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Data Article

Data for β -lactoglobulin conformational analysis after (-)-epigallocatechin gallate and metal ions bindingLiangliang Zhang^{a,*}, Indra Dev Sahu^b, Man Xu^a, Yongmei Wang^a, Xinyu Hu^a^a Key Lab. of Biomass Energy and Material, Jiangsu Province; Institute of Chemical Industry of Forest Products, CAF, Nanjing 210042, China^b Chemistry and Biochemistry Department, Miami University, Oxford, OH 45056, USA

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ABSTRACT

This data article contains complementary results related to the paper “Effect of metal ions on the binding reaction of (-)-epigallocatechin gallate to β -lactoglobulin” (Zhang et al., 2017) [1]. Data was obtained by circular dichroism (CD) spectroscopy to investigate potential β -lactoglobulin (β -Lg) conformational changes with different concentrations of EGCg and Cu^{2+} or Al^{3+} added to β -Lg. 500 μL of the 25 μM β -Lg solution containing EGCg (25 μM) or metal ions (0–500 μM) were measured, and the spectra were recorded. CD spectroscopy data present in this article indicated that the β -Lg-Cu, β -Lg-Al and β -Lg-EGCg interaction resulted in unfolding of the secondary structure of β -Lg.

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Specifications Table

Subject area	Chemistry
More specific subject area	Polyphenol chemistry

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* Corresponding author.

E-mail address: zhli20086@163.com (L. Zhang).<http://dx.doi.org/10.1016/j.dib.2016.12.021>2352-3409/© 2016 The Authors. Published by Elsevier Inc. This is an open access article under the CC BY license (<http://creativecommons.org/licenses/by/4.0/>).

Type of data	Figure
How data was acquired	MOS-500 spectropolarimeter (Bio-Logic, France)
Data format	Analyzed
Experimental factors	CD spectroscopy was performed with the method of Li et al. [2].
Experimental features	All samples were prepared in 20 mM PBS buffer at pH 7.4. 500 μ L of the 25 μ M β -Lg solution containing EGCg (25 μ M) or metal ions (0–500 μ M) were measured, and the spectra were recorded.
Data source location	Nanjing, China
Data accessibility	Data is with this article

Value of the data

- The data provides some additional data on the effects of metal ions on the binding reaction of EGCg to β -Lg.
- The data indicated the conformational change of β -Lg after binding with EGCg or metal ions Cu, Al.
- The interaction between [β -Lg-Cu] and [β -Lg-Al] results in unfolding of the secondary structure of β -Lg.
- This data provide insights in understanding the effects of metal ions on the binding reaction of polyphenol compounds to β -Lg.

1. Data

Fig. 1 reports the CD spectra of β -Lg with different concentrations of EGCg or Cu^{2+} or Al^{3+} . The negative bands at 222 nm could indicate the α -helix structure of the proteins [1,3].

2. Experimental design, materials and methods

2.1. Materials

EGCg ($\geq 95\%$) and β -Lg (A variant, purity $\geq 90\%$) were purchased from Sigma-Aldrich Co. (St. Louis, MO, USA). Working solutions of EGCg (0.25 mM) were prepared by dissolving the EGCg in a 50% methanol solution. The working solution of β -Lg (25 μ M) was prepared in 20 mM PBS buffer, pH 7.4 and stored in a refrigerator prior to use. The β -Lg and EGCg concentrations were determined spectrophotometrically by their extinction coefficients: $\epsilon_{280}(\beta\text{-Lg}) = 17600 \text{ M}^{-1} \text{ cm}^{-1}$ and $\epsilon_{280}(\text{EGCg}) = 9700 \text{ M}^{-1} \text{ cm}^{-1}$ at 280 nm [4,5]. For *in vitro* experiments, the working solutions of Cu^{2+} and Al^{3+} (1.0 mM) were prepared by dissolving $\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$ and AlCl_3 , respectively, in double-distilled water containing 0.1 M HCl to facilitate dissolution. All other reagents and solvents were of analytical reagent grade and used without further purification. All aqueous solutions were prepared using freshly double-distilled water.

