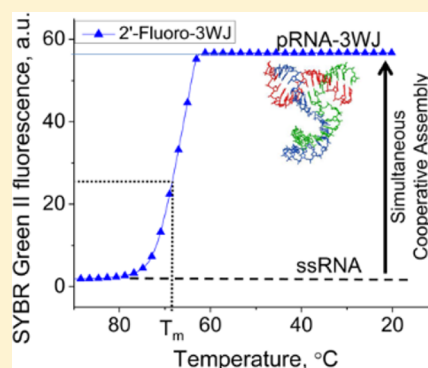


Entropy-Driven One-Step Formation of Phi29 pRNA 3WJ from Three RNA Fragments

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ABSTRACT: The emerging field of RNA nanotechnology necessitates creation of functional RNA nanoparticles but has been limited by particle instability. It has been shown that the three-way junction of bacteriophage phi29 motor pRNA has unusual stability and can self-assemble from three fragments with high efficiency. It is generally believed that RNA and DNA folding is energy landscape-dependent, and the folding of RNA is driven by enthalpy. Here we examine the thermodynamic characteristics of the 3WJ components as 2'-fluoro RNA, DNA, and RNA. It was seen that the three fragments existed either in 3WJ complex or as monomers, with the intermediate of dimers almost undetectable. It seems that the three fragments can lead to the formation of the 3WJ complex efficiently within a rapid time. A low dissociation constant (apparent K_D) of 11.4 nM was determined for RNA, inclusion of 2'-F pyrimidines strengthened the K_D to 4.5 nM, and substitution of DNA weakened it to 47.7 nM. The ΔG°_{37} were -36 , -28 , and -15 kcal/mol for 3WJ $_{2'-F}$, 3WJ $_{RNA}$, and 3WJ $_{DNA}$, respectively. It is found that the formation of the three-component complex was governed by entropy, instead of enthalpy, as usually found in RNA complexes.



Since the proof-of-concept in 1998,¹ RNA nanotechnology has been emerged as a popular field.^{2–10} RNA has the simplistic chemical characteristics of DNA with the complex folding and functionality of proteins.² These attributes make RNA an ideal candidate for creating nanoparticles with diverse functionalities for targeting and treating cancer tumors and viral infections, as well as other applications in nanodevices.^{1,5,11–21} RNA nanotechnology in therapeutics provides many advantages over current technologies:^{2,22} (1) RNA can be produced with a defined shape and stoichiometry as well as high reproducibility.^{1,5,12,23} There are many modified nucleosides within RNA, but it is primarily composed of only four nucleic acid bases, allowing for simplicity in structure and predictable interactions between molecules and formation of structures. (2) RNA can target specific cell groups by targeting cell surface receptors through the use of RNA aptamers that function like protein or chemical ligands.^{24–26} These structures do not induce antibody production, allowing for repeated delivery and therapy.²⁴ (3) RNA nanoparticles that have been produced have a size range of 10–50 nm,^{27–29} the perfect size to be retained within the body and pass through leaky blood vessels in cancer tumors,^{30,31} as well as cell membranes by cell surface receptor endocytosis.²⁹ (4) RNA can be created to harbor multiple therapeutic elements by utilizing branch scaffolds^{21,32–35} and bottom-up construction.^{5,36} Even with these advantages, RNA nanotechnology has been hindered because of the instability of RNA itself, specifically *in vivo*. Dissociation of complex without covalent bonds is an intrinsic property of molecules, for example, RNA molecules, that are thermodynamically unstable in nature.^{37–45}

When RNA nanoparticles are delivered systemically to the body, the particles will exist in low concentrations due to dilution

by circulating blood. Only those RNA particles that do not dissociate at low concentrations are feasible for therapeutic purposes that require systemic delivery. Furthermore, RNA can easily be degraded and cleaved by RNases found throughout the human body.²² Chemically modified nucleotides have been developed to combat the nuclease degradation. Specifically, 2'-F modified nucleotides have been shown to keep the original folding and functionality of the RNA molecules while significantly increasing the half-life.⁴⁶ To overcome the instability issues and push the RNA nanotechnology field to progress further, a stable platform needs to be produced that can remain stable at low concentrations and high temperatures while resisting RNase degradation.^{12–14}

Recently, a three way junction motif (3WJ) in the packaging RNA (pRNA) of the bacteriophage phi29 dsDNA packaging motor was found to be ultrastable.¹² The pRNA-3WJ produced a melting curve indicative of a low Gibbs free energy (ΔG°),^{47,48} as well as a high melting temperature (T_m). It has also been elucidated that the 3WJ is stable in ultralow concentrations, as well as in 8 M urea. This junction serves as the central core of the pRNA linking the helical domain⁴⁹ to the interlocking looped regions⁵⁰ and allows for intermolecular interactions with other pRNA molecules (Figure 1). The core can be formed from three individual RNA oligos with a high efficiency without the presence of metal ions. It has been found that this junction can incorporate RNA functional moieties, such as receptor-binding aptamer,^{51,52} siRNA,^{33,36,53} ribozyme,^{54–56} miRNA,^{57–59} or riboswitch.^{60,61}

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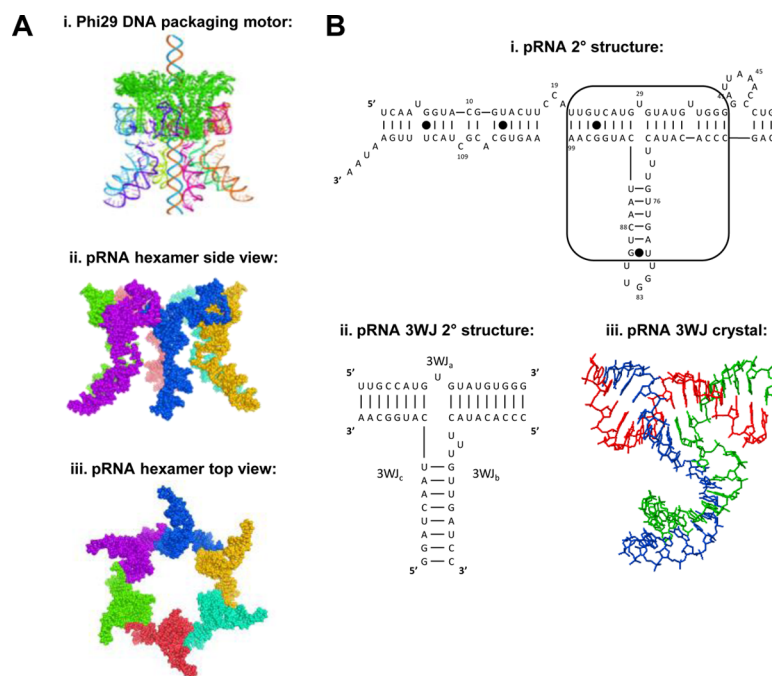


Figure 1. Background information for motor pRNA, pRNA three-way junction (3WJ), and the application of their thermodynamic stability to build RNA nanoparticles. (A) The side-view of the hexameric structure of phi29 DNA packaging motor.⁷⁹ Two bottom panels are side-view and top-view of the hexameric pRNA derived from X-ray crystallography.⁸⁰ (B) Predicted secondary structure of the phi29 packaging RNA (pRNA) with the 3WJ motif outlined and the pRNA-3WJ and pRNA-3WJ secondary structure with solved crystal structure.

