

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

## 5-(Prop-2-yn-1-yl)-5*H*-dibenzo[*b,f*]-azepine: orthorhombic polymorph

 M. M. M. Abdoh,<sup>a</sup> S. Madan Kumar,<sup>b</sup> K. S. Vinay Kumar,<sup>c</sup>  
 B. C. Manjunath,<sup>b</sup> M. P. Sadashiva<sup>c</sup> and N. K. Lokanath<sup>b\*</sup>
<sup>a</sup>Department of Physics, Faculty of Science, An Najah National University, Nabtus West Bank, Palestine, <sup>b</sup>Department of Studies in Physics, Manasagangotri, University of Mysore, Mysore 570 006, India, and <sup>c</sup>Department of Studies in Chemistry, Manasagangotri, University of Mysore, Mysore 570 006, India  
 Correspondence e-mail: lokanath@physics.uni-mysore.ac.in

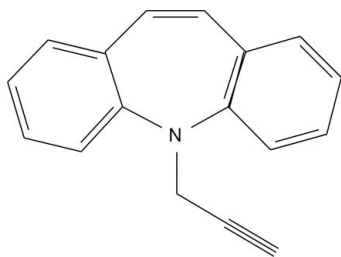
Received 22 November 2012; accepted 28 November 2012

 Key indicators: single-crystal X-ray study;  $T = 103$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.026;  $wR$  factor = 0.067; data-to-parameter ratio = 7.3.

In the title orthorhombic polymorph (space group *Iba*2),  $\text{C}_{17}\text{H}_{13}\text{N}$ , the dihedral angle between the benzene rings is  $55.99$  ( $10$ )° and the azepine ring adopts a boat conformation. In the crystal, molecules are linked by  $\text{C}-\text{H}\cdots\pi$  contacts. The previously-reported polymorph [Yousuf *et al.* (2012). *Acta Cryst. E* **68**, o1101] crystallizes in the monoclinic system (space group  $P2_1/c$ ) with two molecules in the asymmetric unit.

### Related literature

For the previously-reported monoclinic polymorph, see: Yousuf *et al.* (2012). For biochemical background, see: Sadashiva *et al.* (2005).



### Experimental

#### Crystal data

 $\text{C}_{17}\text{H}_{13}\text{N}$ 
 $M_r = 231.28$ 

 Orthorhombic, *Iba*2  
 $a = 16.2444$  (6) Å  
 $b = 21.1700$  (6) Å  
 $c = 7.2399$  (2) Å  
 $V = 2489.76$  (13) Å<sup>3</sup>
 $Z = 8$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.07$  mm<sup>-1</sup>  
 $T = 103$  K  
 $0.35 \times 0.30 \times 0.25$  mm

#### Data collection

 Oxford Diffraction Xcalibur Eos  
 diffractometer  
 10081 measured reflections

 1199 independent reflections  
 1132 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.027$ 

#### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.026$   
 $wR(F^2) = 0.067$   
 $S = 1.09$   
 1199 reflections  
 164 parameters

 1 restraint  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.08$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.10$  e Å<sup>-3</sup>
**Table 1**

Hydrogen-bond geometry (Å, °).

 $\text{Cg1}$  and  $\text{Cg2}$  are the centroids of the  $\text{C2/C3/C12}-\text{C15}$  and  $\text{C6}-\text{C11}$  rings, respectively.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C10}-\text{H10}\cdots\text{Cg2}^{\text{i}}$	0.95	2.75	3.659 (2)	160
$\text{C18}-\text{H18}\cdots\text{Cg1}^{\text{ii}}$	0.95	2.58	3.512 (2)	167

 Symmetry codes: (i)  $-x, y, z - \frac{1}{2}$ ; (ii)  $x, y, z + 1$ .

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2009); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: Mercury (Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXL97*.

SMK thanks UGC-BRS and University of Mysore for the award of a fellowship. MPS gratefully acknowledges financial support (grant No. 37-456/2009[SR]) from the University Grants Commission, New Delhi, India.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6998).

