



Communication The Effects of Temperature on the Growth of a Lead-Free Perovskite-Like (CH₃NH₃)₃Sb₂Br₉ Single Crystal for An MSM Photodetector Application

Chien-Min Hun⁺, Ching-Ho Tien⁺, Kuan-Lin Lee, Hong-Ye Lai and Lung-Chien Chen^{*}

Department of Electro-Optical Engineering, National Taipei University of Technology, Taipei 10608, Taiwan; dreamsmin@gmail.com (C.-M.H.); chtien@mail.ntut.edu.tw (C.-H.T.); t102658016@gmail.com (K.-L.L.); joe80314@gmail.com (H.-Y.L.)

* Correspondence: ocean@ntut.edu.tw; Tel.: +886-2-27712171

+ These authors contributed equally to this work.

Abstract: We have fabricated a photodetector based on $(CH_3NH_3)_3Sb_2Br_9$ (MA₃Sb₂Br₉) lead-free perovskite-like single crystal, which plays an important role in the optoelectronic characteristics of the photodetector as a perovskite-like photosensitive layer. Here, MA₃Sb₂Br₉ single crystals were synthesized by an inverse temperature crystallization process with a precursor solution at three different growth temperatures, 60 °C, 80 °C, and 100 °C. As a result, a MA₃Sb₂Br₉ single crystal with an optimum growth temperature of 60 °C presented a low trap density of 2.63 × 10¹¹ cm⁻³, a high charge carrier mobility of 0.75 cm² V⁻¹ s⁻¹, and excellent crystal structure and optical absorption properties. This MA₃Sb₂Br₉ perovskite-like photodetector displayed a low dark current of 8.09 × 10⁻⁹ A, high responsivity of 0.113 A W⁻¹, and high detectivity of 4.32 × 10¹¹ Jones.

Keywords: lead-free perovskite; MA₃Sb₂Br₉; perovskite single crystal; perovskite photodetector

1. Introduction

Organic-inorganic lead halide composite perovskite materials have received widespread attention because of their excellent photoelectric properties. Due to the advantages of low exciton binding energy, high carrier mobility, long carrier diffusion lengths, wide light absorption range, high light absorption, and tunable optical band gap suitable for solar spectrum, they have quickly attracted widespread attention in the field of highefficiency photovoltaic cells [1–5]. In recent years, the power conversion efficiency (PCE) of photovoltaic cells made of Pb-based organic-inorganic composite perovskite materials has increased from less than 5% to more than 25.5% [6–9]. At the same time, they have shown great application potential in the fields of photodetectors [10,11] and light-emitting diodes (LEDs) [12,13], but the above application research was mostly limited to polycrystalline thin film materials; there were few reports on the preparation and application of singlecrystal materials. In the application of semiconductor materials, single-crystal materials are of great importance. There are a large number of grain boundaries and more defects in polycrystalline materials, so their physical properties are generally not as good as those of single-crystal materials. Optoelectronic devices made of polycrystalline materials, such as photovoltaic cells, generally have lower performance than single-crystal material devices. Therefore, the preparation of single-crystal perovskite materials and the study of their properties and applications have become important in this field [5,14–18].

However, although Pb-based perovskite solar cells have achieved high PCE, the toxicity of heavy metal Pb has a greater impact on human health and the environment. Therefore, the post-processing Pb content has to be considered [19,20]. Although the Pb content per square meter in perovskite solar panels was only a few hundred milligrams, this issue should be given attention when using this material on a large scale. To solve this



Citation: Hun, C.-M.; Tien, C.-H.; Lee, K.-L.; Lai, H.-Y.; Chen, L.-C. The Effects of Temperature on the Growth of a Lead-Free Perovskite-Like $(CH_3NH_3)_3Sb_2Br_9$ Single Crystal for An MSM Photodetector Application. *Sensors* **2021**, *21*, 4475. https:// doi.org/10.3390/s21134475

Academic Editor: Antonio Di Bartolomeo

Received: 9 June 2021 Accepted: 29 June 2021 Published: 30 June 2021

Publisher's Note: MDPI stays neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Copyright: © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). problem, researchers have turned their attention to many low-toxic or lead-free perovskite materials. The development of lead-free organic–inorganic-composite perovskite materials is one of the most important research topics at present and also a challenging problem. So far, tin (Sn) [21,22], germanium (Ge) [23,24], copper (Cu) [25,26], bismuth (Bi) [27,28], and antimony (Sb) [29,30] have been used as replacements for the Pb in Pb-based perovskite materials. Sn was the earliest metal element that was considered to be a more environmentally friendly element with the potential to replace Pb. However, Sn²⁺ and Ge²⁺ were more unstable than Pb²⁺ in the air; they were easily and quickly oxidized into the more stable Sn⁴⁺ and Ge⁴⁺ forms in an oxidizing atmosphere, which led to rapid destruction of the perovskite structure [31–34]. It is important to solve the stability of Sn-based or Ge-based organic–inorganic-composite perovskite materials in the air, and carry out research on the growth and basic performance of new narrow-bandgap lead-free perovskite materials, which is the key to extending the application of perovskite materials in the optoelectronics field.

