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SPECIALTY SECTION

This article was submitted to Nanoscience, a section of the journal Frontiers in Chemistry

RECEIVED 11 July 2022 ACCEPTED 16 August 2022 PUBLISHED 29 September 2022

#### CITATION

Huang S, Fu Y and Mo A (2022), Electrophoretic-deposited MXene titanium coatings in regulating bacteria and cell response for peri-implantitis. *Front. Chem.* 10:991481. doi: 10.3389/fchem.2022.991481

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# Electrophoretic-deposited MXene titanium coatings in regulating bacteria and cell response for peri-implantitis

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**Background:** Two-dimensional(2D)MXenes have continued to receive increasing interest from researchers due to their graphene-like properties, in addition to their versatile properties for applications in electronic devices, power generation, sensors, drug delivery, and biomedicine. However, their construction and biological properties as titanium coatings to prevent periimplantitis are still unclear.

**Materials and methods:** In this work, few-layer  $Ti_3C_2T_x$  MXene coatings with different thicknesses at varied depositing voltages (30, 40, and 50 V) were constructed by anodic electrophoretic deposition without adding any electrolytic ions. *In vitro* cytocompatibility assay was performed on preosteoblasts (MC3T3-E1) cell lines after the characterization of the coating. Meanwhile, the antibacterial activity against bacteria which are closely related to peri-implantitis including *Staphylococcus aureus* (*S. aureus*) and its drug-resistant strain *MRSA* was further investigated.

**Results:** MXene-coated titanium models with different thicknesses were successfully assembled by analyzing the results of characterization. The compounding of  $Ti_3C_2T_x$  could significantly improve the initial adhesion and proliferation of MC3T3-E1 cells. Moreover, the coating can effectively inhibit the adhesion and cell activity of *S. aureus* and *MRSA*, and *MRSA* expressed greater restricting behavior than *S. aureus*. The ability to promote antibacterial activity is proportional to the content of  $Ti_3C_2T_x$ . Its antioxidant capacity to reduce ROS in the culture environment and bacterial cells was first revealed.

**Conclusion:** In summary, this work shows a new avenue for MXene-based nano-biomaterials under the clinical problem of multiple antibiotic resistance.

#### KEYWORDS

 $Ti_3C_2T_x$  coating, MRSA, implant-associated infections, antibacterial ability, cell behavior, nano-biomaterials



## 1 Introduction

Peri-implantitis, a biological complication mediated by impaired immunity at the implant/tissue interface, is attributed to bacterial biofilms on the implant surface (Kotsakis and Olmedo, 2021). The clinical manifestation is severe progressive inflammatory destruction of the periimplant bone leading to the failure of surgery (Konstantinidis et al., 2015; Safioti et al., 2017; Kotsakis and Olmedo, 2021). Many clinical studies have shown that the Gram-positive (Gram (+)) facultative coccus Staphylococcus aureus (S. aureus) is the principal pathogen arousing peri-implantitis (Thurnheer and Belibasakis, 2016; Wu et al., 2022). Furthermore, the bacteria that form biofilms on the titanium surface are resistant to antimicrobial agents, particularly to  $\beta$ -lactams (Wu et al., 2022). Methicillin-resistant Staphylococcus aureus (MRSA) exhibits multidrug resistance to several substances belonging to different antibiotic classes and is prevalent in implantrelated infections (Zhou et al., 2022). Undoubtedly, as in other fields of medicine, anti-adherent and bactericidal surfaces to limit the early adhesion and activity of bacteria on titanium implant surfaces are the main research goals (Cacaci et al., 2020). The surface energy, morphology, wettability, wear resistance, and electrochemical charges of a titanium substrate all influence biofilm adhesion and formation (Song et al., 2015). The most universal way to interfere with the early stages of microbial adhesion is to modify the chemical or micro/nanostructure of the out-layer on the titanium substrate via many methods such as

sol-dip (Safaee et al., 2021), electrospinning nanotechnology (Zhang et al., 2014), silanization (Qiu et al., 2018), and electrophoretic deposition (EPD) (Qiu et al., 2017). The current superficial modifications include the coatings of surfactants (Vacheethasanee and Marchant, 2000), inorganic nanomaterials (Qiu et al., 2017), proteins (An et al., 1996; Sang et al., 2021), antimicrobial peptides (Hwang et al., 2021), and hydrophilic negatively charged polysaccharides such as heparin and hyaluronic acid (Morra and Cassineli, 1999; Hildebrandt, 2002). However, most surface modification materials usually lack multifunctional properties. In recent years, research studies on titanium surface modification biomechanical materials with both advantages and antibacterial properties have been growing, and the pros and cons of different modification methods of titanium-based surfaces have also attracted extensive attention.

Inorganic 2D MXenes with hexagonal structures similar to graphene (Bennett-Jackson et al., 2019) are made up of transition metal carbides, nitrides, or carbonitrides with a general formula of  $M_{n+1}X_nT_x$  (n = 1-3) in which M is an early transition metal (e.g. Sc, Ti, Zr, Hf, V, Nb, Ta, Cr, and Mo), X is a carbon or nitrogen, and  $T_x$  represents abundant surface functional groups (-OH, =O, -F, and rarely, -Cl) opposed to the surface of the M layer during the production of MXene by chemically etching the MAX phase (Kang et al., 2019; Bhardwaj et al., 2022). Due to their outstanding properties (i.e., large surface area, great biocompatibility, antibacterial properties, excellent hydrophilicity and electrical conductivity, flexible surface

