



Article Chemical Constituents of Stems and Leaves of Tagetespatula L. and Its Fingerprint

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Abstract: *Tagetespatula* L. is a widely cultivated herbal medicinal plant in China and other countries. In this study, two new 2, 3-dihydrobenzofuran glucosides (**1**, **2**) and fourteen known metabolites (**3–16**) were isolated from the stems and leaves of *T. patula* (SLT). The chemical structures of the isolated compounds were characterized comprehensively based on one- and two-dimensional NMR spectroscopy and high resolution mass spectrometry. Absolute configurations of compounds **1** and **2** were determined by ECD calculations. Compounds **1** and **2** exhibited moderate in vitro inhibitory activities against human gastric cancer cell lines (AGS) with IC₅₀ values of 41.20 µmol/L and 30.43 µmol/L, respectively. The fingerprint profiles of stems and leaves of *T. patula* with three color types of flowers (Janie Yellow Bright, Jinmen Orange, Shouyao Red and Yellow color) were established by high-performance liquid chromatography (HPLC). Ten different batches of stems and leaves were examined as follow: Shouyao Red and Yellow color (1, 2, 3), Janie Yellow Bright (4, 5, 6, 7) and Jinmen Orange (8, 9, 10). Twenty-two common peaks were identified with similarity values ranging from 0.910 to 0.977. Meanwhile, the average peak area of SLT in the three types of flowers was different and it was the highest in Janie Yellow Bright.

Keywords: Tagetespatula L.; benzofurans; flavonoids; fingerprint

1. Introduction

Tagetes genus belongs to the family Compositae (Asteraceae) and comprises about 55 different species including *Tagetespatula* Linn. (French marigold) which is an annual well-known ornamental and medicinal plant distributed worldwide [1]. *T. patula*, which first originated in Mexico, exhibits cold and drought-resistant ability and, generally, grows on mountains at an altitude of 1350 m [2].

Currently, in China, *T. patula* is a common ornamental and medicinal plant and has been cultivated throughout the country. Traditionally, *T. patula* was used to treat various diseases, such as cough, colic, constipation, diarrhea, rheumatism, and eye problems. Today, the plant is specially used as an antimicrobial, antiseptic, blood purifying, and diuretic agent [3,4]. The flowers of *T. patula* are edible and used in refreshing drinks [5]. Previous phytochemical investigations of flowers of *T. patula* were mainly focused on the flavonoids [6,7], steroids [8], and thiophenes [9]. Its roots and rhizomes contained higher contents of thiophenes [10] and benzofurans [11]. In this study, we report the isolation and structural elucidation of two new euparin glycosides and 14 known compounds (Figure 1) using different chromatographic techniques. The chemical structures of the isolated compounds

have been determined with different spectroscopic techniques including one-dimensional (1D) and two-dimensional (2D) NMR experiments, HR-ESI-MS analysis, as well as ECD. Furthermore, we established an effective fingerprint method for comparison and quality evaluation of SLT with three color types of flowers (Figure A1).

