



Understanding the Role of Histidine in the GHSxG Acyltransferase Active Site Motif: Evidence for Histidine Stabilization of the Malonyl-Enzyme Intermediate

Sean Poust¹, Isu Yoon¹, Paul D. Adams^{4,5}, Leonard Katz^{3,4}, Christopher J. Petzold^{4,5}, Jay D. Keasling^{1,2,3,4,5*}

1 Department of Chemical and Biomolecular Engineering, University of California, Berkeley, California, United States of America, **2** QB3 Institute, University of California, Berkeley, California, United States of America, **3** Synthetic Biology Engineering Research Center, Emeryville, California, United States of America, **4** Joint BioEnergy Institute, Emeryville, California, United States of America, **5** Physical Bioscience division, Lawrence Berkeley National Laboratory, Berkeley, California, United States of America

Abstract

Acyltransferases determine which extender units are incorporated into polyketide and fatty acid products. The ping-pong acyltransferase mechanism utilizes a serine in a conserved GHSxG motif. However, the role of the conserved histidine in this motif is poorly understood. We observed that a histidine to alanine mutation (H640A) in the GHSxG motif of the malonyl-CoA specific yersiniabactin acyltransferase results in an approximately seven-fold higher hydrolysis rate over the wildtype enzyme, while retaining transacylation activity. We propose two possibilities for the reduction in hydrolysis rate: either H640 structurally stabilizes the protein by hydrogen bonding with a conserved asparagine in the ferredoxin-like subdomain of the protein, or a water-mediated hydrogen bond between H640 and the malonyl moiety stabilizes the malonyl-O-AT ester intermediate.

Citation: Poust S, Yoon I, Adams PD, Katz L, Petzold CJ, et al. (2014) Understanding the Role of Histidine in the GHSxG Acyltransferase Active Site Motif: Evidence for Histidine Stabilization of the Malonyl-Enzyme Intermediate. PLoS ONE 9(10): e109421. doi:10.1371/journal.pone.0109421

Editor: Beata G. Vertessy, Institute of Enzymology of the Hungarian Academy of Science, Hungary

Received: May 19, 2014; **Accepted:** August 30, 2014; **Published:** October 6, 2014

Copyright: © 2014 Poust et al. This is an open-access article distributed under the terms of the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

Data Availability: The authors confirm that all data underlying the findings are fully available without restriction. All relevant data are within the paper and its Supporting Information files.

Funding: This work was supported by the Joint BioEnergy Institute which is funded by the Office of Science, Office of Biological and Environmental Research of the U.S. Department of Energy (Contract No. DE-AC02-05CH11231), by the National Science Foundation (award No. EEC-0540879 to the Synthetic Biology Research Center), by the Department of Energy, ARPA-E Electrofuels Program (Contract No. DE-0000206-1577), and by the National Science Foundation Graduate Research Fellowship Program (Grant No. DGE 1106400, to SP). The funders had no role in study design, data collection and analysis, decision to publish, or preparation of the manuscript.

Competing Interests: The authors have declared that no competing interests exist.

* Email: keasling@berkeley.edu

Introduction

During fatty acid and polyketide biosynthesis, acyltransferases (ATs) catalyze transfer of the acyl group from malonyl-, methylmalonyl-, or other short chain acyl-CoAs to acyl carrier proteins (ACPs) using a serine-histidine catalytic dyad. The OH-group of serine acts as a nucleophile to attack the thioester bond of the acyl donor, forming a covalent acyl-O-ester intermediate. The acyl group is subsequently transferred to the ACP via a bi ping pong mechanism (Figure 1 A) [1].

For efficient transfer to the ACP, acyl-AT complexes must be stable in solution. However, ATs have been described as distant relatives of the α/β hydrolase superfamily [1,2], and hydrolysis is an unproductive side reaction, effectively wasting the activated acyl-CoA substrate. The active sites for α/β hydrolases carry a GxSxG motif, whereas ATs possess the highly conserved GHSxG motif in the active site, suggesting that the histidine is important for acyl transfer. These conserved GHSxG and GxSxG motifs are illustrated in sequence logos from pfam PF00698 (Acyl_transf_1) and an evolutionarily related family from the same pfam clan, PF01734 (Patatin), which have phospholipase activity (Figure 1 B

and C) [3]. The sequence logo for the AT family (Figure 1 B) shows that the histidine in the GHSxG motif is highly conserved.

