

Energy Landscapes for Base-Flipping in a Model DNA Duplex

Published as part of *The Journal of Physical Chemistry virtual special issue "Jose Onuchic Festschrift"*.

Nicy,* Debayan Chakraborty,* and David J. Wales*



Cite This: *J. Phys. Chem. B* 2022, 126, 3012–3028



Read Online

ACCESS |



Metrics & More

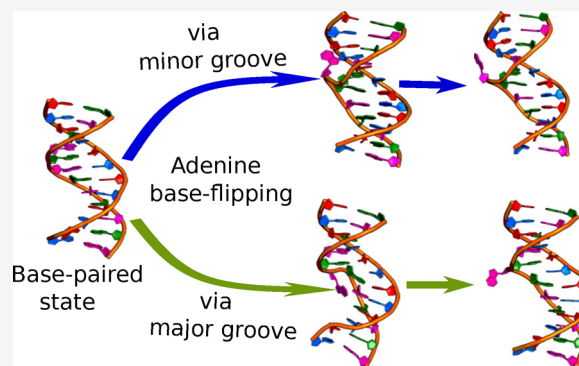


Article Recommendations



Supporting Information

ABSTRACT: We explore the process of base-flipping for four central bases, adenine, guanine, cytosine, and thymine, in a deoxyribonucleic acid (DNA) duplex using the energy landscape perspective. NMR imino-proton exchange and fluorescence correlation spectroscopy studies have been used in previous experiments to obtain lifetimes for bases in paired and extrahelical states. However, the difference of almost 4 orders of magnitude in the base-flipping rates obtained by the two methods implies that they are exploring different pathways and possibly different open states. Our results support the previous suggestion that minor groove opening may be favored by distortions in the DNA backbone and reveal links between sequence effects and the direction of opening, i.e., whether the base flips toward the major or the minor groove side. In particular, base flipping along the minor groove pathway was found to align toward the 5' side of the backbone. We find that bases align toward the 3' side of the backbone when flipping along the major groove pathway. However, in some cases for cytosine and thymine, the base flipping along the major groove pathway also aligns toward the 5' side. The sequence effect may be caused by the polar interactions between the flipping-base and its neighboring bases on either of the strands. For guanine flipping toward the minor groove side, we find that the equilibrium constant for opening is large compared to flipping via the major groove. We find that the estimated rates of base opening, and hence the lifetimes of the closed state, obtained for thymine flipping through small and large angles along the major groove differ by 6 orders of magnitude, whereas for thymine flipping through small angles along the minor groove and large angles along the major groove, the rates differ by 3 orders of magnitude.



INTRODUCTION

The localized distortion within the DNA duplex, in which a single base breaks its hydrogen-bonding with the complementary base and unstacks out of the helix by rotating about 180° into an extrahelical position, is known as base-flipping.^{1–3} This process may occur spontaneously on a sufficiently long time scale (passive flipping), or an enzyme may be required to drive it (active flipping).² Enzymes that are known to interact with a flipped-out base include methyltransferase (which can methylate cytosine (C) or adenine (A)),^{4,5} glycosylase (which removes thymine (T) or uracil (U) from a mismatched base-pair (bp)),^{6–8} endonuclease,⁹ integrase, helicase, polymerase, photolyase,^{2,10} and recombinase.¹ These enzymes facilitate base-flipping out of the helix for several purposes. The first purpose is to access the genetic information contained within the duplex. The second purpose is to chemically modify the base, thereby influencing gene regulation. The most prominent example is base methylation, which plays an important role in epigenetics.¹¹ The third purpose is to recognize and repair the damaged or chemically modified bases. Finally, the last one is to repair the mismatched base-pairs generated due to errors in copying by

polymerases.^{12–15} Base-flipping may also play a role during transcription and replication.^{16,17}

Defects in the working of enzymes associated with base-flipping are linked to several diseases. For example, the DNA repair machinery does not work efficiently in patients with xeroderma pigmentosum.¹⁸ Recently, defects in DNA glycosylases have been linked with colorectal cancer.^{19,20} Furthermore, by selectively hindering the repair pathway, it is possible to obtain improved antibiotics.²¹ For instance, hydrogen peroxide used in ion beam therapy for the treatment of cancer has been shown to stabilize base-pairs, making it difficult for enzymes to flip out the base during nucleotide excision repair (NER) and mismatch repair (MMR).²² Hence, inhibiting the DNA repair machinery can kill cancer cells and may thus provide a route to treatment.^{23–25}

