Crystal structures of (η^4-cycloocta-1,5-diene)bis-(1,3-dimethylimidazol-2-ylidene)iridium(I) iodide and (η^4-cycloocta-1,5-diene)bis(1,3-diethyl-imidazol-2-ylidene)iridium(I) iodide

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The title complexes, (η^4-cycloocta-1,5-diene)bis(1,3-dimethylimidazol-2-yl-idene)iridium(I) iodide, [Ir(C$_5$H$_8$N$_2$)$_2$(C$_8$H$_{12}$)]I, (**1**) and (η^4-cycloocta-1,5-di-ene)bis(1,3-diethylimidazol-2-ylidene)iridium(I) iodide, [Ir(C$_7$H$_{12}$N$_2$)$_2$(C$_8$H$_{12}$)]-I, (**2**), were prepared using a modified literature method. After carrying out the oxidative addition of the amino acid L-proline to [Ir(COD)(IMe)$_2$]I in water and slowly cooling the reaction to room temperature, a suitable crystal of **1** was obtained and analyzed by single-crystal X-ray diffraction at 100 K. Although this crystal structure has previously been reported in the *Pbam* space group, it was highly disordered and precise atomic coordinates were not calculated. A single crystal of **2** was also obtained by heating the complex in water and letting it slowly cool to room temperature. Complex **1** was found to crystallize in the monoclinic space group *C*2/*m*, while **2** crystallizes in the orthorhombic space group *Pccn*, both with *Z* = 4.

1. Chemical context

The Merola group has been interested in the chemistry of electron-rich iridium compounds for many years (Frazier & Merola, 1992; Ladipo *et al.*, 1993; Selnau & Merola, 1993; Merola & Franks, 2013). Recently, we have begun examining the reactivity and catalytic applications of Ir^I N-heterocyclic carbene (NHC) complexes, which have previously been utilized for various transformations including hydrogenation (Hillier *et al.*, 2001), hydrosilylation (Viciano *et al.*, 2006), hydroamination (Sipos *et al.*, 2016), H/D exchange (Cochrane *et al.*, 2014), and C—H bond functionalization (Frey *et al.*, 2006). While investigating the oxidative addition of amino acids to (η^4-cycloocta-1,5-diene)bis(1,3-dimethylimidazol-2-ylidene)iridium(I) iodide in aqueous solution, cooling the reaction to room temperature yielded single crystals of the starting material [Ir(COD)(IMe)$_2$]I, where IMe = 1,3-di-methylimidazol-2-ylidene. Though Herrmann and coworkers previously described the crystal structure of this complex in the space group *Pbam* (Frey *et al.*, 2006), the anisotropic displacement parameters of the COD ligand were highly disordered; thus precise atomic coordinates could not be calculated. In an effort to advance the study of the structural properties and reactivity of Ir^I NHC complexes, we hereby report the single-crystal structure determination of (η^4-cyclo-octa-1,5-diene)bis(1,3-dimethylimidazol-2-ylidene)iridium(I) iodide (**1**) and (η^4-cycloocta-1,5-diene)bis(1,3-diethyl-imidazol-2-ylidene)iridium(I) iodide (**2**).
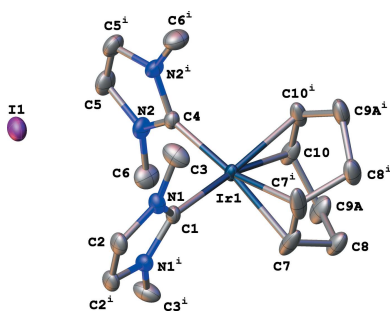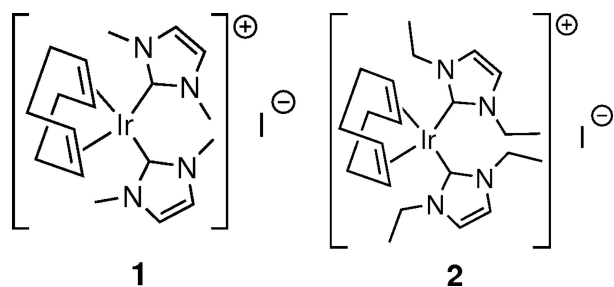
2. Structural commentary

Complex **1** (CCDC ref code 1983640) crystallizes in the monoclinic space group $C2/m$ with $Z = 4$ (Figs. 1 and 2), which differs from Herrmann's original report of the orthorhombic space group $Pbam$. Ir1, C1, C4, and I1 lie in special positions on the mirror plane. The geometry around the metal center is nearly square planar, with the largest angle [C1—Ir1—C4 = 93.14 (10)°] and smallest angle [C7—Ir1—C10 (centroids) = 86.20°] having deviations of 3.14 and 3.80°, respectively, from the ideal 90° geometry. The average Ir—NHC bond length is 2.044 Å [Ir1—C1 = 2.037 (2), Ir1—C4 = 2.051 (2) Å] and the average Ir—C_{COD} bond length is 2.169 Å [Ir1—C7 = 2.163 (2) Å; Ir1—C10 = 2.174 (2) Å] with an Ir—$COD_{centroid}$ distance of 2.047 Å, related by symmetry.

Complex **2** (CCDC ref code 1986045) crystallizes in the orthorhombic space group $Pccn$ with $Z = 4$ (Fig. 3). Atom Ir1 lies in a special position on the twofold rotation axis. Similarly to **1**, the geometry around the metal center is nearly square planar, with the largest angle [C1—Ir1—C1 = 92.93 (12)°] and smallest angle [C8—Ir1—C9 (centroids) = 86.06°] having deviations of 2.92 and 3.94°, respectively, from the ideal 90° geometry. The Ir—NHC bond lengths [2.043 (2) Å] are related by symmetry. The average Ir—C_{COD} bond length is

Figure 2
Displacement ellipsoid plot (50% probability) of (η^4-cycloocta-1,5-diene)bis(1,3-dimethylimidazol-2-ylidene)iridium(I) iodide (**1**), showing part 2 of the disorder for the CH_2 carbon atoms of the COD ring. Symmetry code: (i) $x, 1 − y, z$.

