

## Microwave-assisted three-component domino reaction: Synthesis of indolodiazepinotriazoles

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### Full Research Paper

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### Abstract

A microwave-assisted three-component protocol involving *N*-1 alkylation of 2-alkynylindoles with epichlorohydrin, ring opening of the epoxide with sodium azide, and an intramolecular Huisgen azide–internal alkyne 1,3-dipolar cycloaddition domino sequence has been described. The efficacy of the methodology has been demonstrated by treating various 2-alkynylindoles (aromatic/aliphatic) with epichlorohydrin and sodium azide furnishing annulated tetracyclic indolodiazepinotriazoles in satisfactory yields.

### Introduction

The intermolecular Huisgen azide–alkyne 1,3-dipolar cycloaddition reaction [1-6] for the synthesis of 1,2,3-triazoles in both aqueous [7-10] and organic solvents under either metal-catalyzed [11-13] or metal-free conditions [14-16] has received increasing attention in drug discovery processes [17,18]. The ease of reaction in the intermolecular format has been successfully demonstrated by using both organic/inorganic azides as well as alkynes/diynes [19-21]. In contrast to its employment in an intermolecular format, intramolecular azide–alkyne 1,3-dipolar cycloaddition reactions have been also applied by us and others with the view to synthesize triazole-annulated poly-

heterocycles. Although these cyclizations have been successfully carried out in either one-pot [22-24] or multistep format [25-28], reports involving their application in a three-component domino format are scarce [29,30]. In our laboratory, we had been employing functionalized indoles for the synthesis of annulated indole-based polyheterocycles either in a multicomponent or in a one-pot format [31-35]. In this continuation, we next directed our efforts to the development of a three-component domino strategy for the synthesis of indole-based polyheterocycles by incorporating the intramolecular azide–alkyne 1,3-dipolar cycloaddition reaction as one of the domino steps. Here

we propose a strategy where *N*-1 of 2-alkynylindole [36,37] can be first functionalized with epoxide by reacting 2-alkynylindole with epichlorohydrin. This can then be followed by ring opening of the oxirane by azide to furnish a bis-functionalized indole intermediate having azide and alkyne groups in close proximity. Such an intermediate may then undergo annulation following an intramolecular 1,3-dipolar cycloaddition pathway and in turn lead to the sequential formation of 7- or 5-membered diazepine and triazole rings in a single step. In this communication, we report a versatile microwave-assisted three component domino reaction to furnish annulated tetracyclic indolodiazepinotriazoles in good yields.

## Results and Discussion

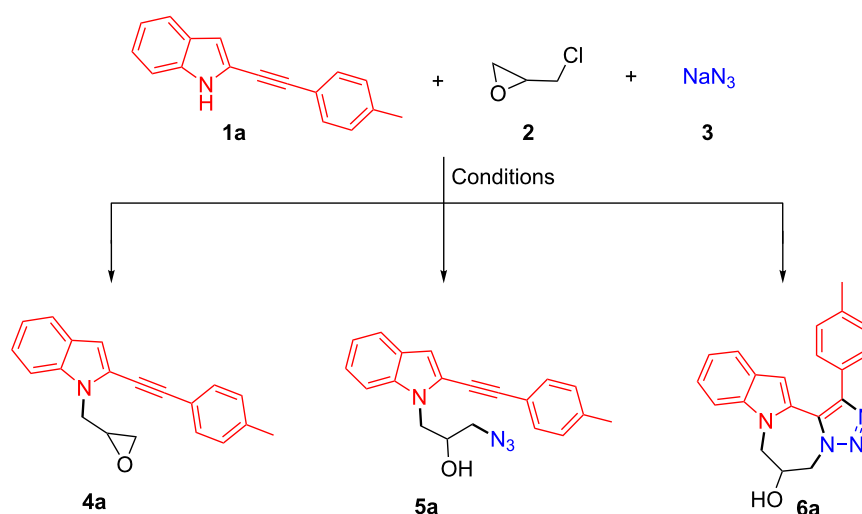
We commenced our studies with the development of a one-pot three-component strategy involving the condensation of the 2-(4-methylphenylethynyl)-1*H*-indole (**1a**) with epichlorohydrin (**2**) and sodium azide (**3**, Scheme 1, Table 1). Initially, a mixture of **1a**, **2** and **3** was allowed to react both in the absence and presence of Cs<sub>2</sub>CO<sub>3</sub> in toluene at rt. The reactants under both the conditions remained unchanged even after prolonged stirring for 15 h (Table 1, entries 1–3) and at higher temperature (110 °C).