Perovskite is a promising material for applications of fast and tunable broadband and narrowband photodetectors [35]. Recently, a material with the same structure as perovskite or a similar one has appeared, $A_3Bi_2X_9$ (A = MA or Cs; X = I, Br, and Cl), a bismuth halide material, which is a more typical perovskite-like material. Since Bi is a low-toxic element adjacent to Pb, A3Bi2X9 becomes a potential new light-absorbing material to replace Pbbased perovskite materials, which have better stability than MAPbI₃ perovskites [36]. As an element of the same main group of Bi, Sb has an outer electron arrangement similar to that of Bi and has become a substitute material for Pb [37]. Han et al. reported that a photodetector based on $MA_3Sb_2I_9$ microcrystals has a high responsivity of 40 A W^{-1} under monochromatic light (460 nm) [38]. Pan et al. reported a highly stable and lead-free Cs₃Bi₂I₉ perovskite nanoplate for visible light photodetection applications, showing a maximum light responsivity of 33.1 mA W^{-1} under the irradiation of a 450-nm laser [26]. Zhai et al. synthesized submillimeter-size monocrystalline lead-free Cs₃Sb₂Br₉ perovskite nanoflakes using an inverse temperature crystallization (ITC) method, demonstrating that the $Cs_3Sb_2Br_9$ perovskite nanoflake photodetector has a fast response speed of 24/48 ms, a high responsivity of 3.8 A W⁻¹, and an excellent detectivity of 2.6×10^{12} Jones [39]. However, the nano/microscale device size is too small and the nanoflakes were deposited on the electrodes at random, which limits the applications in the future. The responsiveness of lead-free perovskite photodetectors reported so far was limited, and the performance of the device was far below the requirements of sensitive photoelectric detection. Therefore, it is urgent to study this issue to develop high-performance, lead-free perovskite photodetectors.

Herein, we successfully prepared millimeter-sized, single-crystal, hexagonal, lead-free MA₃Sb₂Br₉ perovskite via the ITC method. By adjusting the growth temperature and the constant temperature control of the oven, the morphology of the lead-free crystal can be controlled. Moreover, it was revealed that the high quality, smooth surface, and uniform geometric shape enabled the as-prepared crystals to have good optical properties and light absorption function, and allowed for the study of its MA₃Sb₂Br₉ perovskite single crystal in metal–semiconductor–metal (MSM) photodetector applications.

2. Materials and Methods

2.1. Materials

Methylammonium bromide (MABr, 99.9%) was purchased from Lumtec (New Taipei, Taiwan). Antimony (III) bromide (SbBr₃, 99.995%) was purchased from Alfa Aesar (Haverhill, MA, USA). Gamma-butyrolactone (GBL, \geq 99.9%) was purchased from Echo Chemical (Miaoli, Taiwan). C₆₀ Fullerene (C₆₀, 99.95%) was purchased from Sigma-Aldrich (St. Louis, MO, USA). Silver slug (Ag, 99.999%) was purchased from Gredmann (Taipei, Taiwan).

2.2. Preparation of MA₃Sb₂Br₉ Single Crystals and Photodetector Device

For the ITC process in Figure 1, MABr and SbBr₃ precursors were dissolved at a

1.5:1.5 molar ratio in 1 mL GBL solvent. The solution was stirred at 700 rpm for 30 min to ensure proper dissolution. After adding it to the Petri dish, the solution was maintained at 60 °C, 80 °C, or 100 °C until the solution was completely evaporated in the hot air circulation oven. During the heating process, the Petri dish had to be covered to prevent the solution from evaporating too quickly. For a while, a few small MA₃Sb₂Br₉ crystallites appeared in the liquid. The MA₃Sb₂Br₉ single crystals gradually grew into a hexagonal shape and became larger in solution. Moreover, the growth times of the final MA₃Sb₂Br₉ single crystals at 60 °C, 80 °C, and 100 °C were about 120, 72, and 36 h, respectively. As a result, the crystal dimensions (7 \times 8 \times 4 mm), (5 \times 6 \times 3.5 mm), (4 \times 5 \times 3 mm) of the MA₃Sb₂Br₉ single crystals were obtained at 60 °C, 80 °C, and 100 °C, respectively (Figure 2). The lower growth temperature makes the precursor solution volatilize more slowly, which promotes a low growth rate to produce larger and better single crystals. Conversely, the final crystal grown at high temperature becomes smaller and has more defects due to the fast growth rate and faster volatilization of the precursor solution. Finally, 20 nm C_{60} and 300 nm Ag were deposited on the MA₃Sb₂Br₉ single crystals with a finger mask to give the MSM photodetector $(Ag/C_{60}/MA_3Sb_2Br_9/C_{60}/Ag)$ an effective illuminated area of $3.84 \times 10^{-2} \text{ cm}^2$.



Figure 1. MA₃Sb₂Br₉ single crystals' growth process and a schematic diagram of an MSM-structured photodetector.



Figure 2. Photographs of MA₃Sb₂Br₉ single crystals with different growth temperatures.

2.3. Characterization

Morphologies of the $MA_3Sb_2Br_9$ single crystals were inspected using a field-emission scanning electron microscope (FESEM, ZEISS Sigma, ZEISS, Munich, Germany). Crystal crystallization qualities were characterized using an X-ray diffractometer (X'Pert PRO MRD, PANalytical, Almelo, Netherlands). The optical properties were measured by a fluorescence spectrophotometer (F-7000, Hitachi, Tokyo, Japan) and a UV–VIS/NIR spectrophotometer (UH-4150, Hitachi). The electrical characteristics of the devices were measured using a Keithley 2420 sourcemeter under dark and 100 mW cm⁻² illumination conditions. The responsivity characteristics of the MSM photodetectors were collected using a Keithley 2420 sourcemeter to record the photocurrent at a bias voltage of 20 V under the 140 W Xe lamp for illumination. The incident optical power was measured by laser power and energy meters (NOVAII 7Z01550, Ophir Optronics, Jerusalem, Israel). All the characterizations were measured at room temperature.

