## Drug Design

# Synthesis and Evaluation of a 2,11-Cembranoid-Inspired Library 

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#### Abstract

The 2,11-cembranoid family of natural products has been used as inspiration for the synthesis of a structurally simplified, functionally diverse library of octahydroisobenzo-furan-based compounds designed to augment a typical medicinal chemistry library screen. Ring-closing metathesis, lactonisation and $\mathrm{Sml}_{2}$-mediated methods were exemplified and applied to the installation of a third ring to mimic the


#### Abstract

nine-membered ring of the 2,11-cembranoids. The library was assessed for aqueous solubility and permeability, with a chemical-space analysis performed for comparison to the family of cembranoid natural products and a sample set of a screening library. Preliminary investigations in cancer cells showed that the simpler scaffolds could recapitulate the reported anti-migratory activity of the natural products.


## Introduction

The 2,11-cembranoids belong to a structurally complex family of natural products isolated from the Octocorallia species, consisting of a polyoxygenated 2,11-cyclised diterpenoid core scaffold. ${ }^{[1]}$ The structures are further subcategorised into cladiellins 1, briarellins 2 and asbestinins 3 (Figure 1). Due to their struc-

Cladiellin 1


Briarellin 2


Asbestinin 3

Figure 1. Cladiellin, briarellin and asbestinin core scaffolds. Carbon atoms bearing an asterisk are commonly oxygenated or unsaturated within the class.
tural complexity and broad ranging biological activities, the members of these structural class have been and continue to be attractive targets for total synthesis. ${ }^{[2-17]}$ By necessity, such approaches require multi-step routes, with a particular chal-

[^0]lenge being the introduction of the nine-membered cyclic ether ring and the installation of the specific complex substitution patterns associated with each natural product. Methods used to form such a ring include Nozaki-Hiyama-Kishi coupling, ${ }^{[7,8,11,14]}$ Claisen ring-expansion, ${ }^{[6]}$ ring-closing metathesis (RCM) $)^{[4,9,13,15]}$ diazo ketone cyclisations ${ }^{[10,16]}$ and oxidative ringexpansion. ${ }^{[17]}$ A number of these total synthesis routes have been shown to afford access to more than one member of the natural product family ${ }^{[5,7,8,10,11,13,17]}$ allowing the preparation of small libraries of the natural products. The broad range of biological activities of the cembranoid family has been reviewed, ${ }^{[1]}$ and has been found to encompass anticancer, anti-inflammatory, antiviral and antibacterial properties, amongst others.

Natural products have been a great source of inspiration in drug discovery, being used either in their own right as drugs or as the starting point for drug discovery projects. ${ }^{[18]}$ Libraries of natural products, natural-product-inspired or other structurally complex scaffolds have been shown to be a rich source of hits in phenotypic screens, ${ }^{[19]}$ which probe phenotypes arising from biological manipulation with small molecules in cells or small organisms (e.g., zebra fish embryos ${ }^{[20]}$ ). As biological assay screening tools they frequently occupy a complementary chemical space to commercially available synthetic libraries, providing novelty from both a structural and intellectual property perspective. ${ }^{[21]}$ Those natural products found to display interesting biological activities can, however, be stereochemically complex molecules, thus presenting significant synthetic challenges regarding analogue synthesis. Numerous approaches have been taken to overcome such challenges, in particular di-versity-oriented synthesis (DOS), ${ }^{[19,22]}$ biology-oriented synthesis (BIOS) ${ }^{[23]}$ and lead-oriented synthesis. ${ }^{[24]}$ A principal challenge is to retain structural features of the natural products that differentiate them from chemical space explored by traditional small molecule screening libraries, and may be associated with the bioactivities of the natural products, while simplifying the
structures to give easier synthetic access and control of the physicochemical properties.
Within the context of drug discovery, numerous metrics have been suggested as design aids in attempts to predict the likelihood of drug-likeness, ${ }^{[25]}$ one of the most prevalent being Lipinski's rule-of-five. ${ }^{[26]}$ The central tenet of many of these metrics is to predict the ability of a compound to dissolve in aqueous media and to pass through a lipid membrane, both properties that may also be readily measured. To be useful for cell- or organism-based biology, compounds must be sufficiently cell permeable and water soluble. It is therefore desirable to design new libraries around scaffolds where these properties are intrinsic.

Here, we present the synthesis of a 2,11-cembranoid-inspired small library of compounds, designed to retain the biological activities of the natural product class within structurally simplified library analogues, while displaying desirable lead-like properties. The cembranoid class of natural products has demonstrated multiple biological activities in cells and parasites, ${ }^{[1]}$ making the core scaffold an ideal starting point to make libraries of cell-permeable compounds. It was envisaged that the octahydroisobenzofuran core, with and without the third fused medium ring from the 2,11-cembranoid motif, could be used as a privileged structure. ${ }^{[28]}$ It was anticipated that the library of compounds produced from such a privileged core would be able to recapitulate some of the biological activities observed in the natural product class, while reducing the synthetic complexity. The main synthetic challenge, therefore, was to find quick and concise routes to introduce substituent variations around the chosen core and methods to install a third ring system to mimic the nine-membered ring of the 2,11-cembranoids.

