

# Na/H Exchange in Cultured Chick Heart Cells

## *pH<sub>i</sub> Regulation*

DAVID PIWNICA-WORMS, RON JACOB, C. RUSSELL HORRES,  
and MELVYN LIEBERMAN

From the Department of Physiology, Duke University Medical Center, Durham, North Carolina 27710

**ABSTRACT** The purpose of this study was to establish the existence of Na/H exchange in cardiac muscle and to evaluate the contribution of Na/H exchange to pH<sub>i</sub> regulation. The kinetics of pH<sub>i</sub> changes in cultured chick heart cells were monitored microfluorometrically with 6-carboxyfluorescein and correlated with Na<sub>i</sub> content changes analyzed by atomic absorption spectrophotometry; transmembrane H<sup>+</sup> movements were evaluated under pH stat conditions. After induction of an intracellular acid load by pretreatment with NH<sub>4</sub>Cl, a regulatory cytoplasmic alkalization occurred with a *t*<sub>1/2</sub> of 2.9 min. pH<sub>i</sub> regulation required external Na<sup>+</sup> and was concomitant with transmembrane H<sup>+</sup> extrusion as well as a rapid rise in Na<sub>i</sub> content in an Na/H ratio of 1:1. Microelectrode recordings of membrane potential demonstrated directly the electroneutral character of pH<sub>i</sub> regulation. Acid-induced net Na<sup>+</sup> uptake could be either stimulated by further decreasing pH<sub>i</sub> or inhibited by decreasing pH<sub>o</sub>; Na<sup>+</sup> uptake was unaffected by tetrodotoxin (10 μg/ml), quinidine (10<sup>-3</sup> M), DIDS (10<sup>-4</sup> M), Cl<sub>o</sub>-free solution, or HCO<sub>3</sub>-free solution. Amiloride (10<sup>-3</sup> M) maximally inhibited both pH<sub>i</sub> regulation and Na<sup>+</sup> uptake; the ID<sub>50</sub> for amiloride inhibition of Na<sup>+</sup> uptake was 3 μM. Na<sub>o</sub>-dependent H<sup>+</sup> extrusion showed half-maximal activation at 15 mM Na<sub>o</sub>; Li<sup>+</sup>, but not K<sup>+</sup> or choline<sup>+</sup>, could substitute for Na<sup>+</sup> to support H<sup>+</sup> extrusion. Ca<sub>o</sub>-free solution also stimulated acid-induced Na<sup>+</sup> uptake. We conclude that pH<sub>i</sub> regulation following an acid load in cardiac muscle cells is by an amiloride-sensitive, electroneutral Na/H exchange. Stimulation of Na/H exchange up to 54 pmol/cm<sup>2</sup>·s indicates the rapidity of this exchange across cardiac cell membranes. Na/H exchange may also participate in steady state maintenance of pH<sub>i</sub>.

### INTRODUCTION

The intracellular pH (pH<sub>i</sub>) of cardiac muscle in steady state is ~7.1 (Ellis and Thomas, 1976; Roos and Boron, 1981), 1 pH unit more alkaline than the pH<sub>o</sub> expected for electrochemical equilibrium of H<sup>+</sup>. This gradient implies the existence of an H<sup>+</sup> extrusion system; however, mechanisms of pH<sub>i</sub> maintenance in heart cells in steady state have not yet been resolved. Recent evidence in non-

Address reprint requests to Dr. M. Lieberman, Dept. of Physiology, Box 3709, Duke University Medical Center, Durham, NC 27710. Dr. Piwnica-Worms' present address is Dept. of Radiology, Brigham and Women's Hospital, Boston, MA 02115.

steady state indicates that  $\text{Cl}/\text{HCO}_3$  exchange may regulate  $\text{pH}_i$  following cytosolic alkalization (Vaughan-Jones, 1982), but does not appear to participate significantly in  $\text{pH}_i$  regulation following cytosolic acidification (Gonzalez and Clancy, 1981; Vaughan-Jones, 1982; Piwnica-Worms, 1983; Vanheel et al., 1984). We recently reported changes in  $\text{pH}_i$  induced by lowering external  $\text{Na}^+$  consistent with an  $\text{Na}/\text{H}$  exchange mechanism in cultured heart cells (Piwnica-Worms and Lieberman, 1983); evidence suggestive of  $\text{Na}/\text{H}$  exchange has also been reported for the sheep cardiac Purkinje fiber (Deitmer and Ellis, 1980; Ellis and MacLeod, 1983). This mechanism participates in  $\text{pH}_i$  regulation following an acid load in several different tissues: mouse soleus muscle (Aickin and Thomas, 1977), crayfish neuron (Moody, 1981), salamander renal proximal tubule (Boron and Boulpaep, 1983), and rat lymphocytes (Grinstein et al., 1984). Thus, the purpose of this study was to firmly establish the existence of  $\text{Na}/\text{H}$  exchange in the cell membrane of a cardiac preparation and to evaluate the contribution of  $\text{Na}/\text{H}$  exchange to  $\text{pH}_i$  regulation. Monitoring  $\text{pH}_i$  with the fluorescent dye 6-carboxyfluorescein in combination with measurements of transmembrane  $\text{H}^+$  movements,  $\text{Na}_i$  content, and membrane potential demonstrates that electroneutral  $\text{Na}/\text{H}$  exchange regulates  $\text{pH}_i$  following an acid load in heart cells and may participate in steady state  $\text{pH}_i$  maintenance.

## METHODS

### *Tissue Culture*

The techniques for producing polystrands composed chiefly of muscle (contractile) or fibroblasts (noncontractile) from 10–11-d-old chick hearts disaggregated with trypsin have been previously described (Horres et al., 1977; Piwnica-Worms et al., 1983). The contractile preparation consists of 70 strands of muscle cells formed as a sheath around a nylon core, with each strand up to 5 mm in length and an outside diameter typically  $<100 \mu\text{m}$  (tissue dry weight  $\sim 0.25 \text{ mg}$ ).

### *Experimental Solutions*

The compositions of the modified Earle's solutions are given in Table I.  $\text{NH}_4$  solution was made by equimolar substitution of 20 mM  $\text{NH}_4\text{Cl}$  for  $\text{NaCl}$ ; low  $\text{Na}_o$  solution was produced by equimolar substitution of tetramethylammonium-Cl (TMA-Cl) for  $\text{NaCl}$ ;  $\text{Cl}_o$ -free solution was produced by equimolar substitution of methanesulfonate for Cl (see Kenyon and Gibbons, 1977). Bicarbonate-containing solutions were maintained at pH 7.4 by gassing with 96% air/4%  $\text{CO}_2$ , while solutions buffered with various ratios of HEPES/Tris were titrated in air with HCl or NaOH to the indicated pH. Ouabain (Sigma Chemical Co., St. Louis, MO), quinidine HCl (Sigma Chemical Co.), 4,4-diisothiocyanato-2,2'-disulfonic acid stilbene (DIDS; Pierce Chemical Co., Rockford, IL), and amiloride (a gift from Dr. E. Cragoe, Merck, Sharp & Dohme Research Laboratories, West Point, PA) were added directly as solid powder to perfusates. Tetrodotoxin (TTX; Sankyo Co., Tokyo, Japan) was dissolved in distilled  $\text{H}_2\text{O}$  before addition to solutions.

### *Intracellular Ion Content*

Experiments were carried out at  $37^\circ\text{C}$  in an incubator containing either 96% air/4%  $\text{CO}_2$  or 100% air at  $\sim 95\%$  humidity (Horres et al., 1977). To follow rapid  $\text{Na}_i$  content changes, preparations were rinsed continuously in 250 ml of solution for 30–180 s. To follow  $\text{Na}_i$

content changes over longer periods (minutes), preparations were rinsed for 30–45 s, followed by incubation in the same solution. Clearance of extracellular  $\text{Na}^+$  in polystrands occurs with a half-time ( $t_{1/2}$ ) of 4 s (Wheeler et al., 1982).

