



Article A Short and Efficient Total Synthesis of Ficuseptamines A and B

Hani Mutlak A. Hassan

King Fahd Medical Research Center, King Abdulaziz University, P.O. Box 80216, Jeddah 21589, Saudi Arabia; hmahassan@kau.edu.sa

Received: 11 July 2018; Accepted: 23 July 2018; Published: 26 July 2018



Abstract: A rapid and efficient total synthesis of ficuseptamines A and B by a cross metathesis strategy is described.

Keywords: ficuseptamines A and B; cross metathesis; total synthesis

1. Introduction

Ficuseptamines A, B, and C (1a-c) were isolated from the leaves of *Ficus septica* and reported by Shin-ya and co-workers in 2009 (Figure 1) [1]. Ficuseptamines A and B possess an aminocaprophenone structure, while ficuseptamine C contains a pyrrolidine moiety in its structure. After their isolation, these alkaloids were evaluated for cytotoxicity against HeLa (human cervical carcinoma) and ACC-MESO-1 (malignant pleural mesothelioma) cancer cell lines. Ficuseptamine A displayed IC_{50} values of 57 µM and 160 µM against HeLa and ACC-MESO-1 cells lines, respectively. Ficuseptamine B showed better cytotoxicity against the same cell lines (IC₅₀: 23 µM for HeLa; 72 µM for ACC-MESO-1), while ficuseptamine C showed no activity against either cell line. The aryl ketone motif, which is present in ficuseptamines A and B, is frequently found in natural products [2,3] and biologically active molecules [4]. We were interested in devising a novel strategy targeting ficuseptamines A and B for their first total synthesis. We thought that designing an efficient and facile strategy for their rapid total synthesis would offer the possibility of synthesizing ficuseptamine A and B analogues for biological evaluation, given the commercial availability of a wide variety of functionalized terminal olefins, which could be utilized in library design and synthesis. In addition, the incorporation of fluorine atom(s) into ficuseptamines A and B to synthesize fluorinated analogues could also significantly improve their biological activity [5,6].



Figure 1. Structures of ficuseptamine A, B, and C (1a-c).

Olefin metathesis is one of the most useful carbon–carbon bond-forming reactions, and it has found tremendous use in organic chemistry for the construction of a myriad of organic molecules [7–10].

This highly powerful transformation has also been elegantly utilized in the total synthesis of numerous natural products [11,12]. Olefin metathesis catalysts that promote this transformation are displayed in Figure 2. Cross metathesis (CM) is one of the most popular transformations for connecting two independent olefins together to form a more complex olefinic product, which could require several steps to synthesize if a different methodology was employed. Although ring-closing metathesis (RCM) has found extensive use in the total synthesis of numerous natural products, CM has become increasingly popular in the field of total synthesis [13–15], particularly after the discovery of *Z*-selective olefin metathesis catalysts [16]. As part of our interest in the power of olefin metathesis to facilitate useful transformations [17], we report herein a highly efficient methodology for the total synthesis of ficuseptamines A and B, utilizing a cross metathesis approach.



Figure 2. Ruthenium-based olefin metathesis catalysts.

2. Results and Discussion

A retrosynthetic analysis of ficuseptamines A and B is depicted in Scheme 1. We commenced the synthetic work by first targeting ficuseptamine A for total synthesis. Aryl ketone **9a** was prepared in two sequential steps from commercially available alcohol **11** via halogenation and dehydrohalogenation reactions (Scheme 2).



Scheme 1. Retrosynthetic analysis of ficuseptamines A and B.

Thus, primary alcohol **11** was transformed into its bromo counterpart (i.e., **12**), under the influence of hydrobromic acid to furnish **12** in high yield (93%). Subsequent treatment of the brominated

product **12** with DBU (1,8-diazabicyclo[5.4.0]undec-7-ene) delivered CM precursor **9a** in 69% yield through dehydrobromination.



Scheme 2. Bromination of 11 followed by dehydrobromination with DBU to give CM precursor 9a.

