

Review Article

Mannich Bases: An Important Pharmacophore in Present Scenario

Suman Bala, Neha Sharma, Anu Kajal, Sunil Kamboj, and Vipin Saini

M. M. College of Pharmacy, Maharishi Markandeshwar University, Mullana, Ambala, Haryana 133207, India

Correspondence should be addressed to Suman Bala; sumankmj7@gmail.com

Received 8 August 2014; Accepted 8 October 2014; Published 12 November 2014

Academic Editor: Patrick Bednarski

Copyright © 2014 Suman Bala et al. This is an open access article distributed under the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

Mannich bases are the end products of Mannich reaction and are known as beta-amino ketone carrying compounds. Mannich reaction is a carbon-carbon bond forming nucleophilic addition reaction and is a key step in synthesis of a wide variety of natural products, pharmaceuticals, and so forth. Mannich reaction is important for the construction of nitrogen containing compounds. There is a number of aminoalkyl chain bearing Mannich bases like fluoxetine, atropine, ethacrynic acid, trihexyphenidyl, and so forth with high curative value. The literature studies enlighten the fact that Mannich bases are very reactive and recognized to possess potent diverse activities like anti-inflammatory, anticancer, antifilarial, antibacterial, antifungal, anticonvulsant, anthelmintic, antitubercular, analgesic, anti-HIV, antimalarial, antipsychotic, antiviral activities and so forth. The biological activity of Mannich bases is mainly attributed to α , β -unsaturated ketone which can be generated by deamination of hydrogen atom of the amine group.

1. Introduction

Mannich bases, beta-amino ketones carrying compounds, are the end products of Mannich reaction [1, 2]. Mannich reaction is a nucleophilic addition reaction which involves the condensation of a compound with active hydrogen(s) with an amine (primary or secondary) and formaldehyde (any aldehyde) [3]. The schematic representation of general Mannich reaction is given in Scheme 1.

Mannich bases also act as important pharmacophores or bioactive leads which are further used for synthesis of various potential agents of high medicinal value which possess aminoalkyl chain. The examples of clinically useful Mannich bases which consist of aminoalkyl chain are cocaine, fluoxetine, atropine, ethacrynic acid, trihexyphenidyl, procyclidine, ranitidine, biperiden [4–6], and so forth. Mannich bases are known to play a vital role in the development of synthetic pharmaceutical chemistry. The literature studies revealed that Mannich bases are very reactive and can be easily converted to other compounds, for example, reduced to form physiologically active amino alcohols [7]. Mannich bases are known to possess potent activities like anti-inflammatory [8, 9], anticancer [10, 11], antifilarial [8], antibacterial [12, 13], antifungal [13, 14], anticonvulsant [15], anthelmintic

[16], antitubercular [17, 18], analgesic [19], anti-HIV [17], antimalarial [20], antipsychotic [21], antiviral [22] activities and so forth. Along with biological activities Mannich bases are also known for their uses in detergent additives [23], resins, polymers, surface active agents [24], and so forth. Prodrugs of Mannich bases of various active compounds have been prepared to overcome the limitations [25]. Mannich bases (optically pure chiral) of 2-naphthol are employed for catalysis (ligand accelerated and metal mediated) of the enantioselective carbon-carbon bond formation. Mannich bases and their derivatives are intermediates for the synthesis of bioactive molecules [26, 27]. Mannich reaction is widely used for the construction of nitrogen containing compounds [28]. Mannich bases have gained importance due to their application in antibacterial activity [29] and other applications are in agrochemicals such as plant growth regulators.

2. Biological Activities

2.1. Antimicrobial Activity. A novel series of Mannich bases of 3-substituted-4-(5-nitro-2-furfurylidene) amino-5-mercapto-1,2,4-triazoles **I(a–n)** (Figure 1) was synthesized and screened for the antifungal activity against *C. albicans* by employing disc diffusion method using nitrofurazone and

