# Genome mining combined metabolic shunting and OSMAC strategy of an endophytic fungus leads to the production of diverse natural products 

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#### Abstract

Endophytic fungi are promising producers of bioactive small molecules. Bioinformatic analysis of the genome of an endophytic fungus Penicillium dangeardii revealed 43 biosynthetic gene clusters, exhibited its strong ability to produce numbers of secondary metabolites. However, this strain mainly produce rubratoxins alone with high yield in varied culture conditions, suggested most gene clusters are silent. Efforts for mining the cryptic gene clusters in P. dangeardii, including epigenetic regulation and one-strain-many-compounds (OSMAC) approach were failed probably due to the high yield of rubratoxins. A metabolic shunting strategy by deleting the key gene for rubratoxins biosynthesis combining with optimization of culture condition successfully activated multiple silent genes encoding for other polyketide synthases (PKSs), and led to the trace compounds detectable. As a result, a total of 23 new compounds including azaphilone monomers, dimers, trimers with unprecedented polycyclic bridged heterocycle and spiral structures, as well as siderophores were identified. Some compounds showed significant cytotoxicities, anti-inflammatory or antioxidant activities. The attractive dual PKSs gene clusters for azaphilones biosynthesis were mined by bioinformatic analysis and overexpression of a pathway specific transcription factor. Our work therefor provides an efficient approach to mine the chemical diversity of endophytic fungi.


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## 1. Introduction

Endophytic fungi, which live inside plants interacting with the host and promoting plant growth, have been proved to be promising producer of secondary metabolites with novel structures and a remarkable range of activities such as antibacterial, insecticidal, anticancer, plant hormone and antioxidant ${ }^{1,2}$. For instance, Taxomyces andreanae, isolated from the phloem of Taxus brevifolia, could produce taxol as an anti-tumor agent in breast and ovarian cancers ${ }^{3}$. However, advances in genome sequencing and bioinformatics revealed that fungi with many biosynthetic gene clusters have far greater ability to encode secondary metabolites than which were produced under standard laboratory culture conditions. So various genome mining strategies have been applied to activate silent gene clusters for the biosynthesis of natural products $^{4,5}$, including one-strain-many-compounds (OSMAC) ${ }^{6,7}$, cocultivation ${ }^{8}$, chemical epigenetics ${ }^{9}$, ribosome engineering ${ }^{10}$, epigenetic modification ${ }^{11,12}$, overexpression of transcription factor ${ }^{13}$ and heterologous expression ${ }^{14}$, and achieved amazing results.

Penicillium dangeardii is a filamentous fungus isolated from a toxic plant Lysidice rhodostegia (Leguminosae sp.) ${ }^{15}$. Investigation on its chemical compositions mainly led to the isolation of large anhydrides such as rubratoxins $\mathrm{A}-\mathrm{C}(\mathbf{1}-\mathbf{3} \text {, Fig. 1A })^{16,17}$ and penicillactones $A-C^{15}$. Due to the highly oxidized complex structure and bioactivity, the biosynthesis pathway of rubratoxins A and B has been elucidated ${ }^{18}$. However, analysis of genomic information of $P$. dangeardii revealed 43 gene clusters responsible
for natural products biosynthesis, which are significantly more than the number of above known anhydrides metabolites, suggesting that most gene clusters are cryptic.

To activate the cryptic gene clusters in $P$. dangeardii, OSMAC strategy and epigenetic regulation ${ }^{11}$ were firstly applied, respectively. Unfortunately, it failed probably due to the super-high yield of rubratoxins ( $0.65 \mathrm{~g} / \mathrm{L}$ ). Alternatively, a metabolic shunting strategy based on deletion of the key gene for rubratoxins biosynthesis and optimization of culture condition successfully activated multiple cryptic gene clusters involved in the biosynthesis of polyketides, which resulted in the isolation of a large number of new metabolites, including azaphilone monomers, dimers, and trimers featuring unprecedented polycyclic bridged heterocycle and spiral structures, as well as siderophores. All isolated compounds were evaluated for their cytotoxicities, antiinflammatory and antioxidant activities, and the gene clusters for azaphilones biosynthesis were characterized.

## 2. Results and discussion

### 2.1. Bioinformatic analysis of genome of P. dangeardii

Bioinformatic analysis of the genome sequence from $P$. dangeardii showed 43 gene clusters that encode biosynthesis of natural products, including 22 polyketide synthases (PKSs), 8 nonribosomal peptide synthases (NRPSs), 8 PKSs/NRPSs hybrids, and 5 terpene synthases (TSs; Supporting Information Tables


Figure 1 Analyses of metabolites from P. dangeardii. (A) The structures of anhydrides. (B) LC-MS analyses of crude extracts from wild type (WT) and $\Delta r b t J$ mutant strain fermented in PDB and rice culture media. The number on the peaks corresponds to the products shown in Figs. 1A and 2 . The anhydrides are partially present as diacids (labeled $1^{\prime}$ and $2^{\prime}$ ) in aqueous solvent.



20 $R_{1}=R_{2}=\mathrm{CH}_{3}, Y=\mathrm{O}$
21: $R_{1}=\mathrm{CH}_{3}, R_{2}=\mathrm{CH}_{2} \mathrm{OH}, Y=0$
22: $R_{1}=R_{2}=\mathrm{CH}_{2} \mathrm{OH}, Y=\mathrm{O}$ 22: $R_{1}=R_{2}=\mathrm{CH}_{2} \mathrm{OH}, \mathrm{Y}=\mathrm{O}$ 24: $\mathrm{R}_{1}=\mathrm{R}_{2}=\mathrm{CH}_{2} \mathrm{OH}, \mathrm{Y}=\mathrm{NH}$



25: $\mathrm{R}=\mathrm{CH}_{2} \mathrm{OH}$ 26: $\mathrm{R}=\mathrm{COOH}$


28: $\mathrm{R}=\mathrm{COOH}$



36


37


38


34


35


Figure 2 Compounds purified from P. dangeardii $\Delta r b t J$ mutant strain.

S1-S3), which far exceeds the number of the predominant anhydrides rubratoxins A-C (Fig. 1A). This indicated that the title strain possesses potential to produce more diverse natural compounds.

### 2.2. OSMAC strategy and epigenetic modification applied on $P$. dangeardii

To develop its potential for biosynthesis of pleiotropic second metabolites, an OSMAC strategy was applied, where $P$. dangeardii was cultured in different media, including potato dextrose agar (PDA), potato dextrose broth (PDB), Czapek concentrate yeast extract (CY), yeast extract glucose (YG), yeast extract malt extract glucose (YMEG), and rice (Secition 4.1. Strains and cultivation conditions). However, no additional metabolites except rubratoxins were produced, indicated by LC-MS analysis (Supporting Information Fig. S1). Moreover, deletion of the four epigenetic regulation genes separately did not lead to the generation of any new products (data not show). With the high yield of rubratoxins in wild type or mutant strains, we believe that acetyl CoA, the starting unit for polyketide synthase was largely consumed by rubratoxins biosynthesis, which inhibits the biosynthesis of other polyketides and makes efforts to activate
cryptic gene cluster unsuccessful. In addition, the presence of high yield of rubratoxins in P. dangeardii also leads to the trace secondary metabolites undetectable. Therefore, a strategy based on inhibition of rubratoxins production was believed to be efficient for mining the chemical diversity in P. dangeardii.

### 2.3. Combinational genome mining leads to the production of abundant metabolites

To inhibit the production of rubratoxins, the gene rbtJ encoding PKS for rubratoxins biosynthesis was deleted, to give a mutant strain $\Delta r b t J$, which was screened for appropriate culture media. Among the screening rice, PDA, PDB, CY, YG, and YMEG media, rice medium was found to be a perfect medium for strain $\Delta r b t J$, where abundant metabolites were detectable through LCMS analysis (Fig. 1B and Supporting Information Fig. S2). Comprehensive reverse transcription-PCR (RT-PCR) analysis of all 43 core biosynthetic genes exhibited that most core genes in the wild type showed weak to no expression except for the PKS gene for biosynthesis of rubratoxins in contig 20, while inhibiting the production of rubratoxins resulted in significantly increased expression of four additional PKS genes in contigs 4 and 60 (Supporting Information Fig. S3). Therefore, the combinational


Figure $3 \quad{ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY and key HMBC correlations of compounds 4, 27, and $\mathbf{3 5}$.
strategy applied here successfully up-regulated the expression of multiple cryptic genes encoding for PKSs and led to the pleiotropic production of secondary metabolism.

To identify the newly produced compounds in the $\Delta r b t J$ strain, a large-scale fermentation ( 13 kg ) in rice was carried out, following by various chromatography, which resulted in the isolation of a total of 36 compounds from part fractions (Fig. 2), of which 23 compounds possess new structure, including 9 azaphilone monomers (4-9, 18-19 and 34), 8 azaphilone dimers (20-25, 27 and 28), 5 azaphilone trimers (29-33), and 1 unclassified polyketide (35), along with 13 known ones including polyketides Sch $725680(\mathbf{1 0})^{19}$, pinophilin B (11) ${ }^{20}$, pinazaphilone B (12) ${ }^{21}$, Sch $1385568(\mathbf{1 3})^{22}$, talaraculone $F(\mathbf{1 4})^{23}$, $(+)$-mitorubrinic acid $\mathrm{B}(\mathbf{1 5})^{24}$, ( - )-mitorubrin $(\mathbf{1 6})^{25,26}$, ( - -mitorubrinol $(\mathbf{1 7})^{25}$, diazaphilonic acid (26) $)^{27,28},( \pm)$-asperlones A (36) ${ }^{29}$ and orsellinic acid $(\mathbf{3 7})^{30}$, alkaloid terreusinone $(\mathbf{3 8})^{31}$, and a siderophore desferritriacetylfusigen (39) ${ }^{32,33}$. Remarkably, three types of novel azaphilones polymers were discovered for the first time, including 5 unusual methylene-bridged azaphilone dimers (20-24), 2 rare spiro-polycyclic azaphilone dimers ( $\mathbf{2 7}$ and 28), and 5 unprecedented azaphilone trimers ( $\mathbf{2 9}-\mathbf{3 3}$ ) featuring both methylene and polycyclic bridged heterocycle and spiral structures, tetrahydro-3-oxaspiro[bicyclo[3.3.1]nonane-2,2'-pyran] core.

Compound $\mathbf{4}$ was isolated as a yellow amorphous powder. The molecular formula $\mathrm{C}_{21} \mathrm{H}_{22} \mathrm{O}_{7}$ was determined according to its high resolution ESI MS (HR-ESI-MS) spectrum ( $[\mathrm{M}+\mathrm{H}]^{+}, \mathrm{m} / \mathrm{z}$ 387.1438; Calcd. for $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{O}_{7}, 387.1438$ ) combined with the ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR data (Supporting Information Tables S4 and S5). IR spectrum showed the presence of hydroxy $\left(3355 \mathrm{~cm}^{-1}\right)$, conjugated carbonyl $\left(1645 \mathrm{~cm}^{-1}\right)$, and phenyl (1583 and $1447 \mathrm{~cm}^{-1}$ ) groups. The ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of 4 showed two aromatic proton signals at $\delta_{\mathrm{H}} 6.11(\mathrm{~s}, 2 \mathrm{H})$, a methyl protons at $\delta_{\mathrm{H}} 2.12(\mathrm{~s}, 3 \mathrm{H})$, two phenolic hydroxy protons at $\delta_{\mathrm{H}}$ $9.96(\mathrm{~s}), 10.63(\mathrm{~s})$, a conjugated carbonyl carbon at $\delta_{\mathrm{C}} 169.0$, and six aromatic carbon signals in the region of $\delta_{\mathrm{C}} 100.3-161.1$, indicating the existence of 2,4-dihydroxy-6-methyl benzoic acid (orsellinic acid) moiety ${ }^{19,21}$. In addition, the NMR data also displayed resonances attributable to a conjugated ketene at $\delta_{\mathrm{C}} 198.5$ (C-6), two olefinic protons at $\delta_{\mathrm{H}} 5.72(\mathrm{~d}, J=2.1 \mathrm{~Hz}, \mathrm{H}-5), 5.77$ (s, H-4), two E-configured olefinic protons at $\delta_{\mathrm{H}} 6.33$ (dq, $J=15.5,6.9 \mathrm{~Hz}, \mathrm{H}-10), 6.01$ (dd, $J=15.5,1.6 \mathrm{~Hz}, \mathrm{H}-9$ ), two methyl groups at $\delta_{\mathrm{H}} 1.81\left(\mathrm{~d}, J=6.9,1.6 \mathrm{~Hz}, \mathrm{H}_{3}-11\right), 1.33\left(\mathrm{~s}, \mathrm{H}_{3^{-}}\right.$ 12), an oxygenated methylene at $\delta_{\mathrm{H}} 3.75(\mathrm{dd}, J=12.9,10.6 \mathrm{~Hz}$, $\mathrm{H}-1 \mathrm{a}), 4.51$ (dd, $J=10.6,5.2 \mathrm{~Hz}, \mathrm{H}-1 \mathrm{~b})$, $\delta_{\mathrm{C}} 66.8$ (C-1), an oxygen-bearing methine at $\delta_{\mathrm{H}} 5.60(\mathrm{~d}, J=2.8 \mathrm{~Hz}, \mathrm{H}-8), \delta_{\mathrm{C}} 77.2$ (C-8), a methine at $\delta_{\mathrm{H}} 3.43(\mathrm{~m}, \mathrm{H}-8 \mathrm{a}), \delta_{\mathrm{C}} 35.7$ (C-8a), an alcoholic hydroxy proton at $\delta_{\mathrm{H}} 5.36(\mathrm{~s}, 7-\mathrm{OH})$, and three quaternary carbons at $\delta_{\mathrm{C}} 158.5(\mathrm{C}-3), 149.1$ (C-4a), 74.8 (C-7).

Analysis of the ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY NMR data of $\mathbf{4}$ identified the C $1 / \mathrm{C}-8 \mathrm{a} / \mathrm{C}-8$ and C-9/C-10/C-11 moieties (Fig. 3). HMBC correlations from olefin proton $\mathrm{H}-5$ to $\mathrm{C}-7 / \mathrm{C}-8 \mathrm{a}$, from oxygenated methine proton $\mathrm{H}-8$ to ketone carbonyl $\mathrm{C}-6 / \mathrm{C}-4 \mathrm{a} / \mathrm{C}-7$, and from Me-12 to C-6/C-7/C-8 established the 7-methyl cyclohexenone ring. HMBC correlations from olefinic proton $\mathrm{H}-4$ to $\mathrm{C}-3 / \mathrm{C}-5 / \mathrm{C}-$ 8 a , from oxygenated methylene proton $\mathrm{H}-1$ to $\mathrm{C}-3 / \mathrm{C}-4 \mathrm{a} / \mathrm{C}-8 \mathrm{a}$ displayed the presence of a dihydropyran ring. The side chain C-9/ C-10/C-11 was adjacent to C-3 supported by HMBC correlations from H-9 to C-3 and from H-4 to C-9. Thus, the azaphilone skeleton was established. HMBC correlations from oxygenbearing methine proton $\mathrm{H}-8$ to carbonyl carbon $\mathrm{C}-13$ suggested that the orsellinic acid moiety attached to C-8 (Fig. 3). Therefore, the planar structure of $\mathbf{4}$ was determined, which was same as that of a known azaphilone analogue Sch $725680(\mathbf{1 0})^{19}$, but with different relative configuration.

The relative configuration of $\mathbf{4}$ was determined on the basis of ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ coupling constants and NOESY experiments. The coupling constant of $J_{8,8 \mathrm{a}}=2.8 \mathrm{~Hz}$ indicated the syn relationship of $\mathrm{H}-8 / \mathrm{H}-8 \mathrm{a}$. In the NOESY spectrum, NOE correlations of $\mathrm{H}-8$ with $\mathrm{Me}-12$ and $\mathrm{H}-8$ a suggested that these protons were placed on the same side of the molecule (Fig. 4). The absolute configuration of $\mathbf{4}$ was determined by the circular dichroism (CD) exciton chirality method ${ }^{34,35}$. The CD spectrum of 4 showed Cotton effects involved in the conjugated trienone chromophores interacting with benzoate at 360 and $306 \mathrm{~nm}(\Delta \varepsilon-1.61$ and 0.57$)$, suggesting the $8 S$ configuration. Therefore, the absolute configuration of $\mathbf{4}$ was established as $7 S, 8 S, 8 \mathrm{a} R$ and it was named dangelone A .

