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Polyoxygenated Sterols from the South China Sea Soft Coral *Sinularia* sp.

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Abstract: Chemical investigation of the ethanol extract of soft coral *Sinularia* sp. collected from the South China Sea led to the isolation of three new polyoxygenated sterols, (3*S*,23*R*,24*S*)-ergost-5-ene-3 β ,23 α ,25-triol (**1**), (24*S*)-ergostane-6-acetate-3 β ,5 α ,6 β ,25-tetraol (**2**), (24*S*)-ergostane-6-acetate-3 β ,6 β ,12 β ,25-tetraol (**3**) together with three known ones (**4–6**). The structures, including relative configurations of the new compounds (**1–3**), were elucidated by detailed analysis of spectroscopic data (IR, UV, NMR, MS) and by comparison with related reported compounds. The absolute configuration of **1** was further determined by modified Mosher's method. Compound **5** exhibited moderate cytotoxicity against K562 cell line with an IC₅₀ value of 3.18 μ M, but also displayed strong lethality toward the brine shrimp *Artemia salina* with a LC₅₀ value of 0.96 μ M.

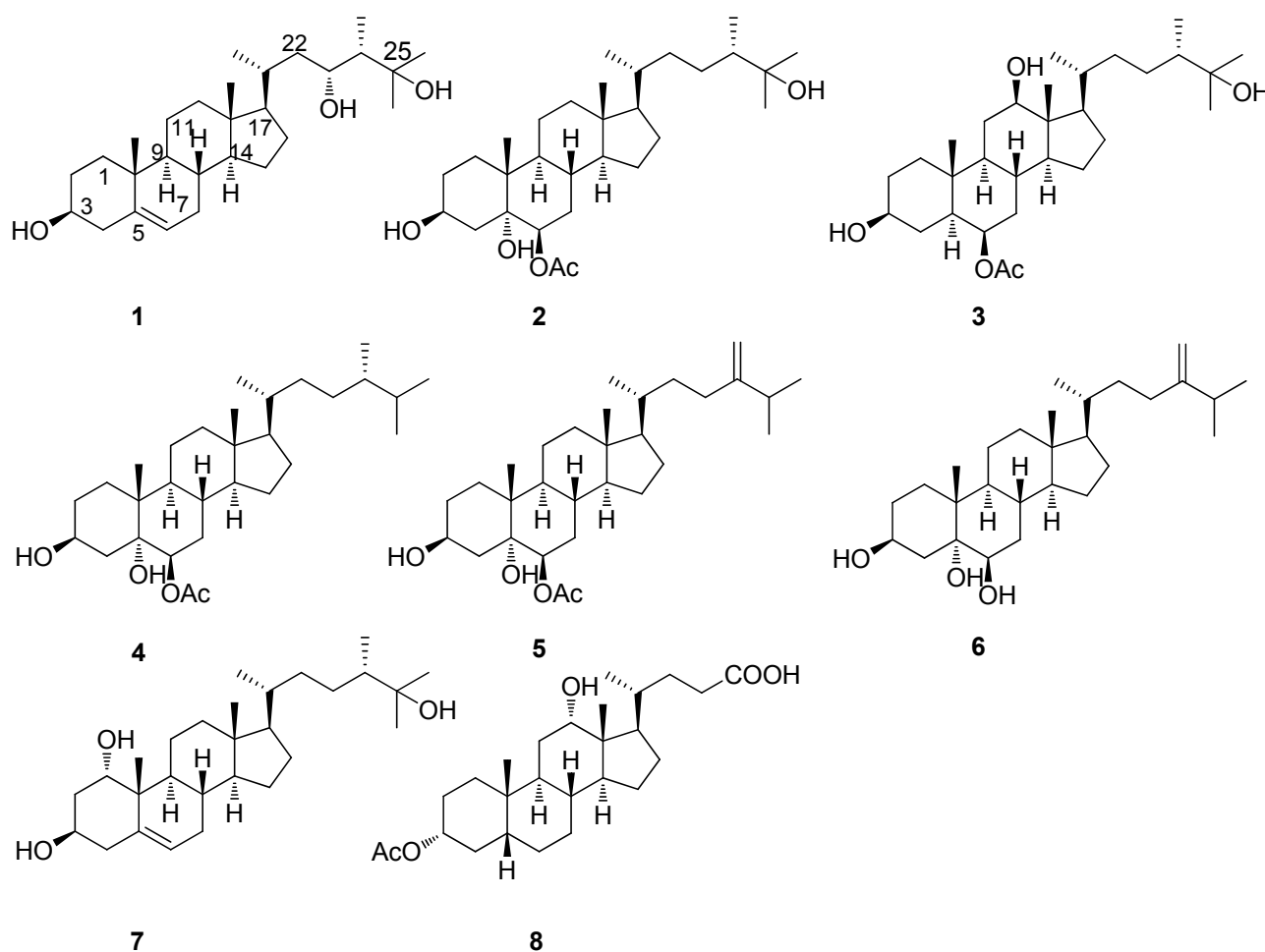
Keywords: soft coral; *Sinularia* sp.; polyoxygenated sterols; cytotoxicity

