



Research article

Explicit molecular dynamics simulation studies to discover novel natural compound analogues as *Mycobacterium tuberculosis* inhibitors

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ARTICLE INFO

Keywords:

Tuberculosis
Natural compounds
Molecular dynamics simulation
Mycobacterium tuberculosis dethiobiotin Synthetase
Butein analogues

ABSTRACT

Tuberculosis (TB) is one of the dreadful diseases present globally. This is caused by *Mycobacterium tuberculosis*. *Mycobacterium tuberculosis* dethiobiotin synthetase (MtDTBS) is an essential enzyme in biotin biosynthesis and is an ideal target to design and develop novel inhibitors. In order to effectively combat this disease six natural compound (butein) analogues were subjected to molecular docking to determine their binding mode and the binding affinities. The resultant complex structures were subjected to 500 ns simulation run to estimate their binding stabilities using GROMACS. The molecular dynamics simulation studies provided essential evidence that the systems were stable during the progression of 500 ns simulation run. The root mean square deviation (RMSD) of all the systems was found to be below 0.3 nm stating that the systems are well converged. The radius of gyration (Rg) profiles indicated that the systems were highly compact without any major fluctuations. The principle component analysis (PCA) and Gibbs energy landscape studies have revealed that the comp3, comp5 and comp11 systems navigated marginally through the PC2. The intermolecular interactions have further demonstrated that all the compounds have displayed key residue interactions, firmly holding the ligands at the binding pocket. The residue Lys37 was found consistently to interact with all the ligands highlighting its potential role in inhibiting the MtDTBS. Our investigation further put forth two novel compounds (comp10 and comp11) as putative antituberculosis agents. Collectively, we propose six compounds as plausible inhibitors to curtail TB and further can act as scaffolds in designing new compounds.

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Received 13 August 2022; Received in revised form 18 January 2023; Accepted 26 January 2023

Available online 30 January 2023

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1. Introduction

One of the ancient infections that has been invading the mankind is the tuberculosis (TB) for nearly about 4000 years. Characteristically, the primary site of TB infection is lungs but can also be seen to infect brain, intestines, kidneys, or the spine [1]. *Mycobacterium tuberculosis* is the causative agent of this disease and spreads through air [1]. According to world health organization (WHO), 1.5 million deaths have occurred due to TB in 2020. This is second to COVID-19 in infectious deaths and 13th leading causes of mortalities (<https://www.who.int/news-room/fact-sheets/detail/tuberculosis>). These facts direct towards the urgency of finding potential drugs that could reverse the condition.

Due to the slow growth and dormancy of some of the bacilli, multitude of reports illuminate on treatments that should be a prolonged or the multidrug therapeutic approach to get rid of the bacteria from the pulmonary and extrapulmonary sites [2]. Besides this, the chemical features also contribute to the success of the treatment regime. It is reported that the drugs in combination can curtail the likelihood of failure and relapse [2]. Furthermore, two effective drugs are to be prescribed to counter the appearance of resistant strains [2].

Several drug have been reported to counter TB and WHO has classified the antiTB drugs into five classes [2,3]. The drug-susceptible tuberculosis agents are grouped into first class and the drugs with uncertain or limited efficacy are grouped into fifth class [2,3], with a possibility of a new group [4]. The drugs namely rifampicin was licensed to administer for humans in 1963. Later bedaquiline and delamanid, have received approval to be used against multi-drug resistant (MDR)-TB [5]. Due to their association with side effects, these drugs are prescribed to be used when no other treatment option is available [5].

In the current study, for discovering the potential inhibitors against *MtDTBS*, we have used various *in silico* methods such as molecular docking, molecular dynamics simulation (MDS) and essential dynamics. These studies have proven to be crucial in swiftly identifying the plausible inhibitors [6–12]. Particularly, a few studies report the use of *in silico* methods in identifying potential inhibitors for *Mt*. Mallavarapu et al., identified the inhibitors for *Mt* MraY (Rv2156c). The target model was built using the homology modelling technique and was then complexed with antibiotic muraymycin D2 (MD2) [13]. Computational methods were applied to discover potential candidates from Asinex database depending on the binding modes and interactions with *Mt* MraY-MD2 [13]. Singh et al., used pharmacophore-based screening, molecular docking and MDS methods to retrieve potential hits against *MtPknG* [14]. *In vitro* evaluation was performed on the identified hits and eventually recommended compound NRB04248 as the potential inhibitor [14]. Sundar et al., used computational drug repurposing approaches to retrieve plausible small molecules against *Mt* Protein kinase A (PknA) [15]. Pitaloka et al., discovered possible drugs against Enoyl-[acyl-carrier-protein] reductase (InhA) of *Mt* adapting computational techniques [16]. Another group has identified prospective inhibitors via *in silico* methods that included structure-based drug design [17].

Nature products have been pivotal in treating TB that includes peptide actinomycin and rapamycin. Several natural products have displayed superior pharmacokinetic properties than the synthetic compounds [18]. In general, there has been a great impact of natural products in the field of medicine and have served as the only means to treat injuries and diseases [19]. A research group has identified natural products isobavachalcone and isoneorautenol and a synthetic chromene as antimycobacterial agents [20]. Another study has reported flavonoids as potential agents [21]. A study has reported the use of marine natural products to selectively kill dormant *Mycobacterium tuberculosis* [22].

A well-known flavonoid, butein (2',3,4,4'-tetrahydrochalcone) is highly regarded for its wide therapeutic potential [23] and widely studied as an anticancer agent [24–27]. In a study, butein is reported to inhibit *Mycobacterium bovis* BCG [28] and has demonstrated anti-mycobacterial activity [29], however no findings are yet published with respect to the butein analogues for *MtDTBS* target. Therefore, in the current study, we have used the butein analogues that are available in the lab to determine its computational efficacy against *Mycobacterium tuberculosis* dethiobiotin synthetase (*MtDTBS*). This enzyme participates in the biosynthesis of biotin [30] and is a promising target to design new drug candidates [30].

