ORIGINAL RESEARCH



Investigation of [³H]diazepam derivatives as allosteric modulators of GABAA receptor $\alpha_1\beta_2\gamma_2$ subtypes: combination of molecular docking/dynamic simulations, pharmacokinetics/drug-likeness prediction, and QSAR analysis

Rachida Djebaili¹ · Samir Kenouche² · Ismail Daoud^{3,4} · Nadjib Melkemi¹ · Ahlem Belkadi¹ · Fouzia Mesli³

Received: 9 March 2022 / Accepted: 2 August 2022

© The Author(s), under exclusive licence to Springer Science+Business Media, LLC, part of Springer Nature 2022

Abstract

In this paper, a data set of [³H] diazepam derivatives was analyzed using various computational methods: molecular docking/dynamic simulations, and QSAR analysis. The main aims of these studies are to understand the binding mechanisms by which benzodiazepines allosterically modulate GABA_A receptor $\alpha_1\beta_2\gamma_2$ subtypes, from inducing neuronal inhibition at lower doses to the anesthetic effect at higher doses, and also, to define the structural requirements that contribute to improving the response of GABA_A/ $\alpha_1\beta_2\gamma_2$ receptor to benzodiazepine drugs. The results of the molecular docking study allowed selecting Ro12-6377 and proflazepam as the best modulators for the four binding sites simultaneously. Subsequently, the stability of the selected complexes was investigated by performing molecular dynamics simulation. The latter confirmed the features of both modulators to exert direct effects on the chloride-channel lining residues. Pharmacokinetics and drug-likeness profile were assessed through in silico tool. Furthermore, a QSAR analysis was conducted using an improved vemolecular dynamics simulations proposed byrsion of PLS regression. The goodness of fit and the predictive power of the resulting PLS model were estimated according to internal and external validation parameters: $R^2 = 0.632$, $R^2_{adj} = 0.584$, F = 12.806; *p*-value = 6.2050e - 07, $Q^2_{loo} = 0.639$, and $Q^2_{F3} = 0.813$. Clearly, the obtained results ensure the predictive ability of the developed QSAR model for the design of new high-potency benzodiazepine drugs.

Keywords Benzodiazepine \cdot GABA_A receptor \cdot Extracellular domain \cdot Transmembrane domain \cdot Chloride channel \cdot TM₂ helix

Introduction

Gamma-aminobutyric acid type A (GABA_A) are fastacting ionotropic receptors that belong to the superfamily of Cys loop-type ligand-gated ion channels (PLGICs)

☑ Ismail Daoud i.daoud@univ-biskra.dz

- ¹ Group of Computational and Pharmaceutical Chemistry, LMCE Laboratory, University of Mohamed Khider, 07000 Biskra, Algeria
- ² Group of Modeling of Chemical Systems Using Quantum Calculations, Applied Chemistry Laboratory (LCA), University of Mohamed Khider, 07000 Biskra, Algeria
- ³ Laboratory of Natural and Bioactive Substances, LASNABIO, University of Abou-BakrBelkaid, Tlemcen, Algeria
- ⁴ Present address: Department of Matter Sciences, University of Mohamed Khider Biskra, BP 145 RP, 07000 Biskra, Algeria

Published online: 11 August 2022

[1]. This type of GABA receptor is found at 20-50% of brain synapses [2].GABA ARs assemble into pentameric isoforms consisting of five heteromeric subunits that surround a selective chloride-conducting pore. The general architecture of each mature subunit is constructed by the sequence of 450 amino acid residues [3]. In total, 200–250 amino acids contribute in the extracellular domain (ECD) to form a long hydrophilic N-terminal α-helix followed by ten β -strands folded into a β -sandwich containing the Cys-loops. A total of 85-255 amino acids contribute in the transmembrane domain (TMD) to form four membrane α -helices (named from TM₁ to TM₄) connected by three loops (short intracellular loop links TM₁-TM₂, short extracellular loop links TM2-TM3, and long cytoplasmic loop links TM₃-TM₄) and terminating with one small extracellular C-terminal [4]. The five $TM_2\alpha$ helices were assembled in the center to form the chloride-channel liner, with a possible contribution from TM_1 [5]. Upon activation, GABA_ARs allow negative chloride ions to flow through

the ion channel into the cell, thereby inhibiting neuronal excitability for a short time (phasic inhibition) or for a long time (tonic inhibition) [3, 6].

In the mammalian brain, GABA_ARs are assembled from 8 different families of subunits containing a total of 19 subtypes: α_{1-6} , β_{1-3} , γ_{1-3} , δ , ε , π , θ , and ρ_{1-3} . The predominant GABA_ARs isoform consists of two α-subtypes, two β -subtypes, and one γ -subtype, or one δ -subtype. γ-containing receptors distribute mainly in the synaptic sites and account for approximately 90% of GABA_ARs in the adult brain. In contrast, δ -containing receptors are abundant in the extrasynaptic sites located in specific brain areas such as the hippocampus, amygdala, neocortex, thalamus, hypothalamus, and the cerebellum [6]. Owing to its complex subunit assembly, GABAARs contain large numbers of allosteric binding sites that make them the most important drug target in the central nervous system (CNS). They modulate by two endogenous molecules (neurosteroids and endocannabinoid 2-arachidonoylglycerol) and a wide range of exogenous molecules; the most important among them are benzodiazepines(BDZs), barbiturates, the intravenous general anesthetics propofol and etomidate, alcohols, the competitive antagonist bicuculline, and the channel blocker picrotoxin [3].

The $\alpha_1\beta_2\gamma_2$ combination constitutes approximately 43–60% of GABA_ARs in the adult brain [6], which makes it the subject of many previous reports attempting to provide high-resolution structural data that illuminate atomic mechanisms of drug recognition. The most important among them are the cryo-electron microscopy structures complemented with mutagenesis, electrophysiology, and

molecular dynamics simulations proposed by Kim et al. [2, 7]. Kim et al. demonstrated, in addition to the classical BDZ site at the ECD α_1^+/γ_2^- interface, the presence of three additional binding sites in the TMD: two at $\beta_2^+/\alpha_1^$ interfaces and a third at the γ_2^+/β_2^- interface (Fig. 1). β_2^+ $/\alpha_1^-$ sites were also previously observed by Masiulis et al. [8] through their study on $\alpha_1\beta_3\gamma_2$ Rs. The pharmacological role of BDZs exerts in the presence of two GABA in their binding sites at the ECD β_2^+/α_1^- interfaces. More precisely, BDZs react as positive allosteric modulators that enhance the affinity of GABARs for GABA, leading to an increase in the conduction of chloride ions through the ionic channel into the cell, hyperpolarization of neurons, and thus inhibition of neuronal excitability. The BDZ binding loci at both TMD β_2^+/α_1^- interfaces were identified to be also the binding sites for etomidate and propofol. Also, the TMD binding locus located at the γ_2^+/β_2^- interface was estimated to overlapped in part with that of the phenobarbital. Etomidate, propofol, and barbiturate are general intravenous anesthetics that differ from the benzodiazepines in their ability—at higher doses—to directly activate GABA_ARs without the need for GABA. In a mechanism similar to that of barbiturate, occupation of the γ_2^+/β_2^- interface by diazepam (DZP) leads to close the gap presented at the interface between the two γ_2 - β_2 subunits, which according to Kim et al., may explain the anesthetic properties observed during the administration of a high dose of DZP. Currently, the [3H]diazepam derivatives are the most common psychoactive drugs used to treat epilepsy, insomnia, muscle spasms, anxiety, alcohol withdrawal, and panic disorder [9].



Fig. 1 Binding sites of the two endogenous agonists (GABA) and diazepam. **a** ECD interface for the binding of diazepam (**a**) and GABA (a' and a''). **b** The three TMD interfaces identified for the binding of diazepam [7]

The strategy of combining molecular docking, molecular dynamics simulation, and QSAR analysis has emerged as a practical tool in the process of drug development through computational techniques. Its main advantage lies in improving the success rate of drug screening in less time and at a lower cost [10]. Molecular docking simulations are designed to determine the best ligand/target binding mode that generates the biological response. In practice, it allows the screening of large libraries of compounds by implementing fast and inexpensive docking algorithms. Its process is based on the production of all possible ligand poses within the binding target, and associates each pose with a score value that approximates its free energy landscape. At the ends of simulations, the best binding modes are ranked based on the values of the latter, and thus the most appropriate complexes are selected. Mostly, after molecular docking simulations, the best-docked complexes are subject to stability investigation through molecular dynamics simulation. Monitoring the dynamic profile of complexes over a certain time range provides the advantage of detecting various internal motions and conformational changes that occur in the binding site. Hence, validate the docking protocols. Otherwise, MD simulations can be generated before performing molecular docking for several objectives such as optimizing the target structure and ensuring its flexibility, quantifying the ligands/ target free binding energies,...etc., as well as, during the docking process, to accurately detect the binding locus and properly dock ligands [11].

Quantitative structure-activity relationships (QSARs) are statistically derived models mainly devoted to predicting the activities of new chemical entities from knowledge of their chemical structures. QSAR models quantitatively correlate the physicochemical and biological properties of compounds with their biological responses [12]. The success of any QSAR model depends on many factors, including the selection of practical statistical techniques for model development and validation strategies. The quality of models is evaluated through an internal validation process usually based on the use of the cross-validation method (CV), whereas the predictive power can be estimated using independent test data that was not involved in model generation [13]. In the literature, considerable researchers have sought to study the BDZ by performing QSAR analysis. Against this background, fifty-seven compounds from the dataset of interest in our study were examined by Maddalena and Johnston [14] using the backpropagation artificial neural network method (ANN) and multilinear regression analysis (MLR). The two-layer ANN model gave an excellent correlation between the binding affinities of the 38 compounds of the training set $(R_t = 0.941)$ and the 10 selected input variables $(\pi_7, F_7,$ MR₁, MR_{2'}, R_1 , F_2 , MR₆, μ_1 , δ_{p8} , and δ_{m3}), where F, R, μ , $\delta_{\rm p}$, and $\delta_{\rm m}$ denote polar constant, resonance constant,

dipole moment, hammett-para constant, and hammett-meta constant, respectively. The predictive power was tested with an external set of 19 compounds, and an optimal cross-validation coefficient (R_{cv} =0.910) was found. Later, So et al. [15] re-examined the data provided by Maddalena et al. with an improved version of the genetic neural network (GNN). The top-ranking variables selected by GNN shared only four variables with the ANN selection (π_7 , F_7 , MR₁, and MR₂). Therefore, an accurate comparison and a further discussion were made between the results of the two statistical methods, as well as the three highly predictive models (T6-2 # 1–3) which were combined with the optimal functional groups proposed by Maddalena to be placed in positions 1, 7, and 2' and used to design 20 new BDZ derivatives with their predicted activities.

This research aims to highlight the binding mechanism by which a data set of classical benzodiazepines allosterically modulates GABA_A receptor $\alpha_1\beta_2\gamma_2$ subtypes. In addition to the well-known ECD binding interface, our study was further expanded to include the three TMD interfaces that were recently discovered. In the first step, a data set of [³H]diazepam derivatives was subjected to fast screening through the molecular docking approach. Subsequently, molecular dynamics simulation and pharmacokinetics/drug-likeness evaluations were performed to refine the best-docked complexes. Finally, an improved version of PLS regression was implemented to quantify structural features that contribute to the improvement of BDZ/ $\alpha_1\beta_2\gamma_2$ Rs responses. Throughout this paper, the results have been interpreted in light of the combination of the cited approaches.

Materials and methods

Biological data

The results in vitro for the 50% inhibition of the binding of [³H]diazepam to homogenates of rat brain cell membranes by BDZs expressed as log (1/C) reported earlier in the review of Hadjipavlou-Litina and Hansch [16] (Table 1) were investigated to perform a molecular docking simulation and to predict QSAR model using PLS analysis. According to Micheli et al. [17], the good structural diversity allows this dataset to be optimal for undergoing QSAR analysis.

Molecular descriptor generation

First, a gradient norm limit of 0.1 kcal/(Å mol) was chosen for the pre-optimization of sixty-five BDZ derivatives, using the molecular mechanics force field (MM⁺) method included in HyperChem package version 8.08 [18]. Then,

Table 1 The classical BDZ data set under study [16]



N°	Name	R ₁	R ₃ / R ₅ /R ₆	R ₇	R ₈ / R ₉	R ₂ ,	R _{6'}	Log (1/C) _{obs}
1	Ro05-4318/ Ro05-3418	CH ₃	$R_{3}, R_{5}, R_{6} = H$	NH ₂	$R_8, R_9 = H$	Н	Н	6.34
2	Ro05-3072	Н	$R_3, R_5, R_6 = H$	NH ₂	$R_8, R_9 = H$	Н	Н	6.41
3	Ro05-4528	CH ₃	$R_3, R_5, R_6 = H$	CN	$R_8, R_9 = H$	Н	Н	6.42
4	Ro05-2921	Н	$R_{3}, R_{5}, R_{6} = H$	Н	$R_8, R_9 = H$	Н	Н	6.45
5	Ro20-7736	CH ₃	$R_3, R_5, R_6 = H$	NHOH	$R_8, R_9 = H$	F	Н	7.02
6	Ro05-4619	Н	$R_3, R_5, R_6 = H$	NH ₂	$R_8, R_9 = H$	Cl	Н	7.12
7	Ro20-5397	Н	$R_3, R_5, R_6 = H$	СНО	$R_8, R_9 = H$	Н	Н	7.37
8	Ro05-3061	Н	$R_3, R_5, R_6 = H$	F	$R_8, R_9 = H$	Н	Н	7.40
9	Ro20-2533	Н	$R_3, R_5, R_6 = H$	C_2H_5	$R_8, R_9 = H$	Н	Н	7.44
10	Ro20-2541	CH ₃	$R_3, R_5, R_6 = H$	CN	$R_8, R_9 = H$	F	Н	7.52
11	Ro20-5747	Н	$R_3, R_5, R_6 = H$	CHCH ₂	$R_8, R_9 = H$	Н	Н	7.62
12	Ro05-4336	Н	$R_3, R_5, R_6 = H$	Н	$R_8, R_9 = H$	F	Н	7.68
13	Ro20-3053	Н	$R_3, R_5, R_6 = H$	COCH ₃	$R_8, R_9 = H$	F	Н	7.74
14	Triflunordazepam	Н	$R_3, R_5, R_6 = H$	CF ₃	$R_8, R_9 = H$	Н	Н	7.89
15	Diazepam	CH ₃	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Н	Н	8.09
16	Ro07-5220	CH ₃	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Cl	Cl	8.26
17	Ro14-3074	Н	$R_3, R_5, R_6 = H$	N_3	$R_8, R_9 = H$	F	Н	8.27
18	Flunitrazepam	CH ₃	$R_3, R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	F	Н	8.42
19	Ro05-3590	Н	$R_3, R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	CF_3	Н	8.45
20	Norflurazepam	Н	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	F	Н	8.7
21	Delorazepam	Н	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Cl	Н	8.74
22	Clonazepam	Н	$R_3, R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	Cl	Н	8.74
23	Fonazepam	Н	$R_3, R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	F	Н	8.82
24	Ro05-6822	CH ₃	$R_3, R_5, R_6 = H$	F	$R_8, R_9 = H$	F	Н	8.29
25	Ro05-4865	CH ₃	$R_3, R_5, R_6 = H$	F	$R_8, R_9 = H$	Н	Н	7.77
26	Ro05-6820	Н	$R_3, R_5, R_6 = H$	F	$R_8, R_9 = H$	F	Н	8.13
27	Nordazepam	Н	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Н	Н	8.03
28	Ro07-3953	Н	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	F	F	8.79
29	Difludiazepam	CH ₃	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	F	F	8.39
30	Ro07-5193	Н	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Cl	F	8.52
31	Ro22-3294	Н	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Cl	Cl	8.15
32	Nitrazepam	Н	$R_3, R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	Н	Н	7.99
33	Methylclonazepam	CH ₃	$R_3, R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	Cl	Н	8.66
34	7-Aminoflunitrazepam	CH ₃	$R_3, R_5, R_6 = H$	NH ₂	$R_8, R_9 = H$	F	Н	7.19
35	Ro12-6377	CH ₃	$R_3, R_5, R_6 = H$	NHCONHCH ₃	$R_8, R_9 = H$	F	Н	6.34
36	Halazepam	CH ₂ CF ₃	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Н	Н	7.04
37	Pinazepam	CH ₂ CCH	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Н	Н	7.03
38	Prazepam	CH ₂ C ₃ H ₅	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Н	Н	6.96
39	Motrazepam	CH ₂ OCH ₃	$R_3, R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	Н	Н	6.37

Table 1 (continued)



N°	Name	R ₁	R ₃ / R ₅ /R ₆	R ₇	R ₈ / R ₉	R ₂ ,	R _{6'}	Log (1/C) _{obs}
40	Ro20-1310	C(CH ₃) ₃	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Н	Н	6.21
41	Ro07-2750	CH ₂ CH ₂ OH	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	F	Н	7.61
42	Ro08-9013	(CH ₂) ₂ OCH ₂ CONH ₂	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	F	Н	7.37
43	Proflazepam	CH ₂ CHOHCH ₂ OH	$R_3, R_5, R_6 = H$	Cl	$R_8, R_9 = H$	F	Н	6.85
44	Ro22-4683	C(CH ₃) ₃	$R_3, R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	Cl	Н	6.52
45	Ro11-4878	Н	$R_3 = (s)CH_3$ R_5,R_6=H	Cl	$R_8, R_9 = H$	F	Η	8.46
46	Meclonazepam	Н	$R_3 = (s)CH_3$ $R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	Cl	Η	8.92
47	Ro11-6896	CH ₃	$R_3 = (s)CH_3$ $R_5, R_6 = H$	NO ₂	$R_8, R_9 = H$	F	Η	8.15
48	L48	CH ₃	$R_3 = (rac)CH_3$ $R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Н	Η	7.31
49	Temazepam	CH ₃	$R_3 = (rac)OH R_5, R_6 = H$	Cl	$R_8, R_9 = H$	Н	Η	7.79
50	L50	CH ₃	$R_3 = (rac)Cl R_5, R_6 = H$	Cl	$R_8, R_9 = H$	F	Η	8.27
51	L51	Н	$R_3, R_5 = H R_6 = CH_3$	CH ₃	$R_8, R_9 = H$	Н	Н	6.77
52	Ro07-4419	Н	$R_3, R_5, R_6 = H$	Н	$R_8, R_9 = H$	F	F	7.72
53	Ro05-4520	CH ₃	$R_3, R_5, R_6 = H$	Н	$R_8, R_9 = H$	F	Н	7.85
54	Ro05-4608	CH ₃	$R_3, R_5, R_6 = H$	Н	$R_8, R_9 = H$	Cl	Н	8.42
55	Ro05-3546	Н	$R_3, R_5 = H R_6 = Cl$	Н	$R_8, R_9 = H$	Η	Н	6.49
56	Ro13-0699	CH ₃	$R_3, R_5 = H R_6 = Cl$	Н	$R_8, R_9 = H$	F	Н	6.82
57	Ro07-6198	Н	$R_3, R_5, R_6 = H$	Н	$R_8 = Cl R_9 = H$	F	F	7.55
58	Ro20-8895	Н	$R_3, R_5, R_6 = H$	Н	$R_8 = CH_3$ $R_9 = H$	F	Η	7.72
59	Ro13-0593	CH ₃	$R_3, R_5, R_6 = H$	Н	$R_8 = H R_9 = Cl$	F	Н	7.14
60	L60	CH ₃	$R_3, R_5 = H R_6 = Cl$	Н	$R_8 = Cl R_9 = H$	F	Η	6.52
61	Ro22-6762	CH ₃	$R_3, R_5, R_6 = H$	Cl	$R_8 = Cl R_9 = H$	Н	Η	7.40
62	Ro20-8065	Н	$R_3, R_5, R_6 = H$	Cl	$R_8 = Cl R_9 = H$	F	Н	8.44
63	Ro20-8552	Н	$R_3, R_5, R_6 = H$	CH ₃	$R_8 = Cl R_9 = H$	F	Η	7.85
64	L64	Н	$R_3, R_5, R_6 = H$	Cl	$R_8 = H R_9 = Cl$	Η	Н	7.43
65	L65	Н	$R_3, R_5, R_6 = H$	Cl	$R_8 = H$ $R_9 = CH_3$	Н	Н	7.28

the molecular geometries were optimized at the DFT/ Ub3lyp/ $6-311^{++}G(d,p)$ level of theory using the Gaussian 09 W software [19]. For all stationary points, there is no imaginary frequency at the optimized molecular geometries ensuring that the optimized structures are at the minimum on the potential energy surface. The atomic charges q_{N1} , $q_{C3},\,q_{N4},\,q_{C6},\,q_{C7},\,q_{C8},\,q_{C9},\,q_{C2'},\,q_{C6'},\,and$ the dipole moment (DM) have been assessed using the ChelpG electronic population scheme [20]. The number of hydrogen-donors (HD), number of hydrogen-acceptors (HA), molecular lipophilicity (Log P), and molar refractivity $(M_{\rm P})$ have been computed using MarvinSketchversion 20.21 [21]. The flexible torsions $(F_{\rm T})$ have been computed using Molegro Virtual Docker version 5.5 [22]. Finally, the hydrophobic constants in positions 7 (π_{C7}) and 2' ($\pi_{C2'}$) have been extracted from the literature [23].

