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## Structure Reports

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1,2:5,6-Di-O-isopropylidene-3-C-methyl- $\alpha$ -D-allofuranose

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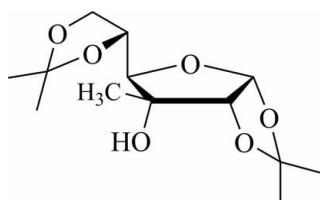
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Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.007$  Å; disorder in main residue;  $R$  factor = 0.046;  $wR$  factor = 0.130; data-to-parameter ratio = 15.2.

The title carbohydrate,  $\text{C}_{13}\text{H}_{22}\text{O}_6$ , is a derivative of D-glucose, in which the furanosidic and isopropylidene rings are in twisted conformations. The mean plane of the furanosidic ring makes a dihedral angle of  $70.32$  ( $18$ ) $^\circ$  with the mean plane of the fused isopropylidene ring. The methyl groups in the other isopropylidene ring are disordered over two sets of sites, with an occupancy ratio of  $0.74$  (6): $0.26$  (6). In the crystal, molecules are linked by  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds into chains with graph-set notation  $C(5)$  along  $[100]$ . Weak  $\text{C}-\text{H}\cdots\text{O}$  interactions also occur.

## Related literature

For background information on this class of compound, see: Bio *et al.* (2004); Canuto *et al.* (2007); Mane *et al.* (2008); Yoneda *et al.* (2011). For details of ring-puckering calculations, see: Cremer & Pople (1975). Graph-set notation for the description of hydrogen-bonding motifs is given by Bernstein *et al.* (1995).



## Experimental

## Crystal data

$\text{C}_{13}\text{H}_{22}\text{O}_6$   
 $M_r = 274.3$   
 Triclinic,  $P1$   
 $a = 5.503$  (4) Å  
 $b = 8.113$  (1) Å  
 $c = 9.122$  (2) Å

$\alpha = 99.65$  (2) $^\circ$   
 $\beta = 103.69$  (3) $^\circ$   
 $\gamma = 98.86$  (2) $^\circ$   
 $V = 382.0$  (3) Å $^3$   
 $Z = 1$   
 Mo  $K\alpha$  radiation

$\mu = 0.09$  mm $^{-1}$   
 $T = 293$  K

 $0.14 \times 0.11 \times 0.08$  mm

## Data collection

Nonius KappaCCD diffractometer  
 9589 measured reflections  
 2758 independent reflections

2107 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.043$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.046$   
 $wR(F^2) = 0.130$   
 $S = 1.04$   
 2758 reflections  
 181 parameters  
 8 restraints  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.21$  e Å $^{-3}$

$\Delta\rho_{\text{min}} = -0.24$  e Å $^{-3}$   
 Absolute structure: Flack  $x$  calculated using 872 quotients  $[(I^+) - (I^-)] / [(I^+) + (I^-)]$  (Parsons & Flack, 2004). There is insufficient information present to define handedness

Table 1

Hydrogen-bond geometry (Å,  $^\circ$ ).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{O}3-\text{H}3\cdots\text{O}4^{\text{i}}$	0.82	2.28	3.022 (4)	152
$\text{C}1-\text{H}1\cdots\text{O}2^{\text{ii}}$	0.98	2.58	3.218 (5)	123
$\text{C}2-\text{H}2\cdots\text{O}6^{\text{iii}}$	0.98	2.54	3.504 (5)	167

Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $x + 1, y, z$ ; (iii)  $x, y, z + 1$ .

Data collection: COLLECT (Nonius, 2004); cell refinement: DIRAX/LSQ (Duisenberg, 1992); data reduction: EVALCCD (Duisenberg *et al.*, 2003); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL2013 (Sheldrick, 2013); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012); software used to prepare material for publication: WinGX (Farrugia, 2012).

This work was supported by the Brazilian agencies FAPERJ, CAPES and CNPq, and by the X-ray diffraction laboratory LDRX-UFF for the data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PK2484).