3. Results and Discussion

Figure 3 shows the FESEM surface morphologies of the $MA_3Sb_2Br_9$ single crystals with different growth temperatures. It can be seen that, when the growth temperature of the perovskite crystal was 60 °C, a dense and crack-free morphology with a large grain size could be obtained. On the other hand, when the growth temperatures of the perovskite crystals were 80 °C and 100 °C, relatively irregular grain size and pinholes were observed on the perovskite crystal. This is because the pinholes or cracks act as nonradiative recombination centers, which could cause poor device performance. Therefore, the optimized growth temperature helps to create high-crystallinity crystals without pinholes and cracks, which can promote the separation and transport process of photo-excited carriers.



Figure 3. Field-emission scanning electron microscope (FESEM) micrographs of MA₃Sb₂Br₉ single crystals with different growth temperatures.

To further understand the situation of a MA₃Sb₂Br₉ single crystal, an X-ray diffraction analysis was carried out; the result is shown in Figure 4. In the range of 10 to 50°, the MA₃Sb₂Br₉ single crystal synthesized under the three growth temperature conditions has a good overlap of diffraction peaks, which proves that their composition was consistent. Moreover, four main peaks appeared at the diffracted peak positions $2\theta = 17.73^{\circ}$, 26.76°, 45.32°, and 55.24°, corresponding to the crystal planes (002), (003), (303), and (042), respectively. Comparing with the crystal data in the literature [38,40], it was found that these were the main characteristic peaks of MA₃Sb₂Br₉ single crystal, which confirmed the formation of trigonal P3-m1 symmetry MA₃Sb₂Br₉ single crystal, and the corresponding lattice parameters a = 7.6918 Å and c = 7.5842 Å. From the three sets of diffraction peak intensities, it could be seen that the main peak of MA₃Sb₂Br₉ single crystal grown at 60 °C was sharp and the full width at half maximums (FWHMs) (0.1053°) was very small, which proves that the crystallinity was very high. Compared with the other two growth temperatures, the crystal quality of the single crystal grown at 100 °C was the worst, and the degree of crystallization was not high. The quality of the single crystal grown at 80 °C was slightly better than that of the one grown at 100 °C.



Figure 4. X-ray diffraction (XRD) patterns of MA₃Sb₂Br₉ single crystals with different growth temperatures.

Figure 5a shows the transmittance spectra of MA₃Sb₂Br₉ single crystals at different growth temperatures. MA₃Sb₂Br₉ single crystals grown at different temperatures of 60 °C, 80 °C, and 100 °C show transmittances of 74.3%, 82.1%, and 85.7% at wavelengths higher than 500 nm, respectively. In addition, the sharp drop in transmittance observed below 500 nm was due to the strong absorption of the MA₃Sb₂Br₉ single crystal. Such a strong absorption undoubtedly promotes the generation of effective exciton within the material. Figure 5b shows the absorption spectra of MA₃Sb₂Br₉ single crystals obtained at different growth temperatures. It can be seen that the MA₃Sb₂Br₉ single crystal synthesized at different growth temperatures of 60 °C, 80 °C, and 100 °C exhibited different absorption levels in the visible light range. Among them, the absorbance of the MA₃Sb₂Br₉ single crystal was the highest at the growth temperature of 60 °C, followed by 80 °C, and lowest at 100 °C. This shows that the MA₃Sb₂Br₉ single crystal synthesized at a growth temperature of 60 °C not only has high crystal quality, but also has the best light absorption. The MA₃Sb₂Br₉ single crystal under 80 °C has better light absorption than that of 100 °C. The band gap of these crystals varied from 2.38 to 2.48 eV and was calculated by a Tauc plot. The UV-Vis absorption of the MA₃Sb₂Br₉ single crystal obtained under the conditions of the three combinations was consistent with the results of the previous SEM and XRD characterization.



Figure 5. (a) Transmittance and (b) absorbance spectra of MA₃Sb₂Br₉ single crystals with different growth temperatures.

After discussing the light absorption properties of MA₃Sb₂Br₉ single crystal with different growth temperatures, its photoluminescence (PL) properties were studied, and the PL spectra were obtained, as shown in Figure 6. It can be seen that the three kinds of MA₃Sb₂Br₉ single crystals with different growth temperatures had a very narrow PL emission peak at the wavelength of 523–527 nm, and FWHMs of 39.9, 42.4, and 47.9 nm. The PL intensities of single crystals at the three growth temperatures were different. The MA₃Sb₂Br₉ single crystal grown at 60 °C had the highest PL intensity, followed by 80 °C, and 100 $^{\circ}$ C was the lowest. This was consistent with the conclusion drawn from the previously analyzed absorbance spectra. For a perovskite material, the narrow PL emission peak may also be caused by the characteristics of the sharp absorption edge, according to Figure 5, except for the crystal quality [41]. It was proven that the MA₃Sb₂Br₉ single crystal with good light absorption performance also had the best PL performance. In addition, the MA₃Sb₂Br₉ single crystal grown at 60 °C corresponded to the narrow PL peak that appeared on both sides (highly symmetrical), indicating that the single crystal MA₃Sb₂Br₉ had a low density of defect states at this growth temperature. This was consistent with the previously proved MA₃Sb₂Br₉ single crystal quality and the best crystallization results.



