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Effects of High Hydrostatic Pressure and Storage Temperature on Fatty Acids and Non-Volatile Taste Active Compounds in Red Claw Crayfish (*Cherax quadricarinatus*)

Chunsheng Liu^{1,2,*}, Meng Li², Yuanyuan Wang³, Yi Yang², Aimin Wang^{1,2} and Zhifeng Gu^{1,2,*}

- State Key Laboratory of Marine Resources Utilization in South China Sea, Hainan University, Haikou 570228, China
- ² Ocean College, Hainan University, Haikou 570228, China
- ³ Coconut Research Institute of Chinese Academy of Tropical Agricultural Sciences, Wenchang 570311, China
- * Correspondence: liucs@hainanu.edu.cn (C.L.); hnugu@163.com (Z.G.); Tel.: +86-13086002911 (C.L.)

Abstract: The effects of high hydrostatic pressure (treated with 200, 400 and 600 MPa) and storage temperatures (4 °C and -20 °C) on the fatty acids and flavor compounds of red claw crayfish were studied. HHP decreased the PUFA, GMP, IMP and AMP, citric and lactic acids, and PO₄³⁻ contents, but the FAA, Ca²⁺ and Cl⁻ contents increased in HHP-treated crayfish compared to untreated crayfish at 0 d. Storage at -20 °C could restrain the fatty acids and flavor contents compared to those stored at 4 °C. The GMP, AMP, citric acid and PO₄³⁻ contents decreased, and Ca²⁺ and Cl⁻ contents increased after storage at 4 °C for 15 d (p < 0.05). HHP at 200 and 400 MPa increased EUC on 0 d. No significant changes in EUC were observed after storage at -20 °C for 15 d, significant decreases were noted at 4 °C than the crayfish stored for 0 d (p < 0.05), except for the untreated group. Generally, HHP at 200 or 400 MPa, and storage at -20 °C is beneficial according to the shelling rates and EUC of crayfish.

Keywords: crayfish; high hydrostatic pressure (HHP); cold storage; fatty acids; non-volatile taste active compounds

1. Introduction

Crayfish, representing ~4% of world aquaculture by its market value, is one of the most important and popular aquatic food products worldwide because it is considered a delicacy and abundant nutrients [1]. Red claw crayfish (*Cherax quadricarinatus*) is one of the largest freshwater crayfish species (commercial size of 100–200 g) and is naturally distributed in the tropical regions of Southeast Papua New Guinea and Queensland in Australia [2,3]. This species has been cultured worldwide and has become an important freshwater aquatic product in aquaculture [4,5]. Since its introduction into China, the annual production of red claw crayfish has reached 3000 tons per year [6,7].

Recently, the consumption of shelled fresh crayfish has increased annually in China [8]. However, aquatic products can be easily contaminated with numerous human pathogens and spoilage organisms [8,9]. These deleterious microorganisms are considered the crucial influencing factors for the safety and quality of aquatic food and can raise health risks [10]. To eliminate these microorganisms, many efficient and internationally accepted post-harvesting processes have been developed, such as high hydrostatic pressure (HHP), ultrasound, irradiation, rapid chilling and chemical preservatives [11,12].

HHP is a non-thermal technology with the potential for food preservation and shelf-life extension while maintaining the natural characteristics of food [13,14]. For instance, the total viable counts in razor clams (*Sinonovacula constricta*) were reduced from 6.71 log₁₀ colony-forming unit (CFU)/g to 1.54 log₁₀ CFU/g at 400 MPa HHP treatment for 10 min [15]. The total plate counts in Pacific white shrimp (*Litopenaeus vannamei*) with HHP-treated (550 MPa/5 min/25 °C) were below the level of detection (undetected level,



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). <1 log₁₀ CFU/g) after storing at 4 °C and 25 °C for 15 d [16]. *Vibrio* spp. counts in Suminoe oyster (*Crassostrea ariakensis*) were significantly reduced from ~1 × 10⁵ CFU/g to undetectable levels at 400 MPa HHP treatment for 3 min [9]. Although the quality of aquatic food treated by HHP is less affected compared to conventional thermal processing, HHP, especially at higher pressure, also changes the appearance, nutrient and flavor compounds [11,17]. Recently, the effects of HHP on the quality of many aquatic animals, such as shrimps, crabs, oysters, bay scallops, squids, and fish, have been reported [18–22]. According to these studies, the optimum HHP conditions were various among different aquatic species. Therefore, it is vital to define proper HHP treatment conditions to improve food safety and prevent damage to the nutrition and flavor components in raw aquatic products.

Red claw crayfish has become one of the most important aquatic products due to its large size, high nutritional value and good taste [23]. Our previous study showed that HHP at \geq 200 MPa could completely shuck the shell of red claw crayfish, and the counts of pathogenic *Vibrio* significantly decreased to an undetected level at \geq 400 MPa [24]. However, few studies have reported the biochemical characteristics of this freshwater crayfish species [25,26]. Furthermore, no study has reported the changes in fatty acids and flavor compounds in red claw crayfish treated with different pressure levels and storage temperatures. Therefore, this study aimed to compare the effects of different pressure levels on the fatty acids and flavor compounds [free amino acid (FAA), 5'-nucleotides, organic acids and inorganic ions] in red claw crayfish and evaluate the stability of these characteristics in HHP-treated crayfish after storage at 4 °C and -20 °C for 15 d.

2. Results and Discussion

2.1. Fatty Acid Compositions

The fatty acid compositions of crayfish treated with different HHP, and storage temperatures are shown in Table 1. C20:5n-3 (EPA) was the major fatty acid, accounting for 18.55–23.49% of total fatty acid, followed by C18:1n-9 (16.59–19.62%) and C18:2n-6 (13.18–16.71%) in all treatment groups. Li et al. [26] reported that the top three fatty acids were EPA (19.72–24.36), C18:1n-9 (17.78–23.78%), and C16:0 (13.34–15.79%) in the muscle of female pre-adult red claw crayfish, while the top three fatty acids reported by Méndez-Martínez et al. [23] were C18:1n-9 (29.30–41.50%), C16:0 (19.54–27.35%) and C18:2n-6 (11.90–19.90%). Polyunsaturated fatty acid (PUFA) contents in red claw crayfish also showed obvious differences. A large fluctuation of fatty acid composition often occurs in aquatic animals, which can be affected by life stage, reproductive periodicity, and culture conditions [27]. In the present study, the body weight of crayfish was ~40 g, which showed significant differences in the other two reports (average body weights were 10.8 g and 19.25 g, respectively), causing the differences in fatty acid profiles.

As shown in Table 1, the EPA contents decreased after HHP treatment, and significant changes were observed when comparing 600 MPa-treated crayfish with the control group at 0 d (p < 0.05). On the contrary, the C18:1n-9 and C16:0 contents in crayfish increased after HHP treatment, and significant differences were observed when comparing 400- and 600-MPa treated crayfish with control groups (p < 0.05). The levels of PUFA decreased with the increase of pressure, while the monounsaturated fatty acids (MUFA) and total saturated fatty acids (SFA) showed opposite results. Significant differences were found between the untreated and 600 MPa-treated crayfish at 0 d (p < 0.05). Furthermore, the n-3 PUFAs (mainly contributed by EPA) in crayfish treated with 600 MPa showed a significant decrease compared to the untreated group (p < 0.05). In this study, the higher pressure $(\geq 600 \text{ MPa})$ could cause some changes in the fatty acid profiles. Yi et al. [16] reported that the fat content of Pacific white shrimp also significantly decreased after HHP treatment (550 MPa/5 min), while there were only minor changes found in the compositions of fatty acid in oysters after HHP treatment [9,14]. Oysters have a stronger shell and can tolerate higher pressure compared to shrimps. Crayfish showed a cooked appearance after 600 MPa treatment [24], while only slight changes were observed in soft tissues of oysters treated with the same pressure [9]. The cooked phenomenon in 600 MPa-treated crayfish might cause the changing of fatty acid profiles.

