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Dithiocarbamates effectively inhibit the α -carbonic anhydrase from *Neisseria* gonorrhoeae

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ABSTRACT

Recently, inorganic anions and sulphonamides, two of the main classes of zinc-binding carbonic anhydrase inhibitors (CAIs), were investigated for inhibition of the α -class carbonic anhydrase (CA, EC 4.2.1.1) from *Neisseria gonorrhoeae*, NgCA. As an extension to our previous studies, we report that dithiocarbamates (DTCs) derived from primary or secondary amines constitute a class of efficient inhibitors of NgCA. K_Is ranging between 83.7 and 827 nM were measured for a series of 31 DTCs that incorporated various aliphatic, aromatic, and heterocyclic scaffolds. A subset of DTCs were selected for antimicrobial testing against *N. gonorrhoeae*, and three molecules displayed minimum inhibitory concentration (MIC) values less than or equal to 8 μ g/mL. As NgCA was recently validated as an antibacterial drug target, the DTCs may lead to development of novel antigonococcal agents.

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Carbonic anhydrase; inhibitor; dithiocarbamate; *Neisseria gonorrhoeae*; antibacterials

1. Introduction

A decase ago, prokaryotic carbonic anhydrases (CAs, EC 4.2.1.1) were proposed as drug targets for development of novel antibacterials¹. CAs catalyse the interconversion between CO₂ and bicarbonate, which generate a pH imbalance; CAs are widespread in bacteria and play an important role in various metabolic functions^{2,3}. Bacteria encode at least four genetic families of CAs, including the α -, β -, γ -, and ι -CAs, with many species containing more than one class and more than one CA isoform; however the functions of these different CAs have only recently started to be understood in detail¹⁻³. Although comprehensive in vitro inhibition studies of bacterial CAs are available^{1,2}, these results have only recently been validated in vivo. Seminal reports of Flaherty's and Seleem's groups showed that in some bacteria, such as in vancomycin-resistant enterococci (VRE) or Neisseria gonorrhoeae, clinically used sulphonamide CA inhibitors (CAIs) possess potent antibacterial activity^{4,5}. N. gonorrhoeae is a sexually transmitted pathogen that is becoming a global health concern due to increased resistance to a wide range of antibioticsincluding next generation cephalosporins^{6,7}. Acetazolamide, the CAI par excellence, and some of its newly designed derivatives were recently shown to be bacteriostatic against N. gonorrhoeae with minimum inhibitory concentration values as low as 0.25 μ g/mL and no toxicity obseved to host cells⁵. Sulphonamides, of which acetazolamide belongs to, are one of the main classes of CAIs, and their

interaction with bacterial CAs from various pathogens has been extensively studied in the last decade^{8–11}. As there is an urgent need for novel antibacterials, including antigonococcal agents, a deeper investigation of CA and profiling various classes of CAIs may be of great interest. A previous study of anion inhibitors found interesting inhibitory effects of *N*,*N*-diethyl-ditiocarbamate [5b], which was as a low micromolar inhibitor of the α -CA *N. gonorrhoeae* (NgCA). Based upon this previous study, we investigated dithiocarbamates as inhibitors of NgCA.

2. Materials and methods

2.1. Enzymology and CA activity and inhibition measurements

An Applied Photophysics stopped-flow instrument was used to assay the CA- catalysed CO_2 hydration activity¹². Phenol red (0.2 mM) was used as a pH indicator, working at the absorbance maximum of 557 nm, with 10 mM HEPES (pH 7.4) as a buffer, and in the presence of 10 mM NaClO₄ to maintain constant ionic strength, in order to follow the initial rates of the CA-catalysed CO_2 hydration reaction for a period of 10–100 s. The CO_2 concentrations ranged from 1.7 to 17 mM for the determination of the kinetic parameters and inhibition constants. For each inhibitor, at least six traces of the initial 5–10% of the reaction were used to determine the initial velocity. The uncatalyzed rates were determined in the same manner and subtracted from the total

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observed rates. Stock solutions of inhibitors (10–20 mM) were prepared in distilled-deionized water, and dilutions up to $0.01 \,\mu$ M were done thereafter with the assay buffer. Inhibitor and enzyme solutions were preincubated together for 15 min at room temperature prior to the assay, in order to allow for the formation of the E-l complex. The inhibition constants were obtained by non-linear least-squares methods using Prism 3 and the Cheng-Prusoff equation, as reported earlier^{13,14}, and represent the mean from at least three different determinations. The NgCA concentration in the assay system was 6.3 nM. The NgCA used was a recombinant enzyme obtained in-house, as described earlier^{5,15,16}.

2.2. Chemistry

DTCs **1–30** were previosuly reported by one of our groups^{17,18} and were of > 99% purity. DTC **31**, acetazolamide, buffers and other reagents are commercially available from Sigma-Aldrich (Milan, Italy).

