SCIENTIFIC REPORTS

Received: 29 March 2016 Accepted: 11 July 2016 Published: 08 August 2016

OPEN Rapid discovery and identification of anti-inflammatory constituents from traditional Chinese medicine formula by activity index, LC-MS, and NMR

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The traditional activity-guided approach has the shortcoming of low accuracy and efficiency in discovering active compounds from TCM. In this work, an approach was developed by integrating activity index (AI), liquid chromatography – mass spectrometry (LC-MS), and nuclear magnetic resonance (NMR) to rapidly predict and identify the potential active constituents from TCM. This approach was used to discover and identify the anti-inflammatory constituents from a TCM formula, Gui-Zhi-Jia-Shao-Yao-Tang (GZJSYT). The AI results indicated that, among the 903 constituents detected in GZJSYT by LC-MS, 61 constituents with higher AI values were very likely to have anti-inflammatory activities. And eight potential active constituents of them were isolated and validated to have significant inhibitory effects against NO production on LPS-induced RAW 264.7 cell model. Among them, glycyrrhisoflavone (836), glisoflavanone (893) and isoangustone A (902) were reported to have anti-inflammatory effects for the first time. The proposed approach could be generally applicable for rapid and high efficient discovery of anti-inflammatory constituents from other TCM formulae or natural products.

Traditional Chinese medicine (TCM) is a valuable resource for new drug and lead compound discovery. The most common way to find active compounds from TCM is activity-guided approach. The whole extract of TCM was systematically separated to several fractions, which then were carried out on activity screening. The compounds in active fractions were isolated and evaluated pharmacologically to find the active compounds. Generally one fraction composes of more than ten constituents. Since the activities of constituents are unknown before biological testing, tremendous amount of time and efforts were spent on isolation of inactive compounds, which significantly hampers rapid discovery of compounds of interest, e.g. those with appreciable pharmacological effects. To overcome this drawback, in our previous study¹, activity index (AI) was proposed to predict the contribution of each constituent detected in TCM formula to the anti-inflammatory effect. The constituents with positive AI values might be active, while the ones with negative or near zero AI values might be inactive or had weak activity. This method had been successfully used to find several anti-inflammatory constituents in Ju-Zhi-Jiang-Tang. However, only the constituents detected in the whole extract of Ju-Zhi-Jiang-Tang were evaluated. The trace constituents that were detected only in the fractions were not involved, which might omit some constituents that had significant activity though their contents in the formula were low.

Liquid chromatography - mass spectrometry (LC-MS) is a rapid, sensitive, and high efficient technique, and has been increasingly used for the identification of constituents in TCM formulae. But the constituents in TCM are complex, and their concentrations in TCM differ by several orders of magnitude. Generally 50 to more than 100 constituents are detected in the whole extract of TCM formula by LC-MS^{2,3}. Preliminary fractionation could enrich the concentrations of trace constituents, which would be helpful for detecting trace constituents in TCM formula by LC-MS analysis^{4,5}. Thus the constituents in the whole extract and fractions could be evaluated on their AIs. Semi-preparative LC guided by LC-MS was used to prepare the potential active constituents for activity validation, and the structures of potential active constituents could be identified by nuclear magnetic resonance (NMR)³.

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Figure 1. The total ion current (TIC) chromatograms of GZJSYT in negative and positive modes by LC-IT-MS.

Gui-Zhi-Jia-Shao-Yao-Tang (GZJSYT), a well-known TCM formula from Shang-Han-Lun, is comprised of *Cinnamomi ramulus* (Guizhi in Chinese), *Paeoniae radix alba* (Baishao in chinese), *Zingiberis rhizoma recens* (Shengjiang in Chinese), *Jujubae fructus* (Dazao in Chinese), and *Glycyrrhizae radix et rhizoma preparata cum melle* (Zhigancao in Chinese). It is often used to treat gastralgia, dysentery, peptic ulcer, tuberculous peritonitis, prosopalgia, and restless legs syndrome in clinic. Pharmacology research indicated that the mechanism of antidiarrheal effect of GZJSYT might be the inhibition of excessively accelerated small intestinal movement and the acetylcholine released by parasympathetic nerves⁶. In addition, it was reported that GZJSYT exerted therapeutic effect against peptic ulcer by alleviating gastro spasm⁷. However, there has been no research on the comprehensive identification of chemical profile of GZJSYT, and the active ingredients are still unclear, which would hamper its broad application⁸.

In this work, the approach based on AI, LC-MS, and NMR was used to discover and identify the anti-inflammatory constituents from GZJSYT. The whole extract and fractions of GZJSYT were analyzed by LC-Q-TOF-MS and LC-IT-MS, and a total number of 903 constituents were detected. In addition, the anti-inflammatory effects of the whole extract and fractions were evaluated on LPS-induced NO production in RAW 264.7 macrophages. The AI method was used to evaluate the contributions of 903 constituents to the anti-inflammatory effects of the whole extract and fractions. Finally, eight potential active constituents were prepared, and their anti-inflammatory activities were validated.

Results

Characterization of chemical constituents in the extract of GZJSYT by LC-MS. In order to separate as many as possible constituents in one chromatographic separation, chromatographic separation conditions, including the elution gradient and flow rate were optimized to obtain good peak capacity. Finally, the optimum chromatographic conditions were used for LC-MS analysis. The extract of GZJSYT was analyzed by LC-Q-TOF-MS and LC-IT-MS method in negative and positive ion modes to acquire the accurate mass and MSⁿ information that was indispensable for profiling the constituents. The total ions current (TIC) chromatograms obtained by LC-IT-MS are shown in Fig. 1. It was found that most constituents were detected in negative ion mode, except several compounds could only be detected in positive ion mode (Table 1). A total number of 72 constituents, including 34 ones from *Glycyrrhizae radix et rhizoma preparata cummelle*, 29 ones from *Paeoniae radix alba*, ten ones from *Jujubae fructus*, six ones from *Cinnamomiramulus*, and three ones from *Zingiberis rhizoma recens*, were detected in the whole extract of GZJSYT. Among them, 58 constituents were identified by comparing with the retention times and MSⁿ data of the reference standards or by referring to the literatures. As shown in

