

RESEARCH LETTER

# Comparative structural analysis provides new insights into the function of R2-like ligand-binding oxidase

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(Received 11 December 2021, revised 10 February 2022, accepted 14 February 2022, available online 4 March 2022)

doi:10.1002/1873-3468.14319

Edited by Dietmar Manstein

**R2-like ligand-binding oxidase (R2lox) is a ferritin-like protein that harbours a heterodinuclear manganese–iron active site. Although R2lox function is yet to be established, the enzyme binds a fatty acid ligand coordinating the metal centre and catalyses the formation of a tyrosine–valine ether cross-link in the protein scaffold upon O<sub>2</sub> activation. Here, we characterized the ligands copurified with R2lox by mass spectrometry-based metabolomics. Moreover, we present the crystal structures of two new homologs of R2lox, from *Saccharopolyspora erythraea* and *Sulfolobus acidocaldarius*, at 1.38 Å and 2.26 Å resolution, respectively, providing the highest resolution structure for R2lox, as well as new insights into putative mechanisms regulating the function of the enzyme.**

**Keywords:** aldehyde deformylating oxygenase; ferritin-like protein; hydroxy fatty acids; long-chain fatty acids; R2-like ligand-binding oxidase; R2lox

The ferritin-like superfamily is composed by a large diversity of proteins sharing a common fold, constituted of an  $\alpha$ -helix bundle core housing a dinuclear metal centre [1,2]. Despite their similar architecture, members of this superfamily differ largely in sequence and perform a variety of functions. Well-studied members of this family include the soluble methane monooxygenase which catalyses the conversion of methane to methanol [3], the ribonucleotide reductase (RNR) that reduces ribonucleotides to deoxyribonucleotides [4], the aldehyde

deformylating oxygenase (ADO) which produces alkanes from fatty aldehydes [5], and fatty acid desaturases involved in the synthesis of many unsaturated lipids essential to cells [6].

The metalloenzyme R2-like ligand-binding oxidase (R2lox) was discovered due to its sequence resemblance with the R2 subunit of the class Ic RNR (R2c) [7,8]. Both R2lox and R2c belong to the ferritin-like superfamily and harbour a heterodinuclear manganese–iron centre [9–11]. R2lox has been identified in various

## Abbreviations

ADO, aldehyde deformylating oxygenase; APCI-LC/MS, atmospheric pressure chemical ionization liquid chromatography–mass spectrometry; ESI-LC/MS, electrospray ionization liquid chromatography–mass spectrometry; GkR2loxI, *Geobacillus kaustophilus* R2lox first homolog; HFA, hydroxy fatty acid; MtR2lox, *Mycobacterium tuberculosis* R2lox; PEG, polyethylene glycol; R2, ribonucleotide reductase R2 subunit; R2lox, R2-like ligand-binding oxidase; RNR, ribonucleotide reductase; SaR2loxI, *Sulfolobus acidocaldarius* R2lox first homolog; SaR2loxII, *Sulfolobus acidocaldarius* R2lox second homolog; SeR2lox, *Saccharopolyspora erythraea* R2lox.

organisms, and a number of organisms encode several isoforms, for example, *Sulfolobus acidocaldarius* encodes four R2lox homologues [9]. In R2c, the high-valent Mn(IV)/Fe(III) centre catalyses a single-electron transfer to produce a free radical essential to the synthesis of deoxyribonucleotides [12]. In R2lox, the mature cofactor was characterized by spectroscopy as a Mn(III)/Fe(III) centre bridged by a  $\mu$ -hydroxo/bis- $\mu$ -carboxylato network [13]. The assembly of the R2lox active site takes place upon dioxygen (O<sub>2</sub>) activation and involves the formation of high-valent intermediates which, unlike R2c, catalyse a two-electron reaction to generate a tyrosine–valine ether cross-link in the proximity of the metal centre [14,15]. Different types of covalent cross-links have been observed in proteins to serve different purposes [16–18], including in another di-metal ferritin-like protein where a valine–phenylalanine cross-link has been proposed to stabilize the metal centre [19]. In addition, it was shown that the redox state of the cofactor impacts R2lox structure, leading to the formation of likely routes for O<sub>2</sub> and substrate access to the active site [20]. Despite these findings, the physiological substrate and function of R2lox remain unknown.

To date, crystal structures of R2lox proteins from three organisms have been solved, that is, from *Mycobacterium tuberculosis* (MtR2lox), *Geobacillus kaustophilus* (GkR2loxI) and *Sulfolobus acidocaldarius* (SaR2loxI) (PDB ID: 3EE4, 4HR0 and 6QRZ, respectively) [8,14,21]. The structures show that, although the R2-protein fold is conserved, R2loxes are extensively remodelled to accommodate an unexpected ligand in interaction with the dinuclear metal centre. An additional study showed by mass spectrometry that GkR2loxI copurifies with a mixture of mainly C<sub>16</sub> and C<sub>18</sub> hydroxy fatty acids (HFAs) when the recombinant protein is produced in *Escherichia coli* [20].

Here, using mass spectrometry-based metabolomics, we further characterized some of the ligands that copurify with GkR2loxI and SaR2loxI. In addition, we solved two novel R2lox structures by X-ray crystallography, the highest resolution structure of an R2lox to date from *Saccharopolyspora erythraea* (SeR2lox) at 1.38 Å resolution and the structure of a second R2lox homolog from *Sulfolobus acidocaldarius* (SaR2loxII, 54% sequence identity with SaR2loxI) at 2.26 Å resolution, which exhibits unusual features. Using a comparative structural approach combined with bioinformatics, we obtained new insights into the potential regulatory mechanisms of R2lox and discuss the function of this enigmatic enzyme.

