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Forwards and backwards — synthesis of *Laurencia* natural products using a biomimetic and retrobiomimetic strategy incorporating structural reassignment of laurefurenynes C-F†

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Laurefurenynes C–F are four natural products isolated from *Laurencia* species whose structures were originally determined on the basis of extensive nuclear magnetic resonance experiments. On the basis of a proposed biogenesis, involving a tricyclic oxonium ion as a key intermediate, we have reassigned the structures of these four natural products and synthesized the four reassigned structures using a biomimetic approach demonstrating that they are the actual structures of the natural products. In addition, we have developed a synthesis of the enantiomers of the natural products laurencin and deacetyllaurencin from the enantiomer of (*E*)-laurefucin using an unusual retrobiomimetic strategy. All of these syntheses have been enabled by the use of tricyclic oxonium ions as pivotal synthetic intermediates.

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Introduction

The C₁₅-acetogenin natural products isolated from Laurencia species display both wide structural diversity and wide ranging biological activity which have piqued the interest of synthetic organic chemists since the initial isolation of (+)-laurencin in 1965.¹⁻³ These natural products generally contain cyclic ethers, and frequently medium-ring ethers, and have served as a testbed for the development of numerous new and efficient methodologies to allow access to complex cyclic ethers.4-7 The structural assignment of many of these acetogenic natural products has come primarily from high-field NMR experiments with single crystal X-ray diffraction being used where possible. The unambiguous assignment of natural products structures with compounds not suitable for single crystal X-ray analysis8 is a challenging task and it is inevitable that incorrect structures are reported in the literature with these structural misassignments frequently being uncovered through total synthesis.9-11 Within the C₁₅ acetogenic natural products isolated from Laurencia species a number of structural misassignments have also occurred that have varied from gross structural reassignments, 12-15 through to atom transpositions 16-19 and stereochemical reassignment.20-22 Key to correcting many of these

structural misassignments have been postulates regarding the biosyntheses of acetogenic Laurencia natural products, many of which have been proposed to proceed through complex oxonium ion intermediates. 15,23-33 These biogenetic postulates have allowed rational prediction of the likely structures of a number of the natural products15,21,22,26,34 and in a number of cases the biogenetic arguments have been augmented by DFT calculations of both proton and carbon NMR chemical shifts14,22,34 for a range of candidate structures with ultimately the structure of the natural products being established through total synthesis.35 Among the acetogenic Laurencia natural products whose structures have been reassigned are laurefucin, 36,37 obtusallenes V, VI and VII, 16,19,26 elatenyne, 12-15,38 laurendecumenyne B,15,39,40 aplysiallene,41,42 a chloroenyne from Laurencia majuscula^{13,34,38,43} and laurefurenynes A and B,²⁰⁻²² additionally the structures of the ocellenynes have been tentatively reassigned on the basis of biogenetic considerations and DFT calculations of NMR chemical shifts. 24,44,45

The laurefurenynes are a series of six acetogenic natural products which were isolated and characterized by Jaspars and co-workers from a sample of *Laurencia* sp. collected in the Philippines in 1991.²⁰ The structures of these six natural products ((Z/E)-1, (Z/E)-2 and (Z/E)-3) were assigned through extensive 1D and 2D NMR experiments (Fig. 1).⁴⁶ In 2013 the Britton group²¹ and our own group²² published structural reassignments of laurefurenynes A and B from (Z/E)-1 to (Z/E)-4 based on total synthesis and DFT calculations of NMR chemical shifts. In the original paper of Jaspars, a plausible biogenesis of the laurefurenynes (((Z/E)-1, (Z/E)-2 and (Z/E)-3) was proposed based on epoxide and/or bromonium ion cyclizations. However, the

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Fig. 1 Originally proposed structures of laurefurenynes A–F along with reassigned structures of laurefurenynes A and B.

reassigned structures of laurefurenynes A and B (Z/E)-4 fit with a proposed biosynthesis involving the previously characterized tricyclic oxonium ions 7 ⁴⁷ which likely arise *via* transannular displacement of bromide or chloride from the halofucins 5 and 6 (Scheme 1), ^{15,48,49} followed by opening of the oxonium ions 7 at C-7 with water and then displacement of the C-12 bromine atom with water with inversion of configuration. ^{21,22} If it is indeed the case that laurefurenynes A and B (Z/E)-4 are biosynthesized from the halofucins 5 and 6 then it would be reasonable to postulate that laurefurenynes C-F are produced on the same

