Scheme 38:** Stereoselective synthesis of *trans*-5-hydroxy-6-substituted 2-piperidinones **139** from chiral imine **138**.

of pain transmission and regulation of immune response (Scheme 38).

A stereoselective synthesis of L-hydroxypipercolic acid **145** was reported recently by Zhang and Sun. Compound **145** is an intermediate for the synthesis of  $\beta$ -lactamase inhibitors. A key step in this synthesis was the hydrocyanation of chiral sulfinyl imine **141**, prepared from commercially available and inexpensive L-glyceraldehyde acetal, with trimethylsilyl cyanide (TMSCN) in THF at -10 °C. The reaction product **142** was obtained in quantitative yield and good diastereomeric ratio. Further hydrolysis of the cyclic acetal, and subsequent epoxidation of the resulting diol under typical Mitsunobu conditions led to epoxide

derivative **143**. The piperidine ring was formed through a 6-*endo-tet* cyclization by treatment of the epoxide **143** with sodium carbonate in toluene at 80 °C. Hydrolysis of the cyano group under acidic conditions of compound **144** led to expected L-hydroxypipercolic acid hydrochloride **145** in high yield (Scheme 39) [129].

In 2018, Wei and co-workers described the diastereoselective synthesis of 1-substituted isoquinolones using one-pot addition–cyclization–deprotection of the imine with Grignard reagents [130]. In this work, the addition to chiral imines **146**, **148** and **150** was performed using 2,2'-dipyridyl- or 4-methylmorpholine (NMM) to promote the complexation with the Grig-



**Scheme 39:** Stereoselective synthesis of L-hydroxypipercolic acid **145** from chiral imine **144**.

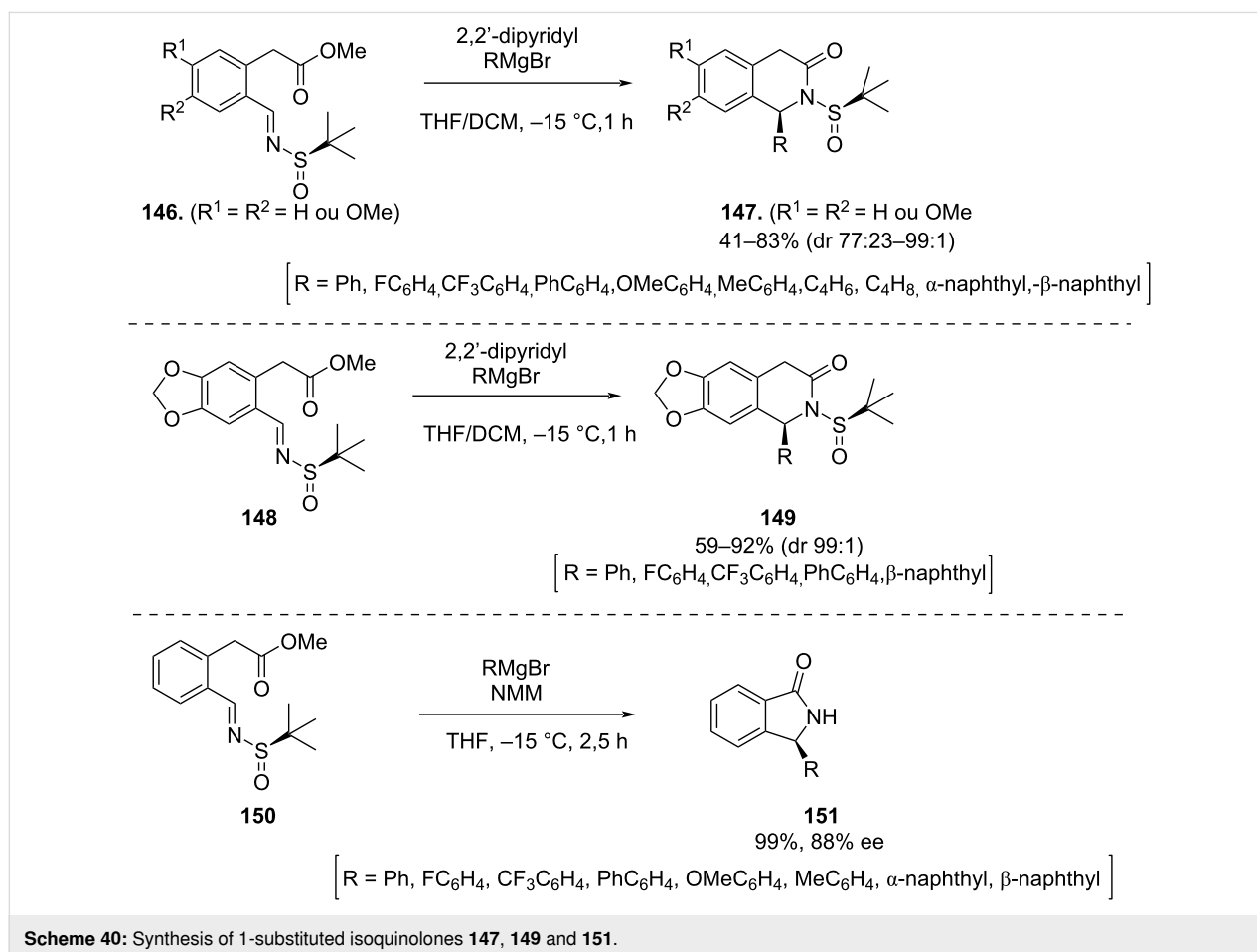
nard reagent. Products **147** and **149** were obtained in excellent yields and high diastereoselectivity and when 4-methylmorpholine (NMM) was used as additive, the heterocycle **151** was obtained in one pot addition–cyclization–deprotection of imine **150** (Scheme 40).

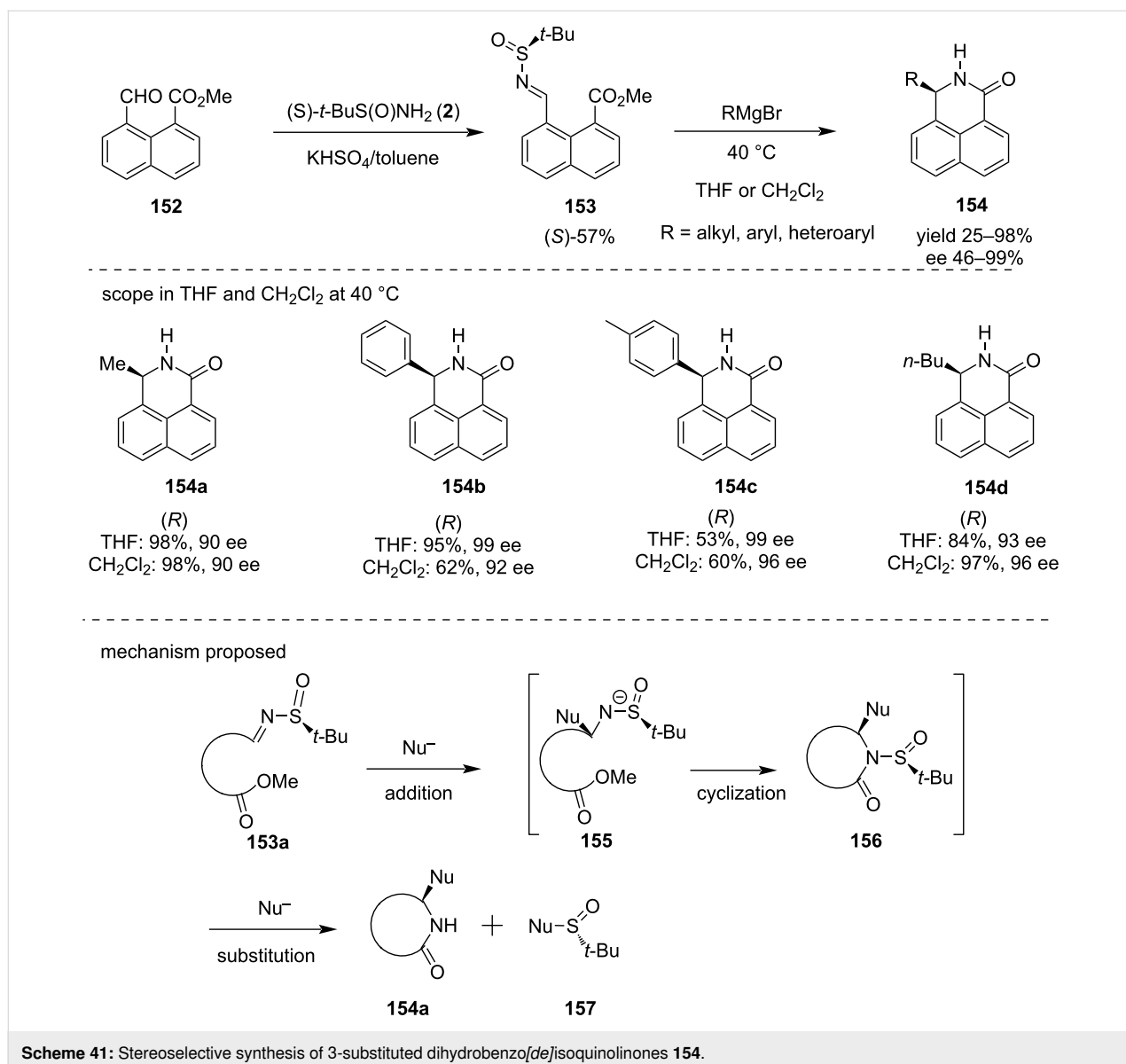
In 2020, Kaczorek and Kawęcki described the stereoselective synthesis of 3-substituted dihydrobenzo[*de*]isoquinolones **154** in both enantiomeric forms in one step. In this study, they reported an addition–cyclization–substitution reaction employing (*Rs*) and (*Ss*) *N*-*tert*-butylsulfinyl imine **14** and Grignard reagents using THF or CH<sub>2</sub>Cl<sub>2</sub> as solvent at 40 °C. The 3-substituted dihydrobenzo[*de*]isoquinolinones **154** were obtained with good yield and with enantiomeric excess of 46–99%. The mechanism was explained by stereoselective addition of the Grignard reagent to the *N*-sulfinyl imine **153a** derived from **152**, in a subsequent cyclization to obtain the intermediate **156** then, a substitution at the sulfur atom occurred to form **154a** and **157** [131] (Scheme 41).

In 2017, Reddy and co-workers described the stereoselective synthesis of (*S*)-1-benzyl-6,7-dimethoxy-1,2,3,4-tetrahydroiso-

quinoline (**163**), (*S*)-1-benzyl-6,7-dimethoxy-*N*-methyl-1,2,3,4-tetrahydroisoquinoline (**164**), (–)-*O*,*O*-dimethylcoclaurine (**165**) and (+)-*O*-methylarmapavine (**166**) alkaloids via chiral *tert*-butylsulfinamide through a haloamide cyclization. The strategy was based on the addition of organomagnesium bromide or chloride to chiral *N*-sulfinyl imine **160**. A subsequent base promoted cyclization of chloroamides (**158** and **162**) and the products **165** and **163** were obtained in 91% and 93% yields respectively. The *N*-methylation of alkaloids **163** and **165** using 37% formaldehyde and sodium borohydride formed the tetrahydroisoquinoline **164** and **166** in high yields of 95% and 94%, respectively [132] (Scheme 42).

Pinto and co-workers reported recently the enantioselective synthesis of natural alkaloids (–)-cermizine **B** **171** and (+)-serratezomine **E** **172**. A key step of the synthetic strategy is the allylation with allylmagnesium bromide of *N*-*tert*-butanesulfinyl imine **168**. At the allylation step, a new chiral center with *S* configuration in compound **169** is formed in 96% yield. After removal the chiral sulfinyl group under acid conditions, treatment with acryloyl chloride produced acrylamide derivative **170**. From this common intermediate, and after several subse-





**Scheme 41:** Stereoselective synthesis of 3-substituted dihydrobenzo[de]isoquinolinones **154**.

quent steps, including ring-closing metathesis [133], (–)-cermizine **171** and (+)-serratezomine **172** were obtained 57% and 72% yield, respectively (Scheme 43) [134,135].

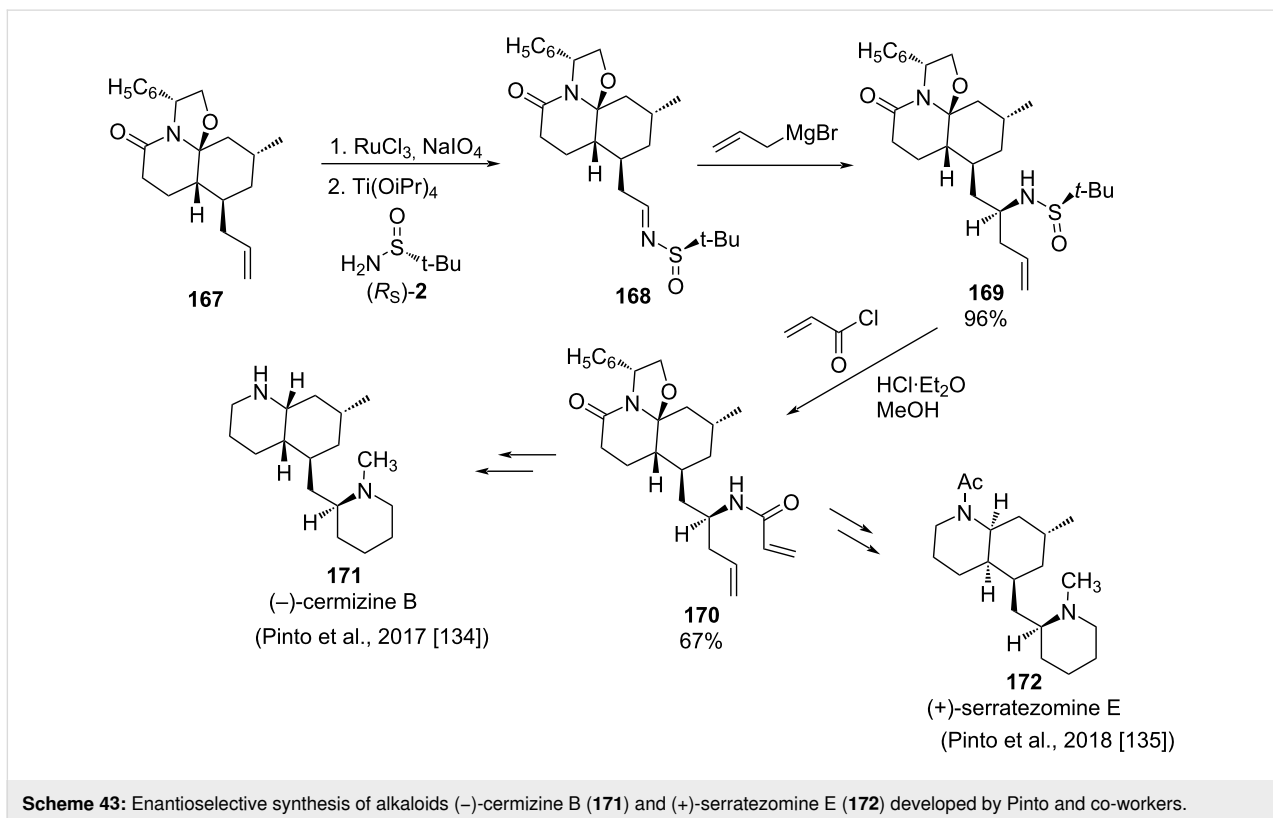
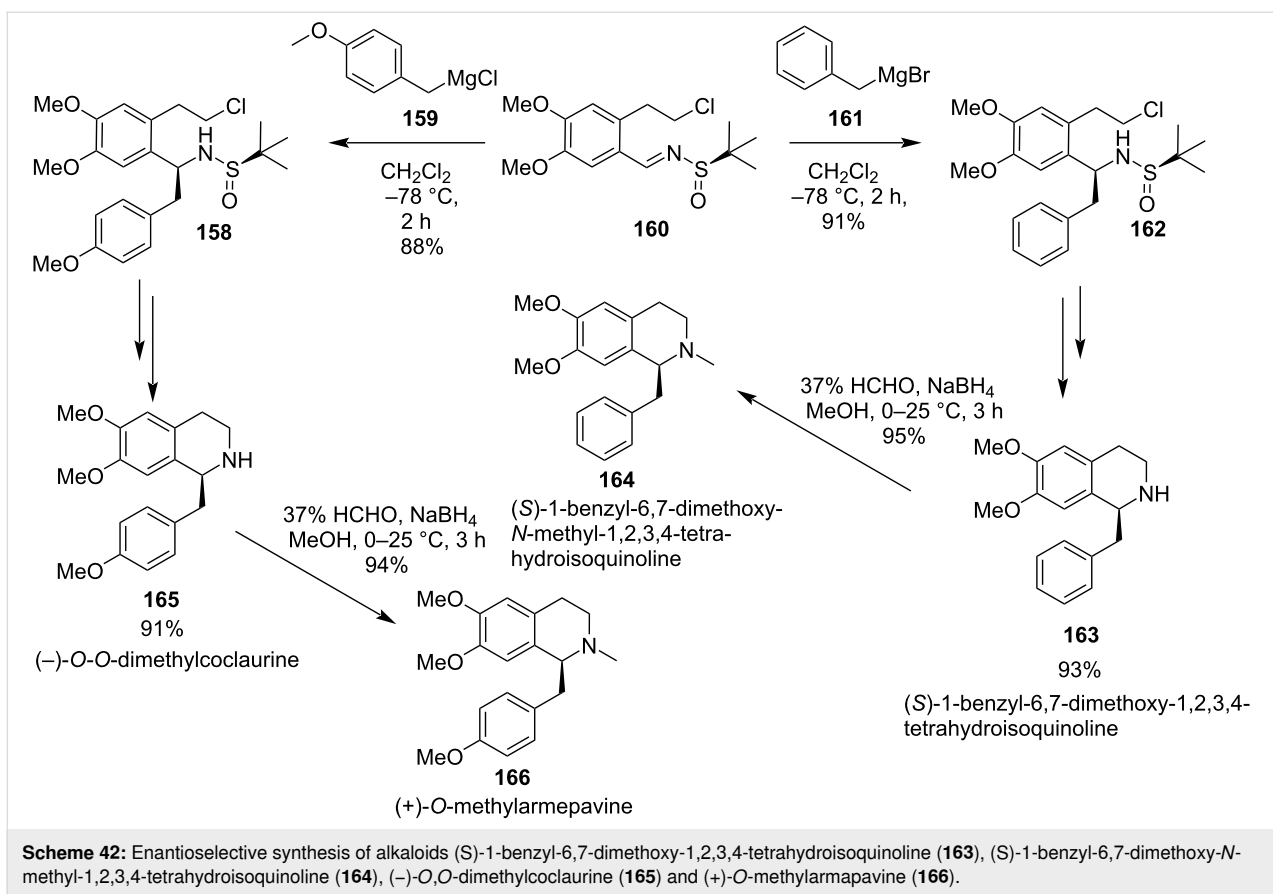
### Cyclizations involving a position in the attacking nucleophile