2.172 Å [Ir1—C8 = 2.197 (2), Ir1—C9 = 2.147 (2) Å] with an Ir—$COD_{centroid}$ distance of 2.058 Å, again related by symmetry.

This discrepancy in Ir—C_{COD} bond lengths and Ir—$COD_{centroid}$ distances between the two complexes is likely due to the conformation of the COD ligand, which is a boat in **1** and a twist-boat in **2**.

Figure 1
Displacement ellipsoid plot (50% probability) of (η^4-cycloocta-1,5-diene)bis(1,3-dimethylimidazol-2-ylidene)iridium(I) iodide (**1**), showing part 1 of the disorder for the CH_2 carbon atoms of the COD ring. Symmetry code: (i) $x, 1 − y, z$.

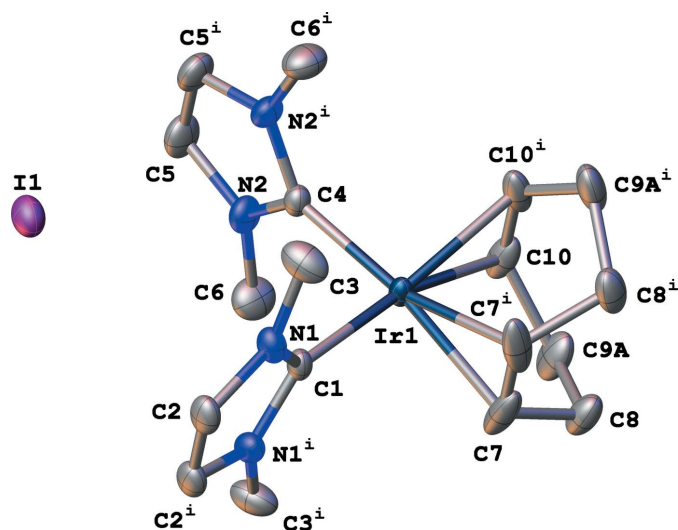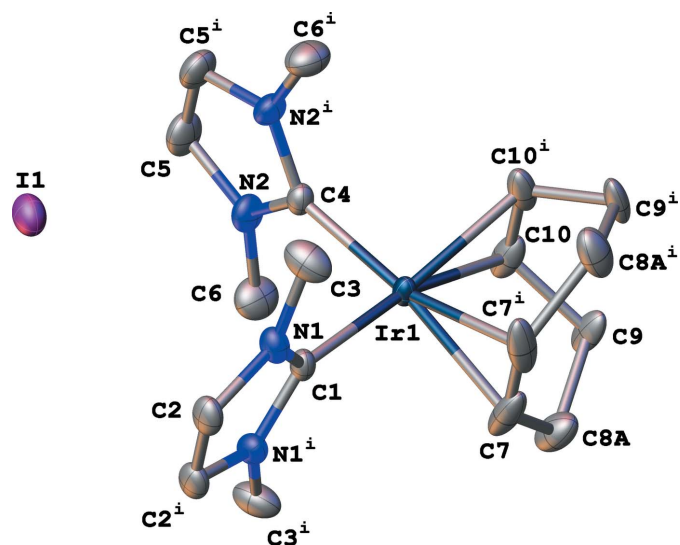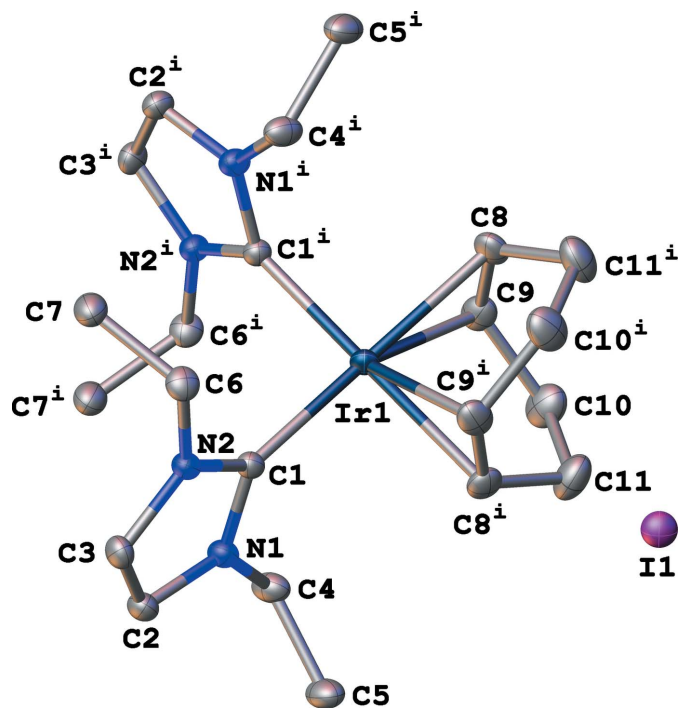**Figure 3**
Displacement ellipsoid plot (50% probability) of (η^4-cycloocta-1,5-diene)bis(1,3-diethylimidazol-2-ylidene)iridium(I) iodide (**2**). Symmetry code: (i) $\frac{3}{2} − x, \frac{3}{2} − y, z$.

Table 1
Experimental details.