Received: January 15, 2022

Revised: March 24, 2022

Published: April 15, 2022

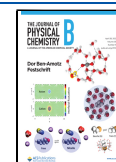


Table 1. Experimental Techniques Used for Studying Base-Flipping in a DNA Duplex

technique (advantage)	observation/prediction of lifetime of closed base-pair	limitations
X-ray (high resolution)	UDG follows major groove pathway ⁷ no lifetime information	- no information on dynamics ³ - low solubility of large macromolecules ³ - different crystal and solution structures ³
NMR imino-H exchange studies (monitor dynamics)	1–5 ms for AT bp ³⁸ 10–50 ms for GC bp ³⁸ 91–122 ms for AT tracts ⁶⁸ <5 ms for GC tracts ⁶⁹	- exact structure being monitored is not known - rate may be the rate of base wobbling ^{2,43–45,67} - uncertain whether the target base, or its partner base, or both, have flipped out ²
FCS/ddFCS ⁵² (monitor dynamics)	even in presence of enzymes, the lifetime obtained is of the order of seconds. ^{46–50} 0.3–20 s for GT mismatched bp ⁵³	- probe may not be specific to a base ² - alteration of natural structure of DNA by ⁷⁰ insertion of probes - indirect observation since probe is placed on base adjacent to the target base ⁷¹
AFM (monitor dynamics at nm resolution)	lifetime of closed bp even in absence of stacking interactions is of the order of seconds ⁵⁴	- results obtained depend on the interactions between the target molecule and the surface it is attached to during AFM study ⁷²
host–guest complexation (monitor dynamics)	around 1000 s using β -cyclodextrin ⁵⁸	- difficult to obtain base-specific host ² - macrocycle used to trap the flipped base may induce base-flipping as in case of Bisacridine

In 1925, Johnson and Coghill reported “The discovery of 5-Methyl-cytosine in tuberculinic acid, the nucleic acid of the tubercle bacillus”.²⁶ However, as their identification was based solely on the optical properties of picrate, their report was met with scepticism.²⁷ In 1948, the paper chromatography studies of Hotchkiss revealed the presence of epicytosine in thymus DNA.²⁸ Finally, in 1950, equipped with Markham and Smith’s technique for detecting ultraviolet-absorbing molecules on a paper chromatogram, Wyatt demonstrated the existence of 5-methylcytosine in plant and animal nucleic acids.^{27,29} Since then, methylation of bases has been studied extensively, mainly for understanding how it influences gene regulation.³⁰ It was during one such investigation in 1994 that Klimašauskas et al. first observed a cytosine base flipped outside a DNA helix. They detected this structure using X-ray crystallography for a ternary complex made up of *HhaI* DNA oligonucleotide, methyltransferase enzyme, and its cofactor.⁴ Earlier hydrogen-exchange kinetic studies using deuterium labeling and nuclear magnetic resonance (NMR) had already hinted at base-opening in nucleic acids.^{31,32}

Base-flipping has since been investigated using a variety of experimental techniques, listed in Table 1. While the X-ray structure reported by Klimašauskas had the target base flipped out into the active site of an enzyme, several other cases have been subsequently reported.^{3,4} The enzyme may interact with the target base by flipping out its partner base,^{3,33} or it may flip both bases in the base-pair.^{34,35} Furthermore, the enzyme may cause a significant distortion of the sugar-phosphate backbone.^{34,36} Although X-ray crystallography successfully captures the static structure of the molecule at high resolution, it does not provide dynamic information.³

NMR imino-proton (imino-H) exchange studies have been employed to monitor both spontaneous and enzymatic base-flipping in nucleic acids.^{37–40} This method is based on a two-state model, with the base either in a closed state or an open state. The imino-H on the N1 atom in guanine and the N3 atom in thymine can be exchanged when the base is in an open state.³⁸ Interestingly, solid-state NMR and F-NMR (fluorine-NMR)

studies have also explored the dynamics of base-flipping.^{3,24,41,42} Computational studies have revealed that the imino-H may be exchanged even when the base has moved slightly out of the helix.^{2,43–45} However, atomistic molecular dynamics simulations with standard force fields are not able to reproduce proton transfer reactions.

Several fluorescence correlation spectroscopy (FCS) studies have used 2-aminopurine and tetramethylrhodamine as fluorescent probes.^{46–51} However, it has not been possible to monitor spontaneous base-flipping effectively because of the time scale. In recent work using diffusion decelerated fluorescence correlation spectroscopy (ddFCS), Yin et al. determined the lifetime of a GT mismatched bp in a DNA duplex to be of the order of seconds.^{52,53} The lifetime of Watson–Crick (WC) base-pairs has been predicted to be longer than for mismatched base-pairs,^{52,53} and guanine–cytosine (GC) pairs are usually longer lived than adenine–thymine (AT) because of the third hydrogen-bond (H-bond).³⁸

Base-flipping has also been analyzed using atomic force microscopy (AFM),^{54–56} host–guest complexation,^{57–59} total internal reflectance fluorescence microscopy (TIRFM),⁶⁰ Förster resonance energy transfer (FRET),⁶¹ photochemical approaches,⁶² and chemical probes.^{63–66}

The fundamental question remains, how does the process of base-flipping occur at an atomic level of detail? What are the open states that are sampled during different experiments that make the flipping rates observed during NMR, and AFM, FCS and ddFCS⁵² differ by almost 4 orders of magnitude?⁵³ How do enzymes recognize specific sequences of DNA, mismatched base-pairs, and chemically modified and damaged bases during NER? Do they capture a base that is already flipped out, or do they drive the process of base-flipping?