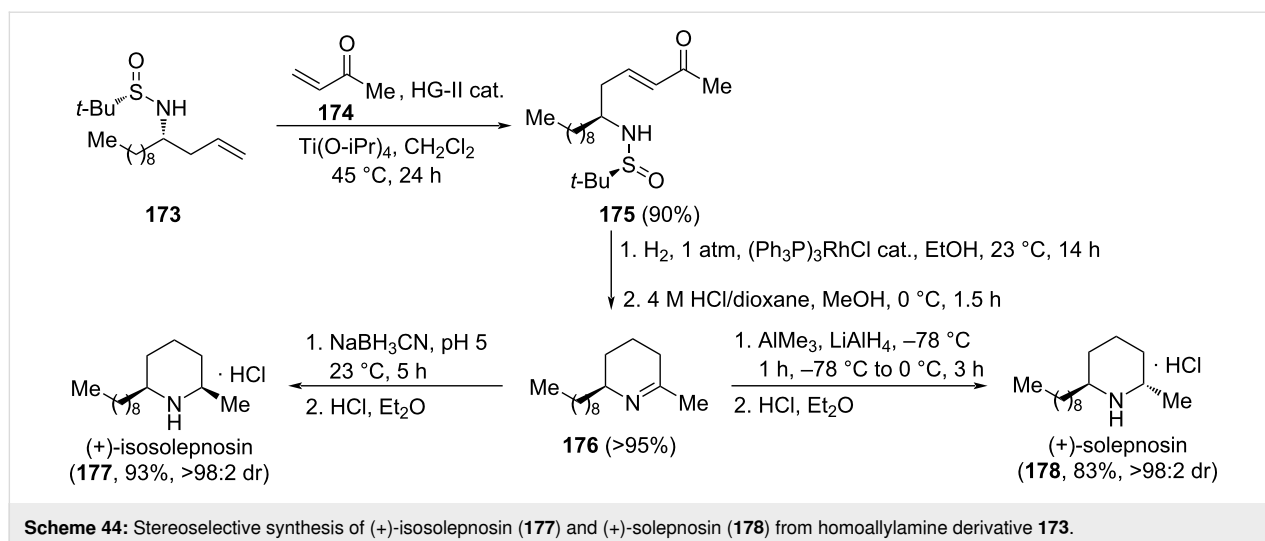
Isosolenopsin (**177**) and solenopsin (**178**) are two isomeric piperidine alkaloids isolated from fire ants (*Solenopsis*) and display hemolytic, insecticide and antibiotic properties. A straightforward synthesis of these natural products from a common imine intermediate was reported by Medjahdi et al. comprising as key steps the indium–titanium-mediated aminoallylation of nonanal with (*R*)-*tert*-butanesulfonamide and allyl bromide, giving rise homoallylamine derivative **173**, a subsequent cross metathesis with methyl vinyl ketone (**174**) cata-

lyzed by Hoveyda–Grubbs second generation catalyst to produce compound **175**, followed by hydrogenation–desulfonation, and final stereoselective reduction of the resulting cyclic imine intermediate **176**. In this diastereodivergent approach, reduction of this imine with sodium borohydride in a citrate-phosphate buffer medium (pH 5) led to (+)-isosolenopsin (**177**) in 93% yield and >98:2 *cis/trans* selectivity. On the other hand, when the reduction of imine **177** was carried out applying H. Yamamoto’s protocol (AlMe<sub>3</sub>/LiAlH<sub>4</sub>), (+)-solenopsin (**178**) was isolated in 83% yield and with excellent diastereoselectivity (>98:2 *trans/cis* selectivity, Scheme 44) [136].

There are many compounds with a 1,2,3,4-tetrahydroisoquinoline structural motif bearing substituents at 1-position which display a wide range of biological activities. However,







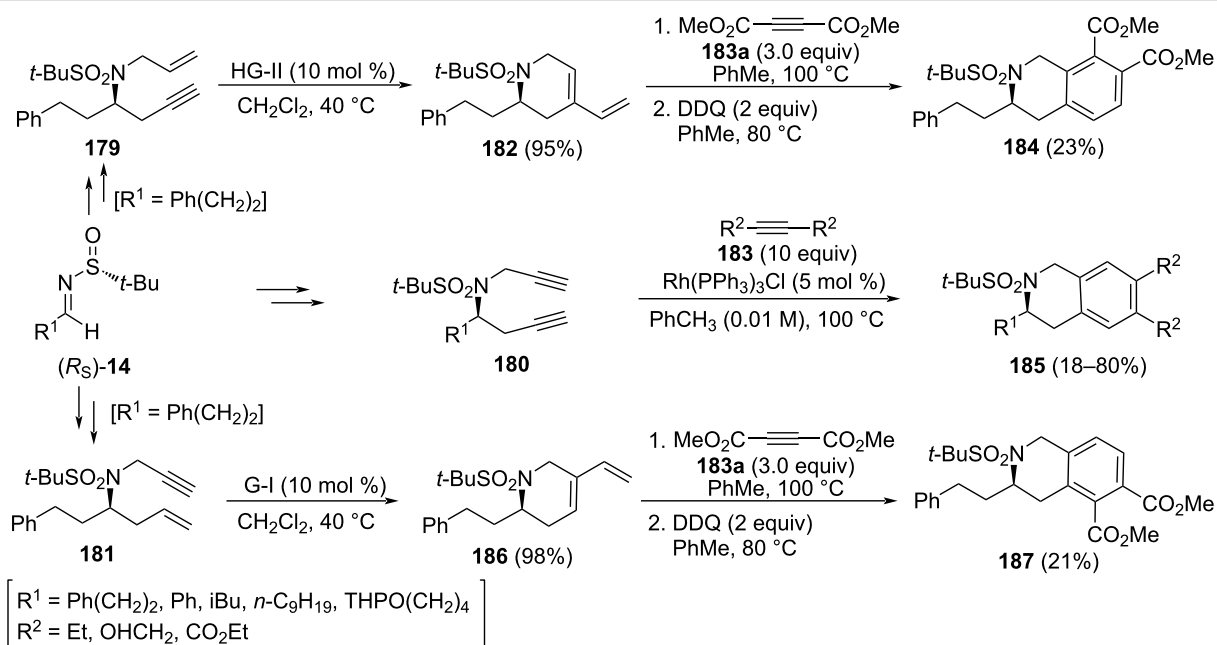
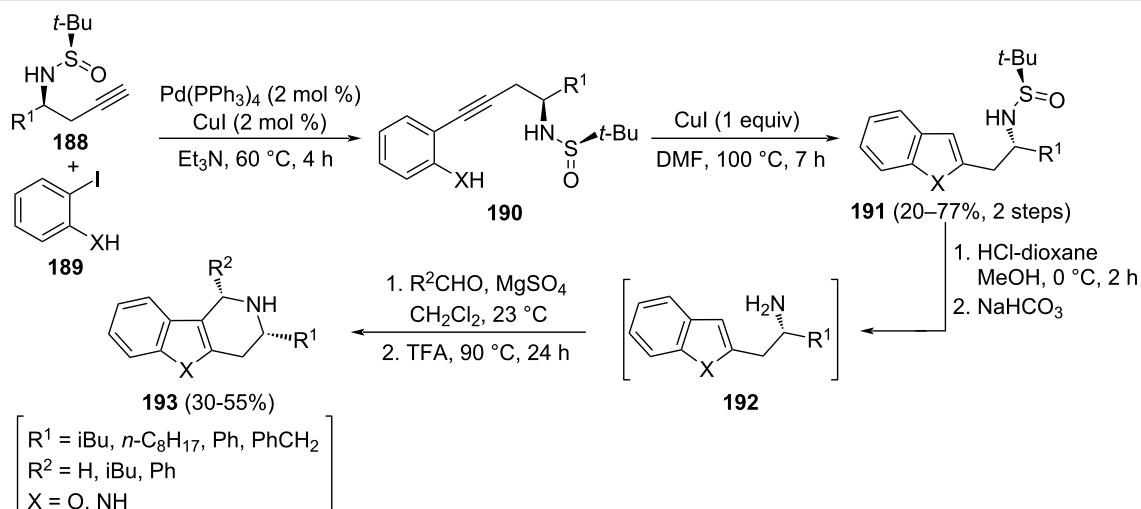
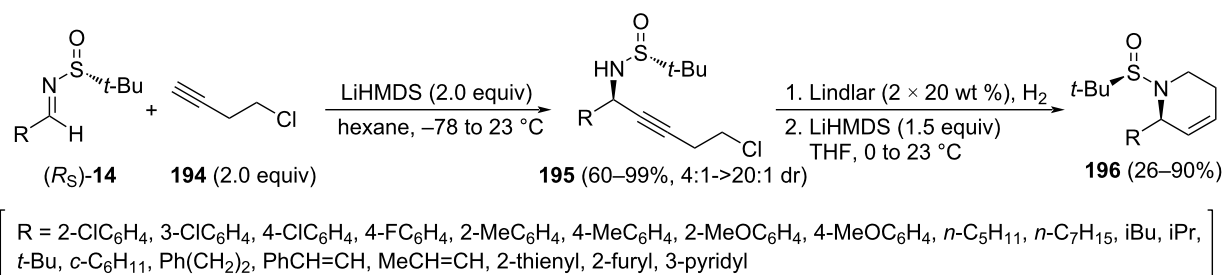
compounds bearing substituents at 3-position are less represented in nature and among pharmaceutical drugs. A multistep methodology to synthesize 1,2,3,4-tetrahydroisoquinolines **185** bearing substituents at 3-, 6- and 7-positions in a highly enantioselective fashion starting from chiral *N*-*tert*-butanesulfonyl imines (*R<sub>S</sub>*)-**14** was reported by Sirvent et al. [137]. The key step of this synthesis is a [2 + 2 + 2] cyclotrimerization by means of Wilkinson catalyst of azadiyne system **180**, which was accessible from imine **14** by consecutive diastereoselective indium-promoted propargylation, selective *N*-propargylation and final oxidation of the sulfonyl group. Sulfonyl imines (*R<sub>S</sub>*)-**14** could be also precursors of tetrahydroisoquinolines with substituents at different positions of the aromatic ring, by combining allylation and propargylation processes as the first steps of this new strategy. The resulting azaenynes **179** and **181** were efficiently transformed by a ruthenium-catalyzed ring-closing metathesis into cyclic 1,3-dienes **182** and **186**, respectively. The best results were obtained by performing the metathesis with Hoveyda–Grubbs second generation catalyst in the synthesis of cyclodiene **182**, and with Grubbs first generation catalyst for compound **186**. When these dienes reacted with dimethyl acetylenedicarboxylate in toluene at 100 °C, followed by dehydrogenation of the resulting [4 + 2] adduct with DDQ, the expected 7,8- or 5,6-bis(methoxycarbonyl)substituted 1,2,3,4-tetrahydroisoquinolines **184** and **187**, were obtained, respectively (Scheme 45).

Many indole alkaloids have been known for years and used in ancient cultures as psychotropic, stimulants and poisons. On the other hand, benzofurans and indoles, with a 2-aminoalkyl substituent at the 2-position, are not common compounds nor are they represented in nature. For that reason, synthetic methodologies to access these systems are of interest in order to explore their biological activity. Homopropargylamine derivatives **188**

were obtained in a highly diastereoselective fashion (>95:5 dr) by nucleophilic addition of allenylindium intermediate to chiral *N*-*tert*-butanesulfonyl imines **14**. Subsequent Sonogashira coupling of compounds **188** with *o*-iodophenol (X = O) or *o*-iodoaniline (X = NH) **189**, led to 2-(2-aminoalkyl)benzofuran (X = O) and -indole (X = NH) derivatives **191**. Further removal of the sulfonyl unit under acidic conditions produced amine derivatives **192**, which were transformed into tetrahydropyrido-benzofuran (X = O) and indole (X = NH) derivatives **193** with relative *cis*-configuration, upon reaction with aldehydes. This Pictet–Spengler condensation was facilitated by the nucleophilic character of the 3-position of the benzofuran or indole moiety (Scheme 46) [138].

Turlington and co-workers reported a stereoselective synthesis in three steps of 2-substituted 1,2,5,6-tetrahydropyridines **196** starting from chiral *N*-*tert*-butanesulfonyl imines (*R<sub>S</sub>*)-**14** [139]. The synthesis commenced with addition of the organolithium compound resulting from the deprotonation of 4-chloro-1-butyne (**194**) to the imine. The propargylamine derivatives **195** were obtained in high yields and diastereoselectivities (>20:1 dr, in most cases). The lowest diastereoselectivity was found for pyridyl-substituted imine **14** (R = 3-pyridyl, 4:1 dr), due probably to competitive coordination of the lithium acetylide by the heteroatoms present in these imines. Reduction of the triple bond with Lindlar catalyst to provide olefin with *cis*-configuration, and cyclization using LiHMDS led to 1,2,5,6-tetrahydropyridines **196** (Scheme 47).

Many methods have been provided to generate thermodynamically stable *cis*-2,6-disubstituted piperidines, but the synthesis of *trans*-derivatives remains elusive. In this regard, Bhattacharjee and co-workers reported a highly efficient large-scale synthesis of 2,6-*trans*-piperidine derivative **199** from easily

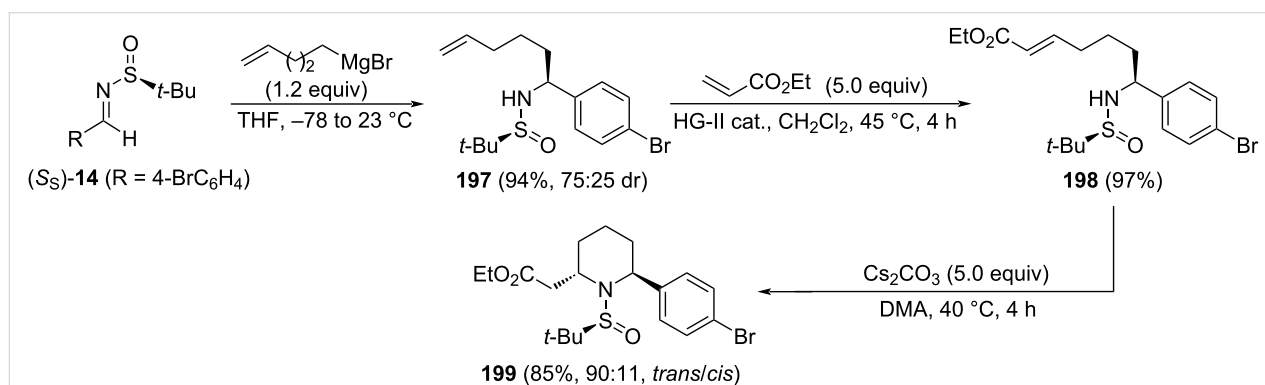
Scheme 45: Stereoselective synthesis of tetrahydroquinoline derivatives **184**, **185** and **187** from chiral imines ( $R_S$ )-**14**.Scheme 46: Stereoselective synthesis of pyridobenzofuran and pyridoindole derivatives **193** from homopropargylamine derivatives **192**.Scheme 47: Stereoselective synthesis of 2-substituted 1,2,5,6-tetrahydropyridines **196** from chiral imines ( $R_S$ )-**14**.

available starting material [140]. Key steps of the synthesis are the diastereoselective addition of 4-pentenylmagnesium bromide to chiral imine ( $S_S$ )-**14** ( $R = 4\text{-BrC}_6\text{H}_4$ ). Two diastereoisomers **197** were isolated in 94% yield and moderate diastereoselectivity. The major component of the mixture was transformed into compound **198** through a Hoveyda–Grubbs second generation cross metathesis with ethyl acrylate. Another key step was the intramolecular aza-Michael reaction promoted by cesium carbonate as base in dimethylacetamide (DMA), leading to 2,6-*trans*-piperidine derivative **199** in high yield (Scheme 48). This compound was an intermediate in the synthesis of a novel class of anti-infective agents.

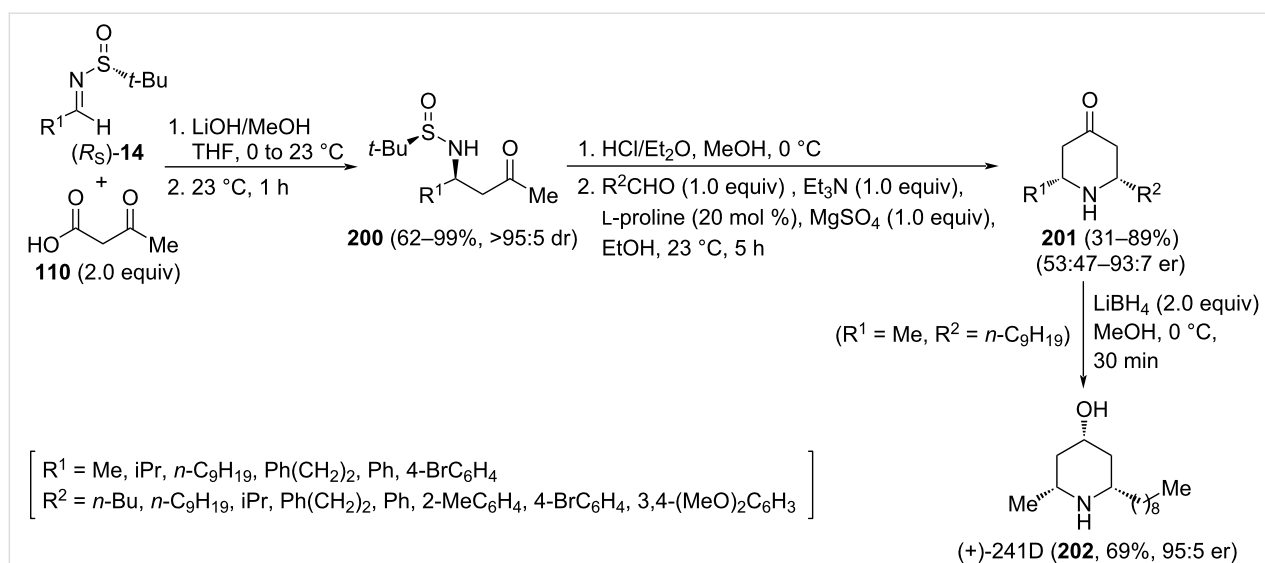
The base-promoted decarboxylative Mannich coupling of chiral imines ( $R_S$ )-**14** [derived from an aldehyde  $R^1\text{CHO}$  and ( $R$ )-*tert*-butanesulfinamide] with these reagents provided  $\beta$ -amino ketone derivative **200** in high yields and diastereoselectivities [120]. These compounds were easily transformed into *cis*-2,6-

disubstituted piperidin-4-ones **201** through a  $\text{L}$ -proline organocatalyzed intramolecular Mannich reaction with a second aldehyde ( $R^2\text{CHO}$ ). Almost no diastereoselectivity was observed when  $R^1$  and  $R^2$  were aromatic rings. On the other hand, aliphatic aldehydes gave in general excellent enantiomeric ratios (>90:10). It is important to note that the order of reaction of carbonyl compounds  $R^1\text{CHO}$  and  $R^2\text{CHO}$  with *tert*-butanesulfinamide to form chiral imine **14**, or in the intramolecular organocatalyzed condensation, determined the absolute configuration of compounds **201** [141]. The usefulness of this methodology was demonstrated in the synthesis of the alkaloid (+)-241D (**202**), isolated from the skin of the Panamanian poison frog *Dendrobates speciosus*, through the reduction of piperidin-4-one **201** ( $R^1 = \text{Me}$ ,  $R^2 = n\text{-C}_9\text{H}_{19}$ ) with lithium borohydride (Scheme 49).