2. Results and Discussion

2.1. Structural Elucidation of the Isolated Compounds

Compound 1 was obtained from the 95% EtOH extract of the T. patula L as a white amorphous powder. The molecular formula was determined as $C_{19}H_{24}O_9$ by HR-ESI-MS showing a pseudomolecular ion peak at m/z 397.1399 [M+H]⁺ correspondent to of. The ¹H-NMR spectrum of compound **1** showed resonances of two dd peaks at δ_H 3.38 (dd, J = 8.0 and 15.5 Hz, H-3) and 3.12 (dd, J = 8.0 and 15.5 Hz, H-3) coupled with a triplet signal at δ_H 5.48 (t, J = 8.5 Hz, H-2), which are the characteristic signals of 2, 3-dihydrobenzofuran derivatives [12]. In addition, one aromatic acetyl methyl group with chemical shift of $\delta_{\rm H}$ 2.54 ppm (3H, s, H-11), two singlet aromatic protons at $\delta_{\rm H}$ 7.68 (s, H-4) and 6.27 (s, H-7) were also observed. Two downfielded signals at $\delta_{\rm H}$ 5.28 and 5.32 assignable to vinylic protons (2H, s, H-13) and two doublets at δ_H 4.22 and 4.49 (d, J = 10.0 and 12.8 Hz, H-14) were also observed. The ¹³C-NMR spectrum of compound 1 showed signals for nineteen carbons, of which eight were speculated to be from the benzofuran skeleton. The assignment of ¹³C-NMR signals of compound **1** was assisted by the analysis of the HSQC spectrum. The carbon signal at $\delta_{\rm C}$ 204.2 (C-10) and $\delta_{\rm C}$ 26.4 (C-11) indicated the presence of an acetyl moiety, while the signals at $\delta_{\rm C}$ 145.8 and $\delta_{\rm C}$ 113.9 could confirm the presence of a double bond. Analysis of the ¹³C-NMR data (Table 1) clearly indicated that compound 1 was closely related to the known compound 2, 3-dihydro-14-isobutyryloxyeuparin and the major difference was that C-14 in compound 1 ($\delta_{\rm C}$ 70.0) was shifted downfield as compared with that known compound (δC 62.3) [13]. In addition, the HMBC correlation between the signals at $\delta_{\rm H}$ 4.28 (1H, d, J = 4.3Hz, glu-1) and $\delta_{\rm C}$ 70.0 (C-14) suggested that the attachment of a sugar moiety at C-14 of aglycone (Figure 2). To determine the absolute configuration of compound 1, the ECD method was used to determine the configuration of C-2 of compound 1. ECD was calculated by the time-dependent density functional theory (TD-DFT) method at the B3LYP/6-31G+(d, p) level utilizing the conductor polarizable continuum model (CPCM) in methanol. By comparing the calculated ECD spectral data with that of experimental data of compound 1, the absolute configuration at C-2 was assigned to be R (Figure 3). Thus, the structure of compound 1 was characterized as $1-\{(2R)-2-[1-(\beta-D-glucopyranoside) vinyl]-6-hydroxy-1-benzofuran-5-yl\}-ethanone.$ It is a new compound and named as $2H\beta$, 3-dihydro-euparin-14-O- β -D-glucoside.

Position		1	2				
rosition -	δ _C	δ _H	δ _C	δ _H			
2	86.4	5.48 (1H, t, 8.5)	90.0	5.27 (1H, d, 5.0)			
3	34.4	3.12 (1H, dd, 8.0, 15.5), 3.38 (1H, dd, 8.0, 15.5)	81.0	5.33 (1H, d, 5.0)			
4	128.6	7.68 (1H, s)	133.1	8.18 (1H, s)			
5	115.0		115.8				
6	166.6		167.7				
7	98.4	6.27 (1H, s)	98.7	6.32 (1H, s)			
8	167.8		167.7				
9	120.4		121.1				
10	204.2		205.0				
11	26.4	2.54 (3H, s)	26.6	2.61 (3H, s)			
12	145.8		144.8				
13	113.9	5.32 (1H, s), 5.28 (1H, s)	114.5	5.36 (1H, s), 5.41 (1H, s)			
14	70.0	4.49 (1H, d, 10.0), 4.22 (1H, d,12.8)	63.6	4.16 (1H, d, 15.0), 4.23 (1H, d, 10.0)			

Table 1. ¹³C-NMR spectroscopic data of compounds **1** and **2** (500 MHz for ¹H and 125 MHz for ¹³C data, J in Hz, recorded in CD₃OD).

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Position 1				2							
δ _C			δ _H		δ_{H}						
	1′		103.7	4.28 (1H, d, 4.3)	105.5		4.54 (1H, d, 5.0)			
	2′		75.1	3.35	(1H, m)	75.3		3.17 (1H, dd, 5.0, 10.0)			
	3′		78.1	3.40	(1H, m)	78.4		3.36 (1H, s)			
	4'		71.7	3.21	(1H, m)	71.6		3.28 (1H, br s)			
	5′		78.0	3.13	(1H, m)	78.1		3	3.36 (1H, s)		
	6'		62.83	3.25	(2H, m)	62.8	3.73(1H	, dd, 5.0, 1	10.0) 4.00 (1	1H, dd, 2.0), 11.5)
1	0 11 10 HO 6	5 4 9 3 8 0 7	13 2 12 12 14 1	он 06'4' 1HO ^{2'3} ОДН	$ \begin{array}{c} 0 \\ 11 \\ 10_{5} \\ HO \\ 6 \\ 7 \\ 8 \\ 0 \\ 1 \end{array} $	OH 0 1' 0 5' 6'4' 3 13 2' OH 3' OH 2 12) 14 HO		R ₁ R ₂ Of		OH R4	
			1		2						
								R₁	R_2	R ₃	R₄
							3	OH	OCH.	OH	OH
							-				
	R ₁	_	R ₄		o	Н	4	ОН	н	ОН	ОН
	R1-	= <u>)</u>			HO	~~~	5	ОН	н	OAra	ОН
	í >	_//			HO-A H	3CO OH	6	ORha	н	он	ОН
	К 3						_	011			
	R.	R	R	R.	0		'	Он	п 	Оп	
	141	112	13	14	$\langle \neg$	00113	8	ОН	н	OGIC	н
12	ОСН₃	OGIc	OCH₃	CH₃	H ₃ CO H	\rangle	9	ОН	н	OAra	н
14	OCH₃	OGIc	OCH₃	н		,	10	ORha	н	ОН	н
15	н	ОСН₀	OGIC	н	HU ↑ OCH₃		11	ОН	н	OXyl	н
	-	3			13		16	OGIc	OCH₃	ОН	н

