



Chromium(II)-catalyzed enantioselective arylation of ketones

Gang Wang^{1,2}, Shutao Sun², Ying Mao², Zhiyu Xie² and Lei Liu^{*1,2,3}

Full Research Paper

Open Access

Address:

¹Shenzhen Research Institute of Shandong University, Shenzhen 518057, China, ²School of Pharmaceutical Sciences, Shandong University, Jinan 250012, China and ³School of Chemistry and Chemical Engineering, Shandong University, Jinan 250100, China

Email:

Lei Liu* - leiliu@sdu.edu.cn

* Corresponding author

Keywords:

arylation; asymmetric catalysis; chromium; ketone; tertiary alcohol

Beilstein J. Org. Chem. **2016**, *12*, 2771–2775.

doi:10.3762/bjoc.12.275

Received: 17 October 2016

Accepted: 06 December 2016

Published: 19 December 2016

This article is part of the Thematic Series "Strategies in asymmetric catalysis".

Guest Editor: T. P. Yoon

© 2016 Wang et al.; licensee Beilstein-Institut.

License and terms: see end of document.

Abstract

The chromium-catalyzed enantioselective addition of carbo halides to carbonyl compounds is an important transformation in organic synthesis. However, the corresponding catalytic enantioselective arylation of ketones has not been reported to date. Herein, we report the first Cr-catalyzed enantioselective addition of aryl halides to both arylaliphatic and aliphatic ketones with high enantioselectivity in an intramolecular version, providing facile access to enantiopure tetrahydronaphthalen-1-ols and 2,3-dihydro-1*H*-inden-1-ols containing a tertiary alcohol.

Introduction

Catalytic enantioselective carbon–carbon bond formation reactions have achieved enormous development during the last few decades as a consequence of the growing demand for enantiopure compounds in modern industry, especially the pharmaceutical industry. The chromium (Cr)-catalyzed enantioselective addition of carbo halides to carbonyl compounds is one of the most reliable methods in organic chemistry for chemoselective and structurally diverse synthesis [1–9]. To date, the Cr-catalyzed enantioselective carbonyl addition reactions mainly focused on allylation, propargylation, alkenylation and alkylation of aldehydes [10,11]. Since the first example of enantioselective allylation of aldehydes catalyzed by a Cr(II)–salen complex in 1999 by Cozzi and co-workers [12], several elegant cat-

alytic enantioselective allylation and propargylation reactions have been developed by the groups of Nakada [13,14], Berkessel [15], Kishi [16], Sigman [17], Yamamoto [18], Guiry [19], Chen [20], Gade [21], White [22], and Zhang [23–25], respectively. The alkenylation and alkylation reactions were mainly explored by the Kishi group [26–30], and they established a toolbox approach to search for the specific ligand with a given substrate in the Cr-catalyzed process [28]. They successfully applied the method to the natural product total synthesis like halichondrin B and norhalichondrin B, and in the subsequent pharmaceutical study, finally leading to the discovery of the anticancer drug Eribulin [31–35]. However, to our knowledge, the Cr-catalyzed enantioselective arylation of carbonyl

compounds has rarely been explored. On the other hand, most of the reactions focused on aldehyde components, while asymmetric addition to ketones remains a big challenge probably due to the decreased reactivity and selectivity [36,37]. A breakthrough was made by the Sigman group who reported the catalytic enantioselective addition of allylic bromides and propargyl halides to arylaliphatic ketones using oxazoline ligands with high enantioselectivity (up to 95% ee) [38–41]. After that, the Chen group also disclosed enantioselective allylation of ketones using spirocyclic chiral borate and chiral bipyridyl alcohol ligands with the ee value ranging from 27% to 97% [42,43]. However, as far as we know, a Cr-catalyzed enantioselective arylation of ketones has never been reported to date [44]. Tetrahydronaphthalen-1-ol bears a chiral tertiary alcohol center and is a common structural motif in numerous biologically active natural products and clinical drugs [45]. The method to

prepare these compounds through intramolecular arylation of ketones would be highly desired. Herein, we report the first Cr-catalyzed enantioselective arylation of ketones in an intramolecular version.