### References

- Macrae, C. F., Edgington, P. R., McCabe, P., Pidcock, E., Shields, G. P., Taylor, R., Towler, M. & van de Streek, J. (2006). *J. Appl. Cryst.* **39**, 453–457.  
 Oxford Diffraction (2009). *CrysAlis PRO*. Oxford Diffraction Ltd, Yarnton, Oxfordshire, England.  
 Sadashiva, M. P., Mallesha, H., Karunakaramurthy, K. & Rangappa, K. S. (2005). *Bioorg. Med. Chem. Lett.* **15**, 1811–1814.  
 Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.  
 Yousuf, S., Khan, M., Fazal, S., Butt, M. & Basha, F. Z. (2012). *Acta Cryst. E* **68**, o1101.

## supplementary materials

*Acta Cryst.* (2013). E69, o17 [doi:10.1107/S1600536812048908]

**5-(Prop-2-yn-1-yl)-5*H*-dibenzo[*b,f*]azepine: orthorhombic polymorph**

**M. M. M. Abdoh, S. Madan Kumar, K. S. Vinay Kumar, B. C. Manjunath, M. P. Sadashiva and N. K. Lokanath**

**Comment**

As part of our studies of new derivatives useful for tailoring of biologically active 5-membered heterocyclic rings such as 3,5-disubstituted isoxazole (Sadashiva *et al.*, 2005), we now describe the title compound.

In the title molecule, C<sub>17</sub>H<sub>13</sub>N (Fig. 1.), benzene rings fused to azepine rings are nearly planar and its geometry is similar to 5-(Prop-2-ynyl)-5*H*-dibenzo[*b,f*]azepine (Yousuf *et al.*, 2012). The dihedral angle between the benzene rings is 55.99 (10)° which is almost equal and large compared to the two molecules respectively for the first polymorph.

Seven-membered azepine ring adopts a boat conformation as indicated by the puckering parameters Q<sub>2</sub> = 0.7126 (18) Å, Q<sub>3</sub> = 0.2154 (17) Å, φ<sub>2</sub> = 177.93 (15)°, φ<sub>3</sub> = 178.1 (5)°, and the total puckering amplitude Q<sub>T</sub> = 0.7444 (17) Å. The title molecule adopts butterfly shape which may be essential for inhibition pocket which is similar to the reported polymorph (Yousuf *et al.*, 2012).

The packing of the title molecules is as shown (Fig. 2.) and features short C—H···π contacts (Table 1).

**Experimental**

5*H*-dibenzo[*b,f*]azepine (1, 0.0026 mol) was taken in a mixture of toluene and water in the ratio 1:1, was added sodium hydroxide (0.026 mol) followed by tetra-*n*-butylammonium bromide (TBAB) (0.00286 mol) at room temperature. After 15 minutes, propargyl bromide was added (0.00286 mol) at room temperature. Then, the resulting reaction mixture was heated at 60°C for 5 h. After completion of reaction (monitored by TLC), the reaction mixture was diluted with water (50 ml). The aqueous layer was extracted with ethyl acetate (3 × 20 ml), the combined ethyl acetate layer was washed with 0.1 N hydrochloric acid (2 × 25 ml), followed by brine solution (2 × 25 ml). Then, the organic layer was dried over anhydrous sodium sulfate, filtered and concentrated under reduced pressure to afford crude 2, which was purified by column chromatography over silica gel (60–120 mesh) using Hexane: Ethyl acetate mixture in 9.5:0.5 ratios as eluent. The pure compound 2 was crystallized in ethyl acetate and hexane to obtain yellow blocks.

<sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 300 MHz): δ 7.33 (t, J=7.2 Hz, 2H), 7.13 (t, J=8.7 Hz, 4H), 7.02 (t, J=7.2 Hz, 2H), 6.75 (s, 2H), 4.51 (d, J=1.8 Hz, 2H), 3.08 (s, 1H).