HO, HOOH

Me JOH

OH

OH

OH

OH

OH

OH

OH

OH

H2O, C-12

Me JOH

OH

Me JOH

C-7

Br

H2O, C-7

Br

H2O, C-7

Br

H2O, C-7

Transannular halide loss

H2O, C-7

Fig. (E/Z)-bromofucins (X = Br)

6; (E/Z)-chlorofucins (X = Cl)

HOOH

HOH

HOH

HOH

HOH

We

(Z/E)-10; reassigned structures of laurefurenyne E and laurefurenyne E and laurefurenyne F

Scheme 1 Proposed biosynthesis of the reassigned structures of laurefurenynes A–F from the halofucins.

biosynthetic pathway.²¹ Here, opening of the oxonium ions 7 with water, or a water equivalent, ^{26,34} at C-10 would produce the reassigned structures of laurefurenynes E and F (Z/E)-9 ⁵⁰ and displacement of the bromine substituent with inversion of configuration by water or a water equivalent would yield the reassigned structures of laurefurenynes C and D (Z/E)-10.

Based on the above analysis, we report the first synthesis and the first biomimetic synthesis of the reassigned structures of laurefurenynes C-F (Z/E)-10 and (Z/E)-9. Confirmation that the structures of the actual natural products are represented by (Z/E)-10 and (Z/E)-9 comes through comparative ¹H and ¹³C nuclear magnetic resonance data for our synthetic material with that of the natural material, 20 Mosher's ester analysis to establish the absolute configuration of a newly installed secondary alcohol, 51,52 single crystal X-ray diffraction data giving the structure of the reassigned natural product laurefurenyne D (E)-10 and the reassigned dioxabicyclo [5.2.1] decane cores 12 of laurefurenynes C-F.53 In combination with the reported reassigned structures of laurefurenynes A and B (Z/E)-4 by Britton and our own group, 21,22 reassignment of the structures of the remaining laurefurenynes (Z/E)-9 and (Z/E)-10 offers chemical evidence to prompt reevaluation of the initially proposed biosynthesis of these natural products.20 In addition, the development of methods necessary for the total synthesis of the reassigned structures of laurefurenynes C-F (Z/E)-9 and (Z/E)-10 enabled the total synthesis of the enantiomers of the natural products laurencin1-3 and deacetyllaurencin54,55 namely entlaurencin and ent-deacetyllaurencin using an unusual retrobiomimetic approach.56

Results and discussion

The synthesis of the reassigned structures of laurefurenynes C–F (Z/E)-10 and (Z/E)-9 began with studies concerning the expansion of the scope of our previously reported bromocyclization–nucleophilic quenching reaction of known enantiopure bromomesylates 11 ⁵⁷ to include oxygen nucleophiles (Fig. 2). In our previous report, only chloride and bromide had been used as nucleophiles in this transformation and extra steps were required for the introduction of oxygen substituents at the desired C-10 position.⁴⁷ The expansion of the scope of nucleophiles would allow a direct bromocyclization–oxygenation

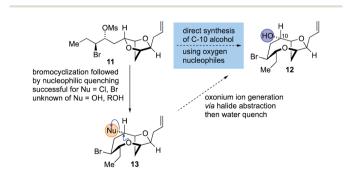
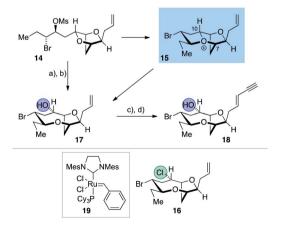


Fig. 2 Proposed strategy for the synthesis of **12** by the direct use of oxygen nucleophiles in the bromocyclization–nucleophilic quenching reaction.