## Materials and methods

### Cloning

Constructs encoding the full-length *Sulfolobus acidocaldarius* R2lox homolog II (accession number WP\_011277966) and full-length *Saccharopolyspora erythraea* R2lox (accession number WP\_009945174) were amplified using polymerase chain reaction from genomic DNA obtained from DSMZ (DSM number 639 and 40517, respectively). Then, constructs were inserted into pET-46 Ek/LIC (Novagen, Merck KGaA, Darmstadt, Germany) in frame with the N-terminal 6xHis tag using the following primers: 5'-GACG ACGACAAGATGGTATTAATTTTCGAAGAATACAA ACATACG-3' (forward) and 5'-GAGGAGAAGCCCGG TTATAAACTCTCTATCACATCCAAGTCCTTTG-3' (reverse) for SaR2loxII, and 5' GACGACGACAAGATGAC GAGCACCGCGACC-3' and 5'-GAGGAGAAGCCCGG TTAGGCCTTTTCCAGCGCG-3' for SeR2lox. The constructs encoding the N-terminally His-tagged full-length R2loxI from *Geobacillus kaustophilus* (accession number yp\_148624) and R2loxI from *S. acidocaldarius* (accession number WP\_011278976) were cloned as described in [14] and [21], respectively.

### Protein production and purification

Expression of SaR2loxII and SeR2lox was carried out in *Escherichia coli* BL21(DE3) (Novagen) in a benchtop bioreactor system (Harbinger). Cells expressing SaR2loxII were cultured in terrific broth medium (Formedium) supplemented with ampicillin (50  $\mu$ g·mL<sup>-1</sup>). Cells expressing SeR2lox were cultured in minimal media (Formedium) supplemented with ampicillin (50  $\mu$ g·mL<sup>-1</sup>), 0.25 mM MnCl<sub>2</sub> and 0.25 mM (NH<sub>4</sub>)<sub>2</sub>Fe(SO<sub>4</sub>)<sub>2</sub>, initially at 37 °C. Expression was induced with 0.5 mM isopropyl  $\beta$ -D-1-thiogalactopyranoside at an OD<sub>600</sub> of 0.8 for 15 h at ambient temperature. Cells were then harvested by centrifugation and stored at -80 °C. Cells expressing SaR2loxII and SeR2lox were disrupted by high-pressure homogenization (Avestin Emulsiflex C3), after resuspension in lysis buffer (25 mM Hepes-Na pH 7.0, 300 mM NaCl and 10 mM imidazole). The lysate was cleared by centrifugation and the supernatant applied to a nickel–nitrilotriacetic acid agarose (Protino) gravity flow column. The sample was washed and eluted using washing buffer (lysis buffer containing 20 mM imidazole) and elution buffer (lysis buffer containing 250 mM imidazole). The protein was concentrated using Vivaspin 20 centrifugal concentrators with a 30,000 molecular weight cut-off polyethersulfone membrane (Sartorius) and applied to a HiLoad 16/60 Superdex 200 prep grade size exclusion column (GE Healthcare Sverige AB, Danderyd, Stockholm, Sweden) equilibrated in a final buffer of 25 mM Hepes-Na pH 7.0 and 150 mM NaCl. Fractions corresponding to the pure protein were pooled, concentrated

(12.5 mg·mL<sup>-1</sup> and 25.6 mg·mL<sup>-1</sup> for *SaR2loxII* and *SeR2lox*, respectively), aliquoted, flash-frozen in liquid nitrogen and stored at -80 °C. The protein concentrations were calculated using theoretical molecular weight for *SaR2loxII* (37010.62) and extinction coefficient at 280 nm of 44350 M<sup>-1</sup> cm<sup>-1</sup>, and *SeR2lox* (37249.35) and extinction coefficient at 280 nm of 53190 M<sup>-1</sup> cm<sup>-1</sup>, as calculated by ProtParam [22]. *SaR2loxI* and the metal-free *GkR2loxI* were produced and purified as described in [21] and [14], respectively.

### Crystallization, X-ray data collection and structure refinement

*SeR2lox* and *SaR2loxII* were crystallized by sitting-drop vapour diffusion method using a Mosquito nanolitre pipetting robot (SPT Labtech, Melbourn, Hertfordshire, UK). For *SeR2lox*, a volume of 200 nL protein solution at 25.6 mg·mL<sup>-1</sup> in 25 mM Hepes-Na pH 7.0 and 150 mM NaCl was mixed with 200 nL of reservoir condition composed of 200 mM lithium sulfate, 100 mM sodium acetate pH 4.6 and 50% (v/v) polyethylene glycol 400, resulting in crystals appearing overnight at 21 °C. For *SaR2loxII*, a volume of 75 nL of a protein solution at 12.5 mg·mL<sup>-1</sup> in 25 mM Hepes-Na pH 7.0 and 150 mM NaCl was mixed with 150 nL of reservoir solution consisting of 24% (w/v) polyethylene glycol 1500, 5% (v/v) formamide, 40 mM sodium propionate, 20 mM sodium cacodylate trihydrate and 40 mM bis-tris propane (pH 9.0). A single crystal grew in a week at 21 °C. Of note, crystallization conditions were optimized from initial hits found using commercial screens, that is, from condition A1 in JCSG plus (Molecular Dimensions) for *SeR2lox* and from condition C3 from PACT premier (Molecular Dimensions) supplemented by condition H5 from Additive Screen (Hampton Research) for *SaR2loxII*. *SeR2lox* and *SaR2loxII* crystals were flash-cooled in liquid nitrogen without additional cryoprotecting agent, and X-ray diffraction data were collected at cryogenic temperature on the beamlines PX14.1 at BESSY (Berlin, Germany) and X06SA at the Swiss Light Source (Villigen, Switzerland), respectively. Data reduction and scaling were performed using XDS [23]. The high-resolution cut-off was determined based on a combination of  $I/\sigma(I)$ ,  $R_{\text{meas}}$  and  $CC_{1/2}$ .

