

Article

## Nonlinear Optical Properties Tuning in *Meso*-Tetraphenylporphyrin Derivatives Substituted with Donor/Acceptor Groups in Picosecond and Nanosecond Regimes

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**Abstract:** *meso*-Tetraphenylporphyrin (TPP) and its two substituted derivatives (*meso*-tetrakis(4-cyanophenyl)porphyrin [TPP(CN)<sub>4</sub>] and *meso*-tetrakis(4-methoxyphenyl)porphyrin [TPP(OMe)<sub>4</sub>]) were synthesized. Their nonlinear absorption and refraction properties were studied using the Z-scan technique in the picosecond (ps) and nanosecond (ns) regimes. The open aperture Z-scan results reveal that TPP and TPP(CN)<sub>4</sub> display an identical reverse saturable absorption (RSA) character in the ps and ns regimes. While TPP(OMe)<sub>4</sub> exhibits a transition from saturable absorption (SA) to RSA in the ps regime and a typical RSA character in the ns regime. The closed aperture Z-scan results show that TPP(CN)<sub>4</sub> and TPP(OMe)<sub>4</sub> have regular enhancement of the magnitude of nonlinear refraction as compared to their parent TPP in both the ps and ns regimes. In addition, the second-order molecular hyperpolarizabilities ( $\gamma$ ) of these three porphyrins are calculated, and the  $\gamma$  values of TPP(CN)<sub>4</sub> and TPP(OMe)<sub>4</sub> are remarkable larger than that of TPP. The introduction of the electron-withdrawing group CN and the electron-donating group OMe into TPP has enhanced its nonlinear refraction and  $\gamma$  value, and tuned its nonlinear absorption (TPP(OMe)<sub>4</sub>), which could be useful for porphyrin-related applications based on the desired NLO properties.

**Keywords:** porphyrins; nonlinear refraction; nonlinear absorption; Z-scan; second hyperpolarizabilities

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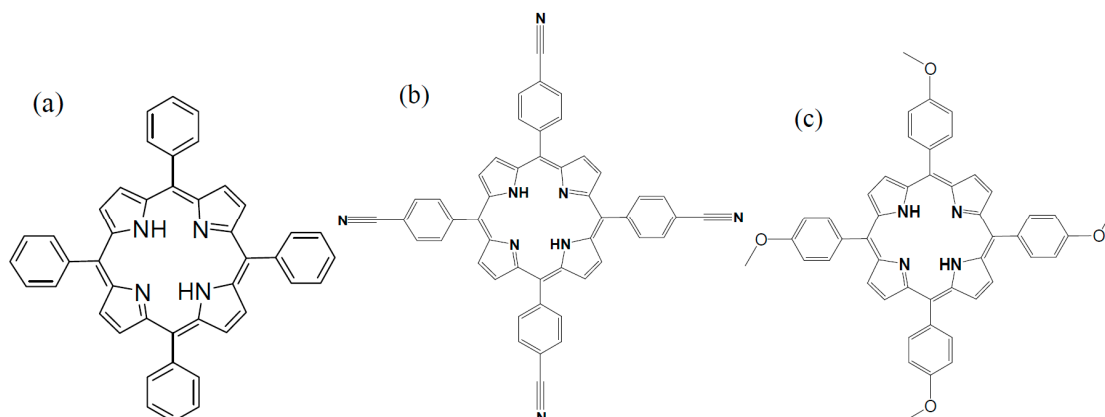
## 1. Introduction