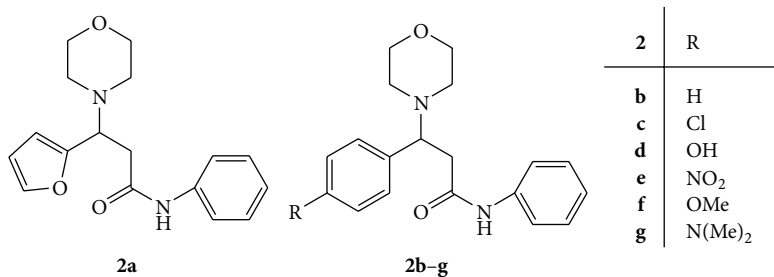


FIGURE 2

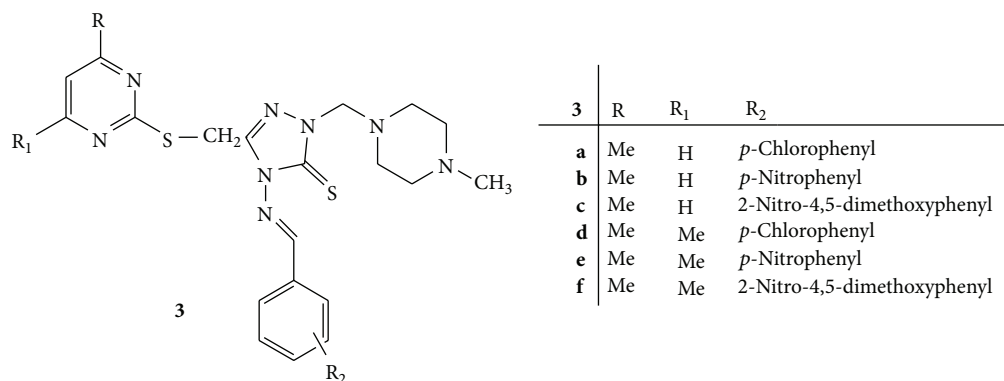


FIGURE 3

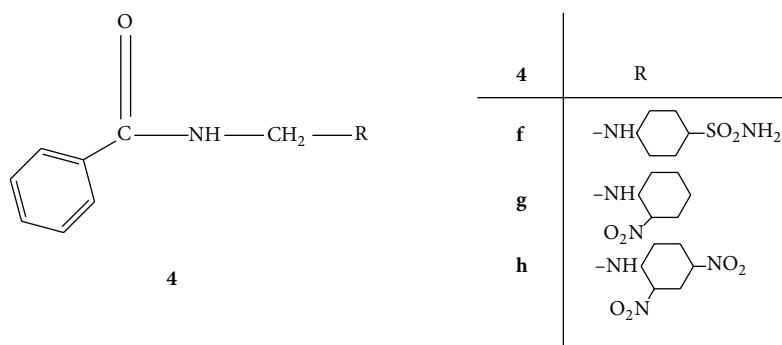


FIGURE 4

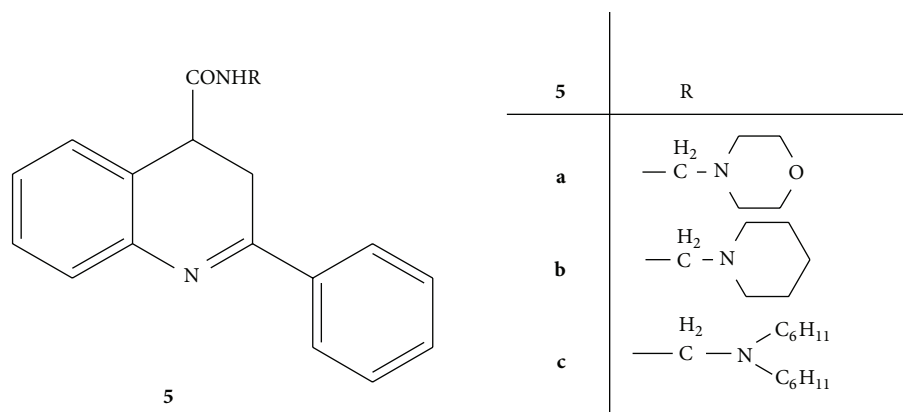


FIGURE 5

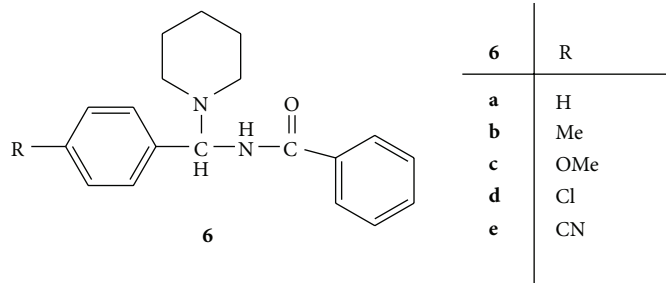


FIGURE 6

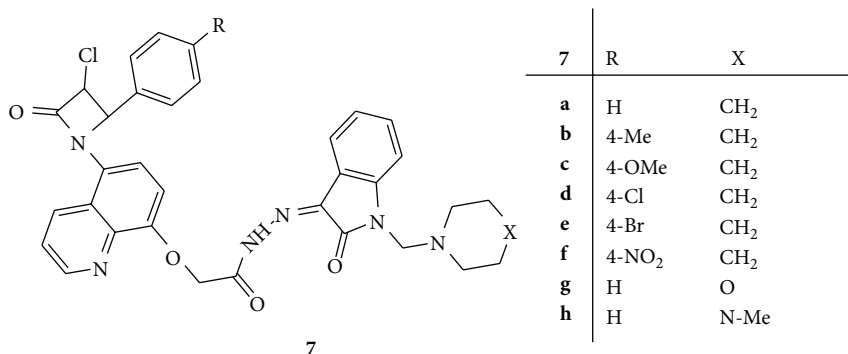


FIGURE 7

potent antimicrobial agents than cinchophen and both of the standard drugs. The compounds **5d** and **5e** were almost equipotent to cinchophen but did not respond to fungal strains [34].

A series of various substituted N-[(1-piperidinobenzyl)benzamide] (PBB) **6(a-e)** (Figure 6) have been synthesized. The antimicrobial activity of synthesized PBB was carried out on the microorganisms *S. aureus*, *B. subtilis*, *E. coli*, and *P. aeruginosa* by well diffusion method using ampicillin as standard drug. It has been found that all the compounds were found to be more active against all bacterial strains than ampicillin whereas compounds **6d** and **6e** substituted with chloro and cyano group were found to most potent. These results revealed that introduction of electron withdrawing groups on the phenyl ring enhances the activity whereas electron releasing groups are less active compared to unsubstituted phenyl ring [35].

Synthesis of novel Mannich bases (Z)-2-(5-(3-chloro-2-oxo-4-*p*-tolylazetid-1-yl)quinolin-8-yloxy)-*N'*-(2-oxo-1-(piperidin-1-ylmethyl)indolin-3-ylidene)acetohydrazides, **7(a-h)**, (Figure 7) was carried out and screened for antibacterial activity against *S. aureus*, *B. cereus*, *E. coli*, and *P. aeruginosa*. Amoxicillin and cefaclor were used as reference drugs. Among all the synthesized compounds **7d**, **7e**, and **7f** were found to be highly active. These results suggested that substitution with the nitro, chloro, and bromo group has led to enhanced antibacterial activity [36].

A series of N-Mannich bases of isatin **8(a-e)** were prepared with 4-amino-*N*-carbamidoyl benzene sulpho-
namide (Figure 8). The synthesized compounds were then

screened for antibacterial activity and 4-amino-*N*-carbamidoyl benzene sulpho-
namide was used as reference drug. From the antibacterial screening it was found that the compound bearing chloro group at 5-position has prominent activity against all the bacterial strains with lowest MIC values [37].

A series of 4-[1H-benzimidazole-yl](substituted benzal) methyl-amino] benzoic acids **9(a-c)** (Figure 9) was synthesized and screened for the antibacterial and antifungal activities against *S. aureus*, *B. subtilis*, *S. typhi*, *E. coli*, *A. niger*, and *C. albicans*, respectively. Ciprofloxacin and ketoconazole were used as standard drugs for antibacterial and antifungal activities, respectively. The results were obtained in terms of MIC values. The compounds **9b** and **9c** substituted with hydroxyl and methoxy group on aromatic ring have shown the highest antibacterial activity whereas compound **9a** substituted with dimethyl amine has shown highest antifungal activity [38].

Mannich bases of thiosemicarbazide **10(a-y)** (Figure 10) were synthesized and evaluated for the antifungal activity against *C. albicans* and *A. niger* by using brain heart infusion (BHI) method to estimate the MIC. Fluconazole was employed as standard drug for comparison. From the results it was concluded that analogs with only alkyl groups have shown comparable activity for both strains as compared to fluconazole whereas acyl derivatives were found to be partly active. Analogues with aromatic and substituted aromatic aldehydes showed least activity, while analogues with aliphatic aldehyde, ketones, and amines showed greater activity in *C. albicans* compared to *A. niger*. The analogs substituted

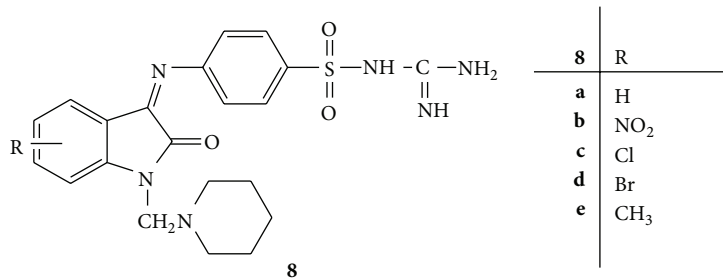


FIGURE 8

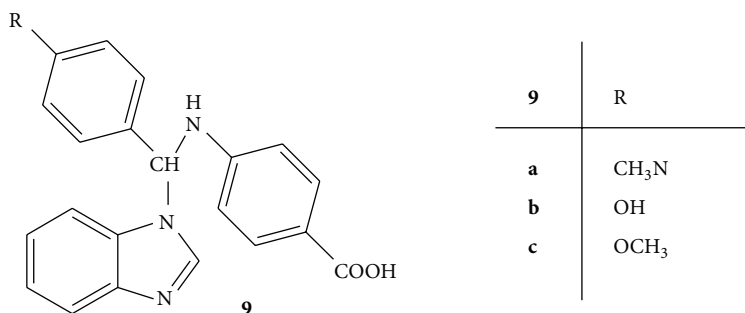


FIGURE 9

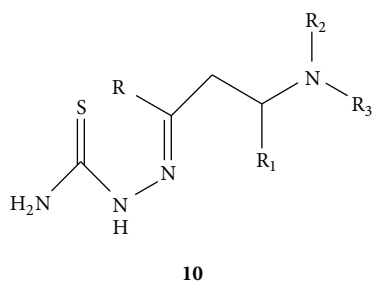


FIGURE 10

with morpholine and aromatic ketone were found to highly active. Compounds **10(q-t)** have shown comparable highest activities [39].

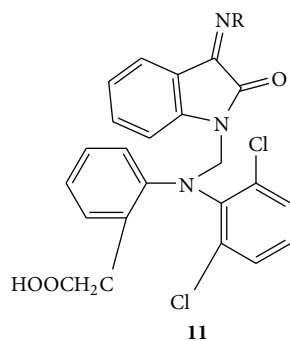
A series of Mannich bases of isatin and its derivatives with 2-[2,6-dichlorophenyl)amino] phenylacetic acid **11(a-f)** (Figure 11) was synthesized. All the derivatives were evaluated for antibacterial and antifungal activities. The results configured that all the synthesized compounds have shown moderate antibacterial activity except **11b**. The reason for moderate activity may be due to steric hindrance in binding of bacteria caused by bulky phenyl acetic acid group present at nitrogen atom of isatin nucleus. The compound **11b** has shown good activity against all the bacteria indicating that the substitution of isatin at the 3-position with a substituted phenyl ring favors antibacterial activity compared to an unsubstituted phenyl ring at the same position. All the compounds have shown significant antifungal activity, especially

compound **11e**, indicating that unsubstituted phenyl ring at 3-position of isatin is desired for antifungal activity compared to substituted phenyl ring [40].

A series of novel Mannich bases **12(a-h)** of 4-amino-3-(N-phthalimido-methyl)-1,2,4-triazole-5-thione (Figure 12) was synthesized and evaluated for the antimicrobial activity against a variety of gram positive, gram negative, and fungal strains. Levofloxacin was used as standard drug. The results revealed that Mannich bases **12h** with electron donating substituents on phenyl ring showed comparable activity to levofloxacin against *E. coli* and *K. pneumonia*. Mannich bases **12e** and **12f** with halogens on phenyl ring have shown activity very close to levofloxacin against *P. aeruginosa*. Mannich base **12b** with 2-hydroxy group has shown highest antifungal activity against *C. albicans*. The results also enlightened the fact that the presence of morpholine ring in heterocyclic molecules increases the antimicrobial activity [41].