Additionally, the resulting structures have the ability to keep the strong folding of the core, while retaining the functionality of the conjugated RNA moieties.¹² Even with the strong stability of the 3WJ, it is still very much susceptible to degradation by RNases; therefore, chemical modifications to the RNA are required.^{40,62,63} The effects on the thermodynamic stability of the pRNA-3WJ core, though, are still unknown.

Previously, thermodynamics of nucleic acids and their folding properties have been studied; however, the majority of the studies have been completed on RNA and DNA duplex sequences.^{38,41–43,64–67} An estimation of thermodynamic parameters for RNA 3WJ and 4WJ have been surmised⁶⁸ from studies completed using two piece designs by incorporating looped regions between helical branches. A gap has remained within the thermodynamic studies of nucleic acids regarding elucidating thermodynamic characteristics of multibranch structures, that is, multistranded motifs. Additionally, the understanding of chemical modifications to the RNA backbone on RNA junctions has remained a mystery, and untouched. To understand the thermodynamic characteristics of such structures and comprehend the governing laws of motif folding, new methods must be developed to be able to see multistranded interactions.

Here we report the measured thermodynamic parameters for 3WJ complexes containing DNA, RNA, and 2'-F U/C modified RNA strands (Figure 1), as well as hybrid complexes by means of comparison of their stability using a real-time polymerase chain reaction (rtPCR) machine and temperature gradient gel electrophoresis (TGGE). Results concluded that use of DNA strands weakened the structure of the pRNA-3WJ, while 2'-F modifications strengthened the RNA motif by elevating the transition temperatures and lowering ΔG° . More importantly, the data appear to show the 3WJ formed in a single step without displaying the presence of dimer species, showing that all three

strands of the pRNA-3WJ formed together into a complex rapidly with the intermediate product undetectable. The assembly was also revealed to be entropy driven.

MATERIALS AND METHODS

Oligonucleotides and Assembly of 3WJs. RNA and DNA oligonucleotides were obtained from Integrated DNA Technologies (IDT). RNA oligonucleotides containing 2'-F U/C modifications were ordered from Trilink BioTechnologies. Assembly of 3WJs was performed by mixing equimolar concentration of corresponding strands in TMS buffer (50 mM TRIS pH = 8.0, 100 mM NaCl, 10 mM MgCl₂), heating to +80 °C, and slowly cooling to +4 °C at a rate of 2.0 °C/min for a total of 37 min. The 3WJ formations were assayed on a 12% native PAGE TBM running buffer (89 mM Tris, 200 mM borate acid, and 5 mM MgCl₂) ran at 100 V at 4 °C for 90 min. Gels were stained with ethidium bromide (EB) and imaged using a Typhoon (GE).

K_D Measurements. Apparent equilibrium dissociation constants (K_D) for 3WJ formations were determined by titration over a range of concentrations from 0.1 nM to 512 nM, as previously described.^{23,69} Concentrations of the 3WJs were calculated using the absorbance of UV light at 260 nm using a Nanodrop 2000 (Thermo Scientific) using an optical density equaling one as 40 $\mu\text{g}/\text{mL}$ and 50 $\mu\text{g}/\text{mL}$ for RNA and DNA, respectively. Briefly, fixed amounts of [³²P] ATP 5'-end labeled 3WJ_b strands of RNA, DNA, and 2'-F modified RNA (0.1 nM final) were mixed with variable amounts of unlabeled 3WJ_a and 3WJ_c strands RNA to make the indicated final concentration (0.1 nM to 512 nM) of each. The resulting 3WJs were then heated to 80 °C for 3 min in TMS buffer and slowly cooled to 4 °C. The resulting gel shifts were measured using Image J software and interpreted with the program Origin 8.0. The fractions (f) for each trimer forming complex was calculated by dividing the

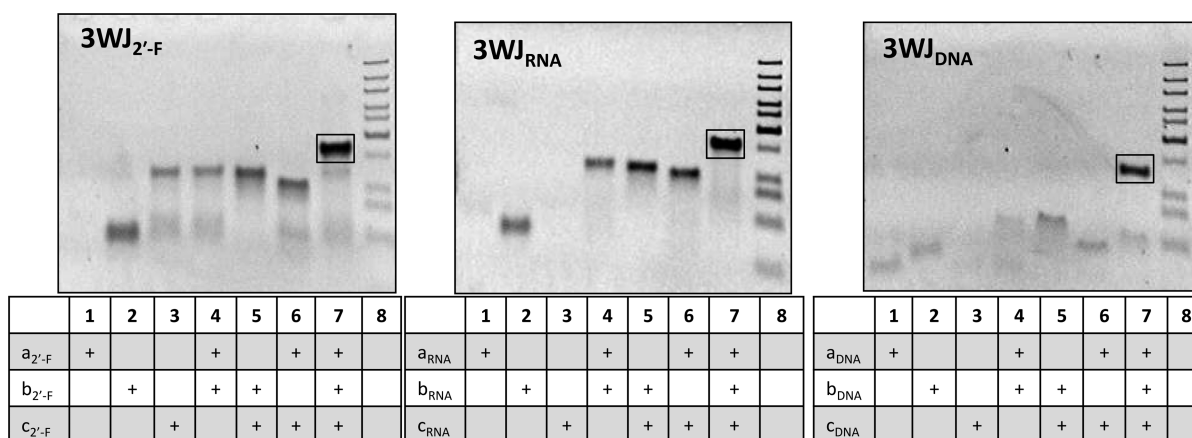


Figure 2. Assembly of pRNA-3WJ core structures. 12% polyacrylamide gels in native conditions displaying the stepwise assembly of the $3WJ_{2'-F}$, $3WJ_{RNA}$, and $3WJ_{DNA}$.

corresponding quantified values for trimers by the total sum of values for all complexes (monomer, dimer, and trimer) presented in the corresponding lane. The combined data from several independent measurements were subjected to nonlinear curve fitting using the equation:

$$K_D = \frac{\left[\left(\frac{C_t}{3} \right)^2 \times (1 - f)^3 \right]}{f^1} \quad (1)$$

where C_t is the total concentration of RNA strands in each lane, and f is the fraction of 3WJ complex to total concentration.

