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# Extending effective microwave absorbing bandwidth of CoNi bimetallic alloy derived from binary hydroxides

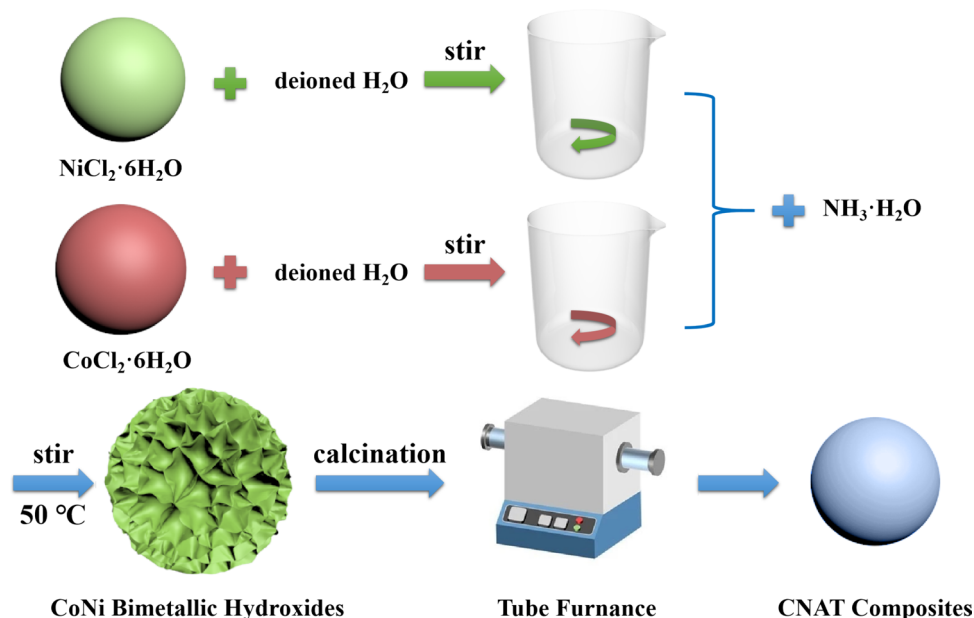
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Effectively broadening microwave absorbing frequency of pure magnetic substances remains a huge challenge. Herein, micro-perspective structures can be controlled through a calcination route. Satisfactorily, the composites prepared at the calcination temperature of 900 °C exhibit excellent microwave attenuation performance with a broad working frequency and appropriate paraffin filling ratio. Remarkably, the composites can reach an extremely high reflection loss (RL) value of  $-49.79$  dB, and the extended effective working frequency range (RL  $< -10$  dB) of 6.84 GHz can also be obtained. Superb magnetic loss, admirable dielectric loss, sufficient dipole polarization, as well as superior impedance matching should be band together for obtaining ideal microwave absorbers. The CoNi hydroxides derived bimetallic alloy composites were fabricated via a cost-effective and facile synthesis process, and this work aroused inspirations of designing high-performance microwave absorbers for maintaining the sustainable development.

Nowadays, rapid development of wireless communications, such as smart phones, portable computers and artificial intelligence, etc. has given rise to the spurting explosion of electromagnetic pollution<sup>1,2</sup>. In consideration of the sustainable development of the human beings, microwave absorbing materials with high-performance have aroused wide attention all over the world<sup>3</sup>. Due to the outstanding ferromagnetic characteristics, magnetic metallic materials such as Co, Ni and their alloys have unparalleled advantages among all kinds of candidates<sup>4</sup>.

Despite the high saturation magnetization and magnetocrystalline anisotropy, narrow effective bandwidth caused by inferior impedance matching has seriously limited their applications in practical situation<sup>5,6</sup>. In recent years, experts and scholars have hammered at broadening absorption bandwidth via introducing novel carbon-based materials, such as grapheme<sup>7</sup>, CNTs<sup>8</sup>, MXene<sup>9</sup>, etc. However, these kinds of raw materials are highly expensive, which have extremely impeded their industrial applications to some extent. Abundant researches on magnetic materials and their alloys have been reported to investigate the synthesis process and microwave absorption properties<sup>10,11</sup>. Take their high magnetic saturation, outstanding electrical conductivity and excellent temperature stabilization into account, magnetic materials and their alloys can be regarded as promising candidates for microwave absorption<sup>12</sup>. However, their stringent preparation conditions with complex and high-cost fabrication process limited the development of magnetic composites. Therefore, it is quite necessary to solve these problems. Bimetallic hydroxides as one kind of inorganic functional materials are attractive for extensive applications, such as supercapacitor<sup>13</sup>, catalyst<sup>14</sup>, and potentiometric sensors<sup>15</sup>, etc. Compared to conventional materials, bimetallic hydroxides materials have superior advantages of cost-effectiveness, chemical stability, environmentally nature, and so forth<sup>16–18</sup>. In this work, we take facile and cost-effective co-precipitation method to prepare the magnetic CoNi bimetallic hydroxides (CoNiBH). Fortunately, CoNi bimetallic hydroxide is beneficial to introduce cobalt and nickel positive ions, thus bringing about dual magnetic loss sources and large specific surface areas. Through controlling the heating temperatures, the CoNi bimetallic hydroxides derived products exhibited diverse morphology and size, which plays a significant role in achieving excellent impedance matching

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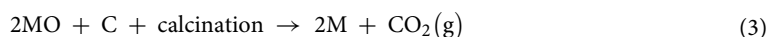
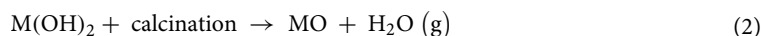
**Figure 1.** Schematic illustration of the synthesis process of CNAT hybrids samples.

properties. Therefore, assembling appropriate ingredients via simple co-precipitation and tunable calcinations route is an effective strategy for extending microwave absorbing bandwidth.

