

Conversion of DNA methyltransferases into azidonucleosidyl transferases via synthetic cofactors

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ABSTRACT

Aziridine-based cofactor mimics have been synthesized and are shown to undergo methyltransferase-dependent DNA alkylation. Notably, each cofactor mimic possesses an azide functionality, to which can be attached an assortment of unnatural groups following methyltransferase-dependent DNA delivery. DNA duplexes modified with these cofactor mimics are capable of undergoing the Staudinger ligation with phosphines tethered to biological functionalities following enzymatic modification. This methodology provides a new tool by which to selectively modify DNA in a methyltransferase-dependent way. The conversion of biological methyltransferases into azidonucleosidyl transferases demonstrated here also holds tremendous promise as a means of identifying, as yet, unknown substrates of methylation.

INTRODUCTION

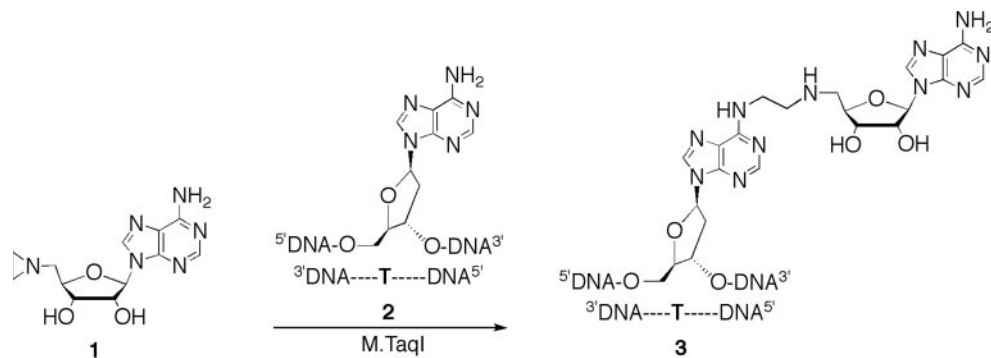
S-adenosyl-L-methionine (SAM)-dependent methylation of nucleic acids and proteins plays an absolutely vital role in the regulation of gene transcription (1–6). Flaws in the activity and expression levels of the eukaryotic DNA methyltransferase (MTase) DNMT1 have been integrally linked to oncogenic potential (7–9). Thus, agents capable of undergoing DNMT1-dependent transfer to DNA might represent an attractive new chemotherapeutic strategy by virtue of altered transcriptional repression mechanisms so often associated with promoter methylation. Such an approach is significantly different from those exemplified by simple inhibition of MTases with SAM analogs. The focal point of efforts described here, substances capable of undergoing transfer to nucleic acids in an MTase-dependent way also hold tremendous promise as biochemical tools by which to dissect and understand biological methylation. Although not elaborated here, such agents might also be used by MTases whose natural

substrates are not nucleic acids. For instance, posttranslational protein methylation plays a large role in transcription regulation and constitutes an important facet of proteomics. The absence of functionality, however, renders the methyl group difficult to identify and isolate from complex biological mixtures. Thus, substances that take part in SAM-dependent MTase pathways may be important proteomic tools in addition to DNA modifying agents. Our interests in the area of cofactor mimicry are reflected by investigations into the synthesis and DNA modification chemistry of **1** (Scheme 1) and related congeners.

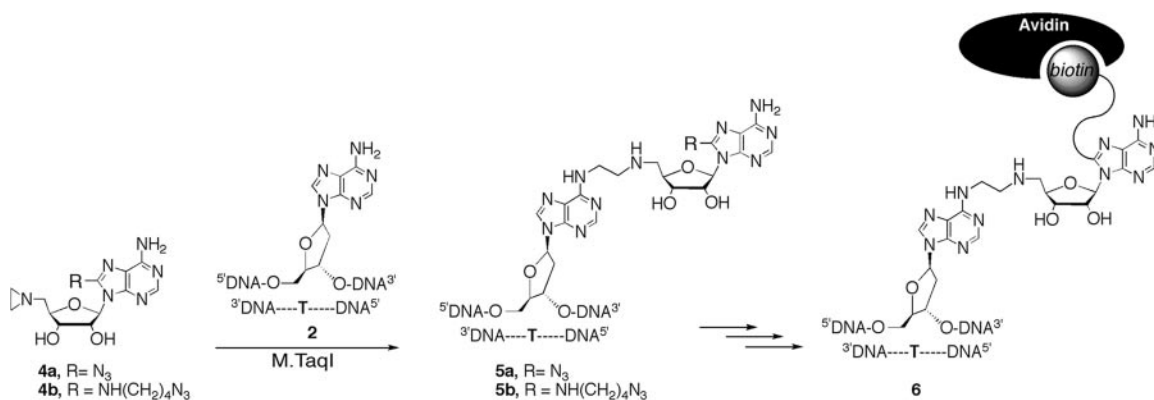
The 5'-aziridine adenylate **1** is a substitute for SAM in the M.TaqI catalyzed alkylation of adenine within the recognition sequence d(TCGA) as depicted in Scheme 1 (10–12). Instead of generating the N6-methyladenine, substrate adenylation is accomplished via ring-opening of the aziridine to yield **3**. This chemistry is tolerant of cofactor C8 modification and has been successfully used to fluorescently tag short oligonucleotides and large plasmid substrates in an M.TaqI-dependent fashion (13). The aziridine nucleoside **1** also undergoes M.HhaI-dependent DNA attachment within the M.HhaI recognition sequence d(GCGC) (11). These findings provide clear evidence of the importance of 5' aziridine adenylates as 'cofactor mimics' of SAM as tools for biology in the short term and potential therapeutic agents in the long term. The further development of these substances requires not only new and more efficient ways by which to construct them, but also an understanding of how these materials might be made compatible with already existing technologies.