In calculating the C_v , the concentration of the labeled $3WJ_b$ strand was not included, as it was in trace amounts and negligible to the concentrations of $3WJ_a$ and $3WJ_c$. The C_t at equilibrium was calculated by interpolation of the fraction (f) at 50%. These values were then in turn used to calculate the equilibrium K_D s.⁴⁸ Marky et al. described this calculation using changing temperature to reach equilibrium, thus requiring equal concentrations of each strand; however, equilibrium was reached here by varying concentrations, as previous performed,^{23,69} thus not requiring equal concentrations of each strand.

Real-Time PCR Annealing Curves of 3WJ Complexes.

Each synthesized pRNA-3WJ strand ($3WJ_a$, $3WJ_b$, $3WJ_c$) was mixed at equimolar concentrations in the presence of 1× SYBR Green II dye (Invitrogen) at equal molar ratios of 10 μ M to create a $3WJ_{RNA}$, $3WJ_{DNA}$, and $3WJ_{2'-F}$. SYBR Green II dye is a reporter dye with higher specificity for RNA but shows binding to both DNA and RNA bases.⁷⁰ Using a Roche Lightcycler 480 real-time PCR machine, with accurate readings of temperatures to $\pm >0.25$ °C, samples were heated to 95 °C for 5 min for an initial denaturing period and then slowly cooled at a rate of 0.11 °C/s until 20 °C. The Roche Lightcycler 480 was able to detect the fluorescence levels and thus monitor the formation of the 3WJ cores. All samples were completed in triplicate to ensure accuracy of the annealing temperatures and profiles. Samples were then subjected to electrophoresis on a 12% polyacrylamide gel to ensure the formation of the 3WJs. All hybrid structures of the pRNA-3WJ core were created using a mix of RNA/DNA, RNA/2'-F RNA, or DNA/2'-F RNA in the same method described as above.

TGGE and Thermodynamic Parameters. All 3WJs were assembled as described above. The TGGE system from BiometraHmbGh, Germany, was used in this study. Temperature was varied perpendicular or parallel to the electrical current.

All experiments were performed in TMS running buffer. The RNA bands were detected in native 15% gel by total RNA stain using EB. The concentrations of 3WJs were typically 10 μ M. All varieties were run a minimum of three times to ensure accuracy of measurement.

Various concentrations of preassembled pRNA-3WJs covering 1000-fold dilutions from 10 μ M to 0.016 μ M were subjected to the perpendicular TGGE for thermodynamic parameter calculations. 3WJ bands were detected with radiolabeled $3WJ_c$ strands at the 5'-end using γ -³²P ATP (Perkin-Elmer). A linear temperature gradient was applied perpendicular to the electric field. The gels were run for 1 h at constant 80 V, dried, and exposed to phosphorimaging screen overnight. Gels were then visualized on a Typhoon (GE), and Image J software was used to quantify the area of bands on each lane by plotting intensity of each band intensity and integrating the area under the curve for each band. Background signal of each lane immediately surrounding the band of interest from the gel was subtracted and removed from the band intensity and not included in calculations. The fractions of 3WJ bands were obtained by subtracting the area of melted bands from the trimer band. Melting temperatures were then calculated from a plot of 3WJ fraction (f) versus temperature, with $f = 0.5$ (50%) corresponding to the T_m .

Calculation of Thermodynamic Parameters. T_m values at different concentration of 3WJs were used to calculate thermodynamic parameters. van't Hoff plots were generated by plotting the T_m versus the concentrations of non-self-complementary three molecular strands according to a previous method:⁴⁸

$$\frac{1}{T_m} = \frac{2R}{\Delta H^\circ} \ln C_t + \frac{\Delta S^\circ - 2R \ln 6}{\Delta H^\circ} \quad (2)$$

where R is the gas constant 1.987 cal mol⁻¹ K⁻¹, C_t is the total concentration of each 3WJs, and ΔH° and ΔS° are enthalpy and entropy changes, respectively. Under this method of calculation, the assumption is made that the heat capacity remains unchanged throughout the entire melting profile.

Fluorescence of pRNA 3WJ-MG Aptamer. The assembled unmodified and 2'-F A/C modified 3WJs (1 μ M) were mixed with MG (2 μ M) in binding buffer containing 100 mM KCl, 5 mM MgCl₂, and 10 mM HEPES (pH 7.4) and incubated at room temperature for 30 min. The fluorescence was measured using a fluorospectrometer (Horiba Jobin Yvon; SPEX Fluolog-3),

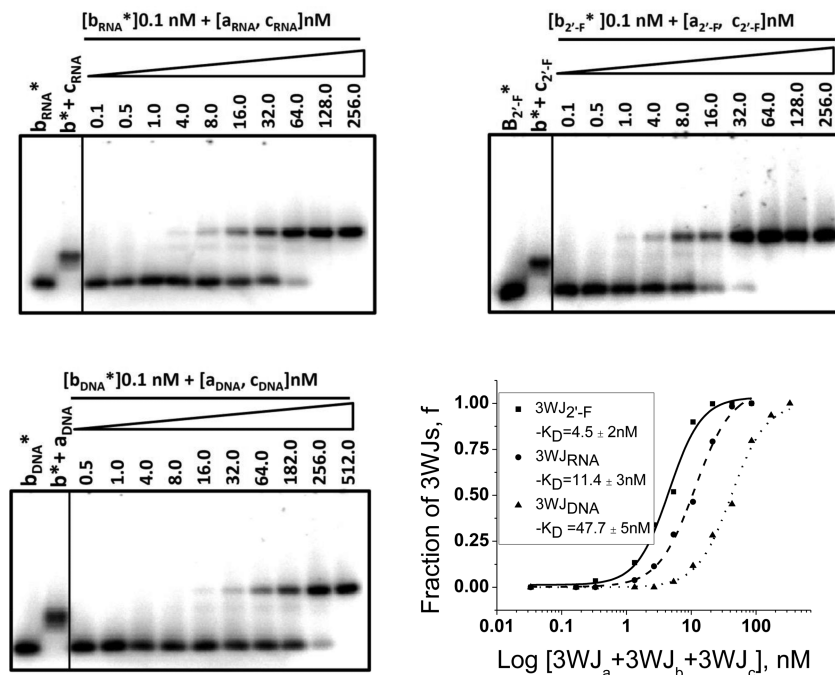


Figure 3. Dissociation constant measurements. Electrophoretic mobility shift assay (EMSA) of ³²P-labeled 3WJ_b (constant concentration) assembling with increasing concentrations of 3WJ_a and 3WJ_c for RNA, 2'-F RNA, and DNA on 12% polyacrylamide gels in native conditions. Plot of percentage of 3WJ formed versus total concentration to solve for *K_D* of each 3WJ.