reaction of 11 to provide the dioxabicyclo[5.2.1]decane core of the reassigned structures of laurefurenvnes E-F (12) more efficiently, obviating the need for the intermediacy of 13 (Nu = Cl, Br) as a synthetic intermediate. We conducted the initial studies on expanding the scope of the bromocyclization/nucleophilic quenching reaction with the known enantiopure bromomesylate 14 57 instead of 11, as the reaction with 14 was consistently found to give exclusive C-10 opening products with halide nucleophiles, 47 therefore simplifying reaction analysis. Thus, exposure of the bromomesylate 14 to our previously described conditions $(AgAl(pftb)_4 \cdot CH_2Cl_2 (pftb = perfluoro-t-butoxy))$ and TiCl₄) at low temperature which likely generates the oxonium ion 15, followed by addition of saturated aqueous NaHCO3 solution, unexpectedly gave what appeared to be the previously reported chloride **16** as the major product by TLC analysis *versus* an authentic sample (Scheme 2).47 However, using excess AgOBz as the oxygen nucleophile in the place of saturated aqueous NaHCO₃, gave the C-10 alcohol 17 in 74% overall yield after methanolysis of the crude benzoate ester; 17 has previously been reported by both Kim and Snyder. 58-60 The formation of 17 with AgOBz rather than aqueous NaHCO3 solution is most likely due to the chloride anion scavenging ability of Ag⁺, rendering chloride anions unable to compete with benzoate anions for nucleophilic attack at the oxonium ion intermediate 15.61 Following Kim's two step procedure⁵⁸ compound 17 was readily transformed into ent-(E)-laurefucin 18.36,37,58,59,62,63 Hence, crossmetathesis of 17 with crotonaldehyde using catalyst 19 and copper(1) iodide,64 followed by the Colvin-Ohira reaction using trimethylsilyldiazomethane and lithium diisopropylamide installed the (E)-enyne in 41% overall yield giving ent-(E)-laurefucin 18 that had identical ¹H NMR data to a sample we had prepared previously. The matching spectroscopic data for our synthetic 17 and 18 with material prepared by Kim and Snyder confirmed that the C-10 alcohol had the configuration shown.⁶⁵



Scheme 2 Synthesis of 17 by the direct use of oxygen nucleophile (AgOBz) and its conversion to ent-(E)-laurefucin 18. Reagents and conditions: (a) AgAl(pftb) $_4$ ·CH $_2$ Cl $_2$, TiCl $_4$, CH $_2$ Cl $_2$, -40°C, 2 h, then AgOBz, -78°C, 1 h, then work-up with sat. NaHCO $_3$ and excess Bu $_4$ N $^+$ I $^-$; (b) K $_2$ CO $_3$, MeOH, 74% (2 steps); (c) crotonaldehyde, cat. 19, cat. Cul, Et $_2$ O, RT, overnight; (d) TMSCHN $_2$, lithium diisopropylamide, -78°C, 1 h, then 0 °C, 1 h, 18 41% E: Z > 20 : 1 (two steps).

The use of AgOBz as the oxygen nucleophile was then applied to the bromocyclization-nucleophilic quenching reaction with known enantiopure bromomesylate 11 (Scheme 3). The desired C-10 oxygenated compound 12 was obtained in 63% overall yield after methanolysis of the crude benzoate ester, without any product from C-7 oxygenation being observed. The structure of alcohol 12 was confirmed by single crystal X-ray diffraction studies⁵³ (Flack x absolute structure parameter =-0.034(5); shown in Scheme 3).66 Alcohol 12 was then transformed into the reassigned structures of laurefurenyne E and F (Z)-9 and (E)-9 using known procedures for the stereoselective synthesis of (Z)- and (E)-enynes. Thus, oxidative cleavage of the terminal olefin in 12,67 followed by Stork-Zhao olefination gave the corresponding (Z)-vinyl iodide,68 followed by Sonogashira coupling with trimethylsilylacetylene and TMS deprotection gave (Z)-9. In a related manner, the reassigned structure of laurefurenyne F (E)-9 was synthesized from 12 by a crossmetathesis, Colvin-Ohira sequence.58 We were unable to obtain X-ray crystal structures of (E)- and (Z)-9, however, (E)- and (Z)-9 were also synthesized from (E)- and (Z)-bromofucin (E)and (Z)-5 via the previously characterized oxonium ions 7 47 suggesting that the synthetic route from 12 to 9 had not resulted in undesired alterations of stereocenters. Hence, the oxonium ions (Z)- and (E)-7, readily formed from the corresponding bromofucins on treatment with a silver(1) salt followed by low temperature NMR analysis, were quenched with water to give (Z)- and (E)-9 respectively.