Our interest in the development of cofactors related to **1** as universal cofactors dictated that the core structure of **1** be equipped with a handle through which any desired molecule (DNA damaging moiety, affinity matrix handle, fluorophore, etc.) could be appended following MTase-dependent anchoring. Such a handle would need to present a minimal disruption to cofactor–MTase interactions, and be capable of rapid couplings to other reagents under biological conditions. Thus, both the cofactor and its ligation partner would need to be abiotic. As highlighted by Bertozzi, Zhou and others, the Staudinger ligation of azides and *o*-methoxycarbonyl functionalized

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Scheme 1. Nucleoside transfer by M.TaqI.



Scheme 2. Azidonucleoside transfer by M.TaqI and envisioned biotin:avidin associations.

triarylphosphines represents an incredibly powerful reaction in which all our criteria could be met (14–17). Moreover, many azides have been shown to be highly active participants in [2 + 3] Huisgen cycloadditions; these ‘Click chemistry’ reactions also are compatible with biological conditions as most appropriately highlighted by Cravatt and co-workers (18–20). Our program to develop these synthetic cofactors into useful biochemical tools motivated us to pursue chemistry depicted by Scheme 2.

The ability of 8-azidoadenosine (and aryl azides in general) to undergo the Staudinger ligation with triarylphosphines (21) coupled with the ability of C8 azido-SAM to retain its cofactor function (22,23), led to our previously reported synthesis of azido cofactor **4a**. However, until now, we have not shown that such a cofactor could be used by DNA methyltransferases or used as an anchoring point for biomolecule biotinylation. We report here that **4a** is very effective in its role as a synthetic cofactor and that DNA modified with this substance undergoes facile Staudinger ligation with a biotinylated reagent as shown in rather generic fashion by Scheme 2. We also demonstrate here the facile construction and utility of an alkyl azide bearing cofactor **4b**. It is significant that both **4a** and **4b** serve as effective cofactors with multiple DNA methyltransferases and that the lesions created (MTase-dependent) can be biotinylated under biological conditions, and thus also immobilized by virtue of biotin:streptavidin associations. The power of organic synthesis to afford azide-bearing cofactors allows the conversion of DNA MTases into azidonucleosidyl transferases.

MATERIALS AND METHODS

General

pUC19 and all enzymes (unless indicated) were obtained from New England Biolabs. The DNA MTase reactions with M.TaqI were run in buffer A [20 mM Tris-OAc (pH 6.0), 50 mM KOAc, 10 mM Mg(OAc)₂, 0.01% Triton X-100]. The DNA MTase reactions with M.EcoRI were run in buffer B [10 mM Tris-HCl (pH 7.4), 50 mM NaCl, 0.5 mM EDTA, 0.01% Triton X-100] (11). All agarose gels were prepared with a high-melt agarose in 1× TAE containing 0.05 μg/μl ethidium bromide. Following electrophoresis, gels were de-stained for 20 min (1 mM MgSO₄, 10 mM 2-mercaptoethanol) prior to visualization. Bands were visualized at 300 nm using a photo-documentation system. Synthetic oligonucleotides were obtained from Sigma-Genosys and gel-purified prior to use. The concentration of the oligonucleotide was determined at 260 nm using the following molar extinction coefficients: 15 400 (A), 11 500 (G), 8700 (T), 7400 (C). The immobilized streptavidin pull-down assay was performed in buffer C (1 M NaCl, 1 mM EDTA and 10 mM sodium phosphate, pH 7.5) (24).

Synthesis of the azido-based cofactor mimics

The synthesis of the aryl azide cofactor mimic, **4a**, has been previously described (25). The experimental procedures and corresponding spectral data for **4b** and the intermediates **7–12** can be found in Supplementary Material. Stock concentrations

were determined using UV/Vis spectroscopy (260 nm). The concentration of **1** was determined as previously described (10). The molar extinction coefficients were determined to be 6633 for the aryl azide cofactor mimic, **4a**, and 8900 for the alkyl azide cofactor mimic, **4b**.

Restriction/protection assay

Commercially available pUC19 was linearized with R.EcoRI according to manufacturer's protocol (final concentration of 0.2 $\mu\text{g}/\mu\text{l}$ or 114 nM). R.EcoRI was heat-inactivated at 65°C for 15 min prior to further plasmid use. Reaction mixtures were prepared by the addition of appropriate stock solutions to a total volume of 20 μl (in buffer A). The final DNA concentration was 14.3 nM; the final concentrations of the cofactors and the M.TaqI varied upon the specific reaction sequence. All reactions were heated at 65°C for 4 h, followed by cooling to 0°C. The extent of methyltransferase-dependent DNA alkylation was analyzed by the addition of R.TaqI (2 U in an additional 10 μl buffer A), followed by incubation at 65°C for 1 h. Upon cooling to 0°C, Proteinase K (Ambion) (0.02 U in 5 μl H₂O) was added to each reaction and incubated at 37°C for 1 h. The extent of alkylation (as indicated by protection from endonuclease cleavage) was visualized by electrophoresis on a 2% agarose gel.