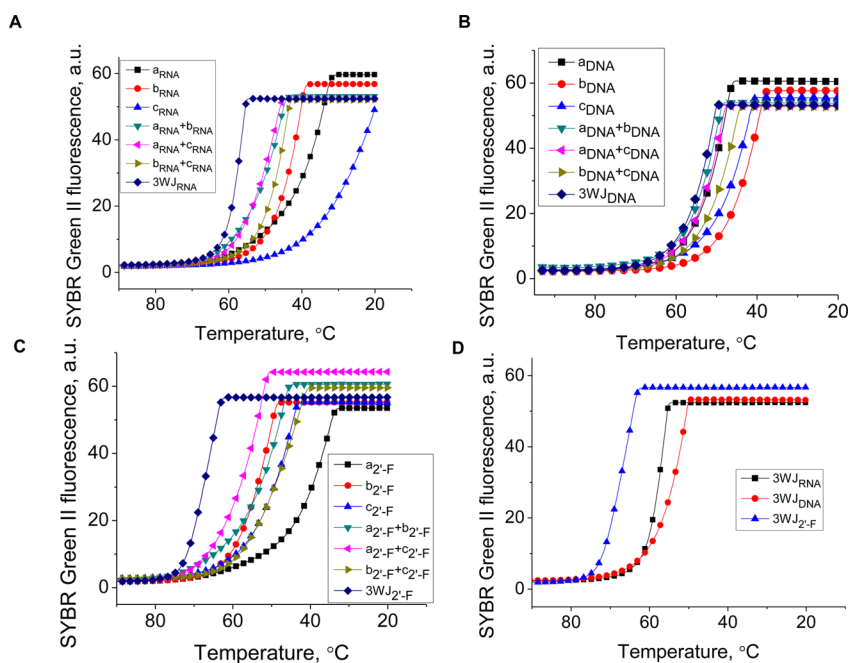


Figure 4. Assembly curves produced from the Roche 480 Lightcycler using SYBR green II reporter dye. (A) 3WJ_{RNA}, (B) 3WJ_{DNA}, (C) 3WJ_{2'-F}, and each of the components of the 3WJs. (D) The three 3WJ species directly compared, showing the differences in annealing temperatures (*T_a*).

excited at 615 nm, and scanning from 625 to 800 nm for emission.⁷¹

RESULTS AND DISCUSSION

Assembly of the Three-Way Junctions. pRNA three way junctions composed of RNA (3WJ_{RNA}), DNA (3WJ_{DNA}), and 2'-F U/C modified RNA (3WJ_{2'-F}) were assembled using equimolar concentrations of each strand of the 3WJ. The formation of the pRNA-3WJs in each of the species was confirmed by 12% native

PAGE (Figure 2). A stepwise assembly through the polyacrylamide gel was observed between the monomer to dimer and finally to the fully assembled trimer band indicating the formation of 3WJs. The assembled RNA, DNA, and 2'-F 3WJs products resulted in high yield (>90%). Any side product is assumed to be contributed to a slight mismatch in strand concentrations loaded during assembly, preventing all nucleic acid strands from forming into the 3WJ complex. The data are in

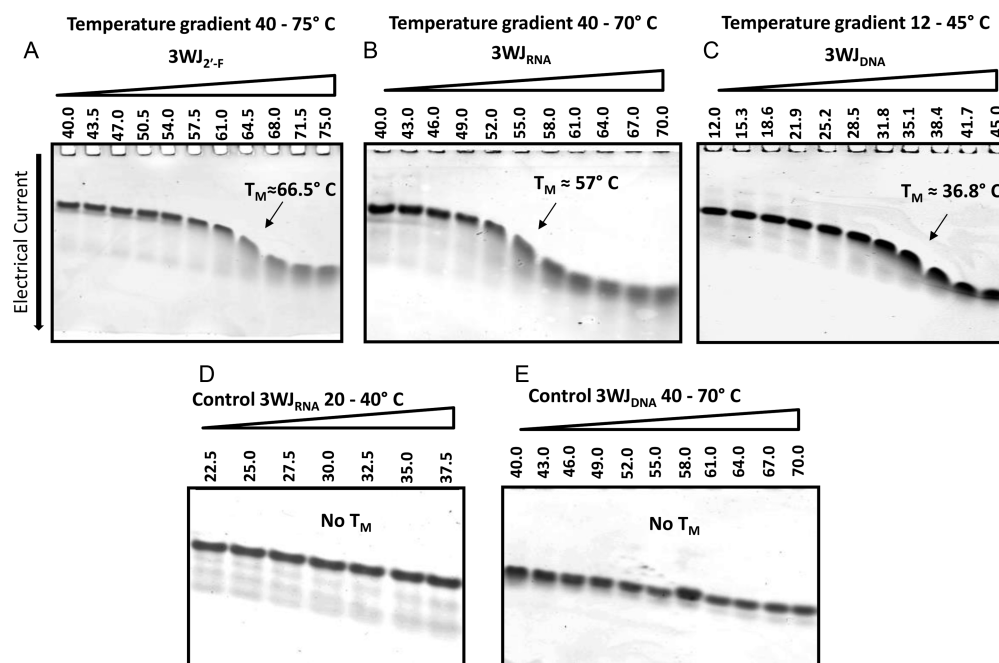


Figure 5. Representative 15% native TGGE with temperature gradient perpendicular to the electrical current. (A) 3WJ_{2'-F}, (B) 3WJ_{RNA}, and (C) 3WJ_{DNA}; total 3WJ concentration in each lane = 10 μ M. Panels (D) and (E) are control gels showing migration of 3WJ_{RNA} (D) before reaching its T_m in the temperature range of 20–40 °C and migration of 3WJ_{DNA} (E) in the temperature range of 40–70 °C that is over its T_m . The 3WJ bands were detected by total nucleic acid stain with EB.

agreement with previously reported results of RNA-3WJ assembly.¹²

Dissociation Constants (K_D) Measurements of the 3WJs. The K_D values of nucleic acids formation are directly related to their stabilities as more stable complexes require lower concentrations of the components for self-assembly resulting in a lower K_D value. We measured the apparent dissociation equilibrium constants for RNA, DNA, and 2'-F modified RNA 3WJ complexes (Figure 3). For the 3WJ_{RNA} and 3WJ_{2'-F}, the K_D values were found to be 11.4 and 4.5 nM, respectively. For the 3WJ_{DNA} complex, this value was about five times higher (47.7 nM). This indicates that 3WJ_{2'-F} and 3WJ_{RNA} were the most stable complexes, and the least stable was 3WJ_{DNA}.

