


Pharmacological Activities and In-Silico Studies of Bioactive Compounds Identified in Organic Fractions of the Methanolic Extract of *Citrullus Colocynthis*

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Abstract

Medicinal plants have been extensively exploited for their immense pharmacological and immune-supporting potential. Fruit of *Citrullus colocynthis* has several active secondary metabolites such as phenolics, flavonoids, and essential oils that are used in traditional medicines as antidiabetic, anti-inflammatory, antioxidant, and antimicrobial agents. In this study, phytoconstituents in organic fractions (*n*-hexane, chloroform, and ethyl acetate) of the methanolic extract of *C. colocynthis* were analyzed and identified by FT-IR, HPLC, and GC-MS analysis. Ethyl acetate fraction showed the highest antioxidant scavenging ($76 \pm .769\%$) and anti-inflammatory ($40 \pm .473\%$) activities at the concentration of 3 mg/mL. Similarly, antidiabetic effect was measured by inhibition of α -amylase where, ethyl acetate fraction ($77 \pm .844\%$) exhibited the highest antidiabetic activity. Among all organic fractions, ethyl acetate exhibited strong antimicrobial potential followed by *n*-hexane and chloroform fractions against selected pathogenic bacteria. Various concentrations of the ethyl acetate extract were tested *in-vivo* for cytotoxicity and results indicated minor morphological changes in liver cells including ballooning, fatty droplets, and slight accumulation of extracellular matrix even at concentrations of 400 mg/kg. *In-silico* study showed that stigmasta-7,16-dien-3-ol had a strong interaction with COX-1 and COX-2 to reduce inflammation. The abovementioned results indicate the pharmacological strengths of *C. colocynthis* to fight several diseases.

Keywords

Citrullus colocynthis, GC-MS, Antioxidant, Anti-inflammatory, Antimicrobial, Cytotoxicity

Introduction

Medicinal plants have been used to improve overall human health and are known to cure multiple ailments. The extracts of medicinal plants are used for the formulation of novel traditional medicines or food supplements.¹ Plant-based chemicals are further classified as primary and secondary metabolites. Primary metabolites consist of proteins, amino acids, purines, pyrimidines, and chlorophyll, whereas, secondary metabolites consist of several bioactive compounds with immense biological potential. *C. colocynthis* has been used for various ethnopharmacological purposes for centuries. A major compound from *C. colocynthis* named as cucurbitin or cucurbit, has shown effectiveness in a wide range of

diseases, including skin, gynecological, pulmonary infections, inflammatory conditions, cardiovascular problems, and immune-related illnesses.²

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Extracts of *C. colocynthis* contain phytochemicals such as phenolics, flavonoids, alkaloids, terpenes, and tannins that show good potential as antimicrobial, antidiabetic,^{3,4} and anticancer agents.⁵ The antioxidant properties of polyphenols enable them to scavenge free radicals and prevent their future formation. Free radical scavenging activities of *C. colocynthis* extract are broad-spectrum because of the large proportion of polyphenols as its phytoconstituent.⁶ All these secondary metabolites from medicinal plants can include phenols, alkaloids, terpenoids, acetogenins, and saponins.⁷ Bioactive compounds derived from *Citrullus* plants, categorized as alkaloids, flavonoids, and phenolic acids also show antimalarial, analgesic, and diuretic activities.^{5,8} Terpenoids in extracts of medicinal plants are known for their anthelmintic, antiviral, antibacterial, antimalarial, anti-inflammatory, and anticancer potential.^{9,10} Phenols and flavonoids exhibit antiallergic, antibacterial, and antioxidant activities. Saponins are reported to have antiviral, anti-inflammatory, and immune responses.⁸ Globally, people use these plants as primary source of medicine based upon their folkloric uses and to have protection against infectious diseases.¹¹ In addition to have remedial effects, these medicinal plants are able to scavenge harmful radicals such as peroxy radical from the body.¹²

Citrullus is commonly consumed by diabetic people. Diabetes is a common metabolic disorder leading to death of large number of people every year. Due to stress and insufficient amount of enzyme, glycation of protein leads to formation of reactive oxygen species (ROS).¹³ Although several synthetic drugs are used to cure diabetes, nonetheless, they are associated with certain limitations and side effects. Natural products are rich source of novel and effective antidiabetic compounds, which significantly cure diabetes and can serve as inhibitors of α -glucosidase and considerably lower blood glucose levels.¹⁴ *C. colocynthis* has been used extensively to cure diabetes in folk medicine and contains various biologically active compounds that cause hypoglycaemic effect by inhibiting activity of α -glucosidase.¹⁵ Phenolic compounds (galloyl epicatechin) derived from *C. colocynthis* exhibit antidiabetic potential with no side effects and are attributed as a very low-cost treatment of diabetes and related disorders.¹⁶

The aim of this research was to perform the detailed and in-depth phytochemical analysis and to explore the biological potential of bioactive compounds isolated from active fractions of methanolic extract of *C. colocynthis*. The study also aimed to evaluate the potentially active compounds with antimicrobial, antioxidant, antidiabetic, and cytotoxic effects *in-vitro*.

Materials and Methods

Collection of Plant Material

C. colocynthis plants were purchased from the local market of Lahore, Pakistan. Plants were cross-verified and approved from a taxonomist of the Department of Botany, University of Central Punjab, Lahore, Pakistan.

Preparation of Methanolic Extract

The *C. colocynthis* fruit (100 g) was washed with sterile water and shade dried. The dried fruit was ground to make a fine powder. Ten-grams of the powdered sample were soaked into 100 mL of methanol for 14 days at room temperature. The mixture was filtered and the residue was resuspended in 100 mL of methanol. The process was repeated three times and the filtrate was combined. The filtrate was subjected to concentration in a rotary evaporator (Rota Vapor R-200 Buchi, Switzerland) at pressure of 100 mbar and 40 °C temperature.¹⁷

Organic Extraction and Fractionation

The methanolic extract was fractionated by using organic solvents including *n*-hexane, chloroform, and ethyl acetate on polarity pattern. After fractionation, samples were concentrated to dryness using a rotary evaporator at a temperature 40–45 °C and stored in a refrigerator at 4 °C for further analysis.¹⁸

Qualitative and Quantitative Analysis

Phytoconstituents Analysis

Qualitative phytochemical profiling of *n*-hexane, chloroform, and ethyl acetate fractions of the methanolic extract of *C. colocynthis* was performed to detect the presence of alkaloids, saponins, tannins, flavonoids, terpenoids, glycosides, and quinones by following the methods as described before [19].

Fourier Transform Infrared (FT-IR) Spectroscopic Analysis

FT-IR analysis of *n*-hexane, chloroform, and ethyl acetate fractions of *C. colocynthis* was performed using Perkin Elmer spectrophotometer system. The characteristic peaks and functional groups at resolution of 4 cm⁻¹ within a peak a range of 500 to 4000 cm⁻¹ were found out.²⁰

Determination of Total Phenolic and Flavonoid Content

Folin-ciocalteu assay was used to determine the total phenolic content by using a UV-visible spectrophotometer at 760 nm, and the total flavonoid content of fractions was estimated by AlCl₃ at 415 nm.^{19,21}

Characterization of Phytochemical Components

High-Performance Liquid Chromatography Analysis of Phenolics and Flavonoids

The phenolics and flavonoids were analyzed using high-performance liquid chromatography (HPLC). The HPLC instrument equipped with a quaternary pump (model 1260 Agilent, USA)

was used in this analysis. A 20- μ L volume of the filtered sample was injected into a Zorbax Eclips Plus reverse phase (C18) column [4.6 \times 250 mm; 5 μ m particle size]. The mobile phase used contained methanol (solvent A) and 1% acetic acid (Solvent B).²² The run time of each sample was 60 minutes with a flow rate of 1 mL/min. Each sample was run in triplicate to ensure validity.