1. Introduction

Soft coral of the genus *Sinularia* has been found to be a rich source of bioactive secondary metabolites [1,2], such as acylated spermidine [3,4], lipids and fatty acids [5], cyclic sesquiterpene

peroxides [6], sterols [7,8], and norditerpenes [9,10]. A number of them showed an array of biological activities such as cytotoxic activities [3,4] and inhibitory effect on LPS-induced TNF- α production [9,10]. As part of our ongoing investigation of new natural bioactive compounds from marine invertebrates in the South China Sea [11–16], the soft coral *Simularia* sp. attracted our attention because the crude extract of *Simularia* sp. showed lethal activity toward brine shrimp *Artemia salina*. Bioassay-guided fractionation of the active extracts led to the isolation of three new polyoxygenated sterols (**1–3**) and three known ones (**4–6**) [17,18] (Figure 1).

Figure 1. Structures of compounds **1–6** from *Simularia* sp.



2. Results and Discussion

Compound **1** was obtained as a white, amorphous powder with $[\alpha]_D^{25} -22.4$ (c 0.05, CH₃OH). IR spectrum showed its absorption at 3410–3584 cm⁻¹ for hydroxy groups. The molecular formula was established as C₂₈H₄₈O₃ based on HRESIMS ($[M + Na]^+$ 455.3498 (calcd for C₂₈H₄₈O₃Na 455.3496)) and NMR spectroscopic data. The ¹H-NMR (Table 1) and ¹³C-NMR (Table 2) spectra implied that compound **1** was a polyhydroxylated sterol. The ¹H-NMR spectrum of **1** revealed four methyl singlet signals at δ_H 0.70 (3H, s, H₃-18), 1.00 (3H, s, H₃-19), 1.23 (3H, s, H₃-26) and 1.23 (3H, s, H₃-27), and two methyl doublet signals at δ_H 0.81 (3H, d, J = 6.6 Hz, H₃-28) and 1.08 (3H, d, J = 6.6 Hz, H₃-21), as well as two oxymethines (δ_H 3.71 and 3.56), and an olefinic proton signal at δ_H 5.34 (br d, J = 4.8 Hz). The ¹³C-NMR and DEPT spectra of **1** displayed 28 signals which were assigned to four

