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Aromatic Constituents from the Stems of Astragalus membranaceus (Fisch.) Bge. var. Mongholicus (Bge.) Hsiao

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Academic Editor: Derek J. McPhee Received: 5 February 2016 ; Accepted: 10 March 2016 ; Published: 16 March 2016

Abstract: Four new aromatic constituents, astraflavonoids A (1), B (2), C (3), and astramemoside A (4), along with sixteen known ones 5–20 were obtained from the stems of *A. membranaceus* (Fisch.) Bge. *var. mongholicus* (Bge.) Hsiao. Their structures were elucidated by chemical and spectroscopic methods. Among the known isolates, 14 was obtained from the Astragalus genus for the first time, while 7–12, 18–20 were isolated from the species for the first time. The effects of the compounds obtained from the plant on glucose consumption were analyzed in differentiated L6 myotubes *in vitro*, whereby compounds 1, 2, 3, 7, 8, 10, 11, 14, 15 and 18 displayed significant promoting effects on glucose consumption in L6 myotubes. Among them, the activities of 1, 2 and 7 were comparable to that of insulin, which suggested that these compounds may be involved in glucose metabolism and transport.

Keywords: *Astragalus membranaceus* (Fisch.) Bge. *var. mongholicus* (Bge.) Hsiao.; stem; structure elucidation; aromatic constituents; L6 cells; glucose consumption

1. Introduction

Astragalus membranaceus (Fisch.) Bge. var. mongholicus (Bge.) Hsiao (AM), belongs to the Astragalus genus of the Leguminosae family. The main chemical constituents of the plant are flavonoids and terpenoids. During the course of our studies, 14 oleanane type saponins, including eight new ones, named astroolesaponins A, B, C₁, C₂, D, E₁, E₂, and F, have been obtained from its stems, and some of them showed depressing effects on triglyceride levels in sodium oleate-induced HepG2 cells [1]. During our continued research on this species, twenty aromatic constituents, including four new ones, were isolated from AM using SiO₂ gel, ODS, Sephadex LH-20 column chromatography (CC) and preparative HPLC chromatography (Prep HPLC), and their structures were clearly determined by chemical and spectroscopic methods (¹H-NMR, ¹³C-NMR, ¹H-¹H COSY, HSQC, HMBC, UV, IR, CD, and MS). Based on the evidences of previous activity reports on the Astragalus genus [1], the glucose consumption effects of the isolates were examined.

2. Results and Discussion

The 70% EtOH extract of AM was subjected to solvent partition, chromatographic isolation, and chemical and spectral analysis. As a result, four new compounds (Figure 1), astraflavonoids A–C (1–3) and astramemoside A (4), along with sixteen known ones (Figure 2), kaempferol (5) [2], kaempferol-3-O- β -D-glucoside (6) [3], kaempferol-3-O-(2-O- α -L-rhamnopyranosyl)- β -D-glucopyranoside (7) [4], kaempferol-3,7-di-O- β -D-glucopyranoside (8) [5], rhamnocitrin-3-O- β -D-glucopyranoside (9) [6], rhamnocitrin-3-O-neohesperoside (10) [7], complanatuside (11) [8], quercetin-3-O- β -D- neospheroside (12) [9], genistein (13) [10], sophorabioside (14) [11], calycosin (15) [12], odoratin-7-O- β -D-glucopyranoside (16) [13], (–)-liquiritigenin (17) [14,15], maltol- β -D-glucopyranoside (18) [16], 2,6-dimethoxy-4-hydroxyphenyl-1-O- β -D-glucopyranoside (19) [17], and benzyl- α -L-arabino-pyranosyl(1" \rightarrow 6')- β -D-glucopyranoside (20) [18] were obtained from it. Among the known isolates, compound 14 was isolated from the Astragalus genus for the first time, and 7–12, 18–20 were isolated from the species for the first time. Herein, the isolation and identification of these compounds are described, as well as their effects on glucose consumption in L6 cells.



Figure 1. The new compounds 1–4 obtained from AM.



Figure 2. The known compounds 5–20 obtained from AM.