The base-catalyzed addition of 4-nitrobutanoates **203** to  $N$ -*tert*-butanesulfinyl imines ( $R_S$ )-**14** under solvent-free reaction condi-



**Scheme 48:** Stereoselective synthesis of 2-substituted *trans*-2,6-disubstituted piperidine **199** from chiral imines ( $S_S$ )-**14** ( $R = 4\text{-BrC}_6\text{H}_4$ ).



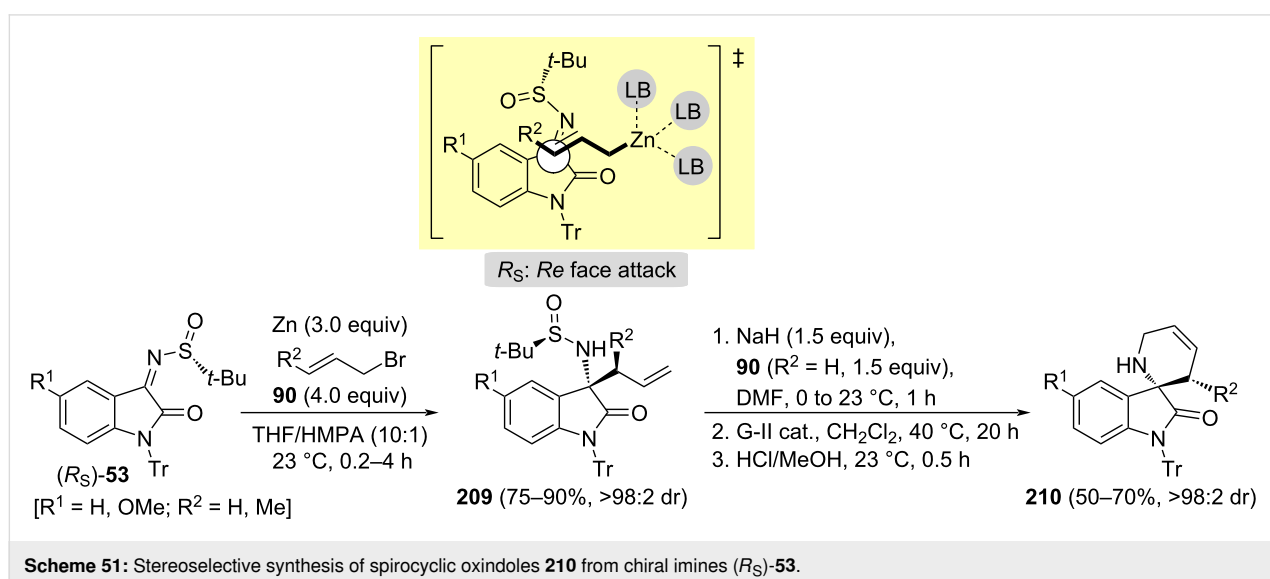
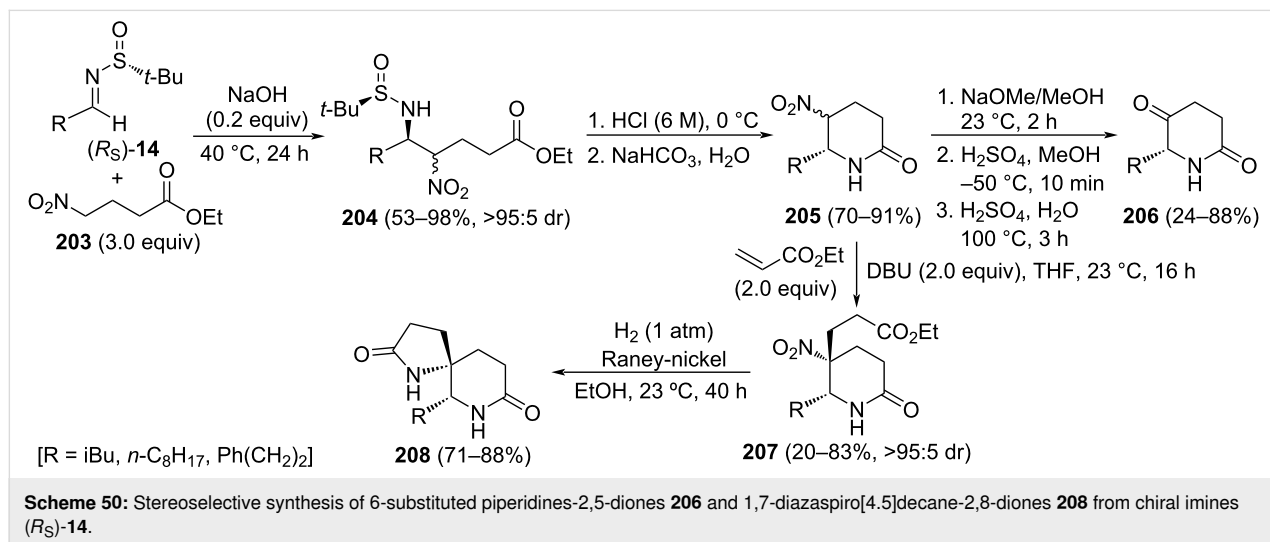
**Scheme 49:** Stereoselective synthesis of *cis*-2,6-disubstituted piperidines **200**, and alkaloid (+)-241D, from chiral imines ( $R_S$ )-**14**.

tions proceeded with high facial diastereoselectivity. The resulting  $\beta$ -nitroamine derivatives **204** were easily transformed into 5-nitro-6-substituted piperidine-2-ones **205**, upon removal of the sulfinyl group with concomitant  $\delta$ -lactam formation. Further transformation of the nitro group under Nef-type reaction conditions led to enantioenriched 6-substituted piperidine-2,5-diones **206** [142]. Interestingly, from compounds **205**, and following a two-step process, involving conjugative addition to ethyl acrylate with formation of **207** as a single diastereoisomer, and final reduction of the nitro group with Raney-nickel, 1,7-diazaspiro[4.5]decane-2,8-diones **208** were accessed in a highly stereoselective fashion (Scheme 50) [143].

A large number of biologically active natural products and synthetic pharmaceutical drugs contain the 3-aminoxindole motif. Chen and Xu demonstrated that the zinc-mediated allylation of

chiral oxindole sulfinyl imines ( $R_S$ )-**53** with allylic bromides proceeded smoothly at room temperature in a mixture of THF and HMPA, and a wide range of highly enantiomerically enriched 3-allyl-substituted 3-aminoxindoles **209** were prepared. The observed diastereofacial selectivity was rationalized by considering an acyclic transition state model. The addition of the allylic reagent occurred to the less hindered *Re* face of the imine with ( $R_S$ )-configuration. *N*-Allylation of compounds **209**, followed by ring-closing metathesis with Grubbs second generation catalyst, and removal of the sulfinyl group, led to chiral spirocyclic aminooxindoles **210** in reasonable yields (Scheme 51) [144].

Azaspirocyclic alkaloids with interesting pharmacological properties have been isolated from skin extracts of dendrobatid frogs, and also from methanol extracts of ants of the species



*Carabella bicolor*. Amongst these alkaloids, (–)-histrionicotoxin displays a potent noncompetitive acetylcholine antagonist activity. It was found that perhydrohistrionicotoxin analogues display similar biological properties. In this regard, Peralta-Hernández and Cordero-Vargas reported the synthesis of an advanced synthetic intermediate **213** of perhydrohistrionicotoxin [145]. The diastereoselective nucleophilic addition of a lithium acetylide to cyclic chiral *N-tert*-butanesulfinyl imine **211** is a key step in this strategy. The addition proceeded with total stereocontrol to give a single diastereoisomer, and further removal of the silyl group provided propargylamine derivative **212**. Partial hydrogenation of this compound under Lindlar conditions led to terminal olefin, which reacted with ethyl iodacetate in refluxing dichloroethane, in the presence of 1.6 equivalents of lauroyl peroxide (DLP), as thermal initiator of the radical process. A spiroactam was isolated in 45% yield, taking place under the essayed reaction conditions successively a radical addition of the enolate to the terminal alkene, lactonization and removal of the sulfinyl group. Final deprotection of the hydroxy group led to compound **213**, a precursor of the 6-(*R*) epimer of perhydrohistrionicotoxin (Scheme 52).

Chiral aromatic sulfinyl imines **214** with a 2-haloethyl substituent at *ortho*-position were effective synthetic intermediates in the stereoselective preparation of 1-aryl-1,2,3,4-tetrahydroisoquinolines, which are compounds that display interesting biological activities. For instance, solifenacin (**216**), a competitive muscarinic acetylcholine receptor antagonist currently used in the treatment of overactive bladders, was prepared from (*R<sub>S</sub>*)-**214a** (X = Br) by addition first of phenylmagnesium bromide at –40 °C in toluene. A 93:7 diastereomeric mixture was obtained and the major diastereoisomer was easily isolated after column chromatography. Subsequent intramolecular cyclization in the presence of NaH in DMF at room temperature gave the pure diastereomer **215**, a precursor of solefinacin (**216**) [146]. Diastereoselective allylation of chlorinated derivative (*R<sub>S</sub>*)-**214b** (X = Cl) with allylmagnesium bromide in dichloromethane gave the corresponding homoallylic sulfinamide as a 9:1 mixture of easily separable diastereoisomers, and the major component of the mixture was further cyclized to give product **217**, which was transformed after 5 steps into almorexant (**218**),

a non-peptide antagonist of the human orexin receptor, which plays a major role in controlling the sleep/wake cycle [147]. The same precursor (*R<sub>S</sub>*)-**214b** (X = Cl) and strategy was followed in the first steps of the synthesis of compound **220**, used as neuroprotective agents in the treatment of neurological diseases, such as epilepsy and ischemia. In this case, the addition of 4-chlorophenylmagnesium bromide to (*R<sub>S</sub>*)-**214b** gave the expected product with 93:7 ratio of diastereoisomers. An intramolecular cyclization of the major diastereoisomer afforded **219** in high yield (Scheme 53) [146].

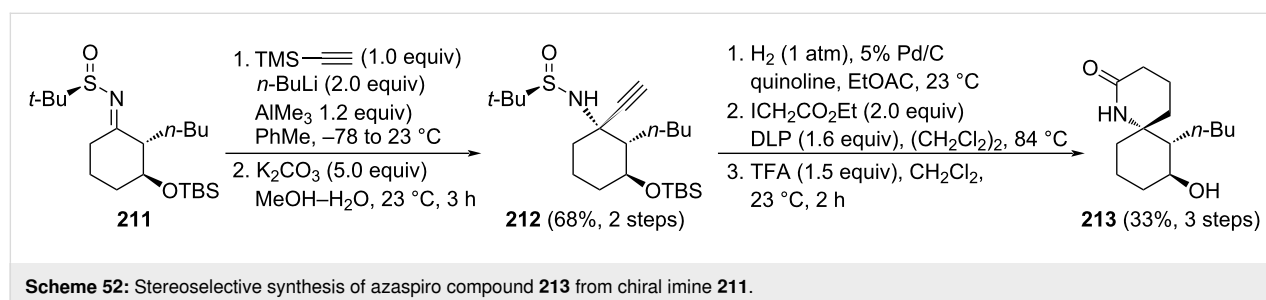
### Asymmetric synthesis of pyrrolizidines, indolizidines and quinolizidines

Bicyclic systems containing bridgehead nitrogen, such as 1-azabicyclo[3.3.0]octanes, 1-azabicyclo[4.3.0]nonanes and 1-azabicyclo[4.4.0]decanes are structural motifs frequently encountered in alkaloids, which can come from quite diverse sources, such as bacteria, fungi, plants and animals, among others. Many of these natural products display extremely potent biological activities and their syntheses, along with that of structurally related analogs, remain of great interest for chemists and pharmacologists. Relevant contributions regarding the synthesis of these bicyclic compounds, involving chiral *N-tert*-butanesulfinyl imines, are compiled in the following paragraphs.

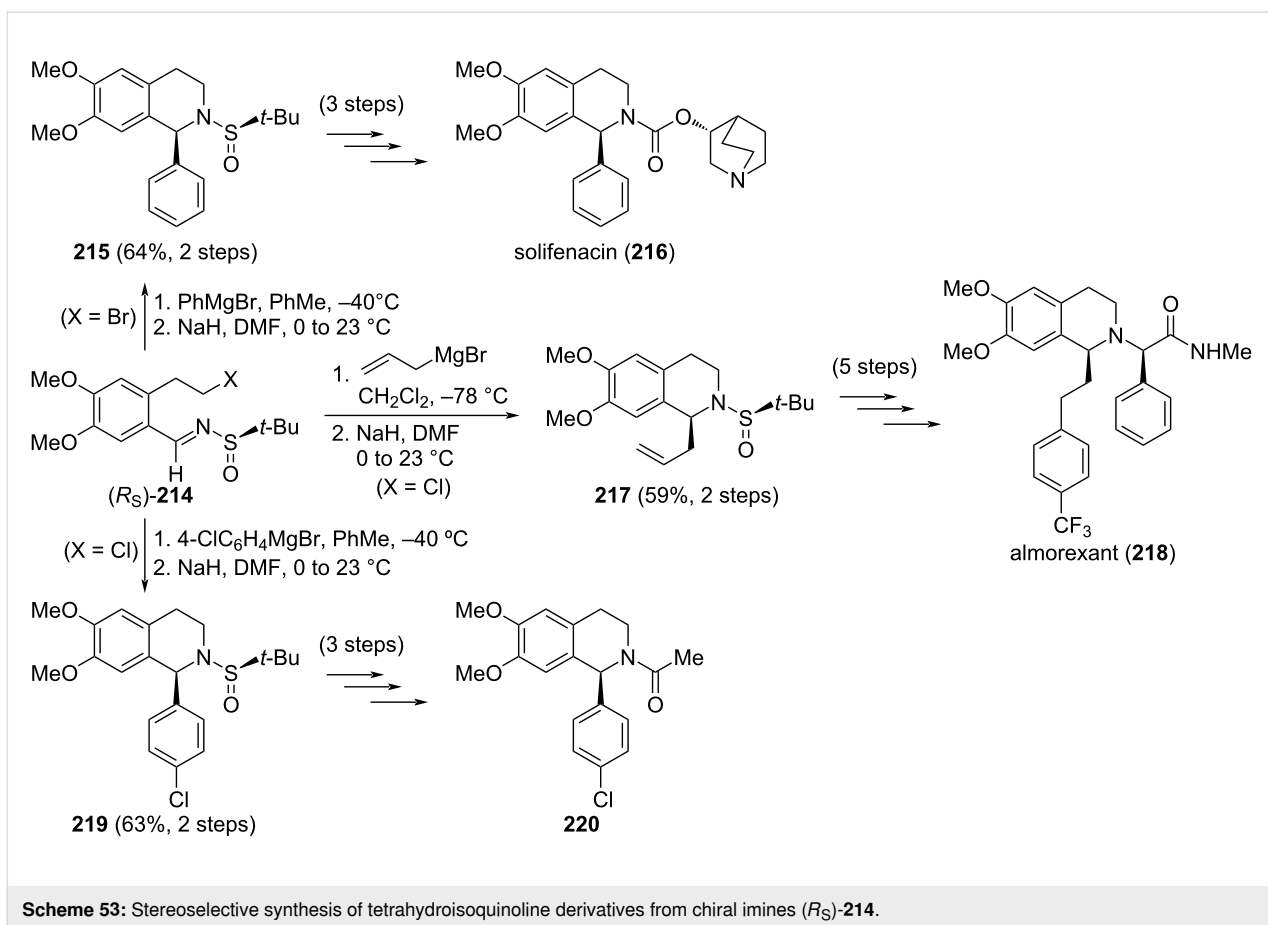
#### Initial stereocontrol by allylation of sulfinyl imines

The allylation of chiral *N-tert*-butanesulfinyl imines is of great synthetic interest because in this reaction together with a new functionality (amino derivative group), a carbon–carbon bond is formed. In addition, the double bond of the allylic moiety can participate in a number of further synthetically useful transformations, including the generation of functional groups prone to participate in intramolecular cyclization processes involving the nitrogen atom of the starting imine. Interestingly, the allylation of these imines can be carried out in a stereoselective fashion with different allylating reagents [66,148,149].

