

# A Review on Microorganisms in Constructed Wetlands for Typical Pollutant Removal: Species, Function, and Diversity

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Wang J, Long Y, Yu G, Wang G, Zhou Z, Li P, Zhang Y, Yang K and Wang S (2022) A Review on Microorganisms in Constructed Wetlands for Typical Pollutant Removal: Species, Function, and Diversity. Front. Microbiol. 13:845725. doi: 10.3389/fmicb.2022.845725 Constructed wetlands (CWs) have been proven as a reliable alternative to traditional wastewater treatment technologies. Microorganisms in CWs, as an important component, play a key role in processes such as pollutant degradation and nutrient transformation. Therefore, an in-depth analysis of the community structure and diversity of microorganisms, especially for functional microorganisms, in CWs is important to understand its performance patterns and explore optimized strategies. With advances in molecular biotechnology, it is now possible to analyze and study microbial communities and species composition in complex environments. This review performed bibliometric analysis of microbial studies in CWs to evaluate research trends and identify the most studied pollutants. On this basis, the main functional microorganisms of CWs involved in the removal of these pollutants are summarized, and the effects of these pollutants on microbial diversity are investigated. The result showed that the main phylum involved in functional microorganisms in CWs include Proteobacteria, Bacteroidetes, Actinobacteria and Firmicutes. These functional microorganisms can remove pollutants from CWs by catalyzing chemical reactions, biodegradation, biosorption, and supporting plant growth, etc. Regarding microbial alpha diversity, heavy metals and high concentrations of nitrogen and phosphorus significantly reduce microbial richness and diversity, whereas antibiotics can cause large fluctuations in alpha diversity. Overall, this review can provide new ideas and directions for the research of microorganisms in CWs.

Keywords: constructed wetlands, functional microorganisms, microbial diversity, pollutant removal, wastewater treatment

# INTRODUCTION

Constructed wetlands (CWs) are passive biological engineering systems that use natural processes for wastewater treatment (Chen et al., 2021b; Zheng et al., 2021a). They have been widely used since the 1960s because of their simple operation, ease of maintenance, low cost, and environmental friendliness, providing a viable alternative to traditional

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wastewater treatment technologies (Zhao et al., 2016; Zheng et al., 2021a,b). They are mainly composed of substrate, plants, and microorganisms that purifying wastewater through the interaction of physical, chemical, and biological processes (Zhao et al., 2020c). Previous studies have shown that CWs can remove most environmental pollutants, including COD, N, P (Lin et al., 2002; Li et al., 2013; Zhao et al., 2019), heavy metals (Zhang et al., 2021e), and antibiotics (Liu et al., 2019), as well as some increasingly emerging pollutants (e.g., pesticides, flame retardants and persistent organic pollutants; Rajan et al., 2019; Long et al., 2021; Vymazal et al., 2021). Consequently, CWs are widely used in the treatment of domestic sewage, industrial wastewater, mine drainage, land leachate, polluted lake water, effluent from the livestock industry and other wastewater (Lin et al., 2002; Zhao et al., 2020c).

In CWs, microorganisms play a key role in pollutants removal, such as the degradation of organic pollutants and the conversion of various nutrients (Wang et al., 2020d). They can even use antibiotics as their sole carbon source (Ricken et al., 2013; Bessa et al., 2017). Regarding heavy metal compounds, which are generally difficult to biodegrade, microorganisms can also remove them from wastewater through biosorption, bioaccumulation and speciation transformation (Si et al., 2019). In addition, microorganisms can improve the tolerance and removal efficiency of CWs to pollutants by enhancing phytoremediation (Syranidou et al., 2018; Vassallo et al., 2020). In this context, in order to further optimize CWs, it is necessary to investigate the functional microorganisms associated with pollutant removal.

In recent years, advances in molecular biotechnology have largely facilitated intensive studies of microbial community structure and diversity (Arroyo et al., 2013). The advent of methods such as 16S sequencing, metagenomics sequencing, and high-throughput sequencing technologies has not only allowed a more accurate assessment of microbial diversity but only the analysis of the relative abundances of different microbial species and of the overall community structure (Zhao et al., 2016; Sanchez, 2017). In CWs, the diversity of microbial communities and the richness of certain species are key factors for efficient wastewater treatment (Zhao et al., 2020c). Therefore, in addition to the need for a summary of functional microorganisms, there is also a great need to study and analyze the effects of different pollutants on microbial diversity.

In recent year, various types of bibliometric analysis have been applied in different fields (Zhao et al., 2020b; Chen et al., 2021a; Kilicaslan et al., 2021). As a quantitative analysis technique, bibliometrics can reveal the current status and trends of a given research field by studying the distribution, quantitative relationships, and changing relationships of literature and information (Mao et al., 2015, 2018b; Liang and Gong, 2017). Overall, bibliometrics is a highly effective method of summarization and analysis and has become a useful tool when dealing with large amounts of scientific data (Mao et al., 2018a). This can help us to study the current state of research and trends in the field of microorganisms in CWs. In this context, the purpose of this review is to first analyze, *via* bibliometrics, microbial-related articles in the field of CWs and several typical pollutants that are closely associated with microorganisms. Subsequently, the main functional microorganisms associated with the removal of targeted pollutants in CWs are systematically summarized and the effects of these pollutants on the diversity of microbial communities in CWs are discussed. This review will help us to further understand and explore the mechanisms of pollutant removal by microorganisms in CWs and the effects of pollutants in wastewater on microorganisms.

## **BIBLIOMETRICS**

Here, the bibliometrics approach is divided into two main parts: plotting of the publication trends of articles related to CWs and microorganisms using the Origin 2021 software and analyzing the keywords found in the publications in terms of microorganisms using the VOSviewer software.

## DATA COLLECTION

Respective data were obtained from the Web of Science (WOS) Core Collection database. And this review only considered the Science Citation Index Expanded (SCI-Expanded). To fully study the changes in the number of publications over the years, only publications from 1900 (the earliest time point that that can be set in the online SCI-Expanded database) to December 2020 were considered.

A search for CWs with the keywords "constructed wetland\*" or "artificial wetland\*" or "man-made wetland\*" or "treatment wetland\*" or "engineered wetland\*" or "reed bed\*" yielded 9,628 documents (the starting year is 1991, as the earliest record of microbial publications is from this year). The keywords "bacteri\* OR microb\* OR microorganism\*" were searched for microbial, yielding a total of 2,764 documents. The year of publication and the bibliographic information of these publications (including authors, titles, source publications, references abstracts, and cited) were exported for subsequent analysis.

## **BIBLIOMETRIC ANALYSIS**

The first aspect is the investigation of the general trend of microbial research by determining the change of articles involving microorganisms in CWs over time. For this, the yearly number of publications was counted and the articles about CWs and microorganisms were indicated separately by different colors to obtain the percentage of articles about microorganisms in the field of CWs (**Figure 1**). As can be seen in **Figure 1**, there is an overall upward trend in the number of yearly publications on CWs and microorganisms. Furthermore, an increasing number of publications involving CWs are related to microorganisms. These results reflect the wide use and study



of CWs in an increasing number of countries as well as reflecting the increasing importance of microorganisms. Particularly in recent years with the advances in the molecular characterization of microbial communities, such as denaturing gradient gel electrophoresis or terminal fragment length polymorphism of PCR-amplified 16S ribosomal RNA gene fragments, as well as metagenomics, have greatly contributed to the development of microbial ecology research (Zhao et al., 2016; Sanchez, 2017; Rajan et al., 2019). This can be corroborated with the increasing trend of cumulative publication volume of microbial articles in **Figure 1**. We fit the cumulative publication volume of microbial articles and found that it in line with the exponential function, with an  $R^2$  value of 0.9987, indicating an exponentially increasing publication trend.

The second aspect is the analysis of keyword co-occurrence by the VOSviewer software, with the aim to understand the connection among author keywords. After exporting the publications related to microbial from the WOS platform in plain text form, they were analyzed using the keyword co-occurrence function of the VOSviewer software and subsequently merged through the thesaurus file. The final results are shown in **Figure 2**. This figure shows the top 50 keywords in terms of the number of occurrences number; the high numbers indicate that there are more studies related to them, facilitating the subsequent analysis and summary. Different circles in the figure represent different keywords, and the circle size indicates the number of times the keywords appear, the larger the circle, the more times it appears. The line between the circles indicates that two keywords have appeared together in an article, and the more times they appear together, the thicker the line is. By selecting the keyword "microorganisms," we could observe the connection between this one and other keywords. Among these keywords, we intercepted the top 10 keywords in terms of number; of these 5 were related to the type of contaminants, namely nitrogen, phosphorus, heavy metals, antibiotics, and nutrients. However, among them, nutrients mainly contain nitrogen and phosphorus (Li et al., 2013), and therefore, this review finally identified nitrogen, phosphorus, heavy metals, and antibiotics as four typical contaminants for a functional microbial summary and diversity analysis.

## FUNCTIONAL MICROORGANISMS

As mentioned above, four pollutants typical for CWs, namely nitrogen, phosphorus, heavy metals, and antibiotics, were



identified. Many functional microorganisms play an important role in the treatment of these polluted water. This review provides a summary and functional analysis of these microorganisms. Because the microbial species present in CWs are highly diverse, only the more abundant functional species reported were considered. In addition, only the phylum and genus are summarized and analyzed in this review, as most studies have analyzed microbial species at these two levels.

# FUNCTIONAL MICROORGANISMS IN NITROGEN REMOVAL

Excess nitrogen discharge into water bodies tends to cause eutrophication and black-odorous, which deteriorates water quality and in turn poses a serious threat to humans and aquatic organisms (Wang et al., 2020d; Zhang et al., 2021c). Biological processes are the key processes in the nitrogen removal mechanisms of CWs, with Tan et al. (2021a) reporting that microorganisms can remove almost 90% of the nitrogen. The main pathways of nitrogen removal by microorganisms in CWs are shown in **Figure 3**.

