

# Attributing Increases in Ozone to Accelerated Oxidation of Volatile Organic Compounds at Reduced Nitrogen Oxides Concentrations

Zekun Zhang<sup>a</sup>, Jiakui Jiang<sup>a</sup>, Bingqing Lu<sup>a</sup>, Xue Meng<sup>a</sup>, Hartmut Herrmann<sup>b,c</sup>, Jianmin Chen<sup>b,a</sup> and Xiang Li<sup>b,a,\*</sup>

<sup>a</sup>Department of Environmental Science & Engineering, Fudan University, Shanghai 200032, China

<sup>b</sup>Institute of Eco-Chongming (IEC), Shanghai, China

<sup>c</sup>Leibniz-Institut für Troposphärenforschung (IfT), Permoserstr. 15, 04318 Leipzig, Germany

\*To whom correspondence should be addressed: Email: [lixiang@fudan.edu.cn](mailto:lixiang@fudan.edu.cn)

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## Abstract

Surface ozone ( $O_3$ ) is an important secondary pollutant affecting climate change and air quality in the atmosphere. Observations during the COVID-19 lockdown in urban China show that the co-abatement of nitrogen oxides ( $NO_x$ ) and volatile organic compounds (VOCs) caused winter ground-level  $O_3$  increases, but the chemical mechanisms involved are unclear. Here we report field observations in the Shanghai lockdown that reveals increasing photochemical formation of  $O_3$  from VOC oxidation with decreasing  $NO_x$ . Analyses of the VOC profiles and  $NO/NO_2$  indicate that the  $O_3$  increases by the  $NO_x$  reduction counteracted the  $O_3$  decreases through the VOC emission reduction in the VOC-limited region, and this may have been the main mechanism for this net  $O_3$  increase. The mechanism may have involved accelerated OH- $HO_2$ - $RO_2$  radical cycling. The  $NO_x$  reductions for increasing  $O_3$  production could explain why  $O_3$  increased from 2014 to 2020 in response to  $NO_x$  emission reduction even as VOC emissions have essentially remained unchanged. Model simulations suggest that aggressive VOC abatement, particularly for alkenes and aromatics, should help reverse the long-term  $O_3$  increase under current  $NO_x$  abatement conditions.

**Keywords:** ozone; VOCs and  $NO_x$  reduction ; photochemical box model

## Significance Statement:

Tropospheric  $O_3$  is an important secondary pollutant affecting climate change and air quality in the atmosphere. However, its response to precursor emission reductions is unclear. It is generally assumed to depend on the levels of  $NO_x$  and VOC reductions. However, we show here that the fast oxidation of VOCs occurs at decreased  $NO_x$  emissions during Shanghai's COVID-19 lockdown and can explain the substantial net  $O_3$  increase. This could explain why  $O_3$  has increased in response to  $NO_x$  emission reduction even as VOC emissions have remained unchanged. The mechanism may involve accelerated OH- $HO_2$ - $RO_2$  radical cycling.

## Introduction

Ground-level ozone ( $O_3$ ) is an air pollutant detrimental both to human health (1) and to terrestrial vegetation (2). In the mid-20th century, Haagen-Smit et al. proved that  $O_3$  is generated by the photochemical reaction of volatile organic compounds (VOCs) and nitrogen oxides ( $NO_x$ ) in sunlight (3). It is now well established that the chemistry of  $O_3$  formation is highly nonlinear and pertains to the coupling of the “ $NO_x$  cycle” and the “ $RO_x$  ( $RO_x = OH + HO_2 + RO_2$ ) radical cycle” (4–6). The  $NO_x$  and  $RO_x$  cycles are terminated by the cross-reaction of  $NO_x$  and/or  $RO_x$ . From the perspective of  $O_3$  pollution reduction, if  $HO_2$  self-reaction and its reaction with  $RO_2$  constitute the dominant termination reaction (known as the “ $NO_x$ -limited” regime), reductions in  $NO_x$  emissions serve the greatest benefit. Meanwhile, if the reactions of  $NO_2$  with OH and  $RO_2$  form the dominant termination reaction (known as the “VOC-limited” regime), reductions in VOC emissions prove

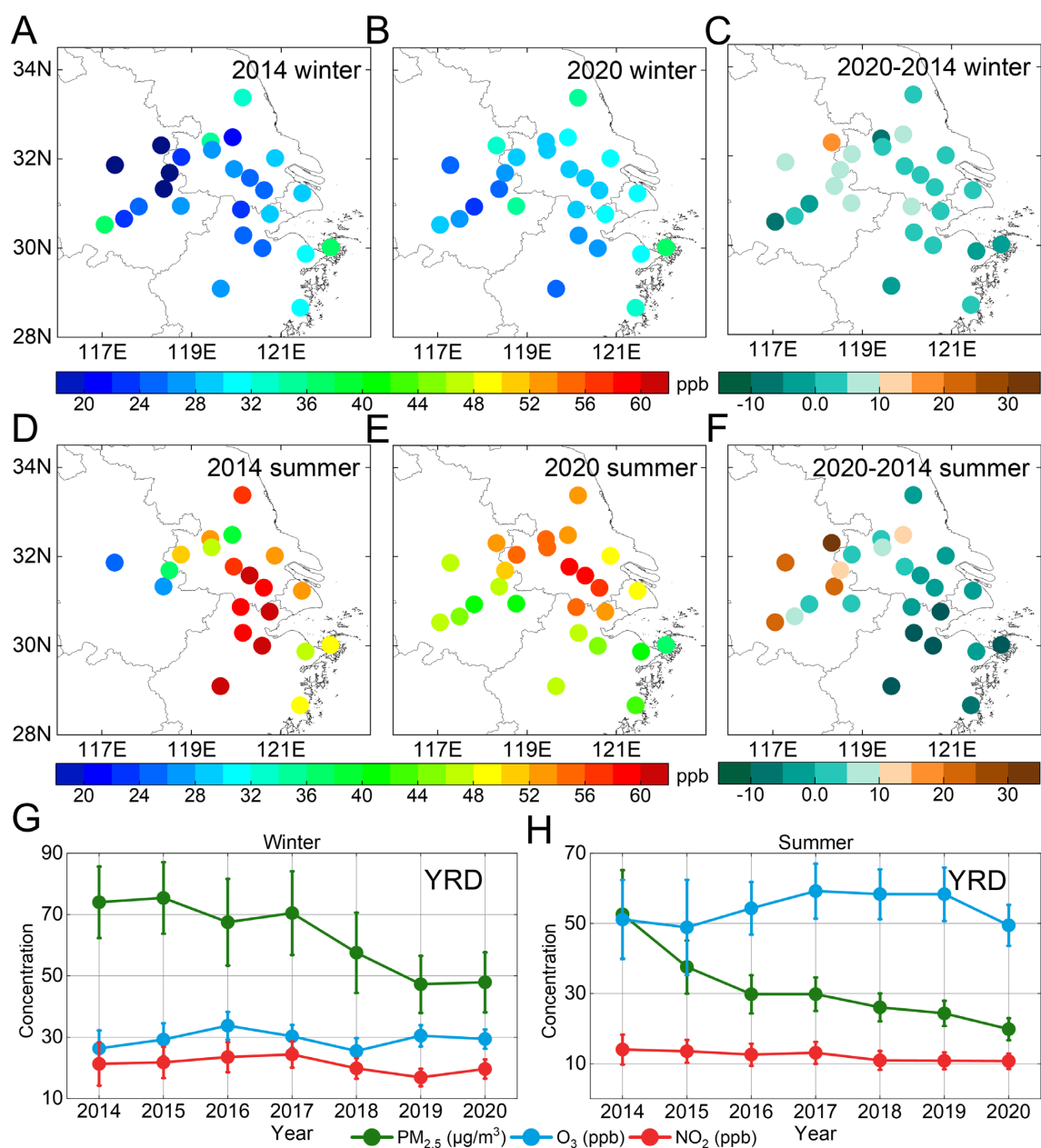
the most effective (5, 7). In the context of China, most city clusters [e.g. the Pearl River Delta, Yangtze River Delta (YRD) and Northern China Plain] are found to be VOC-limited (5), with anthropogenic VOCs dominating, whereas most rural areas are  $NO_x$ -limited in this regard (8).