Fatty Acid (%)		0	d			4 °C /1	15 d		−20 °C /15 d				
Fatty Acid (%)	Control	200 MPa	400 MPa	600 MP	Control	200 MPa	400 MPa	600 MP	Control	200 MPa	400 MPa	600 MP	
C12:0	$0.36 \pm 0.02a$	$0.28\pm0.03ab$	$0.21\pm0.05b$	$0.33 \pm 0.07a$	$0.23 \pm 0.04 \mathrm{b}$	$0.32 \pm 0.05a$	$0.25\pm0.05b$	$0.19 \pm 0.07 \mathrm{b}$	$0.27\pm0.02ab$	0.20 ± 0.04	$0.24 \pm 0.02b$	$0.20 \pm 0.03b$	
C14:0	$0.43 \pm 0.04a$	$0.51 \pm 0.07a$	$0.48\pm0.05a$	$0.58 \pm 0.11a$	$0.49 \pm 0.09a$	$0.56 \pm 0.08a$	$0.51\pm0.04a$	$0.46 \pm 0.08a$	$0.45 \pm 0.02a$	$0.47 \pm 0.05a$	$0.51 \pm 0.06a$	$0.49 \pm 0.04a$	
C15:0	$0.98 \pm 0.04a$	$1.01 \pm 0.06a$	$0.96 \pm 0.13a$	$1.23 \pm 0.09a$	$1.13 \pm 0.14a$	$1.18\pm0.07a$	$0.97 \pm 0.09a$	$0.97 \pm 0.13a$	$1.10 \pm 0.07a$	$1.02 \pm 0.05a$	$1.12\pm0.11a$	$1.07 \pm 0.09a$	
C16:0	$6.30 \pm 0.34c$	$7.04 \pm 0.45 bc$	$7.25 \pm 0.19 bc$	$8.01 \pm 0.45a$	$8.02 \pm 0.25a$	7.88 ± 0.61 ab	$8.22 \pm 0.11a$	7.97 ± 0.29 ab	$7.43 \pm 0.12b$	$7.52 \pm 0.22b$	$8.04 \pm 0.18a$	$8.08 \pm 0.35a$	
C17:0	$1.66 \pm 0.05a$	$1.54\pm0.11a$	$1.35\pm0.06a$	$1.65\pm0.07a$	$1.44 \pm 0.05a$	$1.70\pm0.08a$	$1.53\pm0.06a$	$1.36\pm0.06a$	$1.58\pm0.06a$	$1.31\pm0.12a$	$1.36\pm0.09a$	$1.51 \pm 0.21a$	
C18:0	$6.26\pm0.71a$	$6.05 \pm 0.09a$	$6.24 \pm 0.22a$	$6.54 \pm 0.31a$	$6.68 \pm 0.18a$	$6.08 \pm 0.25a$	$6.04 \pm 0.31a$	$6.36 \pm 0.11a$	$5.93 \pm 0.26a$	$6.51 \pm 0.41a$	$6.05\pm0.18a$	$5.96 \pm 0.25a$	
C20:0	$1.09 \pm 0.04a$	$0.88 \pm 0.03 ab$	$0.65 \pm 0.07 \mathrm{b}$	$1.02 \pm 0.12a$	$0.75 \pm 0.09 ab$	$0.94 \pm 0.05 \mathrm{ab}$	$0.80 \pm 0.07 \mathrm{ab}$	$0.65 \pm 0.07 b$	$0.86 \pm 0.05 ab$	$0.68 \pm 0.04b$	$0.79 \pm 0.08 \mathrm{ab}$	$0.66 \pm 0.06b$	
ΣSFA	$17.08\pm1.03b$	$17.30\pm0.78b$	$17.14\pm0.66\mathrm{b}$	$19.36\pm0.98a$	$18.74\pm0.74 \mathrm{ab}$	$18.67 \pm 1.15 \mathrm{ab}$	$18.32\pm0.66\text{ab}$	$17.96\pm0.80\mathrm{b}$	$17.64\pm0.47\mathrm{b}$	$17.71 \pm 0.93b$	$18.11\pm0.68 \mathrm{ab}$	17.96 ± 0.93 ab	
C16:1n-7	$2.98\pm0.11a$	$2.78 \pm 0.11a$	$2.60\pm0.09a$	$3.14\pm0.21a$	$2.87 \pm 0.17a$	$3.17 \pm 0.22a$	$3.25\pm0.28a$	$2.86 \pm 0.11a$	$3.09 \pm 0.09a$	$2.71 \pm 0.11a$	$2.98 \pm 0.21a$	$2.91 \pm 0.25a$	
C16:1n-5	$0.65\pm0.08a$	$0.51\pm0.03a$	$0.39 \pm 0.06a$	$0.59\pm0.04a$	$0.44\pm0.06a$	$0.58\pm0.02a$	$0.49\pm0.04a$	$0.40\pm0.02a$	$0.53\pm0.05a$	$0.39\pm0.03a$	$0.48\pm0.06a$	$0.42\pm0.02a$	
C17:1n-7	$1.27\pm0.02a$	$1.10\pm0.03a$	$0.93 \pm 0.06a$	$1.28\pm0.07a$	$1.07\pm0.05a$	$1.23\pm0.07a$	$1.13\pm0.09a$	$1.04\pm0.07a$	$1.16\pm0.07a$	$0.95\pm0.10a$	$1.06\pm0.06a$	$1.03 \pm 0.08a$	
C18:1n-9	$16.59 \pm 1.12b$	$17.11 \pm 0.56b$	$18.26 \pm 0.66 ab$	18.56 ± 1.23 ab	19.07 ± 0.83 ab	18.31 ± 0.79 ab	18.93 ± 1.37 ab	$20.64 \pm 0.98a$	$17.50 \pm 1.02 ab$	$18.00 \pm 0.96 \mathrm{ab}$	18.44 ± 1.17 ab	$19.62\pm0.86a$	
ΣMUFA	$21.49 \pm 1.10 \mathrm{c}$	$21.50\pm0.64c$	$22.19\pm0.75 bc$	$23.57\pm0.98 ab$	$23.44\pm0.86\text{ab}$	$23.29\pm0.84ab$	$23.79 \pm 1.30 \mathrm{ab}$	$24.94 \pm 1.04 a$	$22.26 \pm 1.03 bc$	$22.06\pm0.98 bc$	$22.97 \pm 1.26 bc$	$23.98\pm0.95 ab$	
C18:2n-6	$13.18\pm0.78c$	$15.18\pm0.31 \mathrm{ab}$	15.70 ± 1.13 ab	$15.10\pm0.95 \mathrm{ab}$	15.86 ± 0.79 ab	15.04 ± 1.01 abc	15.83 ± 0.79 ab	$15.80 \pm 0.63 ab$	$14.89 \pm 1.04 \mathrm{abc}$	$16.63 \pm 0.77a$	$16.71\pm0.83a$	$16.56\pm0.49a$	
C18:3n-6	$1.36 \pm 0.04a$	$1.05 \pm 0.03 ab$	$0.79 \pm 0.05c$	$1.22 \pm 0.06a$	0.93 ± 0.07 ab	$1.20 \pm 0.04a$	1.00 ± 0.11 ab	$0.81 \pm 0.06c$	$1.09\pm0.07 \mathrm{ab}$	0.82 ± 0.04	$0.95\pm0.07 \mathrm{ab}$	$0.83 \pm 0.05 ab$	
C18:3n-5	$1.81 \pm 0.05a$	$1.36\pm0.05 \mathrm{ab}$	$0.94 \pm 0.08 \mathrm{c}$	1.53 ± 0.14 ab	$1.12 \pm 0.05c$	$1.53\pm0.06ab$	$1.17\pm0.09c$	$0.92 \pm 0.08c$	$1.33\pm0.08ab$	$1.12\pm0.07c$	$1.24 \pm 0.13c$	$0.96 \pm 0.08c$	
C18:3n-3	$2.32\pm0.10a$	$2.11 \pm 0.22a$	$2.73 \pm 0.08a$	$2.01 \pm 0.25a$	$2.01 \pm 0.18a$	$2.43 \pm 0.16a$	$2.20 \pm 0.26a$	$1.76 \pm 0.14a$	$2.21 \pm 0.14a$	$1.81\pm0.12a$	$2.07\pm0.09a$	$2.15\pm0.21a$	
C18:2n-7	$1.60\pm0.06a$	$1.33 \pm 0.05a$	$1.01 \pm 0.08a$	$1.70\pm0.07a$	$1.25\pm0.09a$	$1.59\pm0.05a$	$1.20 \pm 0.07a$	$1.05\pm0.08a$	$1.32\pm0.09a$	$1.08\pm0.05a$	1.24 ± 0.11 a	$1.00 \pm 0.07a$	
C20:2n-6	$2.76\pm0.07a$	$2.88\pm0.06a$	$2.77\pm0.11a$	$3.05\pm0.14a$	$2.78\pm0.06a$	$3.03 \pm 0.21a$	$2.82\pm0.07a$	$2.86 \pm 0.21a$	$2.85\pm0.08a$	$2.88\pm0.09a$	$3.01 \pm 0.26a$	$2.92 \pm 0.17a$	
C20:4n-6 (ARA)	$10.35 \pm 0.33a$	$9.99 \pm 0.23a$	$8.06 \pm 0.36 bc$	$7.60 \pm 0.27 \mathrm{c}$	$8.00 \pm 0.18 \mathrm{bc}$	8.92 ± 0.31 ab	$7.65 \pm 0.15c$	$8.25 \pm 0.22 bc$	$8.65 \pm 0.37 bc$	$7.92 \pm 0.29c$	$7.47\pm0.26c$	$8.11 \pm 0.19 \mathrm{bc}$	
C20:5n-3 (EPA)	$22.23\pm1.09a$	$21.22\pm1.10 \mathrm{ab}$	$22.19 \pm 1.16a$	$19.06\pm0.78 \mathrm{bc}$	20.62 ± 1.22 ab	$18.55\pm0.60\mathrm{c}$	$20.43\pm0.75 ab$	$19.73\pm1.01 \mathrm{bc}$	$22.40\pm0.85a$	$22.09\pm0.77a$	$20.59\pm0.82ab$	$20.10\pm0.46 abc$	
C22:6n-3 (DHA)	$5.81 \pm 0.22a$	$6.07\pm0.18a$	$6.37\pm0.32a$	$5.80 \pm 0.27a$	$5.25\pm0.18a$	$5.75\pm0.36a$	$5.59\pm0.16a$	$5.93 \pm 0.31a$	$5.36 \pm 0.41a$	$5.89 \pm 0.24a$	$5.63 \pm 0.19a$	$5.44 \pm 0.26a$	
ΣPUFA	$61.44 \pm 2.26a$	$61.19 \pm 2.10a$	60.68 ± 2.57 ab	$57.07 \pm 2.18b$	$57.82 \pm 1.68b$	$58.04 \pm 1.96b$	$57.90 \pm 2.14b$	$57.50 \pm 2.41b$	$60.10 \pm 2.51 ab$	60.23 ± 2.50 ab	$58.92 \pm 2.17b$	$58.07 \pm 1.76b$	
Σn-3	30.37 ± 1.23 ab	$29.40 \pm 1.32 \mathrm{ab}$	$31.39 \pm 1.44a$	$27.07 \pm 1.16 \mathrm{c}$	$27.88 \pm 1.37 bc$	$26.73 \pm 1.04 \mathrm{c}$	$28.22 \pm 1.13 bc$	$27.42 \pm 1.34 bc$	29.97 ± 1.32 ab	$29.79 \pm 1.08 \mathrm{ab}$	$28.30 \pm 1.01 \mathrm{bc}$	$27.69\pm0.89 \mathrm{bc}$	
Σn-6	27.65 ± 1.19 ab	$29.10\pm0.57a$	$27.33 \pm 1.36 \mathrm{ab}$	$26.77\pm1.12b$	$27.57\pm0.97 ab$	$28.20 \pm 1.36 \mathrm{ab}$	$27.30 \pm 1.02 ab$	27.71 ± 1.12 ab	$27.47 \pm 1.23 ab$	$28.24 \pm 1.13 \mathrm{ab}$	28.15 ± 1.23 ab	$28.41\pm0.75 ab$	
$\Sigma n-3/\Sigma n-6$	1.10	1.01	1.15	1.01	1.01	0.95	1.03	0.99	1.09	1.05	1.01	0.97	

Table 1. Changes in fatty acid profiles of HHP treated red claw crayfish with different treatments.

