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Aqua(iminodiacetato-κ^3O,N,O')(1,10-phenanthroline-κ^2N,N')zinc(II) sesquihydrate

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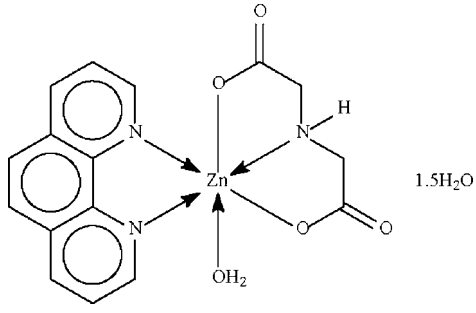
Key indicators: single-crystal X-ray study; T = 100 K; mean σ(C–C) = 0.007 Å; disorder in solvent or counterion; R factor = 0.048; wR factor = 0.146; data-to-parameter ratio = 15.6.

The iminodiacetate dianion in the title compound, $[Zn(C_4H_5NO_4)(C_{12}H_8N_2)(H_2O)]\cdot1.5H_2O$, chelates to the Zn^{II} center with its N and two O atoms. The metal atom is also chelated by the *N*-heterocycle and coordinated by one water molecule, leading to a distorted octahedral environment. The dianion, and coordinated and uncoordinated water molecules interact through O—H···O hydrogen bonds, generating a three-dimensional network. One of the two uncoordinated water molecules has half-site occupancy. The crystal studied was a non-merohedral twin with a 15% twin component.

Related literature

For the structure of zinc bis[iminodiacetate(1-)] tetrahydrate, see: Sinkha *et al.* (1975). For the dihydrated adenine adduct of zinc iminodiacetate, see: Morel *et al.* (2003). For the use of *PLATON* to separate twin fractions from diffraction data, see: Spek (2003).



Experimental

Crystal data

$[Zn(C_4H_5NO_4)(C_{12}H_8N_2)(H_2O)]\cdot1.5H_2O$
M_r = 421.70
Triclinic, $P\overline{1}$
a = 6.5989 (1) Å
b = 10.6440 (1) Å
c = 11.5456 (2) Å
α = 95.156 (1)°
β = 91.845 (1)°
γ = 92.190 (1)°
V = 806.56 (2) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 1.57 mm^{-1}
T = 100 (2) K
0.35 × 0.25 × 0.15 mm

Data collection

Bruker SMART APEX diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.610, T_{max} = 0.799
7242 measured reflections
3640 independent reflections
3455 reflections with $I > 2\sigma(I)$
R_{int} = 0.025

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.048
$wR(F^2)$ = 0.146
S = 1.22
3640 reflections
233 parameters
72 restraints
H-atom parameters constrained
$\Delta\rho_{max}$ = 1.02 e Å^{-3}
$\Delta\rho_{min}$ = −0.88 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O1w—H11···O1^{i}	0.84	2.17	2.801 (4)	132
O1w—H12···O3^{ii}	0.84	1.92	2.757 (4)	172
O2w—H21···O2^{i}	0.84	1.98	2.815 (5)	177
O2w—H22···O4^{iii}	0.84	1.92	2.756 (5)	177
O3w—H31···O2w	0.84	1.94	2.780 (9)	174

Symmetry codes: (i) $x+1, y, z$; (ii) $-x+1, -y+1, -z+2$; (iii) $-x+1, -y+2, -z+2$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2469).

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supplementary materials

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Aqua(iminodiacetato-κ^3O,N,O')(1,10-phenanthroline-κ^2N,N')zinc(II) sesquihydrate

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Comment

(type here to add)

Experimental

An methanol solution of zinc(II) nitrate hexahydrate (0.30 g, 1 mmol) and 1,10-phenanthroline (0.20 g, 1 mmol) was mixed with an aqueous solution of iminodiacetic acid (0.14 g, 1 mmol) and sodium hydroxide (0.08 g, 2 mmol). The mixture was briefly heated. The cool solution yielded a white solid. This was recrystallized from a water-methanol mixture to give colorless crystals.