QSAR analysis

PLS regression

The chemical, physical, topological, and quantum properties are necessarily correlated for a given molecule. This is a reflection of the innate properties of the system, and additional data collected in the same way will show the same collinearity [24]. Indeed, PLS regression is a useful method for multivariate data containing correlated molecular descriptors. This method based on dimension reduction technique builds orthogonal components, often called factors or latent variables, as linear combinations of the original predictor variables [25]. PLS constructs these components while considering the observed response values, leading to a parsimonious model with reliable predictive power [26].

In this work, the number of components used in PLS is chosen by a fivefold cross-validation method [27]. The dataset is randomly divided into training dataset (80%) and testing dataset (20%). Training sets were used for model development and test sets for model external validation. Before conducting PLS analysis, each response variable is scaled to unit variance by dividing it by its standard deviation, and the molecular descriptors are centered by subtracting the average value and scaled to unit variance. First, variable selection by stepwise regression method is used to identify the best subset of molecular descriptors. This is a combination of backward and forward selection [25]. The objective is to use the minimum number of descriptors to develop a good predictive model. Thus, we must select the good subsets of descriptors. However, it should be noted that the subset of molecular predictors that do the best at meeting welldefined objective criteria can be highly variables depending on precisely which observations are included in the training set. In addition, the best training model does not necessarily guarantee a better quality of prediction. This depends on the

training and test sets obtained from the original dataset. For this reason, we conducted a statistical simulation for which 10,000 splits were performed resulting in 10,000 training and test sets. For each simulation, regression diagnostics for detecting possible outliers was carried out by computing leverage values (h_{ii}) for identifying outlying x-variables and studentized deleted residuals (r_i^*) for identifying outlying y -variables. Note that once outliers have been detected the model is regenerated excluding the outlying observations from the dataset. Thereafter, the best model is selected on each training set resulting in 10,000 best training models following the Bayesian Information Criterion (BIC) [28]. This criterion is chosen because it penalizes larger models more heavily and will tend to select a smaller subset of descriptors in comparison to other criteria [29]. The best choice of descriptors will balance fit with model size. Subsequently, among these 10,000 best models, we sought to select the best molecular descriptors according to the highest probability of their occurrence.

To verify a model's predictive ability, the developed QSAR model is quantified using the coefficient of determination (R^2) [30], the adjusted coefficient of determination (R^{2}_{adj}) [31], and the Fisher statistics (*F*) [32]. This latter is computed to judge the overall significance of the regression model. The external predictive ability of the developed QSAR model is determined by computing the leave-one-out cross-validation coefficient (Q^2_{Foo}) [32] and the predictive squared correlation coefficient (Q^2_{F3}) [33–35]. The external validation ensures the predictability of the developed QSAR model for the prediction of untested molecules [13].

Molecular docking protocol

The electron microscopy structure of the human GABA_A receptor $\alpha_1\beta_2\gamma_2$ subtypes in complex with GABA plus the DZP structures (PDB ID:6X3X, Resolution = 2.92 Å) was downloaded from RCSB Database (http://www.rcsb.org). The downloaded PDB file contains nine chains: five chains denote the subunits α_1 (B and D), β_2 (A and C), and γ_2 (E), and four chains denote the Fab-chains (named from I to K).

MOE 2014.0901 software package [36] was used to prepare the four benzodiazepine binding sites: the classical site at the ECDD⁺/E⁻ chain interface and the three TMD sites at the A⁺/ B⁻, C⁺/D⁻, and E⁺/A⁻ chain interfaces. For each binding site, all co-crystallized ligands and non-essential subunits were removed from the $\alpha_1\beta_2\gamma_2$ -DZP complex. Then, after structure correction, protonation at neutral medium (PH=7), and cavity detection, the native co-crystallized DZP structure was re-docked into the selected binding site pocket. The method was validated by giving the best binding pose which has a low RMSD value (root-mean-square deviation). The BDZ structures previously optimized using the DFT method were converted into databaseinput files and docked one by one into the four DZP binding pockets using the semi-flexible docking process of MOE 2014.0901 package [36]. During the process, the conformation of the receptor was fixed, while ligands remained flexible. Here, the best binding poses were selected according to the lowest energy score values, registered in the PDB file, and visualized using BIOVIA Discovery Studio visualizer v20.1.0.19295 package [37].

Molecular dynamics protocol

The best-docked ligand/ $\alpha_1\beta_2\gamma_2$ complexes were subjected to stability tests using molecular dynamics simulation (MD). MD simulation was implemented using the "compute" option included in the MOE 2014.0901 software [36]. First, the selected complexes were prepared by deleting the DZPcocrystallized structure, fixing hydrogens, and fixing charges. Thereafter, the parameters of the "dynamics" tool were adjusted to execute the combination Nosé-Poincaré-Andersen (NPA) algorithm/Merck molecular force field (MMFF94x) [38, 39], with enabling bonding, van der Waals, electrostatics, and restraints. The protocols were settled for an equilibrium period of 100 ps followed by a production period of 900 ps, at a constant temperature of 310 K. Finally, the variations in potential energies, U (kcal/mol), as a function of time, *t* (ps), are retained and plotted using Origin 6.0 software [40].

Pharmacokinetics/drug-likeness prediction

In silico estimation of pharmacokinetic properties and prediction of drug-likeness were carried out by using the free web tool SwissADME [41]. Our study is based on the prediction of the following pharmacokinetic parameters: gastrointestinal absorption (GI), P-glycoprotein (P-gp) substrate, blood-brain barrier (BBB) penetration, and cytochrome enzyme (CYP) inhibition. Indeed, out of 57 human CYP450 enzymes, the CYP1A2, CYP2C9, CYP2C19, CYP2D6, CYP3A4, and CYP2E1 metabolize 90% of drugs [42]. In addition, the drug-likeness prediction is based on several rules such as Lipinski, Ghose, Veber, Egan, and Muegge.

Results and discussion

Molecular docking simulation

Orientations, interactions, and binding affinities of 65 positive allosteric modulators of GABA_ARs (BDZ dataset, Table 1) were investigated in four distinct BDZ binding sites: the classical site at the ECD α_1^+/γ_2^- interface, the TMD sites at the two $\beta_2^+(A)/\alpha_1^-(B)$ and $\beta_2^+(C)/\alpha_1^-(D)$ interfaces and the TMD site at the γ_2^+/β_2^- interface (Fig. 1). Residues involved in each active pocket were detected using the site finder wizard implemented in the MOE 2014.0901 software [36] (Table 2). During the docking process, the conformation of residues remains unchanged, while ligands are altered. The best binding modes of re-docked DZPs were selected based on the given root-mean-square deviation values (RMSD). Generally, docking protocols that are able to generate the same co-crystallized binding modes with an RMSD value of less than 1.5 or 2 A° (depending on ligand size) or even better, less than 1 A° are considered validated [43, 44], whereas the best binding modes of docked ligands were selected according to the given S_{score} values. The binding free energy score (S_{score}) is a quantitative estimate of the most stable binding pose between the target macromolecule and ligand. Among the generated poses, the best ones are those with the most negative energy score values.

 Table 2
 Residues involved in each active pocket

Bindi	ng site	Residues involved in active pockets
ECD	α_1^+/γ_2^- interface (classical binding site)	 (PHE100 PHE101 HIS102 ASN103 GLU138 PRO140 PRO154 LYS156 SER159 TYR160 ALA161 VAL203 GLN204 SER205 SER206 THR207 TYR210) (ASP56 MET57 TYR58 ASN60 SER61 ASP75 PHE77 ALA79 MET130 THR142 ARG144 SER186 GLU189 ASP192 SER195)
TMD	$\beta_2^+(A)/\alpha_1^-(B)$ interface	 (ILE255 VAL258 LEU259 MET261 THR262 ASN265 THR266 ARG269 GLU270 ASP282 LEU285 MET286 PHE289 VAL290) (ILE228 GLN229 LEU232 PRO233 MET236 THR237 LEU240 PHE258 THR261 THR262 THR265 LEU269 SER272)
	$\beta_2^+(C)\!/\!\alpha_1^-(D)$ interface	1: (MET261 THR262 ASN265 THR266 ARG269 ASP282 LEU285 MET286 PHE289 VAL290) 2: (VAL227 ILE228 LEU232 PRO233 MET236 THR237 THR265 LEU269)
	γ_2^{+}/β_2^{-} interface	 (TYR220 PHE221 LEU223 GLN224 THR225 MET227 PRO228 LEU231 ILE232 THR263 ILE264 HIS267 LEU268 THR271 LEU272) (MET276 THR277 SER280 THR281 ALA283 ARG284 LYS285 LYS289 MET296 ASP297 VAL300 SER301 PHE304 ILE305)

Surveying the first five ligands having the lowest energy scores (Table 3), the lowest S_{score} value in the three α_1^+/γ_2^- , $\beta_2^+(A)/\alpha_1^-(B)$, and γ_2^+/β_2^- sites was assigned to Ro12-6377. Also, Ro12-6377 exhibits the second-lowest S_{score} value in the TMD $\beta_2^+(C)/\alpha_1^-(D)$ interface. Therefore, out of the 65 studied ligands, Ro12-6377 is the ligand that exhibited the highest predicted affinity towards the four sites, simultaneously. By similar reasoning, the second highest predicted affinity was rated to proflazepam.

By comparing the affinities of Ro12-6377 towards the four sites, the α_1^+/γ_2^- site was defined to be the principal target for Ro12-6377. So, when a high dose of Ro12-6377 is administered, it mainly acts to bind at the ECD α_1^+/γ_2^- interface as it is the high-affinity binding site, and then, acts to bind at the TMD γ_2^+/β_2^- and the two β_2^+/α_1^- interfaces as they are the second and third high-affinity binding sites, respectively. On the other hand, proflazepam tends to fill thetwo β_2^+/α_1^- interfaces before moving to occupy the γ_2^+/β_2^- site. Recently, from a database of 7922 compounds, proflazepam was selected among the top-100 docked ligands at the binding pocket of SARS-CoV-2 Main Protease crystallized in holo-form [45].

Likewise, the affinities of the remaining 63 ligands (Table S1, Supplementary materials) towards the four binding sites were compared. Unexpectedly, our results are inconsistent with previous findings indicating that the classical site is always the main target of all classical benzodiazepines. As we can see, $\beta_2^+(A)/\alpha_1^-(B)$ and γ_2^+/β_2^- are expected to be, respectively, the main binding sites for (Ro05-4336, Ro07-2750, Ro05-3546) and (L₅₁,Ro13-0699). Moreover, in some cases, both ECD and TMD sites share the same binding affinity towards the bound ligand, as is evident for Ro05-2921, Ro20-5397, Pinazepam, Ro07-4419, Ro05-3546, and

Ro20-8552, which makes it difficult to distinguish, accurately, the main target for the binding. Consequently, after the discovery of the three TMD sites, it became necessary to expand previous findings indicating that the main target of classical benzodiazepines is always located at the ECD α_1^+/γ_2^- interface.

The binding modes of both ligands in each of the four binding interfaces are located at the same level as the cocrystallized DZP (Figs. S1 and S2, Supplementary materials). However, their docking orientations are markedly not equivalents, with the exception of the binding orientation of proflazepam in the γ_2^+/β_2^- interface where it shares an almost perfect superimposition to the DZP-bound structure (Fig. S2 (d) in Supplementary materials). The favorable drug binding pose is determined by the distribution of polar and non-polar regions along the surface of the ligand and its complementary target binding site. Thus, while the non-polar regions create hydrophobic interactions mainly contributing to the binding affinity of the drug towards the biological target, polar regions create electrostatic points contributing to modulating the drug binding kinetics (specificity and orientation) [46]. In order to estimate all possible interactions, the docking-outputs generated by MOE software were converted into (.pdb) files and visualized with the default parameters of BIOVIA DS visualizer v20.1.0.19295 package [37]. As can be seen, the binding interactions for both ligands with the four target-sites residues exhibit the formation of four types of interactions, most of which are of type hydrogen bonds and hydrophobic interactions. The standard values of distances and energy cutoffs for considering the formation of hydrogen bonds with specific target residues are categorized into three subtypes: strong bonds (2.2-2.5 Å, E: 14-40 kcal/ mol), moderate bonds (2.5-3.2 Å, E: 4-15 kcal/mol), and weak bonds (3.2–4.0 Å, E > 4 kcal/mol) [47]. Generally, strong bonds in targeted-ligand interactions are undesirables as they mostly tend to have a covalent character. This later hinders the process of drug liberation from its receptor, thus increasing the risk of drug toxicity. Nevertheless, new insights on how to address the pharmacological advantages/potential risks balance of covalent drugs

Table 3The first five ligandshaving the highest bindingaffinity for the four bindinginterfaces

ECD α_1^+/γ_2^- interface	ce	$ TMD \ \beta_2^+(A) / \alpha_1^-(B) \\ interface $		$TMD \ \beta_2^+(C)/\alpha_1^-(D)$ interface		TMD γ_2^+/β_2^- interface	
Ligand	S _{score}	Ligand	S _{score}	Ligand	S _{score}	Ligand	S _{score}
Ro12-6377	-8.013	Ro12-6377	-7.409	Ro08-9013	-7.455	Ro12-6377	-7.546
Meclonazepam	-7.856	Proflazepam	-7.170	Ro12-6377	-7.265	Ro08-9013	-7.070
Methylclonazepam	-7.831	Ro07-2750	-7.152	Proflazepam	-7.077	Pinazepam	-6.990
Proflazepam	-7.807	Pinazepam	-7.146	Pinazepam	-6.848	Proflazepam	-6.972
Ro11-6896	-7.785	Motrazepam	-7.087	Meclonazepam	-6.774	Ro07-2750	-6.918

have been emerged and discussed in the literature [48, 49]. Referencing to this latter, several standard values of distance cutoffs to consider the formations of hydrophobic interactions have been found. Janiak [50] suggested that the optimum range is between 3.3 and 3.8 A° , while other researchers have suggested a relatively higher range [51–53].

Owing to the lack of previous experimental and theoretical studies on Ro12-6377 and proflazepam, we will attempt to suggest mechanisms for how they modulate $\alpha_1\beta_2\gamma_2Rs$ signaling by binding at the ECD and the TMD interfaces, based on the results obtained from molecular docking and molecular dynamics simulation.