2. Materials and methods

2.1. Selection of the protein

The protein for the current study is *MtDTBS* (PDB 6CZD). This is an X-ray crystal structure with 2.40 Å resolution [31]. The structure was prepared by using 'prepare protein' module obtainable in the discovery studio v16 (DS). This protocol prepares the input proteins and executes actions such as inserting missing atoms in the incomplete residues [32,33], modelling the missing loop regions, deletion of the alternate conformations, dislodges the water molecules, standardizing the atom names, and protonating titratable residues using predicted pKs. The active site of the target was selected around the cocrystallised ligand for all the residue atoms that fall in the radius of 7.9 Å that includes, Thr11, Gly12, Val13, Gly14, Lys15, Thr16, Lys37, Thr41, Asp49, Pro71, Met72, Ala73, Pro74, Glu108, Ala110, Gly111 and Val115. The XYZ coordinates for the binding site sphere are taken as 33.497189 Å, 22.149442 Å, and 33.777841 Å, respectively. To evaluate and validate the docking algorithm, the inbound cocrystallised ligand was docked into the X-ray structure of the target. The redocked binding mode and the native cocrystallised ligand binding mode were found to be similar. The RMSD between the native ligand and the redocked pose was observed to be 0.81 Å (Supplementary Figure 1).

2.2. Selection of the ligands

The six small molecules used for the current study are available in the lab. The three compounds are earlier published from our

Table 1
Binding affinity and intermolecular interactions.

Comp no	Dock score ^a	Hydrogen	π interactions	Carbon hydrogen	Attractive charges	van der Waals
1	51.4398	Lys15, Glu52	–	–	Lys15, Asp49	Thr16, Lys37, Glu108, Gly109, Gly111
2	47.7706	Glu52, Glu108	Asp48	–	–	Lys15, Thr16, Lys37, Asp47, Asp49, Ala110
3	44.6003	Thr16, Asp48	–	Asp47	–	Lys15, Lys37, Gln40, Asp49, Ala110, Gly111
5	40.0093	Thr41	Met72, Ala73, Pro74, Ala110	Pro71	–	Thr11, Lys37, Gln40, Gly42, Asp47, Gly111, Val115
10	45.6705	Thr16, Asp47, Glu52	Asp48	–	–	Lys37, Thr41, Gly42, Asp49, Arg55, Pro71
11	42.5373	Thr16, Asp49, Glu52, Glu108	Ala110	–	–	Thr11, Lys37, Gln40, Thr41, Asp47, Asp48, Ala73, Pro74

^a -CDOCKER interaction energy expressed in kcal/mol.

group [7]. These small molecules were prepared by using the ‘full minimization’ module available with the DS. The CHARMM forcefield was applied with the *Smart Minimizer* algorithm. This is performed by 1000 steps of steepest descent with an RMS gradient tolerance of 3, subsequently followed by conjugate gradient minimization. The well-prepared protein and ligands were upgraded to molecular docking to understand the binding affinities between them.

2.3. Molecular docking to assess the binding affinity

To elucidate the predictive binding mode and to assess the binding affinity of the selected compounds towards the *Mycobacterium tuberculosis*, the CDOCKER available with the discovery studio v18 (DS) was used. In the current study six compounds, three from the previous study were also included [7]. The CDOCKER docking method used here is a grid-based molecular docking method that uses CHARMM. In here, the receptor is fixed in contrast to the ligands that are allowed to move during the refinement. Random conformations of the ligands are then generated. The conformations are correspondingly translated into the binding site. The poses are then generated using random rigid-body rotations followed by simulated annealing. A final minimization is then used to refine the ligand poses. Finally, the molecular docking score is evaluated according to the –CDOCKER interaction energy expressed in kcal/mol. Each ligand was allowed to generate 50 poses. Each pose was ranked from the largest cluster, with good dock score and interactions with the key residues. These complexes were upgraded to molecular dynamics simulation (MDS) to gain insights on the complex stabilities and intermolecular interactions.

2.4. Molecular dynamics simulation studies

To vividly understand the nature of small molecules at the binding pocket of the protein, the molecular dynamics simulations (MDS) were performed using the GROMACS v2016.6 [34]. The main reason for adapting the MDS is to evaluate the binding stabilities of the protein-ligand complex and to understand the interactions at the atomistic level as reported earlier [35,36]. To accomplish this, the topologies of the ligand were generated from SwissParam [37] while using the CHARMM27 all atom force field. The dodecahedron water box was generated and solvated with TIP3P water model and subsequently, the counter ions (Na^+) were supplemented (Supplementary Table 1). The minimization of the system was done. The protein and the ligand were coupled and two step equilibration was done with conserved number of particles (N), system volume (V) and temperature (T) (NVT) and the constant number of particles (N), system pressure (P) and temperature (T) (NPT) for 1 ns each with V-rescale thermostat. The NPT ensembles were upgraded to MDS for 500 ns. During the equilibration process, the protein backbone was restrained. The Parrinello-Rahman barostat [38] was used to monitor the pressure of the system. The LINCS algorithm was used to maintain the geometry of the molecules that constraints the bond length [39]. Long-range electrostatic interactions were computed with Particle Mesh Ewald (PME) technique [40]. The results were analyzed using the visual molecular dynamics (VMD) [41] and DS and studied according to the root mean square deviation (RMSD), root mean square fluctuations (RMSF), radius of gyration (Rg) [42], number of hydrogen bonds, and the mode of ligand binding. Here, the protein backbone was analyzed. Furthermore, the PCA analysis and essential dynamics (ED) was conducted using the *gmx covar* and *gmx sham* to understand the dynamic motion of the protein.

3. Results

3.1. Binding affinity studies

The binding affinity of the selected molecules was conducted adapting the CDOCKER module available with the DS. The results have shown that all the molecules have occupied the defined binding pocket by good dock score (Table 1). It was also noted that several key residues have adhered with the compounds to accommodate them at the binding pocket firmly.