Astraflavonoid A (1) was isolated as a yellow powder with $[\alpha]_D^{25} - 69.8^{\circ}$ (in MeOH). Its molecular formula is C₃₆H₃₆O₁₈, as indicated by HR-Q-TOF-ESI-MS (m/z 755.1795 [M – H]⁻, calcd for C₃₆H₃₅O₁₈, 755.1829). The IR spectrum showed absorption bands due to hydroxyl (3365 cm⁻¹), α , β -unsaturated ester carbonyl (1693 cm⁻¹), α , β -unsaturated ketone (1654 cm⁻¹), aromatic ring (1604, 1512, 1453 cm⁻¹), and ether functions (1071 cm⁻¹). The ¹H- and ¹³C-NMR (DMSO-*d*₆, Table 1) spectra suggested the presence of the following moieties in 1: kaempferol aglycon [δ 6.16 (1H, d, *J* = 1.5 Hz, H-6), 6.24 (1H, d, *J* = 1.5 Hz, H-8), 6.86 (2H, d, *J* = 9.0 Hz, H-3',5'), 7.99 (2H, d, *J* = 9.0 Hz, H-2',6'), 12.58 (1H, br. s, 5-OH)], *trans-p*-feruloyl [δ 3.81 (3H, s, 3'''-OCH₃), 6.21 (1H, d, *J* = 16.0 Hz, H-8''''), 6.93 (1H, dd, *J* = 1.5, 8.0 Hz, H-6''''), 7.14 (1H, *J* = 1.5 Hz, H-2''''), 7.25 (1H, d, *J* = 16.0 Hz, H-7'''')], together with two anomeric proton signals [δ 5.40 (1H, br. s, H-1'''), 5.66 (1H, d, *J* = 7.5 Hz, H-1'')]. Moreover, the ¹H-¹H COSY experiment indicated the presence of partial structure indicated in bold lines (Figure 3). In the HMBC experiment, the long-range correlations

from the following proton to carbon pairs were observed: $\delta_H 5.66 (H-1'')$ to $\delta_C 132.7 (C-3)$; $\delta_H 5.40 (H-1''')$ to $\delta_C 75.9 (C-2'')$; $\delta_H [4.15 (1H, d, J = 11.0 Hz), 4.27 (1H, d, J = 11.0 Hz), H2-5''']$ to $\delta_C 166.2 (C-9'''')$. Meanwhile, the ¹³C-NMR data for the sugar parts were in good agreement with those of isorhamnetin 3-*O*-(5-*O*-*trans*-feruloyl- β -D-apiofuranosyl)-(1 \rightarrow 2)- β -D-glucopyranoside [δ 60.5 (C-6''), 68.0 (C-5'''), 70.2 (C-4''), 73.8 (C-4'''), 76.2 (C-2'''), 75.7 (C-2''), 77.0 (C-3''), 77.4 (C-5''), 77.6 (C-3'''), 98.5 (C-1''), 107.6 (C-1''')] [19]. Consequently, the structure of **1** was elucidated to be kaempferol 3-*O*-[(5-*O*-*trans*-*p*-feruloyl)- β -D-apiofuranosyl](1 \rightarrow 2)- β -D-glucopyranoside.

No.	$\delta_{\mathbf{C}}$	$\delta_{\rm H}$ (J in Hz)	No.	δ _C	$\delta_{\rm H}$ (J in Hz)
2	155.4	-	6''	60.6	3.29 (m, overlapped)
3	132.7	-			3.55 (br. d, ca. 12)
4	177.1	-	1'''	107.7	5.40 (br. s)
5	161.2	-	2'''	76.3	3.75 (br. s)
6	98.5	6.16 (d, 1.5)	3'''	77.5	-
7	163.9	-	4'''	73.6	3.57 (d, 9.5)
8	93.3	6.24 (d, 1.5)			3.95 (d, 9.5)
9	156.0	-	5'''	67.6	4.15 (d, 11.0)
10	103.9	-			4.27 (d, 11.0)
1′	120.9	-	1''''	125.4	-
2′,6′	130.6	7.99 (d, 9.0)	2''''	110.7	7.14 (d, 1.5)
3′,5′	114.9	6.86 (d, 9.0)	3''''	147.7	-
4'	159.8	-	4''''	149.1	-
5-OH	-	12.58 (br. s)	5''''	115.3	6.77 (d, 8.0)
1''	98.3	5.66 (d, 7.5)	6''''	122.9	6.93 (dd, 1.5, 8.0)
2''	75.9	3.49 (dd, 7.5, 9.0)	7''''	144.5	7.25 (d, 16.0)
3''	76.9	3.42 (dd, 9.0, 9.0)	8''''	114.0	6.21 (d, 16.0)
$4^{\prime\prime}$	70.2	3.10 (dd, 9.0, 9.0)	9''''	166.2	-
5''	77.4	3.09 (m)	3''''-OCH ₃	55.5	3.81 (s)

Table 1. ¹H- and ¹³C-NMR data for **1** in DMSO-*d*₆.