Figure 6. Photoluminescence (PL) spectra of MA₃Sb₂Br₉ single crystals with different growth temperatures.

To determine the charge transport and trap-state density of single crystals, the dark current–voltage (*I–V*) curve of the MA₃Sb₂Br₉ single crystal photodetector device was measured by a space-charge-limited current (SCLC) analysis. The defects of perovskite will trap carriers, thereby reducing the concentration of free carriers. Under a lower bias, the current and voltage are linearly related. The injected carriers are continuously captured by defects when the voltage gradually increases. When the defect is filled, the current will increase nonlinearly—that is, a kink-point appears on the *I–V* curve. The voltage corresponding to this point is the trap-filled limit voltage (*V_{TFL}*), which can be obtained from the intersection of the tangent of the linear growth part and the tangent of the nonlinear growth part. Figure 7 presents the dark *I–V* curve at different growth temperatures. Under a lower bias voltage, it was a linear Ohmic region (green line); the high bias region was the trap-filling limit region (purple line). The trap-state density could be calculated by Equation (1), as follows [14]:

$$n_{\rm trap} = \frac{2\varepsilon_0 \varepsilon_r V_{TFL}}{qL^2},\tag{1}$$

where *q* is the elementary charge; n_{trap} is the trap-state density; *L* is the single crystal thickness; and ε_r and ε_0 represent the elative dielectric constant and vacuum permittivity, respectively. After fitting, the corresponding V_{TFLs} of the MA₃Sb₂Br₉ single crystals at growth temperatures of 60 °C, 80 °C, and 100 °C were 9.5, 12.3, and 19.1 V, respectively. The trap-state densities can be calculated as 2.63×10^{11} , 4.44×10^{11} , and 9.40×10^{11} cm⁻³, respectively, indicating that the lower trap-state density of the perovskite MA₃Sb₂Br₉ crystal grown at 60 °C led to better crystalline quality and higher carrier mobility, which are vital to improve the charge transport of an optoelectronic device. In addition, the charge carrier mobility of MA₃Sb₂Br₉ was also estimated according to the Mott–Gurney law [14]:

$$J = \frac{9}{8}\varepsilon_0\varepsilon_r \mu \frac{V^2}{L^3},\tag{2}$$

where J, ε_0 , ε_r , μ , V, and L are the dark current, the relative dielectric constant, the vacuum permittivity, the charge carrier mobility of single crystals, the applied voltage, and the single crystal thickness, respectively. As a result, the calculated charge carrier mobility of the MA₃Sb₂Br₉ single crystals at growth temperatures of 60 °C, 80 °C, and 100 °C was 0.75, 0.18, and 0.09 cm² V⁻¹ s⁻¹, respectively. In particular, the MA₃Sb₂Br₉ single crystal grown at 60 °C was larger than the previously reported value (0.4 cm² V⁻¹ s⁻¹) [37], which indicates a predictable improvement in the photocurrent of the photodetector.

Figure 8a depicts the *I–V* curve of MA₃Sb₂Br₉ single-crystal photodetectors with different growth temperatures under dark and 1 sun illumination. The photocurrents of the MA₃Sb₂Br₉ single crystals at growth temperatures of 60 °C, 80 °C, and 100 °C at 0 V bias were 1.18×10^{-9} , 1.15×10^{-9} , and 1.04×10^{-9} A, respectively. Compared with those at 80 °C (8.95×10^{-9} and 7.15×10^{-8} A) and 100 °C (1.15×10^{-8} and 5.85×10^{-8} A) of the MA₃Sb₂Br₉ single-crystal photodetector, the dark current and photocurrent of the MA₃Sb₂Br₉ single-crystal photodetector grown at 60 °C were 8.09×10^{-9} and 1.04×10^{-7} A (@50 V), respectively. The relatively low dark current could be attributed to the high-quality MA₃Sb₂Br₉ single crystals at growth temperatures of 60 °C, 80 °C, and 100 °C were 10.2, 7.6, and 6.3, respectively. Responsivity (R), external quantum efficiency (EQE), and detectivity (D*) were important performance indicators of photodetectors. The wavelength-dependent responsivity, EQE, and detectivity of MA₃Sb₂Br₉ single-crystal photodetectors with different growth temperatures are presented in Figure 8b–d, measured at a bias voltage of 20 V. R, EQE, and D* were defined as follows [14,37]:

$$R = \frac{I_{ph} - I_{dark}}{P_{in}},\tag{3}$$

$$EQE = R \frac{hc}{e\lambda},$$
(4)

$$D* = \frac{R}{\sqrt{2eI_{dark}/A}},\tag{5}$$

where I_{ph} is the photocurrent, I_{dark} is the dark current, P_{in} is the illumination power, e is the electronic charge constant, h is the Planck's constant, c is the light velocity, λ is the wavelength of the incident light source, and A represents the effective illumination area of the photodetector. We observed that the MA₃Sb₂Br₉ single-crystal photodetector grown at 60 °C attained a maximum responsivity of 0.113 A W⁻¹, EQE of 32.7%, and detectivity of 4.32×10^{11} Jones under monochromatic light with a wavelength of 430 nm, while 0.078 and 0.065 A W⁻¹ responsivity, and 22.6% and 18.8% EQEs, as well as 2.85×10^{11} and 2.11×10^{11} Jones detectivities, were calculated for MA₃Sb₂Br₉ single-crystal photodetectors grown at 80 °C and 100 °C, respectively. The increase in responsivity might be attributed to the lower trap-state density and the high optical absorption of the MA₃Sb₂Br₉ single crystal grown at 60 °C, with the improved carrier injection helping to enhance the photocurrent of the photodetector. On the other hand, reducing the trap-state density could result in a slow recombination rate of the photodetector, which is beneficial in terms of the suppression of the dark current and enhanced detection capability. To compare Cs₃Sb₂Br₉based photodetector with a very narrow band width (480 ± 4 nm only). This work exhibits a wide responsivity and detectivity range, from 400 to 500 nm [42].