quaternary carbons, nine methines, nine methylenes, and six methyls. The olefinic carbon signals appearing at δ_C 140.8 (C) and 121.6 (CH) corresponded to one trisubstituted double bond. The ^{13}C -NMR chemical shifts at δ_C 75.8 (CH), 75.2 (C) and 71.8 (CH) confirmed the presence of three oxygenated carbons. Detailed analysis of the ^1H - ^1H COSY spectrum in combination with HMQC and HMBC (Figure 2) experiments allowed the assignment of all of the chemical shifts in the ^1H and ^{13}C -NMR spectra and led to structure **1**. Comparison of ^1H -NMR and ^{13}C -NMR of **1** with those of the known compound **7** (patusterol A) [19], a hydroxylated steroid from the Kenyan soft coral *Lobophytum patulum*, further confirmed the structure of **1**. The obvious differences between the two compounds are the chemical shifts at δ_H 3.71 (1H, ddd, $J = 12.0, 8.4, 3.0$ Hz, H-23) in **1** vs. 3.71 (1H, br t, $J = 2.6$ Hz, H-1) in **7**, and δ_C 75.8 (CH) in **1** vs. 72.5 (CH) in **7**, indicating that the hydroxyl group in **1** is at C-23 not as **7** at C-1. In addition, the ^1H - ^1H COSY correlations of H-23 with H-24 and H-22, and the HMBC correlations from H₃-28 to C-24, C-23 and C-25 enable the hydroxyl group to be placed at C-23. The relative stereochemistry of **1** was assigned on the basis of 2D NOESY experiment. In the NOE spectrum of **1**, NOE correlations observed from H₃-18 to H-20 indicated that H-20 was in β disposition. The α orientation of H-9 and H-14, and β orientation of H-8 were also determined by NOESY experiment. The absolute configuration at C-3 and C-23 of **1** was tried to determine using the modified Mosher's method [20]. The (*S*)- and (*R*)-MTPA esters **1r** and **1s** were prepared using (*R*)- and (*S*)-MTPA chloride, respectively. The determination of $\Delta\delta$ values ($\delta_S - \delta_R$) (Figure 3) for protons neighboring C-3 and C-23 should lead to the assignment of the configuration at C-3 and C-23 in **1**. The configuration at C-24 in this and congener sterols was suggested as 24*S* on biogenetic grounds, since almost all the 24-methylsterols isolated from corals have 24*S* stereochemistry according to the literature [21,22]. According to the relatively small difference of $\Delta\delta$ values ($\delta_S - \delta_R$), the absolute configuration of **1** was tentatively determined as (3*S*,23*R*,24*S*)-ergost-5-ene-3 β ,23 α ,25-triol.

Table 1. ^1H -NMR data for compounds **1–3**.

H#	1 , δ_H (J in Hz) ^a	2 , δ_H (J in Hz) ^b	3 , δ_H (J in Hz) ^a
1	1.82 (1H, d, $J = 4.8$ Hz, H-ax) 1.12 (1H, m, H-eq)	1.75 (1H, br d, $J = 12.0$ Hz, H-ax) 1.18 (1H, m, H-eq)	1.82 (1H, br d, $J = 4.8$ Hz, H-ax) 1.27 (1H, m, H-eq)
2	2.30 (1H, ddd, $J = 13.2, 4.8, 1.8$ Hz, H-ax) 1.49 (1H, m, H-eq)	2.01 (1H, dt, $J = 12.0, 2.4$ Hz, H-ax) 1.51 (1H, m, H-eq)	1.99 (1H, 1H, dt, $J = 12.6, 2.4$ Hz, H-ax) 1.49 (1H, m, H-eq)
3	3.56 (1H, m)	3.99 (1H, m)	4.09 (1H, m)
4	2.22 (1H, td, $J = 12.6, 2.4$ Hz, H-ax) 1.48 (1H, d, $J = 2.4$ Hz, H-eq)	1.77 (1H, br d, $J = 12.0$ Hz, H-ax) 1.53 (1H, m, H-eq)	1.84 (1H, br d, $J = 12.0$ Hz, H-ax) 1.54 (1H, m, H-eq)
5	–	–	1.61 (1H, m) ^d
6	5.34 (1H, br d, $J = 4.8$ Hz)	4.68 (1H, t, $J = 2.4$ Hz)	4.70 (1H, m)
7	2.02 (1H, dt, $J = 12.6, 4.8$ Hz, H-ax) 1.96 (1H, m, H-eq)	1.68 (1H, dd, $J = 13.8, 2.4$ Hz, H-ax) 1.47 (1H, d, $J = 2.4$, H-eq)	1.64 (1H, m) 1.61 (1H, m) ^d
8	1.52 (1H, m)	1.52 (1H, m)	1.29 (1H, m)
9	1.16 (1H, m)	1.63 (1H, m)	1.68 (1H, m)
10	–	–	–
11	1.43 (1H, m) 1.32 (1H, m)	1.34 (1H, m) 1.32 (1H, m)	1.73 (1H, dd, $J = 14.4, 7.2$ Hz H-ax) 0.77 (1H, m, H-eq)

Table 1. Cont.