Table 1. Cont.

Figure 1. Structure of the isolated compounds from *T. patula*.



Figure 2. HMBC correlations of compounds 1 (a) and 2 (b).



Figure 3. Caculated and experimental ECD spectra of compound 1 (a) and 2 (b).

Compound 2 was isolated as a yellow crystalline amorphous powder and its molecular formula $C_{19}H_{24}O_{10}$ was determined by HR-ESI-MS showing a pseudo molecular ion peak at m/z 435.1269 $[M+Na]^+$. The ¹H-NMR spectrum of compound **2** displayed the presence of the signals of two singlet aromatic protons at $\delta_{\rm H}$ 8.18 (1H, s, H-4) and 6.30 (s, H-7), an AB type of methylenic signals at $\delta_{\rm H}$ 4.23, 4.16 (2H, d, J = 10.0 Hz and J = 15.0 Hz, H-14), one singlet acetyl group at $\delta_{\rm H}$ 2.61 (3H, s, H-11), two coupled heterocyclic protons at δ_H 5.33 (1H, d, J = 5.0 Hz, H-2) and 5.17 (1H, d, J = 5.0Hz, H-3), two germinal olefinic protons at $\delta_{\rm H}$ 5.41 and 5.36 (2H, each s, H-13), and one anomeric proton at $\delta_{\rm H}$ 4.54 (1H, d, J = 5.0 Hz, H-1'). The ¹³C-NMR spectrum of compound **2** showed nineteen carbon signals, of which eight from benzofuran skeleton, and the determination was assisted by analysis of the HSQC spectrum. Signals at $\delta_{\rm C}$ 205.0 (C-10) were assigned to a carbonyl carbon. Comparison of the ¹H and ¹³C-NMR spectrum data of compound **2** (Table 1) with known compounds 1-{(2*R**,3*S**)-3-(β-D-glucopyranosyloxy)-2,3-dihydro-2-[1-(hydroxymethyl)vinyl]-1-benzofuran-5-yl}ethanone showed their close structural relationship [14]. Differences were observed for the C-6 which resonated as an oxygenated carbon in the ¹³C-NMR spectrum. In addition to the two singlet signals of H-4 and H-7, the ¹H-NMR spectrum were found. A β -glucopyranosyl was elucidated to be attached to C-14 based on HMBC correlations of H-1' with C-13 (Table A1). The H-2 and H-3 were deduced to be in a cis conformation via NOESY correlations. The ECD data were calculated based on time-dependent density functional theory (TDDFT) The absolute configurations of C-2 and C-3 were determined as (2R, 3R) by (Figure 3). comparison of the calculated ECD spectra (Figure 3). Thus compound 2 was assigned as 1-{(2R,3R)-3-(β-D-glucopyranoside)-2,3-dihydro-2-[1-(hydroxymethyl)vinyl]-6-hydroxy-1-benzofuran-5-It is also a new compound and named as 14-hydroxy-2,3-dihydro vl}-ethanone. -euparin-3-O-β- D-glucoside.

