

Cooperative Catalysis

Cooperative Catalysis for the Highly Diastereo- and Enantioselective [4+3]-Cycloannulation of *ortho*-Quinone Methides and Carbonyl Ylides

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Dedicated to Professor Mark Lautens on the occasion of his 60th birthday

Abstract: We describe herein a highly diastereo- and enantioselective [4+3]-cycloannulation of ortho-quinone methides and carbonyl ylides to furnish functionalized oxa-bridged dibenzooxacines with excellent yields and stereoselectivity in a single synthetic step. The combination of rhodium and chiral phosphoric acid catalysis working in concert to generate both transient intermediates in situ provides direct access to complex bicyclic products with two quaternary and one tertiary stereogenic centers. The products may be further functionalized into valuable and enantiomerically highly enriched building blocks.

Oxa-bridged heterocyclic skeletons are ubiquitously present in numerous natural products and bioactive molecules.^[1] Therefore, the development of new, efficient, and stereoselective synthetic methods towards their rapid construction is highly desirable. Extensive efforts have previously been directed at carbo-bridged bicyclic frameworks through Lewis-acid-catalyzed intramolecular Diels–Alder (IMDA) reactions, molecular rearrangements involving a ring-opening/closure tandem process, free-radical reactions, and transition-metal-catalyzed annulation reactions.^[2] However, there still remains a high demand for the synthesis of oxa-bridged heterocycles.^[3]

The combination of a transition-metal catalyst and a Lewis acid or organocatalyst to activate two different substrates for a given reaction has attracted significant interest among synthetic organic chemists recently since it potentially enables highly efficient and/or unprecedented complex chemical transformations in a one-pot operation.^[4] The success of this strategy relies upon the simultaneous activation of two reacting partners by two different catalysts that operate in concert in two distinct catalytic cycles.^[4c] A prominent early example is the work of Hu, Gong, and co-

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https://doi.org/10.1002/anie.201913603.

workers on cooperative Rh-/chiral phosphoric acid catalyzed multicomponent reactions of α -diazoesters, alcohols, and imines, which were converted into α -hydroxy β -amino esters with excellent enantio- and diastereocontrol.^[5] In another example, Terada et al. developed an elegant carbonyl ylide formation/reduction sequence towards isochromanones under cooperative Rh-/chiral phosphoric acid catalysis.^[6]

Carbonyl ylides generated from carbonyl compounds and a rhodium carbone complex are classically considered as highly reactive transient species and are widely employed in 1,3-dipolar cycloaddition reactions with a wide variety of 2π systems.^[7,8] However, their reactivity with 4π -systems is still underexplored due to the challenges associated with entropy factors and strain aspects in the formation of seven-membered rings.^[9,10]

Ortho-quinone methides (o-QMs) feature a particularly reactive 4π -system and have increasingly been exploited as versatile synthetic intermediates for the construction of complex heterocycles.^[11] In recent years, we and others have meticulously developed Brønsted acid catalyzed reactions of o-QMs with a wide range of typically 2π -nucleophiles, leading to a broad range of benzannulated oxygen heterocycles with good to excellent stereocontrol.^[12,13]

We now report the first cooperative, catalytic, enantioselective [4+3]-cycloannulation of *o*-QMs and carbonyl ylides to afford complex and enantiomerically highly enriched oxabicyclic dibenzooxacines. We envisioned that a chiral phosphoric acid would easily form hydrogen-bonded *o*-QM **A** starting from *ortho*-hydroxy benzylalcohol **1** in one catalytic cycle, while in a second and separate catalytic cycle, carbonyl ylide **B** would be generated through Rh-catalyzed decomposition of a α -diazoester **2** tethered to an aryl ketone (Scheme 1). The decisive question here was whether both transient intermediates **A** and **B** formed in only very low amounts would have sufficient stability and lifetime to undergo the desired [4+3]-cycloannulation and provide the product **3** with good stereocontrol in the chiral environment provided by the phosphoric acid catalyst.