From the many successful synthetic strategies exemplified for the 2,11-cembranoid natural products, we chose to explore a Diels-Alder reaction to construct the [6,5]-bicyclic core of the library ${ }^{[6,7]}$ We anticipated a rapid and flexible entry to the octahydroisobenzofuran scaffold through a Diels-Alder cyclisation, with the potential to vary the substituent patterns through choice of starting materials or manipulation of intermediates. Such an approach would provide a platform to investigate both the functionalization of the bicycle as well as the installation of a medium or larger ring equivalent to the nine-membered ring of the natural products. The retrosynthetic ap-


Scheme 1. Library design and retrosynthetic analysis.
proach is outlined in Scheme 1. From (S)-5-(hydroxymethyl)fur-an-2 2 H )-one (9), a Diels-Alder reaction with a range of dienes was envisaged to install the $[6,5]$ core of compound 7 , which can be further manipulated to produce bicyclic analogues of the general structure 5, or further cyclised to tricyclics such as compound 4. Following the generation of the library, aqueous solubility, permeability and structural diversity in relation to inhouse screening libraries were assessed. In addition, the ability to recapitulate some of the biological activities of the more complex natural products with simpler structures from the library was demonstrated.

## Results and Discussion

Protection of the commercially available alcohol 9 with tert-butyl(chloro)diphenylsilane (TBDPSCI) gave the lactone 8 in $87 \%$ yield; chiral HPLC confirmed the presence of a single enantiomer (Scheme 2). Reaction of the silyl ether 8 with butadiene in


Scheme 2. Reagents and conditions: a) TBDPSCl, imidazole, DMF, $0^{\circ} \mathrm{C}-\mathrm{RT}$, 2.5 h ; b) butadiene, $\mathrm{TfN}(\mathrm{AlMeCl})_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 60^{\circ} \mathrm{C}$; c) DIBAL-H, $\mathrm{CH}_{2} \mathrm{Cl}_{2},-78^{\circ} \mathrm{C}$; d) allyITMS (TMS = trimethylsilyl), $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2},-78^{\circ} \mathrm{C}$; e) tetrabutylammonium fluoride (TBAF), $0^{\circ} \mathrm{C}$; f) acid chloride, pyridine or $\mathrm{Et}_{3} \mathrm{~N} ; \mathrm{g}$ ) Grubbs II, 0.0005 m $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{RT} ; \mathrm{h}\right) \mathrm{Pd} / \mathrm{C}(5 \mathrm{wt} \%), \mathrm{H}_{2}, \mathrm{EtOAc}(\mathrm{Ac}=$ acetyl), RT.

[^1]which in turn was reacted with allyltrimethylsilane ${ }^{[31,32]}$ to give compound 12 as a single enantiomer. The exclusive cis selectivity from reaction of the bicyclic lactol 11 contrasts with literature examples of monocyclic lactols, where the trans isomer predominates from an intermolecular allyltrimethylsilane addition. ${ }^{[30]}$ The observed difference can be rationalised by looking at a proposed conformation of the oxonium intermediate 13, where the least hindered approach of the nucleophile leads to the cis product 12.

Deprotection of silyl ether 12 using TBAF gave the alcohol 14 in $82 \%$ yield, which was esterified to form esters of various chain lengths, that is, compounds 15, 16, 17 and 18. Following considerable optimisation, ring-closing metathesis (RCM) cyclisation using the Grubbs II catalyst at high dilution in dichloromethane gave the tricyclic compounds 20, 21 and 22. The nine-membered lactone 19 could not be formed, whereas the ten-membered lactone 20 was produced in low yield. The alkene conformation of compound $\mathbf{2 0}$ was presumed to be cis due to the strain of a trans double bond in such a ten-membered lactone. The larger, less strained eleven- and twelvemembered lactones were more readily constructed. The alkene conformation of compound 21 was undetermined due to overlapping NMR signals, whereas compound 22 was assigned to be trans by analogy to the crystal-structure conformations of compounds 40 and 42 (see below). By using a similar reaction sequence, but reducing the alkene moiety of compound 10 earlier in the route, the cyclohexane analogues 26,27 and 28 were also formed. The structure of compound 26 was confirmed by X-ray crystallography (see the Supporting Information).

As lactone 19 could not be formed, speculated to be due to the transoid nature of the ester moiety in compound 15 adding to the strain in forming a nine-membered ring, synthesis of the ether analogue 30 was attempted (Scheme 3). Alkyla-


Scheme 3. Reagents and conditions: a) NaH, allylbromide; b) Grubbs II, $0.0005 \mathrm{~m}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, RT. $0 \%$.
tion of alcohol 14 with allyl bromide gave the cyclisation precursor 29. However, no RCM product could be observed under the above-described optimised conditions, highlighting the difficulty in forming the strained nine-membered ring. Previous RCM approaches to nine-membered ring formation in the context of 2,11-cembranoid synthesis have met with mixed success, ${ }^{[3,4,15,33]}$ and appear highly substrate dependent.