Data are presented as mean \pm standard deviation. Different letters within the same row denote significant differences (p < 0.05). SFA—saturated fatty acids; MUFA—monounsaturated fatty acids; PUFA—polyunsaturated fatty acids. Σ n-3: 8:3n3; 20:4n-3; 20:5n-3; 22:6n-3. Σ n-6: 18:2n6; 18:3n-6; 20:2n6; 20:4n-6.

Physical and biochemical changes occur in aquatic animals during chilled storage. In our study, the fatty acid compositions in all treatment groups showed almost no significant changes after 15-d storage at both 4 °C and -20 °C, except for some kinds of fatty acids in control and 200 MPa-treated crayfish at 4 °C. In control and 200 MPa-treated crayfish, the MUFA significantly increased from 21.49% and 21.50% to 23.44% and 23.29%, respectively, while the PUFA significantly decreased from 61.44% and 61.19% to 57.82% and 58.04%, respectively (p < 0.05). Similarly, decreases in PUFA composition were observed in Bogue (*Boops boops*) and raw oyster after ~two weeks of storage at 4 °C [9,28]. In our previous study, there were numerous spoilage organisms observed in untreated and 200 MPa-treated crayfish [24]. If these spoilage bacteria were not completely killed, the lipid could be deteriorated, which led to changes in fatty acid profiles [29].

2.2. Changes of FAAs and 5'-Nucleotide Contents in Crayfish

As shown in Table 2, twenty FAAs were identified in crayfish. Of these, arginine (Arg, 5.70–8.47 mg/g) content was the most abundant in all crayfish groups at 0 d, followed by glycine (Gly, 1.03–2.22 mg/g), glutamine (Gln, 0.99–1.76 mg/g), and alanine (Ala, 0.99–1.73 mg/g). According to the taste thresholds of different FAAs, the taste activity values (TAV) of Arg, Gly, Ala, and histidine (His) were more than 1 and considered active FAAs in crayfish [9]. As shown in previous reports, the main FAA contents were different in different aquatic species. For example, the top three FAAs were Arg, proline (Pro), and Ala in Japanese flying squid (*Todarodes pacificus*) muscles [21]; Gly, Ala, and glutamic acid (Glu) in Suminoe oysters [9]; Pro, Arg, and Gly in Chinese shrimp (*Fenneropenaeus chinensis*) and mud crab (*Scylla paramamosain*) [30,31]. Therefore, the differences in FAAs content contribute to various tastes of aquatic produces.

In comparing the effects of HHP on the FAA contents of crayfish muscle, we found that the total contents of umami amino acid (UAA) and bitter amino acid (BAA) in crayfish significantly increased after HHP treatment compared to the untreated group at 0 d (p < 0.05). FAAs are produced by proteolysis and certain amino acid metabolic pathways [21,32]. HHP can increase the contents of FAA according to protein denaturation, which has been reported in many aquatic species, such as the Suminoe oyster, squid, and cod (*Gadus morhua*) [9,21,33]. Furthermore, there were almost no significant differences in FAA contents of crayfish after storage for 15 d at 4 °C and -20 °C when compared to the crayfish treated with the same HHP levels, except for some kinds of FAA stored at 4 °C (significantly increased and decreased in tryptophan (Trp) and Arg, respectively, p < 0.05).

In previous reports, the tends of FFA contents were different after a period of storage. For example, the total FAA contents in squids significantly increased after 10-d storage at $4 \degree C (p < 0.05)$, while significant decreases in total FAA contents of Suminoe oyster were observed after 15-d storage at $4 \degree C$ and $-20 \degree C (p < 0.05)$ [9,21]. In hammour (*Epinephelus coioides*), there was a slight decrease in total FAA contents of the HHP-treated group after 30-d storage at $4 \degree C$ when compared to those at 0 d (p > 0.05), which showed the same result as our study [34]. Generally, FAAs in an organism are produced from both proteolysis and certain amino acid metabolic pathways, in which proteolysis increases FAA contents while amino acid metabolism decreases the concentration of specific amino acids [32]. HHP treatment and refrigerated/frozen storage could modify this progress. Therefore, these discrepant contents of FFAs could be explained by the differences in the food system, storage temperature, pressure and substrates supplied by HHP-promoted proteolysis and metabolic rates [21].