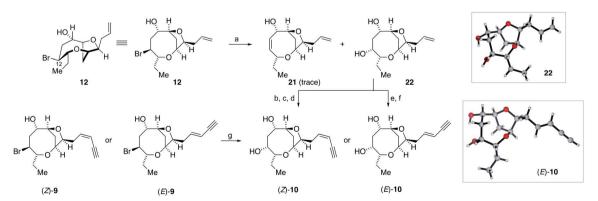
The stereochemical assignment at C-10 of laurefurenynes E and F 9 was further supported by Mosher ester analysis of the C-10 secondary alcohol (blue circle) (see ESI†),51,52 providing further evidence that the absolute configuration of C-10 is (S), in keeping with the X-ray crystal structure of 12. The spectroscopic data of (E)-9 were in accordance with the data for the natural product laurefurenyne F ($[\alpha]_D^{25} = +17.5$ (c = 0.12, MeOH)lit.²⁰ $[\alpha]_{\rm D}^{25} = +17.0 \ (c = 0.10, \text{ MeOH})) \text{ for } (Z)$ -9, the NMR spectroscopic data are in accordance with the published data for the natural product laurefurenyne E; however, the specific rotation of natural laurefurenyne E was reported to be $\left[\alpha\right]_{D}^{25} = +11.0$ (c = 0.10, MeOH) whereas that of our synthetic material was $[\alpha]_D^{25} = -5.9$ (c = 0.12, MeOH) for a sample that has a Z : E ratio of > 15:1. Measurement of the specific rotation of a sample with a lower Z: E ratio of 7.8: 1 gave $[\alpha]_D^{25} = -4.7$ (c = 0.17, MeOH) which is still of a lower magnitude and of opposite sign compared to that reported in the isolation paper.²⁰

The successful generation of laurefurenynes E and F 9 from the bromofucins 5 provides chemical evidence to support their potential biosynthesis via the oxonium ions 7 as with other related natural products (Scheme 1).⁴⁷ As noted above laurefurenynes C and D may well therefore arise biosynthetically from laurefurenynes E and F via an S_N2 reaction at C-12 with an oxygen nucleophile. However, this transformation was expected to be challenging due to the neopentyl-like nature of C-12 coupled with the β -oxygen substituent which is known to retard the rates of nucleophilic substitution.⁶⁹ Under a range of conditions reported for substitution at neopentyl-like centres with oxygen nucleophiles,⁷⁰⁻⁷² substrate 12 was either unreactive, underwent low conversion or yielded the elimination

Scheme 3 Synthesis of the reassigned structures of laurefurenynes E–F 9 via 12. (a) TiCl₄, AgAl(pftb)₄·CH₂Cl₂, CH₂Cl₂, -40 °C, 2 h then add AgOBz, -78 °C, 1 h, then work-up with sat. NaHCO₃ and excess Bu₄NI; (b) K₂CO₃, MeOH, 63% (2 steps); (c) OsO₄, NalO₄, 2,6-lutidine, dioxane, water; (d) ICH₂PPh₃I, NaHMDS, HMPA, THF, -78 °C to r.t.; (e) Me₃SiC \equiv CH, Pd(PPh₃)₄, CuI, Et₃N, then K₂CO₃, MeOH, (Z)-9 35% Z: E > 10 : 1 (four steps); (f) crotonaldehyde, cat. 19, cat. CuI, Et₂O, r.t., overnight; (g) TMSCHN₂, lithium diisopropylamide, -78 °C, 1 h, then 0 °C, 1 h; (h) crude treated with K₂CO₃, MeOH, (E)-9 29% E: Z > 20 : 1 (three steps); (i) AgAl(pftb)₄·CH₂Cl₂, CD₂Cl₂, -40 °C, 1.5 h, then filter at -78 °C and NMR analysis of 7, then retrieve NMR sample, excess H₂O followed by NaHCO₃(s), -78 °C, 15 min then warm to RT (Z)-9 39%; (E)-9 47%.