Dreier and coworkers proposed that the conserved histidine in the GHSxG motif could act as an alternative catalytic nucleophile responsible for the observed transacylation activity of an AT mutant in which the active site serine was changed to alanine [4]. This hypothesis was refuted by Szafranska and coworkers, who found the same serine to alanine mutant had no acyl transfer activity upon additional purification of the AT [5]. We sought to further investigate and clarify the role of the highly conserved histidine in acyl transfer reactions. Specifically, we examined three variants of the GHSxG motif in the malonyl-CoA specific yersiniabactin synthase AT domain [6]: S641A, H640A, and the double mutant H640A+S641A in the context of the full PKS module.

Materials and Methods

His-tagged constructs were expressed in *Escherichia coli* BLR, and purified using Ni-NTA chromatography followed by anion

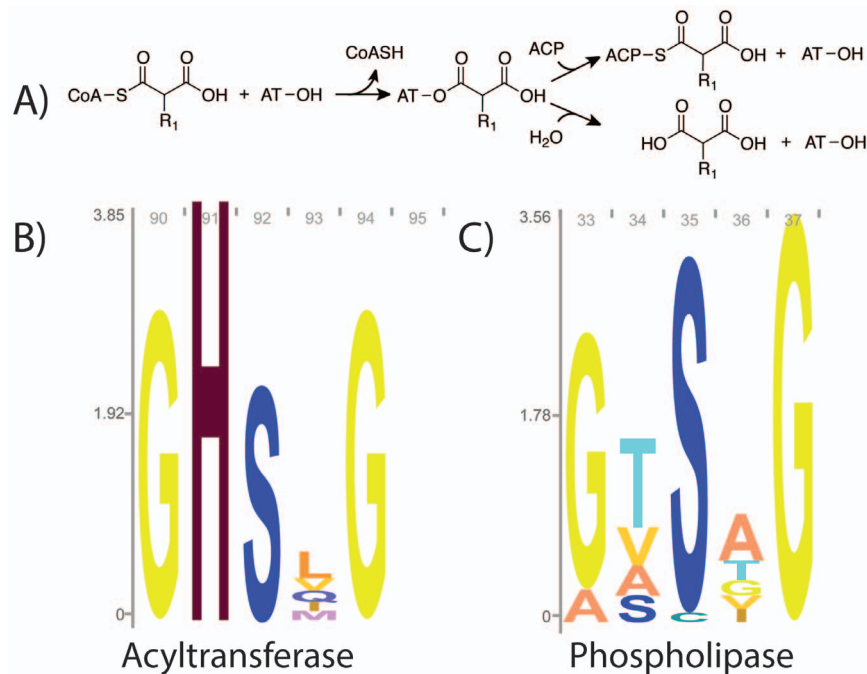


Figure 1. Acyltransferase reaction mechanism and sequence logos. A) AT reaction mechanism: ATs utilize a ping-pong mechanism to transfer acyl groups from CoA to ACP (for most ATs, $R_1 = \text{H}$ or CH_3). Hydrolysis, shown in the bottom branch is a competing, unproductive reaction. B) Sequence logo for the motif containing the active site serine of pfam PF00698, which encompasses ATs from fatty acid and polyketide biosynthesis. C) Sequence logo for pfam PF01734, which are evolutionarily related phospholipases. Logos created using Skyline [12].
doi:10.1371/journal.pone.0109421.g001

exchange chromatography using a HiTRAP Q column (GE Healthcare). We observed that Ni-NTA chromatography alone was insufficient to remove background acyltransferase activity from negative controls (e.g. the double mutant S641A+H640A). For each variant, we measured steady-state rates of malonyl-CoA hydrolysis using a coupled fluorometric assay [7,8], at physiological concentrations of malonyl-CoA [9]. This fluorometric coupled assay has been previously utilized to measure rates of acyl-CoA hydrolysis by ATs [8]. We also investigated the formation of malonyl-O-AT and the subsequent intraprotein transfer to form malonyl-S-ACP. We accomplished this by examining peptide acylation using high-resolution mass spectrometry after trypsin digestion of the protein following the *in vitro* loading/transfer reactions [10]. Detailed information on protocols is provided in the Methods S1.

