Molecular dynamics simulation of Fe-NO₂ At-alpha mangostin as radiopharmaceutical model for detection of fatty acid synthase in cancer

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ABSTRACT

α-mangostin is a xanthone compound isolated from mangosteen pericarp. It is known as an anticancer through induction of apoptotic process by inhibiting fatty acid synthase (FAS) receptor. α-mangostin is a potentially useful ligand for diagnostic purposes in the form of complexes with a radionuclide such as⁶⁸Gallium (⁶⁸Ga). Unfortunately, α-mangostin could not be directly labeled with radionuclides. In order to be labeled, a chelator such as 1,4,7-triazcyclononane-1,4,7-triacetic acid (NOTA), a derivative (NO2At), is required. The aim of this study was to find out the interaction of Fe-NO2At-α-mangostin complex compound against FAS receptor using molecular dynamics software. Both the metals have similar chemical characteristics. The results showed a strong interaction between Fe-NO2At-α-mangostin complex compound and FAS receptor. The molecular dynamic showed the complex compound Fe-NO2At-α-mangostin in FAS-KS which produced a bond-free energy values (ΔG) of – 96.7 kcal/mol, forming hydrogen bonds with amino acid residues Glu 115 and Ser 114. The model of molecular dynamic result could be used as a model for the production of ⁶⁹Ga-α-mangostin in radiopharmaceutical.

Key words: 68Ga, fatty acid synthase, Fe, molecular dynamics, NO₂At, α-mangostin

INTRODUCTION

Positron-emission tomography/computed tomography (PET/CT) scan using ⁶⁸Gallium (⁶⁸Ga)-labeled radiopharmaceutical is a molecular imaging diagnostic modality to detect and localized tumor, as well as and monitoring treatment.^[1]

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Radiometal ⁶⁸Ga-based radiopharmaceuticals usually require chelators that can conjugate with molecules as target receptors. The chelator protects the radiometal ion to avoid transchelation and hydrolysis reaction, so the radiopharmaceutical has specific biological benefits.^[2] Chelators are covalently bonded to the target molecule, producing biologically active radiopharmaceuticals. The most chelator used for ⁶⁸Ga³⁺ compound complex is a derivative of a macrocyclic chelator known as NOTA.^[3]

Ga³⁺ has a small ionic radius (0.62Å)^[3,4] and compatible with the hole size of the NOTA cavity that leads to the stability of compound complex of Ga³⁺NOTA.^[2,5] This phenomenon

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is known as a size-match selectivity.^[4] NOTA and its derivatives are considered to be the gold standard chelator for ⁶⁸Ga^[3,5] since the kinetics of the formation of complex compounds are fast at room temperature^[2] and excellent *in vivo* stability.^[5]

Discovering a new drug for cancer is challenging, such as drugs that can induce apoptosis in cancer cells. In this regard, there is an anticancer compound through induction of apoptotic process by inhibiting fatty acid synthase (FAS) receptor^[6] known as α -mangostin. It is a major isolated compound of *Garcinia mangostana*.^[7] α -mangostin is a potential ligand for cancer theranogtic if it can be labeled with different radionuclides. Unfortunately, α -mangostin could not be directly labeled with radionuclide. In order to be labeled, a chelator NOTA derivates such as NO2At is required.

FAS is an enzyme required to regulate *de novo* biosynthesis of long-chain fatty acids and has seven domains. The catalytic domains include acyl carrier protein, dehydrogenase (DH), enoyl reductase (ER), β -ketoacyl reductase (KR), β -ketoacyl synthase (KS), malonyl/acetyltransferase (MAT), and thioesterase domain (TE).^[8,9] The are two types of FAS. Human FAS and others mammals are type I, which consisted of two identical proteins (homodimerics) and the seven domains forming one binding. FAS in plants is a type II and the seven domains are independent.^[8]

High expression of FAS in cancer cell is a poor indicator in patients with breast, ovary, and prostate cancer.^[10] The inhibitory reaction of α -mangostin (IC₅₀ value) at FAS was 5.54 mM. α -mangostin can inhibit FAS at the acetyl binding site of KS and stronger than MAT.^[11] The study aim was to find out the interaction between Fe-NO2At- α -mangostin complex compound and FAS receptor using molecular dynamics software. In this study, Ga was substituted by Fe since the Ga parameter is not available in the system. Both the metals have similar chemical characteristics.