Results and Discussion

Initially, the Cr-catalyzed asymmetric intramolecular arylation of arylaliphatic ketone 5-(2-iodophenyl)pentan-2-one (**1a**) was selected as the model reaction for optimization employing Kishi's oxazoline/sulfonamides as the chiral ligands. A series of oxazoline/sulfonamide ligands (**L1–L8**) were tested and the results were summarized in Table 1. Four subgroups of R¹ were studied (entries 1–4, Table 1) and isopropyl substituted oxazoline proved to be the best ligand with a 42% ee. Afterwards, R² (Table 1, entries 2, 5 and 6) and R³ (Table 1, entries 6–8) substituents were also examined, and **L8** bearing a methyl

Table 1: Screening conditions of the catalytic enantioselective Cr-mediated arylation of ketone.^a

1a

L1–8

NiCl₂·DMP

2a

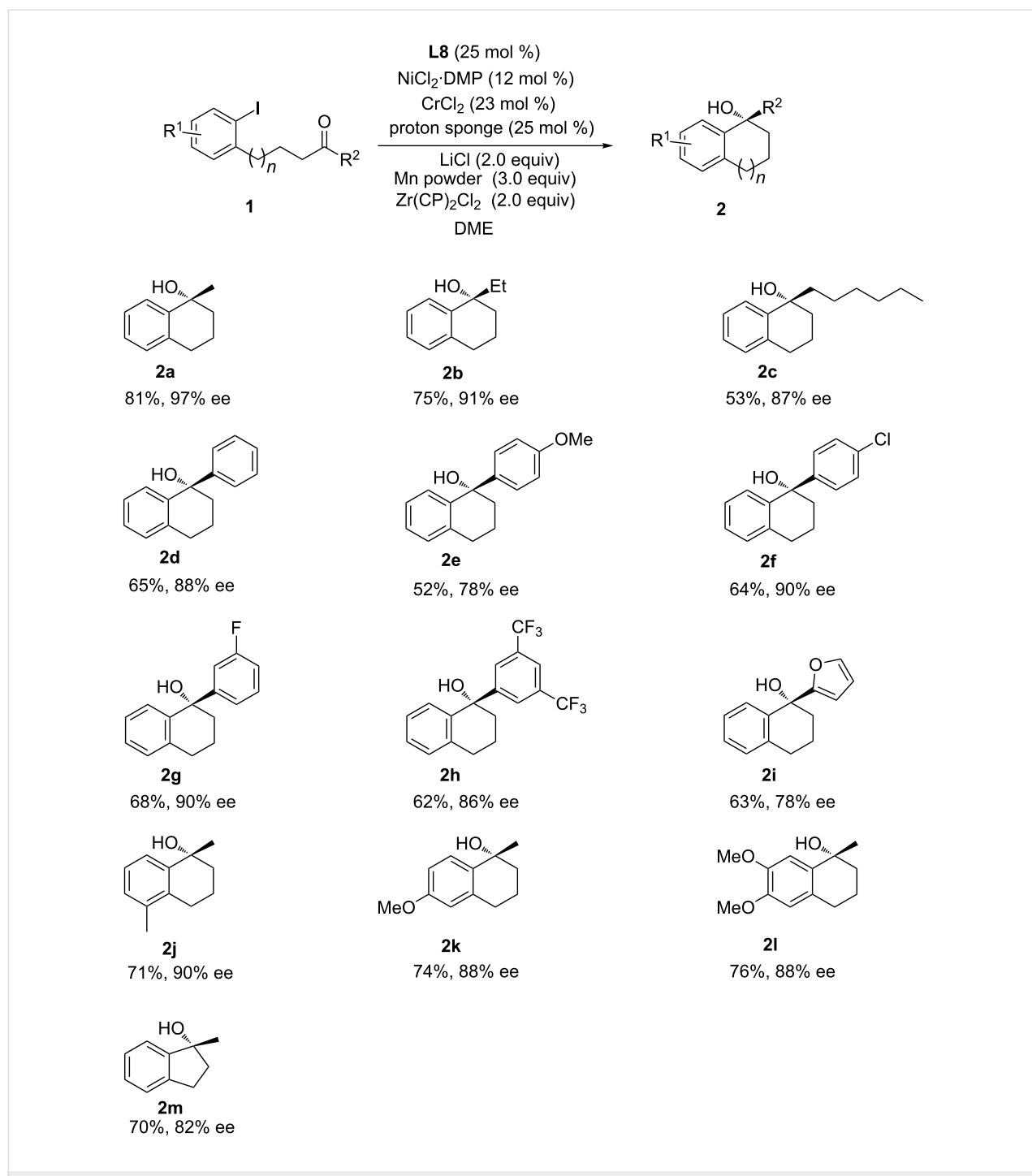
entry	L	R ¹	R ²	R ³	Solvent	Yield (%) ^a	ee ^b
1	L1	Me	Ph	H	MeCN	43	28
2	L2	iPr	Ph	H	MeCN	50	42
3	L3	t-Bu	Ph	H	MeCN	47	25
4	L4	Ph	Ph	H	MeCN	51	21
5	L5	iPr	Bn	H	MeCN	12	37
6	L6	iPr	Me	H	MeCN	49	56
7	L7	iPr	Me	OMe	MeCN	75	75
8	L8	iPr	Me	Me	MeCN	85	82
9	L8	iPr	Me	Me	THF	93	81
10	L8	iPr	Me	Me	DME	90	86
11 ^c	L8	iPr	Me	Me	DME	86	92
12 ^d	L8	iPr	Me	Me	DME	81	97
13 ^e	L8	iPr	Me	Me	DME	51	83
14 ^{d,f}	L8	iPr	Me	Me	DME	35	70