2.2. Anti-Inflammatory Activity. Mannich bases of nicotinamide were synthesized with secondary amines like mefenamic acid, **13** (Figure 13), and diclofenac, **14** (Figure 14), in the presence of formaldehyde and hydrochloric acid. Anti-inflammatory activity was studied by hind paw oedema method using carrageenan as a phlogistic agent on Wistar rats. Nicotinamide, diclofenac, and mefenamic acid were used as standard drugs. Both the Mannich bases showed greater anti-inflammatory activity than the corresponding parent drugs [42].

Nicotinamide showed least anti-inflammatory activity. Mannich bases of diclofenac (MND) showed greater activity than Mannich bases of mefenamic acid (MNM). The order of anti-inflammatory activity was found as MND > MNM



11	R
b	N--SO ₂ NH ₂
e	HNN-

FIGURE 11

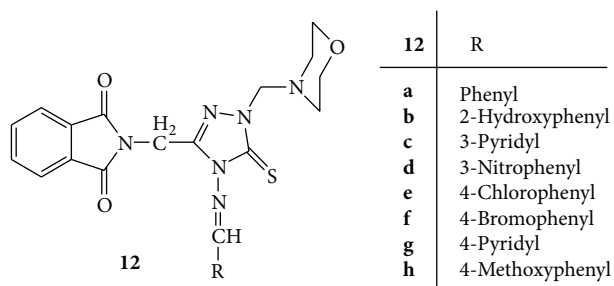


FIGURE 12

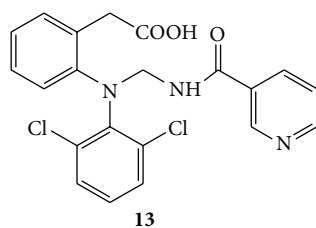


FIGURE 13

> diclofenac > mefenamic acid > nicotinamide. It appears that the synthesized Mannich bases showed synergistic anti-inflammatory activity [42].

Mannich bases of indole, **15**, derivatives were synthesized by reacting different derivatives of indole with various aromatic and heterocyclic amines in the presence of formaldehyde and dimethylformamide (Figure 15). Anti-inflammatory activity was screened on albino rats of Wistar strains by carrageenan paw induced method. Diclofenac sodium was used as standard drug. Carrageenan suspension 0.9% in sodium chloride was injected in plantar region of hind paw and paw volume was measured with the aid of plethysmometer. It was observed that the newly synthesized Mannich bases **15c**, **15f**, and **15g** possessing electron withdrawing groups (nitro and chloro) exhibited better anti-inflammatory activity [43].

A series of 4-[(4-aryl) methylidene] amino-2-(substituted-4-ylmethyl)-5-[1-[4-(2-methylpropyl)phenyl]ethyl]-2,4-dihydro-3H-1,2,4-triazole-3-thiones **16(a-n)** (Figure 16) was synthesized from ibuprofen by a three-component

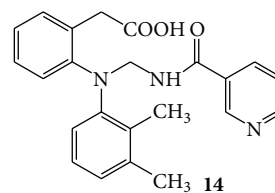
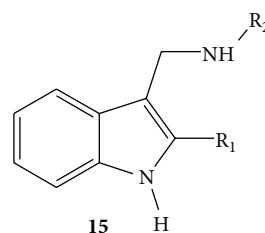


FIGURE 14



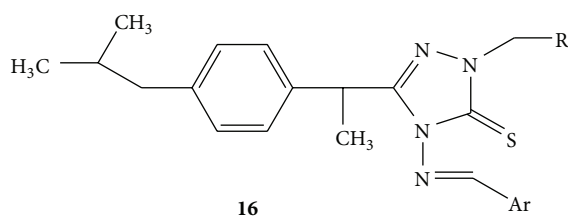
15	R ₁	R ₂
c	H	<i>p</i> -NO ₂ C ₆ H ₅
f	H	ClC ₆ H ₅ NS
g	Me	ClC ₆ H ₅ NS

FIGURE 15

Mannich reaction. The synthesized Mannich bases were screened for their anti-inflammatory activity by using carrageenan induced rat paw oedema model. Diclofenac and ibuprofen were used as standard drug substances. The Mannich bases **16b**, **16f**, **16k**, and **16l** carrying morpholino and *N*-methylpiperazino residues were found to be most promising anti-inflammatory agents [44].

A series of Mannich bases of 5-methyl-2-[(2-oxo-2H-chromen-3-yl)carbonyl]-2,4-dihydro-3H-pyrazol-3-one **17(a-j)** (Figure 17) was synthesized by using conventional and nonconventional (microwave) techniques. The newly synthesized Mannich bases were screened for their anti-inflammatory activity by means of carrageenan induced rat paw oedema model. Indomethacin was used as standard drug substance for comparison of results. Among all the compounds, **17f** and **17g** containing sulfonic group at *para* and *meta* positions, respectively, have shown potent anti-inflammatory activity along with minimum ulcerogenic index. In the compound **17d** introduction of bulky group resulted in the extreme decrease in the activity [45].

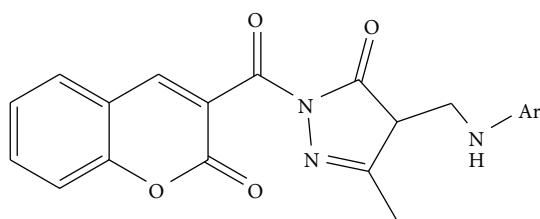
2.3. Anthelmintic Activity. *N*-Mannich bases of benzimidazolyl substituted 1H-isindole-1,3(2H)-dione **18(a-l)** were



16

16	Ar	R
b	4-ClC ₆ H ₄	Morpholine
f	4-CH ₃ C ₆ H ₄	Morpholine
k	4-NO ₂ C ₆ H ₄	Morpholine
l	4-NO ₂ C ₆ H ₄	1-Methylpiperazine

FIGURE 16



17

17	Ar
d	
f	
g	

FIGURE 17

synthesized (Figure 18). All the synthesized compounds were screened for anthelmintic activity by Watkins technique, against common Indian earthworm *P. posthuma*. Piperazine hydrochloride was used as standard drug. All the synthesized derivatives have shown significant anthelmintic activity whereas the derivatives substituted with different groups, that is, **18a** (piperazino-), **18b** (morpholine-), **18c** (diphenylamino-), **18i** (chloro), and **18j-l** (nitro and dinitro) groups, showed better activity than other derivatives [46].

A novel series of eight Mannich bases, 2-(3-chloro-2,5-dioxo-1-(4-substituted phenylamino)-8-(trichloromethyl)-1,6,7-triazaspiro[3.4]oct-7-en-6-yl)-N'-((3-methyl-5-oxo-1-(morpholine/piperidin/N-methylpiperazine-1-ylmethyl)-4,5-dihydro-1H-pyrazol-4-yl)methylene)acetohydrazides **19(a-h)** (Figure 19) were synthesized. All the synthesized Mannich bases were evaluated for their anthelmintic activity against *P. posthuma*. Piperazine citrate was used as standard drug. The compound **19h** containing N-methylpiperazine moiety was found to have significant anthelmintic activity [47].

2.4. Anticonvulsant Activity. Synthesis of two series, **20**, **21(a-c)**, of Mannich bases of 1,5-benzodiazepines was carried out and evaluated for the anticonvulsant activity by isoniazid and thiosemicarbazide induced convulsion model (Figures 20 and 21). Among all the synthesized compounds **20a** with cyclohexane at position 2 and **21a** with methyl group at

positions 2 and 4 were found to most active among all the synthesized compounds. The compounds **20b**, **20b**, and **21c** were found to be least active [48].

A series of various Mannich base derivatives of lamotrigine, **22(a-f)**, were synthesized (Figure 22). All the synthesized compounds were screened for anticonvulsant activity by maximal electroshock (MES) convulsion method. Phenobarbitone sodium and lamotrigine were used as reference drugs and % reduction of time spent by animals in extension, flexion, clonus, and stupor phase were noted. Compounds **22d** and **22f** showed more potent anticonvulsant activity when compared with that of the standard drug [49].