Refinement

Carbon- and nitrogen-bound hydrogen atoms were placed at calculated positions (C–H 0.95–0.98 Å, N–H 0.88 Å) and were treated as riding on their parent atoms, with U(H) set to 1.2 times U_{eq}(C or N). The water H-atoms were placed in chemically-sensible positions on the basis of hydrogen bonding, but were not refined; their temperature factors were tied by a factor of 1.5.

For the three phenanthroline groups, the central six-membered ring was refined as a rigid hexagon of 1.39 Å sides. The temperature factors of the carbon atoms of this fused-ring system were restrained to be nearly isotropic.

The O3*w* atom gave a large temperature factor when allowed to refined at full occupancy. The occupancy could be refined, and this refined to nearly 0.5. As such, the occupancy was then fixed as exactly 0.5. This water molecule was within hydrogen bonding distance of only one other acceptor atom.

The structure is a non-merohedral twin. *PLATON* (Spek, 2003) was used to de-twin the structure. The twin component refined to 15%; the inclusion of the twin law lowered the R index from 6.4%. More importantly, it improved the weighting scheme. The final difference Fourier map was now diffuse, with the largest peak of slighly over 1 e Å^{-3} in the vicinity of C12.

Figures

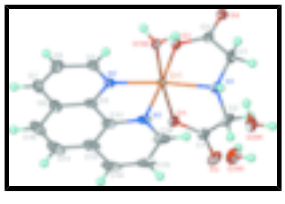


Fig. 1. Thermal ellipsoid plot (Barbour, 2001) of $Zn(H_2O)(C_4H_5NO_4)(C_{12}H_8N_2)\cdot 1.5H_2O$ at the 70% probability level. Hydrogen atoms are drawn as spheres of arbitrary radius. The O3*w* water molecule, which lies on a general position, has 0.5 occupancy.

Aqua(iminodiacetato-κ^3*O*,*N*,*O*')(1,10-phenanthroline- κ^2*N*,*N*')zinc(II) sesquihydrate

Crystal data

$[Zn(C_4H_5NO_4)(C_{12}H_8N_2)(H_2O)]\cdot 1.5H_2O$	$Z = 2$
$M_r = 421.70$	$F_{000} = 434$
Triclinic, $P\overline{1}$	$D_x = 1.736$ Mg m^{-3}
Hall symbol: -P 1	Mo $K\alpha$ radiation $\lambda = 0.71073$ Å
$a = 6.5989$ (1) Å	Cell parameters from 6223 reflections
$b = 10.6440$ (1) Å	$\theta = 2.5$–28.3º
$c = 11.5456$ (2) Å	$\mu = 1.57$ mm^{-1}
$\alpha = 95.156$ (1)º	$T = 100$ (2) K
$\beta = 91.845$ (1)º	Block, colorless
$\gamma = 92.190$ (1)º	0.35 × 0.25 × 0.15 mm
$V = 806.56$ (2) Å^3	

Data collection

Bruker SMART APEX diffractometer	3640 independent reflections
Radiation source: fine-focus sealed tube	3455 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{int} = 0.025$
$T = 100$(2) K	$\theta_{max} = 27.5$º
ω scans	$\theta_{min} = 1.8$º
Absorption correction: Multi-scan (SADABS; Sheldrick, 1996)	$h = -8\rightarrow 8$
$T_{min} = 0.610$, $T_{max} = 0.799$	$k = -13\rightarrow 13$
7242 measured reflections	$l = -14\rightarrow 14$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.048$	H-atom parameters constrained
$wR(F^2) = 0.146$	$w = 1/[\sigma^2(F_o^2) + (0.0241P)^2 + 4.8628P]$ where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.22$	$(\Delta/\sigma)_{max} = 0.001$
3640 reflections	$\Delta\rho_{max} = 1.02$ e Å^{-3}
233 parameters	$\Delta\rho_{min} = -0.88$ e Å^{-3}
72 restraints	Extinction correction: none
Primary atom site location: structure-invariant direct methods	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($Å^2$)