Organic materials such as polymethine dyes [1], chromophores [2], phthalocyanines [3], porphyrins [4], and their related compounds [5] are target materials to study nonlinear optical (NLO) properties. The conjugated  $\pi$ -electron bonding networks in these molecules are the principal reason for their typical NLO properties [6,7]. Recent years the centrosymmetric or asymmetric structure, non-substituted or substituted configuration, and the metal derivatives of these organic compounds have been commonly studied [8,9]. Porphyrins are one of the attractive organic compounds with large nonlinear optical (NLO) properties [10–14]. Moreover, the electronic structure of porphyrins can flexibly be tailored by the variation of peripheral substituents or the ligands of central atoms for tuning their NLO properties [5,15]. Recent years, significant efforts have been devoted to the establishment of structure-NLO properties relationships for porphyrin-based molecules so that materials with improved and optimized NLO properties can be rationally designed and synthesized for particular applications. As expected, substantial progresses have been achieved, especially for the optical limiting (OL) property in the form of reverse saturable absorption (RSA) [5,8,16–20]. It was concluded that the enhancement of OL behavior can be indeed achieved by the variation of peripheral substituents, the extension of  $\pi$ -conjugation and the ligand of central atom [9,21,22]. In addition, saturable absorption (SA) in porphyrin-based compounds was also observed, which are mainly dependent on excitation wavelength, excitation intensity, concentration and molecular structure [23–25].

Aside from the referred nonlinear absorption of RSA and SA, nonlinear refraction is also an important NLO process, which can be caused by nonlinear absorption *via* causality and Kramers-Kronig relations [26]. As nonlinear refraction could lead to a modulation of the phase or amplitude of an intense optical wave in a nonlinear medium, hence it has attracts significant scientific and technological interest for its various applications related to optical limiting, nonlinear spectroscopy, optical switching and optical communications and so on [27,28]. On the other hand, nonlinear refractive index may result in unwanted phase distortion or even optical damage, which should be taken well consideration in the high power beam related experiments and applications. Therefore, the investigation of nonlinear refraction properties is also of great significance for particular applications. Until now, although the nonlinear refraction properties of some porphyrin-related molecules were reported [21,22,29–35], the reports on nonlinear refraction of porphyrins are still sparse.

The Z-scan technique is a simple and effective method to study the nonlinear absorption and refraction properties of nonlinear media [36], which has been widely used in ns, ps, and fs regimes [12,22,29]. In this paper, we reported our recent finding on the nonlinear absorption and refraction properties of *meso*-tetraphenylporphyrin (TPP) and its two substituted derivatives (*meso*-tetrakis(4-cyanophenyl)porphyrin [TPP(CN)<sub>4</sub>] and *meso*-tetrakis(4-methoxyphenyl)porphyrin [TPP(OMe)<sub>4</sub>]) using the Z-scan technique in the picosecond (ps) and nanosecond (ns) regimes. Furthermore, the second molecular

hyperpolarizabilities of these three porphyrins are calculated. The molecular structures of TPP, TPP(CN)<sub>4</sub> and TPP(OMe)<sub>4</sub> are shown in Figure 1.

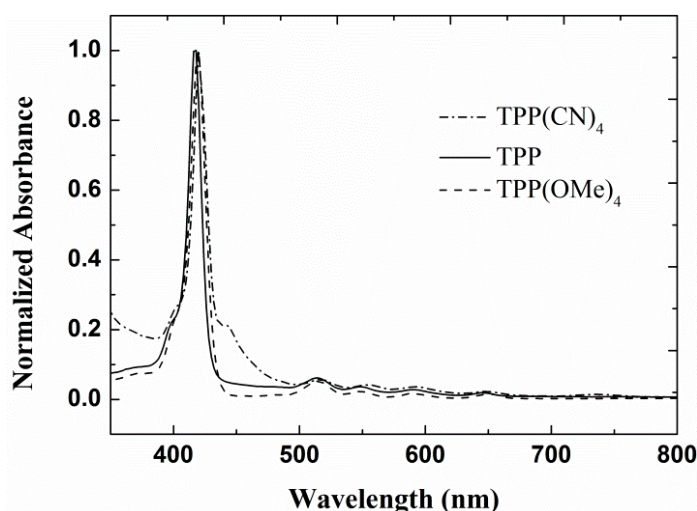


**Figure 1.** The molecular structures of (a) *meso*-tetraphenylporphyrin (TPP); (b) TPP(CN)<sub>4</sub> and (c) TPP(OMe)<sub>4</sub>.