FAA (mg/g)	0 d				4 ° C/15 d								
	Control	200 MPa	400 MPa	600 MPa	Control	200 MPa	400 MPa	600 MPa	Control	200 MPa	400 MPa	600 MPa	Taste Attribut
Aspartic acid	$0.02 \pm 0.00c$	$0.03 \pm 0.01c$	$0.06 \pm 0.01 ab$	$0.08 \pm 0.02a$	$0.03 \pm 0.01 \mathrm{c}$	$0.03 \pm 0.01c$	$0.10 \pm 0.01a$	$0.09 \pm 0.02a$	$0.02 \pm 0.02c$	$0.02 \pm 0.01c$	$0.04 \pm 0.01 bc$	$0.09 \pm 0.01a$	Umami (+)
Glutamic acid	$0.10 \pm 0.01e$	$0.23 \pm 0.02d$	$0.42 \pm 0.02 bc$	0.28 ± 0.01 cd	$0.13 \pm 0.02e$	$0.21 \pm 0.03d$	$0.58 \pm 0.03a$	$0.47 \pm 0.02 ab$	$0.11 \pm 0.02e$	$0.23 \pm 0.03d$	$0.34 \pm 0.03c$	0.27 ± 0.02 cd	Umami (+)
ΣUAA	$0.12 \pm 0.01 f$	0.26 ± 0.03 de	$0.48 \pm 0.03 bc$	0.36 ± 0.02 cd	$0.16 \pm 003 ef$	0.24 ± 0.03 de	$0.68 \pm 0.04a$	$0.56 \pm 0.03ab$	$0.13 \pm 0.03 f$	0.25 ± 0.04 de	0.38 ± 0.04 cd	0.36 ± 0.03 cd	
Threonine	$0.12 \pm 0.02b$	$0.30 \pm 0.03a$	$0.27 \pm 0.04a$	$0.41 \pm 0.02a$	$0.26 \pm 0.03a$	$0.30 \pm 0.04a$	$0.18 \pm 0.02 ab$	$0.30 \pm 0.03a$	$0.23 \pm 0.03 ab$	$0.20 \pm 0.02 ab$	$0.31 \pm 0.05a$	$0.41 \pm 0.03a$	Sweet (+)
Serine	$0.24 \pm 0.02a$	$0.29 \pm 0.02a$	$0.17 \pm 0.02a$	$0.26 \pm 0.03a$	$0.25 \pm 0.01a$	$0.19 \pm 0.02a$	$0.08 \pm 0.03b$	$0.08 \pm 0.03b$	$0.25 \pm 0.02a$	$0.23 \pm 0.03a$	$0.25 \pm 0.01a$	$0.23 \pm 0.04a$	Sweet (+)
Glycine	$2.22 \pm 0.11a$	$1.07 \pm 0.08b$	$1.27 \pm 0.06b$	$1.03 \pm 0.09b$	$0.98 \pm 0.08b$	$1.56 \pm 0.13ab$	$1.47 \pm 0.06 ab$	$1.12 \pm 0.07b$	$0.96 \pm 0.08b$	$1.23 \pm 0.05b$	$1.57 \pm 0.06 ab$	$1.23 \pm 0.04b$	Sweet (+)
Alanine	$0.99 \pm 0.12b$	$1.73 \pm 0.05 ab$	$1.54 \pm 0.05 ab$	$1.47 \pm 0.03 ab$	$1.60 \pm 0.08 ab$	$1.99 \pm 0.06a$	$2.00 \pm 0.22a$	1.77 ± 0.24 ab	$1.09 \pm 0.09b$	$1.20 \pm 0.04b$	$1.64 \pm 0.22 ab$	$1.64 \pm 0.06 ab$	Sweet (+)
Proline	$0.23 \pm 0.03c$	$0.48 \pm 0.04 bc$	$0.54 \pm 0.03 ab$	$0.37 \pm 0.05 bc$	$0.55 \pm 0.03 ab$	$0.57 \pm 0.05 ab$	$0.86 \pm 0.04a$	$0.55 \pm 0.05 ab$	$0.41 \pm 0.05 bc$	$0.42 \pm 0.04 bc$	$0.58 \pm 0.06 ab$	$0.41 \pm 0.05 bc$	Sweet/bitter (+)
Glutamine	$0.99 \pm 0.14c$	$1.56 \pm 0.08 ab$	$1.76 \pm 0.06a$	$1.52 \pm 0.06ab$	$0.84 \pm 0.07 bc$	$0.91 \pm 0.08c$	$1.01 \pm 0.05c$	$1.33 \pm 0.09 bc$	$1.24 \pm 0.12 bc$	$1.35 \pm 0.07 bc$	$1.87 \pm 0.11a$	$1.85 \pm 0.08a$	Sweet (+)
Asparagine	$0.24 \pm 0.04b$	0.40 ± 0.04 ab	$0.53 \pm 0.02ab$	$0.61 \pm 0.05a$	$0.24 \pm 0.04b$	$0.54 \pm 0.03 ab$	$0.38 \pm 0.04b$	$0.49 \pm 0.05 ab$	$0.35 \pm 0.05b$	$0.37 \pm 0.07b$	$0.59 \pm 0.05 ab$	$0.66 \pm 0.06a$	Sweet (+)
SAA	$5.03 \pm 0.44c$	$5.83 \pm 0.37 bc$	$6.08 \pm 0.22 ab$	$5.67 \pm 0.26 bc$	$4.72 \pm 0.34c$	$6.06 \pm 0.30 ab$	$5.98 \pm 0.42ab$	$5.64 \pm 0.54 bc$	$4.53 \pm 0.41c$	$5.00 \pm 0.28c$	$6.81 \pm 0.51a$	$6.52 \pm 0.33 ab$	
aline	$0.10 \pm 0.02c$	$0.24 \pm 0.02ab$	$0.21 \pm 0.02 abc$	$0.23 \pm 0.03 abc$	$0.26 \pm 0.03 ab$	$0.36 \pm 0.02a$	$0.40 \pm 0.04a$	$0.21 \pm 0.02 abc$	$0.16 \pm 0.02 bc$	$0.17 \pm 0.03 bc$	0.23 ± 0.04 abc	$0.21 \pm 0.04 abc$	Bitter/sweet (-)
Aethionine ($0.11 \pm 0.02b$	$0.23 \pm 0.04a$	$0.20 \pm 0.02ab$	$0.23 \pm 0.01a$	$0.18 \pm 0.03 ab$	$0.25 \pm 0.01a$	$0.28 \pm 0.01a$	$0.20 \pm 0.04 ab$	$0.12 \pm 0.01b$	$0.15 \pm 0.03b$	$0.21 \pm 0.04 ab$	$0.24 \pm 0.03a$	Bitter/sweet/sulfurous (-
eucine	$0.11 \pm 0.04c$	$0.28 \pm 0.05 ab$	$0.21 \pm 0.03 bc$	$0.23 \pm 0.05 bc$	$0.31 \pm 0.06 ab$	$0.49 \pm 0.08a$	$0.52 \pm 0.09a$	$0.24 \pm 0.07 bc$	$0.15 \pm 0.03c$	$0.18 \pm 0.06 bc$	$0.22 \pm 0.04 bc$	$0.23 \pm 0.01 bc$	Bitter (-)
ryptophan	$0.24 \pm 0.04 e$	$1.47 \pm 0.06d$	2.36 ± 0.21 cd	$0.15 \pm 0.05e$	2.01 ± 0.23 cd	$3.15 \pm 0.31b$	$5.94 \pm 0.62a$	$2.84 \pm 0.14c$	$0.49 \pm 0.05e$	$1.32 \pm 0.07 d$	$2.86 \pm 0.11c$	$0.22 \pm 0.06e$	Bitter (-)
Phenylalanine	$0.16 \pm 0.03 bc$	$0.13 \pm 0.02c$	$0.24 \pm 0.04b$	$0.25 \pm 0.02b$	$0.25 \pm 0.04b$	$0.39 \pm 0.05a$	$0.44 \pm 0.06a$	$0.31 \pm 0.04 ab$	$0.16 \pm 0.03 bc$	$0.19 \pm 0.06 bc$	$0.23 \pm 0.05b$	$0.25 \pm 0.04b$	Bitter (-)
ysine	$0.19 \pm 0.06c$	$0.48 \pm 0.03b$	$0.44 \pm 0.07b$	$0.46 \pm 0.07b$	$0.27 \pm 0.05 bc$	$0.67 \pm 0.07a$	$0.74 \pm 0.04a$	$0.45 \pm 0.06b$	$0.24 \pm 0.04 bc$	$0.30 \pm 0.05 bc$	$0.45 \pm 0.06b$	$0.58 \pm 0.04 ab$	Bitter/sweet (-)
Argnine	$6.26 \pm 0.56b$	$5.70 \pm 0.36 bc$	$5.83 \pm 0.44 bc$	$8.47 \pm 0.48a$	$3.00 \pm 0.20d$	$3.09 \pm 0.19d$	$2.24 \pm 0.22e$	$5.16 \pm 0.19c$	$5.44 \pm 0.45 bc$	$4.98 \pm 0.26c$	$5.98 \pm 0.41b$	$8.28 \pm 0.59a$	Bitter/sweet (-)
soleucine	$0.06 \pm 0.03b$	$0.16 \pm 0.05 ab$	$0.13 \pm 0.01 ab$	$0.14 \pm 0.04 ab$	$0.17 \pm 0.02 ab$	$0.26 \pm 0.06a$	$0.29 \pm 0.04a$	0.14 ± 0.04 ab	$0.09 \pm 0.06b$	$0.10 \pm 0.04 b$	$0.15 \pm 0.05 ab$	$0.14 \pm 0.03 ab$	Bitter (-)
listidine	$0.34 \pm 0.04b$	$0.49 \pm 0.02a$	$0.32 \pm 0.05b$	$0.37 \pm 0.06 ab$	$0.26 \pm 0.03b$	$0.53 \pm 0.04a$	$0.38 \pm 0.03 ab$	$0.42 \pm 0.06 ab$	$0.31 \pm 0.03b$	$0.34 \pm 0.06b$	$0.37 \pm 0.03 ab$	$0.43 \pm 0.06 ab$	Bitter (-)
yrosine	$0.15 \pm 0.03a$	$0.24 \pm 0.06a$	$0.23 \pm 0.05a$	$0.26 \pm 0.06a$	$0.24 \pm 0.04a$	$0.32 \pm 0.02a$	$0.31 \pm 0.06a$	$0.21 \pm 0.07a$	$0.16 \pm 0.04a$	$0.17 \pm 0.06a$	$0.15 \pm 0.04a$	$0.33 \pm 0.07a$	Bitter (-)
ysteine	$0.03 \pm 0.02a$	$0.05 \pm 0.03a$	$0.05 \pm 0.01a$	ND	$0.03 \pm 0.01a$	$0.04 \pm 0.02a$	$0.04 \pm 0.03a$	$0.03 \pm 0.01a$	$0.03 \pm 0.02a$	$0.03 \pm 0.02a$	$0.04 \pm 0.01a$	ND	Bitter/sweet/sulfurous (-
EBAA	$7.75 \pm 0.87d$	$9.37 \pm 0.70 bc$	$10.22 \pm 0.87 ab$	$10.79 \pm 0.80 ab$	$6.98 \pm 0.66d$	$9.55 \pm 0.78 bc$	$11.58 \pm 1.01a$	$10.21 \pm 0.54 ab$	$7.35 \pm 0.36d$	7.93 ± 0.67 cd	$10.89 \pm 0.76 ab$	$10.91 \pm 0.81 ab$	
LFAA	$12.90 \pm 1.21d$	$15.46 \pm 0.94 bc$	$16.78 \pm 1.01ab$	$16.82 \pm 0.95 ab$	$11.86 \pm 0.89d$	$15.85 \pm 0.94 bc$	$18.44 \pm 1.23a$	$16.41 \pm 1.04ab$	$12.01 \pm 0.72d$	$13.06 \pm 0.89d$	$18.08 \pm 1.13a$	$17.86 \pm 0.99a$	

 Table 2. The contents and taste attributes of FAAs in red claw crayfish with different treatments.

Data are presented as mean \pm standard deviation. Different letters within the same row denote significant differences (p < 0.05). ND: none detected. Σ UAA—total umami amino acid; Σ SAA—total sweet amino acid; Σ BAA—total bitter amino acid; Σ FAA—total free amino acid. +—pleasant taste; -—unpleasant taste.

As shown in Table 3, GMP, IMP, and AMP were the main 5'-nucleotide components in untreated crayfish (57.15, 42.27, and 51.80 mg/100 g, respectively). Furthermore, the contents of GMP, IMP, and AMP decreased after HHP treatment (except for IMP in the 200 MPa-treated group). Significant differences were observed in GMP content of 200, 400, and 600 MPa-treated groups, IMP contents of 400 and 600 MPa-treated groups, and AMP content in 600 MPa-treated group, compared to untreated crayfish at 0 d (p < 0.05). After 15-d storage at both 4 °C and -20 °C, the IMP contents in HHP-treated crayfish groups were almost the same as those at 0 d, while significant increases were observed in HHP-untreated crayfish (p < 0.05). However, there were no obvious differences in GMP and AMP contents of crayfish after storage for 15 d at -20 °C, compared to the same treated crayfish at 0 d. Significant decreases were observed in these two 5'-nucleotides when crayfish were stored at 4 °C for 15 d (p < 0.05).