## References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **35**, 1555–1573.
- Bio, M. M., Xu, F., Waters, M., Williams, J. M., Savary, K. A., Cowden, C. J., Yang, C., Buck, E., Song, Z. J., Tschaen, D., Volante, R. P., Reamer, R. A. & Grabowski, E. J. J. (2004). *J. Org. Chem.* **69**, 6257–6266.
- Canuto, C. V. B. S., Gomes, C. R. B., Marques, I. P., Faro, L. V., Santos, F. C., Frugulhetti, I. C. P. P., Souza, T. M. L., Cunha, A. C., Romero, G. A., Ferreira, V. F. & Souza, M. C. B. V. (2007). *Lett. Drug. Des. Discov.* **4**, 404–409.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Duisenberg, A. J. M. (1992). *J. Appl. Cryst.* **25**, 92–96.
- Duisenberg, A. J. M., Kroon-Batenburg, L. M. J. & Schreurs, A. M. M. (2003). *J. Appl. Cryst.* **36**, 220–229.
- Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.
- Mane, R. S., Kumar, K. S. A. & Dhavale, D. D. (2008). *J. Org. Chem.* **73**, 3284–3287.
- Nonius (2004). COLLECT. Nonius BV, Delft, The Netherlands.
- Parsons, S. & Flack, H. (2004). *Acta Cryst.* **A60**, s61.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Sheldrick, G. M. (2013). SHELXL2013. University of Göttingen, Germany.
- Yoneda, J. D., Velloso, M. H. R., Leal, K. Z., Azeredo, R. B. V., Sugiura, M., Albuquerque, M. G., Santos, F. C., Souza, M. C. B. V., Cunha, A. C., Seidl, P. R., Alencastro, R. B. & Ferreira, V. F. (2011). *J. Mol. Struct.* **985**, 1–4.

## supplementary materials

*Acta Cryst.* (2013). E69, o1512 [doi:10.1107/S1600536813022447]

**1,2:5,6-Di-*O*-isopropylidene-3-*C*-methyl- $\alpha$ -*D*-allofuranose**

**Luana da Silva Magalhães Forezi, Marcos Moitrel Pequeno Silva, Fernanda da Costa Santos, Vitor Francisco Ferreira and Maria Cecília Bastos Vieira de Souza**

**1. Comment**

The structural modification of carbohydrates has been extensively explored for improvement of their pharmacology properties (Bio *et al.*, 2004; Canuto *et al.* (2007); Mane *et al.* 2008; Yoneda *et al.* 2011). As ongoing research in developing potentially new drugs, we report here the structure of 1,2:5,6-Di-*O*-isopropylidene-3-*C*-methyl- $\alpha$ -*D*-allofuranose.

In the title molecule (Fig. 1), the torsion angle formed by atoms O4, C1, C2, O2 is 101.6 (3) $^{\circ}$  and that formed by O1, C1, C2, C3, is -130.8 (3) $^{\circ}$ .

The structure exhibits disorder in a isopropylidene group (atoms C71A, C72A, C71B and C72B) over two positions. Rings A (O1,C1,C2,O2,C8) and B (C1,C2,C3,C4,O4) adopt a twisted conformation, with ring-puckering parameters  $q_2 = 0.223$  (5)  $\text{\AA}$ ,  $\varphi_2 = 277$  (1) $^{\circ}$ ;  $q_2 = 0.376$  (5)  $\text{\AA}$ ,  $\varphi_2 = 89.8$  (6) $^{\circ}$ , respectively. Ring C (O5,C5,C6,O6,C7) also shows a twist conformation, with ring-puckering parameters  $q_2 = 0.404$  (3)  $\text{\AA}$ ,  $\varphi_2 = 196.021$  (1) $^{\circ}$  (Cremer & Pople, 1975).

In the crystal packing, molecules are linked by O—H $\cdots$ O hydrogen bonds into chains with graph-set notation  $C(5)$  along [100] (Bernstein *et al.*, 1995). There are also short C—H $\cdots$ O interactions that form a  $C(7)$  chain motif along [001] direction (Fig. 2).