product 21 (Scheme 4, see ESI† for details). Ultimately, using Fleet's method (cesium trifluoroacetate in DMF at 120 °C for 12 hours)73 gave the desired substitution product 22 in 62% yield along with a trace of elimination product 21. Previously, Fujiwara and Murai have reported that bromination of alcohols with a β-oxygen that is part of a medium ring, can occur with inversion or retention of configuration depending on reaction conditions with retention of configuration being ascribed to neighboring group participation by the medium ring oxygen atom.74 We therefore needed to confirm the relative configuration of 22. Pleasingly, 22 was isolated as a white crystalline solid which allowed single crystal X-ray diffraction analysis (absolute structure parameter of 22: -0.01(9))^{53,66} which demonstrated that the C-12 alcohol in 22 was formed with inversion of configuration and confirmed that the relative configuration was as expected (Scheme 4). Analogous to the synthesis of the reassigned structures of laurefurenynes E and F 9, two separate protocols were used for the synthesis of laurefurenynes C and D 10, with the first synthesis starting from diol 22. Thus, diol 22 was readily converted into laurefurenyne C by modified

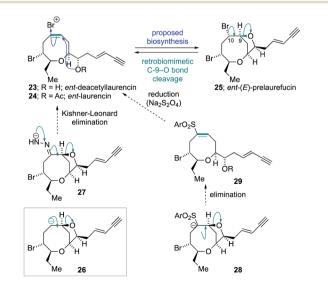
Lemieux-Johnson oxidation67 of the terminal alkene in 22 followed by Stork-Zhao olefination and Sonogashira coupling followed by deprotection. This provided the (Z)-enyne (Z)-10 in 31% yield over the 4 steps with >10:1 Z:E selectivity. Alternatively, the (E)-enyne 10 was readily formed by Kim's procedure⁵⁸ involving cross metathesis of 22 with crotonaldehyde followed by Colvin-Ohira reaction giving laurefurenyne D (E)-10. Laurefurenynes C and D 10 were also synthesized from laurefurenynes E and F 9 by displacement of the secondary bromides with cesium trifluoroacetate (Scheme 4).73 The NMR spectroscopic data for the reassigned structures of laurefurenvnes C and D 10 was identical with that of the natural products. Furthermore, the optical rotations of laurefurenyne C (Z)-**10** ([α]_D²⁵ = + 10.8 (c = 0.13, MeOH), lit.²⁰ [α]_D²⁵ = +20.0 (c = 0.10, MeOH)) and laurefurenyne D (E)-**10** ([α]_D²⁵ = +14.0 (c = 0.10, MeOH), lit.²⁰ $[\alpha]_D^{25} = +32.0$ (c = 0.10, MeOH)) were of the same sign but of reduced magnitude compared to those of the isolated natural products indicating that the absolute configuration of the natural products is likely as shown in Schemes 1 and 4 in keeping with the proposed biosynthesis (Scheme 1).



Scheme 4 Synthesis of the reassigned structures of laurefurenynes C-D 10 via 12. (a) CF₃CO₂Cs, Me₂NCHO, 120 °C, 12 h, 22 62%, 21, trace; (b) OsO₄, NalO₄, 2,6-lutidine, dioxane, water; (c) ICH₂PPh₃I, NaHMDS, HMPA, THF, -78 °C to RT; (d) Me₃SiC \equiv CH, Pd(PPh₃)₄, CuI, Et₃N, then K₂CO₃, MeOH, (Z)-10 32% Z: E > 10 : 1 (four steps); (e) CH₃CH \equiv CHCHO, cat. 19, cat. CuI, Et₂O, RT; (f) TMSCHN₂, nBuLi, -78 °C, 1 h, then 0 °C, 30 min, then 1.0 M HCl (E)-10 81% E: Z > 20 : 1 (two steps); (g) CF₃CO₂Cs, Me₂NCHO, 120 °C, 12 h, (Z)-10 23%, (E)-10 38%.