| No | t _R (min) | Identification | Molecular Formula | Detected ions (<i>m/z</i>) | Error (ppm) | Source | | |
|-----|----------------------|---|---|-------------------------------------|-------------|--------|--|--|
| 1 | 1.15 | Malic acid | C ₄ H ₆ O ₅ | 133.0154 [M-H] ⁻ | 8.7 | GPJZC | | |
| 2 | 1.25 | Unknown | - | — | — | G | | |
| 3 | 1.53 | Unknown | — | — | — | PJ | | |
| 4* | 1.59 | Citric acid | C ₆ H ₈ O ₇ | 191.0208 [M-H] ⁻ | 5.6 | GPJZC | | |
| 5 | 1.71 | Desbenzoyl paeoniflorin | $C_{16}H_{24}O_{10}$ | 421.1344 [M+HCOO] ⁻ | -1.8 | Р | | |
| 6* | 2.07 | Succinic acid | C ₆ H ₄ O ₆ | — | _ | GPJZC | | |
| 7 | 2.15 | Isomaltodebenzoyl paeoniflorin | $C_{22}H_{34}O_{15}$ | 583.1876 [M+HCOO] ⁻ | -0.6 | Р | | |
| 8* | 3.26 | Gallic acid | C ₇ H ₆ O ₅ | 169.0157 [M-H] ⁻ | 8.6 | Р | | |
| 9 | 3.95 | 1-О- β -D-Gucopyranosyl paeonisuffrone | $C_{16}H_{24}O_9$ | 405.1404 [M+HCOO] ⁻ | 0.4 | Р | | |
| 10 | 4.16 | Glucogallin C ₁₃ H ₁₆ O ₁₀ 331.0669 [M–H] ⁻ | | -0.5 | Р | | | |
| 11 | 5.57 | 6-Galloylsucrose or 6'-Galloylsucrose or 1'-Galloylsucrose | C ₁₉ H ₂₆ O ₁₅ | 493.1197 [M-H] ⁻ | 0.4 | Р | | |
| 12 | 7.18 | 6- O - β -D-glucopyranosyl lactinolide | C ₁₆ H ₂₆ O ₉ | 407.1559 [M+HCOO] ⁻ | 0 | Р | | |
| 13 | 8.46 | 6′-O-Galloyldesbenzoyl paeoniflorin | C ₂₃ H ₂₈ O ₁₄ | 527.1403 [M-H]- | 0.6 | Р | | |
| 14 | 8.73 | (p-Hydroxybenzyl)malonic acid $C_{10}H_{10}O_5$ 209.0465 [M-H] ⁻ | | 209.0465 [M-H]- | 4.6 | G | | |
| 15 | 9.67 | Zizybeoside I or its isomer | C ₁₉ H ₂₈ O ₁₁ | 477.1613 [M+HCOO]- | -0.1 | J | | |
| 16 | 9.74 | Unknown | C ₁₆ H ₂₆ O ₈ | 391.1611 [M+HCOO] ⁻ | 0.3 | Р | | |
| 17 | 10.32 | Zizybeoside I or its isomer | C ₁₉ H ₂₈ O ₁₁ | 477.1603 [M+HCOO] ⁻ | -2.2 | J | | |
| 18 | 10.32 | Oxypaeoniflorin | C ₂₃ H ₂₈ O ₁₂ | 541.1560 [M+HCOO] ⁻ | -0.5 | Р | | |
| 19 | 11.09 | Zizybeoside II | C ₂₅ H ₃₈ O ₁₆ | 639.2147 [M+HCOO] ⁻ | 0.8 | J | | |
| 20 | 11.81 | Unknown | _ | _ | — | Р | | |
| 21 | 11.85 | 6'-O-β-Glucopyranosyl albiflorin or its isomer | $C_{29}H_{38}O_{16}$ | 687.2152 [M+HCOO] ⁻ | 1.5 | Р | | |
| 22 | 12.51 | Unknown | $C_{35}H_{48}O_{21}$ | 849.2685 [M+HCOO] ⁻ | 1.8 | Р | | |
| 23 | 12.69 | 6'-О- β -Glucopyranosyl albiflor in or its isomer | $C_{29}H_{38}O_{16}$ | 687.2152 [M+HCOO] ⁻ | 1.5 | Р | | |
| 24 | 12.95 | Isomaltopaeoniflorin | C ₂₉ H ₃₈ O ₁₆ | 687.2146 [M+HCOO] ⁻ | 0.6 | Р | | |
| 25 | 13.19 | Albiflorin | C ₂₃ H ₂₈ O ₁₁ | 525.1616 [M+HCOO] ⁻ | 0.4 | Р | | |
| 26 | 13.53 | 6'-O-β-Glucopyranosyl albiflorin or its isomer | C ₂₉ H ₃₈ O ₁₆ | 687.2152 [M+HCOO] ⁻ | 1.5 | Р | | |
| 27* | 14.21 | Paeoniflorin | C ₂₃ H ₂₈ O ₁₁ | 525.1619 [M+HCOO] ⁻ | 1.0 | Р | | |
| 28* | 16.7 | Liquiritigenin-7-O- β -D-apiofuranosyl-4'-O- β -D-glucopyranoside | C ₂₆ H ₃₀ O ₁₃ | 549.1620 [M-H]- | 1.2 | G | | |
| 29* | 16.75 | Liquiritin | C ₂₁ H ₂₂ O ₉ | 419.1338 [M+H]+ | 0.3 | G | | |
| 30* | 17.06 | Liquiritin apioside | C ₂₆ H ₃₀ O ₁₃ | 549.1625 [M-H] ⁻ | 2.1 | G | | |
| 31* | 17.09 | Liquiritigenin | C ₁₅ H ₁₂ O ₄ | 257.0812 [M+H]+ | 1.4 | G | | |
| 32 | 17.93 | Galloylalbiflorin Apigenin 6-C-β-xylopyranosyl-8-C-(6‴-O-(3-hydroxy-3- | C ₃₀ H ₃₂ O ₁₅ | 631.1674 [M-H]- | 0.9 | P | | |
| 34 | 18.15 | methylglutaroyl)- β -glucopyranoside) | C ₃₂ H ₃₆ O ₁₈ | 707.1842 [M-H] | 1.9 | G | | |
| 35 | 20.55 | Albiflorin R ₁ | C ₂₂ H ₂₈ O ₁₁ | 525.1608 | -1.1 | P | | |
| 36* | 21.53 | Glvcvroside | C ₂₇ H ₃₀ O ₁₃ | [M+HCOO] ⁻ 607.1684 | 2.6 | G | | |
| 37* | 22.17 | Ononin | C ₂₂ H ₂₂ O ₉ | [M+HCOO] ⁻ 475.1254 | 1.7 | G | | |
| 38 | 23.54 | Sacranoside A or its isomer | C ₂₁ H ₃₄ O ₁₀ | [M + HCOO] ⁻ 491.2143 | 1.8 | Р | | |
| 39 | 24.53 | Sacranoside A or its isomer | C ₂₁ H ₃₄ O ₁₀ | 491.2143 | 1.8 | Р | | |
| 40 | 26.25 | Macedonoside A isomer | CHO | 837 3939 [M_H]- | 3.0 | G | | |
| 41 | 26.23 | 24-Hydroxy-licorice-saponin A | C42 ¹¹ 62 ^O 17 | 999.4474 [M_H]- | 3.2 | G | | |
| 42 | 27.08 | 24-Hydroxy-licorice-saponin A ₃ isomer | C48H72O22 | 999.4489 [M-H]- | 4.7 | G | | |
| Co | ntinued | | | | | | | |