China has been confronting serious air pollution for several years (9, 10). In an attempt to improve air quality, China has implemented strict pollution control policies since 2013. However, ground-level  $O_3$  concentrations have not improved following  $NO_x$  abatement (11, 12). Recent studies have shown that VOCs and  $NO_x$  should be controlled simultaneously to reverse  $O_3$  increases (13, 14). These results concur with those of Pusede (15) and Roberts et al. (16), who found that simultaneous VOC and  $NO_x$  abatement could considerably mitigate the peak of  $O_3$  production and prevent surface air quality deterioration.

**Competing interest statement:** The authors declare no competing interest.

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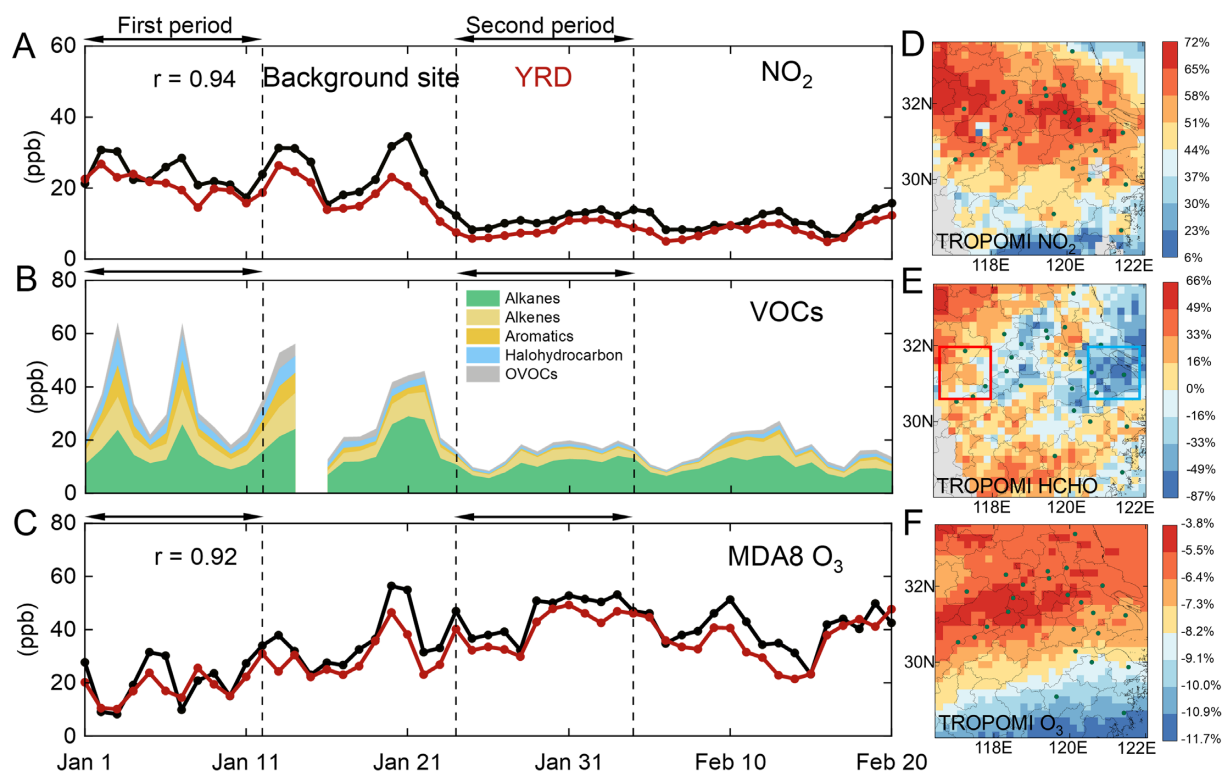


**Fig. 1.** Variations in winter and summer YRD's MDA8 O<sub>3</sub> concentrations from 2014 to 2020. (A and B) Distributions of winter mean MDA8 O<sub>3</sub> concentrations in YRD in 2014 and 2020. (D and E) Distributions of summer mean MDA8 O<sub>3</sub> concentrations in YRD in 2014 and 2020. (C and F) Reductions of winter and summer mean MDA8 O<sub>3</sub> concentrations between 2014 and 2020 (negative values). (G and H) Interannual variation (Error bars: SD) of NO<sub>2</sub>, MDA8 O<sub>3</sub> and PM<sub>2.5</sub> in the YRD during winter and summer.

The large-scale emission reductions seen during the COVID-19 lockdown are unprecedented opportunities for improving our understanding of O<sub>3</sub>'s response to the control of its precursor emissions. The Chinese government adopted countrywide measures from January 23 to February 13, 2020, to prevent the propagation of the disease. According to satellite and surface observations, NO<sub>2</sub> concentrations in China decreased by >50% during the lockdown period compared with those in the pre-lockdown period, and other pollutants such as PM<sub>2.5</sub> and CO also decreased significantly (17–19). However, ground-level O<sub>3</sub> concentrations in parts of China increased substantially (20–22). The increase in O<sub>3</sub> during the lockdown has been tentatively attributed to weakened titration (23, 24), O<sub>3</sub> production (25), and decreased PM<sub>2.5</sub> (19, 24).