These results demonstrated that the incorporation of enzymatically resistant DNA or 2'-F modified RNA strands into the RNA 3WJ motif could increase the resistance of the complex to RNases. However, while the DNA strand incorporation decreases the stability, the 2'-F modified RNA increases the stabilization of the 3WJ complex, as shown by the lower dissociation constant compared to the 3WJ_{RNA}. Therefore, the stability and properties of 3WJ_{RNA} motif can be potentially tuned by alternating the ratio of 2'-F modified RNA and DNA strands.

Determination of the Formation of Complex and T_a of DNA, RNA, and 2'-F RNA-3WJs by rtPCR Machine. The thermostability of 3WJs complexes were compared by measuring their fluorescence intensities as a function of temperature in the presence of SYBR Green II dye on a Roche 480 Lightcycler to obtain T_a .¹² The annealing transitions in Figure 4A–C show that the mixture of three strands (3WJ_a, 3WJ_b, and 3WJ_c) produced the highest annealing temperature compared to any monomer or dimer formation. Within the assembling profile, the slope of the transitions directly correlates with ΔG° , as the steeper slope results in more negative ΔG° values.^{47,48} These results showed that the assembly of the 3WJ was preferred over any dimer

formation of any two strands in each of the three species: RNA, DNA, and 2'-F modified RNA.

From each of the transitions obtained by the Roche 480 Lightcycler, the curves of the each of the completed 3WJ structures were compared (Figure 4D). The 3WJ_{2'-F} was the most stable with $T_a = 69.8$ °C, the 3WJ_{RNA} was the next stable at 59.3 °C, and the 3WJ_{DNA} had the lowest where $T_a = 48.9$ °C. These results correlated with literature reported values of overall thermostability for nucleic acids and followed the order of stability: 2'-F RNA > RNA > DNA.^{43,64,72}

Surprisingly, from the assembly profiles of the pRNA-3WJ species, a single annealing temperature, not two, was seen. This is evidenced by the single slope to the transition profile and hinting that the 3WJ forms rapidly with all three strands producing no byproducts, whereas a two-step association would display a plateau within the melting curve itself or two individual sloped regions, resulting in two annealing temperatures. These results show the three-stranded motif forming together, and a dimer species was undetectable due to the rapid 3WJ formation. This rapid formation from three fragments to form the pRNA-3WJ is highly beneficial because it allows a high yield of assembly while permitting the construction and assembly of RNA nanoparticles without creating side products as a result of the dimer formation.

Comparison of Stability between DNA, RNA, and 2'-F RNA-3WJs by TGGE. TGGE is common technique to measure T_m of large and complex nucleic acids.^{23,73,74} This approach has an advantage over the real-time PCR in that it can be applied to directly measure the T_m of RNA complexes as fractions of RNA versus temperature with no intercalation dye required.

Melting temperatures of 3WJ complexes were determined by measuring the decrease in the yield of 3WJs versus temperature (Figure 5). A temperature gradient was applied perpendicular to the electrical current, with an increasing temperature that resulted in the melting of the structures in the later lanes. Here, the percent 3WJ complex was compared to dimer and monomer

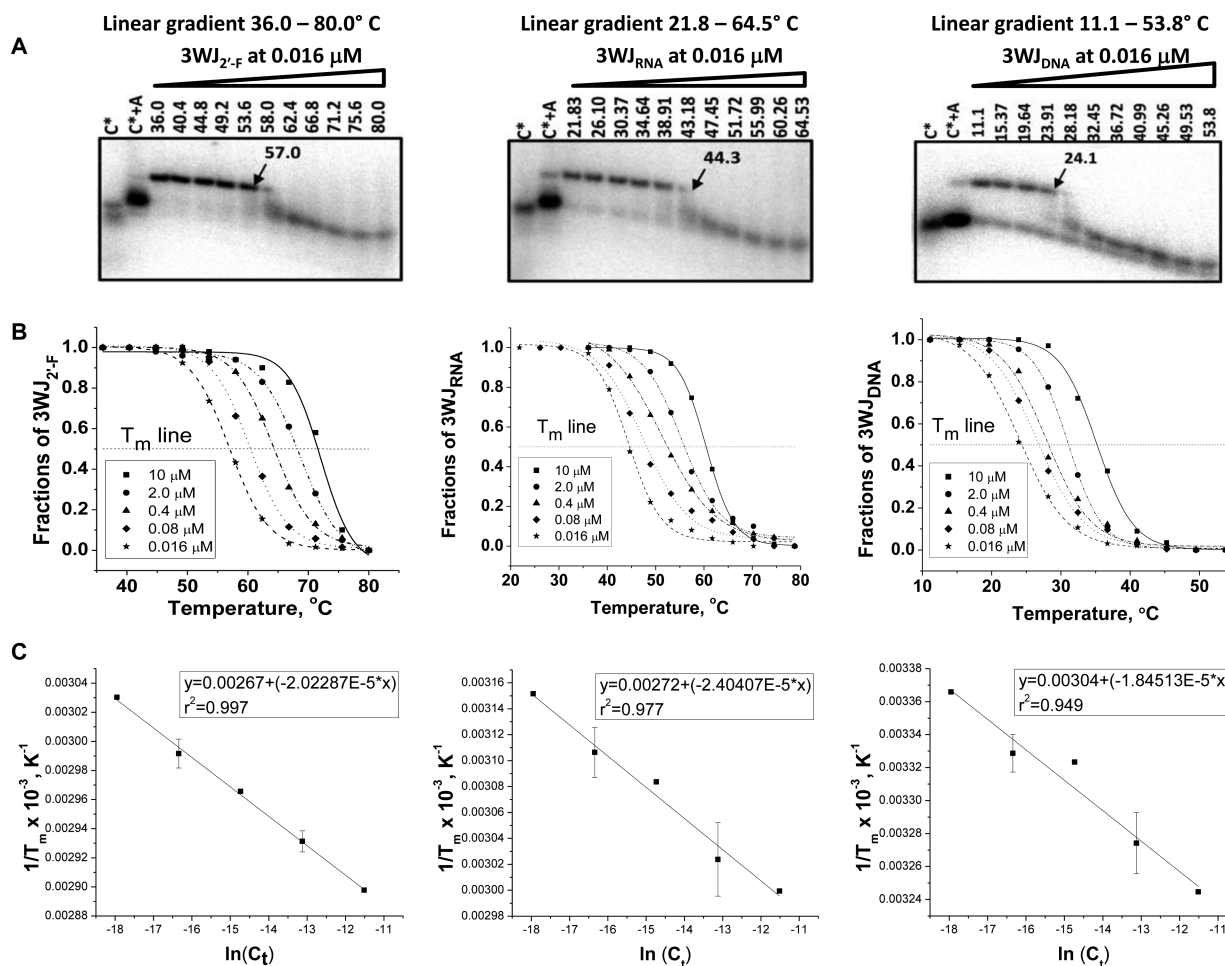


Figure 6. Calculation of thermodynamic parameter for 3WJs formation. (A) Representation of native 15% TGGE for the 3WJ complexes at the lowest concentration (0.016 μM in each lane). The radiolabeled C strand of corresponding 3WJ complexes indicated with asterisk “*”. Lanes 3WJ_c* and 3WJ_c*+3WJ_a served as controls for monomer and dimer location. (B) Melting temperature profiles of RNA, 2'-F RNA, and DNA 3WJs obtained after quantification of the corresponding band from the perpendicular TGGE at various concentrations. (C) Plots of T_m versus 3WJ concentrations (van't Hoff analysis) evaluated from melting curves of the 3WJ complexes obtained at different strand concentrations.

