

Eighteen 5,7-Dihalo-8-quinolinol and 2,2'-Bipyridine Co(II) Complexes as a New Class of Promising Anticancer Agents

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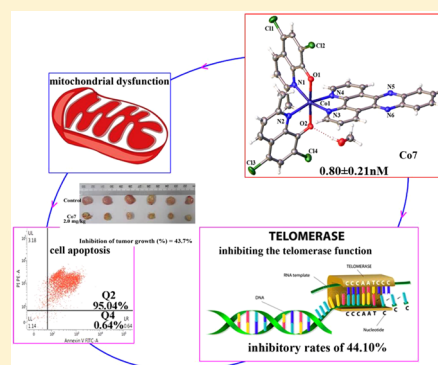
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Supporting Information

ABSTRACT: Here we first report the design of a series of bis-chelate Co(II) 5,7-dihalo-8-quinolinol-phenanthroline derivative complexes, [Co(py)(QL1)₂] (Co1), [Co(py)(QL2)₂] (Co2), [Co(Phen)(QL1)₂] (Co3), [Co(Phen)(QL2)₂] (Co4), [Co(DPQ)(QL1)₂](CH₃OH)₄ (Co5), [Co(DPQ)(QL2)₂] (Co6), [Co(DPPZ)(QL1)₂](CH₃OH) (Co7), [Co(MDP)(QL1)₂](3H₂O) (Co8), [Co(ODP)(QL1)₂](CH₃OH) (Co9), [Co(PPT)(QL1)₂](CH₃OH) (Co10), [Co(CIPT)(QL1)₂] (Co11), [Co(dpy)(QL3)₂] (Co12), [Co(mpy)(QL1)₂] (Co13), [Co(Phen)(QL4)₂] (Co14), [Co(ODP)(QL4)₂] (Co15), [Co(mpy)(QL4)₂]I (Co16), [Co(CIPT)(QL4)₂] (Co17), and [Co(CIPT)(QL5)₂] (Co18), with 5,7-dihalo-8-quinolinol and 2,2'-bipyridine mixed ligands. The antitumor activity of Co1–Co18 has been evaluated against human HeLa (cervical) cancer cells *in vitro* (IC₅₀ values = 0.8 nM–11.88 μM), as well as *in vivo* against HeLa xenograft tumor growth (TIR = 43.7%, *p* < 0.05). Importantly, Co7 exhibited high safety *in vivo* and was more effective in inhibiting HeLa tumor xenograft growth (43.7%) than cisplatin (35.2%) under the same conditions (2.0 mg/kg). In contrast, the H-QL1 and DPPZ ligands greatly enhanced the activity and selectivity of Co7 in comparison to Co1–Co6, Co8–Co18, and previously reported cobalt(II) compounds. In addition, Co7 (0.8 nM) inhibited telomerase activity, caused G2/M phase arrest, and induced mitochondrial dysfunction at a concentration 5662.5 times lower than Co1 (4.53 μM) in related assays. Taken together, Co7 showed low toxicity, and the combination could be a novel Co(II) antitumor compound candidate.

KEYWORDS: 5,7-Dihalo-8-quinolinol, Co(II) complex, antitumor activity, telomerase activity, mitochondrial dysfunction