**2. Experimental**

The reaction for obtaining the title compound was taken under nitrogen in a tribulcate vessel. To it, 2.64 ml (3.95 mmol) of methyl magnesium bromide 3 *M* diluted in THF (7.9 mmol, 2 eq) was added. Then, under vigorous stirring and in a ice bath, 3.96 mmol (1.02 g) diluted in dry THF was poured into the solution. The reaction took place for 5 h in room temperature. After finishing the reaction, it was slowly added dropwise 10 ml of distilled water and 1 g of Celite, filtering the resultant product in a Celite layer. The THF solvent was removed by heat. The aqueous phase was extracted with  $\text{CH}_2\text{Cl}_2$  (3  $\times$  30 ml) and the product washed with distilled water (3  $\times$  20 ml), dried with  $\text{MgSO}_4$  anhydrous and the solvent eliminated on a vacuum rotator evaporator apparatus. The solid was recrystallized in hexane and the yellow solid product was obtained with 72% yield. (m.p. = 104–105 $^{\circ}\text{C}$ ) (Bio *et al.*, 2004).

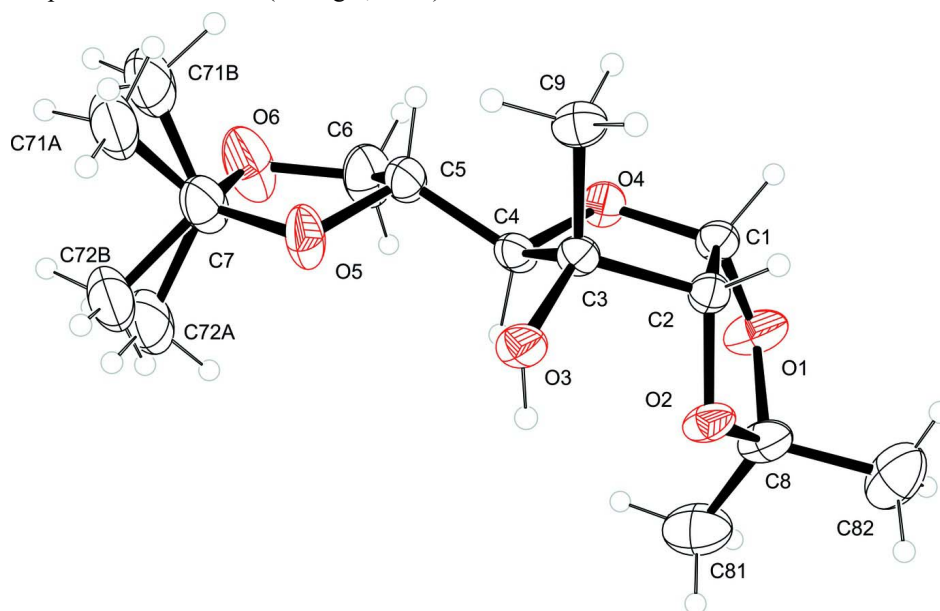
**3. Refinement**

The H atoms were placed at calculated idealized positions and refined using a riding model with individual displacement parameters  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}$  ( $Csp^2$ ) or  $1.5U_{\text{eq}}$  (methyl and hydroxyl groups).

