

Open Access

# Achiral bis-imine in combination with CoCl<sub>2</sub>: A remarkable effect on enantioselectivity of lipase-mediated acetylation of racemic secondary alcohol

K. Arunkumar<sup>1,2</sup>, M. Appi Reddy<sup>1</sup>, T. Sravan Kumar<sup>1</sup>, B. Vijaya Kumar<sup>1</sup>, K. B. Chandrasekhar<sup>2</sup>, P. Rajender Kumar<sup>1</sup> and Manojit Pal<sup>\*3</sup>

## Letter

Address:

<sup>1</sup>Custom Pharmaceutical Services, Dr. Reddy's Laboratories Limited, Bollaram Road Miyapur, Hyderabad 500 049, India, <sup>2</sup>Department of Chemistry, Jawaharlal Nehru Technological University of Anantapur, Anantapur 515002, Andhra Pradesh, India, and <sup>3</sup>Institute of Life Sciences, University of Hyderabad Campus, Gachibowli, Hyderabad 500 046, Andhra Pradesh, India

Email:

Manojit Pal\* - manojitpal@rediffmail.com

\* Corresponding author

Keywords:

acetylation; bis-imine; cobalt chloride; enantioselectivity; lipase

Beilstein J. Org. Chem. **2010**, 6, 1174–1179. doi:10.3762/bjoc.6.134

Received: 26 August 2010 Accepted: 09 November 2010 Published: 10 December 2010

Associate Editor: S. Flitsch

© 2010 Arunkumar et al; licensee Beilstein-Institut. License and terms: see end of document.

#### Abstract

A bis-imine (prepared via a new  $FeCl_3$ -based method) in combination with  $CoCl_2$  facilitated lipase-mediated acetylation of the (R)-isomer of a racemic benzylic secondary alcohol with 91% ee<sub>s</sub>. The methodology was used for the preparation of the known drug rivastigmine.

## Introduction

The development and use of newer synthetic methods for the stereoselective synthesis of chiral molecules have increased enormously in recent years especially in the chemical and pharmaceutical industry [1]. Biocatalysis, being an environmentally friendly process, has attracted particular attention for this purpose [2-6]. For example, high enantioselectivity was observed in lipase-mediated preparation of alcohols and amines [7-9]. These biocatalysts work under mild reaction conditions, and their immobilized forms, being stable in organic solvents, have allowed an easy separation of products and the potential recycling of the enzyme, thereby enhancing their economic

viability [10,11]. Recently, we have observed that achiral bisimines in combination with CoCl<sub>2</sub> improved the enantio-selectivity substantially in CAL-B (*Candida antarctica* lipase B) [12,13] mediated acetylation of a racemic secondary alcohol with vinyl acetate. Here we report our preliminary results on the synthesis and identification of a novel ligand for this process (Scheme 1) and its application in the preparation of the known drug rivastigmine [14]. While the uses of bis-imine/transition-metal complexes have been reported for the enantioselective synthesis of chiral compounds [15-19], their use as activators in an enzymatic reaction has not been previously explored.

**Scheme 1:** Lipase-catalyzed acetylation of racemic benzylic secondary alcohol [(RS)-4] and its application.

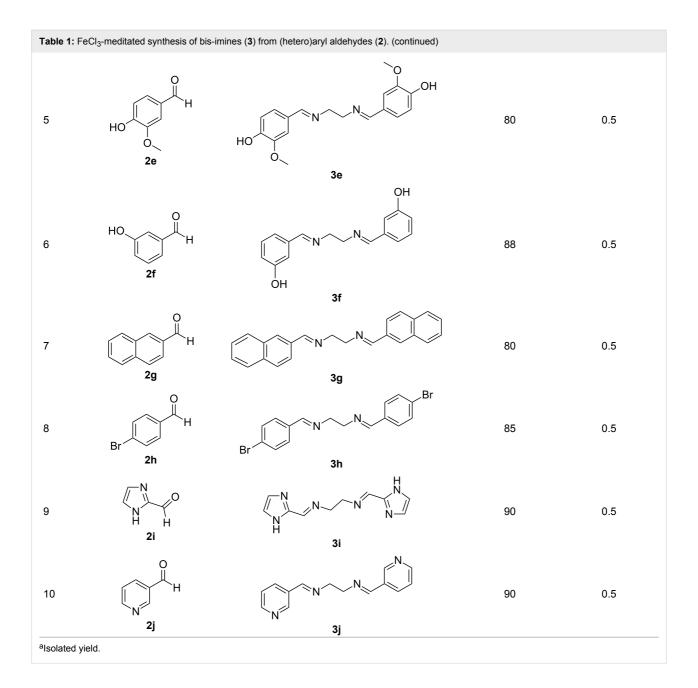
## Results and Discussion

The report that bis-imine/Cu(I)-complexes were able to promote the direct and enantioselective addition of imines to alkylacetylenes [15] prompted us to evaluate a variety of bis-imines in combination with CoCl<sub>2</sub> [16] in the lipase-mediated acetylation of a benzylic secondary alcohols. Accordingly, a number of achiral bis-imines were prepared and used to generate the desired complex. While a number of methods have been