GC-MS Analysis

Volatile bioactive compounds were isolated and identified through GC-MS analysis on the basis of their molecular weight, and profiling of *C. colocynthis* fruit organic fractions was determined by GC-MS System²³ (GC-MS model: 7890B, 5977A, Agilent USA). Inlet temperature was kept at 280 °C and sample volume of 1 μ L was injected in the column (DB 5MS 30 cm, .25 mm, .25 μ m). Helium was used as a carrier gas and flow rate was maintained at 1 mL/min. The temperature of the oven was programmed as 50 °C for 1min, 25 °C to 120 °C for 5min, 10 °C to 160 °C for 0min, 6 °C to 240 °C for 0min, and finally 2 °C to 290 °C for 0min. Run time was maintained at 51.133min. Chromatograms were processed for identification of volatile compounds.

Biological Activities of Active Fractions

Total polyphenol and flavonoid contents in organic fractions of *C. colocynthis* were determined as mg of gallic acid equivalent/g or as mg/100g dry matter. The highest values of polyphenols were recorded in ethyl acetate fraction (718.02 \pm 1.89), while, minute concentrations of flavonoids were recorded in this fraction (28.654 \pm .303) mg/100gm. Previous studies have suggested the presence of polyphenols in the methanolic extract of *C. colocynthis* other than the fruit part. A study has shown polyphenols as 740 mg of gallic acid/100g and flavonoids as 130 mg/100g of catechin in *C. colocynthis* extract.²⁴

Antioxidant Free Radical Scavenging Activity by DPPH Assay

The DPPH assay of organic fractions of *C. colocynthis* was measured by following the previously reported methods.^{21,25} For the reaction, .001 g of DPPH was dissolved into 50 mL of methanol. Next, 1 mL of DPPH solution was added in each fraction of prepared concentrations (.5, 1.0, 1.5, 2.0, 2.5, and 3.0 mg/mL) of the plant extract. Samples were incubated at 28 °C for 30 minutes and observed for color change from purple into yellowish, showing the scavenging of free radicals. Absorbance was measured at 517 nm by using a spectrophotometer. Scavenging activity was expressed as the percentage inhibition using the following formula

$$\begin{aligned} & \% \text{ Radical scavenging activity} \\ & = \left(\frac{\text{Abs of blank} - \text{Abs of sample}}{\text{Abs of blank}} \right) \times 100 \end{aligned}$$

Scavenging Activity of Hydroxyl Radical

Scavenging activity of hydroxyl radical was performed by following the previously published method.²⁶ From each fraction (*n*-hexane, ethyl acetate, and chloroform) of *C. colocynthis*, concentrations ranging from .5, 1.0, 1.5, 2.0, 2.5, and 3.0 mg/mL, were separately added in a test tube. To it, .5 mL of DMSO (80%), .5 mL of iron-EDTA, .25 mL of EDTA solution, and .05 molL⁻¹ phosphate buffer of pH 7.4, were added in each test tube. The test tubes were kept in a water bath for 10 min at 85 °C. The reaction was stopped by adding .5 mL of cold tri-chloroacetic acid (TCA). Finally, .5 mL of Nash's reagent was added in each test tube and incubated for 10 minutes at room temperature. Absorbance was taken at 517 nm using a spectrophotometer. By using the following formula, % age inhibition was determined as:

$$\begin{aligned} & \% \text{ Radical Scavenging activity} \\ & = \left(\frac{\text{Abs of blank} - \text{Abs of sample}}{\text{Abs of blank}} \right) \times 100 \end{aligned}$$

Anti-inflammatory Activity

In-vitro anti-inflammatory activity was examined by following the previously described method.²⁷ Concentrations comprising 0.5, 1.0, 1.5, 2.0, 2.5, and 3.0 mg/mL were tested against diclofenac sodium as a standard drug. The absorbances were measured at 660 nm by using a UV-visible spectrophotometer. The experiment was repeated 3 times and results were compared with the standard drug. The % age of inhibition was found by using the following formula:

$$\% \text{ Inhibition} = \frac{(\text{Abs of control} - \text{Abs of sample})}{\text{Abs of control}} \times 100$$

Antimicrobial Activity

Antimicrobial activity was performed by disc-diffusion method.²⁸ Each fraction was tested individually against five bacterial pathogens including *Bacillus subtilis*, *Enterococcus faecalis*, *Staphylococcus aureus*, *E. coli*, and *Klebsiella pneumoniae*, obtained from the Microbial Culture Bank of University of Central Punjab. Overnight grown fresh bacterial cultures were spread on nutrient agar plates. The discs (6 mm) coated with specific concentration were placed on an agar plate containing test pathogens. Discs coated with DMSO were used as a negative control. Ampicillin and Ciprofloxacin were taken as standard drug controls. Plates were incubated at 37 °C for 24 h. Antibacterial activity was assessed by measuring the zone of inhibition by comparing with standard.

Minimum Inhibitory Concentration

C. colocynthis fractions (ethyl acetate and chloroform) were assessed for minimum inhibitory effect against selected pathogens by microdilution method using resazurin-based assay.²⁹ In a microtitre plate, 100 μ L of nutrient broth was added in each well. Following this, 100 μ L of 250 mg/mL ethyl acetate fraction was added in each first well of row A-D and two-fold diluted till column 10. In each well, 20 μ L of individual bacterial cultures was used as; rows A and E = *E. coli*, rows B and F = *Klebsiella pneumoniae*, rows C and G = *Staphylococcus aureus*, rows D and H = *Bacillus subtilis*. Ciprofloxacin (10 μ L/mL) was taken as positive control in column 12, whereas DMSO was taken as negative control in column 11. The plate was well wrapped with aluminum foil and incubated at 37 °C for 24 hrs. Finally, 20 μ L of resazurin was added in each well and re-incubated for 4 hrs for colorimetric indication. Change in color from purple to light blue was visualized, and the MIC of a compound was recorded.

Antidiabetic Activity

Antidiabetic activity was measured by α -amylase inhibition. Organic fractions of *C. colocynthis* (*n*-hexane, ethyl acetate, and chloroform) were dissolved in DMSO in concentrations ranging from 0.5, 1.0, 1.5, 2.0, 2.5, and 3.0 mg/mL. Ten- μ L of α -amylase solution was supplemented with 390 μ L of DMSO containing each individual concentration and kept at 37 °C in an incubator for 10 minutes. The mixture was re-incubated for one hour following mixing of 100 μ L of starch solution (1%). Following incubation, .1 mL of iodine solution (1%) was added. Absorbance was observed at 565 nm using a spectrophotometer and results were compared with standard of metformin.³⁰ The % inhibition was determined by the given formula:

$$\% \text{ Inhibition} = \left(\frac{\text{Abs of control} - \text{Abs of sample}}{\text{Abs of control}} \right) \times 100$$

Hemolytic Activity

Hemolytic activity was examined by the method described by Shahzadi et al. (2017).^{31,32} Three milliliter of blood sample was centrifuged at 4000 rpm for 5 minutes following the addition of chilled phosphate buffer saline (pH 7.4) for washing of blood cells. From each organic fraction [concentration = 0.5, 1.0, 1.5, 2.0, 2.5, and 3.0 mg/mL], 100 μ L was separately mixed with blood cells. Samples were incubated at 37°C for 30 minutes and chilled for 4–7 minutes and re-centrifuged for 5 minutes at 4000 rpm. Next, 100 μ L of supernatant from each test tube was used and diluted ten times using cold PBS. Triton X-100 (.1%) and PBS were taken as positive control and negative control, respectively. The absorbance was recorded at 576 nm using a spectrophotometer

(HALO DB-20 Dynamica). The % RBCs lysis for each sample was measured as below:

$$\% \text{ Lysis} = \left(\frac{\text{Abs of sample} - \text{Abs of DMSO}}{\text{Abs of Titron X} - 100 (\text{maximum lysis})} \right) \times 100$$

Thrombolytic Activity

As reported previously, for thrombolytic activity, pre-weighed micro-centrifuge tubes (we) were added with 0.5 mL of blood sample and incubated for 40 min at 37 °C to observe blood clotting³³⁻³⁵ Tubes containing clots (wm) were weighed again after removal of the serum to calculate weight of clot before lysis (wm-we). Previously prepared concentration of each organic fraction was separately added in tubes containing clots and incubated for 85 min at 37 °C for lysis of clot. Tubes were left overnight and weighted to measure the clot weight (wl) after lysis. Following lysis, difference of weight between after lysis (wl) and weight of empty tube (we) was recorded as the weight of the clot. Weight before lysis and weight after lysis were indicated (wb) and (wl), respectively. Streptokinase was taken as positive control, and distilled water was taken as negative control.