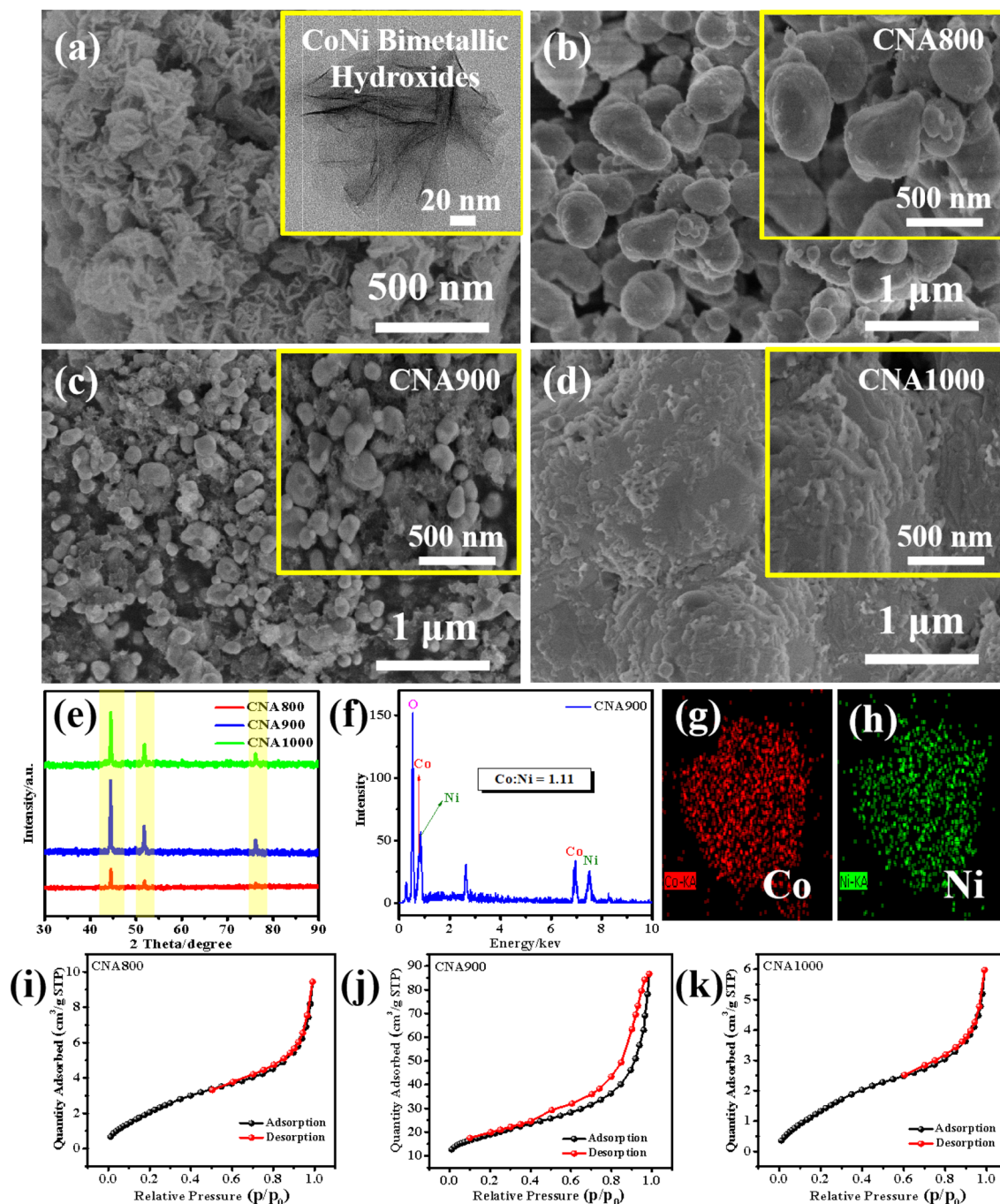
Fortunately, the as-prepared CoNi bimetallic alloys derived from CoNi double hydroxides was obtained. More importantly, appropriate impedance matching performance and extremely strong electromagnetic attenuation capacity were achieved. When tuning annealing temperature to  $900\text{ }^\circ\text{C}$ , the  $\text{RL}_{\min}$  value of  $-49.79\text{ dB}$  appeared at the frequency of  $15.68\text{ GHz}$ , and the broadest bandwidth was up to  $6.84\text{ GHz}$  ( $11.16\text{--}18\text{ GHz}$ ) with a thin layer thickness of  $2.1\text{ mm}$ . The results not only demonstrated that the calcination process has great significance for forming suitable particle shapes with favorable impedance matching, but also provided a novel rational design inspiration into widening working frequency range for practical applications.

## Results

The facile synthesis process of nanoflower-like bimetallic hydroxides is depicted in Fig. 1. Obviously, a series of typical calcinations treatment are carried out to fabricate CoNi alloy composites. The formation of CoNi alloys can be ascribed to two possible factors: (1) The residual ethanol caused by centrifugal washing and blast air drying process may bring a small amount of carbon source, thus, metallic oxides can be reduced to metals; (2) the residual  $\text{NH}_4\text{Cl}$  may produce  $\text{NH}_3$  (gas) and  $\text{HCl}$  (gas) at a high temperature under the nitrogen atmosphere, herein, metallic oxides can also be reduced to metals. The relevant reaction process can be described as the following equations<sup>19,20</sup>:

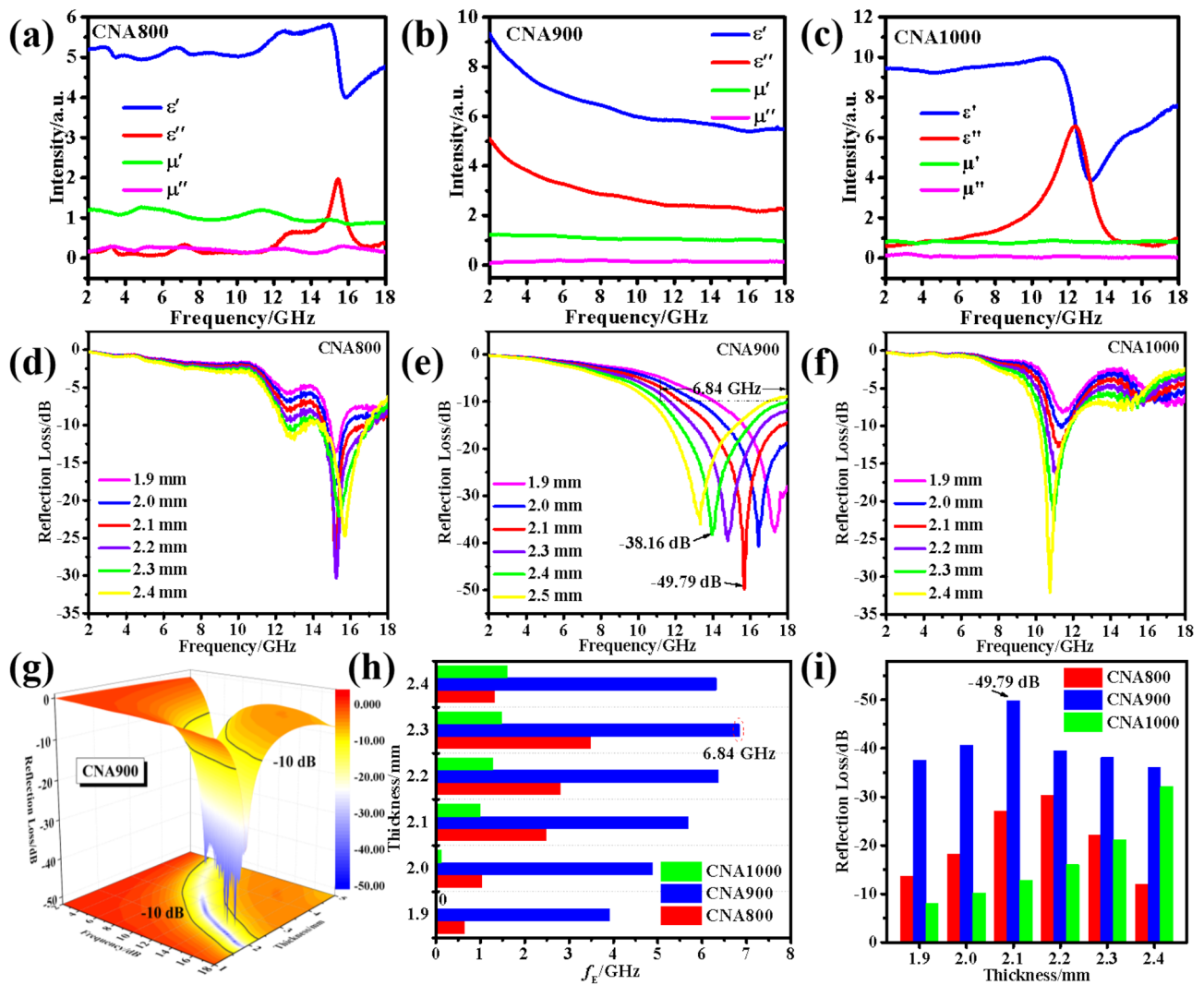


where  $\text{M}(\text{OH})_2$  means CoNi bimetallic hydroxides,  $\text{MO}$  stands for CoNi oxides and  $\text{M}$  signifies CoNi alloys. The morphologies of the CoNi bimetallic hydroxide precursors and final CNAT composites were systematically characterized, which can be seen from Fig. 2a–d. As described in Fig. 2a, flower-like hydroxides precursors with nanosheets intersecting morphology was fabricated initially via co-precipitation. With the increase of annealing temperature, one can find that the size of the final product exhibits an inverted “V” trend. For CNA800 and CNA900, the microstructure remains the form of nanoparticles. However, for CNA1000, particles aggregated together to a block shape, which may result in impedance mismatch. As provided in Fig. 2e, the typical diffraction peaks located between  $30^\circ$  and  $90^\circ$  can be well indexed to the (111), (200) and (220) planes of CoNi alloy, demonstrating that the above reaction equations are of high credibility level. Besides, all the peak positions are consistent with either metal Co (JCPDS No. 89-4301) or metal Ni (JCPDS No. 89-7128). Furthermore, intensity of the XRD patterns demonstrates the high purity and crystallinity of the CNAT composites. At the same time, the energy dispersive X-ray spectrometry image and element mapping results of CNA900 can be observed in



**Figure 2.** (a) SEM image of the CoNi bimetallic hydroxide precursors. Inset: TEM image of the as-prepared precursor. SEM images of (b) CNA800, (c) CNA900 and (d) CNA1000. (e) XRD pattern of CNA700/800/900. (f) EDS of CNA900 and (g,h) element EDS mapping of CNA900. The  $N_2$  adsorption–desorption isotherms of (i) CNA800, (j) CNA900 and (k) CNA1000, respectively.

Fig. 2f–h. It should be noticed that the atom ratio of Co and Ni in Fig. 2f is calculated to be 1.11, which is close to the ingredient ratio of the added raw materials, indicating that CoNi bimetallic particles have been successfully manufactured. Equally important, cobalt and nickel elements are marked by red dots and green dots in Fig. 2g,h, indicating homogeneous distribution of the two metal components. With regard to all samples, the smaller nanoparticles are presented, the more specific surface area is obtained. As shown in Fig. 2i–k, the BET surface areas of CNA800/900/1000 samples are 9.28, 67.04 and 6.50  $m^2/g$ , respectively, indicating that the CNA900 possesses the highest specific surface area. Therefore, CNA900 possesses more contact interfaces as well as superior electron transport channels. In a word, it can be concluded that only by adopting the appropriate annealing temperature, well-sized CNAT composites could be preferably synthesized.