Current research generally agrees that nitrogen removal by microorganisms in CWs is mainly accomplished through ammonification, nitrification and denitrification (Hu et al., 2016; Xie et al., 2016; Zhao et al., 2018). Ammonification is the process of converting organic nitrogen (Org-N) in wastewater into NH4<sup>+</sup>, which is then removed in other processes (e.g., nitrification, volatilization, and plant uptake; Lee et al., 2014; Xie et al., 2016). As shown in Table 1, the popular genera of ammonifying bacteria include Chitinophaga, Isoptericola, Bacillus, and Sinorhizobium. Regarding nitrification and denitrification, the microorganisms use NH4+ as electron donor during nitrification and oxidize it to NO2- and further to NO<sub>3</sub><sup>-</sup>, which is then used as electron acceptor during denitrification and reduced to N<sub>2</sub>O or N<sub>2</sub> (Tan et al., 2021b; Zhang et al., 2021c). The microorganisms involved in nitrification can be divided into two categories, ammonia-oxidizing bacteria (AOB) and archaea (AOA) which convert ammonium to nitrite, and nitrite-oxidizing bacteria (NOB) which convert nitrite to nitrate (Zhang et al., 2021b,c). In particular, AOA has a higher adaptability to low ammonia and high salt environments compared to AOB (Wang et al., 2019b; Zhao et al., 2021). This can help the AOA to become the predominant group more quickly and speed up the process of nitrification, as ammonia oxidation is the first and rate-limiting step in nitrification (Wang et al., 2019b). The popular phyla involved in nitrification include Proteobacteria, Nitrospirae, Nitrospinae, and Thaumarchaeota. Of these, the Thaumarchaeota phylum contains all the currently known AOA (Wang et al., 2019b). Regarding denitrification, Proteobacteria, Bacteroidetes, Firmicutes, and Actinobacteria are popular denitrifying bacteria in CWs.



However, nitrifying bacteria in CWs microbial communities usually face problems of low abundance and weak competitiveness (Tan et al., 2021a). This will result in a longer start-up period required for stable NH4+ oxidation, making nitrification a limiting step for nitrogen removal (Tan et al., 2021a). In this context, recent studies have highlighted the importance of heterotrophic nitrification and aerobic denitrification (HN-AD) bacteria (Liu et al., 2020b; Tan et al., 2021a; Wang et al., 2022). These bacteria can be responsible for  $NH_4^+$  and  $NO_3^$ transformation in the start-up phase of CWs, converting the nitrogen in the aqueous solution to nitrogen gas for complete denitrification (Tan et al., 2021a). Moreover, they grow more rapidly and can dominate quickly (Tan et al., 2021a). The discovery of HN-AD bacteria has changed the traditional theory that nitrification can only be carried out by autotrophic bacteria and denitrification can only take place under anaerobic conditions, which makes it more advantageous in nitrogen removal and organic matter removal (Wang et al., 2022). The HN-AD bacteria reported in the studies mainly included the genera Dechloromonas, Ferribacterium, Hydrogenophaga, Zoogloea, and Aeromonas. Regarding the denitrification process, new nitrogen removal pathways have also been detected, such as sulfur autotrophic denitrification (SAD) and denitrifying anaerobic methane oxidation (DAMO; Huang et al., 2020b; Wang et al., 2020a). During SAD, sulfur-oxidizing bacteria (SOB) reduce NO3<sup>-</sup> to N2, using elemental S, S2<sup>-</sup>, and

 $S_2O_3^{2-}$  as electron donors and  $NO_3^-$  as electron acceptors under anaerobic or anoxic conditions (Li et al., 2020a; Wang et al., 2020a). Hence, this pathway may dominate in removing nitrogen of low C/N ratio water due to available electron donors from sulfur and its compounds (Wang et al., 2020a). Most sulfur autotrophic denitrifying bacteria belong to the phylum *Proteobacteria*, with the popular genera being *Thiobacillus* and *Sulfurimonas*. Regarding DAMO, which can reduce  $NO_2^-$  to  $N_2$  under anaerobic conditions using methane (CH<sub>4</sub>) as the electron donor and sole source of carbon (Huang et al., 2020b; Zhang et al., 2020). DAMO can alleviate the greenhouse effect and contribute to reduce the unnecessary byproduct  $N_2O$  in the nitrogen removal process, thus allowing for more environmental benefits (Huang et al., 2020b; Zhang et al., 2020).

In addition to the traditional nitrification-denitrification process of nitrogen removal, there exists a novel pathway—anaerobic ammonia oxidation (anammox; Hu et al., 2016; Kraiema et al., 2019). In this pathway, nitrite is used as an electron acceptor under anaerobic conditions to convert ammonia directly to N<sub>2</sub> (Zhao et al., 2018; Kraiema et al., 2019). This makes it an alternative denitrification pathway at low oxygen levels and low C/N ratios (Hu et al., 2016). At present, almost all reported anammox bacteria belong to the phylum *Planctomycetes* (Jia et al., 2021; Zhang et al., 2021c). Regarding nitrate, among the different nitrogenous pollutants, nitrate nitrogen is more likely to leach and eventually deteriorate water

### TABLE 1 | Functional microorganisms in CWs for nitrogen removal.

| Function  |  | Phyla   | Genera (Notes)   | Nitrogen transformation process   | References   |
|---|--|---|--|---|--|
| Ammonification  |  | Bacteroidetes,<br>Actinobacteria,<br>Firmicutes,<br>Proteobacteria  | Chitinophaga, Isoptericola, Bacillus,<br>Sinorhizobium   | $Org\text{-}N\!\rightarrow\!NH_4^+$   | Xie et al., 2016   |
|   | Ammonia oxidizing<br>archaea (AOA)   | Thaumarchaeota  | Nitrososphaera, Nitrosopumilus, Candidatus<br>Nitrosotalea, Candidatus_Nitrosoarchaeum,<br>Candidatus_Nitrosopumilus<br>Nitrosomonas, Nitrosospira   | $NH_4^+ \rightarrow NO_2^-$   | Wang et al., 2019b;<br>Zhao et al., 2021   |
| Nitrification   | Ammonia oxidizing<br>bacteria (AOB)  | Proteobacteria  | (Belongs to <i>Betaproteobacteria</i> )<br><i>Rudaea,</i> (Belongs to <i>Gammaproteobacteria</i> )<br><i>Nitrobacter</i> (Belongs to <i>Alphaproteobacteria</i> )  |   | Zhao et al., 2016;<br>Huang et al., 2019a  |
|   | Nitrite oxidizing<br>bacteria (NOB)  | Proteobacteria<br>Nitrospirae<br>Nitrospinae<br>Chloroflexi   | Nitrosococcus (Belongs to<br>Gammaproteobacteria)<br>Nitrospira<br>Nitrospina<br>Nitrolancea<br>Bradyrhizobium, Hyphomicrobium,<br>Rhizobium, Rhodobacter, Rhodoplanes,<br>Paracoccus, Methylobacterium,<br>Gemmobacter, Brevundimonas,<br>Broseobacter, Arospirillium                             | $NO_2^- \rightarrow NO_3^-$   | Zhao et al., 2021<br>Wang et al., 2020d<br>Gao et al. 2017: Zhao   |
|   | Traditional<br>denitrification or<br>dissimilatory nitrate to<br>ammonium (DNRA) | (Belongs<br>Thauera<br>Proteobacteria Denitrati<br>Ferribac<br>(Belongs<br>Enteroba<br>Rhodano<br>Halomor | (Belongs to Alphaproteobacteria)<br>Thauera, Comamonas, Sulfuritalea,<br>Denitratisoma, Azoarcus, Ralstonia,<br>Ferribacterium<br>(Belongs to Betaproteobacteria)<br>Enterobacter, Thermomonas, Arenimonas,<br>Rhodanobacter, Silanimonas, Dokdonella,<br>Halomonas, Zobellella, Thiothrix, Vibrio | $NO_2^-/NO_3^- \rightarrow N_2 \uparrow$<br>(Traditional denitrification)<br>$NO_2^-/NO_3^- \rightarrow NH_4^+$<br>(DNRA) | et al., 2018, 2019,<br>2020d, 2021; Aguilar<br>et al., 2019; Liu et al.,<br>2020b; Wang et al.,<br>2020d; Jia et al., 2021;<br>Tan et al., 2021b;<br>Zhang et al., 2021d |
| Denitrification   |  | Actinobacteria<br>Bacteroidetes   | (Belongs to Gammaproteobacteria)<br>Desulfovibrio, Geobacter, Sulfuricurvum<br>Propionicella, Micropruina<br>Maritimimonas, Chryseolinea, Prolixibacter,<br>Paludibacter, Terrimonas   |   | Zhao et al., 2019<br>Si et al., 2018; Ajibade<br>et al., 2021; Kowal<br>et al., 2021; Zhao et al.,   |
|   |  | Firmicutes<br>Calditrichaeota   | Clostridium<br>Calorithrix<br>Thiobacillus, Thiomonas  |   | 2021<br>Zhao et al., 2019<br>Zhao et al., 2021   |
|   | Sulfur autotrophic denitrification (SAD)   | Proteobacteria  | (Belongs to Betaproteobacteria)<br>Thiohalophilus, Thioalbus<br>(Belongs to Gammaproteobacteria)<br>Sulfurimonas, Sulfurovum   | $S + NO_3^- + NH_4^+ \rightarrow N_2 \uparrow + SO_4^{2-}$  | Wang et al., 2020a;<br>Zhao et al., 2021   |
|   | Denitrifying anaerobic methane oxidation   | Bacteroidetes   | (Belongs to Epsilonproteobacteria)<br>Flavobacteriaceae  |   | Wang et al., 2020a   |
|   |  | candidate division<br>NC10  | Candidatus Methylomirabilis  | $CH_4 + NO_2^- \rightarrow N_2 \uparrow + CO_2 \uparrow$  | Zhang et al., 2020   |
| 11-4  | (DAIVIO)   | Proteobacteria  | Zoogloea, Dechloromonas, Acidovorax,<br>Hydrogenophaga, Ferritrophicum,<br>Propionivibrio<br>(Belongs to Betaproteobacteria)   |   | Liu et al., 2020b; Tan<br>et al., 2021a; Wang  |
| Heterotrophic nitrification and aerobic denitrification (HN-AD) |  | Probic  | Pseudomonas, Acinetobacter, Aeromonas,<br>Klebsiella   | $NH_4^+/NO_3^-/NO_2^-\!\rightarrow\!N_2\uparrow$  | et al., 2022   |
|   |  | Bacteroidetes<br>Firmicutes   | (Belongs to Gammaproteobacteria)<br>Flavobacterium, Pedobacter<br>Bacillus   |   | Zhao et al., 2016  |
| Anaerobic ammo<br>(anammox)                                     | nia oxidation  | Planctomycetes  | Candidatus_Scalindua, Candidatus<br>Kuenenia, Candidatus Brocadia  | $NH_4^+ + NO_2^- \rightarrow N_2 \uparrow$  | Zhao et al., 2021  |

quality (Li et al., 2021b). Therefore, nitrate removal is important to protect freshwater systems and underground water quality (Li et al., 2021b). In addition to denitrification, there is an alternative pathway for the reduction of nitrate, namely the dissimilatory nitrate reduction to ammonium (DNRA; Rahman et al., 2019). DNRA reduces  $NO_3^-$  to available  $NH_4^+$  for use by other microorganisms, such as AOB and AOA (Zhang et al., 2021b,c). It has been reported to be more favorable than denitrification under high salinity conditions in sulfide-rich marine and coastal ecosystems (Zhang et al., 2021c). Many studies have found that some denitrifying genera able to execute the DNRA process, such as Vibrio, Clostridium, and Desulfovibrio (Zhang et al., 2021a,c). However, it is still difficult to distinguish denitrifying bacteria from DNRA bacteria, which requires further development of molecular biotechnology. However, currently, denitrifying bacteria and DNRA bacteria are not well distinguished, which requires further development of molecular biotechnology.