% Clot lysis was calculated by the given formula:

$$\% \text{ lysis} = \text{wb} - \text{wl} / \text{wb} \times 100$$

Cytotoxicity Assay

Hepatotoxicity assay was performed to check the toxic effects of bioactive compounds derived from *C. colocynthis*.³⁶ Swiss-albino male rats of approximately of the same age group, having weight of 110–130 g, were taken from the animal house of the University of Central Punjab, Lahore. Sixteen rats were divided in four groups as one control and three test groups. Different doses of each organic fraction were suspended in normal saline to prepare four doses of concentrations; 100 mg/kg, 200 mg/kg, 300 mg/kg, and 400 mg/kg according to animal body weight. One group was treated with *n*-hexane, second with ethyl acetate, and third group was treated with chloroform fraction. Group I: Control group was incorporated with 1 mL/kg normal saline as oral dose for 15 days; Group II, Group III, and Group IV were subjected to dose concentrations of 100 mg/kg, 200 mg/kg, 300 mg/kg, and 400 mg/kg of *n*-hexane extract, ethyl acetate, and chloroform, respectively. The dose was given orally to rats by gavage for 15 days (once a day). Blood samples of all rats were drawn, rats were slaughtered, and livers were used to examine toxicity.

Statistical Analysis

Statistical data was presented as means \pm SD, and one way ANOVA was applied to observe difference between biological activities of different experimental fractions of *C. colocynthis*.

Throughout the obtained results, * $P < .05$ and *** $P < .001$ were considered statistically significant and highly significant, respectively.

Molecular Docking

Phyto-compounds shown by GC-MS analysis were used as ligands for molecular docking analysis against COX-1 and COX-2 proteins. PubChem database was used to fetch 2D molecular structures of compounds which were converted to PDB file format using PyMOL software. The crystal structure of COX-1 (6Y3C) and COX-2 (SC-558) was retrieved from the RCSB, PDB database, whereas the (3D) structure of selected target proteins was presented by the AutoDock Vina tool.³⁷ Retrieved files were saved as Target.pdb., whereas Kollman partial charges and polar hydrogen atoms were inserted to the 3D structures. By using AutoDock Vina software 1.2.0, docking was performed.³⁸

Molecular Dynamics

The MD complexes were imposed to check any residues which may still be deformed or unstable after MD. Characterization of MD complexes such as deformability, β -factor, and co-variance analysis was performed using iMODS software.³⁹

ADMET Analysis

Assumption of the toxicity and pharmacokinetics of the compounds identified from *C. colocynthis* extract was carried out via SwissADME.^{40,41}

Results and Discussion

Phytochemical Screening

Phytochemicals such as alkaloids, steroids, terpenoids, saponins, and tannins were confirmed in all three organic fractions (*n*-hexane, ethyl acetate, and chloroform). Flavonoids were present in chloroform fraction, however, found missing in ethyl acetate and *n*-hexane fractions. Quinones were detected in ethyl acetate and *n*-hexane fractions, and were absent in chloroform fraction. The results are in line with the previously published literature, where extracts of *C. colocynthis* fruit have shown the presence of secondary metabolites including alkaloids, steroids, terpenoids, saponins, and tannins.²¹ Significant flavonoids in *C. colocynthis* include C-glycosides, quercitrin, iso-vitexin, iso-orientin, iso-orientin-3'-methylether, C-p-hydroxybenzyl derivatives, 8-C-p-hydroxybenzoylisovitexin, 6-C-p-hydroxybenzoylvitexin, 8-C-p-hydroxybenzoyl, and isovitexin-4'-O-glucoside have been reported from stem, fruit, and leaf extracts of the plant. These flavonoids have shown considerable antioxidant potential and can scavenge the toxic radicals.⁴²⁻⁴⁴ Screening of the extracts on gas chromatography revealed the presence of albumin, ascorbic acid, β -carotenes, vicilins, tocotrienols, lutins, 2-tetracyclobutane,

and 2-dodecyclobutane. In addition, presence of various amino acids such as arginine, leucine, isoleucine, and fatty acids including palmitic acid, linoleic acid, and linolenic acid was also observed in the extract. Previously reported literature on *C. colocynthis* has shown the presence of these compounds through chromatographic techniques and their biological potential as antidiabetic, hepatoprotective, and anti-tumor agents.⁴⁵ The results of GC-MS of *C. colocynthis* fractions are summarized in Table S1.

FT-IR Analysis

Characterization of phenolic constituents by spectroscopic analysis provides vital detail about the formulation of plant species and their motif identification. The UV-VIS spectroscopy is a simple technique used for identification and segregation of phytochemicals on the basis of polarity (lipophilic and hydrophilic molecules). Spectroscopic (FT-IR) analysis can thus be used for the analysis of phytoconstituents of the plant extracts.²⁰ For the structural elucidation and identification of biologically active compounds, FT-IR, which is a non-destructive analytical technique, was used offering a prompt analysis.⁴⁶ The results of FT-IR analysis revealed the presence of alkaloids (N-H stretching), polyphenols, flavonoids (O-H stretching), and terpenes (CH group).⁴⁷ The FT-IR spectrum verified the existence of alkanes, alkynes, phenols, alcohols, alkyl halides, aldehydes, carboxylic acids, and amines in *C. colocynthis*. When previously studied through FT-IR for functional groups analysis, *C. colocynthis* has shown the abovementioned compounds with broad-spectrum bioactive potential. The FT-IR analysis in this study served as the confirmational tool to re-evaluate the presence of these functional groups in organic fractions.^{43,48} The FT-IR spectrum profile of *n*-hexane, ethyl acetate, and chloroform fraction has been illustrated in (Figure 1, Table S2).

Total Phenolic and Flavonoid Content

Total phenolic and flavonoid content in *C. colocynthis* fractions is shown in Table 1. The highest concentration of phenolics was observed in ethyl acetate fraction (718.02 ± 1.89 mg/100 gm) followed by chloroform and *n*-hexane fractions. The highest quantity of flavonoids was estimated in ethyl acetate fraction (28.654 ± 0.303 mg/100 gm), whereas, least quantity was present in *n*-hexane fraction. Previous studies on *C. colocynthis* extracts have shown the presence of polyphenols and flavonoids which was also confirmed in this research.⁴³ A similar study was conducted on the extract of *C. colocynthis*, which showed $10.7 + .1$ mg/g of total flavonoid content.⁴⁹ The total amount of phenolic compounds present in the methanolic extract of *C. colocynthis* was found to be 58.76 mg/g. Also, the flavonoids content of *C. colocynthis* extract was determined and found to be 2.01 mg/g and .13 mg/g of rutin equivalent, respectively.⁴⁵