2.5. Anticancer Activity. A series of novel Mannich bases of 2-propoxybenzylideneisonicotinohydrazide, **23** (Figure 23), were screened for the cytotoxicity studies against the A549 human lung adenocarcinoma. Dulbecco's modified eagle medium was used for the growth of cells and supplemented with glutamine, fetal bovine serum, penicillin, and streptomycin. Gemcitabine was used as standard drug. The viability of the cells was assessed by MTT assay. Cells were placed in 96-well plate and after 24 h they were treated with different test compounds. In each well MTT in phosphate buffer saline was added and incubated and then DMSO was added to each well and kept in incubator. From all the compounds synthesized, **23c** and **23k** exhibited potential cytotoxic activity superior to that of the standard drug, gemcitabine [50].

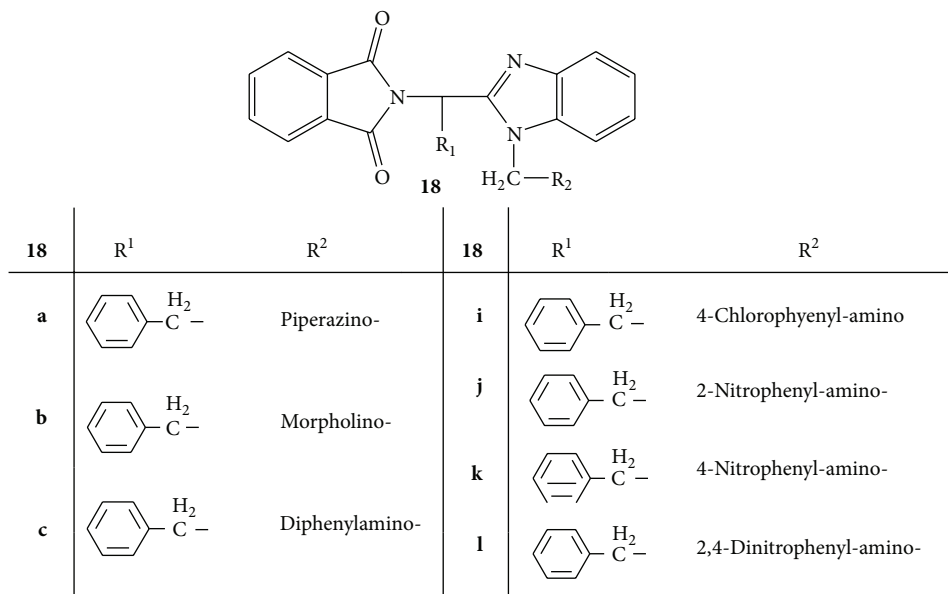


FIGURE 18

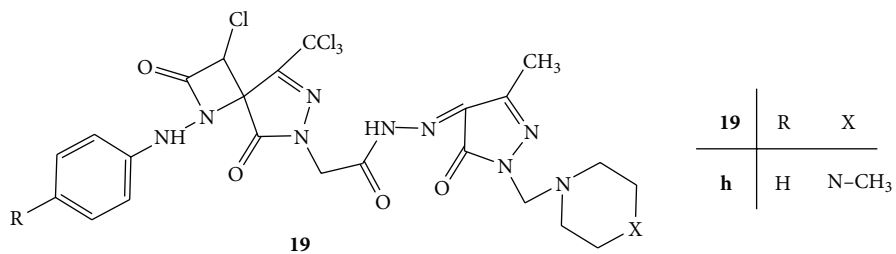


FIGURE 19

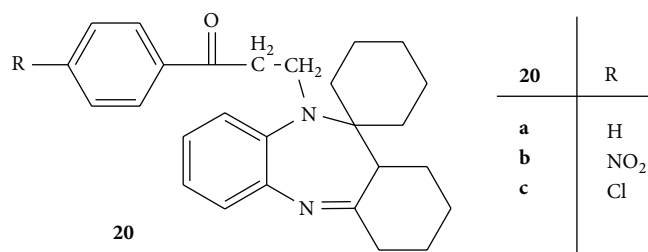


FIGURE 20

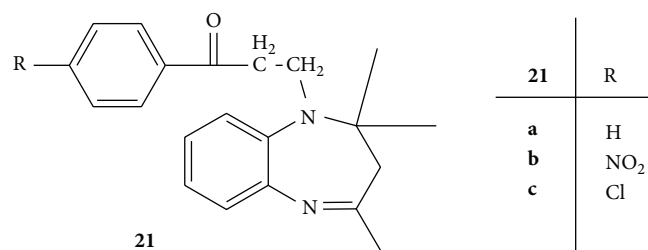


FIGURE 21

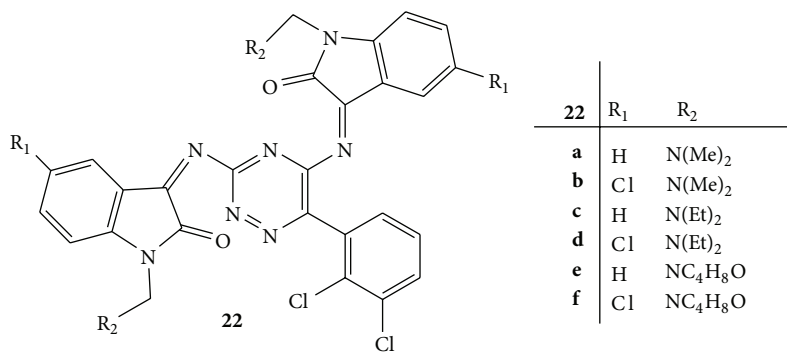


FIGURE 22

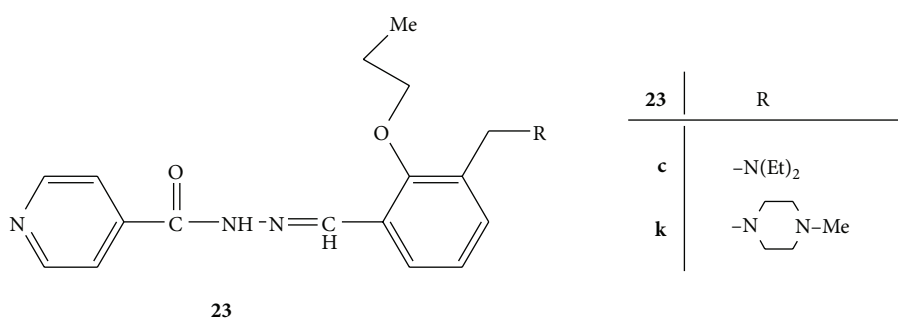


FIGURE 23

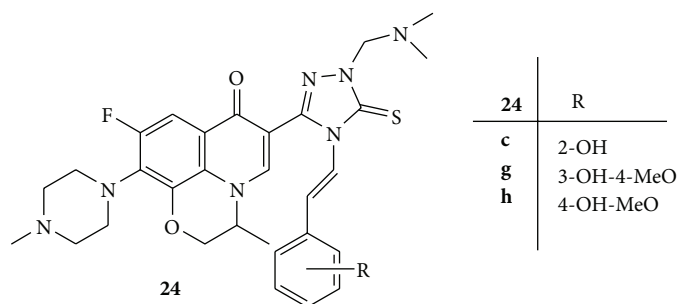


FIGURE 24

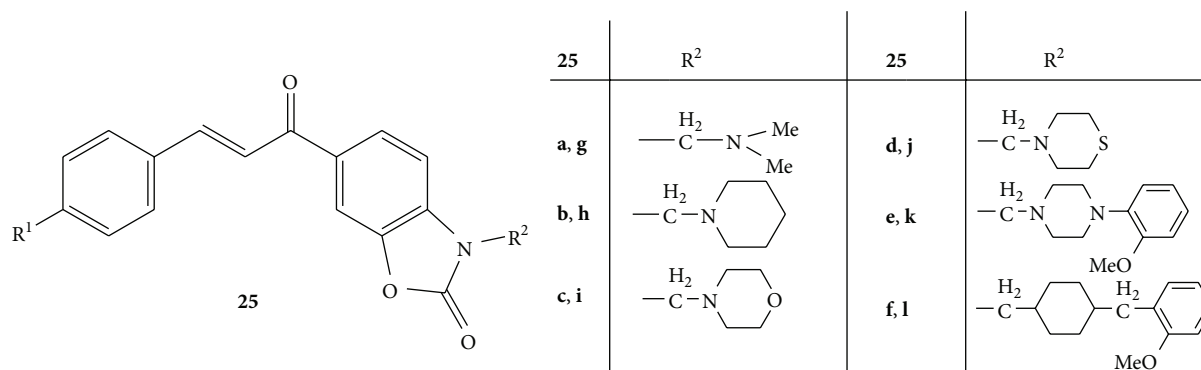
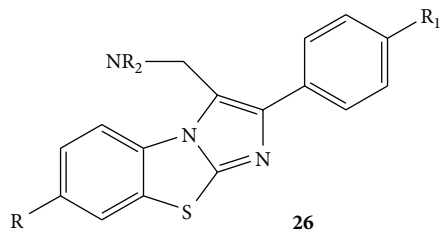
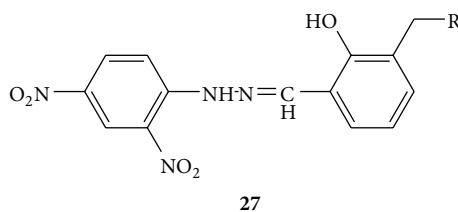