DNA labeling and duplex formation

The synthetic oligonucleotide utilized for M.TaqI reactions contained the sequence d(TGAATCTCGAGCACCC). The 5' ³²P-labeled oligonucleotide was prepared with T4 polynucleotide kinase and [γ -³²P]ATP using standard methods (26). The labeled oligonucleotide was desalted and unincorporated [γ -³²P]ATP removed via Sephadex G-25 spin column (Amersham). The labeled strand was annealed to its complement d(GGGTGCTCGAGATTCAAA) in 1 \times TE buffer by heating to 80°C (5 min) and cooling to 4°C over 4 h. A similar procedure to prepare the synthetic oligonucleotide for M.EcoRI was followed, but utilized the sequence d(TGAATGAA-TTCGACCC) and its complement d(GGGTCGAATTC-ATTCAAA).

M.TaqI and M.EcoRI reactions with synthetic oligonucleotide

Reaction mixtures were prepared by the addition of appropriate stock solutions to a total volume of 20 μl in either buffer A or buffer B. The final DNA concentration was 1 μM ; final concentration of cofactor was 100 μM ; the final concentrations of M.TaqI and M.EcoRI were 6 and 2 μM , respectively. All reactions were incubated at 37°C for 18 h, followed by cooling to 0°C. Proteinase K (0.02U in 2 μl H₂O) was added to each reaction and digestions were carried out for 1 h at 37°C. The resulting alkylated and proteolyzed samples were then processed as indicated below.

Staudinger ligation with cofactor-linked ³²P-labeled oligonucleotide

An aliquot (5.5 μl , 5 pmol) of cofactor-linked ³²P-labeled duplex was combined with 2.5 μl H₂O, 1 μl 50 mM NaOH and 1 μl 10 mM biotin-linked phosphine **13** or **15** (in DMF). The final concentration of DNA in the ligation reaction was 500 nM. Additional control reactions were also prepared

(containing DMF only). The samples were incubated at 37°C for 14 h. The ligation reaction was either analyzed by DPAGE or ethanol-precipitated (26) with 39 μg tRNA (*E.coli*, Type XX, Strain W) before subsequent processing and data acquisition.

Immobilized streptavidin pull-down assay

ImmunoPure[®] Immobilized Streptavidin (Pierce) was prepared by washing twice with 1 M NaCl (total slurry volume of 30 μl) (24). The washed agarose was added to the ligation reaction in a total volume of 20 μl buffer C. The slurry was gently mixed at room temperature for 1 h. The material was transferred to a micro-spin column and unbound DNA was washed away by subsequent resuspension in 1 M NaCl and centrifugation (3 \times 50 μl). The extent of radiolabeled DNA retention on the streptavidin-linked agarose was determined by liquid scintillation counting of the micro-spin column matrices.