Allosteric modulation of the classical binding site

The bulky structure of Ro12-6377, compared toDZP, enabled the NHCONHCH₃ group attached at C₇ and both phenyls (B) and (C) to penetrate deeper into the binding site, whereas the diazepine ring exerts its effect by interacting with the residues located in front of the pocket (Figs. 2a and 3a and Table 4). The methyl group attached at N₁ forms two hydrophobic interactions type Alkyl-Alkyl and Pi-Alkyl with α_1 Val203 and γ_2 Tyr58, respectively. Concurrently, the π -electron clouds of γ_2 Tyr58 and γ_2 Phe77 were involved in two Pi-Pi stacked interactions with the π -electron cloud of phenyl (B). The linear backbone skeleton of NHCONHCH₃



Fig. 2 Binding modes resulting from the molecular docking of Ro12-6377and proflazepam at the interfaces of **a** ECD α_1^+/γ_2^- , **b** TMD $\beta_2^+(A) / \alpha_1^-(B)$, **c** TMD $\beta_2^+(C)/\alpha_1^-(D)$, and **d** TMD γ_2^+/β_2^-



Fig. 3 Binding interactions resulting from the molecular docking of Ro12-6377 at the interfaces of **a** ECD α_1^+/γ_2^- , **b** TMD $\beta_2^+(A)/\alpha_1^-(B)$, **c** TMD $\beta_2^+(C)/\alpha_1^-(D)$, and **d** TMD γ_2^+/β_2^-

group formed an Alkyl-Alkyl interaction with the side chain of γ_2 Ala79, received two moderate H-bonds from α_1 Ser206side chain, and gave two moderate H-bonds to the main carbonyl of γ_2 Phe77. The four moderate H-bonds lead to the formation of two intermolecular pentameric rings that contribute to the stability of Ro12-6377 at the binding site. Also, in addition to the strongH-donor bond received from α_1 His102, an intramolecular interaction was observed between the fluorine atom bound at C₂ andC₂. This interaction could be explained by the strong withdrawing property of the oxygen that madeC₂ a more electrophilic center able to receive the nucleophilic attacks. The pendant phenyl (C)

ECD α_1^+	γ_2^- interfac	e						
Ligand name	Posit ^a	Category	Type of interactions	From	From chemistry	То	To chemistry	Dist ^b (A°)
DZP	2	Hydrogen	Conventional	D:SER206:H	H-donor	D:DZP404:O1	H-acceptor	2.68
	4	bond	H-bond	D:DZP404:H151		D:SER205:OG		2.55
	2		C-H bond	D:SER205:HA		D:DZP404:O1		2.65
	1			D:DZP404:H202		D:GLN204:O		2.94
	4	Hydrogen bond; electrostatic	Pi-Cation; Pi-Donor H-bond	D:DZP404:H152	Positive; H-donor	E:PHE77	Pi-orbitals; Pi-orbitals	3.03
	Ring C	Hydropho-	Pi-Pi stacked	D:TYR160	Pi-orbitals	D:DZP404	Pi-orbitals	5.24
		bic		D:TYR210		D:DZP404		4.22
				E:PHE77		D:DZP404		5.49
			Pi-Pi T-shaped	D:PHE100		D:DZP404		5.36
	1		Alkyl	D:DZP404:C20	Alkyl	D:VAL203	Alkyl	5.21
	7		Pi-Alkyl	D:HIS102	Pi-orbitals	D:DZP404:CL		4.32
	1			E:TYR58		D:DZP404:C20		4.37
	7			E:PHE77		D:DZP404:CL		5.12
	Ring B			D:DZP404		D:VAL203		5.24
Ro12-	7	Hydrogen bond	Conventional	D:SER206:HG	H-donor	:*0:O	H-acceptor	2.57
6377			H-bond	:*0:H		E:PHE77:O		3.06
	2'		C-H bond	D:HIS102:HE1		:*0:F		2.47
	7			D:SER206:HB3		:*0:O		2.70
				:*0:H		E:PHE77:O		2.98
	Internal ^c	Halogen	Halogen (Fluorine)	:*0:C	Halogen acceptor	:*0:F	Halogen	3.66
	Ring B	Hydropho-	Pi-Pi stacked	E:TYR58	Pi-orbitals	:*0	Pi-orbitals	3.93
		bic		E:PHE77		:*0		5.00
	Ring C		Pi-Pi T-shaped	D:PHE100		:*0		5.23
	7		Alkyl	E:ALA79	Alkyl	:*0:C	Alkyl	3.55
	1			:*0:C		D:VAL203		4.55
			Pi-Alkyl	E:TYR58	Pi-orbitals	:*0:C		4.88
PLZ	1	Hydrogen	Conventional	D:THR207:HG1	H-donor	:*0:O	H-acceptor	2.10
		bond	H-bond	D:TYR210:HH		:*0:O		2.85
	4		C-H bond	D:HIS102:HE1		:*0:N		2.57
	Ring B	Hydropho-	Pi-Pi stacked	E:TYR58	Pi-orbitals	:*0	Pi-orbitals	4.73
		bic		E:PHE77		:*0		4.55
	7		Pi-Alkyl	E:TYR58		:*0:CL	Alkyl	3.43

Table 4Detailed binding interactions resulting from the molecular docking of co-crystallized DZP, Ro12-6377, and proflazepam (PLZ) at the
classical site

^aPosition of interaction

^bdistance of interaction

^cinternal interaction

is involved deeper into the DZP pocket where it is delineated by the hydrophobic side chains of γ_2 Phe77, α_1 Phe100, α_1 His102, α_1 Ter160, and α_1 Ter210. Here, a hydrophobic interaction type Pi-Pi T-shaped was observed between the π -electron cloud of ring (C) and the surrounded side chain of α_1 Phe100. Compared with Ro12-6377 and DZP, proflazepam penetrates less within the binding site (Figs. 2a and 4a and Table 4). Phenyl (C) is oriented outside the binding pocket, which explains the absence of any interaction on this pendant ring. Rings (A) and (B) seem to selectively accept rather than accept and donate bonds to the residues



Fig. 4 Binding interactions resulting from the molecular docking of proflazepam at the interfaces of **a** ECD α_1^+/γ_2^- , **b** TMD $\beta_2^+(A)/\alpha_1^-(B)$, **c** TMD $\beta_2^+(C)/\alpha_1^-(D)$, and **d** TMD γ_2^+/β_2^-

of subunits. The α_1 subunit acts through three H-bonds; two are oriented from the hydroxyl groups at the side chains of Thr207 and Tyr210 towards the first hydroxyl group of the CH₂CHOHCH₂OH substitute, and one is oriented from the side chain of His102 towards N₄. The γ_2 subunit prefers to act hydrophobically through two types of interactions: Pi-Pi stacked and Pi-Alkyl. The Pi-Pi stacked is created between the π -electron cloud of phenyl (B) and the Tyr58 and Phe77 side chains, while the Pi-Alkyl interaction links the chlorine atom at C_7 to the side chain of Tyr58.

The binding modes of Ro12-6377 and proflazepam elucidated here agree with the previous finding indicating the importance of His102 in the recognition of classical benzodiazepines. With the exception of the His102Cys mutant in the α_5 subunit, GABA_ARs that contain α His102 mutation to any other residue suffer from total insensitivity to the DZP and its analogs [2]. The topological organization of the $\alpha_4\beta\gamma_2$ and $\alpha_6\beta\gamma_2$ receptors reveals the presence of a natural substitution of His102 by the Arg residue, which leads to steric problems affecting the binding of classical BDZ at their correspondent binding locus. This could explain the selectivity of classical BDZ towards $\alpha\beta\gamma_2$ Rs containing the $\alpha_1, \alpha_2, \alpha_3$, and α_5 subunits rather than those containing the α_4 and α_6 subunits [8, 9]. Else, the γ_2 Phe77Tyr mutant affects less the binding affinity of DZP but more strongly reduces that of its analogs containing the chlorine substitutes at the pendant phenyl (C). This finding was explained by the difference in flexibility between the two pendant phenyls since the presence of the chlorine atoms possibly caused unfavorable steric clashes with the side chain of the tyrosine residue. Furthermore, several other mutation findings were surveyed in detail in the previous researches [54–56].

Allosteric modulation of the three TMD binding sites

Despite the common topological organization between the combinations $\beta_2(A)/\alpha_1(B)$ and $\beta_2(C)/\alpha_1(D)$, the two TMD orthosteric pockets inserted at $\beta_2^+(A)/\alpha_1^-(B)$ and $\beta_2^+(C)/\alpha_1^-(D)$ interfaces are not qualitatively identical and thus may not be functionally equivalent. The binding pocket at $\beta_2^+(A)/\alpha_1^-(B)$ interface was rated as the largest, and estimated to have a higher affinity for Ro12-6377 and proflazepam than the binding pocket at $\beta_2^+(C)/\alpha_1^-(D)$ interface.

The binding modes of Ro12-6377 for both TMD $\beta_2^+/\alpha_1^$ interfaces are equivalents: the NHCONHCH₃ groups and the fused benzodiazepine rings (A and B) are deeply embedded in the binding pockets, whereas the pendant phenyl rings (C) interact with the residues in front of the pockets (Fig. 2b, c). By superposing the binding modes, a rotation of 98.8° was observed between the two poses of the phenyl rings (C) (Fig. S3, Supplementary materials). The binding interactions at $\beta_2^+(C)/\alpha_1^-(D)$ interface exhibit the formation of halogen interaction between the bound fluorine atom at C2' and N1 (Table 6). This intramolecular interaction leads the fluorine to orient towards the principal chain of β_2 Met286 where it stabilizes through the formation of a strong H-bond interaction (Fig. 3c), and consequently, leads the phenyl ring (C) to deviate by an angle of 98.8° from the phenyl plane observed at the $\beta_2^+(A)/\alpha_1^-(B)$ interface (Fig. 3b). At this time, the binding interactions exhibit the formation of additional electrostatic and hydrophobic interactions between the substitute groups at N_1 and C_7 and the side chains of four residues from the β2 subunit: Val258, Leu285, Met286, and Phe289 (Table 5).

Unlike Ro12-6377, the binding modes of proflazepam for both TMD $\beta_2^+(A)/\alpha_1^-(B)$ and $\beta_2^+(C)/\alpha_1^-(D)$ interfaces reveal significant differences in the docking orientations and binding interactions (Fig. 2b, c). At $\beta_2^+(A)/\alpha_1^-(B)$ interface (Fig. 4b and Table 5), the chlorobenzene ring (B) was oriented to interact hydrophobically with side chains of residues located in front of the pocket: β_2 Met286, α_1 Met236, and α_1 Leu232. The pendant phenyl ring (C) was embedded deeper into the binding pocket, in a manner similar to that observed inDZP, and shares Pi-Alkyl interaction at a distance of 4.28A° with side chain of α_1 Pro233. The diazepine ring (A) established a moderate H-bond between one of the hydrogens of C₃ and the main carbonyl of β_2 Leu285. The CH₂CHOHCH₂OH group was oriented towards the ECD where its first hydroxyl group was stabilized by strong intramolecular interaction type H-donor with lone pairs of the oxygen bond at C2, and with three intermolecular H-bonds formed with side chains of Arg269 and Asn265 of subunit β_2 . At $\beta_2^+(C)/\alpha_1^-(D)$ interface (Fig. 4c and Table 6), the pendant phenyl (C) is located at a higher level than rings (A) and (B), with its fluorine atom pointing towards the front of the binding site. This orientation drives the π -electron cloud into Pi-Alkyl interactions with side chains of Pro233 and Leu269 of subunit α_1 , as well as driven the Lone pairs of the fluorine atom to form halogen-halogen interaction with the main oxygen of α_1 lle228 and receive a strong H-bond from the side chain of β_2 Asn265. This latter also shares a strong H-bond with N₄. Phenyl (B) is located between the side rings of β_2 Phe289 and α_1 Pro233, which contributes to the formation of two hydrophobic interaction types Pi-Pi stacked and Pi-Alkyl. Its chlorine atom is oriented towards the bottom of the binding pocket, more precisely towards Thr265 of α_1 subunit. Here, no interactions were observed. The CH₂CHOHCH₂OH group exerts its influence by occupying the front of the binding pocket and sharing two H-bonds with α_1 Ile228 and α_1 Pro233. The length of its backbone skeleton also leads to forming two additional intramolecular bonds with the oxygen atom at C_2 and the fluorine at C_2 .

By examining the binding mode of Ro12-6377 at the γ_2^+ $/\beta_2^-$ interface (Figs. 2d and 3d and Table 7), both the diazepine and phenyl (C) are positioned in front of the pocket. The methyl group attached at N1 forms two types of interactions. The first is a moderate H-donor bond which is given to the carbonyl of γ_2 Ser280. The seconds are three hydrophobic bond type Alkyl-Alkyl formed with side chains of γ_2 Ala283, γ_2 Arg284, and γ_2 Val300. Simultaneously, γ_2 Asp297 established two hydrogen bonds with the diazepine ring (A); strong H-donor has given to the oxygen attached at C2, and moderate H-acceptor bond received from C₃. The phenyl (C) was involved in two hydrophobic interactions; type Pi-PiTshaped with side chain of γ_2 Phe304, and type Pi-Alkyl with the side chain of β_2 Pro228. Its fluorine atom was oriented toward TM_1 : β_2 subunit, which resulted in halogen interaction with the lone pair of the oxygen situated in the main chain of Leu223, and in moderate H-acceptor bond forms with the side chain of Pro228. On the other hand, the phenyl (B) was inserted deeper in the binding pocket as its π -electron cloud shared Pi-Alkyl interaction with side chain of β_2 Pro228. Its

Table 5 Detailed binding interactions resulting from the molecular docking of co-crystallized DZP, Ro12-6377, and proflazepam (PLZ) at the TMD $\beta_2^+(A)\alpha_1^-(B)$ interface

TMD β_2^+	$(\mathbf{A})/\alpha_1^-(\mathbf{B})$ i	nterface						
Ligand Name	Posit ^a	Category	Type of interactions	From	From chemistry	То	To chemistry	Dist ^b (A°)
DZP	1	Hydrogen	C-H bond	A:DZP406:H201	H-donor	B:ILE228:O	H-acceptor	2.93
		bond		A:DZP406:H202		B:ILE228:O		2.89
	4	Electro- static	Pi-Cation	A:DZP406:N15	Positive	A:PHE289	Pi-orbitals	4.22
	1	Hydro-	Alkyl	A:DZP406:C20	Alkyl	A:MET286	Alkyl	4.50
		phobic		A:DZP406:C20		B:ILE228		4.46
	7			A:DZP406:CL		A:MET261		4.42
				A:DZP406:CL		A:LEU285		4.79
			Pi-Alkyl	A:PHE289	Pi-orbitals	A:DZP406:CL		4.86
	Ring C			A:DZP406		B:PRO233		4.28
Ro12-	3	Hydrogen	C-H bond	:*0:H	H-donor	B:ILE228:O	H-acceptor	2.78
6377		bond		:*0:H		B:ILE228:O		2.89
	1			:*0:H		A:LEU285:O		3.04
	7			:*0:H		B:THR237:OG1		2.51
	Ring B	Hydro-	Pi-Pi stacked	A:PHE289	Pi-orbitals	:*0	Pi-orbitals	4.11
	1	phobic	Alkyl	:*0:C	Alkyl	A:LEU285	Alkyl	4.64
				:*0:C		A:MET286		5.39
	7			:*0:C		A:VAL258		5.37
				:*0:C		B:LEU240		5.04
			Pi-Alkyl	A:PHE289	Pi-orbitals	:*0:C		5.30
	Ring B			:*0		B:PRO233		4.65
	Ring C			:*0		A:MET286		4.67
				:*0		B:LEU232		5.20
				:*0		B:MET236		4.66
PLZ	1	Hydrogen	Conventional H-bond	A:ARG269:HH12	H-donor	:*0:O	H-acceptor	2.10
		bond		:*0:H		A:ASN265:OD1		1.97
	Internal ^c			:*0:H		:*0:O		2.44
	1		C-H bond	A:ARG269:HD3		:*0:O		2.72
	3			:*0:H		A:LEU285:O		2.90
	7	Hydro-	Alkyl	:*0:CL	Alkyl	B:LEU232	Alkyl	4.20
		phobic		:*0:CL		B:MET236		3.73
	Ring B		Pi-Alkyl	:*0	Pi-orbitals	A:MET286		4.91
				:*0		B:LEU232		5.42
	Ring C			:*0		B:PRO233		4.28

^aPosition of interaction

^bdistance of interaction

cinternal interaction

NHCONHCH₃ group attached at C₇ was pointed towards the TM₂: β_2 helix. In addition to the two Alkyl-Alkyl interactions shared with side chains of Pro228 and Ile264 of β_2 subunit, the NHCONHCH₃ group also received one moderate H-bond from side chain of γ_2 Thr277 and simultaneously gave four moderate H-bonds, three of which to β_2 Gln224, and thus, leads to the formation of two intermolecular pentameric rings and one butameric ring contribute, as in the classical site, to the stability of Ro12-6377 at the binding site.

Proflazepam adopted a similar binding mode of DZP in the binding locus (Fig. S2 (d), Supplementary material). The binding mode of DZP at the γ_2^+/β_2^- interface was previously discussed in detail by Kim et al. [7]. The uncommon

Table 6	Detailed binding	interactions	resulting from	the molecula	ar docking o	f co-crystallize	ed DZP	, Ro12-6377,	, and proflazepan	n (PLZ)) at the
TMD β_2^+	$\frac{1}{2}(C)/\alpha_1^{-}(D)$ interface	ce									