## 2. Results and Discussion

### 2.1. Linear Absorption Characterization

The UV-visible spectra of TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub> in CH<sub>2</sub>Cl<sub>2</sub> are presented in Figure 2. The Soret band of these compounds is centered around 420 nm, and the Q-bands are all located between 500 and 700 nm [12]. It can be seen from Figure 2 that the incorporation of the donor methoxy group and acceptor cyano group into TPP did not arouse obvious change in its linear absorption spectra.

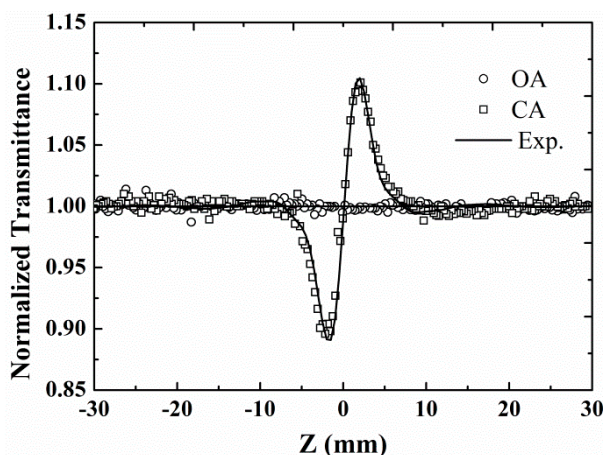


**Figure 2.** UV-visible absorption spectra of TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub> in CH<sub>2</sub>Cl<sub>2</sub>.

### 2.2. Picosecond Z-Scan Measurements

To check the effect of CH<sub>2</sub>Cl<sub>2</sub> solvent on the nonlinear absorption and refraction of the entire porphyrin/CH<sub>2</sub>Cl<sub>2</sub> solution, the Z-scan measurements are first performed on CH<sub>2</sub>Cl<sub>2</sub>, as shown in

Figure 3. The circles show the normalized transmission without an aperture (open aperture, OA), which is varying as a function of distance along the lens axis. The result reveals that no nonlinear absorption of  $\text{CH}_2\text{Cl}_2$  could be observed within the limit of the intensity ( $1.1 \text{ GW}/\text{cm}^2$ ) used in the experiment. However, the closed aperture (CA) result (squares) shows an obvious valley-peak configuration, which indicates a positive nonlinear refraction.