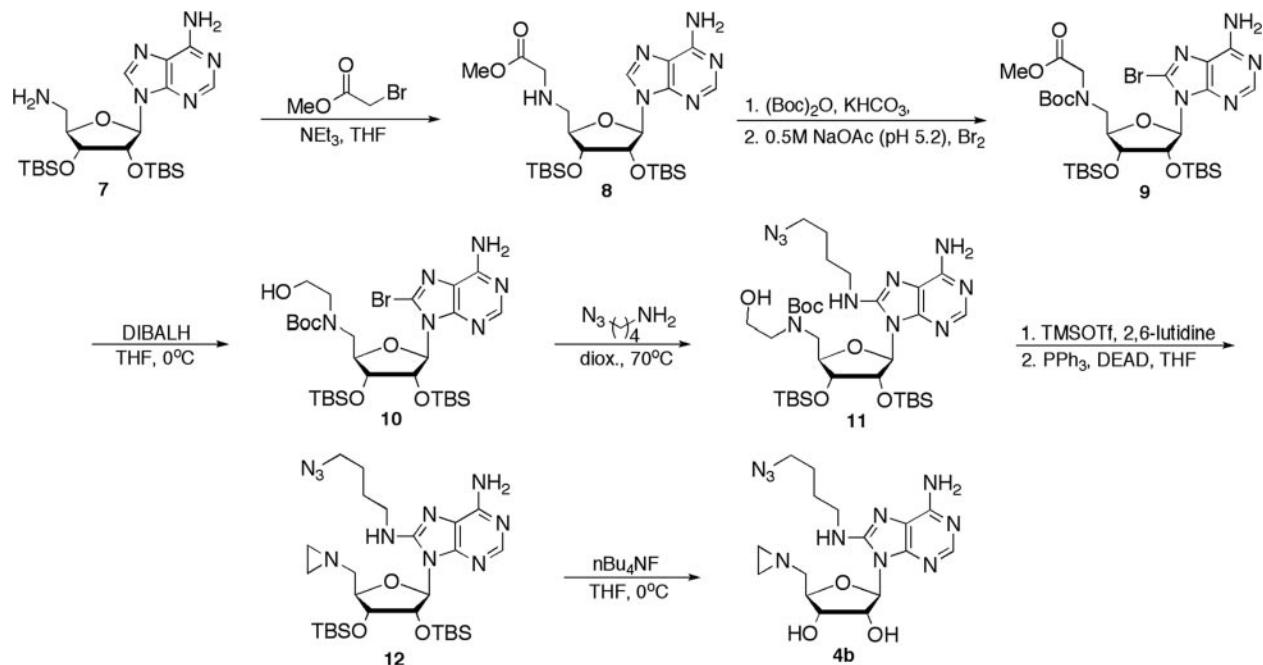
RESULTS

Synthesis of alkyl azide cofactor

The structure and corresponding construction of alkyl azide cofactor **4b** differs significantly from that of aryl azide **4a** (Scheme 3). Installation of the azide-capped butane moiety dictated C5' elaboration prior to base modification. Beginning with an intermediate previously described (27), the primary amine **7** was alkylated with the methyl bromoacetate to provide **8** in 85% yield. Due to anticipated difficulties in achieving effective C8 bromination, the secondary amine functionality of **8** was protected as the *t*-butyl carbamate (28). Subsequent bromination under mildly acidic conditions (29) yielded **9** in 89% yield from **8**. Reduction of the methyl ester to the primary alcohol **10** with DIBALH (30) occurred with a yield of 81%, followed by S_NAr reaction with 4-azido-butylamine (31), to yield guanidine **11** in 49% yield (32). Following Boc deprotection using TMSOTf (33), the crude 5' ethanolamine was converted to aziridine **12** utilizing Mitsunobu conditions (34) in 54% over the two steps. Finally, the silyl ethers were removed with nBu₄NF (35) to yield the desired alkyl azide cofactor mimic **4b**, in 60% yield.

M.TaqI-mediated DNA alkylation

A restriction/protection assay previously described was utilized to analyze the extent of MTase-dependent DNA alkylation by the unsubstituted aziridine cofactor **1** and the two azido-based cofactor mimics **4a** and **4b** (13). Figure 1A illustrates the effect of increasing M.TaqI concentration on the extent of DNA alkylation. In comparing lanes 3–5, 7–9 and 11–13 for the three cofactor mimics at 100 μM , an increase in the amount of DNA alkylation is observed with a gradient ranging from 20 to 200 nM M.TaqI as reflected by the increased protection from the R.TaqI digestion. Significantly, in the absence of M.TaqI, no protection of the DNA is observed (lanes 2, 6 and 10); none of the cofactors tested thus appeared to inhibit R.TaqI nor was non-specific DNA alkylation by any of the cofactors sufficient to render protection from R.TaqI-mediated plasmid scission. Similar trends were observed with an increase in the amount of cofactor (Figure 1B). Increasing cofactor concentration from 10 to



Scheme 3. Azidonucleoside construction.