MATERIALS AND METHODS

Hardware and software

Hardware: the hardware used is a personal computer with Intel Xeon 6 core, NVIDIA GTX 1080Ti, WD Green SSD, and 16GB RAM. Software: The softwares used were as follows: AutoDock 4 (The Scripps Research Institute, USA),^[12] Biovia Discovery Studio 2019 (Dassault Systemes, Sandiego), Protein Data Bank (rcsb.org/pdb/), PubChem (NCBI, NIH, USA), VMD, MGL Tools 1.5, and ChemDraw Ultra 15.1 programs (PerkinElmer Inc., downloaded at http:// www.cambridgesoft.com/).

Materials

The three-dimensional (3D) FAS-KS structure of PDB with PDB ID code 3HHD and 1EK4 was used as a target protein

complex in this study. Meanwhile, the 3D of α -mangostin and its derivative structures was created using Biovia Discovery Studio 2019.

Methods

Recognizing structure of fatty acid synthase-ketoacyl synthase

This study was conducted by recognizing the structure of FAS-KS. The crystal structure of the FAS-KS used in this study was downloaded from the PDB site with PDB ID 3HHD from human and 1EK4 from of *Escherichia coli*. The next step was comparing 3HHD with 1EK4 structures to look for the binding site of FAS-KS.

α-Mangostin modeling

 α -mangostin and the complexed compound of Fe-NO2At- α -mangostin were modeled usingBiovia Discovery Studio program and the results in the form of 3D form were stored in PDB format. The charge and atom types were added to those complex compounds using the MGL tools 1.5.6 program and saved in PDB format.

Molecular docking

Molecular docking simulation of α -mangostin on the active site of FAS-KS 3HHD was tested using Autodock program with the grid box parameter set at 60 × 60 × 60 with a grid spacing of 0.376 Å placed at the midpoint of the active site. The docking was done by 100 times searching conformation using the Lamarckian-Genetic Algorithm protocol.

Molecular dynamics simulation

The results of the docking were carried out on molecular dynamics simulations using Amber18 (University of California, San Francisco, USA). The input files for simulations were prepared using the LeaP module. The antechamber program was used for ligand parameterization by a semi-empirical calculation of AM1-BCC. The receptor was prepared with the PDB4amber module to change the name of amino acid residue to the Amber format and regulate the protonation state of those amino acids. The tleap program was used to connect the receptors to ligands and their parameters, such as ff14SB, gaff, water TIP3P. The next step was adding the water molecule box into the system with the TIP3P water model. The minimization of structure was done by using the steepest descent method and conjugating each gradient for 5000 times, followed by three steps of the increasing grade of heat. Sequentially, the steps were 0-100°K for 20 ps, 100-200°K for 20 ps, and 200-300°K for 20 ps and balanced gradually using the module pmemd for 1 ns. The production phase was carried out for 50 ns. Then, the trajectory of the simulation results was analyzed to obtain root mean square deviation (RMSD), root mean square fluctuation (RMSF), and ligand binding energy with receptors. The bond energy was obtained using the Molecular Mechanic-Generalized Boltzmann Surface Area (MMGBSA) method.

RESULTS

The KS domain of FAS enzyme is a homodimer protein where the active site of this enzyme is located between the two monomers [red circle in Figure 1].

The minimization process of apo to be holo structure

Apo structure changes can be done by inserting dodecanoic acid into the active site of 3HHD, then relaxing it by minimizing the structure in the Amber program by 10,000 steps with 5000 steps steepest descends and 5000 conjugate gradient. There was a shift of amino acids on the active site of the KS-FAS apo structure after the minimization process is presented in Figure 2.

α -mangostin modeling

The modeling of α -mangostin and Fe-NO2At- α -mangostin using the BIOVIA Discovery Studio program is presented in Figure 3.

The molecular docking of α -mangostin to fatty acid synthase-ketoacyl synthase apo

The process of docking α -mangostin to the structure of the FAS-KS apo which was relaxed with the Amber 18 program is presented in Figure 4.

