Published online 2015 February 20.

# Evaluation of Sub-acute Oral Toxicity of Lithium Carbonate Microemulsion (Nano Size) on Liver and Kidney of Mice

# Heibatullah Kalantari<sup>1</sup>; Anayatollah Salimi<sup>2,\*</sup>; Anahita Rezaie<sup>3</sup>; Fereshteh Jazayeri Shushtari<sup>4</sup>; Mehdi Goudarzi<sup>1</sup>

1 Nanotechnology Research Center, Department of Pharmacology and Toxicology, School of Pharmacy, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, IR Iran

<sup>2</sup>Nanotechnology Research Center, Department of Pharmaceutics, School of Pharmacy, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, IR Iran
<sup>3</sup>Department of Pathobiology, School of Veterinary Medicine, Shahid Chamran University, Ahvaz, IR Iran
<sup>4</sup>School of Pharmacy, Arvand International Institute, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, IR Iran

\*Corresponding author: Anayatollah Salimi, Nanotechnology Research Center, Department of Pharmaceutics, School of Pharmacy, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, IR Iran. Tel: +98-6113738381, E-mail: anayatsalimi2003@yahoo.com

Received: July 25, 2014; Revised: September 22, 2014; Accepted: October 4, 2014

Background: The development of drug delivery systems has improved the therapeutic and toxic properties of existing drugs in therapy. Microemulsion systems are novel vehicles for drug delivery, which have been developed in recent years. These systems are currently of interest to the pharmaceutical scientist because of their considerable potential to act as drug delivery vehicles by incorporating into a wide range of drug molecules. Although these systems improved solubility and bioavailability of drugs, they may have potential toxic effects on the body organs.

Objectives: The purpose of this study was to examine a possible hepatotoxic and nephrotoxic effect of lithium carbonate microemulsion (LCME) in a mice model.

Materials and Methods: Eighty male Swiss albino mice were randomly allocated to eight experimental groups, as follows: Group 1, as negative control group were treated orally with normal saline (0.9% NaCl); Group 2, received microemulsion base without drug as control group; Groups 3 to 5, received lithium carbonate (LC) solution in doses of 50, 100, and 200 mg/kg, respectively; Groups 6 to 8, received LCME orally in doses of 50, 100, and 200 mg/kg, respectively. All drugs were administered orally for ten consecutive days. Serum glutamate pyruvate aminotransferase (SGPT), serum glutamate oxaloacetate aminotransferase (SGOT), alkaline phosphatase (ALP), blood urea nitrogen (BUN), and plasma creatinine (Cr), as markers of liver and kidney toxicity in treated mice, were measured. Furthermore, the changes of tissue were assessed by histopathologic examination.

Results: The findings showed that serum activity of ALP, SGOT, and SGPT and the levels of BUN and Cr in microemulsion base group was greater than normal saline group. However, this difference was not significant. Administration of LC and LCME in all doses resulted in a significant increase in the levels of BUN and serum activity of SGOT and SGPT in comparison to normal saline group (P < 0.05). Histopathological changes were observed in mice treated with LC or LCME.

Conclusions: This study showed that subacute oral administration of different doses of LCME with severe toxicity in comparison to the same dose of LC.

Keywords: Lithium Carbonate; Hepatotoxicity; Nephrotoxicity

#### 1. Background

The development of drug delivery systems has been improved the therapeutic and toxic properties of existing drugs (1). In the past few years, engineered nanoparticles have been investigated for many applications due to the enhanced material properties that result from a reduction in particle dimensions (2). Microemulsion (ME) method is a very versatile technique that enables the preparation of diverse nano-materials on their own or in combination with other techniques (3, 4). The MEs are clear, stable, isotropic mixtures of oil, water, and surfactant, frequently in combination with a co-surfactant. These systems are currently of interest to the pharmaceutical scientist because of their considerable potential to act as drug delivery vehicles by incorporating into a wide range of drug molecules (5).