It is evident from experiments that the time scale of base-flipping lies in the range of milliseconds to several hundred seconds or more. Hence, rare event methodology is required.

Some of the earliest investigations using molecular mechanics employed the FLEX force field in combination with the energy optimization program JUMNA (junction minimization of

nucleic acids).^{73–76} Subsequent studies combined all-atom force fields such as AMBER (assisted model building with energy refinement) and CHARMM (chemistry at Harvard macromolecular mechanics) with umbrella sampling.⁷⁷ This approach requires a predefined reaction coordinate (or order parameter), involving backbone torsions, distance and dihedral restraints.^{43,67,78,79} For biomolecular reactions that involve a complex rearrangement of atoms, order parameters may introduce bias,⁸⁰ and regions of configuration space that are separated by large barriers can be incorrectly lumped together.^{81–83} Conformational changes orthogonal to the reaction coordinate can also be important.⁸⁴

An alternative approach is to use collective variables. Conformational flooding⁸⁵ and metadynamics⁸⁶ simulations have exploited this methodology.⁸⁷ An adaptive sampling algorithm, in which the simulation is guided back and forth to obtain multiple paths, has also been used to study base-flipping,⁸⁴ while transition path sampling (TPS) provides another approach that requires order parameters.⁸⁸ Finally, two other schemes have been employed that make it possible to explore the landscape orthogonal to the reaction coordinate, namely the on-the-path random walk method,⁸⁹ which is a generalized ensemble sampling scheme, and selective integrated tempering sampling.⁵³

The potential energy landscape (PEL) framework used in the present work does not employ reaction coordinates. In recent studies,^{90–93} we have successfully exploited the PEL framework to probe complex conformational transitions in nucleic acids, and rationalize key experimental findings. We note that the order parameters used for analysis during this investigation were only used in postprocessing to identify DNA duplex structures with a particular base present in an extrahelical state. However, the calculation of rates does depend on the definition of reactant and product, as discussed below.

METHODS

A DNA duplex with the sequence $d(\text{GA})_6$ was first constructed using the nucleic acid builder (NAB) program in AMBER18.^{94–96} Here, d stands for double stranded, where the complementary base pairing is implicit. The duplex was modeled using the symmetrized version⁹⁷ of the AMBER99BSC0 force field^{98–101} along with torsional Olomouc corrections (χOL4).¹⁰² Symmetrization ensures that the permutational isomers have the same energy. The solvent and salt effects were treated implicitly using a generalized Born model (GB-OBC) based on the parametrization of Onufriev, Bashford, and Case.^{103,104}

This sequence has been analyzed by Giudice et al., who studied base-flipping using umbrella sampling.⁶⁷ We chose the same sequence to compare our results with the existing simulation study. Although there have been studies in the past describing the unusual structures adopted by oligopurine-oligopyrimidine sequences,¹⁰⁵ in our work we have considered a short sequence of 12 base-pairs in the canonical Watson–Crick double helical structure. The structures with bases in a flipped-out state were obtained by employing group rotation moves implemented within the global optimization program GMIN.^{106–108} The group rotation moves were performed by defining two different kinds of pivot points. The first pivot point is based on atoms $\text{C5}'$ and $\text{O3}'$ in the respective base. The second pivot point is based on atoms forming the glycosidic bond, i.e., $\text{C1}'$ and N9 for purines and $\text{C1}'$ and N1 for pyrimidines. We perform group rotation moves by defining

these two kinds of pivot points for each base that needs to be flipped. In the present study we have focused on four central bases, i.e., adenine, guanine, cytosine, and thymine, flipped out one at a time, to reduce edge effects for these relatively small systems.^{109,110} We note that experimental studies using NMR have shown that the length of the duplex and sequence can affect the barriers and flipping rates.¹¹¹

