

Article

Polyoxygenated Klysimplexane- and Eunicellin-Based Diterpenoids from the Gorgonian *Briareum violaceum*

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Abstract: Three new polyoxygenated diterpenoids with a rare 4-isopropyl-1,5,8a-trimethylperhydrophenanthrene structure of the klysimplexane skeleton, briarols A-C (1-3), and one eunicellin-based diterpenoid, briarol D (4), were isolated from *Briareum violaceum*, a gorgonian inhabiting Taiwanese waters. The chemical structures of these compounds were determined by employing extensive analyses of NMR and high-resolution electrospray ionization mass spectrometry (HRESIMS) data. Metabolites 1-3 were found to possess the rarely found skeleton of the diterpenoid klysimplexin T. All isolated compounds showed very weak cytotoxic activity against the growth of three cancer cell lines. A plausible biosynthetic pathway for briarols A-C from the coexisting eunicellin diterpenoid briarol D (4) was postulated.

Keywords: *Briareum violaceum*; klysimplexane; briarol; eunicellin; gorgonian



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1. Introduction

Gorgonian corals belonging to genus *Briareum* (Cnidaria, Octocorallia, Briareidae) inhabiting the western Pacific Ocean and Caribbean waters have been found to be a rich source of diterpenoids [1,2] possessing fused bicarbocyclic structures of briarane [3–7], eunicellin [8–11], and asbestinane [12–14] types, in addition to cembranoids [15,16]. Many of these metabolites exhibit a wide range of bioactivities, including anti-inflammatory [3,11,17–20], cytotoxic [14,21–23], antiviral [14,21,24], anti-malarial [8], antimicrobial [14], and analgesic [20] activities. Our previous study on the chemical constituents of *Briareum violaceum* afforded the isolation of briarellins (2,9:3,16-diepoxyeunicellins), which were shown to possess interesting structures generated from intramolecular cyclization of corresponding cembranoids [10]. In our efforts to discover new natural products from marine organisms, a continuous chemical investigation of *B. violaceum* was carried out. The present study led to the discovery of four new diterpenoids. Three of them, briarols A-C (1-3), were identified as compounds of a rare (4-isopropyl-1,5,8a-trimethylperhydrophenanthrene) skeleton, which was discovered for only one time as klysimplexin T in 2011 [25] and is herein denominated as the klysimplexane skeleton (Figure 1). The structure elucidation of the new metabolites was performed by extensive spectroscopic analyses, including two-dimensional (2D) NMR correlation and high-resolution electrospray ionization mass spectrometry (HRESIMS)

analyses. A plausible biosynthetic pathway was suggested and the cytotoxicity of the new compounds was evaluated.

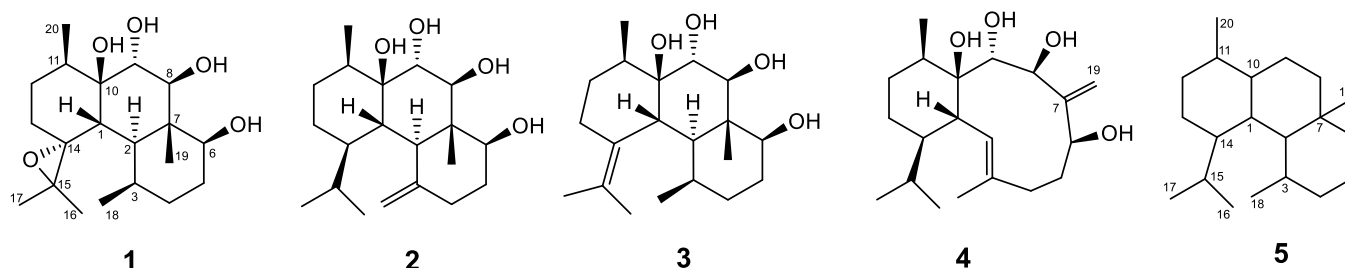


Figure 1. Structures of new diterpenoids isolated from *B. violaceum* (1–4) and the klysimplexane skeleton (5).

2. Results and Discussion

The lyophilized organism was extracted with ethyl acetate (EtOAc) followed by chromatographic fractionation of solvent-free extract on silica (Si) gel. Fractions showing ^1H NMR signals characteristic of polyoxygenated terpenoids were separated mainly by a reverse-phase (RP) column and high-performance liquid chromatography (RP-HPLC), yielding diterpenoids 1–4 (Figure 1). The spectra of these compounds are given in the Supplementary Materials (Figures S4–S40). The IR absorption bands at ν_{max} 3413–3464 cm^{-1} and the four ^{13}C NMR signals, resonating at the region of δ_{C} 70.7 to 81.9 ppm, disclosed the multi-hydroxylated pattern of the isolated compounds.