Figure 7. Dark current–voltage (*I–V*) curves of MA₃Sb₂Br₉ single-crystal photodetector with different growth temperatures.

Figure 9 shows the time-dependent photocurrent curve of the $MA_3Sb_2Br_9$ single-crystal photodetector with different growth temperatures under a light intensity of 100 mW cm⁻² and an applied bias of 20 V in order to measure the response time. The rise and fall times of the detector were defined as the time required for the photocurrent to increase from 10% to 90% and decrease from 90% to 10% of the steady-state saturation current, respectively. The rise and fall times of the MA₃Sb₂Br₉ single-crystal photodetector grown at 60 °C were 47.1 and 1162 ms, which was faster than the rise times (58.1 and 59.5 ms) and fall times (3738 and 4230 ms) of the MA₃Sb₂Br₉ single-crystal photodetectors grown at 80 °C and 100 °C, respectively, indicating that single crystal grown at the lower temperature of 60 °C improves the separation efficiency of photoinduced charge carriers.



Figure 8. (a) Current–voltage (*I–V*) curves of MA₃Sb₂Br₉ single-crystal photodetectors with different growth temperatures under dark and 1 sun illumination. (**b–d**) Wavelength-dependent responsivity, external quantum efficiency, and detectivity of MA₃Sb₂Br₉ single-crystal photodetectors with different growth temperatures under 20 V.

Figure 9. Photocurrent response of the MA₃Sb₂Br₉ single-crystal photodetector with different growth temperatures of MA₃Sb₂Br₉ -60° C (**a**), MA₃Sb₂Br₉ -80° C (**b**), and MA₃Sb₂Br₉ -100° C (**c**) under light illumination of 100 mW cm⁻² (bias @20 V).

We characterized the linear dynamic range (LDR) of the prepared MA₃Sb₂Br₉ singlecrystal photodetector. A plot of the photocurrent versus the illumination power density at 430 nm (bias @20 V) for the MA₃Sb₂Br₉ single-crystal photodetectors is displayed in Figure 10a–c. The minimum detectable light intensities were 4×10^{-5} , 8×10^{-5} , and 2×10^{-4} W cm⁻² for the MA₃Sb₂Br₉ single crystals at growth temperatures of 60 °C, 80 °C, and 100 °C, respectively. Here, LDR was obtained by LDR = 20LOG(I_{upper}/I_{lower}), where I_{upper} and I_{lower} represent the upper and lower detectable currents, respectively. Therefore, the LDRs of the MA₃Sb₂Br₉ single-crystal photodetectors were 67.96, 61.94, and 53.98 dB, respectively. A comparison of the lead-free Sb-based perovskite photodetectors considered in other works is given in Table 1.

Figure 10. Illumination power density-dependent photocurrent of the MA₃Sb₂Br₉ single-crystal photodetector with different growth temperatures of MA₃Sb₂Br₉-60 °C (**a**), MA₃Sb₂Br₉-80 °C (**b**), and MA₃Sb₂Br₉-100 °C (**c**) at 20 V bias (430 nm).

Device Structure	Preparation Method	Rise /Decay Times (ms)	Responsivity (A/W)	EQE (%)	Detectivity (Jones)	Ref.
ITO/MA ₃ Sb ₂ I ₉ single crystals/ITO	Slowly cooling process	0.4/0.9	40	-	10 ¹²	[39]
ITO/MA ₃ Sb ₂ Br9 single crystals/ITO	Slowly cooling process	1000	0.03	-	$5 imes 10^8$	[39]
Au/Ti/Si/SiO ₂ /Cs ₃ Sb ₂ Br ₉ nanoflakes/Ti/Au	ITC process	24/48	3.8	-	$2.6 imes 10^{12}$	[40]
Au/Si/SiO ₂ /Cs ₃ Sb ₂ Br ₉ single crystals/Au	Solvothermal process	0.2/3.0	2.29	18	$3.77 imes 10^{12}$	[42]
Ag/C ₆₀ /MA ₃ Sb ₂ Br ₉ single crystals/C ₆₀ /Ag	ITC process (oven)	47.1/1162	0.113	32.7	4.32×10^{11}	This work

Table 1. Comparison of lead-free Sb-based perovskite photodetectors.