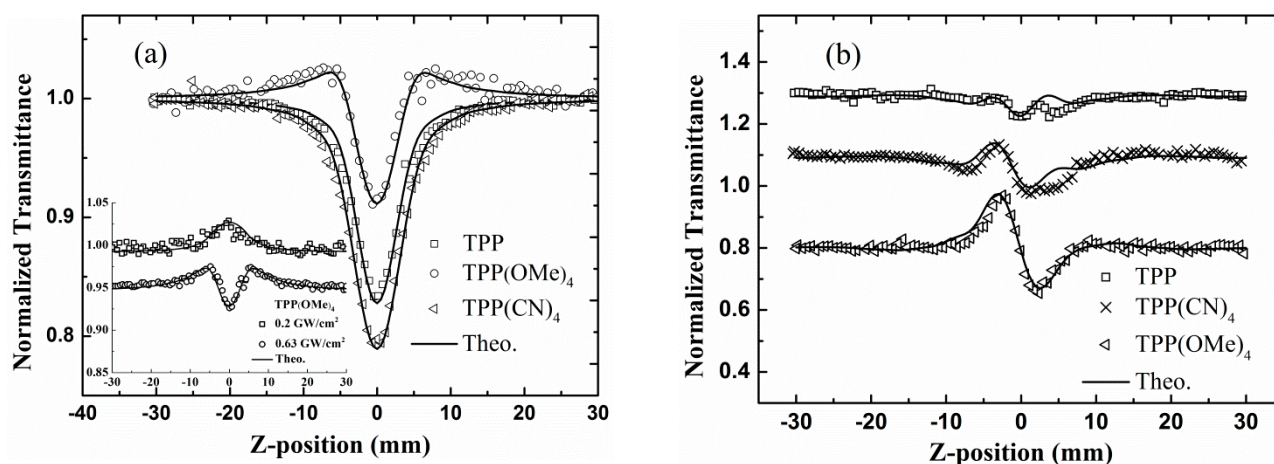


**Figure 3.** Open aperture (OA) and closed aperture (CA) Z-scan data of  $\text{CH}_2\text{Cl}_2$  with the same on-axis intensity  $1.1 \text{ GW}/\text{cm}^2$ .

The Z-scan measurements along with the corresponding fits for TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub> in  $\text{CH}_2\text{Cl}_2$  are shown in Figure 4. Figure 4a presents the OA Z-scan results. TPP and TPP(CN)<sub>4</sub> both exhibit a notable valley at the focus, indicating a reverse saturable absorption (RSA) character with a positive nonlinear absorption coefficient. Whereas TPP(OMe)<sub>4</sub> shows a different absorption property. It experiences a switch from saturable absorption (SA) at a low input intensity to RSA at higher intensities. The inset in Figure 4a clearly provides the SA pattern of TPP(OMe)<sub>4</sub> at the input intensity  $0.2 \text{ GW}/\text{cm}^2$ , and the coexistence of SA and RSA at  $0.63 \text{ GW}/\text{cm}^2$ . It can be concluded that RSA effect will play a dominant role as the input intensity increasing largely enough. In consideration of the negligible nonlinear absorption of  $\text{CH}_2\text{Cl}_2$  (in Figure 3), the nonlinear absorption effects in Figure 4a can be mainly attributed to the solutes rather than  $\text{CH}_2\text{Cl}_2$  solvent. As mentioned above, the occurrence of SA heavily depends on excitation wavelength, excitation intensity, concentration and molecular structure [18–20], and the first three parameters for all the compounds are same in our experiment, thus we attributed the SA-to-RSA behavior in TPP(OMe)<sub>4</sub> to its unique molecular structure. This implies that the incorporation of electron-donating groups (methoxy) into TPP can turn its nonlinear absorption property.

The OA and CA Z-scan curves are fitted based on the coupled equations involving the irradiance and the nonlinear phase shift [36]. Furthermore, the nonlinear absorption switching behavior of TPP(OMe)<sub>4</sub> is described by a nonlinear absorption coefficient of  $\alpha(I) = \frac{\alpha_0}{1+I/I_s} + \beta_{\text{eff}}I$  [37]. Here,  $\alpha_0$  is the linear absorption coefficient of the sample.  $\beta_{\text{eff}}$  is the effective nonlinear absorption coefficient and  $I_s$  is the saturable intensity. For the RSA behavior in TPP and TPP(CN)<sub>4</sub>, the saturable intensity  $I_s \rightarrow \infty$  and hence the nonlinear absorption coefficient can be simplified to  $\alpha(I) = \alpha_0 + \beta_{\text{eff}}I$ . The fittings

in Figure 4a with such nonlinear absorption coefficients yield  $\beta_{\text{eff}}$  values for TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub>, which are 0.65, 1.0, and 4.0 cm/GW, respectively. Furthermore, the saturable intensity  $I_s$  of TPP(OMe)<sub>4</sub> can be obtained by fitting and is  $1.25 \pm 0.25$  GW/cm<sup>2</sup>.