Next, we explored the key CM reaction between **9a** and *N*,*N*-dimethyl-4-pentene-1-amine **10** to afford the α , β -unsaturated CM product **8a**. We initially performed the CM reaction using Grubbs I catalyst **4** (5 mol %) with CH₂Cl₂ as a solvent to afford CM product **8a** in 22% yield (entry 1, Table 1). Increasing the Grubbs I catalyst **4** loading (10 mol %) provided **8a** in a disappointing 15% yield (entry 2, Table 1). We then turned our attention to Grubbs II catalyst **5**, which gave the CM product **8a** in modest 37 and 41% yields using 5 mol % and 10 mol % catalyst loading in CH₂Cl₂, respectively (entries 3 and 4, Table 1). Switching the solvent from CH₂Cl₂ to toluene and performing the reaction with Grubbs II catalyst **5** (10 mol %) at 80 °C improved the yield to 47% (entry 5, Table 1).

Table 1. Optimization of the CM reaction.



^a All reactions were performed using 9a (2 mmol), 10 (1 mmol), solvent, and temperature for 10 h, except entry 9, in which the reaction was carried out using 9a (1 mmol) and 10 (2 mmol). E = entry.

A report by Grubbs and co-workers described how the presence of the phosphine ligand (PCy₃) in olefin metathesis catalysts can attack the carbine through its dissociation from the metal complex via decomposition [18]. This fact turned our attention to Hoveyda-Grubbs II 7, which lacks the PCy₃ ligand. Pleasingly, the use of Hoveyda-Grubbs II 7 (5 mol %) increased the yield significantly, to afford

8a in a 68% yield using CH₂Cl₂ as the solvent (entry 6, Table 1). Treating **9a** and **10** with an increased Hoveyda-Grubbs II **7** loading (10 mol %) in CH₂Cl₂ provided **8a** in an excellent 76% yield (entry 7, Table 1). In contrast, increasing the temperature to 80 °C and changing the solvent from CH₂Cl₂ to toluene diminished the yield to 52% (compare entry 7 vs. 8, Table 1). The double bond geometry of the CM product **8a** was identified as the (*E*)-configured isomer, indicating that CM proceeded selectively to give the thermodynamically stable (*E*)-isomer exclusively. A change in molar ratio of the coupling partners **9a** and **10**, and performing the reaction using 10 mol % of Hoveyda-Grubbs II **7** in CH₂Cl₂, resulted in a reduction of the yield to 42%, with formation of the homodimer of **10** as a competing side-product (entry 9, Table 1). Grubbs and co-workers categorized olefin metathesis substrates as Type I, Type II, Type III, or Type IV [19]. According to Grubbs' selectivity model, terminal olefin **10** is a Type I olefin substrate, which usually undergoes fast homodimerization, while olefin **9a** is a Type II olefin, which undergoes slow homodimerization. Thus, such a change in the ratio of the coupling partners **9a** and **10** has promoted homodimerization of **10** to occur.

With synthesis of compound **8a** accomplished through CM, saturation of the newly formed double bond (Pd/C, H₂, CH₃OH, r.t., 16 h) proceeded smoothly to provide ficuseptamine A in 62% isolated yield. We then envisioned performing one-pot CM/hydrogenation reactions for the synthesis of ficuseptamine A without isolation of the unsaturated CM product **8a**. In fact, a literature screen revealed that such a CM/hydrogenation maneuver had been reported by Cossy and co-workers in their total synthesis of (-)-centrolobine [20]. One-pot sequential reactions are increasingly being utilized in organic synthesis to accelerate the synthesis of target molecules [21]. Motivated by Cossy's work, we subjected aryl ketone **9a** and terminal olefin **10** to our optimized CM conditions, followed by hydrogenation in a one-pot fashion, to afford ficuspetamine A in 65% yield directly from the starting materials **9a** and **10** (Scheme 3). This one-pot CM/hydrogenation sequence proved to be fruitful, thus improving the efficiency of the synthesis of ficuseptamine A.