However, the changes in the chemical mechanisms connecting O<sub>3</sub> levels with the abatement of its precursors are rarely discussed.

Here, we use detailed field observations during the Shanghai lockdown combined with an observation-based model (OBM) to show evidence for the increased photochemical formation of O<sub>3</sub> from VOC oxidation with decreasing NO<sub>x</sub>. We demonstrate that this mechanism can explain the observed O<sub>3</sub> increase. This also helps explain why O<sub>3</sub> concentrations in China increased in recent years as NO<sub>x</sub> emissions have greatly decreased while VOC emissions have remained unchanged (26, 27). We further show that a reasonable ratio of annual NO<sub>x</sub> and VOC reduction decreases O<sub>3</sub> photochemical production, with important implications for developing O<sub>3</sub> pollutant control strategies to decrease O<sub>3</sub> in China.



**Fig. 2.** Air quality changes in YRD in response to the COVID-19 lockdown from January 1 to February 20, 2020. (A and C) The red lines show observed concentrations of 24-h average  $\text{NO}_2$  and MDA8  $\text{O}_3$  at YRD. The  $\text{NO}_2$  and  $\text{O}_3$  observations are averages across the monitoring sites of the MEE. Also shown are observations results sampled at the background site and the corresponding correlation coefficients ( $r$ ). (B) VOCs observation data from background site. (D, E, and F) Relative percentage change [(FP-SP)/FP] in tropospheric  $\text{NO}_2$ , HCHO, and  $\text{O}_3$  column densities measured by the TROPOMI satellite instrument during FP and SP. The TROPOMI aggregation is on a  $5.5 \times 3.5$  km grid.

## Results and Discussion

### Worsening $\text{O}_3$ air quality from 2014 to 2020

Fig. 1A to F shows the spatial distribution of the winter and summer mean maximum daily 8-h average  $\text{O}_3$  (MDA8  $\text{O}_3$ ) concentrations around the YRD (Figure S1) in 2014 and 2020 (maps for 2015 to 2019 are presented in the Figure S2). In both years, high levels of MDA8  $\text{O}_3$  concentrations were observed in the eastern YRD, especially in Shanghai and surrounding regions, where the emission intensity was the highest (28). As illustrated in Fig. 1C and F, a slight increase in MDA8  $\text{O}_3$  concentrations occurred across the YRD from 2014 to 2020 even with considerable reductions in the emissions of major air pollutants (Figures S3 and S4, and Tables S1 to S3). According to an estimate of the temporal trend (Fig. 1G and H), the winter (summer) YRD concentration of MDA8  $\text{O}_3$  increased by  $0.43\% \text{ yr}^{-1}$  ( $1.25\% \text{ yr}^{-1}$ ) from 2014 to 2020. The titration effect caused by the  $2.3\% \text{ yr}^{-1}$  ( $4.1\% \text{ yr}^{-1}$ )  $\text{NO}_x$  decreases along with unchanged VOCs was found to be a significant cause of such increases. Meanwhile, the  $6.2\% \text{ yr}^{-1}$  ( $9.2\% \text{ yr}^{-1}$ ) reductions in  $\text{PM}_{2.5}$  may also be major factors as aerosol particles scavenge hydrogen oxide ( $\text{HO}_x$ ) and nitrogen trioxide ( $\text{NO}_3$ ) radicals, which would otherwise produce  $\text{O}_3$  (29). Similar increases in  $\text{O}_3$  simultaneously recorded with the progressive implementation of clean-air actions were generally observed in various areas of China, including the North China Plain and Pearl River Delta (30, 31). Such results point to potential increases in oxidation capacity in the YRD. Although  $\text{NO}_x$  reduction has traditionally been considered important for controlling  $\text{O}_3$ , recent studies have shown that the co-abatement of VOCs and  $\text{NO}_x$  is more important for suppressing  $\text{O}_3$  photochemical production (32, 33). In this regard, COVID-

19 represents an opportunity that informs our understanding of how VOCs and  $\text{NO}_x$  should be effectively controlled to reverse  $\text{O}_3$  increases.

The national lockdown policy in response to COVID-19 beginning on 2020 January 24, impelled a rapid drop in transportation and industrial emissions, compounded by the Chinese Spring Festival holiday (19, 34). For the present work, the “first period” (FP) comprised the normal emissions from January 1 to January 11, and the “second period” (SP) included the lockdown from January 24 to February 3 when lockdown restrictions were at their height (34). Fig. 2 and Figure S5A show 40% to 80% observed decreases in surface  $\text{NO}_2$  levels and tropospheric  $\text{NO}_2$  columns from the TROPOMI satellite, suggesting rapid falls in  $\text{NO}_x$  emissions in the YRD (the measurement of sites changes are discussed in SI Appendix A, Figure S6 and S7, and Tables S4 to S7). TROPOMI Formaldehyde (HCHO) observations suggest a weaker reduction in VOC emissions (Fig. 2E). For example, TROPOMI HCHO decreased by 12% in the western YRD (red squares), whereas it increased by 38% in the eastern area (blue squares). The surface  $\text{NO}_2$  concentrations came from the Ministry of Ecology and the Environment (MEE) monitoring site network. For the eastern YRD, as laid out in Fig. 2E, the average falls in the course of SP were 40% for surface  $\text{NO}_2$  and 60% for TROPOMI  $\text{NO}_2$ . Conversely, the  $\text{O}_3$  concentrations quantified at MEE sites and TROPOMI satellite increased in the course of the SP (Figure S6B and Fig. 2F). Ground-level MDA8  $\text{O}_3$  in the YRD increased from  $18 \pm 5$  ppb during the FP to  $40 \pm 5$  ppb during the SP.

We utilized a machine learning model to predict  $\text{O}_3$  levels in a “business-as-usual” scenario (Materials and Methods section).