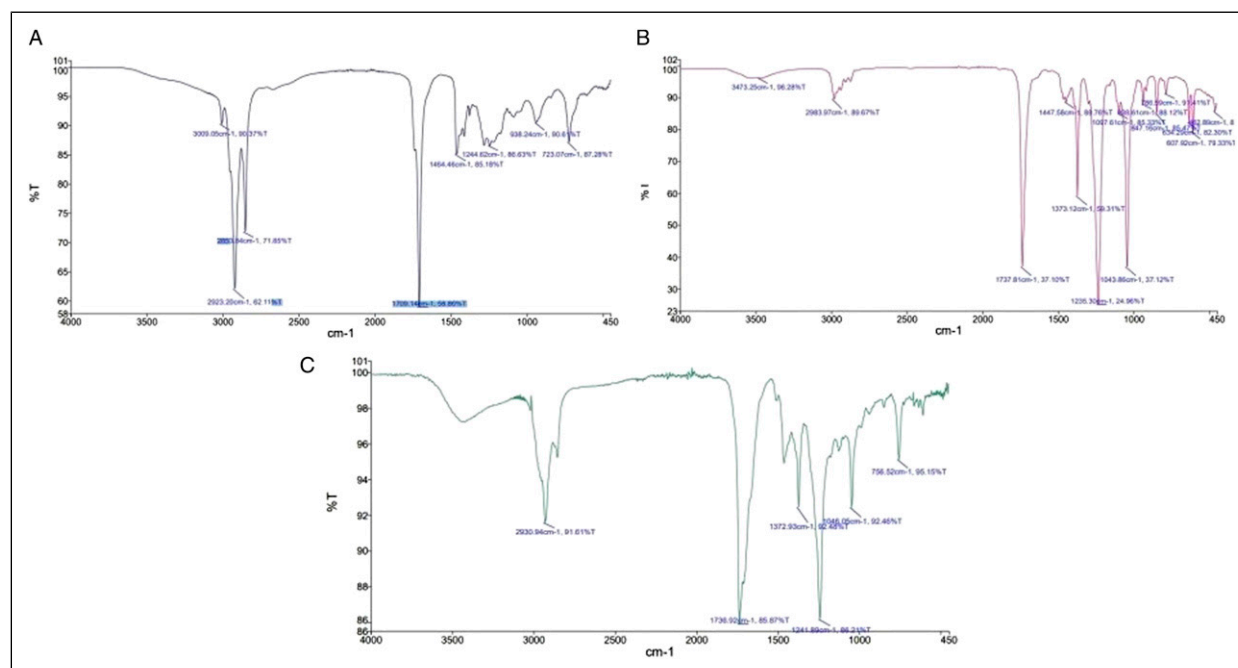


Figure 1. FT-IR Spectrum of methanolic fractions of *C. colocynthis*.

Table 1. Total polyphenol and flavonoid content.

Fractions	Mean \pm S.D TFC as CE mg/100 gm)	Mean \pm S.D TPC as CE mg/100 gm)
<i>n</i> -Hexane	15.142 \pm 0.148	146 \pm 1.89
Ethyl acetate	28.654 \pm 0.303	718.02 \pm 1.89
Chloroform	17.637 \pm 0.084	186.24 \pm 0.145

HPLC Analysis

A detailed phytochemical characterization of the organic fractions (*n*-hexane, chloroform, and ethyl acetate) of the methanolic extracts of *C. colocynthis* fruit confirmed the existence of phenolic and flavonoid metabolites such as quercetin, gallic acid, vanillic acid, chlorogenic acid, syringic acid, *m*-coumaric acid, *p*-coumaric acid, ferulic acid, benzoic acid, cinnamic acid, and sinapic acid. These compounds have been previously explored in *C. colocynthis* fruit extract and were demonstrated for biological potential in previous investigations.^{50,51} Dominant compounds observed through HPLC analysis were caffeic acid, vanillic acid, and benzoic acid. The highest concentration of vanillic acid (173.3708 ppm), caffeic acid (32.3764 ppm), and benzoic acid (129.5395 ppm) was observed in *n*-hexane fraction followed by ethyl acetate and chloroform fractions, respectively (Figure 2, Table S3). One of the recent studies on *C. colocynthis* in Alexandria showed the presence of benzoic acid, rhamnetin, isoquercetin, hydroxycaffeic acids, and caffeic acid in several organic fractions, however, the highest quantities of caffeic acid (32.3764 ppm) in this study are exceptional as compared to previous (10.3764 ppm), where its presence was shown to be minimal.^{52,53}

GC-MS Analysis

Outcome of GC-MS analysis showed the presence of valuable volatile biologically active compounds in all active fractions of *C. colocynthis*. Bioactive constituents identified in *n*-hexane fraction included 9,12-octadecadienoic acid (*Z, Z*)-methyl ester (20.78%) and linoleic acid ethyl ester (20.58%) which were in significantly higher concentration. Eleven different metabolites were quantified in ethyl acetate fraction, and among all constituents, linoleic acid ethyl ester (26.71%) and 9,12-octadecadienoic acid (*Z, Z*)-, methyl ester (26.63%) were significant. In chloroform fraction, linoleic acid ethyl ester (45.51%) and 9,12-octadecadienoic acid (*Z, Z*)-, methyl ester (21.73%) were noted in significant amounts (Figure 3 and Table S4).

Biological Activities of *C. colocynthis* Methanolic Extract

Antioxidant Activity

The free radical scavenging activity was performed by DPPH assay to quantify the percentage inhibitory effect of bioactive compounds. The quotients of antioxidant activity are shown in

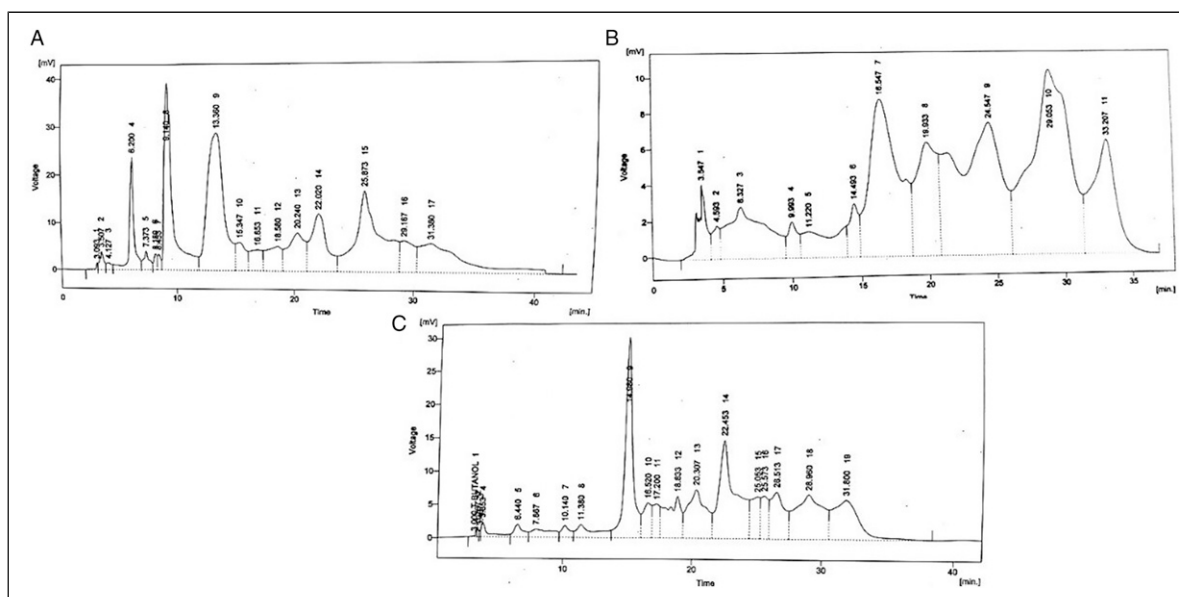


Figure 2. HPLC chromatogram methanolic fractions of *C. colocynthis*.