Each experiment was ended by rinsing polystrands in ice-cold Na-free, K-free solution (Table I) for 30 s to clear extracellular cations. Cold solution inhibits >98% of transmembrane Na efflux (Wheeler et al., 1982). Preparations were then rapidly air-dried, removed from the silver support, dried overnight in a vacuum oven at 110°C (National Appliance

TABLE I  
Solutions

Component*	1 Control	2 Low Na	3 $\text{NH}_4^+$	4 0-Cl	5 0- $\text{HCO}_3^-$	6 pH stat	7 0-Na, 0-K rinse
$\text{Na}^+$	144.8	0.8	124.8	144.8	123.8	144.0	—
TMA <sup>+</sup>	—	118.0	—	—	—	—	—
$\text{K}^+$	5.4	5.4	5.4	5.4	5.4	5.4	—
$\text{Mg}^{2+}$	0.8	0.8	0.8	0.8	0.8	0.8	0.8
$\text{Ca}^{2+}$	2.7	2.7	2.7	2.7	2.7	2.7	2.7
Choline <sup>+</sup>	—	26.0	—	—	—	—	145.0
$\text{NH}_4^+$	—	—	20.0	—	—	—	—
$\text{Cl}^-$	128.8	128.8	128.8	—	133.8	156.4	150.4
$\text{X}^-$	—	—	—	128.0	—	—	—
$\text{HCO}_3^-$	26.0	26.0	26.0	26.0	—	—	—
$\text{H}_2\text{PO}_4^-$	0.8	0.8	0.8	0.8	0.8	—	—
$\text{SO}_4^{2-}$	0.8	0.8	0.8	0.8	0.8	—	0.8
Glucose	5.6	5.6	5.6	5.6	5.6	5.6	5.6
HEPES	—	—	—	—	40–5	0.05	5.4
Tris	—	—	—	—	0–35	—	4.1
Albumin (g/liter)	1.4	1.4	1.4	1.4	1.4	—	—
pH <sub>o</sub>	7.4	7.4	7.4	7.4	5.5–8.5	7.4	7.4

\* Compositions are given in millimolar unless otherwise noted.  $\text{X}^-$  is methanesulfonate (Eastman Kodak Corp., Rochester, NY). TMA<sup>+</sup> (tetramethylammonium) was obtained from ICN Pharmaceuticals, Plainview, NY; HEPES (Ultral Brand) from Calbiochem-Behring, La Jolla, CA; and Tris, bovine serum albumin (fraction V), and choline- $\text{HCO}_3^-$  from Sigma Chemical Co., St. Louis, MO.  $\text{HCO}_3^-$ -free,  $\text{HPO}_4^-$ -free solution was made from solution 5 omitting  $\text{NaH}_2\text{PO}_4$ ; some Na-free solutions were made from solution 2 by substituting  $\text{KH}_2\text{PO}_4$  for  $\text{NaH}_2\text{PO}_4$  while adjusting KCl; Ca-free rinse was made from solution 7 by omitting  $\text{CaCl}_2$ .

Co., Portland, OR), and weighed on an electrobalance (Cahn G-2, Ventron Instruments, Paramount, CA). The contribution of monofilament without tissue was subtracted from the total weight of the preparations to obtain tissue dry weight (Piwnica-Worms et al., 1983). Individual preparations were then extracted for 2 h in 0.5 ml of 1 N  $\text{HNO}_3$  (reagent grade). For  $\text{Na}^+$  analysis, 1.5 ml of Na/K blank solution containing 0.685 g/liter CsCl and 0.125 ml/liter Acationox (Scientific Products, McGraw Park, IL) was added to the extract. (For  $\text{K}^+$  analysis, an aliquot of the resulting solution was further diluted to 1 part to 4 parts of blank solution.) The  $\text{Na}^+$  and  $\text{K}^+$  concentration of each diluted extract was determined with an atomic absorption spectrophotometer (model 460, Perkin-Elmer, Norwalk, CT) along with appropriate Na/K standards using parallel reagent and substrate

controls to correct for possible cation contamination. Intracellular ionic concentrations were calculated using steady state cell water determinations for the polystrand (4.29  $\mu\text{l}/\text{mg}$  dry weight; Horres et al., 1977).

The total cell  $\text{Ca}^{2+}$  content ( $\text{Ca}_i$ ) was also determined by atomic absorption spectrophotometry after rinsing polystrands in ice-cold Ca-free solution for 30 s. After dry weight determination, two polystrands were pooled to increase material for analysis and extracted for 2 h in 200  $\mu\text{l}$  of 1 N  $\text{HNO}_3$ . 250  $\mu\text{l}$  of Ca blank solution containing 0.75 N  $\text{HNO}_3$  and 13 mM  $\text{LaCl}_3$  was then added to each extract and the  $\text{Ca}^{2+}$  concentration was measured together with appropriate reagent blanks and calibration solutions.

Ion content determinations normalized to tissue dry weight, unlike ionic activities, are not complicated by water movement under non-steady state conditions and therefore accurately reflect net transmembrane ionic movements (McManus and Schmidt, 1978). In addition,  $^{42}\text{K}$  and  $^{24}\text{Na}$  tracer experiments do not indicate the existence of hidden pericellular compartments in the polystrand, which, if present, would confound calculations of transmembrane ion flux from measurement of ionic content (Horres and Lieberman, 1977; Wheeler et al., 1982).

#### *Intracellular pH Monitoring*

The fluorescence intensity of 6-carboxyfluorescein (6CF) indicated relative changes in intracellular pH (Piwnica-Worms and Lieberman, 1983). Briefly, preparations were exposed to the nonfluorescent precursor dye 6CF diacetate (Molecular Probes, Inc., Junction City, OR), which diffused through the cell membrane to be cleaved by native intracellular esterases to form 6CF. Preparations were then transferred to a perfusion chamber and fluorescence intensity (excitation, 488 nm; emission, 523 nm) was monitored with a filter microfluorometer. Because the dye concentration and the pathlength of light were constant, the direction and rate of fluorescence intensity changes indicated the direction and response time of changes in  $\text{pH}_i$ . Single-wavelength fluorescence is nonlinear for  $\text{pH} > 7.3$  (Piwnica-Worms and Lieberman, 1983); therefore, response times are reported as apparent  $t_{1/2}$ . For  $\text{pH} < 7.3$  (acid-loaded conditions), apparent  $t_{1/2}$  values are not expected to be greatly distorted by this nonlinearity.

#### *pH Stat*

To measure the transmembrane movement of  $\text{H}^+$  equivalents, polystrands were suspended in plastic cuvettes containing 3.0 ml of test solution weakly buffered with 0.05 mM HEPES ( $\text{pH}$  7.40; Table I). Solutions were continuously mixed with a magnetic stirbar and maintained at 37°C. Acid release by the cells was monitored using a microcombination pH electrode (MI-710, Microelectrodes, Inc., Londonderry, NH) coupled to a pH meter (model 26, Radiometer, Copenhagen, Denmark) and a pH stat titrator (model 11 with an ABU 12 Autoburette, Radiometer). Solution pH was maintained at  $7.40 \pm 0.01$  using 0.5 mM KOH as a titrant and both the rate of titrant addition and solution pH were continuously recorded (model A5223-5, Omniscrite 2 channel, Houston Instrument, Austin, TX). Correction for background  $\text{CO}_2$  absorption (typically 1–2 nmol  $\text{H}^+$  equivalents/min) was made in all experiments. At the end of each experiment, preparations were removed for dry weight determination and solution buffer capacity was measured by addition of KOH. If net alkali, rather than acid, was transiently extruded by preparations, pH stat titration automatically stopped while alkaline drifts in  $\text{pH}_o$  continued to be recorded. This  $\Delta\text{pH}_o$ , multiplied by the solution buffer capacity, was used to measure net extrusion of alkali equivalents.

### *Electrical Recordings*

Transmembrane potentials were recorded using standard microelectrode techniques on spontaneously beating polystrands in a superfusion chamber (Horres et al., 1979; Piwnica-Worms et al., 1983); switching between solutions occurred with a  $t_{1/2}$  of 4 s at a flow rate of 10 ml/min. Records were retained with a chart recorder (7702B, Hewlett-Packard Co., Waltham, MA).

### *Statistics*

Values are presented as means  $\pm$  SEM. Statistical significance was evaluated by the two-tailed Student's *t* test unless otherwise indicated (Remington and Schork, 1970).

## RESULTS

### *pH<sub>i</sub> Regulation*

pH<sub>i</sub> regulation in heart cells was examined following an acid load produced by the NH<sub>4</sub><sup>+</sup> prepulse technique (Boron and De Weer, 1976; Roos and Boron, 1981). NH<sub>4</sub><sup>+</sup>-loaded cells rapidly lose NH<sub>3</sub> on exposure to NH<sub>4</sub><sup>+</sup>-free solution; H<sup>+</sup> released in the cells results in cytoplasmic acidification. Fig. 1A illustrates the pH<sub>i</sub> response induced by a 15-min pretreatment with solution containing 20 mM NH<sub>4</sub><sup>+</sup>. Removal of external NH<sub>4</sub><sup>+</sup> caused a rapid cytosolic acidification, followed by pH<sub>i</sub> recovery with a  $t_{1/2}$  of  $\sim 2.9 \pm 0.5$  min ( $n = 6$ ). Regulation of pH<sub>i</sub> was inhibited by Na<sub>o</sub>-free solution (Fig. 1B); pH<sub>i</sub> rapidly recovered on subsequent re-establishment of an inward Na gradient. The cause of the transient acidification observed during the initial return to control Na<sub>o</sub> is unknown; a similar transient, under these conditions, has also been recorded in renal proximal tubules with pH-selective microelectrodes (Boron and Boulpaep, 1983). The effect of amiloride ( $10^{-3}$  M), a rapid and reversible inhibitor of Na/H exchange in several cell types (for review, Benos, 1982), is illustrated in Fig. 1C. The initial rapid decrease in fluorescence on addition of amiloride reflected, in part, quenching of the 6CF signal by amiloride<sup>1</sup> since the neutral form of the drug can permeate cells (Benos et al., 1983). Amiloride did not block pH<sub>i</sub> changes caused by passive NH<sub>3</sub> movements, but did block the pH<sub>i</sub>-regulatory response following the NH<sub>4</sub><sup>+</sup>-induced acid load ( $n = 3$ ). In sum, these findings suggest an Na/H exchange mechanism involved in pH<sub>i</sub> regulation in heart cells following an acid perturbation.