2.3. Bacterial strains and media

Strains and media used in this study were previously reported by our group^{5,19}. *N. gonorrhoeae* strains used in the study were clinical isolates obtained from the Centres for Disease Control and Prevention (CDC). Media and reagents were purchased commercially: brucella broth, IsoVitaleX, and chocolate II agar plates (Becton, Dickinson and Company, Cockeysville, MD, USA), yeast extract and dextrose (Fisher Bioreagents, Fairlawn, NJ, USA), protease peptone (Oxoid, Lenexa, KS, USA), haematin, pyridoxal, and nicotinamide adenine dinucleotide (NAD) (Chem-Impex International, Wood Dale, IL, USA), and phosphate buffered saline (PBS) (Corning, Manassas, VA, USA).

2.4. Antibacterial activity of DTCs against *N. gonorrhoeae* strains

The (MICs of DTCs compounds were carried out using the broth microdilution method as described previously^{5,19}. Briefly, bacterial strains were grown for 24 h on GC chocolate agar II, at 37° C in presence of 5% CO₂. Then a bacterial suspension equivalent to 1.0 McFarland standard was prepared and diluted in brucella broth supplemented with yeast extract, protease peptone, haematin, pyridoxal, NAD, and IsoVitaleX, to achieve a bacterial concentration of about 1×10^6 CFU/mL. Test agents were added in the 96-

well plates and serially diluted along the plates. Plates were then, incubated for 24 h at 37° C either aerobically or in the presence of 5% CO_2 before determining the MICs as observed visually.

3. Results and discussion

Sulphonamide-type CAIs were first used to inhibit growth of N. gonorrhoeae in vitro in the 1960s; however, it was not untill the 1990s that Carter's group reported the presumed presence of CAs in N. gonorrhoeae by using a monospecific antibody prepared against the purified Neisseria sicca enzyme¹⁵. This enzyme was thereafter purified and characterised in 1997 by Lindskog's group¹⁶, who showed that NgCA is an α -class enzyme that possesses a high catalytic activity, with a k_{cat} for the CO₂ hydration reaction of $1.7 \times 10^6 \text{ s}^{-1}$ ¹⁷. The same group showed that NgCA was inhibited by metal complexing anions such as cyanide, cyanate, thiocyanate, and azide (as determined by using the esterase activity of the enzyme with 4-nitrophenyl acetate as a substrate¹⁶) as well as by the sulphonamide acetazolamide (5-acetamido-1,3,4thiadiazole-2-sulphonamide)¹⁶. Thereafter, we reported a comprehensive anion inhibition study of NgCA [5b], which found that the most effective inhibitors were sulfamide, sulphamic acid, and N,Ndiethyl-dithiocarbamate. This compound possesses the CS₂⁻ zincbinding group (ZBG), also present in trithiocarbonate (TTC)¹⁷, which has been shown via X-ray crystallography on human CAs (hCAs) to bind in a monodentate fashion to the metal ion from the enzyme's active site to displace the nucleophile (water or hydroxide ion) that is essential in the catalytic process¹⁷. The Xray structure of TTC bound to hCA II led thereafter to the discovery of DTCs and their derivatives (monothiocarbamates and xanthates) as potent CAIs^{18,20}. X-ray crystallography of some DTCs bound to hCA II demonstrated that their ZBG is coordinated in a monodentate fashion to the metal ion whereas the organic scaffold participates in a range of favourable interactions with the active site amino acid residues¹⁸ – Figure 1.

Thus, we decided to investigate a series of previously reported DTCs¹⁸, types **1–30** together with the *N*,*N*-diethyl derivative **31**, for their interaction with NgCA (Table 1). The following structureactivity relationship (SAR) may be observed from the data presented in Table 1:

i. The most effective NgCA inhibitors among the investigated DTCs were compounds **1**, **20** and **29**, which showed K₁s in the range of 83.7–136 nM. It is interesting to note that both

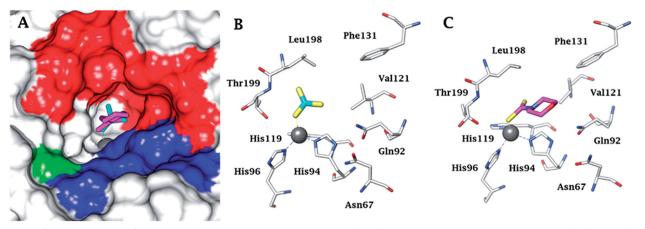


Figure 1. (A) Surface representation of hCA II active site in adduct with superimposed trithiocarbonate (cyan, PDB 3K7K) and the DTC morpholinocarbodithioate 23 (magenta, PDB 3P5A). The hydrophobic half of the CA active site is shown in red, and the hydrophilic one in blue; the proton shuttle residue His64 is shown in green. Cartoon view of hCA II active site in complex with B) trithiocarbonate and C) DTC 23.