functionalization, and targeted therapeutic properties), MXenes have been diffusely used in the biomedical field (Huang et al., 2018; Zamhuri et al., 2021). In tissue engineering, MXenes can promote the proliferation, differentiation, adhesion, and migration of a variety of tissue-engineered stem cells, such as human mesenchymal stem cells (Jang and Lee, 2021) and human neural stem cells (Guo et al., 2022) without obvious cytotoxicity; Ti<sub>3</sub>C<sub>2</sub> is the first negatively charged Maxine synthesized by etching the Al layer in Ti<sub>3</sub>AlC<sub>2</sub> (Naguib et al., 2011). Rasool et al. (2016) found that the antibacterial activity of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> against Gram (-) Escherichia coli and Gram (+) Bacillus subtilis was higher than that of the currently widely used 2D antibacterial agent graphene oxide (GO) (Rasool et al., 2017). The order of antibacterial activity against these two strains was monolayer  $Ti_3C_2T_x \gg$ multilayer  $Ti_3C_2T_x > Ti_3AlC_2$ . Meanwhile, Jastrzebska et al. (2017) also reported that Ti<sub>2</sub>CT<sub>x</sub> MXenes lacked bactericidal properties against Botulinum, S. aureus, and Bacillus. Mild apoptosis of Bacillus occurs only when bacterial cells are drawn into the swelling layer and placed between individual flakes of the expanded TiCT<sub>x</sub>. Thus, the differences in thickness and surface chemistry of MXenes can affect their toxicity and antibacterial activity. At present, the antibacterial applications of  $Ti_3C_2T_x$  mainly focus on composite hydrogels (Yin et al., 2020; Li et al., 2022), composite membranes (Rasool et al., 2017; Pandey et al., 2018; Zha et al., 2019; Wu et al., 2021), composite scaffold (Zhou et al., 2021), etc., and there is no research on the antibacterial use of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> alone in titanium-based coatings without adding any auxiliary materials. It has been reported that layered Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> nanosheets with high hydrophilicity can generate edge-cutting and wrapping effects after absorbing the surface of microorganisms (Rasool et al., 2016). Because of the anionic nature of its surface, Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> may also react with some molecules in the cell wall and cytoplasm of microorganisms, destroying the cell structure and leading to the death of microorganisms (Soleymaniha et al., 2019). This reveals the powerful antibacterial application potential of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>.

EPD is an important craft of colloidal depositing, which can deposit various materials from colloidal suspensions (Hu et al., 2020). In view of its high deposition rate, uniform film formation, simple equipment, high material purity, and unrestricted substrate size, EPD stands out among many titanium-coating methods and is diffusely used in the preparation of medical thin films and coatings (Sikkema et al., 2020). The process parameters in EPD (e.g. suspension concentration, applied voltage value, and deposition time) have a considerable influence on the morphology and performance of the coating (Chavez-Valdez et al., 2013). Therefore, coatings with different coverage, thickness, microstructural morphology, density, and adhesion strength can be obtained by changing the process parameters, and the mechanical properties, corrosion resistance, and bonding strength of the titanium substrate can be adjusted within a certain range. In this study, we selected the few-layer MXene (Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>) colloidal dispersion with a negative charge as the raw material,

and the coatings were prepared on the surface of a pure titanium sheet by anodic EPD for the first time under different deposition voltages. In the circumstance of frequent antibiotic resistance, it is significant to investigate how to apply a new nano-  $Ti_3C_2T_x$ coating without adding any electrolytic ions to the titanium inert surface to solve the clinical problem of peri-implantitis. We plan to carry out a series of surface characterizations and assessments, combining the cell activity evaluation of the mouse pre-osteoblast cell line MC3T3-E1 with the *in vitro* antibacterial property investigation of *S. aureus* and *MRSA*. This work aims to prepare an MXene coating model with superior mechanical and biological properties and provide a new coating modification method for bone defect repairing materials represented by titanium.

## 2 Materials and methods

### 2.1 Specimen preparation

One side of the pure titanium plates with the dimensions  $10 \times$  $10 \times 1 \text{ mm}^3$  was polished step-by-step with series abrasive papers (600-1500 mesh) to a mirror flat and then placed in acetone, absolute ethanol, and ultrapure water for ultrasonic cleaning for 20 min, which were denoted as Ti. The preparation process of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> few-layer dispersion is summarized as follows. In the state of magnetic stirring, 1 g of LiF was added to 20 ml of 9M HCl, and then 1 g of Ti<sub>3</sub>AlC<sub>2</sub> powders (purchased from Jilin 11 technology Co., Ltd.) were slowly poured into the mixture within 20 min. After stirring continuously for 24 h (35°C, 200 r/ min), the resulting etching solution was washed with ultrapure water through centrifugation (3,500 rpm, 5 min for each circle) and mixed by hand alternately for about 8-9 times, until the pH of the supernatant was above 6, collecting the supernatant as the Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> few-layer dispersion and stored at 4°C for use. Thinking over the easy-oxidized characteristics of MXene at high temperature and high voltage, the fresh  $\mathrm{Ti}_3\mathrm{C}_2\mathrm{T}_x$  fewlayer dispersion (0.15 mg/ml) was sonicated at low speed for 30 min in an ice-water bath beforehand and negatively charged  $Ti_3C_2T_x$  was uniformly deposited on the surface of the titanium sheet by anodic electrophoresis at 30, 40, and 50 V for 2 min, which were characterized as M30, M40, and M50, respectively. The samples were cut with a diamond saw blade under ice-water cooling to expose the cross-sectional topography for observation.