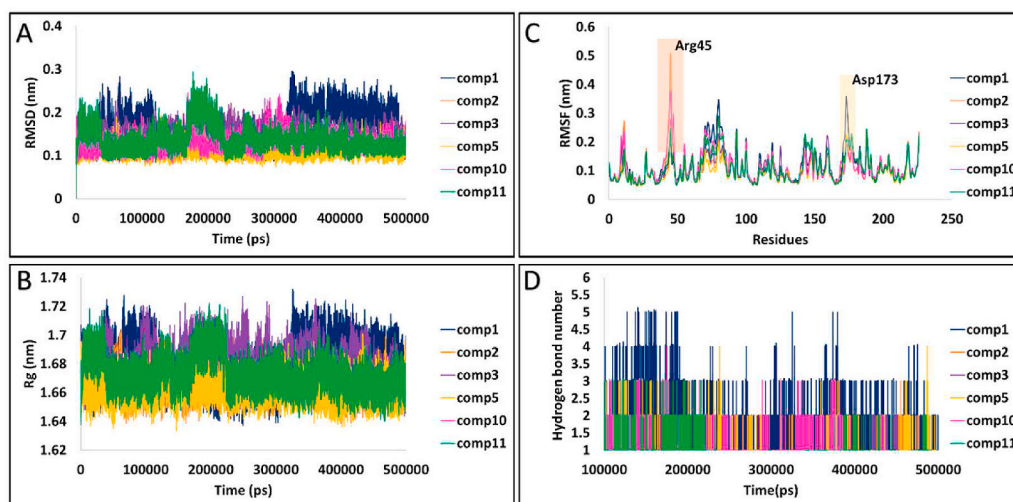


Fig. 1. MDS analysis of six systems conducted for 500 ns. A) RMSD profiles. B) Rg plots. C) RMSF profiles. The residues that showed fluctuations are marked in orange box. D) number of hydrogen bonds during the MD run.

3.2. Molecular dynamics simulation studies

The MDS studies elaborate on the stability of the protein ligand complexes formed during the molecular docking. The top ranked poses from the molecular docking were forwarded to MDS and the results were analyzed according to RMSD, RMFS, Rg, binding mode and hydrogen bond number.

3.3. Stability analysis by RMSD

RMSD records the deviations present between the initial structure and the final structure of the simulation run. Generally, a smaller deviation denotes a relatively stable structure [43,44]. In the current study, the RMSD for the six systems has projected to be stable below 0.3 nm. During the evolution of the simulation, no abnormal behavior was noticed among the six systems. Comp1 and comp11 demonstrated marginal deviations in the plots. Comp1 showed a minor surge at around 300 ns with an average of 0.15 nm, while the comp11 displayed a surge between 160 ns and 230 ns and has remained stable thereafter, with an average of 0.13 nm. The average RMSD for Comp2 is 0.14 nm, comp3 is 0.14 nm, comp5 is 0.11 nm and comp10 is 0.13 nm, respectively. These results state that the systems were stable during 500 ns simulation run representing the RMSD values below 0.3 nm (Fig. 1A).

3.4. The protein backbones were compact during the simulations

The Rg plots elucidates on the compactness of the protein backbone. The Rg plots for all the systems were found to be between 1.64–1.74 nm. This implies that smaller the reading highly compact is the structure. Similar to the RMSD profiles, the comp11 has shown a negligible peak between 160 ns and 230 ns and a clear plateau was noticed later. The other systems were highly compact with an average of 1.68 nm, 1.66 nm, 1.68 nm, 1.66 nm, 1.67 nm and 1.67 nm, respectively for comp1, comp2, comp3, comp5, comp 10 and comp11. These results suggest that the compounds were highly compact during the evolution of the simulation run (Fig. 1B).

3.5. Residue specific fluctuations

The residue specific fluctuations were determined according to the RMSF. The results have shown that the protein backbone has displayed minimum fluctuations except for Arg45 and Asp173. These residues are at a distance from the binding pocket and therefore we speculate that their impact might be negligible on the binding potential. While the key residues that interact with the ligands were relatively stable (Fig. 1C).

3.6. Hydrogen bond interactions

During the course of the simulation run, the presence of the hydrogen bonds were meticulously monitored. It was noticed that the hydrogen bonds were formed during the entire simulation run. Notably, the number was higher until 250 ns for all the complexes (Fig. 1D). At least one hydrogen bond was present during the entire simulation (Fig. 1D).

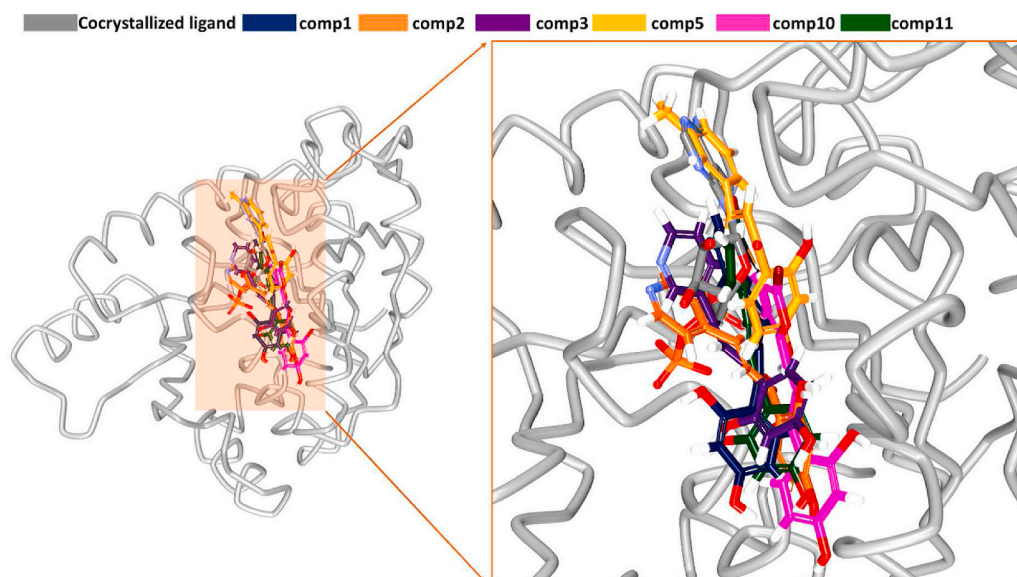


Fig. 2. Accommodation of the small molecules at the binding pocket of the protein. Left panel indicates the protein and the ligands at the binding pocket and the right panel indicates its zoomed version.