However, a change in the nature of solvent from toluene to CH<sub>3</sub>CN, DMF or DMSO produced a dramatic effect on the outcome of the reaction, resulting in the formation of products comprising intermediates (**4a** and/or **5a**) and/or indole-based polyheterocycle indolodiazepinotriazole **6a**. Use of the polar solvent CH<sub>3</sub>CN at 90 °C for 15 h furnished a single product in 65% isolated yield, which was characterized as 2-[2-(4-methyl-

**Table 1:** Optimization of reaction conditions for the synthesis of **6a** in a three-component domino format.

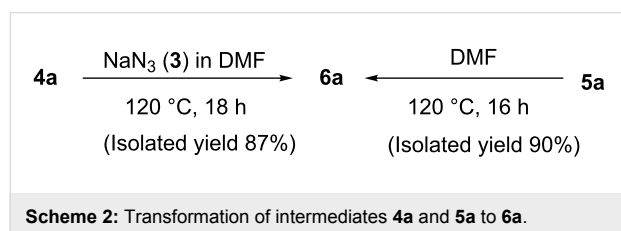
Entry	Base	Solvent	Temp (°C)	Time	Yield (%) <sup>a</sup> of <b>4a/5a/6a</b>
1	–	toluene	rt	15 h	NR
2	Cs <sub>2</sub> CO <sub>3</sub>	toluene	rt	15 h	NR
3	Cs <sub>2</sub> CO <sub>3</sub>	toluene	110	15 h	NR
4	Cs <sub>2</sub> CO <sub>3</sub>	CH <sub>3</sub> CN	90	15 h	65/–/–
5	Cs <sub>2</sub> CO <sub>3</sub>	DMF	rt	15 h	NR
6	Cs <sub>2</sub> CO <sub>3</sub>	DMF	120	1 h	77/–/–
7	Cs <sub>2</sub> CO <sub>3</sub>	DMF	120	4 h	40/30/–
8	Cs <sub>2</sub> CO <sub>3</sub>	DMF	120	15 h	–/15/50
9	Cs <sub>2</sub> CO <sub>3</sub>	DMF	120	18 h	–/–/60
10	Cs <sub>2</sub> CO <sub>3</sub>	DMSO	120	1 h	82/–/–
11	Cs <sub>2</sub> CO <sub>3</sub>	DMSO	120	4 h	42/40/–
12	Cs <sub>2</sub> CO <sub>3</sub>	DMSO	120	10 h	–/20/52
13	Cs <sub>2</sub> CO <sub>3</sub>	DMSO	120	15 h	–/–/64
14	Cs <sub>2</sub> CO <sub>3</sub>	DMSO	120 MW	10 min	80/–/–
15	Cs <sub>2</sub> CO <sub>3</sub>	DMSO	120 MW	30 min	20/45/10 <sup>b</sup>
16	Cs <sub>2</sub> CO <sub>3</sub>	DMSO	120 MW	1 h	–/18/42
17	<b>Cs<sub>2</sub>CO<sub>3</sub></b>	<b>DMSO</b>	<b>120 MW</b>	<b>1.5 h</b>	<b>–/–/71</b>
18	Cs <sub>2</sub> CO <sub>3</sub>	DMF	120 MW	1.5 h	–/–/64
19	Cs <sub>2</sub> CO <sub>3</sub>	CH <sub>3</sub> CN	90 MW	1.5 h	80/–/–
20	Cs <sub>2</sub> CO <sub>3</sub>	CH <sub>3</sub> OH	90 MW	1.5 h	NR
21	K <sub>2</sub> CO <sub>3</sub>	DMSO	120 MW	1.5 h	–/10/54 <sup>b</sup>
22	Na <sub>2</sub> CO <sub>3</sub>	DMSO	120 MW	1.5 h	–/12/52 <sup>b</sup>
23	K <sub>3</sub> PO <sub>4</sub>	DMSO	120 MW	1.5 h	–/–/62
24	<i>t</i> -BuOK	DMSO	120 MW	1.5 h	–/–/65
25	DBU	DMSO	120 MW	1.5 h	–/15/48 <sup>b</sup>
26	TEA	DMSO	120 MW	1.5 h	–/20/45 <sup>b</sup>