### 2.2 Surface characterization

A field-emission scanning electron microscope (FE-SEM, Nova Nano SEM 450, United States) with an accelerating voltage of 20 KV was used to observe Ti, M30, M40, and M50. X-ray diffraction (XRD, Rigaku Smartlab, Bruker) with a wide-angle diffraction-Cu target (5–85°) radiation source was applied to investigate the phase composition of the surface of each sample. Raman spectra of various sample surfaces were obtained using a Raman microscope system (Lab RAM, HR Evolution, France) with an Ar-ion laser (514 nm) for excitation. The chemical composition of various samples was determined by X-ray photoelectron spectroscopy (XPS, Thermo Fischer, ESCALAB 250Xi, United States) using an Al ka excitation light source (hv = 1486.6 eV). The static water contact angle of each sample surface was measured using a contact angle meter (CAM, Kruss DSA100, Germany). Nanoindenter (MCT + UNIT + MST) was used to test the wear resistance of each coating sample surface below 20 GPa. Among them, 15 N is the maximum load of the nano-scratch experiment at a loading speed of 29.98 N/min, the scratch speed is 6 mm/min, and the scratch length is 3 mm.

### 2.3 In vitro cytocompatibility evaluation

### 2.3.1 Cell culture

A mouse pre-osteoblast cell line (MC3T3-E1; ATCC CRL-2593, United States) was cultured at 37°C in a 5% CO2 incubator with the alpha-minimum essential medium (a-MEM, GIBCO, United States) supplemented with 10% fetal bovine serum (FBS, GIBCO, United States) and 1% penicillin-streptomycin (PS, Hy Clone, United States). The culture medium was changed every 2 days, and the MC3T3-E1 cells could be used for further studies when the confluency of cells reached around 80%. The cells were detached by incubating in a trypsin/EDTA (0.25% trypsin, 0.02% EDTA) (Gibco, Invitrogen) solution for 1 min at 37°C, then centrifuged for 5 min at 1,000 r/min and resuspended in the aminimum essential medium for seeding on sample surfaces. A cell density of  $\sim 2 \times 10^4$  cells/well was used for all studies. Before reseeding, the specimens were sterilized in 75% ethanol for 2 h, dried by UV irradiation for 1 h on the super clean bench, and then incubated in the culture medium overnight.

#### 2.3.2 Cell cytotoxicity

The live-dead cell viability/cytotoxicity detection kit (TIANDZ, China) was used to test the cytotoxicity of the samples. After seeding the MC3T3-E1 cells on each sample for 24 h, all cells were stained by Calcein-AM (2  $\mu$ M) and PI solutions (8 $\mu$ M) for 30 min at room temperature in the dark. Photographs of cells under a fluorescence microscope (Olympus, Japan) were taken within 1 h. According to the manufacturer's instructions, live cells were stained green, while dead cells fluoresced red.

### 2.3.3 Cell morphology

The MC3T3-E1 cells were seeded on specimen surfaces for 24 h and washed with PBS, then fixed with 4%

paraformaldehyde solution for 10 min. After permeation with 0.5% Triton X-100 for 5 min and rinsed thrice with PBS, staining was conducted with the Fluorescein isothiocyanate-phalloidin (FITC-phalloidin, Solarbio Biotech Co., Ltd., China) with 1% albumin from bovine serum (BSA, Sigma, United States) for 30 min at room temperature in the dark. Finally, DAPI was added for a further 5 min. The stained actin cytoskeletons (green) and cell nucleus (blue) were visualized with a fluorescence microscope (Olympus, Japan).

#### 2.3.4 Cell adhesion

After seeding the MC3T3-E1 cells on each sample for 4 and 24 h, the cells on various sample surfaces were fixed in 2.5% glutaraldehyde for 4 h at 4°C and then rinsed thrice with PBS. The samples were dehydrated by using ethanol of graded v/v concentration (30, 50, 75, 85, 95, and 100%, 15 min each), and dried overnight under natural ventilation. Lastly, the samples were sputter-coated with gold and observed by SEM.

#### 2.3.5 Cell proliferation

The MC3T3-E1 cells were cultured on specimen surfaces for 1, 3, and 5 days, and proliferation was evaluated by using the cell counting kit-8 (CCK-8) assay (APE×BIO, United States). A highly water-soluble tetrazolium salt, WST-8, is reduced in the presence of an electron-coupling reagent to yield the water-soluble formazan form of the dye. The amount of formazan produced by dehydrogenase has a direct linear relationship with the number of viable cells. At each time point, 350 uL of working solution (WST-8: normal medium = 1:10 (v/v)) was introduced to each sample and three blank wells were added. After incubating at 37°C for 1 h and blowing out the staining solution, 100 ul of staining solution per well was transferred to a 96-well plate, and then the absorbances at 450 nm were examined.

### 2.4 In vitro antibacterial tests

#### 2.4.1 Culture of S. Aureus and MRSA

S. aureus (ATCC 29213) and MRSA (ATCC 43330) were purified by scribing on Luria-Bertani (LB) agar plates and incubated at 37°C overnight. A single colony was selected in a 5 ml LB liquid culture medium and incubated overnight at 37°C to obtain a bacterial suspension. The bacterial suspension was then gradually diluted by a 10% LB liquid culture medium to  $10^7$  CFU/ml.

### 2.4.2 Bacterial integrity test

The samples were first sterilized with 75% (v/v) ethanol for 2 h and dried by UV irradiation for 1 h on the super clean bench. Subsequently, bacterial suspensions (50 uL,  $10^7$  CFU/ml) were



inoculated on different sample surfaces and incubated for 24 h at  $37^\circ \text{C}.$ 

Bacteria adhering to the sample surfaces were stained with a LIVE/DEAD Bacterial Staining Kit (BBcellProbe<sup>®</sup> N01/PI, BestBio) for 15 min. Then the samples were washed with 0.9% saline solution. The distribution of live bacteria (green) and dead bacteria (red) was observed by fluorescence microscopy.