Phasing was performed by molecular replacement using Phaser [24], using the atomic coordinates of *GkR2loxI* (PDB ID: 4HR0 [14]) and *MtR2lox* (PDB ID: 3EE4 [8]) edited with Sculptor [25], as a starting search model for *SaR2loxII* and *SeR2lox*, respectively. A well-contrasted solution was obtained with one molecule per asymmetric unit in space group  $P6_4$  for *SaR2loxII*, and  $P4_32_12$  for *SeR2lox*. Refinement was performed with phenix.refine [26]. Model examination and manual modifications were iteratively conducted with Coot [27]. When all residues visible in the electron density map were built, metal ions and

waters were added, and the model was further refined. Restraints for metal coordination were generated using elbow [28] and manually modified based on previous studies on R2lox structures [8,14]. Metal ions were modelled as Mn(III) and Fe(III) based on previous studies on R2lox [13]. For *SaR2loxII*, the metal ions were refined at an occupancy of 0.35; no electron density could be observed for a putative tyrosine–valine ether cross-link between Tyr170 and Val78. Similarly, no clear electron density could support the presence of a ligand in the protein binding pocket. The protein was modelled from residue 20 to the last protein residue, that is, 308. Residues 126 to 130 could not be modelled in the electron density. The *SeR2lox* structure was modelled from residue Ala6 to Asp296, the metal ions were refined to an occupancy of 1, and electron density connecting the phenolic oxygen of Tyr167 and the C $\beta$  of Val76 was observed. This ether cross-link was restrained to an ideal distance of  $1.5 \pm 0.05$  Å. Quality of models was evaluated using MolProbity [29]. Data reduction and model refinement statistics were compiled using phenix.table\_one. Anomalous difference maps were computed with phenix.maps using Friedel pair reflections unmerged to preserve the contribution of anomalous scattering atoms. All phenix programmes belong to the PHENIX suite version 1.19\_4092 [26]. Figures were prepared using Inkscape 1.0 and the PYMOL Molecular Graphics System, version 2.4, Schrödinger, LLC. Electrostatic protein contact potentials plotted on molecular surfaces shown in Fig. 3 were generated with the vacuum electrostatics tool in PYMOL.

### Ligand identification by ESI-MS and APCI-MS

*GkR2loxI* samples were mixed with 60  $\mu$ L GC-grade chloroform (Sigma-Aldrich, St. Louis, MO, USA) and vortexed for 30 s. A volume of 30  $\mu$ L of LC-grade methanol (Sigma-Aldrich, St. Louis, MO, USA) was subsequently added, and the samples were vortexed for other 30 s and centrifugated at  $1000\times g$  for 10 min. The chloroform phase was finally extracted and dried under N<sub>2</sub> gas in an HPLC vial, before being stored at -80 °C. The frozen samples were re-suspended before analysis in 20  $\mu$ L methanol, HPLC-grade from Fischer Scientific (Waltham, MA, USA), and Milli-Q water before LCMS analysis. Each batch of samples was first analysed in positive mode and finally in negative.

The chromatographic separation was performed on an Agilent 1290 Infinity UHPLC-system (Agilent Technologies, Waldbronn, Germany). A total of 2  $\mu$ L of each sample were injected onto an Acquity UPLC HSS T3, 2.1  $\times$  50 mm, 1.8  $\mu$ m C18 column in combination with a 2.1 mm  $\times$  5 mm, 1.8  $\mu$ m VanGuard precolumn (Waters Corporation, Milford, MA, USA) held at 40 °C. The gradient elution buffers were A (H<sub>2</sub>O, 0.1% formic acid) and B (75/25 MeOH:2-propanol, 0.1% formic acid), and the flow rate was 0.5 mL min<sup>-1</sup>. The compounds were eluted with a linear gradient consisting of 0.1–10% B over 2 min; B was increased to 99% over 5 min

and held at 99% for 2 min; B was decreased to 0.1% for 0.3 min, and the flow rate was increased to 0.8 mL min<sup>-1</sup> for 0.5 min; these conditions were held for 0.9 min, after which the flow rate was reduced to 0.5 mL min<sup>-1</sup> for 0.1 min before the next injection.

The compounds were detected with an Agilent 6546 Q-TOF mass spectrometer equipped with a jet stream electrospray ion source operating in positive or negative ion mode. The settings were kept identical between the modes, with exception of the capillary voltage. A reference interface was connected for accurate mass measurements; the reference ions purine (4  $\mu$ M) and HP-0921 (Hexakis(1H, 1H, 3H-tetrafluoropropoxy)phosphazine) (1  $\mu$ M) were infused directly into the MS at a flow rate of 0.05 mL min<sup>-1</sup> for internal calibration, and the monitored ions were purine  $m/z$  121.05 and  $m/z$  119.03632; HP-0921  $m/z$  922.0098 and  $m/z$  966.000725 for positive and negative modes, respectively. The gas temperature was set to 150 °C, the drying gas flow to 8 L min<sup>-1</sup> and the nebulizer pressure 35 psig. The sheath gas temp was set to 350 °C, and the sheath gas flow 11 L min<sup>-1</sup>. The capillary voltage was set to 4000 V in both positive and negative ion modes. The nozzle voltage was 300 V. The fragmentor voltage was 120 V, the skimmer 65 V and the OCT 1 RF V<sub>pp</sub> 750 V. The collision energy was set to 0 V. The  $m/z$  range was 70–1700, and data were collected in centroid mode with an acquisition rate of 4 scans s<sup>-1</sup> (1977 transients/spectrum).

For APCI ionization, all chromatographic settings were kept the same. The APCI source settings were as follows: the gas temperature was set to 170 °C, the vaporizer to 250 °C, the gas flow 6 L min<sup>-1</sup>, and the nebulizer pressure 40 psig. The capillary voltage was set to 3000 V in both positive and negative ion modes. The Corona current was 4  $\mu$ A in both modes. The fragmentor voltage was 120 V, the skimmer 65 V and the OCT 1 RF V<sub>pp</sub> 750 V. The collision energy was set to 0 V. The  $m/z$  range was 70–1700, and data were collected in centroid mode with an acquisition rate of 4 scans s<sup>-1</sup> (1977 transients/spectrum).

All data processing was performed using the Agilent Masshunter Qual version B.07.00 (Agilent Technologies Inc., Santa Clara, CA, USA). Reference and tuning standards: purine, 4  $\mu$ M, Agilent Technologies (Santa Clara, CA, USA) HP-0921 (Hexakis(1H, 1H, 3H-tetrafluoropropoxy)phosphazine), 1  $\mu$ M, Agilent Technologies (Santa Clara, CA, USA) Calibrant, ESI-TOF, ESI-L Low Concentration Tuning Mix, Agilent Technologies (Santa Clara, CA, USA) HP-0321 (Hexamethoxyphosphazine) and 0.1 mM, Agilent Technologies (Santa Clara, CA, USA).

