# Poster presentation

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# Bioactivation of pentaerythrityl tetranitrate by mitochondrial aldehyde dehydrogenase

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## Background

Organic nitrates represent a class of very effective antiischemic drugs used for the treatment of angina pectoris, myocardial infarction and congestive heart failure [1]. A limitation of their clinical use is the development of nitrate tolerance. A remarkable exception is the long-acting drug pentaerythrityl tetranitrate (PETN), which was shown to cause no tolerance [2]. The mitochondrial isoform of aldehyde dehydrogenase (ALDH2) was shown to be involved in bioactivation of nitroglycerin (GTN) [3] and PETN [4]. The present work was carried out to study the molecular mechanisms underlying the pharmacological differences between GTN and PETN at the level of purified ALDH2.

## Results

As quantified by liquid chromatography-mass spectroscopy (LC-MS) ALDH2 metabolized 50  $\mu$ M PETN to 14.84  $\pm$  4.05 nmol/min/mg pentaerythrityl trinitrate (PETriN) in the absence and 62.38  $\pm$  7.21 nmol/min/mg PETriN in the presence of the reducing agent dithiothreitol. Co-incubation of 2  $\mu$ M [<sup>14</sup>C]GTN with ALDH2 and increasing concentrations of PETN resulted in only small decreases in 1,2-glyceroldinitrate formation as measured by radio thin layer chromatography (control: 7.98  $\pm$  0.20, 1  $\mu$ M PETN: 7.95  $\pm$  0.20, 10  $\mu$ M PETN: 7.41  $\pm$  0.18, 100  $\mu$ M PETN: 6.66  $\pm$  0.50 nmol/min/mg) indicating that PETN does not compete with GTN metabolism. In the presence of 25  $\mu$ g of ALDH2 PETN activated soluble guanylate cyclase (sGC) with an EC<sub>50</sub> of 3.61 ± 0.35  $\mu$ M and a maximum at 30  $\mu$ M PETN (2.67 ± 0.12  $\mu$ mol/min/mg). This effect was enhanced by superoxide dismutase (SOD) resulting in an EC<sub>50</sub> of 0.64 ± 0.08  $\mu$ M and maximal sGC activation (18.44 ± 0.27  $\mu$ mol/min/mg in the presence of 30  $\mu$ M PETN). ALDH2 activity, measured as formation of NADH from NAD+ in the presence of acetaldehyde, was rapidly inactivated by GTN but not by PETN (1 mM each).

### Conclusion

Low and high affinity pathways of PETN may explain the apparent differences between PETN metabolism and bioactivation, respectively. The reaction of PETN with ALDH2 leads to activation of sGC with markedly higher values in the presence of SOD suggesting formation of superoxide as a co-product of PETN metabolism. The lack of vascular tolerance to PETN may be due to significantly lower rates of ALDH2 inactivation as compared to GTN.

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