	1	2
Crystal data		
Chemical formula	$[Ir(C_5H_8N_2)_2(C_8H_{12})]I$	$[Ir(C_7H_{12}N_2)_2(C_8H_{12})]I$
M_r	619.54	675.65
Crystal system, space group	Monoclinic, $C2/m$	Orthorhombic, $Pccn$
Temperature (K)	100	100
a, b, c (Å)	26.6519 (4), 8.3070 (2), 9.7852 (2)	10.6041 (2), 12.3058 (2), 18.2513 (3)
α, β, γ (°)	90, 100.241 (2), 90	90, 90, 90
V (Å^3)	2131.90 (8)	2381.65 (7)
Z	4	4
Radiation type	Mo $K\alpha$	Mo $K\alpha$
μ (mm^{-1})	7.72	6.92
Crystal size (mm)	$0.54 \times 0.22 \times 0.11$	$0.38 \times 0.17 \times 0.12$
Data collection		
Diffractometer	XtaLAB Synergy, Dualflex, HyPix	XtaLAB Synergy, Dualflex, HyPix
Absorption correction	Gaussian (*CrysAlis PRO*;Rigaku OD, 2018)	Gaussian (*CrysAlis PRO*;Rigaku OD, 2018)
T_{min}, T_{max}	0.179, 0.960	0.318, 1.000
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	27006, 5884, 5415	59059, 6352, 3839
R_{int}	0.034	0.062
$(\sin \theta/\lambda)_{max}$ (Å^{-1})	0.871	0.870
Refinement		
$R[F^2 > 2\sigma(F^2)]$, $wR(F^2)$, S	0.024, 0.059, 1.04	0.030, 0.060, 1.00
No. of reflections	5884	6352
No. of parameters	136	130
No. of restraints	12	0
H-atom treatment	H-atom parameters constrained	H-atom parameters constrained
$\Delta\rho_{max}, \Delta\rho_{min}$ (e Å^{-3})	2.40, −1.29	1.54, −0.83

Computer programs: *CrysAlis PRO* (Rigaku OD, 2018), *SHELXT* (Sheldrick, 2015a), *SHELXL* (Sheldrick, 2015b) and *OLEX2* (Dolomanov et al., 2009).

3. Supramolecular features

An examination of the packing diagrams for both title complexes show no unusual supramolecular features.

4. Database survey

In our search for the COD bis-NHC moiety, we were somewhat surprised to find only ten reported IrCOD structures in the Cambridge Structural Database (CSD2019, update 3; Groom *et al.*, 2016) with two monodentate NHCs, including the original disordered structure reported by Herrmann (WEXKOA; Frey *et al.*, 2006). Structures similar to the title compound include a square-planar [(COD)bis(1-ethyl-3-methylimidazol-2-ylidene)iridium(I)] complex (BAHZER; Hintermair *et al.*, 2011) and a complex containing quinoline-functionalized NHC ligands (ROWWUX; Jiménez *et al.*, 2015), both in space group $P2_1/c$ (No. 14). Other closely related structures include an iridium COD complex with pyrazolyl-functionalized NHC ligands (CEMVIA; Messerle *et al.*, 2006), and an iridium COD complex with pentafluoro-benzyl functionalized NHCs (TESGEE; Burling *et al.*, 2006), both of which crystallized in space group $C2/c$ (No. 15).

5. Synthesis and crystallization

The title compounds were synthesized using a modified literature procedure (Köcher & Herrmann, 1997). [Ir(COD)Cl]$_2$ (500 mg, 0.744 mmol) and a magnetic stir bar were added to a flame-dried, nitrogen-purged 100 mL Schlenk flask. Ethanol (20 mL) was added *via* syringe and the red solution was stirred. After 5 minutes, a solution of NaOEt in ethanol (1 *M*, 3.5 mL, 3.50 mmol) was added to the reaction flask dropwise. The solution was stirred for 1 h while the color slowly changed from red to bright yellow, indicating the formation of [Ir(COD)(OEt)]$_2$. The NHC precursor 1,3-dimethylimidazolium iodide (840 mg, 3.75 mmol) or 1,3-diethylimidazolium iodide (945 mg, 3.75 mmol) was dissolved in ethanol (10 mL) and added to the stirring mixture *via* syringe. After 48 h, the bright-orange mixture was filtered through celite. The solvent was removed by rotary evaporation, and the residue was dissolved in minimal dichloromethane.

The crude product was purified *via* column chromatography with silica gel, first using a 1:1 mixture of cyclohexane to ethyl acetate as the mobile phase to collect the bright-yellow iridium mono-NHC complex, followed by 7% methanol in dichloromethane to collect the desired orange iridium bis-NHC product. The solvent was removed by rotary evaporation and the bright-orange solid was dried overnight under vacuum (449 mg, 49% for **1**; 415 mg, 42% for **2**). The products were characterized by ^{1}H and ^{13}C NMR spectroscopy in agreement with previously reported data.

Single crystals of **1** for X-ray crystallography were collected from a subsequent oxidative addition reaction. The title compound, L-proline, and 10 mL of water were added to a 6 dram vial and stirred overnight at 323 K. Upon slowly cooling the reaction mixture to room temperature, bright-orange crystals of the title compound grew and were collected. Single crystals of **2** were grown by dissolving the complex in

water, heating it to 323 K, and letting the solution slowly cool to room temperature.

6. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 1.

Compound **1** was solved with *SHELXS* and refined with *SHELXL* within *OLEX2*. The refinement proceeded quite well although the displacement ellipsoids for the CH_2 carbon atoms of the COD ring were overly elongated, suggesting that there was possible disorder. In *OLEX2*, the disorder tools were utilized to split the carbon atoms while adding *SHELXL* SIMU restraint. The disorder model appeared to refine well with reasonable displacement ellipsoids. Fig. 1 shows part 1 of the disorder and Fig. 2 shows part 2. Both parts show nearly equal occupancies refining to 0.515 (19):0.485 (19). The two parts seem best described as the result of static disorder wherein the saturated portion of the COD ring is slightly twisted. The unsaturated carbon atoms are also likely a part of the disorder, but the positional change is so slight as to not warrant (and to resist) modeling. However, a consequence of this slight disorder is that generating the entire molecule does generate two different hydrogen-atom positions, also refining to 0.515 (19):0.485 (19) relative occupancies.