Scheme 3. Improving ficuseptamine A (1a) synthesis by a one-pot CM/hydrogentation sequence.

With ficuseptamine A (**1a**) synthesis accomplished, we moved forward to target ficuseptamine B (**1b**). We imagined employing a dual ethenolysis [22] and CM approach for its total synthesis. We envisaged that aryl ketone **9b**, which is a different regioisomer to aryl ketone **9a**, could be achieved by ethenolysis of the Claisen-Schmidt product **17**, using a suitable olefin metathesis catalyst to dissect the internal alkene of **17** to give the desired terminal olefin **9b** (Scheme 4). We also thought that this method would allow access to styrene derivatives (e.g., compounds such as **18**), which are important building blocks in organic chemistry.



Scheme 4. Synthetic plan for the synthesis of aryl ketone **9b** via a Claisen-Schmidt reaction followed by ethenolysis.

We began the synthesis towards ficuseptamine B by first preparing the Claisen-Schmidt product 17. We initially combined 3-hydroxy-4-methoxyacetophenone 15 and 3-hydroxy-4-methoxybenzaldehyde 16 under aqueous NaOH conditions in EtOH, which provided chalcone 17 in 46% yield (entry 1, Table 2). The use of cesium carbonate (Cs_2CO_3) as the base in EtOH led to 17 in 41% yield (entry 2, Table 2). However, efficient synthesis of 17 was achieved using SOCl₂/EtOH as a catalyst system (acid catalysis) [23] to afford 17 in 83% yield (entry 3, Table 2).

Table 2. Synthesis of Claisen-Schmidt product **17** under various catalysis conditions. ^a All reactions were carried out using ketone **15** (1 equiv.), aldehyde **16** (1 equiv.), and catalyst in EtOH at 23 °C. ^b NaOH (10% w/v).



With the Claisen-Schmidt product **17** in hand, we employed the conditions reported by Diver and co-workers with slight modification for its ethenolysis [24]. Exposing **17** to ethylene gas (60 psi pressure) with Hoveyda-Grubbs II **7** and using 1,2-DCE (1,2-dichloroethane) as a solvent at 40 °C delivered aryl ketone **9b** in 71% yield (Scheme 5).



Scheme 5. Ethenolysis of chalcone 17 to deliver aryl ketone 9b.

To the best of our knowledge, this is the first time ethenolysis has been applied to cleave a chalcone compound (i.e., structure such as **17**). Importantly, no isomerization [25] of the double bond is possible, given the structure of chalcone **17**, which lacks any methylene groups adjacent to the double bond. Having accomplished the synthesis of aryl ketone **9a** through ethenolysis, we employed the one-pot, two-step CM/hydrogenation protocol previously used in ficuseptamine A synthesis. Gratifyingly, subjecting aryl ketone **9b** and terminal olefin **10** to our optimized CM conditions (Hoveyda-Grubbs II **7** (10 mol %), CH₂Cl₂, 40 °C) followed by one-pot hydrogenation delivered ficuseptamine B in 69% yield (Scheme 6).



Scheme 6. Synthesis of ficuseptamine B (1b) by one-pot CM/hydrogenation sequence.

The spectroscopic data of the synthetic ficuseptamines A and B were identical to those reported by Shin-ya and co-workers [1]. The described chemistry for the synthesis of ficuseptamines A and B through CM allows the generation of various analogues of these natural products in a straightforward manner, which could have other biological activities, such as potential ligands for the 5-HT₇ receptor [26].

3. Conclusions

In summary, we have reported a highly efficient methodology for the first total synthesis of ficuseptamines A and B through a CM strategy. A sequential one-pot, two-step CM/hydrogenation procedure was employed to expediently accomplish their total synthesis.