**Computing details**

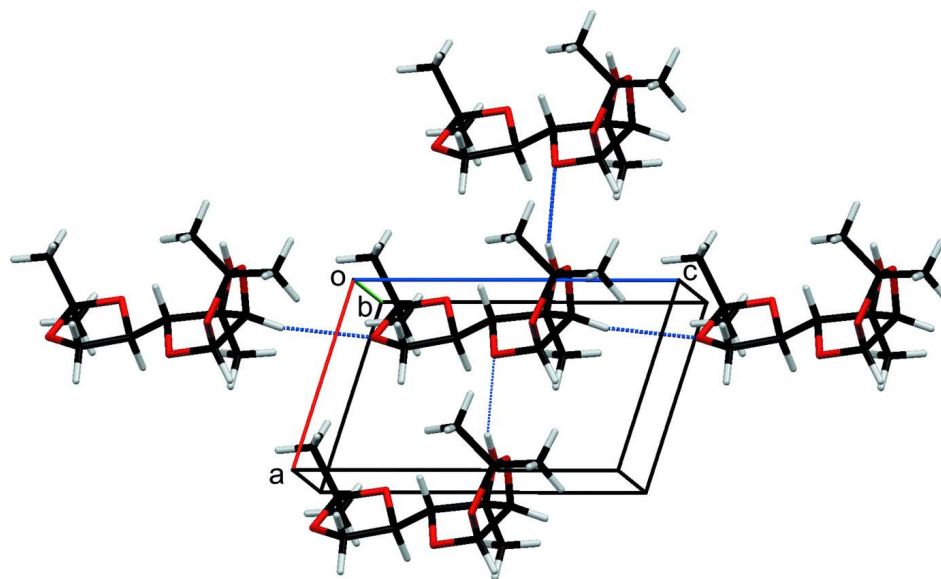
Data collection: *COLLECT* (Nonius, 2004); cell refinement: *DIRAX/LSQ* (Duisenberg, 1992); data reduction: *EVALCCD* (Duisenberg *et al.*, 2003); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2013* (Sheldrick, 2013); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to

prepare material for publication: *WinGX* (Farrugia, 2012).



**Figure 1**

Ellipsoid plot representation of the molecular structure of compound I with displacement ellipsoids drawn at the 30% probability level. Atoms C71A/B and C72A/B are disordered with fractional occupancies of 0.74 (6):0.26 (6) for the A and B components, respectively.



**Figure 2**

A packing diagram of (I), viewed approximately down the *b* axis. Hydrogen-bonds are shown by dashed lines.

1,2:5,6-Di-O-isopropylidene-3-C-methyl- $\alpha$ -D-allofuranose

Crystal data

$C_{13}H_{22}O_6$	$Z = 1$
$M_r = 274.3$	$F(000) = 148$
Triclinic, $P1$	$D_x = 1.192 \text{ Mg m}^{-3}$
Hall symbol: $P1$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$a = 5.503 (4) \text{ \AA}$	Cell parameters from 3284 reflections
$b = 8.113 (1) \text{ \AA}$	$\theta = 3.1\text{--}27.5^\circ$
$c = 9.122 (2) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$\alpha = 99.65 (2)^\circ$	$T = 293 \text{ K}$
$\beta = 103.69 (3)^\circ$	Prism, colourless
$\gamma = 98.86 (2)^\circ$	$0.14 \times 0.11 \times 0.08 \text{ mm}$
$V = 382.0 (3) \text{ \AA}^3$	

Data collection

Nonius KappaCCD diffractometer	2758 independent reflections
Radiation source: Enraf–Nonius FR590	2107 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\text{int}} = 0.043$
Detector resolution: 9 pixels $\text{mm}^{-1}$	$\theta_{\text{max}} = 25.7^\circ$ , $\theta_{\text{min}} = 3.1^\circ$
CCD rotation images, thick slices scans	$h = -6 \rightarrow 6$
9589 measured reflections	$k = -9 \rightarrow 9$
	$l = -11 \rightarrow 11$

Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0729P)^2 + 0.0707P]$
Least-squares matrix: full	where $P = (F_o^2 + 2F_c^2)/3$
$R[F^2 > 2\sigma(F^2)] = 0.046$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$wR(F^2) = 0.130$	$\Delta\rho_{\text{max}} = 0.21 \text{ e \AA}^{-3}$
$S = 1.04$	$\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$
2758 reflections	Extinction correction: <i>SHELXL2013</i> (Sheldrick, 2013)
181 parameters	Extinction coefficient: 0.20 (3)
8 restraints	Absolute structure: Flack $x$ calculated using 872
Hydrogen site location: inferred from neighbouring sites	quotients $[(I^+)-(I^-)]/[(I^+)+(I^-)]$ (Parsons & Flack, 2004). There is insufficient information present to define handedness.
H-atom parameters constrained	Absolute structure parameter: $-0.2 (5)$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
O1	0.2339 (6)	-0.0979 (4)	0.6028 (5)	0.0646 (10)	
O2	-0.0659 (5)	0.0519 (3)	0.6425 (3)	0.0456 (8)	
O3	-0.0994 (5)	0.3540 (4)	0.5720 (3)	0.0434 (7)	
H3	-0.2119	0.2677	0.5383	0.065*	
O4	0.3788 (5)	0.1260 (4)	0.4944 (3)	0.0446 (7)	
O5	0.0585 (6)	0.3883 (5)	0.2656 (4)	0.0633 (10)	

O6	0.2663 (8)	0.3148 (8)	0.0861 (5)	0.1016 (17)	
C1	0.3517 (7)	0.0751 (5)	0.6312 (5)	0.0394 (9)	
H1	0.5169	0.0992	0.7088	0.047*	
C2	0.1689 (7)	0.1724 (5)	0.6883 (4)	0.0357 (9)	
H2	0.2256	0.2197	0.7997	0.054*	
C3	0.1409 (6)	0.3087 (5)	0.5930 (4)	0.0327 (9)	
C4	0.1819 (7)	0.2156 (5)	0.4429 (5)	0.0397 (9)	
H4	0.0255	0.1324	0.3853	0.048*	
C5	0.2672 (8)	0.3188 (6)	0.3367 (5)	0.0486 (11)	
H5	0.4124	0.411	0.3948	0.058*	
C6	0.3295 (12)	0.2203 (10)	0.2017 (6)	0.0787 (18)	
H6A	0.5092	0.2151	0.2255	0.094*	
H6B	0.2282	0.1051	0.1701	0.094*	
C7	0.0949 (10)	0.4150 (9)	0.1214 (6)	0.0727 (17)	
C71A	0.222 (7)	0.598 (3)	0.144 (3)	0.149 (7)	0.74 (6)
H71A	0.0946	0.6662	0.1244	0.223*	0.74 (6)
H71C	0.3259	0.6077	0.0732	0.223*	0.74 (6)
H71B	0.3279	0.6382	0.2478	0.223*	0.74 (6)
C72A	-0.147 (3)	0.345 (4)	-0.0032 (17)	0.116 (6)	0.74 (6)
H72A	-0.1074	0.3085	-0.099	0.174*	0.74 (6)
H72B	-0.2456	0.4326	-0.0132	0.174*	0.74 (6)
H72C	-0.2427	0.2503	0.0222	0.174*	0.74 (6)
C71B	0.204 (18)	0.595 (8)	0.107 (11)	0.149 (7)	0.26 (6)
H71D	0.1085	0.6732	0.145	0.223*	0.26 (6)
H71E	0.1935	0.5954	0.0005	0.223*	0.26 (6)
H71F	0.3798	0.6279	0.1663	0.223*	0.26 (6)
C72B	-0.180 (6)	0.404 (11)	0.027 (6)	0.116 (6)	0.26 (6)
H72D	-0.1971	0.3568	-0.0794	0.174*	0.26 (6)
H72E	-0.2186	0.5155	0.0375	0.174*	0.26 (6)
H72F	-0.2952	0.3311	0.0647	0.174*	0.26 (6)
C8	-0.0148 (8)	-0.1149 (6)	0.6267 (6)	0.0517 (12)	
C81	-0.2032 (11)	-0.2263 (8)	0.4845 (8)	0.0820 (17)	
H81A	-0.1132	-0.2828	0.4198	0.123*	
H81B	-0.3017	-0.1573	0.429	0.123*	
H81C	-0.3152	-0.31	0.5137	0.123*	
C82	-0.0177 (13)	-0.1798 (8)	0.7692 (8)	0.0823 (18)	
H82A	0.1478	-0.2012	0.8139	0.123*	
H82C	-0.1428	-0.2839	0.7441	0.123*	
H82B	-0.0598	-0.0963	0.8418	0.123*	
C9	0.3413 (8)	0.4681 (5)	0.6694 (5)	0.0454 (10)	
H9C	0.3209	0.5524	0.6076	0.068*	
H9B	0.5079	0.4417	0.6796	0.068*	
H9A	0.3227	0.5117	0.7699	0.068*	