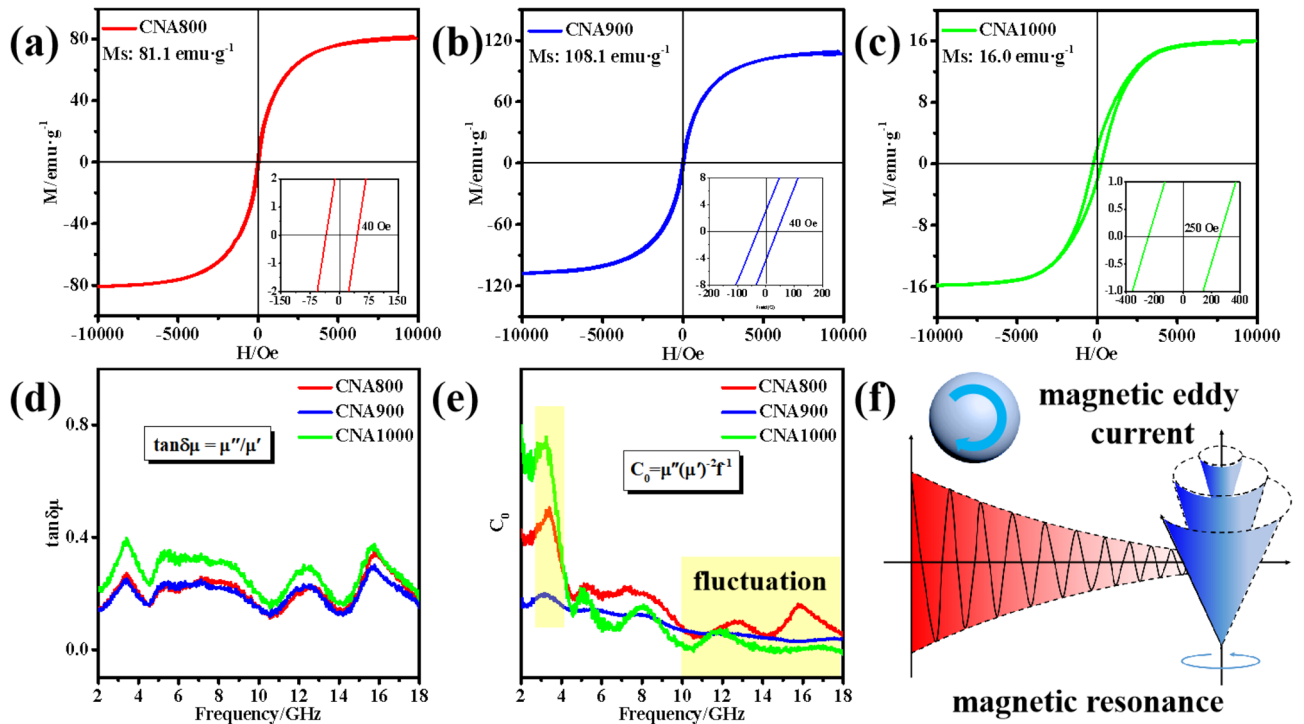


**Figure 3.** The electromagnetic parameters values of (a) CNA800, (b) CNA900 and (c) CNA1000 samples. Reflection loss values versus frequency of (d) CNA800, (e) CNA900 and (f) CNA1000. (g) 3D RL plots of CNA900. (e) Effective bandwidth and (f) minimum reflection loss values at different thicknesses, respectively.

## Discussion

Figure 3a–c exhibit the electromagnetic parameters of CNA800/900/1000 samples, including  $\epsilon'$ ,  $\epsilon''$ ,  $\mu'$ , and  $\mu''$ , respectively, which represent the real and imaginary parts of permittivity and permeability. As a rule, the real parts represent the storage performance of electrical and magnetic energy, while the imaginary parts signify the dissipation capacity of microwave energy<sup>21</sup>. CNA800 exhibits the lowest relative electromagnetic parameters among all the as-prepared composites with the same filling ratio. At the same time, the  $\epsilon'$  value of the CNA900 sample decreased from 9.32 to 5.47 during the working frequency from 2 to 18 GHz, while the  $\epsilon''$  value decreased from 5.10 to 2.25. It can be found that the  $\epsilon'$  and  $\epsilon''$  values of CNA800 and CNA1000 accompanied with dramatic fluctuations during high frequency range, which can be ascribed to dielectric resonance behavior and dipole polarization caused by particles aggregation<sup>22</sup>. Moreover, the average values of the real ( $\mu'$ ) and imaginary ( $\mu''$ ) permeability of CNA900 are 1.08 and 0.15, separately. In addition, the  $\mu'$  and  $\mu''$  values of CNA800 and CNA1000 specimens keep around at 1 and 0, respectively, which are similar to air. Complex permeability  $\mu_r$  values maintain constant during the working frequency, indicating high stability of the magnetic dissipation capacity of the as-made samples. Based on the free electron theory,  $\epsilon'' = 1/\pi\epsilon_0\rho f^2$ , where  $\rho$  signifies the electrical resistivity, higher electron conductivity gives rise to larger imaginary parts of permittivity. Namely, CNA900 possesses the strongest dielectric dissipation capacity among all CNAT samples, which may further promote the microwave attenuation behaviors. For the purpose of evaluating the microwave absorption performance of the as-synthesized CNAT composites, reflection loss values (RL) curves are shown in Fig. 3d–f according to the basis of generalized transmission line theory and metal back model<sup>24</sup>:

$$Z_{in} = Z_0(\mu_r/\epsilon_r)^{1/2} \tanh[j(2\pi fd/c)(\mu_r\epsilon_r)^{1/2}] \quad (5)$$