Ligand Name Posit' Category Type of interactions From From chemistry To To chemistry Disk' (A') DZP 1 Hydrogen Bond C-H bond C:DZP406:H201 H-donor D:ILE228:0 H-acceptor 3.88 4 Hydrogen I bond Pi-Cation; Pi-Donor C:DZP406:H202 Positive; H-donor C:PHE289 Pi-orbitals 2.65 1 Biod Alkyl C:DZP406:C20 D:ILE228:0 H-orbitals 2.60 1 Bic Alkyl C:DZP406:C20 D:ILE228 4.46 C:DZP406:C20 D:ILE228 4.482 C:DZP406:C20 D:ILE228 4.42 7 C:DZP406:C20 D:ILE228 4.41 4.41 4.41 7 C:DZP406:CL C:MET286 5.39 5.30 5.30 8ng C Fi-Alkyl C:HIE289 Pi-orbitals C:DZP406 C:MET286 5.39 8ng B Hydrogen C:H bond :0CH H-donor D:LE228:0 4.31 1 bo	TMD β_2^+	$(\mathbf{C})/\alpha_1^-(\mathbf{D})$ in	terface						
DZP 1 Hydrogen Bond C-H bond C:DZP406:H201 C:DZP406:H202 H-donor D:ILE228:0 DILE228:0 H-acceptor 3.08 2.56 4 Hydrogen Islatic Pi-Cation; Bond; Electron C:DZP406:H202 D:ILE228:0 Pi-orbitals 2.56 1 Bic bic H-bond C:DZP406:H202 Dille228:0 Pi-orbitals 2.60 1 Bic Alky1 C:DZP406:C20 Alky1 C:MET286 Alky1 4.68 7 C:DZP406:C20 D:ILE228 4.76 4.61 4.61 4.61 7 C:DZP406:C20 D:ILE228 4.82 4.61 4	Ligand Name	Posit ^a	Category	Type of interactions	From	From chemistry	То	To chemistry	Dist ^b (A°)
Image: Participant in the section of the sectin of the section of the section of the section of the sec	DZP	1	Hydrogen	C-H bond	C:DZP406:H201	H-donor	D:ILE228:O	H-acceptor	3.08
4 Hydrogen band; band; bic Pi-Cation; pi-Donoin static C:DZP406:1152 Positive; H-donor C:PHE289 Pi-orbitals 2.65 Ring C Hydropho Pi-Sigma D:PR0233:HB2 C-H C:DZP406 Pi-orbitals 2.60 1 bic Alkyl C:DZP406:C20 Alkyl C:MET286 Alkyl 4.68 1 bic Alkyl C:DZP406:C20 D:LE228 4.76 7 F C:DZP406:C20 D:LE0285 4.82 7 F C:DZP406:C20 D:LE0285 4.41 7 F C:DZP406:C20 C:LE0264 4.99 Ring C Pi-Alkyl C:PHE289 Pi-orbitals C:DZP406:C1 C:LE026 5.39 Ring E F C:DZP406 C:DZP406:C1 C:MET286 5.39 Ring E Hydrogen C-H bond; halogen C:MET286:HA H-donor; halogen acceptor 7.01 3.03 7 F C-H bond; halogen C:MET286 H-donor; halogen acceptor 7.01 3.03 7 Mod Halogen (Fluorine) halogen ?0.11E228:O H-d			Bond		C:DZP406:H202		D:ILE228:O		2.56
Ring C Hydropio- bic Pi-Sigma DPR0233:HB2 C-H C:DZP406.C2 CHE C:DE7406.C2 CHE7280 Alkyl 4.68 1 bic Alkyl C:DZP406.C20 DILE23C 4.61 C:DZP406.C20 DILE0232 DILE0232 4.61 7.4 F. C:DZP406.C20 DILE0230 4.61 7.4 F. C:DZP406.C20 CHE7610 4.61 7.4 F. F. C:DZP406.C2 CHE7610 4.61 7.4 F. F. C:DZP406.C2 CHE7610 4.61 7.4 F. F. C:DZP406.C2 CHE7610 5.01 8mg B F. C:DZP406.C1 CHE7610 CHE7610 5.01 8mg B F. C:DZP406.C1 CHE7610 CHE7610 5.01 8mg B Hydrogen CHHond: C:DZP406.C1 CHE7610 5.01 8mg B Hydrogen CHHond: C:DZP406.C1 CHE7610 5.01 8mg B Hydrogen CHHond: C:DZP406.C1 D:HE7280 H-acceptor 7.01 8mg B Hydrogen CHHond: *0.01 H-acceptor 7.01 2.01 10 Hydrogen Halogen (Fluorine) <td></td> <td>4</td> <td>Hydrogen Bond; Electro- static</td> <td>Pi-Cation; Pi-Donor H-Bond</td> <td>C:DZP406:H152</td> <td>Positive; H-donor</td> <td>C:PHE289</td> <td>Pi-orbitals; Pi-orbitals</td> <td>2.65</td>		4	Hydrogen Bond; Electro- static	Pi-Cation; Pi-Donor H-Bond	C:DZP406:H152	Positive; H-donor	C:PHE289	Pi-orbitals; Pi-orbitals	2.65
I bic Alkyl C:D2P406:C20 Alkyl C:MET280 Alkyl 4.68 C:D2P406:C20 D:LE023 4.76 C:D2P406:C20 D:PR0233 4.64 C:D2P406:C20 D:PR0233 4.61 C:D2P406:C20 D:PR0233 4.61 C:D2P406:C20 C:D2P406:C2 C:MET261 4.61 C:D2P406:C20 C:D2P406:C2 C:MET261 4.61 Ring C Pi-Alkyl C:D2P406:C2 C:MET261 4.61 Ring B Pi-Alkyl C:D2P406:C2 D:LE0269 5.00 Ring B Hydrogen bond; hologen C:MET286:HA *0:H H-donor, halogen *0.61 *10.61 Ring B Hydrogen bond; halogen (Fluorine) halogen C:MET286:HA *0:H H-donor, halogen *10.11E228:O H-acceptor halogen 2.95 Ring B Hydrogen bond; halogen (Fluorine) halogen (Fluorine) halogen *0.11E228:O H-acceptor *0.11E228:O 4.00 Ring B Halogen (Fluorine) halogen (Fluorine) halogen *0.01 D:EU2230 4.01 Ring B Halogen (Fluorine) halogen *0.01 D:EU2230 4.01 Ring B Halogen (Fluorine) halogen *0.01 D:EU2232 4.01 <td< td=""><td></td><td>Ring C</td><td>Hydropho-</td><td>Pi-Sigma</td><td>D:PRO233:HB2</td><td>C-H</td><td>C:DZP406</td><td>Pi-orbitals</td><td>2.60</td></td<>		Ring C	Hydropho-	Pi-Sigma	D:PRO233:HB2	C-H	C:DZP406	Pi-orbitals	2.60
R12- 7 C:DZP406:C20 D:LU232 4.82 C:DZP406:C20 D:LU232 4.82 C:DZP406:C20 C:DCP406:C20 C:DE72406:C20 4.81 7 C:DZP406:C20 C:DE72406:C20 4.81 7 Pi-Alky1 C:PHE280 Pi-orbitals C:DZP406:C2 5.00 8ing C D:LU205 5.00 C:DZP406:C2 5.00 5.00 8ing B Fordingen C:DZP406:C2 D:LEU269 5.00 8ing C C:DZP406:C2 C:DZP406:C2 C:DZP406:C2 5.00 8ing B Hydrogen C-H bond; halogen (Fluorine) C:DZP406 C:MET286:HA 1.9 7 C:MET280:HA H-donor; halogen acceptor Pi-CIE228:O H-acceptor 2.01 7 C:MET280:HA H-donor; Halogen 1.1E228:O 3.03 7 C:MET280:HA Pi-Orbitals 0.01 3.00 7 Pi-Alky1 100:H Halogen acceptor 10:LE228:O 4.84 8ing B Hydrogen C:H bond; Pi-Pi stacked 10:PI LE228:O 4.84 8ing B Hydrogen Pi-Alky1 10:C D:EU240 Alky1 8ing B Hydrogen Pi-Alky1 10:C C:MET286 4.84		1	bic	Alkyl	C:DZP406:C20	Alkyl	C:MET286	Alkyl	4.68
7 CiDZP406:C20 D:RC023 4.82 7 CiDZP406:C20 C:MCT261 4.61 7 CiDZP406:C20 C:MCT261 4.61 7 Pi-Alkyl C:PHE289 Pi-orbitals C:DZP406:C1 C:MET281 4.11 8 CiDZP406:C1 C:DZP406:C1 C:DZP406:C1 C:DZP406:C1 4.11 8 GiDZP406:C1 C:DZP406:C1 C:DZP406:C1 C:DZP406:C1 5.01 8 Fig C C:DZP406:C1 C:DZP406:C1 C:DZP406:C1 5.01 8 Fig G:D:TDAT280:F1 H-doror: C:DZP406:C1 5.01 5.01 8 Hydrogen C-H bond; S:01 H-dcorr; halogen 7.01 1.01 7.					C:DZP406:C20		D:ILE228		4.76
7 Pi-Alkyl C:DZP406:C2 C:DE228:C C:DE228:C C:DE1023:C C:DE1023:C C:DE1023:C C:DE1023:C C:DE102:C C:DE102:C <lic:de228:c< li=""> C:DE102</lic:de228:c<>					C:DZP406:C20		D:LEU232		4.82
7 CMET261 CMET261 CMET261 4.61 CMET261 CMET261 CLEU265 4.11 Ring C CMET260 CDZP406CL 4.91 Ring C CDZP406 CMET280 Pi-orbials CMET280 5.01 Ring C CDZP406 CMET280 Pi-orbials CMET280 1.01 5.01 Ring C Hydrogen CH bond; halogen CMET286:HA H-donor; halogen 1.01 1.01 1.01 2.01 7 Y F Sinder CMET280 Pi-donor; halogen 1.01 1.01 2.01 3.01 7 Y Y Pi-donor; halogen Pi-diagen Y 3.01 3.01 3.01 7 Y Y Pi-diagen Pi-donor; halogen Pi-diagen Y 3.01 7 Y Y Pi-stacked CPHE280 Pi-orbials Pi-diagen 4.14 7 Nig Pi-alkyl Pi-O Pi-diagen Y Y 4.14 7 Pi-stacked CPHE280 Pi-orbials CMET280 Y Y 8 Pi-diagen Pi-diagen Pi-diagen Pi-orbials Pi-diagen 8 Pi-diagen					C:DZP406:C20		D:PRO233		4.46
R12 Fi-Alkyl CiD2P406:CL CiLEU285 Fi-CPHE289 Pi-orbitals CiD2P406CL 4.99 Ring C CiD2P406 CiD2P406CL CiD2P406CL 5.03 Ring C CiD2P406 CiD2P406 CiMET286-14A H-donor; halogen CFluorine) CiD2P406 CiMET286-14A H-donor; halogen CFluorine) Pi-acceptor 2.95 7 Cime C 60:H H-donor; Pi-acceptor 2.95 7 Cime C 60:H H-donor; Pi-acceptor 2.95 7 Fi-acceptor 2.95 Fi-acceptor 2.95 7 Fi-acceptor 2.95 Fi-acceptor 2.95 8 Hydrogen Halogen (Fluorine) *0:H Halogen Cime 4.90 8 Hydrogen Halogen (Fluorine) *0:H Halogen Cime 4.90 4.90 8 Hydrogen Halogen (Fluorine) *0:F Halogen 4.90 4.90 8 Hydrogen Halogen (Fluorine) *0:F A.90 5.90 5.90 </td <td></td> <td>7</td> <td></td> <td></td> <td>C:DZP406:CL</td> <td></td> <td>C:MET261</td> <td></td> <td>4.61</td>		7			C:DZP406:CL		C:MET261		4.61
Rol2 Fi-Akyl C:PHE289 Pi-orbitals C:DZ406C L:DE2269 L:DE279					C:DZP406:CL		C:LEU285		4.41
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				Pi-Alkyl	C:PHE289	Pi-orbitals	C:DZP406:CL		4.99
Ring B C:DZP406 C:MET286 :HA H-donor; halogen acceptor :*0:F H-acceptor; halogen 2.19 6377 1/2 Hydrogen bond; halogen C:H bond; halogen (Fluorine) :*0:H H-donor; halogen acceptor :*0:F H-acceptor; halogen 2.19 3 Hydrogen bond; C:H bond; halogen :*0:H H-donor; halogen acceptor D:ILE228:O H-acceptor; halogen 2.95 7 T Halogen (Fluorine) :*0:H Halogen cereptor :*0:F Halogen 3.03 7 T Halogen (Fluorine) :*0:N Halogen cereptor :*0:F Halogen 3.05 Ring B Hydropen Pi-Pi stacked C:PHE289 Pi-orbitals D:ILE2240 Alkyl 4.04 Ring C H-Alkyl :*0:C Alkyl D:LEU240 Alkyl 4.04 Ring C Hydrogen halogen (Fluorine) :*0:C Alkyl :*0:C 2.95 PLZ 1/2 Hydrogen halogen Conventional H-bond halogen (Fluorine) :*0:F :*0:F Hacceptor halogen 2.96 1 Dond Conv		Ring C			C:DZP406		D:LEU269		5.50
Ro12- 6377 2' Hydrogen bond; halogen C-H bond; halogen (Fluorine) halogen (Fluorine) C:MET286:HA halogen acceptor H-donor; halogen acceptor :*0:F H-acceptor; halogen 2.19 3 Hydrogen bond C-H bond :*0:H H-donor; i*0:H D:ILE228:O H-acceptor; 2.95 7 :*0:H D:ILE228:O H-acceptor; 3.03 7 :*0:H D:ILE228:O H-acceptor; 2.95 Internal ^C Halogen Halogen (Fluorine) :*0:N Halogen acceptor; :*0:F Halogen; 3.03 Ring B Hydropho- Pi-Pi stacked C:PHE289 Pi-orbitals :*0:F Halogen; 4.14 7 bic Alkyl :*0:C Alkyl 2.02 4.64 Ring B Hydrogen bond; halogen Pi-Alkyl :*0:C D:PRO233 4.40 ?*0 :*0 D:LEU232 4.94 ?*0 :*0:F H-acceptor; halogen 2.46 ?*0 Pi-Alkyl :*0:F H-acceptor; halogen 2.46 ?*0 Conventional H-bond; halogen (Fluorine) halogen :*0:F H-		Ring B			C:DZP406		C:MET286		5.39
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Ro12- 6377	2'	Hydrogen bond; halogen	C-H bond; halogen (Fluorine)	C:MET286:HA	H-donor; halogen acceptor	:*0:F	H-acceptor; halogen	2.19
bond:*0:HD:ILE228:O3.037:*0:HD:THR237:OG12.79Internal ^e HalogenHalogen (Fluorine):*0:NHalogen acceptor:*0:FHalogen3.50Ring BHydropho-Pi-Pi stackedC:PHE289Pi-orbitals:*0Pi-orbitals4.147bicAlkyl:*0:CAlkylD:LEU240Alkyl4.63Ring BPi-Alkyl:*0:CAlkylD:PRO2334.40Ring C::*0Pi-orbitalsD:REU324.40Ring C::*0D:Atset:*04.54Ring C::*0D:MET2864.54indogen (Fluorine):*0:PiD:MET2364.87PLZ2'Hydrogen bond; halogenConventional H-bond; halogen (Fluorine)C:ASN265:HD21 :*0:HH-donor; halogen acceptor:*0:FH-acceptor halogen1.901bond:*0:H:*0:H:*0:F:*0:F2.58Internal ^e :*0:H:*0:H:*0:F2.58Internal ^e :*0:H:*0:H:*0:F2.58Internal ^e :*0:H:*0:H:*0:F3.072'HalogenHalogen (Fluorine) halogen:*0:H:*0:F3.072'HalogenHalogenD:ILE228:O:*0:F3.08Internal ^e :*0:H:*0:H:*0:F:*0:F3.072'HalogenHalogen (Fluorine) halogen:*0:H:*0:O3.072'		3	Hydrogen	C-H bond	:*0:H	H-donor	D:ILE228:O	H-acceptor	2.95
7 :*0:H D:THR237:OG1 2.79 Internal ^c Halogen Halogen (Fluorine) :*0:N Halogen acceptor :*0:F Halogen 5.00 Ring B Hydropho Pi-Pi stacked C:PHE289 Pi-orbitals :*0:P 4.14 7 b'c Alkyl :*0:C Alkyl D:LEU240 Alkyl 4.63 Ring B F Pi-Alkyl :*0:C Alkyl D:PRO233 4.40 Ring C F :*0 C:MET286 4.54 Ring C F :*0 D:LEU232 4.94 Ring B Hydrogen Conventional H-bonthalogen (Fluorine) C:ASN265:HD21 H-donor, halogen acceptor :*0:F H-acceptor 1.00 1 bond Longen (Fluorine) C:ASN265:HD21 H-donor :*0:F 1.90 2.55 4 Hydrogen Conventional H-bond C:ASN265:HD21 H-donor :*0:F 2.56 1 bond D:PRO233:HD3 :*0:F :*0:F 2.55 1 C-H bond D:PRO233:HD3 :*0:F :*0:F 2.58 1 internal ^c :*0:H :*0:F :*0:F 3.07 2'* Halogen (Fluorine) D:PRO233:HD			bond		:*0:H		D:ILE228:O		3.03
Internal ^c Halogen Halogen (Fluorine) :*0:N Halogen acceptor :*0:F Halogen 5.0 Ring B Hydropho Pi-Pi stacked C:PHE289 Pi-orbitals :*0 Pi-orbitals 4.14 7 bic Alkyl :*0:C Alkyl D:LEU240 Alkyl 4.63 Ring B Pi-Alkyl :*0:C Alkyl D:PR0233 4.40 Ring C Pi-Alkyl :*0 C:MET286 4.54 Ring C :*0 D:LEU232 4.40 PLZ Hydrogen bond; halogen Conventional H-bond; halogen (Fluorine) C:ASN265:HD21 H-donor; halogen acceptor i*0:F H-acceptor; halogen 2.46 1 bond C:Roureinal H-bond; halogen (Fluorine) i*0:H :*0:F 2.58 2.55 1 C-H bond D:PR0233:HD3 H-donor; halogen acceptor :*0:F 2.58 Internal ^c :*0:H :*0:F :*0:F 2.58 Internal ^c :*0:H :*0:F :*0:F 3.30 1		7			:*0:H		D:THR237:OG1		2.79
Ring B Hydropho Pi-Pi stacked C:PHE289 Pi-orbitals :*0 Pi-orbitals 4.14 7 bic Alkyl :*0C Alkyl D:LEU240 Alkyl 4.63 Ring B Pi-Alkyl :*0C Alkyl D:PRO233 4.40 Ring C :*0 C:MET286 4.54 Ring C :*0 D:LEU232 4.94 PLZ Hydrogen bond; halogen Conventional H-bont; halogen (Fluorine) C:ASN265:HD21 H-donor; halogen acceptor i*0:F H-acceptor; halogen 2.46 4 Hydrogen bond; Conventional H-bont halogen (Fluorine) C:ASN265:HD21 H-donor; halogen acceptor i*0:F H-acceptor; halogen 2.46 1 bond Conventional H-bont halogen C:ASN265:HD22 H-donor :*0:F 2.53 Internal* C:H bond D:PRO233:HD3 :*0:F :*0:O 2.53 Internal* :*0:H :*0:H :*0:O 2.53 Ring B Hydropho Pi-Pi stacked C:PHE280 Pi-orbitals :*0:F 3.03 Ring C Halogen Pi-Alkyl		Internal ^c	Halogen	Halogen (Fluorine)	:*0:N	Halogen acceptor	:*0:F	Halogen	3.50
7 bic Alkyl :*0:C Alkyl D:LEU240 Alkyl 4.63 Ring B Pi-Alkyl :*0 D:Drobials D:PRO233 4.40 Ring C :*0 C:MET286 4.54 :*0 D:LEU232 4.94 :*0 D:MET236 4.94 :*0 :*0:P Piconor; :*0:P *10 :*0:P :*0:P 1.90 1 bond; :*0:P :*0:P 2.55 :*0:H :*0:P :*0:P 2.53 Internal* :*0:H :*0:P 2.53 Internal* :*0:P :*0:P 3.07 2' Halogen Pilopentione) D:LE228:O :*0:P 3.07 2' Halogen Pi-		Ring B	Hydropho-	Pi-Pi stacked	C:PHE289	Pi-orbitals	:*0	Pi-orbitals	4.14
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		7	bic	Alkyl	:*0:C	Alkyl	D:LEU240	Alkyl	4.63
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		Ring B		Pi-Alkyl	:*0	Pi-orbitals	D:PRO233		4.40
PLZ2'Hydrogen bond; halogenConventional H-bond; halogen (Fluorine):*0D:LEU2324.944Hydrogen bond; halogenConventional H-bond; halogen (Fluorine)C:ASN265:HD21 halogen acceptorH-donor; halogen acceptorH-acceptor; halogen acceptor2.464Hydrogen bond; 1Conventional H-bond bondC:ASN265:HD22 $:*0:H$ H-donor; halogen acceptorH-acceptor; halogen1.904Hydrogen bondConventional H-bond $:*0:H$ C:ASN265:HD22 $:*0:H$ H-donor; bicH-acceptor; $:*0:H$ 2.25InternalcC-H bondC:ASN265:HD22 $:*0:H$ H-donor; $:*0:O$ H-acceptor; $:*0:O$ 2.58InternalcC-H bondD:PRO233:HD3:*0:OD:ILE228:O2.53Internalc:*0:H:*0:H:*0:O2.532'Halogen bicHalogen (Fluorine)D:ILE228:OHalogen acceptor $:*0:O$:*0:F2.532'Halogen bicHalogen (Fluorine)D:ILE228:OHalogen acceptor $:*0:O$:*0:F3.072'Halogen bicHalogen (Fluorine)D:ILE228:OHalogen acceptor $:*0:O$:*0:F4.142'Halogen bicPi-Pi stacked Pi-Alkyl:*0FHalogen4.34bicPi-Alkyl:*0ED:PRO233Alkyl4.45Fing C:*0:*0D:PRO235Alkyl4.52i*0:*0:*0D:PRO233Alkyl4.52bi		Ring C			:*0		C:MET286		4.54
PLZ2'Hydrogen bond; halogenConventional H-bond; halogen (Fluorine):*0D:MET2364.874Hydrogen halogenConventional H-bond; halogenC:ASN265:HD21 :*0:HH-donor; halogen acceptor:*0:FH-acceptor; halogen2.464Hydrogen bond; 1Conventional H-bond bondC:ASN265:HD22 :*0:HH-donor; halogen acceptor:*0:NH-acceptor; halogen1.901bondC:Nentional H-bond :*0:HC:ASN265:HD22 :*0:HH-donor; :*0:H:*0:NH-acceptor; 2.251.901Let HondD:PRO233:HD3:*0:H:*0:NH-acceptor; 2.252.53Internal ^c -:*0:H:*0:H:*0:O2.532'Halogen Halogen (Fluorine)D:ILE228:OHalogen acceptor :*0:H:*0:F2.532'Halogen Halogen (Fluorine)D:ILE228:OHalogen acceptor :*0:H:*0:F3.38Ring BHydropho- bicPi-Pi stacked 		-			:*0		D:LEU232		4.94
PLZ2'Hydrogen bond; halogenConventional H-bond; halogen (Fluorine)C:ASN265:HD21H-donor; halogen acceptor:*0:FH-acceptor; halogen2.464Hydrogen bondConventional H-bond bondC:ASN265:HD22H-donor:*0:NH-acceptor; halogen1.901bondC:nventional H-bond bondC:ASN265:HD22H-donor:*0:NH-acceptor; halogen1.901bondC:Nventional H-bond bondC:ASN265:HD22H-donor:*0:NH-acceptor; halogen1.901EC-H bondD:PR0233:HD3:*0:NH-acceptor; b:0:H:*0:F2.53Internal ^c E:*0:H:*0:H:*0:F2.53Internal ^c :*0:H:*0:H:*0:O3.072'HalogenHalogen (Fluorine)D:ILE228:OHalogen acceptor:*0:FHalogen2'HalogenPi-pi stackedC:PHE289Pi-orbitals:*0:FHalogen3.38Ring BHydropho bicPi-Pi stackedC:PHE289Pi-orbitals:*04.34Pi-Alkyl:*0:*0D:PR0233Alkyl4.46Ring C:*0:*0D:PR0233Alkyl4.52:*0:*0:*0D:PR0233Alkyl4.52:*0:*0:*0D:PR0233Alkyl4.52:*0:*0:*0D:PR0233Alkyl4.52:*0:*0:*0D:LEU269:519					:*0		D:MET236		4.87
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	PLZ	2'	Hydrogen bond; halogen	Conventional H-bond; halogen (Fluorine)	C:ASN265:HD21	H-donor; halogen acceptor	:*0:F	H-acceptor; halogen	2.46
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		4	Hydrogen	Conventional H-bond	C:ASN265:HD22	H-donor	:*0:N	H-acceptor	1.90
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		1	bond		:*0:H		D:ILE228:O	-	2.25
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$				C-H bond	D:PRO233:HD3		:*0:O		2.58
Internalc:*0:H:*0:O3.072'HalogenHalogen (Fluorine)D:ILE228:OHalogen acceptor:*0:FHalogen3.38Ring BHydropho bicPi-Pi stackedC:PHE289Pi-orbitals:*0Pi-orbitals4.34Ring C:*0:*0D:PRO233Alkyl4.46ring C:*0:*0D:PRO2335.19		Internalc			:*0:H		:*0:F		2.53
2'HalogenHalogen (Fluorine)D:ILE228:OHalogen acceptor:*0:FHalogen3.38Ring BHydrophoPi-Pi stackedC:PHE289Pi-orbitals:*0Pi-orbitals4.34bicPi-Alkyl:*0D:PRO233Alkyl4.46Ring C:*0D:PRO2334.52:*05.19		Internalc			:*0:H		:*0:O		3.07
Ring BHydropho-Pi-Pi stackedC:PHE289Pi-orbitals:*0Pi-orbitals4.34bicPi-Alkyl:*0D:PRO233Alkyl4.46Ring C:*0D:PRO2334.52:*0D:LEU2695.19		2'	Halogen	Halogen (Fluorine)	D:ILE228:O	Halogen acceptor	:*0:F	Halogen	3.38
bic Pi-Alkyl :*0 D:PRO233 Alkyl 4.46 Ring C :*0 D:PRO233 4.52 :*0 D:LEU269 5.19		Ring B	Hydropho-	Pi-Pi stacked	C:PHE289	Pi-orbitals	:*0	Pi-orbitals	4.34
Ring C :*0 D:PRO233 4.52 :*0 D:LEU269 5.19		÷	bic	Pi-Alkyl	:*0		D:PRO233	Alkyl	4.46
:*0 D:LEU269 5.19		Ring C		2	:*0		D:PRO233	,	4.52
		J			:*0		D:LEU269		5.19