4. Materials and Methods

4.1. General Chemistry Experimental

Chemical reactions were performed in over-dried glassware under nitrogen and anhydrous conditions, unless otherwise stated. Reactions were magnetically stirred using a Teflon-coated stir bar and monitored by pre-coated silica gel aluminum plates (0.25 mm thickness) with a fluorescent indicator (254 nm), using UV light as the visualizing agent. Alternatively, oxidative staining using an aqueous basic solution of KMnO₄ and heat was carried out for visualization. Silica gel (60 Å, 200–425 mesh) was used for flash column chromatography. Nuclear magnetic resonance (NMR) spectra were recorded on a Bruker 400 MHz spectrometer (Bruker, Billerica, MA, USA) in acetone- d_6 , CDCl₃, or DMSO- d_6 as the solvent. Chemical shifts are reported in parts per million (ppm) with reference to the hydrogenated residues of the deuterated solvent as the internal standard. Coupling constants (*J* values) are recorded in Hertz (Hz), and signal patterns are expressed as follows: singlet (s), doublet (d), dd (doublet of doublets), triplet (t), quintet (quint), and multiplet (m). Elemental analyses were performed on a 2400 Perkin Elmer Series II analyzer (PerkinElmer, Inc., Waltham, MA, USA).

High-resolution mass spectrometry was conducted using a Micromass Q-ToF mass spectrometer (Waters Corporation, Milford, MA, USA).

4.2. Experimental Procedures for Chemical Synthesis and Characterization Data of Compounds

4.2.1. Synthesis of 1-(4-Hydroxy-3-methoxyphenyl)prop-2-en-1-one (9a)

A round-bottom flask equipped with a magnetic stir bar was charged with 3-hydroxy-1-(4hydroxy-3-methoxyphenyl)propan-1-one 11 (2.52 g, 12.8 mmol) and conc. HBr (30 mL) was then added. The reaction mixture was then stirred at 95 $^\circ$ C for 2 h. The reaction mixture was cooled to room temperature and the resulting precipitate was filtered, washed with ice-cold H₂O, and dried to afford 3-bromo-1-(4-hydroxy-3-methoxyphenyl)propan-1-one 12 (quantitative yield, 3.09 g, 93%) as a white solid, which was judged to be of good purity by TLC analysis and mass spectrometry, and carried forward in crude form to the next step. To a stirred solution of 3-bromo-1-(4-hydroxy-3-methoxyphenyl)propan-1-one 12 (3 g, 11.6 mmol) in dry benzene (50 mL), was added, dropwise, a solution of DBU (2.08 mL, 13.9 mmol) in dry benzene (5 mL). A condenser was attached, and the reaction mixture was stirred at 70 °C for 3 h. The solvent was removed in vacuo, and the crude product was extracted with EtOAc (three times). The combined organic layers were washed with H₂O, brine, dried over MgSO₄, filtered, and concentrated in vacuo. The crude product was then purified by silica gel column chromatography to afford **9a** (1.42 g, 69%) as a white solid. ¹H-NMR $(400 \text{ MHz}, \text{acetone-}d_6): \delta 7.66 \text{ (dd}, J = 8.4, 2.0 \text{ Hz}, 1\text{H}), 7.61 \text{ (d}, J = 2.0 \text{ Hz}, 1\text{H}), 7.40 \text{ (dd}, J = 17.0, 10.4 \text{ Hz}, 10.4 \text{ Hz})$ 1H), 6.95 (d, *J* = 8.4 Hz, 1H), 6.36 (dd, *J* = 17.0, 2.0 Hz, 1H), 5.86 (dd, *J* = 10.4, 2.0 Hz, 1H), 3.93 (s, 3H); HRMS calcd. for C₁₀H₁₁O₃ [M + H]⁺ 179.0708, found 179.0713. Anal. calcd. for C₁₀H₁₀O₃: C, 67.41; H, 5.66. Found: C, 67.31; H, 5.55.

