



## Article

# Low-Temperature Fabrication of IZO Thin Film for Flexible Transistors

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**Abstract:** Solution-processed thin film transistors (TFTs) used in flexible electronics require them to be fabricated under low temperature. Ultraviolet (UV) treatment is an effective method to transform the solution precursors into dense semiconductor films. In our work, high-quality indium zinc oxide (IZO) thin films were prepared from nitrate-based precursors after UV treatment at room temperature. After UV treatment, the structure of IZO thin films was gradually rearranged, resulting in good M–O–M network formation and bonds. TFTs using IZO as a channel layer were also fabricated on Si and Polyimide (PI) substrate. The field effect mobility, threshold voltage ( $V_{th}$ ), and subthreshold swing (SS) for rigid and flexible IZO TFTs are 14.3 and 9.5 cm<sup>2</sup>/Vs, 1.1 and 1.7 V, and 0.13 and 0.15 V/dec., respectively. This low-temperature processed route will definitely contribute to flexible electronics fabrication.

**Keywords:** thin film transistors (TFTs); flexible; low-temperature; ultraviolet (UV)



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## 1. Introduction

Recently, metal oxide semiconductors used as channel layers in thin film transistors (TFTs) have been extensively investigated due to their good transparency and electronic conductivity [1–3]. Increasing demand for next-generation flexible electronics has resulted in increased attention on applications with high performance and low-cost materials and processes. Although TFTs fabricated by conventional vacuum-based methods have advantages, their high cost and area-limited uniformity restrict their application (requiring long processing times in high vacuum environments for successful film deposition). Therefore, the solution-based process shows great potential for next-generation devices owing to its cost-effective fabrication, large-area deposition, and simple manufacturing process [4–8].

To achieve high densification, sol–gel thin films require a very high temperature, which provides enough thermal energy to remove impurities and form a metal–oxygen–metal (M–O–M) structure [9,10]. This temperature is usually higher than 300 °C. Such a high processing temperature is not suitable for flexible polymer substrate. Therefore, exploring a low-temperature solution method is necessary.

Sung Kyu Park et al. provided a new theoretical way to fabricate TFT at room temperature [11]. UV treatment induces the efficient densification of metal oxide thin films without additional high temperature processing. There are two stages after UV exposure for thin films: firstly, rapid chemical condensation, followed by gradual structural reorganization and densification [10]. Subsequently, a high-quality M–O–M network is formed [8]. Thus, UV treatment is an effective way to fabricate metal oxide thin films at low temperature [12]. However, the mobility is not satisfactory due to the low dielectric constant of Al<sub>2</sub>O<sub>3</sub>, which is used as a dielectric layer in TFT devices.

In this work, solution-processed high-performance IZO thin film transistors were fabricated by UV treatment. The IZO thin film was analyzed by spectroscopy. Additionally, IZO TFTs were also studied in detail. After UV exposure, IZO thin film showed a smooth surface morphology with a roughness of 0.42 nm. C-related and N-related groups were obviously decomposed, indicating the formation of high-purity thin film. Furthermore, TFTs using IZO as a channel layer and HfO<sub>2</sub> as a dielectric layer were also fabricated on rigid and flexible substrates. The flexible IZO TFT showed good performance and repeatability, such as an on/off ratio of 10<sup>7</sup>, a mobility of 9.5 cm<sup>2</sup>/Vs, and an SS of 0.15 V/dec. This new process of fabricating IZO TFTs at low temperature creates insight into the applications of flexible and transparent devices.

## 2. Materials and Methods

### 2.1. Precursor Synthesis

IZO solution was prepared by dissolving indium nitrate hydrate (In(NO<sub>3</sub>)<sub>3</sub>·xH<sub>2</sub>O) and zinc nitrate hydrate (Zn(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O) in 2-methoxyethanol (2-ME). The In:Zn ratio was 7:3. The solution was stirred at 70 °C for 2 h and then stirred at room temperature for 12 h. Before use, the precursor solution was filtered through a 0.2 mm PTFE syringe filter.