Pyrroloisoquinoline alkaloid (–)-crispine A (**223**) was isolated from *Carduus crispus* plants which were used in folk medicine for the treatment of different inflammatory diseases, such as bronchitis, stenocardia, gastroenteritis, and rheumatism. In addi-

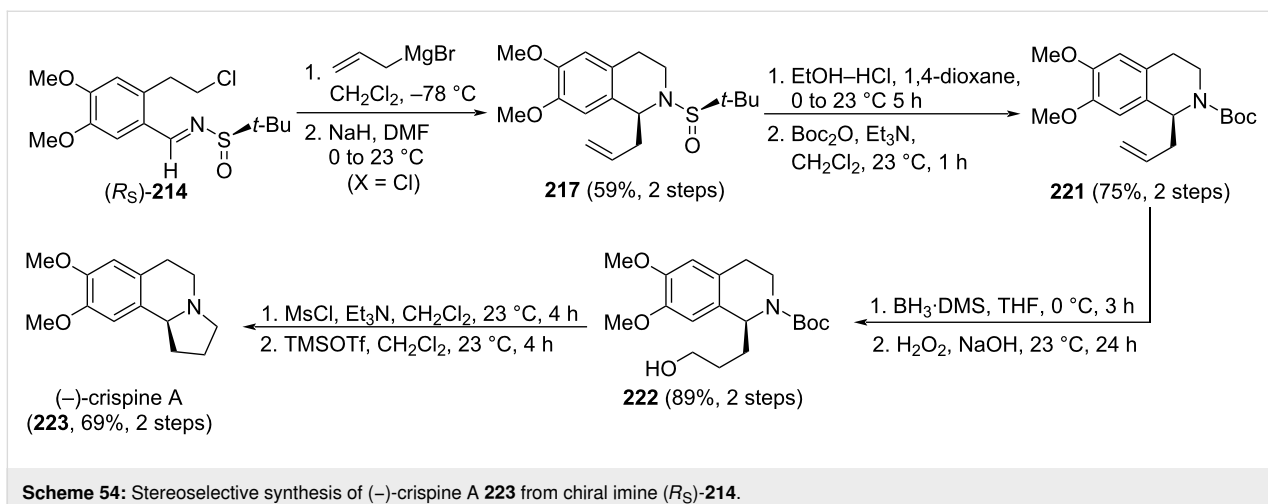


**Scheme 52:** Stereoselective synthesis of azaspiro compound **213** from chiral imine **211**.



tion, it also shows promising biological activity against human cancer cell lines. The allylation of chiral imine (*R<sub>S</sub>*)-**214** with allylmagnesium bromide in dichloromethane at  $-78\text{ }^{\circ}\text{C}$  was the key step in the synthesis reported by Reddy and co-workers of this alkaloid. The allylated product was obtained in 80% yield as 9:1 mixture of diastereoisomers, and the major diastereoisomer was separated from the mixture and cyclized to give tetra-

hydroisoquinoline **217** (see above, Scheme 53). The formation of the 5-membered ring to produce target (–)-crispine A (**223**) was accomplished in six additional steps which comprise removal of the sulfinyl group and subsequent *N*-Boc protection to give **221**, hydroboration–oxidation to produce terminal alcohol derivative **222**, and formation of the mesylate, removal of the Boc group, and final cyclization (Scheme 54) [150].

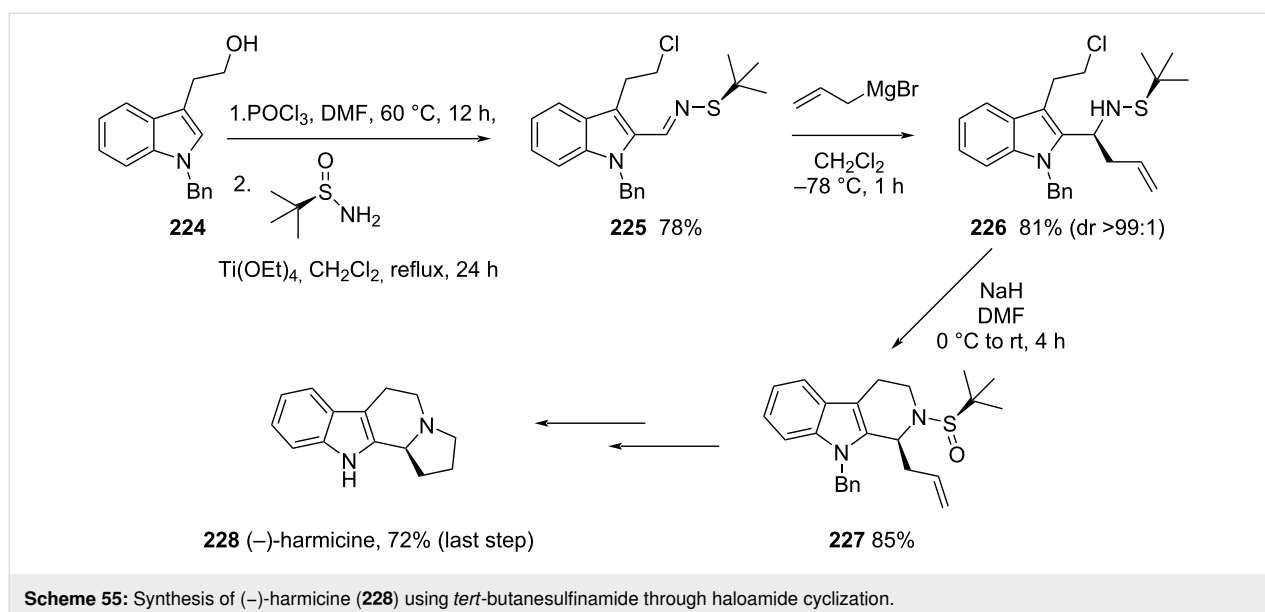


A similar strategy was described by the same authors for the stereoselective synthesis of (–)-harmicine and other tetrahydro- $\beta$ -carboline alkaloids. The allylation of chiral imine (*R*<sub>S</sub>)-**225** with allylmagnesium bromide in dichloromethane at –78 °C was the key step for the synthesis reported by Reddy and co-workers that led to compound **226** in 81% yield (dr > 99:1). After a sequence of similar steps of removal of the sulfinyl group, protection of the amine, hydroboration–oxidation and formation of the mesylate, removal of the Boc group, and final cyclization, the (–)-harmicine (**228**) was obtained in 72% yield in the last step [151] (Scheme 55).

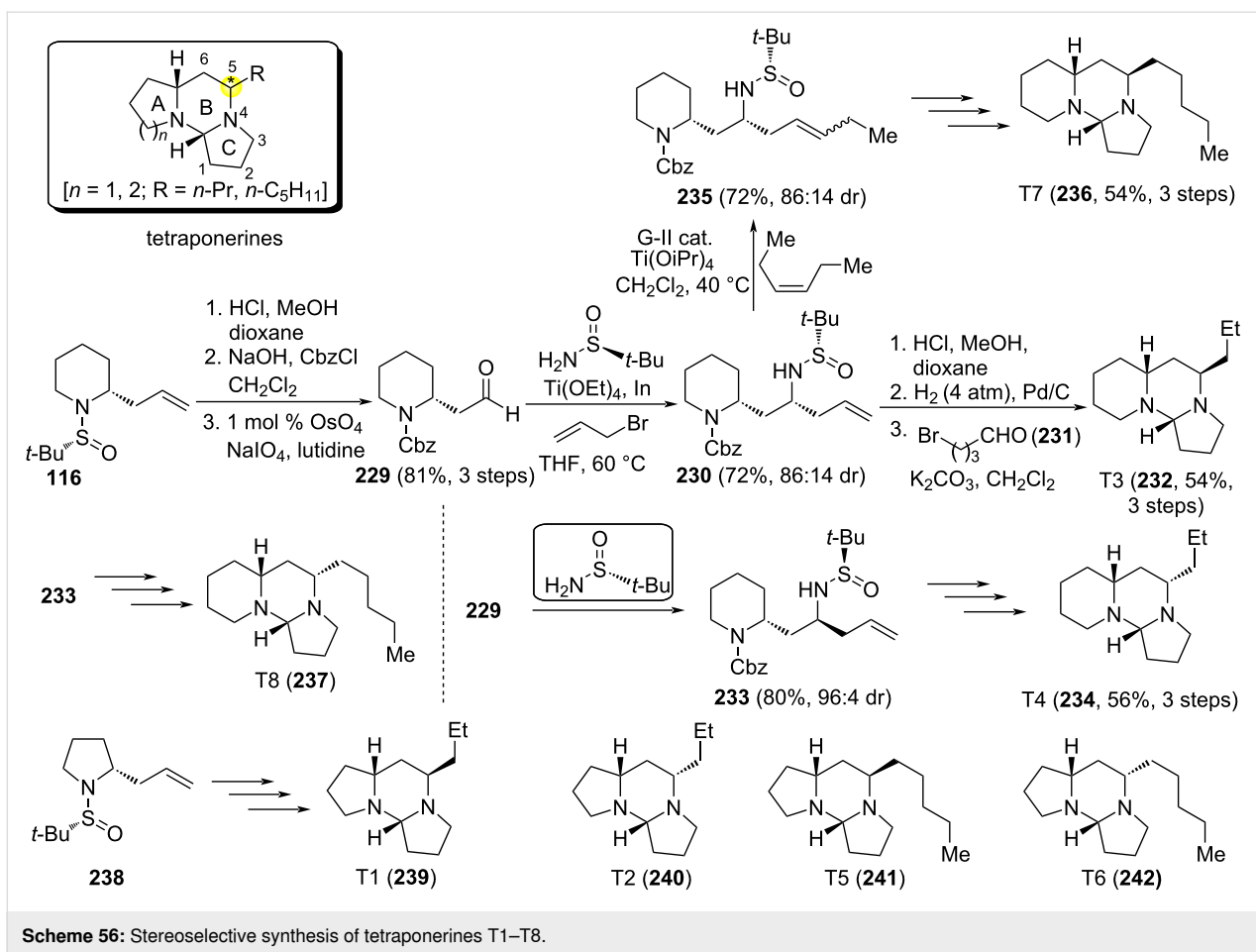
Tetraonerines T1–T8 are tricyclic alkaloids with aminal structure, and depending on the size of the A–B–C rings, they are divided in two groups (5–6–5: T1, T2, T5, T6; or 6–6–5: T3, T4, T7, T8). Other differentiating elements are the alkyl chain at C5-position (*n*-propyl: T1–T4; *n*-pentyl: T5–T8), and the configuration of this stereocenter [(*R*): T1, T3, T5, T7; (*S*): T2, T4, T6, T8] (Scheme 56). The stereocontrolled synthesis of these alkaloids was reported by Bosque et al. Key step transformations in the stereoselective synthesis of each natural tetraonerine are two consecutive indium-mediated aminoallylations of the appropriate stereoisomer of a chiral *N*-*tert*-butyl-sulfinamide. Allylpyrrolidine derivative **238**, which is the precursor of 5–6–5 tetraonerines (T1, T2, T5, T6; **185–187**), was obtained from 4-bromobutanal (**231**) in the first aminoallylation, and allyl piperidine derivative **126** (see Scheme 32; precursor of 6–6–5 tetraonerines: T3, T4, T7, T8; **231**, **233**, **235** and **237**, respectively) was prepared from 5-bromopentanal. Importantly, to prepare tetraonerines T5–T8, with a pentyl group at C5-position, a cross-metathesis reaction involving the allyl group of the second aminoallylation with *cis*-3-hexene was

carried out in order to elongate the side chain [152,153]. The anticancer activity of tetraonerines T5–T8 against four different carcinoma human cell lines was also investigated, observing a promising cytotoxic activity of tetraonerine T7 (**236**) against breast carcinoma cell line MCF-7 [152].

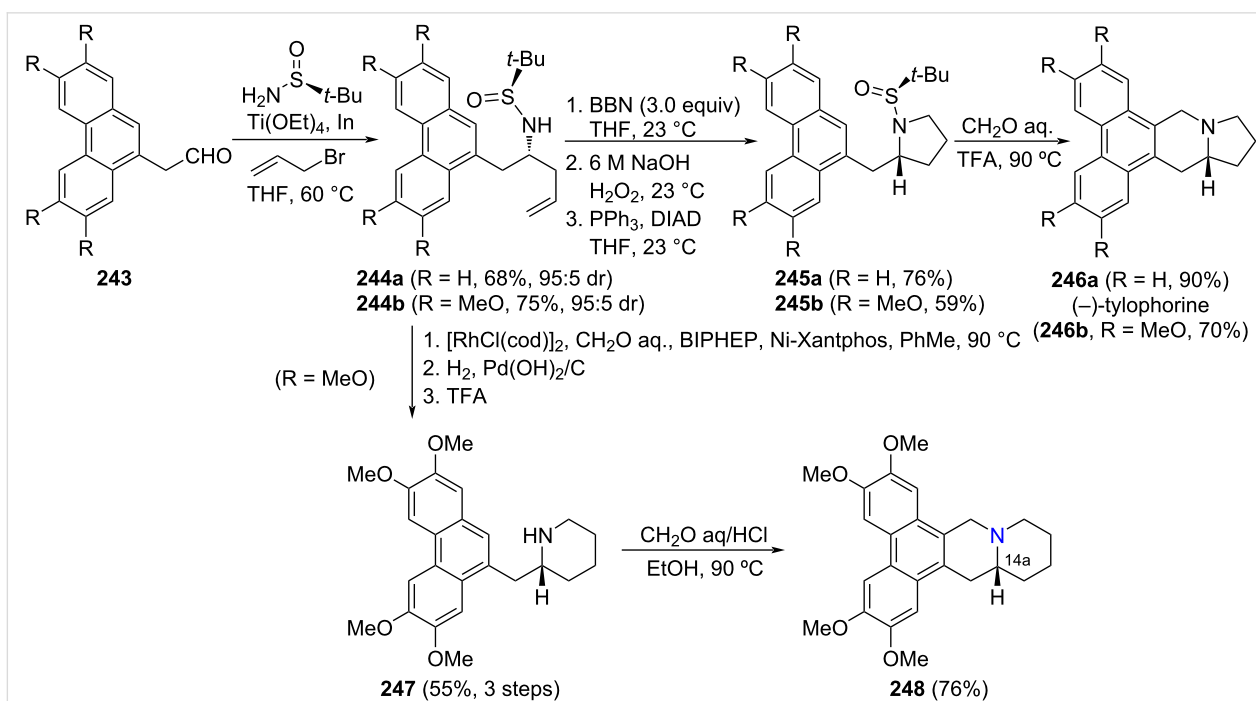
Taking advantage of this highly diastereoselective indium-mediated amino allylation of carbonyl compounds, an efficient stereocontrolled synthesis of phenanthroindolizidines **246** and phenanthroquinolizidine 7-methoxycryptopleurine **248** was accomplished by Antón-Torrecillas et al., using 2-(phenanthren-9-yl)acetaldehydes **243** as starting materials. The initially formed homoallylamine derivatives **244** were transformed first into pyrrolidines **245** (hydroboration–oxidation–intramolecular Mitsunobu *N*-alkylation), and after removal of the sulfinyl group, and a Pictet–Spengler reaction involving formaldehyde, the expected phenanthroindolizidine **246a** and the alkaloid (–)-tylophorine (**246b**) were obtained (Scheme 57) [154]. On the other hand, key chiral homoallylic sulfinamide intermediate **244b** was also transformed in four steps into enantio-enriched 7-methoxycryptopleurine **248**, a rhodium-catalyzed linear hydroformylation being one of the steps involved in the formation of piperidine derivative **247**. Cytotoxic evaluation of both enantiomers of 7-methoxycryptopleurine demonstrated that the compound with (*R*)-configuration shown in Scheme 57 was much more potent than its antipode against four cancer cell lines examined [155]. Phenanthroquinolizidines with a quaternary center at C-14a position, bearing a methyl group instead of the proton, were prepared following the same methodology, and using the corresponding methyl ketone as starting material. These compounds displayed also cytotoxic activity against different human cancer cell lines [156].







Scheme 56: Stereoselective synthesis of tetraponerines T1–T8.



Scheme 57: Stereoselective synthesis of phenanthroindolizidines 246a and (–)-tylophorine (246b), and phenanthroquinolizidine 248.

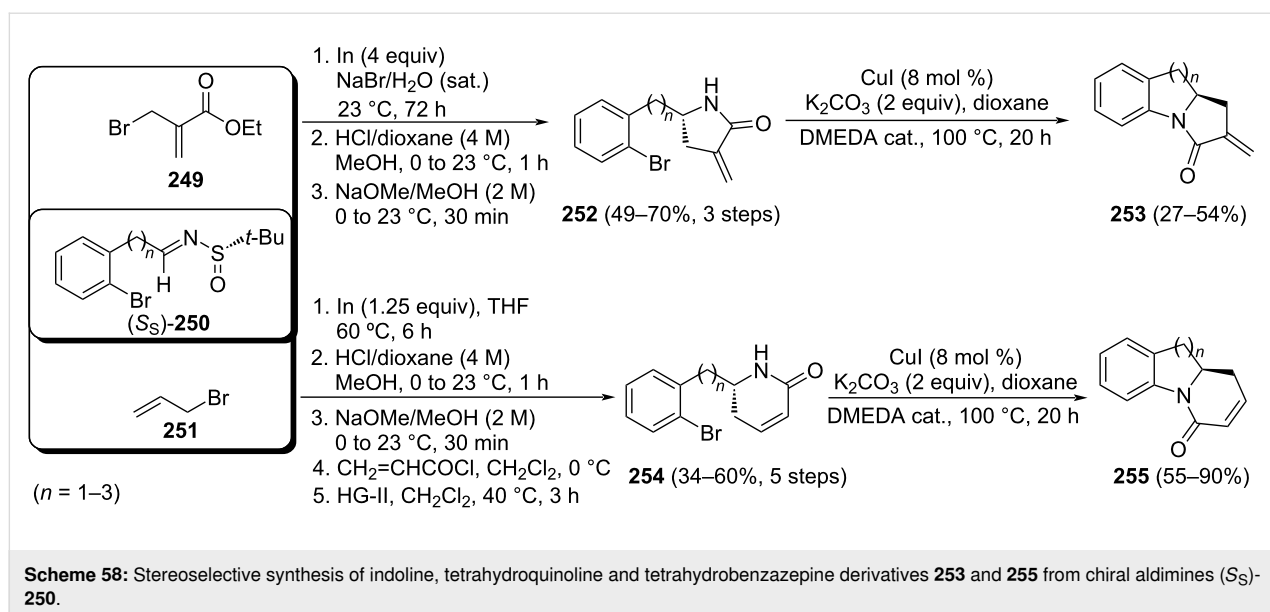
Indium-mediated allylation of sulfinyl imines was also the source of the stereocontrol in the synthesis of benzo-fused 1-azabicyclo[*j.k.0*]alkanes **253** and **255** reported by Sirvent et al.. The starting chiral imines (*S<sub>S</sub>*)-**250** derived from aliphatic aldehydes, with a 2-bromophenyl substituent, and (*S*)-*tert*-butanesulfinamide. When the allylation was carried out with ethyl 2-(bromomethyl)acrylate (**249**), and after removal of the sulfinyl unit, the resulting free amine derivative led to  $\alpha$ -methylene- $\gamma$ -butyrolactams **252**. On the other hand, dihydropyridin-2-ones **254** were obtained after sequential allylation with allyl bromide **251**, desulfinylation, acylation with acryloyl chloride, and ring-closing metathesis. Lactams **252** and **254** were easily transformed into target polycyclic compounds **253** and **255** by performing an intramolecular *N*-arylation using Ullmann-type reaction conditions (Scheme 58) [157].