Table 2. Results suggested that ethyl acetate fraction gave the highest DPPH scavenging activity among all the tested fractions with the value of $76 \pm 0.769\%$ at concentration of 3 mg/mL. It was observed that by increasing the concentration from 0.5 mg/mL to 3 mg/mL, DPPH scavenging activity was significantly increased. IC_{50} values of *n*-hexane, ethyl acetate, and chloroform fractions were 2.29 mg/mL, 1.72 mg/mL, and 2.06 mg/mL, respectively. Previous studies have shown that *C. colocynthis* possesses antioxidant potential and related biological activities. DPPH scavenging activity of the organic leaf extract of *C. colocynthis* has been reported as $88.0 \pm 2.7\%$ at concentration of 250 mg/mL.²⁶ DPPH analysis of some other medicinal plants has demonstrated the highest IC_{50} values of *R. stricta* and *M. crassifolia* as 0.335 and 0.448 mg/mL, respectively.⁵⁴ A recent analysis of *C. colocynthis* showed that the ethanolic extract of roots and leaves had high free radical inhibition activity which was 36.52 and 33.83%, respectively.²¹ Nonetheless, in this research study, the values of antioxidant activity of all three organic fractions were considerably high as compared to previously reported ones.

Hydrogen Peroxide Activity

Free radicals (H_2O_2) scavenging assay was performed by using various concentrations of methanolic fractions of *C. colocynthis* as described in Table 2. The ethyl acetate fraction showed strong scavenging activity ($72 \pm 0.612\%$) at the concentration of 3.0 mg/mL, whereas chloroform and *n*-hexane fraction showed $62 \pm 0.634\%$ and $59 \pm .486\%$ scavenging activities, respectively. The significant difference in percentage inhibition of H_2O_2 of all fractions was noted. H_2O_2 is produced in the human body in various metabolic

processes and may also exist in air and water vapors in response to redox reactions. Free radicals of H_2O_2 decompose into reactive oxygen species (ROS) which initiate the lipid peroxidation and consequently desecration of DNA. The ethyl acetate fraction of *C. colocynthis* was capable to scavenge hydrogen peroxide in the presence of phenolic compounds which donates the electrons to the peroxides and neutralize it into water.⁵⁵

Anti-inflammatory Activity

Results of denaturation of BSA have been summarized in Table 3. *In-vitro* anti-inflammatory activity was determined by comparing the results with standard diclofenac sodium to check biopotential of secondary metabolites found in fractions of *C. colocynthis*. *n*-hexane, ethyl acetate, and chloroform fractions of *C. colocynthis* with various concentrations in the range of 0.5 mg/mL–3 mg/mL were used. The maximum inhibitory activity was recorded at a concentration of 3 mg/mL of ethyl acetate fraction ($40 \pm 0.473\%$), followed by chloroform ($39 \pm 0.336\%$), and *n*-hexane fraction ($37 \pm 0.413\%$). As reported by a previous study, the anti-inflammatory potential of phenolic compounds such as vanillic acid and quercetin in *C. colocynthis* was explored. The data showed that compounds were involved in cure, and demonstrated anti-inflammatory, antimicrobial, antioxidant, and antidiabetic activities.⁵⁶ Major cause of auto-antigens in several arthritic disorders might be due to protein denaturation.⁵⁷ Denaturation of proteins are known to be inhibited by anti-inflammatory antidotes which have been significantly found in organic extracts of *C. colocynthis*. Moreover, besides showing anti-inflammatory activities, some of these compounds additionally possess

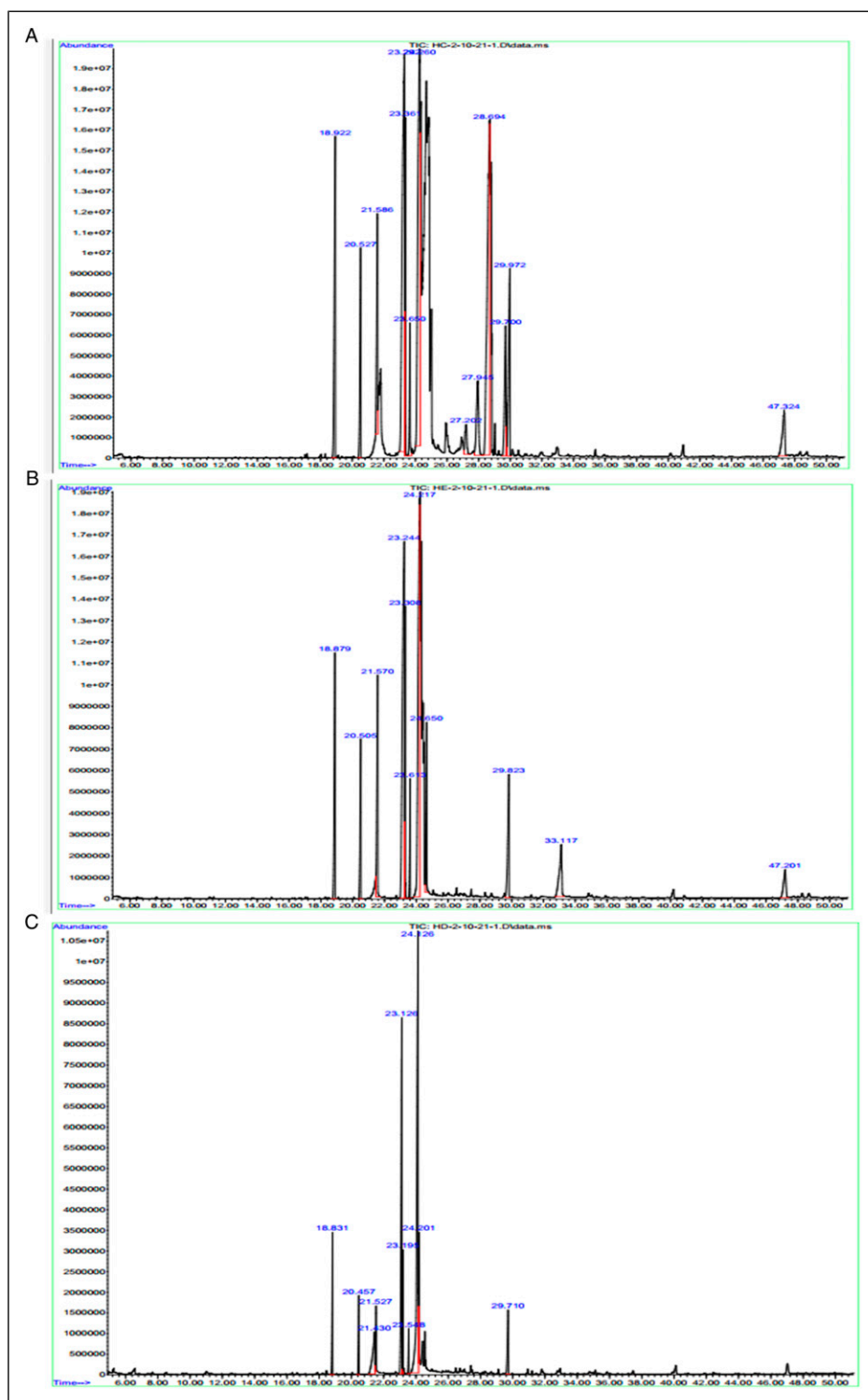


Figure 3. GC-MS chromatogram methanolic fractions of *C. colocynthis*.

antimicrobial potential. Increasing antibiotic resistance and because of evolving pathogens, existing antimicrobials are losing their efficacy. Hence, several therapeutic compounds from medicinal plants have been tested as an alternative source of antimicrobials for pathogenic microbes because of

low side-effects and cost-effectiveness.⁵⁸ *C. colocynthis*, a medicinal plant, simultaneously possesses therapeutic, antimicrobial, and anti-inflammatory potential and can be a preferred choice to fight microbial and inflammatory diseases.¹¹

Table 2. Antioxidant activity of methanolic fractions of *C. colocynthis* by DPPH and H₂O₂ assay.