### 2.2. Film Fabrication and Characterization

The IZO precursor solution was spin coated on Si substrate at 500 rpm for 5 s, followed by 3000 rpm for 30 s. The substrate was precleaned with acetone, alcohol, and deionized water sequentially in an ultrasonicator for 10 min, then dried by N<sub>2</sub> flow. To densify the IZO thin film, the sample was cured under a high-pressure mercury UV lamp for 20 min under N<sub>2</sub> purging. In our work, thin film transistors were fabricated using IZO as the channel layer and HfO<sub>2</sub> as the gate insulator. Using atomic layer deposition (ALD), 50 nm HfO<sub>2</sub> was grown at 200 °C. We also explored flexible thin film transistors using PI as a substrate and ALD-Al<sub>2</sub>O<sub>3</sub> as a buffer layer. Al films deposited by thermal evaporation were used as a source/drain electrode of TFTs with channel widths (*W*) = 1000 μm and channel lengths (*L*) = 100 μm.

### 2.3. Characterizations

The optical transmission was carried out by a double-beam spectrophotometer (U-3900, U-3900, Hitachi, Ltd., Tokyo, Japan). The surface morphology was measured by an atomic force microscope (AFM; nanonaviSPA-400 SPM, SII Nano Technology Inc., Chiba, Japan). The AFM measurement mode used was the tapping mode. The parameters of the AFM tip (Tap150AL-G, Innovative Solutions Bulgaria Ltd., Sofia, Bulgaria) were resonant frequency: 150 KHz and force constant: 5 N/m. The measurement geometry was rectangle and the acquisition time was 4 min. Fourier Transform Infrared Spectroscopy (FTIR) was carried out by the Nicolet 5700. The chemical composition of the thin film was analyzed by X-ray photoelectron spectroscopy (XPS, Thermo Scientific K-Alpha+, Thermo Fisher Scientific Inc., Waltham, MA, USA). The crystal structure of the thin film was investigated by X-ray diffraction (XRD, Rigaku D/max-rB, Rigaku Corporation, Tokyo, Japan). The electrical properties were measured by a semiconductor parameter analyzer (Keithley 4200, Tektronix Inc., Beaverton, OR, USA).

The field-effect mobility (*μ*) and SS were extracted by using the following equations [13]

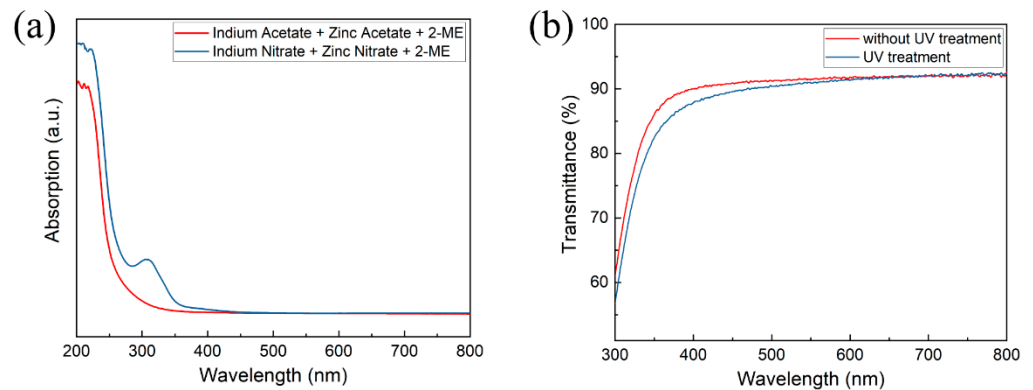
$$I_D = \left( \frac{W}{2L} C_i \mu \right) (V_G - V_{th})^2 \quad (1)$$

$$SS = \frac{dV_G}{d(\text{Log} I_D)} \quad (2)$$

where *W* and *L* are the channel width and length, respectively. *C<sub>i</sub>* is the capacitance per unit area of the insulator; *V<sub>th</sub>* is the threshold voltage; and *V<sub>G</sub>* is the gate voltage.