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0496 (18)	0.0411 (18)	0.120 (3)	0.0181 (14)	0.0450 (19)	0.0236 (18)
O2	0.0326 (14)	0.0361 (15)	0.079 (2)	0.0128 (12)	0.0242 (13)	0.0242 (14)
O3	0.0304 (14)	0.0420 (16)	0.0656 (19)	0.0149 (12)	0.0176 (13)	0.0193 (14)

O4	0.0368 (14)	0.0536 (17)	0.0531 (17)	0.0194 (13)	0.0218 (12)	0.0149 (14)
O5	0.066 (2)	0.102 (3)	0.0433 (17)	0.0418 (19)	0.0250 (15)	0.0373 (17)
O6	0.079 (3)	0.203 (5)	0.059 (2)	0.070 (3)	0.039 (2)	0.056 (3)
C1	0.031 (2)	0.043 (2)	0.052 (2)	0.0133 (17)	0.0156 (17)	0.0175 (19)
C2	0.0308 (19)	0.041 (2)	0.040 (2)	0.0105 (16)	0.0139 (16)	0.0117 (17)
C3	0.0270 (18)	0.037 (2)	0.038 (2)	0.0114 (16)	0.0119 (15)	0.0107 (17)
C4	0.0302 (19)	0.048 (2)	0.041 (2)	0.0087 (17)	0.0079 (16)	0.0101 (18)
C5	0.039 (2)	0.070 (3)	0.042 (2)	0.014 (2)	0.0138 (19)	0.021 (2)
C6	0.078 (4)	0.122 (5)	0.061 (3)	0.045 (4)	0.038 (3)	0.037 (3)
C7	0.055 (3)	0.132 (5)	0.045 (3)	0.030 (3)	0.019 (2)	0.037 (3)
C71A	0.239 (13)	0.140 (8)	0.076 (15)	0.001 (8)	0.059 (11)	0.059 (8)
C72A	0.057 (5)	0.26 (2)	0.048 (6)	0.069 (7)	0.023 (4)	0.033 (8)
C71B	0.239 (13)	0.140 (8)	0.076 (15)	0.001 (8)	0.059 (11)	0.059 (8)
C72B	0.057 (5)	0.26 (2)	0.048 (6)	0.069 (7)	0.023 (4)	0.033 (8)
C8	0.043 (2)	0.041 (2)	0.083 (3)	0.017 (2)	0.031 (2)	0.021 (2)
C81	0.066 (4)	0.056 (3)	0.115 (5)	0.002 (3)	0.025 (3)	0.005 (3)
C82	0.097 (4)	0.073 (4)	0.112 (5)	0.041 (3)	0.055 (4)	0.056 (4)
C9	0.043 (2)	0.038 (2)	0.056 (2)	0.0101 (18)	0.0146 (18)	0.0080 (18)