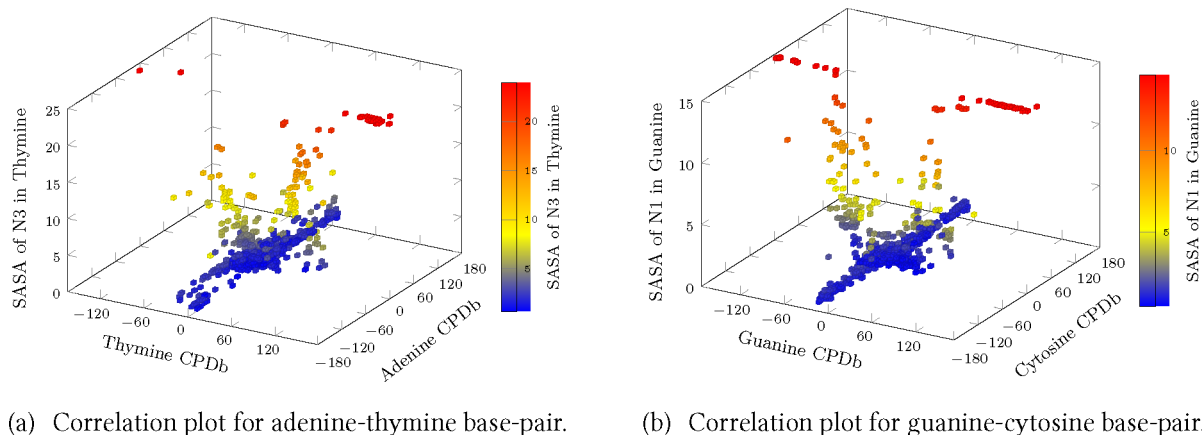
The center-of-mass pseudodihedral angle, CPDb, described by Song et al., was used to diagnose which base in the DNA strand is flipped out, to what extent, and toward which groove (i.e., major or minor).¹¹² For discrete path sampling (see Supporting Information) we require two end points. One end point was chosen as the lowest energy structure with all bases paired and the other was chosen with a base flipped out. Discrete path sampling employs pathway searches between successive pairs of end points in parallel. The structures with different bases flipped out (found using GMIN) were fed into PATHSAMPLE. Finally, several low energy states with the base flipped out to the maximum extent were considered for calculating barriers and rates. The flipped-out states reported in Table 4 are the subset with the lowest barriers and maximum rate constants for base-flipping.

We first need to obtain an initial connected pathway between the selected end point minima. The discrete path sampling approach was used to find pathways in terms of local minima and the transition states that connect them.^{113–116} The algorithms used within this procedure have been reviewed before,^{82,117,118} and we simply summarize the steps here:

- The doubly-nudged^{119,120} elastic band^{121–124} (DNEB) method was used to find candidate transition state geometries.
- Hybrid eigenvector-following was applied to obtain converged transition states from the candidates.¹²⁵
- The limited-memory Broyden–Fletcher–Goldfarb–Shanno (L-BFGS)^{126,127} minimization algorithm was employed to obtain the two local minima directly connected by each transition state.¹¹⁶
- The missing connection algorithm¹²⁸ was then used to choose pairs of minima for further double-ended connection attempts until a fully connected pathway between the initial and final states was obtained.

For a system of N atoms with $3N$ degrees of freedom, the potential energy landscape is a $3N$ -dimensional surface in a $3N + 1$ -dimensional space.¹¹⁶ To visualize this multidimensional surface we employ disconnectivity graphs¹²⁹ where the potential (or free) energy is represented along the vertical axis. The equally spaced nodes located along this axis represent different superbasins. All the minima within the database are divided into these superbasins, which form disjoint sets. The barrier for interconversion of minima lying within the same superbasin is less than or equal to the threshold energy.^{129,130} Branches arising from these nodes represent the individual minima and terminate at the energy of a particular minimum.

Once an initial connected pathway has been found, large barriers may result due to incomplete sampling. Lower barrier pathways exist but have not yet been found, and further sampling is required. Various schemes¹³¹ to locate such pathways were employed in the present work. These schemes are implemented within the PATHSAMPLE program. Convergence of the sampling was monitored via inspection of disconnectivity graphs and computation of interconversion rates between target minima.^{132,133}



(a) Correlation plot for adenine-thymine base-pair.

(b) Correlation plot for guanine-cytosine base-pair.

Figure 1. Correlation plots between the CPDb dihedral and SASA calculated using POPS. The CPDb dihedral angle is measured in degrees, and the SASA is measured in \AA^2 .

Rates were extracted from the stationary point databases using graph transformation.^{134–136} The individual minimum-to-minimum rate constants were calculated using transition state theory in the harmonic approximation, as for local thermodynamic properties. The reactant state was considered to be the closed WC base-paired state, and the product state was taken to be an open state with one of the bases flipped out. The inverse rates in the backward and forward direction were taken as the lifetime of product and reactant, i.e., open and closed states, respectively.