A Increasing *M.TaqI* concentration

B Increasing cofactor concentration

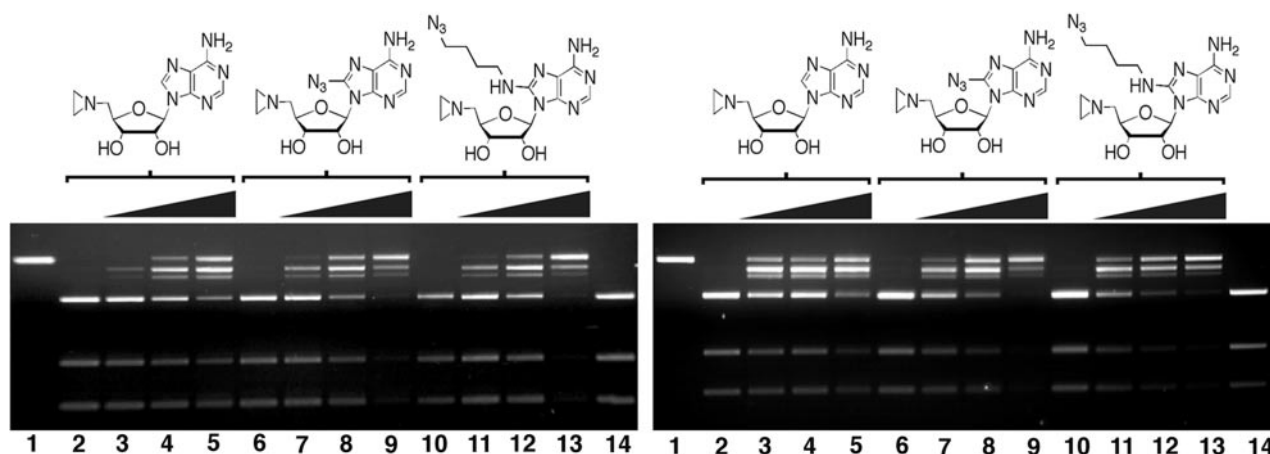


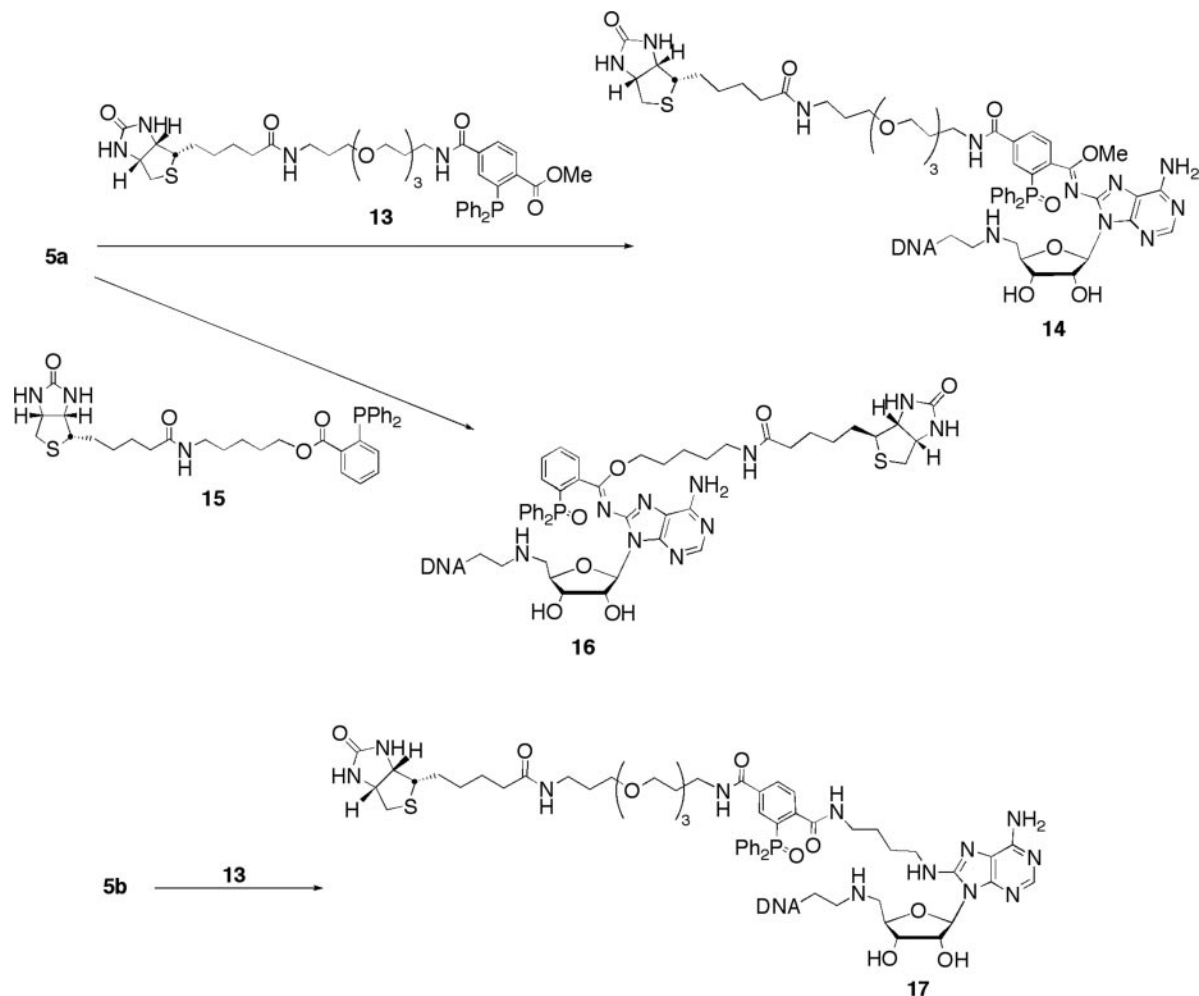
Figure 1. DNA alkylation reactions with pUC19. DNA alkylation reactions of R.EcoRI linearized pUC19 by aziridine cofactor mimics **1**, **4a** and **4b**. Reaction mixtures were prepared by addition of appropriate stock solutions to a total volume of 20 μ l containing 14.3 nM DNA buffered with 20 mM Tris-OAc (pH 6.0), 50 mM KOAc, 10 mM Mg(OAc)₂, 0.01% Triton X-100. The mixtures were analyzed on a 2% agarose gel run at 120 V for 2 h. (A) Increase in *M.TaqI* concentration: (1) DNA; (2) DNA, 100 μ M cofactor **1**, R.Taq ^{α} I; (3) DNA, 100 μ M **1**, 20 nM *M.TaqI*, R.Taq ^{α} I; (4) DNA, 100 μ M **1**, 100 nM *M.TaqI*, R.Taq ^{α} I; (5) DNA, 100 μ M **1**, 200 nM *M.TaqI*, R.Taq ^{α} I; (6–9) same as 2–5, but with aryl azide cofactor **4a**; (10–13) same as 2–5, but with alkyl azide cofactor **4b**; (14) DNA, R.Taq ^{α} I. (B) Increase in cofactor concentration. (1) DNA; (2) DNA, 200 nM *M.TaqI*, R.Taq ^{α} I; (3) DNA, 10 μ M cofactor **1**, 200 nM *M.TaqI*, R.Taq ^{α} I; (4) DNA, 50 μ M **1**, 200 nM *M.TaqI*, R.Taq ^{α} I; (5) DNA, 100 μ M **1**, 200 nM *M.TaqI*, R.Taq ^{α} I; (6–9) same as 2–5, but with aryl azide cofactor **4a**; (10–13) same as 2–5, but with alkyl azide cofactor **4b**; (14) DNA, R.Taq ^{α} I.

100 μ M coincided with increased protection from R.TaqI as seen in lanes 3–5, 7–9 and 11–13. As indicated by lanes 2, 6 and 10, R.TaqI is not inhibited by the methyltransferase.

Staudinger ligation with enzymatically azidated oligonucleotide

We synthesized the two biotin-linked triarylphosphines (**13** and **15**, Scheme 4) to assess the effectiveness of Staudinger

ligations with cofactor-modified DNA (see Supplementary Material for experimental procedures and corresponding spectral data). Based upon previous results, the *o*-methoxycarbonyl-functionalized triarylphosphine (**13**) was expected to ligate with both the aryl and alkyl azide adducts, as depicted in Scheme 4. Alternatively, ester-linked phosphine **15** was anticipated to couple only with the aryl azide-modified DNA based upon previous findings in our lab



Scheme 4. Biotinylation of azidonucleoside-linked DNA substrates.

(21). These expectations were predicated on the knowledge that aryl azides react with *o*-alkoxycarbonyl triarylphosphines to afford highly stable imidate structures such as **14** or **16**. In marked contrast, alkyl azides react with such phosphines to produce amide-linked materials. Alkyl azide reactions with such phosphines invoke dissociation of the ester-derived alkoxide moiety from the ligated species.