*Geometric parameters (Å, °)*

O1—C1	1.405 (5)	C7—C71B	1.53 (3)
O1—C8	1.426 (5)	C7—C72B	1.53 (2)
O2—C8	1.415 (5)	C71A—H71A	0.96
O2—C2	1.421 (5)	C71A—H71C	0.96
O3—C3	1.405 (4)	C71A—H71B	0.96
O3—H3	0.82	C72A—H72A	0.96
O4—C1	1.412 (5)	C72A—H72B	0.96
O4—C4	1.428 (5)	C72A—H72C	0.96
O5—C7	1.422 (6)	C71B—H71D	0.96
O5—C5	1.423 (5)	C71B—H71E	0.96
O6—C7	1.395 (8)	C71B—H71F	0.96
O6—C6	1.415 (8)	C72B—H72D	0.96
C1—C2	1.504 (5)	C72B—H72E	0.96
C1—H1	0.98	C72B—H72F	0.96
C2—C3	1.521 (5)	C8—C82	1.485 (8)
C2—H2	0.98	C8—C81	1.503 (8)
C3—C9	1.506 (6)	C81—H81A	0.96
C3—C4	1.532 (6)	C81—H81B	0.96
C4—C5	1.493 (5)	C81—H81C	0.96
C4—H4	0.98	C82—H82A	0.96
C5—C6	1.492 (7)	C82—H82C	0.96
C5—H5	0.98	C82—H82B	0.96
C6—H6A	0.97	C9—H9C	0.96
C6—H6B	0.97	C9—H9B	0.96
C7—C72A	1.489 (12)	C9—H9A	0.96
C7—C71A	1.502 (16)		
C1—O1—C8	110.7 (3)	O5—C7—C72B	102 (2)
C8—O2—C2	109.1 (3)	C71B—C7—C72B	97 (2)

C3—O3—H3	109.5	C7—C71A—H71A	109.5
C1—O4—C4	109.0 (3)	C7—C71A—H71C	109.5
C7—O5—C5	106.9 (4)	H71A—C71A—H71C	109.5
C7—O6—C6	109.5 (4)	C7—C71A—H71B	109.5
O1—C1—O4	111.8 (4)	H71A—C71A—H71B	109.5
O1—C1—C2	105.2 (3)	H71C—C71A—H71B	109.5
O4—C1—C2	106.8 (3)	C7—C72A—H72A	109.5
O1—C1—H1	110.9	C7—C72A—H72B	109.5
O4—C1—H1	110.9	H72A—C72A—H72B	109.5
C2—C1—H1	110.9	C7—C72A—H72C	109.5
O2—C2—C1	104.0 (3)	H72A—C72A—H72C	109.5
O2—C2—C3	108.0 (3)	H72B—C72A—H72C	109.5
C1—C2—C3	105.1 (3)	C7—C71B—H71D	109.5
O2—C2—H2	113	C7—C71B—H71E	109.5
C1—C2—H2	113	H71D—C71B—H71E	109.5
C3—C2—H2	113	C7—C71B—H71F	109.5
O3—C3—C9	107.8 (3)	H71D—C71B—H71F	109.5
O3—C3—C2	112.6 (3)	H71E—C71B—H71F	109.5
C9—C3—C2	110.7 (3)	C7—C72B—H72D	109.5
O3—C3—C4	112.9 (3)	C7—C72B—H72E	109.5
C9—C3—C4	112.8 (3)	H72D—C72B—H72E	109.5
C2—C3—C4	100.0 (3)	C7—C72B—H72F	109.5
O4—C4—C5	107.2 (3)	H72D—C72B—H72F	109.5
O4—C4—C3	103.8 (3)	H72E—C72B—H72F	109.5
C5—C4—C3	118.7 (3)	O2—C8—O1	105.1 (3)
O4—C4—H4	108.9	O2—C8—C82	110.5 (4)
C5—C4—H4	108.9	O1—C8—C82	110.4 (4)
C3—C4—H4	108.9	O2—C8—C81	108.5 (4)
O5—C5—C6	102.8 (4)	O1—C8—C81	108.6 (4)
O5—C5—C4	108.1 (3)	C82—C8—C81	113.3 (5)
C6—C5—C4	115.4 (4)	C8—C81—H81A	109.5
O5—C5—H5	110.1	C8—C81—H81B	109.5
C6—C5—H5	110.1	H81A—C81—H81B	109.5
C4—C5—H5	110.1	C8—C81—H81C	109.5
O6—C6—C5	103.3 (5)	H81A—C81—H81C	109.5
O6—C6—H6A	111.1	H81B—C81—H81C	109.5
C5—C6—H6A	111.1	C8—C82—H82A	109.5
O6—C6—H6B	111.1	C8—C82—H82C	109.5
C5—C6—H6B	111.1	H82A—C82—H82C	109.5
H6A—C6—H6B	109.1	C8—C82—H82B	109.5
O6—C7—O5	106.7 (4)	H82A—C82—H82B	109.5
O6—C7—C72A	105.2 (14)	H82C—C82—H82B	109.5
O5—C7—C72A	110.0 (8)	C3—C9—H9C	109.5
O6—C7—C71A	107.4 (16)	C3—C9—H9B	109.5
O5—C7—C71A	107.9 (12)	H9C—C9—H9B	109.5
C72A—C7—C71A	118.9 (19)	C3—C9—H9A	109.5
O6—C7—C71B	106 (5)	H9C—C9—H9A	109.5
O5—C7—C71B	119 (4)	H9B—C9—H9A	109.5
O6—C7—C72B	127 (3)		