Another fourteen known compounds were identified as patuletin (3) [15], quercetin (4) [16], quercetin-3-O- α -L-arabinopyranoside (5) [17], quercetin-7-O- α -L-rahmnoside (6) [18], kaempferol (7) [19], kaempferol-3-O- β -D-glucoside (8) [20], kaempferol-3-O- α -L-arabinopyranoside (9) [21], kaempferol-7-O- α -L-rahmnoside (10) [22], kaempferol-3-O- β -D-xylopyranoside (11) [23,24], 1- β -D-glucopyranosyl-2, 6-dimethoxy-4-propenylphenol (12) [25], syringaresinol-4'-O- β -D-glucoside (13) [26], 4-allyl-2, 6-dimethoxy phenyl glucoside (14) [27], 2-methoxy-4-(2-propenyl) phenyl- β -D-glucoside (15) [28], and 6-methoxy–kaempferol-7- β -D-glucoside (16) [29], based on the analysis of their 1D and 2D NMR, MS, as well as other spectroscopies.

2.2. HPLC Fingerprints and Similarity Analysis of the Various SLT Samples

The following ten different batches of stems and leaves of *T. patula* (SLT) were examined: Shouyao Red and Yellow color (1, 2, 3), Janie Yellow Bright (4, 5, 6, 7), and Jinmen Orange The chromatographic fingerprint of (SLT) (Figure 4) exhibited 22 distinct peaks (8, 9, 10). from the common patterns. These chromatograms were very similar apart from the differences in peak areas and lackingof common peak 26. Among these chromatograms, patuletin was chosen as the reference peak to calculate the RRT, because it has moderate retention time, a clear shape, and a high peak area. Eight common peaks (peak 12, 13, 14, 15, 16, 17, 18, and 19) were chosen as the "characteristic fingerprint peaks" to represent the characteristics of the SLT and identified as quercetin-3-O- α -L, arabinopyranoside; kaempferol-3-O- β -D, glucoside; kaempferol-3-O- α -L, arabinopyranoside; kaempferol-3-O- β -D, xylopyranoside; quercetin-7-O- α -L, rahmnoside; kaempferol-7-O- α -L, rahmnoside; patuletin; and kaempferol, respectively. The compounds were identified by comparing their retention time with the reference substances (Figure 5). The new compounds were not recognized due to low peak area. After analyzing, the result of similarity were evaluated for all the samples using to the Similarity Evaluation System for Chromatographic Fingerprint of Traditional Chinese Medicine (Version 2004A). The RPA of characteristic peaks relative to the reference peak were calculated and are shown in Table 2. The RPA data of the characteristic peaks were used to reflect the quantitative expression of the present HPLC fingerprint of the *T. patula*.

The average peak area of patuletinin Janie Yellow Bright was 222.7, followed by the Jinmen Orange type (218.4) and ShouyaoRed and Yellow color (212.8). As shown in Table 3, the comparison between the original chromatograms and the average chromatograms of all samples presented similarity values ranging from 0.910 to 0.979, denoting that the quality of all samples was reliable and stable, however, there is a slight difference in various types of samples. Different types of *T. patula* should be distinguished by variant or ecological type. The results of the precision, stability, and repeatability rates are shown as supplementary data in (Table A2) and the picture of the three types of flower of SLT are shown in (Figure A1).



Figure 4. The chromatographic fingerprint of 10 T. patula samples (S1-S10).



Figure 5. Compound **12**: Quercetin-3-O- α -L, arabinopyranoside; **13**: kaempferol-3-O- β -D, glucoside; **14**: kaempferol-3-O- α -L, arabinopyranoside; **15**: kaempferol-3-O- β -D, xylopyranoside; **16**: quercetin-7-O- α -L, rahmnoside; **17**: kaempferol-7-O- α -L, rahmnoside; **18**: patuletin; **19**: kaempferol.

Peak No.	S 1	S2	S 3	S 4	S 5	S 6	S 7	S 8	S 9	S10
4	0.425	0.416	0.940	0.425	0.243	0.468	0.594	0.130	0.712	0.468
7	0.806	0.694	1.092	0.560	0.432	0.718	0.977	0.283	0.596	0.485
11	0.824	1.674	2.878	1.901	0.707	1.756	2.523	1.001	1.740	1.166
12	0.373	1.116	1.377	0.898	0.346	0.897	1.239	0.483	0.628	0.440
patuletin	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000	1.000
20	0.388	1.841	1.934	2.137	0.683	1.966	3.057	1.223	1.626	2.995
22	0.201	1.304	1.673	1.079	0.697	1.234	2.072	0.370	0.928	0.303
24	0.289	2.620	2.942	1.918	0.721	2.037	3.036	0.369	1.102	0.558
25	0.468	0.629	1.320	1.244	0.696	0.658	0.702	0.362	1.819	0.839

Table 2. Relative peak area (RPA) of the common peaks of the *T. patula*.

