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# The role of $\beta$ -hairpin conformation in ester hydrolysis peptide catalysts based on a TrpZip scaffold†

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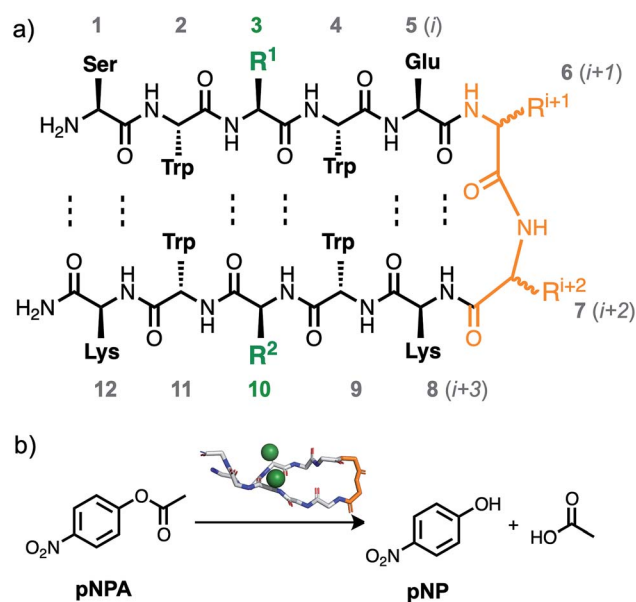
To explore the role of peptide conformation on catalytic activity in the context of ester hydrolysis catalysts, pairs of sequences were designed that contained or lacked  $\beta$ -hairpin character. For the hydrolysis of *para*-nitrophenylacetate in aqueous media, we found small but consistent trends wherein His-containing sequences based on a TrpZip scaffold showed higher catalytic activity without  $\beta$ -hairpin character.

## Introduction

Peptide and peptidomimetic catalysts are important for understanding principles of enzymatic catalysis, and they are also useful catalysts in their own right in a variety of synthetic applications, including in the synthesis of natural products.<sup>1–4</sup> Peptide catalysts provide a middle ground between small molecule organocatalysts and full-length enzymes, retaining the synthetic accessibility of the former, and some aspects of the complex multifunctional, three-dimensional, and chiral environment of the latter. While the rate enhancement and stereoselectivity of most current peptide catalysts is not on the scale of enzymatic catalysis, studying catalytic peptides can increase their synthetic utility.<sup>3,4</sup> Elements of secondary structure have often been used in the design of peptide catalysts,<sup>5–10</sup> which can serve to orient reactive residues near each other in space to mimic the catalytic triads and dyads of enzyme active sites. The study of ester hydrolysis catalysts has been used as a starting point for more challenging hydrolysis reactions,<sup>11</sup> and has important practical applications itself,<sup>12</sup> such as in the depolymerization of polyester plastics.<sup>13,14</sup> Peptide ester hydrolysis catalysts have been reported using a variety of structural scaffolds including  $\beta$ -hairpins,<sup>7,15,16</sup> polyproline helices,<sup>9,17</sup>  $\alpha$ -helices,<sup>5</sup> cyclic peptides,<sup>18</sup> dendrimers,<sup>19</sup> and more.<sup>20</sup> Further investigation of the effect of secondary structure is important for determining the role of conformation on catalysis and in guiding the design of future catalysts.

$\beta$ -Hairpins are among the smallest design motifs that form well-folded monomeric structures.<sup>21,22</sup>  $\beta$ -Hairpins form two

faces, as the side chains alternate above and below the plane of hydrogen bonding, with cross-strand residues oriented on the same face (Scheme 1a). In the case of  $\beta$ -hairpin ester hydrolysis catalysts, catalytic sequences have been reported that were designed *de novo*,<sup>15</sup> and that were identified from combinatorial screening.<sup>7,16</sup> Combinatorial screens have been performed using invariant turn-nucleating residues to create a  $\beta$ -hairpin scaffold, and variable positions in the residues oriented on the



**Scheme 1** (a)  $\beta$ -Hairpin sequences based on the TrpZip sequence (PDB ILE0) were used with modifications in positions 3 and 10 to introduce catalytically active residues and in positions 6 and 7 to bias the sequences towards or against  $\beta$ -hairpin conformation. (b) Catalytic activity of the peptides towards the hydrolysis of *para*-nitrophenylacetate (pNPA) into *para*-nitrophenol (pNP) in water was examined.