Based on the summary in **Table 1**, the phylum *Proteobacteria* contains a large number of species involved in nitrogen transformation. This phylum is widely distributed in CWs and is the most dominant phylum in most systems, playing an important role in nitrogen removal from different wastewaters (Gao et al., 2017; Si et al., 2018; Zhao et al., 2018). The genera *Nitrosomonas, Nitrobacter*, and *Nitrosospira* are associated with nitrification. The genera *Tauera, Thiobacillus, Thermomonas,* and *Arenimonass* are frequently detected among denitrifying bacteria. The class *Alphaproteobacteria, Betaproteobacteria*, and *Gammaproteobacteria* are the dominant class related to nitrogen removal in CWs. They contain large numbers of nitrifying bacteria, AOB, and NOB, which play important ecological functions in CWs and are largely involved in the nitrogen removal (Aguilar et al., 2019; Ajibade et al., 2021).

In addition, there is now a growing number of studies linking functional genes to the functional and quantitative analysis of nitrogen removal microorganisms (Zhao et al., 2020d; Tan et al., 2021b; Zhang et al., 2021b). For example, the abundance of *nrfA*- and *nirK*-carrying microorganisms influenced the denitrification performance of CWs (Zhao et al., 2020d); the abundance of the nitrification functional genes *amoA*-AOA, *amoA*-AOB, and *nxrA* represented the growth status of nitrifying bacteria (Zhao et al., 2021). Currently, the functional gene pools associated with the various processes of nitrogen removal (e.g., nitrification, denitrification, anammox, and DNRA) have been summarized (Tan et al., 2021a,b; Zhang et al., 2021b). Functional genes can essentially analyze the function of microorganisms and provide a feasible approach for us to further study functional microorganisms in CWs.

# FUNCTIONAL MICROORGANISMS IN PHOSPHORUS REMOVAL

Phosphorus is one of the main elements causing eutrophication in water bodies (Du et al., 2017; Wang et al., 2021b). Excess phosphorus discharged into the aquatic environment from domestic, agricultural, and industrial sources can also harm aquatic life by altering the pH, lowering oxygen levels, and promoting algal growth (Du et al., 2017; Wang et al., 2021b). Microorganisms play an important role in the removal of phosphorus from CWs and can influence the form of the phosphorus (Wang et al., 2021b). The main microorganisms associated with phosphorus removal in CWs are shown in **Table 2**.

Biological phosphorus removal in CWs is mainly achieved by phosphorus-accumulating organisms (PAOs), which can absorb phosphate from wastewater and store it in cells under alternating aerobic and anaerobic conditions (Du et al., 2017; Shi et al., 2017; Tian et al., 2017). Under anaerobic conditions, PAOs break down intracellular polyphosphate and take up volatile fatty acids from the environment, which is then stored in the form of polyhydroxyalkanoates (Lv et al., 2021). Under aerobic conditions, PAOs rely on polyhydroxyalkanoates for energy provision and absorb phosphate to form polyphosphate storage (Tian et al., 2017). In general, the amount of phosphorus uptake by PAOs will be greater than the amount of phosphorus released, thus realizing the phosphorus removal process of microorganisms in CWs (Du et al., 2017; Shi et al., 2017; Tian et al., 2017). The main phylum is Proteobacteria, which are largely involved in phosphorus removal (Si et al., 2018, 2019; Huang et al., 2020a). Of these, Alphaproteobacteria, Betaproteobacteria, and Gammaproteobacteria contain most of the microbial species associated with biological phosphorus removal (Shi et al., 2017; Huang et al., 2020a; Wei et al., 2020; Lv et al., 2021). The families Rhodobacteraceae and Rhizobiaceae of the class Alphaproteobacteria can absorb volatile fatty acids under aerobic conditions and convert them into poly-\beta-hydroxyalkanoates, facilitating total phosphorus removal in CWs (Lv et al., 2021). The class Betaproteobacteria mainly contains the genera Candidatus Accumulibacter, Dechloromonas, and Rhodocyclus. Of these, the genus Candidatus Accumulibacter is considered a typical PAOs and the dominant PAOs in full-scale wastewater treatment plants and laboratory-scale reactors (Huang et al., 2019a). The genus Dechloromonas can reduce perchlorate, accumulate polyphosphate, and absorb carbon under anaerobic conditions (Huang et al., 2020a). Rhodocyclus have also been shown to have a significant contribution to phosphorus removal (Li et al., 2017). Regarding Gammaproteobacteria, three genera, namely Klebsiella, Pseudomonas, and Acinetobacter have been identified in relevant studies (Tian et al., 2017). Of these, Pseudomonas has a strong ability to absorb phosphorus from wastewater and store it in its cells as polyphosphate, making it an effective phosphorusremoval microorganism (Huang et al., 2020a). Tian et al. (2017) report that it can remove up to 80.6% of total phosphorus from domestic wastewater. Regarding the genus Acinetobacter, it is the first bacteria isolated from biomass with a high phosphorus removal capacity (Du et al., 2017). In addition to Proteobacteria, other taxa, such as Gemmatimonadacea, that can take up excess phosphate under aerobic conditions (Wang et al., 2021b).

The phosphorus removal efficiency of PAOs mainly depends on the accumulation and consumption of intracellular polyphosphate (Tian et al., 2017), which is directly related to the activities of the enzymes exopolyphosphatase (ppx) and polyphosphate kinase (ppk; Du et al., 2017). The ppx and ppk can catalyze anaerobic phosphorus release and aerobic phosphorus

| Function   | Phyla                              | Genera (Notes)  | Morphology of the removed phosphorus                   | References  |  |
|--|------------------------------------|---|--|---|--|
|  |                                    | Rhodobacteraceae (family),  |  |   |  |
|  |                                    | Rhizobiaceae (family)   |  | Lv et al., 2021   |  |
|  | Proteobacteria                     | (Belongs to Alphaproteobacteria)<br>Candidatus Accumulibacter,<br>Dechloromonas, Rhodocyclus                |  | Li et al., 2017; Huang et al.,<br>2019a, 2020a; Zheng et al., |  |
| PAO  |                                    | (Belongs to Betaproteobacteria)   | Phosphate  | 2021a   |  |
|  |                                    | Pseudomonas, Klebsiella,<br>Acinetobacter   |  | Du et al., 2017; Tian et al.,<br>2017; Huang et al., 2020a;   |  |
|  |                                    | (Belongs to Gammaproteobacteria)  |  | Zheng et al., 2021a   |  |
|  | Chloroflexi, Gemmatimonadetes      | Rhodocyclaceae (family),<br>Gemmatimonadacea (family),<br>Gemmatimonas                                      |  | Wei et al., 2020; Wang et al.,<br>2021b                       |  |
| PSB  | Actinobacteria, Proteobacteria     | Corynebacterium, Enterobacter   | Convert insoluble phosphorus into soluble phosphorus   | Wang et al., 2021b  |  |
| DNPAO  | Proteobacteria                     | Paracoccus (Belongs to<br>Alphaproteobacteria),<br>Pseudomonadaceae (family),<br>Pseudomonas, Dechloromonas | Polyphosphate  | Huang et al., 2019a; Lv et al.,<br>2021; Wang et al., 2021b   |  |
|  | Chloroflexi                        | Anaerolineae (class)  |  | Lv et al., 2021   |  |
| Solubilize vast tricalcium phosphate through secreting organic acids | Proteobacteria                     | <i>Delftia</i> (Belongs to<br><i>Betaproteobacteria</i> )   | Phosphate  | Li et al., 2020b  |  |
| Associated with the P  | Proteobacteria                     | Brevundimonas, Pseudorhodoferax,<br>Variovorax, Panacagrimonas  | Organic phosphoric acid esters/<br>Insoluble phosphate | Wu et al., 2020   |  |
| ,  | Chlorobi, Firmicutes, Spirochaetes | Chlorobaculum, Bacillus, Leptospira   |  | Wu et al., 2020   |  |

uptake, respectively, to achieve biological phosphorus removal (Tan et al., 2021b). However, high temperatures inhibit their activities; according to a previous study, the optimum temperature ranges from  $20.0-35.0^{\circ}$ C (Du et al., 2017).

In addition to PAOs, phosphorus-solubilizing bacteria (PSB) and denitrifying phosphorus-accumulating organisms (DNPAO) have also been found in CWs. Example are the genera *Corynebacterium* and *Enterobacter*, which are PSB that secrete organic acids (e.g., oxalic and citric acids) to convert insoluble phosphorus in the soil into soluble phosphorus for plant uptake (Wang et al., 2021b). Regarding DNPAO, it can use NO<sub>3</sub><sup>-/</sup> NO<sub>2</sub><sup>-</sup> as electron acceptors to absorb polyphosphate under anoxic conditions (Wang et al., 2020b). *Alphaproteobacteria* (e.g., the genus *Paracoccus*) and *Anaerolineae* have been reported to be DNPAO (Lv et al., 2021). The genera *Brevundimonas* and *Chlorobaculum* produce organophosphate hydrolases that hydrolyze organophosphate as a phosphorus source for growth (Wu et al., 2020).

## FUNCTIONAL MICROORGANISMS IN HEAVY METAL REMOVAL

Heavy metals are widely distributed in aquatic systems, difficult to degrade, and can accumulate in the food chain, making them hazardous environmental pollutants (Yu et al., 2020; Chen et al., 2021b). In CWs, microorganisms can be effective in removing heavy metals through mechanisms, such as biosorption, biomineralization, and valence transformation (Si et al., 2019; González Henao and Ghneim-Herrera, 2021). **Figure 4** shows the main pathways of heavy metal removal by microorganisms in CWs. The relevant microbial phyla and genera are summarized in **Table 3**.