Patuletin was taken as the reference peak to calculate RPA.

Table 3. Evaluation results of 10 batches of *T. patula* samples.

Similarity	S 1	S2	S 3	S 4	S 5	S 6	S 7	S 8	S 9	S10	R
S1	1										
S2	0.825	1									
S3	0.887	0.952	1								
S4	0.876	0.954	0.933	1							
S5	0.894	0.907	0.892	0.926	1						
S6	0.868	0.944	0.927	0.949	0.943	1					
S7	0.849	0.963	0.933	0.976	0.898	0.952	1				
S 8	0.884	0.882	0.874	0.922	0.946	0.939	0.897	1			
S9	0.921	0.879	0.952	0.911	0.906	0.897	0.878	0.895	1		
S10	0.785	0.845	0.825	0.872	0.858	0.884	0.841	0.928	0.855	1	
R	0.915	0.964	0.963	0.979	0.958	0.977	0.967	0.958	0.949	0.910	1

2.3. Cytotoxicity Assay

The isolated compounds from *T. patula*were evaluated for their cytotoxicity against human gastric cancer cell (AGS). The cytotoxicity assay was performed using the MTT assay method. Human gastric cancer cell (AGS) were obtained from the Cell Bank of Type Culture Collection of the Chinese Academy of Sciences, Shanghai Institute of Cell biology. Compounds **1** and **2** displayed moderate cytotoxicity against human gastric cancer cell (AGS) with IC₅₀ values of 41.20 μ mol/L and 30.43 μ mol/L, respectively, (Table 4). 5-Fluorouracil was used as a positive control and showed an IC₅₀ value of 11.79 μ mol/L.

Table 4. The human gastric cancer cell inhibit	tory activities of isolates.
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Compounds	IC ₅₀ (µmol/L)
1	41.2 (25.94–70.53)
2	30.4 (9.13-59.06)
5-Fluorouracil	11.8 (-0.63-20.78)

 IC_{50} was afforded with confidence interval (n = 3); C.I., 95% confidence interval; positive control, 5-Fluorouracil.

3. Materials and Methods

3.1. General Experimental Procedures

Optical rotations were measured with a Perkin-Elmer Model 241 polarmeter (Perkin Elmer, Inc. Waltham, MA, USA). The high-resolution electrospray ionization mass spectroscopy (HR-ESI-MS) as was acquired using a Waters Xevo Q-TOF (Waters Corporation, Milford, MA, USA). ECD spectra were recorded on a BioLogic MOS-450 spectrometer. Semipreparative HPLC was performed using a HITACHI 7100 (Hitachi Corporation, Tokyo, Japan) equipped with a YMC-Pack ODS-A, 10 × 250 mm column (YMC, Kyoto, Japan) and detections were performed with a VWD detector at 210 nm. Column

chromatography was performed using silica gel (200–300 mesh, Marine Chemical Factory, Qingdao, China). Prepared silica gel G plates (Marine Chemical Factory, Qingdao, China) were used for TLC analysis. Acetonitrile (HPLC grade) was purchased from Duksan (Anshan, Kyonggi, Korea).

3.2. Plant Material

The aerial part of *T. patula* was collected in Dalian, Liaoning province, China, on March 2016 and authenticated by Professor Bing Wang, the Liaoning University of Traditional Chinese Medicine, China.