FIGURE 25



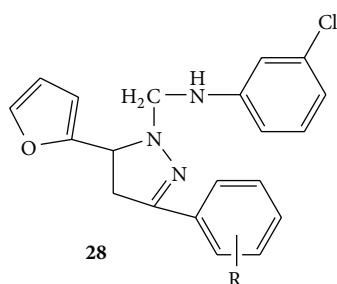
26	R	R ₁	R ₂
c	OC ₂ H ₅	H	4-(2-Pyridinyl)piperazino
f	F	H	Pyrrolidino

FIGURE 26



27	R
e	—N(C ₆ H ₅) ₂
h	—N
i	—N

FIGURE 27



28	R
a	H
b	<i>p</i> -Cl
c	<i>p</i> -Br
d	<i>o</i> -OH
e	<i>p</i> -OH

FIGURE 28

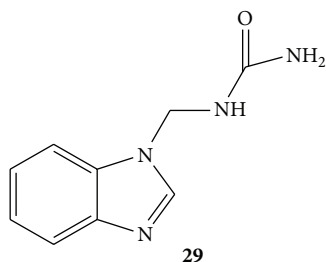


FIGURE 29

Schiff-Mannich bases of fluoroquinolones, **24** (Figure 24), from antibacterial analogs were synthesized and screened for antitumor activity against L1210, CHO, and HL60 cell lines. The cell lines were maintained in RPMI 1640 medium supplemented with fetal bovine serum. The medium containing 5×10^3 cells was seeded into 96-well microplate and compounds to be tested were then added. Plates were incubated and MTT solution in PBS was added to each well. The plates were further incubated for 4 h

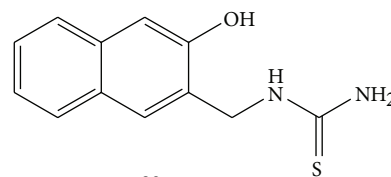


FIGURE 30

and DMSO was added to the wells containing CHO cell lines. Dodecylbenzenesulfonate (10%) was added to cells containing L1210 and HL60. The optimal density of each well was measured at 570 nm. Among the synthesized compounds (**24a–24k**), **24c**, **24g**, and **24h** exhibited significant cytotoxic activity than the parent compound [51].

Twelve new Mannich bases of 6-(3-aryl-2-propenoyl)-2(3H)-benzoxazolones **25** (Figure 25) were synthesized and evaluated for cytotoxicity in human pre-B-cell leukemia cell line BV-173 using MTT-dye reduction assay. The selected compounds were further evaluated on chronic myeloid leukemia K-562 cells. The cytotoxic studies revealed that the compound bearing 4-methoxy function (**25g–i**) at R¹ in B-ring causes significant decrease in potency while non-methoxy substituted analogues (**25a–f**) bearing bulky basic substituent at nitrogen were found to be more potent [52].

Some novel Mannich bases of 2-arylimidazo[2,1-b]benzothiazoles **26(a–h)** (Figure 26) were synthesized and evaluated for their anticancer activity. The anticancer activity was studied on HepG2 (hepatocarcinoma cell line), MCF-7 (breast carcinoma cell line), and HeLa 9 (human cervical carcinoma cell line) cell lines using MTT assay. All the synthesized compounds have shown cytotoxic activity against the used cell lines. The compounds **26c** and **26f**

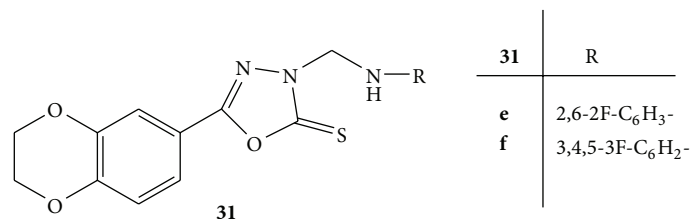


FIGURE 31

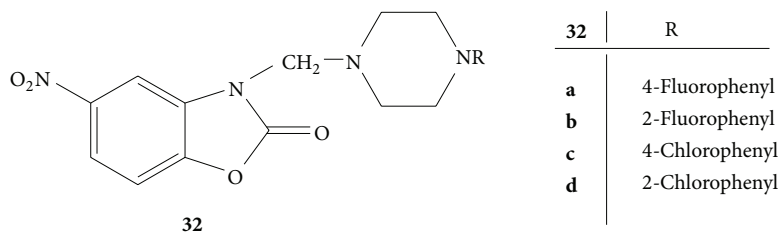


FIGURE 32

substituted with 4-(2-pyridinyl)piperazino and pyrrolidino, respectively, were found to induce G2/M cell cycle arrest with downregulation of cyclin B and upregulation of Chk2 protein. Both compounds have shown characteristic features of apoptosis. The compound **26f** could be considered the potential lead for its development as a novel anticancer agent [53].

2.6. Antioxidant Activity. Ten new Mannich bases were synthesized using (E)-2-{-2-(2,4-dinitrophenyl)hydrazono}methyl}phenol, **27** (Figure 27), as a key intermediate and screened for antioxidant activity. From all the synthesized derivatives **27(a-j)**, the compound **27h** containing morpholine moiety was found to be most active followed by the compound **27i** with piperazine moiety and compound **27e** with diphenylamine moiety [54].

A novel series of Mannich bases of pyrazolines **28(a-e)** was synthesized (Figure 28) and evaluated for the antioxidant activity using DPPH radical and NO radical scavenging methods. Ascorbic acid and rutin were employed as standard drugs for comparison. The results of both the assays suggested that compounds **28d** and **28e** have shown highest scavenging activity. The compound **28e** was found to have highest antioxidant capacity as compared to both of the standard drugs. These results confirmed that phenolic compounds possess high antioxidant ability as compared to nonphenolic compounds [55].

Two Mannich bases of benzamide, that is, 1-((1H-benzodiazole-1-yl)methyl)urea (BIUF) **29** (Figure 29) and 1-((3-hydroxynaphthalen-2-yl)methyl)thiourea (TNTUF) **30** (Figure 30) were synthesized and evaluated for the antioxidant activity. Hydrogen peroxide radical scavenging, DPPH radical scavenging, and reducing power assays were used for antioxidant activity estimation. Ascorbic acid was used as standard drug. Both of the Mannich bases were found to be active antioxidant agents due to presence of electron releasing amide group in them. BIUF was found to more

active than TNTUF due to the presence of two N atoms in the benzimidazole adjoined with amide group [56].

A series of novel Mannich bases of 1,3,4-oxadiazole derivatives having 1,4-benzodioxan **31** (Figure 31) were synthesized. All the synthesized novel compounds were screened for their *in vitro* antioxidant activity by using 2,2'-diphenyl-1-picrylhydrazyl radical (DPPH), 2,2'-azinobis (3-ethylbenzothiazoline-6-sulfonate) cationic radical (ABTS^{•+}), and ferric reducing antioxidant power (FRAP) scavenging assays [57].

All the compounds exhibited good antioxidant activities. BHT and Trolox were used as standard compounds for the comparison. The compounds **31e** and **31f** with multifluoro substitution on benzene ring showed significant radical scavenging ability in all three scavenging assays [57].

2.7. Analgesic Activity. A novel series of Mannich bases of 5-nitro 3-substituted piperazino methyl-2-benzoxazolinones, **32(a-d)** (Figure 32), were prepared and studied for the analgesic activity by *p*-benzoquinone induced writhing test. Among all the synthesized derivatives the compounds bearing electron withdrawing groups at *para* position such as fluoro and chloro showed potent analgesic activity [9].