By using the alkene functionality of lactone 10 as a point of derivatisation, dihydroxylation of the alkene from the least sterically crowded face using catalytic osmium tetroxide (OsEnCat) followed by dimethylacetal protection gave the tricyclic compound 31 in $52 \%$ yield over two steps (Scheme 4). By using a similar sequence to that shown in Scheme 2, lactone


Scheme 4. Reagents and conditions: a) OsEnCat ( $0.5 \mathrm{~mol} \%$ ), N -methylmorpholine $N$-oxide (NMO); b) 2,2-dimethoxypropane, pyridinium tosylate (PPTS); c) DIBAL-H, $-78^{\circ} \mathrm{C}$; d) allyITMS, $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2},-78^{\circ} \mathrm{C}$; e) TBAF; f) hex-5enoyl chloride, pyridine (py), $\mathrm{CH}_{2} \mathrm{Cl}_{2} ;$ g) Grubbs II, $0.0005 \mathrm{M}_{\mathrm{CH}_{2} \mathrm{Cl}_{2}}$; h) HCl , THF.

31 was reduced to lactol 32, reacted with allyltrimethylsilane in the presence of $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ and deprotected to give alcohol 33. Esterification, ring-closing metathesis and acetal deprotection with acid gave the diol 36.

Analogue synthesis by changing the diene component was also demonstrated (Scheme 5). The use of 2-methylbutadiene


Scheme 5. Reagents and conditions: a) $\mathrm{TfN}(\mathrm{AlMeCl})_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 60^{\circ} \mathrm{C}, 100 \%$; b) $\mathrm{BH}_{3} \cdot \mathrm{THF}$, then $\mathrm{H}_{2} \mathrm{O}_{2}, \mathrm{H}_{2} \mathrm{O} 55 \%$.
gave an 85:15 inseparable mix of methyl regioisomers (major isomer 37 shown). The major 5-methyl component was isolated following oxidative hydroborylation to give the alcohol 38. As above, the alcohol 38 was in turn converted to the lactone 39. As further exemplification of diene variations in the DielsAlder reaction, cyclohexa-1,3-diene and cyclopentadiene were used to make the tetracyclic compounds 40, 41 and 42. Confirmation of the structure of compounds 40 and 42 was achieved by X-ray crystallography (see the Supporting Information). During the synthesis of compound 42, it was found necessary to reduce the alkene derived from the diene due to undesired ring-opening occurring in the subsequent RCM reaction, presumably as a consequence of release of inherent ring strain in the embedded cyclopentene ring that was not present in the cyclohexene analogue.

The Mukaiyama aldol reaction on lactol 11 provided a means of introducing further substitution and a handle for elaboration. To this end, adducts of the general structure 43 were formed as single enantiomers by reacting the hemiacetal 11 with a range of silyl enol ethers. As in the example of allyltrimethylsilane addition described above (Scheme 2), the cis products were observed exclusively or as the major component (85:15 in the formation of compound 48). The silyl ethers 44 and 45 were obtained, as well as alcohols 46,47 and 48 following silyl deprotection (Scheme 6).








Scheme 6. Reagents and conditions: a) silyl enol ether, $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ (TMSOTf for compound 46), $\mathrm{CH}_{2} \mathrm{Cl}_{2},-78^{\circ} \mathrm{C} 36-97 \%$.

With the aldehyde 44 in hand, derivatisation was exemplified either by reduction to the diol 49, or through reductive amination with the amines listed in Scheme 7. Deprotection of amines 50-54 with TBAF provided alcohols 55-59, respectively, in good yield.

Modification of the terminal olefin in compound 24 was also exemplified. Oxyarylation ${ }^{[34]}$ by using a gold(I) catalyst in the presence of Selectfluor with phenyltrimethylsilane gave the ether 60 as an inseparable $4: 1 \mathrm{mix}$ of diastereomers (Scheme 7).


Scheme 7. Reagents and Conditions: a) $\mathrm{NaBH}_{4}$, THF, MeOH $82 \%$; b) TBAF, THF, 41-93\%; c) amine, $\mathrm{NaBH}(\mathrm{OAC})_{3}, 1,2$-dichloroethane (DCE), $52-64 \%$; d) $\left.\mathrm{PhSiMe}{ }_{3},\left[\mathrm{AuCl}^{( } \mathrm{PPh}_{3}\right)\right]$, selectfluor, $\mathrm{MeOH}, \mathrm{MeCN}$, then TBAF $29 \%$.

An alternative lactonisation approach to forming an oxygencontaining nine-membered ring system and thus providing library members with closer correspondence to the [6,5,9]-tricyclic core of the natural product family was attempted from aldehyde 44. A Horner-Wadsworth-Emmons reaction gave the $\alpha, \beta$-unsaturated ester 61 (Scheme 8). Reduction of both al-


Scheme 8. Reagents and Conditions: a) triethylphosphonoacetate, NaH, THF; b) Pd/C, EtOAc, $\mathrm{H}_{2}$; c) TBAF, THF; d) LiCl, $\mathrm{H}_{2} \mathrm{O}$, THF; e) 2-methyl-6-nitrobenzoic anhydride, DMAP, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$.
kenes by hydrogenation over palladium gave the ester 62. Deprotection of the silyl group with TBAF followed by ester hydrolysis gave the acid 64 in $71 \%$ over two steps. Lactonisation was initially attempted by using the protocol of Yamaguchi et al., ${ }^{[35]}$ through the intermediate 2,4,6-trichlorobenzoyl anhydride. None of the desired lactone was observed in a number of attempts, with a dimeric species being the only isolated product seen. A more successful approach used conditions developed by Shiina et al., ${ }^{[36]}$ whereby a solution of the acid 64 was added slowly by using a syringe pump over a time period of 15 h to a solution of 2-methyl-6-nitrobenzoic anhydride and 4-dimethylaminopyridine (DMAP) in dichloromethane. By using these conditions, lactone 65 was isolated in $30 \%$ yield. Interestingly, an IR absorption for the carbonyl group was observed at $\tilde{v}=1730 \mathrm{~cm}^{-1}$, suggesting the carbonyl group to be significantly twisted out of plane from the lowest energy lactone conformation (compare with the crystal structure of compound 69, Figure 2).