<sup>a</sup>Isolated yields. <sup>b</sup>Yields based on HPLC (C18 reversed-phase column, 150 × 4.8 mm, 5 μm). NR = no reaction.



**Scheme 1:** One-pot three-component domino reaction furnishing indole derivatives (**4a** and **5a**) and indolodiazepinotriazole **6a**.

phenyl)ethynyl]-1-(oxiran-2-ylmethyl)-1*H*-indole (**4a**, Table 1, entry 4). In contrast, use of the polar aprotic solvent DMF with high dielectric constant produced both intermediates **4a/5a** as well as the annulated product **6a**. Interestingly, a significant increase in the yield of the title compound **6a** was observed by prolonging the reaction. Carrying out the reaction in DMF at rt also failed to promote annulation even after 15 h of prolonged stirring (Table 1, entry 5). Increasing the temperature to 120 °C furnished the intermediate **4a** as a single product in 77% isolated yield within 1 h (Table 1, entry 6). Further stirring up to 4 h at 120 °C led to the partial conversion of **4a** (by ring opening of the epoxide with NaN<sub>3</sub>) into yet another intermediate 1-azido-3-{2-[2-(4-methylphenyl)ethynyl]-1*H*-indol-1-yl}propan-2-ol (**5a**, Table 1, entry 7) in 30% isolated yield. Nonetheless, extending the reaction times up to 15 h, led to the complete disappearance of **4a** and furnished a mixture of the intermediate **5a** in 15% isolated yield and the title compound **6a** characterized as 1-(4-methylphenyl)-6,7-dihydro-5*H*-[1,2,3]triazolo[5',1':3,4][1,4]diazepino[1,2-*a*]indol-6-ol in 50% isolated yield (Table 1, entry 8). The findings clearly suggest that the formation of indole-based annulated product **6a** in the three-component domino format occurs via **4a** and **5a** intermediacy and requires higher temperature and prolonged stirring. This was again evident from the fact that a prolonged stirring up to 18 h led to the complete disappearance of the intermediates **4a** and **5a** and afforded **6a** as a single product in 60% isolated yield (Table 1, entry 9). The role of intermediates **4a** and **5a** in the formation of **6a** was further substantiated by treating **4a** with NaN<sub>3</sub> in DMF at 120 °C and by heating **5a** in DMF at 120 °C. As envisaged, both reactions furnished **6a** as a single product in 87% and 90% isolated yield, respectively (Scheme 2). Replacing DMF with yet another polar aprotic solvent, i.e., DMSO, produced similar results except for a marginal increase in the isolated yield of **6a** to 64% in 15 h (Table 1, entries 10–13).