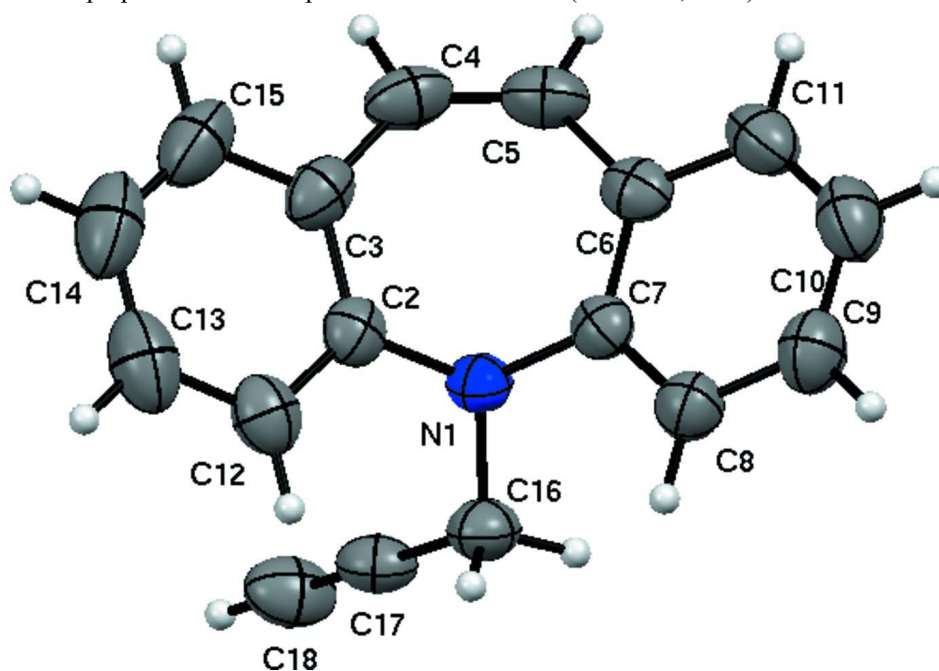
MS (*M*<sup>+</sup>+1): 232. Melting point (°C): 90 (Uncorrected)