Briarol A (**1**) was obtained as a white powder with an optical rotation of $[\alpha]_{\text{D}}^{25} = -101.9$ (c 0.24, CHCl_3). The sodiated ion peak at m/z 377.2300 $[\text{M} + \text{Na}]^+$ in the HRESIMS established a molecular formula of $\text{C}_{20}\text{H}_{34}\text{O}_5$ for **1**, appropriate for four degrees of unsaturation. The IR absorption at ν_{max} 3430 cm^{-1} revealed the presence of hydroxy functionality. As the ^1H NMR spectrum, measured in CDCl_3 , showed nine overlapped proton signals at δ_{H} 1.50–1.80 ppm, we remeasured **1** in C_6D_6 and acetone- d_6 to allow better signal resolution and to facilitate integrated 2D NMR correlation analyses (Figure 2 and Figures S2–S19). The ^{13}C NMR spectrum of **1**, combined with distortionless enhancement by polarization transfer (DEPT) and heteronuclear single quantum correlation (HSQC) spectra, displayed 20 sp^3 -hybridized carbon signals (δ_{C} 10.9–81.9 ppm) assignable for 5 methyl, 4 methylene, 7 methine, and 4 quaternary carbons (Table 1). Therefore, the four degrees of unsaturation identified metabolite **1** as a tetracyclic diterpenoid. Analyzing proton homonuclear correlation spectroscopy (^1H - ^1H COSY) correlations revealed the presence of three partial structures of consecutive proton systems extending from H-1 and H₃-18 to H-6 through to H-3, from H-8 to H-9, and from H₃-20 to H₂-13 through H-11 (Figure 2). The $^2J_{\text{CH}}$ and $^3J_{\text{CH}}$ correlations, as determined by the heteronuclear multiple bond correlation (HMBC) experiments, established the connectivities of the partial structures, and hence the 6-6-6 tricyclic framework of **1** (Figure 2). The four most downfield-shifted carbon signals in the ^{13}C NMR spectrum (δ_{C} 78.9–81.9) were attributable to four hydroxy-bearing carbons. Thus, the remaining oxygen atom in the molecular formula of **1** together with the two upfield-shifted oxycarbons (δ_{C} 70.9, C and 64.2, C) suggested the presence of a tetrasubstituted epoxy ring. Four ^1H singlets (δ_{H} 4.54, 4.12, 2.94, and 2.32; Table 2), lacking HSQC correlations, were assigned to the protons of four hydroxy groups. Two protons of these (δ_{H} 4.12 and 4.54), exhibiting HMBC correlations with C-5 (δ_{C} 25.4, CH_2) and C-9 (δ_{C} 79.8, CH)/C-1 (δ_{C} 43.5, CH), were recognized as 6-OH and 10-OH, respectively (Figure 2). The HMBC correlations from both H-1 and CH_3 -20 to C-10 (δ_{C} 78.9, C) confirmed the presence of a hydroxy group at C-10. Moreover, the HMBC correlations observed for H₃-19 (δ_{H} 1.09 3H, s) with the oxymethine carbons C-6 and C-8 together with the ^1H - ^1H COSY correlation H-8/H-9 are indicative of the hydroxy groups at C-8 and C-9, respectively (Figure 2). Furthermore, the long-range connectivities from the protons of the tertiary methyls H₃-16 and H₃-17 (δ_{H} 1.08 and 0.97, each 3H, s) and from the angular methine proton H-1 (δ_{H}

1.64, d, $J = 12.0$ Hz) to the oxycarbons C-14 (δ_C 70.9) and C-15 (δ_C 64.2) placed the epoxy group at C-14/C-15. The above findings and other detailed 2D NMR correlation analyses unambiguously established the planar structure of **1** (Figure 2).

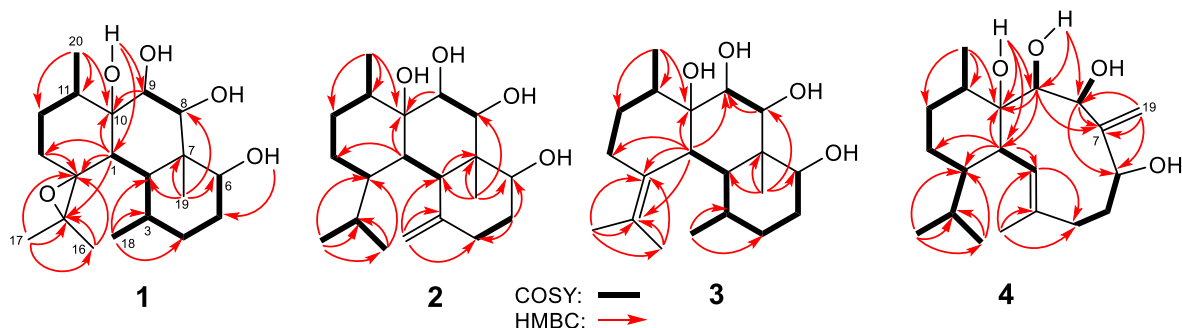


Figure 2. Key proton homonuclear correlation spectroscopy (^1H - ^1H COSY) and heteronuclear multiple bond correlation (HMBC) correlations for (1–4).

Table 1. ^{13}C NMR spectroscopic data of compounds 1–4.