| No | t _R (min) | Identification | Molecular Formula | Detected ions (m/z) | Error (ppm) | Source |
|--------|----------------------|---|--|---|-------------|--------|
| 43 | 27.47 | 22-Hydroxy-licorice-saponin G ₂ | C ₄₈ H ₇₂ O ₂₁ | 853.3895 [M-H] ⁻ | 3.2 | G |
| 44 | 27.99 | Benzoylpaeoniflorin | C ₃₀ H ₃₂ O ₁₂ | 629.1891 [M+HCOO] ⁻ | 2.4 | Р |
| 45* | 28.4 | Licorice saponin A ₃ | C48H72O21 | 983.4535 [M-H] ⁻ | 4.2 | G |
| 46 | 28.62 | Albiflorin R ₃ C ₃₀ H ₃₂ O ₁₂ | | 629.189 [M+HCOO] ⁻ | 2.3 | Р |
| 47 | 29.53 | Yunganoside K ₁ or its isomer | C48H72O21 | 983.4535 [M-H] ⁻ | 3.2 | G |
| 48^* | 29.84 | Macedonoside A | C42H62O17 | 837.3951 [M-H] ⁻ | 4.4 | G |
| 49 | 30.16 | Uralsaponin R/S C ₄₈ H ₇₄ O ₂₀ 969.4723 [M–H] ⁻ | | 2.3 | G | |
| 50 | 30.60 | Yunganoside K ₂ or its isomer | Yunganoside K ₂ or its isomer C ₄₂ H ₆₂ O ₁₇ 837.3950 [M–H] ⁻ | | 4.3 | G |
| 51 | 30.66 | Uralsaponin P C ₄₂ H ₆₄ O ₁₆ 823.4147 [M–H] ⁻ | | 3.1 | G | |
| 52 | 31.4 | Unknown | C ₁₆ H ₃₂ O ₅ | 303.2181 [M-H] ⁻ | 1.3 | J |
| 53* | 31.99 | Licorice saponin G ₂ C ₄₂ H ₆₂ O ₁₇ 837.3950 [M-H] ⁻ | | 837.3950 [M-H] ⁻ | 4.3 | G |
| 54 | 32.77 | Yunganoside L ₁ /J ₁ | C48H72O20 | H ₇₂ O ₂₀ 967.4576 [M-H] ⁻ | | G |
| 55 | 32.96 | Yunganoside K ₂ or its isomer | C42H62O17 | 837.3954 [M-H] ⁻ | 4.7 | G |
| 56 | 33.22 | Yunganoside G ₁ | C48H74O21 | 985.4688 [M-H] ⁻ | 3.9 | G |
| 57 | 33.5 | Yunganoside K ₂ or its isomer | omer C ₄₂ H ₆₂ O ₁₇ 837.3954 [M–H] ⁻ | | 4.5 | G |
| 58 | 33.79 | Yunganoside G ₂ or its isomer | C42H64O17 | 839.4105 [M-H] ⁻ | 4.1 | G |
| 59* | 34.03 | Uralsaponin A | Uralsaponin A C ₄₂ H ₆₂ O ₁₆ 821.4012 [M–H] ⁻ | | 5.7 | G |
| 60 | 34.04 | Unknown | Unknown C ₃₀ H ₄₄ O ₃ 453.3362 [M+H] ⁺ | | 0.3 | J |
| 61 | 35.13 | Yunganoside L ₁ /J ₁ | C ₄₈ H ₇₂ O ₂₀ 967.4590 [M-H] ⁻ | | 4.7 | G |
| 62 | 35.37 | Albiziasaponin B C ₄₈ H ₇₄ O ₂₀ 969.4744 [M–H] ⁻ | | 4.5 | G | |
| 63* | 35.7 | Licorice saponin H ₂ C ₄₂ H ₆₂ O ₁₆ 821.4009 [M–H] ⁻ | | 5.3 | G | |
| 64 | 35.7 | Unknown C ₃₀ H ₄₄ O ₃ 453.3364 [M+H] ⁺ | | 0.2 | J | |
| 65 | 36.27 | Glycyrrhizic acid isomer | С ₄₂ H ₆₂ O ₁₆ 821.4009 [М-Н] ⁻ | | 5.3 | G |
| 66 | 38.09 | Licorice saponin C_2 or its isomer | C42H62O15 | 805.4056 [M-H] ⁻ | 5.0 | G |
| 67 | 38.59 | Actinopolysporin B | C ₁₆ H ₃₀ O ₄ | 285.2077 [M-H] ⁻ | _ | — |
| 68 | 39.21 | Unknown | — | — | _ | С |
| 69 | 39.48 | Unknown | _ | — | _ | С |
| 70 | 39.66 | Unknown | — | — | _ | _ |
| 71 | 39.82 | Unknown | — | — | — | С |
| 72 | 40.06 | Unknown | _ | _ | _ | Р |

Table 1. The constituents detected in the whole extract of GZJSYT by LC-Q-TOF-MS. t_R: retention time; G: Glycyrrhizae radix et rhizoma preparata cum melle; P: Paeoniae radix alba; J: Jujubae fructus; C: Cinnamomi ramulus; Z: Zingiberis rhizoma recens. *Compared with reference standards.

Fig. 2A, the identified constituents mainly included monoterpene glycosides, triterpene saponins, flavonoid and its glycosides, gallic acid derivatives, and organic acids.

A total number of 16 monoterpene glycosides in the root of *Paeoniae radix alba* were observed in negative ion mode. Monoterpene glycosides in the formula mostly possess a "cage-like" pinane skeleton with the hemiketal group or lactone group^{9,10}. In the present work, the monoterpene with similar pinane skeleton of paeoniflorin was defined as paeoniflorin type (PT), and that with similar skeleton of albiflorin was defined as albiflorin type (AT).

Compounds 7, 13, 24, 27, and 44 were identified as PTs. Their fragmentation pathways were summarized in Fig. 3A. The characteristic fragment ions of PTs were the base peak ion $[M - HCHO - H]^-$ in MS² spectra, in which the neutral loss of HCHO (30 Da) was speculatively related to pinane skeleton of PTs due to the instability of hemiketal group. In MS³ spectra, compounds 7 and 13 generated base peak ions at m/z 489 $[M - HCHO - H_2O - H]^-$ and m/z 479 $[M - HCHO - H_2O - H]^-$, respectively. But compounds 24, 27, and 44 produced base peak ions $[M - HCHO - BA - H]^-$ in MS³ spectra. The characteristic fragment ion at m/z 165 was observed in MS³ or MS⁴ spectra of compounds 7, 24, 27, and 44. For compound 44, m/z 165 [M-HCHO-BA-Benzoyl-Glc-H]⁻ was observed as a fragment ion in MS³ spectrum, while for compounds 7, 24, and 27, m/z 165 was generated as base peak ions in MS⁴ spectra by the loss of one glucosyl residue (162 Da) (for compound 27) or two glucosyl residues (324 Da) (for compounds 7 and 24) from the precursor ions. For compound 13, the base peak ion in MS⁴ spectrum was at m/z 271, which was produced by the loss of pinane skeleton and a glucosyl segment (42 Da) from the precursor ions. Finally, compound 27 was identified as paeoniflorin by comparing with the reference standard. Compounds 7, 13, 24, and 44 were supposed to be isomaltodebenzoyl paeoniflorin, 6'-O-galloyl desbenzoyl paeoniflorin, isomaltopaeoniflorin, and benzoylpaeoniflorin, respectively, by referring to the literatures^{11,12}.