Table 3. The concentrations of 5'-nucleotides in red claw crayfish with different treatments.

Nucleotides (mg/100 g)	0 d					4 ° C	/15 d		−20 °C/15 d			
	Control	200 MPa	400 MPa	600 MPa	Control	200 MPa	400 MPa	600 MPa	Control	200 MPa	400 MPa	600 MPa
CMP UMP GMP IMP AMP	$\begin{array}{c} 0.48 \pm 0.21b \\ ND \\ 57.15 \pm 3.48a \\ 42.27 \pm 3.39b \\ 51.80 \pm 2.98ab \end{array}$	$\begin{array}{c} 0.63 \pm 0.51 ab \\ 0.39 \pm 0.09 d \\ 25.46 \pm 3.77 b \\ 69.61 \pm 5.08 a \\ 40.52 \pm 3.61 bc \end{array}$	$\begin{array}{c} 0.74 \pm 0.26 ab \\ 0.69 \pm 0.11 cd \\ 16.14 \pm 2.97 bc \\ 14.31 \pm 2.79 c \\ 46.40 \pm 7.99 bc \end{array}$	$\begin{array}{c} 0.68 \pm 0.06 ab \\ 0.80 \pm 0.13 cd \\ 16.25 \pm 2.31 bc \\ ND \\ 31.68 \pm 5.32 c \end{array}$	$\begin{array}{c} 0.54 \pm 0.16b \\ 1.02 \pm 0.11bc \\ 18.10 \pm 3.05e \\ 77.48 \pm 4.52a \\ 6.90 \pm 3.32d \end{array}$	$\begin{array}{c} 0.54 \pm 0.17b \\ 0.73 \pm 0.09cd \\ 2.83 \pm 1.50e \\ 61.78 \pm 5.11a \\ 3.94 \pm 1.57d \end{array}$	$\begin{array}{c} 0.86 \pm 0.11 ab \\ 1.32 \pm 0.91 ab \\ 4.83 \pm 1.04 de \\ 9.75 \pm 1.55 c \\ 6.80 \pm 2.79 d\end{array}$	$\begin{array}{c} 0.48 \pm 0.16b \\ 0.72 \pm 0.22cd \\ 8.91 \pm 1.88cd \\ 1.04 \pm 0.09d \\ 4.97 \pm 3.52d \end{array}$	$\begin{array}{c} 0.49 \pm 0.09b \\ 0.33 \pm 0.17d \\ 50.02 \pm 2.48a \\ 76.11 \pm 7.32a \\ 65.54 \pm 5.48a \end{array}$	$\begin{array}{c} 0.54 \pm 0.13b \\ 0.45 \pm 0.16d \\ 23.24 \pm 2.49b \\ 63.40 \pm 5.17a \\ 46.26 \pm 4.42bc \end{array}$	$\begin{array}{c} 0.71 \pm 0.33 ab \\ 1.25 \pm 0.24 ab \\ 21.43 \pm 2.99 b \\ 12.82 \pm 3.12 c \\ 64.56 \pm 7.41 a \end{array}$	$\begin{array}{c} 1.13 \pm 0.09a \\ 2.45 \pm 0.91a \\ 17.05 \pm 3.67 bc \\ 1.06 \pm 0.55 d \\ 55.15 \pm 4.32 ab \end{array}$

Data are presented as mean \pm standard deviation. Different letters within the same row denote significant differences (p < 0.05). ND—none detected.

GMP, IMP, and AMP are the three flavor-contributing 5'-nucleotides [9,14]. GMP provides a meaty flavor and can be used as a flavor enhancer [35]. IMP is an umami substance, and its taste can be strongly enhanced by some kinds of sweet amino acids, such as serine (Ser), Gly, and Ala [9]. AMP promotes umami and sweet taste in some seafood [21]. The decreases in 5'-nucleotides after HHP treatment and 4 °C storage were also observed in oysters, which showed the same tendency as our study [9,14].

2.3. Effect of HHP on EUC of Crayfish

As shown in Figure 1, the EUC values of crayfish treated with different HHP pressures and storage temperatures were calculated. The EUC values of crayfish first increased and then decreased with the enhancement of HHP pressure at 0 d. The highest EUC value was observed in 200 MPa-treated crayfish (3.92 g MSG/100 g wet weight), followed by 400 MPa-treated, untreated, and 600 MPa-treated individuals (3.11, 2.44, and 1.53 g MSG/100 g wet weight, respectively). Significant differences were observed among the four treatment crayfish groups (p < 0.05). Comparing the EUC values of crayfish with other aquatic produces, the values in untreated crayfish were similar to Chinese mitten crab (*Eriocheir sinensis*) (2.07–5.42 g MSG/100 g) [36], lower than raw oysters (6.47 g MSG/100 g) [9], and higher than Yangtze (*Coilia ectenes*) (0.33–0.99 g MSG/100 g) [37]. Furthermore, there were no significant differences in EUC values in crayfish stored for 15 d at -20 °C compared to the same treatment groups at 0 d. In comparison, significant decreases were found in crayfish at 4 °C for 15 d compared to those at 0 d (p < 0.05), except of the 600 MPa treatment group.

The EUC values were calculated by mixing two umami amino acids (Asp and Glu) and three 5'-nucleotides (GMP, IMP, and AMP). This parameter has been widely used to evaluate the umami taste of aquatic foods, such as crab, squid, and oyster [9,14,21,36]. In this study, Glu, IMP and GMP were the main umami compositions in crayfish, which influenced the EUC values. In addition, storage at -20 °C could significantly slow down the EUC values compared to 4 °C conditions for crayfish.



Figure 1. The equivalent umami concentration (EUC, g MSG/100 g wet weight) of red claw crayfish with different treatments. Bars with different letters denote significantly different (p < 0.05).

2.4. Comparison of Organic Acids and Betaine in Crayfish

As shown in Table 4, the concentrations of betaine were almost undetectable in untreated crayfish, while citric acid (59.82 mg/g), succinic acid (9.62 mg/g), and lactic acid (2.67 mg/g) were the predominant organic acids. The contents of citric and lactic acids significantly decreased after HHP pressure compared to untreated crayfish (p < 0.05), except for lactic acid in the 200 MPa-treated group. While the succinic acid slightly decreased after HHP pressure, significant differences were observed only in 600 MPa-treated crayfish compared to untreated ones (p < 0.05). In aquatic foods, the organic acids are mainly citric, lactic, acetic, malic, succinic, and propionic acids. The composition and concentrations of these organic acids are significantly different according to species and living conditions [30,38]. The major organic acids in Suminoe oyster were citric and succinic acids in mud crab [30]. Beside acidic taste, organic acids contribute to other flavors. For example, succinic acid could enhance bitter taste and strong salty at different concentrations [39]. Citric acid could contribute to oysters' soft, crisp acidic taste [9].

Table 4. The concentrations of organic acids and betaine in red claw crayfish with different treatments.

Organic	b					4 ° C	/15 d		−20 °C/15 d			
(mg/g)	Control	200 MPa	400 MPa	600 MP	Control	200 MPa	400 MPa	600 MP	Control	200 MPa	400 MPa	600 MP
Malic acid Lactic acid Citric acid Succinic acid Betaine	$\begin{array}{c} 0.16 \pm 0.03 ef \\ 2.67 \pm 0.12 d \\ 59.82 \pm 2.17 a \\ 9.62 \pm 0.22 ab \\ 0.01 \pm 0.00 a \end{array}$	$\begin{array}{c} 0.09 \pm 0.03 f \\ 2.53 \pm 0.21 d \\ 11.47 \pm 1.04 d \\ 8.74 \pm 0.52 b c \\ 0.00 \pm 0.00 a \end{array}$	$\begin{array}{c} 0.07 \pm 0.01 f \\ 1.14 \pm 0.05 e \\ 11.69 \pm 0.95 d \\ 8.32 \pm 0.38 b c \\ 0.00 \pm 0.00 a \end{array}$	$\begin{array}{c} 0.12 \pm 0.02 ef \\ 1.40 \pm 0.08 e \\ 17.08 \pm 4.42 cd \\ 7.94 \pm 0.53 c \\ 0.01 \pm 0.00 a \end{array}$	$\begin{array}{c} 1.41 \pm 0.09a \\ 6.24 \pm 0.35a \\ 2.78 \pm 0.26e \\ 9.75 \pm 0.48ab \\ 0.00 \pm 0.00a \end{array}$	$\begin{array}{c} 0.97 \pm 0.06b \\ 4.22 \pm 0.18c \\ 2.61 \pm 0.44e \\ 11.80 \pm 0.73a \\ 0.00 \pm 0.00a \end{array}$	$\begin{array}{c} 0.63 \pm 0.06c \\ 0.83 \pm 0.08e \\ 2.61 \pm 0.28e \\ 8.87 \pm 0.49bc \\ 0.00 \pm 0.00a \end{array}$	$\begin{array}{c} 0.15 \pm 0.02 ef \\ 1.35 \pm 0.09 e \\ 5.88 \pm 2.15 e \\ 7.60 \pm 0.43 c \\ 0.00 \pm 0.00 a \end{array}$	$\begin{array}{c} 0.32 \pm 0.03d \\ 5.43 \pm 0.36b \\ 36.75 \pm 3.47b \\ 10.47 \pm 1.25ab \\ 0.00 \pm 0.00a \end{array}$	$\begin{array}{c} 0.22\pm 0.02 de\\ 6.41\pm 0.28 a\\ 12.24\pm 2.17 d\\ 11.91\pm 1.39 a\\ 0.00\pm 0.00 a\end{array}$	$\begin{array}{c} 0.02\pm 0.02\\ 1.22\pm 0.12e\\ 12.74\pm 0.48d\\ 9.69\pm 1.04ab\\ 0.00\pm 0.00a \end{array}$	$\begin{array}{c} 0.15 \pm 0.09 ef \\ 1.26 \pm 0.09 e \\ 19.78 \pm 2.86 c \\ 7.26 \pm 0.59 c \\ 0.00 \pm 0.00 a \end{array}$

Data are presented as mean \pm standard deviation. Different letters within the same row denote significant differences (p < 0.05).