Analysis of NMR data (Tables S4 and S5) of compounds 5 and $\mathbf{6}$ suggested that they had the similar azaphilone plane frame as $\mathbf{4}$ and the distinction between them lay the respective side chains at $\mathrm{C}-3$. In the side chain of $\mathbf{5}$, a hydroxymethyl group at $\mathrm{C}-11\left[\delta_{\mathrm{H}}\right.$ $4.09(\mathrm{~d}, J=13.6 \mathrm{~Hz}, 2 \mathrm{H}), \delta_{\mathrm{C}} 60.6$, and $\delta_{\mathrm{H}} 4.98(\mathrm{t}, J=5.1 \mathrm{~Hz}$, $\mathrm{OH}-11$ )] was present instead of $\mathrm{C}-11$ methyl group ( $\delta_{\mathrm{C}} 18.1$ ) in 4. The COSY correlations of $\mathrm{H}-9 / \mathrm{H}-10 / \mathrm{H}-11 / \mathrm{OH}-11$ and HMBC correlations from $\mathrm{H}-11$ to $\mathrm{C}-9 / \mathrm{C}-10$ confirmed the side chain assignment of 5 . In the side chain of $\mathbf{6}$, a free $\mathrm{C}-11$ carboxyl group ( $\delta_{\mathrm{C}} 166.8$ ) replaced the $\mathrm{C}-11$ methyl in 4 , which was verified by HMBC correlation from $\mathrm{H}-9$ to $\mathrm{C}-11$. The relative configurations of 5 and $\mathbf{6}$ were in accordance with that of $\mathbf{4}$ based on NOESY spectra. By comparing the CD spectra and optical rotation with those of 4, the absolute configurations of $\mathbf{5}$ and $\mathbf{6}$ were both determined as $7 S, 8 S, 8 \mathrm{a} R$ and they were named dangelones B and C, respectively.

Compound 7 possessed the molecular formula $\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{O}_{9}$ on the basis of HRESIMS $\left([\mathrm{M}+\mathrm{Na}]^{+}, m / z\right.$ 467.1328; Calcd. for $\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{O}_{9} \mathrm{Na}, 467.1313$ ). The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data (Tables S 4 and S5) of 7 showed an additional acetyl group signals at $\delta_{\mathrm{H}}$


Figure 4 Key NOESY correlations of compounds 4, 27, and 28.
$2.02(3 \mathrm{H}, \mathrm{s}, \mathrm{H}-22)$ and $\delta_{\mathrm{C}} 20.7$ (C-22), 170.6 (C-21) by comparison to those of pinophilin $\mathrm{B}(\mathbf{1 1})^{20}$. HMBC correlations from oxygenated methylene protons $\mathrm{H}_{2}-11\left(\delta_{\mathrm{H}} 4.70\right)$ to carbonyl carbon $\mathrm{C}-21\left(\delta_{\mathrm{C}} 170.6\right)$ and from methyl protons $\mathrm{H}_{3}-22\left(\delta_{\mathrm{H}} 2.02\right)$ to $\mathrm{C}-21$ confirmed that the acetyl group was attached to the terminal C-11 hydroxy in 11. Thus, 7 was an acetylated derivative of 11 and named dangelone D .

Compound $\mathbf{8}$ was established the molecular formula $\mathrm{C}_{21} \mathrm{H}_{22} \mathrm{O}_{9}$ by positive HRESIMS at $m / z 419.1352[\mathrm{M}+\mathrm{H}]^{+}$(Calcd. for $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{O}_{9}, 419.1337$ ), two more hydrogens than that of pinazaphilone B(12) ${ }^{21}$. Comparing the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data (Tables S4 and S5) of $\mathbf{8}$ with those of $\mathbf{1 2}$ indicated that a trans-olefin double bond ( $\delta_{\mathrm{C}} 137.5,125.5$ ) at $\mathrm{C}-9(\mathrm{C}-10)$ in $\mathbf{1 2}$ was replaced by two methylene [ $\delta_{\mathrm{H}} 2.55(2 \mathrm{H}$, overlap, H-9), 2.56 ( 2 H , overlap, $\mathrm{H}-10$ ) and $\delta_{\mathrm{C}} 31.1$ (C-9), 30.5 (C-10)] in 8, suggesting that $\mathbf{8}$ was a hydrogenated derivative of $\mathbf{1 2}$ at $\mathrm{C}-9$ and $\mathrm{C}-10$, which was confirmed by HMBC correlations from methylene protons $\mathrm{H}_{2}-9$ ( $\delta_{\mathrm{H}} 2.55$ ) to $\mathrm{C}-3 / \mathrm{C}-4 / \mathrm{C}-10 / \mathrm{C}-11$. The relative configurations of 7 and $\mathbf{8}$ were established to be same as that of $\mathbf{1 2}$ by the similar $J_{8,8 \mathrm{a}}$ coupling constants and NOESY spectra, respectively. Since the CD spectra of $\mathbf{7}$ and $\mathbf{8}$ agree well with those of $\mathbf{1 0}^{19}$ and $\mathbf{1 2}^{21}$, respectively (Supporting Information Figs. S283 and S284), the absolute configurations of $\mathbf{7}$ and $\mathbf{8}$ were both assigned as $7 S, 8 S$, 8 aS . Then, $\mathbf{8}$ was named dangelone E .

Comparing the NMR data (Tables S4 and S5) of compound 9 with those of $\mathbf{1 2}^{21}$, an oxygenated methine was observed at $\delta_{\mathrm{H}}$ 4.00 (d, $J=4.4 \mathrm{~Hz}, \mathrm{H}-6)$ and $\delta_{\mathrm{C}} 74.3$ (C-6) in 9, with the disappearance of ketone carbonyl carbon at $\delta_{\mathrm{C}} 197.3$ in 12. HMBC correlations from the oxygenated methine proton H-6 to $\mathrm{C}-4 \mathrm{a} / \mathrm{C}-5 / \mathrm{C}-7 / \mathrm{C}-12$, from $\mathrm{H}-12$ to $\mathrm{C}-6 / \mathrm{C}-7 / \mathrm{C}-8$, and from $\mathrm{H}-5$ to C-4/C-7/C-8a indicated that ketone carbonyl group at C-6 in $\mathbf{1 2}$ was reduced to a hydroxy in 9 . The trans-diaxial relationship of $\mathrm{H}-8 / \mathrm{H}-8 \mathrm{a}$ was deduced from the coupling constant $\left(J_{8,8 \mathrm{a}}=10.1 \mathrm{~Hz}\right)$. The NOESY correlations between H-6/H-12/H8 suggested that they were placed on the same side of the molecule. To determine the absolute configuration of $\mathbf{9}$, the CD spectrum was measured in methanol and compared with its calculated electronic circular dichroism (ECD) of $6 R, 7 R, 8 S, 8 a S-9$ and $6 S, 7 S, 8 R, 8 \mathrm{a}-9$ using the quantum chemical method ${ }^{36}$. The predicted ECD spectrum of $6 R, 7 R, 8 S, 8 \mathrm{aS}-9$ exhibited an excellent fit with the experimental one (Supporting Information Fig. S285). Thus, the absolute configuration of 9 was assigned to be $6 R, 7 R$, $8 S, 8 \mathrm{a} S$ and it was named dangelone F .

Compound $\mathbf{1 8}$ was obtained as a yellow amorphous powder and had a molecular formula of $\mathrm{C}_{26} \mathrm{H}_{28} \mathrm{O}_{13}$ on the base of HRESIMS $[\mathrm{M}+\mathrm{Na}]^{+} \mathrm{m} / \mathrm{z} \quad 571.1429$ (Calcd. for $\mathrm{C}_{26} \mathrm{H}_{28} \mathrm{O}_{13} \mathrm{Na}$, 571.1422). The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data (Supporting Information

Table S6) exhibited additional five oxygenated carbons ( $\delta_{\mathrm{C}}$ $101.1,73.0,70.7,88.1,62.9$ ), an anomeric proton at $\delta_{\mathrm{H}} 5.67$ (d, $J=4.5 \mathrm{~Hz}$ ), and six hydrogens between $\delta_{\mathrm{H}} 3.66-4.21$ by comparing their NMR data with those of 6 , indicating the presence of a $\alpha$-ribose ${ }^{37}$, which was confirmed by ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY and HMBC correlations. The $\alpha$-ribose was connected to $\mathrm{C}-17$ in orsellinic acid moiety supported by HMBC correlation from anomeric proton $\mathrm{H}-1^{\prime}$ to $\mathrm{C}-17$. The sugar moiety in $\mathbf{1 8}$ was determined as D-ribose by comparing the retention time in LCMS of the thiocarbamoyl-thiazolidine derivative of the acid hydrolysate of 18 with those of authentic D-ribose and l-ribose (Supporting Information Fig. S7). Due to the same relative configuration and similar CD spectrum by comparing with those of $\mathbf{6}$, the absolute configuration of the aglycone in $\mathbf{1 8}$ was established as $7 S, 8 S, 8 \mathrm{a} R$, which was further confirmed by ECD calculations ${ }^{36}$. The calculated weighted ECD spectrum of $7 S, 8 S, 8 \mathrm{a} R$ $\mathbf{1 8}$ agrees well with the experimental spectrum of $\mathbf{1 8}$ (Supporting Information Fig. S286). Thus, 18 was determined to be $17-O-\alpha$-d-ribosyl-dangelone C and named dangeloside A .

The NMR spectroscopic data (Table S6) of compound 19 closely resembled those of Sch $725680(\mathbf{1 0})^{19}$, except for the presence of an additional ribose moiety, the ${ }^{13} \mathrm{C}$ NMR data of which was in good accordance with those of $\alpha$-ribose moiety in 18. HMBC correlation from anomeric proton $\mathrm{H}-1^{\prime}$ to $\mathrm{C}-17$ suggested that ribose was linked to $\mathrm{C}-17$. Due to the limitation of the trace amount of $\mathbf{1 9}$, the configuration of d-ribose in 19 was determined according to the same biosynthetic origin of suger moiety in 18. The relative configuration of the aglycone in 19 was same as that of $\mathbf{1 0}$ based on NOESY spectrum. The absolute configuration of the aglycone in $\mathbf{1 9}$ was identified as $7 S, 8 S, 8 \mathrm{a} S$ by comparison of the CD spectrum with those of $\mathbf{1 0}$ (Supporting Information Fig. S287). Thus, 19 was determined to be $17-O-\alpha-$ D-ribosyl-Sch 725680 and named dangeloside B.

Compound 20 was isolated as a light yellow powder, HRESIMS analysis of $\mathbf{2 0}$ gave an ion $[\mathrm{M}+\mathrm{H}]^{+}$at $m / z 777.2205$ consistent with a molecular formula of $\mathrm{C}_{43} \mathrm{H}_{36} \mathrm{O}_{14}$ (Calcd. for $\mathrm{C}_{43} \mathrm{H}_{37} \mathrm{O}_{14}, 777.2178$ ). Acquisition of ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$, HSQC NMR spectra of $\mathbf{2 0}$ showed only 21 carbons, the presence of an orsellinic acid moiety, and signals of azaphilone skeleton, including a downfield shifted olefinic proton at $\delta_{\mathrm{H}} 8.08$ (s), two ketone carbonyl at $\delta_{\mathrm{C}}$ 193.8, 193.6, a propenyl group at $\delta_{\mathrm{H}} 5.78$ (dd, $J=15.6,1.3 \mathrm{~Hz}), 6.51(\mathrm{dq}, J=15.6,6.9 \mathrm{~Hz}), 1.67(\mathrm{~d}$, $J=6.9 \mathrm{~Hz}$ ), $\delta_{\mathrm{C}} 124.1,137.0,18.7$ (Supporting Information Tables S7 and S9). These signals were very similar to those of 16, a known compound ( - )-mitorubrin ${ }^{25}$, except for the presence of an additional methylene ( $\delta_{\mathrm{H}} 3.67, \delta_{\mathrm{C}} 19.6$ ) in 20. According to its molecular formula, $\mathbf{2 0}$ was presumed to be composed of two
same azaphilone monomer moieties similar to 16. Furthermore, HMBC correlations from the additional methylene protons $\mathrm{H}_{2}-21$ ( $\delta_{\mathrm{H}} 3.67$ ) to $\mathrm{C}-4 \mathrm{a} / \mathrm{C}-5 / \mathrm{C}-6$, from $\mathrm{H}-1$ to $\mathrm{C}-3 / \mathrm{C}-4 \mathrm{a} / \mathrm{C}-8 / \mathrm{C}-8 \mathrm{a}$, from H-4 to C-3/C-4a/C-8a/C-9, and from H-9 to C-3/C-4/C-10/C-11 confirmed that $\mathbf{2 0}$ was a symmetric azaphilone dimer connected by a C-21 methylene bridge to $\mathrm{C}-5$ and $\mathrm{C}-5^{\prime}$ in two same azaphilone monomer moieties, respectively. Then, $\mathbf{2 0}$ was named didangelone A .

Compound 21 was isolated as yellow powder; its molecular formula was determined as $\mathrm{C}_{43} \mathrm{H}_{36} \mathrm{O}_{15}$ by HRESIMS $\left([\mathrm{M}+\mathrm{H}]^{+}\right.$, $\mathrm{m} / \mathrm{z}$ 793.2121; Calcd. for $\mathrm{C}_{43} \mathrm{H}_{37} \mathrm{O}_{15}, 793.2127$ ), one more oxygen than that of $\mathbf{2 0}$. The ${ }^{13} \mathrm{C}$ NMR data (Table S9) of $\mathbf{2 1}$ were closely related to those of $\mathbf{2 0}$ except that the signal for $\mathrm{C}-11$ at $\delta_{\mathrm{C}} 18.7$ in 20 was shifted downfield to $\delta_{\mathrm{C}} 62.9$ in 21, which indicated that 21 was the 11 -hydroxy derivative of $\mathbf{2 0}$. HMBC correlations from $\mathrm{H}-$ 11 to C-9/C-10 confirmed the location of hydroxy group at C-11. Then, 21 was named didangelone B .

The molecular formula $\mathrm{C}_{43} \mathrm{H}_{36} \mathrm{O}_{16}$ of compound 22 was established by HRESIMS ([M+Na] ${ }^{+}, m / z$ 831.1918; Calcd. for $\mathrm{C}_{43} \mathrm{H}_{36} \mathrm{O}_{16} \mathrm{Na}, 831.1896$ ). Analysis of the NMR spectra indicated that $\mathbf{2 2}$ also was a completely symmetrical azaphilone dimer. The NMR data (Tables S7 and S9) of $\mathbf{2 2}$ showed very similar signals to those of $\mathbf{2 0}$. The differences were that a methyl group ( $\delta_{\mathrm{H}} 1.67, \delta_{\mathrm{C}}$ 18.7) at C-11/C-11' in $\mathbf{2 0}$ was replaced by a hydroxymethyl group ( $\delta_{\mathrm{H}} 4.07, \delta_{\mathrm{C}} 60.6$ ) in 22, respectively, which was elucidated by ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY correlations of $\mathrm{H}-9 / \mathrm{H}-10 / \mathrm{H}-11$ and HMBC correlations from hydroxymethyl proton $\mathrm{H}-11$ to $\mathrm{C}-9 / \mathrm{C}-10$. Then, 22 was named didangelone C .

Compound 23 was isolated as a red amorphous powder. The molecular formula of $23, \mathrm{C}_{43} \mathrm{H}_{37} \mathrm{NO}_{13}$, was established by its HRESIMS spectrum $\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{m} / \mathrm{z}\right.$ 776.2358; Calcd. for $\mathrm{C}_{43} \mathrm{H}_{38} \mathrm{NO}_{13}, 776.2338$ ), which showed that a nitrogen atom replaced an oxygen atom comparing with that of $\mathbf{2 0}$. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra (Tables S7 and S9) showed the typical signals of methylene-bridged azaphilone dimer. Detailed analysis of its NMR spectrum revealed that comparing with $\mathrm{C}-1^{\prime}$ and $\mathrm{C}-3^{\prime}$, the resonances for $\mathrm{C}-1$ and $\mathrm{C}-3$ were shifted up-field by 15.6 and 11.3 ppm , respectively, which suggested that an oxygen atom from one of pyran ring in $\mathbf{2 0}$ was replaced by a nitrogen atom to become a pyridine ring. Then, $\mathbf{2 3}$ was named didangelone D .