formations. T_m values were determined as 50% of 3WJ remaining. Any remaining dimers were not considered as a complex, as it was not complete 3WJ formations.

From the resulting gels, melting temperatures were derived for the 3WJ_{2'-F}, 3WJ_{RNA}, and 3WJ_{DNA} as 66.5 °C, 57 °C, and 35.2 °C, respectively. The melting temperatures for the two RNA species were within the range of error from the T_m found by the fluorescence melting curves, but there was a large difference in T_m between the two methods for the 3WJ_{DNA}. This difference between these two methods was presumably due to differing affinities of SYBR Green II to DNA and RNA stacking bases.⁷⁰ Nevertheless, both methodologies produced data that allowed direct comparison of the stability among 3WJ complexes and demonstrated the trend that 2'-F RNA had a higher melting temperature than RNA, and RNA had a higher T_m than DNA.

Within the TGGE melting gels, the dimer species were again undetected in the melting curves in the 2'-F RNA and RNA 3WJ species, further pointing to the possibility of the unusual assembly pathway. However, the 3WJ_{DNA} seemed to display a dimer intermediate, which was previously not seen from the PCR profiles. It is believed that this difference is because the TGGE showed the melting profile, while the PCR curves showed the association of the molecules. Therefore, the TGGE system of the DNA species was dissociation down to a dimer then a monomer,

possibly because DNA was not the natural species of the pRNA-3WJ and forced the motif into an unnatural conformation. Furthermore, any dimer species that is seen in radiolabeled gel analysis can be attributed to the presence of extreme excess unlabeled strands allowing for a mismatch in concentrations between each individual strand and were therefore ignored.

Thermodynamic Parameters for 3WJs Formation. Non-self-complementary nucleic acids usually exhibit a linear correlation T_m and RNA concentration, as the T_m increased with the increase in nucleic acid strands concentration.^{38,68,75} The TGGE approach was further applied to measure T_m of 3WJ species as a function of concentration using ~1000 fold dilution of the 3WJ_{RNA}, 3WJ_{2'-F}, and 3WJ_{DNA} complexes from 10 μM to 0.016 μM (Figure 6A). T_m's were calculated by finding the temperature at which 50% of the nucleic acids were in trimer formation compared to the total concentration of the bands observed. 3WJ T_m's were used to calculate thermodynamic parameters for three components of nonself complementary sequences, according to Marky et al.⁴⁸ The typical van't Hoff plots for 3WJ_{2'-F}, 3WJ_{RNA}, and 3WJ_{DNA} are represented in Figure 6C. From the obtained ΔH° and ΔS° parameters, the ΔG°₃₇ was calculated using eq 3:

$$\Delta G_{37}^{\circ} = \Delta H^{\circ} - (310.15 \text{ K})\Delta S^{\circ} \quad (3)$$

where ΔH° and ΔS° are enthalpy and entropy, respectively, and K is the abbreviation for Kelvin. All thermodynamic parameters are summarized in Table 1. Here the data produced a near linear

Table 1. Thermodynamic Parameters for 3WJs Formation^a

3WJs	1/T _m vs log (C _i)			T _m ^b (°C)
	ΔG°_{37} (kcal/mol)	ΔH° (kcal/mol)	ΔS°_{37} (e.u.)	
2'-F RNA	-36 ± 0.45	-200 ± 5.7	-520 ± 17	72.1
RNA	-28 ± 0.58	-170 ± 13	-440 ± 39	60.4
DNA	-15 ± 0.71	-220 ± 25	-650 ± 83	35.2

^aParameters derived from 15% native TGGE. ^bT_m values for 3WJ strand concentrations of 10⁻⁶ M.

trend lines with *r*² values very near 1.00. Because this little error is seen in calculating melting temperatures, samples were repeated in a selective fashion due to the high number of gels that would be needed. The repeated experiments allowed for calculation of error in the thermodynamic parameters which remained low. The results indicated that the most thermodynamically stable was the 3WJ_{2'-F} ($\Delta G^\circ_{37} = -36$ kcal/mol) complex, followed by 3WJ_{RNA} ($\Delta G^\circ_{37} = -28$ kcal/mol), and the 3WJ_{DNA} ($\Delta G^\circ_{37} = -15$ kcal/mol). On the basis of the parameters of free energy change for RNA and 2'-F RNA-3WJ assemblies, it can be determined that the energy for these complexes were favored 2-fold compared to the 3WJ_{DNA}. The 3WJ_{DNA} displayed a more favorable decreased enthalpy value of $\Delta H^\circ = -220$ kcal/mol, compared to 3WJ_{2'-F} ($\Delta H^\circ = -200$ kcal/mol) and 3WJ_{RNA} ($\Delta H^\circ = -170$ kcal/mol). However, the comparison of entropy parameters resulted in a significant increase for the RNA and 2'-F RNA complexes. The two RNA species resulted in more negative free energy changes, yet had higher enthalpies when compared to DNA. This data combined with the increased entropy values of the RNA species indicated that the thermodynamic stabilities of the 3WJ_{RNA} and 3WJ_{2'-F} were entropy-driven (Table 1).

The findings of more negative ΔG° values for 2'-F RNA and RNA are consistent with other studies comparing helical DNA, RNA, and 2'-F RNA.^{43,64,65,72,76} However, the notion that was RNA entropically driven, compared to DNA in helical structures, appeared less unanimous as the majority report RNAs to be less entropically favored and that folding is normally driven by a lower enthalpy value.^{43,64,65,76} Here it is believed that the 3WJ_{DNA} provided a more rigid structure, producing a lower internal energy or ΔH compared to the RNAs; however, the flexibility of the RNAs provided more disorder, thus giving strong stability, ease of folding, and a more negative ΔG° . This entropy-driven assembly combined with the one-step assembly expresses the unusual thermodynamic characteristics of the pRNA-3WJ.