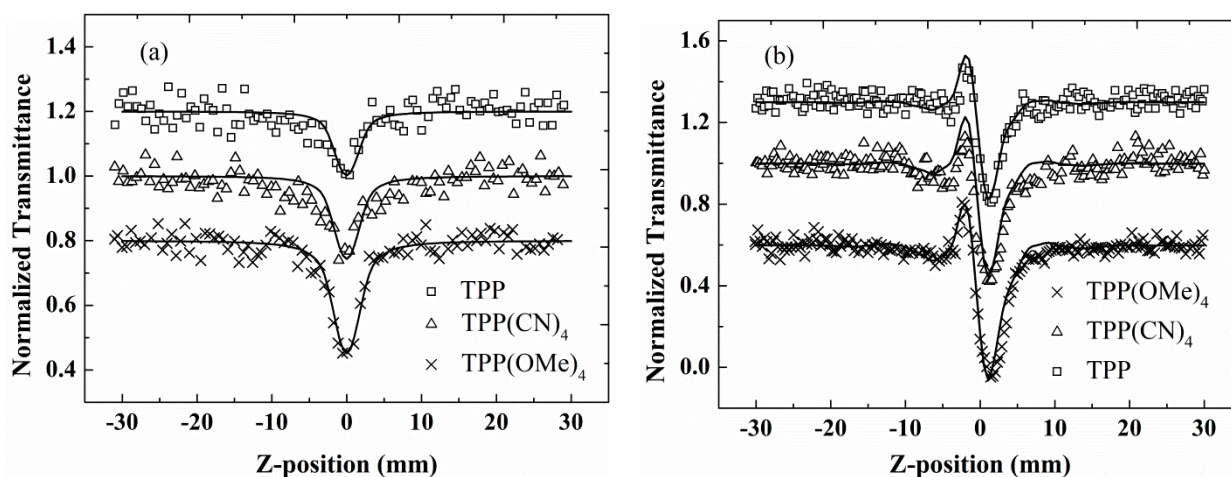
**Figure 4.** Open (a) and closed aperture (b) Z-scan data of TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub> with 21 ps pulses, on-axis intensity of 1.1 GW/cm<sup>2</sup>. Solid lines are theoretical fits. Inset shows OA Z-scan data at 0.2 and 0.63 GW/cm<sup>2</sup>, respectively.

The CA Z-scan curves (Figure 4b) for the three porphyrins in CH<sub>2</sub>Cl<sub>2</sub> exhibit primary peak-valley configurations. Since the CA Z-scan result of CH<sub>2</sub>Cl<sub>2</sub> solvent has a valley-peak configuration (Figure 3), it is obvious that the presence of the curves in Figure 4b is attributed to the combined effect of porphyrins solute and CH<sub>2</sub>Cl<sub>2</sub> solvent. Furthermore, the solvent contribution has reduced the actual magnitude of nonlinear refraction in these porphyrins due to the opposite signs of their nonlinear refraction. Therefore, all the measured porphyrins have a negative nonlinear refractive index. In addition, it is clear in Figure 4b that TPP(OMe)<sub>4</sub> has the largest difference between the normalized transmittance at peak and valley positions. This implies a larger nonlinear refraction index of TPP(OMe)<sub>4</sub> than those of TPP and TPP(CN)<sub>4</sub>. While TPP with the smallest magnitude of normalized transmittance holds a comparative absolute value of nonlinear refraction index compared with that of CH<sub>2</sub>Cl<sub>2</sub>. The theoretical curves are obtained from the best fits with the change in refractive index expressed as  $\Delta n = n_{2\text{eff}} I$ , where,  $n_{2\text{eff}}$  is the effective nonlinear refractive index. The curves are reasonably well matches with the experimental data. The true  $n_{2\text{eff}}$  values for these molecules are extracted dependent on the experimental data in Figure 3 and Figure 4b, which are  $-1.9$ ,  $-3.2$  and  $-9.6$  ( $\times 10^{-6}$  cm<sup>2</sup>/GW) for TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub>, respectively. TPP(CN)<sub>4</sub> and TPP(OMe)<sub>4</sub> have notable nonlinear refraction compared to their parent system TPP. Moreover, the nonlinear refraction of TPP(OMe)<sub>4</sub> has a more remarkable enhancement.