Compound 24 was also a red amorphous powder, its molecular formula was determined as $\mathrm{C}_{43} \mathrm{H}_{37} \mathrm{NO}_{15}$ by HRESIMS ( $[\mathrm{M}+\mathrm{H}]^{+}$, $\mathrm{m} / \mathrm{z}$ 808.2291; Calcd. for $\mathrm{C}_{43} \mathrm{H}_{38} \mathrm{NO}_{15}, 808.2295$ ). The NMR data (Tables S7 and S9) of $\mathbf{2 4}$ were similar to those of 23, indicating that $\mathbf{2 4}$ was also a nitrogenous azaphilone dimer with a pyridine ring. The differences from $\mathbf{2 3}$ were that two methyl group ( $\delta_{\mathrm{H}} 1.51$ and $1.55, \delta_{\mathrm{C}} 18.74$ and 18.65 ) at $\mathrm{C}-11$ and $\mathrm{C}-11^{\prime}$ in 23 was replaced by two hydroxymethyl group ( $\delta_{\mathrm{H}} 3.85$ and $3.81, \delta_{\mathrm{C}} 60.7$ and 60.9) in 24, respectively, which was confirmed by HMBC correlations from $\mathrm{H}-11\left(\delta_{\mathrm{H}} 3.85\right)$ to $\mathrm{C}-9 / \mathrm{C}-10$ and from $\mathrm{H}-11^{\prime}\left(\delta_{\mathrm{H}}\right.$ 3.81 ) to $\mathrm{C}-9^{\prime} / \mathrm{C}-10^{\prime}$. Then, $\mathbf{2 4}$ was named didangelone E .

The absolute configuration of $\mathbf{2 0}$ was determined by ECD calculations ${ }^{36}$. It was found that the calculated weighted ECD spectra of $7 R, 7^{\prime} R$-20 were in good accordance with the experimental CD spectra of $\mathbf{2 0}$ (Supporting Information Fig. S288). Consequently, the absolute configuration of $\mathbf{2 0}$ was identified as $7 R, 7^{\prime} R$. Likewise, the CD curve similarities between 21, 22, 23, 24 and 20 indicated that they bear the same configurations.

Compound 25 was isolated as a light yellow amorphous powder. Its molecular formula $\mathrm{C}_{42} \mathrm{H}_{34} \mathrm{O}_{17}$ was established by HRESIMS $\left([M+H]^{+}, \mathrm{m} / \mathrm{z}\right.$ 811.1860; Calcd. for $\mathrm{C}_{42} \mathrm{H}_{35} \mathrm{O}_{17}$,
811.1869). The 1D NMR (Supporting Information Tables S8 and S9) and HSQC data showed four ketone carbonyl carbons C-6/C-8/C-6'/C-8 ( $\delta_{\mathrm{C}} 194.8 / 194.0 / 195.0 / 193.9$ ), two oxygenated quaternary carbon C-7/C-7' ( $\delta_{\mathrm{C}} 86.6 / 86.2$ ), two downfield shifted olefin protons $\mathrm{H}-1 / \mathrm{H}-1^{\prime}\left(\delta_{\mathrm{H}} 8.15 / 8.23\right)$, and two orsellinic acid moieties. Based on these features, the NMR data of $\mathbf{2 5}$ were closely related to those of a known azaphilone dimer diazaphilonic acid (26) ${ }^{27,28}$ except that the carboxyl group at C-11 ( $\delta_{\mathrm{C}}$ 174.7) in 26 was replaced by a hydroxymethyl group [ $\delta_{\mathrm{H}} 3.65$ (d, $J=7.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), $\delta_{\mathrm{C}} 63.7$ ] in 25, which was confirmed by HMBC correlations from $\mathrm{H}-11\left(\delta_{\mathrm{H}} 3.65\right)$ to $\mathrm{C}-9 / \mathrm{C}-10$ and ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY correlations of $\mathrm{H}-11 / \mathrm{H}-10\left(\mathrm{H}-10^{\prime}\right) / \mathrm{H}-9$. Thus, the planar structure of $\mathbf{2 5}$ was determined. The relative configuration of $\mathbf{2 5}$ was established on the basis of NOESY experiment. The NOE correlations of $\mathrm{H}-11$ with $\mathrm{H}-9 / \mathrm{H}-10^{\prime} / \mathrm{H}-9^{\prime}$ a suggested that these protons were placed on the same side of the molecule. According to its biosynthesis pathway same as that of azaphilone monomer, the absolute configuration of $\mathrm{C}-7$ and $\mathrm{C}-7^{\prime}$ was assigned as $R$. ECD calculations were conducted to clarify the absolute configuration of $\mathbf{2 5}^{36}$. The calculated weighted ECD spectra of $9 R, 10 R, 10^{\prime} R, 7 R, 7^{\prime} R-25$ and $9 S, 10 S, 10^{\prime} S, 7 R, 7^{\prime} R-25$ have obvious differences, revealing that the calculated ECD spectrum of $9 R, 10 R, 10^{\prime} R, 7 R, 7^{\prime} R-\mathbf{2 5}$ agrees well with the experimental one (Supporting Information Fig. S289). Therefore, the absolute configuration of $\mathbf{2 5}$ was assigned as $9 R, 10 R, 10^{\prime} R, 7 R, 7^{\prime} R$ and it was named didangelone H .

Compound 27 was isolated as a light yellow amorphous powder. HRESIMS spectrum established its molecular formula as $\mathrm{C}_{42} \mathrm{H}_{34} \mathrm{O}_{17}\left([\mathrm{M}+\mathrm{H}]^{+}, \mathrm{m} / \mathrm{z}\right.$ 811.1829; Calcd. for $\mathrm{C}_{42} \mathrm{H}_{35} \mathrm{O}_{17}$, 811.1869), showing 26 degrees of unsaturation. 1D NMR (Tables S8 and S9) and HSQC spectra displayed typical signals for azaphilone dimer, including four ketone carbonyl carbons C-6/C-8/C$6^{\prime} / \mathrm{C}-8^{\prime}\left(\delta_{\mathrm{C}} 195.4 / 193.4 / 194.7 / 194.3\right.$ ), two downfield shifted olefin protons $\mathrm{H}-1 / \mathrm{H}-1^{\prime}\left(\delta_{\mathrm{H}} 7.90 / 8.13\right)$, and two orsellinic acid moieties. The ${ }^{13} \mathrm{C}$ NMR data and HMBC interactions from $\mathrm{H}-1$ to $\mathrm{C}-3 / \mathrm{C}-4 \mathrm{a} /$ $\mathrm{C}-8 / \mathrm{C}-8 \mathrm{a}$ and from $\mathrm{H}-1^{\prime}$ to $\mathrm{C}-3^{\prime} / \mathrm{C}-4 \mathrm{a}^{\prime} / \mathrm{C}-8^{\prime} / \mathrm{C}-8 \mathrm{a}^{\prime}$ established two orsellinic acyl substituted azaphilone pyranoquinone bicyclic cores similar to that of $\mathbf{1 6}$. In one of azaphilone cores, a double bond at C-3 and C-4 was reduced, which was supported by HMBC correlations of $\mathrm{H}-1$ to $\mathrm{C}-3\left(\delta_{\mathrm{C}} 104.3\right)$ and of $\mathrm{H}_{2}-4\left(\delta_{\mathrm{H}} 3.19 / 2.82\right)$ to C-3/C-4a/C-5/C-8a. In addition to above mentioned two orsellinic acyl substituted azaphilone cores, the remaining ${ }^{13} \mathrm{C}$ NMR data showed only one unsaturated carboxyl carbon signal at $\delta_{\mathrm{C}} 176.3$ (C-11), indicating two unassigned degrees of unsaturation. Therefore, $\mathbf{2 7}$ was presumed to be dimerized though two rings by two orsellinic acyl substituted azaphilone bicyclic cores. ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY cross-peaks of $\mathrm{H}-9 / \mathrm{H}-10 / \mathrm{H}-10^{\prime}\left(\mathrm{H}-9^{\prime}\right) / \mathrm{H}-11^{\prime}$ and HMBC interactions from methine proton $\mathrm{H}-9$ to $\mathrm{C}-3$, and from oxygenated methylene protons $\mathrm{H}-11^{\prime}$ to $\mathrm{C}-3 / \mathrm{C}-9^{\prime}$ established a tetrahydropyran ring connecting with a dihydropyran ring of azaphilone core though spiral carbon C-3 (Fig. 3). HMBC correlations from $\mathrm{H}-9$ to $\mathrm{C}-3^{\prime} / \mathrm{C}-4^{\prime} / \mathrm{C}-4 \mathrm{a}^{\prime}$, and from methylene proton $\mathrm{H}-9^{\prime}$ to $\mathrm{C}-3^{\prime} /$ $\mathrm{C}-10^{\prime} / \mathrm{C}-11^{\prime}$ indicated a cyclohexene ring fused with a pyran ring of another azaphilone core though double bond at $\mathrm{C}-3^{\prime}\left(\mathrm{C}-4^{\prime}\right)$, then establishing the polycyclic bridged heterocycle and spiral structures, tetrahydro-3-oxaspiro [bicyclo [3.3.1]nonane-2,2'-pyran] core. HMBC correlation from $\mathrm{H}-10$ to $\mathrm{C}-11$ suggested a carboxyl group attached to C-10. Thus, the planar structure of 27 was established. The signals for two active conjugated olefin protons $\mathrm{H}-5$ and $\mathrm{H}-5^{\prime}$ did not show in the ${ }^{1} \mathrm{H}$ NMR spectrum due to deuterium exchange in $\mathrm{CD}_{3} \mathrm{OD}$.

The relative configuration of 27 was assigned by NOESY experiment. NOESY correlations of axial $\mathrm{H}-11^{\prime} \mathrm{a}$ with $\mathrm{H}-1 / \mathrm{H}-10$ and of $\mathrm{H}-10$ with $\mathrm{H}-9 / \mathrm{H}-10^{\prime} / \mathrm{H}-11^{\prime} \mathrm{a}$ indicated that these protons were in the same side of the tetrahydropyran ring, whereas that of $\mathrm{H}-9^{\prime}$ a with $\mathrm{H}-11^{\prime} \mathrm{b}$ suggested the other side orientation (Fig. 4). According to the same biosynthetic pathway for azaphilone monomer, the absolute configuration of $\mathrm{C}-7$ and $\mathrm{C}-7^{\prime}$ was both assigned as $R$. ECD calculations were performed to clarify the absolute configuration of $\mathbf{2 7}{ }^{38}$. The calculated weighted ECD spectra of $3 R, 9 R, 10 S, 10^{\prime} R, 7 R, 7^{\prime} R-27$ and $3 S, 9 S, 10 R, 10^{\prime} S, 7 R, 7^{\prime} R$ 27 have obvious differences from 210 to 300 nm , revealing that the calculated ECD spectrum of $3 R, 9 R, 10 S, 10^{\prime} R, 7 R, 7^{\prime} R-27$ agrees well with the experimental one of 27 (Supporting Information Fig. S290). Therefore, the absolute configuration of 27 was determined as $3 R, 9 R, 10 S, 10^{\prime} R, 7 R, 7^{\prime} R$ and it was named didangelone F .

Compound 28 possessed the molecular formula $\mathrm{C}_{42} \mathrm{H}_{34} \mathrm{O}_{17}$ deduced from its HRESIMS $\left([\mathrm{M}+\mathrm{Na}]^{+}, m / z\right.$ 833.1708; Calcd. for $\mathrm{C}_{42} \mathrm{H}_{35} \mathrm{O}_{17} \mathrm{Na}, 833.1688$ ) and ${ }^{13} \mathrm{C}$ NMR. The planar structure of 28 was found to be same as that of 27 by detailed interpretation of its 1D and 2D NMR data (Tables S8 and S9). 28 differs from 27 only in the stereochemistry at $\mathrm{C}-3 / \mathrm{C}-9 / \mathrm{C}-10 / \mathrm{C}-10^{\prime}$, of which the relative configuration for $\mathrm{C}-3 / \mathrm{C}-9 / \mathrm{C}-10 / \mathrm{C}-10^{\prime}$ was same as that of 27 according to analysis of NOESY spectrum (Fig. 4). Using the same computational methods as that of $\mathbf{2 7}$, the absolute configuration of $\mathbf{2 8}$ was finally determined as $3 S, 9 S, 10 R, 10^{\prime} S, 7 R, 7^{\prime} R$ based on its CD curve similarities with ECD spectra of $3 S, 9 S, 10 R, 10^{\prime} S, 7 R, 7^{\prime} R-28$ (Supporting Information Fig. S290) ${ }^{38}$, which is actually a diastereoisomer of 27 and it was named didangelone G .

Compound 29 was isolated as a yellow amorphous powder. Its molecular formula was identified as $\mathrm{C}_{64} \mathrm{H}_{52} \mathrm{O}_{24}$ according to HRESIMS ( $[\mathrm{M}+\mathrm{H}]^{+}, \mathrm{m} / \mathrm{z}$ 1205.2882; Calcd. for $\mathrm{C}_{64} \mathrm{H}_{53} \mathrm{O}_{24}$, 1205.2921). 1D NMR (Supporting Information Tables S10 and S11) and HSQC data showed three sets of typical azaphilone monomer signals, including six ketone carbonyl carbons, three oxygenated olefin protons, and three orsellinic acid moieties, indicating that 29 was probably an azaphilone trimer. A detailed comparison of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data between 29, 27 and 20 disclosed that except for the presence of an azaphilone dimer moiety similar to 27 , an additional methylene ( $\delta_{\mathrm{H}} 3.25 / 3.12$, $\delta_{\mathrm{C}}$ 20.9) and an orsellinic acyl substituted azaphilone monomer moiety, resemble to the half fragment of $\mathbf{2 0}$, were present in $\mathbf{2 9}$. HMBC correlations from the additional methylene protons $\mathrm{H}_{2}{ }^{-}$ 21 ( $\delta_{\mathrm{H}} 3.25 / 3.12$ ) to $\mathrm{C}-5 / \mathrm{C}-6 / \mathrm{C}-4 \mathrm{a}^{\prime \prime} / \mathrm{C}-5^{\prime \prime}$ revealed that a $\mathrm{C}-21$ methylene bridge connected $\mathrm{C}-5^{\prime \prime}$ in an azaphilone monomer to C-5 in an azaphilone dimer moiety similar to 27 . Thus, the planar structure of 29 was established. According to NOESY correlations, the relative configuration for $\mathrm{C}-3 / \mathrm{C}-9 / \mathrm{C}-10 / \mathrm{C}-10^{\prime}$ of 29 was same as that of 27 . As the CD spectra of 29 with 27 highly coincidence, the absolute configurations of 29 was assigned as $3 R, 9 R, 10 S, 10^{\prime} R, 7 R, 7^{\prime} R, 7^{\prime \prime} R$. Then, 29 was named tridangelone A.

Compound 30 has the same molecular formula as 29, $\mathrm{C}_{64} \mathrm{H}_{52} \mathrm{O}_{24}$, deduced from HRESIMS ([M+Na] ${ }^{+}, m / z$ 1227.2653; Calcd. for $\mathrm{C}_{64} \mathrm{H}_{52} \mathrm{O}_{24} \mathrm{Na}, 1227.2682$ ). Interpretation of 1D and 2D NMR data (Tables S10 and S11) disclose that $\mathbf{3 0}$ and $\mathbf{2 9}$ possess the same planar structure. $\mathbf{3 0}$ differs from 29 only in the stereochemistry at $\mathrm{C}-3 / \mathrm{C}-9 / \mathrm{C}-10 / \mathrm{C}-10^{\prime}$. According to NOESY correlations, the relative configuration for $\mathrm{C}-3 / \mathrm{C}-9 / \mathrm{C}-10 / \mathrm{C}-10^{\prime}$ of $\mathbf{3 0}$ was same as that of 28 . The CD curve similarities between $\mathbf{3 0}$ and $\mathbf{2 8}$ indicated that the absolute configuration of $\mathbf{3 0}$ was established as
$3 S, 9 S, 10 R, 10^{\prime} S, 7 R, 7^{\prime} R, 7^{\prime \prime} R$, which is actually a diastereoisomer of 29 and it was named tridangelone B.

Compounds $\mathbf{3 1}$ and $\mathbf{3 2}$ were assigned to have the same planar structure by analysis of their HRESIMS and NMR data. Comparison of the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data (Tables S10 and S11) between 31 and 29 suggested that 31 was a $11^{\prime \prime}$-hydroxy derivative of 29 , which was supported by HMBC correlations from hydroxymethyl proton $\mathrm{H}-11^{\prime \prime}\left(\delta_{\mathrm{H}} 4.14\right)$ to $\mathrm{C}-9^{\prime \prime} / \mathrm{C}-10^{\prime \prime}$. According to NOESY correlations, the relative configurations for $\mathrm{C}-3 / \mathrm{C}-9 / \mathrm{C}-$ $10 / \mathrm{C}-10^{\prime}$ of $\mathbf{3 1}$ and $\mathbf{3 2}$ were same as those of $\mathbf{2 9}$ and $\mathbf{3 0}$, respectively. The absolute configurations of $\mathbf{3 1}$ and $\mathbf{3 2}$ were respectively determined as $3 R, 9 R, 10 S, 10^{\prime} R, 7 R, 7^{\prime} R, 7^{\prime \prime} R$ and $3 S, 9 S, 10 R$, $10^{\prime} S, 7 R, 7^{\prime} R, 7^{\prime \prime} R$ by comparing their CD spectra with those of 29 and 30. Then, 31 and 32 were named tridangelones C and D, respectively.