Among heavy-metal polluted water, mining-impacted water, especially acid mine drainage (AMD), has attracted widespread attention worldwide (Chen et al., 2021b). The AMD generated during and after mining and smelting activities is characterized by high acidity and sulfate and toxic metallic ion enrichment (Chen et al., 2021b,c). Therefore, for AMD remediation, bacterial sulfate reduction in CWs is a key process as it reduces the acidity of AMD and removes heavy metals by immobilizing them as sulfides, hydroxides, and carbonate precipitates (Chen et al., 2021c). The bacteria involved in sulfate reduction are known as sulfate-reducing bacteria (SRB) and they can drive simultaneous sulfate and metal removal as well as acidity neutralization (Habe et al., 2020; Chen et al., 2021b). The majority of SRB belong to the class Deltaproteobacteria in the phylum Proteobacteria. Among them, the more popular ones include the genera Desulfovibrio, Desulfobacter, Desulfobulbus, and Desulfurobacterium. In addition to SRB, other functional microorganisms with complementary ecological niches are also important for the effective remediation of AMD (Chen et al., 2021b,c). For example, members of the genus Propioniciclava can use a variety of carbohydrates to produce acetate and propionic acids, driving dissimilatory SBR metabolism (Torregrosa et al., 2019). Regarding the genus Cellulomonas, it can protect SRB



community from oxygen exposure and also generate low-molecular-weight compounds through saccharification and fermentation to act as electron donors for SRB (Chen et al., 2021b). Therefore, these phyla and genera, which are mainly involved in organic decomposition and sulfate reduction, are the key microbial groups participating in the treatment of AMD in CWs.

Ironically, metal ions generally negatively impact microorganisms by disrupting cell membranes, inhibiting enzyme activity, destroying DNA, and disturbing cellular function (Yu et al., 2020), making tolerance important for the removal of heavy metals by microorganisms. Fe $^{2+}$  can be oxidized to Fe $^{3+}$ by the genera Thiomonas and Sideroxydans, making it easier to precipitate and thus less hazardous (Chen et al., 2021c). Yu et al. (2020) also found that the genera Serratia and Pseudomonas screened using Cd2+ and Zn2+ concentrations gradients showed resistant to these two heavy metals resulting in an increase in removal rates of 10.13 and 8.57%, respectively. The extracellular polymeric substances synthesized by Pseudomonas can bind heavy metals and block their diffusion

within the biofilm, achieving extracellular sequestration, thereby protecting cells from heavy metal stress (Teitzel and Parsek, 2003; Giovanella et al., 2017). In addition, the cell surface of Pseudomonas and Serratia could also enhance the adsorption of Cd<sup>2+</sup> and Zn<sup>2+</sup> due to the presence of anionic functional groups (Cristani et al., 2012; Limcharoensuk et al., 2015). These findings lead us to infer that the cultivation of resistant microorganisms is a viable approach in heavy metal removal from wastewater and deserves further investigation. However, the way in which Serratia reduces the hazard of heavy metals is through secreting several proteins and enzymes such as heavy metal-binding proteins, transporter proteins, amino acids, histidine-binding proteins, and redox enzymes, which can efflux metal ions (Chen et al., 2019). This cannot contribute to heavy metal removal by CWs. Therefore, resistant microorganisms are not exactly the same as functional microorganisms and further research into the mechanisms of heavy metal removal by microorganisms is required to make a determination. Yu et al. (2020) found that functional microorganisms also evolved in the control group that was not inoculated with resistant

#### TABLE 3 | Functional microorganisms in CWs for heavy metal removal.

| Pollutant type  | Phyla               |                                | Genera (Notes)   | Removal principle  | References   |
|---|---------------------|--------------------------------|--|--|--|
| MIW, Especially AMD<br>(Mainly contains $Fe^{2+}$ , $Cd^{2+}$ , $Zn^{2+}$ , $Cu^{2+}$ , $Cr^{2+}$<br>and other heavy<br>metals) | SRB                 | Proteobacteria                 | Desulfobacterium,<br>Desulforhabdus,<br>Desulfobacca,<br>Desulforegula,<br>Desulforustis,<br>Desulforibrio,<br>Desulfobacter,<br>Desulfobacter,<br>Desulfobalbus,<br>Desulfococcus,<br>Desulfocapsa,<br>Desulfatirhabdium<br>(Belongs to<br>Deltaproteobacteria) | $SO_4^{2-}$ +2CH <sub>2</sub> O → H <sub>2</sub> S+2HCO <sub>3</sub> <sup>-</sup><br>(H <sub>2</sub> S is dissociated into HS <sup>-</sup> +H <sup>+</sup> )<br>H <sub>2</sub> S+Me <sup>2+</sup> → MeS ↓+2H <sup>+</sup><br>(Me <sup>2+</sup> refers to heavy metal ions) | Chen et al., 2016,<br>2021c; Sanchez,<br>2017; Urakawa et al.,<br>2017 |
|   |                     | Firmicutes                     | Desulfotomaculum,<br>Desulfosporosinus   |  | Urakawa et al., 2017;<br>Habe et al., 2020                             |
|   |                     | Aquificae                      | Desulfurobacterium   |  | Urakawa et al., 2017   |
|   | Others              | Actinobacteria, Proteobacteria | Cellulomonas,<br>Propioniciclava,<br>Geobacter   |  | Chen et al., 2021b,c   |
| Fe <sup>2+</sup>  | Proteobacteria      |                                | Thiomonas,<br>Sideroxydans   | Bio-oxidation  | Chen et al., 2021c   |
| Cd <sup>2+</sup> , Zn <sup>2+</sup>   | Proteobacteria      |                                | Serratia, Pseudomonas  | Biosorption/Cellular sequestration   | Yu et al., 2020  |
| Cu <sup>2+</sup>  | Fusobacteria, Baci  | teroidetes, Proteobacteria     | Hydrogenophaga,<br>Rheinheimera  | Desistant to have a matcle   | Guo et al., 2021   |
| Cd <sup>2+</sup>  | Proteobacteria, Ba  | acteroidetes                   | Pseudomonas,<br>Chryseobacterium   | Resistant to neavy metals  | Zhang et al., 2021e  |
| Fe <sup>2+</sup> , Se <sup>4+</sup>   | Firmicutes, Proteol | bacteria                       | Bacillus, Planococcus,<br>Pseudomonas  | Support plant growth   | Vassallo et al., 2020  |
| Zn <sup>2+</sup> , Ni <sup>2+</sup> , Cd <sup>2+</sup>  | Proteobacteria      |                                | Herminiimonas,<br>Methylophilus,<br>Cupriavidus,<br>Novosphingobium  |  | Syranidou et al.,<br>2018  |

microorganisms, albeit over a longer period. Most likely, the microbial community structure in the system was spontaneously altered, facilitating resistance to heavy metal stress. In contrast, systems inoculated with resistant microorganisms can experience a less pronounced microbial community evolution to obtain a dominant strain when encountering environments with heavy metals, saving time for biofilm stabilization (Rahman, 2020).

The interactions between microorganisms and plants also greatly affects the removal of heavy metals (González Henao and Ghneim-Herrera, 2021). Microorganisms and plants have long been growing together and microorganisms have more or less established associations with plants (Vassallo et al., 2020). In particular, rhizobacteria and endophytic bacteria can support plant growth through uptake of nutrients (e.g., N, P, Mg, Fe, and Ca), biosynthesis of plant growth-promoting phytohormones, and tolerance to pollution and stress (Syranidou et al., 2018; Vassallo et al., 2020). This can alleviate the toxic stress of heavy metals on plants and can also facilitate the accumulation of heavy metals by plants (Syranidou et al., 2018; Vassallo et al., 2020). In turn, the main function of plants in CWs is to provide additional oxygen and organic matter for microbial growth (Zhou et al., 2013). Thus, good plant growth also provides a more suitable environment for microbial growth (Sturz et al., 2000). This mutualistic interaction facilitates heavy metal removal by CWs. Syranidou et al. (2016) found that inoculation of the *Juncus acutus* with a selected endophytic bacterial consortium removed emergent pollutants and metals faster and more efficiently compared to uninoculated plants. Similarly, Vassallo et al. (2020) isolated eight bacterial strains (belonging to the genera *Bacillus, Planococcus*, and *Pseudomonas*) from samples taken from the roots of *Phragmites australis*. They grew well in wastewater with high concentrations of heavy metals (45 mg/l for Fe and 0.09 mg/l for Se), and the higher the concentration of heavy metals, the faster they grow (Vassallo et al., 2020). In conclusion, rhizobacteria and endophytic bacteria have been shown to be reliable functional microorganisms for heavy metal removal as they have sufficient resistance to heavy metals and can enhance phytoremediation efficacy.

# FUNCTIONAL MICROORGANISMS IN ANTIBIOTIC REMOVAL

Antibiotics are compounds that inhibit the growth of microorganisms (Chen et al., 2020; Xu et al., 2020). They are widely used in human and animal medicine and as animal growth promotors (Chen et al., 2020; Xu et al., 2020). Based on previous studies, antibiotics are now widely present in

aquatic environment and that their presence and persistence often cause toxic effects, posing a significant threat to humans, animals, and aquatic habitats (Huang et al., 2017; Shan et al., 2020; Lu et al., 2021). Antibiotic contamination can also lead to the spread of resistance genes, thereby increasing the resistance of microorganisms and reducing the therapeutic potential against human and animal pathogens (Troiano et al., 2018; Chen et al., 2020; Shan et al., 2020). It has been reported that CWs are highly suitable for antibiotic removal, with removal efficiencies as high as 91.8 to 99.5% (Xu et al., 2020). The removal of antibiotics in CWs has undergone a series of complex physical, chemical and biological processes, such as adsorption, precipitation, and microbial degradation (Wang et al., 2019a). Among them, microorganisms are considered to be the driving force for the degradation of antibiotics in CWs (Wang et al., 2019a; Shan et al., 2020; Zheng et al., 2021b). The main functional microorganisms involved in antibiotics removal in CWs are summarized in Table 4.