The citric acid contents in all groups significantly decreased after storing for 15 d at 4 °C (p < 0.05). On the contrary, the malic acid content in all treatments and lactic acid content in control and 200 MPa treatment crayfish significantly increased (p < 0.05). For crayfish stored at -20 °C, significant decreases in lactic acid contents were observed in the control and 200 MPa treatment groups (p < 0.05). Furthermore, the citric acid contents in untreated crayfish significantly decreased to 36.75 mg/g of wet weight but still showed significantly higher than those at 4 °C (p < 0.05). Citric acid is the main substrate in the Krebs cycle of living organisms and undergoes degradation [40]. Therefore, its easy

degradability caused by microorganisms may be the main reason for decreased citric acid in crayfish stored at 4 °C.

2.5. Changes of Inorganic Ions in Crayfish

Table 5 shows the contents of inorganic ions in crayfish. Among eight detected inorganic ions, the contents of PO_4^{3-} were the highest, followed by K⁺, Cl⁻, Na⁺, Ca^{2+,} and Mg²⁺, and the trace element Zn²⁺. In previous studies, the contents of inorganic ions varied according to different aquatic species. In Yangtze, the PO_4^{3-} content was the highest, followed by K⁺, Cl⁻, Na⁺, Mg²⁺, and Ca²⁺, which was similarly observed in crayfish in our study [38]. In Suminoe oyster, the contents of PO_4^{3-} , Cl⁻, Ca²⁺, Mg²⁺, Na⁺ and K⁺, showed a decreasing tendency [9]. In squids, the K⁺ content was the highest, followed by Na⁺, PO_4^{3-} , Mg²⁺, Ca²⁺, and Cl⁻ [21]. When comparing the content of Zn²⁺, Guo et al. [41] reported that these trace elements were 43.0 mg/kg in Chinese mitten crab, which was higher than those in crayfish.

Table 5. The concentrations of mineral ions in red claw crayfish with different treatments.

M 17	0 d					4 ° C	/15 d		−20 ° C/15 d			
Mineral lons	Control	200 MPa	400 MPa	600 MPa	Control	200 MPa	400 MPa	600 MP	Control	200 MPa	400 MPa	600 MP
Ca ²⁺ (g/kg)	$0.89\pm0.07d$	1.59 ± 0.36 cd	$1.88\pm0.31c$	$1.80\pm0.09c$	$3.57 \pm 0.22a$	$3.47\pm0.40a$	$2.04\pm0.27b$	$2.29\pm0.25b$	$0.90 \pm 0.09 \mathrm{d}$	$1.36\pm0.08cd$	$1.97\pm0.28c$	$1.92\pm0.16\mathrm{c}$
Na ⁺ (g/kg)	$1.35\pm0.06b$	$1.87 \pm 0.09 ab$	$1.76 \pm 0.26 ab$	$1.58 \pm 0.35b$	$2.12 \pm 024a$	$2.01 \pm 0.17a$	$2.07 \pm 0.15a$	$2.24 \pm 0.33a$	$1.53\pm0.27b$	$1.56 \pm 0.17b$	$1.73 \pm 0.09 ab$	$1.64 \pm 0.08 ab$
K^+ (g/kg)	$3.29 \pm 0.25a$	$2.71 \pm 0.21 ab$	$2.88 \pm 0.35 ab$	$3.07 \pm 0.27a$	$2.43 \pm 0.09 bc$	$2.47 \pm 0.14 bc$	$2.37 \pm 0.17 bc$	$2.17 \pm 0.14 bc$	$3.07 \pm 0.27a$	$2.86 \pm 0.36ab$	$2.75\pm0.09ab$	$2.78 \pm 0.15 ab$
Mg^{2+} (g/kg)	$0.31\pm0.03a$	$0.33\pm0.04a$	$0.29 \pm 0.03a$	$0.32\pm0.04a$	$0.43\pm0.05a$	$0.37 \pm 0.05a$	$0.33\pm0.02a$	$0.27\pm0.06a$	$0.32\pm0.04a$	$0.34 \pm 0.03a$	$0.32\pm0.05a$	$0.30\pm0.03a$
Zn ²⁺ (mg/kg)	$34.65\pm1.13c$	$39.93 \pm \mathbf{0.68ab}$	$39.85\pm1.75ab$	$42.01\pm0.96a$	$36.75\pm1.25bc$	$33.06\pm1.06c$	$38.15\pm0.84 bc$	$42.24 \pm 1.13a$	$32.10\pm1.40c$	$36.14 \pm 1.22 bc$	$41.92\pm0.96a$	$42.46 \pm 1.04a$
$Cl^{-}(mg/g)$	$1.63 \pm 0.05c$	$2.37 \pm 0.06 bc$	$2.12 \pm 0.15 bc$	$1.73 \pm 0.08c$	$2.42 \pm 0.28 ab$	2.53 ± 0.17 ab	$2.60 \pm 0.09 ab$	$3.03 \pm 0.22a$	$1.89\pm0.10c$	$1.99 \pm 0.25 bc$	$2.14\pm0.08 bc$	$1.90 \pm 0.05c$
$PO_4^{3-} (mg/g)$	$6.83\pm0.09a$	$5.07\pm0.25b$	$4.91\pm0.12b$	$5.34\pm0.31\text{ab}$	$0.69\pm0.07c$	$0.77\pm0.07c$	$1.11\pm0.09c$	$0.84\pm0.08c$	$6.15\pm0.13a$	$5.90\pm0.21a$	$5.27\pm0.31 ab$	$5.52\pm0.27ab$

Data are presented as mean \pm standard deviation. Different letters within the same row denote significant differences (p < 0.05).

The PO₄^{3–} contents in HHP-treated crayfish were significantly lower than those in the untreated group (p < 0.05), while the opposite results were observed in Cl⁻ of all HHP-treated groups and Ca²⁺ of 400 and 600 MPa-treated groups (p < 0.05). The Ca²⁺ and Cl⁻ contents in all crayfish groups significantly increased after storing for 15 d at 4 °C, while significant decreases in PO₄^{3–} contents were observed in all crayfish groups (p < 0.05). There were almost no significant changes in inorganic ion contents of crayfish groups stored at -20 °C, except for Zn²⁺ in 200 MPa treatment. HHP could cause changes in the biological membrane permeability of crayfish cells, resulting in the permeation of extracellular substances into tissues and leakage of some metal ions [42,43]. The changes of inorganic ions caused by HHP have been reported in many aquatic foods, such as the Suminoe oyster and squid [9,21].

In living organisms, inorganic ions play important roles in regulating cell osmotic pressure, acting as an important component of functional proteins, and so on. In food products, inorganic ions are essential auxiliary flavor components in aquatic food products. Liu et al. [14] found that the TAVs of K⁺, Na⁺, PO₄^{3–}, and Cl⁻ were >1 and were the main inorganic ions in oysters. In our study, the TAVs of PO₄^{3–}, Cl⁻, and K⁺ were more than 1 in control crayfish, while the contents of Ca²⁺ increased and reached more than its taste threshold (TAV > 1) after HHP treatment. PO₄^{3–} could enhance the intensities of umami and sour tastes and suppress bitterness [39]. K⁺ could contribute to bitter and salty tastes [14]. Cl⁻ could increase umami and sweet tastes and decrease sour taste [14]. Ca²⁺ showed a negative correlation with EUC [9]. In our study, the decrease of PO₄^{3–} after HHP treatment and 4 °C storage and increase of Cl⁻ after HHP treatment and 4 °C storage enhanced the positive taste.

3. Materials and Methods

3.1. Sampling, Packaging, and Pressure Treatment

More than 300 adult red claw crayfish (*C. quadricarinatus*) (~100 mm of total length; ~40 g of wet weight) were purchased from a commercial crayfish farm in Chengmai, Hainan Province, China, in June 2021. A total of 180 crayfish were selected and cleaned.

Before HHP treatments, all crayfish were individually packaged in double poly bags and then sealed with a vacuum. The 180 packed crayfish were randomly divided into control and three HHP-treated (including 200 MPa, 400 MPa, and 600 MPa) groups (45 crayfish for each group), respectively. HHP was performed with HPP.L1-66/5 type ultra-high-pressure equipment with a 5-L cylindrical pressure vessel (Tianjin Huatai-Senmiao Bioengineering Technology Co. Ltd., Tianjin, China). HHP-treated groups were processed for 3 min at 20 °C. The pressure rates were 300 MPa/min and released within 3 s for all HHP treated groups. Crayfish of each group were then randomly divided into three sub-groups (15 crayfish for each sub-group), in which two groups were stored at 4 °C and analyzed after 0 d and 15 d. One group was stored at -20 °C and analyzed after 15 d. For each sub-group, the meat of 5 crayfish was mixed, and stored at -80 °C for analyses.