Figure S8 shows that the  $O_3$  concentration during the lockdown significantly increased compared with that in the business-as-usual scenario. The increase in  $O_3$  might well be explained by unfavorable meteorological conditions and the co-abatement of  $O_3$  precursors. Therefore, we prioritized the use of multiple linear regression (MLR) models to assess the meteorological short-term effects on MDA8  $O_3$  before and after the lockdown (2020 January 1 to 2020 February 29) in the YRD (35). The coefficient of determination ( $R^2$ ), for the MLR model in fitting MDA8  $O_3$  anomalies in the YRD was 0.52 (regression coefficients are summarized in Table S8). We then used the Lindeman, Merenda, and Gold method to quantify the relative importance of each meteorological variable in explaining that the atmosphere-driven  $O_3$  increases (Figure S9). A decrease in fractional total cloud cover during the lockdown (Table S9) was the most significant driver of the  $O_3$  increase seen in the YRD with a contribution of 41%. Less cloud cover favors strong solar radiation, which in turn promotes  $O_3$  production (36). We found that 850-hPa relative humidity (RH850) was negatively correlated with  $O_3$  (regression coefficient:  $-0.3$ ) because wet aerosols formed under high humidity reduce ultraviolet actinic flux and thereby inhibit photochemical reaction rates (37). Meanwhile, the weakening of the 10-m zonal wind (U10, 7%) and 10-m meridional wind (V10, 3%) caused less downwind flux of  $O_3$ , resulting in more  $O_3$  remaining in situ in the YRD (Table S9) (38). These findings further demonstrated the importance of meteorological short-term effects on  $O_3$  (39). However, VOC and  $NO_x$  reduction effects on  $O_3$  formation mechanisms must be further considered to effectively understand the impact of the lockdown on  $O_3$  production.

### Evidence for $O_3$ formation by accelerated photochemistry

The cycle between OH and  $HO_2$  is highly pertinent to  $O_3$  formation, influencing the chain reaction propagation (40). Thus, the levels and reaction routes of free radicals are significant factors in  $O_3$  generation. The simulated diurnal patterns (3-day average) of free radicals at urban and background sites are displayed in Fig. 3A and B, and Figure S10A and B. The highest levels of free radicals were modeled around 12:00 to 13:00 at two sites. The daily peak of OH at the urban site increased by  $\sim 0.69 \times 10^6 \text{ cm}^{-3}$  in the SP relative to the FP but decreased by  $0.86 \times 10^6 \text{ cm}^{-3}$  at the background site. Meanwhile, the daytime-averaged OH concentration (08:00 to 16:00) at the urban site ( $0.89 \times 10^6 \text{ cm}^{-3}$ ) was comparable with that at the background site ( $0.67 \times 10^6 \text{ cm}^{-3}$ ) during the SP. The daytime-averaged OH level simulated in this research was comparable with that ( $1.5 \pm 0.9 \times 10^6 \text{ cm}^{-3}$ ), modeled on a Beijing winter (41).

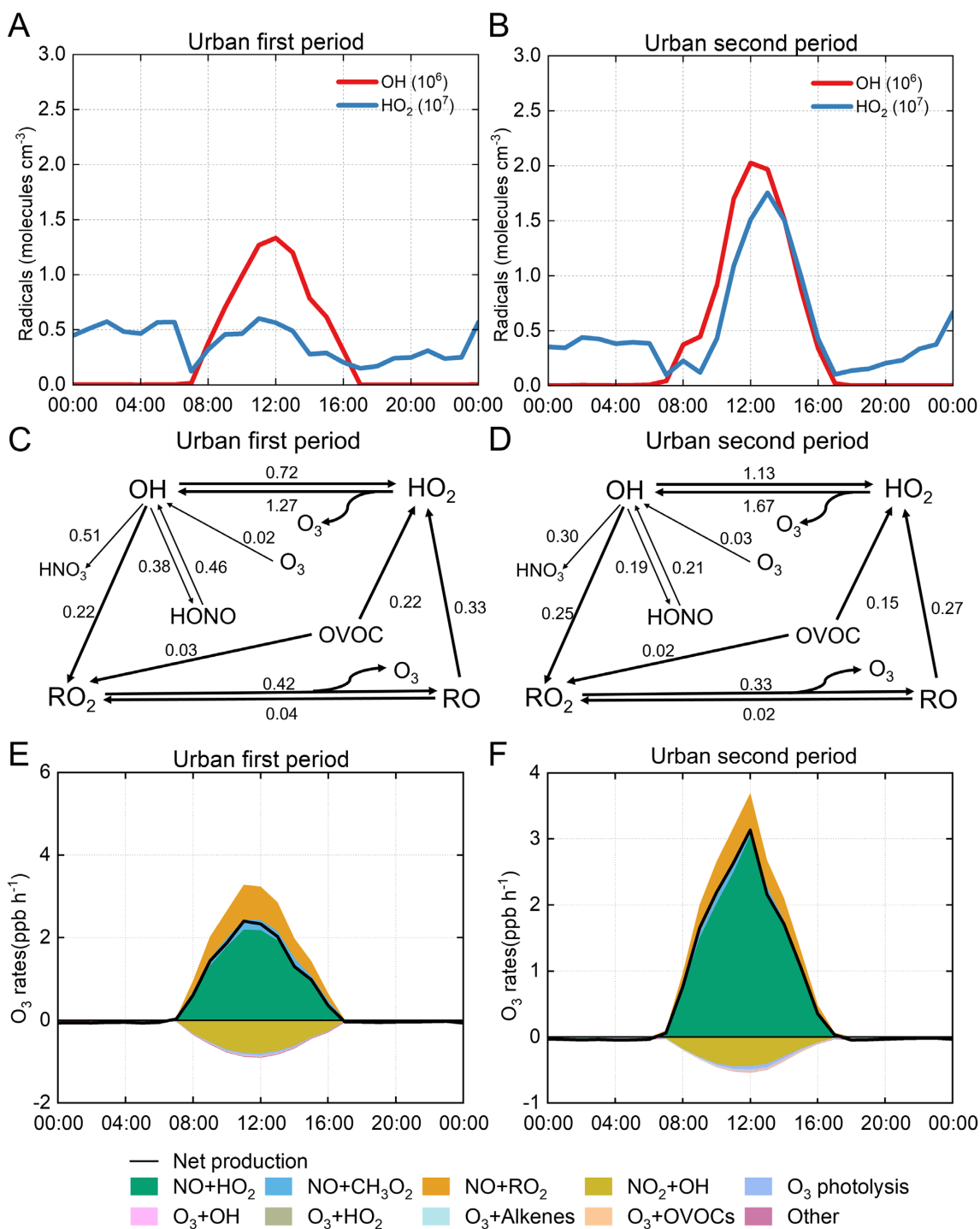
In comparison, the daily peak of  $HO_2$  at the urban site increased by  $\sim 0.95 \times 10^7 \text{ cm}^{-3}$  but decreased by  $0.49 \times 10^7 \text{ cm}^{-3}$  at the background site. The daytime-averaged  $HO_2$  concentration at the background site ( $1.02 \times 10^7 \text{ cm}^{-3}$ ) during the FP was more than double that at the urban site ( $0.41 \times 10^7 \text{ cm}^{-3}$ ), whereas the  $HO_2$  concentrations were similar at both sites during the SP, which was more likely because there were less VOCs at the background site and thus less  $HO_2$  produced through photochemical reactions. The daytime-averaged  $HO_2$  concentration at both sites during the SP was higher than that ( $0.52 \pm 0.23 \times 10^8 \text{ cm}^{-3}$ ) in Beijing during typical winter pollution episodes (the atmospheric oxidation capacity and the OH reactivity are discussed in SI Appendix B and Figures S11 and S12) (42). Table S10 compares the OH and  $HO_2$  levels for different Chinese cities. The differing levels of free radicals in various studies may be attributed to the different air pollutants and weather conditions at each locale sampled (40). The peak OH and