4.2.2. Synthesis of Ficuseptamine A by a One-Pot Cross Metathesis/Hydrogenation Procedure (1a)

To a stirred solution of aryl ketone 9a (315 mg, 1.77 mmol) and N,N-dimethyl-4- pentene-1-amine 10 (100 mg, 0.885 mmol) in dry and degassed CH₂Cl₂ (5 mL), was added Hoveyda-Grubbs second generation catalyst 7 (55 mg, 0.0885 mmol, 10 mol %). The reaction mixture was then deoxygentated by performing vacuum/N₂ cycles four times and stirred at 40 $^{\circ}$ C for 10 h. After the completion of the reaction, Pd/C (10 wt %) was added, and the N_2 atmosphere was substituted with H_2 by performing vacuum/H₂ cycles four times. The reaction mixture was then stirred under a double layer H₂ balloon at 23 °C for 16 h. The Pd/C catalyst was removed by filtration through a pad of Celite[®], followed by washing of the filter cake with CH₂Cl₂ (10 mL), and the filtrate was evaporated in vacuo. The crude product was purified directly by silica gel column chromatography to afford ficuseptamine A (1a) as a white solid (153 mg, 65%). Spectroscopic data for synthetic ficuseptamine A matched literature data of natural ficuseptamine A [1]. ¹H-NMR (400 MHz, acetone- d_6): δ 7.57 (dd, J = 8.4, 2.0 Hz, 1H), 7.54 (d, J = 2.0 Hz, 1H), 6.89 (d, J = 8.4 Hz, 1H), 3.89 (s, 3H), 2.93 (t, J = 7.2 Hz, 2H), 2.21 (t, J = 7.2 Hz, 2H), 2.13 (s, 2 × 3H), 1.68 (quint, J = 7.2 Hz, 2H), 1.47 (quint, J = 7.2 Hz, 2H), 1.37 (quint, J = 7.2 Hz, 2H); ¹³C-NMR (100 MHz, acetone-*d*₆) δ: 198.6, 152.1, 148.3, 130.6, 123.8, 115.3, 111.5, 60.2, 56.2, 45.6, 38.4, 28.3, 27.8, 25.3; HRMS calcd. for $C_{15}H_{24}NO_3$ [M + H]⁺ 266.1756, found 266.1767. Anal. calcd. for C₁₅H₂₃NO₃: C, 67.90; H, 8.74; N, 5.18. Found: C, 67.76; H, 8.70; N, 5.07.

4.2.3. Synthesis of (2E)-1,3-Bis(3-Hydroxy-4-methoxyphenyl)prop-2-en-1-one (17)

Compound 17 was synthesized using a previously reported method, except that it required purification [23]. To a stirred solution of 3-hydroxy-4-methoxyacetophenone 15 (1.20 g, 7.89 mmol) and 3-hydroxy-4-methoxybenzaldehyde 16 (1.31 g, 7.89 mmol) in absolute EtOH (5 mL), was added thionyl chloride (0.58 mL, 7.89 mmol) dropwise and the reaction mixture was stirred at 23 °C for 4 h. The reaction mixture was precipitated by the addition of H₂O (~5 mL), and the resulting precipitate was filtered, washed with ice-cold H₂O, and ice-cold EtOH. After the crude product was allowed to air dry, purification by recrystallization from EtOH afforded 17 (1.97 g, 83%) as a yellow solid. Analytical

data were in accordance with literature data [27]. ¹H-NMR (400 MHz, CDCl₃): δ 7.72 (d, *J* = 16.2 Hz, 1H), 7.63-7.60 (m, 2H), 7.40 (d, *J* = 16.2 Hz, 1H), 7.28-7.27 (m, 1H), 7.12 (dd, *J* = 10.4, 2.0 Hz, 1H), 6.93 (d, *J* = 8.2 Hz, 1H), 6.85 (d, *J* = 8.2 Hz, 1H), 3.98 (s, 3H), 3.94 (s, 3H); ¹³C NMR (100 MHz, DMSO-*d*₆): δ 188.5, 152.7, 150.7, 147.0, 146.9, 144.3, 131.3, 128.1, 122.8, 122.4, 119.8, 115.2, 114.8, 112.5, 111.8, 56.3, 56.2; HRMS calcd. for C₁₇H₁₇O₅ [M + H]⁺ 301.1076, found 301.1065. Anal. calcd. for C₁₇H₁₆O₅: C, 67.99; H, 5.37. Found: C, 67.80; H, 5.41.