The MEs have been shown several advantages for drug delivery such as ease of preparation, increasing drug solubility, controlling drug delivery rate, perfect stability, and improving bioavailability of hydrophilic and lipophilic drugs through different delivery routes (6).

Lithium salts were first used to treat gout. It was noted in the 1880s that lithium was somewhat effective in the treatment of depression and in the 1950s, lithium was seen to improve the symptoms of bipolar disease. Despite extensive researches, the underlying mechanisms of its therapeutic actions have not been fully elucidated;

Copyright © 2015, School of Pharmacy, Ahvaz Jundishapur University of Medical Sciences. This is an open-access article distributed under the terms of the Creative Commons Attribution-NonCommercial 4.0 International License (http://creativecommons.org/licenses/by-nc/4.0/) which permits copy and redistribute the material just in noncommercial usages, provided the original work is properly cited.

however, this is probably related to its effects on other electrolytes such as sodium, potassium, magnesium, and calcium (7, 8).

Today lithium carbonate (LC) is widely used in the treatment and prevention of manic-depressive and bipolar disorders. However, this medication has a low therapeutic index and therefore, many attendant side effects (9-11). A number of clinical studies showed that prolonged treatment with therapeutic levels of LC has a toxic effect on some patients, causing acne form eruptions, sinus node dysfunction, hyperparathyroidism, hypothyroidism, thyrotoxicosis, renal toxicity, and diabetes insipidus (12, 13). Furthermore, acute LC intoxication affects predominantly the central nervous system and the renal system and is potentially lethal (11).

# 2. Objectives

The goal of the present study was to evaluate potential hepatotoxic and nephrotoxic effects of oral administration of LC microemulsion (LCME) in mice model.

# 3. Materials and Methods

## 3.1. Chemicals

The LC powder was purchased from Daroupakhsh company (Iran). Tween 80 and propylene glycol (PG) used for this experiment were obtained from Merck Company, Germany. Sesame oil was obtained from Sabz Kooh, Iran. All chemicals and solvents used in this study were of analytical grade and obtained through commercial sources.

## 3.2. Animals

Male Swiss albino mice (age, 6 - 8 weeks old; weight, 25 - 30 g), were obtained from the animal house of Jundishapur University of Medical Sciences, Ahvaz, Iran. Mice were maintained in 12/12-hour light/dark cycles and had access to regular chow food and drinking water ad libitum. Animal were acclimated to the environment for a minimum of one week prior to the experiment. All experimental procedures were conducted according to the ethical codes and protocols approved by the Committee of Animal Experimentation of Ahvaz Jundishapur University of Medical Sciences, Iran (Ref. No. B-9333).

# 3.3. Microemulsion Preparation Method

To examine concentration area of components for the existing boundary of ME dots, pseudo-ternary phase diagrams were constructed using the water titration method. Phase diagram was prepared with the 3:1 weight ratios of Tween 80 Propylene glycol. Oil phase (Sesame Oil) and the surfactant/surfactant mixture were then mixed at the weight ratios of 1:9, 2:8, 3:7, 4:6, 5:5, 6:4, 7:3, 8:2, and 9:1. These mixtures were diluted drop-wise with double distilled water, under moderate agitation. The samples were classified as MEs when they appeared as clear liquids. The components of the convenient formulations are 2.43% oil, 66.55% surfactant/co-surfactant, 31.02% water, and 0.5%, 1%, and 2% of LC, respectively. LC was added to the mixtures of oil and surfactant/co-surfactant and then an appropriate amount of distilled water was added to the mixture drop by drop and the MEs containing LC were obtained by stirring the mixtures at ambient temperature (14).

The stability of MEs was assessed regarding the temperature stability and centrifugation. The LCME were kept in different temperature (4°C, 25°C, and 37°C) and observed for flocculation, precipitation, and phase separation. Moreover, MEs were centrifuged by high-speed centrifuge (MPV-350R, Poland) at 10000 rpm for 30 minute at 25°C and were evaluated for any change in their homogeneity. Finally, the ME with the highest stability and better mean droplet size was selected and used in the experiments (15).