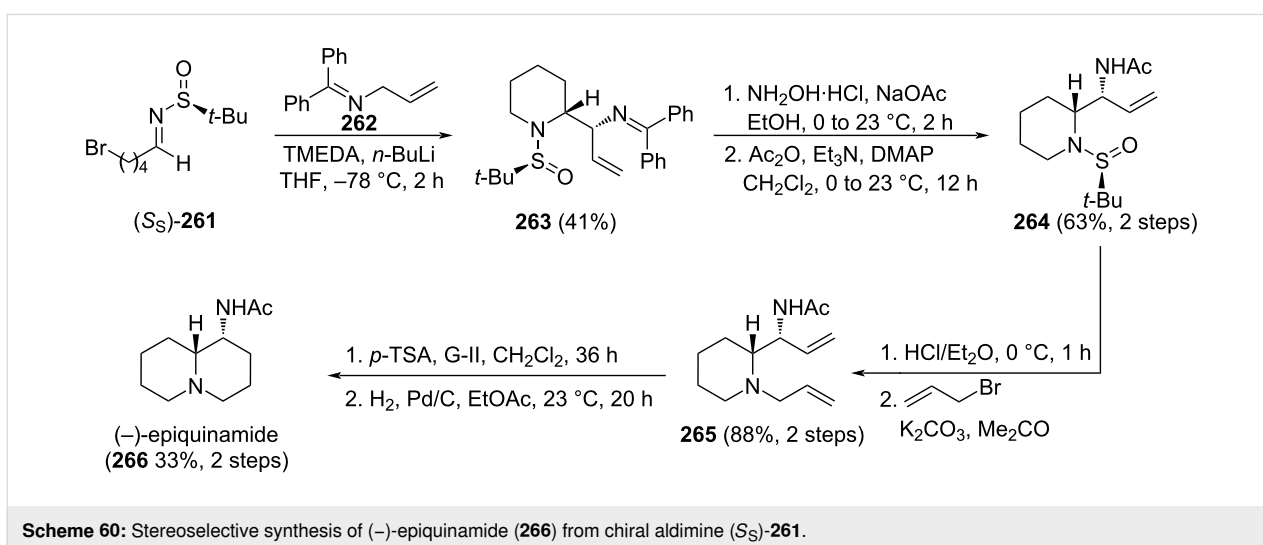
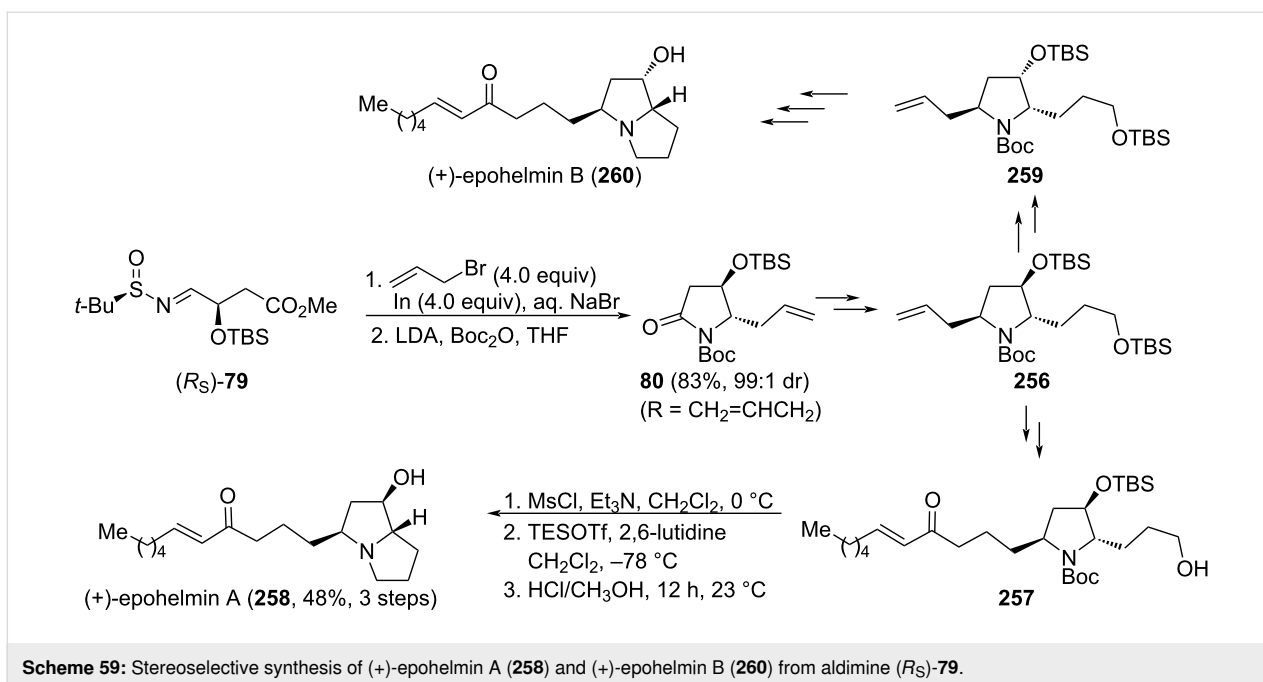
The group of Wei reported a diastereoselective approach for the synthesis of *trans*-4-hydroxy-5-allyl-2-pyrrolidinone **80** through an indium-mediated allylation of  $\alpha$ -chiral aldimine (*R<sub>S</sub>*)-**79** (see above Scheme 24) [100]. Allyl pyrrolidone **80** was an intermediate in the synthesis of alkaloids (+)-epohelmin A (**258**) and B (**260**). These natural products were isolated from an unidentified fungus, and inhibited recombinant lanosterol synthase with low IC<sub>50</sub> values, with potential use as anticholesteraeic drugs to complement or even substitute the now widely used members of the statin family. The second five-membered ring of the pyrrolizidinic system in (+)-epohelmin A (**258**) was constructed from alcohol **257**, upon forming the corresponding mesylate, subsequent removal of the Boc group, promoting cyclization, and final desilylation. Compound **256** was the last common intermediate in the diastereodivergent approach to both epohelmins, the corresponding epimer **259** was the precursor of

(+)-epohelmin B (**260**) by applying the same reaction conditions as for (+)-epohelmin A (**258**, Scheme 59) [158].

The addition of a lithium anion of *N*-(diphenylmethylidene)allylimine (**262**) to chiral sulfinyl imines was investigated by Prasad and co-workers. They found that the reaction with imines derived from aliphatic aldehydes afforded 1,2-diamine derivatives with excellent diastereoselectivity (>99:1). Allylation of chiral imine (*S<sub>S</sub>*)-**261** with concomitant cyclization led to piperidine **263** as a single diastereoisomer. Removal of diphenylmethylidene group and acylation of the resulting free amine provided compound **264**, which after desulfinylation, *N*-allylation, ring-closing metathesis and catalytic hydrogenation produced the quinolizidine alkaloid (–)-epiquinamide (**266**), isolated from the skin of the Ecuadoran frog *Epipedobates tricolor* (Scheme 60) [159].

Fustero and co-workers described for the first time the use of *N-tert*-butanesulfinamide in a desymmetrization-type process involving an intramolecular aza-Michael reaction for obtaining the advanced intermediates **271a** and **271b** in the total synthesis of (–)-hippodamine (**273**) and (+)-*epi*-hippodamine (**272**). The condensation reaction between the symmetric ketone **267** and (*R*)-*N-tert*-butanesulfinamide in the presence of titanium(IV) ethoxide followed by the reductive amination with NaBH<sub>4</sub> and double-direction cross-metathesis reaction led to **268a** in 50% yield and **268b** in 49% yield. These compounds were submitted to the desymmetrization process by an intramolecular aza-Michael reaction using NaH in THF. The applied conditions yielded a mixture of *cis*-**269** and *trans*-**269** diastereoisomers as major product (*cis/trans* 3:1) and a small amount of other possible isomer **270** was detected (**269a/270a** 95:5) and



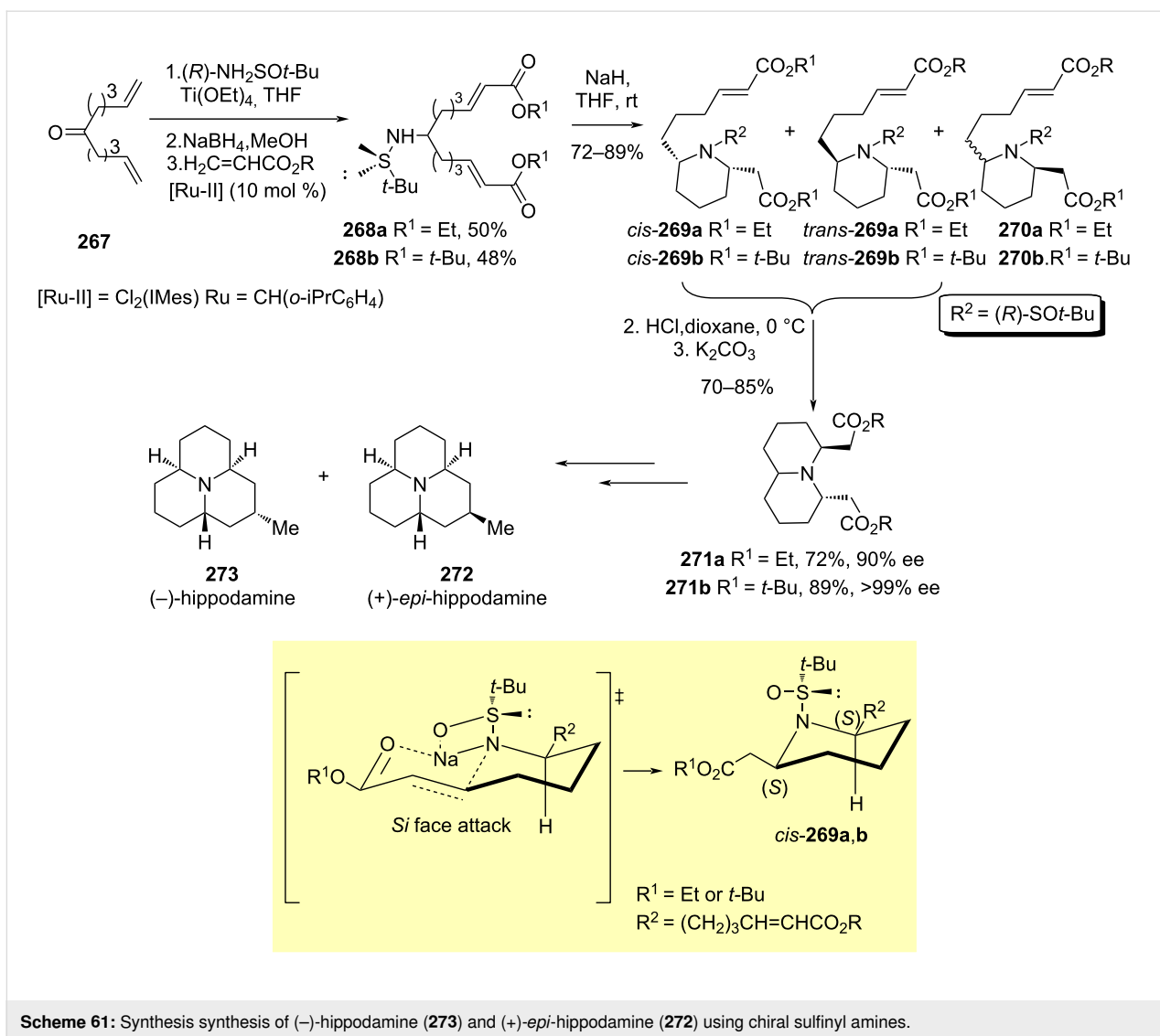


(**269b/270b** 96:4). The addition of nitrogen nucleophile occurred to the *Si* face of the conjugated ester, opposite to the bulky *tert*-butyl group gave *cis*-**269a,b** as major isomer (Scheme 61). The sulfoxide auxiliary was removed under acid conditions, and after a basification process with saturated aqueous  $\text{NaHCO}_3$ , a second intramolecular aza-Michael reaction took place to the products **271a** and **271b** with excellent yields and high diastereoselectivity. Since **271a,b** is  $C_2$ -symmetric, the cyclization of *cis*-**269a,b** or *trans* **269a,b** as soon as **270a,b** gave the same isomer **271**. Consequently, **271a** was obtained in 90% ee and **271b** in >99% ee, as determined by chiral HPLC analysis. After three steps, a mixture of (-)-hippodamine (**273**) and (+)-*epi*-hippodamine (**272**) were obtained in

high yield which was separated by column chromatography [160] (Scheme 61).

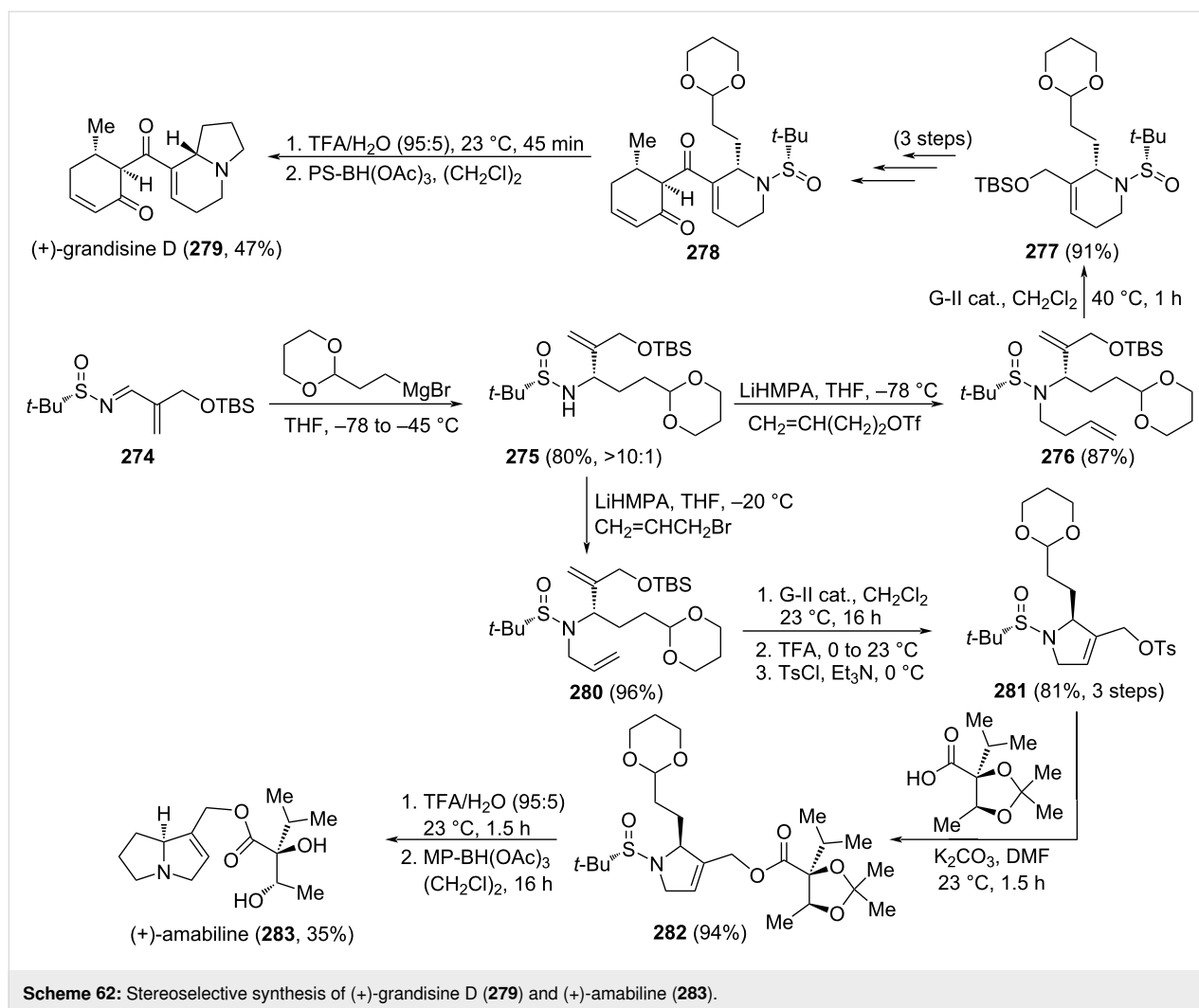
### Stereocontrol synthesis through addition of Grignard reagents to *N*-sulfinyl imines

The group of Lindsley developed new methodologies for the enantioselective synthesis of different azabicyclic systems. These methodologies are based on the diastereoselective addition of Grignard reagents to chiral sulfinyl imines, and applied to the syntheses of the indolizidine alkaloid (+)-grandisine D (**279**) [161], and pyrrolizidine alkaloid (+)-amabiline (**283**) [162], starting from imine **274**. The addition of a masked acetal organomagnesium compound to sulfinyl imine **274** provided



compound **275** in high yield and diastereoselectivity, which after reaction with the appropriate alkylating reagent led to **276** and **280**. These compounds were prone to undergo ring-closing metathesis, generating 6- and 5-membered ring systems, respectively. After several steps, acetal derivatives **278** and **282** were transformed into (+)-grandisine D (**279**) and (+)-amabiline (**283**). Treatment under acidic conditions of **278** and **282** produced removal of the sulfanyl group and hydrolysis of the acetal, revealing the masked aldehyde. In the final step, an intramolecular hydroamination allowed the formation of the five-membered ring, to complete the construction of the azabicyclic arrays (Scheme 62). The indolizidine alkaloids grandisines were isolated from leaves of the Australian rain forest tree *Elaeocarpus grandis*, and these alkaloids display selective human  $\delta$ -opioid receptor affinity. Meanwhile, hepatotoxic (+)-amabiline (**283**) was found in the seeds and flowers of *Borago officinalis*.