^aYield of isolated product. ^bDetermined by HPLC analysis on a chiral column. ^cReaction at 0 °C for 24 h. ^dReaction at –20 °C for 24 h. ^eReaction at –40 °C for 24 h. ^fAryl bromide used instead of aryl iodide.

group in both R² and R³ gave the best enantiocontrol. The solvent effect was then investigated, and 1,2-dimethoxyethane (DME) was identified to be the best choice (Table 1, entries 8–10). Lowering the reaction temperature was found to be beneficial for improving the enantioselectivity, and when the reaction was performed at –20 °C, expected **2a** was isolated in 81% yield with 97% ee (Table 1, entries 10–13). Aryl bromide

proved to be an inferior coupling component, providing **2a** in 35% yield and 70% ee (Table 1, entry 14).

With the optimized conditions in hand, the scope of the ketone component was first explored (Scheme 1). Aliphatic ketones with (longer) alkyl chain such as ethyl (**1b**) and *n*-hexyl ketones (**1c**), were also tolerated albeit with slightly decreased



Scheme 1: Scope of the catalytic enantioselective Cr-mediated arylation of ketones.

yield and selectivity. The asymmetric arylation of various arylaliphatic ketones also went smoothly (**1d–h**). Phenyl ketone **1d** and ketones with electron-withdrawing groups in different substituent patterns gave the expected products with good enantiocontrol, while the enantioselectivity for ketone **1e** bearing an electron-donating group decreased. The mild process exhibited excellent functional group tolerance, with chloride (**2f**), fluoride (**2g**), and CF_3 moieties (**2h**) well tolerated for further manipulation [46,47]. Heteroaryl ketones such as furan-substituted ketone (**1i**) were also suitable substrate, giving product **2i** in 78% ee. The scope of the aryl halide component was next explored (**1j–l**). Aryl halides bearing different substituent patterns were tolerated giving the tetrahydronaphthalen-1-ols with good ee values. When 4-(2-iodophenyl)butan-2-one (**1m**) was used, enantiopure indan-1-ol was obtained in 70% yield and 82% ee.