Compounds 5 and 18 both showed adduct ions $[M + HCOO]^-$ in MS spectra, as well as base peak ions $[M - H]^-$ in MS² spectra. The characteristic fragment ions, including base peak ions $[M - HCHO - H]^-$ in MS³ spectra and the fragment ion at m/z 165 in MS⁴ spectra, manifested the typical fragmentation pattern of PTs, *i.e.*, successive loss of unstable hemiketal group and the substituent group from pinane skeleton. Besides, the



Figure 2. The chemical structures of identified constituents in the whole extract of GZJSYT (**A**) and eight potential active constituents isolated by semi-preparative LC (**B**).

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fragment ion at m/z 327 [M – HCHO – 138 – H][–] in MS³ spectrum of compound 18 was detected. Here, the loss of 138 Da was deduced to be *p*-hydroxybenzoic acid. Base on the MSⁿ fragmentation behaviour mentioned above, compounds 5 and 18 were ascribed to be desbenzoyl paeoniflorin and oxypaeoniflorin, respectively.

Compounds 21, 23, 25, and 26 were identified as ATs with the lactone pinane skeleton. Their fragmentation pathways are shown in Fig. 3B. They all showed adduct ions $[M + HCOO]^-$ in MS spectra, and produced deprotonated molecules $[M - H]^-$ as base peak ions in MS² spectra. Compounds 21, 23, and 26 shared the identical molecular formula $C_{29}H_{38}O_{16}$, and gave the similar fragmentation behavior as compound 25. The four compounds all generated the typical base peak ions $[M - BA - H]^-$ in MS³ spectra and $[M - BA - HCHO - H]^-$ in MS⁴ spectra. According to the MSⁿ data, compound 25 was assigned as albiflorin. Nevertheless, compounds 21, 23, and 26 could not be confirmed only by MSⁿ data, they were tentatively deduced to be 6'-O- β -Glucopyranosyl albiflorin or its isomers.

Compounds 32 and 46 were also identified as ATs. Compound 32, containing a galloyl at C-6', produced the ion at m/z 613 $[M - H_2O - H]^-$ as base peak ion in MS² spectrum, instead of m/z 509 $[M - BA - H]^-$. In MS³ spectrum, the base peak ion was at m/z 271 [Gallyol + 120 - H]⁻ by the loss of monoterpenoid skeleton. The reason might be galloyl group was easier to ionize than monoterpenoid skeleton. For compound 46, base peak ion was at m/z 583 $[M - H]^-$ in MS² spectrum, as well as m/z 461 $[M - BA - H]^-$ in MS³ spectrum. By comparing with the data in literatures^{13,14}, compounds 32 and 46 were tentatively deduced as galloylalbiflorin and albiflorin R₃, respectively.

Compounds 9 and 35 possessed another "cage-like" pinane skeleton that was different from PTs and ATs. Compound 9 exhibited adduct ion at m/z 405 [M + HCOO]⁻ in MS spectrum and base peak ion at m/z 359 [M – H]⁻ in MS² spectrum. Furthermore, its MS³ spectrum showed base peak ion at m/z 197 [M – Glc – H]⁻, as well as the "cage-like" pinane skeleton ion at m/z 179 [M – Glucose – H]⁻ and fragment ion at m/z 161 [M – Glucose – H₂O – H]⁻. Therefore, compound 9 was ascribed to be 1-*O*- β -D-Glucopyranosyl paeonisuf-frone¹⁵. For compound 35, it showed adduct ion at m/z 525 [M + HCOO]⁻ in MS spectrum and base peak ion at m/z 479 [M – H]⁻ in MS² spectrum. In MS³ spectrum, base peak ion at m/z 449 [M – HCHO – H]⁻ was yielded, together with the fragment ions at m/z 461 [M – H₂O – H]⁻ and m/z 357 [M – BA – H]⁻, indicating the existence of "cage-like" pinane skeleton. Additionally, base peak ion at m/z 327 in MS⁴ spectrum corresponded to the loss of benzoic acid from the precursor ion at m/z 449 [M – HCHO – H]⁻. The predominant fragmentation pathway was similar to that in the literature¹⁵, so compound 35 was identified as albiforin R₁.



Figure 3. The proposed fragmentation pathway of PTs (A) and ATs (B) in negative ion mode.

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Compound 12 produced adduct ion at m/z 407 [M + HCOO]⁻ in MS spectrum and base peak ion at m/z 361 [M - H]⁻ in MS² spectrum. The fragment ions at m/z 199 [M - Glc - H]⁻, m/z 163 [M - Glc - 2H₂O - H]⁻, and m/z 155 [M - Glc - CO₂ - H]⁻ were observed in MS³ spectrum. The above MSⁿ data was accordance with that in the literature¹⁵, so compound 12 was presumed as 6-O- β -D-glucopyranosyl lactinolide.

Triterpene saponins were the main constituents of *Glycyrrhizae radix et rhizoma*, and performed a variety of pharmacological activities^{16–18}. In this work, 22 oleanane triterpene saponins were identified in the extract of GZJSYT. These oleanane-type triterpene saponins were analyzed in negative ion mode. Their fragment patterns were similar to that reported in our previous works^{19,20}. They were grouped into four types according to base peak ions in MS² spectra: (a) base peak ion at *m*/*z* 351, deriving from a GluA-GluA unit of C-3 of A ring for compounds 40, 43, 48, 50, 53, 55, 57, 58, 59, 63, 65, and 66; (b) base peak ion at *m*/*z* 497 [2GluA + Rha – H][–], demonstrating the presence of Rha-GluA-GluA group at C-3 of A ring for compounds 47, 54, 56, 61, and 62; (c) base peak ion at *m*/*z* 485 [Aglycone – H][–] for compounds 49 and 51. In MS³ spectra, they produced base peak ion at *m*/*z* 355 via the cleavage of C18-C19 bond and C17-C22 bond at E-ring; (d) base peak ion [M – Glc – H][–], indicating the existence of an ester glycoside at C-30 of E ring for compounds 41, 42, and 45. And they generated base peak ions at *m*/*z* 351 [2GluA – H][–] in MS³ spectra.