Mannich bases of various 2-amino pyridine, **33** (Figure 33), derivatives were synthesized and screened for analgesic activity using acetic acid induced writhing test and hot plate test. A series of Mannich bases were prepared by condensing 2-amino pyridine with isatin and treating the above condensed products with various secondary amines. The analgesic activities were evaluated using central and peripheral analgesic assay. Indomethacin, diclofenac, and morphine were used as reference drugs. Among the various synthesized compounds **33e**, **33g**, **33h**, **33j**, **33k**, and **33l** revealed potent analgesic activity. Compounds **33f** and **33g** were found to be most potent against central and peripheral analgesic assay, respectively [58].

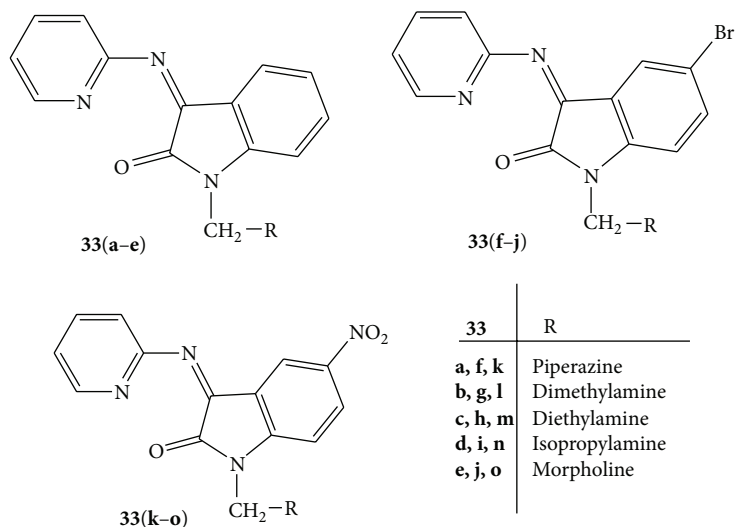


FIGURE 33

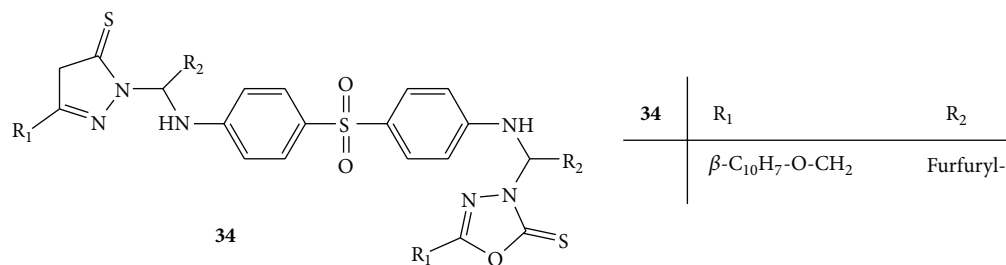


FIGURE 34

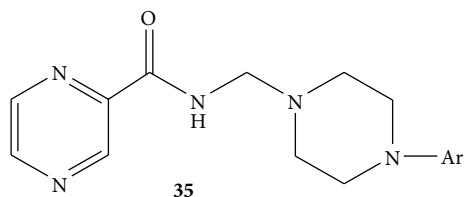
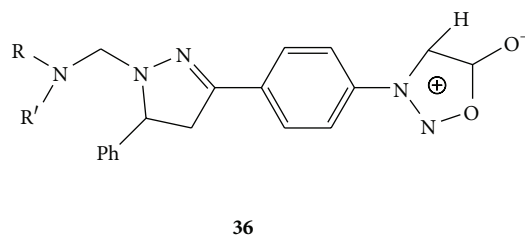


FIGURE 35

2.8. Antimycobacterial Activity. By reacting oxadiazole derivatives, dapson, and suitable aldehyde in the presence of methanol, a series of new oxadiazole Mannich bases **34(a-n)** (Figure 34) was synthesized. The synthesized Mannich bases were evaluated for their antimycobacterial activity against *M. tuberculosis* H37Rv and INH resistant *M. tuberculosis* using agar dilution method. Among all the synthesized compounds, compound **34d** 3-{2-furyl[4-(4-{2-furyl[5-(2-naphthoxy)methyl]-2-thioxo-2,3-dihydro-1,3,4-oxadiazol-3-yl]methylamino}phenylsulfonyl)-anilino]methyl}-5-(2-naphthoxy)methyl)-2,3-dihydro-1,3,4-oxadiazole-2-thione was found to be the most promising compound active against *M. tuberculosis* H37Rv and isoniazid (INH) resistant *M. tuberculosis* with minimum inhibitory concentration (MIC) of 0.1 μ M and 1.10 μ M, respectively [59].

By using microwave irradiation method, a series of Mannich bases of pyrazinamide was synthesized. The synthesized compounds were evaluated for their *in vitro* and *in vivo* antimycobacterial activity. For *in vivo* activity *M. tuberculosis* H37Rv (MTB) strain was used. Among all the synthesized compounds, 1-cyclopropyl-6-fluoro-1,4-dihydro-8-methoxy-7-(3-methyl-4-((pyrazine-2-carboxamido)methyl)piperazin-1-yl)-4-oxoquinoline-3 carboxylic acid **35** (Figure 35) was found to be the most active compound *in vitro* with MIC of 0.39 and 0.2 lg/mL against MTB and multidrug-resistant MTB, respectively. In the *in vivo* animal model 17 decreased the bacterial load in lung and spleen tissues with 1.86 and 1.66 – log₁₀ protections, respectively [60].

A series of Mannich bases of 3-[p-(5-arylpyrazolin-3-yl) phenyl]syndone **36(a-j)** (Figure 36) was synthesized and studied for the antimycobacterial activity. The antimycobacterial activity of the test compounds was evaluated against the standard strain of *M. tuberculosis* H37Rv. Streptomycin and pyrazinamide were used as standards. The results exposed the fact that the compounds **36c**, **36d**, **36e**, **36g**, and **36i** exhibited good activity. The promising activities are accredited to the presence of long alkylating chains with electron donating groups like hydroxyl, amino, methylene, and ethylene through mesomeric effect attached to the pyrazoline moiety [61].



36	R	R'
c	t-Butyl	H
d	CH ₃ CH ₂ CH ₂ CH ₂ ⁻	H
e		H
g	(CH ₂) ₅	H
i		H

FIGURE 36

3. Conclusion

As demonstrated by the frame of work reviewed in this paper, Mannich bases and their derivatives are found to have potent diverse activities. This review summarized various biological activities of Mannich base derivatives in present scenario. It can be concluded that Mannich bases have remarkable biological potential which is remaining unexplored. However this review would expectantly shed light on ways to raise the therapeutic worth and specificity of Mannich bases. This bioactive core has maintained the interest of researchers in gaining the most suggestive and conclusive access in the field of various Mannich bases of medicinal importance from last decades and also endorsed the researchers for design of novel heterocyclic/aryl derivatives for progress of new environment-friendly technology.

Conflict of Interests

The authors declare that there is no conflict of interests regarding the publication of this paper.