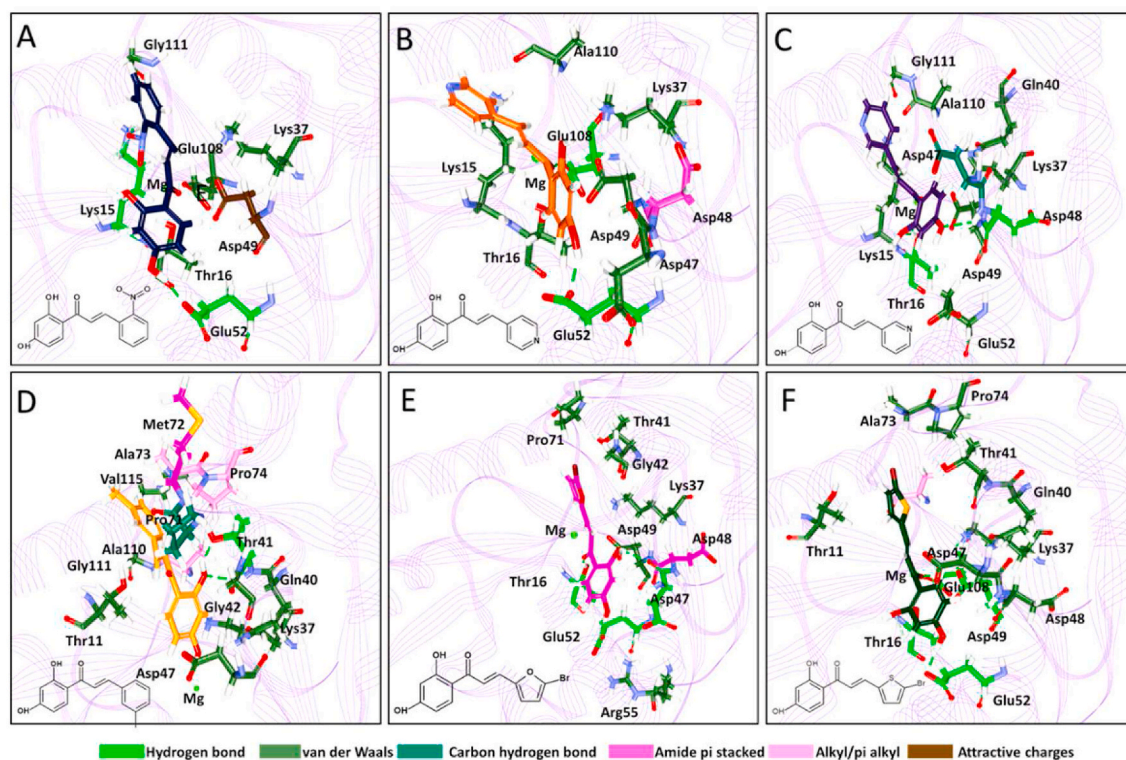


Fig. 3. Intermolecular interactions between the protein and the ligands. A) comp1, B) comp2, C) comp3, D) comp5, E) comp10, F) comp11.

3.7. Binding mode analysis

From the stable RMSD, the representative structures from last 50,000 ps were extracted and were superimposed onto the X-ray structure using the 'align structures' module available on DS. The results have shown that the small molecules have occupied the same position as that of the cocrystallised ligand (Fig. 2). Several key residues have positioned the ligands at the binding pocket of the target.

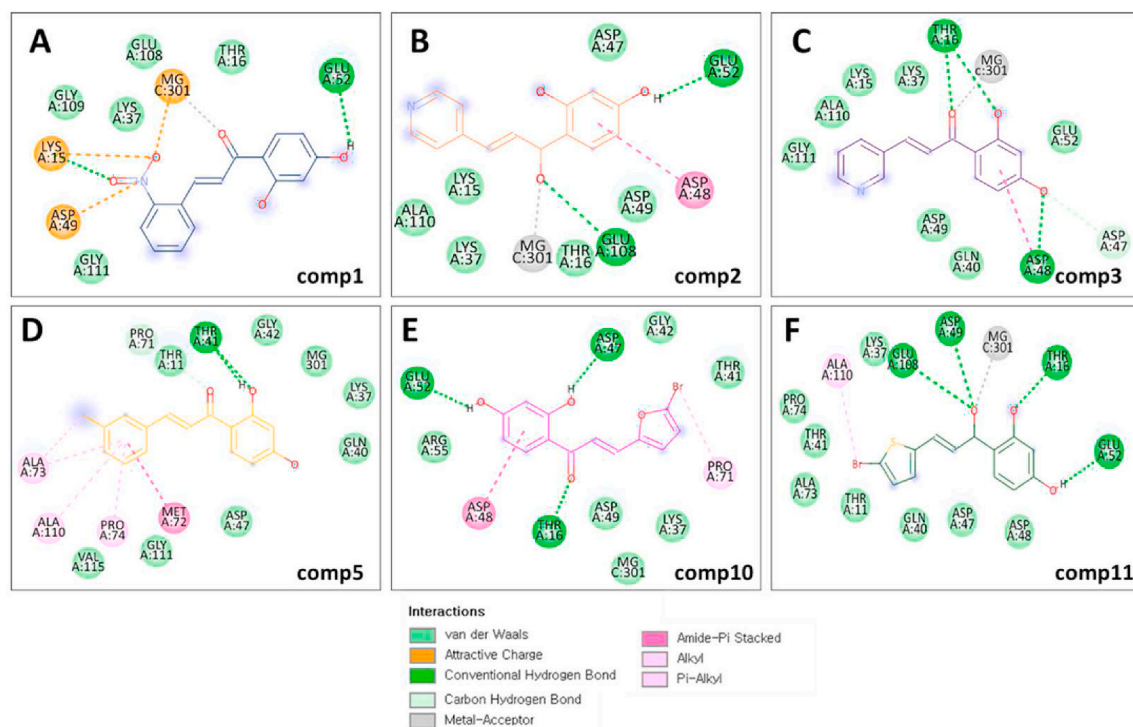


Fig. 4. Intermolecular 2D interactions between the protein and the ligands. A) comp1, B) comp2, C) comp3, D) comp5, E) comp10, F) comp11.

3.8. Intermolecular interactions

Comp1 has formed hydrogen bond interactions with Lys15 and Glu52 residues by an acceptable bond length. Additionally, the residues Lys15, Asp49 have bound to the ligand via the attractive charges. The residues Thr16, Lys37, Glu108, Gly109 and Gly111 have interacted by van der Waals interactions holding the ligand at the binding pocket of the target (Fig. 3A, Fig. 4A and Table 1).