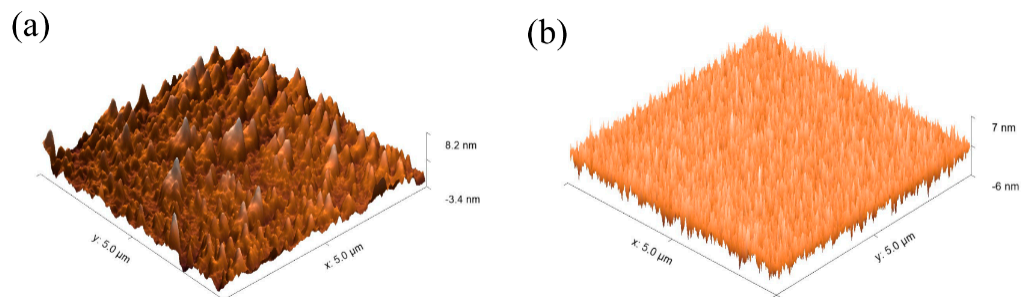
### 3. Results and Discussion

The absorption spectra of IZO thin films fabricated by acetate-based precursors and nitrate-based precursors are shown in Figure 1a. The nitrate-based precursor was chosen due to its higher UV absorptivity. The transmittance of IZO thin films on quartz glass substrates was employed to evaluate the optical performance, as shown in Figure 1b. The transparency of IZO thin films was about 90% in the visible range, favoring their application in transparent electronic devices. An interesting observation is that after UV treatment, the transparency of IZO thin film was improved in the short wavelength.



**Figure 1.** (a) The absorption spectra of different precursor solutions. (b) The transmittance of IZO thin films with and without UV treatment, respectively.

The surface morphology of thin films was characterized by AFM, as shown in Figure 2. The scanning area was  $5 \mu\text{m} \times 5 \mu\text{m}$ . It was found that IZO with UV treatment shows a small root-mean-squared (RMS) value of 0.42 nm and is free of obvious holes. On the contrary, the IZO thin film without UV treatment has a much higher RMS of 2.31 nm. After UV treatment, the densification of the IZO thin film was enhanced significantly. A good interface between the electrode and semiconductor is one of the important parameters to ensure high-performance thin film transistor devices [14,15].



**Figure 2.** AFM images of thin films: (a) IZO without UV treatment, (b) IZO with UV treatment.

The XRD patterns of IZO thin films are presented in Figure 3a. There are no obvious peaks in this pattern, indicating the amorphous structure for thin films. The amorphous phase can contribute to the formation of the semiconductor layer with a smooth surface and excellent uniformity, which is beneficial for the fabrication of TFT devices. The results are consistent with the AFM analysis.

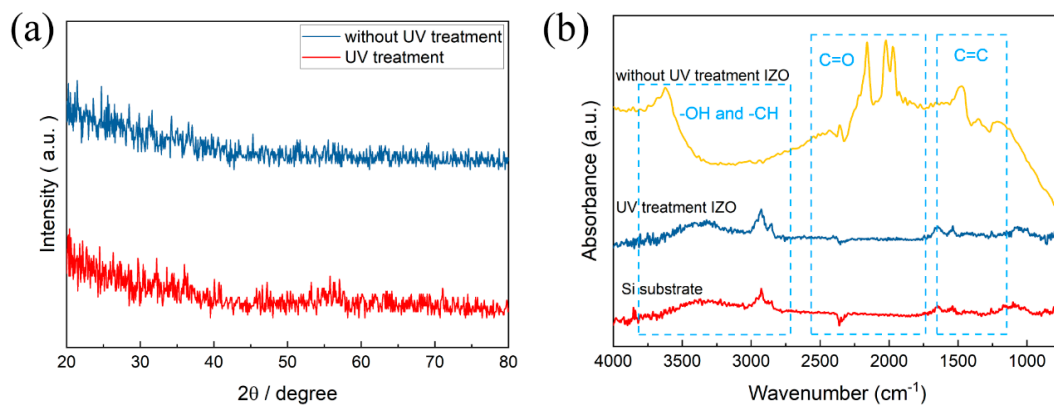


Figure 3. (a) XRD patterns and (b) IR spectra of IZO thin films.