### *pH<sub>i</sub>-induced Changes in Na<sub>i</sub>*

Figs. 2 and 3 show changes in Na<sub>i</sub> content caused by an NH<sub>4</sub><sup>+</sup> prepulse. In control

<sup>1</sup> Despite preliminary results to the contrary (Piwnica-Worms and Lieberman, 1983), a control series of experiments confirmed that amiloride can affect 6CF fluorescence as predicted by equations for either static or dynamic quenching (Lakowicz, 1983): at a fixed 6CF concentration, quenching was proportional to the amiloride concentration such that the fluorescence intensity of 1  $\mu$ M 6CF in control solution was reduced 24% by  $10^{-3}$  M amiloride, 4% by  $10^{-4}$  M amiloride, and  $\sim 1\%$  by  $10^{-5}$  M amiloride. On the other hand, at a fixed amiloride concentration, quenching was a constant proportion of the fluorescence intensity at various 6CF concentrations. Quenching was identical with or without protein (albumin) and in HCO<sub>3</sub><sup>-</sup>- or HEPES-buffered solutions.

polystrands, steady state  $\text{Na}_i$  was  $68.0 \pm 3.6$  nmol/mg dry weight ( $n = 9$ ) or  $15.8 \pm 0.8$  mM, assuming a steady state water content of  $4.29 \mu\text{l/mg}$  dry weight. This value correlated well with previously published measurements of  $\text{Na}_i$  in polystrands (Wheeler et al., 1982). After 15 min in  $\text{NH}_4^+$  solution,  $\text{Na}_i$  was  $48.7 \pm 4.2$  nmol/mg dry weight ( $n = 5$ ). Acidifying the cytosol by shifting  $\text{NH}_4^+$ -loaded preparations to  $\text{NH}_4^+$ -free (control) solution for 1 min caused a rapid increase in  $\text{Na}_i$  to almost twice the steady state value. The difference of  $75.6 \pm 6.9$  nmol/

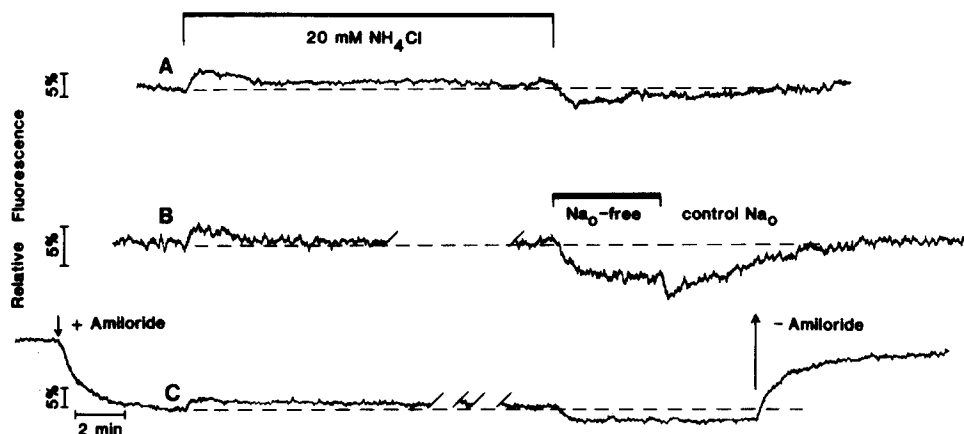


FIGURE 1. Effects of Na-free solution and amiloride on  $\text{pH}_i$  recovery after acid load as monitored microfluorometrically with the intracellular pH-sensitive dye 6CF (excitation, 488 nm; emission, 523 nm). Tracings show relative fluorescence intensity. Top bar shows time of exposure to 20 mM  $\text{NH}_4^+$  for all traces. (A)  $\text{pH}_i$  recovery in control solution. Six experiments on five preparations showed similar results. (B) Effect of Na-free solution. Horizontal dashed lines indicate fluorescence intensity prior to  $\text{NH}_4\text{Cl}$  exposure; breaks in the records indicate where light was scattered by bubbles passing through the chamber. (C) Effect of exposure to amiloride ( $10^{-3}$  M). Each tracing represents a different preparation. Relative fluorescence is expressed as a percentage of total fluorescence due to the dye. Downward deflection of relative fluorescence trace indicates decreasing  $\text{pH}_i$ .

mg dry weight between the latter two values of  $\text{Na}_i$  content (Fig. 3,  $b - a$ ) represented an acid-induced net Na uptake. The 1-min uptake proved to be the most useful measure of acid-induced net Na movements for technical reasons associated with obtaining consistent results at shorter times and because  $\text{Na}_i$  content peaked at this time. The decline in  $\text{Na}_i$  content over longer times (Fig. 2) was ouabain inhibitable ( $10^{-4}$  M; data not shown), which suggests that the Na/K pump secondarily responded to the acid-induced increase in  $\text{Na}_i$ .

$\text{pH}_i$  regulation was previously found to be amiloride sensitive, and therefore  $\text{Na}_i$  changes during  $\text{pH}_i$  regulation were similarly examined. Amiloride ( $10^{-3}$  M) did not significantly affect the fall in  $\text{Na}_i$  during  $\text{NH}_4^+$  exposure; however,

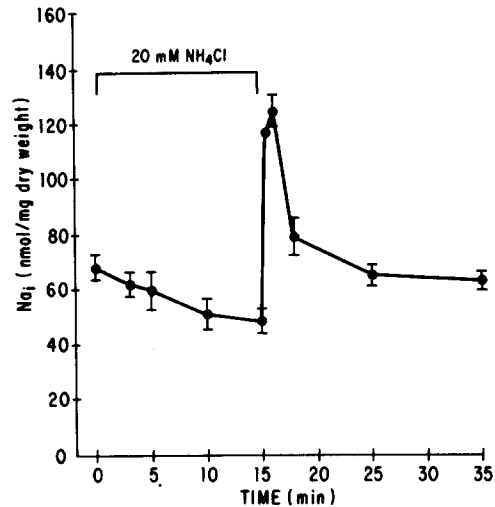


FIGURE 2. Changes in Na<sub>i</sub> content during NH<sub>4</sub><sup>+</sup>-induced acid load. Bars represent means ± SEM of three to six determinations; individual points are means of two experiments. Na<sub>i</sub> content at time zero is the value in control solution; no significant changes occurred in preparations maintained in control solution for times similar to these experiments.

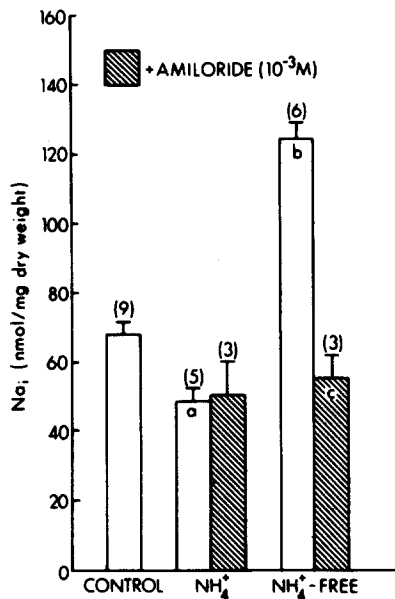


FIGURE 3. Changes in Na<sub>i</sub> content of cultured heart cells produced by NH<sub>4</sub><sup>+</sup>-induced changes in pH<sub>i</sub>. Na<sub>i</sub> content was measured in control solution after a 15-min exposure to 20 mM NH<sub>4</sub><sup>+</sup> (solution 3, Table I) and after a subsequent 1-min return to NH<sub>4</sub><sup>+</sup>-free (control) solution. Bars represent means ± SEM; n is the number of determinations.

exposing polystrands to  $\text{NH}_4^+$ -free solution containing amiloride blocked net Na uptake. The difference of  $68.5 \pm 8.7$  nmol/mg dry weight (Fig. 3, *b - c*) represented the amiloride-sensitive net Na uptake, which was 91% of the acid-induced net Na uptake. Similar results were obtained whether amiloride was applied at the beginning of  $\text{NH}_4^+$  exposure and present throughout the return to  $\text{NH}_4^+$ -free solution or applied simultaneously with a return to  $\text{NH}_4^+$ -free solution (amiloride-sensitive Na uptake:  $68.5 \pm 8.7$  vs.  $67.2 \pm 7.4$  nmol/mg dry weight; degrees of freedom = 7,  $P > 0.50$ ), which suggests that amiloride inhibited Na uptake within seconds.