Synthetic laurefurenyne D (E)-10 was initially isolated as a colorless oil. However, a single crystal suitable for single crystal X-ray diffraction analysis was obtained by slow diffusion of pentane into an ethyl acetate solution of (E)-10. The crystal structure of (E)-10 (shown in Scheme 4) was obtained with an absolute structure parameter of -0.07(9), providing further confirmation of its structure.⁵³

The success of the synthesis of *ent-(E)*-laurefucin **18**, and the reassigned structures of the laurefurenynes C-F 9 and 10, that were enabled by the direct C-10 oxygenation of bromomesylates 11 and 14 in the bromocyclization-nucleophilic quenching reaction, prompted us to further investigate the scope of this reaction. We proposed that the enantiomer of the Δ^4 -oxocene natural products laurencin and deacetyllaurencin,75-78 namely ent-laurencin 24 and ent-deacetyllaurencin 23 could be synthesized via an unusual retrobiomimetic C9-O bond cleavage from ent-(E)-prelaurefucin 25 (Scheme 5). This proposed C9-O bond cleavage requires the formation of a formal anion at C-10 (26) with subsequent E1cB elimination. Reduction of ent-(E)-prelaurefucin 25 with Zn was unlikely to be selective for formation of ent-deacetyllaurencin given the previous precedent in a related system^{37,79} and we therefore elected to investigate two elimination reactions for the formation of 23 and 24, namely a Kishner-Leonard elimination (27)80,81 and elimination to form a vinyl sulfone (28 to 29) which could undergo subsequent selective reduction with the mild reducing agent sodium dithionite.82 In order to put this plan into practice we needed to further expand the scope of the bromocyclization-nucleophilic quenching reaction to include sulfur nucleophiles for the synthesis of sulfones. The requisite sulfone 30 was readily prepared from enantiopure bromomesylate 14. Oxonium ion formation from 14 was achieved as before followed by the addition of 1-phenyl-1H-tetrazole-5-thiol 31 as the sulfur nucleophile to give corresponding C-10 sulfide in 76% yield.83,84 The sulfide was then oxidized to the desired sulfone 30 in 77%



Scheme 5 Proposed routes to *ent*-laurencin by a retrobiomimetic strategy.

Scheme 6 Synthesis of sulfone 30. Reagents and conditions: (a) TiCl₄, AgAl(pftb)₄·CH₂Cl₂, CH₂Cl₂, -40 °C, 2 h; (b) then add 31, -78 °C, 1 h; (c) 30% aq. H₂O₂, cat. Mo₇O₂₄(NH₄)₆·4H₂O, EtOH, 0 °C to RT overnight, 59% from 14.

yield by cat. $Mo_7O_{24}(NH_4)_6 \cdot 4H_2O$ and 30% aqueous H_2O_2 in 77% yield (Scheme 6).

Treatment of 30 with 1.1 equivalents of strong base (NaHMDS) at -78 °C resulted in no conversion of starting material but on warming to ambient temperature, complete conversion of the sulfone was observed giving the two cyclopropane-containing products 32 and 33 in a 1:2.2 ratio in 90% combined yield (Scheme 7). The structures of the sulfones were assigned by extensive NMR experiments including ¹H-¹H NOE experiments (see ESI†). A plausible mechanism for the formation of 32 and 33 is shown below. Thus, deprotonation the acidic proton adjacent to the sulfone in 30 would give the anion 34 from which direct cyclopropanation can occur giving 32 or E1cB elimination can occur giving alkoxide 35. Alkoxide 35 can then undergo intramolecular proton transfer to provide the αsulfonyl anion 36 from which cyclopropanation can readily occur giving 33. The generation of product 33 indicated that the desired E1cB fragmentation was possible although the anion stabilizing properties of the sulfone promoted further reactions.

Our attention therefore turned to investigate the Kishner-Leonard elimination from the C-10 hydrazone **38** (Scheme 8). Oxidation of the previously synthesized alcohol **17** using Dess-Martin periodinane gave ketone **37**, which on treatment with hydrazine hydrate gave the requisite hydrazone **38** (not isolated). Following a literature procedure, ⁸⁵ treatment of the crude hydrazone **38** with dimsyl sodium on a small scale (0.7 mmol NaH – caution – sodium hydride in DMSO is a known explosion hazard) ⁸⁶ gave an inseparable mixture of the cyclopropane **39**, as

Scheme 7 Formation of cyclopropanes 32 and 33 from 30, and a proposed mechanism for product generation. Reagents and conditions: (a) 1.1 eq. NaHMDS, THF, -78 °C to RT, 15 min 32 : 33 = 1 : 2.2 90%.