	x	*y*	*z*	U_{iso}*/U_{eq}	Occ. (<1)
Zn1	0.30260 (7)	0.63033 (5)	0.84683 (4)	0.01073 (15)	
O1	−0.0016 (5)	0.7036 (3)	0.8533 (3)	0.0154 (6)	
O2	−0.1646 (5)	0.8841 (3)	0.8519 (4)	0.0242 (8)	
O3	0.3125 (5)	0.6228 (3)	1.0256 (3)	0.0137 (6)	
O4	0.3975 (5)	0.7409 (3)	1.1911 (3)	0.0168 (7)	
O1W	0.6167 (5)	0.5782 (3)	0.8460 (3)	0.0137 (6)	
H11	0.6902	0.6376	0.8795	0.021*	
H12	0.6294	0.5131	0.8812	0.021*	
O2W	0.5864 (6)	1.0151 (3)	0.7055 (3)	0.0270 (8)	
H21	0.6601	0.9737	0.7475	0.040*	
H22	0.5886	1.0902	0.7349	0.040*	
O3W	0.7199 (12)	1.0113 (7)	0.4793 (7)	0.0287 (16)	0.50
H31	0.6705	1.0115	0.5455	0.043*	0.50
H32	0.7870	1.0790	0.4740	0.043*	0.50
N1	0.3717 (6)	0.8255 (3)	0.8921 (3)	0.0155 (8)	
H1	0.4735	0.8496	0.8507	0.019*	
N2	0.1966 (5)	0.4443 (4)	0.7994 (3)	0.0142 (7)	
N3	0.3044 (5)	0.6254 (4)	0.6611 (3)	0.0157 (7)	
C1	−0.0062 (7)	0.8228 (4)	0.8568 (4)	0.0153 (8)	
C2	0.1934 (7)	0.8985 (4)	0.8650 (5)	0.0207 (10)	
H2A	0.2137	0.9349	0.7901	0.025*	
H2B	0.1849	0.9696	0.9259	0.025*	
C3	0.4372 (7)	0.8367 (4)	1.0155 (4)	0.0178 (9)	
H3A	0.3811	0.9141	1.0538	0.021*	
H3B	0.5869	0.8478	1.0206	0.021*	
C4	0.3754 (6)	0.7252 (4)	1.0842 (4)	0.0127 (8)	
C5	0.1390 (6)	0.3577 (4)	0.8671 (4)	0.0149 (8)	
H5	0.1355	0.3802	0.9485	0.018*	
C6	0.0828 (7)	0.2336 (5)	0.8242 (5)	0.0200 (9)	
H6	0.0424	0.1735	0.8760	0.024*	
C7	0.0864 (7)	0.1993 (5)	0.7068 (4)	0.0198 (9)	
H7	0.0504	0.1151	0.6768	0.024*	
C9	0.1991 (4)	0.4127 (3)	0.67929 (18)	0.0146 (8)	
C8	0.1450 (5)	0.2916 (2)	0.6299 (2)	0.0186 (9)	
C10	0.1399 (5)	0.2654 (2)	0.5097 (3)	0.0265 (11)	
H10	0.1029	0.1827	0.4759	0.032*	
C11	0.1889 (5)	0.3603 (3)	0.43886 (18)	0.0271 (11)	
H11A	0.1854	0.3423	0.3567	0.032*	
C12	0.2430 (5)	0.4813 (3)	0.4883 (2)	0.0213 (10)	
C13	0.2481 (4)	0.5075 (2)	0.6085 (2)	0.0168 (9)	
C14	0.2936 (7)	0.5828 (6)	0.4185 (4)	0.0254 (11)	
H14	0.2912	0.5683	0.3360	0.030*	
C15	0.3446 (7)	0.6986 (5)	0.4710 (4)	0.0220 (10)	
H15	0.3765	0.7668	0.4262	0.026*	
C16	0.3497 (7)	0.7169 (5)	0.5941 (4)	0.0209 (10)	