In order to confirm the effect of UV treatment on the organic group of IZO thin films, the IR spectra were measured, as shown in Figure 3b. According to the group theory, the peaks between  $2500\text{ cm}^{-1}$  to  $4000\text{ cm}^{-1}$  are related to  $\text{-OH}$  and  $\text{-CH}$  stretching vibrations. The  $\text{-OH}$  vibration peak shifted to a high wavenumber after UV treatment, indicating that the free state of  $\text{M-OH}$  bonding was decomposed by the UV/ $\text{O}_3$  treatment [16]. The peaks at approximately  $2000\text{ cm}^{-1}$  were related to asymmetric  $\text{C=O}$  bonding, which promotes the condensation of oxide gel films by chelating with coordination bonding to the metal elements [17]. In addition, the peaks at about  $1500\text{ cm}^{-1}$  were associated with the vibration of  $\text{C=C}$  bonding. After UV treatment, the IR spectra for IZO thin film was found to be similar with the Si substrate. This result demonstrates that UV treatment can effectively reduce the carbon-related organic group and form high-purity thin films.

To further investigate the chemical composition of IZO thin films, XPS characterization was carried out (Figure 4). All peaks were calibrated to C 1s ( $284.5\text{ eV}$ ). The corresponding In and Zn high-resolution scans are shown in Figure 4a,b, respectively. Figure 4a shows that there are two related peaks near  $444\text{ cm}^{-1}$  and  $452\text{ cm}^{-2}$ , respectively. Additionally, the peaks of Zn 2p are at approximately  $1021\text{ cm}^{-1}$  and  $1044\text{ cm}^{-1}$ , respectively. In XPS N 1s narrow scans (Figure 4c), the N signal was not observed for IZO thin film after UV treatment, indicating the complete decomposition of the N-related group.

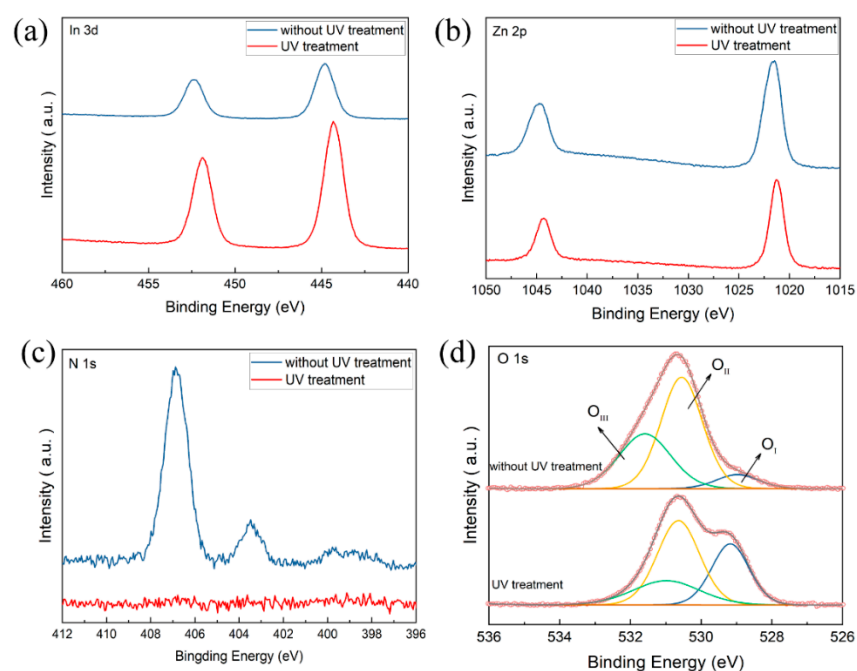
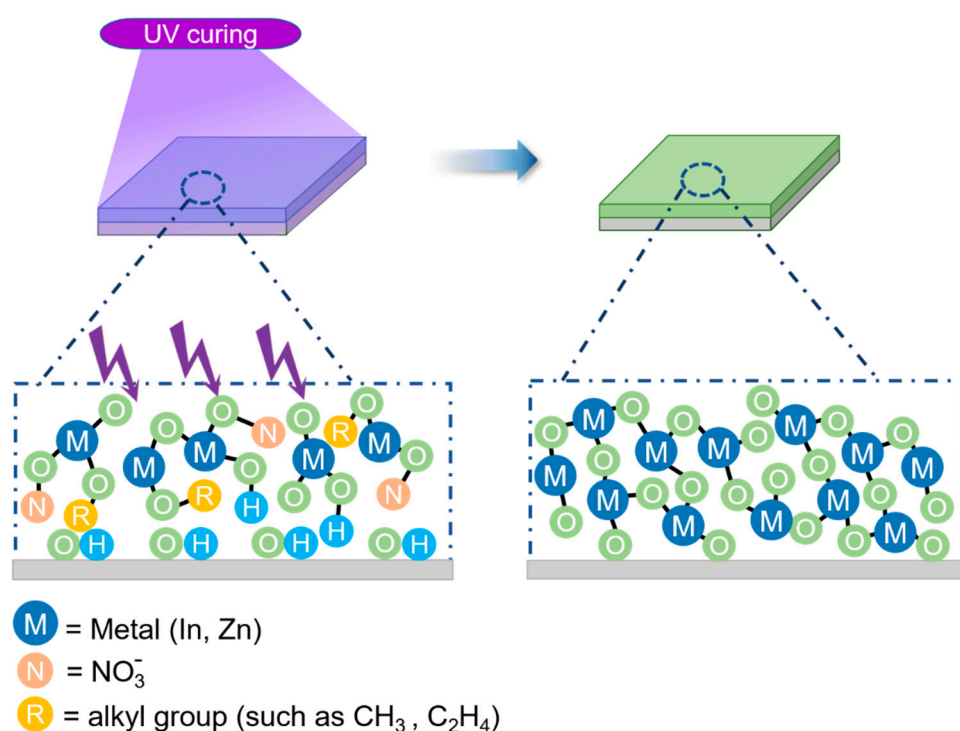