12	1.28 (1H, m) 0.97 (1H, m)	1.58 (1H, m) 1.42 (1H, m)	4.31 (1H, td, $J = 7.2, 0.6$ Hz)
13	–	–	–
14	1.33 (1H, m)	1.25 (1H, m)	1.30 (1H, m)
15	1.44 (1H, m) 0.94 (1H, m)	1.55 (1H, m) 1.03 (1H, m)	1.51 (1H, m) 1.02 (1H, m)
16	1.88 (1H, m) 1.45 (1H, m)	1.89 (1H, m) 1.50 (1H, m)	1.86 (1H, m) 1.40 (1H, m)
17	1.15 (1H, m)	1.05 (1H, m)	1.10 (1H, m)
18	0.70 (3H, s)	0.78 (3H, s)	0.68 (3H, s)
19	1.00 (3H, s)	1.14 (3H, s)	1.16 (3H, s)
20	1.50 (1H, m)	1.36 (1H, m)	1.39 (1H, m)
21	1.08 (3H, d, $J = 6.6$ Hz)	0.94 (3H, d, $J = 6.6$ Hz)	0.93 (3H, d, $J = 6.6$ Hz)
22	1.84 (1H, m) 1.10 (1H, m)	1.62 (1H, m) 1.01 (1H, m)	1.62 (1H, m) 1.08 (1H, m)
23	3.71 (1H, ddd, $J = 12.0, 8.4, 3.0$ Hz)	1.68 (1H, m) 0.77 (1H, m)	1.86 (1H, m) 0.78 (1H, m)
24	1.56 (1H, m)	1.27 (1H, m)	1.28 (1H, m)
25	–	–	–
26	1.23 (3H, s) ^c	1.09 (3H, s)	1.15 (3H, s) ^e
27	1.23 (3H, s) ^c	1.10 (3H, s)	1.15 (3H, s) ^e
28	0.81 (3H, d, $J = 6.6$ Hz)	0.87 (3H, d, $J = 7.2$ Hz)	0.89 (3H, d, $J = 7.2$ Hz)
OAc	–	2.02 (3H, s, CH_3CO-)	2.06 (3H, s, CH_3CO-)

^a Spectra were measured in $CDCl_3$ (600 MHz); ^b Spectra were measured in CD_3OD (600 MHz). ^{c,d,e} Overlapping signals.

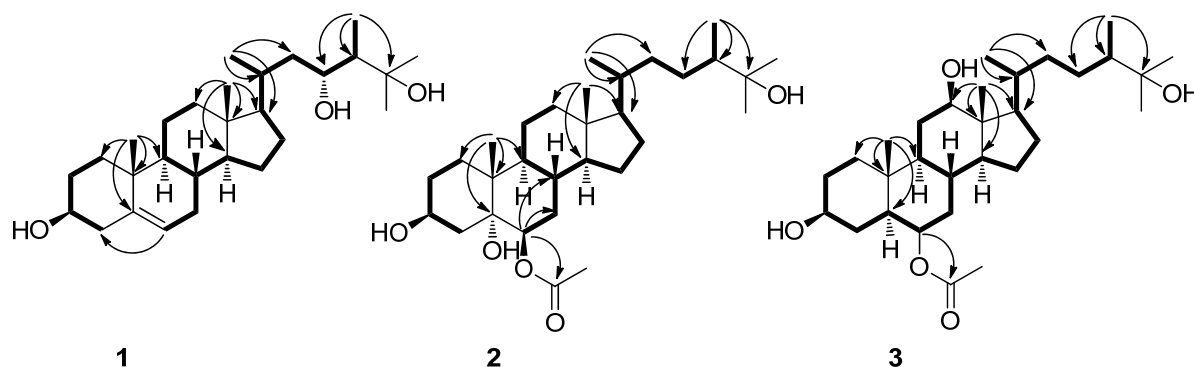
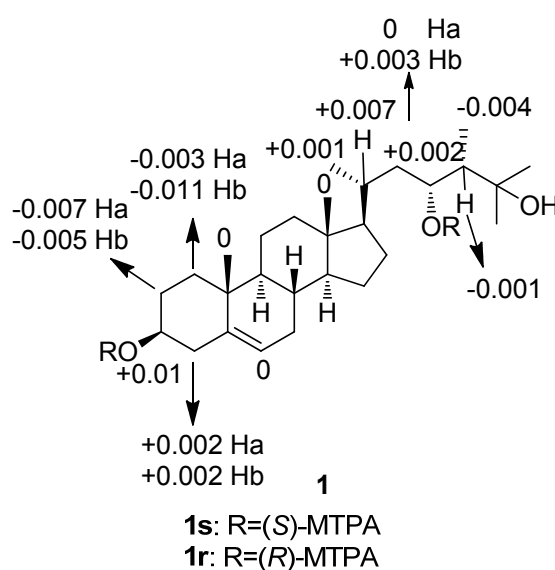
Table 2. ¹³C-NMR data for compounds 1–3.