Spontaneously contractile polystrands contain a nonmuscle component (fibroblast-like cells) that may account for up to 25% of the ionic content (Horres, 1975). No significant difference was found in  $\text{Na}_i$  content between contractile (C) polystrands and noncontractile (NC) polystrands under control conditions (C:  $68.0 \pm 3.6$  [ $n = 9$ ] vs. NC:  $82.1 \pm 11.1$  [ $n = 3$ ] nmol/mg dry weight;  $P > 0.30$ ). Similarly, no differences were found after 15 min in  $\text{NH}_4^+$  (C:  $48.7 \pm 4.2$  [ $n = 5$ ] vs. NC:  $36.7 \pm 3.8$  [ $n = 3$ ] nmol/mg dry weight;  $P > 0.05$ ). However,  $\text{Na}_i$  content in noncontractile polystrands only approached control values 1 min after return to  $\text{NH}_4^+$ -free solution ( $75.3 \pm 10.6$  nmol/mg dry weight [ $n = 3$ ]), and was significantly less than  $\text{Na}_i$  content in contractile polystrands ( $124.3 \pm 5.5$  nmol/mg dry weight [ $n = 6$ ];  $P < 0.02$ ). Thus, nonmuscle cells in contractile polystrands could not account for the rapid increase in  $\text{Na}_i$  content beyond control values. In fact, their presence would cause an underestimation of net Na uptake into cardiac muscle cells.

The effects of pharmacological agents and changes in external ion content on Na uptake induced by an acid load are shown in Table II. Polystrands were exposed for 15 min to 20 mM  $\text{NH}_4^+$  solution (except as noted) and then switched for 1 min to  $\text{NH}_4^+$ -free solutions altered as indicated. Net Na uptake is presented as a percent of Na uptake induced by a return to  $\text{NH}_4^+$ -free (control) solution. Amiloride ( $10^{-3}$  M) significantly inhibited net Na uptake. Ouabain ( $10^{-4}$  M), at a dose known to rapidly and completely inhibit the Na/K pump in polystrands (Lieberman et al., 1982), significantly increased net Na uptake to 138% of control or, accounting for the 1-min uptake period, to a rate of  $103.8 \pm 9.6$  nmol Na/mg dry weight · 1 min ( $n = 7$ ). In the absence of an acid load, ouabain ( $10^{-4}$  M) causes  $\text{Na}_i$  content in cultured heart cells to increase at a rate of only 10–14 nmol/mg dry weight · 1 min (recalculated from Horres et al., 1979, and Wheeler, 1981). Therefore, Na/K pump inhibition alone cannot account for the rapid acid-induced net Na uptake. Simultaneous addition of amiloride and ouabain produced a small net Na uptake. Either ouabain competed for amiloride binding sites, rendering amiloride ineffective, or, more likely, the simultaneous addition of these two agents induced Na uptake by an amiloride- and ouabain-insensitive pathway. For this reason, the acid-induced net Na uptake, unattenuated by Na/K pump activity, was taken to be the difference in Na uptake between each agent acting alone. Tetrodotoxin (10  $\mu\text{g}/\text{ml}$ ) applied throughout the maneuver did not significantly affect acid-induced net Na uptake, which suggests that  $\text{Na}^+$  movements were not mediated by fast Na channels. Quinidine ( $10^{-3}$  M), a putative blocker of Na/Ca exchange (Parker, 1978), applied 1 min



before the return to control solution, caused  $\text{Na}_i$  content changes similar to ouabain (data not shown), which indicates that quinidine also does not inhibit acid-induced net Na uptake.

In control solution, 2 min after application of amiloride ( $10^{-3}$  M),  $\text{Na}_i$  declined to  $57.8 \pm 3.6$  nmol/mg dry weight ( $n = 3$ ); after 5 min,  $\text{Na}_i$  was  $56.8 \pm 3.5$  nmol/mg dry weight ( $n = 4$ ), a value significantly different from control ( $68.0 \pm 3.6$ ;  $P < 0.05$ ). Membrane potential recordings during addition of amiloride ( $10^{-3}$  M) to control solution for 5 min showed maximum diastolic potentials to depolarize by 4–5 mV with no changes in overshoot potentials (data not shown). These findings were inconsistent with a decrease in  $\text{Na}^+$  conductance and

TABLE II  
*Effects of Various Drugs and External Ions on Acid-induced Na Uptake*

Condition	$\Delta\text{Na}_i$ (% control)	<i>n</i>	<i>P</i>
$\text{NH}_4^+$ -free control	100±9	6	—
+ Amiloride ( $10^{-3}$ M)	10±4	10	<0.001
+ Ouabain ( $10^{-4}$ M)	138±13	7	0.05
+ Amiloride ( $10^{-3}$ M) + ouabain ( $10^{-4}$ M)	36±15	4	<0.02
+ TTX (10 µg/ml)	86±6	3	>0.20
+ DIDS ( $10^{-4}$ M)	82±14	4	>0.30
$\text{HCO}_3^-$ -free	96±20	3	>0.50
$\text{Cl}^-$ -free	109±22	4	>0.50

Polystrands were exposed to 20 mM  $\text{NH}_4\text{Cl}$  solution for 15 min and then incubated for 1 min in  $\text{NH}_4^+$ -free (control) solution altered as indicated (see text for details). Net Na uptake in  $\text{NH}_4^+$ -free solution was normalized to the control  $\text{NH}_4^+$ -free maneuver as seen in Fig. 3. Results are presented as means  $\pm$  SEM; *n* is the number of determinations; *P* values are a two-tailed comparison with control.

suggested that part of the steady state  $\text{Na}^+$  “leak” may actually be mediated by Na/H exchange.

#### *Na/H Exchange vs. Na + HCO<sub>3</sub>/Cl + H Exchange*

The results shown thus far are consistent with the presence of an Na/H exchange mechanism. However, the Na uptake induced by an acid load does not distinguish between Na/H exchange and a coupled Na +  $\text{HCO}_3^-/\text{Cl}^- + \text{H}^+$  exchange. The latter occurs in squid axons (Boron and Russell, 1983), snail neuron (Thomas, 1977), and barnacle muscle (Boron et al., 1981), and would also produce net Na uptake in response to an acid load. This exchange is affected by the direction of the  $\text{Cl}^-$  gradient and is inhibited by removal of  $\text{HCO}_3^-$  as well as by the application of stilbene derivatives (Russell et al., 1983). Therefore, to rule out the possibility of Na +  $\text{HCO}_3^-/\text{Cl}^- + \text{H}^+$  exchange, Na uptake following  $\text{NH}_4\text{Cl}$  preincubation was measured under the following conditions (Table II): (a) in nominally  $\text{HCO}_3^-$ -free solution throughout, (b) with a return to  $\text{Cl}^-$ -free solution (which might stimulate Na uptake), and (c) in the presence of DIDS ( $10^{-4}$  M), at a dose 10 times that necessary to inhibit 90% of  $\text{pH}_i$  regulation mediated by Na +  $\text{HCO}_3^-/\text{Cl}^- + \text{H}^+$

exchange in snail neuron (Thomas, 1976). None of these treatments influenced the acid-induced net Na uptake in cultured heart cells.

#### *Amiloride Dose Response*

The dose-response curve for amiloride inhibition of Na uptake induced by an acid load is shown in Fig. 4. At an  $\text{Na}_o$  concentration of 144 mM, half-maximal inhibition of amiloride-sensitive net Na uptake occurred at 3  $\mu\text{M}$ .

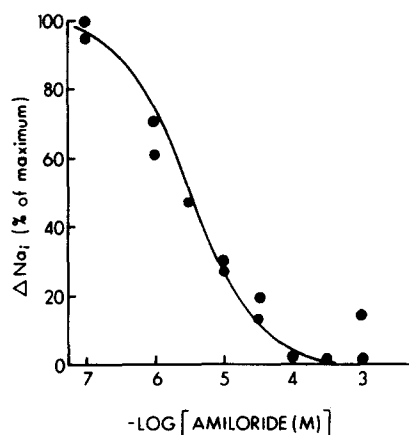


FIGURE 4. Dose-response curve for amiloride inhibition of acid-induced net  $\text{Na}^+$  uptake by cultured heart cells.  $\text{Na}^+$  uptake was defined as the difference between  $\text{Na}_i$  content after a 15-min incubation in 20 mM  $\text{NH}_4^+$  solution and after a subsequent 1-min switch to  $\text{NH}_4^+$ -free (control) solution containing various concentrations of amiloride.  $\text{pH}_o$  was 7.4;  $\text{Na}_o$  was 144 mM. The maximum net  $\text{Na}^+$  uptake in this experiment was 57 nmol/mg dry weight · 1 min. The solid line represents a simple titration curve with an apparent  $K_i$  of 3  $\mu\text{M}$ .