H16	0.3874	0.7985	0.6304	0.025*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Zn1	0.0119 (2)	0.0110 (2)	0.0096 (2)	0.00101 (17)	0.00101 (17)	0.00212 (17)
O1	0.0118 (14)	0.0139 (14)	0.0205 (16)	−0.0007 (11)	0.0005 (12)	0.0029 (12)
O2	0.0160 (16)	0.0179 (16)	0.040 (2)	0.0051 (13)	0.0028 (15)	0.0070 (15)
O3	0.0180 (15)	0.0111 (14)	0.0121 (15)	0.0005 (12)	0.0027 (12)	0.0010 (11)
O4	0.0199 (16)	0.0168 (15)	0.0137 (15)	0.0026 (12)	−0.0020 (12)	0.0013 (12)
O1W	0.0139 (14)	0.0114 (14)	0.0162 (15)	−0.0003 (11)	−0.0003 (12)	0.0035 (12)
O2W	0.036 (2)	0.0172 (17)	0.0267 (19)	0.0008 (15)	−0.0018 (16)	−0.0029 (14)
O3W	0.038 (4)	0.023 (4)	0.026 (4)	0.005 (3)	0.006 (3)	0.003 (3)
N1	0.0130 (18)	0.0134 (17)	0.021 (2)	0.0021 (14)	0.0008 (14)	0.0051 (14)
N2	0.0089 (16)	0.0159 (18)	0.0175 (19)	0.0032 (13)	−0.0017 (14)	−0.0013 (14)
N3	0.0078 (16)	0.027 (2)	0.0135 (18)	0.0044 (14)	0.0017 (13)	0.0069 (15)
C1	0.014 (2)	0.017 (2)	0.016 (2)	0.0014 (16)	0.0028 (16)	0.0052 (17)
C2	0.013 (2)	0.014 (2)	0.037 (3)	0.0018 (16)	0.0014 (19)	0.0100 (19)
C3	0.023 (2)	0.0123 (19)	0.018 (2)	−0.0015 (17)	0.0011 (18)	−0.0002 (16)
C4	0.0091 (18)	0.0126 (19)	0.017 (2)	0.0027 (15)	0.0023 (15)	0.0016 (16)
C5	0.0088 (18)	0.019 (2)	0.016 (2)	0.0008 (15)	−0.0006 (15)	−0.0006 (16)
C6	0.016 (2)	0.018 (2)	0.027 (2)	0.0023 (17)	0.0023 (18)	0.0038 (18)
C7	0.014 (2)	0.019 (2)	0.025 (2)	0.0023 (17)	−0.0019 (18)	−0.0068 (18)
C9	0.0073 (18)	0.024 (2)	0.013 (2)	0.0044 (16)	0.0002 (15)	0.0001 (17)
C8	0.0083 (19)	0.024 (2)	0.023 (2)	0.0015 (16)	0.0000 (16)	−0.0039 (18)
C10	0.011 (2)	0.042 (3)	0.024 (2)	0.004 (2)	−0.0026 (18)	−0.012 (2)
C11	0.013 (2)	0.050 (3)	0.016 (2)	0.005 (2)	−0.0006 (17)	−0.006 (2)
C12	0.0075 (18)	0.041 (3)	0.016 (2)	0.0062 (18)	0.0046 (16)	0.0011 (19)
C13	0.0101 (19)	0.026 (2)	0.015 (2)	0.0059 (17)	0.0009 (15)	0.0034 (17)
C14	0.014 (2)	0.053 (3)	0.010 (2)	0.010 (2)	0.0018 (17)	0.008 (2)
C15	0.012 (2)	0.042 (3)	0.016 (2)	0.0095 (19)	0.0052 (16)	0.015 (2)
C16	0.013 (2)	0.033 (3)	0.018 (2)	0.0052 (18)	0.0019 (17)	0.0109 (19)

Geometric parameters (Å, °)