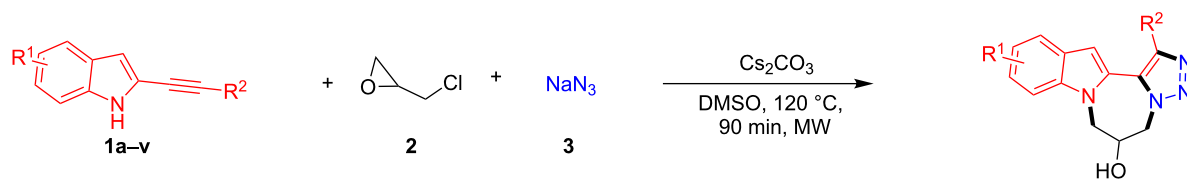
Next, in order to reduce the reaction times and to enhance the isolated yield of the annulated product **6a**, we applied microwave conditions instead of conventional heating and monitored the progress of the reaction at different time intervals. A significant increase in the yield of **6a** resulting from the increase in the reaction times under microwave conditions was observed. Initially, a 10 min irradiation of the reaction mixture

furnished the intermediate **4a** as the only product in 80% isolated yield (Table 1, entry 14), whereas a 30 min irradiation resulted in a mixture of **4a/5a/6a** in 20/45/10% yields as evident from HPLC (Table 1, entry 15). Extending exposure to microwave conditions for 1 h produced a mixture of **5a** and **6a** (Table 1, entry 16); however, a further exposure up to 1.5 h furnished the desired compound **6a** as the only product in 71% isolated yield (Table 1, entry 17). Thus, under microwave irradiation conditions, not only the isolated yield of **6a** increased from 60% under conventional heating to 71%, but the duration of reaction was also reduced from 15 h to 1.5 h. Switching the solvent from DMSO to DMF under microwave conditions furnished **6a** in slightly reduced yield (Table 1, entry 18) while the use of CH<sub>3</sub>CN and CH<sub>3</sub>OH failed to produce the desired product (Table 1, entry 19 and 20). Replacing Cs<sub>2</sub>CO<sub>3</sub> with other bases such as K<sub>2</sub>CO<sub>3</sub>, Na<sub>2</sub>CO<sub>3</sub>, K<sub>3</sub>PO<sub>4</sub>, *t*-BuOK, DBU and TEA either produced a mixture of **5a/6a** or furnished **6a** in reduced yields (Table 1, entries 21–26). The observations clearly suggest that the formation of **6a** in the three-component format involved intermolecular *N*-1 alkylation of the 2-alkynylindole **1a** with epichlorohydrin to form **4a**, ring opening of **4a** with sodium azide to form **5a**, and finally an intramolecular Huisgen azide–internal alkyne 1,3-dipolar cycloaddition reaction.

Once the reaction conditions for the three-component format had been optimized, several 2-alkynylindoles bearing different functional groups were treated with epichlorohydrin and sodium azide in order to establish the scope and limitation of the strategy. In total 22 compounds **6a–v** (Scheme 3) were synthesized, with their isolated yields varying from 54–73%. The findings suggest that although the electronic properties of the substitution (R<sup>1</sup>) on the phenyl ring of the indole had no effect on the outcome of the isolated yield of the final products, the nature of R<sup>2</sup> had a profound effect on the yields. When the aromatic group was used as R<sup>2</sup>, the final products **6a–c** and **6e–q** were obtained in isolated yields ranging from 66–73%, whereas substituting R<sup>2</sup> with aliphatic/trimethylsilyl moieties furnished the cyclized products (**6d** and **6r–v**) in diminished (54–65%) isolated yields.