3.2. Fatty Acid Analysis

The fatty acids of crayfish meat were extracted with chloroform–methanol (2:1, v/v) according to the method of Folch et al. [44]. In detail, the total fatty acids were extracted with chloroform–methanol (2:1, v/v), then saponified using 0.4 M potassium hydroxide (KOH) in methanol, followed by esterification with 25% boron trifluoride ether solution in methanol, and finally the extraction of fatty acid methyl esters (FAMEs) in hexane. Then, the FAMEs were analyzed by gas chromatograph (Hewlett-Packard model HP 5890, Palo Alto, CA, USA). Identification of fatty acids was made after a comparison of their retention times with standards (Supelco 37 Component FAME Mix, Bellefonte, PA, USA).

3.3. Free Amino Acid Assay

FAAs in crayfish samples were determined using HPLC (Waters 2996, Waters Corporation, Milford, MA, USA) according to the method described in Liu et al. [9]. A sample of 2.5 g was homogenized in three volumes of 10% trichloroacetic acid (TCA) and centrifuged at $10,000 \times g$ for 15 min at 4 °C. Supernatants were then analyzed for FAAs by high-performance liquid chromatography (HPLC) in a Waters 2996 (Waters Corporation, Milford, MA, USA). The identity and quantity of each amino acid were assessed by comparing the retention times and peak areas of each amino acid standard (Sigma-Aldrich, St. Louis, MO, USA).

3.4. 5'-Nucleotide Assay

The 5'-nucleotides in crayfish were extracted and analyzed according to the method described by Liu et al. [9]. HPLC conditions for 5'-nucleotide analyses were as follows: injection volumes: 20 μ L, mobile phase: methanol and 0.05% phosphoric acid; flow rate: 1.0 mL/min; column temperature: 30 °C; detector wavelength: 260 nm.

3.5. Equivalent Umami Concentration (EUC)

The EUC [g monosodium glutamate (MSG) per 100 g tissue weight] is the concentration of MSG equivalent to the umami intensity given by the mixture of MSG-like amino acids and the 5'-nucleotides, and is represented by the following equation:

$$Y = \sum a_i b_i + 1218(\sum a_i b_i) (\sum a_j b_j)$$

where *Y* equals g MSG per 100 g; a_i is the concentration (g/100 g) of each umami amino acid (Asp or Glu); a_j is the concentration (g/100 g) of each umami 5'-nucleotide (IMP, GMP and AMP); b_i is the relative umami concentration for each umami MSG to MSG (Glu, 1 and Asp, 0.077); b_j is the relative umami concentration for each umami 5'-nucleotide to

IMP (IMP, 1; GMP, 2.3 and AMP, 0.18); 1218 is a synergistic constant based on the relative umami concentration of g/100 g used.

3.6. Organic Acid and Betaine Assay

Crayfish meat (2 g) was homogenized in 10 mL of purified water for 5 min. Then, the sample was centrifuged at $10,000 \times g$ for 20 min. Malic acid, lactic acid, citric acid, and succinic acid were analyzed according to the previously described method [9]. The HPLC conditions were the same as mentioned in Section 2.4 except for the detector wavelength (215 nm).

Betaine in crayfish was extracted and analyzed with the method described by Liu et al. [9]. HPLC conditions were as follows: injection volumes: 20 μ L; mobile phase: 83% aqueous acetonitrile; flow rate: 0.7 mL/min; column temperature: 20 °C; detector wavelength: 260 nm.

3.7. Inorganic Ion Assay

The concentrations of mineral composition (Ca²⁺, Na⁺, K⁺, Mg²⁺, and Zn²⁺) in crayfish were measured using flame atomic absorption spectrophotometry (AA-6800, Shimadzu Corporation, Tokyo, Japan) following a previously described method [45]. The concentrations of PO_4^{3-} and Cl^- were analyzed using an 882-ion chromatograph system according to the method described by Liu et al. [9].

3.8. Statistical Analysis

Data were presented as mean \pm standard deviation (SD). Data of fatty acid profiles and flavor contents of all treatments were analyzed by two-way analysis of variance (ANOVA), and the means were subsequently separated by Tukey's test. Prior to ANOVA, homogeneity of variances was tested using Levene's test. Statistical treatment of the data was performed using the Data Processing System (DPS) statistical software.

4. Conclusions

This study studied the changes in fatty acids and flavor compounds of red claw crayfish treated with different HHPs and storage temperatures. The level of PUFA in crayfish decreased with the increase of pressure at 0 d, and cold storage (4 °C and 20 °C) for 15 d almost did not cause significant changes in fatty acid profiles. HHP significantly increased the FAA contents in crayfish. There were almost no significant differences in the FAA contents of crayfish stored at 4 $^\circ$ C and -20 $^\circ$ C for 15 d compared to the crayfish treated with the same HHP levels. GMP, IMP, and AMP were the main 5'-nucleotide in crayfish, and their contents decreased after HHP treatment, except for IMP in 200 MPa treatment. The contents of citric and lactic acids decreased after HHP pressure compared to control crayfish, while there was a slight decrease in succinic acid. After storing at 4 °C and -20 °C for 15 d, the citric acid contents in all groups significantly decreased, and the lactic acid contents in control and 200 MPa treatments significantly increased (p < 0.05). The PO₄³⁻ contents in HHP-treated crayfish were significantly lower than those in the untreated group (p < 0.05), while opposite results were observed in Cl⁻ of all HHP-treated groups and Ca²⁺ of 400 and 600 MPa-treated groups (p < 0.05). The Ca²⁺ and Cl⁻ contents in all crayfish groups significantly increased after storage at 4 $^{\circ}$ C (p < 0.05), and significant decreases in PO_4^{3-} contents were observed in all crayfish groups (p < 0.05). However, almost no significant changes in inorganic ion contents were found when stored at -20 °C. In conclusion, HHP at 200 MPa or 400 MPa provides minimal changes considering the contents of fatty acids and non-volatile taste active compounds in crayfish, and storage at -20 °C is more protective compared to at 4 °C.

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References

- Cai, J.; Zhou, X.; Yan, X.; Lucente, D.; Lagana, C. Top 10 Species Groups in Global Aquaculture 2017; FAO Fisheries and Aquaculture Department: Rome, Italy, 2019; CA5224EN/1/06.19.
- Jones, C.M. Production of juvenile redclaw crayfish, *Cherax quadricarinatus* (von Martens) (Decapoda, Parastacidae) I. Development of hatchery and nursery procedures. *Aquaculture* 1995, 138, 221–238. [CrossRef]
- 3. Jones, C.M.; Ruscoe, I.M. Assessment of stocking size and density in the production of redclaw crayfish, *Cherax quadricarinatus* (von Martens) (Decapoda: Parastacidae), cultured under earthen pond conditions. *Aquaculture* **2000**, *189*, 63–71. [CrossRef]
- 4. Nunes, A.L.; Zengeya, T.A.; Hoffman, A.C.; Measey, G.J.; Weyl, O.L.F. Distribution and establishment of the alien Australian redclaw crayfish, *Cherax quadricarinatus*, in South Africa and Swaziland. *PeerJ* **2017**, *5*, e3135. [CrossRef] [PubMed]
- 5. Liu, S.; Qi, C.; Jia, Y.; Gu, Z.; Li, E. Growth and intestinal health of the red claw crayfish, *Cherax quadricarinatus*, reared under different salinities. *Aquaculture* 2020, 524, 735256. [CrossRef]
- 6. Yin, Y.; Yan, W.; Zheng, Y.; Peng, G. Research progress and industrial development prospect in Jiangsu province for *Cherax quadricarinutus*. *J. Aquac.* **2018**, *39*, 9–13, (In Chinese with English Abstract).
- 7. Yuan, J.; Guo, J.; Wang, H.; Guo, A.; Lian, Q.; Gu, Z. Acute toxicity of cypermethrin on the juvenile of red claw crayfish *Cherax quadricarinatus*. *Chemosphere* **2019**, 237, 124468. [CrossRef]
- Du, S.; Zhang, Z.; Xiao, L.; Lou, Y.; Pan, Y.; Zhao, Y. Acidic electrolyzed water as a novel transmitting medium for high hydrostatic pressure reduction of bacterial loads on shelled fresh shrimp. *Front. Microbiol.* 2016, 7, 305. [CrossRef]
- Liu, C.; Gu, Z.; Lin, X.; Wang, Y.; Wang, A.; Sun, Y.; Shi, Y. Effects of high hydrostatic pressure (HHP) and storage temperature on bacterial counts, color change, fatty acids and non-volatile taste active compounds of oysters (*Crassostrea ariakensis*). *Food Chem.* 2022, 372, 1311247. [CrossRef]
- 10. Nirmal, N.P.; Benjakul, S. Effect of catechin and ferulic acid on melanosis and quality of Pacific white shrimp subjected to prior freeze–thawing during refrigerated storage. *Food Control* **2010**, *21*, 1263–1271. [CrossRef]
- Aganovic, K.; Hertel, C.; Vogel, R.F.; Johne, R.; Schlüter, O.; Schwarzenbolz, U.; Jäger, H.; Holzhauser, T.; Bergmair, J.; Roth, A.; et al. Aspects of high hydrostatic pressure food processing: Perspectives on technology and food safety. *Compr. Rev. Food Sci.* 2021, 20, 3225–3266. [CrossRef]
- Rathod, N.B.; Kulawik, P.; Ozogul, Y.; Ozogul, F.; Bekhit, A.E.D.A. Recent developments in non-thermal processing for seafood and seafood products: Cold plasma, pulsed electric field and high hydrostatic pressure. *Int. J. Food Sci. Technol.* 2022, 57, 774–790. [CrossRef]
- Lorido, L.; Estévez, M.; Ventanas, J.; Ventanas, S. Comparative study between Serrano and Iberian dry–cured hams in relation to the application of high hydrostatic pressure and temporal sensory perceptions. *LWT Food Sci. Technol.* 2015, 64, 1234–1242. [CrossRef]
- Liu, C.; Ji, W.; Jiang, H.; Shi, Y.; He, L.; Gu, Z.; Zhu, S. Comparison of biochemical composition and non–volatile taste active compounds in raw, high hydrostatic pressure-treated and steamed oysters *Crassostrea hongkongensis*. *Food Chem.* 2021, 344, 128632. [CrossRef]
- Xuan, X.; Cui, Y.; Lin, X.; Yu, J.; Liao, X.; Ling, J.; Shang, H. Impact of high hydrostatic pressure on the shelling efficacy, physicochemical properties, and microstructure of fresh razor clam (*Sinonovacula constricta*). J. Food Sci. 2018, 83, 284–293. [CrossRef]
- Yi, J.; Zhang, L.; Ding, G.; Hu, X.; Liao, X.; Zhang, Y. High hydrostatic pressure and thermal treatments for ready-to-eat winemarinated shrimp: An evaluation of microbiological and physicochemical qualities. *Innov. Food Sci. Emerg.* 2013, 20, 16–23. [CrossRef]
- 17. Campus, M. High pressure processing of meat, meat products and seafood. Food Eng. Rev. 2010, 2, 256–273. [CrossRef]