Pt-based drugs were extensively used to treat a large number of tumors in the clinic.^{1–10} However, cisplatin (*cis*-[PtCl₂(NH₃)₂]) and its derivatives were limited by drug resistance and some severe side effects,^{11,14} and consequently, the Ru, Ti, Au, Co, Ir, Os, Rh, Fe, Cu, etc., complexes have been designed and attracted attention.^{16–38} In addition, cobalt has emerged as a key of vitamin B12 (cobalamin) metabolism and its metal complexes have been reported as DNA cleavage agents, antiviral, antifungal, antitumor antiproliferative, antioxidant, and have shown antimicrobial activity, such as oxoisoporphine,¹⁵ shydroxamic acids prodrugs,^{16,17} 2-benzimidazole derivatives,¹⁸ acetylene-hexacarbonyldicobalt,¹⁹ 2-acetylpyridine and malonic acid dihydrazide,²⁰ antiulcer drug famotidine,²¹ sparfloxacin,²³ valine-derived Schiff bases,²⁴ *N*-benzoyl-*N'*-dialkylthiourea derivatives,²⁵ nonsteroidal anti-inflammatory drugs,²⁸ nonsteroidal anti-inflammatory drug tolfenamic acid,³⁰ 2-acetoxy-(2-propynyl)benzoate]hexacarbonyldicobalt,³¹ a fluorescent coumarin,³² and tetradentate phenolate-based ligand³⁵ cobalt(II/III) complexes.

Recently, many novel hydroxyquinoline metal compounds, such as Zn, Cu, Ni, Ir, Os, Rh, Au, Co, Fe, Sn, Ru, Pt, Pd, and Ln metal complexes,^{39–60} have proved to be promising anticancer drugs *in vitro* and *in vivo*. Among them, a small amount of copper(II) complexes of hydroxyquinolines were designed and preliminarily identified as anticancer drugs.^{50–60} However, to date, Co(II) complexes bearing 5,7-dihalo-2-methyl-8-quinolinol and *o*-phenanthroline derivative mixed ligands have not been reported.

To gain the mixed chelating cobalt(II) complexes with high anticancer activity *in vitro* and *in vivo*, we first designed 18 novel Co(II) complexes Co1–Co18 with 2,2'-bipyridine (py), 1,10-phenanthroline (Phen), 5,7-dichloro-2-methyl-8-quinolinol (H-QL1), dipyridoquinoxaline (DPQ), 5,7-dibromo-2-methyl-8-quinolinol (H-QL2), dipyridophenazine (DPPZ), 5,6-dimethyl-

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10-phenanthroline (MDP), 4,4'-dimethoxy-2,2'-bipyridyl (ODP), 4,7-diphenyl-10-phenanthroline (PPT), 4,7-dichloro-1,10-phenanthroline (CIPT), 5-chloro-7-iodo-8-hydroxyquinoline (H-QL3), 5,5-dimethyl-2,2-dipyridine (dpy), 5,7-diiodo-8-hydroxyquinoline (H-QL4), 4,4'-dimethyl-2'-bipyridine (mpy), and 5,7-dibromo-8-quinolinol (H-QL5). Additionally, the biological properties of Co1–Co18 have been evaluated.

First, the mononuclear complexes Co1–Co18 were prepared by CH₃CN–CH₃OH (3.5 mL/1.5 mL) reflux of H-QL1 (or H-QL2, H-QL3, H-QL4, H-QL5) and py, Phen, DPQ, DPPZ, MDP, ODP, PPT, CIPT, dpy, and mpy with cobalt(II) acetate (2:1:1) at 80 °C for 24 h, respectively (Scheme 1). These 18 new Co(II) complexes were structurally fully characterized (Figures 1 and S1–S45).