#### *pH<sub>o</sub> Effects on Acid-induced Na Uptake*

The response of Na uptake to changes in  $\text{pH}_o$  (maintaining  $\text{pH}_i$  approximately constant) is summarized in Table III. Preparations were exposed to  $\text{HCO}_3^-$ -free,  $\text{HPO}_4^{2-}$ -free  $\text{NH}_4^+$  solution ( $\text{pH}_o$  7.4) for 15 min before switching to  $\text{HCO}_3^-$ -free,  $\text{HPO}_4^{2-}$ -free,  $\text{NH}_4^+$ -free solutions of various  $\text{pH}_o$ 's. The absence of  $\text{HCO}_3^-$  and  $\text{HPO}_4^{2-}$  in the solutions allowed  $\text{pH}_o$  to be varied widely without precipitating calcium salts; ouabain ( $10^{-4}$  M) was added to  $\text{NH}_4^+$ -free solutions to prevent Na/K pump attenuation of net Na uptake. Increasing  $\text{pH}_o$  from 7.4 to 8.5 in  $\text{NH}_4^+$ -free solution did not significantly change amiloride-sensitive net Na uptake. However, exposing preparations to solutions at  $\text{pH}_o$  6.5 decreased amiloride-sensitive net Na uptake; solutions at  $\text{pH}_o$  5.5 further decreased net Na uptake. Although the decrease in net Na uptake upon decreasing  $\text{pH}_o$  was thermodynamically consistent with Na/H exchange, these data are also compatible with a kinetic effect due to competition between external  $\text{H}^+$  and  $\text{Na}^+$  for a binding site on the exchanger.

*pH<sub>i</sub> Effects on Acid-induced Na Uptake*

The dependence of Na uptake on pH<sub>i</sub> (at constant pH<sub>o</sub>) was examined by exposing polystrands to NH<sub>4</sub><sup>+</sup> solution (pH<sub>o</sub> 7.4) for various times before returning to NH<sub>4</sub><sup>+</sup>-free solution (pH<sub>o</sub> 7.4). Prolonging the duration of NH<sub>4</sub><sup>+</sup> exposure (but less than the time required for NH<sub>4</sub><sup>+</sup> to reach electrochemical equilibrium) increases the magnitude of the acidification produced on switching to NH<sub>4</sub><sup>+</sup>-free solution (Roos and Boron, 1981). Therefore, as shown in Fig. 5, increasing the internal acidification increased amiloride-sensitive net Na uptake. Because ouabain (10<sup>-4</sup> M) was applied during NH<sub>4</sub><sup>+</sup>-free incubation, changes in net Na uptake could not be explained by pH<sub>i</sub> sensitivity of the Na/K pump-mediated Na efflux (Breitwieser and Russell, 1983).

TABLE III  
*The Influence of pH<sub>o</sub> on Acid-induced Na Uptake in Cultured Heart Cells*

pH <sub>o</sub>	Amiloride	ΔNa <sub>i</sub>	
		nmol/mg dry wt. 1 min	Amiloride-sensitive ΔNa <sub>i</sub> nmol/mg dry wt. 1 min
8.5	O*	87.2±7.0 (6)	80.3±7.6 <sup>‡</sup>
	+	6.9±2.9 (4)	
7.4	O	93.9±10.7 (6)	86.4±10.9
	+	7.5±1.9 (3)	
6.5	O	65.7±3.5 (3)	56.6±5.0 <sup>‡</sup>
	+	9.1±3.6 (3)	
5.5	O	37.9±15.2 (3)	29.9±15.5 <sup>‡†</sup>
	+	8.0±3.4 (3)	

Values are means ± SEM of *n* determinations. Net Na uptake was defined as the difference in Na<sub>i</sub> after incubating in HCO<sub>3</sub><sup>-</sup>-free, HPO<sub>4</sub><sup>2-</sup>-free, 20 mM NH<sub>4</sub><sup>+</sup> solution for 15 min, and Na<sub>i</sub> after a subsequent incubation for 1 min in HCO<sub>3</sub><sup>-</sup>-free, PO<sub>4</sub><sup>3-</sup>-free, NH<sub>4</sub><sup>+</sup>-free solution containing either ouabain\* (10<sup>-4</sup> M) or amiloride (10<sup>-3</sup> M).

\* Comparison with pH<sub>o</sub> 7.4, *P* > 0.50.

<sup>‡</sup> Comparison with pH<sub>o</sub> 7.4, *P* < 0.05 (one-tailed).

<sup>†</sup> Comparison with pH<sub>o</sub> 6.5, *P* = 0.07 (one-tailed).

*Na<sub>o</sub>-induced pH<sub>i</sub> Changes*

To examine the relationship between Na<sub>o</sub> and pH<sub>i</sub>, Na gradients were systematically varied while rates of pH<sub>i</sub> changes were monitored with 6CF (Fig. 6). From a stable pH<sub>i</sub> in control solution, lowering Na<sub>o</sub> resulted in increased rates of intracellular acidification up to a minimum *t*<sub>1/2</sub> of ~20 s in 0.8 mM Na<sub>o</sub> solution. Responses occurred only when Na<sub>o</sub> was reduced below 72 mM and, within the accuracy of this technique, a half-maximal response occurred between 20 and 30 mM Na<sub>o</sub>. On switching to progressively lower Na<sub>o</sub>, both the rate and relative magnitude of the cytosolic acidification increased. However, this qualitative change in magnitude could not be analyzed further because the observations in this series involved different polystrands and hence the final dye concentration may have varied (see Piwnica-Worms and Lieberman, 1983). In contrast, heart cells Na<sub>i</sub>-depleted by preincubation in low Na<sub>o</sub> alkalinized on return to normal

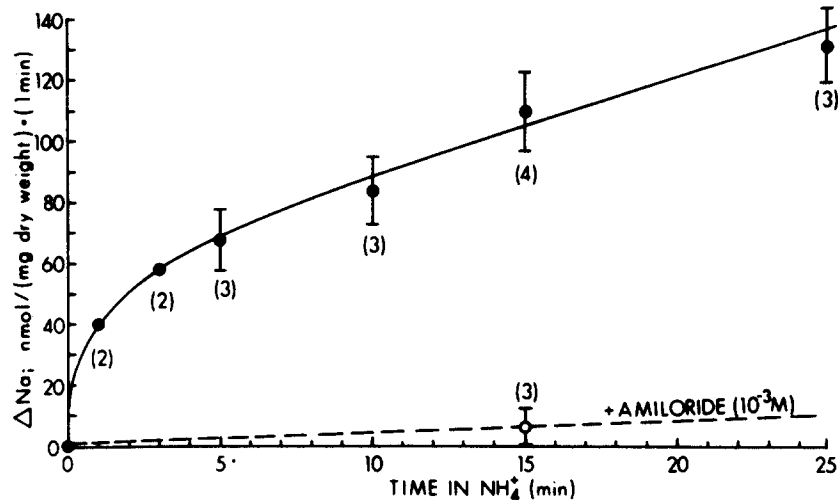


FIGURE 5. Effect of decreasing  $pH_i$  on acid-induced net  $Na^+$  uptake in cultured heart cells. After incubating cells for various times in 20 mM  $NH_4^+$  solution, the net change in  $Na_i$  content was determined after 1 min in  $NH_4^+$ -free (control) solution containing ouabain ( $10^{-4}$  M). Increasing the time in  $NH_4^+$  solution decreased  $pH_i$  during the subsequent exposure to  $NH_4^+$ -free solution.

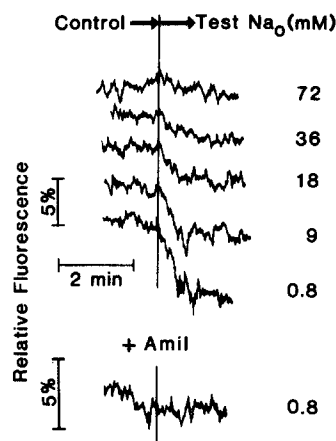
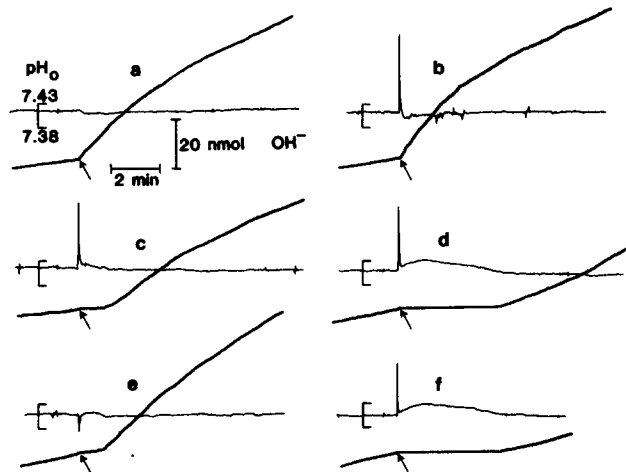


FIGURE 6. Rate of cytosolic acidification in cultured heart cells as a function of  $Na_o$ . Kinetics of  $pH_i$  changes were monitored with 6CF (excitation, 488 nm; emission, 523 nm). Relative fluorescence is expressed as a percentage of total fluorescence due to the dye. Downward deflection of relative fluorescence trace indicates decreasing  $pH_i$ . Following the stable recording of  $pH_i$  in control solution, six preparations were switched to test solutions of different  $Na_o$  (TMA substitution) produced by mixing various ratios of solutions 1 and 2 (Table I). In this series, the diameter of each monitored strand was 100  $\mu m$ . In the last tracing, amiloride ( $10^{-5}$  M) was added 3 min before the switch to low  $Na_o$  solution.