The cell morphology of these two bacteria under SEM was used to compare the antibacterial activities of Ti, M30, M40, and M50. For SEM observation, the bacteria on the sample surfaces were fixed with 2.5% glutaraldehyde solution at 4°C overnight, dehydrated with gradient ethanol solutions (30, 50, 75, 85, 95, and 100% (v/v), 15 min each), and dried with hexamethyl disilazane ethanol solution series.

#### 2.4.3 Agar culture observation

Bacterial counting is to further assess the antibacterial activities of various samples. After culturing for 24 h at  $37^{\circ}$ C, the samples with bacterial suspensions were transferred to test tubes containing 4 ml of 0.9% NaCl solution and shaken well to separate the bacteria from the surface of the samples. Then, the bacterial suspensions were serially diluted 10-fold with 0.9% NaCl solution. Finally, 100 uL of the diluted bacterial suspensions were added to a standard LB agar culture medium and the culture was continued for 18 h at  $37^{\circ}$ C. Finally, photographs of agar culture plates were taken.

### 2.4.4 Evaluation of bacterial viability

The AlamarBlue assay kit was used to assess bacterial viability on different samples. After culturing for 24 h at 37°C, 500 uL of 10% AlamarBlue was introduced to each sample and cultured for another 2 h at 37°C in the dark. Finally, 100 uL of

the medium was transferred to a 96-well black plate and the corresponding fluorescence intensity (FI) was measured with an extinction wavelength of 560 nm and an emission wavelength of 590 nm. The antibacterial rate was calculated as follows:

Antibacterial ratio (%) = 
$$\frac{\text{FIcontrol} - \text{FItest}}{\text{FIcontrol}} \times 100$$
,

where  $FI_{test}$  represents the fluorescence intensity of M30, M40, and M50;  $FI_{control}$  was the fluorescence intensity of Ti.

### 2.4.5 Intracellular reactive oxide assay

Intracellular reactive oxygen species (ROS) levels in bacteria were performed with an ROS detection kit. 2',7'-Dichlorodihydrofluorescein diacetate (DCFH-DA) can be converted to non-fluorescent dichlorofluorescein (DCFH) by deacetylation with intracellular esterases. DFCH can be oxidized by ROS in cells to produce 2',7'-dichlorofluorescein (DCF). Therefore, the fluorescence intensity of DCF represents the ROS level to some extent. Bacteria were cultured on different samples for 24 h at 37°C, and then 500 uL DCFH-DA (10 mM) was introduced into each sample well. After culturing for 20 min at 37°C in the dark, 100 µL of the medium was transferred to a 96-well blank plate, and the fluorescence intensity corresponding to DCF was detected by using a microplate reader with an extinction wavelength of 488 nm and an emission wavelength of 535 nm. The ROS levels were expressed by calculating the ratio of (F test-F blank)/(F control-F blank), in which F test was the fluorescence intensity of the coated samples; F control was the fluorescence intensity of pure Ti; F blank was the fluorescence intensity of the 24-well plate without bacteria and samples. At last, the fold increase was obtained by normalizing the control group of pure Ti.





# 2.5 Statistical analysis

The data were expressed as the mean  $\pm$  standard deviation. One-way analysis of variance (ANOVA) and Tukey's multiple

comparison tests were used to analyze the statistical significance of the difference. Statistical analysis was assessed using GraphPad Prism 9.0 software. A value of p < 0.05 was considered statistically significant.



## 3 Results and discussion

### 3.1 Surface characterization

Figures 1A,B; 2 show the surface and section morphologies of the  $Ti_3C_2T_x$  MXene film, Ti, M30, M40, and M50. From the surface of the samples, the surface of the MXene film is a wrinkled nanostructure with uniform morphology and no obvious voids and the surface of pure Ti is relatively flat after treatment, while M30, M40, and M50 display uneven surfaces such as waves, owing to the  $Ti_3C_2T_x$  deposited on the titanium surface uniformly. As for the cross-section, the cross-section of the MXene film showed a layered and tightly stacked structure similar to an "accordion" after etching from  $Ti_3AlC_2$  powders, and a 2D nanosheet structure with few layers of  $Ti_3C_2T_x$  and uniform size is obtained. The cross-section morphologies of the coated samples exhibit a distinct and tightly fitted layered MXene structure as shown in Figure 2. With the increase in deposition voltages, the film thickness becomes thicker, which means an increase in the number of MXene layers.

The samples were further characterized by XRD and Raman spectroscopy. Regarding the phase composition of various samples, it can be seen from the XRD patterns (Figure 3A) compared with  $Ti_3AlC_2$  that the position of the (002) main peak of  $Ti_3C_2T_x$  is shifted to the left by a certain angle, and the peak intensity becomes shallower and wider, indicating the decreasing of crystallization degree and the increasing of the interlayer space after etching (Allen-Perry et al., 2021). All coated samples exhibited typical  $\alpha$ -Ti features (Jin et al., 2014), but the intensity of the main Ti phase peak was lower than that of



pure titanium. Moreover, the representative (002) peak of  $Ti_3C_2T_x$  at  $2\theta = 7.5^{\circ}$  remained clear and obvious in coatings. No shift to the low-angle side was observed. The resultant measurements indicated the characteristic loading of  $Ti_3C_2T_x$  on titanium sheets. Figure 3B shows the Raman spectra of various samples. The characteristic peaks of  $Ti_3C_2T_x$  at 210, 382, and 600 cm<sup>-1</sup> assigned as Ti-C vibrations are highly consistent with previously reported data (Xie et al., 2019). These characteristic peaks also emerge on all coated samples, but not on pure titanium.