Data reduction, solution and refinement for **2** presented some interesting issues that are discussed here. The data were collected on a XtaLAB Synergy, Dualflex, HyPix diffract-ometer. Data reduction was performed with *CrysA-lisPro171.40_64.67a* (Rigaku OD, 2018). The crystal was of good quality and peak searching found 9425 peaks that were merged to 5446 profiles. Unit-cell calculations fit 98.2% of the peaks to the cell 9.1397 (5), 10.6193 (7), 12.3249 (6), 89.980 (5). 89.988 (4), 89.965 (6). Further refinement and space group determination led to the finalization of the data in orthorhombic *P*. *SHELXT* within *OLEX2* was used for structure solution and several non-centrosymmetric space groups were identified with nearly equal figure of merit. Attempts were made to refine the structure in all five of the proffered space groups and the only one that provided a reasonable solution was $P2_12_12$. However, while the structure refinement parameters were 'reasonable', several displace-ment ellipsoids in the finalized model were elongated along strange directions. The data were reexamined and a close view of the Ewald sphere showed weak, but clearly present peaks between the axes. The ~9 Å axis was doubled and now all peaks were aligned fully with the new axes of 18.2790 (10), 10.6196 (7), 12.3245 (6), 89.979 (5), 89.985 (4), 89.965 (5). With those particular settings in *CrysAlis*, the only reasonable unit cell found was triclinic.

Moving into *OLEX2* again, a solution was found in $P\bar{1}$ that refined into a solution with excellent figures of merit and well-shaped displacement ellipsoids with $Z = 4$. However, it was noted that the heavy atoms, iridium and iodine all had coor-dinates that suggested they sat on special positions, *e.g. x* = 0.7500. ADDSYM in *PLATON* (Spek, 2020) was used to search for higher symmetries and the result suggested that

Pccn was an appropriate high-symmetry space group. The newly created data and instruction files from *PLATON* were used in *OLEX2* and the structure in *Pccn* solved and refined cleanly into the final structure. With this result in hand, the raw data were re-reduced, the originally found *x* axis was again doubled and space-group analysis was re-performed with slightly larger angle tolerances (0.03 *vs* 0.015). *Pccn* was then clearly identified as the top match for the space group. The data and instruction files were once more used in *OLEX2* and *SHELXT* used as the solution program, which determined that *Pccn* was the best space group. Refinement led to the final structure solution reported in this paper.

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supporting information

Crystal structures of (η^4-cycloocta-1,5-diene)bis(1,3-dimethylimidazol-2-yl-idene)iridium(I) iodide and (η^4-cycloocta-1,5-diene)bis(1,3-diethylimidazol-2-ylidene)iridium(I) iodide

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Computing details

For both structures, data collection: *CrysAlis PRO* (Rigaku OD, 2018); cell refinement: *CrysAlis PRO* (Rigaku OD, 2018); data reduction: *CrysAlis PRO* (Rigaku OD, 2018). Program(s) used to solve structure: SHELXT (Sheldrick, 2015a) for (1); ShelXT (Sheldrick, 2015a) for (2). For both structures, program(s) used to refine structure: *SHELXL* (Sheldrick, 2015b); molecular graphics: *OLEX2* (Dolomanov *et al.*, 2009); software used to prepare material for publication: *OLEX2* (Dolomanov *et al.*, 2009).

(η^4-Cycloocta-1,5-diene)bis(1,3-dimethylimidazol-2-ylidene)iridium(I) iodide (1)

Crystal data

[Ir(C$_5$H$_8$N$_2$)$_2$(C$_8$H$_{12}$)]I	$F(000) = 1176$
$M_r = 619.54$	$D_x = 1.930$ Mg m^{-3}
Monoclinic, $C2/m$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$a = 26.6519$ (4) Å	Cell parameters from 17811 reflections
$b = 8.3070$ (2) Å	$\theta = 2.6$–$38.3°$
$c = 9.7852$ (2) Å	$\mu = 7.72$ mm^{-1}
$\beta = 100.241$ (2)°	$T = 100$ K
$V = 2131.90$ (8) Å^3	Prism, orange
$Z = 4$	$0.54 \times 0.22 \times 0.11$ mm

Data collection

XtaLAB Synergy, Dualflex, HyPix diffractometer	27006 measured reflections
	5884 independent reflections
Radiation source: micro-focus sealed X-ray tube, PhotonJet (Mo) X-ray Source	5415 reflections with $I > 2\sigma(I)$
Mirror monochromator	$R_{int} = 0.034$
ω scans	$\theta_{max} = 38.2°$, $\theta_{min} = 3.1°$
Absorption correction: gaussian (CrysAlisPro;Rigaku OD, 2018)	$h = -45 \rightarrow 44$
	$k = -14 \rightarrow 13$
$T_{min} = 0.179$, $T_{max} = 0.960$	$l = -16 \rightarrow 16$

Refinement

Refinement on F^2	5884 reflections
Least-squares matrix: full	136 parameters
$R[F^2 > 2\sigma(F^2)] = 0.024$	12 restraints
$wR(F^2) = 0.059$	Hydrogen site location: inferred from neighbouring sites
$S = 1.04$	

H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0317P)^2 + 1.4859P]$
where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} = 0.002$
$\Delta\rho_{max} = 2.40$ e Å^{-3}
$\Delta\rho_{min} = -1.29$ e Å^{-3}