**Refinement**

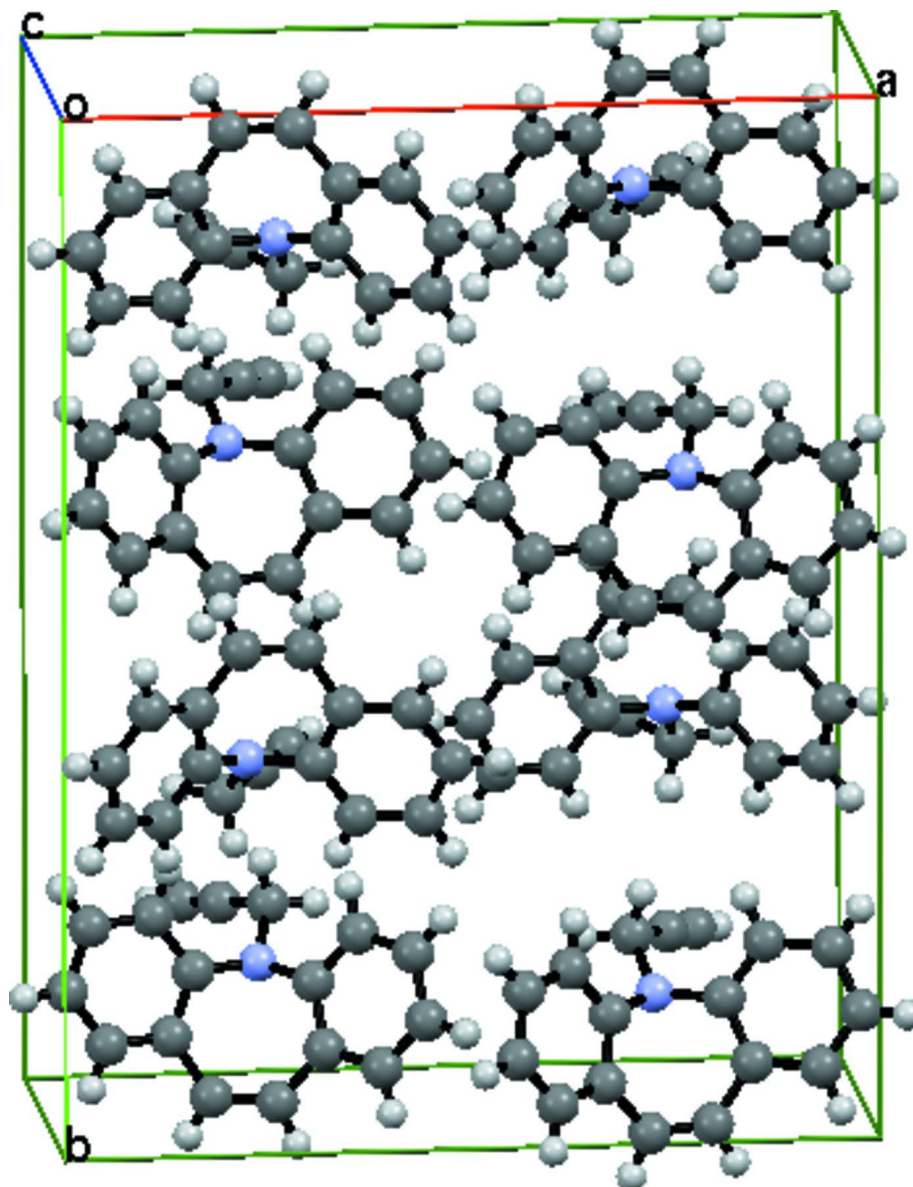
In the absence of significant anomalous dispersion effects Friedel pairs were merged. All the hydrogen atoms of the compound are fixed geometrically (C—H = 0.93–0.97 Å) and allowed to ride on their parent atoms. The poorly fitted reflections (2 0 0) and (1 1 0) were omitted during refinement.

**Computing details**

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2009); cell refinement: *CrysAlis PRO* (Oxford Diffraction, 2009); data reduction: *CrysAlis PRO* (Oxford Diffraction, 2009); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008).

**Figure 1**

The molecular structure of the title molecule with 50% probability ellipsoids.



**Figure 2**

Packing diagram of molecule, viewed along the crystallographic *a* axis. Dashed lines indicates short contacts.

### 5-(Prop-2-yn-1-yl)-5H-dibenzo[*b,f*]azepine

#### Crystal data

$C_{17}H_{13}N$

$M_r = 231.28$

Orthorhombic, *Iba*2

Hall symbol: I 2 -2c

$a = 16.2444$  (6) Å

$b = 21.1700$  (6) Å

$c = 7.2399$  (2) Å

$V = 2489.76$  (13) Å<sup>3</sup>

$Z = 8$

$F(000) = 976$

$D_x = 1.234$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 1199 reflections

$\theta = 3.2$ – $25.0^\circ$

$\mu = 0.07$  mm<sup>-1</sup>

$T = 103$  K

Block, yellow

$0.35 \times 0.30 \times 0.25$  mm

*Data collection*

Oxford Diffraction Xcalibur Eos  
diffractometer  
Radiation source: fine-focus sealed tube  
Graphite monochromator  
Detector resolution: 16.0839 pixels mm<sup>-1</sup>  
 $\omega$  scans  
10081 measured reflections

1199 independent reflections  
1132 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.027$   
 $\theta_{\text{max}} = 25.0^\circ$ ,  $\theta_{\text{min}} = 3.2^\circ$   
 $h = -19 \rightarrow 19$   
 $k = -25 \rightarrow 25$   
 $l = -8 \rightarrow 8$