### Ligand pocket comparison

The PDB entry 4HR0 of *GkR2loxI* and the refined structures of *SaR2loxII* and *SeR2lox* published in this work were run on the ConSurf server (<https://consurf.tau.ac.il>) [30,31] using the homolog search algorithm HMMER, 1

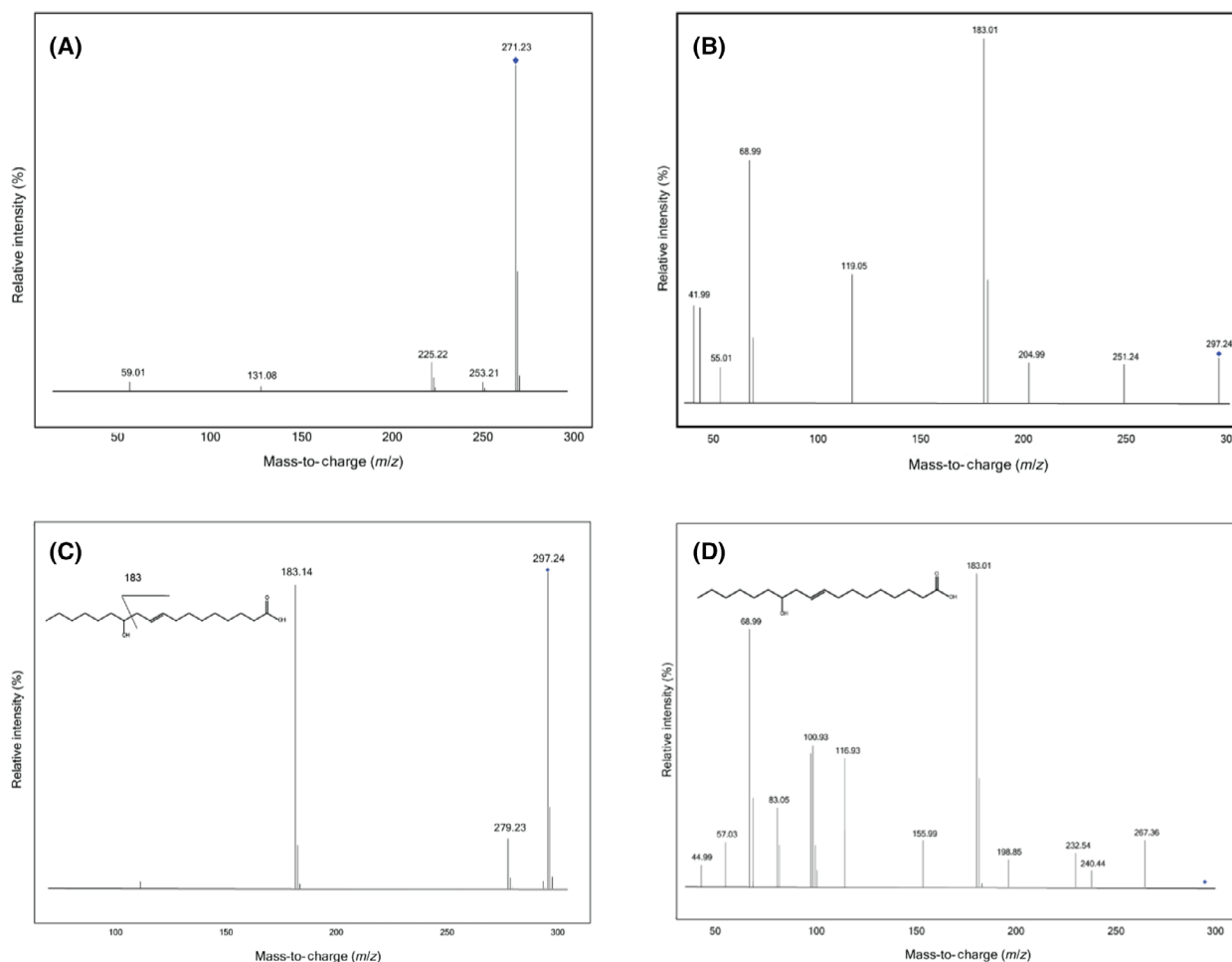
iteration, UniRef90, number of putative homologous sequences to 150 and PSI-BLAST E-value cut-off to 0.0001. The maximal %ID between sequences was set to 95% and the minimal %ID for homologs at 35%. The alignment method was set to MAFFT-L-INS-i.

## Results

### *GkR2loxI* and *SaR2loxI* copurify with a mixture of long-chain HFAs

The ligand copurifying with *GkR2loxI* was previously identified by mass spectrometry as a mixture of long-chain HFAs. However, the position of the hydroxy group could not be determined [20]. In this study, we analysed the pool of ligands in *GkR2loxI* and *SaR2loxI* with electrospray ionization liquid chromatography-mass spectrometry (ESI-LCMS) and compared them to a pool of HFA standards to identify the position of the hydroxy group. Our first observation was that both proteins copurify with a similar mixture of ligands (Fig. S1). We performed ESI-LCMS analysis on the copurified ligands of *GkR2loxI* and can confirm that they mainly are HFAs with a hydroxy group further away from the carboxyl group and in some cases includes a double bond. Chen *et al.* (2016) and Yang *et al.* (2013) show that the position of the hydroxy group and any double bond have a large impact on the fragmentation patterns of HFAs [32,33]. We identified two major liquid chromatography peaks at 271.23  $m\cdot z^{-1}$  and 297.24  $m\cdot z^{-1}$ . The peak at 271.23  $m\cdot z^{-1}$  corresponds to a C<sub>16</sub> HFA, but the MSMS spectra acquired give poor structural information. The fragmentation spectra only confirm the presence of a signal at 225.22  $m\cdot z^{-1}$  corresponding to the loss of a carboxylic group (Fig. 1A). The peak at 297.24  $m\cdot z^{-1}$  corresponds to a C<sub>18</sub> HFA with one double bond. The fragment 251.24  $m\cdot z^{-1}$  corresponds to the loss of the carboxylic acid (Fig. 1B). The presence of a peak at 183.01  $m\cdot z^{-1}$  indicates that the location of the hydroxy group is on position C<sub>12</sub> by comparison with the fragmentation of the standard 12-hydroxy-9(*E*)-octadecenoic acid at 20 eV (Fig. 1C). However, the ligand shows a slightly different fragmentation pattern than 12-hydroxy-9(*E*)-octadecenoic acid at 40 eV (Fig. 1D). This indicates that the double bond is in a different position between the hydroxy and the carboxyl groups. In addition, we performed atmospheric pressure chemical ionization liquid chromatography-mass spectrometry (APCI-LCMS) to identify possible aldehydes, but were unable to find other compounds than the ones already visible in the ESI-LC chromatogram. Finally, in order to search for eventual enzyme products formed