Seven flavonoids were identified in the extract of GZJSYT, which were all from *Glycyrrhizae radix et rhizoma*. By comparison with the reference standards, compounds 28, 29, 30, 31, 36, and 37 were explicitly characterized as liquiritigenin-7-*O*- β -D-apiofuranosyl -4'-*O*- β -D-glucopyranoside, liquiritin, liquiritin apioside, liquiritigenin, glycyroside, and ononin, respectively. Compound 33 showed *m/z* 707 [M – H]⁻ in MS spectrum, and gave the characteristic fragment ions at *m/z* 605 [M – 102 – H]⁻, *m/z* 563 [M – 144 – H]⁻, *m/z* 473 $[M - 144 - 90 - H]^-$, and m/z 443 $[M - 144 - 120 - H]^-$ (base peak) in MS² spectrum. Here, the loss of 102 Da and 144 Da indicated the presence of 3-hydroxy-3-methylglutaryl, and 90 Da and 120 Da was from glucosyl segment. In MS³ spectrum, the ions at m/z 425 $[M - 144 - 120 - H_2O - H]^-$, 383 $[M - 144 - 120 - 60 - H]^-$, 353 $[M - 144 - 120 - 90 - H]^-$, and 323 $[M - 144 - 120 - 120 - H]^-$ were detected. So, compound 33 was characterized as apigenin 6-*C*- β -xylopyranosyl-8-*C*-(6'''-O-(3-hydroxy-3-methylglutaroyl)- β -glucopyranoside) by comparing the above MSⁿ data with that in the literature¹⁹.

Four organic acids (compounds 1, 4, 8, and 14) and two organic acid glycosides (compounds 10 and 11) were identified. The characteristic neutral loss of H_2O , HCOOH (or CH_3COOH), and CO_2 occurred in MS^2 or MS^3 spectra of four organic acids. Among them, compounds 4 and 8 were confirmed as citric acid and gallic acid, respectively, by comparing with the reference standards. Compounds 1 (m/z 133) and 14 (m/z 209) were inferred as malic acid and (p-hydroxybenzyl)malonic acid, respectively. Compound 10 exhibited deprotonated molecule $[M - H]^-$ at m/z 331. It produced base peak ion at m/z 271 $[M - 60 - H]^-$ in MS^2 spectrum, as well as the fragment ions at m/z 241 $[M - 90 - H]^-$ and m/z 169 $[M - 162 - H]^-$. In MS^3 spectrum, it generated base peak ion at m/z 169 $[M - H - 60 - 102]^-$. Finally, compound 10 was deduced as glucogallin according to the above MS^n data. Compound 11 gave deprotonated molecule $[M - H]^-$ at m/z 493. The base peak ion at m/z 313 $[M - 180 - H]^-$ in MS^2 spectrum indicated the existence of glucose or fructose in the molecular structure. In MS^3 spectrum, the base peak ion at m/z 169 $[M - 180 - 144 - H]^-$ (gallic acid) and fragment ion at m/z 223 $[M - 180 - 90 - H]^-$ were observed. Based on the MS^n data mentioned above, compound 11 was deduced to be galloylsucrose. However, the linkage position of saccharide groups was not clear only by MS^n data.

Besides the above mentioned types of compounds, three other types of compounds (15, 17, and 19) from *Jujubae fructus* were identified. Both compounds 15 and 17 showed molecular formulae of $C_{19}H_{28}O_{11}$ and gave adduct ions at $m/z 477 [M + HCOO]^-$ in MS spectra. Though they had identical base peak ions, *i.e.*, deprotonated molecules at $m/z 431 [M - H]^-$ in MS² spectra and base peak ions at $m/z 269 [M - Glc - H]^-$ in MS³ spectra, they had some different fragment ions in MS² and MS³ spectra. By comparing the MSⁿ data with those in the literature²¹, compounds 15 and 17 were deduced to be zizybeoside I or its isomers. Compound 19 gave adduct ion at $m/z 639 [M + HCOO]^-$ in MS spectrum. It produced base peak ion at $m/z 593 [M - H]^-$ in MS³ spectrum, and base peak at $m/z 431 [M - Glc - H]^-$ and fragment ion at $m/z 269 [M - 2Glc - H]^-$ in MS³ spectrum. Finally, compound 19 was identified as zizybeoside II²¹.

Prediction of the potential anti-inflammatory constituents by AI method. The whole extract of GZJSYT was separated to 64 fractions by macroporus resin column chromatography and preparative LC. The fractions (except fractions D20, E19, and E20 due to little amount) were analyzed by LC-IT-MS to detect the chemical constituents in these fractions, and to acquire the peak area values of the constituents. As a result, a total number of 903 constituents were detected in the fractions. The anti-inflammatory activities of the whole extract and fractions of GZJSYT were evaluated on LPS-induced RAW 264.7 macrophages at corresponding concentrations without cytotoxicity. As shown in Fig. 4, the whole extract of GZJSYT (A01) exhibited lower inhibition rate against NO production at about 26%. But fractions B03, B04, C19, D04–D08, D11–D19, E01, and E03–E18 performed better inhibition rate than indomethacin (positive control). In particular, fractions B04, D07, D13, D14, E07–E15, E17, and E18 exerted inhibition rate against NO production at more than 80%.

The AIs of the constituents in GZJSYT were calculated according to the equation (1). As shown in Fig. 5, 61 constituents had higher AIs, which existed mainly in fractions E15, E16, E17, and E18. The greater the AI value is, the more likely the constituent is active¹. Therefore, the 61 constituents were speculated to have anti-inflammatory activities.

Isolation and identification of eight potential active constituents. In order to verify the predicted results of AI method, eight constituents with higher AI values were isolated from *Glycyrrhizae radix et rhizoma preparata cum melle* with semi-preparative LC guided by LC-MS. Based on LC-MS, ¹H and ¹³C NMR (Supplementary information), they were identified as glycycoumarin (833), glycyrrhisoflavone (836), euchrestaflavanone A (838), licoisoflavone A (840), licochalcone A (883), licoisoflavone B (890), glisoflavanone (893), and isoangustone A (902). Their chemical structures are shown in Fig. 2B.

Validation of anti-inflammatory activities of eight potential active constituents. The anti-inflammatory activities of eight constituents were determined using LPS-induced RAW 264.7 macrophages at a series of concentrations without cytotoxicity. The results are shown in Fig. 6. Glycycoumarin (833), glycyrrhiso-flavone (836), euchrestaflavanone A (838), licoisoflavone A (840), licochalcone A (883), licoisoflavone B (890), gli-soflavanone(893), and isoangustone A (902) possessed IC_{50} values of $30.5 \pm 1.1 \,\mu$ M, $11.7 \pm 1.0 \,\mu$ M, $29.0 \pm 1.9 \,\mu$ M, $19.1 \pm 1.6 \,\mu$ M, $9.2 \pm 1.4 \,\mu$ M, $18.6 \pm 1.6 \,\mu$ M, $19.7 \pm 1.4 \,\mu$ M, and $10.8 \pm 1.9 \,\mu$ M, respectively. Compared with indomethacin (positive control, IC_{50} : $31.6 \pm 2.0 \,\mu$ M)²², eight compounds all had high inhibition rates against NO production. Among them, glycycoumarin (833), euchrestaflavanone A (838), licoisoflavone A (840), licochalcone A (883) and licoisoflavone B (890) had been reported to have anti-inflammatory activity²³⁻²⁵, but glycyrrhisoflavone (836), glisoflavanone (893), and isoangustone A (902) were reported to have anti-inflammatory activity for the first time in this study.