Results and Discussion

Mutating H640 to alanine (mutant H640A) resulted in an approximately 7-fold increase in the malonyl-CoA hydrolysis rate over the wildtype enzyme (WT) rate (Table 1). Steady-state hydrolysis rates were measured at a concentration of 35 μM malonyl-CoA, which is the concentration of malonyl-CoA in exponentially growing *E. coli* as measured by Bennett and coworkers [9]. These values likely represent k_{cat} values, as the K_m for cognate acyl-CoA hydrolysis by other acyltransferases has been previously measured to be in the low μM range [8]. As expected, when the active site serine was mutated to an alanine, hydrolysis was not detectable (Table 1, S641A and S641A+H640A). Despite the increased rate of malonyl-CoA hydrolysis, the H640A mutant was still capable of transferring the malonyl moiety to the ACP when incubated with 35 μM malonyl-CoA. Loading/transfer reactions were quenched with 50% acetonitrile after 20 seconds and we detected similar amounts of malonyl-S-

Table 1. Hydrolysis rates for yersiniabactin AT mutants at a concentration of 35 μM malonyl-CoA.

Variant	$v/[E]_o$ (min^{-1})	Fold increase over wildtype
Wildtype	0.39 \pm 0.09	1
H640A	3.01 \pm 0.32	7.7
S641A	n.d.	
H640A+S641A	n.d.	

n.d. = below limit of detection; error bars are the standard deviation of 3 replicates.
doi:10.1371/journal.pone.0109421.t001

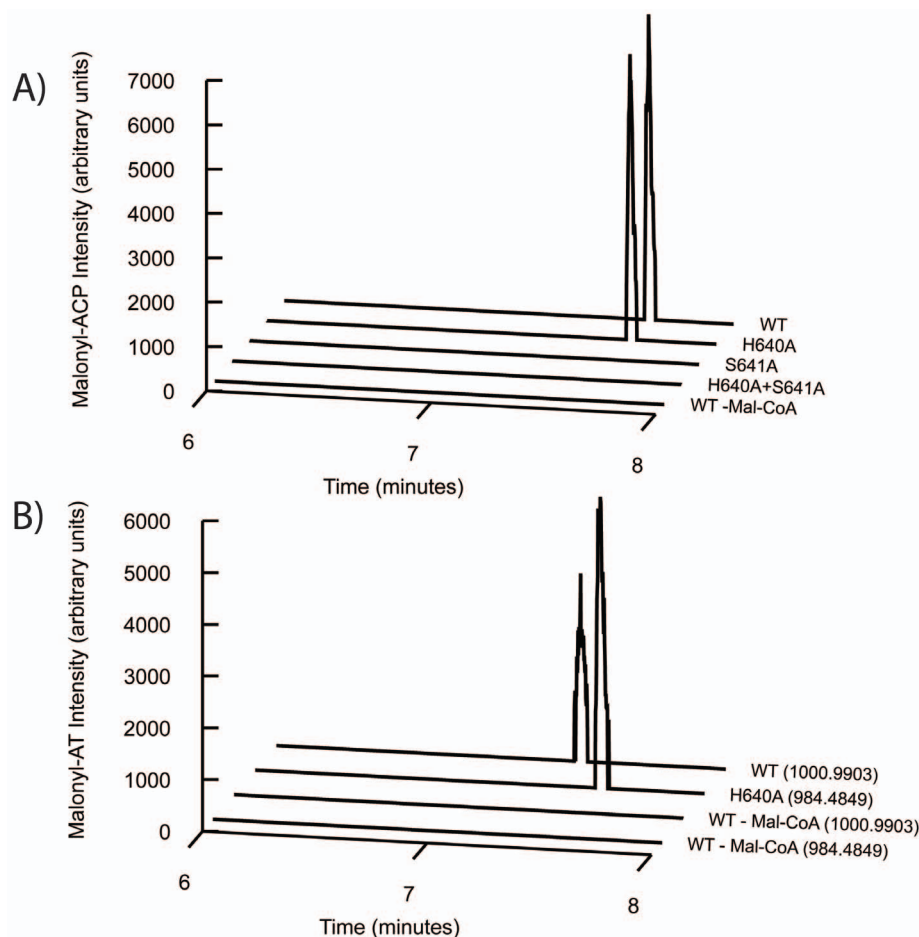


Figure 2. Acylation and transacylation activities of WT, H640A, S641A, and H640A+S641A. A) Transacylation activity as observed by high-resolution LC/QTOFMS. Data for the variants (WT, H640A, S641A, H640A+S641A) incubated with malonyl-CoA and a wildtype negative control without malonyl-CoA (WT - Mal-CoA). B) Formation of malonyl-AT complex for wildtype (monoisotopic peptide $m/z = 1000.9903$, $z = 4$) and H640A (monoisotopic peptide $m/z = 984.4849$, $z = 4$) as observed by high-resolution LC/QTOFMS. Data for wildtype and H640A (WT, H640A) incubated with malonyl-CoA as well as a wildtype negative control without malonyl-CoA (WT - Mal-CoA) are shown. The mass for each chromatogram is shown in parenthesis to the right. Additional details on chromatogram preparation in the Methods S1. doi:10.1371/journal.pone.0109421.g002