3.3. Extraction and Isolation

The air-dried stems and leaves (15.0 Kg) of T. patula was refluxed with 60% EtOH for three times (3 \times 120 L, 2 h each) at low temperature (45 °C), then filtered and the solvent was evaporated in vacuum to furnish the crude extract. Then, it was partitioned with petroleum and ethyl acetate successively several times to give the ethyl acetate layer and H₂O layer which was subjected to column chromatography (CC) over AB-8 macroporous adsorptive resins and eluted with EtOH/H₂O (0:100, 60:40 and to 95:5). The 60% EtOH eluate and ethyl acetate layer (total 300 g) were combined to column chromatography on silica gel and eluted with CH₂CL₂/MeOH (100:0 to 0:100) to afford six fractions (Fr. A1-A6). Fr. A4 was passed over a silica gel column and eluted with CH₂CL₂/MeOH (100:0 to 0:100) to give six fractions (Fr. B1-B6). Compound 3 (27.9 mg) was further crystallized in MeOH from Fr. B3 (CH₂CL₂/MeOH, 20:1). Then, Fr. B4 (eluted with CH₂CL₂/MeOH, 10:1) was subjected to an ODS column and eluted with MeOH/H₂O. The 30% fraction from ODS column was 927 mg and gave six fractions C1-6. Fraction C1 was further purified by HPLC (eluted with MeOH/H₂O, 45:55) to give compound 2 (8 mg) and compound 16 (9 mg). Similarly, Fraction C3 was further subjected to an ODS column and gave three subfractions D1-D3. Fraction D1 was purified on an HPLC (eluted with MeOH/H₂O, 37:63) to afford compound 12 (8 mg). In the same way, compound 13 (14.8 mg) was isolated from fraction D2 by HPLC (eluted with MeOH/H₂O, 40:60), whereas compound 1 (8.2 mg), compound 14, and compound 15 (8 mg) were isolated from fraction D3 by an HPLC (eluted with MeOH/H₂O, 42:58). Finally, compound 4 (9 mg) was obtained from Fraction B6. Fr. A6 was applied to a silica gel (CH₂CL₂/MeOH from 15:1 to 0:100) to give seven subfractions E1-E7. Fraction E2 was further passed through polyamide column (CH₂CL₂/MeOH from 1:0 to 0:1) to obtain eight fractions F1-F8. Fraction F4 was further purified by ODS column equipped with HPLC eluting with MeOH/H₂O (45:55), resulting in optically pure compounds of 5 (16 mg), 6 (18 mg), 8 (34 mg), 11 (10 mg). In the same manner, 7 (12 mg), 9 (10 mg), 10 (20 mg) were isolated from fraction F5.

3.4. Acid Hydrolysis of Compounds 1 and 2

Compounds 1 and 2 (2 mg each) were hydrolyzed with 10% HCL-dioxane (1:1, 1 mL) for 4 h at 80 °C in a water bath. The mixture was neutralized with Ag2CO3 and extracted with CH2CL2. After the aqueous layer was condensed, both of them were examined by HPLC and compared with authentic samples. Analytical HPLC was performed on Agilent ZORBAX NH2 column (5 μ m, 4.6 mm × 150 mm) with isocratic elution using MeCN-H₂O (4:1) for 40 min at a flow rate 1 mL/min. The peaks were detected with evaporative light scattering detector and D-glucose (Sigma, St. Louis., MO, USA) showed the retention time at 9.277 min.

3.5. Chromatographic Conditions

The HPLC fingerprint analysis was conducted on an Agilent HPLC system with a diode array detector (DAD) and Thermo C18 column ($4.6 \times 250 \text{ mm}$, 5 µm) maintained at 30 °C. The mobile phases were acetonitrile (A) and 0.1% phosphoric acid in deionized water (B) (v/v) and a flowrate of 1.0 mL/min was utilized. The gradient elution program was as follows: 0–5 min, 10% to 17% A; 5–15 min, 17% to 17% A; 15–29 min, 17% to 57% A; 29–36 min, 57% to 80% A; 36–43 min, 80% to 80% A; An injection volume of 20 µL was used for each run. The absorption spectra of the samples

were recorded in the range of 190–400 nm and the detection wavelength was set at 254 nm for the establishment of fingerprints.

3.6. MTT Assay

The cytotoxicity of compounds isolated from the T. patula, against human gastric cancer cell (AGS) was assessed by MTT colorimetric method. Cells in DMEM medium supplemented with 10% fetal bovine serum (FBS) were incubated in a 96-well plate in the presence of different concentrations of test compounds (2.5, 5, 10, 20, 40, 80 μ mol/L) at 37 °C in a 5% CO2 incubator for 24 h. Formazan crystals resolved with DMSO and then the ODS value at 492 nm of each well was measured. 5-FU were included as a positive control.

4. Conclusions

In this work, two new compounds (1 and 2) and 14 known compounds were isolated and identified. Compounds 1 and 2 showed moderate cytotoxicity against human gastric cancer cell lines (AGS). Furthermore, chromatographic fingerprint analysis was performed for different types of leave and stem extracts to evaluate the quality and differentiate various types of SLT.