Zn1—O3	2.072 (3)	C2—H2B	0.9900
Zn1—N2	2.093 (4)	C3—C4	1.535 (6)
Zn1—N1	2.124 (4)	C3—H3A	0.9900
Zn1—N3	2.141 (4)	C3—H3B	0.9900
Zn1—O1W	2.166 (3)	C5—C6	1.401 (6)
Zn1—O1	2.182 (3)	C5—H5	0.9500
O1—C1	1.267 (5)	C6—C7	1.373 (7)
O2—C1	1.255 (5)	C6—H6	0.9500
O3—C4	1.278 (5)	C7—C8	1.433 (6)
O4—C4	1.233 (6)	C7—H7	0.9500
O1W—H11	0.8399	C9—C8	1.3900
O1W—H12	0.8400	C9—C13	1.3900
O2W—H21	0.8400	C8—C10	1.3900
O2W—H22	0.8400	C10—C11	1.3900

O3W—H31	0.8401	C10—H10	0.9500
O3W—H32	0.8399	C11—C12	1.3900
N1—C3	1.469 (6)	C11—H11A	0.9500
N1—C2	1.475 (6)	C12—C13	1.3900
N1—H1	0.8800	C12—C14	1.440 (6)
N2—C5	1.315 (6)	C14—C15	1.350 (8)
N2—C9	1.398 (4)	C14—H14	0.9500
N3—C16	1.329 (6)	C15—C16	1.415 (7)
N3—C13	1.376 (5)	C15—H15	0.9500
C1—C2	1.513 (6)	C16—H16	0.9500
C2—H2A	0.9900		
O3—Zn1—N2	98.01 (14)	N1—C3—H3A	108.3
O3—Zn1—N1	83.15 (14)	C4—C3—H3A	108.3
N2—Zn1—N1	172.78 (14)	N1—C3—H3B	108.3
O3—Zn1—N3	175.75 (14)	C4—C3—H3B	108.3
N2—Zn1—N3	79.24 (16)	H3A—C3—H3B	107.4
N1—Zn1—N3	99.99 (16)	O4—C4—O3	125.7 (4)
O3—Zn1—O1W	88.31 (12)	O4—C4—C3	117.1 (4)
N2—Zn1—O1W	92.50 (13)	O3—C4—C3	117.2 (4)
N1—Zn1—O1W	94.66 (13)	N2—C5—C6	122.7 (4)
N3—Zn1—O1W	88.57 (13)	N2—C5—H5	118.6
O3—Zn1—O1	90.74 (12)	C6—C5—H5	118.6
N2—Zn1—O1	93.69 (13)	C7—C6—C5	119.5 (5)
N1—Zn1—O1	79.15 (13)	C7—C6—H6	120.3
N3—Zn1—O1	92.67 (13)	C5—C6—H6	120.3
O1W—Zn1—O1	173.81 (12)	C6—C7—C8	119.5 (4)
C1—O1—Zn1	114.4 (3)	C6—C7—H7	120.3
C4—O3—Zn1	114.8 (3)	C8—C7—H7	120.3
Zn1—O1W—H11	109.5	C8—C9—C13	120.0
Zn1—O1W—H12	109.4	C8—C9—N2	121.7 (2)
H11—O1W—H12	109.5	C13—C9—N2	118.2 (2)
H21—O2W—H22	108.3	C10—C8—C9	120.0
H31—O3W—H32	110.0	C10—C8—C7	122.4 (3)
C3—N1—C2	114.7 (4)	C9—C8—C7	117.5 (3)
C3—N1—Zn1	105.8 (3)	C8—C10—C11	120.0
C2—N1—Zn1	109.4 (3)	C8—C10—H10	120.0
C3—N1—H1	108.9	C11—C10—H10	120.0
C2—N1—H1	108.9	C12—C11—C10	120.0
Zn1—N1—H1	108.9	C12—C11—H11A	120.0
C5—N2—C9	119.1 (4)	C10—C11—H11A	120.0
C5—N2—Zn1	128.5 (3)	C13—C12—C11	120.0
C9—N2—Zn1	112.3 (3)	C13—C12—C14	118.0 (3)
C16—N3—C13	118.5 (4)	C11—C12—C14	122.0 (3)
C16—N3—Zn1	129.7 (4)	N3—C13—C12	121.9 (2)
C13—N3—Zn1	111.8 (2)	N3—C13—C9	118.1 (2)
O2—C1—O1	125.1 (4)	C12—C13—C9	120.0
O2—C1—C2	116.7 (4)	C15—C14—C12	119.5 (4)
O1—C1—C2	118.3 (4)	C15—C14—H14	120.2
N1—C2—C1	114.4 (4)	C12—C14—H14	120.2