Sulfonamides (SAs), including sulfamethoxazole (SMX) and sulfadiazine (SDZ), are widely used in animal agriculture and human health care and are the most common residual antibiotics in almost all environmental compartments (Ouyang et al., 2021). They can significantly inhibit bacterial populations, such as Desulfarculus, denitrifying bacteria, and Syntrophobacter, affecting the sulfur and nitrogen cycles (Man et al., 2020). Microbial-mediated degradation can significantly contribute to the removal of SAs in CWs, both under aerobic and anaerobic conditions (Chen et al., 2020). For example, under aerobic conditions, Bacillus and Geobacter can degrade SAs (Chen et al., 2020). The genus Bacillus, belonging to the phylum Firmicutes, can be enriched under SAs stress, degrading SMX to  $NH_4^+$  and further to  $NO_3^-$  (Liu et al., 2018). Geobacter, a member of the phylum Proteobacteria, is considered a potential SDZ degrader (Zhang et al., 2019a). The genus Microbacterium in the phylum Actinobacteria can also use SMX as the sole carbon source under aerobic conditions (Sauvetre et al., 2020; Ouyang et al., 2021). The molecular mechanism of SMX catabolism by Microbacterium is initiated by ipso-hydroxylation, followed by NADH-dependent hydroxylation of the carbon atom attached to the sulfonyl group, which leads to the release of sulfite, 3-amino-5-methylisoxazole, and benzoguinone imine, of which the latter is converted to 4-aminophenol (Ricken et al., 2013). As for the genus Bradyrhizobium can accelerate SAs removal under anaerobic conditions (Chen et al., 2020). In CWs, the three main phyla involved in the degradation of SAs are Proteobacteria, Actinobacteria and Firmicutes, and some key genera, such as Pseudomonas, may metabolize glucose and subsequently attenuate SMX by co-metabolism of organic matter and SMX (Zheng et al., 2021b). In addition, Desulfovibrio also plays a key role in SMX transformation and can transform SMX alone (Ouyang et al., 2021).

In addition to SAs, common antibiotics include fluoroquinolones (FQ) and cephalosporin (CP), two of which are the most widely used antimicrobials drugs worldwide (Alexandrino et al., 2017). Among them, FQ include ciprofloxacin (CIP), ofloxacin (OFL), and enrofloxacin (ENR); CP include ceftiofur (CEF). For these types of antibiotics, Amorim et al. (2014) investigated the soil bacterium *Labrys portucalensis F11* in minimal medium supplemented with acetate as an additional carbon source and demonstrated its ability to degrade a range of FQ (e.g., CIP). Similarly, Lin et al. (2016) suggested that the genus *Arthrobacter* can dissipate FQ (e.g., OFL) as an additional carbon and energy source. The genus *Dysgonomonas* has also been shown to biodegrade ENR and CEF (Alexandrino et al., 2017).

In addition, the first Watch List of the EU Water Framework Directive [European Commission (EC)2015] identifies the antiinflammatory diclofenac (DCF) and the antibiotic SMX as two emerging contaminants (Sauvetre et al., 2020). For DCF, Bessa et al. (2017) demonstrated that strain *Brevibacterium* sp. *D4* could biodegrade 35% of 10 mg/l of DCF as the sole carbon source. Moreira et al. (2018) also reported that the bacterial strain *Labrys. portucalensis F11* degraded 70% of 34  $\mu$ M of DCF, supplied as the sole carbon source, after 30 days of cultivation. Regarding the microorganisms associated with the degradation of other antibiotics, such as ampicillin, tetracycline, triclosan (TCS), and ceftiofur (CEF) are listed in **Table 4**.

As shown in **Table 4**, the vast majority of functional microorganisms related to antibiotic removal belong to the *Proteobacteria*, *Acidobacteria*, and *Bacteroidetes* phyla, probably due to the presence of degradation genes (Liu et al., 2019). According to Huang et al. (2017), the phylum most significantly related to antibiotic removal is *Proteobacteria* phylum, followed by *Bacteroidetes* and *Actinobacteria*. Of these, *Betaproteobacteria* of the phylum *Proteobacteria* have been shown to be effective in addressing the global antibiotic resistance issue (Alexandrino et al., 2017; Shan et al., 2020).

However, functional microorganisms may develop antibiotic resistance in the process of degrading antibiotics and may even cover functions other than antibiotic resistance (Alonso et al., 2001; Troiano et al., 2018). Antibiotic resistance may be inherent to microorganisms or may arise through horizontal gene transfer from donor bacteria, phages, or free DNA (Alonso et al., 2001; Troiano et al., 2018; Santos et al., 2019a). An increase in antibiotic resistant microorganisms may lead to a decrease in the therapeutic potential of antibiotics, thus making it more difficult to treat microorganisms' infections (Santos et al., 2019a). Notably, an induction of antibiotic resistance genes has been reported with the effective removal of antibiotics by CWs (Liu et al., 2019). Therefore, it is a great challenge for CWs to avoid the induction of antibiotic resistance genes while effectively removing antibiotics.

## **EMERGING POLLUTANTS**

In addition to the above four typical pollutants, CWs are also used to remove some emerging pollutants, such as hormones, pesticides, food additives, flame retardants, nanoparticles, and persistent organic pollutants (e.g., polychlorobiphenyls and polycyclic aromatic hydrocarbons; Rajan et al., 2019; Vymazal et al., 2021; Yin et al., 2021). Biodegradation is generally considered as one of the important processes responsible for these emerging pollutants removal

### TABLE 4 | Functional microorganisms in CWs for antibiotic removal.

| Antibiotic Category |     | Phyla                            | Genera (Notes)   | Removal principle                                     | References  |  |
|---------------------|-----|----------------------------------|--|---|---|--|
|                     |     |                                  | Sphingomonas,<br>Bradyrhizobium,<br>Sphingorhabdus, Reyranella,<br>Ochrobactrum, Sphingobium,<br>Hyphomicrobium<br>(Belongs to<br>Alphaproteobacteria)<br>Acidovorax, Ralstonia, | Biodegradation or use of antibiotics as carbon source | Syranidou et al., 2017; Man et al., 2020;<br>Sauvetre et al., 2020; Zheng et al.,<br>2021b        |  |
|                     |     | Proteobacteria                   | Azonexus<br>(Belongs to  |   | Syranidou et al., 2017; Chen et al., 2020   |  |
| SAs                 | SMX |                                  | Betaproteobacteria)<br>Desulfovibrio (Belongs to<br>Deltaproteobacteria)<br>Pseudomonas, Luteimonas,<br>Enterobacter, Acinetobacter  |   | Ouyang et al., 2021   |  |
|                     |     |                                  | (Belongs to<br>Gammaproteobacteria)  |   | 2021; Zheng et al., 2021b   |  |
|                     |     | Actinobacteria                   | Rhodococcus,<br>Microbacterium,<br>Arthrobacter, Gordonia,<br>Nocardioides, Streptomyces   |   | Syranidou et al., 2017; Wegrzyn and<br>Felis, 2018; Sauvetre et al., 2020;<br>Ouyang et al., 2021 |  |
|                     |     | Firmicutes                       | Bacillus, Virgibacillus  |   | Syranidou et al., 2017; Chen et al., 2020   |  |
|                     | SDZ | Proteobacteria                   | Geobacter (Belongs to<br>Deltaproteobacteria)<br>Labrys, Bradyrhizobium  |   | Chen et al., 2020   |  |
|                     | CIP | proteobacteria                   | (Belongs to<br>Alphaproteobacteria)<br>Pseudoxanthomonas   |   | Syranidou et al., 2017; Santos et al.,<br>2019b; Chen et al., 2020                                |  |
|                     | 01  |                                  | (Belongs to  |   | Santos et al., 2019b  |  |
| FQ                  |     | Actinobacteria,<br>Bacteroidetes | Nocardioides,<br>Dysgonomonas<br>Rhizobacter,  |   | Alexandrino et al., 2017; Santos et al.,<br>2019b   |  |
|                     | OFL | Proteobacteria                   | Uliginosibacterium<br>(Belongs to<br>Betaproteobacteria)   |   | Lu et al., 2021   |  |
|                     |     | Actinobacteria,<br>Bacteroidetes | Arthrobacter, Bacteroides  |   | Tong et al., 2020; Lu et al., 2021  |  |
|                     | ENR | Bacteroidetes                    | Alkaliflexus, Dysgonomonas   |   | Alexandrino et al., 2017; Santos et al.,<br>2019b   |  |
|                     |     | Proteobacteria                   | Labrys, Sphingobium<br>(Belongs to<br>Alphaproteobacteria)   |   | Wegrzyn and Felis, 2018; Sauvetre et al., 2020  |  |
| DCF                 |     | Actinobacteria                   | Microbacterium,<br>Brevibacterium,<br>Streptomyces   |   | Wegrzyn and Felis, 2018; Sauvetre et al., 2020  |  |
|                     |     | Fungi                            | Trametes<br>Luteimonas,  |   | Wegrzyn and Felis, 2018   |  |
| Ampicillin          |     | Proteobacteria                   | Pseudoxanthomonas<br>(Belongs to   |   | Santos et al., 2019b; Zheng et al.,<br>2021b  |  |
| Tetracycline        |     | Proteobacteria                   | Gammaproteobacteria)<br>Novosphingobium<br>Pseudomonas, Alcaligenes,<br>Stenotrophomonas,  |   | Santos et al., 2019b; Lu et al., 2021   |  |
| TCS                 |     | Proteobacteria                   | <i>Methylococcales</i> (order)<br>(Belongs to  |   | Liu et al., 2016  |  |
| CEF                 |     | Bacteroidetes                    | Gammaproteobacteria)<br>Dysgonomonas   |   | Alexandrino et al., 2017  |  |

(Vymazal et al., 2021). Zhang et al. (2021e) found that Firmicutes, Clostridia, and Acetobacterium were able to tolerate abiotic stresses and thus degrade chlorpyrifos into carbon sources. Liu et al. (2020b) demonstrated that Pseudomonas, Duganella, and Sphingobium are resistant to the threat of organophosphate flame retardants [tris (2-chloroethyl) phosphate, tris (1-chloro-2-propyl) phosphate, and tricresyl phosphate] and have the ability to biodegrade. Ahmad et al. (2019) also showed that various genera, such as Flavobacteriaceae, Novosphingobium, and Mycobacterium can degrade polycyclic aromatic hydrocarbons in a diverse environment. However, there is still relatively little research on these emerging pollutants removed by microorganisms in CWs, so this section is not the focus of this review and more research is needed in the future to focus on the degradation and removal mechanisms of these emerging pollutants by microorganisms.