Analysis of 3WJ Hybrid Formations and Thermostability. The hybrid composition (2'-F RNA/RNA; RNA/DNA; DNA/2'-F RNA) within the RNA 3WJs are of great interest due to their ability to maintain the diverse functionality of structured RNA molecules, while incorporating the chemical stability of 2'-F RNA and DNA. To test for hybrid 3WJ viability, the 2'-F RNA/RNA; RNA/DNA; DNA/2'-F RNA hybrids of the 3WJs complexes were characterized by parallel TGGE and fluorescence annealing temperature experiments. Using the TGGE, a temperature gradient was applied in parallel to the electrical current (Figure 7). As the samples migrated through the gel, the temperature increased from 20 to 70 °C, therefore melting the hybrid structures as they migrated further into the gel. A less stable 3WJ complex migrates further as the elevated temperature melts the structure to smaller single strands, thus causing an increased rate of migration and separating the stable hybrids from unstable hybrids.

The TGGE analysis demonstrates that each hybrid structure forms correctly and is stable at lower temperature ranges 20–40 °C (A). However, at a higher range of 40–70 °C, the RNA/DNA and DNA/2'-F RNA-3WJ hybrid structures melted, resulting in a more rapid migration rate compared to 2'-F RNA/RNA hybrids (Figure 7B). This direct comparison between hybrid stability demonstrates weakness in the thermostability of hybrids involving DNA strands. In addition, the 3WJs hybrids followed

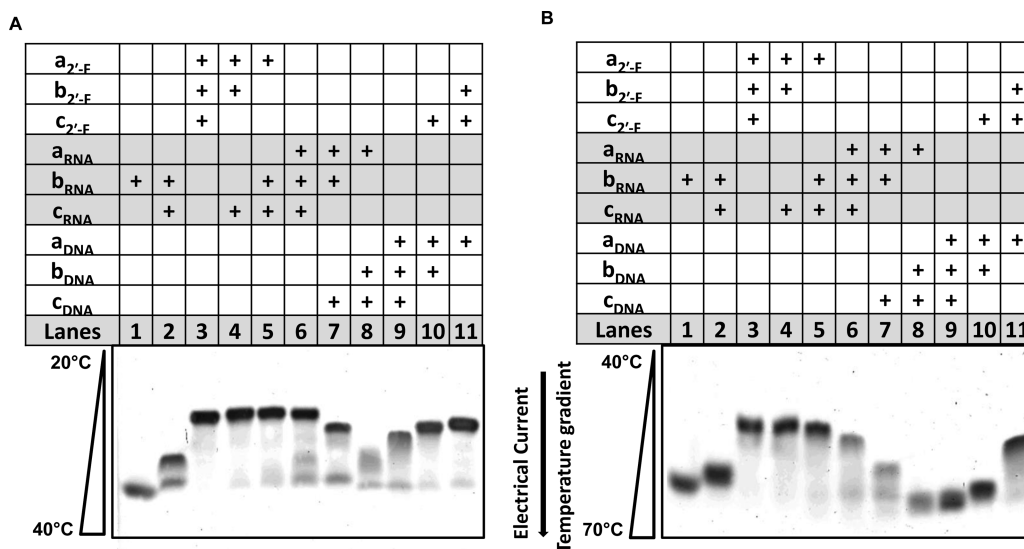


Figure 7. Native 15% TGGE of some hybrids 3WJs with temperature gradient in parallel of the electrical current. As the 3WJs migrated into the gels, weaker structures melted due to the elevating temperatures, resulting in a more rapid migration; stable structures migrated at slower rates. Concentration of hybrids in each lane = 10 μM; the bands were detected by total nucleic acid stain with EB. (A) Hybrids analyzed in a linear temperature gradient of 20–40 °C and (B) the same samples but the temperature range was 40–70 °C.

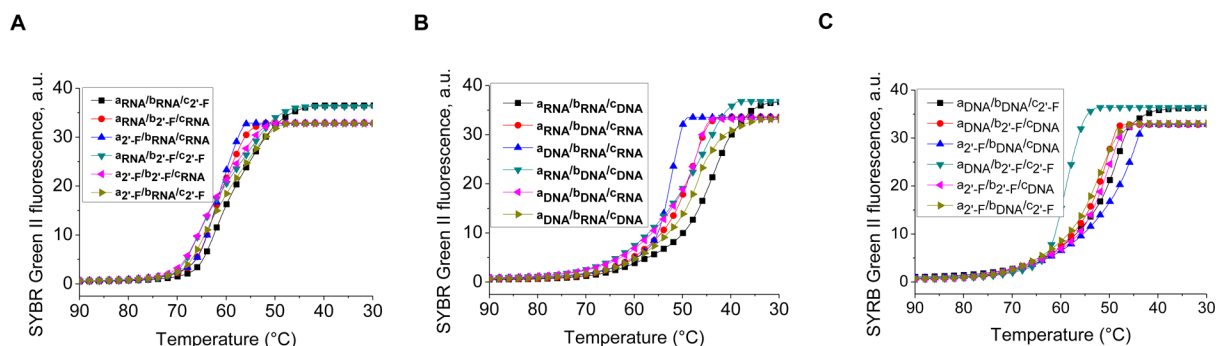


Figure 8. Comparison of pRNA-3WJ hybrid structures. Assembly curves produced from the Roche 480 Lightcycler of the pRNA-3WJ (A) RNA/DNA hybrids, (B) RNA/2'-F RNA hybrids, and (C) DNA/2'-F RNA hybrids.

Table 2. Annealing Temperature for 3WJ Hybrid Formation^a

2'-F RNA to RNA	T _a (°C)	RNA to DNA	T _a (°C)	DNA to 2'-F RNA	T _a (°C)
a _{2'-F} /b _{2'-F} /c _{2'-F}	69.8 ± 2.0	a _{RNA} /b _{RNA} /c _{RNA}	59.3 ± 1.7	a _{DNA} /b _{DNA} /c _{DNA}	48.9 ± 3.2
a _{2'-F} /b _{2'-F} /c _{RNA}	65.4 ± 0.1	a _{RNA} /b _{RNA} /c _{DNA}	42.6 ± 2.2	a _{DNA} /b _{DNA} /c _{2'-F}	48.4 ± 1.6
a _{2'-F} /b _{RNA} /c _{2'-F}	64.1 ± 0.2	a _{RNA} /b _{DNA} /c _{RNA}	48.6 ± 1.5	a _{DNA} /b _{2'-F} /c _{DNA}	51.6 ± 0.4
a _{RNA} /b _{2'-F} /c _{2'-F}	65.5 ± 0.2	a _{DNA} /b _{RNA} /c _{RNA}	53.1 ± 0.1	a _{2'-F} /b _{DNA} /c _{DNA}	47.2 ± 1.5
a _{2'-F} /b _{RNA} /c _{RNA}	62.1 ± 0.1	a _{RNA} /b _{DNA} /c _{DNA}	44.5 ± 2.6	a _{DNA} /b _{2'-F} /c _{2'-F}	59.5 ± 0.2
a _{RNA} /b _{2'-F} /c _{RNA}	62.7 ± 0.2	a _{DNA} /b _{RNA} /c _{DNA}	45.9 ± 2.4	a _{2'-F} /b _{DNA} /c _{2'-F}	52.4 ± 0.6
a _{RNA} /b _{RNA} /c _{2'-F}	61.9 ± 0.4	a _{DNA} /b _{DNA} /c _{RNA}	47.3 ± 0.4	a _{2'-F} /b _{2'-F} /c _{DNA}	51.2 ± 1.8

^aAnnealing temperatures calculated at 10 μM total strand concentration in TMS buffer.