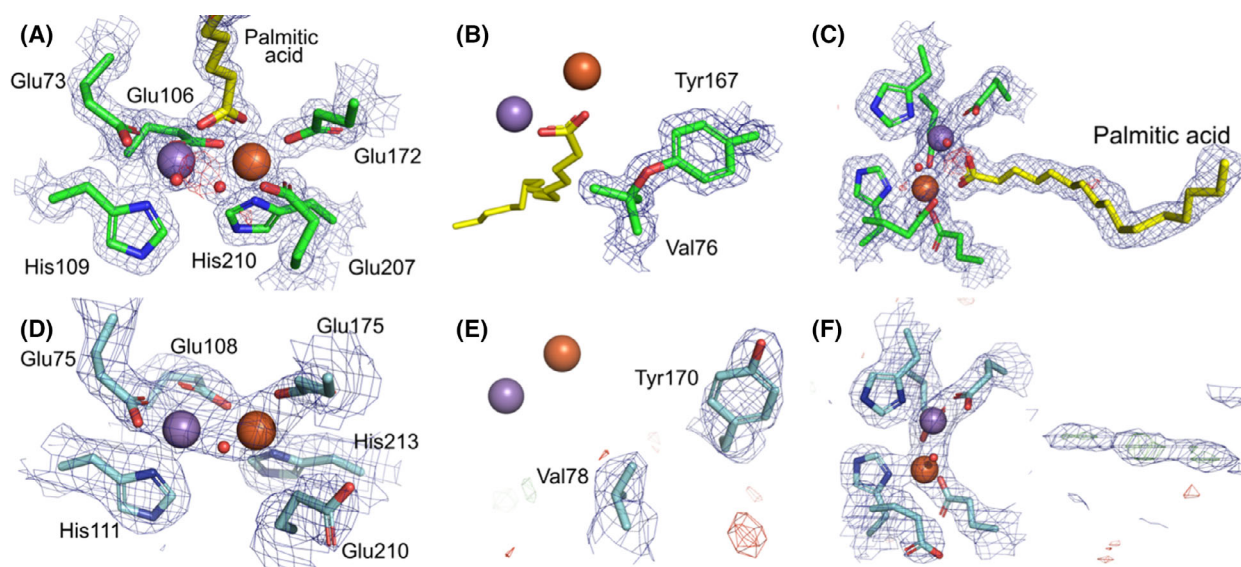
**Fig. 1.** ESI-LCMS fragmentation of copurified ligands from *Gkr2loxI* and the standard 12-hydroxy-9(*E*)-octadecenoic acid. (A) Fragmentation of ion  $271 \text{ m}\cdot\text{z}^{-1}$  at 20 eV. The signal is weak and hard to compare to any standard. The only identifiable ion at  $225 \text{ m}\cdot\text{z}^{-1}$  indicates the loss of the carboxyl group. (B) Fragmentation of ion  $297 \text{ m}\cdot\text{z}^{-1}$  at 40 eV. The fragment  $183.01 \text{ m}\cdot\text{z}^{-1}$  indicates the fragmentation around the hydroxy group, which was identified by comparing with the (C) fragmentation spectra of the standard 12-hydroxy-9(*E*)-octadecenoic acid at 20 eV, which shows that fragmentation around the hydroxy group in position  $\text{C}_{12}$  gives a peak at  $183.14 \text{ m}\cdot\text{z}^{-1}$ . (D) Fragmentation spectra of the standard 12-hydroxy-9(*E*)-octadecenoic acid at 40 eV which, however, shows a different fragmentation to the R2lox ligand in Fig. 1B. The ligand is, therefore, most likely a 12-hydroxy-octadecenoic acid, but the position of the double bond was not identifiable.

upon redox chemistry, we chemically reduced and oxidized *Gkr2loxI* and analysed the ligands with ESI-LCMS and APCI-LCMS. Regrettably, no new compounds were identified following this procedure (Fig. S2).

### ***SeR2lox* high-resolution structure**

The gene coding for *SeR2lox* was recombinantly expressed in *E. coli* using minimal medium supplemented with equimolar amounts of Mn(II) and Fe(II). The crystal structure of the protein was solved to 1.38 Å resolution (Table S1) and currently represents the highest resolution for an R2lox structure. *SeR2lox*

exhibits a canonical ferritin-like fold, along with a ligand-binding pocket accommodating a putative fatty acid molecule (Fig. 2A). The metal site is identical to previously solved *Gkr2loxI* and *Mtr2lox* structures. Mn(III) and Fe(III) ions were modelled at full occupancy in sites 1 and 2, respectively, based on X-ray anomalous diffraction data (Fig. S3A) consistently with previous structural studies [8,14]. The metal ions are bridged by a  $\mu$ -hydroxo ligand and coordinated by the carboxyl group of the fatty acid ligand. Electron density corresponding to the ether cross-link between the phenolic oxygen of Tyr167 and the  $\text{C}\beta$  of Val76 is also clearly observed (Fig. 2B). Palmitic acid ( $\text{C}_{16}$ ) was modelled into the binding pocket, although weak



**Fig. 2.** SeR2lox and SaR2loxII crystal structures. SeR2lox displays a well-defined metal centre (A), tyrosine–valine ether cross-link (B), and ligand modelled as palmitic acid in the binding pocket (C). SaR2loxII exhibits a metal centre at low occupancy (D), no cross-link between Tyr170 and Val78 (E), and its binding pocket is mostly ligand-free (F). Refined electron density maps are shown in mesh: 2F<sub>o</sub>–F<sub>c</sub> maps are coloured in blue and contoured at 2  $\sigma$  in (A) and (B) and at 1  $\sigma$  in (C) to (F); F<sub>o</sub>–F<sub>c</sub> maps are in green (positive) and red (negative) and contoured at 4  $\sigma$ . Manganese and iron ions are depicted as purple and orange spheres, respectively.

electron density extending further away from the pocket possibly suggests the presence of a fatty acid with a longer chain, such as C<sub>18</sub>, at low occupancy (Fig 2C and S3B). However, similar to other R2lox structures, no clear electron density allowed us to model a hydroxy group in position C<sub>12</sub>.