4.2.4. Synthesis of 1-(3-Hydroxy-4-methoxyphenyl)prop-2-en-1-one (9b)

Compound **9b** was synthesized using the slightly modified reaction conditions of Diver et al. [24]. To an inert and oven-dried high-pressure flask, was added compound **17** (800 mg, 2.67 mmol), dry and degassed 1,2-dichloroethane (35 mL), and Hoveyda-Grubbs second generation catalyst **7** (42 mg, 0.067 mmol). The flask was purged with ethylene for 5 min, pressurized to 60 psi, and stirred at 40 °C for 12 h, after which time TLC analysis indicated consumption of the internal alkene **17**. The pressure was then slowly vented, and the solvent was evaporated in vacuo. The crude product was then purified directly by silica gel column chromatography, to afford **9b** (337 mg, 71%) as a white solid. ¹H-NMR (400 MHz, acetone-*d*₆): δ 7.59 (dd, *J* = 8.4, 2.0 Hz, 1H), 7.50 (d, *J* = 2.0 Hz, 1H), 7.33 (dd, *J* = 17.0, 10.4 Hz, 1H), 7.07 (d, *J* = 8.4 Hz, 1H), 6.33 (dd, *J* = 17.0, 2.0 Hz, 1H), 5.84 (dd, *J* = 10.4, 2.0 Hz, 1H), 3.94 (s, 3H); ¹³C NMR (100 MHz, acetone-*d*₆): δ 188.9, 153.0, 147.7, 133.2, 131.8 128.9, 122.8, 115.7, 111.8, 56.5; HRMS calcd. for C₁₀H₁₁O₃ [M + H]⁺ 179.0708, found 179.0719. Anal. calcd. for C₁₀H₁₀O₃: C, 67.41; H, 5.66. Found: C, 67.35; H, 5.59.

4.2.5. Synthesis of Ficuseptamine B by a One-Pot Cross Metathesis/Hydrogenation Procedure (1b)

The experimental procedure for the synthesis of ficuseptamine A (**1a**) was followed, except that aryl ketone **9b** was used as the cross metathesis partner with *N*,*N*-dimethyl-4-pentene-1-amine **10** to afford ficuspetamine B (**1b**) (162 mg, 69%) as a white solid. Spectroscopic data for synthetic ficuseptamine B matched literature data of natural ficuseptamine B [1]. ¹H-NMR (400 MHz, acetone-*d*₆): δ 7.52 (dd, *J* = 8.5, 2.0 Hz, 1H), 7.46 (d, *J* = 2.0 Hz, 1H), 7.01 (d, *J* = 8.5 Hz, 1H), 3.90 (s, 3H), 2.90 (t, *J* = 7.2 Hz, 2H), 2.24 (t, *J* = 7.2 Hz, 2H), 2.15 (s, 2 × 3H), 1.66 (quint, *J* = 7.2 Hz, 2H), 1.48 (quint, *J* = 7.2 Hz, 2H), 1.37 (quint, *J* = 7.2 Hz, 2H); ¹³C-NMR (100 MHz, acetone-*d*₆): δ 198.9, 152.4, 147.3, 131.7, 121.7, 115.2, 111.5, 60.1, 56.3, 45.5, 38.5, 28.1, 27.7, 25.2; HRMS calcd. for C₁₅H₂₄NO₃ [M + H]⁺ 266.1756, found 266.1741. Anal. calcd. for C₁₅H₂₃NO₃: C, 67.90; H, 8.74; N, 5.28. Found: C, 67.78; H, 8.66; N, 5.19.

Author Contributions: Conceived of and designed the experiments: H.M.A.H. Performed the experiments: H.M.A.H. Analyzed the data: H.M.A.H. Wrote the paper: H.M.A.H.