reported to prepare Schiff bases by reacting an amine with a carbonyl compound [20-24], in our hands the reaction of 1,2-amines with aromatic aldehydes under these reaction conditions provided a mixture of mono and bis-Schiff bases. We therefore developed a new and efficient method for the preparation of bis-imines 3 by reacting ethane-1,2-diamine (1) with a number of aryl and heteroaryl aldehydes 2 in the presence of anhydrous FeCl<sub>3</sub> (Scheme 2, Table 1). Aldehydes containing electron donating (e.g., methoxy, hydroxy, fluoro and bromo) or withdrawing groups (e.g., nitro) were found to be equally effective in terms of product yields. The reactions were completed within 30 min in most cases.

All bis-imines prepared were then screened in combination with CoCl<sub>2</sub> for CAL-B mediated acetylation of a racemic secondary alcohol. Thus, 3-(1-hydroxyethyl)phenyl ethyl(methyl)carba-

Table 1: FeCl <sub>3</sub> -meditated synthesis of bis-imines (3) from (hetero)aryl aldehydes (2).				
Entry	Aldehyde (2)	Product (3)	Yield <sup>a</sup> (%)	Reaction Time
1	O H	3a	90	0.5
2	O <sub>2</sub> N 2b	$O_2N$ $N$ $N$ $N$ $N$ $N$ $N$ $N$ $N$ $N$	90	0.15
3	OH 2c	OH OON NOT NOT NOT NOT NOT NOT NOT NOT NOT	88	0.5
4	O H 2d	F 3d	80	0.5



mate [(RS)-4], prepared via the reaction of 3-hydroxyacetophenone (6) with N-ethyl-N-methylcarbamoyl chloride followed by reduction with NaBH<sub>4</sub> (Scheme 3), was selected for our purpose. Subsequently, the enzymatic processes were carried out and the isolated reaction mixture was analysed by chiral HPLC.

Initially, the CAL-B mediated acetylation of (RS)-4 was carried out in the absence of any ligand and CoCl<sub>2</sub>. Vinyl acetate was used as a solvent as well as the acyl donor. No reaction was observed at room temperature even after 48 h. An increase in reaction temperature to 50–55 °C for 24 h facilitated the acetylation, however, the selectivity was not greater than 30%. In

HO

$$K_2CO_3$$
, acetone reflux, 5 h, 85%

 $RS$ )-4

 $RS$ )-4

**Scheme 3:** Preparation of racemic 3-(1-hydroxyethyl)phenyl ethyl(methyl)carbamate [(*RS*)-**4**]

~46

>300

order to achieve better selectivity, we assessed the use of achiral bis-imines in combination with  $CoCl_2$  (Table 2). The reactions were complete within 10 h when diarylidene-ethane-1,2-diamines were used (entries 1–8, Table 2). While 35% enantiomeric excess was achieved in some of these cases (entries 2, 3 and 5, Table 2), the best results, however, were obtained with bis(heteroarylmethylene)ethane-1,2-diamines (entries 9 and 10, Table 2), especially  $\bf 3i$ . The bis-imine  $\bf 3i$  facilitated enantioselective acetylation of the ( $\it R$ )-isomer over the ( $\it S$ )-antipode with high enantiomeric excess (91% ee<sub>s</sub>) and yield (80%). The reac-

10

3j

tion was complete within 12 h. The absolute configuration of the resolved chiral alcohol and its acetate was in accordance with Kazlauskas' rule [25] (see Supporting Information File 1 for optical rotation values).