$HO_2$  concentrations responded differently at the two sites, highlighting the complex effects of the levels of  $NO_x$  and VOC reductions on photochemistry (23, 43).

Because OH- $HO_2$ - $RO_2$  is to a great extent responsible for the oxidation capacity of the atmosphere, calculating its radical sources and sinks offers an insightful understanding of the effects of the co-abatement of  $NO_x$  and VOCs on atmospheric chemical processes. The average production and loss rates for OH- $HO_2$ - $RO_2$  radicals at the urban and background sites are shown in Fig. 3C and D, and Figure S10C to D. The reaction  $HO_2 + NO$  was the main OH production source, with reaction rates reaching 1.27 and 2.71  $\text{ppb h}^{-1}$  for urban and background sites during the FP, respectively. However, this reaction increased to 1.67  $\text{ppb h}^{-1}$  at the urban site and decreased to 0.84  $\text{ppb h}^{-1}$  at the background site during the SP. Next, came the photolysis of HONO. The OH production rate from this reaction weakened by 48% and 66%, respectively, for the two sites. The main mechanism of OH consumption was the production of  $HNO_3$  and  $HO_2$ . During the FP, the high OH +  $NO_2$  reaction rate (to produce  $HNO_3$ ) was for the most part due to the relatively high daily average  $NO_x$  level at the urban (39.1 ppb) and background (26.6 ppb) sites (44). With  $NO_x$  abatement, this reaction fell by 41% and 59% at the urban and background sites, respectively. Unlike the OH +  $NO_2$  response, the rate of  $HO_2$  production from OH rose by 57% at the urban site but fell by 58% at the background site during the SP, probably due to different levels of  $NO_x$  and VOC co-abatement in the two locations.

In addition to OH, RO and oxygenated VOCs (OVOCs) were also primary sources of  $HO_2$ . These two chemical pathways decreased by 18% and 32%, respectively, at the urban site during the SP but decreased by 73% and 74%, respectively, at the background site. The major  $HO_2$  removal mechanism was its reduction reaction with NO, which produced OH radicals. Some research has also revealed that  $HO_2$  radicals can be taken up by aerosols (45). The  $RO_2$  formation rates of OH and OVOCs did not change significantly for the urban site but decreased by 36% and 67%, respectively, at the background site. Substantial VOCs reductions are important factors in the decline of the background site  $RO_2$  formation rates. We keenly point out that the reactions  $NO + HO_2$  and  $RO_2 + NO$  are the two major pathways for radical cycling at the two sites during the FP and SP, which are common in  $NO_x$ -rich environments (46, 47). Obviously, our results show that the OH- $HO_2$ - $RO_2$  cycling at urban stations remained at high levels during the lockdown, with particularly enhanced cycling between OH and  $HO_2$ . Meanwhile, significant weakening was observed at the background sites, which may have depended on the levels of VOC abatement.

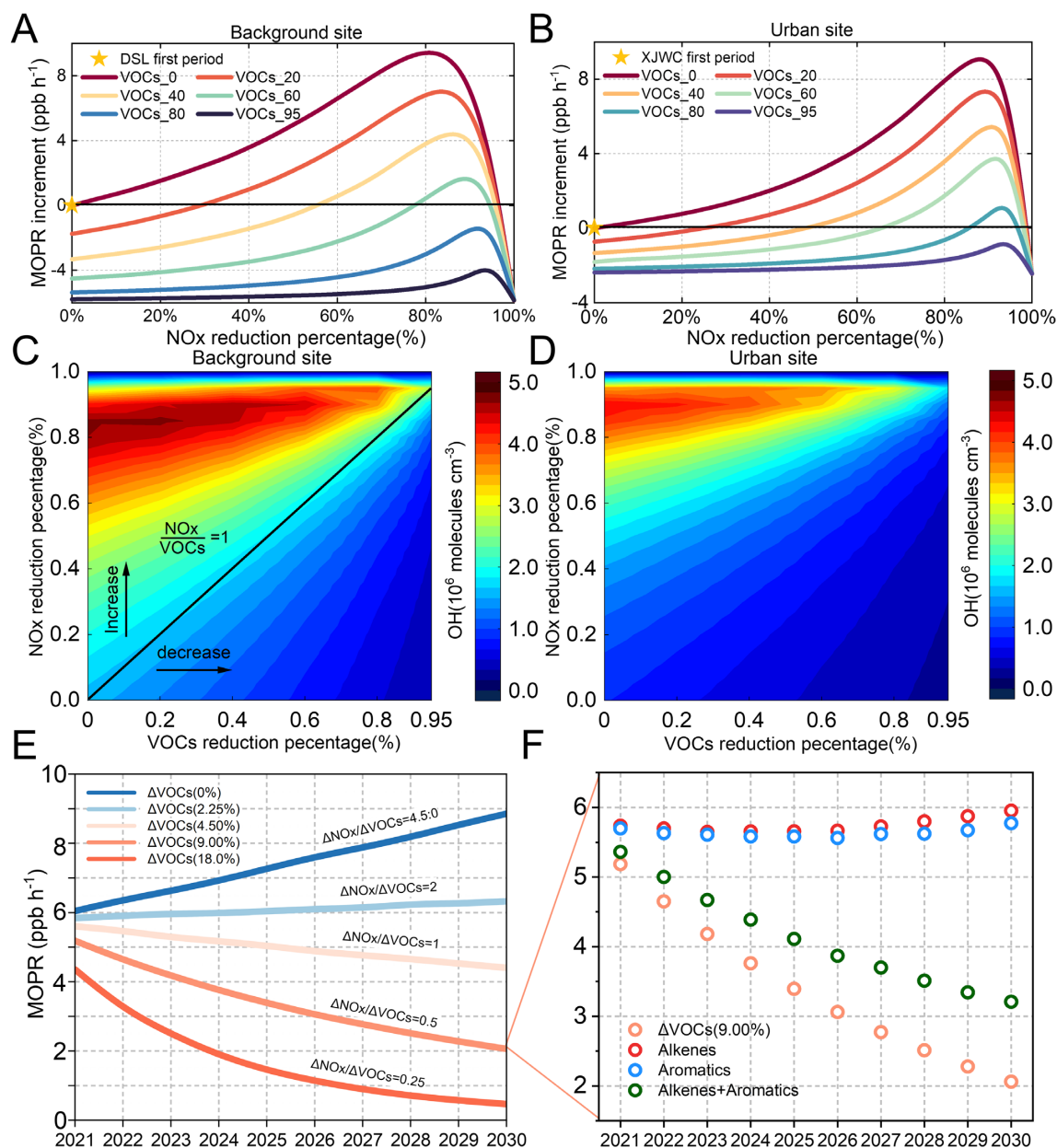
We further examined the  $O_3$  formation mechanisms for the lockdown.  $O_3$  was mainly produced through the oxidation of NO by  $HO_2$  and  $RO_2$  (48). For the  $O_3$  loss, the predominant scavenging pathways were the photolysis of  $O_3$  and the  $NO_2 + OH$  reaction. Fig. 3E and F, and Figure S10E and F show the detailed  $O_3$  chemical budgets before and after the lockdown at the urban and background sites. The major photochemical production of  $O_3$  was clearly demonstrated, with net daytime  $O_3$  average generation rates of 1.5 and 3.8  $\text{ppb h}^{-1}$  for the urban and background sites, respectively, during the FP. The  $O_3$  production intensity at the two sites was lower than that derived from an urban site in Wuhan autumn (8.8  $\text{ppb h}^{-1}$ ) (44) and comparable with that in Beijing's clean episode (2  $\text{ppb h}^{-1}$ ) (41). During the SP, the net daytime  $O_3$  average generation rates increased to 1.8  $\text{ppb h}^{-1}$  for urban sites but decreased to 0.8  $\text{ppb h}^{-1}$  for background sites. This may have had something to do with the  $O_3$  production rate increase from