Scheme 1. General Synthetic Pathway for Co1–Co18

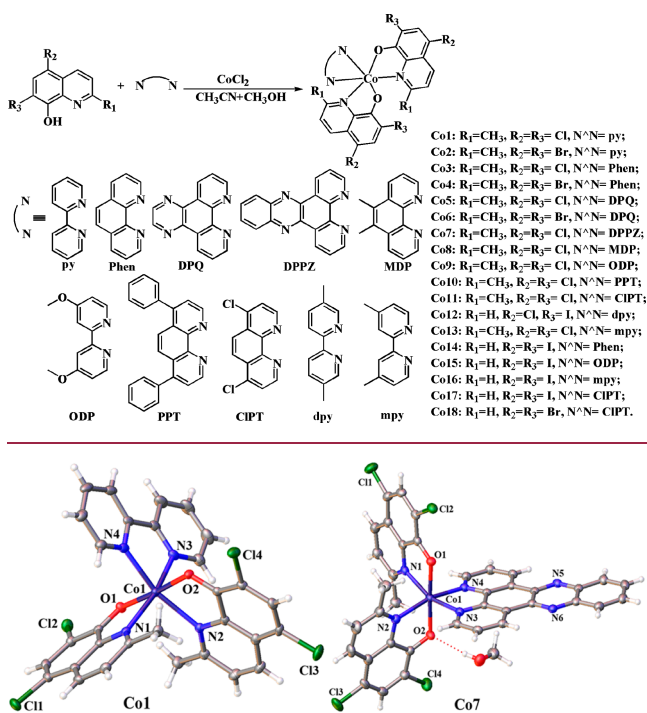


Figure 1. Crystal structures of Co1 and Co7.

The Co(II) atoms in Co1–Co18 were six-coordinated and surrounded by two deprotonated QL ligands (N^O-ligand) and one second ligand molecule (N^N-ligand) showing a distorted octahedral geometry (Figures 1 and S1–S16). In addition, a diagram of Co1–Co18 is shown in Figures 1 and S1–S16, and selected bond distances (Å) and angles (deg) are listed in Tables S1–S54, and the bond lengths (Å) of Co1–Co18 remained normal.

MTT assay was carried out to gauge the *in vitro* anticancer activity of Co1–Co18, H-QL1, H-QL2, py, Phen, DPQ, DPPZ, MDP, ODP, PPT, CIPT, H-QL3, dpy, H-QL4, CoCl₂·6H₂O, mpy, and H-QL5 using BEL-7404 (hepatocellular), Hep-G2 (hepatocellular), HeLa (cervical), MCF-7 (breast) cancer cells, and normal HL-7702 (hepatocyte) cells. As a general observation (Table S55), Co7 was more active (IC₅₀ values = 0.80 nM) than the Co1–Co6, Co8–Co18, cisplatin, H-QL1, H-QL2, py, Phen, DPQ, DPPZ, MDP, ODP, PPT, CIPT, H-QL3, dpy, H-QL4, CoCl₂·6H₂O, mpy, and H-QL5 in all tested cells, and the cytotoxicity of Co1–Co18 against HeLa cells

followed the order Co7 > Co5 > Co8 > Co11 > Co3 > Co10 > Co9 > Co13 > Co1 > Co6 > Co4 > Co2 > Co12 > Co18 > cisplatin > Co15 > Co17 > Co14 > Co16. The higher *in vitro* anticancer activity for Co7 may be due to the H-QL1 and DPPZ ligands. Such observed, different antitumor effects may be due to the electronic effect of the methyl group and the halogenated and more extended planar ligand of H-QL1 and DPPZ. Interestingly, Co1–Co18 showed low toxicity (IC₅₀ > 60.0 μM) to normal HL-7702 (hepatocyte) cells. Compared with previously reported 8-hydroxyquinoline metal complexes (IC₅₀ ≥ 1.00 nM),^{39–60} Co7 exhibited higher cytotoxicity against HeLa cells (IC₅₀ = 0.80 ± 0.21 nM).

Thus, ICP-MS assay showed that the Co(II) concentrations of Co7 ((22.96 ± 0.15 nmol of Co)/10⁶ cells) and Co1 ((18.06 ± 0.05 nmol of Co)/10⁶ cells) were significantly above those of control groups and cisplatin ((4.11 ± 0.59 nmol of Pt)/10⁶ cells),¹⁵ with Co7 (0.80 nM) showing the highest cell accumulation of Co ((22.96 ± 0.15 nmol of Co)/10⁶ cells) and a high extent in HeLa nuclear fraction (Table S56). Thus, Co7 (0.80 nM) showed higher toxicity on the HeLa (cervical) cancer cells possibly due to its better cellular uptake.