Fractions	Conc. mg/mL	DPPH Scavenging %	H ₂ O ₂ Scavenging %	IC ₅₀ (mg/mL)	
				DPPH	H ₂ O ₂
<i>n</i> -Hexane	0.5	11 ± 0.050	9 ± 0.103	2.22	2.40
	1	24 ± 0.234	21 ± 0.275		
	1.5	32 ± 0.367	27 ± 0.219		
	2	45 ± 0.388	42 ± 0.465		
	2.5	54 ± 0.514	53 ± 0.514		
	3	64 ± 0.693	59 ± 0.486		
Ethyl acetate	0.5	22 ± 0.241	18 ± 0.168	1.86	2.03
	1	29 ± 0.218	27 ± 0.242		
	1.5	41 ± 0.366	40 ± 0.345		
	2	54 ± 0.485	49 ± 0.485		
	2.5	63 ± 0.571	57 ± 0.529		
	3	76 ± 0.769	72 ± 0.612		
Chloroform	0.5	12 ± 0.109	11 ± 0.050	2.22	2.47
	1	27 ± 0.251	19 ± 0.101		
	1.5	39 ± 0.305	33 ± 0.301		
	2	46 ± 0.415	43 ± 0.540		
	2.5	55 ± 0.503	50 ± 0.514		
	3	65 ± 0.594	62 ± 0.634		
Ascorbic acid	1	91 ± 0.878	82 ± 0.792		

Table 3. Anti-inflammatory, antidiabetic, hemolytic, and thrombolytic effect of methanolic fractions *C. colocynthis* fruit.

Fractions	Conc. mg/mL	Anti-inflammatory% Inhibition	Antidiabetic % Inhibition	% Hemolysis	% Thrombolysis
<i>n</i> -Hexane	0.5	11 ± 0.112	8 ± 0.007	.41 ± 0.343	0.35
	1	12 ± 0.106	14 ± 0.137	0.82 ± 0.055	0.7
	1.5	22 ± 0.212	33 ± 0.326	1.16 ± 0.195	0.7
	2	26 ± 0.286	39 ± 0.321	1.68 ± 0.577	1.4
	2.5	31 ± 0.369	45 ± 0.419	2.24 ± 0.967	2.45
	3	37 ± 0.413	55 ± 0.528	2.26 ± 0.945	3.85
Ethyl acetate	0.5	13 ± 0.102	13 ± 0.137	0.34 ± .092	1.4
	1	13 ± 0.160	23 ± 0.220	1.05 ± 0.140	2.1
	1.5	21 ± 0.247	37 ± 0.329	1.80 ± 0.092	4.91
	2	27 ± 0.286	51 ± 0.547	2.40 ± 0.140	8.07
	2.5	32 ± 0.321	65 ± 0.526	2.92 ± 0.092	11.22
	3	40 ± .0473	77 ± 0.844	3.37 ± 0.159	13.68
Chloroform	0.5	9 ± 0.052	8 ± 0.134	0.37 ± 0.367	0.7
	1	13 ± 0.189	16 ± 0.086	0.52 ± 0.103	1.4
	1.5	21 ± 0.237	30 ± 0.313	0.99 ± 0.266	1.4
	2	26 ± 0.236	39 ± 0.359	1.51 ± 0.594	1.75
	2.5	31 ± 0.310	44 ± 0.423	1.97 ± 0.885	2.45
	3	39 ± 0.336	57 ± 0.573	2.23 ± 0.992	4.91
Diclofenac Na	1	84.8 ± 0.059	—	—	—
Metformin	1	—	91 ± 1	—	—
Triton X-100	1	—	—	89 ± 1	—
Streptokinase	1	—	—	—	86

Antidiabetic Activity

Results showed that methanolic fraction of *C. colocynthis* contained active phytochemicals, which were characterized by various spectroscopic analyses, and validated for antidiabetic potential through α -amylase assay. The highest inhibition of α -amylase was noted by ethyl acetate fraction

(77 ± 0.844%), followed by chloroform fraction (57 ± 0.573%), and *n*-hexane fraction (55 ± 0.528%). A previous experimental work done to evaluate antidiabetic potential by reduction of enzyme α -amylase and α -glucosidase showed significant antidiabetic potential of *C. colocynthis*.⁵⁹ Studies showed that an *in-vitro* α -glucosidase inhibition assay indicated marked inhibition of α -glucosidase

by *C. colocynthis* extract at 355.5 µg/mL concentration.¹⁵ Both of these enzymes play a significant role to increase the blood glucose levels, and lead to diabetes. Results of this study also confirm the antidiabetic activities of all organic fractions and demonstrated that the antidiabetic effect subsequently increased with the increase in the concentration. This study along with previously reported literature provides the scientific basis to use the whole fruit of *C. colocynthis* as a drug against hyperglycemia and dyslipidemia (Table 3).

Hemolytic Activity

At the highest concentration of 3 mg/mL, *n*-hexane, ethyl acetate, and chloroform fractions showed $2.26 \pm 0.945\%$, $3.37 \pm 0.159\%$, and $2.23 \pm 0.992\%$ hemolysis of blood cells, respectively. The reference Triton X-100 showed $89 \pm 1\%$ blood cell lysis. This data indicates that ethyl acetate fraction showed the significant blood cell lysis followed by *n*-hexane and chloroform fractions. A similar study was performed to test hemolytic potential of various medicinal plants and showed that the methanolic extract of *E. officinalis* exhibited least hemolytic activity ($5.89 \pm 0.78\%$), whereas *T. chebula* exhibited highest percent hemolysis ($13.12 \pm 1.2\%$).⁶⁰ However, the hemolytic activity of *C. colocynthis* extract is in safe range and does not affect its therapeutic potential (Table 3). The methanolic extract of *E. officinalis* exhibited the highest radical scavenging potential with percent inhibition of 90.11 ± 7.2 than the other tested methanolic extracts of selected medicinal plants in that study.⁶⁰ While the lowest percent inhibition of free radical production was shown by methanolic extract with an average \pm SD value of $63.23 \pm 5.8\%$.⁶⁰

Thrombolytic Activity

Highest thrombolytic activity was shown by ethyl acetate fraction with the values of 13.68% followed by chloroform (4.91%) and *n*-hexane (3.85%). The reference streptokinase showed 86% clot lysis at 30,000 IU concentration. This data indicates that ethyl acetate fraction showed the significant clot lysis demonstrating the potential of *C. colocynthis* fruit as a good choice as thrombolytic agent. A previous study on *in-vitro* thrombolytic activity of various medicinal plants using methanolic extracts of *Buchholzia coriacea* (27.062%) at a dose concentration of 100 µg/mL highlighted their thrombolytic potential; however, there are less reports on the thrombolytic potential of *C. colocynthis*.⁶¹ Thrombolysis potential of medicinal plants suggest that these plants can be employed for remedial purposes and have advantages over traditional therapeutic drugs. Besides thrombolysis potential, anticoagulant activity of plant extracts has also been evaluated by the PT, INR, and PTT assays, using normal human plasma and showed promising results complementing their therapeutic roles and can be a future research prospect for *C. colocynthis*.