^aPosition of interaction

^bdistance of interaction

^cinternal interaction

Table 7 Detailed binding interactions resulting from the molecular docking of co-crystallized DZP, Ro12-6377, and proflazepam (PLZ) at the TMD γ_2^+/β_2^- interface

Ligand Name	Posit ^a	Category	Type of interactions	From	From chemistry	То	To chemistry	Dist ^b (A°)
DZP	3	Hydrogen bond	C-H bond	E:DZP403:H171	H-donor	E:SER280:OG	H-acceptor	2.77
	1			E:DZP403:H202		E:THR281:OG1		2.80
	Ring C	Hydrophobic	Pi-Pi T-shaped	E:PHE304	Pi-orbitals	E:DZP403	Pi-orbitals	5.81
	1		Alkyl	E:DZP403:C20	Alkyl	A:PRO228	Alkyl	4.25
	7			E:DZP403:CL		A:LEU223		4.64
				E:DZP403:CL		E:VAL300		4.38
	Ring B		Pi-Alkyl	E:DZP403	Pi-orbitals	E:VAL300		4.92
Ro12-6377	7	Hydrogen bond	Conventional H-bond	:*0:H	H-donor	A:GLN224:O	H-acceptor	2.43
	2'		C-H bond	A:PRO228:HD3		:*0:F		2.61
	7			E:THR277:HB		:*0:O		2.73
	2			E:ASP297:HA		:*0:O		2.28
	3			:*0:H		E:ASP297:OD1		2.89
	1			:*0:H		E:SER280:O		2.71
	7			:*0:H		A:GLN224:O		2.97
				:*0:H		E:THR281:OG1		2.78
				:*0:H		A:GLN224:O		2.94
	2'	Halogen	Halogen (Fluorine)	A:LEU223:O	Halogen acceptor	:*0:F	Halogen	2.56
	Ring C	Hydrophobic	Pi-Pi T-shaped	E:PHE304	Pi-orbitals	:*0	Pi-orbitals	5.40
	1		Alkyl	E:ALA283	Alkyl	:*0:C	Alkyl	4.28
				:*0:C		E:ARG284		3.96
				:*0:C		E:VAL300		3.90
	7			:*0:C		A:PRO228		4.75
				:*0:C		A:ILE264		3.81
	Ring B		Pi-Alkyl	:*0	Pi-orbitals	A:PRO228		5.11
	Ring C			:*0		A:PRO228		5.29
PLZ	1	Hydrogen bond	Conventional	A:GLN224:HE21	H-donor	:*0:O	H-acceptor	2.32
			H-bond	A:GLN224:HE22		:*0:O		2.59
				:*0:H		E:THR281:OG1		1.92
	2'		C-H bond	A:PRO228:HD3		:*0:F		2.80
	2			E:THR277:HA		:*0:O		2.79
	7			E:ARG284:HD3		:*0:CL		2.83
	1			:*0:H		E:SER280:OG		2.74
				:*0:H		A:GLN224:O		2.81
				:*0:H		E:THR281:OG1		2.37
	7	Halogen	Halogen (Cl, Br, I)	E:ASP297:OD1	Halogen acceptor	:*0:CL	Halogen	3.18
	2'		Halogen (Fluorine)	A:LEU223:O		:*0:F		2.90
	Ring C	Hydrophobic	Pi-Pi T-shaped	E:PHE304	Pi-orbitals	:*0	Pi-orbitals	5.82
	7		Alkyl	:*0:CL	Alkyl	A:LEU223	Alkyl	4.29
	Ring B		Pi-Alkyl	:*0	Pi-orbitals	E:VAL300		4.92
	Ring C			:*0		A:LEU231		5.35
	1	Unfavorable	Unfavorable Accep- tor-Acceptor	A:GLN224:O	H-acceptor	:*0:O	H-acceptor	2.83

^aPosition of interaction

^bdistance of interaction

^cinternal interaction

binding interactions observed between the both modulators are related to the presence of two distinct binding groups on the structure of proflazepam: the fluorine atom at $C_{2'}$ and the long backbone skeleton of CH₂CHOHCH₂OH at N₁. The phenyl (C) is placed in front of the pocket (Fig. 2d and 4d and Table 7), its π -electron cloud establishes two interactions of types Pi-PiT-shaped and Pi-Alkyl with side chains of γ_2 Phe304 and β_2 Leu231, respectively. Its fluorine atom points towards the TM_1 : β_2 subunit, where it receives a moderate H-bond from the side chain of Pro228 and establishes halogen interaction with the main oxygen of Leu223. The phenyl (B) binds at a higher level than rings (A) and (C), with its chlorine atom pointing towards the TM_3 : γ_2 subunit. This orientation results in Pi-Alkyl interaction between the $\pi\text{-}electron$ cloud and $\gamma_2Val300,$ and also leads the chlorine to accept the moderate H-bond from γ_2 Arg284, forms halogen interaction with the side chain of γ_2 Asp297, and interacts hydrophobically with β_2 Leu223. The binding modes of the oxygen atom attached at C₂ in the three $\alpha_1^+/\gamma_2^-,\beta_2^+(A)/\alpha_1^-(B)$, and $\beta_2^+(C)/\alpha_1^-(D)$ interfaces reflect the insensitivity of Ro12-6377 and proflazepam to participate in any interaction with the neighboring residues using this position. Otherwise, the γ_2^+/β_2^- interface reflects the contribution of the oxygen atom at this position to enable the receptor potentiation by accepting, respectively, strong and moderate H-bonds from Asp297 and Thr277 of subunit γ_2 . The CH₂CHOHCH₂OH group penetrates deep into the binding site. Its pose and orientation towards the β_2 :A:TM₂ helix are identical to that observed for the NHCONHCH₃ group of Ro12-6377. The length of its backbone skeleton allowed it to simultaneously influence the γ_2 :TM₂helix and the β_2 :A:TM₁ by forming six H-bonds with γ_2 :TM₂:Ser280, γ_2 :TM₂:Thr281, and β_2 :A:TM₁:Gln224.

The detailed mechanisms of interactions by which proflazepam modulates the two TMD $\beta_2^+(A)/\alpha_1^-(B)$ and $\beta_2^+(C)/\alpha_1^-(D)$ sites are appearing to be complementary to each other. As can be seen, the $\beta_2^+(A)/\alpha_1^-(B)$ interface has been predicted to specifically modulate by the first hydroxyl group of CH₂CHOHCH₂OH, C₃, the chlorine atom at C₇, and the π -electron clouds of both phenyls (B) and (C), whereas the $\beta_2^+(C)/\alpha_1^-(D)$ interface has been predicted to specifically modulate by the second hydroxyl group of CH₂CHOHCH₂OH, N_4 , the fluorine atom at C_2 , and the π -electron cloud of both phenyls (B) and (C). This selectivity in binding interactions between the two orthosteric binding sites allows us to hypothesize that they may be functional in a complementary manner, as previously observed for the two agonist (GABA) binding sites which we do not yet know an explanation for why they are structurally identical and functionally not equivalent [8]. Furthermore, the bulky structure of proflazepam allowed it to directly induce influence on the pore-lining helices β_2 :A:TM₂, β_2 :C:TM₂, and α_1 :D:TM₂ by creating H-bonds with β_2 :A:Arg269, β_2 :A:Asn265, β_2 :C:Asn265, and interacting hydrophobically with α_1 :D:Leu269. Likewise,

three hydrophobic interactions with β_2 :A:TM₂:Met261, β_2 :C:TM₂:Met261, and α_1 :D:TM₂:Leu269 are observed for the DZP, whereas just one hydrophobic interaction with β_2 :A:TM₂:Val258 is established for Ro12-6377. Evidently, rings (C) of both proflazepam and DZP adopt a similar hydrophobic interaction with α_1 :D:TM₂:Leu269.

Obviously, at γ_2^+/β_2^- interface, the binding mode of Ro12-6377 is the most influential on pore-lining by sharing five interactions with the residues of γ_2 :TM₂helix and one interaction with β_2 :A:TM₂ helix. Likewise, the binding mode of proflazepam is connected to the pore-lining by participating in five interactions with γ_2 :TM₂ helix. Accordingly, both modulators exhibit common interactions with γ_2 :TM₂residues: Thr281, Thr277, Arg284, and Ser280. Otherwise, by examining the binding interactions of DZP, its structure predicted to enrich the skeleton of γ_2 :TM₂ through two moderate H-bonds originating from the methyl group attached at N₁ to Ser280 and from C₃ to Thr281. These two interactions are identical between the three modulators.

The feature of interacting with the residues of the TM_2 helices is of great importance as it leads both Ro12-6377 and proflazepam to directly induce motions in the chloridechannel lining, thereby, possibly contributing to the expansion of its diameter by opening the 9' gate by orienting the β_2 :C:Leu259 side chain towards one of the two adjacent α subunits. As mentioned earlier, this rotation is the main factor in the activation of the pLGICs family [7, 57].

Molecular dynamics simulation

MD simulation is used as a complementary tool to validate the docking results before they are approved in the drugdesign process. MD simulation offers the peculiarity of treating biological systems as flexible entities. This flexibility allows for free integrations between the macromolecule binding site and the binding-ligand, resulting in binding modes (poses or interactions) that confirm or refute the results of molecular docking [58]. Or in some cases, it may lead to the release of the ligand from the binding site, and this is an undesirable defect especially if the ligand shows the highest stability in the docking simulation [59].

For these reasons, Ro12-6377 and proflazepam in complex with the four binding interfaces were subjected to MD simulations using the settled parameters cited in the "Materials and methods" section. Their dynamic behaviors were investigated by evaluating the response of the potential energy, U (Kcal/mol), over a time period of 1000 picoseconds (ps) (Fig. 5). During the first 500 ps of the simulation, the ECD α_1^+/γ_2^- interface exhibits higher stability in complex with Ro12-6377 than in complex with proflazepam. Thereafter, from 500 ps until the end of the simulation, both complexes tend to have equivalent stability (6270.7 kcal/mol for Ro12-6377 and 6226.86 kcal/mol for proflazepam) (Fig. 5a).



Fig. 5 Evaluation the response of potential energy, U (kcal/mol), as function of time, t(ps), for Ro12-6377 and proflazepam in complex with **a** ECD $\alpha_1^+ \gamma_7^-$, **b** TMD $\beta_2^+ (A)/\alpha_1^-(B)$, **c** TMD $\beta_2^+ (C)/\alpha_1^-(D)$, and **d** TMD γ_2^+ / β_2^- interfaces

Moreover, within the three TMD binding interfaces, the two modulators exhibited stability equivalence throughout all the simulation periods (Fig. 5b–d).

Later, deep analyses of binding modes, binding orientations, and binding interactions of Ro12-6377 and proflazepam within the four binding interfaces were performed and discussed between the two simulations.

MD simulation analysis of the classical binding site

At ECD α_1^+/γ_2^- interface, the binding poses of Ro12-6377 are equivalents for the two simulations (Figs. 6a and 7a and Table 8). However, significant differences between the two binding orientations were detected, notably in substitutes at C₂, C₅, and C₇. The new binding orientation predicted for phenyl (C) is driving the C_{2'} bound fluorine atom to move away from the diazepine ring (A), which leads to the

disappearance of the intramolecular interaction formed with C_2 . All the H-bonds formed with α_1 Ser206, γ_2 Phe77, and α_1 His102 were vanished and replaced by H-bonds given from side chain of α_1 Lys156 to the oxygen atom at C_2 , and from the NHCONHCH₃ group at C_7 to both α_1 Tyr160 and γ_2 Asp56. Similarly, the hydrophobic interactions suggested with Phe100 and Val203 of α_1 subunit were also replaced by hydrophobic interactions created between the π -orbitals of Tyr160 and Tyr210 of the same subunit and the π -electron cloud of phenyl (C). In contrast, the hydrophobic interactions established with the subunit γ_2 residues (Tyr58, Phe77, and Ala79) were preserved as the same.

The binding modes of proflazepam are inconsistent in the binding poses for the two simulations. The adequate binding pose generated by MD simulation was placed less deeply within the binding locus than that generated by molecular docking simulation (Fig. 8a). Unlike molecular



Fig. 6 Binding modes resulting from the molecular dynamics simulation of Ro12-6377 at the interfaces of **a** ECD α_1^+/γ_2^- , **b** TMD $\beta_2^+(A)/\alpha_1^-(B)$, **c** TMD $\beta_2^+(C)/\alpha_1^-(D)$, and **d** TMD γ_2^+/β_2^- interfaces

docking simulation, this binding position emerged its structure to react as donor and acceptor with neighboring residues. The most notable differences in the binding orientations appear in the substitutes groups at N₁,C₅, and C₇. By examining the binding interactions (Fig. 9a and Table 8), the moderate H-bond established between the first hydroxyl group of CH₂CHOHCH₂OH and the side chain of α_1 Tyr210 was preserved between the two simulations, whereas the remaining interactions were completely vanished and replaced by five interactions with the residues of the same subunit: side chain of Lys156 gives two H-donors to N₄ and the fluorine atom bond at C₂, and were also involved in Pi-Cation interaction with the π -electron cloud of phenyl (C). The main chain of Ser159 receives one H-acceptor bond from the second hydroxyl group of CH₂CHOHCH₂OH. A moderate intramolecular H-bond was formed between the oxygen atom at C₂ and the CH₂CHOHCH₂OH group. Finally, hydrophobic interaction type Pi-Alkyl was observed between the π -electron cloud of phenyl (B) and the side chain of Val203.