Figure 3. The main ¹H-¹H COSY and HMBC correlations of 1.

Astraflavonoid B (**2**) displayed a negative optical rotation $([\alpha]_D^{25} - 91.7^\circ \text{ in MeOH})$. Its molecular formula, C₄₃H₄₈O₂₃, was determined from the molecular ion peak at *m/z* 931.2525 [M – H][–] by HR-Q-TOF-ESI-MS measurement. Acid hydrolysis of **2** with 1 M HCl yielded D-glucose and L-rhamnose, which were identified on the basis of their retention times (HPLC) and optical rotations [20] The ¹H-, ¹³C-NMR (DMSO-*d*₆, Table 2) and various 2D NMR spectra indicated the presence of a kaempferol aglycon [δ 6.51 (1H, d, *J* = 1.5 Hz, H-6), 6.81 (1H, d, *J* = 1.5 Hz, H-8), 6.87 (2H, d, *J* = 9.0 Hz, H-3',5'), 8.06 (2H, d, *J* = 9.0 Hz, H-2',6'), 12.63 (1H, br. s, 5-OH)], *trans-p*-feruloyl [δ 3.80 (3H, s, 3'''''-OCH₃), 7.28 (1H, *J* = 1.5 Hz, H-2'''''), 6.77 (1H, d, *J* = 8.5 Hz, H-5'''''), 7.04 (1H, d, *J* = 1.5, 8.5 Hz, H-6'''''), 7.54 (1H, d, *J* = 16.0 Hz, H-7'''''), 6.46 (1H, d, *J* = 16.0 Hz, H-8''''')], along with two β-D-glucopyranosyl [δ 5.18 (1H, d, *J* = 7.0 Hz, H-1''''), 5.67 (1H, d, *J* = 7.5 Hz, H-1'')]

No.	δ _C	δ _H (J in Hz)	No.	δ _C	δ _H (J in Hz)
2	155.8	-	2′′′	70.5	3.75 (br. s)
3	132.9	-	3'''	70.4	3.46 (dd, 3.0, 9.0)
4	177.4	-	4'''	71.8	3.14 (dd, 9.0, 9.0)
5	160.9	-	5'''	68.2	3.72 (m)
6	99.6	6.51 (d, 1.5)	///	17.2	0.79 (d, 6.0)
7	162.6	-	1''''	99.1	5.18 (d, 7.0)
8	94.4	6.81 (d, 1.5)	2''''	73.0	3.32 (dd, 7.0, 9.0)
9	156.6	-	3''''	76.1	3.36 (dd, 9.0, 9.0)
10	105.6	-	4''''	69.3	3.31 (m, overlapped)
1'	120.6	-	5''''	73.8	3.80 (m)
2′,6′	130.7	8.06 (d, 9.0)	6''''	63.0	4.20 (dd, 5.5, 12.0)
3′,5′	115.0	6.87 (d, 9.0)			4.44 (br. d, ca. 12)
4'	160.0	-	1'''''	125.4	-
5-OH	-	12.63 (br. s)	2''''	110.9	7.28 (d, 1.5)
1''	98.2	5.67 (d, 7.5)	3''''	147.8	-
2''	77.5	3.46 (dd, 7.5, 9.0)	4''''	149.3	-
3''	77.2	3.40 (dd, 9.0, 9.0)	5'''''	115.4	6.77 (d, 8.5)
$4^{\prime\prime}$	70.1	3.09 (dd, 9.0, 9.0)	6'''''	123.1	7.04 (dd, 1.5, 8.5)
5''	77.4	3.09 (m)	7''''	145.2	7.54 (d, 16.0)
6''	60.7	3.28 (dd, 4.5, 11.5)	8''''	114.0	6.46 (d, 16.0)
		3.56 (br. d, ca. 12)	9'''''	166.5	-
1'''	100.5	5.10 (br. s)	3'''''-OCH ₃	55.6	3.80 (s)

Table 2. ¹H- and ¹³C-NMR data for **2** in DMSO-*d*₆.