### 2.3. Nanosecond Z-Scan Measurements

To examine the nonlinear properties of these porphyrins in nanosecond regime, we performed the Z-scan measurements with 6 ns laser pulses. Figure 5a gives the open aperture Z-scan curves. Each of the compounds displays RSA characters under the excitation of nanosecond pulses. The CA curves of these three porphyrins are all characterized by peak-valley configurations (Figure 5b), which indicate

the negative nonlinear refractions due to the coexistence of electronic effect and thermal-lensing effect [38]. The buildup time of thermal lensing can be determined by  $\omega_0/v_s$  [39].  $\omega_0$  is the laser beam radius at focus and it is  $\sim 19 \mu\text{m}$  in our Z-scan with 6 ns laser pulses. The velocity of sound  $v_s$  in  $\text{CH}_2\text{Cl}_2$  is 1090 m/s [39]. Hence, the calculated buildup time of thermal nonlinearity in our experiment is  $\sim 17.4$  ns. Although it is longer than the pulse duration we used, the thermal lensing effect may even come into play, which belongs to the problem of the thermally induced optical nonlinearities in liquids in the transient regime [40]. Since the nonlinear absorption and refraction of the solvent under the identical experimental conditions are extremely weak, the both of them could be neglected in the ns regime.



**Figure 5.** Open (a) and closed aperture (b) Z-scan data of TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub> with 6 ns pulses, on-axis intensity 27 MW/cm<sup>2</sup>. Solid lines are theoretical fits.

In the thin sample and slowly varying envelope approximations, the beam propagating equations separately governing the irradiance ( $I$ ) and the phase ( $\phi$ ) are:

$$\frac{dI}{dz'} = -(\alpha_0 + \beta_{\text{eff}} I) \quad (1)$$

$$\frac{d\phi}{dz'} = k(n_{2\text{eff}} I + \Delta n_{\text{th}}) \quad (2)$$

Here,  $z'$  is the propagation length in the sample. Parameters  $\alpha_0$  and  $\beta_{\text{eff}}$  denote the linear and nonlinear absorption coefficients, respectively.  $k$  is the wavevector, and  $n_2$  is the nonlinear refraction coefficient.  $\Delta n_{\text{th}}$  is the transient thermal index change induced by the heating of the solution as a result of the absorption of laser pulses. It is obtained by solving the photoacoustic wave equation [38,40], and it is expressed as Equation (3) (see [38,40]).

$$\frac{\partial^2 \Delta n_{\text{th}}}{\partial t^2} - v_s^2 \nabla^2 (\Delta n_{\text{th}}) = (dn/dT) v_s^2 \nabla^2 (\Delta T) \quad (3)$$

Herein,  $dn/dT$  is the rate of change of the index with respect to temperature and  $\Delta T$  is the temperature change in the sample induced by the absorption of laser radiation, which is described by:

$$\Delta T = \frac{1}{\rho_0 C_p} \int_{-\infty}^t (\alpha_0 + \beta_{\text{eff}} I) I dt \quad (4)$$

The related parameters in the simulations are taken from Ref. [39], which are  $v_s = 1090$  m/s,  $dn/dT = 5.5 \times 10^{-4}$  K $^{-1}$ ,  $C_p = 1.6 \times 10^3$  J/(cm $^3$  K), respectively. The density of CH $_2$ Cl $_2$  is  $\rho_0 = 1.33 \times 10^3$  kg/m $^3$ . Based on the theory and the related parameters above, the nonlinear coefficients of these molecules could be extracted, which are 100, 150 and 360 cm/GW for TPP, TPP(CN) $_4$ , and TPP(OMe) $_4$ , respectively. Meanwhile the nonlinear refractive index of TPP, TPP(CN) $_4$ , and TPP(OMe) $_4$  are  $-1.3$ ,  $-1.7$  and  $-4.6$  ( $\times 10^{-3}$  cm $^2$ /GW), respectively. Note that the electronic effect plays a dominant role in the nonlinear refraction of these three porphyrins. It can be seen that the electron-withdrawing substituent CN and electron-donating substituent OMe both have led to the enhancement of nonlinear refraction in the ps and ns regimes, whereas their effects on the nonlinear absorption property are weak as compared with that of TPP due to the different extent of linear absorption, except that TPP(OMe) $_4$  has a SA-to-RSA switch in the case of ps pulses.