4. Conclusions

We have successfully fabricated lead-free perovskite-like MA₃Sb₂Br₉ single crystals using an ITC method. Compared with the other crystal growth temperatures, the obtained MA₃Sb₂Br₉ single crystal at 60 °C presented a dense and crack-free morphology with large grain size, excellent crystallinity, better charge carrier mobility (0.75 cm² V⁻¹ s⁻¹), and lower trap state density (2.63×10^{11} cm⁻³). Moreover, the MA₃Sb₂Br₉ single-crystal photodetector showed a lower dark current (8.09×10^{-9} A), high responsivity (0.113 A W⁻¹), and high detectivity (4.32×10^{11} Jones). In addition, the perovskite single crystal had extremely low defect density and few interface defects, which was conducive to the design of more stable optoelectronic devices. Lead-free Sb-based perovskite materials have a similar band gap as Pb-based perovskite materials (~2.2 eV). The operational temperature and water sensitivity should be investigated in the future.

Author Contributions: L.-C.C. carried out the experiments, designed the study, and gave suggestions on the whole manuscript. C.-H.T. and C.-M.H. conceived the idea, performed the data analysis and interpretation, and wrote the manuscript. K.-L.L. and H.-Y.L. prepared the samples and performed all measurements. All authors have read and agreed to the published version of the manuscript.

Funding: This research was funded by MOST Nos. 109-2218-E-027-003-MY2 and 109-2622-E-027-031.

Institutional Review Board Statement: Not applicable.

Informed Consent Statement: Not applicable.

Data Availability Statement: Not applicable.

Acknowledgments: This research was supported by the Ministry of Science and Technology (Taiwan) under Contract Nos. 109-2218-E-027-003-MY2 and 109-2622-E-027-031.

Conflicts of Interest: The authors declare no conflict of interest.

References

- Lee, J.W.; Seol, D.J.; Cho, A.N.; Park, N.G. High-Efficiency Perovskite Solar Cells Based on the Black Polymorph of HC(NH2)2PbI3. Adv. Mater. 2014, 26, 4991–4998. [CrossRef] [PubMed]
- 2. Saliba, M.; Correa-Baena, J.P.; Graetzel, M.; Hagfeldt, A.; Abate, A. Perovskite Solar Cells: From the Atomic Level to Film Quality and Device Performance. *Angew. Chem. Int. Ed.* **2018**, *57*, 2554–2569. [CrossRef] [PubMed]
- Stranks, S.D.; Eperon, G.E.; Grancini, G.; Menelaou, C.; Alcocer, M.J.; Leijtens, T.; Herz, L.M.; Petrozza, A.; Snaith, H.J. Electron-Hole Diffusion Lengths Exceeding 1 Micrometer in an Organometal Trihalide Perovskite Absorber. *Science* 2013, 342, 341–344. [CrossRef]
- Li, F.; Zhang, C.; Huang, J.; Fan, H.; Wang, H.; Wang, P.; Zhan, C.; Liu, C.M.; Li, X.; Yang, L.; et al. A Cation-Exchange Approach for the Fabrication of Efficient Methylammonium Tin Iodide Perovskite Solar Cells. *Angew. Chem. Int. Ed.* 2019, *58*, 6688–6692. [CrossRef]
- 5. Yao, F.; Peng, J.; Li, R.; Li, W.; Gui, P.; Li, B.; Liu, C.; Tao, C.; Lin, Q.; Fang, G. Room-temperature liquid diffused separation induced crystallization for high-quality perovskite single crystals. *Nat. Commun.* **2020**, *11*, 1–9. [CrossRef]
- Kojima, A.; Teshima, K.; Shirai, Y.; Miyasaka, T. Organometal Halide Perovskites as Visible-Light Sensitizers for Photovoltaic Cells. J. Am. Chem. Soc. 2009, 131, 6050–6051. [CrossRef]
- Yao, J.; Qiu, B.; Zhang, Z.G.; Xue, L.; Wang, R.; Zhang, C.; Chen, S.; Zhou, Q.; Sun, C.; Yang, C.; et al. Cathode engineering with perylene-diimide interlayer enabling over 17% efficiency single-junction organic solar cells. *Nat. Commun.* 2020, *11*, 1–10. [CrossRef]
- 8. Chen, L.C.; Tien, C.-H.; Lee, K.L.; Kao, Y.T. Efficiency Improvement of MAPbI3 Perovskite Solar Cells Based on a CsPbBr3 Quantum Dot/Au Nanoparticle Composite Plasmonic Light-Harvesting Layer. *Energies* **2020**, *13*, 1471. [CrossRef]