#	1 ^a	2 ^b	3 ^b	4 ^c
1	43.5 (CH) ^d	41.6 (CH)	45.2 (CH)	50.1 (CH)
2	44.3 (CH)	45.6 (CH)	42.3 (CH)	125.5 (CH)
3	28.5 (CH)	144.1 (C)	27.9 (CH)	137.7 (C)
4	32.3 (CH ₂)	36.6 (CH ₂)	31.2 (CH ₂)	38.1 (CH ₂)
5	25.4 (CH ₂)	33.3 (CH ₂)	24.8 (CH ₂)	39.0 (CH ₂)
6	81.9 (CH)	81.5 (CH)	81.4 (CH)	76.2 (CH)
7	44.5 (C)	46.4 (C)	43.8 (CH)	162.8 (C)
8	80.2 (CH)	78.4 (CH)	79.8 (CH)	70.7 (CH)
9	79.8 (CH)	78.2 (CH)	78.8 (CH)	80.9 (CH)
10	78.9 (C)	76.0 (C)	77.2 (C)	78.3 (C)
11	32.9 (CH)	32.1 (CH)	32.3 (CH)	33.2 (CH)
12	30.6 (CH ₂)	33.6 (CH ₂)	33.8 (CH ₂)	33.8 (CH ₂)
13	25.8 (CH ₂)	22.0 (CH ₂)	25.3 (CH ₂)	26.1 (CH ₂)
14	70.9 (C)	43.9 (CH)	128.3 (C)	44.4 (CH)
15	64.2 (C)	28.5 (CH)	128.0 (C)	31.4 (CH)
16	23.6 (CH ₃)	23.3 (CH ₃)	20.9 (CH ₃)	22.0 (CH ₃)
17	20.8 (CH ₃)	21.9 (CH ₃)	20.4 (CH ₃)	22.5 (CH ₃)
18	17.2 (CH ₃)	111.8 (CH ₂)	15.2 (CH ₃)	18.3 (CH ₃)
19	10.9 (CH ₃)	9.2 (CH ₃)	9.9 (CH ₃)	113.8 (CH ₂)
20	17.9 (CH ₃)	17.2 (CH ₃)	17.1 (CH ₃)	17.8 (CH ₃)

Spectrum recorded at ^a 100 MHz in C_6D_6 , ^b 125 MHz in CDCl_3 , and ^c 150 MHz in acetone- d_6 . ^d Attached protons were deduced by distortionless enhancement by polarization transfer (DEPT) and heteronuclear single quantum correlation (HSQC) experiments.

The relative configurations of the 10 chiral carbons in **1** were mostly deduced by examining nuclear Overhauser effect (NOE) correlations (Figure 3). The large $^3J_{\text{H-H}}$ value of the ring juncture protons H-1 and H-2 (12.0 Hz) must be due to *anti* orientation of the two axial protons, which were assumed to be on the β - and α -faces of the molecule, respectively. Therefore, the key NOE interactions of H-1 with H₃-18, H₃-19, H-9, and 10-OH (red-colored arrows) revealed these protons to be cofacial, indicating the α -oriented hydroxy group at C-9 and hence the S^*,R^*,S^*,R^* , R^* -configurations at C-1, C-3, C-7, C-9, and C-10, respectively. Consequently, the NOE correlations found for H-3 with H-2, H-2 with H-6 and H-8, and H-8 with H-11 (blue-colored arrows) designated the S^*,S^*,S^*,R^* -configurations at C-2, C-6, C-8, and C-11, respectively. However, the NOE interaction of H-1 with H₃-16 could not be used for effective elucidation of the relative configuration at C-14. Fortunately, the NOE correlations for H₃-20/H-12 β and H-12 β /H₃-17 were observed in the Nuclear Overhauser Effect Spectroscopy NOESY spectra of **1**, measured in both CDCl_3 and acetone- d_6 , and estab-

lished the α -orientation of the 14,15-epoxy group. Therefore, briarol A (**1**) could be defined as (1S*,2S*,3R*,6S*,7S*,8S*,9R*,10R*,11R*,14R*)-14:15-epoxy-klysimplexan-6,8,9,10-tetrol.

Table 2. ^1H NMR spectroscopic data of compounds 1–4.

No	1 ^a	2 ^b	3 ^b	4 ^c
1	1.64 d (12.0) ^d	2.08 m	2.81 d(7.5)	2.52 dd (10.0, 3.0)
2	1.23 dd (12.0, 3.6)	2.32 d (12.5)	1.62 m	5.35 d (10.0)
3	1.40 m	-	1.54 m	-
4 α	1.21 m	2.23 dd (7.5, 5.0)	1.52 m	2.15 m
4 β	1.28 m	2.00 m	1.52 m	2.26 m
5 α	1.80 m	1.93 dd (12.0, 5.0)	1.61–1.64 m	1.91 m
5 β	1.80 m	2.34 m	1.61–1.64 m	2.04 m
6	3.54 dd (8.0, 8.0)	3.80 dd (11.5, 5.0)	3.61 dd (11.0,4.5)	4.08 d (7.5)
8	3.44 d (11.2)	3.84 d (11.0)	3.64 d (11.0)	4.21 d (4.5)
9	3.49 d (11.2)	3.56 d (11.0)	3.66 d (11.0)	3.46 d (5.5)
11	1.79 m	1.99 m	2.14 m	1.89 m
12 α	1.79 m	1.32 m	1.55 m	1.41 m
12 β	1.81 m	1.54 m	1.14 m	1.46 m
13 α	1.06 m	1.56 m	2.47 m	1.59 m
13 β	1.22 m	1.56 m	1.64 m	1.59 m
14	-	1.73 m	-	1.54 m
15	-	2.11 m	-	1.25 m
16	1.08 3H, s	0.92 3H, d (6.5)	1.81 3H, s	0.82 3H, d (6.0)
17	0.94 3H, s	0.73 3H, d (6.5)	1.74 3H, s	0.81 3H, d (6.0)
18	0.68 3H, d (7.2)	5.02 s 4.81 s	0.80 3H, d (7.5)	1.57 3H, s
19	1.09 3H, s	0.93 3H, s	1.11 3H, s	5.43 d (1.2) 5.37 d (1.2)
20	1.28 3H, d (5.6)	1.04 3H, d (6.5)	1.07 3H, d (6.0)	0.92 3H, d (6.6)
6-OH	4.12, br s		-	3.62 br s
8-OH	2.94, br s		-	4.26 br d (4.5)
9-OH	2.32, br s		2.91 br s*	4.00 d (5.5)
10-OH	4.54, br s		3.18 br s*	3.57, s

Spectrum recorded at ^a 400 MHz in C₆D₆, ^b 500 MHz in CDCl₃, and ^c 600 MHz in acetone-*d*₆. ^d *J* values (Hz). * Exchangeable data.