RESULTS AND DISCUSSION

Geometric Characterization of Base-Flipping. An order parameter was required to diagnose which base in the DNA strand was flipped out, to what extent it was flipped out, and toward which of the major or minor grooves it was oriented. The geometric parameters used for studying base-flipping are generally based on distances, angles and dihedrals, such as the distance between the N1 atom in purine and the N3 atom in pyrimidine.^{137,138} This parameter was also used in a recent computational study in which the authors compared their calculations with NMR observations without assuming that flipping was favored toward a particular groove.⁸⁴ An alternative distance parameter is based on hydrogen-bonding atoms.⁵³ The base plane rotation angle has also been employed.^{76,79}

For these distance and angle parameters, it is not possible to identify whether the base has flipped toward the major or minor groove. While a larger distance indicates an open base, it is not possible to say exactly which of the two bases in a base-pair has flipped out, or if both the bases have flipped out simultaneously.² To overcome this limitation, dihedral angles were used to classify flipped-out states.⁶⁷ Several previous studies^{43,112} have utilized pseudotorsions, which successfully predict the flipping out process, and indicate in which direction the flipping occurs. Although there might be some issues when using these pseudotorsion angles as reaction coordinates, in our work we have only used them in postanalysis. Usually, one point in the dihedral is taken to be the base-pair flanking the base of interest, or the target base that is flipped out. This formulation emerges from one of the earlier studies on an adenine bulge.¹³⁹ The first dihedral that was widely utilized was the center-of-mass (COM) pseudodihedral, also called the CPD.⁴³ However, later it was found that this definition suffered from several limitations, i.e., conformations with similar CPDs differed significantly.¹¹² Song

et al. later defined another set of dihedrals: CPDa and CPDb. The CPDb dihedral was reported to work better and hence was used during the present analysis. The four sets of atoms considered for CPDb are the two base-pairs flanking the base of interest, the phosphate group on the 3' side of the flipping base, the phosphate group on the 5' side of the flipping base, and the ring atoms of the flipping base. It is important to note that in the case of purine flipping, only atoms of the five-membered ring are considered for the fourth point in the dihedral.¹¹² Additionally, one of the possible reasons for CPDb being a better order parameter is that the second and third points in the CPDb dihedral lie close to the target base, whereas in CPDa the sugars of the neighboring base that are considered as the second and third points lie further away from the target base.

There are several torsion angles defined within the DNA backbone: α , β , γ , δ , ϵ , and ζ , and the glycosidic torsion angle χ .¹⁴⁰ The changes in these dihedrals as a function of base-flipping have also been investigated in the past.^{43–45,67,141} However, these torsion angles have not been evaluated in the present study, primarily because the DNA structures that have been sampled in the current database may have local distortions in the backbone. Analyzing torsion angles localized on the flipping base, when any part of the entire backbone may be deformed, produces ambiguities. Analysis of the torsional profiles with base-flipping could provide important information about which torsions are crucial in this process. Furthermore, it could also yield information about any potential force field artifacts.

More recently, the solvent-accessible surface area (SASA) has been calculated to compare computational studies with NMR observations,^{142,143} highlighting limitations of NMR imino-H exchange results. Even when a base is flipped out slightly (approximately 30°), its SASA is large enough for the imino-H to be exchanged with the solvent.² The magnitude of error in SASA calculations can be comparable to the SASA itself,^{144–149} which makes it less useful as an order parameter.

Energetic parameters are based upon the interactions between different atoms in the system, such as the flipping base and its partner, stacking interactions between adjacent bases, and interactions between hydrogen-bonding groups.^{67,88} The interaction between the base of interest and solvent has also been investigated using hydration number and solvent distribution.^{67,88} However, energetic factors were not employed in the present work.

The CPDb angles for the structures in the database were calculated using the AMBER trajectory analysis tool CPPTRAJ.¹⁵⁰ The SASA of the N1 atom in guanine and N3 atom in thymine were calculated using the parameter optimized surfaces (POPS) program.^{147–149} These specific atoms were chosen because they correspond to the imino-H, where the exchange can be monitored during NMR studies.

The correlation between CPDb and SASA is shown in Figure 1. While similar correlation plots have been presented in the past,¹⁵¹ here, they are explicitly for CPDb. The key observation is that the SASA of N1 in G (or N3 in T) may be high even when CPDb is relatively small (around 30–40°). The SASA may also be high even when the partner base, i.e., C (or A) is flipped out. Therefore, SASA alone cannot be used to identify if a particular base is flipped out or if both bases are flipped out.

In the present work, positive CPDb angles correspond to the minor groove and negative CPDb angles to the major groove. These signs are the reverse of what has been reported before¹¹² due to the definition of the CPDb dihedral and the method used to calculate dihedral angles implemented within the CPPTRAJ tool.¹⁵⁰

Topography of Energy Landscapes. The free energy landscapes for flipping adenine, guanine, cytosine and thymine are shown using four disconnectivity graphs (Figures 2 and 3),^{152,153} colored on the basis of the CPDb dihedral angle for individual bases and an overall disconnectivity graph for the entire landscape (Figure 4). The graphs in Figures 2 and 3 were constructed by removing flipped-out minima for three of the four bases in each case.