2.2. Experimental design

CD spectroscopy was performed using a MOS-500 spectropolarimeter (Bio-Logic, France) with the modified method of Li et al. [2]. The CD spectra of the β -Lg, [β -Lg-EGCg] and [β -Lg-metal] systems were recorded between 190 and 250 nm by scanning the spectrum at 25 $^{\circ}\text{C}$, with a scanning speed of 100 nm min^{-1} , 2 s response time, and 1.0 nm step size. All samples were prepared in 20 mM PBS buffer at pH 7.4. To investigate the effect of EGCg, Cu^{2+} and Al^{3+} on the secondary structure of β -Lg, 500 μ L of the 25 μ M β -Lg solution containing EGCg (25 μ M) or metal ions (0–500 μ M) were measured, and the spectra were recorded. The samples were loaded into a rectangular quartz cuvette with a path

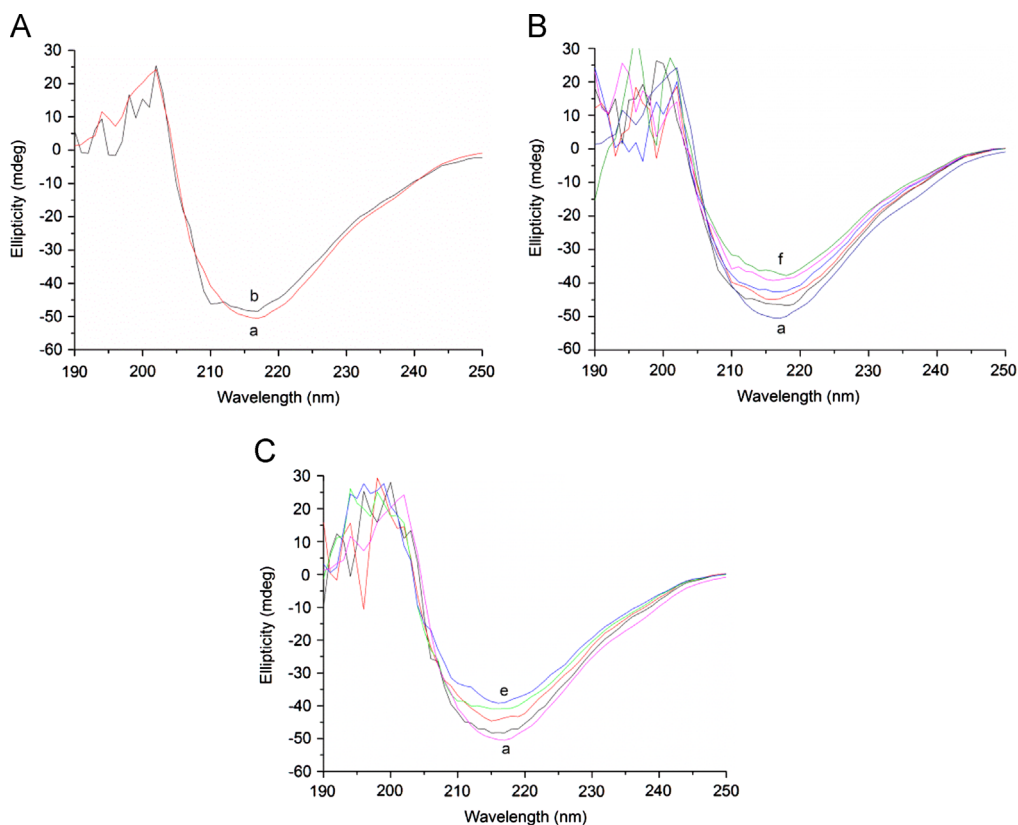


Fig. 1. (A) CD spectra of β -Lg-EGCg system. a, (25 μ M β -Lg), b (25 μ M EGCg); (B) CD spectra of β -Lg-Cu system. a, (25 μ M β -Lg), c(Cu^{2+}): a (0), b (100 μ M), c (200 μ M), d (300 μ M), e (400 μ M), f (500 μ M); (C) CD spectra of β -Lg-Al system. a (25 μ M β -Lg), c(Al^{3+}): a (0), b (100 μ M), c (200 μ M), d (300 μ M), e (400 μ M).

length of 1 mm. The spectra of three consecutive scans were averaged and corrected by subtracting the solvent/buffer spectra.

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Transparency document. Supplementary material

Transparency data associated with this article can be found in the online version at <http://dx.doi.org/10.1016/j.dib.2016.12.021>.

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