Figure 4. The main ¹H-¹H COSY and HMBC correlations of 2.

Astraflavonoid C (3) was obtained as a white powder with negative optical rotation ($[\alpha]_D^{25} - 26.7^{\circ}$ in MeOH). Its molecular formula, C₂₃H₂₈O₁₁, was established by HR-Q-TOF-ESI-MS (*m*/*z* 479.1578 [M – H]⁻; calcd for C₂₃H₂₇O₁₁, 479.1559). Its acid hydrolysis yielded D-glucose [21]. The ¹H-, ¹³C-NMR (CD₃OD, Table 3) and DEPT spectra suggested the presence of one ABX-type aromatic ring, one pentasubstituted aromatic ring, one β -D-glucopyranosyl, and one oxygenated methylene, together with one methylene, one methane, and two methoxy groups. The ¹H-¹H COSY experiment indicated the presence of the partial structure indicated in bold lines (Figure 5). On the other hand, the long-range correlations from H₂-2 to C-9, C-1'; H₂-4 to C-5, C-9, C-10, C-1'; H-5 to C-4, C-7, C-9; H-6 to C-7, C-8, C-10; H-8 to C-6, C-7, C-10; H-6' to C-3, C-2', C-4', C-5'; 3'-OCH₃ to C-3'; 5'-OCH₃ to C-5'; H-1'' to C-4' were clearly found in the HMBC spectra. Furthermore, the NOE correlations between $\delta_{\rm H}$ 4.85 (H-1'') and $\delta_{\rm H}$ 3.83 (5'-OCH₃), 3.90 (3'-OCH₃) observed in the NOESY spectrum confirmed the substituted positions of two methoxy groups and the β -D-glucopyranosyl unit. Thus, the planar structure of **3** was identified as that of isoflavan glycoside. Finally, the CD spectrum of **3** displayed

negative Cotton effect at 285 nm ($\Delta \varepsilon$: -11.84), which indicated the absolute configuration of 3-position was *S* [20].

No.	δ _C	$\delta_{\rm H}$ (J in Hz)	No.	δ _C	δ _H (J in Hz)
2	71.6	3.84 (dd, 8.5, 9.0)	4′	141.3	-
		4.33 (dd, 2.0, 9.0)	5'	141.6	-
3	32.4	3.80 (m)	6'	109.8	6.42 (s)
4	32.5	2.75 (m)	1''	105.6	4.85 (d, 9.0)
5	131.3	6.85 (d, 8.0)	2''	75.6	3.43 (m, overlapped)
6	109.2	6.32 (dd, 2.5, 8.0)	3''	77.9	3.43 (m, overlapped)
7	157.6	-	$4^{\prime\prime}$	71.5	3.36 (dd, 6.5, 9.5)
8	103.9	6.25 (d, 2.5)	5''	78.2	3.24 (m)
9	156.3	-	6''	62.8	3.70 (dd, 5.0, 11.5)
10	114.9	-			3.85 (dd, 2.0, 11.5)
1'	133.1	-	3'-OCH ₃	61.9	3.90 (s)
2′	148.8	-	5'-OCH ₃	61.3	3.83 (s)
3'	147.6	-	U		

Table 3. ¹H- and ¹³C-NMR data for **3** in CD₃OD.



Figure 5. The main ¹H-¹H COSY, HMBC, and NOE correlations of 3.

Astramemoside A (4) was a white powder. Its molecular formula was deduced as $C_{18}H_{22}O_{11}$ from the $[M - H]^-$ quasi-molecular ion at m/z 413.1082 (calcd for $C_{18}H_{21}O_{11}$, 413.1089) in the negative-ion HRESI-TOF-MS spectrum. The IR spectrum showed absorption bands due to hydroxyl (3373 cm⁻¹), ester carbonyl (1721 cm⁻¹), aromatic ring (1609, 1511, 1451 cm⁻¹), and ether (1081 cm⁻¹) functions. The ¹H-, ¹³C-NMR (DMSO- d_6 , Table 4) and various 2D NMR spectra (Figure 6) suggested the presence of a *p*-hydroxybenzoic acid unit [δ_H 6.97 (2H, d, *J* = 9.0 Hz, H-3,5), 7.53 (2H, d, *J* = 9.0 Hz, H-2,6), δ_C 114.8 (C-3,5), 122.3 (C-1), 130.9 (C-2,6), 160.2 (C-4), 164.6 (C-7)].