Acknowledgments: This work was funded by the Deanship of Scientific Research (DSR), King Abdulaziz University, under grant number G-1436-141-202. The author, therefore, would like to thank DSR for financial support.

Conflicts of Interest: The author declares no conflict of interest.

References

- 1. Ueda, J.-Y.; Takagi, M.; Shin-ya, K. Aminocaprophenone- and pyrrolidine-type alkaloids from the leaves of *Ficus septica*. *J. Nat. Prod.* **2009**, *72*, 2181–2183. [CrossRef] [PubMed]
- Battaglia, U.; Moody, C.J. A short synthesis of the triazolopyrimidine antibiotic essramycin. *J. Nat. Prod.* 2010, 73, 1938–1939. [CrossRef] [PubMed]
- 3. Wang, W.; Zeng, Y.H.; Osman, K.; Shinde, K.; Rahman, M.; Gibbons, S.; Mu, Q. Norlignans, acylphloroglucinols, and a dimeric xanthone from *Hypericum chinense*. J. Nat. Prod. **2010**, 73, 1815–1820. [CrossRef] [PubMed]
- Pais, G.C.G.; Zhang, X.; Marchand, C.; Neamati, N.; Cowansage, K.; Svarovskaia, E.S.; Pathak, V.K.; Tang, Y.; Nicklaus, M.; Pommier, Y.; et al. Structure activity of 3-aryl-1,3-diketo-containing compounds as HIV-1 integrase inhibitors. *J. Med. Chem.* 2002, 45, 3184–3194. [CrossRef] [PubMed]
- 5. Purser, S.; Moore, P.R.; Swallow, S.; Gouverneur, V. Fluorine in medicinal chemistry. *Chem. Soc. Rev.* 2008, *37*, 320–330. [CrossRef] [PubMed]