## MICROBIAL ALPHA DIVERSITY ANALYSIS

In addition to the composition and structure of microorganisms, the diversity of microbial communities can also influence the performance of CWs in removing pollutants (Zhang et al., 2019b). Therefore, this review explored the effect of four typical pollutants on the microbial diversity of CWs by counting the values of microbial diversity in different studies. As alpha diversity can reflect the species diversity of microbial communities within a given region, this review has chosen to represent the diversity of microbial communities through the value of alpha diversity (Pitacco et al., 2019; Laliberte et al., 2020).

To better reflect the influence of pollutant concentration on microbial diversity, the pollutants nitrogen and phosphorus were classified into high and low concentrations. Since CWs are generally used for deep treatment of the tailwater of wastewater treatment plants, wastewater with total nitrogen concentrations exceeding 20 mg/l and total phosphorus concentrations exceeding 1.5 mg/l was designated as high concentrations with reference to the Discharge Standard of Pollutants for Municipal Wastewater Treatment Plant (GB 18918–2002) - level B standard. **Table 5** shows the specific values of microbial alpha diversity in CWs considering the four typical pollutants (nitrogen, phosphorus, heavy metals, and antibiotics). To better visualize and compare, box plots (**Figure 5**) were generated using the median value of alpha diversity in **Table 5** as the data.

## EFFECTS OF NITROGEN AND PHOSPHORUS ON ALPHA DIVERSITY

As shown in **Figure 5**, nitrogen and phosphorus have similar effects on alpha diversity, so we put them together for analysis. Regarding richness, at high concentrations of nitrogen and phosphorus, both the Chao1 index and the ACE index were significantly lower than at low concentrations, suggesting

that microbial richness is severely reduced in environments with high concentrations of these pollutants. Most likely, microorganisms not involved in pollutant removal are eliminated or suppressed in these extreme environments, resulting in fewer microbial species (Xiao et al., 2020). The closer data for high concentrations of nitrogen and phosphorus also indicate that certain specific microbial species may have formed, resulting in similar richness levels in different studies. On the contrary, at low concentrations of nitrogen and phosphorus, although the values of both Chao1 and ACE indexes were higher, the data were more scattered. This may be due to the dominance of other influencing factors, such as C/N ratio (Jia et al., 2021) and substrate type (Ajibade et al., 2021).

Regarding diversity, it is evident from the Simpson index that high concentrations of nitrogen lead to a decrease in diversity, i.e., to the emergence of dominant populations. This corresponds with the richness analysis. Apparently, high nitrogen concentrations facilitate the growth of microorganisms associated with nitrogen cycling and suppress the growth of microorganisms of other functions (Xiao et al., 2020). Interestingly, the Shannon index was high but the Simpson index was low for high phosphorus concentrations, with the opposite pattern for low phosphorous levels. This may be explained by the high sensitivity of the Shannon index to community richness (Zhang et al., 2019b). Low phosphorous concentrations allow for a consistently high level of species diversity, reflecting no large differences in abundance among microorganisms with different functions in the absence of significantly prominent contaminants.

The combined richness and diversity indices show that high concentrations of nitrogen and phosphorus lead to varying reductions in both indices, reflecting the presence of significantly dominant populations. These populations are the functional microorganisms that play an important role in the removal of nitrogen and phosphorus.

# EFFECTS OF HEAVY METALS ON ALPHA DIVERSITY

Heavy metals can significantly decrease microbial richness. This can be explained by the toxicity of heavy metals to microorganisms and the inability of microorganisms to directly degrade heavy metals (Bianchi et al., 2020). In this sense, the presence of heavy metals had a selection effect on microorganisms in CWs, and microbial succession occurred in CWs over time, where the enrichment and structural optimization of dominant species may lead to the reduction in richness and diversity (Xiao et al., 2021). However, in contrast to the case of high concentrations of nitrogen and phosphorus, the difference between the data for heavy metals is larger, resulting in a larger box in the box plot. The most likely reasons for this are that the toxicity of different types of heavy metals to microorganisms may vary and that different microorganisms have different resistance levels to heavy metals.

### TABLE 5 Alpha diversity of microorganisms in CWs treated different pollutant.

| Alpha<br>Diversity           | Rich   | ness   | Diversity                                      |  | Concentration of pollutants (mg/l)   | References                             |
|------------------------------|--|--|--|--|--|--|
| Pollutant type               | Chao1  | ACE  | Shannon  | Simpson                                    |  |  |
|                              | 266.500 ± 62.500                             | 269.000 ± 59.000                                 | 3.39500 ± 0.58500                              |  | $NH_{4^{+}} = 35.000, TN = 40.000$<br>$NO_{4^{-}} = 4.395 \pm 0.695$   | Zhao et al., 2016                      |
|                              | 621.250 ± 115.150                            |  | 4.07000 ± 0.71000                              |  | $NH_{4^+} = 79.945 \pm 1.805,$<br>$TN = 87.100 \pm 2.620$  | Li et al., 2019                        |
|                              | 841.500 ± 163.500                            | 1048.950 ± 190.230                               | $6.72500 \pm 0.17500$                          | 0.96814 ± 0.00316                          | NH <sub>4</sub> <sup>+</sup> =89.200   | Xu et al., 2021                        |
|                              | 1237.117 ± 230.722                           | 1224.746 ± 245.1905                              | 6.66750 ± 1.25450                              | $0.96100 \pm 0.02800$                      | $NH_4^+ = 18.000, NO_3^- = 6.000,$<br>TN = 24.000  | Xia et al., 2020                       |
|                              | 1296.915 ± 59.345<br>1375.965 ± 257.945      | 1292.191 ± 62.723<br>1422.860 ± 244.730          | 5.27000 ± 0.17400<br>5.78500 ± 0.78500         | 0.98615 ± 0.00345                          | $NO_{3}^{-} = 50.000$<br>$NH_{4}^{+} = 226.284$  | Zhao et al., 2019<br>Lv et al., 2021   |
|                              | 1586.500 ± 214.500                           |  | 7.67250 ± 0.13250                              |  | $NH_4^+ = 20.140 \pm 0.420,$<br>$NO_3^- = 39.290 \pm 0.730$  | Deng et al., 2019                      |
| High<br>concentration        | 1725.500 ± 108.500                           | 1725.500 ± 108.500                               | 7.84500 ± 0.09500                              | 0.97331 ± 0.00690                          | $NH_{4^{+}} = 115.000,$<br>$NO_{3^{-}} = 182.000$  | Zhao et al., 2020d                     |
| of nitrogen                  | 1755.075 ± 92.235                            |  | 7.68319 ± 0.43451                              | 0.97313 ± 0.00814                          | $NH_{4^+} = 26.250 \pm 11.250,$<br>$NO_{3^-} = 10.500 \pm 4.500,$<br>$TN = 43.750 \pm 18.750$                                    | Xiao et al., 2020                      |
|                              | 2001.110 ± 883.090                           | 2091.605 ± 913.475                               | 7.15500 ± 2.15500                              | 0.89171 ± 0.10240                          | $NH_{4^{+}} = 112.580$   | Wang et al.,<br>2021b                  |
|                              | 2274.150 ± 56.660                            |  | 7.76500 ± 0.20500                              | 0.97500 ± 0.00500                          | $NH_{4^{+}} = 15.170 \pm 0.804,$<br>$TN = 21.910 \pm 1.190$  | Huang et al., 2018                     |
|                              | 2299.615 ± 366.595                           |  | 5.31000 ± 0.42100                              | $0.96165 \pm 0.02155$                      | $NH_{4^+} = 20.000, NO_{3^-} = 1.200,$<br>TN = 35.000  | Yang et al., 2021                      |
|                              | 4398.800 ± 804.200                           | 4960.050 ± 535.150                               | 5.48500 ± 1.35500                              | 0.93350 ± 0.06250                          | $NH_{4^{+}} = 29.900, TN = 39.000$   | Zheng et al., 2020                     |
|                              | 4422.000 ± 315.000                           |  | 6.51850 ± 0.25450                              | 0.99235 ± 0.00295                          | $NH_{4^{+}} = 18.680, TN = 60.360$   | Pelissari et al.,<br>2018              |
|                              | 2133.745 ± 127.930                           | 2144.504 ± 139.730                               | 5.84550 ± 0.15450                              | 0.99150 ± 0.00250                          | $NH_{4^+} = 5.315 \pm 2.345,$<br>$TN = 13.425 \pm 5.785$<br>$NH_{4^+} = 1.290 \pm 0.020,$  | Li et al., 2021a                       |
|                              | 2346.600 ± 216.230                           | 2330.600 ± 228.940                               | 6.28000 ± 0.23000                              |  | $\label{eq:NO_3^-} \begin{split} &NO_3^- = 7.380 \pm 0.130, \\ &NO_2^- = 0.110 \pm 0.010, \\ &TN = 14.680 \pm 0.250 \end{split}$ | Jia et al., 2021                       |
|                              | 2468.095 ± 837.095                           | 2602.635 ± 971.635                               | 8.0350 ± 1.0650                                | 0.94488 ± 0.03924                          | TN=20.000  | Huang et al.,<br>2019a                 |
|                              | 2686.500 ± 317.500                           | 2655.500 ± 303.500                               | 6.24500 ± 0.74500                              | 0.97360 ± 0.02460                          | $NH_{4^{+}} = 0.170, TN = 2.480$   | Wang et al.,<br>2020c                  |
| Low                          | 3119.050 ± 215.950                           | 3110.500 ± 252.140                               | $5.60500 \pm 0.27500$                          | $0.98225 \pm 0.00595$                      | $NO_{3}^{-} = 12.000, NH_{3}^{+} = 8.000$  | Ajibade et al.,<br>2021                |
| concentration<br>of nitrogen | 3574.000 ± 75.000                            |  | 10.89500 ± 0.03500                             | 0.99874 ± 0.00002                          | $NH_{4^+} = 4.000, NO_3^- = 10.000$<br>$NH_{4^+} = 1.630 \pm 0.090,$   | Qin et al., 2021                       |
|                              | 4592.500 ± 269.500                           | 4765.500 ± 259.500                               | 6.47000 ± 0.22000                              | 0.98900 ± 0.00100                          | $NO_3^- = 10.410 \pm 1.660,$<br>TN = 12.680 ± 1.320  | Tong et al., 2019                      |
|                              | 4932.250 ± 175.350                           |  | 9.80000 ± 0.33000                              | 1.00000 ± 0.00000                          | NH <sub>4</sub> <sup>+</sup> = 9.110, NO <sub>3</sub> <sup>-</sup> = 9.530,<br>TN = 19.050                                       | Wei et al., 2020                       |
|                              | 6924.040 ± 1255.720                          | 8110.050 ± 935.750                               | $7.16000 \pm 0.20000$                          | $0.99730 \pm 0.00170$                      | $NH_{4}^{+} = 0.960$   | Zhao et al., 2020c                     |
|                              | 4393.430                                     |  | 9.00625  | 0.99018                                    | $NH_{4^{+}} = 1.500, NO_{3^{-}} = 10.500$  | Zhang et al.,<br>2021e                 |
|                              | 7972.000 ± 186.000                           |  | 6.81750 ± 0.35750                              | 0.92500 ± 0.01500                          | $NH_{4^+} = 2.408 \pm 2.350,$<br>$NO_3^- = 1.885 \pm 0.925,$<br>$NO_3^- = -0.105 \pm 0.091$                                      | Ma et al., 2018                        |
| High                         | 621.250 ± 115.150                            |  | 4.07000 ± 0.71000                              |  | $TP = 10.525 \pm 0.715$  | Li et al., 2019                        |
| concentration                | 841.500 ± 163.500                            | 1048.950 ± 190.230                               | $6.72500 \pm 0.17500$                          | 0.96814 ± 0.00316                          | $PO_4^{3-} = 44.000$   | Xu et al., 2021                        |
| of phosphorus                | $1375.965 \pm 257.945$<br>1725.500 + 108.500 | $1422.860 \pm 244.730$<br>$1725.500 \pm 108.500$ | $5.78500 \pm 0.78500$<br>$7.84500 \pm 0.09500$ | 0.97331 + 0.00690                          | $PO_4^{3-} = 19.554$<br>$PO_4^{3-} = 4.387$  | Lv et al., 2021<br>Zhao et al., 2020d  |
|                              | 1755.075 ± 92.235                            |  | 7.68319 ± 0.43451                              | $0.97313 \pm 0.00813$                      | $TP = 5.250 \pm 2.250$   | Xiao et al., 2020                      |
|                              | 1788.145 ± 157.145                           |  | $7.47000 \pm 0.50000$                          | 0.95000 ± 0.04000                          | TP=3.000   | Huang et al.,<br>2020a                 |
|                              | 2001.110 ± 883.090                           | 2091.605 ± 913.475                               | 7.15500 ± 2.15500                              | 0.89171 ± 0.10240                          | $PO_4^{3-} = 17.500$   | Wang et al., 2021b                     |
|                              | 2274.150 ± 56.660<br>2299 615 + 366 595      |  | $7.76500 \pm 0.20500$<br>5.31000 + 0.42100     | $0.97500 \pm 0.00500$<br>0.96165 + 0.02155 | $IP = 2.810 \pm 0.170$<br>TP = 5.000   | Huang et al., 2018<br>Yang et al. 2021 |
|                              | $2468.095 \pm 837.095$                       | 2602.635 ± 971.635                               | 8.03500 ± 1.06500                              | $0.94488 \pm 0.03924$                      | TP=3.000   | Huang et al., 2019a                    |
|                              | 4398.800 ± 804.200                           | 4960.050 ± 535.150                               | 5.48500 ± 1.35500                              | 0.93350 ± 0.06250                          | $TP = 3.600 \pm 0.900$   | Zheng et al., 2020                     |
|                              |  |  |  |  |  | (Continued)                            |