Table 1. Inhibition constants (K ₁ s) of DTC inhibitors 1-31 against hCA I, II, and NgCA by a stopped flow CO ₂ hydration assay, using aceta	azolamide (AAZ) as the
standard drug ¹² .	

		K _i (nM) ^a		
DTC	Structure	hCA I	hCA II	NgCA
1	H ₂ NO ₂ S N SNa H	97.5	48.1	83.7
2	SNa NH SNa	425	107.0	259
3	N N H SK	85.9	35.8	568
4	HO	295	24.3	438
5	но К	706	41.7	413
6	→ 0 → N → SK	683	13.2	538
7	NaS ONa	485	80.1	827
8	NaO NaO SNa	337	78.7	514
9	NaO NaO SNa	290	45.4	297
10	HONSNa	428	60.7	367
11	N SNa	615	65.9	473
12	N SNa	494	48.7	482
13	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}\\ \begin{array}{c} \end{array}\\ \end{array}\\ \end{array}$	496	80.5	242
14		109	8.9	335
14		107	0.7	(continued)

(continued)

Table 1. Continued.

		K _i (nM) ^a		
DTC	Structure	hCA I	hCA II	NgCA
	NaO N SNa			
15	H N N SK	910	47.9	451
16	H N S SNa	240	18.9	518
17	S N SK	252	30.1	731
18	NaO O S N SNa	84.7	78.5	672
19	S N SNa	434	60.2	723
20	O N H S S	415	67.2	84.4
21	HO	66.5	17.3	454
22	N SNa	0.97	0.95	554
23	N SNa	0.88	0.95	483
24	N SNa	69.9	25.4	654
25	N SNa	43.1	50.9	460
26	N SNa	1838	55.5	522
	\sim			

Table 1. Continued.

		K, (nM) ^a		
DTC	Structure	hCA I	hCA II	NgCA
DTC 27	N SNa	157	27.8	577
28	SNa HN SNa	31.9	13.5	276
29		12.6	0.92	136
30	NC SNa	48.4	40.8	365
31 ^b	N SNa	790	3100	5100
AAZ	_	250	12.0	75.0

^aMean from three different assays, determined using a stopped flow technique (errors were in the range of \pm 5–10% of the reported values); ^bfrom ref. [5b].

20 and 29 possess the same scaffold of piperazine-dithiocarbamate. However, in the case of 29 a second DTC function is incorporated, whereas for 20, a bulkier cyclohexyl-aminocarbonylmethyl moiety is present. This leads to an increased inhibitory effect in the case of 20 compared to 29 (84.4 versus 136 nM, Table 1), probably due to favourable contacts between the bulky tail and amino acid residues from the active site. The second observation pertains to compounds 1 and 2. Derivative 1 incorporated two ZBGs, the DTC and the sulphonamide ones, whereas the second structurally related derivative (2) lacks the sulphonamide moiety. It is likely in the case of 1 that sulphonamide is the dominant interacting group and participates in the enzyme inhibition process by binding to the zinc ion in the active site. This is however impossible for 2, which exhibited 3.1 times weaker NgCA inhibitory activity compared to 1. However, derivative 2 still significantly inhibited the NgCA CO₂ hydrase activity with a K_l of 259 nM.

- ii. Another small group of DTCs, including 2, 9, 13, and 28 showed K₁s in the range of 242 297 nM, which indicates that they are effective NgCA inhibitors. The next most effective inhibitors showed K₁s between 300 and 500 nM and included 4, 5, 10–12, 14, 15, 21, 23, 25, and 30. These compounds incorporated a variety of diverse aliphatic, aromatic, and heterocyclic scaffolds, and are derivatives of both primary and secondary amines. This proves that many diverse chemical entities may lead to the development of efficient DTC inhibitors of NgCA (Table 1).
- iii. The least effective inhibitors were 3, 6–8, 16–19, 22, 26, and 27, which showed K_is in the range of 514–827 nM. Finally, 31, the lead compound was the least effective DTC inhibitor,

with a K_1 of 5100 nM. In contrast, acetazolamide, a sulphonamide derivative, was an effective NgCA inhibitor, with an activity in the same range as the most effective DTCs mentioned above (Table 1).

iv. Many of the investigated DTCs were much more effective as inhibitors against hCA II than NgCA, whereas their activity on hCA I was in the same range as against the bacterial enzyme, i.e. in the high nanomolar range.