Compound 33 was isolated as a yellow amorphous powder. The molecular formula of 33 was determined to be $\mathrm{C}_{64} \mathrm{H}_{54} \mathrm{O}_{24}$ on the basis of HRESIMS data ( $[\mathrm{M}+\mathrm{Na}]^{+}, m / z$ 1229.2950, Calcd. for $\mathrm{C}_{64} \mathrm{H}_{54} \mathrm{O}_{24} \mathrm{Na}, 1229.2897$ ), one fewer oxygen and two more hydrogens than 31. The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra (Tables S10 and S11) of $\mathbf{3 3}$ and $\mathbf{3 1}$ showed very similar signals, with the exception that the carboxyl carbon signal at $\delta_{\mathrm{C}} 172.6$ (C-11) in 31 was replaced by the signals assigned to hydroxymethyl group at $\delta_{\mathrm{H}}$ $3.27 / 3.19$ and $\delta_{\mathrm{C}} 60.9$ in 33. These data indicated that the carboxyl group at $\mathrm{C}-11$ in $\mathbf{3 1}$ was substituted by a hydroxymethyl group in 33, which was supported by ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY correlation of $\mathrm{H}-11 / \mathrm{H}-$ 10 and HMBC correlation from $\mathrm{H}-11$ to $\mathrm{C}-9 / \mathrm{C}-10$. According to NOESY correlations, the relative configuration for C-3/C-9/C-10/ $\mathrm{C}-10^{\prime}$ of $\mathbf{3 3}$ was same as that of $\mathbf{3 1}$. Likewise, the CD curve similarities between $\mathbf{3 3}$ and $\mathbf{3 1}$ suggest that they bear the same absolute configurations and 33 was named tridangelone E.

Compound 34 was obtained as a yellow powder. HRESIMS analysis gave an $[\mathrm{M}+\mathrm{Na}]^{+}$ion at 585.1382 (Calcd. for $\mathrm{C}_{30} \mathrm{H}_{26} \mathrm{O}_{11} \mathrm{Na}, 585.1367$ ), consistent with a molecular formula of $\mathrm{C}_{30} \mathrm{H}_{26} \mathrm{O}_{11}$. Comparison of the NMR data (Supporting Information Table S12) of $\mathbf{3 4}$ with those of (-)-mitorubrin $(\mathbf{1 6})^{25}$ indicated the presence of three fragments, an orsellinic acyl substituted azaphilone monomer moiety similar to $\mathbf{1 6}$ (fragment A), an additional methylene ( $\delta_{\mathrm{H}} 3.40 / 3.76, \delta_{\mathrm{C}} 17.8$ ), and a 3substituted 2,4-dihydroxy-6-methylbenzoic acid moiety (fragment B). HMBC correlations from the methylene protons H-21 ( $\delta_{\mathrm{H}} 3.40 / 3.76$ ) to C-5/C-4a/C-6/C-22/C-23/C-27 suggested that the methylene bridge at $\mathrm{C}-21$ respectively linked to $\mathrm{C}-5$ in fragment A and C-22 in fragment B. Thus, the planar structure of $\mathbf{3 4}$ was established. The absolute configuration of $\mathbf{3 4}$ was determined as $7 R$ by comparing its CD spectra with those of $(-)$-mitorubrin (16) ${ }^{25,39}$ and it was named dangelone G.

Compound 35 was obtained as a yellow amorphous powder. Its molecular formula of $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{O}_{6}$ was determined on the basis of HRESIMS $\left([M-H]^{-}, m / z \quad 353.1032\right.$; Calcd. for $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{O}_{6}$, 353.1030). The ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$, and HSQC NMR (Table S12) spectra showed the presence of an aldehyde group at $\delta_{\mathrm{H}} 9.81(\mathrm{~s}), \delta_{\mathrm{C}} 195.0$, two methyl groups at $\delta_{\mathrm{H}} 1.86 / 2.00$ and $\delta_{\mathrm{C}} 21.2 / 7.2$, six aromatic protons at $\delta_{\mathrm{H}} 6.07 / 6.11 / 6.21 / 5.89 / 6.34 / 7.46$, and a methine at $\delta_{\mathrm{H}}$ 6.42 and $\delta_{\mathrm{C}} 39.3$. HMBC correlations (Fig. 3) from methyl proton $\mathrm{H}-13\left(\delta_{\mathrm{H}} 2.00\right)$ to C-9/C-10/C-11, from aldehyde proton $\mathrm{H}-14\left(\delta_{\mathrm{H}}\right.$ $9.81)$ to $\mathrm{C}-8 / \mathrm{C}-9$, and from aromatic proton $\mathrm{H}-12\left(\delta_{\mathrm{H}} 6.07\right)$ to $\mathrm{C}-7 /$ C-8/C-10/C-11 established a 6-substituted 2,4-dihydroxy-3methylbenzaldehyde (fragment C ). ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY correlations of $\mathrm{H}-4 / \mathrm{H}-5 / \mathrm{H}-6$ and HMBC correlations from $\mathrm{H}-4$ to $\mathrm{C}-2 / \mathrm{C}-6$ and from $\mathrm{H}-6$ to $\mathrm{C}-2 / \mathrm{C}-4 / \mathrm{C}-5$ confirmed a 2 -substituted furan ring (fragment D). Additionally, the ${ }^{1} \mathrm{H}$ NMR spectrum showed the
presence of two meta-position protons of benzene ring at $\delta_{\mathrm{H}} 6.21$ (d, $J=2.3 \mathrm{~Hz}, \mathrm{H}-17$ ), 6.11 (d, $J=2.3 \mathrm{~Hz}, \mathrm{H}-19$ ). HMBC correlations from methyl protons $\mathrm{H}-21$ ( $\delta_{\mathrm{H}} 1.86$ ) to $\mathrm{C}-15 / \mathrm{C}-19 / \mathrm{C}-20$, from $\mathrm{H}-17$ to $\mathrm{C}-15 / \mathrm{C}-18 / \mathrm{C}-19$, and from $\mathrm{H}-19$ to $\mathrm{C}-15 / \mathrm{C}-17 / \mathrm{C}-18$ established a 4 -substituted 1,3-dihydroxy-5-methylbenzene (fragment E). The key HMBC correlations from methine proton $\mathrm{H}-1$ ( $\delta_{\mathrm{H}} 6.42$ ) to C-2/C-6/C-7/C-8/C-12/C-15/C-16/C-20 indicated that the $\mathrm{C}-1$ methine bridge connected to $\mathrm{C}-7$ in fragment $\mathrm{C}, \mathrm{C}-2$ in fragment D , and $\mathrm{C}-15$ in fragments E , respectively (Fig. 3). Thus, the planar structure of $\mathbf{3 5}$ was established. Finally, ECD computational methods were used to clarify the absolute configuration of 35. The conformers of $1 R-35$ and $1 S-35$ were calculated by using the Gaussian $09^{36}$ software (Revision C.01, Gaussian Inc., Wallingford, CT, USA, 2010). It was found that the calculated weighted ECD spectra of $1 R-\mathbf{3 5}$ are in good accordance with the experimental CD spectrum of 35 (Supporting Information Fig. S291). Consequently, the $1 R$ configuration of $\mathbf{3 5}$ was assigned and it was named dangelone H .

### 2.4. Bioactivities evaluation

Cytotoxic activities of all isolated compounds were evaluated against two human cancer lines HepG-2 and MCF-7 in the MTT assay. Taxol was used as a positive control, with $\mathrm{IC}_{50}$ values of 1.28 (HepG2) and 2.31 (MCF-7) $\mu \mathrm{mol} / \mathrm{L}$, respectively. Compounds 14, 5, 10, and 11 exhibited cytotoxicities against HepG2 cells with $\mathrm{IC}_{50}$ values of $2.83,6.82,7.78$, and $11.4 \mu \mathrm{~mol} / \mathrm{L}$, respectively. Compounds $\mathbf{1 4}, \mathbf{1 0}$, and 5 showed potent inhibition against MCF-7 cells with $\mathrm{IC}_{50}$ values of $5.53,14.63$, and $14.98 \mu \mathrm{~mol} / \mathrm{L}$, respectively, while the other compounds were inactive ( $\mathrm{IC}_{50}>20 \mu \mathrm{~mol} / \mathrm{L}$, Supporting Information Table S13). Comparing the structure of compounds $\mathbf{1 4}, \mathbf{5}$, and $\mathbf{1 1}$ with that of the other azaphilone monomers, orsellinic acid esterification at C8 ( $\mathbf{1 4}$ vs. 17), double bond between C-8a and C-1 ( $\mathbf{1 4}$ vs. $\mathbf{5}$ and $\mathbf{1 1})$, and hydroxy at $\mathrm{C}-11(\mathbf{1 4}, \mathbf{5}, \mathbf{1 1} \mathrm{vs}$. the others) significantly increase the activity, indicating the important contribution effects on cytotoxicity.

In addition, compounds 4-39 were also assessed for their antiinflammatory and antioxidant inhibitory activities, with curcumin as a positive control (inhibitory rates $93.7 \pm 2.5 \%$ at $10^{-5} \mathrm{~mol} / \mathrm{L}$ and $91.2 \pm 3.2 \%$ at $10^{-4} \mathrm{~mol} / \mathrm{L}$, respectively). At a concentration of $10^{-5} \mathrm{~mol} / \mathrm{L}$, compound $\mathbf{1 3}$ was found to be significant inhibitory effect on NO production induced by lipopolysaccharide (LPS) in murine microglial BV2 cell with inhibitory rate of $96.0 \pm 4.6 \%$ (Supporting Information Table S14). Four compounds 21, 31, 34, and 36 were found to possess potent antioxidant activities against lipid peroxidation induced by $\mathrm{Fe}^{2+}$-cystine in rat liver microsomal with inhibitory rates of $68.7 \pm 4.2 \%$, $64.8 \pm 2.6 \%, 66.7 \pm 4.5 \%$, and $63.5 \pm 5.2 \%$, respectively, at a concentration of $10^{-4} \mathrm{~mol} / \mathrm{L}$ (Supporting Information Table S15). Interestingly, although the azaphilone monomers 16 and 17 showed no antioxidant activities and cytotoxicities, the azaphilone dimer 21 and trimer 31 exhibited potent antioxidant activities.

The antioxidant activities of compounds $4-39$ were evaluated by measuring the inhibitory rates of malondialdehyde (MDA) in rat liver microsomal lipid peroxidation induced by $\mathrm{Fe}^{2+}$-cystine in vitro according to the reported anti-inflammatory activities of compounds $\mathbf{4 - 3 9}$ were assessed by measuring the inhibitory effects on NO production induced by lipopolysaccharide (LPS) in murine microglial BV2 cell line according to the reported procedures.

### 2.5. Characterization of the gene cluster for azaphilones biosynthesis

To understand the biosynthesis of these azaphilones, we analyzed the PKSs in the genome of P. dangeardii. BLAST search against the non-reducing PKS (NR-PKS) AzaA in the aza cluster ${ }^{40}$ and MrpigA in the MonAzPs cluster ${ }^{41}$, which were responsible for the biosynthesis of azanigerones and Monascus azaphilone pigments (MonAzPs), respectively, revealed two NR-PKSs (DanE and $\mathrm{DanE}^{\prime}$ ) with high protein identities ( $>55 \%$ ) on Contig 60. A detailed bioinformatic analysis of adjacent genes of these two NRPKSs identified two attractive neighbouring dual PKSs gene clusters, which were designated as dan1 and dan2 (Supporting Information Fig. S292A). Since multiple PKS genes located in a same gene cluster are common for fungal genomes ${ }^{42}$ and it has been reported that two or three PKSs synthesize a polyketide product either organized in sequence ${ }^{43,44}$ or in convergence ${ }^{45}$, the combination of dan1 and dan2 clusters responsible for the biosynthesis of isolated azaphilones is reasonable. In the dan1 and dan2 clusters, the predicted domains of NR-PKSs DanE and DanE ${ }^{\prime}$ were same as starter unit: ACP transacylase (SAT), ketosynthase (KS), acyltransferase (AT), product template (PT), Cmethyltransferase (MT), acyl carrier protein (ACP) and reductase (R; Supporting Information Tables S16 and S17). Another NRPKS (DanV) with SAT-KS-AT-PT-ACP-ACP domains in the dan2 cluster was proposed to involve in the formation of orsellinic acid. However, the fatty acid synthase DanG with $\alpha / \beta$ subunits in the danl cluster was supposed to inactive since the absence of ACPS domain. Most of the genes in the dan1 and dan2 clusters have one or more corresponding homolog in the aza gene cluster ${ }^{40}$ (Table S16). However, no key AzaH-like flavin-dependent monooxygenase, which is involved in C-7 hydroxylation and pyranring formation to afford the characteristic pyrano-quinone bicyclic core in azaphilones ${ }^{40}$, was presented in the dan2 cluster. These features suggest that the pair of gene clusters might redundantly or collaboratively biosynthesize the novel azaphilone derivatives with structure diversity.

To verify the gene clusters responsible for azaphilones biosynthesis, a pathway specific transcriptional factor danS was overexpressed through a $g p d A$ gene as a strong promoter based on the homologous recombination method. The metabolites from the overexpressed mutant $\operatorname{gpdA}:: d a n S$ were extracted and analyzed by LC-MS (Supporting Information Fig. S4). Comparing to the metabolites from wild type $P$. dangeardii which did not produce any azaphilones, various azaphilones were detected in the mutant gpdA::danS, suggested that the dan gene clusters were responsible for azaphilones biosynthesis.

### 2.6. Proposed biosynthetic pathway for azaphilones from the dan1 and dan2 clusters

A putative biosynthetic pathway of azaphilones was proposed (Supporting Information Fig. S292B). Started from the NR-PKS DanE or $\operatorname{DanE}^{\prime}$ assembling the hexaketide precursor, an aldehyde intermediate was released. Then, the terminal ketone is reduced to alcohol $\mathbf{i}$ by the ketoreductase $\operatorname{DanU}(61 \%$ identity to AzaE). The FAD-dependent monooxygenase DanB, a homology of AzaH ( $59 \%$ identity) which is the only enzyme responsible for benzaldehyde hydroxylation and the formation of pyran ring during azanigerone biosynthesis ${ }^{40}$, is proposed to catalyze the hydroxylation of $\mathrm{C}-7 \mathrm{in} \mathbf{i}$, which triggers the pyran ring formation to yield ii. Then, $O$-acetyltransferase DanN/DanN ${ }^{\prime}$ catalyzes the
acetylation of $\mathbf{i i}$ at $10-\mathrm{OH}$, followed by the acetic acid elimination catalyzed by the dehydratase DanW to afford iii. In parallel, NRPKS DanV is responsible for the generation of orsellinic acyl chain and $O$-acetyltransferase DanC/DanC ${ }^{\prime}$ directly transferred the orsellinic acyl-ACP to the C-7 hydroxy of iii. Previous studies demonstrated that cytochrome P450 oxygenases are able to catalyze the terminal methyl into alcohol, aldehyde, or acid through one to three consecutive oxidation steps, examples including P450-1 and P450-4 for gibberellin biosynthesis, CYP71AV1 for artemisinic acid biosynthesis, and AsCYP51H10 for the biosynthesis of antimicrobial triterpenes ${ }^{46}$. The terminal C-11 methyl in 16, 13, 4, and $\mathbf{1 0}$ could be hydroxylated by cytochrome P450 DanM to afford 17, 14, 5, and 11, respectively, and further oxidized to yield 15, 6, and $\mathbf{1 2}$ catalyzed by DanM.

For the biosynthesis of azaphilone polymers, we hypothesize that the C-5(4a) double bond of the azaphilone monomer $\mathbf{I}$ is reduced by aldoketoreductase DanK (Fig. S292B), an orthologue of $\mathrm{NAD}(\mathrm{P}) \mathrm{H}$-dependent oxidoreductase $\mathrm{MrPigE}^{41}$ ( $69 \%$ identity), to yield intermediate II. The aldolase DanY is proposed to catalyze the enolization of C-6 ketone in II and introduce a hydroxymethyl at C-5 from formaldehyde, followed by dehydration to generate the reactive enone intermediate III. The enzyme with similar function has been reported, such as the $\alpha$-methylserine aldolase from Variovorax paradoxus, which is responsible for conversion from L-alanine to $\alpha$-methyl-L-serine accompanied by hydroxymethyl transfer from formaldehyde ${ }^{47}$. Michael addition between I and III yields the methylene-bridged azaphilone dimers (20-24). While non-enzymatic intermolecular Diels-Alder cycloaddition between two molecular of I results in the production of azaphilone dimers. Finally, azaphilone trimers (29-33) could be generated from combination of Michael addition and Diels-Alder cycloaddition between different azaphilone dimers with a third azaphilone monomer.