References

- [1] *Advanced Organic Chemistry: Reactions, Mechanisms, and Structure*, John Wiley & Sons, New York, NY, USA, 3rd edition, 1985.
- [2] V. J. Belinelo, G. T. Reis, G. M. Stefani, D. L. Ferreira-Alves, and D. Piló-Veloso, "Synthesis of 6 α ,7 β -dihydroxyvouacapan-17 β -oic acid derivatives. Part IV: mannich base derivatives and its activities on the electrically stimulated guinea-pig ileum preparation," *Journal of the Brazilian Chemical Society*, vol. 13, no. 6, pp. 830–837, 2002.
- [3] S. Joshi, N. Khosla, and P. Tiwari, "In vitro study of some medicinally important Mannich bases derived from anti-tubercular agent," *Bioorganic & Medicinal Chemistry*, vol. 12, no. 3, pp. 571–576, 2004.
- [4] L. Racane, V. T. Kulenovic, L. F. Jakic, D. W. Boykin, and G. K. Zamola, "Synthesis of bis-substituted amidino-benzothiazoles as potential anti-HIV agents," *Heterocycles*, vol. 55, pp. 2085–2098, 2001.
- [5] E. Kashiyama, I. Hutchinson, M.-S. Chua et al., "Antitumor benzothiazoles. 8.1 Synthesis, metabolic formation, and biological properties of the C- and N-oxidation products of antitumor 2-(4-aminophenyl)-benzothiazoles," *Journal of Medicinal Chemistry*, vol. 42, no. 20, pp. 4172–4184, 1999.
- [6] S. R. Bhusare, R. P. Pawar, and Y. B. Vibhute, "Synthesis and antibacterial activity of some new 2-(substituted phenyl sulfonamido)-6-substituted benzothiazoles," *Indian Journal of Heterocyclic Chemistry*, vol. 11, no. 1, pp. 79–80, 2001.
- [7] N. Raman, S. Esthar, and C. Thangaraja, "A new Mannich base and its transition metal (II) complexes—synthesis, structural characterization and electrochemical study," *Journal of Chemical Sciences*, vol. 116, no. 4, pp. 209–213, 2004.
- [8] B. Kalluraya, R. M. Chimbalkar, and J. C. Hegde, "Anti-convulsant activity of nicotinyli/isonicotinyl substituted 1,2,4-triazol-5-thione Mannich bases," *Indian Journal of Heterocyclic Chemistry*, vol. 15, no. 1, pp. 15–18, 2005.
- [9] M. Köksal, N. Gökhan, E. Küpeli, E. Yesilada, and H. Erdogan, "Analgesic and antiinflammatory activities of some new Mannich bases of 5-nitro-2-benzoxazolinones," *Archives of Pharmacological Research*, vol. 30, no. 4, pp. 419–424, 2007.
- [10] Y. Ivanova, G. Momekov, O. Petrov, M. Karaivanova, and V. Kalcheva, "Cytotoxic Mannich bases of 6-(3-aryl-2-propenoyl)-2(3H)-benzoxazolones," *European Journal of Medicinal Chemistry*, vol. 42, no. 11-12, pp. 1382–1387, 2007.
- [11] H. I. Gul, J. Vepsalainen, M. Gul, E. Erciyas, and O. Hanninen, "Cytotoxic activities of mono and bis Mannich bases derived from acetophenone against Renca and Jurkat cells," *Pharmaceutica Acta Helveticae*, vol. 74, no. 4, pp. 393–398, 2000.
- [12] M. Ashok, B. S. Holla, and B. Poojary, "Convenient one pot synthesis and antimicrobial evaluation of some new Mannich bases carrying 4-methylthiobenzyl moiety," *European Journal of Medicinal Chemistry*, vol. 42, no. 8, pp. 1095–1101, 2007.
- [13] S. N. Pandeya, D. Sriram, G. Nath, and E. De Clercq, "Synthesis, antibacterial, antifungal and anti-HIV activities of norfloxacin Mannich bases," *European Journal of Medicinal Chemistry*, vol. 35, no. 2, pp. 249–255, 2000.
- [14] B. N. Singh, S. K. Shukla, and M. Singh, "Synthesis and biological activity of sulphadiazine Schiff's bases of isatin and their N-mannich bases," *Asian Journal of Chemistry*, vol. 19, no. 7, pp. 5013–5018, 2007.
- [15] S. C. Vashishtha, G. A. Zello, K. H. Nienaber et al., "Cytotoxic and anticonvulsant aryloxyaryl Mannich bases and related

- compounds," *European Journal of Medicinal Chemistry*, vol. 39, no. 1, pp. 27–35, 2004.
- [16] E. Bennet-Jenkins and C. Bryant, "Novel sources of anthelmintics," *International Journal for Parasitology*, vol. 26, no. 8-9, pp. 937–947, 1996.
- [17] D. Sriram, D. Banerjee, and P. Yogeewari, "Efavirenz Mannich bases: synthesis, anti-HIV and antitubercular activities," *Journal of Enzyme Inhibition and Medicinal Chemistry*, vol. 24, no. 1, pp. 1–5, 2009.
- [18] J. S. Mulla, A. Y. Khan, S. I. Panchamukhi, M. A. Khazi, M. B. Kalashetti, and I. M. Khazi, "Synthesis and antitubercular activity of Mannich bases of imidazo [2,1-b] [1,3,4] thiadiazoles," *Indian Journal of Novel Drug Delivery*, vol. 3, no. 4, pp. 289–295, 2011.
- [19] W. Malinka, P. Świątek, B. Filipek, J. Sapa, A. Jezierska, and A. Koll, "Synthesis, analgesic activity and computational study of new isothiazolopyridines of Mannich base type," *Farmaco*, vol. 60, no. 11-12, pp. 961–968, 2005.
- [20] G. B. Barlin and C. Jiravinya, "Potential antimalarials . X. Di-Mannich Bases of 4-(7'-Trifluoromethyl-1',5'-naphthyridin-4'-ylamino)phenol and N-(4'-Diethylamino-1'-methylbutyl)-7-trifluoromethyl-1,5-naphthyridin-4-amine," *Australian Journal of Chemistry*, vol. 43, no. 7, pp. 1175–1181, 1990.
- [21] M. K. Scott, G. E. Martin, D. L. DiStefano et al., "Pyrrole mannich bases as potential antipsychotic agents," *Journal of Medicinal Chemistry*, vol. 35, no. 3, pp. 552–558, 1992.
- [22] M. L. Edwards, H. W. Ritter, D. M. Stemerick, and K. T. Stewart, "Mannich bases of 4-phenyl-3-buten-2-one: a new class of antiherpes agent," *Journal of Medicinal Chemistry*, vol. 26, no. 3, pp. 431–436, 1983.
- [23] R. E. Karll and R. J. Lee, "Process and compositions," US Patent no. 4,384,138, 1983.
- [24] F. P. Otto, US Patent, US 3 649 229, 1972.
- [25] J. R. Dimmock and P. Kumar, "Anticancer and cytotoxic properties of Mannich bases," *Current Medicinal Chemistry*, vol. 4, no. 1, pp. 1–22, 1997.
- [26] J.-X. Ji, L.-Q. Qiu, C. Wing Yip, and A. S. C. Chan, "A convenient, one-step synthesis of optically active tertiary aminonaphthol and its applications in the highly enantioselective alkenylations of aldehydes," *The Journal of Organic Chemistry*, vol. 68, no. 4, pp. 1589–1590, 2003.
- [27] P.-J. J. Huang, D. Youssef, T. S. Cameron, and A. Jha, "Microwave-assisted synthesis of novel 2-naphthol bis-Mannich Bases," *Arkivoc*, vol. 2008, no. 16, pp. 165–177, 2008.
- [28] M. Arend, B. Westermann, and N. Risch, "Modern variants of the Mannich reaction," *Angewandte Chemie—International Edition*, vol. 37, no. 8, pp. 1045–1070, 1998.
- [29] B. S. Holla, M. K. Shivananda, M. S. Shenoy, and G. Antony, "Studies on arylfuran derivatives. Part VII. Synthesis and characterization of some Mannich bases carrying halophenylfuryl moieties as promising antibacterial agents," *Farmaco*, vol. 53, no. 8-9, pp. 531–535, 1998.
- [30] M. K. Shivananda and P. Prakash Shet, "Antifungal activity studies of some mannich bases carrying nitrofurantoin moiety," *Journal of Chemical and Pharmaceutical Research*, vol. 3, no. 2, pp. 303–307, 2011.
- [31] A. Idhayadhulla, R. Surendra Kumar, A. Jamal Abdul Nasser, and A. Manilal, "Synthesis and antimicrobial activity of some new mannich base derivatives," *Journal of Chemical and Pharmaceutical Research*, vol. 3, no. 4, pp. 904–911, 2011.
- [32] B. Lingappa, K. S. Girisha, B. Kalluraya, N. Satheesh Rai, and N. S. Kumari, "Regioselective reaction: synthesis of novel Mannich bases derived from 3-(4,6-disubstituted-2-thiomethylpyrimidyl)-4-amino-5-mercapto-1,2,4-triazoles and their antimicrobial properties," *Indian Journal of Chemistry—Section B Organic and Medicinal Chemistry*, vol. 47, no. 12, pp. 1858–1864, 2008.
- [33] S. Bala, N. Sharma, A. Kajal, and S. Kamboj, "Design, synthesis, characterization, and computational studies on benzamide substituted mannich bases as novel, potential antibacterial agents," *The Scientific World Journal*, vol. 2014, Article ID 732141, 9 pages, 2014.
- [34] P. P. Jumade, S. J. Wadher, A. J. Chourasia, U. V. Kharabe, D. Mude, and P. G. Yeole, "Synthesis of newer Mannich bases of quinoline derivative for antimicrobial activity," *International Journal of Chemical Sciences*, vol. 7, no. 3, pp. 1518–1530, 2009.
- [35] S. Ravichandran and S. S. Kumar, "Synthesis, characterization and antibacterial activity of Mannich base, N-[(1-piperidinobenzyl)]benzamide: a structure and reactivity study," *Asian Journal of Biochemical and Pharmaceutical Research*, vol. 2, no. 1, pp. 136–142, 2011.
- [36] G. Madhu, K. N. Jayaveera, L. K. Ravindra Nath, B. Santosh Kumar, and P. Nagarjuna Reddy, "Synthesis and structure activity relationship of new antibacterial active multi substituted quinoline-azetidinone mannich bases," *Der Pharma Chemica*, vol. 4, no. 3, pp. 1033–1040, 2012.
- [37] U. K. Singh, S. N. Pandeya, A. Singh, B. K. Srivastava, and M. Pandey, "Synthesis and antimicrobial activity of Schiff's and N-Mannich bases of isatin and its derivatives with 4-amino-n-carbamimidoyl benzene sulphonamide," *International Journal of Pharmaceutical Sciences and Drug Research*, vol. 2, no. 2, pp. 151–154, 2010.
- [38] M. Vijey Aanandhi, A. K. Verma, R. Sujatha, and R. Kamal Raj, "Synthesis and characterization of novel mannich bases of benzimidazole derivatives for antibacterial and antifungal activity," *International Journal of Pharmacy and Pharmaceutical Sciences*, vol. 5, no. 2, pp. 295–297, 2013.
- [39] S. A. Pishawikar and H. N. More, "Synthesis, docking and *in-vitro* screening of mannich bases of thiosemicarbazide for antifungal activity," *Arabian Journal of Chemistry*, 2013.
- [40] V. Ravichandran, S. Mohan, and K. S. Kumar, "Synthesis and antimicrobial activity of Mannich bases of isatin and its derivatives with 2-[(2,6-dichlorophenyl)amino]phenylacetic acid," *ARKIVOC*, vol. 2007, no. 14, pp. 51–57, 2007.
- [41] U. Yunus, M. H. Bhatti, N. Rahman, N. Mussarat, S. Asghar, and B. Masood, "Synthesis, characterization, and biological activity of novel Schiff and Mannich bases of 4-amino-3-(N-phthalimidomethyl)-1,2,4-triazole-5-thione," *Journal of Chemistry*, vol. 2013, Article ID 638520, 8 pages, 2013.
- [42] R. Babbar, D. P. Pathak, N. Jain, and S. Jain, "Synthesis and anti-inflammatory activity of Mannich bases of Nicotinamide with diclofenac and mefenamic acid," *Der Pharma Chemica*, vol. 4, no. 5, pp. 2024–2028, 2012.
- [43] P. Lohitha, D. Giles, C. Sreedhar et al., "Synthesis and pharmacological evaluation of schiff's and Mannich bases of indole derivatives," *RUGHHS Journal of Pharmaceutical Sciences*, vol. 1, no. 1, pp. 70–78, 2011.
- [44] K. V. Sujith, J. N. Rao, P. Shetty, and B. Kalluraya, "Regioselective reaction: synthesis and pharmacological study of Mannich bases containing ibuprofen moiety," *European Journal of Medicinal Chemistry*, vol. 44, no. 9, pp. 3697–3702, 2009.