C#	1, ^a δ_C , type	2, ^b δ_C , type	3, ^a δ_C , type
1	37.2, CH ₂	33.2, CH ₂	34.9, CH ₂
2	24.3, CH ₂	22.2, CH ₂	21.1, CH ₂
3	71.8, CH	67.9, CH	67.3, CH
4	42.3, CH ₂	31.6, CH ₂	28.2, CH ₂
5	140.8, C	75.5, C	30.7, CH
6	121.6, CH	77.8, CH	76.1, CH
7	31.7, CH ₂	32.5, CH ₂	31.4, CH ₂
8	31.9, CH	32.2, CH	31.9, CH
9	50.1, CH	46.2, CH	45.2, CH
10	36.5, C	39.6, C	38.5, C
11	21.1, CH ₂	29.1, CH ₂	40.5, CH ₂
12	39.7, CH ₂	41.0, CH ₂	73.7, CH
13	42.5, C	43.9, C	42.7, C
14	56.6, CH	57.3, CH	55.8, CH
15	21.0, CH ₂	25.2, CH ₂	24.1, CH ₂
16	28.5, CH ₂	29.3, CH ₂	29.7, CH ₂
17	57.3, CH	57.4, CH	55.9, CH
18	11.8, CH ₃	12.6, CH ₃	12.2, CH ₃

Table 2. Cont.

19	19.4, CH ₃	17.1, CH ₃	16.5, CH ₃
20	35.0, CH	37.8, CH	36.3, CH
21	23.2, CH ₃	15.3, CH ₃	19.0, CH ₃
22	44.2, CH ₂	36.3, CH ₂	39.9, CH ₂
23	75.8, CH	29.1, CH ₂	30.6, CH ₂
24	48.9, CH	46.4, CH	45.4, CH
25	75.2, C	74.2, C	75.3, C
26	30.7, CH ₃	26.0, CH ₃	26.2, CH ₃
27	30.7, CH ₃	27.2, CH ₃	27.2, CH ₃
28	14.1, CH ₃	19.6, CH ₃	14.8, CH ₃
CH ₃ CO	–	21.4, CH ₃	21.4, CH ₃
CH ₃ CO	–	172.1, C	164.5, C

^a Spectra were measured in CDCl₃ (150 MHz); ^b Spectra were measured in CD₃OD (150 MHz).

Figure 2. ¹H-¹H COSY (—) and HMBC (→) correlations for compounds 1–3.Figure 3. $\Delta\delta$ values ($\delta_{(S)} - \delta_{(R)}$) for the MTPA esters of compound 1.

Compound 2 was obtained as a white, amorphous powder with $[\alpha]_D^{25} -45.6$ (*c* 0.90, CH₃OH). Its positive ion HRESIMS revealed a pseudo molecular ion peak at m/z 515.3711 $[M + Na]^+$ (calcd for C₃₀H₅₂O₅Na 515.3712), corresponding to the molecular formula C₃₀H₅₂O₅, possessing five degrees of

unsaturation. IR spectrum showed its absorption at 3234–3587 cm^{-1} for hydroxyl groups, and 1652 cm^{-1} for carbonyl (acetate) group. Comparison of the $^1\text{H-NMR}$ (Table 1) and $^{13}\text{C-NMR}$ (Table 2) data of **2** with those of the known compound **4** [17], revealed that **2** shares the same structure nucleus as **4**, differing from **4** only at the side chain where the C-25 was oxygenated, in agreement with the mass data. The oxygenation of the C-25 caused the $^{13}\text{C-NMR}$ resonance of C-26 and C-27 to be shifted significantly downfield (from δ_{C} 21.5/21.6 to 26.0/27.2) and two singlet methyl signals appeared in **2** [δ_{H} 1.09 (3H, s), 1.10 (3H, s)] instead of two doublet methyls in **4** [δ_{H} 0.77 (3H, d, $J = 6.6$ Hz), 0.78 (3H, d, $J = 6.6$ Hz)] in the $^1\text{H-NMR}$ spectrum. According to these data, compound **2** was assigned as the 25-OH derivative of **4**. In addition, the HMBC (Figure 2) correlation from H-6 to ester carbonyl carbon at δ_{C} 172.1 (CH_3CO), suggesting that the acetoxy group was positioned at C-6. The assigned relative configuration at C-6 was confirmed by that H-6 (δ_{H} 4.68 (1H, t, $J = 2.4$ Hz)) was coupled with H-7 α (equatorial) (δ_{H} 1.47) with a small coupling constant of 2.4 Hz. Consequently, H-6 was in equatorial orientation, indicating the location of the acetoxy group was at axial position. Furthermore, the NOE cross-peaks observed between H₃-19 (δ_{H} 1.14) and both H-4 β (δ_{H} 1.77) and H-2 β (δ_{H} 2.01), and the absence of NOE correlations between H₃-19 and both H-3 and H-6 implied that H-3 and H-6 are both α oriented. At the same time the NOE correlations observed between H-4 α and both H-3 and H-6 further confirmed that H-3 and H-6 were α oriented. The configuration of the side chain of **2** was also confirmed by NOE correlations from H₃-18 to H-20. So the structure of compound **2** was determined as (24*S*)-ergostane-6-acetate-3 β ,5 α ,6 β ,25-tetraol.