Thus, the induction of the level of c-myc, hTERT, and thus, telomerase in HeLa cells by Co1 (4.53 μM) and Co7 (0.80 nM) was investigated using a TRAP-silver staining assay and Western blot. As shown in Figure 2, Co7 (0.80 nM) showed a more

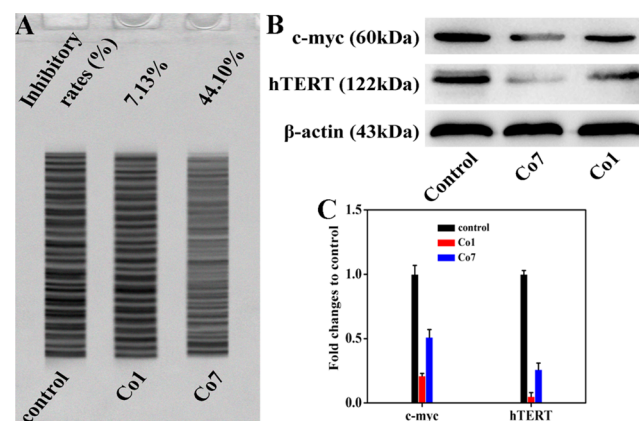


Figure 2. Level of telomerase (A) and related factors (B,C) in HeLa cells induced by Co1 (4.53 μM) and Co7 (0.80 nM) at 24 h.

inhibitory effect on c-myc, hTERT, and telomerase activity than that of Co1 (4.53 μM), suggesting that Co1 (4.53 μM) and Co7 (0.80 nM) inhibited c-myc and hTERT, and thus, telomerase levels were related to a variety of malignant cancers.^{61–65} Importantly, inhibition of telomerase in Co7 (0.80 nM)-treated cells was 44.10%, while that caused by Co1 (4.53 μM) only reached 7.13%.

Furthermore, a G2/M population of 22.89% was observed in Co7 (0.80 nM)-treated cells, while the other corresponding G2/M populations of 20.27% and 12.35% were observed in the Co1 (4.53 μM) treated and control cells (Figure 3), suggesting that Co1 (4.53 μM) and Co7 (0.80 nM) caused G2/M cycle arrest. In addition, Co7 (0.80 nM) and Co1 (4.53 μM) could inhibit the expression of cyclin B1 and CDK1 in HeLa (cervical) cells (Figure S46), mainly due to the G2/M phase arrest (Figure 3) that they cause and their ability to inhibit telomerase.^{66–68} Further, immunofluorescence (Figure S47) and Western blot assays (Figure S46) were carried out. Clearly, Co7 (0.80 nM) and Co1 (4.53 μM) could up-regulate the H2A.X and cleaved-

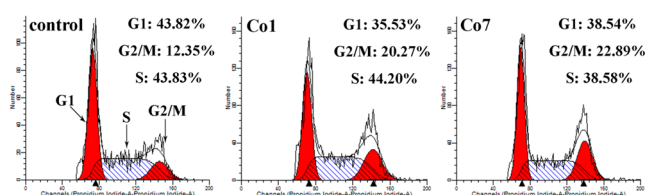


Figure 3. Co7 (0.80 nM) and Co1 (4.53 μ M) caused G2/M phase arrest for 24 h.

PARP levels, indicating that Co7 (0.80 nM) and Co1 (4.53 μ M) remarkably induced DNA damage (Figures S46 and S47) and caused G2/M phase arrest following the order of Co7 > Co1.

In addition, Co7 (0.80 nM) mainly accumulated in a nuclear fraction (Table S56) but also was distributed in the mitochondria. Thus, Co7 (0.80 nM) and Co1 (4.53 μ M) caused obviously up-regulated reactive oxygen species (ROS, Figure S48), intracellular $[Ca^{2+}]$ (Figure 4), and caspase-3/9