Enantioselective [4+3]-cycloadditions of o-QMs were first described independently by the groups of Scheidt and Ye under chiral NHC catalysis to produce benzoxopinones.^[14a,b] Very recently, the first example of a phosphoric acid catalyzed enantioselective reaction of o-QMs with 2indolylmethanols as 1,3-dipoles toward indolylbenzoxepins was established by Shi et al.^[14c] An interesting study by the Lautens group described a purely Brønsted acid catalyzed

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Scheme 1. Design plan for the reaction between *o*-QMs and carbonyl ylides through cooperative Rh/phosphoric acid catalysis.

diastereoselective synthesis of oxa-bridged oxazocines through cycloaddition with isomünchnones. $^{\left[14c\right] }$

To test our hypothesis, we initiated our investigations with the model reaction between benzhydryl alcohol **1a** and α diazoester **2a** in the presence of 5 mol% of Rh₂(OAc)₄ and chiral phosphoric acid **PA1** (10 mol%) in CHCl₃ at room temperature. We were delighted to obtain the desired product **3a** in 77% yield after 12 h with moderate diastereo- and enantioselectivity (Table 1, entry 1). Importantly, diazoester **2a** had to be added slowly for a period of 1 h using a syringe pump to avoid side reactions of the transient carbonyl ylide.

Extensive screening of suitable chiral phosphoric acid catalysts^[15] revealed that $Rh_2(OAc)_4$ (5 mol%) and 10 mol% of (*R*)-**PA7** provided the best combination, which afforded **3a** in 79% yield with 20:1-diastereoselectivity and with 92:8 e.r. (entry 7). A short study of reaction conditions revealed CHCl₃ to be the solvent of choice and that both chemical yield and enantioselectivity were further improved by using 3 Å molecular sieves (MS) as dehydrating agent. When using these conditions, **3a** was eventually obtained in 96% yield with 20:1 diastereoselectivity and with 96:4 e.r. (entry 12). Interestingly, lowering the catalyst loading of $Rh_2(OAc)_4$ and (*R*)-**PA7** did not decrease the enantiomeric ratio, but led to a decrease in the diastereoselectivity of the product (see the Supporting Information for more details).

With optimized conditions in hand, we set out to examine the substrate scope of the reaction. Initially, a series of α diazoesters **2a-k** was tested with benzhydryl alcohol **1a** as a model *ortho*-quinone methide precursor. Pleasingly, the reaction worked well with all substrates and afforded products **3a-k** in good to excellent yields and with excellent enantioselectivity of up to 97:3 e.r. (Scheme 2). The diastereoselectivity appeared to be dependent on the electronic character of the aryl substituent, with electron-rich aryl groups generally giving rise to almost perfect selectivity. In particular, the thiophene-substituted diazoester **2k** gave rise to product **3k** in 92 % yield as a single diastereomer and with 95:5 e.r. On the other hand, substrates **2g-j**, which carry electron-poor substituents (such as halogen and CF₃ groups) furnished products **3g-j** with diminished diastereoselectivity, Table 1: Catalyst screening and optimization studies.^[a]



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Entry	PA	Solvent	Time [h]	3 a (%) ^[0,0]	er	dr
1	1	CHCl3	12	77	66:34	16:1
2	2	CHCl₃	12	71	82:18	20:1
3	3	CHCl₃	12	75	69:31	10:1
4	4	CHCl₃	12	79	83:17	20:1
5	5	CHCl₃	12	80	84:16	20:1
6	6	CHCl₃	12	73	88:12	20:1
7	7	CHCl₃	12	79	92:8	20:1
8	7	CH_2Cl_2	12	83	83:17	20:1
9	7	1,2-DCE	12	75	83:17	15:1
10	7	PhMe	48	58	85:15	8:1
11	7	CPME	48	trace	ND	ND
12 ^[f]	7	CHCl₃	12	96	96:4	20:1