Compound **3** was obtained as a white, amorphous powder with $[\alpha]_{\text{D}}^{25} -26.6$ (c 0.50, CH_3OH). It was found to have the same molecular formula ($\text{C}_{30}\text{H}_{52}\text{O}_5$) as **2**, as determined from high-resolution mass measurements which revealed a pseudo molecular ion peak at m/z 515.3710 [$\text{M} + \text{Na}$] $^+$, (calcd for $\text{C}_{30}\text{H}_{52}\text{O}_5\text{Na}$ 515.3712). Both compounds (**2** and **3**) showed similarity in the ^1H NMR and ^{13}C NMR spectra (Tables 1 and 2), with the most significant difference being the chemical shifts of C-5 (δ_{C} 75.5, C in **2** vs. δ_{C} 30.7, CH in **3**) and C-12 (δ_{C} 41.0, CH_2 in **2** vs. δ_{C} 73.7, CH in **3**). This indicated that the location of a hydroxyl group in **3** was different from that of in **2**. The established planar structure of **3** was further supported by the 2D NMR spectra. The diagnostic HMBC correlation from H₃-18 to C-12 and ^1H - ^1H COSY correlations between H-12 and H-11 (Figure 2) led the location of the hydroxyl group at C-12. The relative configuration of C-12 was established by comparison with the known compound **8** (3 α -acetoxy-12 α -hydroxy-5 β -cholan-24-oic acid). The small coupling constant of H-12 (δ_{H} 3.99, 1H, t, $J = 2.6$ Hz) in **8** means that H-12 is at equatorial position according to the literature [23]. Whereas the large coupling constant of H-12 (δ_{H} 4.31, 1H, td, $J = 7.2, 0.6$ Hz) in **3** supported the axial position of H-12. Moreover, NOE correlations observed from both H-14 and H-9 to H-12 and the absence of NOE correlations between H₃-18 and H-12 also implied H-12 was at α orientation. The chemical shift of H-3 (δ_{H} 4.09) suggested that 3-OH was β oriented and H-5 was α oriented by comparison with the $^1\text{H-NMR}$ data of 3 β -hydroxy-5 α -oxygenated A/B *trans* sterols [24,25]. Based on the above analysis, the relative configuration of **3** was assigned, and the structure was elucidated as (24*S*)-ergostane-6-acetate-3 β ,6 β ,12 β ,25-tetraol.

The structures of compounds **4**, **5** and **6** were identified as 24(*S*)-methylcholestane-3 β ,5 α ,6 β -triol-6-monoacetate [17], 24-methylenecholestane-3 β ,5 α ,6 β -triol-6-monoacetate [17], and ergost-24(28)-en-3 β ,5 α ,6 β -triol [18], respectively, by comparison of their spectroscopic data with those in the literature.

All the isolated compounds (**1–6**) were evaluated for their cytotoxic activity against a panel of five human tumor cell lines (Hela, HL-60, K562, A-549 and SMMC-7721) and lethality toward brine shrimp *A. salina*. Only compound **5** exhibited moderate cytotoxicity against K562 cell line with an IC₅₀ value of 3.18 μM. Moreover, compound **5** also displayed strong lethality toward brine shrimp *A. salina* with a LC₅₀ value of 0.96 μM. For the other compounds, no cytotoxic activity at the concentration of 10 μM and no lethality toward brine shrimp at 25 μg/mL were found.