Figure 4. XPS spectra of (a) In 3d, (b) Zn 3d, (c) N 1s, and (d) O 1s.

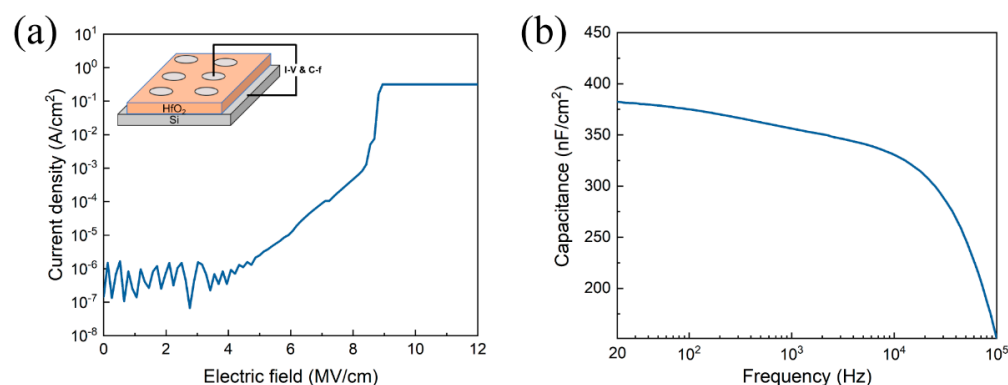
As shown in Figure 4d, The O 1s spectra could be fitted by three component peaks:  $O_I$  (528.7 eV) is related to oxygen ions ( $O^{2-}$ ) combined with metal cations in IZO thin film,  $O_{II}$  (530.6 eV) is associated with the oxygen vacancy, and  $O_{III}$  (531.5 eV) is attributed to bonded oxygen, such as  $H_2O$  or  $O_2$  [18]. The relative area of  $O_I$  is  $O_I/O$ , where  $O = O_I + O_{II} + O_{III}$ . After UV treatment, the ratio of  $O_I$  peak increases from 5% to 31%. This suggests that UV treatment can enhance the formation of M–O bonding in the gel films effectively [19,20].

The mechanism of UV-treated IZO formation is shown in Figure 5. The nitrate-based precursor exhibits high UV absorptivity. The original IZO solution is in disorder, leading to slower densification of thin films. UV treatment can lower the activation barrier for M–O bond-forming condensation, resulting in very fast film densification. The organic impurities are removed, and trap defects are greatly reduced. The M–O–M networks are reorganized without high-temperature annealing. Subsequently, a high-density M–O–M framework is formed [20,21].