Conclusion

In summary, we have developed the first Cr-catalyzed enantioselective arylation of ketones in an intramolecular version using oxazoline/sulfonamide **L8** as the catalyst. Both aliphatic and arylaliphatic ketones proceeded smoothly, providing corresponding tetrahydronaphthalen-1-ols bearing a tertiary alcohol center with good enantioselectivities (up to 97% ee).

Experimental

General procedure for the chromium(II) catalyzed enantioselective arylation of ketones: The solution of **L8** (0.25 equiv, 0.025 mmol), proton sponge (0.25 equiv, 0.025 mmol) and CrCl_2 (0.23 equiv, 0.023 mmol) in DME (1.0 mL) was stirred at room temperature in a glove-box for 1 h. Then the substrate **1** (1.0 equiv, 0.1 mmol), LiCl (2.0 equiv, 0.2 mmol), Mn powder (3.0 equiv, 0.3 mmol), $\text{NiCl}_2\text{-DMP}$ (0.12 equiv, 0.012 mmol) and $\text{Zr}(\text{CP})_2\text{Cl}_2$ (2.0 equiv, 0.2 mmol) were added successively and the mixture was stirred at indicated temperature for 24 h. After that, the mixture was filtered through a short pad of celite and purified by flash chromatography using silica gel or alumina (200–300 mesh) to give the product **2**.

Supporting Information

Supporting Information File 1

Experimental procedures, analytical data for products, copies of NMR spectra and HPLC chromatograms.
[<http://www.beilstein-journals.org/bjoc/content/supporting/1860-5397-12-275-S1.pdf>]

Acknowledgements

We thank the National Science Foundation of China (21472112), Fok Ying Tung Education Foundation (151035),

the Program for New Century Excellent Talents in University (NCET-13-0346), Shenzhen Strategic Emerging Industry Development Special Funds (JCYJ20150430160921949), the Shandong Science Fund for Distinguished Young Scholars (JQ201404), and the Fundamental Research Funds of Shandong University (2015JC035) for financial support.