## 3.3.1. Viscosity Measurement

The viscosities of MEs were measured at 25°C with a Brookfield viscometer (DV-II + Pro Brookfield,) using spindle No. 34 with a shear rate of 100 rpm (14).

## 3.3.2. Particle Size Determination

The mean droplet size of MEs was determined by Scatter Scope 1 Quidix (South Korea) at 25°C (15).

## 3.3.3. The pH Measurement

The pH values of the MEs were obtained by pH-meter (Mettler Toledo Seven Easy, Switzerland) at 25°C.

# 3.4. Experimental Design

Animals were randomly allocated to eight groups of 10 mice to receive assigned treatments orally for ten days. Group 1, as negative control group, received 0.9% saline; group 2 received ME without drug as blank group; groups 3 to 5 received LC solution in doses of 50, 100, and 200 mg/kg, respectively. Groups 6 to 8 received LCME through oral gavage in doses of 50, 100, and 200 mg/kg, respectively. Then on the day 11, 24 hours after the last administration, animals were killed. Blood was withdrawn by cardiac puncture to determine serum glutamate pyruvate transaminase (SGPT), glutamate oxaloacetate transaminase (SGOT), alkaline phosphatase (ALP), blood urea nitrogen (BUN), and creatinine (Cr) levels. Liver and kidney were removed for histopathologic examination.

# 3.5. Statistical Analysis

The collected data were analyzed using SPSS 18.0 (SPSS Inc, Chicago Illinois, the United States). All results were expressed as the mean  $\pm$  SD. The data obtained by one-

way ANOVA followed by Tukey's post hoc test. The level of significance was set at P < 0.05.

## 4. Results

#### 4.1. Microemulsion Results

The components of suitable formulation were as follows: Oil, 2.43%; surfactant/co-surfactant, 66.55%; and water, 31.02% (Figure 1). The MEs formulations had the mean droplet size of  $34.5 \pm 1.2$ ,  $6.32 \pm 0.7$ , and  $22.1 \pm 0.09$  nm for 0.5%, 1%, and 2% solutions, respectively. The ME formulations had the mean viscosity of  $88.6 \pm 4.5$  cps and mean pH of  $5.7 \pm 0.1$ .

#### 4.2. Biochemical Results

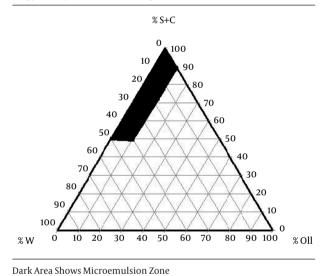
The effects of treatment with different doses of LC and LCME on biochemical parameters of liver and kidney are shown in Table 1. Nephrotoxicity and renal damages is usually associated with increased Cr and BUN levels. Data showed that, there was no difference in the level of Cr between the control group and microemulsion base group. Groups 3 through 8, which were treated with LC and LCME, showed increase in the level of Cr in all doses.

Administration of LC and LCME in all doses resulted a significant increase in the levels of BUN and Serum activity of SGOT and SGPT in comparison to normal saline group (P < 0.05). Serum activity of ALP, SGOT, and SGPT and levels of BUN and Cr in ME base group were greater than those in normal saline group were. However, this difference was not significant.