The NO₂At-Fe conjugation in α-mangostin structure

The NO₂At-Fe conjugation in the α -mangostin structure is carried out in the blue-circled group in Figure 4b and conjugated at C number 13 of α -mangostin compound [Figure 3a]. The interaction of these substituents with the active site of FAS-KS occurs in the hydrophilic region. NO₂ At-Fe is a polar compound; thus, the polar environment is appropriate for the interaction between these molecules. The docking was carried out for Fe-NO₂At-alpha-mangostin in FAS-KS and pose was obtained is presented in Figure 5.

Molecular dynamic simulation

RSMF showed that the fluctuation of amino acid residues that made up receptors during simulation process could represent residual flexibility.^[13] The area of 400–500 Ao on the RMSF chart at Figure 6a, FAS-KS bound to the ligand showed lower fluctuations compared to the apo structure. FAS-KS amino acid residues that bound to the ligands in the region 100–300 A° showed a little bit higher fluctuation than apo structure (monomer 1) due to the position of monomer 1 which is the location, the binding site, of the FAS-KS bound to the ligand. The result of monomer-2 of RMSF in the area 850 A° as a binding site between amino acid located in KSb portion and the ligand showed that the fluctuation was lower than the apo structure because the NO₂At-Fe group was interacting with KSb portion, as presented in Figure 6b.

RMSD is an atomic displacement that occurs during a simulation compared to a native ligand structure.^[13] RMSD results showed that the condition of binding between FAS-KS and the ligand

tends to be more stable than the apo structure [Figures 7 and 8]. The RSMD ligand graphic showed a spike of 1 A° on frame 228–240 for α -mangostin system [Figure 9].



Figure 1: The active side of fatty acid synthase-ketoacyl synthaseenzyme



Figure 2: Amino acids on the active side of the fatty acid synthaseketoacyl synthase apo structure after the minimization process with an root mean square deviation of 2.5 Ao (before relaxation silver and after relaxation blue)



Figure 3: (a) Two-dimensional structure α -mangostin (b) threedimensional structure α -mangostin, (c) The three-dimensional structure Fe-NO2At- α -mangostin



Figure 4: (a) α -mangostin interactions on the active site of fatty acid synthase-ketoacyl synthase, (b) the FAS-KS active surface based on hydrophobicity properties



Figure 6: (a) RSMF of fatty acid synthase-ketoacyl synthasemonomer 1, (b) RSMF of fatty acid synthase-ketoacyl synthasemonomer 2

The calculation of ligand binding energy with energy calculation receptor was done using MMGBSA in Amber 18. The results of molecular dynamic of α -mangostin in FAS-KS produced a free binding energy with the value of ΔG – 57.4 kcal/mol and the hydrogen bonds that were

Figure 5: (a) complex compound Fe-NO2At- α -mangostin interactions on the active site of fatty acid synthase-ketoacyl synthase, (b) the fatty acid synthase-ketoacyl synthaseactive surface based on hydrophobicity properties

Figure 7: Root mean square deviation fatty acid synthase-ketoacyl synthasemonomer 1

formed with amino acid residues Glu 333 and Glu 115, as shown in Figure 10. The molecular dynamic results of the complex compound of Fe-NO2At- α -mangostin in FAS-KS produced bond energy with values of (ΔG) – 96.7 kcal/mol and the hydrogen bonds that were formed with amino acid residues Glu 115 and Ser 114, as shown in Figure 11. The interactions of Fe-NO2At- α -mangostin compounds with FAS-KS were similar to the interactions with α -mangostin in the presence of hydrogen bonding at the residues of Glu 115 and Phe 395.

Moreover, the NO₂At consists of two carboxylic groups. Both the oxygen atoms from the carboxylic group must be bound to Fe metal, but from the simulation molecular dynamic, only one oxygen atom was bound to Fe, while

Figure 8: Root mean square deviation fatty acid synthase-ketoacyl synthasemonomer 2

Figure 10: Results of molecular dynamic α -mangostin in fatty acid synthase-ketoacyl synthase

Figure 12: The interaction of complex compounds with the amino acid Glu118 in fatty acid synthase-ketoacyl synthase

one Fe was bound to Glu 118, as the oxygen atoms were released from Fe and stabilized by the amino acid residue of Ser 114 [Figure 12].