*Refinement*

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.026$   
 $wR(F^2) = 0.067$   
 $S = 1.09$   
1199 reflections  
164 parameters  
1 restraint  
Primary atom site location: structure-invariant  
direct methods  
Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0354P)^2 + 0.3725P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.08 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.10 \text{ e } \text{\AA}^{-3}$   
Extinction correction: *SHELXL97* (Sheldrick,  
2008),  $\text{FC}^* = \text{KFC}[1 + 0.001\text{XFC}^2\Lambda^3/\text{SIN}(2\Theta)]^{-1/4}$   
Extinction coefficient: 0.0046 (6)

*Special details*

**Geometry.** Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement on  $F^2$  for ALL reflections except those flagged by the user for potential systematic errors. Weighted  $R$ -factors  $wR$  and all goodnesses of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The observed criterion of  $F^2 > \sigma(F^2)$  is used only for calculating  $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.23053 (8)	0.36797 (6)	0.6618 (2)	0.0409 (4)
C2	0.31311 (10)	0.36894 (7)	0.5957 (3)	0.0414 (5)
C3	0.34432 (10)	0.42627 (8)	0.5292 (3)	0.0492 (6)
C4	0.29519 (12)	0.48379 (8)	0.5170 (3)	0.0560 (6)
C5	0.21579 (11)	0.48775 (7)	0.4806 (3)	0.0538 (6)
C6	0.15967 (10)	0.43569 (8)	0.4423 (3)	0.0450 (5)
C7	0.16937 (9)	0.37583 (7)	0.5219 (2)	0.0399 (5)
C8	0.11790 (10)	0.32691 (8)	0.4686 (3)	0.0483 (6)
C9	0.05556 (11)	0.33719 (9)	0.3419 (3)	0.0592 (7)
C10	0.04355 (11)	0.39633 (10)	0.2688 (3)	0.0635 (7)
C11	0.09515 (11)	0.44467 (9)	0.3192 (3)	0.0571 (6)
C12	0.36307 (11)	0.31589 (9)	0.5992 (3)	0.0532 (6)
C13	0.44347 (12)	0.31917 (11)	0.5374 (3)	0.0705 (8)
C14	0.47460 (12)	0.37508 (13)	0.4711 (4)	0.0791 (9)
C15	0.42559 (12)	0.42769 (11)	0.4679 (4)	0.0691 (8)
C16	0.21109 (11)	0.32276 (8)	0.8082 (3)	0.0491 (6)