A $\mathrm{Sml}_{2}$-mediated Reformatsky reaction approach was also found successful in the synthesis of a nine-membered lactone. The Reformatsky reaction has been shown previously to give high yields and a controlled stereochemistry in the synthesis of a variety of natural product ring systems, ${ }^{[37]}$ and more pertinently in medium ring synthesis. ${ }^{[38]}$ Here, the alcohol 23 was subjected to the Mukaiyama aldol/TBAF deprotection sequence described above to give the unstable alcohol 67 (Scheme 9). Bromoacetyl bromide acylation followed by $\mathrm{Sml}_{2}$ induced cyclisation gave the alcohol 69 in $45 \%$ yield. In addition, the alcohol 69 could be further derivatised if required, such as to the acetate 70. The structure and absolute stereochemistry of compound 69 was confirmed by X-ray crystallography analysis (Figure 2). As with lactone 65, an unusual IR carbonyl stretch at $\tilde{v}=1723 \mathrm{~cm}^{-1}$ was observed for compound 69

c)

b)



Figure 2. X-ray crystal structure of compounds a) 42, b) $\mathbf{4 0}$, and c) $\mathbf{6 9}$. d) Partial view of compound 69 along the $\mathrm{O}-\mathrm{C}$ lactone single bond.


Scheme 9. Reagents and conditions: a) $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$, trimethyl((2-methylprop-1-en-1-yl)oxy)silane, $\mathrm{CH}_{2} \mathrm{Cl}_{2},-78^{\circ} \mathrm{C}-\mathrm{RT}$; b) TBAF, THF; c) bromoacetyl bromide, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 34 \%$ over three steps; d) $\mathrm{Sml}_{2}$, THF; e) $\mathrm{Ac}_{2} \mathrm{O}$, DMAP, RT.
as a consequence of a $33^{\circ}$ twist out of conjugation with the ester oxygen atom.

With different synthetic routes to nine-membered lactones in hand, our focus returned to accessing nine-membered ethers. Starting from alcohol 25, synthesised as shown in Scheme 2, oxidation to the aldehyde 71 and subsequent reaction with but-3-en-1-ylmagnesium bromide gave alcohols 72 and 73 as a 5:3 separable mixture of isomers (Scheme 10). Following TBS protection, only the major isomer 72 was found to cyclise in the RCM reaction, giving the tricyclic compound 75


Scheme 10. Reagents and conditions: a) Dess-Martin reagent, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; b) but-3-en-1-ylmagnesium bromide, THF; c) trimethylsilyl trifluoromethanesulfonate (TBSOTf), diisopropylethylamine (DIPEA), $\mathrm{CH}_{2} \mathrm{Cl}_{2}$; d) Grubbs II (10 $\mathrm{mol} \%$ ), $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, heating to reflux; e) $\mathrm{HCl}, \mathrm{EtOH}, \mathrm{Et}_{2} \mathrm{O}$.
in $97 \%$ yield. A NOESY correlation between both the [6,5] bridgehead hydrogen atoms and the siloxysubstituted $\mathrm{C}-\mathrm{H}$ hydrogen atom in compound 75 as highlighted in Figure 3 was used to confirm the sterochemistry of the Grignard addition. Deprotection with TBAF


75
Figure 3. ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ NOESY correlations in the tricyclic compound 75 . gave the alcohol 76 in $55 \%$ yield.

The 2,11-cembranoid based library of 44 compounds synthesised by using the routes and methods outlined above is shown in full in Figure 4. As the compounds were intended for the use in drug discovery screening programmes, an assessment of the library in terms of structural diversity, physicochemical properties and potential biological activity was undertaken. An analysis of the molecular properties of the compounds given in Figure 4 is compiled in Table 1. For comparison, the properties of 183 cembranoid nat-

Table 1. Arithmetic mean and standard deviation of the molecular properties of the cembranoid-inspired library, the 2,11-cyclised cembranoid natural products and a random sampling set from the Institute of Cancer Research screening collection.

|  | Cembranoid- <br> inspired <br> library | 2,11-Cyclised <br> cembranoids | ICR screening <br> collection <br> sample |
| :--- | :--- | :--- | :--- |
| no. of compound | 44 | 183 | 358 |
| MW | $275(28)$ | $441(77)$ | $332(59)$ |
| Alog $P$ | $2.6(0.8)$ | $3.1(1.1)$ | $2.6(1.1)$ |
| no. of hydrogen acceptors | $3.2(0.7)$ | $6.4(2.0)$ | $4.1(1.4)$ |
| no. of hydrogen donors | $0.7(0.8)$ | $1.0(0.9)$ | $1.1(0.8)$ |
| no. of rings | $2.6(0.6)$ | $3.4(0.6)$ | $2.9(0.9)$ |
| no. of aromatic rings | $0.07(0.26)$ | $0(0)$ | $2.0(1.0)$ |
| no. of oxygen atoms | $3.1(0.7)$ | $6.4(2.0)$ | $2.4(1.4)$ |
| no. of hydrogen atoms | $25(2.9)$ | $39(5.9)$ | $18(5.3)$ |
| no. of stereoatoms | $4.7(0.9)$ | $8.9(1.5)$ | $0.04(0.33)$ |
| no. of bonds | $21(2.2)$ | $34(5.4)$ | $25(4.7)$ |
| no. of aromatic bonds | $0.43(1.6)$ | $0(0)$ | $11(5.2)$ |
| no. of rotatable bonds | $4.8(3.2)$ | $5.9(3.1)$ | $4.5(2.0)$ |