References

- Okude, Y.; Hirano, S.; Hiyama, T.; Nozaki, H. *J. Am. Chem. Soc.* **1977**, *99*, 3179–3181. doi:10.1021/ja00451a061
- Okude, Y.; Hiyama, T.; Nozaki, H. *Tetrahedron Lett.* **1977**, *18*, 3829–3830. doi:10.1016/S0040-4039(01)83365-7
- Fürstner, A.; Shi, N. *J. Am. Chem. Soc.* **1996**, *118*, 2533–2534. doi:10.1021/ja9537538
- Fürstner, A.; Shi, N. *J. Am. Chem. Soc.* **1996**, *118*, 12349–12357. doi:10.1021/ja9625236
- Avalos, M.; Babiano, R.; Cintas, P.; Jiménez, J. L.; Palacios, J. C. *Chem. Soc. Rev.* **1999**, *28*, 169–177. doi:10.1039/a806117c
- Takai, K.; Nozaki, H. *Proc. Jpn. Acad., Ser. B* **2000**, *76*, 123–131. doi:10.2183/pjab.76.123
- Fürstner, A. *Chem. Rev.* **1999**, *99*, 991–1046. doi:10.1021/cr9703360
- Hargaden, G. C.; Guiry, P. J. *Adv. Synth. Catal.* **2007**, *349*, 2407–2424. doi:10.1002/adsc.200700324
- Tian, Q.; Zhang, G. *Synthesis* **2016**, *48*, 4038–4049. doi:10.1055/s-0036-1589457
- Choi, H.-w.; Nakajima, K.; Demeke, D.; Kang, F.-A.; Jun, H.-S.; Wan, Z.-K.; Kishi, Y. *Org. Lett.* **2002**, *4*, 4435–4438. doi:10.1021/o1026981x
- Namba, K.; Kishi, Y. *Org. Lett.* **2004**, *6*, 5031–5033. doi:10.1021/o1047661b
- Bandini, M.; Cozzi, P. G.; Melchiorre, P.; Umani-Ronchi, A. *Angew. Chem.* **1999**, *111*, 3558–3561. doi:10.1002/(SICI)1521-3757(19991115)111:22<3558::AID-ANGE3558>3.0.CO;2-6 *Angew. Chem. Int. Ed.* **1999**, *38*, 3357–3359. doi:10.1002/(SICI)1521-3773(19991115)38:22<3357::AID-ANIE3357>3.0.CO;2-W
- Inoue, M.; Suzuki, T.; Nakada, M. *J. Am. Chem. Soc.* **2003**, *125*, 1140–1141. doi:10.1021/ja021243p
- Inoue, M.; Nakada, M. *Org. Lett.* **2004**, *6*, 2977–2980. doi:10.1021/o1048826j
- Berkessel, A.; Menche, D.; Sklorz, C. A.; Schröder, M.; Paterson, I. *Angew. Chem.* **2003**, *115*, 1062–1065. doi:10.1002/ange.200390240 *Angew. Chem., Int. Ed.* **2003**, *42*, 1032–1035. doi:10.1002/anie.200390265
- Kurosu, M.; Lin, M.-H.; Kishi, Y. *J. Am. Chem. Soc.* **2004**, *126*, 12248–12249. doi:10.1021/ja045557j
- Lee, J.-Y.; Miller, J. J.; Hamilton, S. S.; Sigman, M. S. *Org. Lett.* **2005**, *7*, 1837–1839. doi:10.1021/o1050528e
- Xia, G.; Yamamoto, H. *J. Am. Chem. Soc.* **2006**, *128*, 2554–2555. doi:10.1021/ja058454p
- McManus, H. A.; Cozzi, P. G.; Guiry, P. J. *Adv. Synth. Catal.* **2006**, *348*, 551–558. doi:10.1002/adsc.200505332
- Huang, X.-R.; Pan, X.-H.; Lee, G.-H.; Chen, C. *Adv. Synth. Catal.* **2011**, *353*, 1949–1954. doi:10.1002/adsc.201100023
- Deng, Q.-H.; Wadeohl, H.; Gade, L. H. *Chem. – Eur. J.* **2011**, *17*, 14922–14928. doi:10.1002/chem.201102375