- [45] K. K. Sivakumar, A. Rajasekaran, P. Senthilkumar, and P. P. Wattamwar, "Conventional and microwave assisted synthesis of pyrazolone Mannich bases possessing anti-inflammatory, analgesic, ulcerogenic effect and antimicrobial properties," *Bioorganic and Medicinal Chemistry Letters*, vol. 24, no. 13, pp. 2940–2944, 2014.
- [46] R. Rita and P. Shrivastva, "Synthesis and characterization of some N-Mannich bases as potential antimicrobial, anthelmintic and insecticidal agents," *Chemical Science Transaction*, vol. 1, no. 2, pp. 431–439, 2012.
- [47] K. R. Raju, A. Guru Prasad, B. Kumar, L. Rao, and K. R. Ravindranath, "Synthesis and medicinal evaluation of Mannich bases carrying azetidinone moiety," *Journal of Clinical and Analytical Medicine*, pp. 1–5, 2014.
- [48] N. S. Pandeya and N. Rajput, "Synthesis and anticonvulsant activity of various Mannich and Schiff bases of 1,5-benzodiazepines," *International Journal of Medicinal Chemistry*, vol. 2012, Article ID 237965, 10 pages, 2012.
- [49] A. A. Kulkarni, S. B. Wankhede, N. D. Dhawale, P. B. Yadav, V. V. Deore, and I. D. Gonjari, "Synthesis, characterization and biological behavior of some Schiff's and Mannich base derivatives of Lamotrigine," *Arabian Journal of Chemistry*, 2013.
- [50] M. Malhotra, M. Arora, A. Samad, K. Sahu, P. Phogat, and A. Deep, "Synthesis and evaluation of some novel derivatives of 2-propoxybenzylideneisonicotinohydrazide for their potential antimicrobial activity," *Journal of the Serbian Chemical Society*, vol. 77, no. 5, pp. 589–597, 2012.
- [51] G. Hua, T. Cao, H. Wenlong et al., "Design, synthesis and antitumor activities of fluoroquinolone C-3 heterocycles (IV): S-triazole Schiff-Mannich bases derived from ofloxacin," *Acta Pharmaceutica Sinica B*, vol. 2, no. 3, pp. 312–317, 2012.
- [52] Y. Ivanova, G. Momekov, O. Petrov, M. Karaivanova, and V. Kalcheva, "Cytotoxic Mannich bases of 6-(3-aryl-2-propenoyl)-2(3H)-benzoxazolones," *European Journal of Medicinal Chemistry*, vol. 42, no. 11-12, pp. 1382–1387, 2007.
- [53] R. M. Kumbhare, K. Vijay Kumar, M. J. Ramaiah et al., "Synthesis and biological evaluation of novel Mannich bases of 2-arylimidazo[2,1-b]benzothiazoles as potential anti-cancer agents," *European Journal of Medicinal Chemistry*, vol. 46, no. 9, pp. 4258–4266, 2011.
- [54] M. Malhotra, R. Sharma, M. Sanduja, R. Kumar, E. Jain, and A. Deep, "Synthesis, characterization and evaluation of Mannich bases as potent antifungal and hydrogen peroxide scavenging agents," *Acta Poloniae Pharmaceutica: Drug Research*, vol. 69, no. 2, pp. 355–361, 2012.
- [55] P. C. Jagadish, N. Soni, and A. Verma, "Design, synthesis, and in vitro antioxidant activity of 1,3,5-trisubstituted-2-pyrazolines derivatives," *Journal of Chemistry*, vol. 2013, Article ID 765768, 6 pages, 2013.
- [56] K. Chakkaravarthi, K. Gokulkrishnan, T. Suman, and D. Tamilvendan, "Synthesize, spectral, antimicrobial and antioxidant studies of diamide mannich base derivatives," *International Journal of Pharmacy and Pharmaceutical Sciences*, vol. 6, no. 1, pp. 492–495, 2013.
- [57] L. Ma, Y. Xiao, C. Li et al., "Synthesis and antioxidant activity of novel Mannich base of 1,3,4-oxadiazole derivatives possessing 1,4-benzodioxan," *Bioorganic and Medicinal Chemistry*, vol. 21, no. 21, pp. 6763–6770, 2013.
- [58] C. K. Kumar and S. N. Pandeya, "Synthesis, characterization and analgesic activity of various pyridine derivatives," *International Journal of Ayurveda and Pharmacy*, vol. 2, no. 6, pp. 1763–1767, 2011.
- [59] M. A. Ali and M. Shaharyar, "Oxadiazole mannich bases: synthesis and antimycobacterial activity," *Bioorganic and Medicinal Chemistry Letters*, vol. 17, no. 12, pp. 3314–3316, 2007.
- [60] D. Sriram, P. Yogeewari, and S. P. Reddy, "Synthesis of pyrazinamide Mannich bases and its antitubercular properties," *Bioorganic and Medicinal Chemistry Letters*, vol. 16, no. 8, pp. 2113–2116, 2006.
- [61] T. Taj, R. R. Kamble, T. M. Gireesh, and R. K. Hunnur, "Facile syntheses of Mannich bases of 3-[p-(5-arylpyrazolin-3-yl)phenyl] sydnones, as anti-tubercular and anti-microbial agents, under ionic liquid/tetrabutylammonium bromide catalytic conditions," *Journal of the Serbian Chemical Society*, vol. 76, no. 8, pp. 1069–1079, 2011.