A subset of DTCs were selected for antibacterial testing against three clinical strains of *N. gonorrhoeae*. It has previously been established that bacteria will become less susceptible to CAIs in conditions that contain elevated levels of CO_2^{21} . Molecules were assayed in both ambient air conditions as well as conditions containing 5% CO₂ to assess for activity at the proposed intracellular NgCA. The three strains tested displayed reduced susceptibility towards the molecules under elevated CO₂ conditions suggesting that inhibition of NgCA is, at least partially, responsible for the antimicrobial activity of these molecules. The control antibiotic azithromycin, which has a different mechanism of action, did not display differential activity based on the culture conditions. This result provides confidence that the difference in CO₂ levels did not have unintended effects on the bacteria that would result in non-specific reduced susceptibility to the test agent.

It was observed that in this cohort, three DTCs, **1**, **22**, and **24** exhibited moderate antigonococcal activity. DTC **1** was the most potent molecule with a MIC value of $1-2 \mu g/mL$ against *N. gonor-rhoeae* (Table 2). This was followed by **22** (MIC = $2-4 \mu g/mL$) and **24** (MIC = $4-8 \mu g/mL$). DTCs **23** and **25** each displayed weak antibacterial activity against *N. gonorrhoeae* with MIC values ranging from 8 to $32 \mu g/mL$. It is interesting to note that while **1** was the

Table 2. Minimum inhibitor	concentrations of DTCs versus N.	gonorrhoeae clinical isolates.

	MIC values versus Neisseria gonorrhoeae strains (µg/mL)					
Test agents/	CDC 178		CDC 181		CDC 194	
Test agents/ Control antibiotics	5% CO ₂ ^a	Ambient air ^b	5% CO ₂ ^a	Ambient air ^b	5% CO ₂ ^a	Ambient air ^b
1	16	2	16	1	16	2
3	>64	>64	>64	>64	>64	64
22	>64	4	>64	2	>64	4
23	>64	32	>64	16	>64	8
24	32	8	32	4	32	8
25	>64	32	>64	16	>64	16
28	>64	>64	>64	>64	>64	64
29	>64	>64	>64	>64	>64	64
30	>64	64	>64	64	>64	64
AAZ	>64	2	>64	4	>64	2
Azithromycin	2	2	>64	>64	1	0.5

^aIndicates incubation in presence of 5% CO₂. ^bIndicates in ambient air.

most potent molecule against both NgCA and N. gonorrhoeae, the DTCs that exhibited moderate potency against N. gonorrhoeae (22 and 23) were among the weaker analogues versus NgCA ($K_1s >$ 500 nM). Moreover, the weakest DTCs, in terms of antigonococcal activity, were 23, 25, 28, 29, and 30 with MIC values $> 8 \mu g/mL$; however, these molecules were more potent inhibitor of NgCA with activities in the range of 136 - 460 nM. Several of these molecules contain polar functional groups such as morpholine (23), piperazine (28) and Di-DTC (29) moieties that may have an adverse effect on molecule accumulation within the Gram-negative bacterial cell, thus leading to reduced antigonococcal activity. As for DTC 25, this molecule contains hydrophobic linear alkyl chains that give rise to additional rotatable bonds that also may have an adverse effect on accumulation into Gram-negative bacterial cells^{22,23}. In summary, while the DTCs displayed moderateto-weak antibacterial activity against the N. gonorrhoeae strains tested, the data does suggest that the DTC functionality may be a useful modification to incorporate into a drug design campaign for development of new anti-gonococcal agents.

4. Conclusions

NgCA, a high-activity α -CA present in the genome of *N. gonor*rhoeae, was investigated for potential inhibition by a series of 31 DTCs derived from both primary and secondary amines. NgCA was inhibited by all investigated derivatives, with K_Is in the range of 83.7 nM - 5.1 μ M. The most effective NgCA inhibitors were contained piperazine-dithiocarbamates that showed activity with K₁s < 140 nM; however, these molecules did not display antibacterial activity in vitro against N. gonorrhoeae. Conversely, DTCs containing more hydrophobic amines did exhibit moderate antibacterial activity even though these analogs possessed reduced NgCA activity. This data suggests that DTCs could be incorporated as the zinc-binding groups in place of sulphonamides, into more traditional CAI molecular scaffolds. Since antibiotic resistance is well documented against many N. gonorrhoeae strains worldwide, finding alternative chemotypes to presently used drugs is relevant. Our study provides interesting steps regarding developing these types of enzyme inhibitors.

Disclosure statement

The authors have no relevant affiliations of financial involvement with any organisation or entity with a financial interest in or financial conflict with the subject matter or materials discussed in the manuscript. CT Supuran is Editor-in-Chief of the Journal of Enzyme Inhibition and Medicinal Chemistry. He was not involved in the assessment, peer review, or decision-making process of this paper. The authors have no relevant affiliations of financial involvement with any organisation or entity with a financial interest in or financial conflict with the subject matter or materials discussed in the manuscript. This includes employment, consultancies, honoraria, stock ownership or options, expert testimony, grants or patents received or pending, or royalties.

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