### SaR2loxII is mainly ligand-free and lacks the tyrosine–valine ether cross-link

Recombinant SaR2loxII was produced and purified from *E. coli* cultured in rich medium. The crystal structure of SaR2loxII was solved to 2.26 Å resolution (Table S1). The protein displays the ferritin-like fold and ligand-binding pocket conserved among R2lox. Neither the first 19 residues of the protein nor residues 126 to 130 were visible in the electron density. In contrast, the C-terminus was modelled until the last residue. Besides, although the first coordination sphere residues of the metal centre are conserved, Glu210 exhibits an unusual outward conformation and the protein appears to be mainly metal-free, as the Mn (III) and Fe(III) ions were both refined to an occupancy of 0.35 (Fig. 2D). Furthermore, the protein does not show a tyrosine–valine ether cross-link, as Tyr170 displays an uncommon conformation, facing away from Val78 (Fig. 2E). Another atypical feature of SaR2loxII is that no clear electron density could

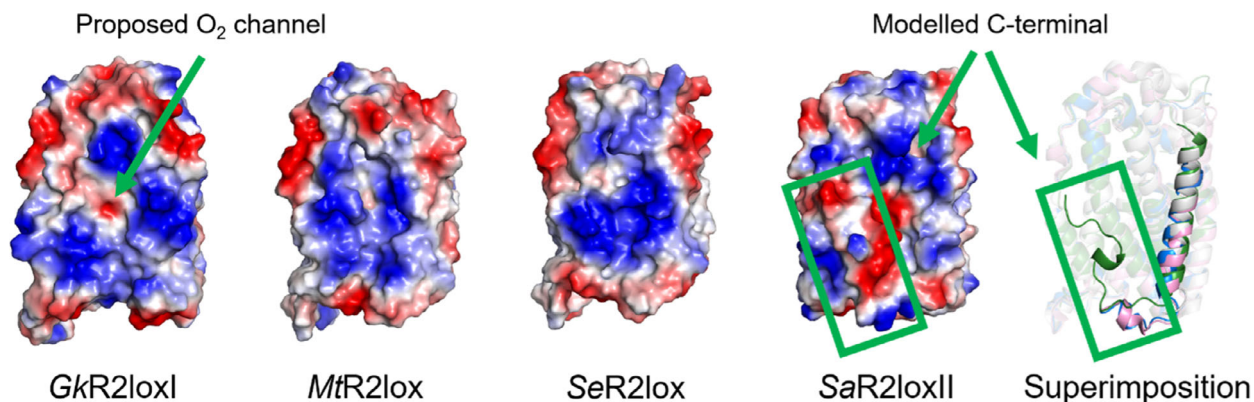
support the presence of a fatty acid ligand at high occupancy in the protein binding pocket (Fig. 2F). In similar expression conditions in *E. coli* cultured in rich media, MtR2lox, GkR2loxI and SaR2loxI are copurified with putative fatty acid ligands [8,14,34].

### Ordered C-terminus in SaR2loxII modifies the surface electrostatics of the protein

Two apparently recurrent features of R2lox crystal structures are the disordered C-terminal tail and a positively charged patch on the enzyme's surface proposed to be an interface for interaction with either a protein partner or with the cell membrane [20,35]. In the structure of SaR2loxII, the C-terminus forms a well-defined coil which covers the positive patch (Fig. 3). In addition, the outward conformation of the conserved metal-binding residue Glu210 (Fig. 2D) extends towards the surface of the protein and contributes to further disrupting this positive patch.

### The closing loop of the ligand-binding tunnel conserves its flexibility

Griese *et al.* (2015) proposed that a flexible loop gates the entrance of the ligand pocket in GkR2loxI [20]. By superimposing the structures, we can see that the configuration of this loop, located between helices G and H,



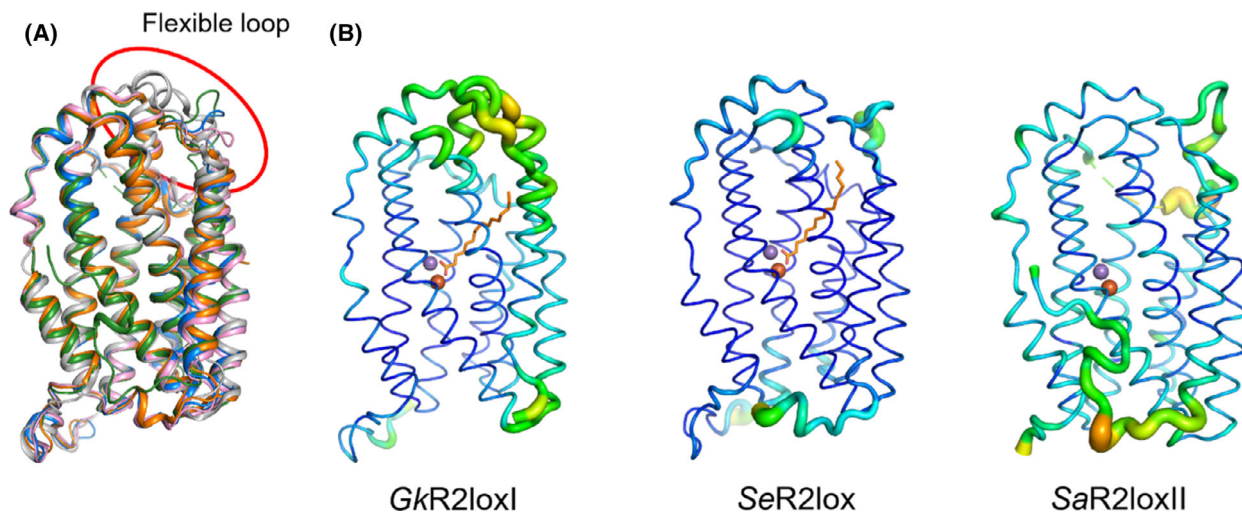
**Fig. 3.** The positive patch of *SaR2loxII* is disrupted by the ordered C-terminus. A comparison of the electrostatic surface (positive in blue and negative in red) of *GkR2loxI*, *MtR2lox*, *SeR2lox* and *SaR2loxII* and a superimposition of the four structures highlighting their modelled C-terminus (PDB ID: [4HR0](#), [3EE4](#), [7QBP](#) and [7QBK](#), respectively). *GkR2loxI* is coloured in white, *MtR2lox* in pink, *SeR2lox* in blue and *SaR2loxII* in green. The C-terminus in the *SaR2loxII* structure shields the positive electrostatic patch hypothesized to be an interaction surface with protein partners or the cell membrane.