$\text{Na}_o$ . Amiloride ( $10^{-3}$  M) completely and reversibly inhibited both low  $\text{Na}_o$ -induced acidification (Fig. 6;  $n = 3$ ) and, in  $\text{Na}_o$ -depleted cells, normal  $\text{Na}_o$ -induced alkalinization. Ouabain ( $10^{-4}$  M) did not inhibit  $\text{Na}_o$ -induced  $\text{pH}_i$  changes (data not shown).



**FIGURE 7.** Rate of acid release by cultured heart cells under pH stat conditions. Polystrands were preincubated for 15 min in pH stat control solution alone (solution 6; Table I), modified as indicated below and by addition of 5 mM HEPES to maintain pH 7.4. At the arrow, polystrands were transferred to the pH stat cuvette containing lightly buffered control solution alone (solution 6) or modified as indicated, and the rate of acid release was recorded as the rate of KOH addition to maintain  $\text{pH } 7.40 \pm 0.01$ . The initial portion of each pH stat tracing shows background  $\text{CO}_2$  absorption by the solution. The thinner tracing in each panel is  $\text{pH}_o$ . The vertical and horizontal scales apply to all panels. Each panel is representative of four to six identical experiments on different preparations, except panel *e*, which was one observation (see Table II for confirmation of DIDS effect). The final acid extrusion rates are normalized to tissue dry weight (see text). (*a*) Preincubation: control solution; pH stat: control solution. (*b*) Preincubation: control solution plus amiloride ( $10^{-4}$  M); pH stat: control solution plus amiloride ( $10^{-4}$  M). (*c*) Preincubation: 20 mM  $\text{NH}_4\text{Cl}$  solution; pH stat: control solution. (*d*) Preincubation: 20 mM  $\text{NH}_4\text{Cl}$  solution plus amiloride ( $10^{-4}$  M); pH stat: control solution plus amiloride ( $10^{-4}$  M). (*e*) Preincubation: 20 mM  $\text{NH}_4\text{Cl}$  solution plus DIDS ( $10^{-4}$  M); pH stat: control solution plus DIDS ( $10^{-4}$  M). (*f*) Preincubation: 20 mM  $\text{NH}_4\text{Cl}$ ,  $\text{Na}_o$ -free solution; pH stat:  $\text{Na}_o$ -free solution.

#### *Transmembrane $\text{H}^+$ Transport*

pH stat experiments confirmed that  $\text{pH}_i$  regulation following an acid load involved the transmembrane extrusion of  $\text{H}^+$  equivalents rather than an  $\text{Na}$ -dependent intracellular buffering mechanism. Preparations preincubated for 15 min in control solution, when transferred to the pH stat cuvette containing control solution, showed a rapid phase of acid release of unknown origin (Fig. 7*a*); after  $\sim 5$  min, a steady rate of acid extrusion was achieved ( $22.8 \pm 1.4$  nmol

H<sup>+</sup> equivalents/mg dry weight · min;  $n = 12$ ). Amiloride ( $10^{-4}$  M) did not inhibit the initial acid transient (Fig. 7*b*), but reduced the steady rate to  $16.4 \pm 1.7$  nmol H<sup>+</sup> equivalents/mg dry weight · min ( $n = 5$ ). Preparations preincubated for 15 min in NH<sub>4</sub>Cl did not show a rapid phase of acid release (Fig. 7*c*); almost no net acid extrusion occurred during the first 20–40 s in control solution (Fig. 7). NH<sub>4</sub>Cl preincubation with amiloride ( $10^{-4}$  M) or Na<sub>o</sub>-free solution completely inhibited net acid extrusion for an extended period (Fig. 7, *d* and *f*, respectively); an amiloride- or Na<sub>o</sub>-insensitive acid release appeared only after 5–6 min in control solution containing amiloride or in Na<sub>o</sub>-free solution. DIDS ( $10^{-4}$  M) had no effect on the pattern of acid release following NH<sub>4</sub>Cl preincubation (Fig. 7*e*).

pH<sub>o</sub> remained at  $\sim 7.41$  in control solution with or without amiloride, drifted briefly alkaline after NH<sub>4</sub>Cl preincubation alone, and shifted strongly alkaline for an extended period following NH<sub>4</sub>Cl preincubation in the presence of amiloride or Na<sub>o</sub>-free solution (Fig. 7). NH<sub>4</sub>Cl preincubation experiments were interpreted as follows: the preparation and external solution in the cuvette represented a closed two-compartment system. Thus, on return to control solution, brief alkaline drifts in pH<sub>o</sub> indicated that NH<sub>3</sub> diffused across the cell membrane to bind extracellular H<sup>+</sup> at a rate only slightly faster than H<sup>+</sup> was extruded by Na/H exchange; the combined effect during the first minute resulted in little net extrusion of acid equivalents (Fig. 7*c*). Inhibiting Na/H exchange with amiloride (or Na<sub>o</sub>-free solution) revealed the external alkalizing effect of transmembrane NH<sub>3</sub> efflux acting alone (Fig. 7, *d* and *f*). Therefore, the difference in the maximum pH<sub>o</sub> changes under each condition, combined with information on the buffer capacity of the solutions, yielded the net amiloride-sensitive acid extrusion. Correcting for background CO<sub>2</sub> absorption, this method gave  $70.9 \pm 9.1$  nmol/mg dry weight ( $n = 5$ ) of amiloride-sensitive H<sup>+</sup> equivalents extruded across the cell membrane in the first minute after transfer to NH<sub>4</sub><sup>+</sup>-free (control) solution. This value, compared with  $68.5 \pm 8.7$  nmol/mg dry weight of amiloride-sensitive net Na uptake (see above), indicates a 1:1 Na/H stoichiometry.

The pH stat technique detected a Na<sub>o</sub>-insensitive acid release (Fig. 8*a*) 6–8 min after NH<sub>4</sub>Cl preincubation in Na<sub>o</sub>-free solution of  $9.4 \pm 0.7$  nmol/mg dry weight · min ( $n = 9$ ).<sup>2</sup> Acid extrusion could be stimulated by addition of Na<sup>+</sup> to the extracellular solution; this effect was amiloride sensitive (Fig. 8*a*). Li<sup>+</sup>, but not K<sup>+</sup> or choline<sup>+</sup>, could substitute for Na<sup>+</sup> in augmenting acid extrusion. The dose response for Na<sub>o</sub>-dependent acid extrusion showed a half-maximal effect of  $\sim 15$  mM Na<sub>o</sub> (Fig. 8*b*). The reduced maximum rate of H<sup>+</sup> extrusion under this condition compared with that immediately after NH<sub>4</sub> pretreatment probably reflects partial dissipation of the acid load by amiloride- or Na<sub>o</sub>-insensitive mechanisms.

<sup>2</sup> Assuming steady state values for cell H<sub>2</sub>O and intracellular buffering capacity (see Discussion), this would be equivalent to a change in pH<sub>i</sub> at the rate of  $\sim 0.035$  pH unit/min; the relatively insensitive single-wavelength fluorescence technique could not detect this slow rate of Na<sub>o</sub>-insensitive pH<sub>i</sub> recovery over the short times tested (Fig. 1).

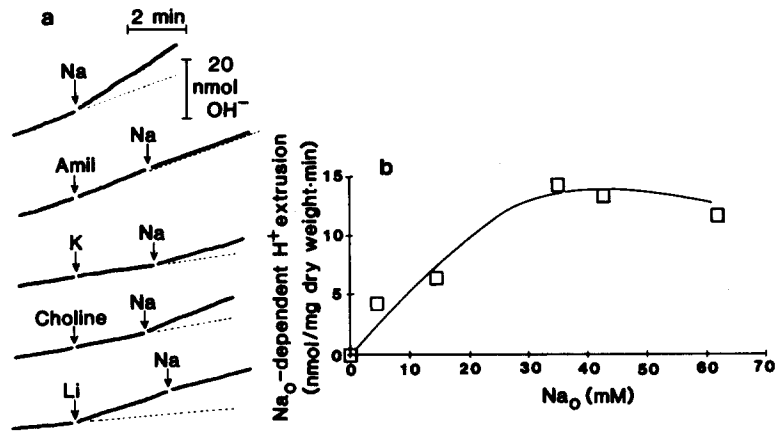


FIGURE 8. Na<sub>o</sub>-dependent H<sup>+</sup> extrusion in cultured heart cells. (a) Preparations were preincubated for 15 min in 20 mM NH<sub>4</sub>Cl, Na<sub>o</sub>-free (TMA) solution (based on solution 6, Table I), with 5 mM HEPES to maintain pH 7.4, and transferred to Na<sub>o</sub>-free pH stat solution (pH 7.4). The initial portion of each tracing shows Na<sub>o</sub>-independent acid release (plus background CO<sub>2</sub> absorption) ~5–6 min after transfer; the dashed line is an extension of this rate. At the arrow, concentrated NaCl, LiCl, KCl, or choline Cl in 0.05 mM HEPES (pH 7.4) was added such that the final concentration of cation was 40 mM; the final concentration of amiloride was 10<sup>-4</sup> M. The break in each tracing represents 10–20 s. The vertical and horizontal scales apply to each tracing. (b) Na<sub>o</sub>-dependent H<sup>+</sup> extrusion based on results obtained by the method illustrated in panel a. Each point in the series represents a single observation corrected for background CO<sub>2</sub> absorption.