The addition of alkynyl or alkenyl Grignard reagents to chiral sulfanyl imine (*S<sub>S</sub>*)-**126** led to 6-substituted *trans*-5-hydroxy-2-piperidinones with high diastereoselectivity, which is controlled mainly for the configuration of the stereocenter bearing the silyloxy group with alkenyl reagents, and by the coordination of the silyloxy substitution and stereochemistry of the sulfonamide with alkynyl organomagnesium compounds. The addition of vinylmagnesium bromide to (*S<sub>S</sub>*)-**126** led, after *N*-Boc protection to 2-piperidinone **284**, a precursor of diolefin **285**, which through ring closing metathesis allowed the synthesis of quinolizidinone **286**, from which (-)-epiquinamide (**266**) was prepared after three steps. On the other hand, the reaction of imine (*S<sub>S</sub>*)-**126** with ethynylmagnesium bromide provided compound **288**, which was transformed into the *N*-allyl-substituted piperidinone **289**. Again, a ring-closing metathesis allowed the formation of the bicyclic indolizidinone system **290**, an advanced synthetic intermediate of alkaloid (+)-swainsonine (**291**,



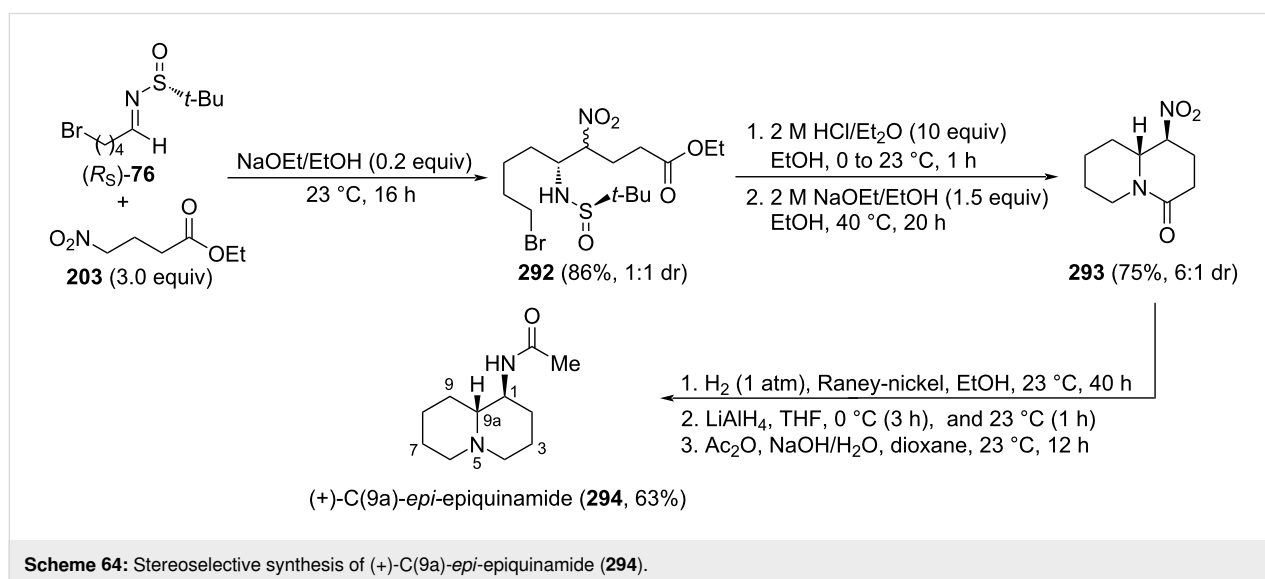
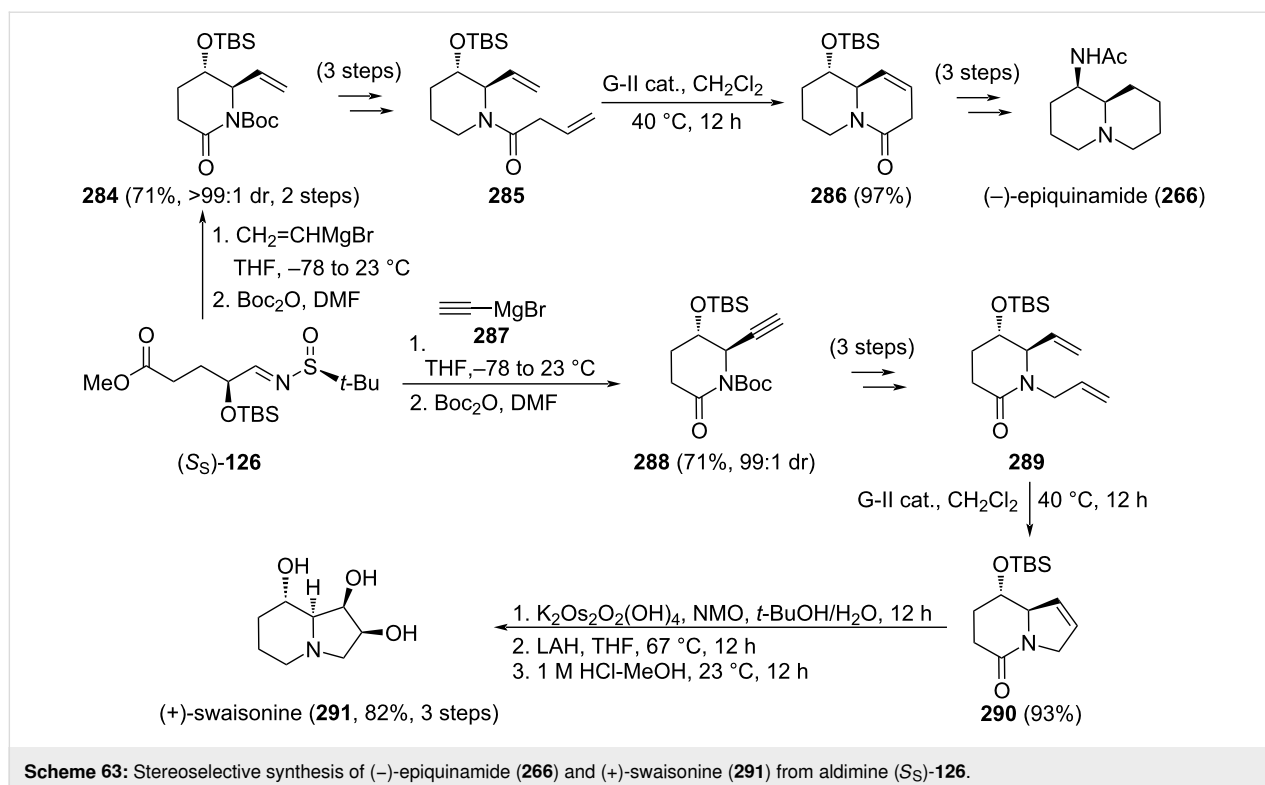
Scheme 63) [163]. This alkaloid was found in the fungus *Rhizoctonia leguminicola* and in plants such as the Australian *Swainsona canescens* and the North American locoweed *Astragalus lentiginosus*. It displays a wide range of biological activities, including alteri neoplastic growth and metastasis.

### Stereocontrol synthesis through Mannich, nitro-Mannich and Mukaiyama–Mannich reactions of *N*-sulfinyl imines

Heterocyclic compounds can be obtained through strategies using Mannich reactions [164–166]. It was previously commented that the addition of ethyl 4-nitrobutanoate (**203**) to chiral sulfinyl imines under basic solvent-free conditions proceeded with high facial diastereoselectivity (Scheme 64 [143]). When the reaction was performed with the imine of 5-bromopentanal, (*R*<sub>S</sub>)-**76**, the nitro-Mannich adduct **292** was formed in 86% yield as a 1:1 mixture of diastereoisomers, considering the stereocenter bearing the nitro group. Taking advantage of this methodology, Benlahrech et al. reported the

synthesis of (+)-*C*(9a)-*epi*-epiquinamide (**294**) [167]. Desulfinylation of compound **292** led to quinolizidine derivative **293**, taking place two consecutive cyclizations involving the resulting free amino group, the ester and the alkyl bromide functionalities. Although compound **292** was isolated in a 1:1 dr, quinolizidine **293** was formed as a 6:1 mixture of diastereoisomers, due to a rapid epimerization under basic conditions, leading to the thermodynamic most stable isomer. Sequential reductions of the nitro group, the lactam and final *N*-acetylation of the primary amine were the last steps of this synthesis (Scheme 64).

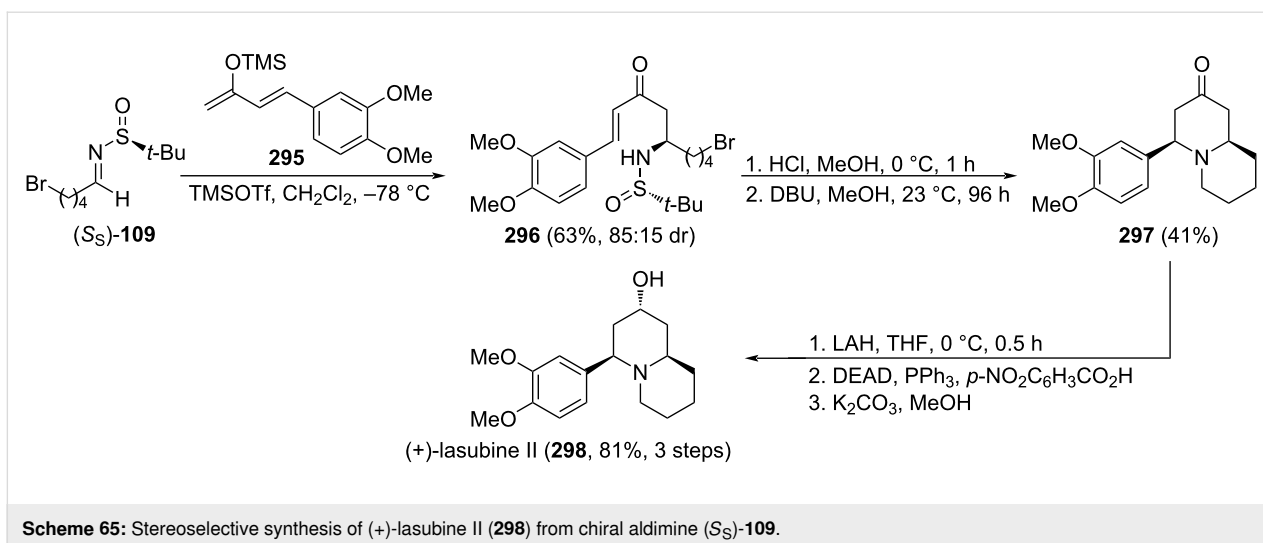
The Lewis acid-promoted addition of silyl enol ethers to chiral sulfinyl imines afforded  $\beta$ -amino ketones in high yields and diastereoselectivities (see above Scheme 24 [100]). Similar levels of stereoselectivity were found performing this type of Mukaiyama–Mannich reactions with enol ethers derived from methyl-conjugated enones. The reaction of chiral imine (*S*<sub>S</sub>)-**109** with silyl enol ether **295** led to  $\beta$ -amino enone **296**. Removal of the sulfinyl group under acidic conditions and



subsequent basic treatment with DBU led, after *N*-alkylation and intramolecular conjugate addition, to *cis*-quinolizidinone **297** in 41% yield (35% of the *trans*-isomer). Reduction of the ketone functionality with LAH gave 2-*epi*-lasubine II, and Mitsunobu inversion of the configuration provide (+)-lasubine II (**298**) in 81% yield (Scheme 65) [168].

A straightforward synthesis of the alkaloids (-)-epimyrtine (**300a**) and (-)-lasubine II (*ent*-**302**) from  $\beta$ -amino ketone

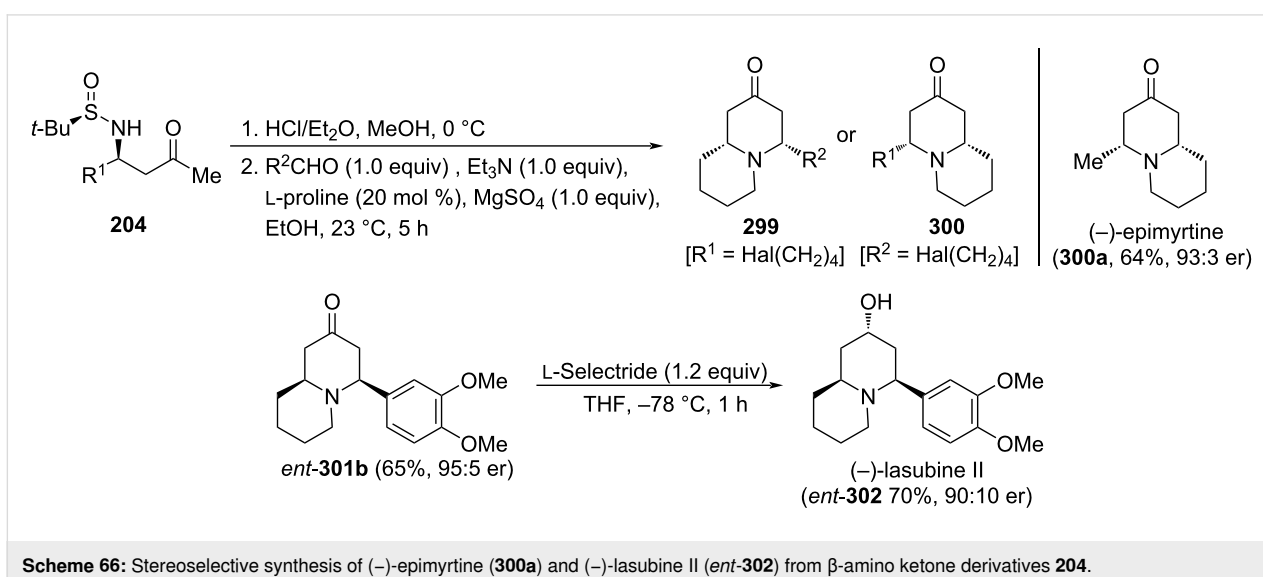
derivatives **204** was reported by Lahosa et al. These compounds participate in a *L*-proline organocatalyzed intramolecular Mannich reaction with a second aldehyde ( $R^2\text{CHO}$ ) to give *cis*-2,6-disubstituted piperidin-4-ones **205** (see above Scheme 49 [141]). However, when one of the aldehydes involved in the formation of  $\beta$ -amino ketone **204** ( $R^1\text{CHO}$ ) or in the intramolecular Mannich reaction ( $R^2\text{CHO}$ ) which is 5-chloro- or 5-bromopentanal, quinolizidinone derivatives **299** or **300** are formed. In both cases, after



formation of the piperidine ring through the expected Mannich condensation, a subsequent intramolecular *N*-alkylation involving the carbon–halogen bond occurred to produce the quinolizidinic systems. The natural alkaloid (–)-epimyrtiline (**300a**), isolated from bilberry (*Vaccinium myrtillus*) was prepared following this methodology, along with *ent*-**299b**, which was transformed in a single step into (–)-lasubine II (*ent*-**302**), by diastereoselective reduction with *L*-selectride (Scheme 66) [141].

Andrade [8,169,170] and Zhao reported an elegant and efficient methodology for the rapid assembling of chiral *N*-*tert*-butanesulfinyl imine ( $R_S$ )-**303** with methyl 2-ethylacrylate to give tricyclic derivative **305**, in high yield and diastereoselectivity. Treatment of ( $R_S$ )-**303** with LiHMDS in THF at  $-78$  °C led to the formation of lithiodienamine intermediate **304**, which

reacted with methyl 2-ethylacrylate, leading to the Michael addition product, and the resulting enolate participated in an intramolecular Mannich reaction to form a six-membered ring. Final addition of an excess of allyl bromide yielded tetrahydrocarbazole **305** in 90% yield. After convenient functional group transformation, diolefin **306** was submitted to ring-closing metathesis to form tetracycle **307**. Further removal of the sulfinyl group and *N*-alkylation with 2-bromoethanol gave compound **308**, which is an ideal substrate for the construction of the five-membered ring leading to spiro compound **309**, upon reaction with potassium *tert*-butoxide (Bosch–Rubiralta spirocyclization). Treatment of **309** with LDA at  $-78$  °C and quenching the resulting metalloenamine with methyl cyanofornate furnished (–)-tabersonine (**310**), and selective hydrogenation of **310** led to (–)-vincadifformine (**311**). In addition, double hydrogenation of **309** produced (–)-aspidospermidine (**312**) in 75% yield. All



these alkaloids are members of the *Aspidosperma* family (Scheme 67) [171,172].

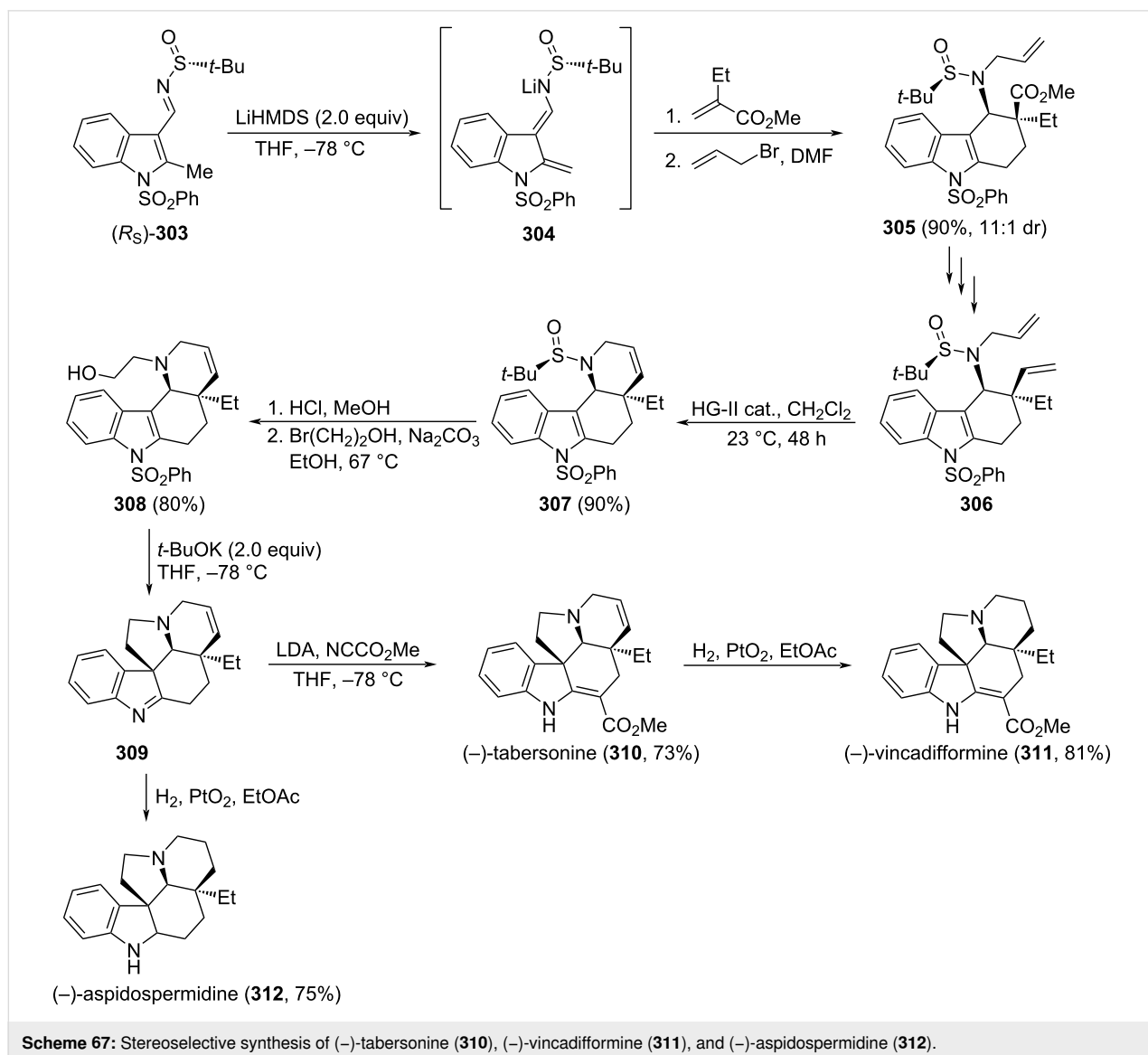
### Stereocontrol by pinacol-type coupling reactions involving N-sulfinyl imines

The alkaloids (+)-epohelmin A (**258**) and (+)-epohelmin B (**260**) were prepared from chiral aldehyde **313** and chiral sulfinyl imine **314**, through samarium iodide-induced reductive cross-coupling reaction leading to compound **315** in 76% yield, and total diastereoselectivity. After several steps, enone amino triol derivative **316** was prepared and directly transformed into (+)-epohelmin A (**258**), by applying the same reaction conditions shown in Scheme 59 [158] for the transformation of pyrrolidine derivative **257** into (+)-epohelmin A (**258**). The epimer (+)-epohelmin B (**260**) was also prepared from **258**, carrying out an oxidation of the alcohol and a diastereoselec-

tive reduction with L-selectride of the resulting cyclic ketone (Scheme 68) [173].