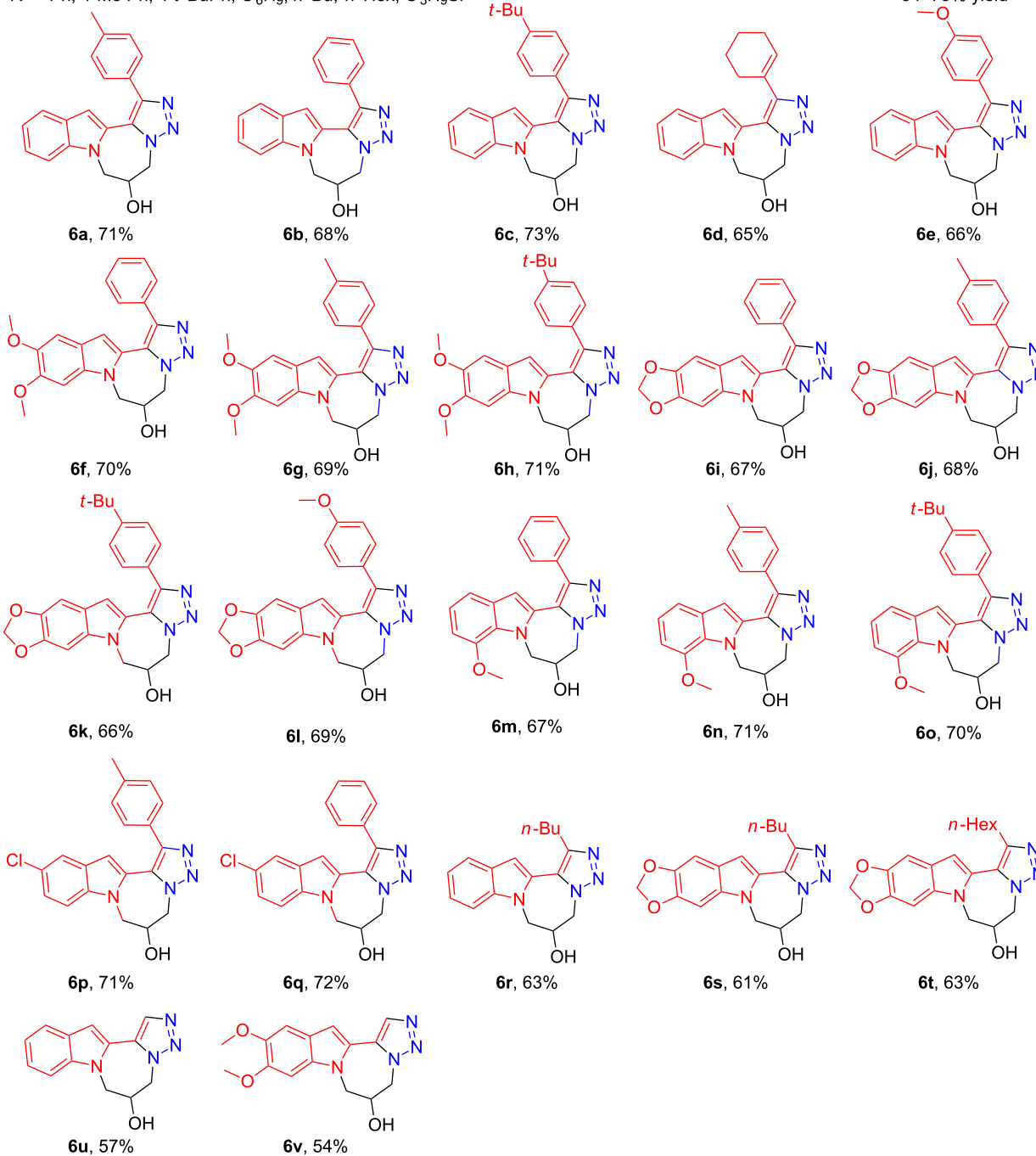
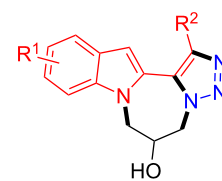
## Conclusion

In conclusion, we have developed a simple and efficient three-component domino reaction for the synthesis of highly substituted indolodiazepinotriazoles in good yields under microwave conditions. The domino sequence comprising *N*-1 alkylation, ring opening of the epoxide, and intramolecular Huisgen azide–internal alkyne 1,3-dipolar cycloaddition reaction, led to the generation of the diazepine and triazole rings annulated to the indole through the formation of four new sigma bonds in a single step.