#### *Electrical Response to pH<sub>i</sub> Regulation*

The electroneutrality of Na/H exchange was confirmed by membrane potential recordings (Fig. 9). Application of NH<sub>4</sub>Cl depolarized polystrands, possibly because of NH<sub>4</sub><sup>+</sup> permeability through K<sup>+</sup> channels (Roos and Boron, 1981). Application of ouabain (10<sup>-4</sup> M) inhibited secondary activation of the electrogenic Na/K pump during pH<sub>i</sub> regulation. The lack of a significant depolarization

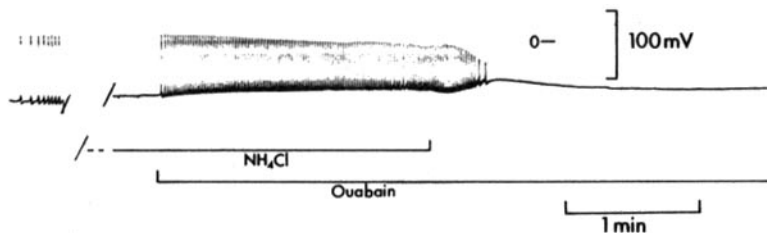


FIGURE 9. Continuous recording of membrane potential of cultured heart cells during pH<sub>i</sub> regulation. The initial tracing shows a polystrand in control solution. The solution was then switched to one containing 20 mM NH<sub>4</sub>Cl for 15 min; ouabain (10<sup>-4</sup> M) was added 2 min before the switch back to NH<sub>4</sub><sup>+</sup>-free (control) solution. The break in the record represents 13 min.

in the presence of a massive uptake of  $\text{Na}^+$  (see Table II) provided direct evidence for electroneutral Na/H exchange.

#### *Ca<sub>i</sub> Content Changes*

The interaction of  $\text{H}^+$  and  $\text{Na}^+$  with  $\text{Ca}^{2+}$  is of considerable interest in cardiac muscle (Bers and Ellis, 1982; Vaughan-Jones et al., 1983; Piwnica-Worms and Lieberman, 1983). Therefore, total cell Ca ( $\text{Ca}_i$ ) was also determined during an  $\text{NH}_4^+$ -induced acid load. Steady state  $\text{Ca}_i$  was  $6.6 \pm 0.6$  nmol/mg dry weight ( $n = 5$ ), and declined by  $\sim 30\%$  during  $\text{NH}_4^+$  exposure. Switching to  $\text{NH}_4^+$ -free solution produced a rise in  $\text{Ca}_i$  back toward the control value ( $n = 3$ ) with a  $t_{1/2}$

TABLE IV  
*Effect of Ca<sub>o</sub> on Acid-induced Na Uptake*

Ca <sub>o</sub> <sup>2+</sup> <i>mM</i>	Mg <sup>2+</sup> <i>0.8 mM</i>	HPO <sub>4</sub> <sup>2-</sup> <i>0.8 mM</i>	Ouabain <i>10<sup>-4</sup> M</i>	Acid-induced $\Delta\text{Na}_i$ <i>nmol/mg dry wt · 1 min</i>
2.7	+	+	-	73.7 ± 9.6 (9)
	+	-	-	43.7 ± 6.6 (3)
	+	-	+	64.8 ± 13.0 (3)
21.8	+	-	+	44.9 ± 7.6 (3)
10.8	+	-	-	58.4 ± 16.9 (3)
0.27	+	+	-	78.3 ± 10.4 (3)
0	+	-	+	125.6 ± 5.7 (3)*
0	-	+	-	145.8 ± 18.3 (3)*

Values are means  $\pm$  SEM.  $n$  is the number of determinations. The asterisk indicates a significant difference from the control value (top of table) at  $P < 0.05$ . Acid-induced net Na uptake was determined as shown in Fig. 3. Both control and  $\text{NH}_4\text{Cl}$ -containing solutions had the indicated  $\text{HPO}_4$  concentrations; Ca and Mg were adjusted on return to control solution; ouabain, if present, was added on return to control solution. Amiloride ( $10^{-5}$  M) inhibited  $\sim 90\%$  of acid-induced Na uptake in  $\text{Ca}_o$ -free solution and at  $\text{Ca}_o$  concentrations of 21.8, 10.8, and 2.7 mM (not tested at 0.27 mM  $\text{Ca}_o$ ).

on the order of 4 min. The response time of  $\text{Ca}_i$  changes indicated that net Ca uptake correlated best with net Na extrusion following peak  $\text{Na}_i$  values (see Fig. 2), which suggests that the Na/Ca exchange that exists in the polystrand preparation (Lieberman et al., 1984) may assist the Na/K pump in re-establishing the Na gradient subsequent to  $\text{pH}_i$  regulation.

#### *External Ca*

The effects of the external divalent ions  $\text{Ca}^{2+}$ ,  $\text{Mg}^{2+}$ , and  $\text{HPO}_4^{2-}$  on acid-induced Na uptake are shown in Table IV. Of the various conditions tested, only nominally  $\text{Ca}_o$ -free or  $\text{Mg}_o$ -free,  $\text{Ca}_o$ -free solutions significantly increased acid-induced Na uptake; on the other hand, the presence of these external cations showed an inhibitory trend. The net increase in acid-induced Na uptake in  $\text{Ca}_o$ -free compared with control  $\text{Ca}_o$  was more than could be accounted for by Na/Ca exchange, even if all steady state  $\text{Ca}_i$  exchanged for  $\text{Na}_o$  at a 4:1 stoichiometry (hypothetical increment in Na uptake: 26 nmol Na/mg dry weight · 1 min vs. measured increment: 52 nmol Na/mg dry weight · 1 min).



## DISCUSSION

*Characterization of Na/H Exchange in Heart Cells*

Several experiments demonstrate the coupled movements of  $\text{Na}^+$  and  $\text{H}^+$  consistent with Na/H exchange: following the introduction of an  $\text{NH}_4^+$ -induced acid load, (a)  $\text{Na}_o$ -sensitive  $\text{pH}_i$  regulation mediated by extrusion of  $\text{H}^+$  across the cell membrane results in a return of  $\text{pH}_i$  toward control values, (b)  $\text{pH}_i$  regulation promotes a rapid net uptake of  $\text{Na}^+$ , (c) larger decreases in  $\text{pH}_i$  produce larger increases in net  $\text{Na}^+$  uptake, and (d) decreasing  $\text{pH}_o$  inhibits acid-induced  $\text{Na}^+$  uptake. Also, without an acid load, (e) reduction of  $\text{Na}_o$ , which reverses the  $\text{Na}^+$  chemical gradient, causes intracellular acidification, and (f) alkalization of  $\text{pH}_i$  occurs when an inward  $\text{Na}^+$  chemical gradient is imposed across cell membranes of  $\text{Na}_i$ -depleted preparations.

Several conclusions regarding the kinetics of Na/H exchange are suggested by these studies. (a) The external  $\text{Na}^+$  binding site is saturated at  $\text{Na}_o$  above 72 mM and, on switching to low  $\text{Na}_o$  solution, a half-maximal rate of cytosolic acidification occurs between 20 and 30 mM  $\text{Na}_o$ . (b) Similarly, the external  $\text{Na}^+$  binding site, evaluated with the pH stat method by addition of  $\text{Na}_o$  to  $\text{Na}_i$ -depleted cells, shows a half-maximal,  $\text{Na}_o$ -dependent rate of  $\text{H}^+$  extrusion at  $\sim 15$  mM  $\text{Na}_o$ . (c) The amiloride-sensitive rate of acid-induced  $\text{Na}^+$  uptake, evaluated at saturating  $\text{Na}_o$  concentrations, is  $\text{pH}_o$  dependent. (d) Internal acidification activates Na/H exchange. These data are consistent with the kinetic mechanisms of Na/H exchange described in sheep cardiac Purkinje fibers (Ellis and MacLeod, 1983) and in other tissues (Rindler and Saier, 1981; Villereal, 1981; Aronson et al., 1982; Vigne et al., 1982; Boron and Boulpaep, 1983; Aronson et al., 1983; Grinstein et al., 1984).