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### TABLE 5 | Continued

| Alpha<br>Diversity             | Richness  |  | Diversity  |  | Concentration of pollutants (mg/l)   | References  |
|--------------------------------|---|--|--|--|--|---|
| Pollutant type                 | Chao1   | ACE  | Shannon  | Simpson                                |  |   |
|                                | 4592.500 ± 269.500<br>2346.600 ± 216.230<br>2133.745 ± 127.930              | 4765.500 ± 259.500<br>2330.600 ± 228.940<br>2144.504 ± 139.730   | 6.47000 ± 0.22000<br>6.28000 ± 0.23000<br>5.84550 ± 0.15450            | 0.98900 ± 0.00100<br>0.99150 ± 0.00250 | $PO_4^{3-} = 0.800 \pm 0.070$<br>TP = 0.210 ± 0.010<br>TP = 0.505 ± 0.325  | Tong et al., 2019<br>Jia et al., 2021<br>Li et al., 2021a   |
| Low                            | 2686.500 ± 317.500  | 2655.500 ± 303.500   | 6.24500 ± 0.74500  | 0.97360 ± 0.02460                      | TP=0.260   | Wang et al.,  |
| concentration<br>of phosphorus | 7492.815 ± 1030.225<br>2355.050 ± 745.750<br>1796.650 ± 461.450<br>4393.430 | 10929.450 ± 1071.540<br>2442.000 ± 656.000<br>2269.500 ± 380.500 | 5.74500 ± 0.46500<br>4.95600 ± 0.93300<br>4.03650 ± 0.52850<br>9.00625 | 0.98350 ± 0.00890<br>0.99018           | TP=0.550 ± 0.300<br>TP=1.200 ± 0.300<br>TP=1.500   | Kang et al., 2017<br>Wang et al.,<br>2020a<br>Zhang et al., |
| Heavy Metal                    | 7972.000 ± 186.000<br>961.900   | 969.000  | 6.81750 ± 0.35750<br>7.08000   | 0.92500 ± 0.01500<br>0.97900           | $PO_4^{3-} = 0.407 \pm 0.025$<br>Control group (NH <sub>4</sub> <sup>+</sup> = 114.600, PO <sub>4</sub> ^{3-} = 17.900)  | 2021e<br>Ma et al., 2018<br>Liu et al., 2020a               |
|                                | 510.000 ± 109.600   | 527.600 ± 108.600  | 4.79950 ± 1.34850  | 0.87500 ± 0.09600                      | Ni=2.000, 5.000, 10.000,<br>30.000   |   |
|                                | 5296.025 ± 164.585  |  | 9.86543 ± 0.10423  | 0.99410 ± 0.00457                      | Control group (NH <sub>4</sub> <sup>+</sup> = 141.580,<br>NH <sub>2</sub> <sup>-</sup> = 17.140, NO <sub>3</sub> <sup>-</sup> = 43.330,<br>PO <sub>4</sub> <sup>3-</sup> = 13.170) | Xiao et al., 2021   |
|                                | 2443.830 ± 238.130<br>516.425 ± 201.145                                     | 509.965 ± 192.635  | 8.89639 ± 0.67463<br>5.14000 ± 0.43000                                 | 0.99107 ± 0.00491<br>0.90500 ± 0.01500 | Ni = 0.100, 1.000<br>Control group (NO <sub>3</sub> <sup>-</sup> = 50.500,<br>NH <sub>4</sub> <sup>+</sup> = 75.428, PO <sub>4</sub> <sup>3-</sup> = 6.581)                        | Wang et al.,<br>2020b                                       |
|                                | 532.555 ± 132.355<br>3186.590 ± 456.720                                     | 545.470 ± 135.490<br>3177.470 ± 474.670                          | 5.32000 ± 0.25000<br>6.56500 ± 0.36500                                 | 0.93000 ± 0.02000<br>0.99425 ± 0.00325 | Zn = 24.769<br>Control group (NO <sub>3</sub> <sup>-</sup> = 4.195, NH <sub>4</sub> <sup>+</sup> = 2.904, PO <sub>4</sub> <sup>3-</sup> = 1.727)                                   | Wang et al.,<br>2021a                                       |
|                                | 2602.910 ± 851.850<br>1288.690 ± 581.040                                    | 2621.995 ± 887.785<br>1353.650 ± 622.490                         | 6.03500 ± 0.36500<br>7.69500 ± 1.47500                                 | 0.98605 ± 0.00755<br>0.96500 ± 0.02500 | $\label{eq:cr} \begin{split} &Cr = 0.100  mmol/l \\ &Control  group  (NO_3^- = 50.500, \\ &NH_4^+ = 75.428,  PO_4^{3-} = 6.581) \end{split}$                                       | Zhao et al., 2020a  |
|                                | 1353.210 ± 806.080<br>73.350 ± 42.150                                       | 1327.970 ± 478.900   | 7.12500 ± 1.77500<br>2.76750 ± 0.72550                                 | 0.93000 ± 0.06000<br>0.88450 ± 0.09650 | Pb=5.000<br>As=20.000, Zn=15.000   | Arroyo et al., 2013   |
|                                | 1962.290 ± 48.290   | 2032.945 ± 118.605   | 7.48500 ± 0.75500  | 0.96200 ± 0.01600                      | $\begin{array}{l} Cu = 2.000 \pm 0.170, \\ Zn = 4.000 \pm 0.210, \\ Cd = 0.100 \pm 0.010, \\ Co = 2.000 \pm 0.230, \\ Ni = 0.500 \pm 0.240, \\ Pb = 0.500 \pm 0.270 \end{array}$   | Si et al., 2019   |
|                                | 3387.500 ± 461.500  | 3491.000 ± 453.000   | 7.12000 ± 0.82000  | 0.97500 ± 0.01500                      | Cu=4.880 $\pm$ 0.080,<br>Zn=5.060 $\pm$ 0.210,<br>Cd=5.170 $\pm$ 0.170,<br>Cr=5.650 $\pm$ 0.580  | Chen et al., 2021c  |
|                                | 1469.050 ± 182.450  | 1540.050 ± 182.300   | 7.44500 ± 0.39500  | 0.98500 ± 0.00500                      | Cr=0.500, 1.000, 2.000,<br>4.000, 8.000, 16.000  | Zhang et al.,<br>2019b                                      |
|                                | 4217.190<br>4019.520  |  | 8.77863<br>8.88151   | 0.98390<br>0.98791                     | Cd = 200.000 µg/l<br>Cd = 200.000 µg/l, chlorpyrifos<br>=200.000 µg/l  | Zhang et al.,<br>2021e                                      |
| Antibiotic                     | 4592.500 ± 269.500  | 4765.500 ± 259.500   | 6.47000 ± 0.22000  | 0.98900 ± 0.00100                      | Control group<br>$(NO_3^- = 10.410 \pm 1.660,$<br>$NH_4^+ = 1.630 \pm 0.090,$<br>$TN = 12.680 \pm 1.320,$<br>$PO_4^{3-} = 0.800 \pm 0.070)$  | Tong et al., 2019   |
|                                | 4088.500 ± 702.500  | 4184.000 ± 755.000   | 6.35000 ± 0.26000  | 0.98700 ± 0.00400                      | Ofloxacin = 0.100, 10.000,<br>1000.000 μg/l  |   |
|                                | 1415.915 ± 733.675  | 1471.010 ± 759.730   | 4.61000 ± 0.73000  | 0.95220 ± 0.03370                      | Control group (NO <sub>3</sub> <sup>-</sup> = 10.590,<br>NH <sub>4</sub> <sup>+</sup> = 20.630, TN = 32.340,<br>PO <sub>4</sub> <sup>3-</sup> = 4.310)                             | Zheng et al.,<br>2021b                                      |
|                                | 1794.070 ± 641.170  | 1805.390 ± 620.340   | 5.17000 ± 0.51000  | 0.98010 ± 0.00720                      | Sulfamethoxazole<br>=100.000 µg/l  |   |
|                                | 485.356 ± 186.811   | 497.745 ± 173.830  | 3.72600 ± 0.64200  | 0.93750 ± 0.02150                      | Control group (NO <sub>3</sub> <sup>-</sup> = 50.000,<br>NH <sub>4</sub> <sup>+</sup> = 76.420,<br>TN = 12.680 $\pm$ 1.320,<br>PO <sub>4</sub> <sup>3-</sup> = 6.600)              | Lu et al., 2021   |