### 2.7. Discussion

Due to the potential of endophytic fungi to produce structurally diverse and biologically active secondary metabolites, various strategies for inducing the expression of silent biosynthetic gene clusters (BGCs) have been developed. However, no any strategies could be universal for all strains. The metabolic shunting strategy applied here is basically adapted from "Robin Hood" who robbed the wealthy to relieve the indigent. Inhibiting the biosynthesis of main metabolites in wild type strain would save the common building blocks (such as acetyl-CoA and malonyl-CoA for PKSs) to be readily competitively acquired by other BGCs, and induce the expression of cryptic gene clusters responsible for the biosynthesis of natural products required the saved building blocks. Results from RT-PCR (Fig. S3) also supported the upregulated expression of four additional genes encoding PKS in the mutant strain. Since polyketide precursors such as acetyl-CoA and malonyl-CoA were reserved by elimination of rubratoxins biosynthesis, three genes encoding PKS in contig 60 responsible for azaphilones biosynthesis were activated and showed high expression, which consequently led to the predominant azaphilones production in the mutant strain. In addition, trace metabolites could become detectable without the presence of main metabolites. Consequently, a siderophore peptide desferritriacetylfusigen (39) was found only from the mutant strain $\Delta r b t J$ although the gene encoding NRPS in contig 23 was expressed in both wild type and mutant strain.

Previously, an approach of genetic dereplication has been reported ${ }^{48-50}$ to discover novel natural products (especially trace metabolites) from fungi, which eliminated the production of major metabolites and made minor compounds detectable. Our strategy based on metabolic shunting and OSMAC is different in principle from genetic dereplication, and leads to the activation of cryptic gene clusters, besides letting trace metabolites detectable. The combinational application of metabolic shunting and OSMAC strategy is helpful to predict the biosynthesis and accumulation of specific types of natural products according to the deleted core gene type.

Although azaphilone monomers have been isolated from many other Penicillium strains ${ }^{21}$, such as Penicillium commune ${ }^{51}$ and Penicillium pinophilum ${ }^{20,52}$, inducing the expression of silent BGCs in $P$. dangeardii led to the production of many structurally complex azaphilone dimers and trimers with methylene and polycyclic bridged heterocycle and spiral structures.

Therefore, our established combinational strategy based on metabolic shunting and OSMAC is efficient and readily handled, and would be a perfect complement to other approaches of globally activating silent BGCs. This strategy has large potential to be applied for genome mining of other fungi or bacteria for discovering numerous novel secondary metabolites with a range of attractive bioactivities.

## 3. Conclusions

An approach combined metabolic shunting and OSMAC strategy was applied to successfully activate cryptic BGCs in the endophytic fungus $P$. dangeardii, which led to the production of numerous new compounds with diverse structure and biological activities. Remarkably, a group of unprecedented methylenebridged and polycyclic azaphilone dimers and trimers were found for the first time. The biosynthetic gene cluster of these azaphilones was identified through overexpression of a pathway transcription factor in azaphilones cluster. Bioinformatic analysis of the interesting pair of dan1 and dan2 gene clusters helps to uncover some intriguing aspects of azaphilones biosynthetic pathway. Consequently, our work successfully develops a new strategy to rapidly and globally activate cryptic gene clusters and discover the amazing novel bioactive natural products in endophytic fungi.

## 4. Experimental

### 4.1. Strains and cultivation conditions

The endophytic fungus $P$. dangeardii was grown on PDA (BD Difco, Sparks, NV, USA) plate at $28^{\circ} \mathrm{C}$ for 6 days for sporulation. $P$. dangeardii wild type and its mutant strains ( $\Delta r b t J$ and gpdA::danS ) were stationarily cultured in PDA, PDB (BD Difco), CY ( $0.01 \% ~(v / v)$ Czapek concentrate, $0.5 \%$ yeast extract, $3 \%$ sucrose, $0.1 \% \quad \mathrm{~K}_{2} \mathrm{HPO}_{4}$ ), YG ( $0.5 \%$ yeast extract, $2 \%$ glucose, $0.04 \%(v / v)$ trace element), YMEG ( $0.4 \%$ yeast extract, $1 \%$ malt extract, $0.4 \%$ glucose, $0.2 \% \mathrm{CaCO}_{3}$ ) and rice at $28{ }^{\circ} \mathrm{C}$ for 6 days for production of secondary metabolites. Escherichia coli XL1Blue was used for standard gene cloning. Saccharomyces cerevisiae YEH02 was used for yeast homologous recombination in vitro.

### 4.2. Gene knockout in P. dangeardii

The deletion of the target gene $r b t J$ (PKS) in $P$. dangeardii was carried out by homologous recombination using the $S$. cerevisiae YEH02. The genomic DNA of $P$. dangeardii was prepared using the plant genomic DNA extraction kit (Tiangen, Beijing, China). Primers for gene knockout were listed in Supporting Information Table S18. Pairs of primers, KC78uFor/KC78uRev, KC78dFor/ KC78dRev and hphFor/hphRev were used for amplification of the upstream ( 2.5 kb ) and downstream ( 2.5 kb ) homologous regions of $r b t J$ and $h p h$ (hygromycin resistance). These PCR products were ligated into linearized vector backbone derived from plasmid pRS423, introduced to YEH02 using PEG/LiAc mediated transformation, and selected on uracildropout semisynthetic media, then constructed the deletion cassette. The plasmid in the correct transformant screened by colony PCR was rescued using Zymoprep Yeast Plasmid Miniprep Kit (Zymo Research, Orange County, CA, USA) and transferred to E. coli for gene cloning. The correct plasmids were used as template to amplify the knockout cassette. The PCR products of knockout cassette were recovered by gel recycling kit (Axygen, Union, CA, USA) and dissolved in STC buffer ( $1.2 \mathrm{~mol} / \mathrm{L}$ sorbitol, $10 \mathrm{mmol} / \mathrm{L}$ calcium chloride and $10 \mathrm{mmol} / \mathrm{L}$ Tris-HCl, pH 7.5).
$P$. dangeardii spores were induced to young germ tubes in 50 mL PDB with $1 \%$ yeast extract at $28^{\circ} \mathrm{C}$ for 17 h with 180 rpm agitation (Zhicheng ZWY-103 B, Shanghai, China). Cells were collected by centrifuging at $4^{\circ} \mathrm{C}$ ( 5000 rpm ) for 5 min (Sigma 318 K , Gettingen, Germany), washed twice with osmotic buffer ( $1.2 \mathrm{~mol} / \mathrm{L}_{\mathrm{MgCl}}^{2}$, $10 \mathrm{mmol} / \mathrm{L}$ sodium phosphate, pH 5.8 ), and resuspended in the 12 mL enzyme cocktail solution ( 36 mg lysing enzyme and 24 mg yatalase in $1.2 \mathrm{~mol} / \mathrm{L} \mathrm{MgSO}_{4}, 10 \mathrm{mmol} / \mathrm{L}$ sodium phosphate buffer) in 150 mL flask at $30^{\circ} \mathrm{C}$ for 6 h with 80 rpm (Zhicheng ZWY-103B). After the digestion, the enzyme solution was transferred to a new tube and an equal volume of trapping buffer ( $0.6 \mathrm{~mol} / \mathrm{L}$ sorbitol, $10 \mathrm{mmol} / \mathrm{L}$ Tris- $\mathrm{HCl}, \mathrm{pH} 7.0$ ) was added before centrifuging at $4^{\circ} \mathrm{C}$ ( 5500 rpm , Sigma 3-18 K) for 10 min . The protoplast layer was collected and washed twice with two volumes of STC buffer and centrifuged ( 5000 rpm , Sigma 3-18 K) for 5 min at $4{ }^{\circ} \mathrm{C}$. Protoplast precipitated in the lower layer suspension with $100 \mu \mathrm{~L}$ STC buffer and gently mixed with $10 \mu \mathrm{~g}$ of knockout cassette DNA fragment and incubated for 1 h on the ice. 1.2 mL of PEG 4000 solution ( $60 \%$ PEG 4000, $50 \mathrm{mmol} / \mathrm{L} \mathrm{CaC1} 2,50 \mathrm{mmol} / \mathrm{L}$ Tris- $\mathrm{HCl}, \mathrm{pH} 7.5$ ) was added to reaction mixture, incubated at $28{ }^{\circ} \mathrm{C}$ for 20 min and plated on the regeneration selection medium (PDA, $1.2 \mathrm{~mol} / \mathrm{L}$ sorbitol, $800 \mathrm{mg} / \mathrm{L}$ hygromycin B). After culturing at $28^{\circ} \mathrm{C}$ for $4-5$ days, the transformants were stationarily inoculated on PDB for 3 days and then extracted the genomic DNA and confirming the genotype by PCR.

### 4.3. RNA preparation of $P$. dangeardii and reverse transcription-PCR (RT-PCR)

P. dangeardii wild type and the $\Delta r b t J$ mutant were grown on rice at $25^{\circ} \mathrm{C}$ for 6 days. The mycelia were collected and used for RNA extraction by TRIZOL® Reagent (Invitrogen, Carlsbad, CA, USA). Residual genomic DNA in the extracts was digested by DNase I ( $2 \mathrm{U} / \mu \mathrm{L}$, NEB) at $37{ }^{\circ} \mathrm{C}$ for 45 min . The cDNA was synthesized using the Transcriptor First Strand cDNA Synthesis Kit (Roche, Mannheim, Germany) with Oligo-dT primers following directions from the user manual. PCR was performed with $2 \times$ Hieff® $®$ PCR Master Mix (YEASEN, Shanghai, China)
with cDNA. Primers for RT-PCR were listed in Supporting Information Table S19.

### 4.4. Overexpression of transcription factor danS in $P$. dangeardii

The overexpression cassette was constructed in plasmid pRS423 including a $g p d A$ gene which as a strong promoter replaced the native promoter in transcription factor danS, an upstream ( 2.5 kb ) and a downstream ( 2.5 kb ) homologous regions of the native promoter, and a selection marker $h p h$. The genes of $g p d A$ and $h p h$, upstream and downstream homologous regions were amplified with the primers of OETF52uFor/OETF52uRev, hphFor/hphRev, gpdaFor/gpdaRev and OETF52dFor/OETF52dRev. Primers for overexpression of transcription factor danS were listed in Supporting Information Table S20. The yeast recombinant in vitro, preparation of the protoplasts and transformation and screening were performed as above described in gene knockout.

### 4.5. General experimental procedures

Optical rotations were measured in MeOH using a JASCO P-2000 spectropolarimeter (Tokyo, Japan). IR spectra were recorded using a Thermo Nicolet 5700 FT-IR spectrophotometer (Waltham, MA, USA). UV spectra were determined with JASCO V-650 instrument. High-resolution electrospray ionization mass spectra (HRESIMS) were carried out using an Agilent Technologies 6520 Accurate Mass Q-TOF LC/MS spectrometer (Santa Clara, CA, USA). The CD spectra were obtained on a JASCO J-815 spectrometer. Medium pressure liquid chromatography (MPLC) was performed using ISCO-CombiFlash®Rf 200 (Teledyne Isco, Lincoln, NE, USA) with different types C18 column. Semipreparative HPLC was carried out on a SSI instrument (Scientific Systems Inc., Philadelphia, PA, USA) with a Series 1500 PDA detector and using a YMC-Pack ODS-A column (YMC, $5 \mu \mathrm{~m}$, $250 \mathrm{~mm} \times 10 \mathrm{~mm}$ ) or a phenyl-hexyl column (phenomenex, $5 \mu \mathrm{~m}$, $250 \mathrm{~mm} \times 10 \mathrm{~mm}$ ). Column chromatography (CC) was performed using middle chromatogram isolated (MCI) gel CHP 20P/P120 (Mistubishi Chemical Co., Tokyo, Japan) and Sephadex LH-20 (GE Healthcare Bio-Sciences AB, Uppsala, Sweden). Thin layer chromatography (TLC) was performed on GF254 (Qingdao Marine Chemical Factory, Qingdao, China). All LC-MS analyses were carried out on a Waters ACQUITY H-Class UPLC-MS (Milford, MA, USA) with QDA mass detector (ACQUITY UPLC® ${ }^{8} \mathrm{BEH}, 1.7 \mu \mathrm{~m}, 50 \mathrm{~mm} \times 2.1 \mathrm{~mm}, \mathrm{C} 18$ column) using positive and negative mode electrospray ionization with a linear gradient of $5 \%-99 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$ ( $\mathrm{v} / \mathrm{v}, 0.02 \%$ formic acid) in 8 min followed by $99 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(v / v, 0.02 \%$ formic acid) for 4 min with a flow rate of $0.4 \mathrm{~mL} / \mathrm{min} 1 \mathrm{D}$ and 2D NMR spectra were determined with Bruker AVIII-600 spectrometer (Rheinstetten, Germany) with a 5 mm dual cryoprobe or a Bruker AVIII500 spectrometer or Bruker AVANCE III 800 MHz with a 5 mm broadband probe. All solvents and chemicals used in this study are of analytical grade (for extraction) or LC-MS grade (for LC-MS analysis).

### 4.6. Analysis, extract, and isolation of metabolites

For metabolites analysis, spores of wild type and mutant strains were inoculated in different culture media and stationarily cultured for 7 days at $28^{\circ} \mathrm{C}$. The mycelia and medium were extracted with ethyl acetate (EtOAc) and analyzed by UPLC-MS.

The strain was incubated on PDA at $28^{\circ} \mathrm{C}$ for 6 days to afford spores and the spores were inoculated into the PDB at $28^{\circ} \mathrm{C}$ for 4 days with 180 rpm (Zhicheng ZWY-103 B) shake to prepare the seed culture. 5 mL seed culture was inoculated into 130 Erlenmeyer flasks ( 500 mL ), previously sterilized by autoclaving, each containing 100 g rice and 100 mL distilled water. All flasks were incubated at $28{ }^{\circ} \mathrm{C}$ for 30 days. The culture was extracted three times with EtOAc at room temperature and the EtOAc was removed under reduced pressure to yield a crude extract. The crude extract was dissolved with petroleum ether (PE) to remove the fatty oil ( 27 g ) and the residue which is insolvable in PE was evaporated to dryness and separated by MCI CC with a gradient of $\mathrm{H}_{2} \mathrm{O}, 30 \%, 70 \%$ and $100 \% \mathrm{MeOH}$, acetone to obtain five fractions (Frs. A-E).

Fr. B ( 5 g ) was further separated with MCI CC eluted with a gradient of $10 \%, 20 \%, 50 \%$ and $100 \% \mathrm{MeOH}$ to obtain four fractions (B1-B4). Fr. B4 (1.1 g) was applied to a Sephadex LH20 CC eluted with MeOH to obtain 10 fractions (Frs. B4a-j), based on TLC analysis.

Fr. B4e was purified by semi-preparative HPLC on phenylhexyl column eluted with $29 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$ ( $\mathrm{v} / \mathrm{v}, 0.02 \%$ formic acid) at a flow rate of $2.0 \mathrm{~mL} / \mathrm{min}$ to yield $6(5.0 \mathrm{mg}$, $\left.t_{\mathrm{R}}=42.0 \mathrm{~min}\right), 8\left(2.5 \mathrm{mg}, t_{\mathrm{R}}=50.0 \mathrm{~min}\right), 9(2.4 \mathrm{mg}$, $\left.t_{\mathrm{R}}=26.0 \mathrm{~min}\right)$ and $15\left(4.0 \mathrm{mg}, t_{\mathrm{R}}=34.0 \mathrm{~min}\right)$.

Fr. B4f was purified by semi-preparative HPLC on phenylhexyl column eluted with $34 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$ ( $v / v, 0.02 \%$ formic acid) at a flow rate of $2.0 \mathrm{~mL} / \mathrm{min}$ to afford $\mathbf{1 8}(5.0 \mathrm{mg}$, $\left.t_{\mathrm{R}}=23.0 \mathrm{~min}\right)$.

Fr. C ( 12 g ) was divided into two parts and separated by MPLC eluted with a linear gradient of $5 \%-95 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}(v / v, 0.02 \%$ formic acid) in 130 min and $100 \% \mathrm{MeOH}$ for 30 min at a flow rate of $50 \mathrm{~mL} / \mathrm{min}$ on C18 column (SepaFlash C18, 20-45 $\mu \mathrm{m}$, SW330, Changzhou, China) to give twenty fractions (Frs. C1-20).