3. Experimental Section

3.1. General Experimental Procedures

Optical rotations were measured in methanol using a JASCO P-1020 digital polarimeter. UV spectra were recorded on a Beckman DU 640 spectrophotometer. IR spectra were measured on a Bruker VECTOR 22 spectrophotometer. ¹H- and ¹³C-NMR spectra were recorded on a JEOL Eclips-600 spectrometer. ESIMS and HRESIMS were measured on a Q-TOF Ultima Global GAA076 LC mass spectrometer. Silica gel (Qing Dao Hai Yang Chemical Group Co.; 200–300 and 300–400 mesh), octadecylsilyl silica gel (Unicorn; 45–60 μm) and Sephadex LH-20 (Amersham Biosciences) were used for column chromatography (CC). Precoated silica gel plates (Yan Tai Zi Fu Chemical Group Co.; G60, F-254) were used for thin layer chromatography (TLC). Semi-preparative HPLC was performed on a Waters 1525 system using a semi-preparative C18 (Kromasil 7 μm, 10 × 250 mm) column coupled with a Waters 2996 photodiode array detector.

3.2. Animal Materials

Soft coral *Sinularia* sp. was collected from the coral reef of Weizhou Island in the South China Sea in September 2008, and was identified by Prof. Hui Huang, South China Sea Institute of Oceanology, Chinese Academy of Sciences of China. The voucher specimen (No. GX-WZ-2008002-4) was deposited in the Key Laboratory of Marine Drugs, the Ministry of Education, Ocean University of China, Qingdao, China.

3.3. Extraction and Isolation

The frozen animals (dry weight 559.7 g) were cut into small pieces and exhaustively extracted with EtOH once (3000 mL) and CHCl₃–CH₃OH (1:1) for six times (3000 mL × 6) successively at room temperature. The organic extracts were evaporated to give a residue, which was suspended into H₂O and partitioned with ethyl acetate. The ethyl acetate fraction was concentrated under reduced pressure to give a residue (28.0 g), which was subjected to gradient silica gel chromatography, eluting with 0%–100% ethyl acetate in light petroleum ether and 20%–100% CH₃OH in CHCl₃ to afford nine fractions (Fr. 1–Fr. 9). Fr. 6 was fractionated on silica gel column chromatography eluting with petroleum ether–ethyl acetate (5:1–1:2), and then chromatographed on Sephadex LH-20 eluted with petroleum ether–CHCl₃–MeOH (2:1:1) to give Fr.61. Further purification of Fr. 61 by semi-preparative HPLC yielded compound **4** (8.4 mg). Fr. 7 and Fr. 8 were firstly isolated by repeated silica gel chromatography, further purified by Sephadex LH-20 (CHCl₃–MeOH 1:1) and reversed-phase silica gel chromatography, and finally subjected to RP-HPLC to yield **5** (27.8 mg) and **6** (22.9 mg),

respectively. Fr. 9 was chromatographed on silica gel eluting with petroleum ether–EtOAc (1:3), and further purified by RP-HPLC (MeOH/H₂O 90:10, flow rate of 2.0 mL/min) to afford **1** (1.8 mg, $t_R = 26.5$ min), **2** (11.3 mg, $t_R = 27.5$ min), and **3** (1.3 mg, $t_R = 32.7$ min) successively.

Compound (**1**): White amorphous powder; $[\alpha]_D^{25} -22.4$ (c 0.05, CH₃OH); IR (KBr) ν_{\max} 3410–3584 cm⁻¹; UV (MeOH) λ_{\max} : 198 nm, ¹H NMR (CDCl₃, 600 MHz) and ¹³C NMR (CDCl₃, 150 MHz) data in Tables 1 and 2; HRESIMS m/z 455.3498 [M + Na]⁺ (calcd for C₂₈H₄₈O₃Na, 455.3496).

Compound (**2**): White amorphous powder; $[\alpha]_D^{25} -45.6$ (c 0.90, CH₃OH); IR (KBr) ν_{\max} at 3234–3587, 1652 cm⁻¹; UV (MeOH) λ_{\max} : 195 nm, ¹H NMR (CD₃OD, 600 MHz) and ¹³C NMR (CD₃OD, 150 MHz) data in Tables 1 and 2; HRESIMS m/z 515.3711 [M + Na]⁺ (calcd for C₃₀H₅₂O₅Na, 515.3712).