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	*y*	*z*	U_{iso}*/U_{eq}	Occ. (<1)
Ir1	0.35715 (2)	0.500000	0.71082 (2)	0.01615 (3)	
N1	0.36047 (6)	0.6287 (2)	0.41903 (16)	0.0216 (3)	
N2	0.46647 (6)	0.3712 (2)	0.79484 (18)	0.0242 (3)	
C1	0.36077 (9)	0.500000	0.5047 (2)	0.0184 (4)	
C2	0.36009 (8)	0.5804 (3)	0.28291 (19)	0.0254 (4)	
H2	0.359874	0.648608	0.204874	0.031*	
C3	0.36109 (11)	0.7956 (3)	0.4619 (2)	0.0346 (5)	
H3A	0.394582	0.842411	0.458346	0.052*	
H3B	0.354380	0.802066	0.557045	0.052*	
H3C	0.334734	0.855247	0.399422	0.052*	
C4	0.43494 (9)	0.500000	0.7697 (3)	0.0189 (4)	
C5	0.51693 (8)	0.4201 (3)	0.8329 (2)	0.0336 (5)	
H5	0.545889	0.352001	0.854780	0.040*	
C6	0.45066 (10)	0.2032 (3)	0.7875 (3)	0.0353 (5)	
H6A	0.463915	0.149514	0.712161	0.053*	
H6B	0.413344	0.197190	0.769749	0.053*	
H6C	0.464082	0.149835	0.875787	0.053*	
C7	0.27918 (8)	0.4158 (4)	0.6632 (2)	0.0436 (7)	
H7A	0.268531	0.362122	0.571156	0.052*	0.485 (19)
H7B	0.270988	0.372499	0.566457	0.052*	0.515 (19)
C8	0.2597 (3)	0.3445 (15)	0.7821 (6)	0.0346 (18)	0.485 (19)
H8A	0.237419	0.251494	0.749983	0.041*	0.485 (19)
H8B	0.239134	0.425165	0.822295	0.041*	0.485 (19)
C9	0.2980 (3)	0.3377 (15)	0.9218 (8)	0.0330 (18)	0.485 (19)
H9A	0.274812	0.415708	0.955269	0.040*	0.485 (19)
H9B	0.303203	0.246025	0.987461	0.040*	0.485 (19)
C10	0.34831 (8)	0.4171 (3)	0.9159 (2)	0.0329 (5)	
H10	0.380731	0.371865	0.969269	0.040*	0.485 (19)
H10A	0.378957	0.364754	0.972248	0.040*	0.515 (19)
I1	0.59789 (2)	0.000000	0.87332 (2)	0.03089 (5)	
C9A	0.3044 (3)	0.2893 (16)	0.8926 (10)	0.0404 (19)	0.515 (19)
H9AA	0.317748	0.186061	0.863449	0.049*	0.515 (19)
H9AB	0.292215	0.270733	0.981216	0.049*	0.515 (19)
C8A	0.2740 (4)	0.2775 (14)	0.7760 (6)	0.0386 (18)	0.515 (19)
H8AA	0.237614	0.251443	0.773250	0.046*	0.515 (19)
H8AB	0.291638	0.178533	0.753713	0.046*	0.515 (19)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ir1	0.01272 (4)	0.02483 (5)	0.01113 (4)	0.000	0.00271 (2)	0.000
N1	0.0225 (7)	0.0278 (8)	0.0148 (6)	0.0037 (6)	0.0047 (5)	0.0019 (5)
N2	0.0164 (6)	0.0314 (9)	0.0245 (7)	0.0044 (6)	0.0025 (5)	0.0026 (6)
C1	0.0153 (9)	0.0264 (12)	0.0142 (9)	0.000	0.0044 (7)	0.000
C2	0.0251 (8)	0.0384 (11)	0.0139 (6)	0.0012 (7)	0.0064 (6)	0.0023 (7)
C3	0.0534 (14)	0.0269 (10)	0.0228 (9)	0.0045 (10)	0.0050 (9)	0.0032 (8)
C4	0.0154 (9)	0.0261 (12)	0.0155 (9)	0.000	0.0036 (7)	0.000
C5	0.0168 (7)	0.0498 (14)	0.0334 (11)	0.0050 (8)	0.0019 (7)	0.0024 (9)
C6	0.0324 (11)	0.0291 (11)	0.0430 (13)	0.0065 (9)	0.0032 (9)	0.0033 (9)
C7	0.0195 (8)	0.092 (2)	0.0185 (8)	−0.0201 (11)	0.0007 (6)	0.0036 (10)
C8	0.022 (2)	0.057 (5)	0.024 (2)	−0.012 (3)	0.0043 (17)	0.005 (3)
C9	0.022 (3)	0.063 (5)	0.014 (2)	−0.014 (3)	0.0043 (16)	0.009 (2)
C10	0.0205 (8)	0.0624 (15)	0.0154 (7)	−0.0086 (9)	0.0018 (6)	0.0091 (8)
I1	0.03408 (10)	0.03389 (10)	0.02760 (9)	0.000	0.01334 (7)	0.000
C9A	0.029 (3)	0.069 (6)	0.025 (3)	−0.011 (3)	0.009 (2)	0.012 (3)
C8A	0.034 (3)	0.055 (5)	0.027 (2)	−0.019 (3)	0.006 (2)	0.004 (2)

Geometric parameters (Å, º)

Ir1—C1	2.037 (2)	C6—H6C	0.9800
Ir1—C4	2.051 (2)	C7—C7^i	1.399 (7)
Ir1—C7^i	2.163 (2)	C7—H7A	1.0000
Ir1—C7	2.163 (2)	C7—H7B	1.0000
Ir1—C10^i	2.174 (2)	C7—C8	1.480 (6)
Ir1—C10	2.174 (2)	C7—C8A	1.616 (8)
N1—C1	1.357 (2)	C8—H8A	0.9900
N1—C2	1.389 (2)	C8—H8B	0.9900
N1—C3	1.448 (3)	C8—C9A	1.530 (12)
N2—C4	1.356 (2)	C9—H9A	0.9900
N2—C5	1.391 (3)	C9—H9B	0.9900
N2—C6	1.456 (3)	C9—C10	1.505 (8)
C2—C2^i	1.336 (5)	C9—C8A	1.540 (10)
C2—H2	0.9500	C10—C10^i	1.378 (6)
C3—H3A	0.9800	C10—H10	1.0000
C3—H3B	0.9800	C10—H10A	1.0000
C3—H3C	0.9800	C10—C9A	1.567 (10)
C5—C5^i	1.328 (5)	C9A—H9AA	0.9900
C5—H5	0.9500	C9A—H9AB	0.9900
C6—H6A	0.9800	C8A—H8AA	0.9900
C6—H6B	0.9800	C8A—H8AB	0.9900
C1—Ir1—C4	93.14 (10)	Ir1—C7—H7B	111.1
C1—Ir1—C7	89.97 (9)	C7^i—C7—Ir1	71.14 (10)
C1—Ir1—C7^i	89.97 (9)	C7^i—C7—H7A	116.5
C1—Ir1—C10	161.15 (8)	C7^i—C7—H7B	111.1