**Figure 5.** The condensation mechanism of IZO thin films by UV treatment.

Before the fabrication of TFT devices, the electrical property of the  $HfO_2$  dielectric was evaluated, as shown in Figure 6. Figure 6a shows the leakage current density of the  $HfO_2$  thin film. Additionally, inset is the schematic of the capacitor. The  $HfO_2$  dielectric represents a low leakage current density of  $10^{-6}$  A/cm<sup>2</sup> at 2 MV/cm. In addition, the breakdown field is as high as 9 MV/cm. Figure 6b shows the areal capacitance as a function of frequency for  $HfO_2$  capacitor. The areal capacitance for  $HfO_2$  is 380 nF/cm<sup>2</sup> at 20 Hz. Youn Sang Kim et al. discussed the relationship between field effect mobility and capacitance [22]. In oxide semiconductors, although the overlap of the spherical S-orbital of the metal ion gives an efficient path for carrier transport, the electron transport is governed by dense localized states between the energy bandgap. The  $HfO_2$  dielectric represents a very high capacitance and a strong electric field. When the  $HfO_2$  was used as a gate dielectric in TFT devices, abundant electrons quickly filled the upper-lying localized states, and the electrons jumped into the transport band easily, leading to the high mobility of TFTs.



**Figure 6.** (a) The leakage current density and (b) capacitance–frequency characteristics of HfO<sub>2</sub> dielectric. Inset: the schematic of the capacitor.

This theory can be summarized as Equation (3) [20]:

$$\mu \cong \mu^0 \left( \frac{C_i}{C_i^0} \right)^{\gamma-2} \quad (3)$$

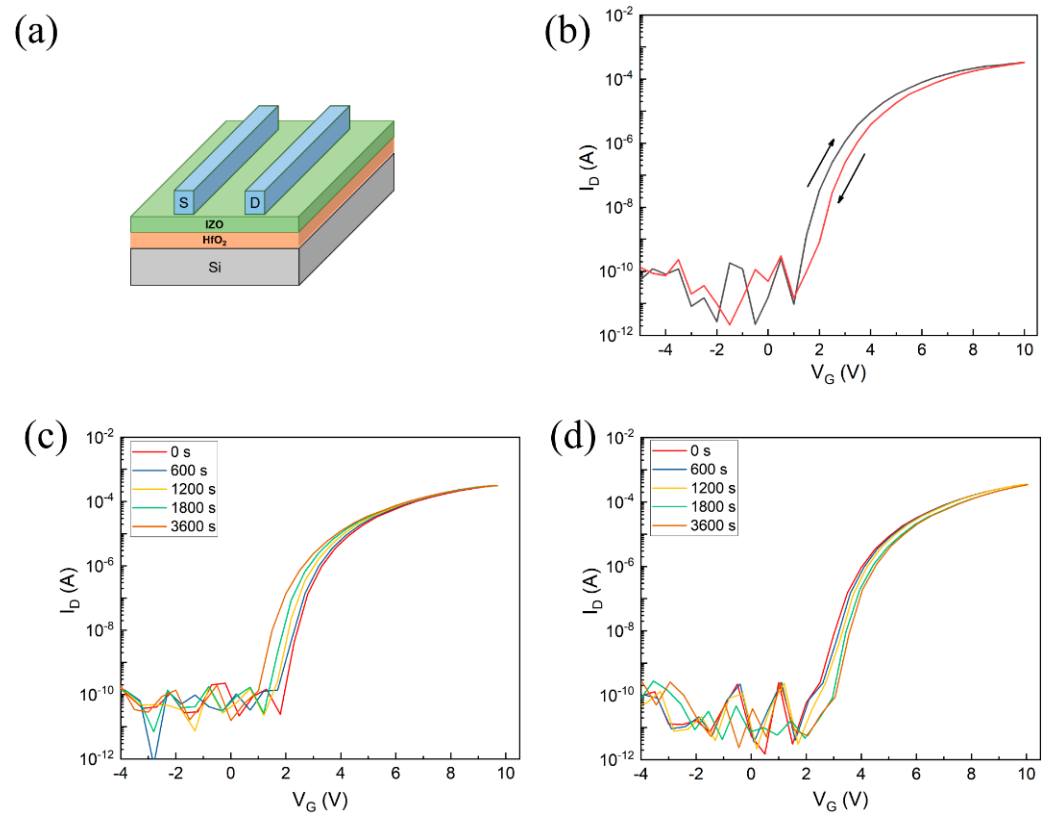
where  $\mu^0$  is the field effect mobility at  $C_i = C_i^0$  and  $\gamma$  is a constant related to the material. Thus, device mobility can be enhanced by a high-capacitance dielectric.