- 9. National Renewable Energy Laboratory. Best Research-Cell Efficiencies. Available online: https://www.nrel.gov/pv/assets/pdfs/best-research-cell-efficiencies.20200104.pdf (accessed on 4 January 2021).
- 10. Ahmadi, M.; Wu, T.; Hu, B. A Review on Organic–Inorganic Halide Perovskite Photodetectors: Device Engineering and Fundamental Physics. *Adv. Mater.* **2017**, *29*, 1605242. [CrossRef]
- 11. Chen, L.C.; Lee, K.L.; Lee, K.Y.; Huang, Y.W.; Lin, R.M. Study of Metal–Semiconductor–Metal CH3NH3PbBr3 Perovskite Photodetectors Prepared by Inverse Temperature Crystallization Method. *Sensors* 2020, 20, 297. [CrossRef]
- 12. Sun, M.; Zhang, F.; Luo, X.; Zhang, D.; Duan, L. Mixed 2D/3D perovskite with fine phase control modulated by a novel cyclopentanamine hydrobromide for better stability in light-emitting diodes. *Chem. Eng. J.* **2020**, 393, 124787. [CrossRef]
- Kang, D.H.; Kim, S.-G.; Kim, Y.C.; Han, I.T.; Jang, H.J.; Lee, J.Y.; Park, N.-G. CsPbBr3/CH3NH3PbCl3 Double Layer Enhances Efficiency and Lifetime of Perovskite Light-Emitting Diodes. ACS Energy Lett. 2020, 5, 2191–2199. [CrossRef]
- Maculan, G.; Sheikh, A.D.; Abdelhady, A.L.; Saidaminov, M.; Haque, A.; Murali, B.; Alarousu, E.; Mohammed, O.F.; Wu, T.; Bakr, O.M. CH3NH3PbCl3 Single Crystals: Inverse Temperature Crystallization and Visible-Blind UV-Photodetector. *J. Phys. Chem. Lett.* 2015, *6*, 3781–3786. [CrossRef]
- 15. Ju, D.; Jiang, X.; Xiao, H.; Chen, X.; Hu, X.; Tao, X. Narrow band gap and high mobility of lead-free perovskite single crystal Sn-doped MA3Sb2I9. *J. Mater. Chem. A* **2018**, *6*, 20753–20759. [CrossRef]
- 16. Dong, Q.; Fang, Y.; Shao, Y.; Mulligan, P.; Qiu, J.; Cao, L.; Huang, J. Electron-hole diffusion lengths > 175 μm in solution-grown CH3NH3PbI3single crystals. *Science* **2015**, *347*, 967–970. [CrossRef] [PubMed]
- 17. Xing, J.; Zhao, C.; Zou, Y.; Kong, W.; Yu, Z.; Shan, Y.; Dong, Q.; Zhou, D.; Yu, W.; Guo, C. Modulating the optical and electrical properties of MAPbBr3 single crystals via voltage regulation engineering and application in memristors. *Light. Sci. Appl.* **2020**, *9*, 1–11. [CrossRef] [PubMed]
- 18. Song, Y.; Bi, W.; Wang, A.; Liu, X.; Kang, Y.; Dong, Q. Efficient lateral-structure perovskite single crystal solar cells with high operational stability. *Nat. Commun.* **2020**, *11*, 1–8. [CrossRef]
- 19. Babayigit, A.; Ethirajan, A.; Muller, M.; Conings, B. Toxicity of organometal halide perovskite solar cells. *Nat. Mater.* **2016**, *15*, 247–251. [CrossRef]
- 20. Giustino, F.; Snaith, H.J. Toward Lead-Free Perovskite Solar Cells. ACS Energy Lett. 2016, 1, 1233–1240. [CrossRef]
- Chen, L.-C.; Tien, C.-H.; Jhou, Y.-C.; Lin, W.-C. Co-Solvent Controllable Engineering of MA0.5FA0.5Pb0.8Sn0.2l3 Lead–Tin Mixed Perovskites for Inverted Perovskite Solar Cells with Improved Stability. *Energies* 2020, 13, 2438. [CrossRef]
- 22. Jiang, X.; Wang, F.; Wei, Q.; Li, H.; Shang, Y.; Zhou, W.; Wang, C.; Cheng, P.; Chen, Q.; Chen, L.; et al. Ultra-high open-circuit voltage of tin perovskite solar cells via an electron transporting layer design. *Nat. Commun.* 2020, *11*, 1–7. [CrossRef] [PubMed]
- 23. Hima, A.; Lakhdar, N. Enhancement of efficiency and stability of CH3NH3GeI3 solar cells with CuSbS2. *Opt. Mater.* 2020, 99, 109607. [CrossRef]
- 24. Kanoun, A.-A.; Kanoun, M.B.; Merad, A.E.; Goumri-Said, S. Toward development of high-performance perovskite solar cells based on CH3NH3GeI3 using computational approach. *Sol. Energy* **2019**, *182*, 237–244. [CrossRef]