Comp2 compound has generated hydrogen bond interactions with Glu52, Glu108 and a π -alkyl interaction with the residue Asp48. The residues Lys15, Thr16, Lys37, Asp47, Asp49 and Ala110 have prompted van der Waals interactions accommodating the ligand at the binding pocket of the target (Figs. 3B and 4B and Table 1).

Comp3 has formed hydrogen bond interactions with the residues Thr16, Asp48 and a carbon hydrogen bond with the residue Asp47. The residues Lys15, Lys37, Gln40, Asp49, Ala110 and Gly111 have formed van der Waals interactions accommodating the ligand at the binding pocket of the target protein (Figs. 3C and 4C and Table 1).

Comp5 has formed a two hydrogen bond interaction Thr41. The residues Ala73, Pro74 and Ala110 have formed π -alkyl interaction and the residue Met72 has prompted an amide π stacked interaction with the ligand. Additionally a carbon hydrogen bond was noticed with Pro71. The residues Thr11, Lys37, Gln40, Gly42, Asp47, Gly111 and Val115 have formed van der Waals interactions adhering the ligand at the binding pocket (Figs. 3D and 4D and Table 1).

Comp10 has generated hydrogen bond interactions with Thr16, Asp47 and Glu52 residues. An amide π stacked interaction was noticed with Asp48 residue. The residues Lys37, Thr41, Gly42, Asp49, Arg55 and Pro71 have held the ligand firmly by van der Waals interactions (Figs. 3E and 4E and Table 1).

Comp11 has formed hydrogen bonds with the residues Thr16, Asp49, Glu52, Glu108 and a π alkyl interaction with the residue Ala110. The residues Thr11, Lys37, Gln40, Thr41, Asp47, Asp48, Ala73 and Pro74 have communicated with the ligand via the van der Waals interaction to efficiently secure the ligand at the binding pocket of the protein (Figs. 3F and 4F and Table 1). The comprehensive 2D interactions between the target and the ligands are given in Fig. 4.

On focusing on the interactions between Mg ion and the ligands, it was observed that all the ligands have formed interaction with the Mg ion. Comp1 has interacted via the attractive charge. Comp2, comp3 and comp11 have formed the metal-acceptor interaction, while comp5 and comp10 have prompted the van der Waals interaction (Fig. 4). A similar attractive charge was noticed between the Mg ion and the native inbound ligand and the docked pose (Supplementary Figure 2B and 2C). These interactions have also stabilized the ligands to be placed at the binding pocket of the protein.

3.9. Protein ligand interaction energy

We further estimated the interaction energy between the protein and ligand in terms of short-range Lennard-Jones energy (LJ-SR).

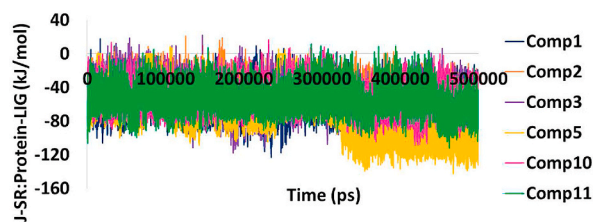


Fig. 5. LJ-SR interaction energy between the target and the ligands.

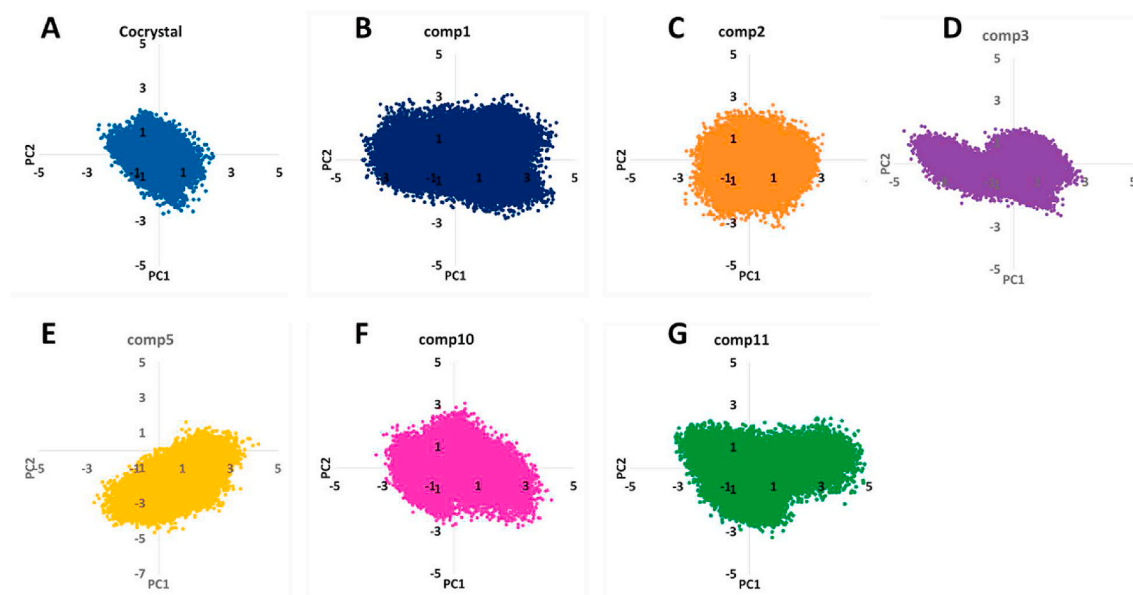


Fig. 6. Principle component analysis (PCA) of six systems in comparison with the cocrystallised ligand system.

The LJ-SR of all the complexes ranges to -140 kJ/mol. The average LJ-SR of comp1 is -61.80 kJ/mol, comp2 is -45.85 kJ/mol, comp3 is -54.18 kJ/mol, comp5 is -73.93 kJ/mol, comp10 is -55.36 kJ/mol and comp11 is -55.49 kJ/mol, respectively (Fig. 5). Comparatively, the cocrystallised ligand demonstrated a better interaction energy of -108.93 kJ/mol (Supplementary Figure 2A).