Mechanistically [12], the special H-bonding rearrangement of the "catalytic triad" (i.e., serine, histidine, and aspartate) at the active site of CAL-B increases the nucleophilicity of the serine residue. This then interacts with the carbonyl group of the vinyl acetate to form the "acyl-enzyme intermediate" **T-1** (Scheme 4)

Table 2: Screening of bis-imines as achiral ligands in CAL-B mediated acetylation of (RS)-4 (step 1, Scheme 1)a. Entry Ligand 3 Time (h) Conversion<sup>b</sup> (%) Ec ees eep 1 3a 10 0 0 0 0 2 3b 5 35 ~26 >99 >200 5 35 3 3с >99 ~26 >200 4 3d 10 0 0 0 0 5 5 35 >99 ~26 >200 3e 6 3f 5 0 0 0 0 7 5 0 0 0 0 3g 5 8 3h 0 0 0 n 9 3i 12 91 >99 48 >500

<sup>a</sup>All the reactions were carried out at 1.0 g scale of (RS)-4 with vinyl acetate (20 mL) as acyl donor, in the presence of CAL-B (150 mg), bis-imine (3, 0.3 mmol) and CoCl<sub>2</sub> (0.3 mmol); ee<sub>s</sub> = enantiomeric excess of substrate (the ee<sub>s</sub> is mentioned as the enzyme is active with only one enantiomer) and ee<sub>p</sub> = enantiomeric excess of product. Both ee<sub>s</sub> and ee<sub>p</sub> were determined by HPLC [column: chiralpak IC (250 x 4.6 mm, 5.0 µm), mobile phase A: 0.05% TFA in water, mobile phase B: n-hexane: IPA (80:20), concentration: 0.5 mg/mL, diluent: ethanol, run time: 30.0 min, temperature: 25 °C, flow: 1.0 mL/min, UV: 220 nm]. <sup>b</sup>Conversion = ee<sub>s</sub>/(ee<sub>s</sub> + ee<sub>p</sub>). <sup>c</sup>E = {ln[ee<sub>p</sub>(1-ee<sub>s</sub>)]/(ee<sub>p</sub>+ee<sub>s</sub>)}/{ln[ee<sub>p</sub>(1 + ee<sub>s</sub>)]/(ee<sub>p</sub> + ee<sub>s</sub>)}.

>99

HO 
$$K_2CO_3$$
, acetone reflux, 5 h  $K_2CO_3$ , acetone reflux,

which finally transfers the acyl group to the substrate alcohol 4 via  $\mathbf{T}$ -2, affording the desired product 5. The  $\operatorname{CoCl}_2$  in combination with  $\mathbf{3i}$  perhaps forms a tight complex with  $\mathbf{T}$ -1 as well as 4 which facilitates the acyl transfer process (Scheme 4). However, the reason for selective acylation was not clearly understood. It was speculated that the orientation of the hydroxy group of the (R)-isomer was possibly in the proximal position of the acyltransfer site and the imidazole moiety for proton abstraction.

Finally, application of this methodology was demonstrated in preparing the well-known drug rivastigmine which has been used to treat mild to moderate dementia associated with Alzheimer's and Parkinson's disease. Thus the enantiopure acetate (*R*)-5 was treated with excess of dimethylamine in toluene to afford the desired (*S*)-8 [(*S*)-rivastigmine] in 60% yield (final step, Scheme 5). Notably, the earlier method for the synthesis of (*S*)-8 involved asymmetric reduction of the ketone 6 to give the alcohol with the required chirality followed by mesylation and subsequent treatment with dimethylamine [26,27].

#### Conclusion

We have developed a novel lipase-based method for acetylation of a benzylic secondary alcohol with high enantio-selectivity and yield. The methodology involves the use of CoCl<sub>2</sub> in combination with a bis-imine (prepared via a new FeCl<sub>3</sub>-based method) and its application has been demonstrated in preparing rivastigmine.

# Supporting Information

## Supporting Information File 1

Experimental procedures and spectral data. [http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-6-134-S1.pdf]

## Acknowledgements

The authors (K.A.) thank Dr. V. Dahanukar, Mr. A. Mukherjee for encouragemets and the analytical group of DRL for spectral support.