- Li, X.; Zhong, X.; Hu, Y.; Li, B.; Sheng, Y.; Zhang, Y.; Weng, C.; Feng, M.; Han, H.; Wang, J. Organic–Inorganic Copper(II)-Based Material: A Low-Toxic, Highly Stable Light Absorber for Photovoltaic Application. *J. Phys. Chem. Lett.* 2017, *8*, 1804–1809. [CrossRef] [PubMed]
- Zeng, F.; Guo, Y.; Hu, W.; Tan, Y.; Zhang, X.; Feng, J.; Tang, X. Opportunity of the Lead-Free All-Inorganic Cs3Cu2I5 Perovskite Film for Memristor and Neuromorphic Computing Applications. ACS Appl. Mater. Interfaces 2020, 12, 23094–23101. [CrossRef] [PubMed]
- Jain, S.M.; Phuyal, D.; Davies, M.; Li, M.; Philippe, B.; de Castro, C.S.; Qiu, Z.; Kim, J.; Watson, T.; Tsoi, W.C.; et al. An effective approach of vapour assisted morphological tailoring for reducing metal defect sites in lead-free, (CH3NH3)3Bi2I9 bismuth-based perovskite solar cells for improved performance and long-term stability. *Nano Energy* 2018, 49, 614–624. [CrossRef]
- 28. Qi, Z.; Fu, X.; Yang, T.; Li, D.; Fan, P.; Li, H.; Jiang, F.; Li, L.; Luo, Z.; Zhuang, X.; et al. Highly stable lead-free Cs3Bi2I9 perovskite nanoplates for photodetection applications. *Nano Res.* **2019**, *12*, 1894–1899. [CrossRef]
- 29. Correa-Baena, J.-P.; Nienhaus, L.; Kurchin, R.C.; Shin, S.S.; Wieghold, S.; Hartono, N.T.; Layurova, M.; Klein, N.D.; Poindexter, J.R.; Polizzotti, A.; et al. A-Site Cation in Inorganic A3Sb2I9 Perovskite Influences Structural Dimensionality, Exciton Binding Energy, and Solar Cell Performance. *Chem. Mater.* **2018**, *30*, 3734–3742. [CrossRef]
- Singh, A.; Najman, S.; Mohapatra, A.; Lu, Y.-J.; Hanmandlu, C.; Pao, C.-W.; Chen, Y.-F.; Lai, C.-S.; Chu, C.W. Modulating Performance and Stability of Inorganic Lead-Free Perovskite Solar Cells via Lewis-Pair Mediation. ACS Appl. Mater. Interfaces 2020, 12, 32649–32657. [CrossRef] [PubMed]
- 31. Krishnamoorthy, T.; Ding, H.; Yan, C.; Leong, W.L.; Baikie, T.; Zhang, Z.; Sherburne, M.; Li, S.; Asta, M.; Mathews, N.; et al. Lead-free germanium iodide perovskite materials for photovoltaic applications. *J. Mater. Chem. A* **2015**, *3*, 23829–23832. [CrossRef]
- Waleed, A.; Tavakoli, M.M.; Gu, L.; Wang, Z.; Zhang, D.; Manikandan, A.; Zhang, Q.; Zhang, R.; Chueh, Y.-L.; Fan, Z. Lead-Free Perovskite Nanowire Array Photodetectors with Drastically Improved Stability in Nanoengineering Templates. *Nano Lett.* 2016, 17, 523–530. [CrossRef]
- Liu, J.; Ozaki, M.; Yakumaru, S.; Handa, T.; Nishikubo, R.; Kanemitsu, Y.; Saeki, A.; Murata, Y.; Murdey, R.; Wakamiya, A. Lead-Free Solar Cells based on Tin Halide Perovskite Films with High Coverage and Improved Aggregation. *Angew. Chem. Int. Ed.* 2018, *57*, 13221–13225. [CrossRef] [PubMed]
- 34. Liu, C.; Tu, J.; Hu, X.; Huang, Z.; Meng, X.; Yang, J.; Duan, X.; Tan, L.; Li, Z.; Chen, Y. Enhanced Hole Transportation for Inverted Tin-Based Perovskite Solar Cells with High Performance and Stability. *Adv. Funct. Mater.* **2019**, *29*, 1808059. [CrossRef]
- 35. Wolanyk, J.; Xiao, X.; Fralaide, M.; Lauersdorf, N.J.; Kaudal, R.; Dykstra, E.; Huang, J.; Shinar, J.; Shinar, R. Tunable perovskitebased photodetectors in optical sensing. *Sen. Actuators B Chem.* **2020**, *321*, 128462. [CrossRef]
- 36. Park, B.-W.; Philippe, B.; Zhang, X.; Rensmo, H.; Boschloo, G.; Johansson, E.M.J. Bismuth Based Hybrid Perovskites A3Bi2I9(A: Methylammonium or Cesium) for Solar Cell Application. *Adv. Mater.* **2015**, *27*, 6806–6813. [CrossRef] [PubMed]
- 37. Ahmad, K.; Mobin, S.M. Recent Progress and Challenges in A3Sb2X9-Based Perovskite Solar Cells. *ACS Omega* 2020, *5*, 28404–28412. [CrossRef]
- 38. Yang, B.; Li, Y.J.; Tang, Y.X.; Mao, X.; Luo, C.; Wang, M.S.; Deng, W.Q.; Han, K.L. Constructing Sensitive and Fast Lead-Free Single-Crystalline Perovskite Photodetectors. J. Phys. Chem. Lett. 2018, 9, 3087–3092. [CrossRef]
- Zheng, Z.; Hu, Q.; Zhou, H.; Luo, P.; Nie, A.; Zhu, H.; Gan, L.; Zhuge, F.; Ma, Y.; Song, H.; et al. Submillimeter and lead-free Cs3Sb2Br9 perovskite nanoflakes: Inverse temperature crystallization growth and application for ultrasensitive photodetectors. *Nanoscale Horizons* 2019, *4*, 1372–1379. [CrossRef]
- 40. Yang, J.M.; Choi, E.S.; Kim, S.Y.; Kim, J.H.; Park, J.H.; Park, N.G. Perovskite-related (CH3NH3)3Sb2Br9 for forming-free memristor and low-energy-consuming neuromorphic computing. *Nanoscale* **2019**, *11*, 6453–6461. [CrossRef]
- 41. Li, L.; Deng, Y.; Bao, C.; Fang, Y.; Wei, H.; Tang, S.; Zhang, F.; Huang, J. Self-Filtered Narrowband Perovskite Photodetectors with Ultrafast and Tuned Spectral Response. *Adv. Opt. Mater.* **2017**, *5*, 1700672. [CrossRef]
- 42. Liu, P.; Liu, Y.; Zhang, S.; Li, J.; Wang, C.; Zhao, C.; Nie, P.; Dong, Y.; Zhang, X.; Zhao, S.; et al. Lead-Free Cs 3 Sb 2 Br 9 Single Crystals for High Performance Narrowband Photodetector. *Adv. Opt. Mater.* **2020**, *8*, 2001072. [CrossRef]