4. PCA and ED

Essential dynamics plays a key role in understanding the overall motion of the proteins in the conformational spaces during the course of the simulation run [44–46]. Logically, a fair amount of flexibility and rigidity is necessary for a given protein to be functional [44,45]. More particularly, this may be required at the binding pocket. Here, we have studied the conformational changes of the protein backbone from the principle component analysis (PCA) [44–46]. The PCA findings have shown that the systems were largely stable (Fig. 6). The cocrystal, comp1, comp2 and comp10 systems were highly stable (Fig. 6A, 6B, 6C and 6F) showing that the systems were well converged. However, comp3, comp5 and comp11 systems have navigated along the PC2 before attaining the equilibrium (Fig. 6D, 6E and 6G). This pattern is in line with the RMSD and the Rg profiles (Fig. 1). Upon viewing at the Gibbs free energy landscapes a varied pattern was observed. Except for comp2 (Fig. 7C) and comp5 (Fig. 7E) systems all the other systems (Fig. 7A, 7B, 7D, 7F and 7G) have displayed two energy minima during the process of attaining the equilibrium state (Fig. 7). From the Gibbs plots it can be revealed that the red spots represent the favourable conformations while the dark blue spots indicate the unfavourable conformations (Fig. 7).

5. Discussion

To find effective therapeutics against the causative agent of TB, the computational framework was undertaken. Computer-aided drug discovery/design (CADD) approaches have paved the way in the progress of discovering, expanding and retrieving the small molecules with therapeutic potential [47]. CADD methods have been instrumental in identifying inhibitors against a varied range of diseases [47].

Natural products/analogues have immensely contributed to the field of pharmacotherapy, in particular to treat cancer and

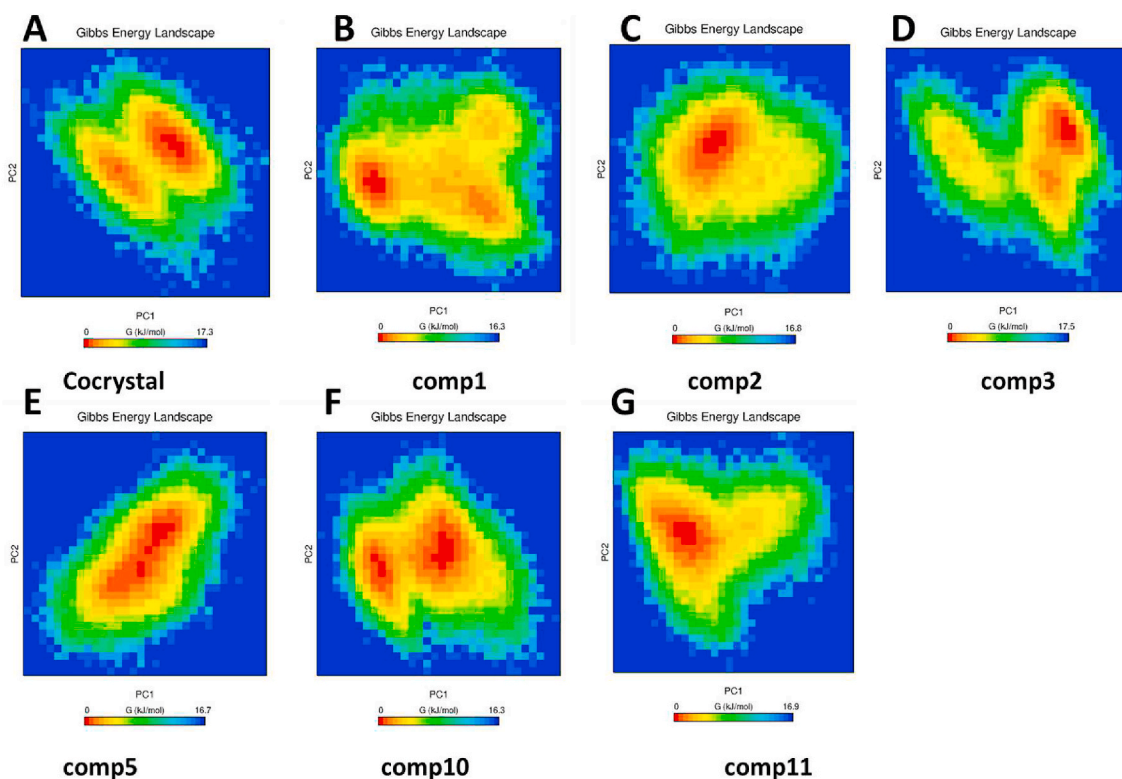


Fig. 7. Gibbs energy landscape of six systems in comparison with the cocrystallised ligand system.

infectious diseases [48]. These compounds offer several advantages and are enriched with medicinal values [48]. Flavonoids occupy a special place in providing medicinal formulation to various diseases [49]. These contain the phenolic group and are present in vegetables, bark, stems, roots, fruits, tea, flowers and wine [50]. One such flavonoid, butein has been extensively studied for its therapeutic potential against cancer [51–53] along with anti-inflammatory and antioxidant properties [24]. In the current study, we have designed the butein compounds computationally and have used it against TB target.

Six compounds were used against the validated target *MtDTBS*. Out of the six compounds, two compounds are novel and have shown anticancer activity in our previous study [7]. The target structure chosen for the current study is complexed with adenosine-5'-diphosphate and Mg ion. The binding pocket can be divided into the nucleoside-binding pocket and the phosphate-binding loop [31]. The compounds used in the study have inclined more towards the nucleoside-binding pocket. These compounds have interacted with the key residue Lys37 via the van der Waals interaction. The interaction with this residue was reported in previous literature [31]. Several other interactions present within the crystal structure were also noted with the selected compounds. The MSD studies conducted for 500 ns have shown that the systems were stable throughout the simulations displaying the stable profiles, read according to RMSD, RMSF and Rg.

Additionally, the modified butein compounds have shown a similar MDS results as that of the cocrystallised ligand. The RMSD profiles of all the modified butein systems were below 0.3 nm, with an average between 0.11 nm and 0.15 nm (Fig. 1A), while the cocrystallised ligand has demonstrated an average RMSD of 0.09 nm without any deflection during the entire simulation run (Supplementary Figure 3A). The Rg profiles of the modified systems ranged between 1.63 and 1.72 nm (Fig. 1B) while the cocrystallised ligand was relatively compact (Supplementary Figure 3B). The RMSF plots of the modified butein systems were recorded to be ranging between 0.09 nm to 0.5 nm, respectively (Fig. 1C), while that of the cocrystallised ligand displayed an average RMSF of 0.15 nm (Supplementary Figure 3C). Interestingly, the RMSF of the cocrystallised ligand complex had shown lower fluctuations as compared to the modified compounds including the residues, Arg45, Asp173 (Supplementary Figure 3C).