MD simulation analysis of the three TMD binding sites

The binding modes of Ro12-6377 for both TMD $\beta_2^+(A)/\alpha_1^-(B)$ and $\beta_2^+(C)/\alpha_1^-(D)$ interfaces are consistent in docking poses for the two simulations (Fig. 6b, c). However,



Fig.7 Binding interactions resulting from the molecular dynamics simulation of Ro12-6377 at the interfaces of **a** ECD α_1^+/γ_2^- , **b** TMD $\beta_2^+(A)/\alpha_1^-(B)$, **c** TMD $\beta_2^+(C)/\alpha_1^-(D)$, and **d** TMD γ_2^+/β_2^- interfaces

notable differences were observed between the binding orientations of C₅ and C₇ substitutes. The binding interactions at both interfaces exhibit the tendency of Ro12-6377 to enhance its interactions with the residues of β_2 subunit relative to those established with subunit α_1 (Figs. 7b, c and Tables 9 and 10). Briefly, at the $\beta_2^+(A)/\alpha_1^-(B)$ interface, all interactions created with residues of subunit β_2 :A were preserved and stabilized by removing the moderate H-bond and

the Pi-Alkyl interaction linking, respectively, the substitutes at N₁ and C₇ with the side chains of Leu285 and Phe289, and creating, instead of them, new moderate H-bond oriented from the methyl group at N₁ towards the main carbonyl group of Asp282. In contrast, only interactions with two residues from α_1 :B subunit (Met236 and Thr237) were preserved. The Pi-Alkyl interaction bond of the side chain of Met236 to phenyl (C) was vanished and replaced with two

Table 8 D	Detailed binding interac	tions resulting from the	molecular dynamics sin	nulation Ro12-6377 a	and proflazepam (PLZ) at the classical site
-----------	--------------------------	--------------------------	------------------------	----------------------	----------------------	-------------------------

ECD α_1^+/γ_2^- int	erface							
Ligand Name	Posit ^a	Category	Type of interactions	From	From chemistry	То	To chemistry	Dist ^b (A°)
Ro12-6377	2	Hydrogen bond	Conventional hydro-	D:LYS156:HZ3	H-donor	:*0:O	H-acceptor	1.67
	7		gen bond	:*0:H		D:TYR160:OH		2.23
				:*0:H		D:TYR160:OH		2.27
			C-H bond	:*0:H		E:ASP56:OD1		2.63
	Ring C	Hydrophobic	Pi-Pi stacked	D:TYR160	Pi-orbitals	:*0	Pi-orbitals	4.46
	Ring B			E:TYR58		:*0		4.29
	Ring C		Pi-Pi T-shaped	D:TYR210		:*0		5.39
	Ring B			E:PHE77		:*0		4.68
	7		Alkyl	E:ALA79	Alkyl	:*0:C	Alkyl	3.59
	1		Pi-Alkyl	E:TYR58	Pi-orbitals	:*0:C	Alkyl	4.47
PLZ	4	Hydrogen bond	Conventional hydro-	D:LYS156:HZ2	H-donor	:*0:N	H-acceptor	1.57
	1		gen bond	:*0:H		D:SER159:O		1.79
	2'		C-H bond	D:LYS156:HE2		:*0:F		2.51
	Internal ^c			:*0:H		:*0:O		2.51
	Ring C	Electrostatic	Pi-Cation	D:LYS156:NZ	Positive	:*0	Pi-orbitals	4.25
	1	Hydrogen bond	Pi-Donor hydrogen bond	:*0:H	H-donor	D:TYR210		2.69
	Ring B	Hydrophobic	Pi-Alkyl	:*0	Pi-orbitals	D:VAL203	Alkyl	4.72

^aPosition of interaction

^bdistance of interaction

cinternal interaction

H-bonds oriented from C₇ substitute towards the main carbonyl group, and Pi-Sulfur interaction given from the sulfur atom in the side chain to phenyl (C). At the $\beta_2^+(C)/\alpha_1^-(D)$ interface, the binding interactions with Thr237, Leu240, and Leu232 of α_1 :D subunit were vanished and replaced by new interactions with residues from the β_2 :C subunit: Arg269, Leu285, and Val258. Side chains of both Leu285 and Val258 were, respectively, involved in Alkyl-Alkyl interactions with the substitutes at N_1 and C_7 , while the side chain of Arg269 established a strong H-bond with the oxygen atom at C₂. Likewise, the intramolecular interaction observed between N_1 and the fluorine atom at $C_{2'}$ also vanished and created, instead of it, additional interaction between this latter and the main carbonyl of Met286. At the α_1 :D subunit, likewise to B:Met236, D:Met236 was replacing the Pi-Alkyl interaction connected its side chain to phenyl (C) by two moderate H-bonds and one Pi-Sulfur interaction. The two moderate H-bonds are oriented from the NHCONHCH₃ group at C₇ towards the main carbonyl group, whereas the Pi-Sulfur interaction appears between the sulfur atom in the side chain and the phenyl (B).

The binding modes of proflazepam for the TMD $\beta_2^+(A)/\alpha_1^-(B)$ interface are inconsistent in docking poses for the two simulations (Fig. 8b). Accordingly, the differences

between the two binding orientations were detected for the entire structure of proflazepam. The adequate binding pose resulting from the MD simulation was inserted more deeply into the binding locus in such a way that the fused benzodiazepine rings and the substitute at N1 have facial alignment with the β_2 :TM₃ helix. The binding interactions after the MD simulation (Fig. 9b and Table 9) suggested the lack of the two H-bonds established with β_2 Leu285 and β_2 Arg269, as well as the three Pi-Alkyl interactions which were created with β_2 Met286, α_1 Leu232, and α_1 Pro233. Alternatively, α_1 Ile228, β_2 Phe289, α_1 Pro233, and β_2 Met286 were estimated to participate in six interactions with phenyl (B) and the substitutes at N_1 and C_7 . The second hydroxyl group of CH₂CHOHCH₂OH gave a strong H-bond to the main carboxyl function of α_1 Ile228. The chlorine at C₇ was involved in three hydrophobic interactions, two are of the Alkyl-Alkyl type created with side chains of β_2 Met286 and α_1 Pro233 and one is of the Pi-Alkyl type created with the Pi-orbitals of β_2 Phe289. Finally, the Pi-orbitals of phenyl (B) have participated through Pi-Pi stacked and Pi-Alkyl interactions with the side chains of β_2 Phe289 and α_1 Pro233, respectively.

The binding modes of proflazepam for the TMD $\beta_2^+(C)/\alpha_1^-(D)$ interface are equivalents in the docking poses for the two simulations (Fig. 8c). The most pronounced



Fig. 8 Binding modes resulting from the molecular dynamics simulation of proflazepam at the interfaces of **a** ECD α_1^+/γ_2^- , **b** TMD $\beta_2^+(A)/\alpha_1^-(B)$, **c** TMD $\beta_2^+(C)/\alpha_1^-(D)$, and **d** TMD γ_2^+/β_2^- interfaces

differences in the binding orientations have appeared for the substitutes groups at N₁, C₅, and C₇. The molecular docking and MD simulations overlap in that the residues β_2 Asn265, β_2 Phe289, α_1 Ile228, α_1 Pro233, and α_1 Leu269 are essential parts of the $\beta_2^+(C)/\alpha_1^-(D)$ modulation by proflazepam (Fig. 9c and Table 10). However, this does not necessarily mean that they have retained the same nature and orientation of their binding interactions shared with the proflazepam structure for both simulations, which is an observation common to all the binding sites studied in this paper. The binding pose after the MD simulation was stabilized by three intramolecular

interactions orienting from the CH₂CHOHCH₂OH group and N₁ towards the fluorine atom bond at C_{2'}. The four hydrophobic interactions created for both phenyls (B) and (C) and the H-bond linked α_1 Ile 228 to the CH₂CHOHCH₂OH group were preserved as the same. β_2 Asn265 reduced the two H-bonds established with N₄ and the fluorine atom at C_{2'} to a single strong H-bond directed from its amide group towards the bound oxygen atom at C₂. α_1 Ile228 loses the halogen interaction established with the fluorine atom at C_{2'} and conserves the H-bond accepted from the CH₂CHOHCH₂OH group. α_1 Pro233 tends to be involved in hydrophobic



Fig. 9 Binding interactions resulting from the molecular dynamics simulation of proflazepam at the interfaces of **a** ECD α_1^+/γ_2^- , **b** TMD $\beta_2^+(A) / \alpha_1^-(B)$, **c** TMD $\beta_2^+(C)/\alpha_1^-(D)$, and **d** TMD γ_2^+/β_2^- interface

interactions using its side chain. Thus, in addition to the two Pi-Alkyl interactions created with phenyl (B) and phenyl (C), it prefers to create an Alkyl-Alkyl interaction with the chlorine atom at C₇ rather than the strong H-bond created with the substitute at N₁. β_2 Phe289 shows an additional hydrophobic interaction type Pi-Alkyl with the chlorine atom at C₇. Finally, three interactions with two new residues were observed for phenyl (B) and the substitutes at N₁ and C₇. The second hydroxyl group of CH₂CHOHCH₂OH accepts a strong H-bond from the side chain of β_2 Arg269, and both the π -electron cloud of phenyl (B) and the chlorine at C₇ formed, respectively, Pi-Sulfur and Alkyl-Alkyl interactions with the side chain of α_1 Met236.

At TMD γ_2^+/β_2^- interface, the binding modes of Ro12-6377 show significant differences for the two simulations. The adequate binding mode generated by MD simulation was inserted more deeply into the binding locus, so that phenyl (B) is positioned between β_2 :TM₁ and γ_2 :TM₂ helices and phenyl (C) has facial alignment with the β_2 :TM₂ and γ_2 :TM₃ helices (Fig. 6d). MD simulation reduced the number of

Table 9 Detailed binding interactions resulting from the molecular dynamics simulation of Ro12-6377 and proflazepam (PLZ) at the TMD $\beta_2^+(A)/\alpha_1^-(B)$ interface

TMD β_2^+	A) $/\alpha_1^-$ (B) in	terface						
Ligand Name	Posit ^a	Category	Type of interactions	From	From chemistry	То	To chemistry	Dist ^b (A°)
Ro12-	7	Hydrogen	Conventional hydro-	:*0:H	H-donor	B:MET236:O	H-acceptor	2.51
6377		bond	gen bond	:*0:H		B:MET236:O		1.85
	1		C-H bond	:*0:H		A:ASP282:O		2.79
	7			:*0:H		B:THR237:OG1		3.06
	Ring C	Other	Pi-Sulfur	B:MET236:SD	Sulfur	:*0	Pi-orbitals	3.90
	Ring B	Hydro-	Pi-Pi stacked	A:PHE289	Pi-orbitals	:*0		4.62
	1	phobic	Alkyl	:*0:C	Alkyl	A:LEU285	Alkyl	5.05
				:*0:C		A:MET286		4.81
	7			:*0:C		A:VAL258		5.23
	Ring C		Pi-Alkyl	:*0	Pi-orbitals	A:MET286		4.84
PLZ	Internal ^c Hydrogen	Hydrogen	Conventional hydro-	:*0:H	H-donor	:*0:O	H-acceptor	1.63
	1	bond	gen bond	:*0:H		B:ILE228:O	-	1.85
			C-H bond	A:ARG269:HD2		:*0:O		2.82
				:*0:H		A:ASN265:OD1		2.71
	Ring B	Hydro-	Pi-Pi stacked	A:PHE289	Pi-orbitals	:*0	Pi-orbitals	4.73
	7	phobic	Alkyl	:*0:CL	Alkyl	A:MET286	Alkyl	5.41
				:*0:CL		B:LEU232		4.88
				:*0:CL		B:PRO233		4.64
				:*0:CL		B:MET236		4.14
	7		Pi-Alkyl	A:PHE289	Pi-orbitals	:*0:CL		4.39
	Ring B			:*0		B:PRO233		4.39

^aPosition of interaction

^bdistance of interaction

^cinternal interaction

interactions suggested by molecular docking to less than half (Fig. 7d and Table 11), or else, suggested three interactions with γ_2 Lys285 and β_2 Leu268 instead of those established with β_2 Pro228, β_2 Leu223, γ_2 Thr277, γ_2 Thr281, γ_2 Phe304, and γ_2 Ala283. Both γ_2 Lys285 and β_2 Leu268 interact with the substitution group at C₇ through two H-bonds and one Alkyl-Alkyl interaction orienting, respectively, from their side chains towards the lone pairs of the oxygen atom and the methyl group. On the other hand, the number of interactions created with γ_2 Asp297 and β_2 Gln224 is reduced by the factor of one H-bond for each. In contrast, interactions with γ_2 Ser280, γ_2 Arg284, γ_2 Val300, and β_2 Ile264 residues are kept identical between the two simulations.

Unlike Ro12-6377, the binding modes of proflazepam for the TMD γ_2^+/β_2^- interface are equivalents for the two simulations (Fig. 8d). However, notable differences in the binding orientations of the substitution groups at C₅ and C₇ have appeared. As previously noted for Ro12-6377, after MD simulation, the number of interactions was almost reduced to half (Fig. 9d and Table 11). All the interactions established with β_2 Pro228, γ_2 Arg284, γ_2 Asp297, β_2 Leu231, and γ_2 Phe304 have vanished, and instead of them, Pi-Sulfur interaction was observed between the sulfur atom at β_2 Met227 side chain and the Pi-orbitals of phenyl (C). The three H-bonds linked β_2 Gln224 to the CH₂CHOHCH₂OH group were reduced into one moderate H-acceptor bond oriented from the second hydroxyl group towards the main carbonyl group of β_2 Gln224. The halogen interaction established between β_2 Leu223 and the fluorine at C_{2'} vanished, and the H-bond connected C₂ to γ_2 Thr277 was replaced by two moderate H-bonds given from the CH₂CHOHCH₂OH group to the main carbonyl group of the same residue. γ_2 Val300 was involved in additional hydrophobic interaction type Alkyl-Alkyl with the chlorine atom at C₇. Finally, interactions with γ_2 Thr281 and γ_2 Ser280 residues are kept identical between the two simulations.

Overall, the MD results do not agree with those observed in the molecular docking analysis that showed interactions with α_1 His102. As mentioned earlier, this residue ensures the recognition of classical BDZ by the ECD interface. The MD results coincided well with those

Table 10 Detailed binding interactions resulting from the molecular dynamics simulation of Ro12-6377 and proflazepam (PLZ) at the TMD $\beta_2^+(C)/\alpha_1^-(D)$ interface

$\frac{\text{TMD } \beta_2^+(\mathbf{C})/\alpha_1^-(\mathbf{D}) \text{ interface}}{2}$									
Ligand Name	Posit ^a	Category	Type of interactions	From	From chemistry	То	To chemistry	Dist ^b (A°)	
Ro12-	2	Hydrogen bond	Conventional hydrogen	C:ARG269:HH12	H-donor	:*0:O	H-acceptor	1.62	
6377	7		bond	:*0:H	H-donor	D:MET236:O		2.55	
	2'	Hydrogen bond; Halogen	C-H bond; halogen (Fluorine)	C:MET286:HA	H-donor; halogen acceptor	:*0:F	H-acceptor; halogen	2.53	
	3	Hydrogen bond	C-H bond	:*0:H	H-donor	D:ILE228:O	H-acceptor	3.03	
				:*0:H		D:ILE228:O		2.75	
	7			:*0:H		D:MET236:O		2.81	
	2'	Halogen	Halogen (Fluorine)	C:MET286:O	Halogen acceptor	:*0:F	Halogen	3.32	
	Ring B	Other	Pi-Sulfur	D:MET236:SD	Sulfur	:*0	Pi-orbitals	5.26	
		Hydrophobic	Pi-Pi stacked	C:PHE289	Pi-orbitals	:*0		4.95	
	1		Alkyl	:*0:C	Alkyl	C:LEU285	Alkyl	4.68	
				:*0:C		C:MET286		4.54	
	7			:*0:C		C:VAL258		4.89	
	Ring B		Pi-Alkyl	:*0	Pi-orbitals	D:PRO233		4.88	
	Ring C			:*0		C:MET286		5.12	
PLZ	2	Hydrogen bond	Conventional hydrogen	C:ASN265:HD22	H-donor	:*0:O	H-acceptor	2.20	
	1		bond	C:ARG269:HH12		:*0:O		1.42	
	Internal ^c	Hydrogen bond; halogen	Conventional hydro- gen bond; halogen (Fluorine)	:*0:H	H-donor; halogen acceptor	:*0:F	H-acceptor; halogen	1.81	
	1	Hydrogen bond	Conventional hydrogen bond	:*0:H	H-donor	D:ILE228:O	H-acceptor	1.65	
	Internal ^c		C-H bond	:*0:H	H-donor	:*0:F		2.50	
	Internal ^c	Halogen	Halogen (Fluorine)	:*0:N	Halogen acceptor	:*0:F	Halogen	3.44	
	Ring B	Other	Pi-Sulfur	D:MET236:SD	Sulfur	:*0	Pi-orbitals	5.49	
		Hydrophobic	Pi-Pi stacked	C:PHE289	Pi-orbitals	:*0		4.64	
	7		Alkyl	:*0:CL	Alkyl	D:PRO233	Alkyl	4.81	
				:*0:CL		D:MET236		5.26	
			Pi-Alkyl	C:PHE289	Pi-orbitals	:*0:CL		4.94	
	Ring B			:*0		D:PRO233		4.44	
	Ring C			:*0		D:PRO233		5.31	
				:*0		D:LEU269		5.36	

^aPosition of interaction

^bdistance of interaction

^cinternal interaction

noted in the molecular docking analysis that indicated the features of Ro12-6377 and proflazepam to directly connect the pore-lining residues. The bulky structure of Ro12-6377 and proflazepam is the key factor in the deep penetration towards the TM₂ helices, in particular, the long backbone skeleton of the NHCONHCH₃ and CH₂CHOHCH₂OH groups, where most interactions with the TM₂ helices have been observed. At TMD $\beta_2^+(C)/\alpha_1^-(D)$, the MD simulation generated two interactions for Ro12-6377 with

 $\beta_2:C:TM_2:Arg269$ and $\beta_2:C:TM_2:Val~258$ residues. In addition, at TMD γ_2^+/β_2^- interface, three interactions were generated with $\beta_2:A:TM_2:Leu268$ and $\gamma_2:TM_2:Lys285$ instead of those previously observed with TM_2:Thr277, TM_2:Thr281, and TM_2:Ala283of subunit γ_2 . Likewise, at TMD $\beta_2^+(C)/\alpha_1^-(D)$ interface, it suggested an additional interaction for the proflazepam with $\beta_2:C:TM_2:Arg269$, and at TMD γ_2^+/β_2^- interface, suggested a lack of interactions with $\gamma_2:TM_2:Arg~284$.

Table 11 Detailed binding interactions resulting from the molecular dynamics simulation of Ro12-6377 and proflazepam (PLZ) at the TMD γ_2^+ / β_2^- interface

TMD γ_2^+/β_2^- in	terface							
Ligand Name	Posit ^a	Category	Type of interactions	From	From chemistry	То	To chemistry	Dist ^b (A°)
Ro12-6377	7	Hydrogen bond	Conventional hydrogen	E:LYS285:HZ2	H-donor	:*0:0	H-acceptor	2.09
			bond	:*0:H		A:GLN224:O		2.94
				:*0:H		A:GLN224:O		1.74
			C-H bond	E:LYS285:HE2		:*0:O		2.73
	2			E:ASP297:HA		:*0:O		2.40
	1			:*0:H		E:SER280:O		2.96
	Hydropho	Hydrophobic	Alkyl	:*0:C	Alkyl	E:ARG284	Alkyl	4.71
				:*0:C		E:VAL300		4.13
	7			:*0:C		A:ILE264		4.75
				:*0:C	A:LEU268		4.66	
PLZ	1	Hydrogen bond	Conventional hydrogen	:*0:H	H-donor	E:THR281:OG1	H-acceptor	1.81
			bond	:*0:H		A:GLN224:O		2.06
			C-H bond	:*0:H		E:THR277:O		2.86
				:*0:H		E:THR277:O		2.94
				:*0:H		E:SER280:OG		2.51
				:*0:H		E:THR281:OG1		2.60
	Ring C	Other	Pi-Sulfur	A:MET227:SD	Sulfur	:*0	Pi-orbitals	5.18
	7	Hydrophobic	Alkyl	:*0:CL	Alkyl	A:LEU223	Alkyl	4.37
				:*0:CL		E:VAL300		4.07
	Ring B		Pi-Alkyl	:*0	Pi-orbitals	E:VAL300		5.15

^aPosition of interaction

^bdistance of interaction

Pharmacokinetic and drug-likeness prediction

As shown in Table S2, Supplementary materials, all BDZ compounds are estimated to have high gastrointestinal absorption and nearly all are able to cross the blood-brain barrier. Almost all BDZ molecules are not affected by p-gp efflux pump. Besides, all tested compounds respect Lipinski, Veber, Egan, Ghose, and Muegge rules.