Table 4.	¹ H- and	¹³ C-NMR	data for	4 in	DMSO-d ₆	<u>.</u>

No.	δ _C	$\delta_{\rm H}$ (J in Hz)	No.	δ _C	δ _H (J in Hz)
1	122.3	-	6′	60.2	3.45 (dd, 5.5, 12.0)
2,6	130.9	7.53 (d, 9.0)			3.65 (br. d, ca. 12)
3,5	114.8	6.97 (d, 9.0)	1''	107.7	5.34 (br. s)
4	160.2	-	2''	76.3	3.65 (br. s)
7	164.6	-	3''	77.5	-
1'	96.8	5.18 (d, 7.0)	4''	73.4	3.80 (d, 9.5)
2'	75.3	3.53 (dd, 7.0, 9.0)			4.31 (d, 9.5)
3′	76.7	3.52 (dd, 9.0, 9.0)	5''	68.0	3.87 (d, 11.0)
4'	69.5	3.20 (dd, 9.0, 9.0)			4.32 (d, 11.0)
5'	76.7	3.52 (m)			



Figure 6. The main ¹H-¹H COSY and HMBC correlations of 4.

Comparison of the ¹H- and ¹³C-NMR spectroscopic data of **4** with those of **1** identified the sugar moiety to be β -D-apiofuranosyl(1 \rightarrow 2)- β -D-glucopyranose. Moreover, the 5-position of the β -D-apiofuranosyl moiety was substituted too. Meanwhile, the long-range correlations from H-2,6 to C-4, C-7; H-1' to C-4; H-1'' to C-2'; H₂-5'' to C-7 observed in the HMBC experiment further proved the correctness of the above deductions.

The effects of the compounds obtained from AM on glucose consumption were analyzed in differentiated L6 myotubes *in vitro*. To create the assay method, standardization of basic parameters like differentiation time, the number of cells to seed, amount of D-glucose to be used and time of incubation were determined (data not shown). As shown in Figure 7, insulin increased the glucose consumption in L6 myotubes to about $4.76 \pm 0.33 \ \mu\text{g}/\text{well}$ (p < 0.001) and the percentage of the raise reached about 9.01%, which serves as a positive control for our study. Among the tested compounds, **1**, **2**, **3**, **7**, **8**, **10**, **11**, **14**, **15** and **18** possessed significant promoting effects on glucose consumption in L6 myotubes, the glucose consumption of which were 5.02 ± 0.29 , 4.92 ± 0.36 , 3.20 ± 0.58 , 4.86 ± 0.67 , 3.37 ± 0.62 , 3.04 ± 0.86 , 2.94 ± 0.60 , 3.96 ± 1.21 , 2.44 ± 0.59 and $3.15 \pm 1.00 \ \mu\text{g}/\text{well}$, respectively. At the concentration of 30 \u03c4 mol/L, compounds **1**, **2** and **7** led to 9.67%, 9.48% and 9.33% increments in glucose consumption, respectively, which was comparable to the effects of insulin. However, the other isolates showed no obvious effect on glucose consumption. These results indicated that some of the constituents in AM can stimulate glucose consumption in L6 myotubes to various degrees.



Figure 7. Effects of compounds **1–11** and **13–20** on glucose consumption in L6 myotubes. L6 myoblasts $(1 \times 10^4 \text{ cells/well})$ were subcultured into 48-place multiwell plates in 2% FBS/DMEM for 7 days to form myotubes. The differentiated myotubes were kept in HBS with no serum or glucose for 2 h, and then were continue incubated in HBS containing 1 mg/mL D-glucose and 2% FBS with or without insulin (Ins, 2 µmol/L) or obtained compounds (30 µmol/L) for another 4 h. Then the glucose concentrations in the supernatant were detected using glucose assay kit and the percentage of glucose consumption in each well was calculated to express the results. Each value represents the mean \pm S.E.M., n = 6. *** p < 0.001, ** p < 0.01, * p < 0.05 vs. control group (Ctrl).