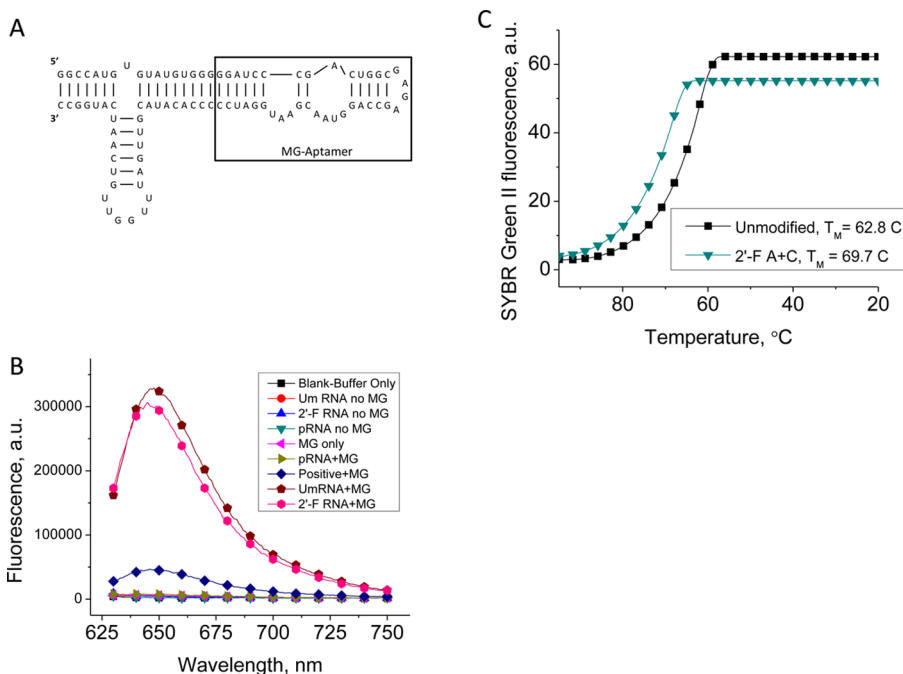


Figure 9. Thermostability of functional pRNA-3WJ nanoparticle. (A) Secondary structure and sequence of the 3WJ-MG aptamer. (B) Fluorescence emission of 3WJ-MG aptamer, demonstrating binding of the nanoparticle to MG in both RNA and 2'-F species. (C) Melting curves for the 3WJ-MG and 2'-F A/C RNA-3WJ.

the general trend that more strands with 2'-F modifications equate to a higher stability overall. These results were further confirmed by the annealing temperatures produced for each hybrid on the Roche 480 Lightcycler as shown in Figure 8 and Table 2. Combining the TGGE gels along with the annealing temperatures provided from the fluorescence annealing curves, the data further support the findings that 2'-F modifications strengthen the thermostability of the pRNA-3WJ, while DNA

substitutions only weaken the 3WJ complex. In this case, even limited modifications lead to a difference in the thermostability.

MG-Aptamer Functionality Assay and Stability. To ensure that the added stability of the 2'-F modifications to the pRNA-3WJ was true for a functional, more complex RNA nanoparticle, a fluorescence assay was performed. The pRNA-3WJ used in this study harbored the Malachite Green (MG) RNA aptamer that binds to Malachite green triphenylmethane dye causing the chemical to fluoresce.⁷⁷ Malachite green itself

emits very low fluorescence; therefore, a change in the fluorescence can be used to confirm binding and complexation between the RNA aptamer and chemical.^{30,71,77,78}

It has been previously shown that the MG aptamer remains active in binding when nucleotides remain unmodified, as well as when using 2'-F cytosine and adenine modified nucleotides.⁷¹ Therefore, we constructed the pRNA-3WJ-MG using unmodified RNA strands and 2'-F A/C modified RNA strands (see Figure 9A for 2D structure). The resulting particles were then tested for binding to the MG to show the folding and functionality of the MG aptamer by measuring its fluorescence emissions (Figure 9B). Both nanoparticles showed almost identical fluorescence values indicating the formation of correctly folded structures.

Further, we used real-time PCR to measure the T_a of pRNA-3WJ-MG and 3WJ-MG_{2',F} constructs (Figure 9C). The annealing curves show that the 2'-F A/C modified 3WJ resulted in the annealing temperature of 69.7 °C, higher than unmodified RNA variant with T_a value of 62.8 °C. This is consistent with the trend of increased RNA 3WJ stability using 2'-F modifications. Thus, the data demonstrate the effectiveness of the fluorine modifications regardless of the complexity of the structure, while retaining the functional conformation of the pRNA-3WJ.

CONCLUSIONS

In this paper, we obtained thermodynamic parameters for the pRNA-3WJ using real-time PCR and TGGE approaches. It was seen that the three fragments existed either in 3WJ complex or as monomers, with the intermediate of dimers almost undetectable. It seems that the three fragments can lead to the formation of 3WJ complex efficiently within a rapid time. It is also found that the formation of the three-component complex was governed by entropy, instead of enthalpy, as usually found in RNA complexes. By combining this beneficial assembly with the improved thermodynamic characteristics by 2'-fluoro modifications to the pRNA, 3WJ stable RNA nanoparticles can be constructed with high yield for the treatment of cancers and viral infections.

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Notes

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ABBREVIATIONS

pRNA, packaging RNA; 2'-F, 2'-fluoro chemical modification; K_D , dissociation constant; 3WJ, three-way junction; T_m , melting temperature; T_a , annealing temperature; rtPCR, real-time polymerase chain reaction; TGGE, temperature gradient gel electrophoresis; ΔG° , change of Gibbs free energy; ΔH° , change in enthalpy; ΔS° , change in entropy; EB, ethidium bromide; MG, Malachite green

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