Fr. C7 (340 mg) was separated on Sephadex LH-20 CC (50\% $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ ) to obtain 16 additional fractions (Frs. C7a-p). Then Fr. C7m ( 70 mg ) purified by semi-preparative HPLC on phenylhexyl column ( $2.0 \mathrm{~mL} / \mathrm{min}, 29 \% \mathrm{MeCN}, \mathrm{v} / \mathrm{v} 0.02 \%$ formic acid) to yield $5\left(9 \mathrm{mg}, t_{\mathrm{R}}=32 \mathrm{~min}\right), \mathbf{1 4}\left(9 \mathrm{mg}, t_{\mathrm{R}}=27 \mathrm{~min}\right)$ and Fr . C7o ( 37 mg ) purified by semi-preparative HPLC on C18 column ( $2.0 \mathrm{~mL} / \mathrm{min}, \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, 36 \%, v / v 0.02 \%$ formic acid) to yield $17\left(4.4 \mathrm{mg}, t_{\mathrm{R}}=26 \mathrm{~min}\right)$.

Fr. C9 ( 888 mg ) was separated on Sephadex LH-20 CC ( $50 \%$ $\mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}$ ) to obtain 11 additional fractions (Frs. C9a-k). Fr. C9b ( 590 mg ) was further purified by semi-preparative HPLC on C 18 column eluted with $44 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}(v / v, 0.02 \%$ formic acid) at a flow rate of $5.0 \mathrm{~mL} / \mathrm{min}$ to yield $39(150 \mathrm{mg}$, $t_{\mathrm{R}}=73.0 \mathrm{~min}$ ).

Fr. C10 ( 440 mg ) was subjected to a Sephadex LH-20 CC eluted with MeOH to yield 6 subfractions (Frs. C10a-f), based on TLC analysis. Fr. C10a ( 43 mg ) was purified by semi-preparative HPLC on C18 column eluted with $37 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(v / v, 0.02 \%$ formic acid) at a flow rate of $2.0 \mathrm{~mL} / \mathrm{min}$ to yield $38(7.7 \mathrm{mg}$, $\left.t_{\mathrm{R}}=17.0 \mathrm{~min}\right)$.

Fr. C11 ( 67 mg ) was further purified by semi-preparative HPLC on C18 column eluting with $35 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(v / v, 0.02 \%$ formic acid) at a flow rate of $2.0 \mathrm{~mL} / \mathrm{min}$ to yield $11(2.2 \mathrm{mg}$, $\left.t_{\mathrm{R}}=23.0 \mathrm{~min}\right), 12\left(15.5 \mathrm{mg}, t_{\mathrm{R}}=26.0 \mathrm{~min}\right), \mathbf{3 7}(20 \mathrm{mg}$, $t_{\mathrm{R}}=11.0 \mathrm{~min}$ ), respectively.

Fr. C12 (1.24 g) was separated on a Sephadex LH-20 CC $(\mathrm{MeOH})$ to obtain 15 additional fractions (Frs. $\mathrm{C} 12 \mathrm{a}-\mathrm{o}$ ) and then Fr. C12j ( 154 mg ) purified by semi-preparative HPLC on

C 18 column ( $5.0 \mathrm{~mL} / \mathrm{min}, 54 \% \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}, v / \mathrm{v} 0.02 \%$ formic acid) to yield $25\left(48.2 \mathrm{mg}, t_{\mathrm{R}}=54.0 \mathrm{~min}\right), 35(2.2 \mathrm{mg}$, $t_{\mathrm{R}}=95.0 \mathrm{~min}$ ). Fr. C12k ( 88 mg ) purified by semi-preparative HPLC on C 18 column ( $2.0 \mathrm{~mL} / \mathrm{min}, 38 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v$ $0.02 \%$ formic acid) to yield $26\left(36 \mathrm{mg}, t_{\mathrm{R}}=36.5 \mathrm{~min}\right)$. Fr. C121 ( 33 mg ) purified by semi-preparative HPLC on C18 column ( $2.0 \mathrm{~mL} / \mathrm{min}, 41 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%$ formic acid) to yield $28\left(9.8 \mathrm{mg}, t_{\mathrm{R}}=41 \mathrm{~min}\right)$. Fr. C12n ( 33 mg ) purified by semi-preparative HPLC on C18 column ( $2.0 \mathrm{~mL} / \mathrm{min}, 42 \%$ $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%$ formic acid) to yield 27 ( 4.1 mg , $\left.t_{\mathrm{R}}=44 \mathrm{~min}\right)$.

Fr. C13 (1.15 g) was applied to a Sephadex LH-20 column eluted with MeOH to give 11 subfractions (Frs. C13a-k), based on TLC analysis. Then Subfraction Fr. C13f ( 45 mg ) was purified by semi-preparative HPLC on C18 column eluted with $42 \%$ $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}(\mathrm{v} / \mathrm{v}, 0.02 \%$ formic acid) at a flow rate of $2.0 \mathrm{~mL} / \mathrm{min}$ to yield $19\left(1.0 \mathrm{mg}, t_{\mathrm{R}}=19.0 \mathrm{~min}\right)$. Fr. C13g ( 247 mg ) was purified by semi-preparative HPLC on phenyl-hexyl column eluted with $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}\left(2.0 \mathrm{~mL} / \mathrm{min}, 47 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%\right.$ formic acid) to yield $\mathbf{1 0}\left(10.2 \mathrm{mg}, t_{\mathrm{R}}=27 \mathrm{~min}\right), \mathbf{1 6}(11.7 \mathrm{mg}$, $t_{\mathrm{R}}=35 \mathrm{~min}$ ).

Fr. C14 ( 1.0 g ) was subjected to a Sephadex LH-20 column eluted with MeOH to give 14 subfractions (Frs. C14a-n), based on TLC analysis. Fr. C141 ( 120 mg ) was further separated by Sephadex LH-20 CC eluted with MeOH to give 6 subfractions (Frs. C1411-6). Fr. C1414 (29 mg) was purified by semipreparative HPLC on C18 column ( $2.0 \mathrm{~mL} / \mathrm{min}, 45 \%-48 \%$ $\mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O} 40 \mathrm{~min}, 48 \%-60 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O} 10 \mathrm{~min}$, $v / v 0.02 \%$ formic acid) to yield 24 ( $3.5 \mathrm{mg}, t_{\mathrm{R}}=15 \mathrm{~min}$ ). Fr. C1416 ( 12 mg ) was purified by semi-preparative HPLC on C18 column ( $2.0 \mathrm{~mL} / \mathrm{min}, 46 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%$ formic acid) to yield 22 $\left(8.0 \mathrm{mg}, t_{\mathrm{R}}=34 \mathrm{~min}\right)$.

Fr. C15 (483 mg) was applied to Sephadex LH-20 CC eluted with MeOH to obtain eight subfractions (Frs. C15a-h). Fr. C15f ( 75 mg ) was purified by semi-preparative HPLC on phenyl-hexyl column ( $2.0 \mathrm{~mL} / \mathrm{min}, 46 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%$ formic acid) to yield $\mathbf{3 1}\left(3.5 \mathrm{mg}, t_{\mathrm{R}}=33 \mathrm{~min}\right), \mathbf{3 3}\left(3.0 \mathrm{mg}, t_{\mathrm{R}}=39 \mathrm{~min}\right)$, and $\mathbf{3 2}$ $\left(16.4 \mathrm{mg}, t_{\mathrm{R}}=43 \mathrm{~min}\right)$.

Fr. C17 (293 mg) was subjected to Sephadex LH-20 CC eluted with MeOH to get 6 subfractions (Frs. C17a-f). Fr. C17d ( 41 mg ) was purified by semi-preparative HPLC on phenyl-hexyl column ( $2.0 \mathrm{~mL} / \mathrm{min}, 50 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%$ formic acid) to yield 29 ( $3.9 \mathrm{mg}, t_{\mathrm{R}}=41 \mathrm{~min}$ ) and $30\left(6.2 \mathrm{mg}, t_{\mathrm{R}}=52 \mathrm{~min}\right)$.

Fr. C18 ( 481 mg ) was further separated by Sephadex LH-20 CC eluted with MeOH to obtain eight subfractions (Frs. C18a-h). Fr. C18f ( 42 mg ) was purified by semi-preparative HPLC on phenyl-hexyl column $\left(2.0 \mathrm{~mL} / \mathrm{min}, 53 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%\right.$ formic acid) to yield $21\left(11.2 \mathrm{mg}, t_{\mathrm{R}}=43 \mathrm{~min}\right)$ and $23(4.4 \mathrm{mg}$, $\left.t_{\mathrm{R}}=33 \mathrm{~min}\right)$. Fr. C18g ( 22 mg ) was purified by semi-preparative HPLC on phenyl-hexyl column ( $2.0 \mathrm{~mL} / \mathrm{min}, 55 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$, $v / v 0.02 \%$ formic acid) to yield $34\left(1.5 \mathrm{mg}, t_{\mathrm{R}}=27 \mathrm{~min}\right)$.

Fr. C20 ( 275 mg ) was purified by semi-preparative HPLC on phenyl-hexyl column $\left(2.0 \mathrm{~mL} / \mathrm{min}, 60 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%\right.$ formic acid) to yield $20\left(13 \mathrm{mg}, t_{\mathrm{R}}=47 \mathrm{~min}\right)$.

Fr. D ( 2.0 g ) was applied to Sephadex LH-20 CC eluted with MeOH to obtain 14 subfractions (Frs. D1-14). Fr. D7 (80 mg) was purified by semi-preparative HPLC on phenyl-hexyl column ( $2.0 \mathrm{~mL} / \mathrm{min}, 47 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}, v / v 0.02 \%$ formic acid) to yield 4 $\left(4.0 \mathrm{mg}, t_{\mathrm{R}}=31 \mathrm{~min}\right), 7\left(0.9 \mathrm{mg}, t_{\mathrm{R}}=24 \mathrm{~min}\right)$, and $\mathbf{1 3}(0.9 \mathrm{mg}$, $\left.t_{\mathrm{R}}=26 \mathrm{~min}\right)$. Fr. D12 ( 80 mg ) was purified by semi-preparative HPLC on phenyl-hexyl column ( $2.0 \mathrm{~mL} / \mathrm{min}, 48 \% \mathrm{MeCN} / \mathrm{H}_{2} \mathrm{O}$, $v / v 0.02 \%$ formic acid) to yield $36\left(2.0 \mathrm{mg}, t_{\mathrm{R}}=31 \mathrm{~min}\right)$.
4.6.1. Dangelone A (4)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-181.3(c 0.8, \mathrm{MeOH})$; $\mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon) 215$ (4.28), 265 (4.03), 346 (4.21) nm. IR (KBr) $\nu_{\max } 3355$, $2918,1645,1583,1447,1255,1204,1169,1104,962,797 \mathrm{~cm}^{-1}$; $\mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 227(-3.29), 259(+0.49), 275(-0.31), 306$ $(+0.57), 360(-1.61) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR $\left(800 \mathrm{MHz}\right.$, in DMSO- $\left.d_{6}\right)$ and ${ }^{13} \mathrm{C}$ NMR ( 200 MHz , in DMSO- $d_{6}$ ), Tables S4 and S5. ESI-MS $\mathrm{m} / \mathrm{z} 387.16[\mathrm{M}+\mathrm{H}]^{+}$and $385.34[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 387.1438 $[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{O}_{7}, 387.1438$.

### 4.6.2. Dangelone B (5)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-132.6(c 0.4, \mathrm{MeOH})$; $\mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ 211, 262, 346 nm . IR (KBr) $\nu_{\max } 3305,2978,1649,1585,1456$, $1257 \mathrm{~cm}^{-1} ; \mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 226(-7.4), 249(+0.98), 272$ $(-1.45), 307(+0.71), 359(-3.76) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR ( 800 MHz , in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 200 MHz , in DMSO- $d_{6}$ ), Tables S4 and S5. ESI-MS m/z $403.38[\mathrm{M}+\mathrm{H}]^{+}$and $401.32[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $403.1387[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{O}_{8}, 403.1387$.

### 4.6.3. Dangelone C(6)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-173.7(c 0.8, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon) 215$ (4.14), 265 (3.96), 347 (4.10) nm. IR (KBr) $\nu_{\max } 3295$, $2917,1649,1591,1257 \mathrm{~cm}^{-1} ; \mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 230(-2.0)$, $255(+0.91), 275(-0.27), 358(-1.63) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR $(800 \mathrm{MHz}$, in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 200 MHz , in DMSO- $d_{6}$ ), Tables S4 and S5. ESI-MS $m / z 417.17[\mathrm{M}+\mathrm{H}]^{+}$and $415.23[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $417.1190[\mathrm{M}+\mathrm{H}]^{+}$Calcd. for $\mathrm{C}_{21} \mathrm{H}_{21} \mathrm{O}_{9}, 417.1180$.

### 4.6.4. Dangelone $D$ (7)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}+52.9(c 0.3, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max } 215$, 263, 349 nm . IR (KBr) $\nu_{\max } 3386,2934,1716,1647,1619,1585$, $1447,1381,1255,1202,1171,1054 \mathrm{~cm}^{-1} ; \mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon)$ : $210(+2.27), 225(-0.35), 258(+1.92), 301(-3.95), 331$ $(+1.63), 361(+1.51) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}\right.$, in acetone- $\left.d_{6}\right)$ and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in acetone- $d_{6}$ ), Tables S4 and S5. ESI-MS $m / z 445.22[\mathrm{M}+\mathrm{H}]^{+}$and $443.38[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 467.1328 $[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{O}_{9} \mathrm{Na}, 467.1313$.
4.6.5. Dangelone $E(8)$

Yellow powder $[\alpha]_{\mathrm{D}}^{20}+43.1(c \quad 1.4, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\text {max }}$ $(\log \varepsilon) 214$ (4.03), 268 (3.77), 314 (3.87) nm. IR (KBr) $\nu_{\max } 3325$, $2918,1712,1644,1588,1449,1255,1168 \mathrm{~cm}^{-1} ; \mathrm{CD}(\mathrm{MeOH})$ $\lambda_{\max }(\varepsilon): 230(-1.52), 266(+2.67), 296(-1.11), 320(+1.72)$, $352(+1.38) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR ( 600 MHz , in acetone- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in acetone- $d_{6}$ ), Tables S4 and S5. ESI-MS $\mathrm{m} / \mathrm{z}$. $419.17[\mathrm{M}+\mathrm{H}]^{+}, 417.21 \quad[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 419.1352 $[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{O}_{9}, 419.1337$.

### 4.6.6. $\quad$ Dangelone $F(9)$

Yellow powder $[\alpha]_{\mathrm{D}}^{20}+60.7(c 1.9, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\text {max }}$ $(\log \varepsilon) 215$ (4.13), 266 (3.92), 308 (3.85) nm. IR (KBr) $\nu_{\max } 3363$, $2975,1696,1645,1450,1259,1204 \mathrm{~cm}^{-1} ; \mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon)$ : $212(+1.87), 254(+1.95), 288(-2.11), 320(+2.21), 374(-0.18)$ $\mathrm{nm} .{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}\right.$, in acetone- $\left.d_{6}\right)$ and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in acetone- $d_{6}$ ), Tables S4 and S5. ESI-MS $m / z 441.16[\mathrm{M}+\mathrm{Na}]^{+}$ and $417.26[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $441.1138[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{21} \mathrm{H}_{22} \mathrm{O}_{9} \mathrm{Na}, 441.1156$.

### 4.6.7. Dangeloside $A$ (18)

Light yellow powder $[\alpha]_{\mathrm{D}}^{20}-52.9(c 0.2, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH})$ $\lambda_{\max }(\log \varepsilon) 206$ (4.04), 264 (3.51), 346 (3.02) nm. IR (KBr) $\nu_{\max }$ 3383, 2920, 1647, 1614, 1419, 1257, $1203 \mathrm{~cm}^{-1}$; CD (MeOH)
$\lambda_{\max }(\varepsilon): 209(-3.65), 232(-0.46), 255(+1.07), 276(-0.45)$, $305(+0.64), 358(-2.58) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}\right.$, in acetone- $\left.d_{6}\right)$ and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in acetone- $d_{6}$ ), Table S6. ESI-MS $\mathrm{m} / \mathrm{z}$ $549.17[\mathrm{M}+\mathrm{H}]^{+}$and $547.23[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 571.1429 $[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{26} \mathrm{H}_{28} \mathrm{O}_{13} \mathrm{Na}, 571.1422$.

### 4.6.8. Dangeloside $B$ (19)

Light yellow powder $[\alpha]_{\mathrm{D}}^{20}+134.5(c 0.6, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH})$ $\lambda_{\text {max }}(\log \varepsilon) 214$ (3.73), 262 (3.42), 348 (3.55) nm. IR (KBr) $\nu_{\text {max }}$ $3409,2924,1722,1650,1582,1453,1257,1051 \mathrm{~cm}^{-1} ; \mathrm{CD}$ $(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 214(+5.25), 262(+5.37), 305(-2.29), 372$ $(+4.04) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR ( 500 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 125 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ), Table S6. ESI-MS m/z $519.17[\mathrm{M}+\mathrm{H}]^{+}$ and $517.23[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $519.1868[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{26} \mathrm{H}_{31} \mathrm{O}_{11}, 519.1861$.