On the basis of the activity results of kaempferol and its glycosides, we can deduce that the glucosylation of 3-hydroxyl group would increase the effect on glucose consumption in L-6 cells, and

disaccharide substitution at 3-hydroxyl showed a stronger activity than monosaccharide substitution. Due to the limited number of compounds, detailed studies are in progress to evaluate more kaempferol glycosides to clarify these structure-activity relationships.

3. Experimental Section

3.1. General Information

The following instruments were used to obtain physical data: optical rotations were recorded on an Autopol IV automatic polarimeter (*l* = 50 mm; Rudolph Research Analytical, Hackettstown, NJ, USA). IR and UV spectra were determined on 640-IR FT-IR (Varian Australia Pty Ltd., Mulgrave, Australia) and Cary 50 UV-Vis spectrophotometers, respectively (Varian, Inc., Hubbardsdon, MA, USA). ¹H- and ¹³C-NMR spectra were measured on a Bruker 500 MHz NMR spectrometer at 500 MHz for ¹H- and 125 MHz for ¹³C-NMR (Bruker BioSpin AG Industriestrasse 26 CH-8117, Fällanden, Switzerland). Negative-ion HR-ESI-Q-TOF-MS were recorded on an Aglient 6520 Q-TOF mass spectrometer (Agilent Corp., Santa Clara, CA, USA).

The following supports were used for chromatography: a macroporous synthetic resin (D101) (Haiguang Chemical Co., Ltd., Tianjin, China), SiO₂ gel (74–149 µm, Qingdao Haiyang Chemical Co., Ltd., Qingdao, China), and ODS (50 µm, YMC Co., Ltd., Tokyo, Japan). Prep HPLC was performed on an ODS column (Cosmosil 5C18-MS-II, Tokyo, Japan; $\Phi = 20 \text{ mm}$, l = 250 mm, flow rate 9.0 mL/min), and the eluate was monitored with a UV detector (Shimadzu RID-10^a UV-vis, Shimadzu Co. Ltd., Kyoto, Japan).

3.2. Plant Material

The stems of *Astragalus membranaceus* (Fisch.) Bge. var. *mongholicus* (Bge.) Hsiao. were collected from Gansu Province, China, and identified by Dr. Li Tianxiang (Experiment Teaching Department, Tianjin University of Traditional Chinese Medicine, Tianjin, China). The voucher specimen was deposited at the Academy of Traditional Chinese Medicine of Tianjin University of TCM.

3.3. Extraction and Isolation

The stems of AM were dealt with by the method described before [1] to give the 95% EtOH eluate from D101 macroporous resin CC. The 95% EtOH eluate (90.6 g) was subjected to SiO₂ gel CC to yield fourteen fractions (Fr. 1–14). Fraction 2 (0.16 g) was purified by Prep HPLC $[MeOH-H_2O (55:45, v/v)]$ to afford calycosin (15, 13.2 mg). Using the same isolation conditions as fraction 2, kaempferol (5, 8.7 mg), genistein (13, 7.2 mg), and (-)-liquiritigenin (17, 8.8 mg) were obtained from fraction 3 (0.78 g). Fraction 6 (7.0 g) was isolated by ODS CC, and purified by Prep HPLC to yield astraflavonoid C (3, 25.0 mg), kaempferol-3-O- β -D-glucoside (6, 43.8 mg), rhamnocitrin-3-O-β-D-glucopyranoside (9, 52.5 mg), and odoratin-7-O-β-D-glucopyranoside (16, 14.4 mg). Fraction 7 (6.9 g) was subjected to SiO₂ gel, ODS, Sephadex LH-20 CC and Prep HPLC to furnish astraflavonoid A (1, 8.5 mg), maltol- β -D-glucopyranoside (18, 59.0 mg), and 2,6-dimethoxy-4-hydroxyphenyl-1-O- β -D-glucopyranoside (19, 12.2 mg). Fraction 8 (7.4 g) was subjected to ODS and Sephadex LH-20 CC, along with Prep HPLC to afford benzyl- α -L-arabinopyranosyl($1'' \rightarrow 6'$)- β -D-glucopyranoside (**20**, 46.4 mg). Fraction 9 (7.0 g) was subjected to ODS CC, and purified by Prep HPLC to give rhamnocitrin-3-O-neohesperoside (10, 46.3 mg) and sophorabioside (14, 38.1 mg). Fraction 10 (12.0 g) was isolated by Sephadex LH-20 CC and Prep HPLC to yield complanatuside (11, 12.4 mg). Fraction 12 (8.2 g) was separated by Prep HPLC to provide astraflavonoid B (2, 22.2 mg), astramemoside A (4, 6.7 mg), kaempferol-3-O-(2-O- α -L-rhamnopyranosyl)- β -D-glucopyranoside (7, 14.5. mg) and kaempferol-3,7di-O-β-D-glucopyranoside (8, 11.2 mg). Fraction 13 (12.1 g) was subjected to Sephadex LH-20 CC and further purified by Prep HPLC to afford quercetin-3-O- β -D-neospheroside (**12**, 2.2 mg).