As shown in Figure 2A, a pronounced shift of the DNA mobility in the presence of the three cofactor mimics and M.TaqI (lanes 3, 7 and 12) highlights the covalent modification of interest on the synthetic oligonucleotide. It is important to note that non-specific alkylation is not visible for the three cofactor mimics tested here (lanes 2, 6 and 11). Subjecting of **5a** to biotinylated triaryldiazotized phosphine **13** produced a significant reduction in DNA mobility (lanes 8 and 13) consistent with triaryldiazotized phosphine ligation. Additionally, the biotinylated triaryldiazotized phosphine **15** only ligated to the product derived from **5a**, as seen in lane 9 versus 14. The observation of a new band is evident in lane 14 (reaction of **5b** with **15**) and is probably the product resulting from the reduction of the azide to the primary amine. This is consistent with the fact that **15** does not ligate to alkyl azides, but rather effects their water-dependent reduction (21). Indeed, the product obtained from treatment with **15** migrates identically to that formed by

treatment of **5b** with triphenylphosphine (lane 15). Importantly, no shift of **3** occurs in the presence of either phosphine (lanes 4 and 5) and indicates that the observed shift resulting from the addition of the triaryldiazotized phosphines is, in fact, due to the Staudinger ligation involving the azides of **4a** and **4b**. In the presence of the phosphines tested here, no alteration in the band mobility occurs for either the DNA or the non-specific alkylation controls (data not shown).

The attachment of cofactors **1**, **4a** and **4b** to DNA is not restricted to the M.TaqI system. As shown in Figure 2B, we briefly investigated the ability of the enzyme M.EcoRI to transfer these cofactors to a radiolabeled synthetic oligo duplex bearing the sequence d(GAATTC). As evidenced by investigation of lane 2, M.EcoRI is clearly capable of transferring **1**. As assessed by independent analysis, the slow mobility material observed in lane 2 moves with a different mobility than that observed in the absence of M.EcoRI and the amount of alkylation observed with enzyme is far greater than that observed in the absence of enzyme (data not shown). Perhaps more intriguing is that the mobility pattern of products formed in lanes 3–12 of panel B are very similar to those of lanes 6–15 in panel A. The activity of M.EcoRI with cofactors **4a** and **4b** closely parallels that observed with M.TaqI. Careful analysis of lane 5 (panel B) clearly indicates however that

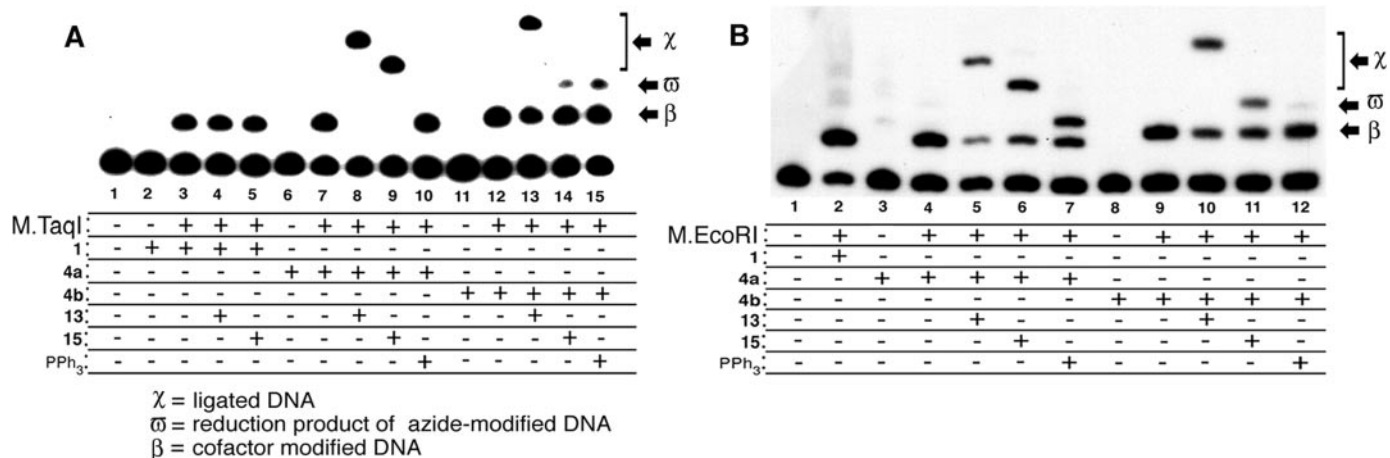


Figure 2. DNA alkylation reaction of synthetic oligonucleotide by aziridine cofactor mimics **1**, **4a** and **4b**, and the Staudinger ligation of resulting alkylation products. (A) Reaction mixtures were prepared by addition of appropriate stock solutions to a total volume of 20 μ l containing 1 μ M DNA buffered with 20 mM Tris-OAc (pH 6.0), 50 mM KOAc, 10 mM Mg(OAc)₂, 0.01% Triton X-100, 100 μ M specified cofactor and 6 μ M M.TaqI. (B) Reaction mixtures were prepared by addition of appropriate stock solutions to a total volume of 20 μ l containing 1 μ M DNA buffered with 10 mM Tris-HCl (pH 7.4), 50 mM NaCl, 0.5 mM EDTA, 0.01% Triton X-100, 100 μ M specified cofactor and 2 μ M M.EcoRI. Samples that were subjected to ligation conditions were brought to a final DNA concentration of 500 nM and contained a 20-fold excess of triarylphosphine. The mixtures were analyzed on a 20% DPAGE ran at 1800 V for 2 h. For the above figure, components that are inclusive in the reaction are denoted by a '+'; components that are exclusive are denoted by a '-'.
χ = ligated DNA
ω = reduction product of azide-modified DNA
β = cofactor modified DNA

Staudinger ligation of the **4a**-modified M.EcoRI substrate does not proceed with the same efficiency observed with the analogous M.TaqI case. Also noteworthy is that treatment of the **4a**-linked M.EcoRI substrate with PPh₃, unlike the case seen in lane 10 of panel A, affords a band with altered mobility relative to the **4a**-linked strand (lane 7, panel B). We believe these subtle differences to result from DNA sequence moderated hydrogen-bonding networks characteristic of M.TaqI and M.EcoRI products. Although slight differences exist between M.TaqI and M.EcoRI-derived reactions and products, it is highly significant that **4a** and **4b** are compatible with multiple DNA methyltransferases.