Discussion

The chemical profile of GZJSYT was firstly reported in this work. The study of fragmentation rules of MSⁿ spectra for different types of constituents was critical for the rapid identification of their chemical structures by LC-MS. The fragment pathway of monoterpenoids, the main constituents in GZJSYT, was investigated in detail.



Figure 4. The inhibition rate against NO production of the whole extract of GZJSYT and its fractions at the corresponding concentrations without cytotoxicity.





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It was found that paeoniflorin type monoterpenoids mainly eliminated HCHO due to the unstable hemiketal group of pinane skeleton in MS^n spectra, and then fragmented the substituent group R_2 , such as BA and Glc, from the mother skeleton under higher collision energy to form typical fragment ion at m/z 165. However, albiflorin type monoterpenoids firstly fragmented the substituent group R_2 , and then lost the lactone group of pinane skeleton under higher collision energy, which was different from the fragmentation behavior of paeoniflorin type monoterpenoids.

Prefraction was not only helpful to decrease the chemical complex of TCM formulae, but also necessary to find the trace constituents that had significant activity. In this work, 903 constituents were detected in GZJSYT fractions while only 72 constituents in the whole extract of GZJSYT. Considering that the identification of chemical structures of those 903 constituents in the fractions was time-consuming even by LC-MS, only the potential active constituents were isolated and identified unambiguously by NMR.

Among the 61 fractions, 40 fractions had anti-inflammatory effects. Many constituents with positive AI values might have contributions to the anti-inflammatory effects. The low inhibition rate against NO production of A01 might be due to the weak activity of constituents with higher contents. However, in fractions, the trace active constituents were enriched, so many fractions had higher inhibition rate against NO production.



Figure 6. The inhibition rates against NO production of eight isolated compounds in different concentrations. 833: glycycoumarin, 836: glycyrrhisoflavone, 838: euchrestaflavanone A, 840: licoisoflavone A, 883: licochalcone A, 890: licoisoflavone B, 893: glisoflavanone, and 902: isoangustone A.

These active fractions contained hundreds of constituents. It was not easy to find the active constituents rapidly without the guidance of AI method. It is generally considered that the main constituents in active fraction may be active, but sometime it is not the case. Here we take the active fractions D05 as an example. Fractions D05 exhibited 53% of inhibition rate against NO production. The LC-MS chromatogram of D05 is shown in Fig. 7. It was found that paeoniflorin (27) was the main constituent of fraction D05. It was easy to think that paeoniflorin (27) might be active. As shown in Table 2, the AI value of paeoniflorin (27) was -0.18. According to our previous work¹, negative AI value indicated that the constituents might have no anti-inflammatory activity or weak activity. In fact, paeoniflorin (27) was proved experimentally to have no anti-inflammatory activity at 50 μ M, only had weak anti-inflammatory effect at 200 μ M (NO inhibition rate of 38%)²⁶. The result indicated the contribution of paeoniflorin (27) to the anti-inflammatory of fraction D05 was weak. Therefore, activity index could predict the probability of the constituents being active to some extent.

The AI results indicated that 61 constituents with higher AI values might have anti-inflammatory activities, and most of them were from *Glycyrrhizae radix etrhizoma*. The structures of eight active constituents were confirmed by NMR, which included one coumarin and seven flavonoids. Interestingly, most of their structures contained isopentene groups, which might be important for their significant anti-inflammatory activities.

Conclusion

In summary, an approach based on AI, LC-MS, and NMR was developed to rapidly discover and identify the anti-inflammatory constituents from GZJSYT in this study. A total number of 903 constituents were detected in the whole extract and fractions by LC-MS. The AI predicted results showed that 61 constituents with higher AIs were very likely to have anti-inflammatory activities. Eight potential active constituents of them were validated to have significant anti-inflammatory activity in LPS-induced RAW 265.7 macrophages. And more importantly, three of them were first reported to have anti-inflammatory activity in this work.

Calculating activity index of constituents requirs the global analysis of the fractions from TCM formulae, not just the active fractions, which is different from the traditional activity-guided method that only focuses in active fractions. The results indicated that the approach was efficient and feasible. This approach would be generally applicable for the rapid discovery of anti-inflammatory constituents from other TCMs or medical plants.



Figure 7. The total ion chromatograms (TICs) of the fraction D05 by LC-IT-MS in negative ion mode.

| No | t _R (min) | Detected <i>m/z</i> | Compounds | AIs |
|--------|----------------------|---------------------|--|-------|
| 183 | 10.07 | 635.2ª | 1,2,6-Tri-O-galloyl-β-D-glucose | 0.76 |
| 216 | 11.14 | 635.3ª | 1,3,6-Tri-O-galloyl-β-D-glucose | 0.38 |
| 25 | 12.29 | 525.3 ^b | Albiflorin | 0.00 |
| 27^* | 14.15 | 525.2 ^b | Peoniflorin | -0.18 |
| 277 | 14.55 | 405.2ª | Unknown | 0.20 |
| 287 | 14.82 | 787.1ª | 1,2,3,6-Tetra-O-galloyl-β-D-glucose | 0.50 |
| 28^* | 15.15 | 595.0 ^b | $\label{eq:liquiv} Liquiritigen in -7 - O - \beta - D - a pi of uranosyl - 4' - O - \beta - D - glucopyranoside$ | 0.46 |
| 320 | 15.81 | 786.9ª | 1,2,4,6-Tetra-O-galloyl-β-D-glucose | 0.75 |
| 352 | 17.15 | 939.2ª | Pentagalloylglucose | 1.14 |
| 35 | 18.82 | 525.1 ^b | Albiflorin R ₁ | 0.07 |
| 423 | 19.84 | 549.3ª | Licuraside | 0.49 |
| 425 | 20.33 | 549.3ª | 2′-O-[β -D-Apiofuranosyl-(1→2)- β -D-glucopyranosyl] isoliquiritigenin | 0.54 |
| 567 | 24.5 | 999.7ª | 24-Hydroxy-licorice- saponin A ₃ isomer | 0.48 |
| 622 | 26.82 | 837.5ª | Macedonoside A isomer | -0.22 |
| 695 | 27.84 | 969.6ª | Albiziasaponin B isomer | 0.66 |
| 700 | 28.53 | 823.5ª | Uralsaponin P isomer | 0.72 |

Table 2. Activity indexes of compounds in the fraction D05 of GZJSYT. ${}^{a}[M-H]^{-}$. ${}^{b}[M+HCOO]^{-}$. *Compared with reference standards.