## 4.3. Results of Histopathologic Examination

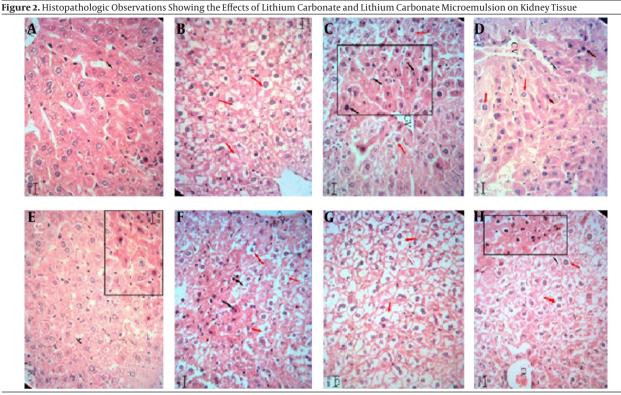
The hepatotoxic and nephrotoxic effect of the LCME was evaluated by the histopathologic examination of the kidney and liver. The histopathologic study of liver in the negative control group with normal saline showed a normal hepatic architecture with distinct hepatic cells, sinusoidal spaces, and a central vein (Figure 2 A). The ME base-treated group exhibited ballooning degeneration and hepatocyte swelling (Figure 2 B). In LC-treated group (50 mg/kg), cell swelling, several foci of coagulation necrosis, and different stages of necrotic cell nucleus were observed (Figure 2 C). Treatment with doses of 100 and 200 mg/kg of LC showed cell swelling with focal coagulation necrosis of hepatocytes (Figure 2 D, 2E). LCM-treated group (50 mg/kg) showed cell swelling in the hepatocytes with greater intensity and multiple foci of coagulation necrosis (Figure 2 F). The LCM-treated group (100 mg/kg) showed cell swelling with high intensity in hepatocytes (Figure 2 G). Cell swelling and necrosis were observed in LCM-treated group (200 mg/kg) (Figure 2 H). The histopathologic study of kidney in the negative control group and the second group with only ME base showed a normal architecture. In these groups, the structure of glomuses was clear, capsular space were small, the structure of epithelial cells in proximal convoluted tubules and distal convoluted tubules was normal, and the boundaries of the visceral layer and parietal layer of renal capsule were clear (Figure 3 A, 3B). In LC-treated group (50 mg/kg), swelling in the living cells of the proximal tubular and coagulation necrosis were seen (Figure 3 C). In LC-treated group (100 mg/kg), cell swelling, associated with protein deposits in the renal tubular lumen, was observed (Figure 3 D). In LC-treated group (200 mg/kg), urinary protein, deposits, and glomerulonephritis were seen in the lumen of the tube (Figure 3 E). In LCM-treated group (50 mg/kg), cell swelling and coagulation necrosis of proximal tubular cells were observed (Figure 3 F). Groups treated with 100 and 200 mg/kg of LCME showed glomerulonephritis and urinary protein deposits (Figure 3 G, 3H).

Figure 1. The Pseudo-Ternary Phase Diagram of the Oil-Surfactant/Cosurfactant Mixture-Water System at the 3:1 Weight Ratio of Tween 80 Propylene Glycol at Ambient Temperature



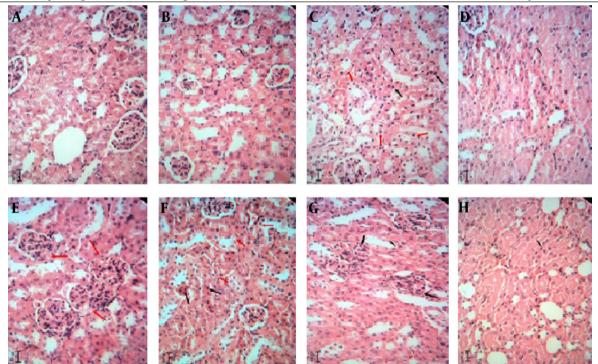
#### 5. Discussion

Liver is one of the largest organs in human body and the major site for metabolism and excretion. It has a wide range of functions including detoxification, protein synthesis, and production of biochemicals necessary for digestion (16). On the other hand, kidney plays a primary role in the excretion of metabolic wastes and in the regulation of extracellular fluid volume, electrolyte composition, and acid-base balance (17). Toxic effects medications on the liver and kidney are common and expected, and could disrupt any or all of these functions (18). Very small particles have the ability to enter, accumulate within, and damage living organisms. This ability results primarily from their small size, which allows them to penetrate physiologic barriers and travel within the circulatory systems (19). Earlier studies have shown that nano-forms of different particles are more toxic than their micro-counterparts after acute exposure via the oral