C17	0.26199 (11)	0.33371 (8)	0.9703 (3)	0.0474 (6)
C18	0.30303 (14)	0.34250 (9)	1.1020 (3)	0.0616 (7)
H4	0.32340	0.52250	0.53740	0.0670*
H5	0.19270	0.52900	0.47920	0.0650*
H8	0.12550	0.28600	0.51950	0.0580*
H9	0.02100	0.30320	0.30540	0.0710*
H10	-0.00010	0.40370	0.18410	0.0760*
H11	0.08650	0.48550	0.26830	0.0690*
H12	0.34190	0.27700	0.64430	0.0640*
H13	0.47730	0.28260	0.54070	0.0850*
H14	0.52970	0.37720	0.42790	0.0950*
H15	0.44760	0.46630	0.42260	0.0830*
H16A	0.15230	0.32680	0.84240	0.0590*
H16B	0.22030	0.27930	0.76240	0.0590*
H18	0.33610	0.34960	1.20820	0.0740*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0432 (7)	0.0429 (7)	0.0367 (8)	-0.0037 (6)	0.0005 (6)	0.0061 (6)
C2	0.0424 (8)	0.0505 (9)	0.0314 (8)	-0.0034 (8)	-0.0036 (8)	-0.0045 (7)
C3	0.0494 (9)	0.0594 (10)	0.0389 (10)	-0.0153 (8)	-0.0019 (8)	-0.0043 (8)
C4	0.0762 (12)	0.0439 (10)	0.0479 (11)	-0.0187 (8)	0.0020 (10)	-0.0015 (8)
C5	0.0747 (11)	0.0393 (9)	0.0474 (11)	0.0003 (8)	0.0066 (11)	0.0039 (8)
C6	0.0506 (9)	0.0464 (8)	0.0380 (9)	0.0061 (7)	0.0079 (8)	0.0023 (8)
C7	0.0385 (8)	0.0449 (8)	0.0362 (9)	0.0031 (7)	0.0050 (7)	-0.0001 (7)
C8	0.0464 (9)	0.0486 (9)	0.0498 (11)	-0.0009 (8)	-0.0006 (9)	-0.0001 (8)
C9	0.0498 (10)	0.0693 (12)	0.0584 (13)	-0.0044 (9)	-0.0072 (10)	-0.0051 (10)
C10	0.0506 (10)	0.0866 (14)	0.0533 (13)	0.0091 (11)	-0.0072 (10)	0.0044 (12)
C11	0.0574 (11)	0.0628 (11)	0.0511 (11)	0.0143 (9)	0.0018 (10)	0.0122 (9)
C12	0.0539 (10)	0.0610 (11)	0.0448 (11)	0.0056 (9)	-0.0074 (9)	-0.0068 (9)
C13	0.0560 (11)	0.1000 (17)	0.0554 (13)	0.0212 (12)	-0.0066 (11)	-0.0157 (13)
C14	0.0453 (11)	0.131 (2)	0.0610 (14)	-0.0082 (12)	0.0035 (11)	-0.0112 (15)
C15	0.0540 (11)	0.0919 (15)	0.0615 (13)	-0.0257 (11)	0.0031 (11)	-0.0036 (13)
C16	0.0545 (10)	0.0483 (9)	0.0446 (11)	-0.0065 (8)	-0.0006 (9)	0.0095 (8)
C17	0.0632 (11)	0.0407 (8)	0.0384 (10)	-0.0004 (8)	0.0064 (10)	0.0056 (8)
C18	0.0885 (15)	0.0558 (11)	0.0404 (11)	-0.0061 (11)	-0.0077 (11)	0.0014 (9)

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

N1—C2	1.424 (2)	C14—C15	1.369 (3)
N1—C7	1.429 (2)	C16—C17	1.454 (3)
N1—C16	1.463 (2)	C17—C18	1.178 (3)
C2—C3	1.401 (2)	C4—H4	0.9500
C2—C12	1.386 (2)	C5—H5	0.9500
C3—C4	1.459 (2)	C8—H8	0.9500
C3—C15	1.393 (3)	C9—H9	0.9500
C4—C5	1.319 (3)	C10—H10	0.9500
C5—C6	1.457 (2)	C11—H11	0.9500
C6—C7	1.401 (2)	C12—H12	0.9500