3.4. Compound Characterization

Astraflavonoid A (1): Yellow powder; $[\alpha]_D^{25}$ –69.8° (*c* = 0.61, MeOH); UV (MeOH) λ_{max} (log ε) 268 (3.99), 292 (3.94, sh), 329 (4.06); IR (KBr) ν_{max} 3365, 2926, 2855, 1693, 1654, 1604, 1512, 1453, 1361, 1275, 1178, 1123, 1071, 1026, 839 cm⁻¹; ¹H- and ¹³C-NMR (DMSO-*d*₆) data see Table 1; Negative-ion mode HR-Q-TOF-ESI-MS *m*/*z* 755.1795 [M – H]⁻ (calcd for C₃₆H₃₅O₁₈, 755.1829).

Astraflavonoid B (2): Yellow powder; $[\alpha]_D^{25} -91.7^\circ$ (c = 0.87, MeOH); UV (MeOH) λ_{max} (log ε) 267 (4.19), 292 (4.08, sh), 325 (4.20); IR (KBr) ν_{max} 3365, 2926, 1698, 1652, 1599, 1512, 1453, 1346, 1275, 1210, 1179, 1124, 1071, 840 cm⁻¹; ¹H- and ¹³C-NMR (DMSO- d_6) data see Table 2; Negative-ion mode HR-Q-TOF-ESI-MS m/z 931.2525 [M – H]⁻ (calcd for C₄₃H₄₇O₂₃, 931.2514).

Astraflavonoid C (3): White powder; $[\alpha]_D^{25} - 26.7^\circ$ (*c* = 0.89, MeOH); CD (*c* = 0.0021 M, MeOH) Δε (λnm) – 11.84 (285); UV (MeOH) λ_{max} (log ε) 228 (4.07, sh), 284 (3.68); IR (KBr) ν_{max} 3367, 2938, 2848, 1622, 1595, 1508, 1461, 1367, 1156, 1109, 1071, 1024, 847 cm⁻¹; ¹H- and ¹³C-NMR (CD₃OD) data see Table 3; Negative-ion mode HR-Q-TOF-ESI-MS m/z 479.1578 [M – H]⁻ (calcd for C₂₃H₂₇O₁₁, 479.1559).

Astramemoside A (4): White powder; $[\alpha]_D^{25} - 21.0^\circ$ (*c* = 0.39, MeOH); UV (MeOH) λ_{max} (log ε) 249 (3.41); IR (KBr) ν_{max} 3373, 2924, 2889, 1721, 1609, 1511, 1451, 1401, 1374, 1270, 1242, 1184, 1081, 1003, 848 cm⁻¹; ¹H- and ¹³C-NMR (DMSO-*d*₆) data see Table 4; Negative-ion mode HR-Q-TOF-ESI-MS *m*/*z* 413.1082 [M – H]⁻ (calcd for C₁₈H₂₁O₁₁, 413.1089).

3.5. Acid Hydrolysis of 2 and 3

A solution of compounds **2** and **3** (each 1.5 mg) was treated with HCl using the same method and conditions as reported in [21]. Identification of L-rhamnose (i) from **2**; and D-glucose (ii) from **2** and **3** by comparison of its retention time and optical rotation with those of authentic sample, t_R : (i) 7.5 min (L-rhamnose, negative optical rotation); and (ii) 14.1 min (D-glucose, positive optical rotation).