C1—Ir1—C10ⁱ	161.16 (8)	C7ⁱ—C7—C8	113.6 (5)
C4—Ir1—C7ⁱ	160.90 (9)	C7ⁱ—C7—C8A	135.3 (4)
C4—Ir1—C7	160.90 (9)	C8—C7—Ir1	114.8 (3)
C4—Ir1—C10	90.61 (8)	C8—C7—H7A	116.5
C4—Ir1—C10ⁱ	90.61 (8)	C8A—C7—Ir1	106.1 (3)
C7ⁱ—Ir1—C7	37.71 (19)	C8A—C7—H7B	111.1
C7—Ir1—C10ⁱ	92.50 (9)	C7—C8—H8A	109.7
C7ⁱ—Ir1—C10	92.49 (9)	C7—C8—H8B	109.7
C7ⁱ—Ir1—C10ⁱ	80.72 (8)	C7—C8—C9A	109.7 (5)
C7—Ir1—C10	80.72 (8)	H8A—C8—H8B	108.2
C10ⁱ—Ir1—C10	36.95 (15)	C9A—C8—H8A	109.7
C1—N1—C2	111.26 (18)	C9A—C8—H8B	109.7
C1—N1—C3	125.22 (17)	H9A—C9—H9B	108.2
C2—N1—C3	123.51 (18)	C10—C9—H9A	109.8
C4—N2—C5	110.93 (19)	C10—C9—H9B	109.8
C4—N2—C6	125.66 (17)	C10—C9—C8A	109.4 (5)
C5—N2—C6	123.40 (19)	C8A—C9—H9A	109.8
N1ⁱ—C1—Ir1	127.97 (11)	C8A—C9—H9B	109.8
N1—C1—Ir1	127.97 (11)	Ir1—C10—H10	112.1
N1—C1—N1ⁱ	103.9 (2)	Ir1—C10—H10A	115.8
N1—C2—H2	126.6	C9—C10—Ir1	114.7 (3)
C2ⁱ—C2—N1	106.78 (12)	C9—C10—H10A	115.8
C2ⁱ—C2—H2	126.6	C10ⁱ—C10—Ir1	71.53 (8)
N1—C3—H3A	109.5	C10ⁱ—C10—C9	116.0 (5)
N1—C3—H3B	109.5	C10ⁱ—C10—H10	112.1
N1—C3—H3C	109.5	C10ⁱ—C10—H10A	115.8
H3A—C3—H3B	109.5	C10ⁱ—C10—C9A	132.7 (5)
H3A—C3—H3C	109.5	C9A—C10—Ir1	106.3 (4)
H3B—C3—H3C	109.5	C9A—C10—H10	112.1
N2—C4—Ir1	127.89 (11)	C8—C9A—C10	111.6 (7)
N2ⁱ—C4—Ir1	127.89 (11)	C8—C9A—H9AA	109.3
N2ⁱ—C4—N2	104.2 (2)	C8—C9A—H9AB	109.3
N2—C5—H5	126.5	C10—C9A—H9AA	109.3
C5ⁱ—C5—N2	106.96 (13)	C10—C9A—H9AB	109.3
C5ⁱ—C5—H5	126.5	H9AA—C9A—H9AB	108.0
N2—C6—H6A	109.5	C7—C8A—H8AA	109.8
N2—C6—H6B	109.5	C7—C8A—H8AB	109.8
N2—C6—H6C	109.5	C9—C8A—C7	109.3 (6)
H6A—C6—H6B	109.5	C9—C8A—H8AA	109.8
H6A—C6—H6C	109.5	C9—C8A—H8AB	109.8
H6B—C6—H6C	109.5	H8AA—C8A—H8AB	108.3
Ir1—C7—H7A	116.5		
Ir1—C7—C8—C9A	−21.8 (8)	C5—N2—C4—N2ⁱ	−0.9 (3)
Ir1—C7—C8A—C9	46.8 (7)	C6—N2—C4—Ir1	−1.4 (3)
Ir1—C10—C9A—C8	−43.0 (7)	C6—N2—C4—N2ⁱ	177.51 (16)
C1—N1—C2—C2ⁱ	0.08 (17)	C6—N2—C5—C5ⁱ	−177.88 (18)
C2—N1—C1—Ir1	−176.04 (16)	C7ⁱ—C7—C8—C9A	−101.0 (6)

C2—N1—C1—N1ⁱ	−0.1 (3)	C7ⁱ—C7—C8A—C9	−32.5 (9)
C3—N1—C1—Ir1	4.7 (3)	C7—C8—C9A—C10	43.1 (7)
C3—N1—C1—N1ⁱ	−179.40 (16)	C10—C9—C8A—C7	−44.8 (8)
C3—N1—C2—C2ⁱ	179.37 (17)	C10ⁱ—C10—C9A—C8	36.7 (8)
C4—N2—C5—C5ⁱ	0.58 (19)	C8A—C9—C10—Ir1	21.6 (8)
C5—N2—C4—Ir1	−179.85 (17)	C8A—C9—C10—C10ⁱ	102.2 (6)

Symmetry code: (i) x, $-y+1$, z.