**Figure 4.** Hysteresis loops of (a) CNA800, (b) CNA900 and (c) CNA1000. (d) Magnetic loss tangents and (e)  $C_0$  values of the as-prepared samples. (f) Schematic illustrations of magnetic loss mechanism in CNAT composites.

$$RL = 20 \log |(Z_{in} - Z_0)/(Z_{in} + Z_0)| \tag{6}$$

Herein,  $Z_{in}$ ,  $Z_0$ ,  $\mu_r$ ,  $\epsilon_r$ ,  $f$ ,  $d$ , and  $c$  stand for the input impedance value of the absorber, the impedance value of free space, complex permeability, complex permittivity, the whole measuring frequency, the thickness of the microwave absorber coating layer and the velocity of light, respectively. Obviously, when the coating thickness varies from 1.9 to 2.4 mm, RL values below  $-10$  dB could be achieved by all samples, which is appropriate for practical application (90% effective microwave absorption)<sup>25</sup>. As for CNA800 and CNA1000, the values of  $RL_{min}$  are  $-30.33$  dB and  $-32.11$  dB, separately. Evidently, the 3D representation of the RL performance of CNA900 is directly exhibited in Fig. 3g and the black bold lines stands for the value of  $-10$  dB. It should be noted that the  $RL_{min}$  value of CNA900 is  $-49.79$  dB at 15.68 GHz with 2.1 mm (Fig. 3e,i). In order to intuitively compare the microwave attenuation capacity of the as-prepared samples, the effective bandwidth and RL values at thin matching thickness are shown in Fig. 3h,i. In detail, the effective working frequency of CNA800 at different thicknesses are all less than 2 GHz, and the largest bandwidth of CNA1000 is less than 4 GHz. Compared with CNA800 and CNA1000, CNA900 exhibits an extremely broad effective bandwidth of 6.84 GHz (from 11.16 to 18 GHz) at a thin layer thickness of 2.3 mm. On the contrary, CNA800 and CNA1000 show narrower band range in the full working frequency, which can be caused by inferior impedance matching and unsuitable microscopic dimensions and morphologies. Therefore, the outstanding performance brings CNA900 a bright prospect for practical applications as ideal electromagnetic absorbers with strong absorbing properties, thin coating layer thickness and broad bandwidth.

In consideration of the presence of Co and Ni magnetic ingredients of the microwave absorbing materials, magnetic loss mechanisms have been provided in Fig. 4. As depicted in Fig. 4a–c, the hysteresis loops with relatively thin S-shape indicates that the CNAT composites exhibit soft magnetic characteristics. The saturated magnetization values ( $M_s$ ) of CNA800/900/1000 are 81.1, 108.1 and 16.0  $\text{emu g}^{-1}$ , respectively. Based on the  $\mu$ - $M$  equation<sup>26</sup>:

$$\mu' = 1 + (M/H) \cos \sigma \tag{7}$$

$$\mu'' = 1 + (M/H) \sin \sigma \tag{8}$$

where  $M$  is the magnetization,  $H$  represents the intensity of external magnetic field and  $\sigma$  signifies the phase lag angle of magnetization behind applied magnetic field. It can be deduced that higher  $M$  value result in higher complex permeability  $\mu_r$  values. Besides, the size shrinkage in CNA900 may lead to more magnetic dissipation behaviors, which may result in augmented values of  $M_s$ . Furthermore, the coercivity values ( $H_c$ ) of the as-prepared CNA800/900/1000 composites are 40 Oe, 40 Oe and 250 Oe, respectively. The higher  $H_c$  value (250 Oe) is related

to the larger shape and surface anisotropy of the CNA1000 sample<sup>27,28</sup>. As shown in Fig. 4d, the average values of magnetic loss tangents ( $\tan \delta_\mu = \mu''/\mu'$ ) for CNA800/900/1000 are 0.20, 0.21 and 0.27, respectively, indicating similar magnetic loss ability of all as-prepared samples. As a rule, domain wall resonance usually exists in MHz frequency and hysteresis loss usually can be ignored in weak electromagnetic field. Therefore, magnetic attenuation mechanisms can be closely related to not only eddy current loss, but also natural resonance and exchange resonance<sup>29</sup>. Typically, eddy current loss ( $C_0$ ) can be assessed based on the following formula<sup>30</sup>:

$$C_0 = \mu''(\mu')^{-2}f^{-1} = 2\pi\mu_0d^2\sigma \quad (9)$$

Herein,  $\mu_0$ ,  $d$ , and  $\sigma$  signify the permeability in vacuum, thickness and electrical conductivity, separately. If the eddy current loss is the dominating reason in magnetic loss mechanism, the  $C_0$ - $f$  curve will present a horizontal line during the testing frequency. With regard to CNA900, the  $C_0$  values almost keep a constant at high frequency range, demonstrating the positive effect of eddy current loss (Fig. 4e). However, for CNA800 and CNA1000, the  $C_0$  values fluctuate a lot, so the eddy current loss is not the main reason for magnetic attenuation. However, it still works to some extent. According to the natural resonance equation,  $2\pi f_y = \gamma H_k$ <sup>31,32</sup>, where  $f_y$  means the natural resonance frequency,  $\gamma$  is electronic gyro-magnetic ration and  $H_k$  stands for the magneto-crystalline anisotropy field. Generally, the natural resonance occurs in low frequency range (0.1–10 GHz), an obvious peak at around 3.23 GHz can be seen from the highlighted part in Fig. 4e. Besides, exchange resonance often appears at high frequency part. Furthermore, schematic diagrams of magnetic eddy current loss and magnetic resonance are exhibited in Fig. 4f. Consequently, relevant magnetic loss mechanisms, such as eddy current loss, natural resonance and exchange resonance, correspond with relative permeability characteristics and contribute to microwave attenuation.