Methods

Materials and reagents. Paeoniae radix alba, Cinnamomi ramulus, Jujubae fructus and Glycyrrhizae radix etrhizoma preparata cum melle, were obtained from Zhejiang Chinese Medical University Medical Pieces Co., Ltd. (Hangzhou, China), and Zingiberis rhizome recens was purchased from the local market.

The sources of reference standards were as follows. Gallic acid and citric acid were purchased from Sinopharm Chemical Reagent Co., Ltd. (Shanghai, China). Succinic acid was acquired from Sangon Biotech (Shanghai) Co., Ltd. (Shanghai, China). Liquiritin apioside and liquiritigenin were obtained from Shanghai Winherb Medical Technology Co., Ltd. (Shanghai, China). Paeoniflorin was obtained from Aladdin Reagent Co., Ltd. (Shanghai, China). Liquiritin, ononin, glycyroside, liquiritigenin-7-O- β -D-apiofuranosyl-4'-O- β -D-glucopyranoside, macedonoside A, uralsaponin A, licorice saponin G₂, licorice saponin H₂, and licorice saponin A₃ were isolated from *Glycyrrhizae radix et rhizoma* in our laboratory, and their structures were identified by UV, MS, and NMR. The purity of the each reference standard was greater than 96%.

HPLC grade of acetonitrile and methanol were purchased from Merck (Darmstadt, Germany). Formic acid (HPLC grade) was obtained from ROE Scientific Inc. (Newark, DE, USA). Deionized water was purified by Milli-Q system from Millipore (Molsheim, France). Acetonitrile for preparative HPLC was purchased from Amethyst Chemicals (J&KScientific Ltd., Beijing, China). Anhydrous ethanol and 95% ethanol were purchased from Shanghai Lingfeng Chemical Co., Ltd. (Shanghai, China) and Zhejiang Changqing Chemical Co., Ltd. (Hangzhou, China), respectively. D101 macroporous resin was bought from Tianjin Haiguang Chemical Co. Ltd. (Tianjin, China).

Dulbecco's modified Eagle's medium (DMEM), Fetal Bovine Serum (FBS), trypsin-EDTA and the penicillin-streptomycin were purchased from Gibico BRL (Grand Island, NY, USA). Lipopolysaccharides (LPS), dimethylsulfoxide (DMSO) and indomethacin (purity \geq 99%) were acquired from Sigma-Aldrich (St. Louis, MO, USA).

Preparation of sample and standard solutions. According to the recipe of GZJSYT in Shang-Han-Lun, *Cinnamomi ramulus* (27.3 g), *Paeoniae radix alba* (54.6 g), *Zingiberis rhizoma recens* (27.3 g), *Jujubae fructus* (72.7 g), and *Glycyrrhizae radix et rhizoma preparata cum melle* (18.2 g) were immersed in 1.2 L water (6-fold) overnight. Then, they were extracted twice for 1 h of each time and the filtrates were combined. Subsequently, the combined solution was concentrated under reduced pressure using a rotary evaporator, and the residue was dissolved in appropriate volume of water. Finally, the solution was analyzed by LC-MS after it was filtered through 0.22 µm filter membrane.

The fractions of GZJSYT were prepared as previously reported¹. Firstly, the extract of GZJSYT was loaded onto a glass column (4.6 cm \times 30 cm) packed with the D101 macroporous resin. The column was rinsed with H₂O first to get fraction B01, and then in turn eluted with 20% ethanol, 40% ethanol, 95% ethanol to give fractions B02, B03, and B04, respectively. Fractions B02, B03 and B04 were subject to preparative HPLC to obtain new subfractions. Subfractions preparation was carried out on a 1200 series LC system (Agilent, Palo Alto, USA) and a Zorbax SB-C18 column (21.2 mm \times 250 mm, 7 µm, Agilent) with the mobile phase consisted of A (H₂O) and B (acetonitrile) at a flow rate of 10 mL/min. Subfractions were collected each three minutes from the fifth minute to the sixty-fifth minute. The elute gradient used for fraction B02 was as follows: 0 min, 3% B; 50 min, 20% B; 64 min, 50% B; 67 min, 100% B; 70 min, 100% B. The obtained subfractions were named as fractions C01-C20. Fraction B03 was applied to preparative HPLC to yield fractions D01-D20 with the following gradient elution: 0 min, 15% B; 40 min, 30% B; 55 min, 45% B; 60 min, 80% B; 64 min, 100% B; 70 min, 100% B. The elute gradient used for fractions by provide the following gradient elution: 0 min, 15% B; 40 min, 30% B; 20 min, 30% B; 40 min, 45% B; 55 min, 90% B; 60 min, 100% B. The elute gradient used for fractions were concentrated and freeze-dried. Finally, the samples were used for LC-IT-MS and evaluated on anti-inflammatory activity except the fractions D20, E19, E20 due to the very low amount of them.

Reference standards were dissolved in methanol at the concentration of about 0.1 mg/mL and stored at 4 °C. They were filtered through 0.22 μ m filter membrane before use.

LC-IT-MS conditions. HPLC was performed on an Agilent 1100 HPLC system (Agilent, Waldbronn, Germany) containing a binary pump, an auto-sampler, a column compartment and a photo-diodearray (DAD) detector. Separation was carried out on a Zorbax SB-C18 Rapid Resolution HT column ($4.6 \text{ mm} \times 50 \text{ mm}$, $1.8 \mu \text{m}$, Agilent) at a flow rate of 0.6 mL/min and column temperature of 30 °C. The mobile phase was prepared using two solutions, A and B, where A was 0.05% formic acid in water, and B was acetonitrile. The gradient elution was optimized as follows: 0 min, 3% B; 35 min, 41% B; 40 min, 100% B; 50 min 100% B. The injection volume was $3 \mu \text{L}$. The DAD detector scanned range 190–400 nm, and the samples were detected at 254 and 280 nm.

IT-MS analysis was operated on a Finnigan LCQ Deca $\hat{X}P^{plus}$ ion trap mass spectrometer (Thermo Finnigan, San Jose, CA, USA) with ESI interface and an ion trap mass (IT-MS) analyzer. The analysis was performed on both negative (ESI⁻) and positive (ESI⁺) ion modes under the following operation parameters: auxiliary/sweep gas (N₂), 20 L/min; sheath gas (N₂), 60 L/min; collision gas (He); ESI spray voltage, -3 kV (ESI⁻) and 4 kV (ESI⁺); capillary voltage, -15 V (ESI⁻) and 19 V (ESI⁺); capillary temperature, 350 °C; tube lens offset voltage, -30 V (ESI⁻) and 25 V (ESI⁺); scan range, *m/z* 100–1500. The collision energy was set at 35–45% to give appropriate fragmentations.