Briarol B (**2**) was obtained as a white powder. It possessed the molecular formula of C₂₀H₃₄O₄ as indicated by the adduct ion peak at *m/z* 361.2348 [M + Na]⁺ in its HREIMS, with 16 mass units fewer than that of **1**. A comparison of the ¹³C and ¹H NMR data of **2** with those of **1** (Tables 1 and 2, respectively) revealed the presence of another klysimplexane-based metabolite. However, the NMR spectroscopic data of **2** showed the appearance of an olefinic double bond (δ_{C} 144.1, C and 111.8, CH₂; δ_{H} 5.02 and 4.81, each 1H, s) and the absence of the epoxy group. Thus, the two oxycarbons (δ_{C} 70.9 and 64.2, each C) and the methylated methine carbons (δ_{C} 17.2, CH₃ and 28.5, CH) in **1** were replaced by the carbons of two methines (δ_{C} 43.9 and 28.5, each CH) and a 1,1-disubstituted double bond in **2**, respectively. These carbons were then assigned as C-14, C-15, C-3, and C-18, respectively, from the 2D NMR correlation analyses of **2** (Figure 2). Therefore, the gross structure of compound **2** was recognized as klysimplexan-3(18)-en-6,8,9,10-tetrol. The investigation of NOE correlations of **2** (Figure 3) resulted in the same relative configurations at C-1, C-6, C-7, C-8, C-9, C-10, and C-11 as those of **1**. Furthermore, the NOE interactions found for the β -oriented H-1 (δ_{H} 2.08, m) with H₃-16 (δ_{H} 0.92, 3H, d, *J* = 6.5 Hz), H₃-16 with one of the exo-methylene protons (δ_{H} 4.81, s, H-18b), and H-18b with H-1 favored the β -orientation for the 14-isopropyl group and thus the *R** configuration at C-14. These findings together with detailed 2D NMR correlations (Figures 2 and 3) unambiguously established compound **2** as (1S*,2R*,6S*,7S*,8S*,9R*,10R*,11R*,14R*)-klysimplexan-3(18)-en-6,8,9,10-tetrol.

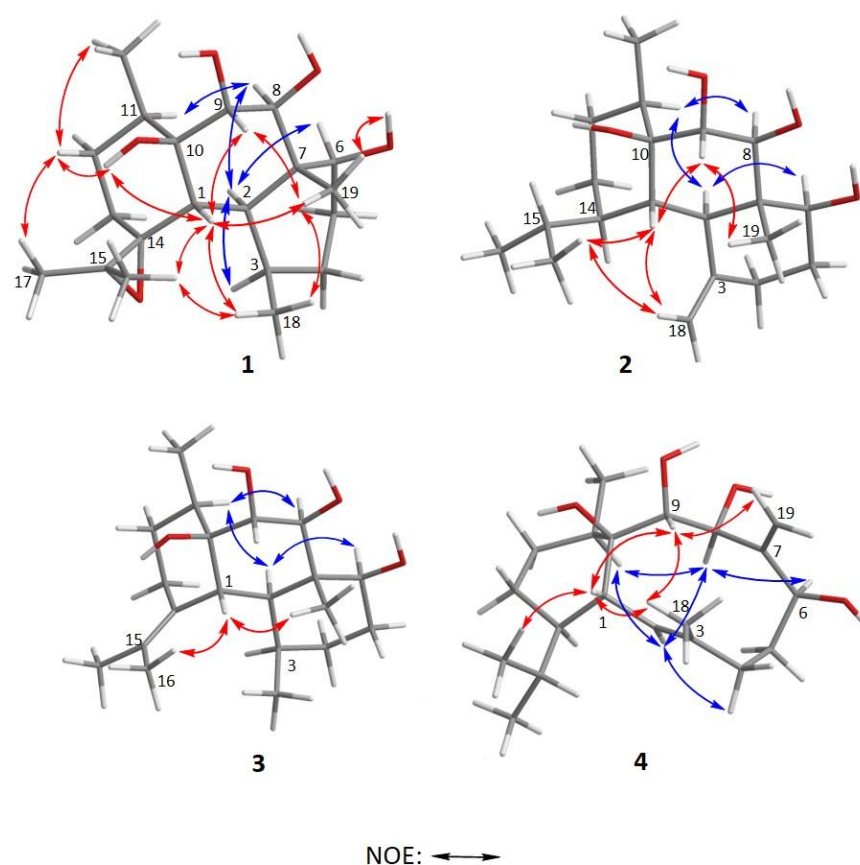


Figure 3. Selected nuclear Overhauser effect (NOE) correlations for (1–4).