**Fig. 3.** Relevant simulation results (3-day averaged) for urban site during FP and SP. (A and B) Simulated average diurnal variations in OH and HO<sub>2</sub>. (C and D) Daytime (08:00 to 16:00 LT) average OH-HO<sub>2</sub>-RO<sub>2</sub> budget. Units: ppb h<sup>-1</sup>. (E and F) Model-simulated average chemical budgets of O<sub>3</sub>.

NO<sub>x</sub> reduction exceeding the decrease rate associated with VOC reduction at urban sites, causing an increase in net O<sub>3</sub> production rate and, thus, a substantial O<sub>3</sub> increase (Figure S13A and B). Conversely, the NO<sub>x</sub> reductions at the background sites had less impact on O<sub>3</sub> than the control of the VOCs, leading to the decrease in O<sub>3</sub> production rates (Figure S13C and D). The O<sub>3</sub> production rate peaks around midday (literally at around 12:00 LT), only to be succeeded by a major drop during the afternoon, consistent with the overall findings from other works that reported noon or afternoon peaks (49, 50).

The links between O<sub>3</sub> and precursor species are elucidated more by the relative incremental reactivity (RIR) process using the OBM (Materials and Methods section). As can be seen in Figure S14, the simulation results for the two sites were the same during the FP and SP. The RIR values of each VOC category were positive for all scenarios, and only NO<sub>x</sub> was negative, indicating that O<sub>3</sub> formation was consistently in the VOC-limited regime (51). During the FP, the two sites had high RIR for aromatics and alkenes, which showed that they control the production of O<sub>3</sub> to a larger extent. Notably, we found that O<sub>3</sub> production at both sites was



**Fig. 4.** Simulation in background and urban sites according to the emission changes of NO<sub>x</sub> and VOCs, with first period (yellow pentagon) as the simulation scenario. (A and B) MOPR increment in background and urban site, the black line represents zero MOPR increment. (C and D) Isoleths of daytime OH concentration in background and urban sites. (E) In a 4.5% annual NO<sub>x</sub> reduction scenario, simulation of MOPR under five  $\Delta\text{NO}_x/\Delta\text{VOCs}$  reduction ratios from 2021 to 2030. (F) MOPR is simulated for alkenes, aromatics, and alkenes and aromatics at an annual reduction rate of 9% for the scenario.

more sensitive to alkenes and CO during the SP, suggesting that future targets of O<sub>3</sub> control should emphasize reducing not only the active species of VOCs but also CO emissions.

### Implications for air pollution control

O<sub>3</sub> formation was determined to be in the VOC-limited regime at the background and urban sites. However, how much VOCs should be controlled to achieve the optimal O<sub>3</sub> reduction was nonetheless uncertain, particularly in a future emission scenario where VOCs and NO<sub>x</sub> tend to be controlled simultaneously. In the pursuit of this objective and to offer detailed information on the requisite decreases in VOCs and NO<sub>x</sub>, we simulated the maximum O<sub>3</sub> production rate (MOPR) increment with the reduction of both VOCs and NO<sub>x</sub> (Fig. 4A and B). We found that at the background (ur-

ban) site, when the mixing ratios of VOCs were lowered from 0% to 95%, the appropriate reduction percentages of NO<sub>x</sub> tended to fall between 0% and 78% (85%) or in the range of 94% (96%) to 100% for the zero MOPR increment. Admittedly, reducing NO<sub>x</sub> by 94% to 100% may prove impracticable. Therefore, this section looked primarily at the range 0% to 78% (85%) in NO<sub>x</sub> reduction to produce the best possible VOC and NO<sub>x</sub> control measures. The researchers found that the MOPR could be reduced when the NO<sub>x</sub>/VOC coabatement ratio was <1.5 for the background site, resulting in the suppression of O<sub>3</sub> production, and 1.25 for the urban site (i.e. the cutting ratios of NO<sub>x</sub>/VOCs where the curves meet the MOPR zero increment line). However, the simulation results for OH and HO<sub>2</sub> radicals at both sites showed that the radical concentrations can be effectively reduced when NO<sub>x</sub>/VOCs <1 (Fig. 4C and D and Fig-

ure S15). This indicates that the weakening of OH and HO<sub>2</sub> experiences lags relative to the MOPR, which may influence atmospheric PM<sub>2.5</sub> formation (52). Similar simulation results for the two sites indicate the consistency of future co-abatement strategies for Shanghai and the YRD.