(η^4-Cycloocta-1,5-diene)bis(1,3-diethylimidazol-2-ylidene)iridium(I) iodide (2)

Crystal data

[Ir(C₇H₁₂N₂)₂(C₈H₁₂)]I
$M_r = 675.65$
Orthorhombic, *Pccn*
$a = 10.6041$ (2) Å
$b = 12.3058$ (2) Å
$c = 18.2513$ (3) Å
$V = 2381.65$ (7) Å³
$Z = 4$
$F(000) = 1304$

$D_x = 1.884$ Mg m⁻³
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 17338 reflections
$\theta = 2.7–37.9°$
$\mu = 6.92$ mm⁻¹
$T = 100$ K
Block, orange
$0.38 \times 0.17 \times 0.12$ mm

Data collection

XtaLAB Synergy, Dualflex, HyPix
   diffractometer
Radiation source: micro-focus sealed X-ray
   tube, PhotonJet (Mo) X-ray Source
Mirror monochromator
Detector resolution: 10.0000 pixels mm⁻¹
ω scans
Absorption correction: gaussian
   (CrysAlisPro;Rigaku OD, 2018)

$T_{min} = 0.318$, $T_{max} = 1.000$
59059 measured reflections
6352 independent reflections
3839 reflections with $I > 2\sigma(I)$
$R_{int} = 0.062$
$\theta_{max} = 38.2°$, $\theta_{min} = 2.5°$
$h = -16\rightarrow17$
$k = -21\rightarrow21$
$l = -30\rightarrow31$

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.030$
$wR(F^2) = 0.060$
$S = 1.00$
6352 reflections
130 parameters
0 restraints

Hydrogen site location: inferred from
   neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.022P)^2 + 1.4351P]$
   where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} = 0.002$
$\Delta\rho_{max} = 1.54$ e Å⁻³
$\Delta\rho_{min} = -0.83$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	U_{iso}*/U_{eq}
Ir1	0.750000	0.750000	0.50409 (2)	0.01428 (3)
N1	0.66768 (19)	0.56037 (15)	0.40954 (10)	0.0170 (4)

N2	0.5533 (2)	0.69802 (15)	0.38262 (10)	0.0160 (4)
C1	0.6489 (2)	0.66696 (18)	0.42698 (11)	0.0147 (4)
C2	0.5863 (2)	0.52747 (19)	0.35491 (12)	0.0197 (5)
H2	0.582250	0.458717	0.333882	0.024*
C3	0.5145 (2)	0.61300 (19)	0.33789 (13)	0.0193 (4)
H3	0.450781	0.615208	0.302881	0.023*
C4	0.7633 (2)	0.49015 (18)	0.44314 (14)	0.0217 (5)
H4A	0.812864	0.532569	0.477466	0.026*
H4B	0.819765	0.463618	0.405332	0.026*
C5	0.7056 (3)	0.3938 (2)	0.48314 (15)	0.0258 (5)
H5A	0.666738	0.345931	0.448297	0.039*
H5B	0.643277	0.419285	0.517187	0.039*
H5C	0.770442	0.355409	0.509178	0.039*
C6	0.5011 (2)	0.80817 (19)	0.37857 (13)	0.0197 (5)
H6A	0.521838	0.847322	0.423096	0.024*
H6B	0.409999	0.804105	0.374896	0.024*
C7	0.5529 (3)	0.86951 (19)	0.31300 (13)	0.0238 (5)
H7A	0.531961	0.831132	0.268855	0.036*
H7B	0.642915	0.875246	0.317200	0.036*
H7C	0.516639	0.940945	0.311475	0.036*
C8	0.8334 (2)	0.8517 (2)	0.59035 (13)	0.0218 (5)
H8	0.877875	0.916010	0.572020	0.026*
C9	0.9006 (2)	0.7551 (2)	0.58265 (13)	0.0232 (4)
H9	0.983639	0.765350	0.560128	0.028*
C10	0.8939 (3)	0.6559 (2)	0.63123 (15)	0.0317 (6)
H10A	0.940977	0.669868	0.675826	0.038*
H10B	0.933711	0.595331	0.606313	0.038*
C11	0.7595 (3)	0.6247 (2)	0.65118 (15)	0.0329 (6)
H11A	0.756815	0.547796	0.662661	0.039*
H11B	0.734462	0.664384	0.694739	0.039*
I1	0.250000	0.750000	0.71451 (2)	0.02175 (4)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ir1	0.01317 (5)	0.01477 (5)	0.01492 (5)	−0.00400 (6)	0.000	0.000
N1	0.0177 (10)	0.0156 (8)	0.0176 (9)	−0.0029 (7)	−0.0019 (7)	−0.0005 (7)
N2	0.0179 (10)	0.0145 (9)	0.0155 (8)	−0.0021 (7)	−0.0004 (8)	−0.0007 (7)
C1	0.0151 (10)	0.0148 (9)	0.0144 (9)	−0.0025 (7)	0.0000 (7)	−0.0004 (8)
C2	0.0237 (12)	0.0175 (10)	0.0178 (10)	−0.0052 (9)	−0.0024 (9)	−0.0011 (8)
C3	0.0213 (12)	0.0216 (11)	0.0149 (10)	−0.0029 (9)	−0.0039 (9)	−0.0012 (9)
C4	0.0212 (14)	0.0171 (9)	0.0267 (11)	−0.0022 (9)	−0.0066 (10)	0.0008 (8)
C5	0.0333 (14)	0.0170 (11)	0.0271 (13)	−0.0048 (10)	−0.0070 (11)	0.0032 (10)
C6	0.0194 (12)	0.0193 (11)	0.0205 (11)	0.0035 (9)	0.0006 (9)	−0.0013 (9)
C7	0.0325 (15)	0.0189 (11)	0.0200 (11)	−0.0005 (10)	−0.0016 (10)	0.0007 (9)
C8	0.0190 (12)	0.0247 (12)	0.0216 (11)	−0.0084 (9)	−0.0032 (9)	−0.0039 (10)
C9	0.0175 (9)	0.0292 (11)	0.0230 (10)	−0.0072 (12)	−0.0054 (8)	0.0031 (13)
C10	0.0270 (15)	0.0384 (15)	0.0297 (14)	−0.0029 (12)	−0.0084 (11)	0.0090 (12)

C11	0.0319 (16)	0.0418 (14)	0.0249 (11)	−0.0040 (14)	−0.0006 (13)	0.0124 (11)
I1	0.01710 (8)	0.02436 (9)	0.02378 (10)	−0.00028 (11)	0.000	0.000