(Continued)

TABLE 5 | Continued

| Alpha<br>Diversity | Richness                                 |                     | Diversity                              |  | Concentration of pollutants (mg/l)  | References                                  |
|--------------------|--|---------------------|--|--|---|---|
| Pollutant type     | Chao1                                    | ACE                 | Shannon                                | Simpson                                |   |   |
|                    | 509.085 ± 253.085                        | 528.041 ± 240.405   | 3.48900 ± 1.31900                      | 0.80000 ± 0.18100                      | Levofloxacin=0.100, 0.200,<br>0.300, 0.500, 1.000, 10.000,<br>100.000 µg/l  |   |
|                    | 3425.000 ± 275.000                       | 3420.000 ± 220.000  | 7.63000 ± 0.16000                      | 0.93450 ± 0.00850                      | Ciprofloxacin =<br>99.400 ± 8.300 µg/l,<br>Azithromycin =<br>1313.900 ± 63.600 µg/l,<br>Oxytetracycline<br>=972.350 ± 39.950 µg/l | Wang et al., 2019a                          |
|                    | 6048.560 ± 1435.820<br>3804.525 ± 72.185 | 7528.120 ± 1563.730 | 6.06500 ± 0.53500<br>7.87500 ± 0.53500 | 0.98371 ± 0.00952<br>0.97140 ± 0.01420 | Triclosan = 60.000 µg/l<br>ciprofloxacin<br>hydrochloride = 50.000 µg/l,<br>Sulfamethoxazole = 50.000 µg/l                        | Liu et al., 2016<br>Yuan et al., 2020       |
|                    | 1748.150 ± 55.850<br>2987.970 ± 261.210  | 2965.010 ± 250.950  | 6.53450 ± 0.73150<br>6.38100 ± 0.30700 | 0.94500 ± 0.02800                      | Sulfadiazine = 4.000<br>Enrofloxacin =<br>46.550 ± 20.850 ng/l,<br>Sulfamethoxazole =1<br>37.600 ± 73.600 ng/l                    | Song et al., 2018<br>Huang et al.,<br>2019b |

Alpha diversity is mainly related to two factors. One is richness, i.e., the number of species; and the other is evenness, i.e., the relative abundance of different species. The Chao1 and ACE indices reflect richness; and the Shannon and Simpson indices are a combination of richness and evenness that reflects diversity. Higher values of these indices represent a higher richness or diversity of microbial communities.

Regarding the effect on diversity, although the median of heavy metals can be high, the data are scattered, with a relatively large gap between the maximum and the minimum values. The occurrence of low values is easy to understand because heavy metals screen and selectively enrich microbial communities (Jia et al., 2021; Xiao et al., 2021), and most studies on heavy metals select microorganisms isolated from plant roots or screened from wastewater with high concentrations of heavy metals as functional microorganisms, resulting in the formation of dominant populations and in a lower diversity index. The high values can be explained by the microbial community being under pressure from heavy metals and the species within the community all evolving towards high heavy metal resistance (Yu et al., 2020). As a result, the abundance of different species is gradually increasing and the community as a whole is more even, so that there are no clearly dominant populations. In addition, during the treatment of heavy metals, key microorganisms require a variety of other microorganisms to cooperate and complement them (Chen et al., 2021b), which can also lead to a higher diversity. For example, the section "Functional Microorganisms in Heavy Metal Removal" mentioned a variety of microorganisms that can enhance the metabolism of SRB or provide them with electron donors, thus enhancing the bacterial sulfate reduction process. Abed et al. (2018) suggested that although SRB play a key role in AMD remediation, they account for only a small fraction of the total bacteria in the CWs.

Since nitrogen and phosphorus are essential elements for CWs, we also counted the microbial alpha diversity of the control group. We found that after the addition of heavy metals, the richness and diversity of microbial community in CWs decreased to varying degrees. This suggests that heavy metals exert more selective pressure on microorganisms compared to nitrogen and phosphorus, resulting in lower richness and more pronounced dominant populations. The experiments by Zhang et al. (2021e) also demonstrated that the combined pollution of antibiotics and heavy metals had a greater effect on microbial richness than that of a single contaminant, resulting in a decrease in richness values; the higher values of diversity may be due to the variety of pollutants and the need for a wider variety of functional microorganisms to deal with these pollutants.

Overall, the addition of heavy metals causes different degrees of reduction in the alpha diversity of microbial communities in CWs; in particular, the effect on microbial richness is obvious. This is can be seen in **Table 5**. And the higher the concentration of heavy metals, the greater the impact on alpha diversity.

# EFFECT OF ANTIBIOTICS ON ALPHA DIVERSITY

Antibiotics, due to their mechanism of action, result in lower richness values appear in microbial communities (Tong et al., 2019). However, the median value of the richness of antibiotics is located at a high level. In this case, the increase in microbial richness may be attributed to antibiotics acting as signaling molecules that stimulate the metabolic activity of microorganisms and thereby stimulate the growth of certain microbial species (Li et al., 2021c). With prolonged incubation time, microorganisms were able to gradually adapt to the environment and accumulate,



(HN: high concentration of phosphorus; LP: low concentration of phosphorus; HM: heavy metal; Ant: antibiotic).

leading to an increase in the richness of the microbial community. This is similar to the effect of heavy metals on microbial richness, but the difference is that the functional microorganisms involved in antibiotic removal can use antibiotics as carbon sources, resulting in a higher microbial richness (Zheng et al., 2021b). Based on the data presented in **Table 5**, in some studies, after the addition of antibiotics, the richness values were even higher than those in the control group.

Regarding diversity, different antibiotics correspond to different functional microorganisms and therefore easily lead to the formation of dominant population (Yuan et al., 2020). This explains the low values of both diversity indices in the presence of antibiotics.

Overall, antibiotics, like heavy metals, can have a significant impact on microbial alpha diversity, resulting in large differences among the various studies considered here. This reflects the fact that microbial communities may both take longer to remain stable and acquire the corresponding resistance or ability to remove contaminants in the presence of antibiotics or heavy metals (Yu et al., 2020; Zheng et al., 2021b).

## CONCLUSION AND PERSPECTIVES

The microbial community, as an important component of CWs, plays a critical role in the removal of pollutants. According to the results of this review, research on microorganisms is gaining increased attention with the advancement of molecular bioanalysis techniques, and studies on microorganisms in CWs have gained considerable importance. This review provided a summary of the functional microorganisms involved in the removal of nitrogen, phosphorus, heavy metals, and antibiotics, the most frequently studied typical pollutants in CWs. This can help researchers to find links between functional microorganisms and pollutants, as well as facilitate the discovery of more relevant functional microorganisms. By summarizing the main functional microorganisms in CWs, we found that the phylum Proteobacteria is the dominant one, containing microorganisms with a wide range of functions. In addition, the phyla Bacteroidetes, Actinobacteria, and Firmicutes are also frequently detected in CWs. These functional microorganisms can remove pollutants from CWs by catalyzing chemical reactions, biodegradation, biosorption, and supporting plant growth, etc. The complexity of the microbial community structure and limitations of microbial analysis techniques make it difficult to draw other general conclusions. Regarding the effects of different pollutants on microbial diversity, we found that different microorganisms respond in different ways. When CWs contain high nitrogen and phosphorus levels, functional microorganisms associated with nitrogen and phosphorus removal become dominant in the system, and numerous cross-over phyla or genera of functional microorganisms have been identified. This indicates that research on nitrogen and phosphorous in CWs is advanced and that the removal mechanisms are well understood. In the case of heavy metals or antibiotics, the system can evolve microorganisms adapted to these substances. However, because studies on these two pollutants are scarce, the results cannot be generalized. Overall, heavy metals and high concentrations of nitrogen and phosphorus decrease both microbial richness and diversity in CWs, whereas antibiotics cause large fluctuations in alpha diversity.

Research on the microbial treatment of pollutants in CWs has achieved tremendous breakthroughs and advances with the development of various technologies, but some aspects deserve further investigations:

- Functional microorganisms in CWs should be studied and analyzed more frequently. They play an important role in the removal of pollutants but are often not the dominant microbes. Many studies have focused on the analysis of the overall microbial profile in the system, whereas research on the profile of functional microorganisms in the system is still lacking. In the future, it may be possible to focus on functional microorganisms and to investigate more deeply the composition and diversity of these microorganisms and the influence of different factors on their growth and development.
- Plant-microbe interactions are critical in the removal of contaminants in CWs, and many functional microorganisms associated with the removal of heavy metals, antibiotics, and organic pollution are rhizobacteria or endophytic bacteria isolated from plants. Therefore, further research on plant-associated microorganisms can deepen our understanding of the role of specific microorganisms and plants acting together in the removal of contaminants from CWs.
- There is a lack of studies on microorganisms involved in heavy metal, antibiotics, and some emerging pollutants (such as pesticides, flame retardants, and Polychlorobiphenyls) removal, and often, some specific strains are cultured or

isolated to improve the removal efficiency. Therefore, research on these aspects needs to be intensified.

• In the future, more emphasis should be placed on the study of microorganisms at the genetic level, determining their functional enzymes or functional genes. However, such an approach is dependent on the technological advances.

## AUTHOR CONTRIBUTIONS

JW was responsible for data curation and formal analysis. JW and YL wrote the manuscript. GY contributed to conceptualization, the formal analysis, and visualization. GW, ZZ, PL, YZ, KY, and SW provided feedback on the manuscript.

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