According to Havriliak–Negami function and Debye theory, polarization relaxation and conduction loss play significant roles in explaining dielectric loss mechanisms in the whole measuring frequency, which can be described as the following equations<sup>33,34</sup>:

$$\varepsilon_r = \varepsilon_\infty + (\varepsilon_s - \varepsilon_\infty)/(1 + j2\pi f\tau) = \varepsilon' - j\varepsilon'' \quad (10)$$

$$\varepsilon' = \varepsilon_\infty + (\varepsilon_s - \varepsilon_\infty)/(1 + (2\pi f)^2\tau^2) \quad (11)$$

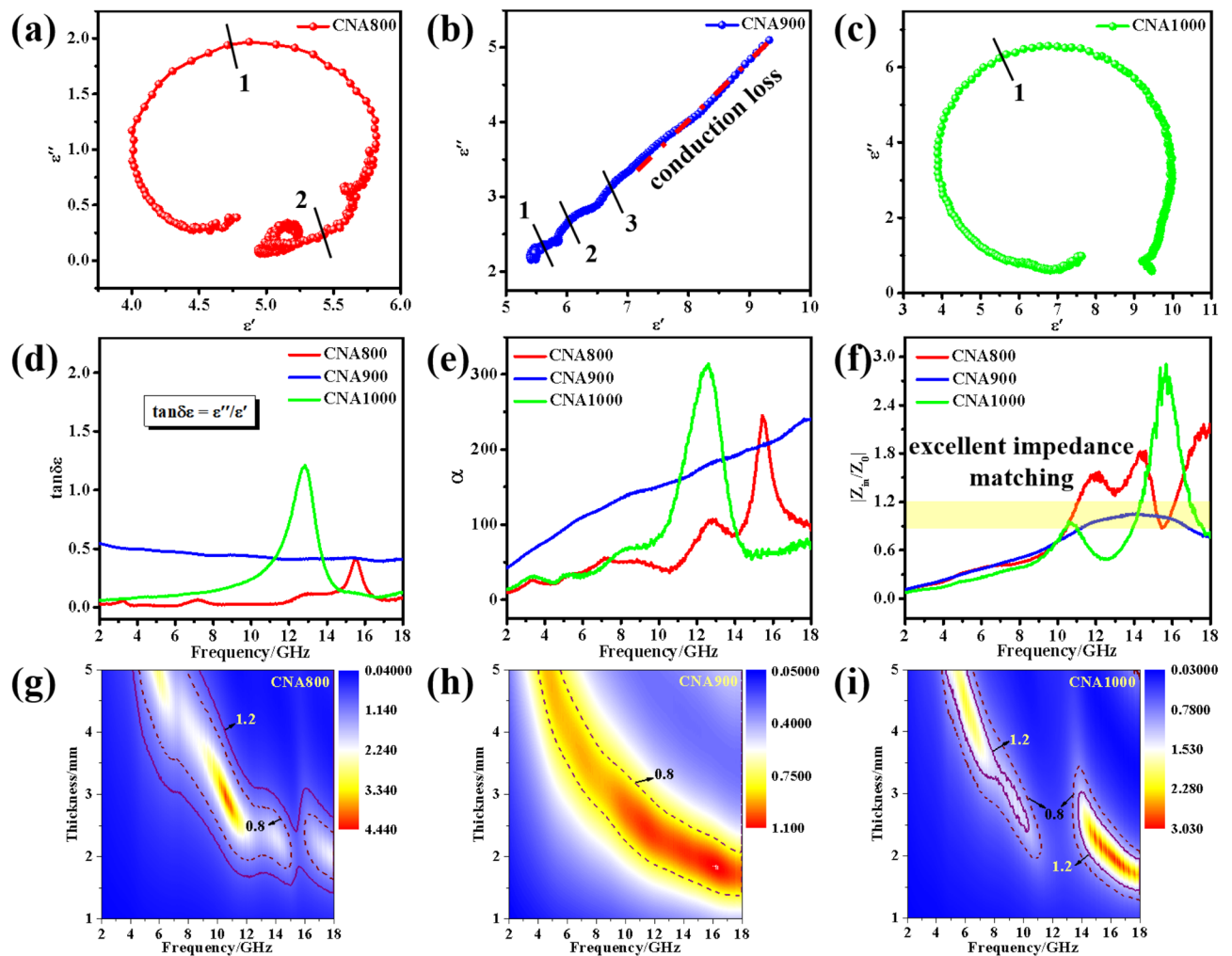
$$\varepsilon'' = 2\pi f\tau(\varepsilon_s - \varepsilon_\infty)/(1 + (2\pi f)^2\tau^2) \quad (12)$$

$$(\varepsilon' - \varepsilon_\infty)^2 + (\varepsilon'')^2 = (\varepsilon_s - \varepsilon_\infty)^2 \quad (13)$$