Interestingly, the distribution of dihedral angles in the free energy landscapes of adenine and guanine is similar to that of their partner bases, i.e., thymine and cytosine. For both adenine and thymine, the bases flipped out slightly toward the minor groove lie at the bottom of the energy landscape. In contrast, for guanine and cytosine, the bases flipped out by small angles (30–60°) along the major groove lie lowest. However, for guanine, some of the minima with bases flipped into the minor groove are also low in energy. Similarly, for thymine, some of the configurations with bases flipped by small and large angles into the major groove are relatively favorable. Purines, i.e., adenine and guanine, apparently may prefer to flip via the minor groove. The local distortions of the backbone and widening of the minor groove facilitate this pathway for the sequence under consideration. We find that configurations with the bases flipped out by large angles via the major groove usually have lower free energy than for a minor groove pathway. This observation is consistent with previous work, which reported that the energy of bases flipped out via the minor groove was slightly higher than those that had flipped via the major groove. However, spontaneous conversion between the two flipped-out states was not possible, as the backbone conformations differed significantly.⁷⁹

In all four disconnectivity graphs, minima with similar dihedral angles are found at different energy levels. There are various possible explanations for this observation. First, each color represents a range of angles spanning over 40°. This spread implies that the same colored branch present in a higher energy region may have the base flipped out by a larger angle. This possibility was checked by decreasing the range of angles grouped together and recoloring the disconnectivity graphs. Second, for the base flipped out toward the major groove, the pathway that was sampled in the database might involve the base flipping via the minor groove, and by traversing an angle greater

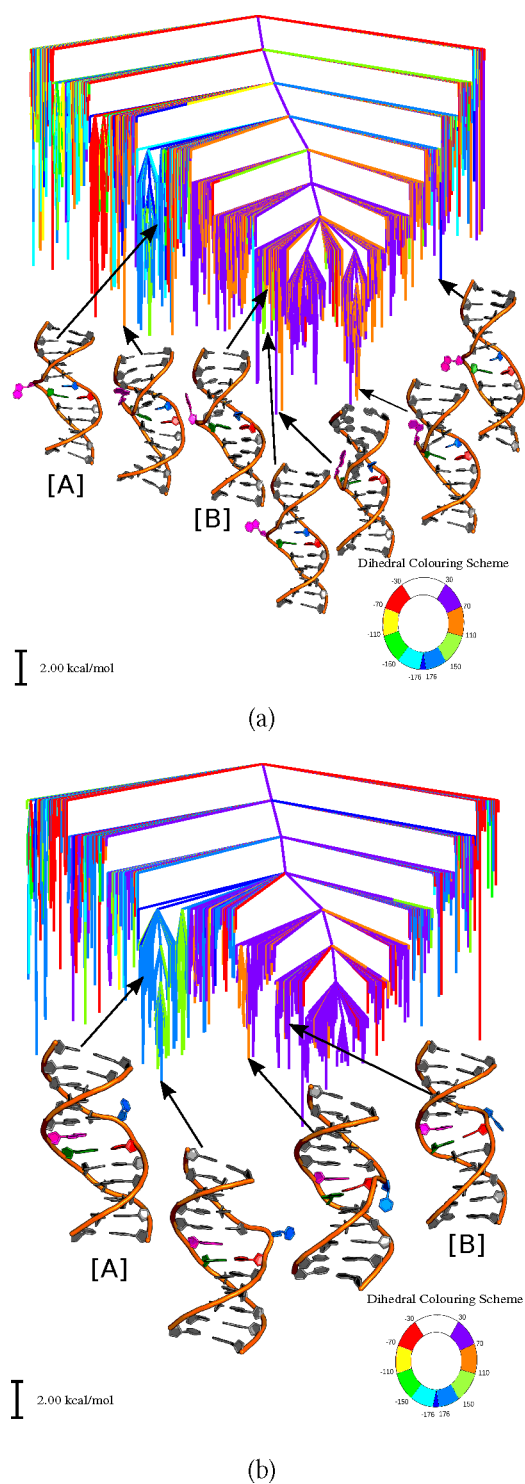


Figure 2. Free energy disconnectivity graphs for (a) adenine and (b) thymine bases flipped out of a DNA duplex at 300 K. The DNA structures labeled as [A] and [B] in the graphs represent the final flipped-out states considered in the plots in Figure 7. [A] and [B] represent the base flipped out via major and minor grooves, respectively.

than 180°, the base could have reached a dihedral associated with the major groove. A similar case was observed for bases flipped out toward the minor groove, where the pathway was found to be via the major groove. In both cases, the minima corresponding to such pathways lie higher in energy. In future