C6—C11	1.389 (3)	C13—H13	0.9500
C7—C8	1.386 (2)	C14—H14	0.9500
C8—C9	1.384 (3)	C15—H15	0.9500
C9—C10	1.373 (3)	C16—H16A	0.9900
C10—C11	1.372 (3)	C16—H16B	0.9900
C12—C13	1.382 (3)	C18—H18	0.9500
C13—C14	1.374 (4)		
C2—N1—C7	114.53 (14)	C3—C4—H4	117.00
C2—N1—C16	117.15 (13)	C5—C4—H4	117.00
C7—N1—C16	116.10 (13)	C4—C5—H5	117.00
N1—C2—C3	117.96 (14)	C6—C5—H5	116.00
N1—C2—C12	122.26 (15)	C7—C8—H8	120.00
C3—C2—C12	119.77 (16)	C9—C8—H8	120.00
C2—C3—C4	123.10 (16)	C8—C9—H9	120.00
C2—C3—C15	118.11 (17)	C10—C9—H9	120.00
C4—C3—C15	118.76 (17)	C9—C10—H10	120.00
C3—C4—C5	126.88 (16)	C11—C10—H10	120.00
C4—C5—C6	127.00 (15)	C6—C11—H11	119.00
C5—C6—C7	122.38 (16)	C10—C11—H11	119.00
C5—C6—C11	119.39 (16)	C2—C12—H12	120.00
C7—C6—C11	118.21 (15)	C13—C12—H12	120.00
N1—C7—C6	118.38 (14)	C12—C13—H13	120.00
N1—C7—C8	122.01 (14)	C14—C13—H13	120.00
C6—C7—C8	119.58 (15)	C13—C14—H14	120.00
C7—C8—C9	120.57 (16)	C15—C14—H14	120.00
C8—C9—C10	120.19 (17)	C3—C15—H15	119.00
C9—C10—C11	119.39 (18)	C14—C15—H15	119.00
C6—C11—C10	121.97 (18)	N1—C16—H16A	109.00
C2—C12—C13	120.44 (18)	N1—C16—H16B	109.00
C12—C13—C14	120.3 (2)	C17—C16—H16A	109.00
C13—C14—C15	119.5 (2)	C17—C16—H16B	109.00
C3—C15—C14	121.9 (2)	H16A—C16—H16B	108.00
N1—C16—C17	110.96 (14)	C17—C18—H18	180.00
C16—C17—C18	179.8 (2)		
C7—N1—C2—C3	69.0 (2)	C4—C3—C15—C14	-178.4 (2)
C7—N1—C2—C12	-112.4 (2)	C3—C4—C5—C6	-0.8 (4)
C16—N1—C2—C3	-149.95 (18)	C4—C5—C6—C7	32.0 (3)
C16—N1—C2—C12	28.6 (3)	C4—C5—C6—C11	-146.4 (2)
C2—N1—C7—C6	-72.36 (18)	C5—C6—C7—N1	7.6 (3)
C2—N1—C7—C8	110.12 (17)	C5—C6—C7—C8	-174.84 (18)
C16—N1—C7—C6	146.21 (16)	C11—C6—C7—N1	-174.02 (16)
C16—N1—C7—C8	-31.3 (2)	C11—C6—C7—C8	3.6 (3)
C2—N1—C16—C17	58.9 (2)	C5—C6—C11—C10	175.77 (19)
C7—N1—C16—C17	-160.75 (14)	C7—C6—C11—C10	-2.7 (3)
N1—C2—C3—C4	-2.9 (3)	N1—C7—C8—C9	175.51 (16)
N1—C2—C3—C15	178.7 (2)	C6—C7—C8—C9	-2.0 (3)
C12—C2—C3—C4	178.5 (2)	C7—C8—C9—C10	-0.6 (3)

C12—C2—C3—C15	0.1 (3)	C8—C9—C10—C11	1.6 (3)
N1—C2—C12—C13	-178.59 (19)	C9—C10—C11—C6	0.1 (3)
C3—C2—C12—C13	0.0 (3)	C2—C12—C13—C14	-0.3 (3)
C2—C3—C4—C5	-33.6 (4)	C12—C13—C14—C15	0.5 (4)
C15—C3—C4—C5	144.8 (2)	C13—C14—C15—C3	-0.4 (4)
C2—C3—C15—C14	0.1 (4)		

*Hydrogen-bond geometry (Å, °)*

Cg1 and Cg2 are the centroids of the C2/C3/C12–C15 and C6–C11 rings, respectively.

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
C10—H10 $\cdots$ Cg2 <sup>i</sup>	0.95	2.75	3.659 (2)	160
C18—H18 $\cdots$ Cg1 <sup>ii</sup>	0.95	2.58	3.512 (2)	167

Symmetry codes: (i)  $-x, y, z-1/2$ ; (ii)  $x, y, z+1$ .