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same face of the  $\beta$ -hairpin (either in the hydrogen bonding or non-hydrogen bonding positions). Sequences that are active ester hydrolysis catalysts can result in rate transformation through a number of effects, including binding to the substrate, altering side chain  $pK_a$ , and directly playing a role in the catalysis of the transformation as a nucleophile or base. Thus, studies that used different solvents and ester substrates may not result in sequences that can be directly compared. However, it is notable that prior reports in different reaction conditions are consistent in identifying sequences with multiple His residues as the most active catalysts<sup>7,16,20</sup> even in work where it was hypothesized that a ‘catalytic triad’ similar to the classical His Ser Asp might emerge from screening. Imidazole functional groups, present in His side chains, are known to catalyze ester hydrolysis,<sup>23,24</sup> so we were curious about the role played by the peptide conformation. Therefore, we sought to explore closely the role of  $\beta$ -hairpin conformation for histidine-containing ester hydrolysis catalysts. We hypothesized that peptides with a  $\beta$ -hairpin conformation would be more active catalysts than closely related analogues without  $\beta$ -hairpin conformation, yet we found that this was not the case for the sequences we investigated as catalysts for the hydrolysis of *para*-nitrophenylacetate (**pNPA**) in water (Scheme 1b).

## Results and discussion

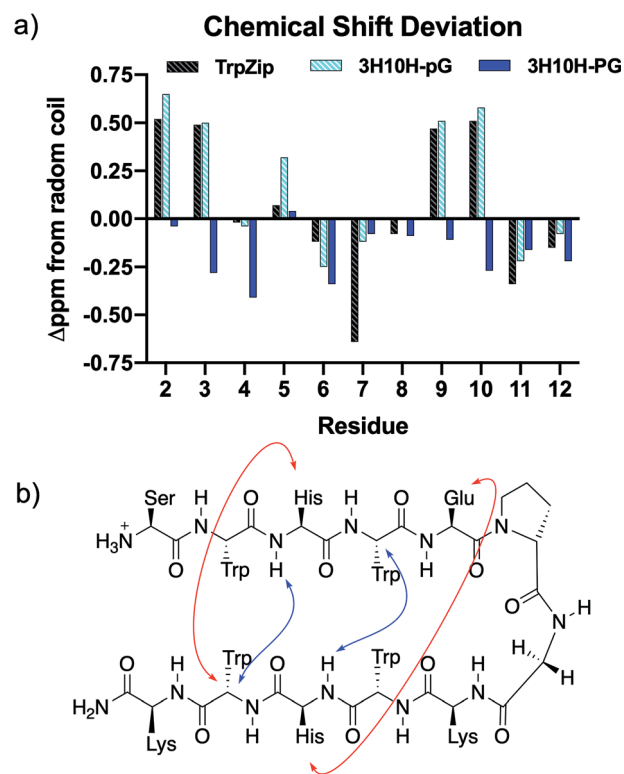
To design sequences that would adopt a  $\beta$ -hairpin conformation, we used a highly stable **TrpZip**  $\beta$ -hairpin sequence as a scaffold. **TrpZip** is a sequence that has been reported to form unusually stable  $\beta$ -hairpins, and contains four tryptophan residues in the non-hydrogen bonding positions.<sup>25</sup> To turn this scaffold sequence into an ester hydrolysis catalyst, we substituted two histidine residues in cross-strand positions, 3 and 10 (Scheme 1a). Other cross-strand pairings were also tested with His in position 3 and either Lys or Ala in position 10. For each of these cross-strand dyads, two versions were compared that varied in the  $i + 1$  and  $i + 2$  turn residues in order to study the role of  $\beta$ -hairpin conformation. To further bias the **TrpZip** scaffold toward a  $\beta$ -hairpin conformation, *D*-Pro-Gly (pG, the lowercase one-letter code refers to a *D* amino acid) was used in the  $i + 1$  and  $i + 2$  turn residues (residues 6 and 7), since this heterochiral pG combination is known to be strongly  $\beta$ -hairpin nucleating.<sup>26–28</sup> For comparison sequences that would not have  $\beta$ -hairpin character, analogue peptides were investigated with Pro-Gly (PG) in residues 6 and 7, as the homochiral combination of Pro and Gly has been found to eliminate  $\beta$ -hairpin formation in peptides.<sup>26,27,29</sup> These design principles resulted in pairs of sequences (Table 1) identical in every respect save for a single chiral center (pG vs. PG). Analogues of these sequences with a truncation of the N- and C-termini were also studied (see ESI†).