Liver sections stained with Hematoxylin and Eosin, magnification × 100. A, Normal saline-treated group; B, microemulsion base-treated group; C, D, and E, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectivel

Figure 3. Histopathologic Observations Showing the Effects of Lithium Carbonate and Lithium Carbonate Microemulsion on Kidney Tissue



Kidney sections stained with Hematoxylin and Eosin, magnification × 100. A, Normal saline-treated group; B, microemulsion base-treated group; C, D, and E, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respectively showing groups treated with 50, 100, and 200 mg/kg of lithium carbonate; F, G, and H, respective

Groups	SGPT, µkat/L	SGOT, μkat/L	ALP, µkat/L	BUN, mmol/L	<b>Creatinine,</b> μ <b>mol/L</b>
Normal saline	$0.64\pm0.05$	$1.58\pm0.13$	$0.86\pm0.08$	$3.18\pm0.77$	37.13 ± 1.77
Base ME	$0.68\pm0.13$	$1.84\pm0.14$	$0.95\pm0.08$	$4.02\pm0.42$	$38.90\pm2.65$
LC					
50 mg/kg	$1.08 \pm 0.11$ <sup>C</sup>	$2.15 \pm 0.13$ <sup>c</sup>	$1.08\pm0.07$	$5.88 \pm 0.74$ <sup>c</sup>	$41.55 \pm 2.65$
100 mg/kg	$1.28\pm0.12$ c, d	$2.51 \pm 0.15$ c	$1.15 \pm 0.07$ <sup>c</sup>	$6.67 \pm 0.54$ <sup>c</sup>	$47.54 \pm 2.65$ <sup>c</sup>
200 mg/kg	$1.64 \pm 0.16$ <sup>C</sup>	$2.82 \pm 0.12$ <sup>c,d</sup>	$1.28 \pm 0.10^{\text{ C}}$	$7.16 \pm 0.70$ <sup>C</sup>	$53.92 \pm 4.42$ <sup>C</sup>
LCME					
50 mg/kg	$1.39 \pm 0.12$ <sup>C</sup>	$2.42\pm0.10~^{\rm C}$	$1.12 \pm 0.09$ <sup>c</sup>	$6.25 \pm 0.43$ <sup>c</sup>	$45.08 \pm 2.65$ <sup>c</sup>
100 mg/kg	$1.64 \pm 0.10$ <sup>c,d</sup>	$2.76\pm0.13~^{\rm C}$	$1.20\pm0.11^{\text{ C}}$	$6.82 \pm 0.64$ <sup>C</sup>	$50.39 \pm 4.42$ <sup>C</sup>
200 mg/kg	$1.94 \pm 0.14$ <sup>C</sup>	3.18 ± 0.16 <sup>c,d</sup>	$1.34\pm0.08~^{\rm C}$	$8.19\pm0.38~^{\rm C}$	$59.23 \pm 2.65$ <sup>c</sup>

Table 1. Effects of the Treatment With Lithium Carbonate and Lithium Carbonate Microemulsion on the Activity and Concentration of Serum Markers of Liver and Kidney Function <sup>a, b</sup>

а Abbreviations: ALP, alkaline phosphatase; BUN, blood urea nitrogen; LC, lithium carbonate; LCM, lithium carbonate microemulsion; ME, Microemulsion; SGOT, serum glutamate oxaloacetate aminotransferase; and SGPT, serum glutamate pyruvate aminotransferase. <sup>b</sup> Data are presented as mean  $\pm$  SD for n = 10.

<sup>C</sup> Significant different with the control (P < 0.05).