Briarol C (3) was also isolated as a white powder that gave a pseudomolecular ion peak at 361.2347 $[M + Na]^+$ in the HRESIMS, consistent with the molecular formula $C_{20}H_{34}O_4$ and four degrees of unsaturation, as with 2. The ^{13}C NMR spectroscopic data of 3 were found to be in accordance with those of 1 from C-1 to C-13 and C-16 to C-20, except for the presence of a tetrasubstituted double bond (δ_C 128.3 and 128.0, each C) instead of the tetrasubstituted epoxy group in 1 (Table 1). Thus, the $^2J_{CH}$ and $^3J_{CH}$ correlations displayed by the two olefinic methyl protons (δ_H 1.81 and 1.74, each 3H, s) with the olefinic carbons (δ_C 128.3 and 128.0, each C), which in turn were correlated with H-1 (δ_H 2.81, d, $J = 7.5$ Hz), confirmed the presence of a 14,15-double bond (Figure 2). The relative configuration of 2 was deduced from NOESY correlations, as illustrated in Figure 3. Furthermore, it was found that the ^{13}C NMR chemical shifts of C-1 to C-11 in 1 and 1H NMR data of H-6, H-8, and H-9 in 1 and 2 (Table 2) were analogous to those of 3, reflecting the same β -orientation for H-1, 6-OH, 8-OH, 10-OH, H₃-18, H₃-19, and H₃-20, and the α -orientation for H-2 and 9-OH. Therefore, compound 3 was clearly identified as (1*R**,2*S**,3*R**,6*S**,7*S**,8*S**,9*R**,10*R**,11*R**)-klysimplexan-14(15)-en-6,8,9,10-tetrol.

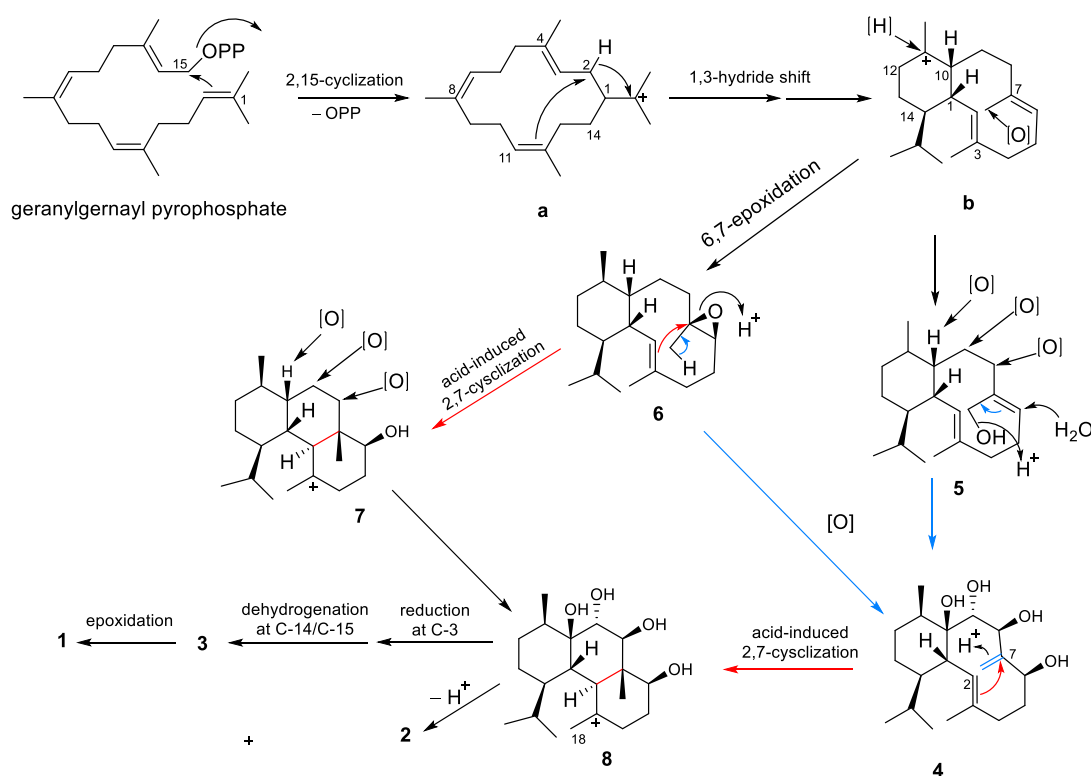
Briarol D (4) was obtained as a white powder and gave a sodiated ion peak at m/z 361.2349 $[M + Na]^+$ by HREIMS, appropriate for a molecular formula of $C_{20}H_{34}O_4$ and four degrees of unsaturation. The ^{13}C NMR and DEPT spectra indicated the presence of 20 carbon signals (Table 1) corresponding to 4 methyls, 5 methylenes (including 1 exomethylene), 8 methines (including 1 olefinic and 3 oxymethines), and 3 quaternary carbons (2 olefinic and 1 oxycarbon) of a diterpenoid. The NMR spectroscopic data (Tables 1 and 2) revealed the presence of a trisubstituted [δ_C 125.5, CH, 137.7, C; δ_H 5.35 (d, $J = 10.0$ Hz)] and an 1,1-substituted [δ_C 162.8, C, 113.8, CH₂ and δ_H 5.43, 5.37 (each d, $J = 1.2$ Hz)] double bond. The remaining two degrees of unsaturation were thus attributed to a bicyclic structure for 4. This was further substantiated by the NMR data comparison of 4 with those of 1–3, which showed the substitution of one ring-juncture methine (2-CH) with an olefinic methine [δ_H/δ_C 5.35 (d, $J = 10.0$ Hz)/125.5, CH] in 4. However, the 1H and