LC-Q-TOF-MS conditions. LC-Q-TOF-MS analysis was manipulated on an AcquityTM ultra performance LC system (Waters Corp., Milford, MA, USA) coupled with a high resolution Triple TOF 5600⁺ (AB SCIEX, Framingham, MA, USA), which was equipped with an ESI source. The chromatographic separation conditions were the same as that in LC-IT-MS. The sample was analyzed in both positive and negative mode, and the parameters in the source were set as follows: curtion gas (CUR), 30 psi; ion source GS1, 50 psi; ion source GS2, 50 psi; source temperature, 550 °C (ESI⁻) and 600 °C (ESI⁺); ionspray voltage floating (ISVF), -4500 V(ESI⁻) and 5500 V (ESI⁺); collision energy (CE), 10 V; declustering potential (DP), ± 100 V; ion release delay (IRD) at 67; ion release width (IRW) at 25. The mass range was *m/z* 100–1500. The accurate mass acquired were processed by means of the elemental composition calculator incorporated in the PeakView[®] software (AB SCIEX).

Isolation and identification of potential active constituents. The *Glycyrrhizae radix et rhizoma preparata cum melle* (5 kg) was extracted with 95% ethanol (8-fold) for 1 h at the first time. Then the extraction was repeated with 95% ethanol (6-fold) for 1 h. The two extracts were mixed together, and condensed to appropriate concentration for separation experiments. The extract was loaded on a glass column ($10 \text{ cm} \times 250 \text{ cm}$) packed with the D101 macroporous resin, and eluted with H₂O, 40% ethanol and 95% ethanol, respectively. Effluent of 95% ethanol was collected and condensed to appropriate volume. Subsequently, the sample was further separated by medium-pressure column chromatography ($10 \text{ cm} \times 50 \text{ cm}$) loaded with ODS. The mobile phases consisted of H₂O (A) and methanol (B) and the flow speed at 60 mL/min. Isocratic elution with 55% B was carried out, and the eluate was collected and dried for semi-preparation.

Semi-preparation was carried out on an Agilent 1100 equipped with a degasser, a quaternary pump, an auto-sampler, a column compartment and a UV detector. The Zorbax SB-C18 column (Semi-Preparative, 9.4 mm \times 250 mm, 5 μ m, Agilent) at a column temperature of 30 °C was utilized and eluted with the gradient profile of A (water) and B (acetonitrile) at a flow rate of 3.0 mL/min. the gradient elution was: 0 min, 15% B; 40 min, 34% B; 80 min. 37% B; 120 min, 40% B; 150 min, 45% B; 160 min, 48% B; 205 min, 54% B; 206 min, 100%

B; 216 min, 100% B. The injection volume was 100 μ L and UV detection wavelength at 230 nm. NMR experiments of isolated compounds were dissolved in methanol- d_4 and performed on a Bruker Avance III 500 NMR spectrometer (¹H: 500 MHz, ¹³C: 125 MHz; Bruker Corporation, Billerica, MA, USA).

Evaluation of the anti-inflammatory activity. The murine macrophage cell line RAW 264.7 (the Type Culture Collection of the Chinese Academy of Sciences, Shanghai, China) was cultured in 90% Dulbecco's Modification Eagle's Medium (DMEM) with 4.5 g/L glucose, L-glutamine and sodium pyrurate supplemented with 10% inactivated Fetal Bovine Serum (FBS) and 1% 10000 U/mL of penicillin, 10000 μ g/mL of streptomycin at 37 °C in a 5% CO₂ humidified atmosphere.

To determine the cytotoxicity of each constituent, RAW 264.7 cells (5×10^4 cells/mL, 100μ L/well) plated in 96 well plates were cultured for 24 h at 37 °C. The samples dissolved in DMSO were diluted with DMEM supplemented with 10% inactivated FBS, penicillin G (100 U/mL) and streptomycin (100μ g/mL) before co-incubated with the cells for 24 h. Then, MTT solution (100μ L, 0.5 mg/mL in per well) was added and incubated for 4 h. The solution in per well was drawn and the formazan crystals were adequately dissolved in DMSO (100μ L/well) by shaking for 10 min at 37 °C. The absorbance at 580 nm was measured in a microplate reader (Bio-Tek ELX800, Winooski, VT, USA). Fresh culture medium was used as a blank in every experiment. The experiments were implemented for three times in parallel.

For the determination of the anti-inflammatory activity, RAW 264.7 cells were seeded into a 96-well plate at a density of 2×10^4 cells/well for 24 h. Then, the fresh culture medium containing samples and LPS (200 ng/mL) were added to each well (140μ L/well) and further incubated for 24 h. Subsequently, 100μ L of supernatant from each well was mixed with 100μ L of Griess reagent (0.5% sulfanilamide and 0.05% naphthylene diamide dihydrochloride in 2.5% H₃PO₄) in a separate 96-well plate. After an incubation of 10 min, the absorbance of NO was determined at 535 nm with a microplate reader. Indomethacin (50μ M) was adopted as a positive control and fresh culture medium was used as the blank in all experiments. The experiments were carried out for three times in parallel.

Calculation of AI values of constituents. The AI values of constituents were calculated using mathematical equations reported in the previous work¹.

$$AIj = \sum_{i}^{m} Is, \, iAn, \, i, \, jCi \tag{1}$$

AI_j standed for AI value of constituent *j*.

 $I_{s,i}$ represented standardized value of NO inhibition rate of fraction *i*, it was defined as Is, $i = \frac{Ii - \overline{I}}{S}$. I_i standed for value of NO inhibition rate of fraction *i*; \overline{I} represented mean value of NO inhibition rate of all fractions; *S* standed for standard deviation of NO inhibition rate of all fractions.

standed for standard deviation of NO infinition rate of an fractions. $A_{n,i,j}$ represented normalized value of peak area of constituent *j* in fraction *I*, it was defined as An, *i*. $j = \frac{Ai, j}{\sum_{i=1}^{n} Ai, j}$ $A_{i,i}$ was peak area of constituent *j* in fraction *i*; *m* was the numbers of fractions in the experiment.

 C_{i}^{j} weight of the anti-inflammation concentration of fraction *i*, it was defined as $C_i = c_{max}/c_i$. c_{max} was the highest value of the anti-inflammation concentrations of fractions, and c_i was the anti-inflammation concentrations of fraction *i*.

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Acknowledgements

The work is supported by the National S&T Major Project (2013ZX09402203).

Author Contributions

S.W. conceived and designed the experiments, H.W. and Y.L. performed the experiments, S.W. and H.W. wrote the manuscript, Y.W. gave help to experiments, X.F. reviewed the manuscript, Y.C. gave help to designation of manuscript.

Additional Information

Supplementary information accompanies this paper at http://www.nature.com/srep

Competing financial interests: The authors declare no competing financial interests.

How to cite this article: Wang, S. *et al.* Rapid discovery and identification of anti-inflammatory constituents from traditional Chinese medicine formula by activity index, LC-MS, and NMR. *Sci. Rep.* **6**, 31000; doi: 10.1038/srep31000 (2016).

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