### Asymmetric synthesis of lysergic acid

In 2013, Liu and colleagues used *N-tert*-butanesulfinamide in the total synthesis of (+)-lysergic acid (**323**). The first step was the aminoallylation of aldehyde **317** with allyl bromide and sulfinamide (*R<sub>S</sub>*)-**1** by means of indium metal, leading to homoallyl derivative **318**. Further steps included an *N*-alkylation and ring-closing metathesis with Grubbs second generation catalyst to generate tetrahydropyridine **320** in high yields. Removal of the silyl protecting group, followed by oxidation of the resulting terminal alcohol and palladium-catalyzed coupling of the aldehyde with 3-chloro-2-iodoaniline afforded indole derivative **322**, which is an advanced precursor of target (+)-lysergic acid (**323**, Scheme 69) [174].



**Scheme 67:** Stereoselective synthesis of (-)-tabersonine (**310**), (-)-vincadifformine (**311**), and (-)-aspidospermidine (**312**).





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## References

- Robak, M. T.; Herbage, M. A.; Ellman, J. A. *Chem. Rev.* **2010**, *110*, 3600–3740. doi:10.1021/cr900382t
- Ferreira, F.; Botuha, C.; Chemla, F.; Pérez-Luna, A. *Chem. Soc. Rev.* **2009**, *38*, 1162–1186. doi:10.1039/b809772k
- Morton, D.; Stockman, R. A. *Tetrahedron* **2006**, *62*, 8869–8905. doi:10.1016/j.tet.2006.06.107
- Wakayama, M.; Ellman, J. A. *J. Org. Chem.* **2009**, *74*, 2646–2650. doi:10.1021/jo9001883
- Aggarwal, V. K.; Barbero, N.; McGarrigle, E. M.; Mickle, G.; Navas, R.; Suárez, J. R.; Unthank, M. G.; Yar, M. *Tetrahedron Lett.* **2009**, *50*, 3482–3484. doi:10.1016/j.tetlet.2009.03.020
- Xia, M.-M.; Li, F.-X.; Ma, Y.-P.; Song, L.-L.; Hou, Y.-N.; Shi, Z.-F.; Cao, X.-P. *Adv. Synth. Catal.* **2020**, *362*, 1112–1124. doi:10.1002/adsc.201901543
- Ye, W.; Zhang, L.; Ni, C.; Rong, J.; Hu, J. *Chem. Commun.* **2014**, *50*, 10596–10599. doi:10.1039/c4cc05042h
- Kokkonda, P.; Andrade, R. B. *Org. Lett.* **2019**, *21*, 9594–9597. doi:10.1021/acs.orglett.9b03762
- Liu, H.; Zhang, X.; Shan, D.; Pitchakuntla, M.; Ma, Y.; Jia, Y. *Org. Lett.* **2017**, *19*, 3323–3326. doi:10.1021/acs.orglett.7b01504
- Kazak, M.; Priede, M.; Shubin, K.; Bartrum, H. E.; Poisson, J.-F.; Suna, E. *Org. Lett.* **2017**, *19*, 5356–5359. doi:10.1021/acs.orglett.7b02635
- Shao, L.-P.; Si, C.-M.; Mao, Z.-Y.; Zhou, W.; Molinski, T. F.; Wei, B.-G.; Lin, G.-Q. *Org. Chem. Front.* **2017**, *4*, 995–1004. doi:10.1039/c7qo00052a
- Tao, W.; Zhou, W.; Zhou, Z.; Si, C.-M.; Sun, X.; Wei, B.-G. *Tetrahedron* **2016**, *72*, 5928–5933. doi:10.1016/j.tet.2016.08.038
- Zhou, W.; Nie, X.-D.; Zhang, Y.; Si, C.-M.; Zhou, Z.; Sun, X.; Wei, B.-G. *Org. Biomol. Chem.* **2017**, *15*, 6119–6131. doi:10.1039/c7ob01395g
- Cogan, D. A.; Liu, G.; Kim, K.; Backes, B. J.; Ellman, J. A. *J. Am. Chem. Soc.* **1998**, *120*, 8011–8019. doi:10.1021/ja9809206
- Cogan, D. A.; Ellman, J. A. *J. Am. Chem. Soc.* **1999**, *121*, 268–269. doi:10.1021/ja983217q
- Liu, G.; Cogan, D. A.; Owens, T. D.; Tang, T. P.; Ellman, J. A. *J. Org. Chem.* **1999**, *64*, 1278–1284. doi:10.1021/jo982059i
- Jiang, Z.-Y.; Chan, W. H.; Lee, A. W. M. *J. Org. Chem.* **2005**, *70*, 1081–1083. doi:10.1021/jo048579e
- Higashibayashi, S.; Tohmiya, H.; Mori, T.; Hashimoto, K.; Nakata, M. *Synlett* **2004**, 457–460. doi:10.1055/s-2004-815409
- Huang, Z.; Zhang, M.; Wang, Y.; Qin, Y. *Synlett* **2005**, 1334–1336. doi:10.1055/s-2005-865234
- Collados, J. F.; Toledano, E.; Guijarro, D.; Yus, M. *J. Org. Chem.* **2012**, *77*, 5744–5750. doi:10.1021/jo300919x
- Tabet, S.; Rodeville, N.; Boiteau, J.-G.; Cardinaud, I. *Org. Process Res. Dev.* **2016**, *20*, 1383–1387. doi:10.1021/acs.oprd.6b00188
- Davis, F. A.; Friedman, A. J.; Kluger, E. W. *J. Am. Chem. Soc.* **1974**, *96*, 5000–5001. doi:10.1021/ja00822a055
- Davis, F. A.; Reddy, R. T.; Reddy, R. E. *J. Org. Chem.* **1992**, *57*, 6387–6389. doi:10.1021/jo00050a007
- Davis, F. A.; Reddy, R. E.; Szweczyk, J. M.; Portonovo, P. S. *Tetrahedron Lett.* **1993**, *34*, 6229–6232. doi:10.1016/s0040-4039(00)73717-8
- Davis, F. A.; Zhou, P.; Reddy, G. V. *J. Org. Chem.* **1994**, *59*, 3243–3245. doi:10.1021/jo00091a001
- Davis, F. A.; Liu, H.; Zhou, P.; Fang, T.; Reddy, G. V.; Zhang, Y. *J. Org. Chem.* **1999**, *64*, 7559–7567. doi:10.1021/jo990907j
- Davis, F. A.; Reddy, R. E.; Portonovo, P. S. *Tetrahedron Lett.* **1994**, *35*, 9351–9354. doi:10.1016/s0040-4039(00)78540-6
- Davis, F. A.; Srirajan, V.; Titus, D. D. *J. Org. Chem.* **1999**, *64*, 6931–6934. doi:10.1021/jo990947n
- Jiang, J.; Schumacher, K. K.; Joullié, M. M.; Davis, F. A.; Reddy, R. E. *Tetrahedron Lett.* **1994**, *35*, 2121–2124. doi:10.1016/s0040-4039(00)76775-x
- Davis, F. A.; McCoull, W.; Titus, D. D. *Org. Lett.* **1999**, *1*, 1053–1055. doi:10.1021/ol990855k
- Davis, F. A. *J. Org. Chem.* **2006**, *71*, 8993–9003. doi:10.1021/jo061027p
- Cainelli, G.; Panunzio, M.; Giacomini, D. *Tetrahedron Lett.* **1991**, *32*, 121–124. doi:10.1016/s0040-4039(00)71234-2
- Davis, F. A.; Zhang, Y.; Rao, A.; Zhang, Z. *Tetrahedron* **2001**, *57*, 6345–6352. doi:10.1016/s0040-4020(01)00500-2
- Borg, G.; Cogan, D. A.; Ellman, J. A. *Tetrahedron Lett.* **1999**, *40*, 6709–6712. doi:10.1016/s0040-4039(99)01351-9
- Kochi, T.; Ellman, J. A. *J. Am. Chem. Soc.* **2004**, *126*, 15652–15653. doi:10.1021/ja044753n
- Colyer, J. T.; Andersen, N. G.; Tedrow, J. S.; Soukup, T. S.; Faul, M. M. *J. Org. Chem.* **2006**, *71*, 6859–6862. doi:10.1021/jo0609834
- Gatling, S. C.; Jackson, J. E. *J. Am. Chem. Soc.* **1999**, *121*, 8655–8656. doi:10.1021/ja991784n
- Wigfield, D. C.; Gowland, F. W. *J. Org. Chem.* **1977**, *42*, 1108–1109. doi:10.1021/jo00426a048
- Sirvent, A.; Foubelo, F. *Org. Chem.* **2018**, *15*, 345–348. doi:10.2174/1570178614666171130162244
- Mejuch, T.; Gilboa, N.; Gayon, E.; Wang, H.; Houk, K. N.; Marek, I. *Acc. Chem. Res.* **2013**, *46*, 1659–1669. doi:10.1021/ar4000532
- Zimmerman, H. E.; Traxler, M. D. *J. Am. Chem. Soc.* **1957**, *79*, 1920–1923. doi:10.1021/ja01565a041
- Yamamoto, Y.; Asao, N. *Chem. Rev.* **1993**, *93*, 2207–2293. doi:10.1021/cr00022a010
- Yamamoto, Y.; Komatsu, T.; Maruyama, K. *J. Org. Chem.* **1985**, *50*, 3115–3121. doi:10.1021/jo00217a020
- Plobbeck, N.; Powell, D. *Tetrahedron: Asymmetry* **2002**, *13*, 303–310. doi:10.1016/s0957-4166(02)00099-x

45. Pflum, D. A.; Krishnamurthy, D.; Han, Z.; Wald, S. A.; Senanayake, C. H. *Tetrahedron Lett.* **2002**, *43*, 923–926. doi:10.1016/s0040-4039(01)02294-8
46. Almansa, R.; Guijarro, D.; Yus, M. *Tetrahedron: Asymmetry* **2008**, *19*, 603–606. doi:10.1016/j.tetasy.2008.01.038
47. Almansa, R.; Guijarro, D.; Yus, M. *Tetrahedron: Asymmetry* **2008**, *19*, 2484–2491. doi:10.1016/j.tetasy.2008.10.012
48. Almansa, R.; Guijarro, D.; Yus, M. *Tetrahedron Lett.* **2009**, *50*, 4188–4190. doi:10.1016/j.tetlet.2009.05.008
49. Almansa, R.; Collados, J. F.; Guijarro, D.; Yus, M. *Tetrahedron: Asymmetry* **2010**, *21*, 1421–1431. doi:10.1016/j.tetasy.2010.03.046
50. Gonzalez-Gomez, J. C.; Medjahdi, M.; Foubelo, F.; Yus, M. *J. Org. Chem.* **2010**, *75*, 6308–6311. doi:10.1021/jo101379u
51. McCoull, W.; Davis, F. A. *Synthesis* **2000**, 1347–1365. doi:10.1055/s-2000-7097
52. Bulut, A.; Aslan, A.; Dogan, Ö. *J. Org. Chem.* **2008**, *73*, 7373–7375. doi:10.1021/jo8010073
53. Chi, Y.; Scroggins, S. T.; Boz, E.; Frechet, J. M. J. *J. Am. Chem. Soc.* **2008**, *130*, 17287–17289. doi:10.1021/ja806584q
54. Song, X.; Hua, Y.-Z.; Shi, J.-G.; Sun, P.-P.; Wang, M.-C.; Chang, J. *J. Org. Chem.* **2014**, *79*, 6087–6093. doi:10.1021/jo500796w
55. Pieczonka, A. M.; Jarzyński, S.; Wujkowska, Z.; Leśniak, S.; Rachwalski, M. *Tetrahedron Lett.* **2015**, *56*, 6506–6507. doi:10.1016/j.tetlet.2015.10.027
56. Rachwalski, M.; Jarzyński, S.; Jasiński, M.; Leśniak, S. *Tetrahedron: Asymmetry* **2013**, *24*, 689–693. doi:10.1016/j.tetasy.2013.04.019
57. Tanner, D. *Angew. Chem., Int. Ed. Engl.* **1994**, *33*, 599–619. doi:10.1002/anie.199405991
58. Sweeney, J. B. *Chem. Soc. Rev.* **2002**, *31*, 247–258. doi:10.1039/b006015l
59. Degennaro, L.; Trinchera, P.; Luisi, R. *Chem. Rev.* **2014**, *114*, 7881–7929. doi:10.1021/cr400553c
60. Ohno, H. *Chem. Rev.* **2014**, *114*, 7784–7814. doi:10.1021/cr400543u
61. Sabir, S.; Kumar, G.; Jat, J. L. *Asian J. Org. Chem.* **2017**, *6*, 782–793. doi:10.1002/ajoc.201700056
62. Morton, D.; Pearson, D.; Field, R. A.; Stockman, R. A. *Synlett* **2003**, 1985–1988. doi:10.1055/s-2003-42028
63. Fernández, I.; Valdivia, V.; Gori, B.; Alcudia, F.; Álvarez, E.; Khair, N. *Org. Lett.* **2005**, *7*, 1307–1310. doi:10.1021/ol050080h
64. Li, Y.; Ma, Y.; Lu, Z.; Wang, L.; Ren, X.; Sun, Z. *Tetrahedron Lett.* **2012**, *53*, 4711–4714. doi:10.1016/j.tetlet.2012.06.100
65. García Ruano, J.; Fernández, I.; Prado Catalina, M. d.; Cruz, A. A. *Tetrahedron: Asymmetry* **1996**, *7*, 3407–3414. doi:10.1016/s0957-4166(96)00448-x
66. Yus, M.; González-Gómez, J. C.; Foubelo, F. *Chem. Rev.* **2013**, *113*, 5595–5698. doi:10.1021/cr400008h
67. Foubelo, F.; Yus, M. *Eur. J. Org. Chem.* **2014**, 485–491. doi:10.1002/ejoc.201301068
68. Liu, W.-J.; Zhao, Y.-H.; Sun, X.-W. *Tetrahedron Lett.* **2013**, *54*, 3586–3590. doi:10.1016/j.tetlet.2013.04.046
69. Maciá, E.; Foubelo, F.; Yus, M. *Heterocycles* **2019**, *99*, 248–266. doi:10.3987/com-18-s(f)18
70. Moragas, T.; Churcher, I.; Lewis, W.; Stockman, R. A. *Org. Lett.* **2014**, *16*, 6290–6293. doi:10.1021/ol502967x
71. Zhang, Y.; Tian, H.; Luo, Z.; Liu, X.; Zhao, Y.; Chen, W.; Zhang, H. *Tetrahedron Lett.* **2018**, *59*, 3357–3360. doi:10.1016/j.tetlet.2018.07.056
72. Chogii, I.; Das, P.; Delost, M. D.; Crawford, M. N.; Njardarson, J. T. *Org. Lett.* **2018**, *20*, 4942–4945. doi:10.1021/acs.orglett.8b02074
73. Huang, Z.-A.; Liu, H.; Lu, C.-D.; Xu, Y.-J. *Org. Lett.* **2015**, *17*, 4042–4045. doi:10.1021/acs.orglett.5b01954
74. Yang, Y.; Huang, Y.; Qing, F.-L. *Tetrahedron Lett.* **2013**, *54*, 3826–3830. doi:10.1016/j.tetlet.2013.05.048
75. Qin, S.; Yao, L.; Luo, Y.; Liu, T.; Xu, J.; Sun, Y.; Wang, N.; Yan, J.; Tang, B.; Yang, G.; Yang, C. *Org. Chem. Front.* **2020**, *7*, 3132–3136. doi:10.1039/d0qo00603c
76. Reidl, T. W.; Anderson, L. L. *Asian J. Org. Chem.* **2019**, *8*, 931–945. doi:10.1002/ajoc.201900229
77. Andresini, M.; Degennaro, L.; Luisi, R. *Org. Biomol. Chem.* **2020**, *18*, 5798–5810. doi:10.1039/d0ob01251c
78. Arya, N.; Jagdale, A. Y.; Patil, T. A.; Yeramwar, S. S.; Holikatti, S. S.; Dwivedi, J.; Shishoo, C. J.; Jain, K. S. *Eur. J. Med. Chem.* **2014**, *74*, 619–656. doi:10.1016/j.ejmech.2014.01.002
79. Gandhi, D.; Kalal, P.; Agarwal, S. *Chem. Biol. Interface* **2017**, *7*, 79–101.
80. Zachariah, S. M.; Ramkumar, M.; George, N.; Ashif, M. S. *Int. J. Pharm. Sci. Rev. Res.* **2015**, *30*, 211–218.
81. Reddy, L. R.; Waman, Y.; Kallure, P.; Nalivela, K. S.; Begum, Z.; Divya, T.; Kotturi, S. *Chem. Commun.* **2019**, *55*, 5068–5070. doi:10.1039/c9cc00863b
82. Reddy, L. R.; Waman, Y.; Nayak, K.; Baharooni, K.; Kotturi, S. *Org. Lett.* **2019**, *21*, 3481–3484. doi:10.1021/acs.orglett.9b00564
83. Lima, L. M.; da Silva, B. N. M.; Barbosa, G.; Barreiro, E. J. *Eur. J. Med. Chem.* **2020**, *208*, 112829. doi:10.1016/j.ejmech.2020.112829
84. Delgado, O.; Monteagudo, A.; Van Gool, M.; Trabanco, A. A.; Fustero, S. *Org. Biomol. Chem.* **2012**, *10*, 6758–6766. doi:10.1039/c2ob25845e
85. Edupuganti, R.; Davis, F. A. *Org. Biomol. Chem.* **2012**, *10*, 5021–5031. doi:10.1039/c2ob25345c
86. Albrecht, B. K.; Gehling, V. S.; Hewitt, M. C.; Vaswani, R. G.; Côté, A.; Leblanc, Y.; Nasveschuk, C. G.; Bellon, S.; Bergeron, L.; Campbell, R.; Cantone, N.; Cooper, M. R.; Cummings, R. T.; Jayaram, H.; Joshi, S.; Mertz, J. A.; Neiss, A.; Normant, E.; O'Meara, M.; Pardo, E.; Poy, F.; Sandy, P.; Supko, J.; Sims, R. J., III; Harmange, J.-C.; Taylor, A. M.; Audia, J. E. *J. Med. Chem.* **2016**, *59*, 1330–1339. doi:10.1021/acs.jmedchem.5b01882
87. Yang, X.-d.; Dong, C.-m.; Chen, J.; Ding, Y.-h.; Liu, Q.; Ma, X.-y.; Zhang, Q.; Chen, Y. *Chem. – Asian J.* **2013**, *8*, 1213–1222. doi:10.1002/asia.201300051
88. Su, L.; Xu, M.-H. *Synthesis* **2016**, *48*, 2595–2602. doi:10.1055/s-0035-1561426
89. Dema, H. K.; Foubelo, F.; Yus, M. *Helv. Chim. Acta* **2012**, *95*, 1790–1798. doi:10.1002/hlca.201200303
90. Lin, Y.-C.; Ribaucourt, A.; Moazami, Y.; Pierce, J. G. *J. Am. Chem. Soc.* **2020**, *142*, 9850–9857. doi:10.1021/jacs.0c04091
91. Adrio, J.; Carretero, J. C. *Chem. Commun.* **2019**, *55*, 11979–11991. doi:10.1039/c9cc05238k
92. Nájera, C.; Sansano, J. M. *Pure Appl. Chem.* **2019**, *91*, 575–596. doi:10.1515/pac-2018-0710
93. Upadhyay, P.; Srivastava, V. *Curr. Organocatal.* **2016**, *3*, 243–269. doi:10.2174/2213337202666150812230640
94. Rajput, A. P.; Nagarale, D. V. *J. Chem. Pharm. Res.* **2016**, *8*, 557–575.
95. Reddy, L. R.; Gupta, A. P.; Villhauer, E.; Liu, Y. *J. Org. Chem.* **2012**, *77*, 1095–1100. doi:10.1021/jo2024224