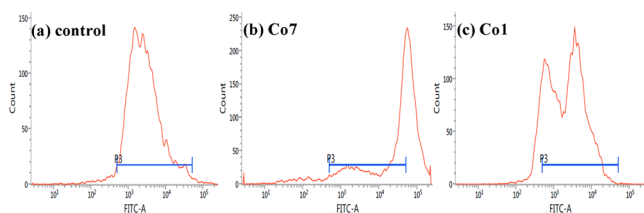


Figure 4. Effects of Co7 (0.80 nM) and Co1 (4.53 μ M) on the $[Ca^{2+}]$ level in HeLa cells at 24 h.

levels (Figure S49 and S50), and down-regulated mitochondrial membrane potential ($\Delta\Psi_m$) (Figure S51) in HeLa cells, illustrating that the $\Delta\Psi_m$, ROS generation, intracellular $[Ca^{2+}]$, and caspase-3/9 played a key role in cancer mitochondrial function damage and apoptosis.^{15,69–72}

Therefore, to clarify the antimigration effects of Co7 (0.80 nM) and Co1 (4.53 μ M) in HeLa cells (Figure 5a–d), the transwell migration assay was carried out. It was found that Co7 (0.80 nM) could significantly induce cell migration at 0.80 nM than that of Co1 (4.53 μ M).

For this, to further investigate the ability of Co7 (0.80 nM) and Co1 (4.53 μ M) to induce HeLa cell apoptotic by flow cytometry (FCM). As shown in Figure 5e–g, the percentages of apoptotic cells treated with Co7 (0.80 nM) and Co1 (4.53 μ M) were 95.68% and 34.42%, respectively, suggesting that Co complexes could cause cell death at higher rates than other 8-quinolinolate metal complexes.^{39–60}

Furthermore, treatment of HeLa xenografts by Co7 (2.0 mg/kg/q2d) was related to significant reduction (TIR = 43.7%, $p < 0.05$) in tumor growth (Figure 6 and Tables S57–S59), which represents approximately 1.3-fold reduction compared with cisplatin-treated groups (IR = $35.2 \pm 5.8\%$, $p < 0.05$).^{51,55,56,72–74} In addition, Co7 (2.0 mg/kg/q2d)-treated mice displayed no obvious signs of toxicity (Figure 6 and Tables S57–S59) as indicated by a relatively stable mouse body weight ($m_{end} = 20.5 \pm 1.3$ g) compared to solvent control ($m_{end} = 20.7 \pm 1.4$ g).

In conclusion, we have shown Co1–Co18 complexes containing mixed 5,7-dihalo-8-quinolinol (H-QL1–H-QL5) and 2,2'-bipyridine derivative-based ligands as potential Co(II) complexes with superior cytotoxicity compared with previously reported Co compounds. MTT studies have demonstrated that Co7 was ca. 300.6 times more cytotoxic than cisplatin ($15.03 \pm$