To verify the structural and functional integrity of biotin following the Staudinger ligation, a pull-down assay utilizing agarose-immobilized streptavidin was performed using products derived from M.TaqI reactions. As shown in Figure 3A, scintillation counting of the 5' end-labeled modified strands revealed a high degree of radioisotopic retention to the streptavidin-agarose in those reactions involving **4a** and either **13** or **15** relative to a DMF control devoid of phosphine. Additionally, Figure 3B illustrates the retention of radioactivity for the ligation product of **5b** with **13**. This observation supports the notion that **5b** did not ligate to **15** (lane 14, Figure 2A), consistent with previous efforts in our labs demonstrating that ester-linked triarylphosphines, such as **15**, form stable ligation products with aryl azides but not alkyl azides.

DISCUSSION

The conversion of biological methyltransferases into azidonucleosidyl transferases represents a significant advancement by virtue of now widely recognized and useful abiotic chemoselective ligation methods. The production of synthetic cofactors similar in structure to SAM but allowing for elaborate post-enzymatic modifications permits site-specific

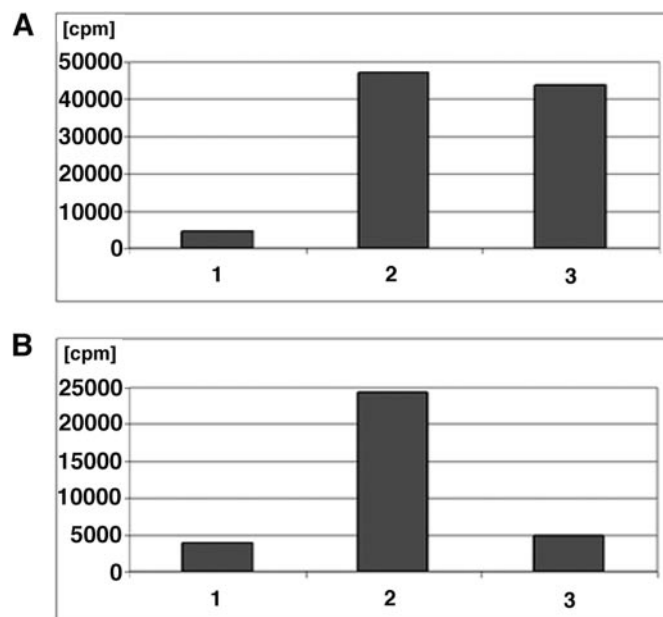


Figure 3. Immobilization of DNA-biotin conjugates on streptavidin-agarose. Reaction mixtures were incubated for 1 h, followed by washing with 1 M NaCl ($\times 3$). The amount of DNA retained was quantitated by scintillation counting. (A) Binding reactions performed on **5a**. Bar 1. **5a** + DMF, Bar 2. **5a** + **13**, Bar 3. **5a** + **15**. (B) Binding reactions performed on **5b**. Bar 1. **5b** + DMF, Bar 2. **5b** + **13**, Bar 3. **5b** + **15**.

modification of biomolecules in a way not previously known. Importantly, MTase-dependent azidation of large biomolecular substrates may be used not only to modify known substrates of methylation, but also to aid in the isolation and identification of currently unknown substrates of methylation, be they nucleic acids or perhaps proteins. Indeed, as attention at the chemistry biology interface continues to intensify on

understanding the roles of DNA and protein methylation, particularly in the realm of transcriptional regulation, synthetic agents capable of intervening in biosynthetic processes will become attractive tools to more readily answer biological questions.

M.TaqI and M.EcoRI both promote DNA azidation upon presentation with an appropriate substrate and either **4a** or **4b**. Qualitative analysis of M.TaqI promoted alkylation of linearized pUC19 can be made by observing the intensity and location of modified and/or unmodified DNA bands in Figure 1. In comparing the efficiency of M.TaqI-dependent DNA alkylation with the three cofactors, it appears that **4a** and **4b** are better cofactors for M.TaqI than is the unsubstituted cofactor **1**. This can be deduced from the almost complete disappearance of the smaller restriction fragments in reactions containing **4a** and **4b**, relative to **1**. Using a small synthetic oligonucleotide, it was possible to assess the activity of M.EcoRI with **1**, **4a** and **4b**. As noted before, M.EcoRI, like M.TaqI, is highly amenable to DNA modification with all three cofactors. These results implicate both **4a** and **4b** as potentially powerful tools by which to modify both known and unknown biological substrates of adenine MTases. Notably, azide-bearing cofactors are not restricted to use by adenine methyltransferases. In addition to M.TaqI and M.EcoRI, we examined two cytosine C5 methylases M.HhaI and M.HpaII. Of the two enzymes, M.HhaI was capable of catalyzing the transfer of aryl azide **4a** and alkyl azide **4b**, albeit with a lower efficiency than M.TaqI. M.HpaII was compatible with the unsubstituted cofactor mimic **1**, but was not able to catalyze the transfer of either azido-based cofactor (unpublished results). From these early efforts, it was clear that the nucleic acid chemistry of azide-bearing cofactors is, in reality, not restricted to adenine methyltransferases. But can azide-bearing DNA substrates undergo ligation chemistries as initially hypothesized?