- Ginson, J.; Kamalakanth, C.; Bindu, J.; Venkateswarlu, R.; Das, S.; Chauhan, O.; Gopal, T.K.S. Changes in K value, microbiological and sensory acceptability of high pressure processed Indian white prawn (*Fenneropenaeus indicus*). *Food Bioprocess Technol.* 2013, 6, 1175–1180. [CrossRef]
- Martínez, M.A.; Velazquez, G.; Cando, D.; Núñez-Flores, R.; Borderías, A.J.; Moreno, H.M. Effects of high pressure processing on protein fractions of blue crab (*Callinectes sapidus*) meat. *Innov. Food Sci. Emerg.* 2017, 41, 323–329. [CrossRef]
- 20. Yi, J.; Xu, Q.; Hu, X.; Dong, P.; Liao, X.; Zhang, Y. Shucking of bay scallop (*Argopecten irradians*) using high hydrostatic pressure and its effect on microbiological and physical quality of adductor muscle. *Innov. Food Sci. Emerg.* **2013**, *18*, 57–64. [CrossRef]
- Yue, J.; Zhang, Y.; Jin, Y.; Deng, Y.; Zhao, Y. Impact of high hydrostatic pressure on non-volatile and volatile compounds of squid muscles. *Food Chem.* 2016, 194, 12–19. [CrossRef]
- 22. Yagiz, Y.; Kristinsson, H.G.; Balaban, M.O.; Welt, B.A.; Ralat, M.; Marshall, M.R. Effect of high pressure processing and cooking treatment on the quality of Atlantic salmon. *Food Chem.* **2009**, *116*, 828–835. [CrossRef]
- Méndez-Martínez, Y.; Ceseña, C.E.; Luna-González, A.; García-Guerrero, M.U.; Martinez-Porchas, M.; Campa-Cordova, Á.I.; Cortés-Jacinto, E. Effects of different dietary protein energy ratios on growth, carcass amino acid and fatty acid profile of male and female *Cherax quadricarinatus* (von Martens, 1868) pre-adults. *Aquacult. Nutr.* 2021, 27, 2481–2496. [CrossRef]
- Lin, X.; Liu, C.; Cai, L.; Yang, J.; Zhou, J.; Jiang, H.; Shi, Y.; Gu, Z. Effect of high hydrostatic pressure processing on biochemical characteristics, bacterial counts, and color of the red claw crayfish *Cherax quadricarinatus*. J. Shell. Res. 2021, 40, 177–184. [CrossRef]
- 25. Li, J.; Guo, Z.; Gan, X.; Wang, Q.; Zhao, Y. Biochemical changes during vitellogenesis in the red claw crayfish, *Cherax quadricarinatus* (von Martens). *Aquac. Res.* **2010**, *41*, 446–455. [CrossRef]
- Li, J.; Guo, Z.; Gan, X.; Wang, D.; Zhang, M.; Zhao, Y. Effect of different dietary lipid sources on growth and gonad maturation of pre–adult female *Cherax quadricarinatus* (von Martens). *Aquacult. Nutr.* 2011, 17, e853–e860. [CrossRef]
- MacDonald, G.A.; Hall, B.I.; Vlieg, P.; Fox, S.; Osborne, T. Seasonal Changes in Proximate Composition and Glycogen Content of Greenshell Mussels; Mussel Industry Council Seminar, Crop & Food Research Ltd.: Nelson, New Zealand, 2000.
- Šimat, V.; Bogdanović, T.; Poljak, V.; Petričević, S. Changes in fatty acid composition, atherogenic and thrombogenic health lipid indices and lipid stability of bogue (*Boops boops* Linnaeus, 1758) during storage on ice: Effect of fish farming activities. J. Food Compos. Anal. 2015, 40, 120–125. [CrossRef]
- Odeyemi, O.A.; Burke, C.M.; Bolch, C.C.J.; Stanley, R. Seafood spoilage microbiota and associated volatile organic compounds at different storage temperatures and packaging conditions. *Int. J. Food Mocrobiol.* 2018, 280, 87–99. [CrossRef]
- 30. Liu, C.; Meng, F.; Tang, X.; Shi, Y.; Wang, A.; Gu, Z.; Pan, Z. Comparison of nonvolatile taste active compounds of wild and cultured mud crab *Scylla paramamosain*. *Fish. Sci.* **2018**, *84*, 897–907. [CrossRef]
- 31. Chen, L.; Zeng, W.; Rong, Y.; Lou, B. Characterisation of taste active compositions, umami attributes and aroma compounds in Chinese shrimp. *Int. J. Food Sci. Technol.* **2021**, *56*, 6311–6321. [CrossRef]
- Shigematsu, T.; Hayashi, M.; Nakajima, K.; Uno, Y.; Sakano, A.; Murakami, M.; Narahara, Y.; Ueno, S.; Fujii, T. Effects of high hydrostatic pressure on distribution dynamics of free amino acids in water soaked brown rice grain. *J. Phys. Confer. Ser.* 2010, 215, 012171. [CrossRef]
- Zhang, Y.; Wang, Q.; Bi, Y.; Cheng, K.; Chen, F. Nutritional and functional activities of protein from steamed, baked, and high hydrostatic pressure treated cod (*Gadus morhua*). Food Control 2019, 96, 9–15. [CrossRef]
- Ahmed, J.; Habeebullah, S.F.K.; Thomas, L.; Mulla, M.Z.; Jacob, H.; Alagarsamy, S. Effect of high-pressure treatment and refrigerated storage on the amino acid profile, color, and texture of hammour (*Epinephelus coioides*) fillets. *J. Food Process. Pres.* 2021, 45, 15977. [CrossRef]
- Kong, L.; Cai, C.; Ye, Y.; Chen, D.; Wu, P.; Li, E.; Chen, L.; Song, L. Comparison of non-volatile compounds and sensory characteristics of Chinese mitten crabs (*Eriocheir sinensis*) reared in lakes and ponds: Potential environmental factors. *Aquaculture* 2012, 364–365, 96–102. [CrossRef]
- 36. Chen, D.; Zhang, M. Non-volatile taste active compounds in the meat of Chinese mitten crab (*Eriocheir sinensis*). *Food Chem.* 2007, 104, 1200–1205. [CrossRef]
- Zheng, J.; Tao, N.; Gong, J.; Gu, S.; Xu, C. Comparison of non-volatile taste-active compounds between the cooked meats of preand post-spawning Yangtze *Coilia ectenes*. *Fish. Sci.* 2015, *81*, 559–568. [CrossRef]
- 38. Sarower, M.G.; Abul, F.M.H.; Biswas, B.; Abe, H. Taste producing components in fish and fisheries products: A review. *Int. J. Food Ferm. Technol.* **2012**, *2*, 113–121.
- Liu, Y.; Zhang, C.; Chen, S. Comparison of Active Non-volatile Taste Components in the Viscera and Adductor Muscles of Oyster (Ostrea rivularis Gould). Food Sci. Technol. Res. 2013, 19, 417–424. [CrossRef]
- Lopez–Garcia, R. Citric acid. In Kirk–Othmer Encyclopedia of Chemical Technology; John Wiley & Sons: Hoboken, NJ, USA, 2000; pp. 1–25.
- 41. Guo, Y.; Gu, S.; Wang, X.; Zhuang, K.; Wang, S.; Shi, J. Nutrients and non-volatile taste compounds in Chinese mitten crab by–products. *Fish. Sci.* **2015**, *81*, 193–203. [CrossRef]
- Rastogi, N.K.; Niranjan, K. Enhanced mass transfer during osmotic dehydration of high pressure treated pineapple. *J. Food Sci.* 1998, 63, 508–511. [CrossRef]
- San Martín, M.F.B.A.; Barbosa-Cánovas, G.V.; Swanson, B.G. Food processing by high hydrostatic pressure. *Crit. Rev. Food Sci.* 2002, 42, 627–645. [CrossRef]

45. Bai, J.; Fan, Y.; Zhu, L.; Wang, Y.; Hou, H. Characteristic flavor of Antarctic krill (*Euphausia superba*) and white shrimp (*Penaeus vannamei*) induced by thermal treatment. *Food Chem.* **2022**, *378*, 132074. [CrossRef]