ural products taken from reviews ${ }^{[1,39]}$ and a random sampling set of 358 compounds taken from a substantially larger 75000 screening library at the Institute of Cancer Research (ICR) are also listed. The ability of a random sampling set of a large screening library to represent the entirety has been demonstrated by Feher and Schmidt, ${ }^{[21]}$ and has been applied here. The ICR screening library itself was chosen by typical criteria for lead-like/drug-like libraries, including limits on predicted/ calculated physicochemical parameters and choice of scaffolds, with a high frequency of heterocyclic aromatic and aliphatic scaffolds ${ }^{[26,27]}$ As a means of visually assessing the diversity between the groups, a principal component analysis (PCA) ${ }^{[41]}$ was performed. PCA has been used widely in chemoinformatics ${ }^{[42]}$ and particularly in the description of chemical space. ${ }^{[22,43]}$ The analysis, carried out by using SIMCA-P $+{ }^{[44]}$ and visualised by using Spotfire, ${ }^{[45]}$ showed (Figure 5) that the ICR screening compounds (red in Figure 5) cover a wide range of chemical


15


78


42


47


58


16


26


84


49


59


17

28


85


18


79


29


80


20

82


21


81


87


22


83


77


41


60


64




39


89


65

Figure 4. Cembranoid-like library.


Figure 5. Principal component analysis (PCA) of the cembranoid-inspired library (green, 44 structures), the 2,11-cyclised cembranoid natural products (blue, 183 structures) and the ICR screening sample (red, 358 structures). Performed by using SIMCA-P $+{ }^{[44]}$ and displayed by using Spotfire. ${ }^{[45]}$
space as would be expected from a well-designed screening deck; ${ }^{[40]}$ however, little overlap exists with the cembranoid natural products (blue in Figure 5), which occupied a discrete and well-defined region defined by the PCA. By contrast, the cem-
branoid-inspired library as synthesised here (green in Figure 5) was found to occupy an area of chemical space between the natural products and the ICR screening collection, showing that the cembranoid-inspired library begins to effectively bridge the chemical space between classical drug discovery screening compounds and the natural products.
The library prepared was assessed for both solubility and permeability, with representative results shown in Table 2. Due

|  | Compound |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | 28 | 15 | 65 | 76 |
| solubility [ $\mu \mathrm{m}$ ] | 550 | 500 | 550 | 500 |
| permeability ( Pe ) $\left[\times 10^{-6} \mathrm{~cm}^{-1}\right]$ | 52 | 25 | > 150 | not detected ${ }^{[\text {a] }}$ |
| Solubility measured by nephelometry. [a] Undetectable due to low ionisation. |  |  |  |  |

to a lack of UV chromophore and/or poor mass spectrometry ionisation, nephelometry ${ }^{[46]}$ was used to measure the solubility. PAMPA (parallel artificial membrane permeability assay) ${ }^{[47]}$ was used to assess the permeability. Generally, the compounds were found to have high aqueous solubility and permeability. Of the 44 compounds given in Figure 4, all were measured to have greater than $100 \mu \mathrm{~m}$ aqueous solubility, with 34 greater than $500 \mu \mathrm{~m}$ and 21 greater than $1000 \mu \mathrm{~m}$. In regard to the 28 compounds that ionised sufficiently well to be detected by the

PAMPA, 25 were found to have a high permeability, greater than $25 \times 10^{-6} \mathrm{cms}^{-1}$. For the specific example of compound 28 in addition to a good aqueous solubility and high PAMPA, the compound showed moderate permeability across a monolayer of Caco-2 human intestinal cancer cells, ${ }^{[48]}$ ( $16 \times$ $10^{-6} \mathrm{cms}^{-1}$ ) with no evidence of transporter efflux.

A selection of fifteen of the cembranoid-inspired compounds was screened against a panel of 60 cancer cell lines at the National Cancer Institute screening panel (see the Supporting Information). ${ }^{[49]}$ This cell-based assay aims to discover compounds with a strong cytostatic or cytotoxic growth inhibition over 48 h . Unfortunately, only low levels (up to $20 \%$ ) of cell growth inhibition were seen when screened at a single concentration of $10 \mu \mathrm{~m}$. Of these, compound $\mathbf{2 8}$ gave $\mathbf{1 7 \%}$ inhibition of proliferation in the PC3 human prostate cancer cell line. A full dose response curve obtained for compound 28 (Figure 6) found a $\mathrm{Gl}_{50}$ value of $63 \mu \mathrm{~m}$. It is important to note