ACP (monoisotopic peptide $m/z = 1141.0562$, $z = 4$) for both wildtype and H640A (Figure 2 A). As with the hydrolysis assay, the S641A and S641A+H640A mutants had no detectable transacylation ability. Additionally, we observed malonyl-O-AT complex formation for both the wildtype (monoisotopic peptide $m/z = 1000.9903$, $z = 4$) and H640A (monoisotopic peptide $m/z = 984.4849$, $z = 4$) PKSs (Figure 2 B, additional details on chromatogram preparation in the Methods S1).

The question immediately arises as to why the removal of H640 increases the rate of hydrolysis. To address this issue, we examined the structure of the AT from DynE8, which has a covalently bound malonate (Figure 3) [11]. DynE8 is an iterative PKS involved in enediynes biosynthesis, and the DynE8 AT has 28% amino acid identity to the yersiniabactin AT. In this structure, the histidine in the GHSxG motif forms a hydrogen bond with a conserved asparagine in the ferredoxin-like subdomain (Figure 3). Through this hydrogen bond, the histidine may structurally stabilize the protein, reducing the hydrolysis rate. A water-mediated hydrogen bond between histidine in the GHSxG motif and the carbonyl oxygen in the AT ester bond is also present in the DynE8 structure (Figure 3), potentially stabilizing the malonyl-O-

AT complex and slowing the rate of hydrolysis. We propose two possible mechanisms for this stabilization: either the water-mediated hydrogen bond favors the sp^2 hybridization of the carbonyl oxygen in the malonyl ester bond over the sp^3 hybridization of the tetrahedral transition state for hydrolysis; or the ordering of water provided by the hydrogen bonding network prevents water molecules from attacking the nearby carbonyl. We speculate that as the functional oxyanion hole forms upon the AT binding the ACP to facilitate transacylation (as proposed by Keatinge-Clay and coworkers) [1], the stabilizing hydrogen bonding interaction would be disrupted. Future crystallographic studies of AT active site mutants may elucidate the exact mechanism of stabilization.

In summary, we have proposed a stabilizing role of the conserved histidine in the GHSxG active site motif of the yersiniabactin synthase AT domain. To generalize, this suggests that ATs have evolved to protect acyl intermediates, functionally diverging from their α/β hydrolase relatives. Future work examining analogous histidine to alanine mutations in other AT domains would further support the role of the conserved histidine in stabilizing the malonyl-O-AT complex.