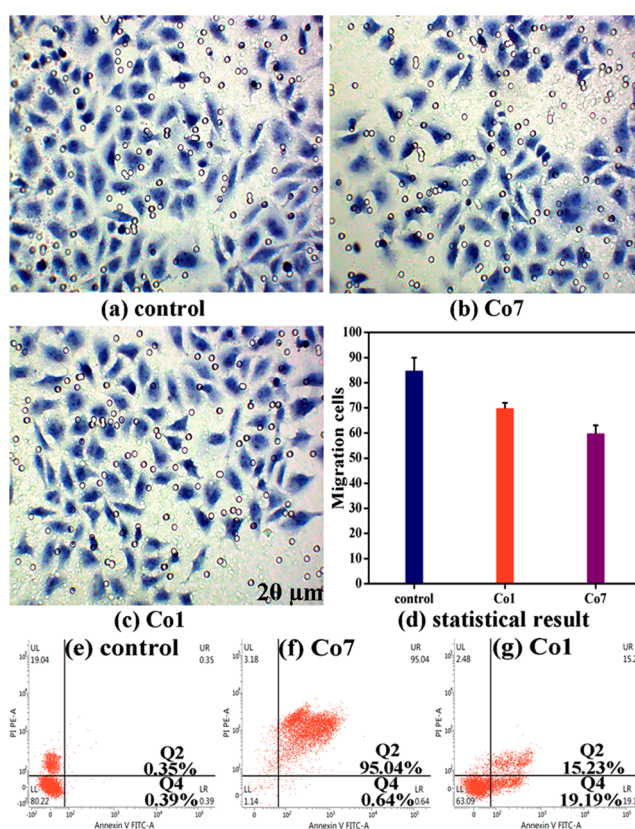


Figure 5. Antimigration (a–d) and apoptosis (e–g) effect of Co7 (0.80 nM) and Co1 (4.53 μ M) on HeLa cells for 24 h.

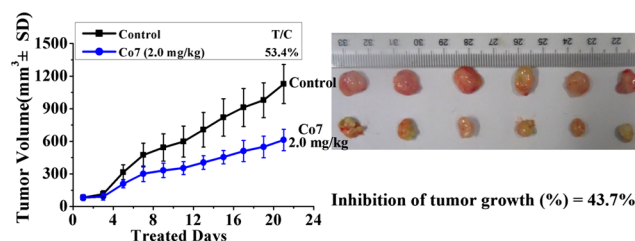


Figure 6. Tumor volume (A, mm³ \pm SD) and images (B) of the HeLa tumors of Co7 ($n = 6$) for 21.0 days.

1.05 μ M), with IC₅₀ values of 0.80 ± 0.21 nM while maintaining a high extent in HeLa nuclear fraction and targeting telomerase; thus, it was considerably less cytotoxic to normal HL-7702 (hepatocyte) cells. In addition, the antitumor activity of Co7 has been evaluated *in vivo* against HeLa xenograft growth (TIR = 43.7%, $p < 0.05$). It also inhibited telomerase activity, caused G2/M phase arrest, and induced mitochondrial dysfunction at a 104.6-fold lower concentration than Co1 (4.53 μ M) in related assays. Thus, the superior cytotoxicity (IC₅₀ = 0.80 ± 0.21 nM) and selectivity index of Co7 in comparison to cisplatin could make it a novel antitumor cobalt(II)-based drug.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsmchemlett.9b00356.

Detailed experiments of Co1–Co18 (PDF)

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Notes

The authors declare no competing financial interest.

The CCDC numbers for Co1–Co18 are 1916696–1916700, 1916702, 1916703, 1917994, 1917997, 1917998, 1918000, 1918001, 1918004, 1918006, 1918007, 1918010, 1918012, and 1918014.

Biographies

Qi-Pin Qin received his Ph.D. degree from Guangxi Normal University in 2017 under the guidance of Prof. Zhen-Feng Chen and Hong Liang. Subsequently, he joined the faculty at Yulin Normal University in 2017 as a professor. His current research interest is focused on metal-based anticancer complexes, nanodrugs, and bioimaging agents. He has designed and synthesized over 300 new coordination compounds. The results have been published in over 70 research papers.

Hua-Hong Zou obtained his Ph.D. from Guangxi Normal University under the supervision of Prof. Fu-Pei Liang in 2015. Currently, he works at the State Key Laboratory for Chemistry and Molecular Engineering of Medicinal Resources, School of Chemistry & Pharmacy of Guangxi Normal University. His research is focused on the design and mechanism of action of metal-based anticancer complexes, and assembly process and mechanism of polynuclear clusters in solution. He has published more than 120 papers related to the above research areas.

Fu-Pei Liang received his Ph.D. degree from the University of Zurich in 2001 and completed postdoctoral research at the University of Zurich in 2002. He is currently a Professor of the State Key Laboratory for Chemistry and Molecular Engineering of Medicinal Resources, School of Chemistry & Pharmacy of Guangxi Normal University. His research interests include the design and mechanism of action of metal-based anticancer complexes, multifunctional complexes, and synthesis and properties of high-nuclearity lanthanide clusters.

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ABBREVIATIONS

SD, standard deviation; TIR, tumor growth inhibition rate

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