^{13}C NMR data pointed out the presence of three hydroxy-bearing methines [$\delta_{\text{H}}/\delta_{\text{C}}$ 4.21 (d, $J = 4.5$ Hz, H-8)/70.7; 4.08 (d, $J = 7.5$ Hz, H-6)/76.2; and 3.46 (d, $J = 5.5$ Hz, H-9)/80.9], as in the case of compounds 1–3. Moreover, two protons resonating at δ_{H} 4.26 (d, $J = 4.5$ Hz) and 4.00 (d, $J = 5.5$ Hz) exhibited COSY correlations with H-8 and H-9 due to 8-OH and 9-OH, respectively. The gross structure of 4 as a eunicellin-derived diterpenoid [26], including the positions of the two olefinic bonds and the four hydroxy groups, was further resolved by the study of the long-range proton–carbon correlations (Figure 2). In particular, the HMBC correlations found from the only available ring-juncture proton (δ_{H} 2.52, dd, $J = 10.0, 3.0$ Hz, H-1) to C-2 (δ_{C} 125.5, CH) and C-10 (δ_{C} 78.3, C), from the olefinic methyl protons (δ_{H} 1.57, s, H₃-18) to C-2, C-3 (δ_{C} 137.7, C), and C-4 (δ_{C} 38.1, CH₂) and from the exomethylene protons (δ_{H} 5.43, 5.37, each d, $J = 1.2$ Hz, H₂-19) to C-6 (δ_{C} 76.2, CH), C-7 (δ_{C} 162.8, C), and C-8 (δ_{C} 70.7, CH), positioned the trisubstituted and 1,1-disubstituted double bonds at C-2/C-3 and C-7, respectively. Based on the above findings and detailed 2D NMR correlations (Figure 2), the molecular framework of 4 was established.

An inspection of NOESY correlations (Figure 3) enabled us to assign the relative configurations of the seven chiral carbons C-1, C-6, C-8, C-9, C-10, C-11, and C-14 in 4. The NOE correlations observed for the β -oriented ring-juncture proton H-1 with the protons of the 9-oxymethine and one of the 14-isopropyl methyls reflected the α -orientation of H-14 and 9-OH. Furthermore, the NOE observed for H-1/H₃-18 and H-2/H-4 combined with upfield chemical shift ($\delta_{\text{C}} < 20$ ppm) observed for C-18 (δ_{C} 18.3 ppm) determined the *E*-geometry of the olefinic bond [27] at C-2/C-3. This finding placed the olefinic H-2 on the α -face of the molecule. Consequently, the NOE interactions found for H-2 with H-8 and H-8 with both H-6 and H-11 revealed the β -orientation for H-6, H-8, H-10, and H₃-20. Compound 4 was thus unambiguously identified as (1*S**,2*E*,6*S**,8*S**,9*R**,10*R**,11*R**)-eunicellin-2,7(19)-dien-6,8,9,10-tetrol.

Based on the above discoveries, it is proposed that compounds 1–4 can be derived from the common eunicellin intermediate (b) after the 2,11-cyclization and 1,3-hydride shift of a cembranoid cation (a). Oxidation of CH₂-8, CH₂-9, CH-10, and CH₃-18 followed by acid-catalyzed hydroxylation at the olefinic C-6 with a subsequent formation of an exomethylene at C-7 in the intermediate 5 yields 4. Furthermore, the 6,7-epoxidation for the intermediate b gives 6 as the intermediate of metabolite 4 and the tricyclic carbonium ion 7. Both 4 and 7 could be further converted into carbonium ion 8, as shown in Scheme 1. Deprotonation at C-18 in 8 can produce 2, while reduction at C-3 and dehydrogenation at C-14/C-15 in 8 gives 3. Subsequently, the epoxidation of the olefinic double bond in metabolite 3 affords 1 (Scheme 1). To the best of our knowledge, the biosynthesis of the klysimplaxane- and eunicellin-type diterpenoids is limited to marine invertebrates, and there are no analogous structures in terrestrial natural products.

The *in vitro* cytotoxicity of the new diterpenoid metabolites (1–4) was assessed against the cancer cell lines of human colon cholangiocellular carcinoma (HuCC-T1), human colon carcinoma (HT-29), and human colon adenocarcinoma (DLD-1). The results showed that all compounds only exhibited very weak cytotoxicity against the tested cancer cells, with the IC₅₀ values ranging from 220.75 to 238.88 μM as compared to doxorubicin hydrochloride (IC₅₀ 1.38 to 2.24 μM). Because of the low yield (< 2.5 mg) and the consumption of the isolated metabolites in measurements of spectroscopic data and cytotoxicity, we suggest that further investigation on other biological activities should be carried out once these tetrahydroxylated diterpenoid molecules, in particular those with the rare klysimplaxane skeleton, can be obtained in sufficient quantities.



Scheme 1. A plausible biosynthetic pathway for 1–4.