To study the electrical properties of IZO thin films with UV treatment, TFT was fabricated using IZO as the channel layer and HfO<sub>2</sub> as the gate dielectric, as shown in Figure 7a. The transfer curves are illustrated in Figure 7b. IZO TFT exhibits typical n-type characteristics: a high on/off ratio of 10<sup>7</sup>, a good mobility of 14.3 cm<sup>2</sup>/Vs, a small SS of 0.13 V/dec., and a  $V_{th}$  of 1.1 V. It is noted that the hysteresis of transfer curve is negligible, indicating a good interface between the channel layer and dielectric layer in the TFT device [23]. The electrical performance of UV-treated IZO TFT are comparable to those of the thermally annealed devices at a high temperature in air [12,24,25]. What calls for special attention is that the IZO thin films were treated by the UV treatment system under N<sub>2</sub> purging. Sung Kyu Park et al. discussed differences between metal oxide thin films annealing in N<sub>2</sub> and air [26]. In the air, the photoactivation efficiency from the mercury lamp is significantly attenuated, mainly owing to absorption by molecular oxygen. Therefore, the energy is insufficient for metal alkoxides and results in poor densification, leading to inactive TFT operation.

Figure 7c,d show the positive bias stress (PBS, +10 V) and negative bias stress (NBS, −10 V), respectively.  $V_{th}$  shifts ( $\Delta V_{th}$ ) of 1.12 and 0.86 V are recorded after PBS and NBS testing for 3600 s, which is comparable with parameters from the literature [27–31]. The negligible  $\Delta V_{th}$  under bias stress is due to charge trapping at the gate dielectric near the interface [30]. Furthermore, we also calculated trapping states ( $N_{trap}$ ) by [32]:

$$SS = \frac{k_B T \ln 10}{q} \left[ 1 + \frac{q^2}{C_{ox}} N_{trap} \right] \quad (4)$$

where  $k_B$  is Boltzmann's constant,  $T$  is the temperature in Kelvin, and  $q$  is the electron charge. The  $N_{trap}$  is  $1.82 \times 10^{11} \text{ eV}^{-1} \text{ cm}^{-2}$  for IZO TFT, which is lower than previous reports [16,22–24]. The low  $N_{trap}$  corresponds with negligible hysteresis. Table 1 summarizes the performance parameters of various methods for the fabrication process of IZO TFTs. The electrical performance of IZO TFTs fabricated at room temperature are comparable to those of the thermally annealed devices.