Moreover, Ro12-6377 and proflazepam can successfully penetrate the blood–brain barrier. They are also estimated to be active effluxes by P-glycoprotein transporter and to act as non-inhibitors towards CYPisoenzymes, except for CYP2D6 in proflazepam. Interestingly, these two compounds share one favorable characteristic in which they do not inhibit CYP2C19 and CYP3A4 enzymes that might be responsible for the hepatic clearance or the formation of active metabolites of BDZ. Indeed, it is well established that the BDZs are primarily metabolized via CYP2C19 and CYP3A4 and the inhibition of them can cause drug-drug interactions (DDIs) [60, 61].

QSAR analysis

The QSAR study was carried out on the BDZ data set previously investigated through molecular docking simulation (Table 1). The data set includes a total of 65 compounds that were used to generate the PLS regression model and evaluate its performance. Using fivefold cross-validation, we randomly split the 65 observations into two sets, a training set containing 52 of the data points, and a test set containing the remaining 13 observations. The molecular descriptors are coded into the term of variables: $x_1 = q_{N1}$, $x_2 = q_{C3}$, $x_3 = q_{N4}$, $x_4 = q_{C6}$, $x_5 = q_{C7}$, $x_6 = q_{C8}$, $x_7 = q_{C9}$, $x_8 = q_{C2}$, $x_9 = q_{C6}$, $x_{10} = \pi_{C7}$, $x_{11} = \pi_{C2}$, $x_{12} = HA$, $x_{13} = HD$, $x_{14} = DM$, $x_{15} = Log P$, $x_{16} = M_R$, $x_{17} = F_T$. The numerical values are summarized in Table S3, supplementary materials.

Table 12 reports the observed (\bar{y}_i) and predicted (\hat{y}_i) biological activities with the corresponding studentized deleted residual (r_i^*) and the leverage (h_{ii}) values of the training and test set compounds of a sample among the 10,000 simulations generated in this study. In this case, six outliers

Table 12 Studentized deleted residual values (r_i^*) and the leverage values (h_{ii}) of the training and test set compounds

Comp N°	Y _i norm	Y _{hat}	r_i^*	h _{ii}	Comp N°	Y _i norm	Y _{hat}	r_i^*	h _{ii}
Training	set								
1	8.2954	8.1891	0.1713	0.2220	35	8.2954	8.5488	-0.4167	0.2493
3	8.4000	9.1331	-1.1372	0.1352	36	9.2113	9.1223	0.1354	0.1266
4	8.4393	9.7341	-2.0056	0.0804	37	9.1982	9.5757	-0.5616	0.0800
6	9.3159	9.8727	-0.9261	0.2553	39	8.3346	8.8803	-0.8621	0.1764
7	9.6430	9.2510	0.5873	0.0922	40	8.1253	8.9133	-1.2514	0.1698
8	9.6823	10.0519	-0.5490	0.0775	43	8.9627	9.6633	-1.4642	0.5146
9	9.7346	9.7370	-0.0036	0.1082	44	8.5309	9.4693	- 1.5789	0.2455
10	9.8393	9.5026	0.4970	0.0669	45	11.0692	10.9866	0.1345	0.2361
11	9.9701	9.8377	0.1953	0.0698	47	10.6636	9.8739	1.2419	0.1540
13	10.1271	9.7364	0.5867	0.0962	48	9.5645	10.1277	-0.8436	0.0843
14	10.3234	10.4628	-0.2098	0.1072	49	10.1926	9.7383	0.6890	0.1121
15	10.5851	9.9585	0.9361	0.0763	50	10.8206	8.6790	4.5068	0.3328
16	10.8075	11.0797	-0.4154	0.1287	51	8.8580	9.5543	- 1.0520	0.0922
17	10.8206	10.0065	1.2313	0.0859	52	10.1010	10.2745	-0.2602	0.1000
19	11.0561	10.9483	0.1691	0.1788	53	10.2711	9.9006	0.5519	0.0827
20	11.3832	10.6639	1.0698	0.0625	54	11.0169	10.7783	0.3697	0.1553
21	11.4356	11.6426	-0.3182	0.1425	55	8.4916	9.3474	-1.2840	0.0683
22	11.4356	11.1475	0.4431	0.1420	56	8.9234	9.4633	-0.7967	0.0581
23	11.5402	10.0976	2.2560	0.0778	58	10.1010	9.8097	0.4441	0.1263
25	10.1664	10.7639	-1.0675	0.3503	59	9.3421	9.2794	0.0929	0.0785
27	10.5066	10.2804	0.3321	0.0600	60	8.5309	9.3314	-1.2208	0.1014
29	10.9776	10.8453	0.2004	0.1183	61	9.6823	9.8128	-0.1930	0.0745
30	11.1477	11.5546	-0.6271	0.1413	62	11.0430	10.4101	0.9384	0.0621
31	10.6636	11.4656	-1.2521	0.1410	63	10.2711	10.0263	0.3632	0.0791
32	10.4542	9.7895	1.0285	0.1353	64	9.7215	9.7755	-0.0789	0.0547
33	11.3309	9.9877	2.2297	0.1836	65	9.5253	9.8470	-0.4706	0.0503
Test set									
2	8.5167	8.6018	-1.2800	0.3654	34	9.5531	9.1409	0.3076	0.3440
5	9.3272	8.8671	0.5421	0.3889	38	9.2475	9.0331	-0.5152	0.9222
12	10.2041	10.2062	-0.3840	0.5716	41	10.1111	10.3151	-2.3002	0.4636
18	11.1873	9.7623	1.8961	0.5383	42	9.7922	9.4752	0.8742	0.8557
24	11.0146	10.1997	0.7054	0.3238	46	11.8516	12.0887	-1.4227	0.9560
26	10.8020	10.4640	-0.1934	0.2681	57	10.0314	10.2422	-0.7454	0.4415
28	11.6789	10.8360	1.6756	0.5610					

(marked in bold) were detected for the activity response. Four observations have r_i^* greater than the threshold |2| and three observations with large h_{ii} values. Consequently, the training and test sets were reduced to 47 and 12 observations, respectively.

Figures 10 and 11 show the results of the 10,000 simulations obtained for the selected biological activity. According to Fig. 10, the best subsets of variables are those with the highest probability of occurrence according to the BIC criterion. As a result, the best subset of variables to select is that containing six variables. The intercept is systematically included whatever the model. The next step is determining the variables retained in the subset. According to Fig. 11, it appears that the six best variables are x_3 , x_{11} , x_{14} , x_{10} , x_{15} , and x_{16} . This ranking order is a function of their probability of occurrence. By combining the results of Figs. 10 and 11, the best variable subset contains: q_{N4} , $\pi_{C2'}$, DM, π_{C7} , log P, and M_R (Table 13).

In order to highlight the weight of each molecular descriptor, the PLS regression models are written with scaled variables. The standardized regression coefficient value of each descriptor highlights the relative importance of the descriptors in determination of biological activity of the compounds.





The final QSAR model with 95% confidence interval of the regression coefficient is:

 $y^{'} \!=\! \overset{10.0747}{_{(\pm 0.3427)}} \! -\! \overset{0.3546x^{'}_{3}}{_{(\pm 0.0908)}} \! +\! \overset{0.4694x^{'}_{10}}{_{(\pm 0.0679)}} \! +\! \overset{0.4460x^{'}_{11}}{_{(\pm 0.0665)}} \! -\! \overset{0.2381x^{'}_{14}}{_{(\pm 0.0676)}} \! -\! \overset{0.1675x^{'}_{15}}{_{(\pm 0.0882)}} \! -\! \overset{0.3665x^{'}_{16}}{_{(\pm 0.0496)}} \!$

y' is the biological activity of [³H] diazepam derivatives, where $y' = \log(1/C)_{obs}/s_y$ and $x'_j = x_j/s_{x_j}$. S_y : the standard deviation corresponding to the biological response. S_{x_j} : the standard deviation corresponding to the *j*th descriptor.





Table 13	The optimal	variables to	generate the	PLS model
----------	-------------	--------------	--------------	-----------

Comp N°	$q_{ m N4}$	π_7	π _{2'}	MD (debye)	Log P	M _R	Comp N°	$\overline{q}_{\mathrm{N4}}$	π ₇	<i>π</i> _{2'}	MD (debye)	Log P	M _R
1	-0.608	-1.230	0.000	6.301	1.643	79.708	34	-0.665	-1.230	0.140	6.893	1.786	79.924
2	-0.618	-1.230	0.000	6.583	1.779	76.592	35	-0.655	-1.030	0.140	4.302	1.743	93.660
3	-0.590	-0.570	0.000	2.121	2.328	80.729	36	-0.573	0.710	0.000	4.083	4.031	85.262
4	-0.620	0.000	0.000	5.061	2.608	71.891	37	-0.607	0.710	0.000	3.104	3.304	87.392
5	-0.655	-1.340	0.140	6.721	2.123	82.211	38	-0.584	0.710	0.000	3.049	3.857	91.754
6	-0.698	-1.230	0.710	6.791	2.383	81.396	39	-0.569	-0.280	0.000	1.785	2.475	87.181
7	-0.620	-0.650	0.000	3.137	2.320	78.475	40	-0.587	0.710	0.000	3.222	4.130	93.618
8	-0.613	0.140	0.000	3.369	2.751	72.108	41	-0.642	0.710	0.140	3.078	2.529	86.321
9	-0.619	1.020	0.000	5.662	3.566	81.533	42	-0.640	0.710	0.140	5.437	1.843	98.979
10	-0.634	-0.570	0.140	3.218	2.471	80.945	43	-0.634	0.710	0.140	5.140	1.898	92.283
11	-0.631	0.820	0.000	4.936	3.345	81.578	44	-0.657	-0.280	0.710	4.125	4.070	99.938
12	-0.657	0.000	0.140	5.752	2.751	72.108	45	-0.736	0.710	0.140	3.539	3.923	81.406
13	-0.670	-0.550	0.140	3.274	2.308	82.510	46	-0.766	-0.280	0.710	2.404	3.721	87.510
14	-0.614	0.880	0.000	1.909	3.486	77.865	47	-0.698	-0.280	0.140	3.063	3.123	86.038
15	-0.601	0.710	0.000	3.228	3.076	79.812	48	-0.660	0.710	0.000	2.958	3.645	84.306
16	-0.679	0.710	0.710	3.462	4.284	89.421	49	-0.609	0.710	0.000	4.751	2.787	81.010
17	-0.659	0.460	0.140	4.057	4.085	82.395	50	-0.541	0.710	0.140	6.653	4.124	84.764
18	-0.622	-0.280	0.140	3.209	2.555	81.544	51	-0.647	0.560	0.000	5.677	3.635	81.974
19	-0.673	-0.280	0.880	3.692	3.426	84.185	52	-0.674	0.000	0.140	5.429	2.893	72.324
20	-0.665	0.710	0.140	4.140	3.355	76.912	53	-0.648	0.000	0.140	5.768	2.615	75.224
21	-0.696	0.710	0.710	3.938	3.816	81.501	54	-0.666	0.000	0.710	5.557	3.076	79.812
22	- 0.696	-0.280	0.710	2.488	3.152	83.016	55	-0.605	0.000	0.000	4.500	3.212	76.696
23	-0.649	-0.280	0.140	2.378	2.691	78.428	56	-0.635	0.000	0.140	5.583	3.219	80.028
24	-0.637	0.140	0.140	4.373	2.757	75.440	57	-0.684	0.000	0.140	4.474	3.497	77.129
25	-0.597	0.140	0.000	3.338	2.615	57.224	58	-0.676	0.000	0.140	6.296	3.264	77.149
26	-0.654	0.140	0.140	4.242	2.893	72.324	59	-0.620	0.000	0.140	5.954	3.219	80.028
27	-0.624	0.710	0.000	3.257	3.212	76.696	60	-0.632	0.000	0.140	4.116	3.823	84.833
28	-0.689	0.710	0.140	3.645	3.497	77.129	61	-0.611	0.710	0.000	2.420	3.680	84.617
29	-0.700	0.710	0.140	3.646	3.361	80.245	62	-0.662	0.710	0.140	3.363	3.959	81.717
30	-0.681	0.710	0.710	3.557	3.959	81.717	63	-0.668	0.560	0.140	5.148	3.868	81.954
31	-0.710	0.710	0.710	3.486	4.420	86.306	64	-0.614	0.710	0.000	3.589	3.816	81.501
32	-0.615	-0.280	0.000	1.381	2.548	78.212	65	-0.617	0.710	0.000	3.336	3.725	81.737
33	-0.663	-0.280	0.710	3.365	3.016	97.151							

The best subset of molecular descriptors providing a good prediction of the response variable corresponds to: $x_3 = q_{N4}$, $x_{10} = \pi_{C7}$, $x_{11} = \pi_{C2}$, $x_{14} = DM$, $x_{15} = Log P$, $x_{16} = M_R$.

Table 14 summarizes the statistical indicators used for internal and external validation. According to the goodness of fit statistics, 63.2% of the variability in BDZ activity

around its mean is explained by the PLS regression equation. The quality of models can be judged and compared based on the R_{adj}^2 values. The F-statistics reveal the significance of the PLS regression equations. The obtained *p*-value shows that the model is statistically highly significant at 95%. Moreover, it is well known that cross-validation is

of	fit	stati	stics,	63.2%	of	the	variability	ir

Table 14 Quality and validation

metrics

Model	Goodnes	s of fit	Goodness of prediction			
	$\overline{R^2}$	R_{adj}^2	F	<i>p</i> -value	Q_{loo}^2	Q_{F3}^2
$Y(n_{\rm tr} = 52; n_{\rm ts} = 13)$	0.632	0.584	12.806	6.2050e – 07	0.639	0.813

 n_{tr} is the training set used for building the PLS regression equation and n_{ts} is the test set used to verify a model's predictive ability for new untested molecules

useful for overcoming the problem of overfitting [32]. This problem refers to a situation when the model requires more information than the data can provide. Indeed, in our case the difference between R^2 and Q_{loo}^2 is much less than the threshold of 0.30, confirming that the PLS regression models are not overfitted. Additionally, our model exhibits a high value of Q_{F3}^2 (0.813). This result confirms that the resulting QSAR model has good external predictability and robustness.

Molecular descriptors with positive regression coefficients indicate positive correlations with observed activity. Therefore, their increase will improve the $GABA_A/BDZ$ response. Our model demonstrates a positive contribution from the hydrophobic constants of the substitutes at C_7 and C_2 , as well as demonstrates a negative contribution from molecular lipophilicity (log P), molecular polarity (DM), molecular size (M_R), and net charge of $N_4(q_{N4})$.

According to the standardized regression coefficient values, the hydrophobicity in positions 7 and 2' are the main properties for determining the biological activities of the studied compounds. This result is in good agreement with previous reports that the 7 and 2' positions are, respectively, the first and second key positions in the BDZ structures to influence binding with GABAARs [14]. Otherwise, the lipophilic behavior shows the least effect on activity. Molecular lipophilicity is directly influenced by both molecular hydrophobicity and polarity [62]. In our case, the positive effects of hydrophobicity at positions 7 and 2' on the observed activity seem to be two times more significant than the negative effect of molecular polarity. The lipophilic feature plays a pivotal role in understanding the pharmacokinetics parameters, pharmacodynamics, and toxicological profile of drugs. Besides, it exhibits an important influence on host-guest interaction and drug binding affinity [46]. BDZ derivatives are relatively weak bases with highly lipophilic characters. At physiological pH, this lipophilicity explains their strong binding to plasma proteins (70-99%) as well as their high penetration through the blood-brain barrier [63].

 N_4 net charge and molar refractivity exhibit about twice the negative influence of molecular lipophilicity on activity. Reducing the negative charge of N_4 is important to avoid the formation of water-unstable BDZ salts. The imine group (N_4) is the most basic nitrogen in the classical BDZ structure. Since the amide group at positions 1 and 2 has a nonbasic character, the lone pair of N_4 can easily protonate when placed in a strongly acidic environment. Thus, it leads to the formation of BDZ salts (iminium ion). Unfortunately, the salts of strong acids are unstable in an aqueous medium and are therefore undergoing sequential hydrolysis of the imine and amide groups, respectively. The imine hydrolysis reaction is reversible. In contrast, hydrolysis of amide leads to the formation of inactive products and, consequently, eliminates activity towards GABA_A receptors [64]. Most of the classical BDZ agents we study here provide unstable salts and are relatively water-insoluble than drugs formed from heterocyclic BDZs. In the latter, the amide group is protected in the form of heterocyclic groups such as ImidazoBDZs and TriazoloBDZs. Hence, hydrolysis of the amide does not occur and the reaction does not lead to inactive products. To improve the solubility of classicalBDZ salts in water-soluble injections, in addition to water, it is necessary to use co-solvents such as PEG 400, propylene glycol, 10% ethyl alcohol, and 2% benzyl alcohol [65].

Our results support those of Maddalena and So [14, 15] by asserting that increased hydrophobicity at position 7 is necessary to obtain highly potent BDZ analogs. Besides, it provides complementary insights on how hydrophobicity at C_2 . position, the net charge of N_4 , and molar refractivity, polarity, and lipophilicity of the entire molecules would be ameliorated to achieve the optimal activity. In particular, the challenge is to provide structures meeting simultaneously the requirements of low lipophilicity/low polarity, since they are naturally anti-correlated. Here, our model suggests that the influence of molecular polarity on activity outweighs that of molecular lipophilicity by 7%. Accordingly, efforts should be devoted primarily to reducing molecular polarity rather than molecular lipophilicity.

After excluding outliers, a comprehensive analysis of the detailed binding interactions with the four binding interfaces was performed for the training and test data sets. Subsequently, the hydrophobic interactions with C_7 and $C_{2'}$ substitutes and the electrostatic interactions with N₄ were selected and collected in Tables S4 to S7, in supplementary materials. As can be seen, the richness of the C_7 position by hydrophobic interactions is attributed in most cases to the presence of chlorine atoms that exhibited Alkyl-Alkyl or Pi-Alkyl interactions with neighboring residues. Significantly, the response of chlorine to establish hydrophobic interactions with the four binding interfaces is different between ECD and TMD. At ECD α_1^+/γ_2^- interface, it tends to act as an acceptor of interactions, while at the three TMD interfaces it acts as a generator of interactions. $C_{2'}$ position shows a total lack of hydrophobic interactions with the four binding interfaces, except for the case of the CF₃ group in Ro05-3590 and the chlorine atom in Ro05-4608. This deficiency maybe because most of the compounds of our data set contains small substituents (H and F) which mainly tend to interact electrostatically rather than hydrophobically. The N₄ atom tends to react as a hydrogen-acceptor from neighboring residues. Importantly, in the ECD α_1^+/γ_2^- interface, all interactions observed for N_4 are received from α_1 His 102 side chain. As reported previously in the molecular docking section, this residue is known to be important in the recognition of classical BDZ. This result does not correlate with previous analyses that support cationic interactions at this position [66].