Compound (**3**): White amorphous powder; $[\alpha]_D^{25} -26.6$ (c 0.50, CH₃OH); IR (KBr) ν_{\max} 3434, 3214, 1638 cm⁻¹; UV (MeOH) λ_{\max} : 197 nm, ¹H NMR (CDCl₃, 600 MHz) and ¹³C NMR (CDCl₃, 150 MHz) data in Tables 1 and 2; HRESIMS m/z 515.3710 [M + Na]⁺ (calcd for C₃₀H₅₂O₅Na, 515.3712).

3.4. Preparation of the (S)-and (R)-MTPA Esters of **1**

Compound **1** (0.5 mg) was dissolved in 500 μ L of pyridine, and dimethylaminopyridine (2.0 mg) and (R)-MTPACl (10 μ L) were then added in sequence. The reaction mixture was stirred for 24 h at room temperature, and 1 mL of H₂O was then added. The solution was extracted with 5 mL of CH₂Cl₂ and the organic phase was concentrated under reduced pressure. Then the residue was purified by semi-preparative HPLC (100% MeOH) to yield (S)-MTPA ester **1s** (0.3 mg, $t_R = 23.40$ min). By the same procedure, the (R)-MTPA ester **1r** (0.3 mg, $t_R = 26.21$ min) was obtained from the reaction of **1** (0.5 mg) with (S)-MTPACl (10 μ L). Selected ¹H NMR (CDCl₃, 600 MHz) of (S)-MTPA ester (**1s**): δ 7.08–7.43 (5H, Ph), 5.348 (1H, m, H-23), 5.30 (1H, s, H-6), 5.17 (1H, m, H-3), 2.769 (1H, m, H-2a), 2.766 (1H, m, H-4a), 2.046 (1H, m, H-1a), 2.045 (1H, m, H-20), 2.01 (1H, m, H-22), 1.994 (1H, m, H-2b), 1.986 (1H, m, H-4b), 1.350 (1H, m, H-1b), 1.314 (1H, m, H-22), 1.269 (1H, m, H-24), 1.25 (6H, s, H₃-26 and H₃-27), 0.994 (3H, d, $J = 6.6$ Hz, H₃-21), 0.99 (3H, s, H₃-19), 0.879 (3H, d, $J = 6.0$ Hz, H₃-28), 0.87 (3H, s, H₃-18); selected ¹H NMR (CDCl₃, 600 MHz) of (R)-MTPA ester (**1r**): δ 7.08–7.43 (5H, Ph), 5.346 (1H, m, H-23), 5.30 (1H, s, H-6), 5.16 (1H, m, H-3), 2.776 (1H, m, H-2a), 2.764 (1H, m, H-4a), 2.049 (1H, m, H-1a), 2.038 (1H, m, H-20), 2.01 (1H, m, H-22), 1.999 (1H, m, H-2b), 1.984 (1H, m, H-4b), 1.361 (1H, m, H-1b), 1.311 (1H, m, H-22), 1.270 (1H, m, H-24), 1.25 (6H, s, H₃-26 and H₃-27), 0.993 (3H, d, $J = 6.6$ Hz, H₃-21), 0.99 (3H, s, H₃-19), 0.883 (3H, d, $J = 6.0$ Hz, H₃-28), 0.87 (3H, s, H₃-18).

3.5. Cytotoxicity Assay

The cytotoxicity against Hela (cervical cancer cells), HL-60 (Human promyelocytic leukemia cells), A-549 (human lung epithelial carcinoma), SMMC-7721 (human hepatocellular carcinoma cell line), and K562 (human immortalised myelogenous leukemia line) cell lines were evaluated by using SRB [26] and MTT [27] methods, respectively, according to the protocols described in previous literature. The test of brine shrimp toxicity on *A. salina* was performed according to standard protocols [28,29].

4. Conclusions

In our continuing discovery for biological secondary metabolites from marine invertebrates in the South China Sea, this study provided a series of polyoxygenated sterols. The discovery of new compounds **1–3** has added to an extremely diverse and complex array of soft coral sterols. Further studies should be conducted to unambiguously establish their absolute configurations by total synthesis as well as to understand their biological/ecological roles in the life cycle of the animal.

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