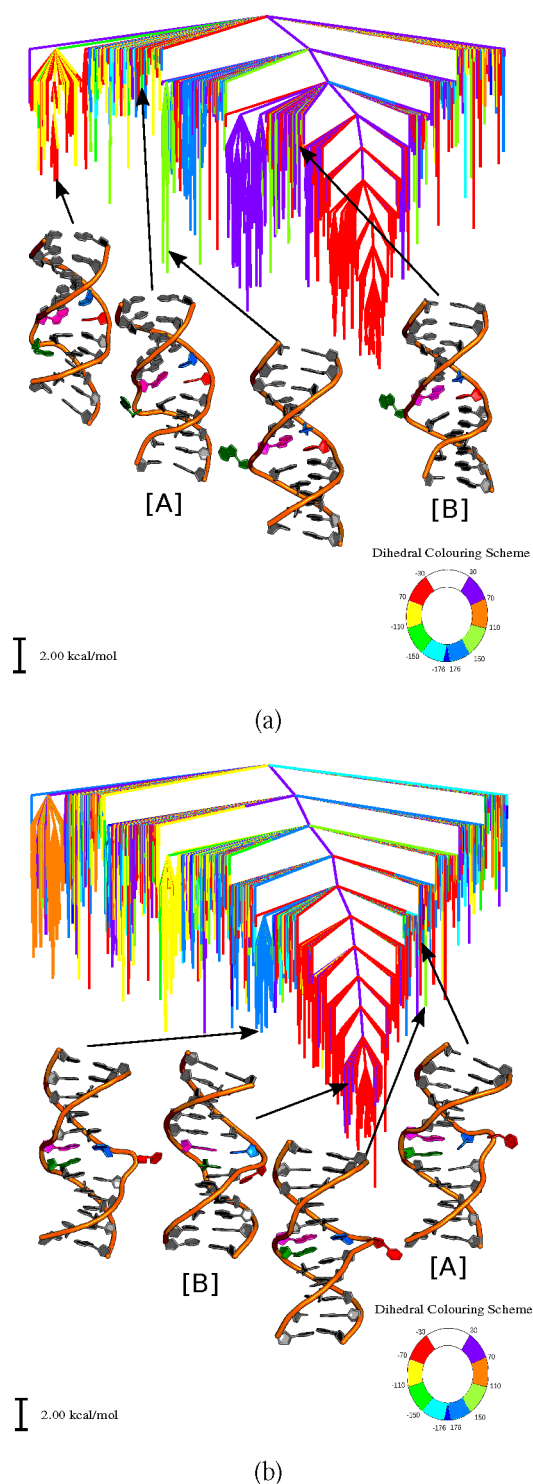


Figure 3. Free energy disconnectivity graphs for (a) guanine and (b) cytosine bases flipped out of a DNA duplex at 300 K. The DNA structures labeled as [A] and [B] in the graphs represent the final flipped-out states considered in the plots in Figure 7. [A] and [B] represent the base flipped out via major and minor grooves, respectively.

work, it would be interesting to check if there is any difference in the backbone conformations of the bases flipped via the major and minor groove, as suggested in a previous study.⁷⁹ Third, a base flipped out by a smaller angle may be located higher in energy. This result is possible when the pathway involves an

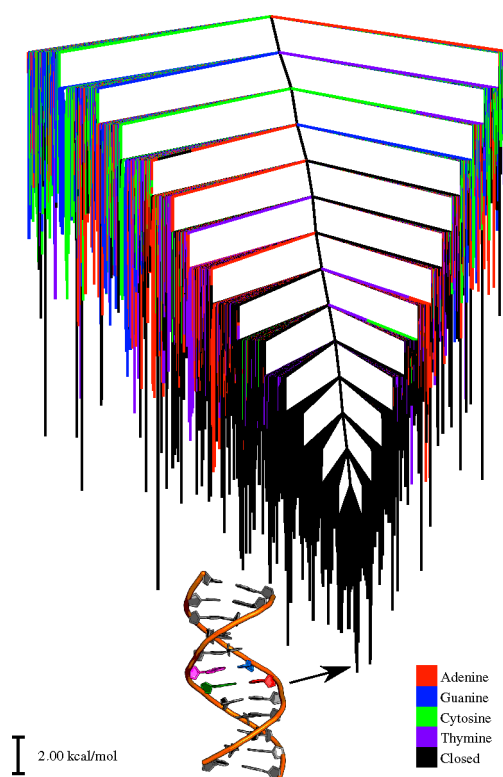


Figure 4. Free energy disconnectivity graph (at $T = 300$ K) including closed states and single base flips for all the four bases. The bases adenine, guanine, cytosine and thymine have been considered to be open if their CPDb dihedral is greater than 30 or less than -30 degree. The disconnectivity graphs in Figures 2 and 3 represent the landscape for flipping of individual bases separately. The same landscape is shown with different colors for minima featuring the four alternative flipped-out bases.

intermediate state in which the base was flipped out by a larger angle, and then returned back to the final smaller angle. This effect may indicate a need for further sampling. Fourth, for a base flipped via the minor groove, the minima that are located toward the bottom of the graph usually have their backbones distorted, and a broad and distorted minor groove. In the absence of such distortions, the pathways lie higher in energy. In future work, we will examine the impact of bending on the energy landscapes for base-flipping more systematically.