Scheme 8 Kishner–Leonard fragmentation for Δ^4 -oxocene synthesis and the synthesis of *ent*-deacetyllaurencin 23 and *ent*-laurencin 24. Reagents and conditions: (a) Dess–Martin periodinane, CH₂Cl₂, RT, 1 h, 37 85%; 42 quant.; (b) N₂H₄·H₂O, EtOH, RT 2 h; (c) 3.5 eq. NaH, DMSO, then add crude 38 RT, 60% 4.5 : 1 mixture of 39 and 40 (from 37); (d) 1.1 eq. NaHMDS, THF, -78 °C to RT, 41 42% and 40 trace (two steps); (e) 2.0 eq. NaHMDS, THF, -78 °C to RT, 23 32% and 43 11% (two steps); (f) Ac₂O, DMAP, CH₂Cl₂, r.t., 1 h, 91%.

the major product, along with the diene 40 (39:40, 4.5:1); the structures of 39 and 40 were assigned by extensive ¹H and ¹³C NMR experiments (Scheme 8). Both the cyclopropane 39 and the diene 40 most likely arise from the hydrazone anion corresponding to 27 with formation of the diene requiring a further elimination of HBr. We were disappointed that the cyclopropane 39 was the major product of this reaction. However, when the crude hydrazone 38 was treated with 1.1 equivalents of NaHMDS in THF at -78 °C with gradual warming to room temperature, during which gas evolution was observed, the desired Δ^4 -oxocene 41 was isolated along with a trace of the 1,3-diene 40 as a side product (42% combined yield over 2 steps). This procedure was then applied to the natural product ent-(E)-laurefucin 18 for the synthesis of ent-laurencin 24. Thus, oxidation of alcohol 18 gave the ketone 42 from which the corresponding hydrazone was prepared. In order to effect the Kishner-Leonard elimination on the hydrazone derived from ketone 42 it was necessary to increase in the equivalents of NaHMDS from 1.1 to 2.0 equivalents presumably due to the deprotonation of the acidic acetylene C-H. Using this procedure gave ent-deacetyllaurencin 23 and diene 43 in a 3:1 ratio in 43% combined yield over 2 steps. The natural product ent-laurencin 24 was obtained by acetylation of ent-deacetyllaurencin 23 in 93% yield. The spectroscopic data of 23 and 24 are in accordance with published data^{2,55} (synthetic entlaurencin: $\left[\alpha\right]_{\mathrm{D}}^{25}=-61.0$ (c=0.10, CHCl₃), lit.² natural (+)-laurencin $\left[\alpha\right]_{\mathrm{D}}^{23} = +70.2$ (c = 1.0, CHCl₃); synthetic *ent*-deacetyllaurencin $\left[\alpha\right]_{D}^{25} = -34.7$ (c = 0.05, CHCl₃), lit.⁵⁵ natural deacetyllaurencin $\left[\alpha\right]_{D}^{17} = +46.1 \ (c = 1.15, \text{CHCl}_{3})$).

Conclusions

In conclusion, we have reported structural reassignment of laurefurenynes C-F 9-10 confirmed through the first synthesis and the first biomimetic synthesis of the natural products based on their proposed biosynthesis. Confirmation of the reassigned natural product structures came from extensive spectroscopic

and crystallographic evidence. The structural reassignment and biomimetic generation of the laurefurenynes enables them to be considered in line with a series of structurally analogous Laurencia natural products, which further lend weight to their proposed biosynthesis from the bromofucins via complex oxonium ions such as 7. In addition, the expansion of the scope of nucleophiles compatible with the bromocyclization-nucleophilic quenching reactions enabled the preparation of a series of substrates for the investigations that led to the total synthesis of ent-deacetyllaurencin 23 and ent-laurencin 24 through a retrobiomimetic C9-O cleavage, which was ultimately achieved using the Kishner-Leonard elimination reaction. The use of a broad two directional biomimetic and retrobiomimetic strategy has facilitated the synthesis of six natural products. Further development of oxonium ion chemistry in general with applications in natural product synthesis are ongoing and will be reported in due course.

Conflicts of interest

There are no conflicts to declare.

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