3.6. Glucose Consumption Assay of Compounds Obtained from AM

3.6.1. Materials

L6 rat skeletal myoblasts cell line (Cell Resource Center, IBMS, CAMS/PUMC, Beijing, China), Dulbecco's modified Eagle's medium (DMEM, Thermo Scientific, Logan, UT, USA), fetal bovine serum (FBS, Thermo Scientific), streptomycin and penicillin G (Thermo Scientific), glucose assay kit (BioSino Bio-technology and Science Inc., Beijing, China), insulin (Sigma, St. Louis, MO, USA), D-glucose (Solarbio Bio-technology and Science Inc., Beijing, China China).

3.6.2. Cell Culture and Differentiation

Stock cultures of L6 myoblasts were grown in 10% (v/v) FBS/DMEM supplemented with streptomycin (100 µg/mL) and penicillin G (100 U/mL) at 37°C under 5% CO₂ atmosphere and maintained below 70% confluence. For differentiation into L6 myotubes, cells were cultured at a density of 1 × 10⁴ cells/well on 48-place multiwell plates (Costar, Washington, DC, USA) and the medium was switched to 2% (v/v) FBS/DMEM for 7 days, with medium changes every second day prior to use in our experiments.

3.6.3. Determination of Glucose Consumption in L6 Myotubes

This method was based on some literatures [22–24] with slight modifications. Briefly, the 7-day-old myotubes were serum and glucose-deprived kept in HEPES-buffered saline (HBS, 20 mmol/L HEPES, 2 mmol/L sodium pyruvate, 136 mmol/L NaCl, 2.7 mmol/L KCl, 2 mmol/L KH₂PO₄, 10 mmol/L Na₂HPO₄, 2.5 mmol/L MgSO₄, and 1 mmol/L CaCl₂, pH 7.4) for 2 h, and were thereafter incubated in HBS containing 1 mg/mL D-glucose and 2% FBS with or without insulin (Ins, 2 µmol/L) or obtained compounds (1–11, 13–20, 30 µmol/L each) for another 4 h. Then glucose concentrations

in the supernatant were determined using glucose assay kit (GOD-POD colorimetric method) and the ratio of glucose consumption in each well was calculated for further comparison. Results were expressed as a percentage of glucose consumption:

Percentage of glucose consumption (%) = [(glucose surplus of control – glucose surplus of sample)/glucose surplus of control] $\times 100\%$ (1)

4. Conclusions

In summary, four new aromatic constituents, astraflavonoids A–C (1–3), and astramemoside A (4), along with sixteen known ones 5–20 were obtained from the 70% EtOH extract of AM. Among the known isolates, 14 was isolated from the Astragalus genus for the first time, and compounds 7–12, 18–20 were isolated from the species for the first time. Their structures were elucidated by chemical and spectroscopic methods. The effects of the compounds obtained from AM on glucose consumption were analyzed in differentiated L6 myotubes *in vitro*. As results, compounds 1, 2, 3, 7, 8, 10, 11, 14, 15 and 18 possessed significant promotion effects on glucose consumption in L6 myotubes. Among them, the activities of 1, 2 and 7 were comparable to that of insulin, which suggested that these AM compounds may be involved in glucose metabolism and transportat. On the basis of the activity results, the structure-activity was discussed. Glucose consumption plays a role in cellular energy homeostasis. This process includes glucose uptake, translocation, glucose storage, involves many key kinase, including AMP-activated protein kinase, phosphoinositide 3-kinase, glycogen synthase kinase, and so on. Further studies will be carried out to elucidate the mechanism of action of these and other kaempferol derivatives on glucose consumption.

Acknowledgments: Part of this research was supported by Programs for New Century Excellent Talents in University (NCET-12-1069), Tianjin Innovative Research Team in University (TD12-5033), and Changjiang Scholars and Innovative Research Team in University (PCSIRT IRT_14R41).

Author Contributions: Yi Zhang designed the research and wrote the manuscript; Jia Hao, Jian Li, and Xiaoxia Li performed the experimental work; Yanxia Liu retrieved literature; Jingya Ruan perfected language. All authors discussed, edited and approved the final version.

Conflicts of Interest: The authors declare no conflict of interest.

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Sample Availability: Samples of the compounds are available from the authors.



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