The element types and chemical states of samples were investigated by the XPS technique and the results are shown in Figure 4A. The F-segment peak at 685.2 eV of  $Ti_3C_2T_x$  clearly appeared on each coated sample, but not on the pure titanium sheet. The titanium (Ti2p), carbon (C1s), and oxygen (O1s) that are corresponding to the binding energy 460.9, 285.1, and 532.4 eV, respectively, coexisted on the surface of coatings

(Awasthi et al., 2020). By analyzing the elemental content table, it can be seen that with the increase in the deposition voltage, the titanium composition on the surface of the coating samples decreased step-by-step, which may be caused by the increase in the thickness of the  $Ti_3C_2T_x$  coating, which increases the coverage of the bottom pure titanium sheet. Combining the results of the XRD patterns, Raman spectra, and XPS spectra implies that  $Ti_3C_2T_x$  was successfully introduced to the titanium sheet surface by a simple electrophoretic deposition technique.

The static water contact angle was measured to evaluate whether the addition of  $Ti_3C_2T_x$  could improve the hydrophilicity of the material. As shown in Figure 4B, the pure titanium sheet exhibits mild hydrophobic behavior with a water contact angle of 97.02  $\pm$  0.8893, which is slightly larger than in previous studies (Yamauchi et al., 2017; Sharifi et al., 2020). This may be ascribed to the reduction of the roughness and oxygen terminal group content of the titanium sheet after



grinding and polishing. With the increase in the coating thickness from the rising of the deposition voltage, the surface contact angle decreases to  $80.12 \pm 1.379^\circ$ ,  $76.45 \pm 1.767^\circ$ , and  $69.58 \pm 1.616^{\circ}$  in turn after the introduction of  $Ti_3C_2T_x$  as the surface coating. The results can be explained by changes in the physical and chemical composition of the sample surfaces, such as roughness, surface charge, and different contents of oxygen terminal groups (Lee et al., 2015; Liu et al., 2017; Zhou et al., 2017). In general, the equilibrium contact angle for hydrophilic surfaces is below 90°, while the contact angle for hydrophobic surfaces is greater than 90°. Surfaces with moderate hydrophilicity are more prone to bind cells than extremely hydrophobic or hydrophilic surfaces (Agarwalla et al., 2020). Since the hydrophilicity of biomaterials has a considerable impact on cell adhesion, proliferation, and differentiation (Bacakova et al., 2011; Qi et al., 2012), a slight decline in the water contact angle is of great importance.

In the nano-scratch experiment, the elastic deformation stage (I) and plastic deformation stage (II) of the samples' stressinduced failure have been marked in Figure 5. For stage (I), the probe indenter was in contact with the sample surface and moved forward, and then the friction force was generated. Under the action of friction and positive pressure, elastic deformation occurred. At this moment, the friction increased linearly with

the increasing load; in stage (II), the pressure and friction continued to increase. When the combined force of the pressure and friction exceeded the elastic deformation limit of the coating, it reached the plasticity deformation stage. The scratches kept deepening and the friction coefficient (lateral force/normal force) also increased. When the needle tip of the indenter scratched the surface and reached the substrate, the friction curve had an inflection point; this load is the critical load for the failure of the adhesion of the film substrate (Zhang et al., 2020). The critical load is affected by the hardness and elastic modulus of the coating, the substrate, the structure, and thickness of the film, etc. (Rustamov et al., 2019). After the end of stage (II), the sample surface was ruptured. The positive pressure and scratch depth continued to increase. At this time, the coating began to peel off in large pieces. In addition, the scratch width becomes wider. The friction force and plastic deformation suddenly increase, and the friction force curve appeared as a violent fluctuation rather than a relatively smooth rising straight line like stages (I) and (II). As shown in Figure 5, the surface of the pure titanium sheet reaches the critical load when the normal pressure is 4.652 N and the friction force is 2.511 N. The test values of the coated specimens (M30/40/50) are 4.923, 5.451, and 5.458 N (the normal pressure) and 1.781, 2.087, 1.794 N (the friction force), respectively. The results show that the surface



friction force and friction coefficient of the coated samples are lower than those of pure titanium sheets; the thicker the coating, the greater will be the critical load. At the same time, the longer it takes to be worn through by the same load, that is, the better the wear resistance.

### 3.2 In vitro cytocompatibility evaluation

### 3.2.1 Cell cytotoxicity

After being coated with  $Ti_3C_2T_x$ , the biological properties of the samples were investigated preliminarily by using MC3T3-E1. Cells were seeded on Ti, M30, M40, and M50 for 24 h, immediately followed by live/dead double staining to evaluate the cell cytotoxicity. As shown in Figure 6, compared with the pure titanium sheet, the number of living cells on the coated samples increased to a certain extent, which may be related to different proliferation activities. There existed sporadic red fluorescence and mainly green fluorescence on each sample surface. This result indicated that the coatings are non-toxic, which may result from the low concentration and flat surface of the  $Ti_3C_2T_x$  MXene. Excessive oxidizing residue ( $Ti_2O_3$ ) can result in rupturing of the cell membrane (Lim et al., 2021). The synthesis method of  $Ti_3C_2T_x$  in this study is mild without using HF for decreasing the degrees of  $Ti_2O_3$  in the final product.  $Ti_3C_2T_x$  shows a dose- and chemical composition-dependent cytotoxicity (Lim et al., 2021). Scheibe et al. indicated that  $Ti_3C_2T_x$  MXene is more than 80% cytocompatible with non-malignant cells if the concentration is between 10 and 400 µg/ml (Scheibe et al., 2019). The fresh  $Ti_3C_2T_x$  few-layer dispersion (150 ug/mL) deposited on pure Ti is biocompatible.