To confirm that the pair of sequences contained one with  $\beta$ -hairpin conformation and one without, the conformational behaviours of **TrpZip**, **3H10H-pG**, and **3H10H-PG** in solution were determined using NMR spectroscopy. TOCSY and NOESY/ROESY 2D-NMR experiments were performed for sequential proton resonance assignment (ESI Fig. S5–S13 and Tables S2–

**Table 1** Peptide sequences studied. The N-termini were free amines and the C-termini were amidated. Lowercase one letter code indicates use of *D* amino acid

Peptide	Sequence
<b>TrpZip</b>	NH <sub>2</sub> -SWTWEGNKWTWK-NH <sub>2</sub>
<b>3H10H-pG</b>	NH <sub>2</sub> -SWHWEPGKWHWK-NH <sub>2</sub>
<b>3H10H-PG</b>	NH <sub>2</sub> -SWHWEPGKWHWK-NH <sub>2</sub>
<b>3H10K-pG</b>	NH <sub>2</sub> -SWHWEPGKWKWK-NH <sub>2</sub>
<b>3H10K-PG</b>	NH <sub>2</sub> -SWHWEPGKWKWK-NH <sub>2</sub>
<b>3H10A-pG</b>	NH <sub>2</sub> -SWHWEPGKAWWK-NH <sub>2</sub>
<b>3H10A-PG</b>	NH <sub>2</sub> -SWHWEPGKAWWK-NH <sub>2</sub>

S5†). Determination of the secondary structure was accomplished using chemical shift deviation (CSD) analysis,<sup>30–32</sup> in which the  $\alpha$ -proton chemical shift for each residue was compared to random coil, with positive deviations in the strands indicative of  $\beta$ -sheet conformation. As seen in Fig. 1a, the CSDs of peptide **3H10H-pG** were very similar to the **TrpZip** sequence which has been reported to form a stable  $\beta$ -hairpin,<sup>25</sup> indicating  $\beta$ -hairpin character, while the **3H10H-PG** sequence did not show any signs of  $\beta$ -hairpin character *via* positive CSDs.



**Fig. 1** (a) Chemical shift deviation analysis comparing the chemical shift of the  $\alpha$ -proton in each residue to random coil values. The **3H10H-pG** sequence had similar CSD values to the **TrpZip** peptide, while the **3H10H-PG** sequence did not show  $\beta$ -hairpin character. (b) Selective NOE interactions observed in **3H10H-pG** provided further evidence of  $\beta$ -hairpin formation. Red arrows indicate long-range NOEs involving side chains. Blue arrows indicate long-range backbone NOEs.

Additionally, cross-strand NOEs were observed for **3H10H-pG**, shown in Fig. 1b, further indicating  $\beta$ -hairpin conformation. With this confirmation of the effect of the pG and PG turn residues on conformation (as well as circular dichroism, ESI Fig. S25<sup>†</sup>), the conformation of other sequences with D-Pro-Gly in the turn positions were assumed to also adopt  $\beta$ -hairpin conformation similarly to **3H10H-pG**, and the sequences with Pro-Gly in the turn positions were assumed to not have  $\beta$ -hairpin character similarly to the **3H10H-PG** sequence.

Each peptide sequence was evaluated for its catalytic efficiency in the hydrolysis of pNPA by monitoring the absorbance of the resulting product *para*-nitrophenol (pNP) to determine the initial velocity of the reaction ( $V_0$ ). The reactions were performed using 5 mM pNPA and 0.1 mM peptide catalyst (2 mol%) in 100 mM HEPES buffer (pH 7.4) with 5% acetonitrile for substrate solubility (Fig. 2). The catalytic performance of the

individual amino acid His was used as a comparison. We found that the peptide **3H10H-PG**, which lacks a  $\beta$ -hairpin conformation, led to increased rate enhancement of pNPA hydrolysis compared to the identical sequence containing a  $\beta$ -hairpin conformation, **3H10H-pG** (Fig. 2a).