A noteworthy observation was noticed in the number of hydrogen bond interactions. The cocrystallised ligand had shown higher number of hydrogen bonds (Supplementary Figure 3D) than the modified compounds (Fig. 1D). This may be due to the larger size of the cocrystallised ligand than the modified compounds. Overall, the results of the modified compounds were comparable to cocrystallised ligand.

We further analyzed to know if any deflections have occurred between the docked pose and the MD derived pose by initiating the 'align structures' module available on the DS. This utilizes the 3DMA program to align the protein sequences in accordance with the 3D structural similarity. This algorithm identifies the segments that are matching with the protein by comparing the C-alpha pseudo-torsion angle and subsequently completes the alignment by discriminative extension of the matching segments. Furthermore, the multiple structure alignment is performed by a progressive pairwise alignment depending on the initial guide tree derived from all-

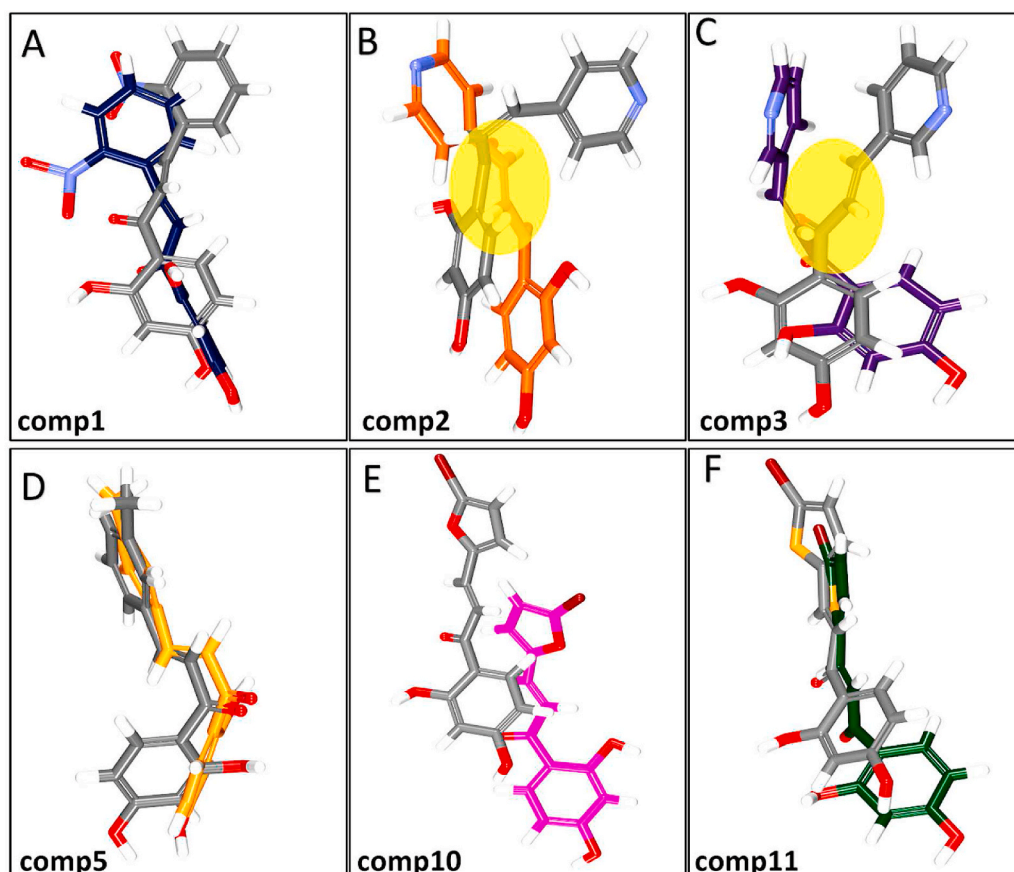


Fig. 8. Alignment of the docked pose and the MD pose. A) comp1, B) comp2, C) comp3, D) comp5, E) comp10, F) comp11. The linear wobbled structure in comp2, comp3, and shown in yellow oval.

Table 2

Binding affinity studies of the compounds from the literature with the target of interest.

Smiles	-CDOCKER interaction energy (kcal/mol)
<chem>O=C(CC[C@H](NC(=O)NOCc1cccc1)NC(=O)OCc2cccc2)NOCc3cccc3</chem>	70.2172
<chem>CCNC(=O)N[C@H](CC(=O)NCC)NC(=O)OCc1cccc1</chem>	64.7585
<chem>O=C(CC[C@H](NC(=O)OCc1cccc1)NC(=O)N2CCCC2)N3CCCC3</chem>	64.0073
<chem>CC(O)(C)C(=O)c1ccc(cc1)n2cc(CSc3nnc(Cc4cccc4)n3\N=C\Cc5ccc(cc5)N6CCOCC6)nn2</chem>	61.6879
<chem>O=C(N[C@H](CCNC(=O)N1CCOCC1)C(=O)N2CCOCC2)OCc3cccc3</chem>	59.8071
<chem>O=C(CC[C@H](NC(=O)OCc1cccc1)NC(=O)N2CCOCC2)N3CCOCC3</chem>	59.5927
<chem>OC(=O)c1ccc(cc1)n2cc(CSc3nnc(Cc4cccc4)n3\N=C\Cc5ccc(cc5)N6CCOCC6)nn2</chem>	57.0524
<chem>OC(=O)c1ccc(cc1)n2cc(CSc3nnc(Cc4cccc4)n3\N=C\Cc5ccc(cc5)N6CCOCC6)nn2</chem>	56.7373
<chem>CCNC(=O)NCC[C@H](NC(=O)OCc1cccc1)C(=O)NCCC</chem>	55.7313
<chem>O=C(N[C@H](CCNC(=O)N1CCOCC1)C(=O)N2CCCC2)OCc3cccc3</chem>	55.4408
<chem>OC(=O)c1ccc(cc1)n2cc(CSc3nnc(Cc4cccc4)n3\N=C\Cc5ccc(cc5)N6CCOCC6)nn2</chem>	53.4039
<chem>CC(O)(C)C(=O)c1ccc(cc1)n2cc(CSc3nnc(Cc4cccc4)n3\N=C\Cc5ccc(cc5)N6CCOCC6)nn2</chem>	52.6891
<chem>C#CCc1nnc(Cc2cccc2)n1\N=C\C[C@H]3CC[C@H](CC3)N4CCOCC4</chem>	52.2451
<chem>CC(O)(C)C(=O)c1ccc(cc1)n2cc(CSc3nnc(Cc4cccc4)n3\N=C\Cc5ccc(cc5)N6CCOCC6)nn2</chem>	51.9265
<chem>O=C(NC1=CC(=O)NC1=O)OCc2cccc2</chem>	44.3654
<chem>Sc1nnc(Cc2cccc2)n1\N=C\Cc3ccc(cc3)N4CCOCC4</chem>	43.6639
<chem>Sc1nnc(Cc2cccc2)n1\N=C\Cc3ccc(cc3)N4CCOCC4</chem>	42.8059
<chem>C#CCc1nnc(Cc2cccc2)n1\N=C\Cc3ccc(cc3)N4CCOCC4</chem>	39.651
<chem>C#CCc1nnc(Cc2cccc2)n1\N=C\CC3=CC=CC3</chem>	36.7983
<chem>Sc1nnc(Cc2cccc2)n1\N=C\Cc3ccc(cc3)</chem>	34.3048