We obtained a reasonable ratio above by controlling the total abatement of NO<sub>x</sub> and VOCs. However, to achieve the National Ambient Air Quality Standard for O<sub>3</sub>, NO<sub>x</sub> and VOC emissions must be reduced to a large degree (14, 53), which cannot be accomplished in the short-term. Thus, reasonable annual reduction plans should be developed to ensure their enforceability. Accordingly, NO<sub>x</sub> emissions are assumed to annually decrease by 4.5% ( $\Delta\text{NO}_x = 4.5\%$ ). The reduction rate in NO<sub>x</sub> was referenced to the trend of winter NO<sub>2</sub> concentration in Shanghai, Nanjing, Hangzhou, and the YRD region from 2014 to 2020 (Figure S16). Working with this postulate, we looked at the associated benefits of various VOC emission reduction rates for 2021 to 2030. As presented in Fig. 4E, we designed five control scenarios with different VOC control rates. For every scenario, the annual emission reduction rate was the same during the emission control period. The results show the additional lowering of the MOPR with faster VOC emission control than those under lower VOC falls. Taking the annual VOC reduction rates ( $\Delta\text{VOCs}$ ) of 4.5%, 9%, and 18% as examples, the MOPR reduction rates were 4.5, 6.7, and 8.5 ppb h<sup>-1</sup>, respectively, compared with the zero VOC reduction (0%) in 2030. Apparently, controlling VOCs to large extents can guarantee the effectiveness of NO<sub>x</sub> control in lowering O<sub>3</sub>. Such reduced O<sub>3</sub> levels arising from extensive VOC control strategies can introduce other positive effects through the reduction of their negative effects on humans and nature (14).

The above results indicate that the MOPR decreases yearly when annual  $\Delta\text{VOCs} \geq 4.5\%$ . Considering the lag of free radical decrease relative to MOPR (Figure S17) and economic feasibility, we considered annual  $\Delta\text{VOCs}$  of 9% to be reasonable. Meanwhile, previous results showed that alkenes and aromatics have stronger control effects on O<sub>3</sub> formation. To highlight the role of reactive VOCs in future controls, we conducted sensitivity simulations by removing the effects of other VOCs (Fig. 4F). We found that the reduction of alkenes and aromatics alone can only offset the negative impact of NO<sub>x</sub> reduction relative to zero VOC reduction conditions. For the condition with both reductions, we see from Fig. 4F that the weakening of MOPR is also significant relative to the condition with all VOCs reduced. This implies that future VOC plans should prioritize the reduction of alkenes and aromatics (54).

In summary, we have shown evidence from field observations in the Shanghai lockdown that reveal the increasing photochemical formation of O<sub>3</sub> from VOC oxidation under decreased NO<sub>x</sub> emission to promote O<sub>3</sub> production. The evidence suggests that the O<sub>3</sub> increase by the NO<sub>x</sub> reduction counteracted the O<sub>3</sub> decrease by the VOC emission reduction in the VOC-limited region. Therefore, O<sub>3</sub> control requires more aggressive VOC reduction under current NO<sub>x</sub> reduction conditions, particularly for alkenes and aromatic VOCs. What we present thus explains the O<sub>3</sub> increase in the lockdown despite the reductions in its precursors. This also helps explain why O<sub>3</sub> has increased from 2014 to 2020 in response to NO<sub>x</sub> emission reduction even as VOC emissions have remained unchanged. Although the national lockdowns are not sustainable, O<sub>3</sub> response to the precursor emission reductions could be used as a reference for future VOC and NO<sub>x</sub> control. The reduction associated with O<sub>3</sub> formation may be important for aerosol climate effects and in understanding pollution–weather feedback loops, most particularly the mitigation of PM formation through VOC reduction (55, 56).

## Materials and Methods

### Observations

The YRD, located in eastern China adjacent to the East China Sea, is one of the biggest city conglomerates in the world. Its location places it in a subtropical monsoon climate. Although the YRD covers only about 1.1% of China's total land area, it is home to ~7.6% of China's total population and accounted for 16.5% of the country's Gross Domestic Product in 2016 (<http://data.stats.gov.cn/>). The high population density and widespread heavy industries lead to intensive energy consumption and emission of pollutants. Therefore, the YRD region has experienced severe O<sub>3</sub> pollution and is considered the key prevention and control area for O<sub>3</sub> pollution in China (57, 58). In this study, 26 cities in the YRD were considered (Figure S1A): Shanghai, Nanjing, Wuxi, Changzhou, Suzhou, Nantong, Yancheng, Yangzhou, Zhenjiang, Taizhou, Hangzhou, Ningbo, Jiaxing, Huzhou, Shaoxing, Jinhua, Zhoushan, Taizhou, Hefei, Wuhu, Maanshan, Tongling, Anqing, Chuzhou, Chizhou, and Xuancheng. The above data of routine air pollutants (PM<sub>2.5</sub>, NO<sub>2</sub>, CO, O<sub>3</sub>) at ground level for winter (December, January, and February) and summer (June, July, and August) 2014 to 2020 in 26 cities were obtained from the China National Environmental Monitoring Center (CNEMC). Monitoring data were collected using continuous automated equipment. All equipment passed the applicability test by the CNEMC, and all monitoring data were subjected to validity audits.

The field experiment was conducted at the background (121.021°E, 31.238°N) and urban site (121.507°E, 31.329°N) in YRD region, China. The background site (Figure S1B) is surrounded by Dianshan Lake and several villages with low population density and is about 40 km away from the built-up center of Shanghai. The region is frequently defined as a reference for the study of air pollution in the YRD (59, 60). The urban site is surrounded by a concentration of residential areas, commercial areas, schools and hospitals, which was under strong influence by urban pollutant emissions.

At the background and urban observation sites, NO<sub>x</sub> (NO/NO<sub>2</sub>) and 60 VOCs were measured. To be specific, hourly concentrations of 60 VOC species were monitored by an online gas chromatography analyzer (Focused Photonics, GC955-615), VOC species are detailed in Table S6. Ambient pollutants were monitored by Thermo Environmental Instruments. NO, NO<sub>2</sub>, and NO<sub>x</sub> were detected by the NO-NO<sub>2</sub>-NO<sub>x</sub> analyser (Model 42i). For routine gaseous pollutant concentration (PM<sub>2.5</sub>, O<sub>3</sub>, NO<sub>2</sub>, SO<sub>2</sub>, and CO) monitoring, we used monitoring data from the CNEMC sites nearest to the background and urban site. In addition, meteorological data (temperature, humidity, etc.) are collected hourly from the ERA5 reanalysis dataset.