*Effect of Amiloride*

Corroborative evidence for the existence of Na/H exchange in cardiac muscle relies on inhibitory actions of the diuretic amiloride. In the polystrand preparation, amiloride ( $10^{-3}$  M) blocks: (a)  $\text{pH}_i$  recovery following an acid load, (b) net  $\text{Na}^+$  uptake induced by an acid load, (c) cytosolic acidification induced by low  $\text{Na}_o$  solution, and (d) cytosolic alkalization produced by re-establishment of an inward  $\text{Na}^+$  gradient in  $\text{Na}_i$ -depleted cells. Maximal inhibition of Na/H exchange by amiloride is similar in dose and reversibility to that reported for several other tissue types (for review, Benos, 1982; Aicken and Thomas, 1977; Moolenaar et al., 1981; Aronson et al., 1983). Furthermore, the half-maximal inhibitory dose ( $\text{ID}_{50}$ ) of amiloride is  $\sim 3$   $\mu\text{M}$  in 144 mM  $\text{Na}_o$ , a value that correlates with  $\text{ID}_{50}$  for amiloride inhibition of Na/H exchange in Chinese hamster lung fibroblast cells (Pouyssegur et al., 1982), renal membrane vesicles (Kinsella and Aronson, 1981), chick skeletal muscle cells in culture (Vigne et al., 1982), and dog red blood cells (Parker, 1983).

*Na/H Exchange vs. Na/Ca Exchange*

Distinguishing the effects of Na/Ca exchange from Na/H exchange in cardiac muscle has received recent attention (Bers and Ellis, 1982; Vaughan-Jones et al.,

1983; Piwnica-Worms and Lieberman, 1983). In this study, the acid-induced net Na uptake is too large to be attributed to Na/Ca exchange, a mechanism also known to exist in the cell membrane of polystrand preparations (Lieberman et al., 1984), as well as other preparations of cardiac muscle (for review, Langer, 1982). The Na/Ca exchange hypothesis suggests that increasing the intracellular  $H^+$  concentration by an acid load would increase cytosolic free  $Ca^{2+}$  both through direct displacement of  $Ca^{2+}$  from intracellular binding sites and through  $H^+$  exchange for  $Ca^{2+}$  across mitochondrial membranes (Vercesi et al., 1978). The resulting increase in cytosolic free Ca would favor Na entry via Na/Ca exchange. However,  $Ca_i$  measurements show, on a mass balance basis, that the initial net Na uptake cannot result from Na/Ca exchange across the cell membrane. The control  $Ca_i$  content was 6.6 nmol/mg dry weight, the majority being highly compartmentalized within subcellular organelles (Murphy et al., 1983). Assuming (a) that all depots of  $Ca_i$  exchange rapidly with external Na, and (b) an Na/Ca 4:1 stoichiometry (Mullins, 1981), then the maximum net Na uptake possible from Na/Ca exchange following the acid load would be 26.4 nmol/mg dry weight. This value represents only 35% of the acid-induced Na uptake (Fig. 2). Because the most positive value observed for  $E_m$  was  $-50$  mV (Fig. 9;  $E_m$  actually remained far negative to that value throughout most of the return to  $NH_4^+$ -free solution), Ca uptake via "slow" channels (for review, McDonald, 1982) is unlikely to enable further Na uptake to occur via the consequent Na/Ca exchange. The purpose of these calculations is not to address the electrogenicity or stoichiometry of Na/Ca exchange, but to illustrate that Na/Ca exchange could not be the mechanism mediating the rapid amiloride-sensitive Na uptake induced by an acid load.

On the other hand, because direct measurement of the membrane potential indicated electroneutral Na uptake following an acid load, an Na/H stoichiometry of 1:1 can be used to estimate the extent of the  $pH_i$  recovery. For example, assuming the mean intracellular buffering capacity of adult chicken cardiac muscle (58 mmol/liter · pH unit; Lykkeboe and Johansen, 1975) is comparable to chick embryo heart cells, the intracellular pH change expected from the 1-min  $H^+$  extrusion of 103.8 nmol/mg dry weight (Table II) is 0.42 pH unit. This calculated  $pH_i$  change is a reasonable value and is in good agreement with that observed in other preparations of muscle under similar conditions (Aickin and Thomas, 1977; Reber and Weingart, 1982).

The presence of  $Ca_o$  is partially inhibitory to Na/H exchange in cultured heart cells (Table IV); a similar effect has been reported in human fibroblasts (Villereal, 1982) and A431 epidermoid cells (Rothenberg et al., 1983; however, see Cala, 1983). Whether the stimulatory effect produced by removing external  $Ca^{2+}$  is by direct action on Na/H exchange or indirect via  $Ca_i$  is unproven. Nevertheless, these data raise a point of caution to investigators evaluating Na/Ca exchange in cardiac muscle. Na uptake promoted by  $Ca_o$ -free solution may be due in part to stimulation of Na/H exchange rather than to Na/Ca exchange alone.

#### *Comparison with Other Cardiac Tissues*

Evidence suggesting Na/H exchange has been described in one other cardiac preparation, the quiescent sheep Purkinje fiber (Deitmer and Ellis, 1980). Major

differences in the magnitude and rate of Na/H exchange exist between the polystrand preparation of cultured heart cells and sheep Purkinje fibers. For example,  $\text{Na}_o$ -sensitive regulation of  $\text{pH}_i$  following a 17-min  $\text{NH}_4^+$ -induced acid load in Purkinje fibers occurs with a  $t_{1/2}$  of  $\sim 10$  min, whereas  $\text{pH}_i$  regulation in polystrands, under similar conditions, occurs with a  $t_{1/2}$  of  $\sim 2.9$  min. In addition, with a 1:1 exchange stoichiometry, the maximum rate of acid-induced net  $\text{Na}^+$  uptake in Purkinje fibers is 2.5 mmol/liter·min (Deitmer and Ellis, 1980), 10 times less than the acid-induced net Na uptake found in polystrands (24.2 mmol/liter·min; Table II). The rapidity of Na/H exchange in polystrands may reflect the minimal extracellular diffusion delays of the preparation. Complex extracellular spaces of naturally occurring cardiac muscle tend to lessen experimentally measured transport rates of ionic fluxes (Horres and Lieberman, 1977; Atwell et al., 1979). In addition, spontaneously contracting polystrands could have an enhanced rate of  $\text{H}^+$  production and consequently require a greater rate of Na/H exchange compared with quiescent Purkinje fibers.

#### *Steady State $\text{pH}_i$ Maintenance*

Data obtained following an acid load in heart cells strongly indicate the activation of Na/H exchange during  $\text{pH}_i$  regulation. Less conclusive is whether Na/H exchange is active at normal  $\text{pH}_i$  values to participate in steady state  $\text{pH}_i$  maintenance. Data obtained from cultured heart cells in support of this include: (a) the  $\text{Na}_i$  content decreases a small but significant amount upon application of amiloride ( $10^{-3}$  M), (b) transmembrane acid extrusion rates in  $\text{HCO}_3^-$ -free solution decrease by 6.4 nmol/mg dry weight·min in amiloride ( $10^{-4}$  M) (Fig. 7), and (c) from control  $\text{pH}_i$  levels, exposure to low  $\text{Na}_o$  solutions promotes cytosolic acidification (Fig. 6). In addition,  $\text{pH}_i$  and  $\text{Na}_i$  activity decline in sheep Purkinje fibers upon application of amiloride ( $10^{-3}$  M) (Deitmer and Ellis, 1980). Thus, the evidence tentatively suggests that a small basal rate of Na/H exchange may participate in steady state  $\text{pH}_i$  maintenance in cardiac muscle. Small but detectable steady state rates of Na/H exchange also exist in neuroblastoma cells (Moolenaar et al., 1981) and *Necturus* gallbladder (Weinman and Reuss, 1982); in contrast, steady state Na/H exchange is undetectable in cultured skeletal muscle cells (Vigne et al., 1982) and thymic lymphocytes (Grinstein et al., 1984).

#### *Relative Magnitude of Na/H Exchange*

The ability of Na/H exchange to mediate rapid net Na influx can be examined by calculating net transmembrane Na flux from  $\text{Na}_i$  content changes. For example, the most rapid net Na uptake rate measured in normal  $\text{Ca}_o$  was 131 nmol/mg dry weight·min (Fig. 5). Assuming no major changes in the volume-to-surface area ratio under these conditions ( $1.06 \times 10^{-4}$  cm; Horres et al., 1977), the net transmembrane Na influx associated with Na/H exchange is 54 pmol/cm<sup>2</sup>·s. This value is 3.5 times larger than steady state ouabain-sensitive (Na/K pump-mediated) unidirectional <sup>24</sup>Na efflux (16 pmol/cm<sup>2</sup>·s; Wheeler et al., 1982) and  $\sim 25$  times the estimated steady state unidirectional Na influx through passive conductive channels (1–2 pmol/cm<sup>2</sup>·s; Wheeler, 1981). The data illustrate that stimulation of Na/H exchange in cardiac muscle can mediate remarkably rapid transmembrane Na movements.

In conclusion, a rapid, electroneutral, amiloride-sensitive Na/H exchange has been demonstrated in the polystrand preparation of cultured heart cells. The exchange is stimulated by intracellular acidification and appears to be a major pathway for pH<sub>i</sub> regulation in these cardiac muscle cells.

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