### 3.2.2 Cell adhesion and morphology

To investigate the initial adhesion and morphology of MC3T3-E1, cells cultured on the sample surface for 4 and 24 h were observed under an SEM (Figure 8A). After culturing for 4 h, MC3T3-E1 cells adhered tightly to the sample surface (pointed by white arrows). The expression of filopodia and lamellipodia was more pronounced on the coated samples than in the relatively round cell morphology on pure titanium sheets. When it comes to 24 h, compared with pure titanium sheets, the cells on the coated samples' surface had a larger spreading area and were more closely connected with each other. We also confirmed this result with fluorescence staining images (Figure 7), where most cells on pure titanium sheets exhibited spindle-shaped morphology with some filopodia.



However, cells on the coated samples exhibited polygonal morphology with sufficient lamellipodia. The expressions of F-actin on the coated samples were slightly higher than that on pure Ti. With the increasing  $Ti_3C_2T_x$  layers, the expressions of F-actin showed a slightly upward trend. The aforementioned phenomenon may indicate that the charge interaction and the enhanced wettability between the surface of  $Ti_3C_2T_x$  and cells are beneficial to the adhesion and growth of cells (Bacakova et al., 2011; Guo et al., 2016). The number of adherent cells increased along with the water contact angle of the sample surface reducing, which was similar to the findings of Arima and Iwata (2007).

### 3.2.3 Cell proliferation

The cell proliferation capacity of MC3T3-E1 on various samples was investigated by comparing the number of viable cells seeded on sample surfaces for 1, 3, and 5 days with the cck-8 assay. As shown in Figure 8B, the absorbance values of all four groups gradually increased with the extension of the culture time, indicating that the cells were in a suitable growth environment.

The cell proliferation rate on the coated samples was significantly higher than that on the pure titanium sheet (p < 0.05). Among the individual coating samples, M50 exhibited the highest cell proliferation, followed by M30 and M40. The trend was especially vivid when the cells were cultured for 1 day (p < 0.001).

### 3.3 In vitro antibacterial tests

### 3.3.1 Bacterial integrity testing

The bacterial viability after coming into contact with the samples was evaluated by fluorescence microscopy observation. The bacteria were stained with an N01/PI fluorescent probe at first. The fluorescent probe N01 could pass through the cell membrane of live bacteria and exhibited bright green fluorescence after excitation. However, the PI could not enter the live bacteria, and thus, could stain the dead bacteria with a damaged cell membrane alone and exhibited red fluorescence after excitation. As shown in Figure 9, more dead bacteria were observed on the coated sheets, whereas a relatively small number

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Live/dead fluorescent staining images of bacteria adhered to the Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>-coated titanium sheets and titanium sheet: (A) S. aureus; (B) MRSA (live bacteria were stained fluorescent green and dead red). The scale bar is 100 µm.

of dead and mostly live bacteria were observed on the titanium sheet. This result verified the antibacterial ability of  $Ti_3C_2T_x$  coatings and the destruction of bacterial membrane integrity. *S. aureus* (Figure 9A) showed small-scale aggregation on the samples' surface, while *MRSA* (Figure 9B) showed a more dispersed distribution.

The antibacterial activity of Ti, M30, M40, and M50 was further assessed with morphological changes of Gram (+) S. aureus and MRSA. After culturing for 24 h at 37°C, the results are shown in Figure 10. At a low magnification of 5 K, many bacteria adhered to the pure Ti surface and aggregated together, while the amount and distribution of bacteria decreased with the increasing thickness of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> coatings. But for S. aureus, there was an increase in the agglomeration stackability between bacteria on the coated samples. The antibacterial properties of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> may be partly attributed to the negative charge and good electrical conductivity on its surface (Rasool et al., 2016), which is associated with the presence of terminal groups such as = O, –OH, and–F, and a TiO2 surface-passivation layer. A titanium sheet is a conductive metal, and Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> with a high negative surface charge is tightly coated on the surface of the titanium sheet, which synergistically enhances the conductivity of the coated sample. Simultaneously, both chemical reactions and physical repulsion may occur owing to the negatively charged bacteria surface (Simanski et al., 2013). As the deposition voltage increases, the surface of the coating material may have a higher cathodic potential, thus showing a stronger anti-adhesion ability (Canty et al., 2019). Under high magnification at 40 K, it can be clearly seen that both bacteria

grow well on pure titanium sheets with typical features of smooth, spherical, and intact cell morphology. As for the cell body morphology of S. aureus, merely small-sized bacterial cell bodies with no obvious agglomeration were shrunken, and there were no observable bacterial cell bodies with a severely deformed cell membrane; while disorganized MRSA with disrupted cell bodies could all be found on the surface of M30, M40, and M50. This may be the result of the reduced edge-cutting effect caused by the state change of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> (Zamhuri et al., 2021). Differences in surface chemistry may affect the toxicity and antibacterial activity of the Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene (Lim et al., 2021). Compared to bacteria adhered to pure Ti with intact cell morphology, bacteria adhered to coatings are captured or encapsulated by nanometerthick Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> and form aggregates. Likewise, the bacterial surface morphology gradually became shrunken and rough. Understanding the interaction between the surface of the Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub>-coating material and the cell envelope is of great significance for evaluating its application as a bacteriostatic agent in implantation clinics. The great electrical conductivity of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> films has been confirmed (Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> monolayer: 4600  $\pm$ 1100 S/cm) (Ling et al., 2014; Lipatov et al., 2019; Tian et al., 2019), which is similar to or even exceeds that of reduced graphene oxide. The main component of the Gram (+) bacterial cell wall is peptidoglycan (PG), which is composed of the anionic sugar polymer wall called teichoic acid (WTA) (Xiong et al., 2022). These WTAs are critical in maintaining bacterial structure, replication, and other major cellular functions (Silhavy et al., 2010; Rohde, 2019). The anionic backbone on WTAs makes them always negatively charged (Brown et al.,