against-all pairwise comparison. Correspondingly, the sequences are clustered into alignments depending on their similarities.

Our results have shown that the protein backbones have well aligned as expected. We further focussed on the ligands to understand if there is any major change from the initial pose. From the results, it can be deduced that comp1, comp5 and comp11 were relatively

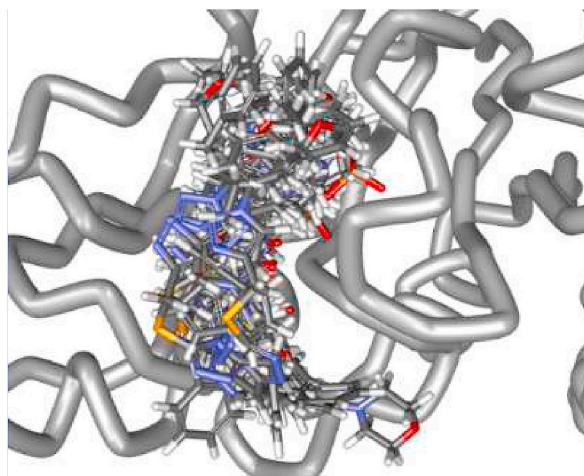


Fig. 9. Accommodation of the compounds at the binding pocket of the target.

Table 3

Binding energy comparison between the reported *Mt* inhibitors and six compounds.

Compound name	-CDOCKER interaction (kcal/mol)	Binding energy (kcal/mol)
comp1	51.4398	-138.971
comp2	47.7706	-150.121
comp3	44.6003	-135.839
comp5	40.0093	-83.1029
comp10	45.6705	-153.886
comp11	42.5373	-149.82
Compound 2	50.1015	-127.768
Compound 7a	45.6095	-119.457

stable with marginal movement from the initial pose (Fig. 8A, 8D, and 8F). In case of comp2 and comp3, the linear structure present between both the peripheral rings has seemed to wobble (Fig. 8B and 8C). Comp10 has moved upwards from the initial position; however, it has still rested in the defined binding pocket (Fig. 8E).

We further made an analogy with the other compounds retrieved from the recently published literature [54,55]. Correspondingly 20 compounds were sketched on Biovia Draw 2017. The compounds were minimized and subjected to molecular docking using the same protocol as described before. The results show that the molecular docking score ranges from 34.30 kcal/mol to 70.21 kcal/mol (Table 2). This high dock score could be because of the size of the small molecules which are slightly bigger than our compounds. These compounds have also showed a curved pattern while accommodating at the binding pocket (Fig. 9), while our compounds of interest have rather displayed a linear form. This again could be because of the smaller size of the compounds making them to easily fit into the binding pocket.

Recent report discovered two inhibitors towards the *Mycobacterium tuberculosis* dethiobiotin synthase (MtDTBS) [30]. One compound, cyclopentylacetic acid (compound 2) has demonstrated a KD of 3.4 ± 0.4 mM and compound 7a, an optimized compound of compound 2 has shown a KD of 57 ± 5 nM [30]. Structurally, compound 2 appears to be similar as that of our compounds while compound 7a has a slightly bigger structure. Here, we performed a binding energy comparison after the molecular docking studies. The docking results of compound 2 and compound 7a are comparable with our compounds while the binding energies are better for our compounds except for comp 5. Interestingly, the dock score of comp1 and compound 2 are highly similar, while the binding energy of comp1 is better. From this analysis we speculate that the KD of our compounds might also be better than compound 2 and compound 7a (Table 3).

The findings from the study are an elegant evidence that supports our objective that the selected butein analogues can act against TB. Taken together, our results present six compounds as prospective candidate inhibitors that could inhibit TB. Although our computational results demonstrate that the butein analogues could be potential MtDTBS inhibitors, it is essential to test these compounds *in vitro* and also to specifically target MtDTBS to understand its inhibitory effect.

6. Conclusion

In order to find effective therapeutics towards the infectious TB, we have used several computational methods such as molecular docking and molecular dynamics simulations. The ligands used are the butein analogues that are present in the lab. These compounds have shown good binding affinity potential and have established stable RMSD profiles. The key residue Lys37 was noticed with all

compounds. In conclusion, we put forth six compounds as potential antiTB drugs. These compounds can also act as starting structures for designing and developing new candidate drugs.

Author contribution statement

Shailima Rampogu, Baji Shaik, Ju Hyun Kim, Tae Sung Jung, Min Woo Ha, Keun Woo Lee: Conceived and designed the experiments; Performed the experiments; Analyzed and interpreted the data; Contributed reagents, materials, analysis tools or data; Wrote the paper.

Funding statement

This work was supported by two National Research Foundation of Korea (NRF) grants funded by the Korean government (MSIT) (Grant Nos. NRF-2021R1F1A1059797 and NRF-2021R1A2B5B02002220).

Data availability statement

Data included in article/supp. material/referenced in article.

Declaration of interest's statement

The authors declare no conflict of interest.

Appendix A. Supplementary data

Supplementary data related to this article can be found at <https://doi.org/10.1016/j.heliyon.2023.e13324>.

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