Antimicrobial Activity

Antimicrobial effect of (*n*-hexane, ethyl acetate, and chloroform) fractions of *C. colocynthis* fruit against five microbial strains (*Enterococcus faecalis*, *Bacillus subtilis*, *Staphylococcus aureus*, *Escherichia coli*, and *Klebsiella pneumoniae*) was observed. Ethyl acetate fraction showed the highest zone of inhibition against *K. pneumoniae* ($19.63 \pm .32$ mm). The other fractions (*n*-hexane and chloroform) exhibited comparably less antimicrobial potential, whereas *Enterococcus faecalis* showed resistance against *n*-hexane fraction. *K. pneumoniae* showed resistance against chloroform fraction. The results have been illustrated in Figure S1 and Table S7. Organic fractions of *C. colocynthis* fruits have shown significant antimicrobial effects in previous research studies, and hence, the said plant can be used against several disease-causing microbes. Previously, the presence of phenolics such as quercetin, myricetin, glycosides, and terpenoids has been shown in several medicinal plants by biochemical profiling of leaves.⁶² Hence, these phenolic compounds can be the potential reason for showing antimicrobial activities. Similar experimental work was performed using *C. colocynthis* organic fractions to assess the antimicrobial activity against different strains.⁶³ In that, *n*-hexane and ethyl acetate fraction showed its effect only against *B. cereus*, *L. monocytogenes*, and *S. aureus*, whereas butanol and methanol showed no activity against the bacterial strains. Maximum antimicrobial activity was observed using chloroform fraction which was selected to isolate the active compounds of *C. colocynthis*. Highest zone of inhibition against *A. fumigatus* of 11 mm (at 25 mg/1 mL DMSO) and 21 mm (at 100 mg/1 mL DMSO) was observed. The highest zone of inhibition of seed extract of *C. colocynthis* against *B. subtilis* was noted as 11 mm (at 25 mg/1 mL DMSO) and 22 mm (at 100 mg/1 mL DMSO). Highest zone of inhibition against *F. oxysporum* was 15 mm (at 25 mg/1 mL DMSO) and 21 mm (at 100 mg/1 mL DMSO). Both of strains showed greater zones of inhibition by increasing the concentrations of the extracts. Gram-negative bacteria *Pseudomonas aeruginosa* showed sensitivity against the ethanolic extract of *C. colocynthis* contrary.⁶⁴

Minimum Inhibitory Concentration

MIC was assessed against two Gram positive (*S. aureus* and *B. subtilis*) and two Gram-negative bacteria (*E. coli* and *K. pneumoniae*) using ethyl acetate and chloroform fraction of the methanolic extract. MIC of ethyl acetate fraction was observed as 125 µg/mL, 62.5 µg/mL, 31.25 µg/mL, and 31.25 µg/mL against *E. coli*, *K. pneumoniae*, *S. aureus*, and *B. subtilis*, respectively. Similarly, MIC values of chloroform fraction of the methanolic extract were assessed as 250 µg/mL, 125 µg/mL, and 125 µg/mL for *E. coli*, *S. aureus* and *B. subtilis*, respectively. *K. pneumoniae* indicated resistance against chloroform fraction. Ciprofloxacin was taken as positive control that exhibits MIC against all tested

pathogens except *K. pneumoniae*. According to previous studies, the extract of *C. colocynthis* fruits exhibited .10 mg/mL MIC against *Candida albicans*, .20 mg/mL MIC against *E. coli*, and .23 mg/mL MIC against *P. aeruginosa*.⁶³ Similarly, another research conducted using *C. colocynthis* extract showed MIC of less than 2 mg/mL against *B. cereus* and *B. subtilis* which the lowest MIC values (.5 mg/mL and 2.0 mg/mL, followed by *Micrococcus* spp. (1.5 mg/mL) and *S. aureus* (2.0 mg/mL), respectively.⁶⁵

Cytotoxicity Assay

Hepatotoxicity was performed to confirm the biopotential of natural compounds derived from *C. colocynthis*. It was observed that ethyl acetate fraction showed the promising profile of biologically active compounds among the selected fractions of the methanolic extract. Cytotoxic effect of secondary metabolites was observed by performing histopathological examination of liver tissues of albino rats, which showed minor changes in the liver tissues such as ballooning, fatty droplets, and accumulation of extracellular matrix at various concentration of ethyl acetate, proving that bioactive compounds in ethyl acetate fraction have no toxic effect. Furthermore, studies indicate that ethyl acetate fraction has good concentration of antioxidants which may impart hepatoprotective effect in reducing free radicals. So, it is inferred that all biologically active compounds found in ethyl acetate fractions of *C. colocynthis* are safe to be used (Figure 4). The ethanolic extract possesses a promising antioxidant power with an IC₅₀ value of 4.56 mg/mL. These results are highly justified by the presence of bioactive compounds in the plant extract characterized with a GC-MS, which showed the presence of

polyphenols and flavonoids as responsible agents for the antioxidant activity. Iso-orientin 3-O-methylether, iso-saponarin, and isovitexin isolated from *C. colocynthis* possess a strong antioxidant power with an IC₅₀ value ranging from 5.62×10^{-4} to 7.13×10^{-2} mg/mL.⁶⁶

Molecular Docking

Deformability, β -Factor, and Co-variance Computations

The major deformability was observed in the stigmasta-7,16-dien-3-ol, (3. beta. ,5. alpha) of COX-1 and COX-2 complex and had many peaks of approximate value 1.0 deformability index. The results of MD are showed in Figure 5 that demonstrates the normal mode analysis (NMA) of stigmasta-7,16-dien-3-ol, (3. beta. ,5. alpha) of COX-1 and COX-2 complex. The graph of deformability (5A) of complexes of COX-1 and COX-2 showed the peaks in the graph correspond to the deformability region of protein. Comparison of NMA and PDB field has been illustrated in the β -factor graph in Figure 5(B). Variance, co-variance, residue index, and atom index have been shown in Figures 5C-5F. Co-variance provided sufficient results at elastic network map.

ADMET Analysis

SwissADME analysis showed that stigmasta-7,16-dien-3-ol, (3. beta. ,5. alpha.) is the good anti-inflammatory compound. It had fair lipophilicity, Lop *P* greater than 4.94, and was slightly water soluble. Physicochemical properties showed that it had a molecular formula of C₂₉H₄₈O and molecular weight of

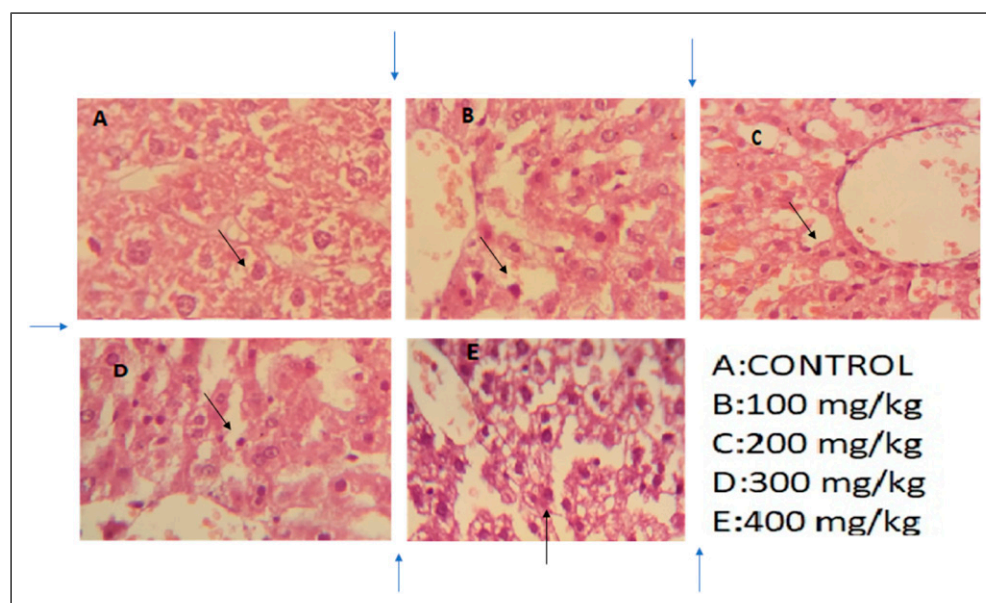


Figure 4. Hepatoprotective effect of ethyl acetate fractions of *C. colocynthis*.