3. Materials and Methods

3.1. General Experimental Procedures

IR spectra and optical rotations were measured on JASCO FT/IR-4100 spectrophotometer and JASCO P-1020 polarimeter (JASCO Corporation, Tokyo, Japan), respectively. LRESIMS and HRESIMS spectra were measured on Bruker APEX II mass spectrometer (Bruker, Bremen, Germany). ^1H and ^{13}C NMR spectra were measured on Varian Unity INOVA 600 FT-NMR (or 500 or 400 FT-NMR) instruments (Varian Inc., Palo Alto, CA, USA) at 600 MHz (or 500 or 400 MHz) for ^1H and 150 MHz (or 125 or 100 MHz) for ^{13}C in CDCl_3 or CD_3OD or acetone- d_6 . Silica (Si) gel (230–400 mesh) (Merck, Darmstadt, Germany) and C18 reverse-phase Si gel (RP-18; 40–63 μm) (Parc-Technologique Blvd, Quebec, Canada) were used for column chromatography. Thin-layer chromatography (TLC) analyses were achieved using precoated Si gel (Kieselgel 60 F-254, 0.2 mm) plates (Merck, Darmstadt, Germany). Further purification and the separation of compounds were performed by reverse-phase high-performance liquid chromatography (RP-HPLC) on a Hitachi L-2455 HPLC instrument with a Supelco C18 column (250 \times 21.2 mm, 5 μm) (Supelco Inc., Bellefonte, PA, USA).

3.2. Animal Material

The soft coral *B. violaceum* was collected from Jihui Fish Port, Taitung, Taiwan, identified, and extracted as described before [10]. A voucher specimen was taken and deposited at the Department of Marine Biotechnology and Resources, National Sun Yat-sen (NSYSU) University, Kaohsiung.

3.3. Extraction and Isolation

The lyophilized bodies of soft coral (500 g, wet weight) were crushed and extracted with EtOAc. The EtOAc extract (3.9 g) was fractionated with Si gel column chromatography (CC) using EtOAc-hexane (0:100 to 100:0, gradient). Polar fractions eluted with EtOAc-hexane (10:1), which showed the diagnostic ^1H NMR (methyl and oxymethine) signals

of polyoxygenated terpenoids, were combined and subfractionated on Si gel CC using acetone-hexane (1:2.5), affording the subfractions F1 and F5. Subfraction F4 was separated on RP-18 Si gel CC using acetyl nitrite (CH₃CN)-H₂O (1.5:1 then 1.2:1) to give compounds **2** (1.5 mg), **3** (2.0 mg), and **4** (2.2 mg), respectively. Compound **1** (2.4 mg) was obtained from subfraction F5 with a 3-step purification process with RP-18 Si gel CC using MeOH-H₂O (1.5:1 then 5:1), RP-HPLC using CH₃CN-H₂O (1:2), and then on Si gel CC using acetone-hexane (1:5).

Briarol A (**1**). White powder; $[\alpha]_D^{25} -101.9$ (c 0.24, CHCl₃); IR (neat) ν_{\max} 3430, 2927, 2853, and 1382 cm⁻¹; ¹³C NMR (100 MHz, C₆D₆) and ¹H NMR (400 MHz, C₆D₆). See Tables 1 and 2, respectively. ¹³C NMR (100 MHz, CDCl₃) δ_C 81.3 (CH, C-6), 79.3 (CH, C-8), 78.7 (CH, C-9), 78.3 (C, C-10), 70.4 (C, C-14), 64.1 (C, C-15), 43.7 (C, C-7), 43.7 (CH, C-2), 42.6 (CH, C-1), 32.0 (CH, C-11), 31.5 (CH₂, C-4), 29.6 (CH₂, C-12), 29.6 (CH₃, C-17), 27.6 (CH, C-3), 25.1 (CH₂, C-13), 24.2 (CH₂, C-5), 23.2 (CH₃, C-16), 17.0 (CH₃, C-20), 16.6 (CH₃, C-18), 9.9 (CH₃, C-19); ¹H NMR (400 MHz, CDCl₃) δ_H 4.33, 3.94, 3.23, and 2.65 (each 1H, br s, 6-OH, 8-OH, 9-OH, and 10-OH), 3.64 (1H, d, $J = 11.2$ Hz, H-8), 3.61 (1H, dd, $J = 10.4, 5.2$ Hz, H-6), 3.56 (1H, d, $J = 11.2$ Hz, H-9), 2.16 (1H, m, H-11), 1.79 (1H, d, $J = 6.0$ Hz, H-1), 1.73 (1H, m, H-3), 1.72 (1H, m, H-5 β), 1.69 (1H, m, H-12 β), 1.68 (1H, m, H-5 α), 1.67 (1H, m, H-13 β), 1.56 (2H, m, H₂-4), 1.55 (1H, m, H-2), 1.54 (1H, m, H-12 α), 1.45 (3H, s, H₃-16), 1.39 (1H, m, H-13 α), 1.33 (3H, s, H₃-17), 1.15 (3H, d, $J = 6.4$ Hz, H₃-20), 1.06 (3H, s, H₃-19), 0.97 (3H, d, $J = 7.6$ Hz, H₃-18); ¹³C NMR (100 MHz, acetone-*d*₆) δ_C 81.9 (CH, C-6), 80.2 (CH, C-8), 79.0 (CH, C-9), 78.6 (C, C-10), 70.0 (C, C-14), 64.2 (C, C-15), 44.2 (C, C-7), 44.0 (CH, C-2), 43.5 (CH, C-1), 32.4 (CH, C-11), 31.7 (CH₂, C-4), 30.5 (CH₂, C-12), 28.2 (CH, C-3), 25.6 (CH₂, C-13), 25.1 (CH₂, C-5), 23.2 (CH₃, C-16), 20.3 (CH₃, C-17), 17.5 (CH₃, C-20), 16.8 (CH₃, C-18), 10.0 (CH₃, C-19); ¹H NMR (400 MHz, acetone-*d*₆) δ_H 4.30 and 3.88 (each 1H, br s, 8-OH and 9-OH), 4.19 (1H, br s, 6-OH), 4.11 (1H, br s, 10-OH), 3.66 (1H, m, H-6), 3.65 (1H, d, $J = 11.2$ Hz, H-8), 3.49 (1H, br d, $J = 11.2$ Hz, H-9), 2.25 (1H, m, H-11), 1.84 (1H, d, $J = 6.0$ Hz, H-1), 1.79 (1H, m, H-3), 1.72 (1H, d, $J = 6.8$ Hz, H-2), 1.67 (1H, m, H-4 β), 1.57 (1H, m, H-5 β), 1.53 (1H, m, H-12 β), 1.52 (1H, m, H-5 α), 1.50 (1H, m, H-4 α), 1.49 (1H, m, H-12 α), 1.48 (3H, s, H₃-16), 1.41 (1H, m, H-13 β), 1.37 (1H, m, H-13 α), 1.365 (3H, s, H₃-17), 1.16 (3H, d, $J = 6.8$ Hz, H₃-20), 1.06 (3H, s, H₃-19), 1.03 (3H, d, $J = 6.8$ Hz, H₃-18). ESIMS m/z 377 [M + Na]⁺; HRESIMS m/z 377.2300 [M + Na]⁺ (calcd for C₂₀H₃₄O₅Na, m/z 377.2299).