22. White, J. D.; Shaw, S. *Org. Lett.* **2011**, *13*, 2488–2491.
doi:10.1021/ol2007378
23. Chen, W.; Yang, Q.; Zhou, T.; Tian, Q.; Zhang, G. *Org. Lett.* **2015**, *17*, 5236–5239. doi:10.1021/acs.orglett.5b02597
24. Xiong, Y.; Zhang, G. *Org. Lett.* **2016**, *18*, 5094–5097.
doi:10.1021/acs.orglett.6b02540
25. Tian, Q.; Bai, J.; Chen, B.; Zhang, G. *Org. Lett.* **2016**, *18*, 1828–1831.
doi:10.1021/acs.orglett.6b00559
26. Namba, K.; Cui, S.; Wang, J.; Kishi, Y. *Org. Lett.* **2005**, *7*, 5417–5419.
doi:10.1021/ol052084s
27. Namba, K.; Cui, S.; Wang, J.; Kishi, Y. *Org. Lett.* **2005**, *7*, 5421–5424.
doi:10.1021/ol052085k
28. Guo, H.; Dong, C.-G.; Kim, D.-S.; Urabe, D.; Wang, J.; Kim, J. T.; Liu, X.; Sasaki, T.; Kishi, Y. *J. Am. Chem. Soc.* **2009**, *131*, 15387–15393. doi:10.1021/ja905843e
29. Liu, X.; Li, X.; Chen, Y.; Hu, Y.; Kishi, Y. *J. Am. Chem. Soc.* **2012**, *134*, 6136–6139. doi:10.1021/ja302177z
30. Peng, J.; Kishi, Y. *Org. Lett.* **2012**, *14*, 86–89. doi:10.1021/ol202878v
31. Aicher, T. D.; Buszek, K. R.; Fang, F. G.; Forsyth, C. J.; Jung, S. H.; Kishi, Y.; Matelich, M. C.; Scola, P. M.; Spero, D. M.; Yoon, S. K. *J. Am. Chem. Soc.* **1992**, *114*, 3162–3164. doi:10.1021/ja00034a086
32. Namba, K.; Kishi, Y. *J. Am. Chem. Soc.* **2005**, *127*, 15382–15383.
doi:10.1021/ja055966v
33. Kim, D.-S.; Dong, C.-G.; Kim, J. T.; Guo, H.; Huang, J.; Tiseni, P. S.; Kishi, Y. *J. Am. Chem. Soc.* **2009**, *131*, 15636–15641.
doi:10.1021/ja9058475
34. Dong, C.-G.; Henderson, J. A.; Kaburagi, Y.; Sasaki, T.; Kim, D.-S.; Kim, J. T.; Urabe, D.; Guo, H.; Kishi, Y. *J. Am. Chem. Soc.* **2009**, *131*, 15642–15646. doi:10.1021/ja9058487
35. Liu, X.; Henderson, J. A.; Sasaki, T.; Kishi, Y. *J. Am. Chem. Soc.* **2009**, *131*, 16678–16680. doi:10.1021/ja9079308
36. Okude, Y.; Hirano, S.; Hiyama, T.; Nozaki, H. *J. Am. Chem. Soc.* **1977**, *99*, 3179–3181. doi:10.1021/ja00451a061
37. Chen, C.; Tagami, K.; Kishi, Y. *J. Org. Chem.* **1995**, *60*, 5386–5387.
doi:10.1021/jo00122a011
38. Miller, J. J.; Sigman, M. S. *J. Am. Chem. Soc.* **2007**, *129*, 2752–2753.
doi:10.1021/ja068915m
39. Miller, J. J.; Sigman, M. S. *Angew. Chem., Int. Ed.* **2008**, *47*, 771–774.
doi:10.1002/anie.200704257
40. Harper, K. C.; Sigman, M. S. *Science* **2011**, *333*, 1875–1878.
doi:10.1126/science.1206997
41. Harper, K. C.; Vilardi, S. C.; Sigman, M. S. *J. Am. Chem. Soc.* **2013**, *135*, 2482–2485. doi:10.1021/ja4001807
42. Huang, X.-R.; Chen, C.; Lee, G.-H.; Peng, S.-M. *Adv. Synth. Catal.* **2009**, *351*, 3089–3095. doi:10.1002/adsc.200900595
43. Chen, R.-Y.; Dhondge, A. P.; Lee, G.-H.; Chen, C. *Adv. Synth. Catal.* **2015**, *357*, 961–966. doi:10.1002/adsc.201400945
44. Chen, C. *Synlett* **1998**, 1311–1312. doi:10.1055/s-1998-1984
45. Chopra, I.; Roberts, M. *Microbiol. Mol. Biol. Rev.* **2001**, *65*, 232–260.
doi:10.1128/MMBR.65.2.232-260.2001
46. Xu, T.; Mu, X.; Peng, H.; Liu, G. *Angew. Chem., Int. Ed.* **2011**, *50*, 8176–8179. doi:10.1002/anie.201103225
47. Wang, F.; Qi, X.; Liang, Z.; Chen, P.; Liu, G. *Angew. Chem., Int. Ed.* **2014**, *53*, 1881–1886. doi:10.1002/anie.201309991

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (<http://creativecommons.org/licenses/by/4.0>), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions:
(<http://www.beilstein-journals.org/bjoc>)

The definitive version of this article is the electronic one which can be found at:
doi:10.3762/bjoc.12.275