varies between the different R2lox proteins, compared with the rest of the scaffold (Fig. 4A). However, a common aspect is that this loop seems to display a high flexibility in the previously solved structures of *GkR2loxI*, *MtR2lox* and *SaR2loxI* as well as in our new structures of *SeR2lox* and *SaR2loxII*. Indeed, this loop always appears among the most flexible segments of the protein, as is evident when representing the atomic B factors of the structures (Fig. 4B). Interestingly, a motif of three sequential proline residues is found on this gating loop in the structures of *SaR2loxI* (Pro257, Pro262 and Pro267), *SaR2loxII* (Pro257, Pro263 and Pro268) and

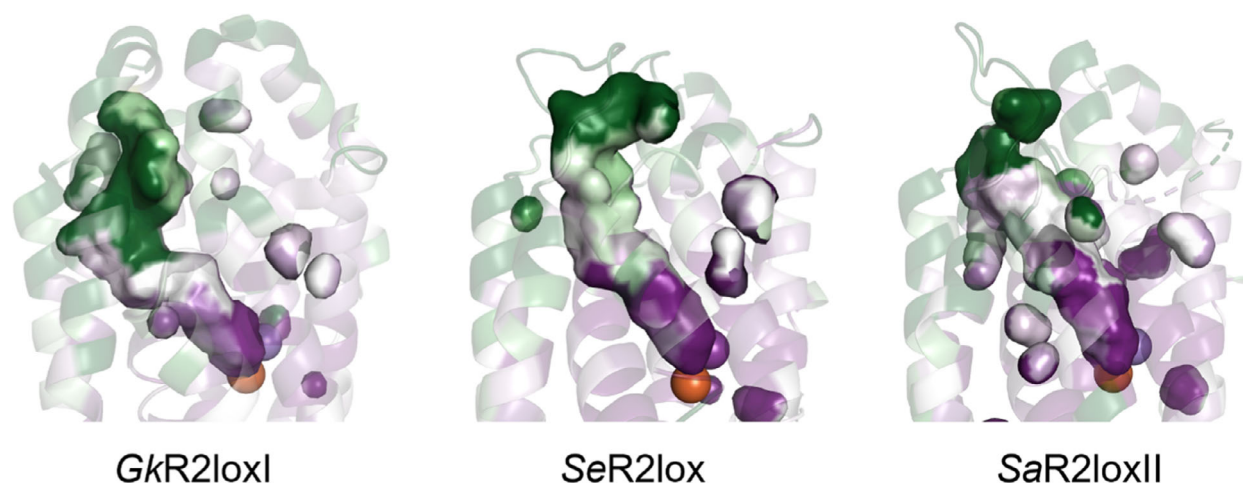
*SeR2lox* (Pro251, Pro255 and Pro259). Similarly, *MtR2lox* also exhibits a pair of proline residues (Pro253 and Pro254). However, no proline residues are observed on this loop in *GkR2loxI*, probably leading to a difference in flexibility as prolines are known to introduce rigidity in flexible regions [36].

#### R2lox ligand-binding pockets are highly diverse within the orthologues

A comparison of the R2lox ligand-binding pockets reveals a large diversity in length and shape across the



**Fig. 4.** The closing loop of the ligand-binding tunnel is highly flexible. (A) Superimposition of *GkR2loxI* in white, *MtR2lox* in pink, *SaR2loxI* in orange, *SeR2lox* in blue and *SaR2loxII* in green (PDB ID: [4HR0](#), [3EE4](#), [6QRZ](#), [7QBP](#) and [7QBK](#), respectively). The highlighted portion of the structure shows the highly flexible loop covering the entrance to the fatty acid ligand channel. (B) B factor representation of *GkR2loxI*, *SeR2lox* and *SaR2loxII*. B factor values are depicted by the thickness of the ribbon and by the colour range from blue to orange. Structures in (A) and (B) are shown with identical orientation.



**Fig. 5.** The ligand pocket of R2lox with conservation scores for each residue. The highly conserved residues of the binding pocket of *GkR2loxI*, *SeR2lox* and *SaR2loxII* are shown in dark purple, and the not conserved residues are shown in dark green (PDB ID: 4HR0, 7QBP and 7QBK, respectively).

orthologues (Fig. 5). Furthermore, we compared the evolutionary conservation of each residue of the ligand-binding pocket with the ConSurf server (<https://consurf.tau.ac.il>) [30,31]. Our results show that residues are highly conserved in proximity to the metal cofactor (dark purple in Fig. 5), but the remaining length of the pocket, especially in the vicinity of the entrance, is highly variable (dark green in Fig. 5).

## Discussion

In this study, we have shown by mass spectrometry that *GkR2loxI* and *SaR2loxI* copurify with a similar mixture of HFAs, mainly  $C_{16}$  and  $C_{18}$ . The fact that we did not observe the formation of a product after subjecting *GkR2loxI* to redox cycles might indicate that the fatty acids bound are not substrates of the enzyme. However, it seems that these ligands mimic the binding of R2lox substrates and thus are trapped in the binding pocket of recombinant proteins from the heterologous expression host *E. coli* where  $C_{16}$  and  $C_{18}$  fatty acids are abundant [37,38]. A similar feature can be observed in the ferritin-like enzyme ADO which also copurifies with a mixture of  $C_{16}$  and  $C_{18}$  fatty acids that are not substrates of the enzyme [39]. Instead, ADO catalyses the conversion of  $C_n$  fatty aldehydes into the corresponding  $C_{n-1}$  alkanes or alkenes [5] for a large range of chain length, including  $C_4$  to  $C_{18}$  [39]. Unlike ADO, fatty aldehydes do not seem to be the substrates of R2lox because *MtR2lox* and *SeR2lox* do not show ADO activity as reported by a recent study (proteins #65 and #75 in this study, respectively) [40]. In addition, fatty acids copurifying

with ADO are not hydroxylated, and thereby the hydroxy group found in R2lox ligands might be physiologically relevant. Nonetheless, the variety of copurified HFAs evidenced by mass spectrometry in our study suggests a putative promiscuous substrate specificity of R2lox, possibly comparable to ADO. Besides, the diversity of binding pocket shapes, previously discussed by Xu *et al.* for a smaller subset of structures [21], together with the low residue conservation in its distal part might indicate distinct substrate specificity between R2lox homologs.