[a] Reactions were carried out with 0.10 mmol of 1a, 0.11 mmol of 2a and Rh₂(OAc)₄ (5 mol%) in the presence of catalyst PA (10 mol%) in CHCl₃ (3 mL). [b] Yield of isolated product of both diastereomers after chromatographic purification. [c] Decomposition accounts for remainder of mass balance. [d] Enantiomeric ratios (er) were determined by chiral HPLC. [e] Diastereomeric ratios (dr) were determined from ¹H NMR of crude reaction mixture. [f] With 3 Å MS (35 mg).

albeit in excellent yields and up to 96:4 e.r. *Ortho*-substituted aryl groups had a detrimental effect on both the diastereoand enantioselectivity, as shown for 3e (6:1 d.r., 89:11 e.r.), probably for steric reasons. Most importantly, this cycloannulation is not limited to aryl- and heteroaryl-substituted diazoketones but could be extended to an alkyl-substituted substrate as well, since dibenzooxacine 3l was obtained in high yield and stereoselectivity similar to the other products.

We then turned our attention to reactions of α -diazoester **2a** with various substituted *o*-hydroxy benzhydryl alcohols **1** as *ortho*-quinone methide precursors (Scheme 3). Gratifyingly, a broad variety of substrates with both electrondonating and electron-withdrawing substituents in the *o*-QM component were readily converted into products **4a–n** at slightly elevated temperature. Yields ranged from moderate to excellent and the diastereo- and enantioselectivity were generally very high. Here again, a dependence of reaction outcome on the electron-rich benzhydryl alcohols furnished products with very good yields (e.g., **4a–4f**), electron-poor substrates afforded products with only moderate chemical yield, albeit excellent enantioselectivity (e.g., **4g** and **4h**).

Structural variation of the quinone moiety was more readily tolerated irrespective of the electronic character, and products **4j**–**4n** with alkyl and various halogen substituents were obtained with synthetically useful yields and very good diastereo- and enantioselectivity (Scheme 3). Unfortunately,

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Scheme 2. Substrate scope for reactions of *ortho*-hydroxy benzhydryl alcohol **1 a** with various α -diazoesters (**2 a**–l). Reactions were carried out with 0.1 mmol of **1 a**, 0.11 mmol of **2**, 3 Å MS (35 mg) and Rh₂(OAc)₄ (5 mol%) in the presence of catalyst **PA7** (10 mol%) in CHCl₃ (3 mL).

the *i*Pr-substituted benzhydryl alcohol **10** failed to deliver product **40** because the transient *o*-QM generated in situ from **10** was apparently too unstable to successfully engage the transient carbonyl ylide in the cycloannulation event. The X-ray structure analysis of the major diastereomer of product **3k** confirmed both the relative and absolute configuration, which was assigned to all other products accordingly (Figure 1).^[16,17]

To gain more insight into the mechanism of this cycloannulation process, some control experiments were conducted. Under the standard conditions, O-methyl-protected benzhydryl alcohol **1p** failed to react with **2a**, thereby underlining the importance of the *o*-QM structure for this reaction [Scheme 4, Eq. (1)]. Furthermore, neither in the presence of the phosphoric acid alone with $Rh_2(OAc)_4$ absent (case A) nor in the presence of $Rh_2(OAc)_4$ alone with the phosphoric acid absent (case B) was a successful reaction observed [Eq. (2)]. We therefore conclude that both catalysts actively participate in this reaction by generating both the *o*-QM and the carbonyl ylide as transient intermediates. These control experiments strongly support our initial reaction



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Scheme 3. Expansion of substrate scope for the reaction of *ortho*hydroxy benzhydryl alcohols **1** with α -diazoester **2a**. Reactions were carried out with 0.1 mmol of **1**, 0.11 mmol of **2a**, 3 Å MS (35 mg) and Rh₂(OAc)₄ (5 mol%) in the presence of catalyst **PA7** (10 mol%) in CHCl₃ (3 mL).