### 4.6.9. Didangelone A (20)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-299.4(c 0.2, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon) 216$ (4.53), 267 (4.43), 300 (4.22), 361 (4.25) nm. IR (KBr) $\nu_{\max } 3367,2918,1716,1620,1587,1449,1262,1169 \mathrm{~cm}^{-1} ; \mathrm{CD}$ $(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 216(-5.56), 262(+9.64), 295(+11.63), 352$ $(+8.52), 402(-8.35) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}\right.$, in $\left.\mathrm{CD}_{3} \mathrm{OD}\right)$ and ${ }^{13} \mathrm{C}$ NMR ( 125 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ), Tables S7 and S9. ESI-MS $\mathrm{m} / \mathrm{z}$ $777.44[\mathrm{M}+\mathrm{H}]^{+} 775.41[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $777.2205[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{43} \mathrm{H}_{37} \mathrm{O}_{14}, 777.2178$.

### 4.6.10. Didangelone $B$ (21)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-226.2(c 0.2, \mathrm{MeOH})$; $\mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon) 215$ (4.56), 267 (4.43), 299 (4.22), 360 (4.24) nm. IR (KBr) $\nu_{\max } 3349,3249,2928,2854,1716,1613,1442,1262,1165 \mathrm{~cm}^{-1}$; $\mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 215(-2.08), 262(+6.57), 295(+10.50)$, 328 (-3.76), 356 (+3.99), 404 ( -5.42 ) nm. ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , in $\left.\mathrm{CD}_{3} \mathrm{OD}\right)$ and ${ }^{13} \mathrm{C}$ NMR ( 125 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ), Tables S 7 and S9. ESI-MS m/z $793.29[\mathrm{M}+\mathrm{H}]^{+}$and $791.74[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 793.2121 $[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{43} \mathrm{H}_{37} \mathrm{O}_{15}, 793.2127$.

### 4.6.11. Didangelone $C$ (22)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-261.8(c 0.9, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon) 211$ (4.43), 265 (4.25), 295 (4.02), 358 (3.98) nm. IR (KBr) $\nu_{\max } 3358,2919,1715,1619,1449,1262,1170 \mathrm{~cm}^{-1} ; \mathrm{CD}$ $(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 221(+2.73), 244(-0.53), 295(+5.30), 331$ ( -7.43 ), $389(-4.55) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}\right.$, in DMSO- $\left.d_{6}\right)$ and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in DMSO- $d_{6}$ ), Tables S7 and S9. ESI-MS $m / z 809.36[\mathrm{M}+\mathrm{H}]^{+}$and $807.53[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 831.1918 $[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{43} \mathrm{H}_{36} \mathrm{O}_{16} \mathrm{Na}, 831.1896$.

### 4.6.12. Didangelone $D(23)$

Red powder $[\alpha]_{\mathrm{D}}^{20}-333.7(c 1.6, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon)$ 208 (4.28), 266 (4.05), 301 (3.89), 364 (3.71) nm. IR (KBr) $\nu_{\max }$ $3281,2921,1713,1620,1448,1323,1262,1168 \mathrm{~cm}^{-1}$; CD $(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 214(-6.24), 234(+3.45), 295(+16.97), 350$ $(+6.50), 391(-12.70) \mathrm{nm} .{ }^{1} \mathrm{H} \operatorname{NMR}\left(600 \mathrm{MHz}\right.$, in $\left.\mathrm{CD}_{3} \mathrm{OD}\right)$ and ${ }^{13} \mathrm{C}$ NMR $\left(150 \mathrm{MHz}\right.$, in $\left.\mathrm{CD}_{3} \mathrm{OD}\right)$, Tables 57 and S9. ESI-MS $\mathrm{m} / \mathrm{z}$ $776.34[\mathrm{M}+\mathrm{H}]^{+}$and $774.45[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 776.2358 $[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{43} \mathrm{H}_{38} \mathrm{NO}_{13}, 776.2338$.

### 4.6.13. Didangelone $E$ (24)

Red powder $[\alpha]_{\mathrm{D}}^{20}-248.6(c 0.2, \mathrm{MeOH}) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon)$ 215 (4.53), 267 (4.33), 300 (4.16), 363 (3.96) nm. IR (KBr) $\nu_{\max }$ $3275,2921,1710,1618,1595,1452,1322,1264 \mathrm{~cm}^{-1} ; \mathrm{CD}$ $(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 212(-2.33), 232(+1.31), 296(+6.31), 387$ $(-5.60) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR ( 600 MHz , in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR
( 150 MHz , in DMSO- $d_{6}$ ), Tables S7 and S9. ESI-MS $m / z 808.30$ $[\mathrm{M}+\mathrm{H}]^{+}$and $806.49[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $808.2291[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{43} \mathrm{H}_{38} \mathrm{NO}_{15}, 808.2295$.
4.6.14. Didangelone $H$ (25)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-365.4$ (c $\left.0.5, \mathrm{MeOH}\right)$; $\mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }$ $(\log \varepsilon) 216$ (4.75), 267 (4.44), 333 (4.43) nm. IR (KBr) $\nu_{\text {max }} 3258$, 2920, 1719, 1621, 1545, 1451, 1323, $1262 \mathrm{~cm}^{-1}$; CD (MeOH) $\lambda_{\max }(\varepsilon): 232(+14.07), 303(+48.21), 347(-64.22) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR ( 500 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 125 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ), Tables S8 and S9. ESI-MS $m / z 811.40[\mathrm{M}+\mathrm{H}]^{+}$and 809.38 $[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $811.1860[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{42} \mathrm{H}_{35} \mathrm{O}_{17}$, 811.1869.

### 4.6.15. Didangelone $F$ (27)

Light yellow powder $[\alpha]_{\mathrm{D}}^{20}-259.5$ (c 0.3, MeOH); UV (MeOH) $\lambda_{\max }(\log \varepsilon) 214$ (4.34), 269 (4.04), 307 (3.97) nm. IR (KBr) $\nu_{\max }$ 3286, 2919, 2852, 1717, 1620, 1452, 1261, $1110 \mathrm{~cm}^{-1}$; CD $(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 223(+10.76), 295(+21.75), 335(-25.94) \mathrm{nm}$. ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ), Tables S8 and S9. ESI-MS $\mathrm{m} / \mathrm{z} 811.41[\mathrm{M}+\mathrm{H}]^{+}$and $809.34[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $811.1829[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{42} \mathrm{H}_{35} \mathrm{O}_{17}, 811.1869$.
4.6.16. Didangelone $G$ (28)

Light yellow powder $[\alpha]_{\mathrm{D}}^{20}-182.5(c 0.8, \mathrm{MeOH})$; UV (MeOH) $\lambda_{\text {max }}(\log \varepsilon) 215$ (4.49), 268 (4.19), 307 (4.10) nm. IR (KBr) $\nu_{\text {max }}$ 3363, 2920, 2851, 1715, 1620, 1452, 1324, 1263, $1171 \mathrm{~cm}^{-1}$; CD $(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 211(+7.91), 268(+12.63), 302(+19.28), 336$ $(-45.34) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR ( 600 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ), Tables S8 and S9. ESI-MS m/z 811.38 $[\mathrm{M}+\mathrm{H}]^{+}$and $809.37[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $833.1708[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{42} \mathrm{H}_{34} \mathrm{O}_{17} \mathrm{Na}$, 833.1688.

### 4.6.17. Tridangelone $A$ (29)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-102.1$ (c 0.4, MeOH); UV (MeOH) $\lambda_{\text {max }}$ ( $\log \varepsilon$ ) 214 (4.54), 268 (4.29), 305 (4.16), 346 (4.01) nm. IR (KBr) $\nu_{\max } 3357,3288,2917,1715,1626,1449,1323,1262,1169 \mathrm{~cm}^{-1}$; $\mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 227(+16.86), 255(-0.33), 295(+26.64)$, 342 ( -31.29 ) nm. ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in DMSO- $d_{6}$ ), Tables S10 and S11. ESI-MS $\mathrm{m} / \mathrm{z}$ $1205.38[\mathrm{M}+\mathrm{H}]^{+}$and $1203.54[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 1205.2882 $[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{64} \mathrm{H}_{53} \mathrm{O}_{24}, 1205.2921$.

### 4.6.18. Tridangelone B (30)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-167.4$ (c $\left.1.8, \mathrm{MeOH}\right)$; UV $(\mathrm{MeOH}) \lambda_{\text {max }}$ $(\log \varepsilon) 214$ (4.66), 268 (4.44), 305 (4.32), 347 (4.17) nm. IR (KBr) $\nu_{\max } 3362,2920,2852,1715,1621,1449,1324,1262,1168$, $1105 \mathrm{~cm}^{-1} ; \mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 209(+11.07), 245(-2.09), 306$ $(+27.55), 345(-27.86) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR ( 600 MHz , in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in DMSO- $d_{6}$ ), Tables S10 and S11. ESIMS $m / z$ 1205.43 $[\mathrm{M}+\mathrm{H}]^{+}$and $1203.54[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $1227.2653[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{64} \mathrm{H}_{52} \mathrm{O}_{24} \mathrm{Na}, 1227.2682$.

### 4.6.19. Tridangelone C (31)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-143.2$ (c 0.4, MeOH); UV (MeOH) $\lambda_{\text {max }}$ ( $\log \varepsilon$ ) 205 (4.45), 269 (4.15), 308 (4.02), 349 (3.87) nm. IR (KBr) $\nu_{\max } 3394,2920,2851,1734,1645,1466,1325,1262,1103$, $801 \mathrm{~cm}^{-1} ; \mathrm{CD}(\mathrm{MeOH}) \lambda_{\text {max }}(\varepsilon): 227$ (+25.79), 294 (+33.52), 340 (-39.96) nm. ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in DMSO- $d_{6}$ ), Tables S10 and S11. ESI-MS $\mathrm{m} / \mathrm{z}$
$1221.33[\mathrm{M}+\mathrm{H}]^{+}$and $1219.40[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 1221.2883 $[\mathrm{M}+\mathrm{H}]^{+}$, Calcd. for $\mathrm{C}_{64} \mathrm{H}_{53} \mathrm{O}_{25}, 1221.2870$.

### 4.6.20. Tridangelone $D$ (32)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-203.8$ ( c 1.3, MeOH); UV $(\mathrm{MeOH}) \lambda_{\text {max }}$ $(\log \varepsilon) 215$ (4.59), 268 (4.37), 307 (4.24), 348 (4.10) nm. IR (KBr) $\nu_{\text {max }} 3247,2920,2851,1619,1449,1324,1262,1169 \mathrm{~cm}^{-1}$; CD $(\mathrm{MeOH}) \lambda_{\text {max }}(\varepsilon): 211(+8.40), 246(-1.68), 306(+22.93), 344$ (-24.60) nm. ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in DMSO- $d_{6}$ ), Tables S10 and S11. ESI-MS $\mathrm{m} / \mathrm{z}$ $1221.26[\mathrm{M}+\mathrm{H}]^{+}$and $1219.39[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 1243.2625 $[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{64} \mathrm{H}_{52} \mathrm{O}_{25} \mathrm{Na}, 1243.2690$.

### 4.6.21. Tridangelone E (33)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-191.3$ (c 1.2, MeOH); UV (MeOH) $\lambda_{\text {max }}$ $(\log \varepsilon) 214$ (4.43), 268 (4.21), 305 (4.06), 346 (3.90) nm. IR (KBr) $\nu_{\text {max }} 3328,2269,2917,1714,1619,1450,1323,1261,1104 \mathrm{~cm}^{-1}$; $\mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 226(+16.04), 258(-0.79), 296(+20.80)$, 339 (-29.02) nm. ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in DMSO- $d_{6}$ ), Tables S10 and S11. ESI-MS $m / z$ $1207.42[\mathrm{M}+\mathrm{H}]^{+}$and $1205.57[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: 1229.2950 $[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{64} \mathrm{H}_{54} \mathrm{O}_{24} \mathrm{Na}, 1229.2897$.

### 4.6.22. Dangelone G (34)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-186.3$ (c $\left.0.9, \mathrm{MeOH}\right)$; UV $(\mathrm{MeOH}) \lambda_{\text {max }}$ $(\log \varepsilon) 216$ (4.15), 262 (3.96), 301 (3.70), 355 (3.63) nm. IR (KBr) $\nu_{\max } 3199,2924,2851,1715,1642,1619,1449,1325,1261 \mathrm{~cm}^{-1}$; $\mathrm{CD}(\mathrm{MeOH}) \lambda_{\max }(\varepsilon): 228(+3.62), 255(-2.77), 300(+4.78), 352$ (-4.69) nm. ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , in DMSO- $d_{6}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in DMSO- $d_{6}$ ), Table S12. ESI-MS $\mathrm{m} / \mathrm{z} 563.12$ $[\mathrm{M}+\mathrm{H}]^{+}$and $561.23[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $585.1382[\mathrm{M}+\mathrm{Na}]^{+}$, Calcd. for $\mathrm{C}_{30} \mathrm{H}_{26} \mathrm{O}_{11} \mathrm{Na}, 585.1367$.

### 4.6.23. Dangelone $H$ (35)

Yellow powder $[\alpha]_{\mathrm{D}}^{20}-54.1(c \quad 0.9, \mathrm{MeOH})$; $\mathrm{UV}(\mathrm{MeOH}) \lambda_{\text {max }}$ $(\log \varepsilon) 205$ (4.15), 270 (3.60), 300 (3.56) nm. IR (KBr) $\nu_{\text {max }} 3331$, 2918, 1730, 1618, 1455, 1261, $1102 \mathrm{~cm}^{-1} ; \mathrm{CD}(\mathrm{MeOH}) \lambda_{\text {max }}(\varepsilon)$ : $217(-2.00), 247(+0.26), 301(+1.30), 342(-2.05) \mathrm{nm} .{ }^{1} \mathrm{H}$ NMR ( 600 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ) and ${ }^{13} \mathrm{C}$ NMR ( 150 MHz , in $\mathrm{CD}_{3} \mathrm{OD}$ ), Table S12. ESI-MS $\mathrm{m} / \mathrm{z} 353.25[\mathrm{M}-\mathrm{H}]^{-}$. HRESIMS: $353.1032[\mathrm{M}-\mathrm{H}]^{-}$, Calcd. for $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{O}_{6}, 353.1030$.
4.7. ECD calculations of compounds $9,18,20,25,27,28$, and 35

Conformational analyses of compounds $\mathbf{9}, \mathbf{1 8}, \mathbf{2 0}, \mathbf{2 5}$, and $\mathbf{3 5}$ were performed via Monte Carlo searching with the MMFF94 force field using the Spartan 10 software (Wavefunction, Inc., Irvine, CA, USA). All conformers with relative energies within $9 \mathrm{kcal} / \mathrm{mol}$ were optimized for density functional theory (DFT) calculations at the B3LYP/6-31G(d) level in vacuum using the Gaussian 09 program ${ }^{36}$. Subsequently, the B3LYP/6-31G(d)optimized conformers were re-optimized at the WB97XD/ DGDZVP level in methanol. The WB97XD/DGDZVP harmonic vibrational frequency analyses confirmed the stability of the conformers. ECD computations for the WB97XD/DGDZVPoptimized conformers were performed using time-dependent DFT (TD-DFT) at the CAM-B3LYP/DGDZVP level in methanol. The predicted ECD spectra were simulated by the Gaussum 2.25 program $^{53}$. The final ECD spectra were acquired on the base of the Boltzmann distribution theory and the relative Gibbs free energy $(\Delta G)$.

Conformational analyses of compounds 27 and $\mathbf{2 8}$ were performed in Yinfo Cloud Platform (https://cloud.yinfotek.com/) using Confab ${ }^{54}$ with systematic algorithm at MMFF94 force field with RMSD threshold of $0.5 \AA$ and energy window of $7 \mathrm{kcal} / \mathrm{mol}$. The theoretical calculations were carried out using Gaussian $09^{38}$ (Revision D.01, Gaussian Inc., Wallingford, CT, USA, 2009). At first, conformers were optimized at PM6 and HF/6-31G(d) theory levels, consecutively. Room-temperature equilibrium populations were calculated on the basis of Boltzmann distribution law and dominative conformers with the value greater than $1 \%$ were saved. The chosen conformers were finally optimized at B3LYP/6$31 \mathrm{G}(\mathrm{d})$ in gas phase. Vibrational frequency analysis confirmed the stable structures. ECD calculations were conducted in methanol with IEFPCM model using TD-DFT. Rotatory strengths for 100 excited states were calculated. The ECD spectrum was simulated using the ECD/UV analysis tool in Yinfo Cloud Platform by overlapping Gaussian functions for each transition.