Satellite observations of O<sub>3</sub>, NO<sub>2</sub> and HCHO columns from the TROPOMI instrument were accessed from <https://s5phub.copernicus.eu/dhus/>. TROPOMI provides daily global coverage with 5.5 × 3.5 km<sup>2</sup> pixel resolution. TROPOMI data have been applied in tracking anthropogenic emission changes during the lockdown. In this study we used the TROPOMI level 2 observations (Table S11).

### The LightGBM model

LightGBM is a state-of-the-art gradient-boosting framework that uses tree-based learning algorithms. It is designed with faster training speed, lower memory usage, better accuracy and capability of handling large-scale data (61). We trained the LightGBM model with 17 features and ozone labels from 1 January 2019 year to 23 January 2020 (Table S12). To achieve better performance, we performed grid tuning of the parameters by a randomly crossval-

idated search on parameter settings. We finally selected the following hyperparameters: max depth = 16, num estimators = 127, reg alpha = 5, learning rat = 0.1. Notably, the emission levels for the “business-as-usual” scenario in the model were calculated using the methodology of shi et al (25).

### Observation-based model (OBM)

The OBM for the investigation of atmospheric oxidative capacity and photochemistry (OBM), as a zero-dimensional box model, has been extensively used to simulate atmosphere chemical processes (62–65). It is based on the latest version of the Master Chemical Mechanism (MCM v3.3.1) (66), which contains more than 5800 chemical species and 17 000 reactions, especially, over 200 reactions about VOCs oxidation by chlorine radical. Expect for gas-phase reactions, several heterogeneous processes and physical processes are considered in simulation (67, 68). However, the impact of changes in aerosol loading on HO<sub>2</sub> and O<sub>3</sub> production is not parameterized in the model. The latest study by Ivatt et al. (45) shows that a novel “aerosol suppression” regime is suppressing surface O<sub>3</sub> production in China and suggests removing the uptake of HO<sub>2</sub> into aerosols increases O<sub>3</sub> concentrations in the North China Plain by 20% to 30%. This result is based on the 2014 emission scenario. PM<sub>2.5</sub> concentrations in the YRD region are significantly decreased in 2020 compared to 2014 (Fig. 1G and H), therefore we suggest that the sensitivity of surface O<sub>3</sub> to HO<sub>2</sub> uptake in China has currently been reduced. This is in line with Ivatt et al., who consider that O<sub>3</sub> sensitivity to HO<sub>2</sub> uptake had reduced in Europe, North America and Japan as policy interventions had reduced particulate-matter concentrations.

In this study, we used the field-measured VOCs, O<sub>3</sub>, NO, NO<sub>2</sub>, SO<sub>2</sub>, and CO, and physical parameters to constrain the boundary of the model and input them into the model for calculation. The photolysis frequencies data were obtained from the Tropospheric Ultraviolet and Visible (TUV, v5.3.1) model combined with the European Centre for Medium-Range Weather Forecasts (ECMWF) products ERA5 reanalysis dataset and observed data. Moreover, the missing VOCs data were interpolated using the MLR with other hour data of the day or average the close hour data. If the data miss more than three hours in a day, the average value of the corresponding hour during this period will represent the hourly value of that day.

OBM is able to simultaneously quantify the O<sub>3</sub> production rate, atmospheric oxidation capacity (AOC), OH reactivity, and the primary production, recycling, and termination rates of OH–HO<sub>2</sub>–RO<sub>2</sub> radicals (62). Here, we focus on AOC, O<sub>3</sub> budget, and OH–HO<sub>2</sub>–RO<sub>2</sub> budget. Before each simulation, the model starts at 00:00 LT and pre-run eight days under constraints of input data to reach a steady-state for unmeasured species in the sampling field.

### Relative incremental reactivity (RIR)

The RIR is an important index to evaluate the sensitivity and control factors of ozone generation, which can also be calculated by OBM (69–71). The definition of RIR is the ratio of the percentage change in ozone formation rate to the percentage change in ozone specific precursor source strength after changing ozone specific precursor source strength, seeing equation.

$$\text{RIR} = \frac{P_{\text{O}_3-\text{NO}}(X) - P_{\text{O}_3-\text{NO}}(X - \Delta X)}{\Delta S(X)/S(X)} \quad (1)$$

Where X represents either a specific precursor or a group of VOCs,  $\Delta X$  is the hypothetical change of X (20% was adopted in this study).  $P_{\text{O}_3-\text{NO}}(X)$  and  $P_{\text{O}_3-\text{NO}}(X - \Delta X)$  mean the production rate

of simulated O<sub>3</sub> based on measured concentration and emission reduction concentration.  $S(X)$  and  $S(\Delta X)$  are the concentration of these corresponding specific precursors and the hypothetical changed concentration, with the unit of per part billion.

### Multiple linear regression (MLR)

We performed a stepwise MLR model applied to daily MDA8 O<sub>3</sub> concentrations by considering many candidate meteorological variables from the ERA5 reanalysis. The MLR model is in the following form:

$$y = \beta_0 + \sum_{k=1}^{10} \beta_k x_k + \text{interaction terms}, \quad (2)$$

where y is the daily MDA8 O<sub>3</sub> concentrations, and ( $x_1, \dots, x_8$ ) are the eight meteorological variables (Table S8) that were selected by a stepwise method featuring the best model fit. All y and  $x_k$  data need to be normalized by subtracting the mean and dividing by the SD.

We next use the LMG (Lindeman, Merenda, and Gold) method to quantify the relative importance of each of the ten selected meteorological variables in explaining the atmospherically driven increase in O<sub>3</sub> in Shanghai over pre-lockdown and COVID-19 lockdown. The LMG method can divide the total R<sup>2</sup> explained by the MLR model into each non-negative individual R<sup>2</sup> contribution of each relevant regressor. This approach has been used in many previous studies to investigate the relative importance of model predictors in PM<sub>2.5</sub> concentrations (72), clouds, and changes in radiative forcing (38).

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### Supplementary Material

Supplementary Material is available at [PNAS Nexus](#) online.

### Authors' Contributions

Z. Z., X. L., J. C., and H. H. designed research; Z. Z. and J. J. performed research; Z. Z., B. L., and X. M. analyzed data; Z. Z., and X. L. wrote the paper.

### Data Availability

All study data are included in the article and Supplementary Material.

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