$\text{R}^1 = \text{H, 7-O-Me, 5,6-OCH}_2\text{O, 5,6-diOMe, 5-Cl}$

$\text{R}^2 = \text{Ph, 4-Me Ph, 4-}t\text{-BuPh, C}_6\text{H}_9, n\text{-Bu, } n\text{-Hex, C}_3\text{H}_9\text{Si}$



**Scheme 3:** Three-component domino reaction for the synthesis of tetracyclic indolodiazepinotriazole compounds based on **6a**. Reaction conditions: **1a** (1.0 mmol), **2** (1.1 mmol), **3** (1.5 mmol) and  $\text{Cs}_2\text{CO}_3$  (1.5 mmol) in DMSO (2.5 mL) at 120 °C, MW under  $\text{N}_2$  atmosphere.

## Supporting Information

### Supporting Information File 1

Experimental section, copies of  $^1\text{H}$ ,  $^{13}\text{C}$  NMR and HRMS spectra of starting and final compounds **1e**, **1h**, **1j–1l**, **1n–1t**, **1v**, **4a**, **5a** and **6a–6v**.

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-9-41-S1.pdf>]

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## References

- Rostovtsev, V. V.; Green, L. G.; Fokin, V. V.; Sharpless, K. B. *Angew. Chem., Int. Ed.* **2002**, *41*, 2596. doi:10.1002/1521-3773(20020715)41:14<2596::AID-ANIE2596>3.0.CO;2-4
- Prescher, J. A.; Bertozzi, C. R. *Nat. Chem. Biol.* **2005**, *1*, 13. doi:10.1038/nchembio0605-13
- Miaoa, T.; Wang, L. *Synthesis* **2008**, 363. doi:10.1055/s-2008-1032037
- Wangler, C.; Schirrmacher, R.; Bartenstein, P.; Wangler, B. *Curr. Med. Chem.* **2010**, *17*, 1092. doi:10.2174/092986710790820615
- Walsh, J. C.; Kolb, H. C. *Chimia* **2010**, *64*, 29. doi:10.2533/chimia.2010.29
- Lal, S.; Díez-González, S. *J. Org. Chem.* **2011**, *76*, 2367. doi:10.1021/jo200085j
- Saha, B.; Sharma, S.; Sawant, D.; Kundu, B. *Synlett* **2007**, 1591. doi:10.1055/s-2007-982543
- Li, P.; Wang, L. *Letf. Org. Chem.* **2007**, *4*, 23. doi:10.2174/157017807780037513
- Kumar, I.; Rode, C. V. *Chem. Lett.* **2007**, *36*, 592. doi:10.1246/cl.2007.592
- Liu, M.; Reiser, O. *Org. Lett.* **2011**, *13*, 1102. doi:10.1021/ol103134c
- Zhang, L.; Chen, X.; Xue, P.; Sun, H. H. Y.; Williams, I. D.; Sharpless, K. B.; Fokin, V. V.; Jia, G. *J. Am. Chem. Soc.* **2005**, *127*, 15998. doi:10.1021/ja054114s
- Yoo, E. J.; Ahlquist, M.; Bae, I.; Sharpless, K. B.; Fokin, V. V.; Chang, S. *J. Org. Chem.* **2008**, *73*, 5520. doi:10.1021/jo800733p
- Meldal, M.; Tornøe, C. W. *Chem. Rev.* **2008**, *108*, 2952. doi:10.1021/cr0783479
- Sau, M.; Rodríguez-Esrich, C.; Pericàs, M. A. *Org. Lett.* **2011**, *13*, 5044. doi:10.1021/ol201869y
- Kwok, S. W.; Fotsing, J. R.; Fraser, R. J.; Rodionov, O. V.; Fokin, V. V. *Org. Lett.* **2010**, *12*, 4217. doi:10.1021/ol101568d
- Becer, C. R.; Hoogenboom, R.; Schubert, U. S. *Angew. Chem., Int. Ed.* **2009**, *48*, 4900. doi:10.1002/anie.200900755