Briarol B (**2**). White powder; $[\alpha]_D^{25} -32.0$ (c 0.15, CHCl₃); IR (neat) ν_{\max} 3464, 2923, 2854, and 1381 cm⁻¹; ¹³C NMR (125 MHz, CDCl₃) and ¹H NMR (500 MHz, CDCl₃). See Tables 1 and 2, respectively. ESIMS m/z 361 [M + Na]⁺, 339 [M + H]⁺ HRESIMS m/z 361.2348 [M + Na]⁺ (calcd for C₂₀H₃₄O₄Na, m/z 361.2349).

Briarol C (**3**). White powder; $[\alpha]_D^{25} -21.3$ (c 0.22, CHCl₃); IR (neat) ν_{\max} 3413, 2925, 2858, and 1374 cm⁻¹; ¹³C NMR (100 MHz, CDCl₃) and ¹H NMR (500 MHz, CDCl₃). See Tables 1 and 2, respectively. ESIMS m/z 361 [M + Na]⁺; HRESIMS m/z 361.2347 [M + Na]⁺ (calcd for C₂₀H₃₄O₄Na, m/z 361.2349).

Briarol D (**4**). White powder; $[\alpha]_D^{25} -29.7$ (c 0.22, CHCl₃); IR (neat) ν_{\max} 3418, 2923, 2853, and 1381 cm⁻¹; ¹³C NMR (150 MHz, acetone-*d*₆) and ¹H NMR (600 MHz, acetone-*d*₆), see Tables 1 and 2, respectively. ESIMS m/z 361 [M + Na]⁺; HRESIMS m/z 361.2349 [M + Na]⁺ (calcd for C₂₀H₃₄O₄Na, m/z 361.2349).

3.4. Cytotoxicity Assay

Cancer cell lines (HT-29, HuCC-T1, and DLD-1) were obtained from the American Type Culture Collection (ATCC). Compounds 1-4 were evaluated for the cytotoxic activity using an Alamar blue assay as previously described [28,29]. The intensity of the produced color was measured at 570 nm using an ELISA plate reader.

4. Conclusions

Three new polyoxygenated diterpenoids of the rare klysimplexane-skeleton, along with a non-ether bridged eunicellin diterpenoid, were discovered from the gorgonian coral *Briareum violaceum* and named briarols A-D, respectively. A possible biosynthetic pathway

for briarols A-C from the coexisting eunicellin diterpenoid was postulated for the first time. Although the compounds did not show potent cytotoxic activity against the tested cancer lines, other possible bioactivities for these metabolites might be worthwhile for further screening. It is noteworthy to mention that this is the first discovery of these rare klysimplexane-type metabolites from a gorgonian coral since the isolation of klysimplexin T from the cultured soft coral *Klyxum simplex* a decade ago.

Supplementary Materials: Figure S1. HRESIMS spectrum of 1; Figures S2–S7: 1D and 2D NMR spectra of 1 in C₆D₆; Figures S8–S13: 1D and 2D NMR spectra of 1 in CDCl₃; Figures S14–S19: 1D and 2D NMR spectra of 1 in acetone-*d*₆; Figure S20. HRESIMS spectrum of 2; Figures S21–S26: 1D and 2D NMR spectra of 2 in CDCl₃; Figure S27. HRESIMS spectrum of 3; Figures S28–S33: 1D and 2D NMR spectra of 3 in CDCl₃; Figure S34. HRESIMS spectrum of 4; Figures S35–S40: 1D and 2D NMR spectra of 4 in acetone-*d*₆.

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