#### 2.4. Second-Order Molecular Hyperpolarizabilities

As controlling of the second-order molecular hyperpolarizabilities ( $\gamma$ ) is crucial for many applications in electronic and photonic devices, we calculated the  $\gamma$  values of the porphyrins investigated by the formulation Equation (5) [41]:

$$\gamma = \frac{\chi^{(3)}}{N_c L} \quad (5)$$

where  $N_c$  is the molecular number density per cubic centimeter,  $L$  is the local-field correction factor, which is approximately to be  $\left(\frac{n_0^2 + 2}{3}\right)^4$ .  $\chi^{(3)}$  is the nonlinear susceptibility, which can be obtained by the real ( $\text{Re } \chi^{(3)}$ ) and imaginary ( $\text{Im } \chi^{(3)}$ ) parts of it by:

$$\text{Re } \chi^{(3)} = \frac{cn_0^2}{120\pi} n_{2\text{eff}} \quad (6)$$

$$\text{Im } \chi^{(3)} = \frac{c^2 n_0^2}{240\pi\omega} \beta_{\text{eff}} \quad (7)$$

Here,  $c$  is the speed of light,  $n_0$  is the linear refractive index of the sample, and  $\omega$  is the angular frequency of the laser pulses.  $n_{\text{eff}}$  and  $\beta_{\text{eff}}$  have been extracted above. In the picosecond regime, the  $\gamma$  values of TPP, TPP(CN) $_4$ , and TPP(OMe) $_4$  are  $1.1 \times 10^{-30}$ ,  $1.7 \times 10^{-30}$ , and  $5.8 \times 10^{-30}$  esu, respectively. Both TPP(CN) $_4$  and TPP(OMe) $_4$  have an enhanced  $\gamma$  value as compared to that of TPP. While in the nanosecond regime, the  $\gamma$  values for TPP, TPP(CN) $_4$ , and TPP(OMe) $_4$  are 441, 585, and 1563 ( $\times 10^{-30}$  esu), respectively.

Recently, the second-order molecular hyperpolarizabilities of some porphyrin-related compounds in picosecond regime have been reported. Venugopal Rao *et al.* [42] have reported  $\gamma$  values at 532 nm, 35 ps from the DFWM measurements of sixteen porphyrins with different metal ions in the ring. The highest value of  $1.432 \times 10^{-30}$  esu was observed for CoTPP. The difference of nonlinear optical nonlinearities between the porphyrins is attributed to the higher electron delocalization in the molecules. Chen *et al.* [41] have reported the second molecular hyperpolarizabilities of a azobenzene-phthalocyanine dyad measured by a Z-scan technique at 532 nm with a pulse duration of 25 ps. which are  $3.87 \times 10^{-30}$  and  $4.82 \times 10^{-30}$  esu before and after illumination with 365 nm UV light, respectively. Shi *et al.* [30]

reported the second hyperpolarizabilities of four porphyrin-heteropolyoxometalate hybrid system and TPP, which are characterized by Z-scan measurements at 532 nm with a pulse duration of 20 ps. The  $\gamma$  value they calculated for TPP is  $0.84 \times 10^{-30}$  esu. The other four compounds possess  $\gamma$  values between  $0.49\text{--}1.2 \times 10^{-30}$  esu. Obviously, the  $\gamma$  value of TPP(OMe)<sub>4</sub> in our study is remarkable larger than those of the related compounds mentioned above in the ps regime. The obtained results imply that the introduction of the cyano or methoxy groups into TPP has obviously tuned its refraction and absorption properties, leads to the enhancements of the nonlinear refraction and the second molecular hyperpolarizabilities, which is expected to provide useful reference for exploring porphyrin derivatives with much higher second hyperpolarizabilities [43].