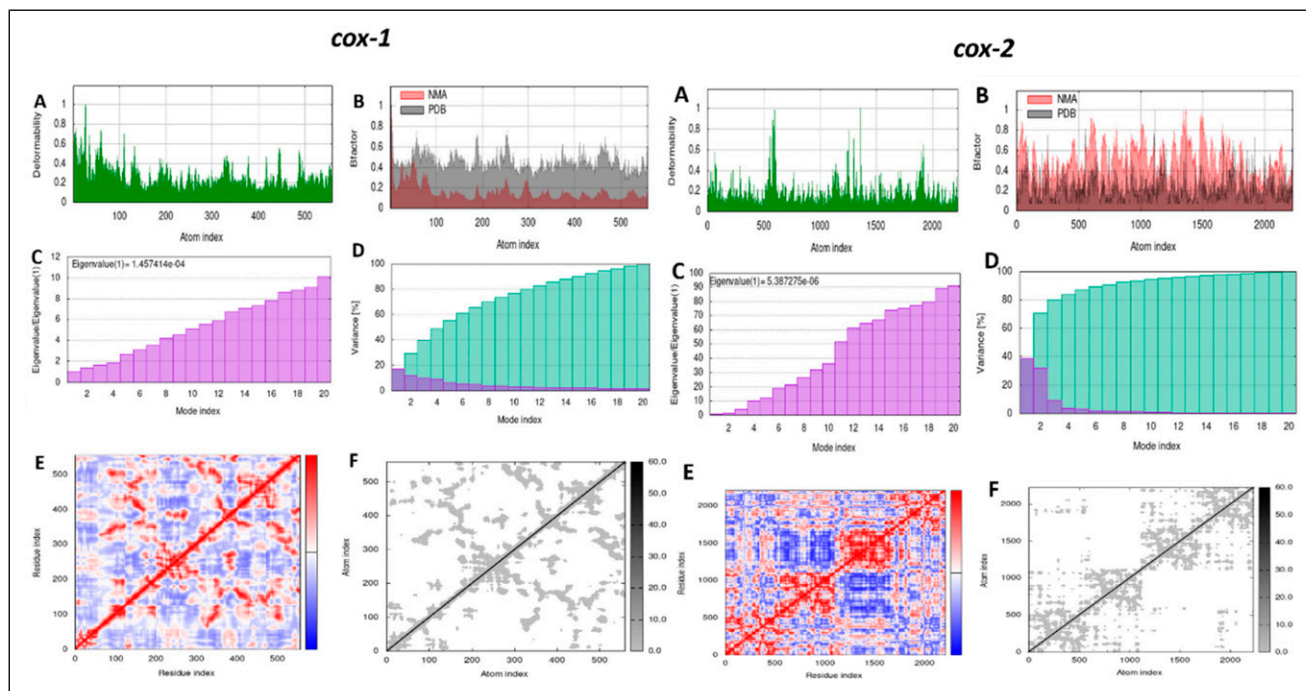


Figure 5. Molecular dynamics of *COX-1* and *COX-2*. (A) Molecular dynamics of stigmas-7,16-dien-3-ol, (3. beta. ,5. alpha.) showed the docking compound. (B) presents a low level of deformation of all residues. (C) shows the beta factor. (D) Eigon values are shown. (E) shows the variance explained in both red and green. (F) shows the variance and elastic of complex.

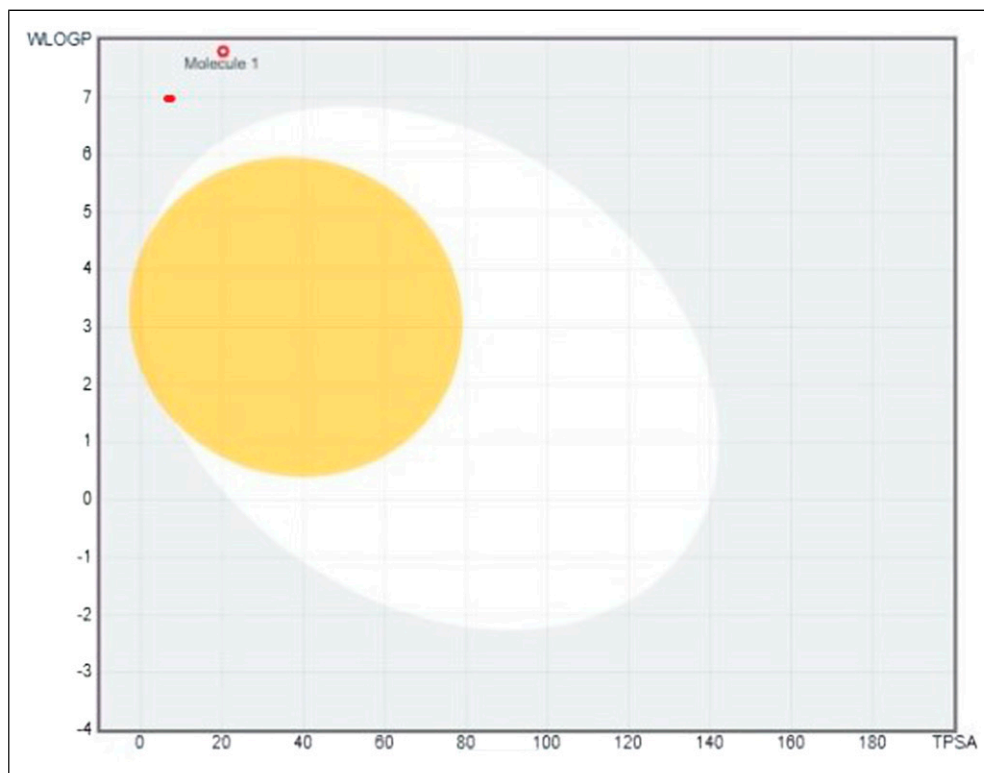


Figure 6. SWISS ADMET analysis of Stigma sterol. The yolk indicates the blood-brain barrier and the white area indicates gastrointestinal absorption.

412.69 g/mol. The pharmacokinetic analysis gave moderate GI absorption and did not violate Lipinski's rule. Compounds interaction and pharmacokinetics evaluated low glycoprotein, permeability, and no inhibitory effect on CYP2C19 and CYP2C9. The boiled-egg image of stigmasta-7,16-dien-3-ol, (3. beta. ,5. alpha.) complex is presented in [Figure 6](#).

Molecular Docking

For the preparation of binding site, PDB format of the drug was used and AutoDock Vina was used for virtual screening. Molecular docking was performed to visualize interaction between ligands and target proteins following the assessment of their binding affinities. The binding energies are demonstrated in [Table S8](#) and [Figure S2](#). The top two ligands with the least binding energy were hexadecanoic acid methyl ester and 9,12-octadecadienoic acid (Z, Z)-, methyl ester, showing scores of -4.9 and -5.1 . The highest binding energy was shown by stigmasta-7,16-dien-3-ol, (3. beta. ,5. alpha), which was -7.5 and -8.4 , 10 E,12(Z)-conjugated linoleic acid -7 and -6.5 was the most effective inhibitor. Further molecular simulation (MD) of top compounds with COX-1 and COX-2 was performed by iMOD simulation which showed the stability of the docked complexes between the phytochemicals and protein through strong hydrogen bonding.⁶⁷ According to the ADMET analysis and drug-like characteristics of the molecules as describe above, GIT bioavailability of the molecules is moderate, however, they are impermeable through the blood-brain barrier (BBB). An efficient drug possesses six physicochemical characteristics such as polarity, flexibility, lipophilicity, saturation, size, and solubility.⁶⁸ The *In-silico* data also indicated that stigmasta-7,16-dien-3-ol, (3. beta. ,5. alpha) might be good in anti-inflammation.

Conclusions

This study summarizes the biological activities of *C. colocynthis* fruits, and a comprehensive analysis of phytochemicals such as phenolics, flavonoids, and primary and secondary metabolites. Organic fractions of methanolic extract showed significant antioxidant, antimicrobial, antidiabetic, and anti-inflammatory activities. Hepatoprotective effect was evaluated by using animal modeling test which showed its safety to be used as medicine. The research comprehends the previously reported literature with new evidences and can be a potential cure for the local patients and pharmaceutical industries. The aforesaid study appraised that the fruit of *C. colocynthis* has potential for diverse biological activities and can be used to cure various ailments and may be used in future to design different drugs by in-silico approach.

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Supplemental Material

Supplemental material for this article is available online.

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