Since the closer spatial proximity of the His residues in the  $\beta$ -hairpin catalyst **3H10H-pG** did not show increased catalytic activity relative to **3H10H-PG**, we wondered how the effect of the His-His cross-strand interaction would compare to the effects of pairings of His in position 3 across from other amino acids. Therefore, we investigated  $\beta$ -hairpin sequences that paired His in position 3 across from Lys or Ala in position 10 (**3H10A-pG** and **3H10K-pG**). When comparing the catalytic performance of these three  $\beta$ -hairpin catalysts (striped bars in Fig. 2b), the  $V_0$  values were very similar, but **3H10H-pG** contained two His residues per peptide, while **3H10A-pG** and **3H10K-pG** contained one each. Therefore, we compared the initial velocities per His residue in Fig. 2c, which showed the His-His pairing led to lower rate enhancement than His-Ala or His-Lys in a  $\beta$ -hairpin structure on a per His residue basis (striped bars). This enforced the finding from the **3H10H-pG** and **3H10H-PG** comparison that for this combination of catalyst scaffold, substrate, and reaction conditions, the positioning of His residues in close spatial proximity had a detrimental effect on catalytic activity. When compared on a per-His equivalent, the His-His containing peptides **3H10H-pG**, and **3H10H-PG** were actually less active than the single amino acid.

The **3H10A-pG** and **3H10K-pG** sequences were also compared to nearly identical sequences without  $\beta$ -hairpin conformation, **3H10A-PG** and **3H10K-PG**. In all of the cross-strand pairings (His-His, His-Ala, His-Lys) the sequence without  $\beta$ -hairpin character (PG turn) had slightly higher rate enhancement than the analogous hairpin sequence (pG turn), a small effect but perceptible relative to the error tolerance. In a linear peptide, the identity of residue 10 would not be expected to affect the reactivity of residue 3, such that on a per His equivalence basis all of the non-hairpin (PG containing) sequences (solid bars) would be expected to have similar rate enhancement. Since the reactions catalyzed by sequences with a single His residue (**3H10A-PG** and **3H10K-PG**) had higher initial velocities per His residue than the sequence with multiple His residues (**3H10H-PG**), the apparent influence of position 10 on position 3 indicates some secondary structure may be forming. However, the CSD analysis indicates that it is not a  $\beta$ -hairpin (Fig. 1a). Similar reactivity trends were observed for the truncated sequences (data in ESI<sup>†</sup>), indicating that the trends, while small, are consistent.