The structure of *SaR2loxII* solved in this study exhibits several unusual features for R2lox that could shed some light on the enzymatic mechanisms of R2lox. First, we observe outward conformations for Tyr170, canonically forming an ether cross-link with Val78 in R2loxes, as well as for Glu210 which usually coordinates the metal centre. These characteristics cannot be fully explained by the low occupancy of the metal ions, as metal-free *GkR2loxI* does not display such outward conformations [14]. Instead, it could represent a mechanism regulating the access of  $O_2$  and/or metal ions allowing the correct maturation of the cofactor under specific conditions. Moreover, the fact that *SaR2loxII* did not trap a fatty acid ligand unlike *SaR2loxI* although both recombinant proteins were produced under similar conditions, that is, in *E. coli* cultured in terrific broth media [34], suggests that paralogous R2loxes might have different substrate specificity and thus could be active under different environmental conditions.

*SaR2loxII* displays an ordered C-terminus shielding a conserved positively charged patch on the protein surface that could potentially be involved in



protein–protein or protein–membrane interactions. A comparable feature was previously observed in only one particular structure of *MtR2lox* (PDB ID: 4AC8), although here the C-terminal segment forms a long  $\alpha$  helix [35]. Even if we cannot exclude a crystallographic artefact, the apparent dynamic behaviour between an ordered and disordered state of the R2lox C-terminal tail likely plays a role in the physiological function of the enzyme, as previously suggested [35]. A parallel can be drawn with the phylogenetically related RNR R2 subunit, in which the C-terminus is disordered in crystal structures of R2 alone, but it is crucial for interaction with its partner R1 [41]. Similarly, we can hypothesize that the C-terminal tail of R2lox might modulate the interaction with a putative partner or the cell membrane, for example, for the delivery of long-chain aliphatic compounds that are poorly water-soluble. For comparison, ADO interacts with an acyl–acyl carrier protein reductase producing fatty aldehydes which are transferred to ADO via a putative channel between the two proteins in order to efficiently synthesize alkanes [42]. In R2lox, such hypothetical transfer could occur through reorganization of the conserved flexible loop closing the substrate-binding tunnel. Although a pH contribution on the flexibility and conformation of the loop from the different crystallization conditions cannot be ruled out, a comparison of the R2lox structures crystallized at similar pH suggests that pH is not the determining factor for the differences in loop conformation. The fact that this loop structure shows the largest variability between structures hints that it is involved in substrate recognition and/or gating.

Interestingly, genomic studies of haloarchaea revealed that R2lox is often coded in a cluster of genes putatively encoding ferredoxin, sterol carrier protein and sterol-binding protein in these organisms [43]. Notably, ADO activity is supported by a ferredoxin–ferredoxin reductase system to provide electrons required for catalysis [44]. Besides, it was also shown that gene coding for R2lox is up-regulated in the hydrocarbon-degrading haloarchaeon *Halorientalis hydrocarbonoclasticus* when cultured in the presence of hexadecane, thereby suggesting a role for R2lox in the biodegradation of this C<sub>16</sub> alkane [43]. According to our present work, a substrate such as hexadecane, or a related compound, is conceivable as the length is compatible with our mass spectrometry and crystallography data.

Still, R2lox's function and substrate remain to be determined. Despite copurifying with C<sub>16</sub> and C<sub>18</sub> HFAs, we believe that these ligands occupy the binding pocket by being trapped from the heterologous

expression host organism and could be misleading in the search for R2lox's actual substrate. One of the biggest challenges in the discovery of R2lox's function is the identification of its physiological substrate. The best way to achieve this goal might be expressing R2lox in their native organisms. The subsequent *in vitro* activation of the enzyme and identification of the product should finally become a more achievable task. The substrate identification might also be tackled by finding a putative partner of R2lox. Finally, the physiological proximity of R2lox to the lipid membrane should be tested since the positive patch on the enzyme surface might indicate the interaction with the negative membrane surface.

## Acknowledgements

We thank the staff from the beamline X06SA at the Swiss Light Source, Paul Scherrer Institut, Villigen, Switzerland, and beamline 14.1 of BESSY, Berlin, Germany, as well as members of Martin Högbom's and Pål Stenmark's groups for their help with data collection. We acknowledge financial support from the Knut and Alice Wallenberg Foundation (2017.0275 and 2019.0436), the Swedish Research Council (2017-04018) and the European Research Council (HIGH-GEAR 724394). The funding sources had no involvement in study design; in the collection, analysis and interpretation of data; in the writing of the report; and in the decision to submit the article for publication.

## Conflict of interest

The authors declare no conflict of interest.

## Author contributions

RD contributed to protein production, mass spectrometry, bioinformatics and structural analysis and manuscript writing; VS contributed to protein production and crystallization, data processing and analysis; AJ contributed to mass spectrometry and data analysis; AN contributed to mass spectrometry and data analysis; JJG contributed to cloning; HL contributed to protein production and crystallization, data processing and analysis, and manuscript writing; MH contributed to project design, data analysis and manuscript writing.

## Data accessibility

The atomic coordinates of the *SeR2lox* and *SaR2loxII* structures are deposited in the Protein Data Bank

under accession codes [7QBP](#) and [7QBK](#), respectively. Other raw data are available upon request.

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## Supporting information

Additional supporting information may be found online in the Supporting Information section at the end of the article.

**Fig. S1.** Extracted ion chromatograms (EICs) for *GkR2loxI* and *SaR2loxI* showing the presence of the same ligands in both proteins.

**Fig. S2.** Extracted ion chromatograms (EICs) of the ligands identified in *GkR2loxI* after reduction-oxidation cycle.

**Fig. S3.** Metal center in *SeR2lox* and ligands modeled in R2lox proteins.

**Table S1.** Data collection and refinement statistics.