96. Fustero, S.; Herrera, L.; Lázaro, R.; Rodríguez, E.; Maestro, M. A.; Mateu, N.; Barrio, P. *Chem. – Eur. J.* **2013**, *19*, 11776–11785. doi:10.1002/chem.201301591
97. Redford, J. E.; McDonald, R. I.; Rigsby, M. L.; Wiensch, J. D.; Stahl, S. S. *Org. Lett.* **2012**, *14*, 1242–1245. doi:10.1021/ol3000519
98. Zhou, Q.-R.; Wei, X.-Y.; Li, Y.-Q.; Huang, D.; Wei, B.-G. *Tetrahedron* **2014**, *70*, 4799–4808. doi:10.1016/j.tet.2014.05.037
99. Si, C.-M.; Mao, Z.-Y.; Liu, Y.-W.; Du, Z.-T.; Wei, B.-G.; Lin, G.-Q. *Org. Chem. Front.* **2015**, *2*, 1485–1499. doi:10.1039/c5qo0250h
100. Si, C.-M.; Shao, L.-P.; Mao, Z.-Y.; Zhou, W.; Wei, B.-G. *Org. Biomol. Chem.* **2017**, *15*, 649–661. doi:10.1039/c6ob02523d
101. Chogii, I.; Njardarson, J. T. *Angew. Chem., Int. Ed.* **2015**, *54*, 13706–13710. doi:10.1002/anie.201506559
102. Xu, C.-P.; Huang, P.-Q.; Py, S. *Org. Lett.* **2012**, *14*, 2034–2037. doi:10.1021/ol300550x
103. Maciá, E.; Foubelo, F.; Yus, M. *Tetrahedron* **2016**, *72*, 6001–6010. doi:10.1016/j.tet.2016.07.020
104. Procopiou, G.; Lewis, W.; Harbottle, G.; Stockman, R. A. *Org. Lett.* **2013**, *15*, 2030–2033. doi:10.1021/ol400720b
105. Mendes, J. A.; Merino, P.; Soler, T.; Salustiano, E. J.; Costa, P. R. R.; Yus, M.; Foubelo, F.; Buarque, C. D. *J. Org. Chem.* **2019**, *84*, 2219–2233. doi:10.1021/acs.joc.8b03203
106. Li, Y.; Xu, M.-H. *Org. Lett.* **2012**, *14*, 2062–2065. doi:10.1021/ol300581n
107. Green, B. T.; Lee, S. T.; Panter, K. E.; Brown, D. R. *Food Chem. Toxicol.* **2012**, *50*, 2049–2055. doi:10.1016/j.fct.2012.03.049
108. Kittakoop, P.; Mahidol, C.; Ruchirawat, S. *Curr. Top. Med. Chem.* **2014**, *14*, 239–252. doi:10.2174/1568026613666131216105049
109. Zhao, G.; Canterbury, D. P.; Taylor, A. P.; Cheng, X.; Mikochik, P.; Bagley, S. W.; Tong, R. *Org. Lett.* **2020**, *22*, 458–463. doi:10.1021/acs.orglett.9b04220
110. Bari, A.; Iqbal, A.; Khan, Z. A.; Shahzad, S. A.; Yar, M. *Synth. Commun.* **2020**, *50*, 2572–2589. doi:10.1080/00397911.2020.1776878
111. Brahmachari, G.; Das, S. *Tetrahedron Lett.* **2012**, *53*, 1479–1484. doi:10.1016/j.tetlet.2012.01.042
112. Fustero, S.; Jiménez, D.; Moscardó, J.; Catalán, S.; del Pozo, C. *Org. Lett.* **2007**, *9*, 5283–5286. doi:10.1021/ol702447y
113. Lebold, T. P.; Leduc, A. B.; Kerr, M. A. *Org. Lett.* **2009**, *11*, 3770–3772. doi:10.1021/ol901435k
114. Dobbs, A. P.; Guesné, S. J. *J. Synlett* **2005**, 2101–2103. doi:10.1055/s-2005-871956
115. Sales, M.; Charette, A. B. *Org. Lett.* **2005**, *7*, 5773–5776. doi:10.1021/ol052436v
116. Merino, P.; Tejero, T.; Greco, G.; Marca, E.; Delso, I.; Gómez-SanJuan, A.; Matute, R. *Heterocycles* **2012**, *84*, 75. doi:10.3987/rev-11-sr(p)1
117. Buffat, M. G. P. *Tetrahedron* **2004**, *60*, 1701–1729. doi:10.1016/j.tet.2003.11.043
118. Laschat, S.; Dickner, T. *Synthesis* **2000**, 1781–1813. doi:10.1055/s-2000-8218
119. Sirvent, J. A.; Foubelo, F.; Yus, M. *Heterocycles* **2014**, *88*, 1163–1174. doi:10.3987/com-13-s(s)75
120. Lahosa, A.; Soler, T.; Arrieta, A.; Cossío, F. P.; Foubelo, F.; Yus, M. *J. Org. Chem.* **2017**, *82*, 7481–7491. doi:10.1021/acs.joc.7b01178
121. Bosque, I.; González-Gómez, J. C.; Foubelo, F.; Yus, M. *J. Org. Chem.* **2012**, *77*, 780–784. doi:10.1021/jo202211u
122. González-Gómez, J. C.; Foubelo, F.; Yus, M. *Synlett* **2008**, 2777–2780. doi:10.1055/s-0028-1083535
123. Reddy, A. A.; Prasad, K. R. *J. Org. Chem.* **2017**, *82*, 13488–13499. doi:10.1021/acs.joc.7b02611
124. Si, C.-M.; Huang, W.; Du, Z.-T.; Wei, B.-G.; Lin, G.-Q. *Org. Lett.* **2014**, *16*, 4328–4331. doi:10.1021/ol5020812
125. Han, P.; Si, C.-M.; Mao, Z.-Y.; Li, H.-T.; Wei, B.-G.; Du, Z.-T. *Tetrahedron* **2016**, *72*, 862–867. doi:10.1016/j.tet.2015.12.057
126. Wang, C.; Liu, Y.-W.; Zhou, Z.; Si, C.-M.; Sun, X.; Wei, B.-G. *Tetrahedron* **2018**, *74*, 2158–2165. doi:10.1016/j.tet.2018.03.022
127. Si, C.-M.; Mao, Z.-Y.; Zhou, Z.; Du, Z.-T.; Wei, B.-G. *Tetrahedron* **2015**, *71*, 9396–9402. doi:10.1016/j.tet.2015.10.059
128. Liu, Y.-W.; Mao, Z.-Y.; Ma, R.-J.; Yan, J.-H.; Si, C.-M.; Wei, B.-G. *Tetrahedron* **2017**, *73*, 2100–2108. doi:10.1016/j.tet.2017.02.057
129. Zhang, Z.; Sun, Z. *Synlett* **2020**, *31*, 355–358. doi:10.1055/s-0039-1690771
130. Zhou, W.; Zhang, Y.-X.; Nie, X.-D.; Si, C.-M.; Sun, X.; Wei, B.-G. *J. Org. Chem.* **2018**, *83*, 9879–9889. doi:10.1021/acs.joc.8b01282
131. Kaczorek, D.; Kawęcki, R. *Tetrahedron Lett.* **2020**, *61*, 152034. doi:10.1016/j.tetlet.2020.152034
132. Reddy, Y. V.; Biradar, D. O.; Reddy, B. J. M.; Rathod, A.; Himabindu, M.; Reddy, B. V. S. *Nat. Prod. Commun.* **2017**, *12*, 1599–1603. doi:10.1177/1934578x1701201019
133. Frederico, D.; Brocksom, U.; Brocksom, T. J. *Quim. Nova* **2005**, *28*, 692–702. doi:10.1590/s0100-40422005000400024
134. Pinto, A.; Griera, R.; Molins, E.; Fernández, I.; Bosch, J.; Amat, M. *Org. Lett.* **2017**, *19*, 1714–1717. doi:10.1021/acs.orglett.7b00492
135. Pinto, A.; Picciché, M.; Griera, R.; Molins, E.; Bosch, J.; Amat, M. *J. Org. Chem.* **2018**, *83*, 8364–8375. doi:10.1021/acs.joc.8b00983
136. Medjahdi, M.; González-Gómez, J. C.; Foubelo, F.; Yus, M. *Heterocycles* **2012**, *86*, 727–734. doi:10.3987/com-12-s(n)27
137. Sirvent, A.; García-Muñoz, M. J.; Yus, M.; Foubelo, F. *Eur. J. Org. Chem.* **2020**, 113–126. doi:10.1002/ejoc.201901590
138. García-Muñoz, M. J.; Foubelo, F.; Yus, M. *J. Org. Chem.* **2016**, *81*, 10214–10226. doi:10.1021/acs.joc.6b01047
139. Penk, D. N.; Robinson, N. A.; Hill, H. M.; Turlington, M. *Tetrahedron Lett.* **2017**, *58*, 470–473. doi:10.1016/j.tetlet.2016.12.061
140. Chowdhury, S.; Mao, J.; Martynow, J.; Zhao, H.; Duffy, E.; Wu, Y.; Thakur, V.; Sirasani, G.; Tang, Y.; Collin, F.; Sinishtaj, S.; Bhattacharjee, A. *Tetrahedron Lett.* **2019**, *60*, 371–374. doi:10.1016/j.tetlet.2018.12.061
141. Lahosa, A.; Yus, M.; Foubelo, F. *J. Org. Chem.* **2019**, *84*, 7331–7341. doi:10.1021/acs.joc.9b01008
142. García-Muñoz, M. J.; Foubelo, F.; Yus, M. *Heterocycles* **2015**, *90*, 1419–1432. doi:10.3987/com-14-s(k)103
143. Hernández-Ibáñez, S.; Soares do Rego Barros, O.; Lahosa, A.; García-Muñoz, M. J.; Benlahrech, M.; Behloul, C.; Foubelo, F.; Yus, M. *Tetrahedron* **2020**, *76*, 130842. doi:10.1016/j.tet.2019.130842
144. Chen, D.; Xu, M.-H. *Chem. Commun.* **2013**, *49*, 1327–1329. doi:10.1039/c2cc38600c
145. Peralta-Hernández, E.; Cordero-Vargas, A. *Synthesis* **2016**, *48*, 4237–4245. doi:10.1055/s-0035-1562612
146. Babu, R. A.; Reddy, N. S. S.; Reddy, B. J. M.; Reddy, B. V. S. *Synthesis* **2014**, *46*, 2794–2798. doi:10.1055/s-0034-1378515
147. Reddy, N. S. S.; Reddy, B. V. S. *Tetrahedron Lett.* **2014**, *55*, 3157–3159. doi:10.1016/j.tetlet.2014.03.130
148. Yus, M.; González-Gómez, J. C.; Foubelo, F. *Chem. Rev.* **2011**, *111*, 7774–7854. doi:10.1021/cr1004474
149. Foubelo, F.; Yus, M. *Chim. Oggi* **2016**, *34* (4), 45–49.
150. Reddy, N. S. S.; Reddy, B. J. M.; Reddy, B. V. S. *Tetrahedron Lett.* **2013**, *54*, 4228–4231. doi:10.1016/j.tetlet.2013.05.132

151. Reddy, N. S. S.; Babu, R. A.; Reddy, B. V. S. *Synthesis* **2016**, *48*, 1079–1086. doi:10.1055/s-0035-1561562
152. Bosque, I.; Gonzalez-Gomez, J. C.; Loza, M. I.; Brea, J. *J. Org. Chem.* **2014**, *79*, 3982–3991. doi:10.1021/jo500446f
153. Bosque, I.; González-Gómez, J. C.; Guijarro, A.; Foubelo, F.; Yus, M. *J. Org. Chem.* **2012**, *77*, 10340–10346. doi:10.1021/jo302045y
154. Anton-Torrecillas, C.; Gonzalez-Gomez, J. C. *Org. Biomol. Chem.* **2014**, *12*, 7018–7025. doi:10.1039/c4ob01133c
155. Anton-Torrecillas, C.; Bosque, I.; Gonzalez-Gomez, J. C.; Loza, M. I.; Brea, J. *J. Org. Chem.* **2015**, *80*, 1284–1290. doi:10.1021/jo502660r
156. Anton-Torrecillas, C.; Loza, M. I.; Brea, J.; Gonzalez-Gomez, J. C. *Org. Biomol. Chem.* **2016**, *14*, 2264–2271. doi:10.1039/c5ob02624e
157. Sirvent, J. A.; Foubelo, F.; Yus, M. *J. Org. Chem.* **2014**, *79*, 1356–1367. doi:10.1021/jo402759v
158. Liu, Y.-W.; Han, P.; Zhou, W.; Mao, Z.-Y.; Si, C.-M.; Wei, B.-G. *Org. Biomol. Chem.* **2016**, *14*, 10714–10722. doi:10.1039/c6ob02212j
159. Uphade, M. B.; Reddy, A. A.; Khandare, S. P.; Prasad, K. R. *Org. Lett.* **2019**, *21*, 9109–9113. doi:10.1021/acs.orglett.9b03507
160. Guerola, M.; Sánchez-Roselló, M.; Mulet, C.; del Pozo, C.; Fustero, S. *Org. Lett.* **2015**, *17*, 960–963. doi:10.1021/acs.orglett.5b00054
161. Fadeyi, O. O.; Senter, T. J.; Hahn, K. N.; Lindsley, C. W. *Chem. – Eur. J.* **2012**, *18*, 5826–5831. doi:10.1002/chem.201200629
162. Senter, T. J.; Fadeyi, O. O.; Lindsley, C. W. *Org. Lett.* **2012**, *14*, 1869–1871. doi:10.1021/ol300466a
163. Si, C.-M.; Mao, Z.-Y.; Dong, H.-Q.; Du, Z.-T.; Wei, B.-G.; Lin, G.-Q. *J. Org. Chem.* **2015**, *80*, 5824–5833. doi:10.1021/acs.joc.5b00803
164. Zhang, W.; Sha, W.; Zhu, Y.; Han, J.; Soloshonok, V. A.; Pan, Y. *Eur. J. Org. Chem.* **2017**, 1540–1546. doi:10.1002/ejoc.201601645
165. Thaisrivongs, D. A.; Miller, S. P.; Molinaro, C.; Chen, Q.; Song, Z. J.; Tan, L.; Chen, L.; Chen, W.; Lekhal, A.; Pulicare, S. K.; Xu, Y. *Org. Lett.* **2016**, *18*, 5780–5783. doi:10.1021/acs.orglett.6b01793
166. Hugelshofer, C. L.; Palani, V.; Sarpong, R. *J. Org. Chem.* **2019**, *84*, 14069–14091. doi:10.1021/acs.joc.9b02223
167. Benlahrech, M.; Lahosa, A.; Behloul, C.; Foubelo, F.; Yus, M. *Heterocycles* **2018**, *97*, 1191–1202. doi:10.3987/com-18-s(t)66
168. Reddy, A. A.; Reddy, P. O.; Prasad, K. R. *J. Org. Chem.* **2016**, *81*, 11363–11371. doi:10.1021/acs.joc.6b01541
169. Chatare, V. K.; Andrade, R. B. *Angew. Chem., Int. Ed.* **2017**, *56*, 5909–5911. doi:10.1002/anie.201702530
170. Walia, M.; Teijaro, C. N.; Gardner, A.; Tran, T.; Kang, J.; Zhao, S.; O'Connor, S. E.; Courdavault, V.; Andrade, R. B. *J. Nat. Prod.* **2020**, *83*, 2425–2433. doi:10.1021/acs.jnatprod.0c00310
171. Zhao, S.; Andrade, R. B. *J. Org. Chem.* **2017**, *82*, 521–531. doi:10.1021/acs.joc.6b02551
172. Zhao, S.; Andrade, R. B. *J. Am. Chem. Soc.* **2013**, *135*, 13334–13337. doi:10.1021/ja408114u
173. Si, C.-M.; Liu, Y.-W.; Mao, Z.-Y.; Han, P.; Du, Z.-T.; Wei, B.-G. *Tetrahedron* **2016**, *72*, 8091–8098. doi:10.1016/j.tet.2016.10.047
174. Liu, Q.; Zhang, Y.-A.; Xu, P.; Jia, Y. *J. Org. Chem.* **2013**, *78*, 10885–10893. doi:10.1021/jo4018777

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