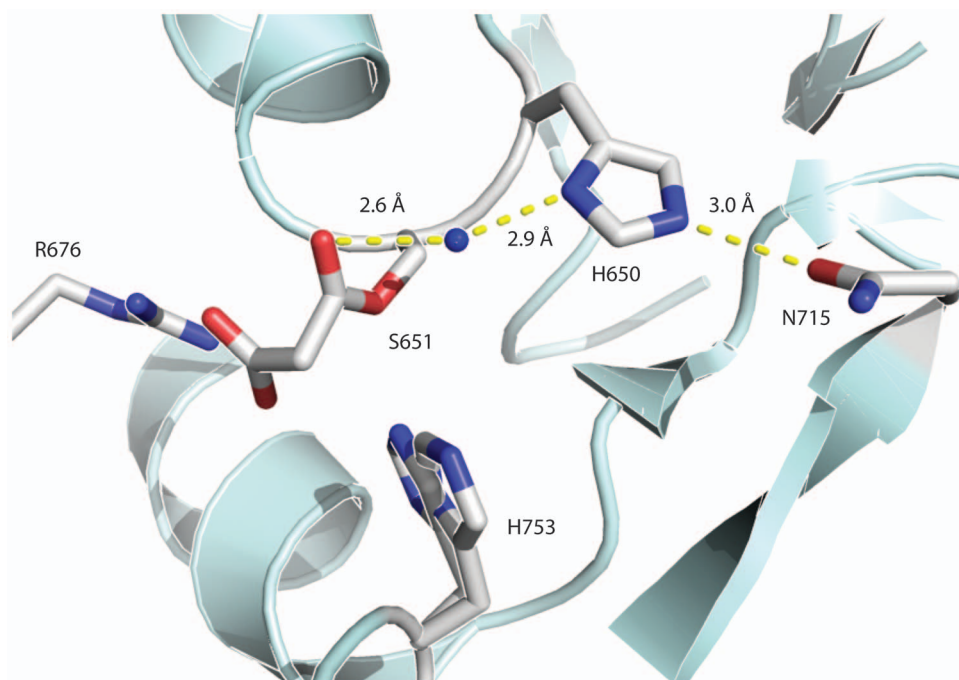


Figure 3. Crystal structure with malonate of the acyltransferase from DynE8, an iterative type I PKS. Residues involved in catalysis are labeled and shown as sticks. The hydrogen bonding water highlighted in the text is shown as a sphere. Figure 3 was prepared using Pymol from the PDB entry 4AMP [11].

doi:10.1371/journal.pone.0109421.g003

Supporting Information

Methods S1 Detailed materials and methods. (DOCX)

Author Contributions

Conceived and designed the experiments: SP LK CJP JDK. Performed the experiments: SP IY. Analyzed the data: SP PDA LK CJP JDK. Contributed reagents/materials/analysis tools: PDA. Wrote the paper: SP LK CJP JDK.

References

- Keatinge-Clay AT, Shelat AA, Savage DF, Tsai S-C, Miercke IJW, et al. (2003) Catalysis, Specificity, and ACP Docking Site of *Streptomyces coelicolor* Malonyl-CoA:ACP Transacylase. *Structure* 11: 147–154.
- Serre L, Verbree EC, Dauter Z, Stuijje AR, Derewenda ZS (1995) The *Escherichia coli* Malonyl-CoA:Acyl Carrier Protein Transacylase at 1.5-Å Resolution: Crystal Structure of a Fatty Acid Synthase Component. *J Biol Chem* 270: 12961–12964.
- Bateman A, Coin L, Durbin R, Finn RD, Hollich V, et al. (2004) The Pfam protein families database. *Nucleic Acids Res* 32: D138–D141.
- Dreier J, Li Q, Khosla C (2001) Malonyl-CoA:ACP Transacylase from *Streptomyces coelicolor* Has Two Alternative Catalytically Active Nucleophiles. *Biochemistry* 40: 12407–12411.
- Szafrańska AE, Hitchman TS, Cox RJ, Crosby J, Simpson TJ (2002) Kinetic and Mechanistic Analysis of the Malonyl CoA:ACP Transacylase from *Streptomyces coelicolor* Indicates a Single Catalytically Competent Serine Nucleophile at the Active Site. *Biochemistry* 41: 1421–1427.
- Miller DA, Luo L, Hillson N, Keating TA, Walsh CT (2002) Yersiniabactin Synthetase: A Four-Protein Assembly Line Producing the Nonribosomal Peptide/Polyketide Hybrid Siderophore of *Yersinia pestis*. *Chem Biol* 9: 333–344.
- Molnos J, Gardiner R, Dale GE, Lange R (2003) A continuous coupled enzyme assay for bacterial malonyl-CoA:acyl carrier protein transacylase (FabD). *Anal Biochem* 319: 171–176.
- Dunn BJ, Cane DE, Khosla C (2013) Mechanism and Specificity of an Acyltransferase Domain from a Modular Polyketide Synthase. *Biochemistry* 52: 1839–1841.
- Bennett BD, Kimball EH, Gao M, Osterhout R, Van Dien SJ, et al. (2009) Absolute metabolite concentrations and implied enzyme active site occupancy in *Escherichia coli*. *Nat Chem Biol* 5: 593–599.
- Bumpus SB, Kelleher NL (2008) Accessing natural product biosynthetic processes by mass spectrometry. *Curr Opin Chem Biol* 12: 475–482.
- Liew CW, Nilsson M, Chen MW, Sun H, Cornvik T, et al. (2012) Crystal Structure of the Acyltransferase Domain of the Iterative Polyketide Synthase in Eneidyne Biosynthesis. *J Biol Chem* 287: 23203–23215.
- Wheeler TJ, Clements J, Finn RD (2014) Skylign: a tool for creating informative, interactive logos representing sequence alignments and profile hidden Markov models. *BMC Bioinf* 15: 7.