Figure 1. X-ray crystal structure of product 3 k.[16]

design of a cooperative catalytic activation of both nucleophile and electrophile in a one-pot process.

GDCh





Scheme 4. Control experiments.

In order to further shine light on the origin of the enantioselectivity, we conducted reactions of **1a** and **2a** in the presence of a chiral rhodium catalyst and both an achiral and a chiral phosphoric acid [Scheme 4, Eq. (3)]. Whereas the enantioselectivity of the latter reaction was virtually unchanged in comparison to the reaction with $Rh_2(OAc)_4$ reported above, no enantioselectivity was observed with diphenyl phosphate as a Brønsted acid catalyst. Moreover, reaction of the stable *ortho*-quinone methide **1q** and **2a** in the presence of the chiral rhodium catalyst alone delivered dibenzooxacine **4q** in low yield and as a racemic mixture, indicating once again the critical role of the chiral phosphoric acid for the enantioselectivity of the process [Eq. (4)].

Finally, we attempted some structural modifications of the products and identified the acetal moiety of 3a as a good starting point for further synthetic elaborations. Under BF₃ activation, the acetal was readily cleaved to the corresponding oxonium ion, which was trapped with allyltributylstannane to furnish isobenzofuran 5 with good yield and complete diastereocontrol. Phenol 5 was then lactonized with *p*-TsOH to produce the highly congested spirocyclic dihydrocoumarin 6, again with good yield (Scheme 5). Product 6 was obtained in 72 % yield over two steps as a single diastereomer and with 98:2 e.r. On the other hand, the oxa-bridged products 3 are sufficiently stable as to easily tolerate further post-modifications such as a Suzuki–Miyaura cross coupling reaction, which proceeded in very good yield in the case of 3i.

In conclusion, we have developed a novel and highly stereoselective [4+3]-cycloannulation of transient carbonyl ylides with in situ generated o-QMs through cooperative Rh/ phosphoric acid catalysis. The reaction enables the catalytic enantio- and diastereoselective synthesis of oxa-bridged heterocycles featuring two quaternary and one tertiary stereogenic centers in a one-pot operation. The benzannulated O-heterocycles were obtained in typically high yields (up to 96%) and excellent stereoselectivity (up to > 20:1 d.r.



Scheme 5. Synthetic elaborations of oxa-bridged dibenzooxacines 3.

and up to 97:3 e.r.). Moreover, the products may be successfully manipulated to access valuable synthetic building blocks. The striking feature of this process is the separate catalytic activation of nucleophile and electrophile, with a chiral phosphoric acid enabling the formation of a transient hydrogen-bonded *ortho*-quinone methide and $Rh_2(OAc)_4$ delivering the transient carbonyl ylide in a one-pot operation. Further extensions of this process are currently underway in our laboratory.

Acknowledgements

This work was generously supported by the Deutsche Forschungsgemeinschaft (SCHN 441/11-2) and through gifts of chemicals from BASF and Evonik. We thank Dr. M. Schlegel for helpful discussions and Dr. P. Lönnecke (both University of Leipzig) for solving the X-ray crystal structure.

Conflict of interest

The authors declare no conflict of interest.

Keywords: asymmetric synthesis · carbonyl ylides · cooperative catalysis · cycloannulation · *ortho*-quinone methides

How to cite: Angew. Chem. Int. Ed. 2020, 59, 5536–5540 Angew. Chem. 2020, 132, 5580–5585

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- [16] CCDC 1946746 (3k) contains the supplementary crystallographic data for this paper. These data are provided free of charge by The Cambridge Crystallographic Data Centre.
- [17] We assume a similar transition-state assembly in this reaction as in previous examples of phosphoric acid catalyzed reactions of *ortho*-quinone methides (see Ref. [12j] for example) based upon the identical absolute configuration of the products.

Manuscript received: October 24, 2019 Revised manuscript received: December 12, 2019 Accepted manuscript online: January 2, 2020 Version of record online: January 23, 2020