Geometric parameters (Å, º)

Ir1—C1ⁱ	2.043 (2)	C5—H5B	0.9600
Ir1—C1	2.043 (2)	C5—H5C	0.9600
Ir1—C8	2.197 (2)	C6—H6A	0.9700
Ir1—C8ⁱ	2.197 (2)	C6—H6B	0.9700
Ir1—C9ⁱ	2.147 (2)	C6—C7	1.518 (3)
Ir1—C9	2.147 (2)	C7—H7A	0.9600
N1—C1	1.364 (3)	C7—H7B	0.9600
N1—C2	1.379 (3)	C7—H7C	0.9600
N1—C4	1.467 (3)	C8—H8	0.9800
N2—C1	1.353 (3)	C8—C9	1.393 (4)
N2—C3	1.389 (3)	C8—C11ⁱ	1.513 (4)
N2—C6	1.466 (3)	C9—H9	0.9800
C2—H2	0.9300	C9—C10	1.510 (4)
C2—C3	1.336 (3)	C10—H10A	0.9700
C3—H3	0.9300	C10—H10B	0.9700
C4—H4A	0.9700	C10—C11	1.520 (4)
C4—H4B	0.9700	C11—H11A	0.9700
C4—C5	1.521 (3)	C11—H11B	0.9700
C5—H5A	0.9600		
C1ⁱ—Ir1—C1	92.93 (12)	H5A—C5—H5B	109.5
C1—Ir1—C8ⁱ	89.85 (9)	H5A—C5—H5C	109.5
C1ⁱ—Ir1—C8ⁱ	171.79 (9)	H5B—C5—H5C	109.5
C1—Ir1—C8	171.79 (9)	N2—C6—H6A	109.4
C1ⁱ—Ir1—C8	89.85 (9)	N2—C6—H6B	109.4
C1ⁱ—Ir1—C9ⁱ	149.88 (10)	N2—C6—C7	111.30 (19)
C1ⁱ—Ir1—C9	93.15 (9)	H6A—C6—H6B	108.0
C1—Ir1—C9	149.88 (10)	C7—C6—H6A	109.4
C1—Ir1—C9ⁱ	93.15 (9)	C7—C6—H6B	109.4
C8ⁱ—Ir1—C8	88.46 (13)	C6—C7—H7A	109.5
C9ⁱ—Ir1—C8	80.66 (9)	C6—C7—H7B	109.5
C9—Ir1—C8	37.37 (10)	C6—C7—H7C	109.5
C9ⁱ—Ir1—C8ⁱ	37.37 (10)	H7A—C7—H7B	109.5
C9—Ir1—C8ⁱ	80.66 (9)	H7A—C7—H7C	109.5
C9ⁱ—Ir1—C9	96.21 (13)	H7B—C7—H7C	109.5
C1—N1—C2	111.1 (2)	Ir1—C8—H8	114.1
C1—N1—C4	124.74 (19)	C9—C8—Ir1	69.36 (13)
C2—N1—C4	124.18 (19)	C9—C8—H8	114.1
C1—N2—C3	111.16 (19)	C9—C8—C11ⁱ	124.8 (2)
C1—N2—C6	125.05 (19)	C11ⁱ—C8—Ir1	111.88 (17)
C3—N2—C6	123.70 (19)	C11ⁱ—C8—H8	114.1
N1—C1—Ir1	124.39 (16)	Ir1—C9—H9	113.1
N2—C1—Ir1	131.63 (17)	C8—C9—Ir1	73.27 (14)

N2—C1—N1	103.98 (19)	C8—C9—H9	113.1
N1—C2—H2	126.5	C8—C9—C10	127.3 (2)
C3—C2—N1	107.1 (2)	C10—C9—Ir1	109.49 (17)
C3—C2—H2	126.5	C10—C9—H9	113.1
N2—C3—H3	126.6	C9—C10—H10A	109.0
C2—C3—N2	106.7 (2)	C9—C10—H10B	109.0
C2—C3—H3	126.6	C9—C10—C11	112.9 (2)
N1—C4—H4A	109.1	H10A—C10—H10B	107.8
N1—C4—H4B	109.1	C11—C10—H10A	109.0
N1—C4—C5	112.4 (2)	C11—C10—H10B	109.0
H4A—C4—H4B	107.8	C8^i—C11—C10	112.7 (2)
C5—C4—H4A	109.1	C8^i—C11—H11A	109.0
C5—C4—H4B	109.1	C8^i—C11—H11B	109.0
C4—C5—H5A	109.5	C10—C11—H11A	109.0
C4—C5—H5B	109.5	C10—C11—H11B	109.0
C4—C5—H5C	109.5	H11A—C11—H11B	107.8
Ir1—C8—C9—C10	−102.1 (2)	C3—N2—C6—C7	−76.3 (3)
Ir1—C9—C10—C11	−39.1 (3)	C4—N1—C1—Ir1	−0.3 (3)
N1—C2—C3—N2	0.1 (3)	C4—N1—C1—N2	179.6 (2)
C1—N1—C2—C3	−0.6 (3)	C4—N1—C2—C3	−179.4 (2)
C1—N1—C4—C5	118.4 (2)	C6—N2—C1—Ir1	2.5 (3)
C1—N2—C3—C2	0.4 (3)	C6—N2—C1—N1	−177.47 (19)
C1—N2—C6—C7	100.0 (3)	C6—N2—C3—C2	177.2 (2)
C2—N1—C1—Ir1	−179.19 (16)	C8—C9—C10—C11	44.4 (4)
C2—N1—C1—N2	0.8 (3)	C9—C10—C11—C8^i	33.9 (3)
C2—N1—C4—C5	−62.9 (3)	C11^i—C8—C9—Ir1	102.9 (2)
C3—N2—C1—Ir1	179.27 (17)	C11^i—C8—C9—C10	0.8 (4)
C3—N2—C1—N1	−0.7 (3)		

Symmetry code: (i) −x+3/2, −y+3/2, z.