#### References

- Carey, J. S.; Laffan, D.; Thomson, C.; Williams, M. T. Org. Biomol. Chem. 2006, 4, 2337–2347. doi:10.1039/b602413k
- Straathof, J. J.; Panke, S.; Schmid, A. Curr. Opin. Biotechnol. 2002, 13, 548–556. doi:10.1016/S0958-1669(02)00360-9
- Panke, S.; Held, M.; Wubbolts, M. Curr. Opin. Biotechnol. 2004, 15, 272–279. doi:10.1016/j.copbio.2004.06.011
- Gotor-Fernandez, V.; Gotor, V. Use of lipases in organic synthesis. In Industrial enzymes: Structure, function and applications; Polaina, J.; MacCabe, A. P., Eds.; Chapter 18; Springer: Dordrecht, The Netherlands, 2007; pp 301–315.
- Ran, N.; Zhao, L.; Chen, Z.; Tao, J. Green Chem. 2008, 10, 361–372. doi:10.1039/b716045c
- Woodley, J. M. Trends Biotechnol. 2008, 26, 321–327. doi:10.1016/j.tibtech.2008.03.004
- Gotor-Fernandez, V.; Gotor, V.; Rebolledo, F. Preparation of Chiral Pharmaceuticals through Enzymatic Acylation of Alcohols and Amines. In *Biocatalysis in the Pharmaceutical and Biotechnology Industries*; Patel, R. N., Ed.; Chapter 7; CRC Press: Boca Raton, 2007; pp 203–248.
- Gotor-Fernandez, V.; Brieva, R.; Gotor, V. J. Mol. Catal. B: Enzym. 2006. 40. 111–120. doi:10.1016/j.molcatb.2006.02.010
- Patel, R. N. Coord. Chem. Rev. 2008, 252, 659–701. doi:10.1016/i.ccr.2007.10.031
- 10. Ghanem, A.; Aboul-Enein, H. Y. *Tetrahedron: Asymmetry* **2004**, *15*, 3331–3351. doi:10.1016/j.tetasy.2004.09.019
- 11. Ghanem, A. *Tetrahedron* **2007**, *63*, 1721–1754. doi:10.1016/i.tet.2006.09.110
- Pàmies, O.; Bäckvall, J.-E. Chem. Rev. 2003, 103, 3247–3262. doi:10.1021/cr020029q
- García-Urdiales, E.; Rebolledo, F.; Gotor, V. Tetrahedron: Asymmetry 2001, 12, 3047–3052. doi:10.1016/S0957-4166(01)00532-8
- 14. Jann, M. W. *Pharmacotherapy* **2000**, *20*, 1–12. doi:10.1592/phco.20.1.1.34664
- Colombo, F.; Benaglia, M.; Orlandi, S.; Usuelli, F.; Celentano, G.
   J. Org. Chem. 2006, 71, 2064–2070. doi:10.1021/jo052481g

- Iqbal, J.; Mukhopadhyay, M.; Mandal, A. K. Synlett 1997, 876–886. doi:10.1055/s-1997-924
- Saito, T.; Takekawa, K.; Nishimura, J.-I.; Kawamura, M. J. Chem. Soc., Perkin Trans. 1 1997, 2957–2960. doi:10.1039/a703590j
- Evans, D. A.; Lectka, T.; Miller, S. J. Tetrahedron Lett. 1993, 34, 7027–7030. doi:10.1016/S0040-4039(00)61588-5
- Li, Z.; Conser, K. R.; Jacobsen, E. N. J. Am. Chem. Soc. 1993, 115, 5326–5327. doi:10.1021/ja00065a067
- Vazzana, I.; Terranova, E.; Mattioli, F.; Sparatore, F. ARKIVOC 2004, v. 364–374
- Sridhar, S. K.; Saravanan, M.; Ramesh, A. Eur. J. Med. Chem. 2001, 36. 615–625. doi:10.1016/S0223-5234(01)01255-7
- Karupaiyan, K.; Puranik, V. G.; Deshmukh, A. R. A. S.; Bhawal, B. M. Tetrahedron 2000, 56, 8555–8560. doi:10.1016/S0040-4020(00)00800-0
- 23. Kunz, H.; Pfrengle, W.; Rück, K.; Sager, W. Synthesis 1991, 1039–1042. doi:10.1055/s-1991-26641
- Kunz, H.; Sager, W. Angew. Chem., Int. Ed. Engl. 1987, 26, 557–559. doi:10.1002/anie.198705571
- Kazlauskas, R. J.; Weissfloch, A. N. E.; Rappaport, A. T.; Cuccia, L. A. J. Org. Chem. 1991, 56, 2656–2665. doi:10.1021/jo0008a016
- Gaitonde, A.; Mangle, M.; Pawar, S. R. Novel processes for the preparation of aminoalkyl phenylcarbamates. World Pat. Appl. WO2005/061446, July 7, 2005.
- Fieldhouse, R. Process for the preparation of tertiary amines attached to a secondary carbon centre. World Pat. Appl. WO2005/058804, June 30, 2005.

## License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/2.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions:

(http://www.beilstein-journals.org/bjoc)

The definitive version of this article is the electronic one which can be found at:

doi:10.3762/bjoc.6.134