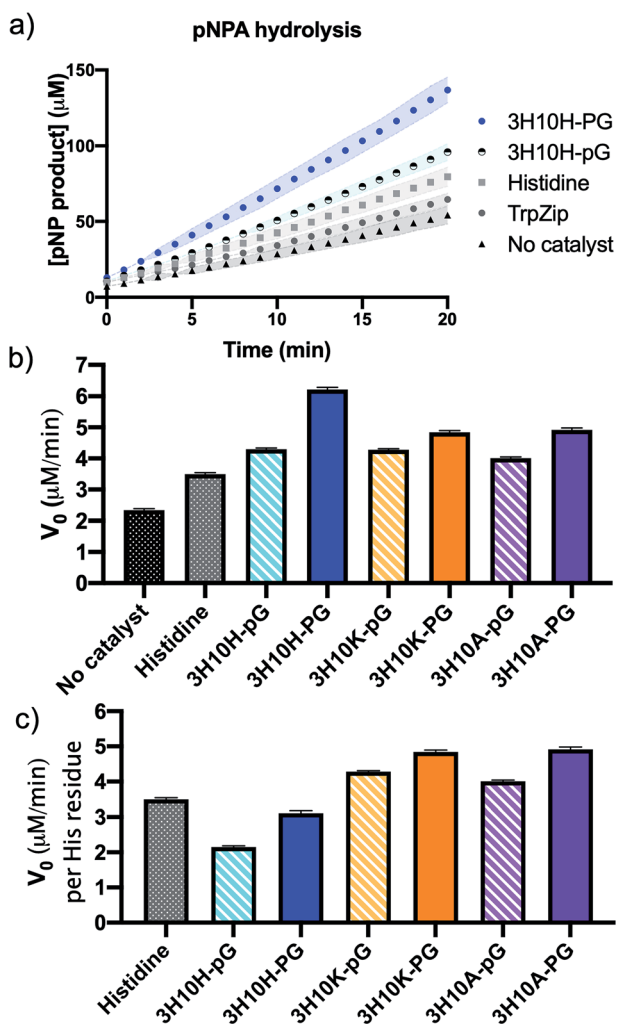


Fig. 2 (a) pNPA hydrolysis plot. The shaded areas represent the area between error bars from 9 replicates in three separate experiments. (b) A comparison of initial velocity of each sequence shows **3H10H-PG** as the catalyst resulting in the highest rate enhancement, however in (c) when the initial velocity is compared on a per histidine equivalent basis, the sequences containing His-His dyads do not show cooperative enhancement.

## Conclusions

In this work, we applied design principles from previous reports on ester hydrolysis catalysts – namely the use of a  $\beta$ -hairpin scaffold, the positioning of catalytically active residues on one face, and the inclusion of multiple His residues as the reactive groups – and studied the effect of these sequences on the rate of the hydrolysis of pNPA in aqueous media. Through the careful design of analogous sequences for comparison, we observed

small yet interesting and consistent trends. A difference in conformation between otherwise identical sequences (caused by the difference in a single chiral center) was observed to lead to differences in catalytic performance. Under the reaction conditions studied here, we found that the sequences without  $\beta$ -hairpin character resulted in increased rate enhancement of **pNPA** hydrolysis compared to the corresponding  $\beta$ -hairpin sequences. Furthermore, while we found sequences with greater His content had rate enhancements, when comparing  $V_0$  per His equivalent, they were less active catalysts. The potential for cooperative activity based on His proximity should be further explored for other substrate, media, and structural scaffolds, but this study indicates that cooperativity should not be assumed – an important consideration in the design of future combinatorial libraries. Future studies will examine His pairings with aromatic groups,<sup>33</sup> investigate the role of pH and His accessibility, and explore further the solution conformation of the non-hairpin peptides.

## Experimental section

Reagents for the *para*-nitrophenylacetate (**pNPA**) hydrolysis assay were purchased from Sigma Aldrich unless otherwise noted. Peptides were purchased from Biomatik with >95% HPLC purity, and the quality control data (mass and HPLC) were verified using a Waters Aquity Arc UHPLC with UV/Vis and QDa detector Liquid Chromatography Mass Spectrometer (LCMS), data in the ESI- $\dagger$  NMR experiments were carried out on a Bruker Advance III-HD-600 NMR spectrometer. UV-Vis absorbance measurements (well plate and cuvette) were carried out using a Tecan M Nano Plus multimode plate reader. Water was purified using a Milli-Q Advantage A10 Water Purification System. The molecular rendering in Scheme 1b was made using the PyMOL Molecular Graphics System, Version 2.0 Schrödinger, LLC.

The hydrolysis of **pNPA** was monitored by measuring the absorbance of the resulting *para*-nitrophenol (**pNP**) product in a multiwell plate. Each catalyst was evaluated nine times – three separate experiments with each catalyst tested in triplicate in three wells. Each well had contained 0.1 mM catalyst and 5 mM **pNPA** in aqueous media buffered by 100 mM HEPES (pH 7.4) with 5% acetonitrile for substrate solubility. The concentration of the peptide catalysts was determined using the absorbance at 280 nm and converting to concentration using the extinction coefficient for the four Trp residues. A standard curve containing known concentrations of **pNP** was used in each plate to convert the absorbance measurements to concentrations. To prepare each multiwell plate experiment, 190  $\mu$ L of catalyst dissolved in HEPES buffer was added to each well, and 10  $\mu$ L of **pNPA** substrate dissolved in acetonitrile was added last to all wells immediately before measuring the absorbance at 405 nm in the plate reader every minute for twenty minutes at 25 °C.

## Author contributions

X. L. and L. S. W. designed the sequences and experiments. X. L. performed the NMR experiments, L. S. W., H. E. G., R. W., G. T.

B. and K. M. B. performed the **pNPA** hydrolysis experiments. L. S. W. and X. L. analyzed the data and wrote the manuscript.

## Conflicts of interest

There are no conflicts to declare.

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