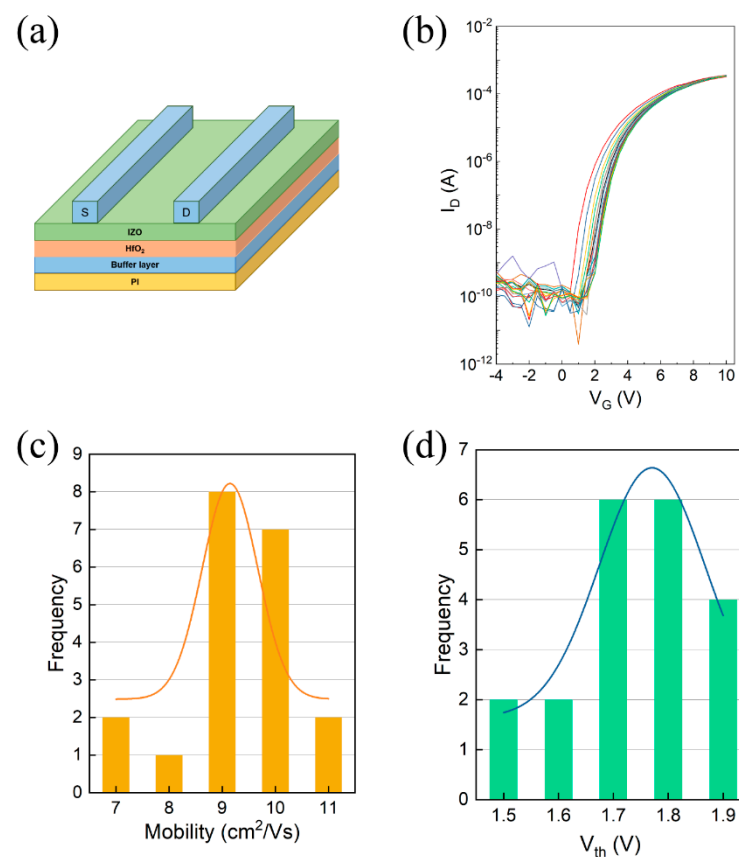


**Figure 7.** (a) Schematic structure of TFT device. (b) Transfer curve of TFT. (c) PBS (+10 V) and (d) NBS (−10 V) test for TFTs.

**Table 1.** The electrical performance of IZO TFTs.

Dielectric Layer	$V_{th}$ (V)	$\mu$ ( $\text{cm}^2/\text{Vs}$ )	$I_{on}/I_{off}$	SS (V/dec.)	Anneal Temperature ( $^{\circ}\text{C}$ )	Ref.
$\text{SiO}_2$	2.55	0.15	$2.8 \times 10^6$	0.86	300	[33]
$\text{SiO}_2$	-	2.3	$3 \times 10^4$	0.96	300	[34]
$\text{Al}_2\text{O}_3$	0.41	30.88	$10^4$	-	250	[35]
$\text{Al}_2\text{O}_3$	1.32	7.23	$3.05 \times 10^6$	-	500	[36]
$\text{SiO}_2$	-	4.94	$10^7$	-	180	[31]
$\text{HfO}_2$	1.1	14.3	$10^7$	0.13	RT	This work

Inspired by the good performance of TFTs based on Si substrate, flexible IZO TFTs were investigated on PI substrate using  $\text{Al}_2\text{O}_3$  as the buffer layer, as shown in Figure 8a. We fabricated 20 IZO TFT devices. The transfer curves for 20 TFT devices are shown in Figure 8b. From Figure 8c,d, these TFT devices show outstanding uniformity and reproducibility: they have a good average mobility of approximately  $9.5 \text{ cm}^2/\text{Vs}$ , a  $V_{th}$  of 1.7 V, and an SS of 0.15 V/dec., which is acceptable compared to previously reported devices. This result suggests that IZO with UV treatment is an ideal optional semiconductor material for flexible transparent electronics.



**Figure 8.** (a) The schematic of flexible TFT device. (b) Transfer curves of 20 flexible TFTs. Histogram of (c) mobility and (d)  $V_{th}$  for 20 devices.

#### 4. Conclusions

In this work, IZO thin film and related TFT devices were fabricated by UV treatment at room temperature. After UV treatment, the IZO thin film showed a lower RMS value of 0.42 nm. Oxygen ions combined with metal cations increased from 5% to 31%, indicating that UV treatment can enhance the formation of M–O bonding in the gel films effectively. After the successful fabrication of high-performance IZO TFT based on Si substrate, the flexible TFT with PI substrate was further integrated. The flexible IZO TFT showed good performance parameters and repeatability, such as an on/off ratio of 10<sup>7</sup>, a mobility of 9.5 cm<sup>2</sup>/Vs, and an SS of 0.15 V/dec. In a word, such a low-temperature fabricated semiconductor material offers a simple and useful approach for the exploration of high-performance, large scale, and low-cost flexible transparent electronics.

**Author Contributions:** Conceptualization, X.D.; methodology, X.D. and B.Y.; validation, B.Y. and H.X.; formal analysis, X.D., B.Y., H.X., J.Q. and X.L.; investigation, X.D., B.Y., H.X. and J.Q.; resources, J.Q., X.L. and J.Z.; data curation, X.D., X.L. and J.Z.; writing—original draft preparation, X.D.; writing—review and editing, B.Y. and H.X.; visualization, X.D., B.Y. and H.X.; supervision, X.D., X.L. and J.Z.; project administration, X.L. and J.Z.; funding acquisition, J.Z. All authors have read and agreed to the published version of the manuscript.

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**Conflicts of Interest:** The authors declare no conflict of interest.



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