C8—O1—C1—O4	-111.3 (4)	C7—O5—C5—C4	-153.6 (4)
C8—O1—C1—C2	4.3 (5)	O4—C4—C5—O5	171.2 (4)
C4—O4—C1—O1	100.8 (3)	C3—C4—C5—O5	-71.9 (4)
C4—O4—C1—C2	-13.8 (4)	O4—C4—C5—C6	56.8 (5)
C8—O2—C2—C1	24.8 (4)	C3—C4—C5—C6	173.7 (4)
C8—O2—C2—C3	136.1 (3)	C7—O6—C6—C5	-21.4 (7)
O1—C1—C2—O2	-17.4 (4)	O5—C5—C6—O6	31.7 (6)
O4—C1—C2—O2	101.6 (3)	C4—C5—C6—O6	149.1 (4)
O1—C1—C2—C3	-130.8 (3)	C6—O6—C7—O5	2.5 (7)
O4—C1—C2—C3	-11.8 (4)	C6—O6—C7—C72A	-114.4 (10)
O2—C2—C3—O3	39.9 (4)	C6—O6—C7—C71A	118.0 (13)
C1—C2—C3—O3	150.4 (3)	C6—O6—C7—C71B	130 (3)
O2—C2—C3—C9	160.6 (3)	C6—O6—C7—C72B	-117 (3)
C1—C2—C3—C9	-88.8 (4)	C5—O5—C7—O6	18.7 (6)
O2—C2—C3—C4	-80.2 (3)	C5—O5—C7—C72A	132.4 (15)
C1—C2—C3—C4	30.3 (4)	C5—O5—C7—C71A	-96.4 (17)
C1—O4—C4—C5	160.1 (3)	C5—O5—C7—C71B	-101 (5)
C1—O4—C4—C3	33.7 (4)	C5—O5—C7—C72B	154 (3)
O3—C3—C4—O4	-158.5 (3)	C2—O2—C8—O1	-22.4 (5)
C9—C3—C4—O4	78.9 (4)	C2—O2—C8—C82	96.6 (5)
C2—C3—C4—O4	-38.6 (3)	C2—O2—C8—C81	-138.5 (4)
O3—C3—C4—C5	82.7 (4)	C1—O1—C8—O2	10.6 (5)
C9—C3—C4—C5	-39.8 (4)	C1—O1—C8—C82	-108.5 (5)
C2—C3—C4—C5	-157.4 (3)	C1—O1—C8—C81	126.6 (4)
C7—O5—C5—C6	-31.1 (6)		

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O3—H3 $\cdots$ O4 <sup>i</sup>	0.82	2.28	3.022 (4)	152
C1—H1 $\cdots$ O2 <sup>ii</sup>	0.98	2.58	3.218 (5)	123
C2—H2 $\cdots$ O6 <sup>iii</sup>	0.98	2.54	3.504 (5)	167

Symmetry codes: (i)  $x-1, y, z$ ; (ii)  $x+1, y, z$ ; (iii)  $x, y, z+1$ .