Author Contributions: Y.-M.W. and D.-Q.D. planned and designed the experiment; Y.-M.W., X.-K.R., M.R., M.Y., Q.C., D.-Q.D., A.M.M., T.-G.K., and D.-C.C. conducted the experiments; Y.W. analyze the data and prepared the figures; Y.-M.W. and D.-Q.D. wrote the paper; A.M.M. modified the paper; all authors agreed to the final version of this manuscript.

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Conflicts of Interest: The authors declared no conflict of interest.

Appendix A

HMBC spectral data of compound 1 and compound 2

Table A1. HMBC spectroscopic data of compounds **1** and **2** (500 MHz for ¹H and 125 MHz for ¹³C Data, J in Hz, recorded in CD3OD).

Desition	-	1	2			
rosition	$\delta_{\rm H}$	НМВС	δ _H	НМВС	NOESY	
2	5.48 (1H, t, 8.5)	C-3,C-9,C-12,C-13,C-14,	5.17 (1H, d, 5.0)	C-12,C-13,C-14	H-3,H-14, H-1'	
3	3.12 (1H, dd, 8.0, 15.5), 3.38 (1H, dd, 8.0, 15.5)	C-2,C-4,C-8,C-9,C-12	5.33 (1H, d, 5.0)	C-1′,C-8	H-3, H-4, H-13, H-14	
4	7.68 (1H, s)	C-3, C-5, C-6, C-7, C-8, C-9	8.18 (1H, s)	C-8,C-10	H-11, H-1′	
5						
6						
7	6.27 (1H, s)	C-5,C-6,C-8,C-9	6.32 (1H, s)	C-5,C-6,C-9		
8						
9						
10						
11	2.54 (3H, s)	C-5, C-10	2.61 (3H, s)	C-5,C10		
12						
13	5.32 (1H, s), 5.28 (1H, s)	C-2, C-12, C-14	5.36 (1H, s), 5.41 (1H, s)	C-2,C12,C-14	H-14	
14	4.49 (1H, d, 10.0), 4.22 (1H, d, 12.8)	C-1′, C-2, C-12, C-13	4.16 (1H, d, 15.0), 4.23 (1H, d, 10.0)	C2,C-12,C-13	H-2, H-13, H-1'	
1'	4.28 (1H, d, 4.3)	C-14, C-2'	4.54 (1H, d, 5.0)	C-3	H-3, H-4, H-14	
2′	3.35 (1H, m)	C-1′, C-3′	3.17 (1H, dd, 5.0, 10.0)	C-1′, C-3′		
3′	3.40 (1H, m)	C-2′, C-4′	3.36 (1H, s)	C-2′		
4'	3.21 (1H, m)	C-3′, C-5′	3.28 (1H, br s)	C-3′, C-6′		
5'	3.13 (1H, m)	C-4′, C-5′, C-6′	3.36 (1H, s)	C-4′		
6′	3.25 (2H, m)	C-5′	3.73(1H, dd, 5.0, 10.0) 4.00 (1H, dd, 2.0, 11.5)	C-4′, C-5′	H-4′	

Method Validation

The precision was measured by analyzing the five replicates of the sample solutions (Sample 1) continuously within one day. The relative standard deviation (RSD) values of relative retention time (RRT) and peak area (RPA) of the fourteen markers did not exceed 0.49% and 6.22%, respectively. The stability test was determined and the RSD values of RRT and RPA were below 0.52% and 6.25%. Moreover, five independently solutions of the same sample (Sample 1) were analyzed to verify the repeatability. The RSD values of RRT and RPA were less than 0.59% and 4.70%, respectively. The results of the precision, stability, and repeatability rates are shown in Table A2.

Table A2. Analytical results of precision, stability, and repeatability rates of fourteen components in the *T. patula*.

Peak No.]	RSD of RPA (%	6)	RSD of RRT (%)				
	Precision	Stability	Repeatability	Precision	Stability	Repeatability		
5	4.397	4.769	3.401	0.492	0.458	0.246		
6	3.504	5.883	1.574	0.321	0.238	0.258		
11	3.343	6.253	3.652	0.375	0.522	0.586		
15	4.840	4.976	4.701	0.125	0.131	0.098		
18(S)	0.000	0.000	0.000	0.000	0.000	0.000		
25	6.224	1.351	3.806	0.064	0.049	0.052		

The pictures of three types of flowers from stems and leaves of T. patula



Figure A1. (a) Janie Yellow Bright, (b) Jinmen Orange, (c) Shouyao Red and Yellow color.

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Sample availability: The samples of the compounds are not available from the authors.



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