Figure 6. a) Concentration-response curve for the antiproliferative effect of compound 28 on PC3 cells in vitro 96 h assay. b) Compound 28 inhibiting the migration properties of PC3 TEM cancer cells. FCS $=$ fetal calf serum. The $y$ axis shows the fold-changes in the cell migration normalised to the absence of FCS.
that the cembranoid family itself is not overtly cytotoxic, with most examples screened to date ranging from single figure micromolar $\mathrm{IC}_{50}$ values to inactive in the cytotoxic assays used. ${ }^{[1]}$ One obvious exception to this is sclerophytin A, with an in vitro $\mathrm{IC}_{50}$ value of 3 nm in L 1210 mouse lymphocytic leukemia cells ${ }^{[51]}$ although the natural product showed no effect on proliferation in PC3 cells at $50 \mu \mathrm{~m}$. Therefore, within this context,
compound 28 displays typical cytotoxicity for the natural product family, ${ }^{[39]}$ with reduced structural and synthetic complexity.
Whereas cembranoid natural products may not be inherently cytotoxic, an appreciable number do display an ability to inhibit the migratory and invasive properties of cancer cells. ${ }^{[50]}$ The effect of compound $\mathbf{2 8}$ on a PC3-based migratory assay was therefore determined. The tricyclic compound 28 at 3and 5 -fold $\mathrm{GI}_{50}$ concentrations demonstrated a reduction to near basal levels of migration (Figure 6). Although less potent than a number of the 2,11-cembranoid natural products (most active being sclerophytin A, with $85 \%$ inhibition of migration at $50 \mu \mathrm{~m}){ }^{[50]}$ it does demonstrate that the anti-migratory activity can be retained in a greatly simplified structural motif.

## Conclusion

The 2,11-cembranoid class of natural products has served as a rich source of inspiration from both a synthetic and biological perspective. Here, we have developed a number of synthetic approaches to produce a library of structurally simplified cembranoid-inspired derivatives. An assessment of the molecular properties of the octahydroisobenzofuran-based library found them to be complementary to both the cembranoid natural products as well as a sample set of an in-house medicinal chemistry screening collection, providing potential access to novel chemical space. A screen of the physicochemical properties found the library to have desirable properties, namely aqueous solubility and cell membrane permeability, essential to be useful in both biochemical and cell-based assays. The illustrative synthetic routes developed could readily be expanded to more densely populate the chemical space occupied by the small library and the natural products, or indeed to follow up any hits from future screening. A preliminary screen of a sample set of the library found compound 28 to weakly inhibit the growth of PC3 cells, as well as having an inhibitory effect when screened in a cell migration assay. These data suggest that some of the bioactivities observed in cancer cells for the structurally complex 2,11-cembranoid natural products can be recapitulated with much simpler scaffolds derived from the embedded octahydroisobenzofuran core.

## Experimental Section

All synthetic methods are included in the Supporting Information. CCDC 1010197 (26), 1011320 (40), 1435146 (42) and 1435148 (69) contain the supplementary crystallographic data. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.

## Acknowledgements

This work was supported by CRUK grant C19524A8027 (PhD studentship to A.J.W.), the CRUK grant C309A11566, and The Institute of Cancer Research, London. We also acknowledge NHS funding to the NIHR Biomedical Research Centre. We thank Amin Mirza and Sarah Langdon for helpful discussions.
G.J.T., M.B.P. and S.J.C. thank the EPSRC for funding the UK National Crystallography Service. The authors declare no competing financial interest.