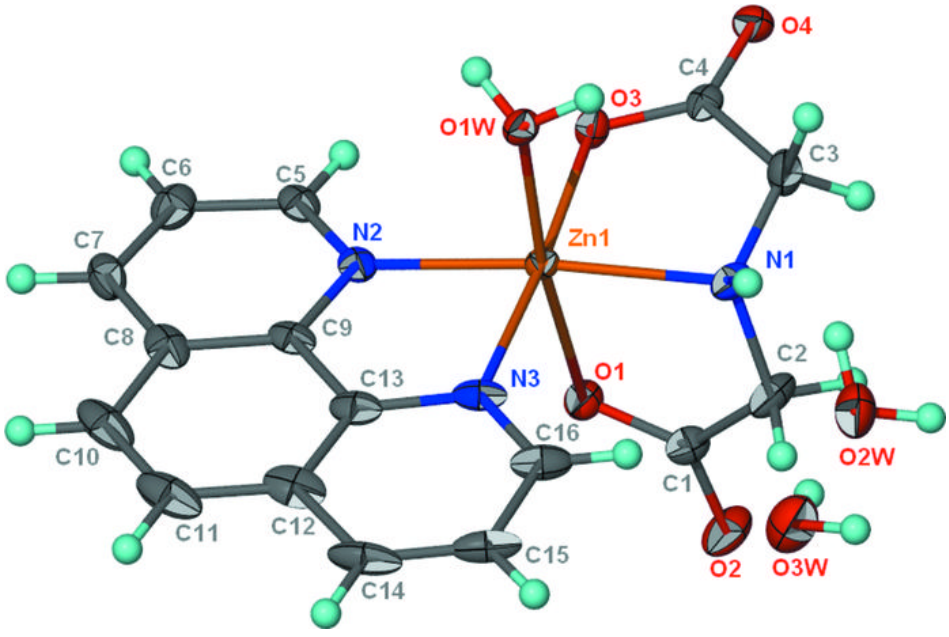
N1—C2—H2A	108.7	C14—C15—C16	119.0 (5)
C1—C2—H2A	108.7	C14—C15—H15	120.5
N1—C2—H2B	108.7	C16—C15—H15	120.5
C1—C2—H2B	108.7	N3—C16—C15	123.0 (5)
H2A—C2—H2B	107.6	N3—C16—H16	118.5
N1—C3—C4	115.9 (4)	C15—C16—H16	118.5
O3—Zn1—O1—C1	95.8 (3)	Zn1—O3—C4—C3	−1.0 (5)
N2—Zn1—O1—C1	−166.1 (3)	N1—C3—C4—O4	−167.9 (4)
N1—Zn1—O1—C1	12.9 (3)	N1—C3—C4—O3	14.5 (6)
N3—Zn1—O1—C1	−86.8 (3)	C9—N2—C5—C6	1.0 (6)
N2—Zn1—O3—C4	179.4 (3)	Zn1—N2—C5—C6	−175.9 (3)
N1—Zn1—O3—C4	−7.8 (3)	N2—C5—C6—C7	−0.2 (7)
O1W—Zn1—O3—C4	87.1 (3)	C5—C6—C7—C8	−0.9 (7)
O1—Zn1—O3—C4	−86.8 (3)	C5—N2—C9—C8	−0.8 (5)
O3—Zn1—N1—C3	14.2 (3)	Zn1—N2—C9—C8	176.55 (15)
N3—Zn1—N1—C3	−162.9 (3)	C5—N2—C9—C13	176.0 (3)
O1W—Zn1—N1—C3	−73.6 (3)	Zn1—N2—C9—C13	−6.6 (3)
O1—Zn1—N1—C3	106.2 (3)	C13—C9—C8—C10	0.0
O3—Zn1—N1—C2	−110.0 (3)	N2—C9—C8—C10	176.8 (3)
N3—Zn1—N1—C2	72.9 (3)	C13—C9—C8—C7	−177.0 (3)
O1W—Zn1—N1—C2	162.3 (3)	N2—C9—C8—C7	−0.2 (4)
O1—Zn1—N1—C2	−17.9 (3)	C6—C7—C8—C10	−175.9 (3)
O3—Zn1—N2—C5	5.3 (4)	C6—C7—C8—C9	1.0 (5)
N3—Zn1—N2—C5	−178.0 (4)	C9—C8—C10—C11	0.0
O1W—Zn1—N2—C5	94.0 (4)	C7—C8—C10—C11	176.9 (3)
O1—Zn1—N2—C5	−85.9 (4)	C8—C10—C11—C12	0.0
O3—Zn1—N2—C9	−171.7 (2)	C10—C11—C12—C13	0.0
N3—Zn1—N2—C9	5.0 (2)	C10—C11—C12—C14	−179.6 (3)
O1W—Zn1—N2—C9	−83.1 (3)	C16—N3—C13—C12	1.8 (5)