### 3. Experimental Section

5,10,15,20-Tetraphenylporphyrin (TPP), 5,10,15,20-tetrakis(4-cyanophenyl)porphyrin [TPP(CN)<sub>4</sub>] and 5,10,15,20-tetrakis(4-methoxyphenyl)porphyrin [TPP(OMe)<sub>4</sub>] were synthesized following the method used by Capitosti *et al* [44].

The nonlinear absorption and nonlinear refraction of TPP, TPP(CN)<sub>4</sub> and TPP(OMe)<sub>4</sub> were measured by Z-scan techniques in ps and ns regime. The light source of 21 ps (FWHM) pulses is generated from a Q-switched Nd: YAG laser (PL2143A, EKSPLA, Vilnius, Lithuania) with a wavelength of 532 nm. The laser system for nanosecond Z-scan was a Surelite Q-switched Nd:YAG 532 nm laser (Continuum, Santa Clara, CA, USA) with a pulse width of 6 ns (FWHM). The spatial profiles of both ps and ns pulses have a top-hat distribution. The Z-scan techniques were arranged as that reported by Yang *et al.* [45]. The beam waist at the focus is  $\sim 26$   $\mu\text{m}$  for ps and  $\sim 19$   $\mu\text{m}$  for ns, respectively. The porphyrins are dissolved in dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>) with the same concentration of  $8.14 \times 10^{-5}$  M and placed in 2 mm-thickness quartz cells, the corresponding linear absorption coefficients are 0.8, 1.1, and 5.8  $\text{cm}^{-1}$  for TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub>. The incident and transmitted pulses energies were measured simultaneously by the same type energy detectors (RjP-765a, Laser Probe Inc., Utica, NY, USA). The energy meter we used is a Laser Probe Inc., Rj-7620 ENERGY RATIONOMETER.

### 4. Conclusions

The nonlinear absorption and nonlinear refraction properties of TPP, TPP(CN)<sub>4</sub>, and TPP(OMe)<sub>4</sub> are measured by the Z-scan technique at 532 nm in the ps and ns regimes. TPP and TPP(CN)<sub>4</sub> display an identical reverse saturable absorption (RSA) character in the ps and ns regimes. While TPP(OMe)<sub>4</sub> exhibits a transition from saturable absorption (SA) to RSA in the ps regime, and a typical RSA character in the ns regime. Furthermore, TPP(CN)<sub>4</sub> and TPP(OMe)<sub>4</sub> show regular enhancement of the magnitude of nonlinear refraction compared to their parent TPP in both the ps and ns regimes. The obtained results related to the three porphyrins in both the ps and ns regimes suggest that the introduction of the cyan or methoxy group into TPP has obviously tuned its NLO properties, leads to the enhancements of the nonlinear refraction and the second molecular hyperpolarizabilities, which is useful for the development of porphyrin-related applications upon desired nonlinear optical properties.



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## Author Contributions

Xueru Zhang and Yinglin Song conceived and designed the experiments; Guanghong Ao and Zhengguo Xiao carried out the Z-scan measurement of these compounds and Guanghong Ao wrote the manuscript; Xuemin Qian synthesized and provided these samples; Zhongguo Li and Yuxiao Wang provided technical support and gave professional guidance. They all have approved the final version of the manuscript.

## Conflicts of Interest

The authors declare no conflict of interest.

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*Sample Availability:* Samples of the compounds investigated here are available from the authors.

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