### 4.8. Determination of the absolute configuration of ribose moieties in compound 18

To verify the absolute configurations of ribose moiety in 18, a modified method based on UPLC-MS analysis was adopted, where the parent compounds were hydrolyzed to afford the sugar samples, the retention times of which were compared with those of standard sugars ( $\mathrm{D} / \mathrm{L})^{55}$.

Acid hydrolysis reaction: dissloving 18 ( 1.4 mg, each) in 1 mL $\mathrm{HCl}(1 \mathrm{~mol} / \mathrm{L})$ treated with $3 \mathrm{~mL} \mathrm{MeOH} / \mathrm{H}_{2} \mathrm{O}(v / v, 2: 1)$ under $70^{\circ} \mathrm{C}$ for 4 h . The reaction mixture was evaporated to dry extract, and extracted three times with EtOAc after extract dissloving in $2 \mathrm{~mL} \mathrm{H}_{2} \mathrm{O}$. The water layer was evaporated to afford a hydrolytic monosaccharide.

Preparation of monosaccharide derivatives: solutions of Lribose ( 2.0 mg ), D-ribose ( 2.0 mg ) and hydrolytic monosaccharides $(0.4 \mathrm{mg}$, each) in 1 mL pyridine were individually treated with L-cysteine methyl ester hydrochloride ( 2.0 mg ) under $60^{\circ} \mathrm{C}$ for 2 h , and then added $O$-tolyl isothiocyanate ( 2.0 mg ) in each reaction system under $60^{\circ} \mathrm{C}$ for 1 h . A blank control group was preapared at the same time.

Each reaction mixture was analyzed by UPLC-MS. As shown in Fig. S7, the retention time of hydrolyzed monosaccharide derivative was in accordance with that of D-ribose derivative. Thus, the ribose in $\mathbf{1 8}$ was determined as D-configuration.

### 4.9. Cytotoxicity assays

The cytotoxicities of compounds 4-39 against human cancer cell lines HepG-2 and MCF-7 (NCI-60, Bethesda, USA) as well as the murine microglial BV2 cell line were measured using the MTT assay according to the reported procedures ${ }^{56}$. Taxol was used as the positive control.

### 4.10. Anti-inflammatory assays

The anti-inflammatory activities of compounds 4-39 were assessed by measuring the inhibitory effects on NO production induced by lipopolysaccharide (LPS) in murine microglial BV2 cell line according to the reported procedures ${ }^{56}$. Curcumin was selected as the positive control (inhibitory rate $93.7 \pm 2.5 \%$ at $10^{-5} \mathrm{~mol} / \mathrm{L}$ ). The murine microglia BV2 cell line was purchased from the Cell Culture Center at the Institute of Basic Medical

Sciences, Chinese Academy of Medical Sciences and Peking Union Medical College (Beijing, China).

### 4.11. Antioxidant assays

The antioxidant activities of compounds 4-39 were evaluated by measuring the inhibitory rates of malondialdehyde (MDA) in rat liver microsomal lipid peroxidation induced by $\mathrm{Fe}^{2+}$-cystine in vitro according to the reported procedures ${ }^{57}$. Curcumin was selected as the positive control (inhibitory rate $91.0 \pm 3.2 \%$ at $10^{-4} \mathrm{~mol} / \mathrm{L}$ ). Rat liver microsomes are prepared from male Sprague-Dawley rats (Hua Fu Kang Bioscience, Beijing, China).

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## Author contributions

Qian Wei conducted most of the experiment and wrote the manuscript; Jian Bai analyzed data, elucidated the chemical structures, and wrote the manuscript; Daojiang Yan analyzed the BGCs of anzaphilones; Xiuqi Bao, Dan Zhang, Wenting Li and Xiangbing Qi performed anti-inflammatory, antioxidant assays and cytotoxicity assays; Bingyu Liu involved in NMR data analysis; Dequan Yu involved in discussion and design; Youcai Hu designed and guided the entire project and edited the manuscript.

## Conflicts of interest

The authors declare no conflicts of interest.

## Appendix A. Supporting information

Supporting information to this article can be found online at https://doi.org/10.1016/j.apsb.2020.07.020.

## References

1. Zhang HW, Song YC, Tan RX. Biology and chemistry of endophytes. Nat Prod Rep 2006;23:753-71.
2. Nisa H, Kamili AN, Nawchoo IA, Shafi S, Shameem N, Bandh SA. Fungal endophytes as prolific source of phytochemicals and other bioactive natural products: a review. Microb Pathog 2015;82:50-9.
3. Stierle A, Strobel G, Stierle D. Taxol and taxane production by Taxomyces andreanae, an endophytic fungus of Pacific yew. Science 1993; 260:214-6.
4. Rutledge PJ, Challis GL. Discovery of microbial natural products by activation of silent biosynthetic gene clusters. Nat Rev Microbiol 2015;13:509-23.
5. Gross H. Strategies to unravel the function of orphan biosynthesis pathways: recent examples and future prospects. Appl Microbiol Biotechnol 2007;75:267-77.
6. Scherlach K, Hertweck C. Discovery of aspoquinolones A-D, prenylated quinoline-2-one alkaloids from Aspergillus nidulans, motivated by genome mining. Org Biomol Chem 2006;4:3517-20.
7. Scherlach K, Schuemann J, Dahse HM, Hertweck C. Aspernidine A and B , prenylated isoindolinone alkaloids from the model fungus Aspergillus nidulans. J Antibiot 2010;63:375-7.
8. Schroeckh V, Scherlach K, Nutzmann HW, Shelest E, SchmidtHeck W, Schuemann J, et al. Intimate bacterial-fungal interaction triggers biosynthesis of archetypal polyketides in Aspergillus nidulans. Proc Natl Acad Sci 2009;106:14558-63.
9. Asai T, Yamamoto T, Shirata N, Taniguchi T, Monde K, Fujii I, et al. Structurally diverse chaetophenol productions induced by chemically mediated epigenetic manipulation of fungal gene expression. Org Lett 2013;15:3346-9.
10. Hosaka T, Ohnishi-Kameyama M, Muramatsu H, Murakami K, Tsurumi Y, Kodani S, et al. Antibacterial discovery in actinomycetes strains with mutations in RNA polymerase or ribosomal protein S12. Nat Biotechnol 2009;27:462-4.
11. Mao XM, Xu W, Li D, Yin WB, Chooi YH, Li YQ, et al. Epigenetic genome mining of an endophytic fungus leads to the pleiotropic biosynthesis of natural products. Angew Chem Int Ed 2015;54: 7592-6.
12. Bai J, Mu R, Dou M, Yan D, Liu B, Wei Q, et al. Epigenetic modification in histone deacetylase deletion strain of Calcarisporium arbuscula leads to diverse diterpenoids. Acta pharma Sin B 2018;8: 687-97.
13. Zhang T, Wan J, Zhan Z, Bai J, Liu B, Hu Y. Activation of an unconventional meroterpenoid gene cluster in Neosartorya glabra leads to the production of new berkeleyacetals. Acta pharma Sin B 2018;8:478-87.
14. Biggins JB, Liu X, Feng Z, Brady SF. Metabolites from the induced expression of cryptic single operons found in the genome of Burkholderia pseudomallei. J Am Chem Soc 2011;133:1638-41.
15. Liu Y, Ding G, Li Y, Qu J, Ma S, Lv H, et al. Structures and absolute configurations of penicillactones A-C from an endophytic microorganism, Penicillium dangeardii Pitt. Org Lett 2013;15:5206-9.
16. Buchi G, Snader KM, White JD, Gougoutas JZ, Singh S. Structures of rubratoxins A and B. J Am Chem Soc 1970;92:6638-41.
17. Chen RD, Yan Z, Zou JH, Wang N, Dai JG, Rubratoxin C. A new nonadride derivative from an endophytic fungus Penicillium sp. F-14. Chin Chem Lett 2014;25:1308-10.
18. Bai J, Yan D, Zhang T, Guo Y, Liu Y, Zou Y, et al. A cascade of redox reactions generates complexity in the biosynthesis of the protein phosphatase-2 inhibitor rubratoxin A. Angew Chem Int Ed 2017;56: 4782-6.
19. Yang SW, Chan TM, Terracciano J, Patel R, Patel M, Gullo V, et al. A new hydrogenated azaphilone Sch 725680 from Aspergillus sp. J Antibiot 2006;59:720-3.
20. Myobatake Y, Takeuchi T, Kuramochi K, Kuriyama I, Ishido T, Hirano K, et al. Pinophilins A and B, inhibitors of mammalian A-, B-, and Y-family DNA polymerases and human cancer cell proliferation. $J$ Nat Prod 2012;75:135-41.
21. Liu Y, Yang Q, Xia G, Huang H, Li H, Ma L, et al. Polyketides with alpha-glucosidase inhibitory activity from a mangrove endophytic fungus, Penicillium sp. HN29-3B1. J Nat Prod 2015;78:1816-22.
22. Yang SW, Chan TM, Terracciano J, Loebenberg D, Patel M, Gullo V, et al. Sch 1385568, a new azaphilone from Aspergillus sp. J Antibiot 2009;62:401-3.
23. Ren J, Ding SS, Zhu A, Cao F, Zhu HJ. Bioactive azaphilone derivatives from the fungus Talaromyces aculeatus. J Nat Prod 2017;80: 2199-203.
24. Natsume M, Takahashi Y, Marumo S. (-)-Mitorubrinic acid, a morphogenic substance inducing chlamydospore-like cells, and its related new metabolite, $(+)$-mitorubrinic acid B, isolated from Penicillium funiculosum. Agric Biol Chem 1985;49:2517-9.
25. Buechi G, White JD, Wogan GN. The structures of mitorubrin and mitorubrinol. J Am Chem Soc 1965;87:3484-9.
26. Zhu J, Porco JAJ. Asymmetric syntheses of (-)-mitorubrin and related azaphilone natural products. Org Lett 2006;8:5169-71.
27. Tabata Y, Ikegami S, Yaguchi T, Sasaki T, Hoshiko S, Sakuma S, et al. Diazaphilonic acid, a new azaphilone with telomerase inhibitory activity. J Antibiot 1999;52:412-4.
28. Hoshiko S, Sasaki T, Yaguchi T, Tabata Y, Sakuma S, Ikegami S, inventors. Meiji Seika Kaisha Ltd. and Meiji Milk Prod Co. Ltd., assignees. New chemical compound PF1195 and its manufacturing method. 1999 Oct 05. Japan patent JP 11269164 A.
29. Xiao Z, Lin S, Tan C, Lu Y, He L, Huang X, et al. Asperlones A and B, dinaphthalenone derivatives from a mangrove endophytic fungus Aspergillus sp. 16-5C. Mar Drugs 2015;13:366-78.
30. Lopes TI, Coelho RG, Yoshida NC, Honda NK. Radical-scavenging activity of orsellinates. Chem Pharm Bull 2008;56:1551-4.
31. Lee SM, Li XF, Jiang H, Cheng JG, Seong S, Choi HD, et al. Terreusinone, a novel UV-A protecting dipyrroloquinone from the marine algicolous fungus Aspergillus terreus. Tetrahedron Lett 2003;44: 7707-10.
32. Moore RE, Emery T. Nalpha-acetylfusarinines: isolation, characterization, and properties. Biochemistry 1976;15:2719-23.
33. Middleton AJ, Cole DS, Macdonald KD. A hydroxamic acid from Aspergillus nidulans with antibiotic activity against Proteus species. J Antibiot 1978;31:1110-5.
34. Koreeda M, Harada N, Nakanishi K. Exciton chirality methods as applied to conjugated enones, esters, and lactones. J Am Chem Soc 1974;96:266-8.
35. Itabashi T, Ogasawara N, Nozawa K, Kawai KI. Isolation and structures of new azaphilone derivatives, falconensins E-G, from Emericella falconensis and absolute configurations of falconensins A-G. Chem Pharm Bull 1996;44:2213-7.
36. Frisch MJ, Trucks GW, Schlegel HB, Scuseria GE, Robb MA, Cheeseman JR, et al. Gaussian 09, revision C.01. Wallingford, CT, USA: Gaussian Inc.; 2010.
37. Gorin PAJ, Mazurek M. Further studies on the assignment of signals in ${ }^{13} \mathrm{C}$ magnetic resonance spectra of aldoses and derived methyl glycosides. Can J Chem 1975;53:1212-23.
38. Frisch MJ, Trucks GW, Schlegel HB, Scuseria GE, Robb MA, Cheeseman JR, et al. Gaussian 09, revision D.01. Wallingford, CT, USA: Gaussian Inc.; 2009.
39. Steyn PS, Vleggaar R. The structure of dihydrodeoxy-8-epi-austdiol and the absolute configuration of the azaphilones. J Chem Soc Perkin Trans 1976;1:204-6.
40. Zabala AO, Xu W, Chooi YH, Tang Y. Characterization of a silent azaphilone gene cluster from Aspergillus niger ATCC 1015 reveals a hydroxylation-mediated pyran-ring formation. Chem Biol 2012;19: 1049-59.
41. Chen W, Chen R, Liu Q, He Y, He K, Ding X, et al. Orange, red, yellow: biosynthesis of azaphilone pigments in Monascus fungi. Chem Sci 2017;8:4917-25.
42. Sanchez JF, Somoza AD, Keller NP, Wang CCC. Advances in Aspergillus secondary metabolite research in the post-genomic era. Nat Prod Rep 2012;29:351-71.
43. Chiang YM, Szewczyk E, Davidson AD, Keller N, Oakley BR, Wang CCC. A gene cluster containing two fungal polyketide synthases encodes the biosynthetic pathway for a polyketide, Asperfuranone, in Aspergillus nidulans. J Am Chem Soc 2009;131: 2965-70.
44. Hui Z, Qiao K, Gao Z, Meehan MJ, Li JWH, Zhao X, et al. Enzymatic synthesis of resorcylic acid lactones by cooperation of fungal iterative polyketide synthases involved in hypothemycin biosynthesis. J Am Chem Soc 2010;132:4530-1.
45. Xie X, Meehan MJ, Xu W, Dorrestein PC, Tang Y. Acyltransferase mediated polyketide release from a fungal megasynthase. J Am Chem Soc 2009;131:8388-9.
46. Zhang X, Li S. Expansion of chemical space for natural products by uncommon P450 reactions. Nat Prod Rep 2017;34:1061-89.
47. Nozaki H, Kuroda S, Watanabe K, Yokozeki K. Gene cloning of $\alpha$ methylserine aldolase from Variovorax paradoxus and purification and characterization of the recombinant enzyme. Biosci Biotechnol Biochem 2008;72:2580-8.
48. Chiang YM, Ahuja M, Oakley CE, Entwistle R, Asokan A, Zutz C, et al. Development of genetic dereplication strains in Aspergillus nidulans results in the discovery of aspercryptin. Angew Chem Int Ed 2016;55:1662-5.
49. Chen L, Wu H, Liu H, Li E, Ren J, Wang W, et al. Genetic dereplication of trichoderma hypoxylon reveals two novel polycyclic lactones. Bioorg Chem 2019;91:103185.
50. Sanchez JF, Entwistle R, Hung JH, Yaegashi J, Jain S, Chiang YM, et al. Genome-based deletion analysis reveals the prenyl xanthone biosynthesis pathway in Aspergillus nidulans. J Am Chem Soc 2011;133:4010-7.
51. Gao SS, Li XM, Zhang Y, Li CS, Cui CM, Wang BG. Comazaphilones A-F, azaphilone derivatives from the marine sediment-derived fungus Penicillium commune QSD-17. J Nat Prod 2011;74:256-61.
52. Zhao DL, Shao CL, Zhang Q, Wang KL, Guan FF, Shi T, et al. Azaphilone and diphenyl ether derivatives from a gorgonian-derived strain of the fungus Penicillium pinophilum. J Nat Prod 2015;78:2310-4.
53. O'Boyle NM, Tenderholt AL, Langner KM. Cclib: a library for package-independent computational chemistry algorithms. J Comput Chem 2008;29:839-45.
54. O'Boyle NM, Vandermeersch T, Flynn CJ, Maguire AR, Hutchison GR. Confab-systematic generation of diverse low-energy conformers. J Cheminf 2011;3:8.
55. Tanaka T, Nakashima T, Ueda T, Tomii K, Kouno I. Facile discrimination of aldose enantiomers by reversed-phase HPLC. Chem Pharm Bull 2007;55:899-901.
56. Qu J, Fang L, Ren XD, Liu Y, Yu SS, Li L, et al. Bisindole alkaloids with neural anti-inflammatory activity from Gelsemium elegans. J Nat Prod 2013;76:2203-9.
57. Hu Y, Ma S, Li J, Yu S, Qu J, Liu J, et al. Targeted isolation and structure elucidation of stilbene glycosides from the bark of Lysidice brevicalyx Wei guided by biological and chemical screening. J Nat Prod 2008;71:1800-5.

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