2013). The bacterial cell membrane is characterized by phospholipids, comprising different ionic molecules (Halder et al., 2015). The presence of membrane potential and ionic molecules enables electrostatic interactions on the bacterial surface (Kim et al., 2019). The membrane permeability and membrane potential of bacteria have a remarkable variation when exposed to electric fields (Canty et al., 2019). The possible contact-active antibacterial process of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> coating mainly included three steps. First, the bacterial cells contacted the surface of hydrophilic Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> under the static state (Yang et al., 2019). Subsequently, the electrostatic repulsion between  $Ti_3C_2T_x$ and WTA creates a stress pressure, which is able to rupture the cell wall when a threshold value is reached. The size of this small defect may diverge, leading to the formation of stomata that ultimately mediates the formation of Gram (+) wall defects (Rauch and Leigh, 2015; Alimohammadi et al., 2018). When the outer cell wall is damaged, the strong electrostatic interactions between the negatively charged  $Ti_3C_2T_x$  surface and positively charged phosphatidylcholine lipids lead to the damage of the integrity of the cell membrane (Ou et al., 2016). The mechanism, in this case, could be explained by the formation of conductive bridges over the insulating lipid bilayer, mediating electron transfer within bacterial cells (Vecitis et al., 2010; Li et al., 2014), possibly causing some molecules in the inner side of the cell membrane and cytoplasm to react, leading to apoptosis

and necrosis of bacterial cells (Muraih et al., 2011; Bayer et al., 2013; Kim et al., 2019). After that, the dead bacterial cells could be easily washed off from the membrane surface because of the great hydrophilicity of  $Ti_3C_2T_x$ . Thus, the  $Ti_3C_2T_x$  coatings exhibited prominent anti-biofouling properties.

#### 3.3.2 Agar culture

For bacteria counting, after collecting and diluting the bacterial culture solution on the sample surfaces, 100 ul of the mixed bacterial solution was cultured on the agar plate for another 18 h. As shown in Figure 11, for MRSA (Figure 11B), a large number of bacterial colonies can be seen on the pure titanium sheet at a density of 106 CFU/ml. Compared to Ti, the number of bacterial colonies on the coated samples was significantly reduced, especially for M40 and M50. Moreover, the number of colonies gradually decreased with the increasing coating thickness. After serial dilution, the trend could be more clearly recognized as M50 had the best antibacterial activity, followed by M40. The splitting and growth of MRSA were significantly inhibited in the aforementioned two coating samples; nevertheless, the reduction of S. aureus (Figure 11A) on the CFU of 106 coated samples compared to pure titanium was not as pronounced as MRSA. After successive ten-fold dilution, it can be relatively distinct to find out that the number of colonies of M30 is higher than that of the other two coating samples. The



Photographs of recultivated (A) *S. aureus* and (B) *MRSA* colonies on agar culture plates; (C) counting analysis of reduction percentages of bacterial colonies. \*p < 0.05, \*\*p < 0.01, and \*\*\*p < 0.001.



antibacterial activity of  $Ti_3C_2T_x$  films on Ti may be surface- and layer-dependent rather than edge-dependent. Thus, the antimicrobial effectiveness of the  $Ti_3C_2T_x$  films could increase with the increase in the number of layers. This result is similar to the antibacterial properties of flat GO sheets on a PET substrate invented by Mangadlao et al. (2015). Penicillin-binding proteins (PBPs) are peptidase enzymes located in the cell membrane necessary for cell wall biosynthesis and maintenance (Meroueh et al., 2006; Lade and Kim, 2021), which catalyzes the cross-linking of adjacent stem peptides to synthesize the PG backbone, to warrant mechanical strength and flexibility during all stages of bacterial growth (Turner et al., 2014; Egan et al., 2020). *S. aureus* has four native PBPs: PBP1, PBP2, PBP3, and PBP4 while *MRSA* contains a fifth PBP, known as PBP2a, a

transpeptidase encoded by the mecA gene (Fishovitz et al., 2014; Ferrer-González et al., 2021). Inhibition of PG biosynthesis and disruption of PG integrity will result in bacterial growth ceasing (Nikolaidis et al., 2014; Naclerio and Sintim, 2020). Likewise, it is speculated that WTA can diffuse out of the PG layer with a pronounced charge dependence (Baur et al., 2014; Kurokawa et al., 2016), while WTAs play a vital role in antibiotic resistance in MRSA (Farha et al., 2013); they increase the vulnerability of bacteria to metals, ions, antimicrobials, and peptides (Romero-Urbina et al., 2015). Due to the electrostatic interaction between  $Ti_3C_2T_x$  and the bacterial surface, the collapse of the bacterial outer cell wall scaffold caused the exposure of the cell membrane. Oxygen-containing groups on the Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene can bind with hydrogen bonds between lipopolysaccharide chains of the cell membrane to inhibit bacterial growth by preventing nutrient uptake (Rasool et al., 2016). Consequently, the phenomenon that the inhibitory activity of MRSA is higher than that of S. aureus may be correlated to the alteration in the cell wall thickness and structure caused by the different cross-linking methods of peptidoglycan on these two bacteria surfaces (Romero-Urbina et al., 2015; Rasool et al., 2017).