O1—Zn1—N2—C9	97.0 (3)	Zn1—N3—C13—C12	−178.48 (15)
N2—Zn1—N3—C16	176.8 (4)	C16—N3—C13—C9	−179.4 (3)
N1—Zn1—N3—C16	4.1 (4)	Zn1—N3—C13—C9	0.3 (3)
O1W—Zn1—N3—C16	−90.4 (4)	C11—C12—C13—N3	178.8 (3)
O1—Zn1—N3—C16	83.5 (4)	C14—C12—C13—N3	−1.6 (4)
N2—Zn1—N3—C13	−2.9 (2)	C11—C12—C13—C9	0.0
N1—Zn1—N3—C13	−175.6 (2)	C14—C12—C13—C9	179.6 (3)
O1W—Zn1—N3—C13	89.9 (3)	C8—C9—C13—N3	−178.8 (3)
O1—Zn1—N3—C13	−96.2 (3)	N2—C9—C13—N3	4.2 (3)
Zn1—O1—C1—O2	174.2 (4)	C8—C9—C13—C12	0.0
Zn1—O1—C1—C2	−4.4 (5)	N2—C9—C13—C12	−176.9 (3)
C3—N1—C2—C1	−97.2 (5)	C13—C12—C14—C15	0.2 (5)
Zn1—N1—C2—C1	21.6 (5)	C11—C12—C14—C15	179.8 (3)
O2—C1—C2—N1	169.4 (4)	C12—C14—C15—C16	0.9 (7)
O1—C1—C2—N1	−11.9 (7)	C13—N3—C16—C15	−0.6 (6)
C2—N1—C3—C4	101.9 (4)	Zn1—N3—C16—C15	179.8 (3)
Zn1—N1—C3—C4	−18.9 (4)	C14—C15—C16—N3	−0.8 (7)
Zn1—O3—C4—O4	−178.3 (3)		

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N1—H1···O2w	0.88	2.64	3.388 (5)	143
O1w—H11···O1^{i}	0.84	2.17	2.801 (4)	132
O1w—H12···O3^{ii}	0.84	1.92	2.757 (4)	172
O2w—H21···O2^{i}	0.84	1.98	2.815 (5)	177
O2w—H22···O4^{iii}	0.84	1.92	2.756 (5)	177
O3w—H31···O2w	0.84	1.94	2.780 (9)	174

Symmetry codes: (i) *x*+1, *y*, *z*; (ii) −*x*+1, −*y*+1, −*z*+2; (iii) −*x*+1, −*y*+2, −*z*+2.

Fig. 1


O4
C4
O3
O1W
C3
C6
C5
N2
Zn1
N1
C7
C8
C9
C2
C13
N3
O1
C16
C10
C1
O2W
C11
C12
C15
O2
O3W
C14