Figure 12 shows the interaction of complex compounds with the amino acid Glu 118 in FAS-KS.

Figure 9: Root mean square deviation ligand

Figure 11: Results of molecular dynamic the complex compound Fe-NO2At- α -mangostin in fatty acid synthase-ketoacyl synthase

Figure 13: The difference of 3HDD and 1EK4 structures after superpose, 1EK4 (gray), 3HHD (green)

DISCUSSION

In this study, the FAS-KS-MAT domain, a human FAS, as the target protein of complex compound Fe-NO2At- α -mangostin is based on Pappenberger *et al.* study. They showed that the human FAS with high-resolution crystal structure is KS-MAT domain which consists of a tandem domain KS which is linked by the linker domain (LD) to the malonyltransferase MAT.^[14] Mammalian FAS can also be

used as a target, but there are some limitations in size and flexibility and its crystal structure resolution is medium.

The KS domain consists of a sequence of amino acid residues (1–852), and the final position of the KS domain is a series of 28 amino acid residues from the sequence 825–852 (KSb). KSb locations are far from the previous residual sequence because of an insertion of MAT and LD in the KS domain.^[14] The KS domain of FAS enzyme is a homodimer protein where the active site of this enzyme is located between two monomers. A previous study by Quan *et al.* 2012 showed that the α -mangostin can inhibit FAS by competing with the acetyl group in the KS domain.^[11] Since the conjugation between crystalline structure of FAS-KS and α -mangostin is not available, an α -mangostin docking simulation must be performed on the FAS-KS.

The 3D structure of 3HHD comes from human FAS-MAT-KS in the form of apo that has no ligand with a resolution of 2.15 Aº. The 3D structure of EEK4 derived from E. coli FAS-KS binds with dodecanoic acid as a ligand with a resolution of 1.85 Aº (holo). In this case, the dodecanoic acid acts as an inhibitor in the acetyl domain. Both FAS-KS domains derived from E. coli and human have a high amino acid sequence identity on the active site. This shows that the FAS-KS domain of *E. coli* can act as a surrogate to design human FAS inhibitors.^[14] The difference of these two structures is shown in Figure 13. α -mangostin interactions on the active site of FAS-KS can be seen more apparently in Figure 4. It shows that the appearance of the FAS-KS active site surface is based on hydrophobes. The results of the α -mangostin docking were parallel to the dedocanoic acid. It means that the conformation results of α -mangostin molecule docking were almost the same as the dodecanoic acid conformation. Figure 6b shows the active surface of FAS-KS based on hydrophobicity properties, while the alkyl substituents on α -mangostin occupy the hydrophilic region (pink surface).

In this study, iron was used instead of gallium metal since the parameter of gallium metal is not available in the Amber program. Fe and Ga have similar characteristics. The ionic radius characteristics of gallium (III) are similar to iron (III). The octahedral radii of Ga³ + and Fe³ + are 0.620 A° and 0.645 A°, respectively. The tetrahedral ion radii of Ga³ + and Fe³ + are 0.47 A° for Ga³ + and 0.49 A°, respectively.^[15] In this regard, Cusnir *et al.* conducted a study to develop the hydroxypyridinone ligand into Fe³ + chelation and how the design of chelators could be suitable for replacing⁶⁸Ga as a radiopharmaceutical for PET.^[16]

CONCLUSION

The NO₂At-Fe compound could be conjugated in C Number 13 of α -mangostin compounds. The Fe-NO₂At- α -mangostin forms a binding hydrogen with amino acid residues of Glu 115 and Ser 114, with a bond strength (Δ G) of – 96.7 kcal/mol.

Based on the dynamic molecular computation, α -mangostin and Fe complex compounds with NO₂At chelator were the models for the development of PET radiopharmaceutical according to ⁶⁸Ga³⁺ toradionuclide since Fe³⁺ and Ga³⁺ have the same coordination number.

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Conflicts of interest

There are no conflicts of interest.

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