<sup>d</sup> Significant different with the corresponding group (P < 0.05).

route (20).Biodistribution experiments have determined liver, kidney, and spleen as the target organs for engineered nanoparticles after absorption by the gastrointestinal tract (20). This study represented the first attempt to delineate the potential hepatotoxic and nephrotoxic effects of LCME formulations following subacute oral administration. The authors evaluated liver and kidney through biochemical parameters and histopathologic examination to detecting organs abnormalities after LCMEuptake by the gastrointestinal tract. The results of present study indicated that LC- and LCME-induced subacute liver and kidney toxicity in mice, shown by an increase in serum liver enzyme activities and kidney biomarker (Table 1). Moreover, the liver and kidney histopathologic findings confirmed the hepatotoxic and nephrotoxic effects of LC and LCME by demonstrating centrilobular necrosis, fatty changes (steatosis), and scattered lymphocytes infiltrate in hepatic parenchyma as well as glomerulonephritis and urinary protein deposition in kidney. Although the intensity of injury on histopathologic examination was slightly more severe in LCME groups (6-8), there was no significant difference between LC and LCME groups when biochemical parameters were considered (Table 1).

Moreover, group 2 that received only ME base (once daily, for 10 days) did not have any significant differences with negative control group based on biochemical parameters (AST, ALT, ALP, BUN and Cr). In addition, only mild inflammation and hepatocyte swelling were seen in histopathologic observation of liver and there were no histopathologic changes in kidney tissue (Figure 2 B, 3B).

In similar study, Pouladi et al. investigated a low dose of LCME to improve motor function and rescues striatal pathology and its toxicity in the YAC128 mouse model of Huntington's disease. They treated mice with 0, 20, or 40  $\mu$ g/kg of LC in the form of NP03 ME, for ten months. Their

findings demonstrated multiple positive effects of lowdose LC in the form of NP03 ME on behavioral and neuropathologic endpoints in an animal model. Additionally, there were no obvious toxic adverse effects associated with long-term LC treatment in YAC128 or WT animals (21).

In the study by Keown et al. the safety and tolerability of the cyclosporine ME (Neoral), the incidence of acute rejection episodes, and graft survival were compared to the conventional cyclosporine formulation (SIM) in patient with renal transplantation. Their finding revealed no difference between two formulations in the safety profile or tolerability, kidney function, and other laboratory parameters throughout the study. They concluded that the pharmacokinetic characteristics of the cyclosporine ME formulation lead to more efficient immunosuppressant after transplantation without any deleterious effect on clinical safety (22).

In conclusion, our results demonstrated that subacute oral administration of different doses of LCME causes slightly more severe signs of toxicity in comparison to LC. Administration of increasing doses of LC and LCME led to increased levels of biochemical parameters and histopathologic changes. Although the intensity of toxicity was higher in LCME, the difference was not significant. Future studies are suggested, particularly for use of LCME in the pharmaceutical product.

#### Acknowledgements

This paper was extracted from PharD thesis of Fereshteh Jazaveri Shushtari.

## Authors' Contributions

All authors were involved in all steps of the manuscript preparation.

## **Funding/Support**

The work was financially supported by the Arvand International Institute, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, Iran.