- Han, C.; Salyer, A.E.; Kim, E.H.; Jiang, X.; Jarrad, R.E.; Powers, M.S.; Kirchhoff, A.M.; Salvador, T.K.; Chester, J.A.; Hockerman, G.H.; et al. Evaluation of difluoromethyl ketones as agonists of the γ-aminobutyric acid type B (GABA_B) receptor. *J. Med. Chem.* 2013, *56*, 2456–2465. [CrossRef] [PubMed]
- 7. Vougioukalakis, G.C.; Grubbs, R.H. Ruthenium-based heterocyclic carbene-coordinated olefin metathesis catalysts. *Chem. Rev.* **2010**, *110*, 1746–1787. [CrossRef] [PubMed]
- 8. Hassan, H.M.A. Recent applications of ring-closing metathesis in the synthesis of lactams and macrolactams. *Chem. Commun.* **2010**, *46*, 9100–9106. [CrossRef] [PubMed]
- Connon, S.J.; Blechert, S. Recent developments in olefin cross-metathesis. *Angew. Chem. Int. Ed.* 2003, 42, 1900–1923. [CrossRef] [PubMed]
- 10. Hoveyda, A.H.; Zhugralin, A.R. The remarkable metal-catalysed olefin metathesis reaction. *Nature* **2007**, *450*, 243–251. [CrossRef] [PubMed]
- 11. Nicolaou, K.C.; Bulger, P.G.; Sarlah, D. Metathesis reactions in total synthesis. *Angew. Chem. Int. Ed.* **2005**, *44*, 4490–4527. [CrossRef] [PubMed]
- 12. Fürstner, A. Metathesis in total synthesis. Chem. Commun. 2011, 47, 6505–6511. [CrossRef] [PubMed]
- Chatare, V.K.; Andrade, R.B. Total synthesis of (-)-albocycline. *Angew. Chem. Int. Ed.* 2017, 56, 5909–5911. [CrossRef] [PubMed]
- 14. Herbert, M.B.; Marx, V.M.; Pederson, R.L.; Grubbs, R.H. Concise syntheses of insect pheromones using Z-selective cross metathesis. *Angew. Chem. Int. Ed.* **2013**, *52*, 310–314. [CrossRef] [PubMed]
- 15. Mann, T.J.; Speed, A.W.H.; Schrock, R.R.; Amir, H.; Hoveyda, A.H. Catalytic Z-selective cross-metathesis with secondary silyl- and benzyl-protected allylic ethers: Mechanistic aspects and applications to natural product synthesis. *Angew. Chem. Int. Ed.* **2013**, *52*, 8395–8400. [CrossRef] [PubMed]
- 16. Herbert, M.B.; Grubbs, R.H. Z-selective cross metathesis with ruthenium catalysts: Synthetic applications and mechanistic implications. *Angew. Chem. Int. Ed.* **2015**, *54*, 5018–5024. [CrossRef] [PubMed]
- 17. Hassan, H.M.A.; Brown, F.K. A convenient approach to acyclic unsaturated amino acids via ring-closing metathesis. *Chem. Commun.* **2010**, *46*, 3013–3015. [CrossRef] [PubMed]
- Hong, S.H.; Wenzel, A.G.; Salguero, T.T.; Day, M.W.; Grubbs, R.H. Decomposition of ruthenium olefin metathesis catalysts. J. Am. Chem. Soc. 2007, 129, 7961–7968. [CrossRef] [PubMed]
- 19. Chatterjee, A.K.; Choi, T.-L.; Sanders, D.P.; Grubbs, R.H. A general model for selectivity in olefin cross metathesis. *J. Am. Chem. Soc.* **2003**, *125*, 11360–11370. [CrossRef] [PubMed]
- 20. Boulard, L.; BouzBouz, S.; Cossy, J.; Franck, X.; Figadère, B. Two successive one-pot reactions leading to the expeditious synthesis of (–)-centrolobine. *Tetrahedron Lett.* **2004**, *45*, 6603–6605. [CrossRef]
- 21. Hayashi, Y. Pot economy and one-pot synthesis. Chem. Sci. 2016, 7, 866–880. [CrossRef] [PubMed]
- 22. Bidange, J.; Fischmeister, C.; Bruneau, C. Ethenolysis: A green catalytic tool to cleave carbon—Carbon double bonds. *Chem. Eur. J.* **2016**, *22*, 12226–12244. [CrossRef] [PubMed]
- Petrov, O.; Ivanova, Y.; Gerova, M. SOCl₂/EtOH: Catalytic system for synthesis of chalcones. *Catal. Commun.* 2008, 9, 315–316. [CrossRef]
- 24. Clark, J.R.; French, J.M.; Diver, S.T. Alkene metathesis approach to β-unsubstituted anti-allylic alcohols and their use in ene-yne metathesis. *J. Org. Chem.* **2012**, *77*, 1599–1604. [CrossRef] [PubMed]
- 25. Hong, S.H.; Sanders, D.P.; Lee, C.W.; Grubbs, R.H. Prevention of undesirable isomerization during olefin metathesis. *J. Am. Chem. Soc.* **2005**, *127*, 17160–17161. [CrossRef] [PubMed]
- Perrone, R.; Berardi, F.; Colabufo, N.A.; Lacivita, E.; Leopoldo, M.; Tortorella, V. Synthesis and structure-affinity relationships of 1-[ω-(4-aryl-1-piperazinyl) alkyl]-1-aryl ketones as 5-HT₇ receptor ligands. *J. Med. Chem.* 2003, 46, 646–649. [CrossRef] [PubMed]
- Leow, P.-C.; Bahety, P.; Boon, C.P.; Lee, C.Y.; Tan, K.L.; Yang, T.; Ee, P.-L.R. Functionalized curcumin analogs as potent modulators of the Wnt/β-catenin signaling pathway. *Eur. J. Med. Chem.* 2014, *71*, 67–80. [CrossRef] [PubMed]

Sample Availability: Samples of the compounds are not available from the author.



© 2018 by the author. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/).