#### 3.3.3 Bacterial viability assessment

We used the AlamarBlue assay kit for bacterial viability detection on each sample. After incubating for 24 h at 37°C, 500 uL of 10% AlamarBlue was added and cultured for another 2 h. At last, 100 uL of the medium was transferred to a 96-well black plate, and the corresponding fluorescence intensity was detected. The results are shown in Figure 12A, for both bacteria, M50 exhibited the lowest bacterial viability (p < 0.001 vs. Ti). After conversion into the bacteriostatic rate, in the tested results of S. aureus, the antibacterial rates of the coating samples (M30/ 40/50) were 31.03, 35.63, and 45.98%, respectively. As for MRSA, the antibacterial rates were successively 17.39, 59.70, and 62.03%. The ability to inhibit MRSA of M30 is not as obvious as that against S. aureus, yet the antibacterial rates against MRSA of M40 and M50 are all over 50%, which is higher than that against S. aureus. Interestingly, combining the results of the previous two experiments, it can be seen that the antibacterial effect of the coated samples against MRSA is slightly better than that against S. aureus on the whole.

#### 3.3.4 Intracellular reactive oxide species assay

Previous studies have pointed to oxidative stress as the identical mechanism for the antimicrobial activity of several metals, metaloxides, and carbon-based nanomaterials (Zhang et al., 2010; Li et al., 2012). The condition for oxidative stress of cells to occur is exposure to higher levels of reactive oxygen species (ROS) such as free radicals,  $O_2^{\bullet}$ ,  ${}^{\bullet}OH$ , and  $H_2O_2$ . To figure out whether ROS production led to bacteria death, we measured the ROS levels using the DCFH-DA assay. The results are displayed in the data plot (Figure 12B). Compared with pure Ti, the ROS tested by these two bacteria on the coated samples showed a decreasing trend with increasing coating thickness, and the M50 was the lowest. It indicated that the antibacterial effect of the coated samples was not directly associated with the generation of ROS. This is a peculiar phenomenon compared to previous studies (Rasool et al., 2016; Yang et al., 2021). Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> and its reaction surface have strong reducing activity which is likely to reduce the reactive oxygen species in the culture environment and bacterial cells (Mashtalir et al., 2014). From another aspect, the  $Ti_3C_2T_x$  coating on the Ti surface may form an electron-enriched physiological environment around the implant, thereby reducing the bacterial invasion of the surrounding tissue. Meanwhile, it can also reduce the damage degree to cells by oxidative stress, showing that the coating composite materials have certain antioxidant properties. In the results of S. aureus, the ROS level of M50 was significantly lower than that of pure titanium sheets (p < 0.001), and the reducing extent of the ROS level in MRSA between the coated samples and pure Ti was not as evident as that in S. aureus. Except for ROSdependent oxidative stress, another mechanism of MXene toxicity is that strong attachment between MXene and cell membrane through the ionic interaction, hydrophobic, Van der Waals forces, and receptor-ligand binding led to membrane destabilization and loss of cell integrity. The direct contact between the MXene and cell membrane also results in MXene accumulation and ultimately cell death (Ganguly et al., 2018). That being so, the specific mediating mechanism for the antibacterial activity of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> titanium coating remains to be further explored.

### 4 Conclusion

In this research, a series of chemical characterization results (XRD, Raman, and XPS) confirmed that few-layer Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene was directly compounded on the surface of pure titanium sheets by anodic electrophoretic deposition without adding any electrolytic ions. The construction of a 2D MXene titanium coating model was successfully completed. Under SEM, the thickness of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> coating increases with the deposition voltages. In the meantime, the static water contact angle test and nano-scratch experiment show that the coatings have excellent hydrophilicity and bonding strength, which brought up a prerequisite for the antibacterial activity of the titanium coating. Then, the results of in vitro cytocompatibility experiments confirmed that the coating material has no cytotoxicity, and the composite of Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> can significantly improve the initial adhesion and proliferation of MC3T3-E1 cells, and the degree of improvement is in positive proportion to the content of  $Ti_3C_2T_x$  to some extent. The surface of the  $Ti_3C_2T_x$ coating exhibits a good antibacterial effect, among which M50 is the best. They can inhibit adhesion and prevent the formation of a biofilm of S. aureus and its drug-resistant strain MRSA, which are closely related to the occurrence of peri-implantitis. This effect may result from the pressure of the cell wall through electrostatic repulsion that destroys the bacterial envelope

leading to the exposure of the cell membrane. The direct contact between the MXene and cell membrane (through the ionic interaction, hydrophobic, Van der Waals forces, and receptor-ligand binding) causes MXene accumulation and damage to the integrity of the cell membrane, and ultimately leads to cell death. The specific antibacterial-mediated mechanism remains to be further explored. It can be reasonably envisaged that stronger biological activity and antibacterial property of coatings may be obtained by increasing the voltage value within an allowable scope (30-100 V). But coincidentally, we discovered the antioxidant capacity to reduce ROS in the culture environment and bacterial cells, which may reduce the degree of oxidative stress damage to cells. Through the comparative experimentations under the same condition, it was found that the antibacterial activity of the composites against MRSA was better than that of S. aureus, which may involve the difference in the cross-linking mode and thickness of peptidoglycan on the surface of these two bacteria. Therefore, under the unfavorable situation of multiple antibiotic resistance, the application of the new nano- $Ti_3C_2T_x$  coating on titanium inert surfaces to solve the clinical problem of periimplantitis in the biomedical field is more promising.

### Data availability statement

The original contributions presented in the study are included in the article/Supplementary Material; further inquiries can be directed to the corresponding author.

### Author contributions

SH: conceptualization, resources, formal analysis, investigation, software, data curation, and writing-original

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draft. YF: resources, data curation, and writing—original draft. AM: methodology, supervision, project administration, and funding acquisition.

# Funding

This work was supported by the Key Research and Development Program of Sichuan Province (Grant No. 2021YFS0084).

# Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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### Supplementary material

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fchem. 2022.991481/full#supplementary-material

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