- Agalave, S. G.; Maujan, S. R.; Pore, V. S. *Chem.–Asian J.* **2011**, *6*, 2696. doi:10.1002/asia.201100432
- Qin, A.; Lam, J. W. Y.; Tang, B. Z. *Chem. Soc. Rev.* **2010**, *39*, 2522. doi:10.1039/b909064a
- Mandadapu, A. K.; Sharma, S. K.; Gupta, S.; Krishna, D. G. V.; Kundu, B. *Org. Lett.* **2011**, *13*, 3162. doi:10.1021/ol201092k
- Aizpurua, J. M.; Azcune, I.; Fratila, R. M.; Balentova, E.; Sagartzazu-Aizpurua, M.; Miranda, J. I. *Org. Lett.* **2010**, *12*, 1584. doi:10.1021/ol1003127
- Fiandanese, V.; Bottalico, D.; Marchese, G.; Punzi, A.; Quarta, M. R.; Fittipaldi, M. *Synthesis* **2009**, 3853. doi:10.1055/s-0029-1217017
- Arigela, R. K.; Mandadapu, A. K.; Sharma, S. K.; Kumar, B.; Kundu, B. *Org. Lett.* **2012**, *14*, 1804. doi:10.1021/ol300399y
- Guggenheim, K. G.; Toru, H.; Kurth, M. *J. Org. Lett.* **2012**, *14*, 3732. doi:10.1021/ol301592z
- Kunick, C. *Liebigs Ann. Chem.* **1993**, 1141. doi:10.1002/jlac.1993199301182
- Akripoulou-Zanze, I.; Gracias, V.; Djuric, S. W. *Tetrahedron Lett.* **2004**, *45*, 8439. doi:10.1016/j.tetlet.2004.09.117
- Oliva, A. I.; Christmann, U.; Font, D.; Cuevas, F.; Ballester, P.; Buschmann, H.; Torrens, A.; Yenes, S.; Pericàs, M. A. *Org. Lett.* **2008**, *10*, 1617. doi:10.1021/ol800291t
27. Fajumdar, K. C.; Ray, K.; Ganai, S. *Synthesis* **2010**, 2101. doi:10.1055/s-0029-1218763
- Donald, J. R.; Martin, S. F. *Org. Lett.* **2011**, *13*, 852. doi:10.1021/ol1028404
- Conrad, W. E.; Rodriguez, K. X.; Nguyen, H. H.; Fettinger, J. C.; Haddadin, M. J.; Kurth, M. *J. Org. Lett.* **2012**, *14*, 3870. doi:10.1021/ol3015804
- Gracias, V.; Darczak, D.; Gasielki, A. F.; Djuric, S. W. *Tetrahedron Lett.* **2005**, *46*, 9053. doi:10.1016/j.tetlet.2005.10.090
- Sharma, S. K.; Mandadapu, A. K.; Saifuddin, M.; Gupta, S.; Agarwal, P. K.; Mandwal, A. K.; Gauniyal, H. M.; Kundu, B. *Tetrahedron Lett.* **2010**, *51*, 6022. doi:10.1016/j.tetlet.2010.09.054
- Sharma, S. K.; Gupta, S.; Saifuddin, M.; Mandadapu, A. K.; Agarwal, P. K.; Gauniyal, H. M.; Kundu, B. *Tetrahedron Lett.* **2011**, *52*, 65. doi:10.1016/j.tetlet.2010.10.147
- Gupta, S.; Kumar, B.; Kundu, B. *J. Org. Chem.* **2011**, *76*, 10154. doi:10.1021/jo201994v
- Gupta, S.; Sharma, S. K.; Mandadapu, A. K.; Gauniyal, H. M.; Kundu, B. *Tetrahedron Lett.* **2011**, *52*, 4288. doi:10.1016/j.tetlet.2011.06.021
- Sharma, S. K.; Mandadapu, A. K.; Kumar, B.; Kundu, B. *J. Org. Chem.* **2011**, *76*, 6798. doi:10.1021/jo201228t
- Nagamochi, M.; Fang, Y.-Q.; Lautens, M. *Org. Lett.* **2007**, *9*, 2955. doi:10.1021/ol071370w
- Fiandanese, V.; Bottalico, D.; Marchese, G.; Punzi, A. *Tetrahedron* **2008**, *64*, 7301. doi:10.1016/j.tet.2008.05.059

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