#### References

- Orive G, Hernández RM, Gascón AR, Pedraz JL. Micro and nano drug delivery systems in cancer therapy. *Cancer Therapy*. 2005;3(1):131-8.
- D'Addio SM, Prud'homme RK. Controlling drug nanoparticle formation by rapid precipitation. *Adv Drug Deliv Rev.* 2011; 63(6):417-26.
- Zeng J, Lee J. Effects of preparation conditions on performance of carbon-supported nanosize Pt-Co catalysts for methanol electro-oxidation under acidic conditions. *J Power Sour.* 2005; 140(2):268–73.
- Raghuveer V, Ferreira P, Manthiram A. Comparison of Pd–Co–Au electrocatalysts prepared by conventional borohydride and microemulsion methods for oxygen reduction in fuel cells. *Electrochem Commun.* 2006;8(5):807-14.
- 5. Lawrence MJ, Rees GD. Microemulsion-based media as novel drug delivery systems. *Adv Drug Deliv Rev.* 2000;**45**(1):89–121.
- Ghosh PK, Murthy RS. Microemulsions: a potential drug delivery system. Curr Drug Deliv. 2006;3(2):167–80.
- 7. Hamamura T, Lee Y, Ohashi K, Fujiwara Y, Miki M, Suzuki H, et al. A low dose of lithium chloride selectively induces Fos protein in the central nucleus of the amygdala of rat brain. *Prog Neuropsychopharmacol Biol Psychiatry*. 2000;**24**(2):285–94.
- Imam SK, Hasan A, Shahid SK. Lithium-induced nephrogenic diabetes insipidus. J Pak Med Assoc. 2005;55(3):125–7.
- Simard M, Gumbiner B, Lee A, Lewis H, Norman D. Lithium carbonate intoxication. A case report and review of the literature. *Arch Intern Med.* 1989;**149**(1):36–46.
- Flaherty B, Krenzelok EP. Neonatal lithium toxicity as a result of maternal toxicity. Vet Hum Toxicol. 1997;39(2):92–3.
- 11. Oktem F, Ozguner F, Sulak O, Olgar S, Akturk O, Yilmaz HR, et al.

Lithium-induced renal toxicity in rats: protection by a novel antioxidant caffeic acid phenethyl ester. *Mol Cell Biochem.* 2005;**277**(1-2):109–15.

- Lewicki M, Paez H, Mandalunis PM. Effect of lithium carbonate on subchondral bone in sexually mature Wistar rats. *Exp Toxicol Pathol*. 2006;**58**(2-3):197–201.
- Thakur SC, Thakur SS, Chaube SK, Singh SP. Subchronic supplementation of lithium carbonate induces reproductive system toxicity in male rat. *Reprod Toxicol.* 2003;17(6):683–90.
- Salimi A, Sharif Makhmal Zadeh B, Moghimipour E. Preparation and characterization of cyanocobalamin (vit B12) microemulsion properties and structure for topical and transdermal application. *Iran J Basic Med Sci.* 2013;16(7):865–72.
- Moghimipour E, Salimi A, Eftekhari S. Design and characterization of microemulsion systems for naproxen. Adv Pharm Bull. 2013;3(1):63–71.
- Forouzandeh H, Azemi ME, Rashidi I, Goudarzi M, Kalantari H. Study of the Protective Effect of Teucrium polium L. Extract on Acetaminophen-Induced Hepatotoxicity in Mice. *Iran J Pharm Res.* 2013;**12**(1):123-9.
- 17. Smith HW. The Kidney: Structure and Function in Health and Disease.New York: Oxford University Press; 1951.
- Choudhury D, Ahmed Z. Drug-associated renal dysfunction and injury. Nat Clin Pract Nephrol. 2006;2(2):80-91.
- Wang J, Zhou G, Chen C, Yu H, Wang T, Ma Y, et al. Acute toxicity and biodistribution of different sized titanium dioxide particles in mice after oral administration. *Toxicol Lett.* 2007;168(2):176–85.
- 20. Dhawan A, Sharma V. Toxicity assessment of nanomaterials: methods and challenges. *Anal Bioanal Chem.* 2010;**398**(2):589– 605.
- 21. Pouladi MA, Brilluad E, Xie Y, Franciosi S, Zhang WN, Zapala M, et al. B13 NP03, a low dose lithium microemulsion, improves motor function and rescues striatal pathology without toxicity in the YAC128 mouse model of Huntington's disease. *J Neurol Neurosurg Psychiat*. 2010;**81**(Suppl 1).
- Keown P, Niese D. Cyclosporine microemulsion increases drug exposure and reduces acute rejection without incremental toxicity in de novo renal transplantation. International Sandimmun Neoral Study Group. *Kidney Int.* 1998;**54**(3):938–44.