where  $\varepsilon_\infty$  represents the relative dielectric permittivity at infinite frequency,  $\varepsilon_s$  is the static permittivity, and  $\tau$  stands for the polarization relaxation time, respectively. Thus, it can be deduced that the  $\varepsilon'$ - $\varepsilon''$  curves (Cole–Cole plots) should be semicircles. Generally, the dielectric polarization mainly comes from ion polarization, molecular polarization, atomic polarization and electron polarization, which is closely related to microwave frequency<sup>35</sup>. Ionic polarization usually occurs in low frequency range ( $< 10^5$  Hz), so it cannot play a part in the testing frequency from 2 to 18 GHz. As a matter of fact, the essence of molecular polarization can be regarded as dipole polarization, since molecular polarization comes from the change of dipole moments. Compared with dipole polarization and ionic polarization, atomic polarization and electronic polarization have very limited effect on dielectric loss, so they can be ignored during the frequency range. Thus, in this work, we just talk about the dipole polarization in the testing frequency range. In order to intuitively observe the semicircles variation trend, Cole–Cole curves are provided in Fig. 5a–c. Obviously, there appear more distorted semicircles for CNA900 than for CNA800 and CNA1000. Namely, CNA900 possesses more Debye relaxation properties than the other two samples, which may be caused by more dipole polarization behaviors of CNA900. Owing to the shrinkage of the size of the as-prepared samples, interfaces between not only CoNi particles but also Co and Ni elements may lead to the formation of abundant dipoles. In addition, the Cole–Cole plot of CNA900 exhibits another straight line, which is related to conduction loss caused by CoNi species<sup>36</sup>. Therefore, for CNA900, both conduction loss and polarization loss are beneficial to dielectric loss. Different from CNA900 sample, CNA800 and CNA1000 samples with fewer semicircles and none straight-line part reveal that conduction loss has a limited effect while polarization relaxation plays a dominant role<sup>37</sup>. This is well corresponding to the resonance peaks (the fluctuation of the  $\varepsilon'$  and  $\varepsilon''$  values at high frequency) that appeared in CNA800 and CNA1000 samples, which can be seen from Fig. 3a,c. As described in Fig. 5d, the average values of dielectric loss tangents ( $\tan \delta_\varepsilon = \varepsilon''/\varepsilon'$ ) for CNA800, CNA900 and CNA1000 are 0.07, 0.45, and 0.25. For CNA900, average value of  $\tan \delta_\varepsilon$  (0.45) is about twice as high as average value of  $\tan \delta_\mu$  (0.21), so it is not hard to conclude that dielectric loss plays a decisive role in enhancing microwave dissipation performance. For the purpose of further understanding the microwave attenuation performance of the as-made samples, attenuation constants  $\alpha$  can be gained by the following formula<sup>38,39</sup>:

$$\alpha = \frac{\sqrt{2}\pi f}{c} \sqrt{(\mu''\varepsilon'' - \mu'\varepsilon') + \sqrt{(\mu''\varepsilon'' - \mu'\varepsilon')^2 + (\mu'\varepsilon'' - \mu''\varepsilon')^2}} \quad (14)$$

Evidently, the calculated average  $\alpha$  values of CNA800/900/1000 are 70.17, 148.74 and 87.07, respectively, indicating the as-prepared CNA900 show the strongest electromagnetic attenuation ability among all final products



**Figure 5.** Cole–Cole curves of (a) CNA800, (b) CNA900 and (c) CNA1000. (d) Dielectric loss tangent of the as-synthesized samples. (e) Attenuation constants and (f) comparison of impedance matching at the layer thickness of 2.1 mm of the as-prepared samples. (g) 2D representations of  $Z$  values for (g) CNA800, (h) CNA900 and (i) CNA1000 in the testing frequency range.

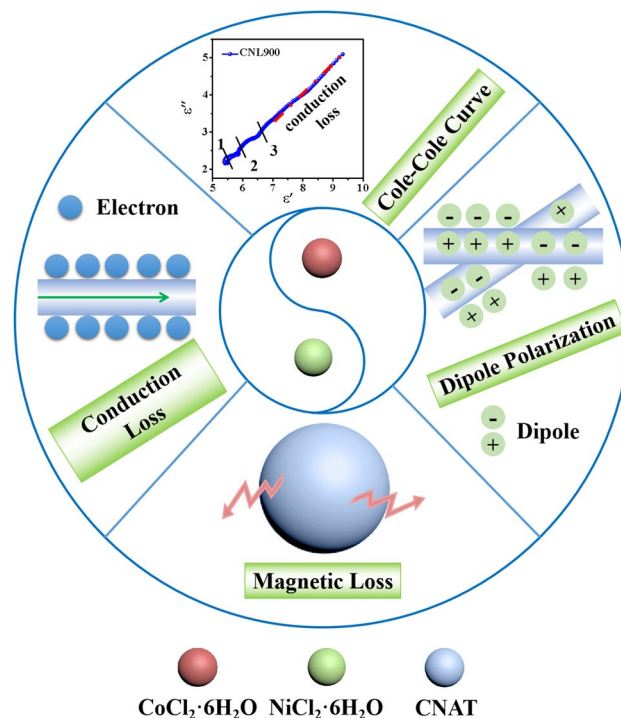
(Fig. 5e). Apart from the above factors, the impedance matching property which can be beneficial for microwave to enter into the as-synthesized materials is another factor that needs to be taken into account. The impedance matching property can be described as follows<sup>40</sup>:

$$Z = |Z_{in}/Z_0| \quad (15)$$

$$Z_{in} = (\mu_r/\epsilon_r)^{1/2}Z_0 \quad (16)$$

It should be pointed out that if there exhibits 0 reflection between air and the surface of the microwave absorber (where the value of  $Z$  is near 1), optimum impedance matching property could be achieved. At the layer thickness of 2.1 mm, effective  $|Z_{in}/Z_0|$  values (0.8–1.2) of CNA900 cover the working frequency range from 10.68 GHz to 17.48 GHz (Fig. 5e), and this frequency range is similar to effective bandwidth (where  $RL < -10$  dB). For comparison, Fig. 5g–i provide 2D representations of the  $Z$  values for CNAT products, and the values near 1 (0.8–1.2) are marked by purple lines. In detail, the purple dash lines stand for  $Z=0.8$  while full lines signify  $Z=1.2$ . Evidently, the  $Z$  values (from 0.8 to 1.2) of both CNA800 and CNA1000 cover narrower frequency range than that of CNA900, which may be caused by more contact interfaces and larger attenuation ability of CoNi particles. Therefore, preserving suitable impedance matching properties and outstanding attenuation capacity may be an effective way to make CNA900 composite desirable microwave absorbers.