Conclusion

In this investigation, a combination of in silico approaches including molecular docking/dynamic simulations, and QSAR analysis have been performed to achieve two purposes: elucidate the binding mechanism by which a dataset of [³H]diazepam derivatives allosterically modulates GABA_A receptor $\alpha_1\beta_2\gamma_2$ subtypes and identify the structural details that contribute to ameliorate the $\alpha_1\beta_2\gamma_2/BDZ$ response. Examination of binding affinities revealed that the known ECD is the target for the majority of classical benzodiazepines. However, the tendency of the remainder to bind mainly on the binding interfaces included in TMD, or in some cases, to act on both ECD and TMD binding sites simultaneously, cannot be overlooked. This result opens the way for further studies that may combine binding in these binding sites with diversity in the activities of benzodiazepines. Binding affinity-based screening identified Ro12-6377and proflazepam as the best modulators for the four binding interfaces. By monitoring the dynamic behaviors over a time period of 1000 ps, the two modulators were observed to have equivalent stability within the four binding sites under study. Binding modes after MD simulation were altered from the structures generated by molecular docking. Thus, several differences in the binding interactions with key residues were detected between the two simulations. Importantly, interactions with pore-lining residues have been suggested for both modulators. The combination of ADMEprediction/Drug-likeness prediction shows their good pharmacokinetic properties as well as their compliance with all drug-likeness rules. Furthermore, the developed QSAR model yielded satisfactory statistical results that explain 63.2% of the variability in benzodiazepine activity. Its stability and predictive power were ensured based on internal and external validation indicators: $R_{adi}^2 = 0.584$, F = 12.806; p-value = 6.2050e - 07, Q_{loo}^2 =0.639, and $Q_{E_2}^2$ =0.813. The model equation demonstrates a positive contribution from the hydrophobicity of the substitutes at C7 and C2', as well as demonstrates a negative contribution from molecular lipophilicity, molecular polarity, molecular size, and net charge of N₄. The model results agree well with previous findings indicating that the increase in hydrophobicity at 7-position mainly contributes to the enhancement of BDZ activity. Finally, the combination of the results of both methods: QSAR and molecular docking, shows that the hydrophobic interactions at the 7-position are mostly attributed to the substitutions of chlorine atoms. The latter tends to act as an acceptor of interactions in the ECD binding interfaces and as a generator of interactions in the three TMD binding interfaces.

Supplementary Information The online version contains supplementary material available at https://doi.org/10.1007/s11224-022-02029-4. Author contribution RD: conceptualization, software, validation, investigation, writing—original draft, writing—review & editing. SK: supervision, conceptualization, software, validation, investigation, writing original draft, formal analysis, resources, writing—review & editing. ID: conceptualization, software, validation, resources, writing—review & editing. NM: conceptualization, software, supervision, validation, investigation, resources, writing—review & editing. AB: writing—original draft, writing—review & editing. FM: writing—review & editing.

Funding This research did not receive any specific grant from funding agencies in the public, commercial, or not-for-profit sectors.

Declarations

Conflict of interest The authors declare no competing interests.

References

- Bergmann R, Kongsbak K, Sørensen PL, Sander T, Balle T (2013) A unified model of the GABAA receptor comprising agonist and benzodiazepine binding sites. PLoS One 8. https://doi. org/10.1371/journal.pone.0052323
- Zhu S, Noviello CM, Teng J, Walsh RM, Kim JJ, Hibbs RE (2018) Structure of a human synaptic GABAA receptor. Nature 559:67– 88. https://doi.org/10.1038/s41586-018-0255-3
- Sigel E, Steinmann ME (2012) Structure, function, and modulation of GABAA receptors. J Biol Chem 287:40224–40231. https:// doi.org/10.1074/jbc.R112.386664
- Miller PS, Aricescu AR (2014) Crystal structure of a human GABAA receptor. Nature 512:270–275. https://doi.org/10.1038/nature13293
- Olsen RW, Sieghart W (2008) International Union of Pharmacology. LXX. Subtypes of γ-aminobutyric acidA receptors: classification on the basis of subunit composition, pharmacology, and function. Update Pharmacol Rev 60:243–260. https://doi.org/10.1124/pr.108.00505
- Chuang SH, Reddy DS (2018) Genetic and molecular regulation of extrasynaptic GABA-A receptors in the brain: therapeutic insights for epilepsy. J Pharmacol Exp Ther 364:180–197. https:// doi.org/10.1124/jpet.117.244673
- Kim JJ, Gharpure A, Teng J, Zhuang Y, Howard RJ, Zhu S, Noviello CM, Walsh RM, Lindahl E, Hibbs RE (2020) Shared structural mechanisms of general anaesthetics and benzodiazepines. Nature 585:303–308. https://doi.org/10.1038/s41586-020-2654-5
- Masiulis S, Desai R, Uchański T, Serna Martin I, Laverty D, Karia D, Malinauskas T, Zivanov J, Pardon E, Kotecha A, Steyaert J, Miller KW, Aricescu AR (2019) GABAA receptor signalling mechanisms revealed by structural pharmacology. Nature 565:454– 459. https://doi.org/10.1038/s41586-018-0832-5
- Richter L, De Graaf C, Sieghart W, Varagic Z, Mörzinger M, De Esch IJP, Ecker GF, Ernst M (2012) Diazepam-bound GABAA receptor models identify new benzodiazepine binding-site ligands. Nat Chem Biol 8:455–464. https://doi.org/10.1038/nchembio.917
- Xu Y, He Z, Liu H, Chen Y, Gao Y, Zhang S, Wang M, Lu X, Wang C, Zhao Z, Liu Y, Zhao J, Yu Y, Yang M (2020) 3D-QSAR, molecular docking, and molecular dynamics simulation study of thieno[3,2-: B] pyrrole-5-carboxamide derivatives as LSD1 inhibitors. RSC Adv 10:6927–6943. https://doi.org/10.1039/ c9ra10085g
- Alonso H, Bliznyuk AA, Gready JE (2006) Combining docking and molecular dynamic simulations in drug design. Med Res Rev 26:531–568. https://doi.org/10.1002/med.20067
- Verma J, Khedkar V, Coutinho E (2010) 3D-QSAR in drug design

 a review. Curr Top Med Chem 10:95–115. https://doi.org/10. 2174/156802610790232260

- Roy K (2007) On some aspects of validation of predictive quantitative structure-activity relationship models. Expert Opin Drug Discov 2:1567–1577. https://doi.org/10.1517/17460441.2.12.1567
- Maddalena DJ, Johnston GAR (1995) Prediction of receptor properties and binding affinity of ligands to benzodiazepine/GABAA receptors using artificial neural networks. J Med Chem 38:715– 724. https://doi.org/10.1021/jm00004a017
- So SS, Karplus M (1996) Genetic neural networks for quantitative structure-activity relationships: Improvements and application of benzodiazepine affinity for benzodiazepine/GABAA receptors. J Med Chem 39:5246–5256. https://doi.org/10.1021/ jm9605360
- Hadjipavlou-Litina D, Hansch C (1994) Quantitative structure-activity relationships of the benzodiazepines. A review and reevaluation. Chem Rev 94:1483–1505. https://doi.org/10. 1021/cr00030a002
- Micheli A, Sperduti A, Starita A, Bianucci AM (2001) Analysis of the internal representations developed by neural networks for structures applied to quantitative structure-activity relationship studies of benzodiazepines. J Chem Inf Comput Sci 41:202– 218. https://doi.org/10.1021/ci9903399
- HyperChem v8. Molecular Modelling System, Hypercube Inc., 1115 NW 4th Street, Gainesville, FL 32601, USA (2009).
- Frisch MJ, Trucks GW, Schlegel HB, Scuseria GE, Robb MA, Cheeseman JR, Scalmani G, Barone V, Petersson GA, Nakatsuji H, Li X, Caricato M, Marenich A, Bloino J, Janesko BG, Gomperts R, Mennucci B, Hratchian HP, Ortiz JV, Izmaylov AF, Sonnenberg JL, Williams-Young D, Ding F, Lipparini F, Egidi F, Goings J, Peng B, Petrone A, Henderson T, Ranasinghe D, Zakrzewski VG, Gao J, Rega N, Zheng G, Liang W, Hada M, Ehara M, Toyota K, Fukuda R, Hasegawa J, Ishida M, Nakajima T, Honda Y, Kitao O, Nakai H, Vreven T, Throssell K, Montgomery JA, Peralta Jr JE, Ogliaro F, Bearpark M, Heyd JJ, Brothers E, Kudin KN, Staroverov VN, Keith T, Kobayashi R, Normand J, Raghavachari K, Rendell A, Burant JC, Iyengar SS, Tomasi J, Cossi M, Millam JM, Klene M, Adamo C, Ca R, Fox DJ, Gaussian 09W (2010)
- Chirlian LE, Francl MM (1987) Atomic charges derived from electrostatic potentials: a detailed study. J Comput Chem 8:894– 905. https://doi.org/10.1002/jcc.540080616
- 21. MarvinSketch was used for calculating molecular descriptors, Marvin (2020). https://www.chemaxon.com
- 22. Molegro A. MVD 5.0 molegro virtual docker. DK8000 Aarhus C, Denmark 2011.
- Gupta SP, Saha RN, Mulchandani V (1993) Quantitative structure– activity relationship studies on benzodiazepine receptor binding: recognition of active sites in receptor and modelling of interaction. J Mol Recognit 5:75–80. https://doi.org/10.1002/JMR.300050205
- 24. Rawlings JO, Pantula SG, Dickey DA (1999) Applied regression analysis: a research tool, Second Edition
- 25. Rawlings JO, Pantula SG, Dickey DA (1998) Applied regression analysis : a research tool, Second Edition, Second Edi
- De Jong S (1993) SIMPLS: an alternative approach to partial least squares regression. Chemom Intell Lab Syst 18:251–263. https://doi.org/10.1016/0169-7439(93)85002-X
- Wiens TS, Dale BC, Boyce MS, Kershaw GP (2008) Three way k-fold cross-validation of resource selection functions. Ecol Modell 212:244–255. https://doi.org/10.1016/j.ecolmodel.2007.10.005
- James G, Witten D, Hastie T (2013;) Robert Tibshirani, An INTRODUCTION TO STATISTICAL LEARNING - WITH APPLICATIONS IN R, New York: springer
- 29. Schwarz G (1978) Estimating the dimension of a model. Ann Stat 6:461–464. http://www.jstor.org/stable/2958889
- Renaud O, Victoria-Feser MP (2010) A robust coefficient of determination for regression. J Stat Plan Inference 140:1852–1862. https://doi.org/10.1016/j.jspi.2010.01.008

- Harel O (2009) The estimation of R2 and adjusted R2 in incomplete data sets using multiple imputation. J Appl Stat 36:1109–1118. https://doi.org/10.1080/02664760802553000
- Veerasamy R, Rajak H, Jain A, Sivadasan S, Varghese CP, Agrawal RK (2011) Validation of QSAR models - strategies and importance. Int J Drug Des Disocovery 2:511–519
- Consonni V, Ballabio D, Todeschini R (2009) Comments on the definition of the Q2 parameter for QSAR validation. J Chem Inf Model 49:1669–1678. https://doi.org/10.1021/ci900115y
- Consonni V, Ballabio D, Todeschini R (2010) Evaluation of model predictive ability by external validation techniques. J Chemom 24:194–201. https://doi.org/10.1002/cem.1290
- Consonni V, Todeschini R, Ballabio D, Grisoni F (2019) On the misleading use of QF32 for QSAR model comparison. Mol Inform 38:2–6. https://doi.org/10.1002/minf.201800029
- Molecular Operating Environment (MOE), 2014.09; Chemical Computing Group Inc., 1010 Sherbooke St. West, Suite #910,. Montreal, QC, Canada, H3A 2R7, 2014.
- 37. BIOVIA (2020) Dassault Systèmes, Discovery Studio Visualizer
- Bond SD, Leimkuhler BJ, Laird BB (1999) The Nosé-Poincaré Method for Constant Temperature Molecular Dynamics. J Comput Phys 151:114–134. https://doi.org/10.1006/jcph.1998.6171
- Halgren TA (1996) Performance of MMFF94*. J Comput Chem 17:490–519. http://journals.wiley.com/jcc
- OriginPro 9.1. OriginLab Corporation, One Roundhouse Plaza, Suite 303, Northampton, MA 01060, United States.
- 41. SwissADME (2021). http://www.swissadme.ch (accessed 15 May 2021)
- 42. Belkadi A, Kenouche S, Melkemi N, Daoud I, Djebaili R (2021) K-means clustering analysis, ADME/pharmacokinetic prediction, MEP, and molecular docking studies of potential cytotoxic agents. Struct Chem. https://doi.org/10.1007/s11224-021-01796-w
- 43. Nasution MAF, Toepak EP, Alkaff AH, Tambunan USF (2018) Flexible docking-based molecular dynamics simulation of natural product compounds and Ebola virus Nucleocapsid (EBOV NP): a computational approach to discover new drug for combating Ebola. BMC Bioinformatics 19:137–176. https://doi.org/10.1186/s12859-018-2387-8
- Hevener KE, Zhao W, Ball DM, Babaoglu K, Qi J, White SW, Lee RE (2009) Dihydropteroate synthase. J Chem Inf Model 444–460. https://doi.org/10.1021/ci800293n
- 45. Durdagi S, Aksoydan B, Dogan B, Sahin K, Shahraki A, Birgul-Iyison N (2020) Screening of clinically approved and investigation drugs as potential inhibitors of SARS-CoV-2 main protease and spike receptorbinding domain bound with ACE2 COVID19 target proteins: a virtual drug repurposing study. ChemRxiv. https://doi.org/10.26434/chemrxiv. 12032712.v2
- Ginex T, Vazquez J, Gilbert E, Herrero E, Luque FJ (2019) Lipophilicity in drug design: an overview of lipophilicity descriptors in 3D-QSAR studies, Future. Med Chem 11:1177–1193. https:// doi.org/10.4155/fmc-2018-0435
- 47. Jeffrey GA (1997) An introduction to hydrogen bonding, Oxford University Press
- Singh J, Petter RC, Baillie TA, Whitty A (2011) The resurgence of covalent drugs. Nat Rev Drug Discov 10:307–317. https://doi.org/10. 1038/nrd3410
- 49. Raschka S, Wolf AJ, Bemister-Buffington J, Kuhn LA (2018) Protein-ligand interfaces are polarized: discovery of a strong trend for intermolecular hydrogen bonds to favor donors on the protein side with implications for predicting and designing ligand complexes. BioRxiv. https://doi.org/10.1101/260612
- Janiak C (2000) A critical account on n-n stacking in metal complexes with aromatic nitrogen-containing ligands. J Chem Soc Dalt Trans 3885–3896. https://doi.org/10.1039/b003010o
- 51. Burley SK, Petsko GA (1985) Aromatic-aromatic interaction: a mechanism of protein structure stabilization. Science (80-.) 229:23–28. https://doi.org/10.1126/science.3892686

- 52. Piovesan D, Minervini G, Tosatto SCE (2016) The RING 2.0 web server for high quality residue interaction networks. Nucleic Acids Res 44:W367–W374. https://doi.org/10.1093/nar/gkw315.
- 53. Peikert M, Chen X, Chi L, Brezesinski G, Janich S, Würthwein EU, Schäfer HJ (2014) Phase behavior and molecular packing of octadecyl phenols and their methyl ethers at the air/water interface. Langmuir 30:5780–5789. https://doi.org/10.1021/la404340h
- 54. Tan KR, Gonthier A, Baur R, Ernst M, Goeldner M, Sigel E (2007) Proximity-accelerated chemical coupling reaction in the benzodiazepine-binding site of γ-aminobutyric acid type A receptors: superposition of different allosteric modulators. J Biol Chem 282:26316–26325. https://doi.org/10.1074/jbc.M702153200
- 55. Derry JMC, Dunn SMJ, Davies M (2004) Identification of a residue in the γ-aminobutyric acid type A receptor α subunit that differentially affects diazepam-sensitive and -insensitive benzodiazepine site binding. J Neurochem 88:1431–1438. https://doi. org/10.1046/j.1471-4159.2003.02264.x
- Rudolph U, Möhler H (2004) Analysis of GABAA receptor function and dissection of the pharmacology of benzodiazepines and general anesthetics through mouse genetics. Annu Rev Pharmacol Toxicol 44:475–498. https://doi.org/10.1146/annurev.pharmtox. 44.101802.121429
- Dämgen MA, Biggin PC (2020) A refined open state of the glycine receptor obtained via molecular dynamics simulations. Structure 28:130-139.e2. https://doi.org/10.1016/j.str.2019.10.019
- Bliznyuk AA, Gready JE (2006) Combining docking and molecular dynamic simulations in drug design. 26. https://doi.org/10. 1002/med.20067
- Daoud I, Melkemi N, Salah T, Ghalem S (2018) Combined QSAR, molecular docking and molecular dynamics study on new Acetylcholinesterase and Butyrylcholinesterase inhibitors. Elsevier Ltd. https://doi.org/10.1016/j.compbiolchem.2018.03.021

- Dean L (2020) Diazepam Therapy and CYP2C19 Genotype. Med Genet Summ 1–7. http://www.ncbi.nlm.nih.gov/pubmed/28520370
- Fukasawa T, Suzuki A, Otani K (2007) Effects of genetic polymorphism of cytochrome P450 enzymes on the pharmacokinetics of benzodiazepines. J Clin Pharm Ther 32:333–341. https://doi. org/10.1111/j.1365-2710.2007.00829.x
- Tsopelas F, Giaginis C, Tsantili-Kakoulidou A (2017) Lipophilicity and biomimetic properties to support drug discovery. Expert Opin Drug Discov 12:885–896. https://doi.org/10.1080/17460441.2017. 1344210
- Saari TI, Uusi-Oukari M, Ahonen J, Olkkola KT (2011) Enhancement of GABAergic activity: neuropharmacological effects of benzodiazepines and therapeutic use in anesthesiology. Pharmacol Rev 63:243–267. https://doi.org/10.1124/pr.110.002717
- 64. DeRuiter J (2004) GABA receptor ligands and the benzodiazepines, Princ. Drug Action 2. Fall 1–27
- Porterfield L (1988) Principles of drug action. AD Nurse 3:11–12. https://doi.org/10.4135/9781483326818.n3
- Greco G, Novellino E, Silipo C, Vittoria A (1992) Study of benzodiazepines receptor sites using a combined QSAR-CoMFA approach. Quant Struct Relationships 11:461–477. https://doi. org/10.1002/qsar.2660110403

Publisher's Note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.

Springer Nature or its licensor holds exclusive rights to this article under a publishing agreement with the author(s) or other rightsholder(s); author self-archiving of the accepted manuscript version of this article is solely governed by the terms of such publishing agreement and applicable law.