Keywords: medicinal chemistry • medium-ring compounds molecular diversity • oxygen heterocycles • synthesis design
[1] A. J. Welford, I. Collins, J. Nat. Prod. 2011, 74, 2318-2328.
[2] a) J. M. Ellis, M. T. Crimmins, Chem. Rev. 2008, 108, 5278-5298; b) M. E. Jung, J. Pontillo, J. Org. Chem. 2002, 67, 6848-6851.
[3] M. E. Jung, J. Pontillo, Tetrahedron 2003, 59, 2729-2736.
[4] B. Wang, A. P. Ramirez, J. J. Slade, J. P. Morken, J. Am. Chem. Soc. 2010, 132, 16380-16382.
[5] H. Kim, H. Lee, J. Kim, S. Kim, D. Kim, J. Am. Chem. Soc. 2006, 128, 15851-15855.
[6] a) L. A. Paquette, O. M. Moradei, P. Bernardelli, T. Lange, Org. Lett. 2000, 2, 1875-1878; b) P. Bernardelli, O. M. Moradei, D. Friedrich, J. Yang, F. Gallou, B. P. Dyck, R. W. Doskotch, T. Lange, L. A. Paquette, J. Am. Chem. Soc. 2001, 123, 9021-9032.
[7] F. Gallou, D. W. C. MacMillan, L. E. Overman, L. A. Paquette, L. D. Pennington, J. Yang, Org. Lett. 2001, 3, 135-137.
[8] a) D. W. C. MacMillan, L. E. Overman, L. D. Pennington, J. Am. Chem. Soc. 2001, 123, 9033-9044.
[9] a) M. T. Crimmins, B. H. Brown, J. Am. Chem. Soc. 2004, 126, 10264 10266; b) M. T. Crimmins, J. M. Ellis, J. Am. Chem. Soc. 2005, 127, 17200 17201; c) M. T. Crimmins, M. C. Mans, A. D. Rodríguez, Org. Lett. 2010, 12, 5028-5031; d) M. J. Campbell, J. S. Johnson, J. Am. Chem. Soc. 2009, 131, 10370-10371.
[10] J. S. Clark, R. Berger, S. T. Hayes, L. H. Thomas, A. J. Morrison, L. Gobbi, Angew. Chem. Int. Ed. 2010, 49, 9867-9870; Angew. Chem. 2010, 122, 10063-10066.
[11] O. Corminboeuf, L. E. Overman, L. D. Pennington, J. Org. Chem. 2009, 74, 5458-5470.
[12] D. W. C. MacMillan, L. E. Overman, J. Am. Chem. Soc. 1995, 117, 10391 10392.
[13] a) M. T. Crimmins, B. H. Brown, H. R. Plake, J. Am. Chem. Soc. 2006, 128, 1371-1378; b) M. T. Crimmins, J. M. Ellis, J. Org. Chem. 2008, 73, 16491660.
[14] L. E. Overman, L. D. Pennington, Org. Lett. 2000, 2, 2683-2686.
[15] a) J. Becker, K. Bergander, R. Fröhlich, D. Hoppe, Angew. Chem. Int. Ed. 2008, 47, 1654-1657; Angew. Chem. 2008, 120, 1678-1681; b) G. Yue, Y. Zhang, L. Fang, C.-C. Li, T. Luo, Z. Yang, Angew. Chem. Int. Ed. 2015, 54, 6531-6535; Angew. Chem. 2015, 127, 6631-6635.
[16] a) J. S. Clark, S. T. Hayes, C. Wilson, L. Gobbi, Angew. Chem. Int. Ed. 2007, 46, 437-440; Angew. Chem. 2007, 119, 441-444; b) J. S. Clark, L. Delion, L. J. Farrugia, Org Lett 2014, 16, 4300-4303; c) J. S. Clark, L. Delion, L. J. Farrugia, Chem. Eur. J. 2015, 21, 4772-4780.
[17] a) G. A. Molander, D. J. St. Jean, J. Haas, J. Am. Chem. Soc. 2004, 126, $1642-1643$; b) J. S. Clark, R. Berger, S. T. Hayes, H. Martin Senn, L. J. Farrugia, L. H. Thomas, A. J. Morrison, L. Gobbi, J Org Chem 2013, 78, 673 696.
[18] a) D. J. Newman, G. M. Cragg, J. Nat. Prod. 2007, 70, 461-477; b) A. L. Harvey, Drug Discovery Today 2008, 13, 894-901; c) A. L. Harvey, R. Edrada-Ebel, R. J. Quinn, Nat. Rev. Drug Discovery 2015, 14, 111-129; d) D. J. Newman, G. M. Cragg, J. Nat. Prod. 2012, 75, 311-335.
[19] I. Collins, A. M. Jones, Molecules 2014, 19, 17221-17255.
[20] a) B. Dasari, S. Jogula, R. Borhade, S. Balasubramanian, G. Chandrasekhar, S. S. Kitambi, P. Arya, Org. Lett. 2013, 15, 432-435; b) M. Aeluri, C. Pramanik, L. Chetia, N. V. Mallurwar, S. Balasubramanian, G. Chandrasekhar, S. S. Kitambi, P. Arya, Org. Lett. 2013, 15, 436-439; c) M. Aeluri, J. Gaddam, D. V. K. S. Tinath, G. Chandrasekhar, S. S. Kitambi, P. Arya, Eur. J. Org. Chem. 2013, 3955-3958; d) S. Chamakuri, S. K. R. Guduru, S. Pamu, G. Chandrasekhar, S. S. Kitambi, P. Arya, Eur. J. Org. Chem. 2013, 39593964.
[21] M. Feher, J. M. Schmidt, J. Chem. Inf. Comput. Sci. 2002, 42, 218-227.
[22] a) G. L. Thomas, E. E. Wyatt, D. R. Spring, Curr. Opin. Drug Discovery Devel. 2006, 9, 700-712; b) W. R. J. D. Galloway, A. Bender, M. Welch, D. R. Spring, Chem. Commun. 2009, 2446-2462.
[23] a) R. Breinbauer, I. R. Vetter, H. Waldmann, Angew. Chem. Int. Ed. 2002, 41, 2878-2890; Angew. Chem. 2002, 114, 3002-3015; b) D. Brohm, S. Metzger, A. Bhargava, O. Müller, F. Lieb, H. Waldmann, Angew. Chem. Int. Ed. 2002, 41, 307-311; Angew. Chem. 2002, 114, 319-323.
[24] a) R. G. Doveston, P. Tosatti, M. Dow, D. J. Foley, H. Y. Li, A. J. Campbell, D. House, I. Churcher, S. P. Marsden, A. Nelson, Org. Biomol. Chem. 2015, 13, 859-865; b) R. Doveston, S. Marsden, A. Nelson, Drug Discovery Today 2014, 19, 813-819.