According to the previous analysis, the reasons why CNA900 exhibits superb microwave absorption are as follows (Fig. 6): (1) the metal soft magnetic material brings natural resonance, exchange resonance and eddy current loss mechanisms that strengthen the microwave absorption to some extent, (2) the conduction loss induced by conductive bimetallic components that allows more microwave to be attenuated, (3) the dipole polarization caused by migration speed of dipoles are out of step of external electric field can contribute to improve dielectric loss ability, (4) the superior impedance matching performance that allows abundant electromagnetic waves to



**Figure 6.** Schematic illustration of microwave absorption mechanisms of the CNAT composites.

Sample	Filling ratio (wt %)	Minimum RL value		RL ≤ -10 dB		References
		$d_m$ (mm)	RL <sub>min</sub> (dB)	$d_m$ (mm)	$f_c$ (GHz)	
CoNi microspheres	60	2.5	-41	1.5	4.16	Ref. <sup>41</sup>
CoNi microflowers	20	2	-28.5	2	6.5	Ref. <sup>42</sup>
Co <sub>0.7</sub> Ni <sub>0.33</sub> /Ni <sub>0.6</sub> Zn <sub>0.4</sub> Fe <sub>2</sub> O <sub>4</sub>	60	2.1	-43.8	2.1	5	Ref. <sup>43</sup>
CoNi-CuO	50	2.5	-25.1	2.5	3.4	Ref. <sup>44</sup>
CoNi@TiO <sub>2</sub>	30	2.0	-25	2.0	6.2	Ref. <sup>45</sup>
CoNi@(CoO-NiO)	50	5	-38.1	2	5.9	Ref. <sup>46</sup>
CoNi@SiO <sub>2</sub> @C	50	2.2	-46	2.2	5.6	Ref. <sup>47</sup>
Co-Ni-Sn/PANI	25	3.2	-36.9	3.2	2.46	Ref. <sup>48</sup>
CNA900	55	2.1	-49.79	2.3	6.84	This work

**Table 1.** Microwave absorption performance of relative materials.

transmit into the absorption coating layer rather than being reflected. Accordingly, the annealing temperature plays a significant role in forming right-sized particles for achieve enough contact interfaces. For comparison, the crucial indices of relative CoNi-based materials for microwave absorption performance are summarized in Table 1. The relevant scholars indeed have made remarkable contributions in previous works. Nevertheless, the as-prepared CNA900 sample exhibits the minimum reflection loss and the broadest effective frequency bandwidth among all relative materials. Thus, CNA900 in this work is supposed to be a promising candidate for effectively exhaustion and absorption of microwave.

## Conclusion

In summary, a facial co-precipitation method and an effective calcinations process have been taken to fabricate a series of CNAT composites. The composites obtained at different temperatures present diverse morphology with change in size. Moreover, it also showed excellent microwave absorption performance with strong attenuation abilities and appropriate impedance matching properties. Remarkably, both the working bandwidth and RL<sub>min</sub> values have been satisfactorily enlarged, the CNA900 sample can achieve the optimum RL value of -49.79 dB with the matching thickness of merely 2.1 mm and the broadest working frequency of 6.84 GHz at 2.3 mm. Therefore, this work opens a novel pathway for the applications of CNAT materials derived from nanoflower-like CoNi bimetallic hydroxides in high-performance microwave absorbing field.



## Methods

**Materials.** Nickel (II) chloride hexahydrate ( $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ ), cobalt (II) chloride hexahydrate ( $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$ ) and ammonium hydroxide ( $\text{NH}_3 \cdot \text{H}_2\text{O}$ ) were all bought from the Nanjing chemical reagent Co., Ltd. All of the chemicals were analytical pure and employed without further purification.

**Synthesis of CNAT.** As depicted in Fig. 1, CNAT composites were fabricated via facile co-precipitation method and calcinations process. In a typical synthesis, 0.71 g of  $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$  along with 0.71 g of  $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$  were dissolved in 60 mL of deionized water and rapidly stirred for 10 min. Then, concentrated ammonium hydroxide was diluted to  $1 \text{ mol} \cdot \text{L}^{-1}$  for the purpose of further utilization. After that, the as-pretreated ammonia hydroxide was added dropwise into the above uniform solution at  $50^\circ\text{C}$  under continuous stirring. After further centrifugal washing (with deionized water and anhydrous ethanol) and blast air drying process, the CoNi bimetallic hydroxides (CoNiBH) can be formed. Finally, CoNi alloys products were gained from pyrolysis of the hydroxide precursors under inert atmosphere. In detail, a certain amount CoNi bimetallic hydroxide powders were put into a combustion boat and annealed at three different temperatures in tube furnace for 2 h under nitrogen atmosphere with a slow heating rate of  $2^\circ\text{C} \cdot \text{min}^{-1}$ . The corresponding composites were defined as CNAT, where T was the annealing temperature ( $T = 800, 900, 1000^\circ\text{C}$ ).

**Characterization.** The morphology and microstructure were characterized by field emission scanning electron microscopy (FE-SEM, Hitachi S4800) and transmission electron microscopy (TEM, Tecnai G2F30 S-TWIN). The composition and phase of the as-prepared samples were obtained by a Bruker D8 ADVANCE X-ray diffractometer (XRD) using  $\text{Cu K}\alpha$  radiation ( $\lambda = 1.5604 \text{ \AA}$ ). Electromagnetic parameters were successfully obtained by an Agilent PNA N5244A vector network analyzer (VNA). The coaxial rings ( $\varphi_{\text{out}}$  of 7.00 mm and  $\varphi_{\text{in}}$  of 3.04 mm) for testing were prepared by mixing 55 wt % CNAT power with 45 wt % paraffin wax matrix. Vibrating sample magnetometer (VSM, Lakeshore, Model 7400 series) was used to form the hysteresis loops of the specimens at an external magnetic field of 10,000 Oe.

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## Author contributions

W.G. designed the experimental works and wrote the manuscript. J.C. and Y.Z. helped in the data analysis. G.W., F.W. and T.Z. helped in the characterization part. B.Z. provided constructive comments to the manuscript.

## Competing interests

The authors declare no competing interests.

## Additional information

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