Initial ligation model studies focused on the ability of cofactor precursors with aryl and alkyl azides to undergo the Staudinger ligation and Huisgen [2 + 3] cycloaddition reactions under biological conditions [(21), unpublished results]. HPLC analysis of an aryl azide derivative indicated fast reaction times and high yields with triarylphosphines bearing the core of **13** and **15**. The alkyl azide derivative also underwent ligation to a terminal alkyne in a similarly facile manner. However, we found the aryl azide to be incompatible with alkyne ligations under a wide array of reaction conditions. Thus, it was concluded that the aryl azide cofactor mimic, **4a**, would be well suited for the Staudinger ligation, whereas the alkyl azide cofactor mimic, **4b**, would be compatible with both the Staudinger ligation (although not as fast as **4a**) and Click chemistry.

Recent applications of Cu(I)-catalyzed Click chemistry emphasize the reliable Huisgen dipolar cycloaddition between an azide and alkyne. This coupling, in which both components are abiotic, has become an important method for coupling subunits together with a high thermodynamic driving force. Applications have ranged from modifications of enzyme active sites to the cell surfaces of *E.coli* (18,19), and continue to evolve on an almost daily basis. Although the methodology has been successful with non-nucleic acid substrates, the Cu(I)-catalyzed cycloaddition in the presence of DNA leads to rapid destruction of nucleic acids via Haber Weiss

chemistry (36,37). As a result, this ordinarily useful methodology was deemed unsuitable for the purposes highlighted here and emphasis was placed on the Staudinger ligation.

Initial efforts to couple biotinylated reagents **13** and **15** with azide-linked pUC19 failed to yield discernible ligation via the streptavidin pull-down assay. Two hypotheses were developed to explain this shortcoming: the first involved the low DNA concentration and the second took into account the sterics imparted by the larger piece of DNA. Thus, our attention was turned to the use of a synthetic oligonucleotide duplex to perform the MTase-dependent alkylations, followed by the Staudinger ligation. Not only did this ultimately allow us to perform the ligations at a higher DNA concentration, but even subtle structural differences between DNA adducts could be readily visualized with denaturing polyacrylamide gel electrophoresis.

The ability of the *o*-methoxycarbonyl functionalized triarylphosphine **13** and the triarylphosphine **15** to undergo the Staudinger ligation on DNA modified with the two azido cofactor mimics has been demonstrated. Based upon previous observations and mechanistic rationale (16,21), the ability of **13** to couple with both the alkyl and aryl azide cases was validated. As expected, the extent of ligation to the alkyl azide-modified DNA was lower than that observed with the aryl azide. This is perhaps best rationalized when one considers the significantly greater electron density (and thus, reduced electrophilic character) of alkyl azides relative to the more electron poor aryl azides. Also noteworthy is that formation of the imidate structures characteristic of purported ligation adducts **14** and **16** (Scheme 4) calls for a significantly abbreviated mechanistic pathway than that involved in the formation of amide structures like **17** (21). Although phosphine **15** is compatible only with aryl azides, its ligation to azide-modified DNA is remarkably efficient. The principal advantage to **15** lies in its ease of preparation. Unlike the phosphine developed by Bertozzi, which requires multiple synthetic manipulations and tedious recrystallizations to purify intermediates (14), **15** can be readily produced from a commercially available triphenylphosphinic acid via attachment to one's linker of choice through simple esterification procedures (21) [see Supplementary Material for experimental procedures and corresponding spectral data].

Having validated the hypothesis that enzymatically azidated DNA was compatible with the Staudinger ligation, it was essential to demonstrate the utility of this methodology in identifying and isolating biological molecules. The high binding affinity between avidin and biotin prompted us to choose this affinity matrix handle to demonstrate this methodology (38). One could easily exploit the avidin-biotin interaction by performing an electrophoretic mobility shift (gel shift) assay or by one of several technologies based on the strength of this interaction. We opted to use an immobilized streptavidin that would allow us to wash away unbound molecules, leaving the biotinylated product bound to the agarose (24). Selective 5' end-labeling of the DNA duplex allowed the application of liquid scintillation counting for detection of avidin-bound materials. The results obtained reveal the cofactor, methylase and ligation-dependent retention of radiolabeled DNA to the immobilized agarose relative to the DMF control. Data obtained from ethanol precipitations must be considered relative to necessary controls as the precipitations are not

quantitative and ~20% is lost. This loss is, however, consistent for all samples regardless of conditions to which they have been subjected.

The data presented here not only validates the integrity of the azido-based cofactor mimics following enzymatic transfer to substrate DNA, but also verifies that affinity-tagged triarylphosphines undergo ligation to azide-linked DNA under biological conditions. The conversion of biological methylases into azidonucleosidyl transferases through cofactor mimicry has thus been established. Although demonstrated for nucleic acids here, we project that azide-bearing cofactors hold tremendous promise as new tools for those at the chemical biology interface.

SUPPLEMENTARY MATERIAL

Supplementary Material is available at NAR Online. Contained in these materials are preparation and characterization details for substances 7–12, 4b, 13 and 15.

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Conflict of interest statement. None declared.

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