[25] E. H. Kerns, L. Di in Drug-Like Properties: Concepts, Structure Design and Methods, Vol. 1, Elsevier, Amsterdam, 2008.
[26] C. A. Lipinski, F. Lombardo, B. W. Dominy, P. J. Feeney, Adv. Drug Delivery Rev. 1997, 23, 3-25.
[27] D. J. Huggins, A. R. Venkitaraman, D. R. Spring, ACS Chem. Biol. 2011, 6, 208-217.
[28] K. C. Nicolaou, J. A. Pfefferkorn, A. J. Roecker, G.-Q. Cao, S. Barluenga, H. J. Mitchell, J Am Chem. Soc. 2000, 122, 9939-9953.
[29] a) J. Mann, A. Thomas, Chem. Commun. 1985, 737-738; b) M. G. B. Drew, J. Mann, A. Thomas, J. Chem. Soc. Perkin Trans. 1 1986, 22792285.
[30] A. Saito, H. Yanai, T. Taguchi, Tetrahedron Lett. 2004, 45, 9439-9442.
[31] R. A. Pilli, V. B. Riatto, Tetrahedron: Asymmetry 2000, 11, 3675-3686.
[32] J. P. Wolfe, M. B. Hay, Tetrahedron 2007, 63, 261-290.
[33] D. Joe, L. E. Overman, Tetrahedron Lett. 1997, 38, 8635-8638.
[34] L. T. Ball, M. Green, G. C. Lloyd-Jones, C. A. Russell, Org. Lett. 2010, 12, 4724-4727.
[35] a) J. Inanaga, K. Hirata, H. Saeki, T. Katsuki, M. Yamaguchi, Bull. Chem. Soc. Jpn. 1979, 52, 1989-1993; b) P. Basabe, O. Bodero, I. S. Marcos, D. Diez, A. Blanco, M. de Roman, J. G. Urones, J. Org. Chem. 2009, 74, 7750-7754; c) M. Hikota, H. Tone, K. Horita, O. Yonemitsu, J. Org. Chem. 1990, 55, 7-9.
[36] a) I. Shiina, M. Kubota, H. Oshiumi, M. Hashizume, J. Org. Chem. 2004, 69, 1822-1830; b) I. Shiina, M. Kubota, R. Ibuka, Tetrahedron Lett. 2002, 43, 7535-7539; c) I. Shiina, H. Oshiumi, M. Hashizume, Y. Yamai, R. Ibuka, Tetrahedron Lett. 2004, 45, 543-547.
[37] D. J. Edmonds, D. Johnston, D. J. Procter, Chem. Rev. 2004, 104, 3371 3404.
[38] G. A. Molander, J. B. Etter, L. S. Harring, P. J. Thorel, J. Am. Chem. Soc. 1991, 113, 8036-8045.
[39] P. Bernardelli, L. A. Paquette, Heterocycles 1998, 49, 531-556.
[40] A. A. Shelat, R. K. Guy, Nat. Chem. Biol. 2007, 3, 442-446.
[41] a) S. Wold, K. Esbensen, P. Geladi, Chemom. Intell. Lab. Syst. 1987, 2, 37 52; b) J. E. Jackson in A User's Guide to Principal Components, Wiley, New York, 1991.
[42] K. Varmuza, P. Filzmoser in Multivariate Statistical Analysis in Chemometrics, CRC Press, London, 2009.
[43] a) C. Lipinski, A. Hopkins, Nature 2004, 432, 855-861; b) C. M. Dobson, Nature 2004, 432, 824-828; c) N. J. Rosén, J. Gottfries, S. Muresan, A. Backlund, T. I. Oprea, J. Med. Chem. 2009, 52, 1953-1962.
[44] Simca-P + 12.0, Umertrics, Umeâ, Sweden, http://www.umetrics.com/.
[45] Spotfile Decision Site, Tibco, Palo Alto, CA, USA, http://spotfire.tibco. com/.
[46] C. D. Bevan, R. S. Lloyd, Anal. Chem. 2000, 72, 1781-1787.
[47] M. Kansy, F. Senner, K. Gubernator, J. Med. Chem. 1998, 41, 1007-1010.
[48] a) P. Artursson, K. Palm, K. Luthman, Adv. Drug Delivery Rev. 2001, 46, 27-43; b) E. H. Kerns, L. Di, S. Petusky, M. Farris, R. Ley, P. Jupp, J. Pharm. Sci. 2004, 93, 1440-1453.
[49] a) R. H. Shoemaker, Nat. Rev. Cancer 2006, 6, 813-823; b) http://dtp.nci. nih.gov.
[50] H. M. Hassan, M. A. Khanfar, A. Y. Elnagar, R. Mohammed, L. A. Shaala, D. T. A. Youssef, M. S. Hifnawy, K. A. El Sayed, J. Nat. Prod. 2010, 73, 848 853.
[51] P. Sharma, M. Alam, J. Chem. Soc. Perkin Trans. 1 1988, 2537-2540.

Received: December 18, 2015
Published online on March 1, 2016


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    - Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/chem.201505093.
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[^1]:    the presence of $\mathrm{AlCl}_{3}$ in dichloromethane at $55^{\circ} \mathrm{C}$ in a sealed tube for one week under literature conditions ${ }^{[29]}$ gave the bicyclic compound 10 in $40 \%$ yield. Although forming the desired product, yields were found to be highly variable. Screening of a number of Lewis acids found $\mathrm{TfN}(\mathrm{AlMeCl})_{2}^{[30]}$ ( $\mathrm{Tf}=$ triflate) to give a consistently high yield of around $70-80 \%$ after a two day reaction. Furthermore, it was found possible to use a solution of $30 \%$ butadiene in toluene as the diene source, avoiding the inconvenience and hazards associated with condensing gaseous butadiene. Reduction of the lactone 10 with diisobutylaluminum hydride (DIBAL-H) gave the lactol 11 in high yield,