It is evident that local distortions of the DNA backbone and subsequent broadening of the minor groove is necessary to facilitate base opening via the minor groove pathway. A similar trend has been reported in a previous study.¹⁵⁴ Bending and opening of bases may be synergistically related, i.e. the bent backbone decreases the barrier, and the base-flipping makes the backbone more flexible, facilitating further bending.¹⁵⁴ Bending decreases the energy needed to overcome the base–base interactions. In a bent backbone, there may be an accumulation of energy in the form of strain, which base-flipping may help to relieve.⁷³ We note that the CPDb dihedral is no longer reliable in classifying whether the base lies toward major or minor groove side when the backbones are bent. In addition, for angles close to 170° , it is difficult to classify whether the base has followed a major or minor groove pathway. Again, this ambiguity is because of associated distortions observed at some stage during most of the base flipping pathways.

A recent study suggested that it is possible to have comparable barriers for adenine and thymine flipping in a DNA duplex.⁸⁴ In the present work, the barrier for flipping adenine is higher than for thymine for major groove flipping pathways. However, the barriers for flipping A and T by large angles are comparable for minor groove flipping pathways. In an earlier report the barrier for flipping guanine was found to be higher than for cytosine.⁴³ A contradictory observation was made in a subsequent study,¹⁵⁵ and another recent computational investigation again suggests that cytosine is more prone to flipping in a GC base-pair.⁸⁴ Results from the conformational flooding approach suggested that the barriers for flipping guanine and cytosine were comparable for the minor groove pathway. In the present work, the barrier for guanine flipping via the minor groove was found to be comparable to cytosine flipping via the major groove.⁸⁵ The barrier for spontaneous flipping of undamaged bases was reported to be comparable by Zheng et al.¹⁵⁶

To the best of our knowledge the only research in which exactly the same sequence was considered used umbrella sampling with a reaction coordinate that did not allow for DNA backbone distortions.⁶⁷ Our estimated barriers may be quantitatively different from previous studies because the values depend on the sequence, the force field, the sampling method, and the choice of reactant and product states.⁸⁴

Mechanisms of Base-Flipping. Base-flipping in a DNA duplex is a multistep process that involves a sequence of events occurring in a specific order. However, this order is not always strictly obeyed because several steps may occur in a concerted fashion. We distinguish eight distinct pathways for spontaneous flipping of adenine, guanine, cytosine, and thymine via the major and minor grooves, as summarized in Table 2 and Table 3. Table 2 lists the observed events, with an alphabetical code to define the pathways in Table 3. Some of the intermediates and transition states are also illustrated (Figures 5 and 6).

A key finding of the present analysis is that there may be a strong correlation between the sequence and the groove along which the base flips, so the flipping rate may be sequence dependent. When the base flips into the minor groove, it interacts with the base pairs on its 5' side. In contrast, the flipping base usually interacts with the base pairs on its 3' side when following the major groove pathway.

The events that occur during base-flipping can be classified into five broad categories: breaking of hydrogen-bonds, coupled motion of neighboring bases, alignment of the flipping base with its backbone, interaction of the open base with the backbone and other nearby bases, and bending (local distortion) of the backbones accompanied by distortion of grooves.

Recent computational studies suggest that base-flipping starts with the loss of hydrogen-bonding in the WC base-pair, i.e., base-pair opening.⁸⁴ This opening may occur either by linear separation of strands, leading to base-plane elongation, or by twisting the base out of its plane.^{88,89} The separation of backbones by increasing the interphosphate distance before base pop-out has also been indicated in an earlier X-ray crystallography study on an enzyme–DNA complex.¹⁵⁷ A similar observation has been made for cytosine flipping via the major groove. Conversely, strand separation during replication may be initiated by base-flipping.⁴³ Once the base has opened, it may then unstack. Previous simulations have proposed that unstacking may occur after or simultaneously with the base-pair opening.⁸⁸

The flipping of an individual base may be accompanied by the coupled motion of one or more of three other bases: the WC

Table 2. Different Steps Observed During Base-Flipping with an Alphabetical Code Assigned to Each One, for Use in Table 3

event	description
	breaking of hydrogen-bonds may take place via,
S	linear separation of backbones, resulting in base plane elongation
Bb	bending of backbone containing the base being flipped out
C	concerted motion and bending (local distortion) of both backbones
U	unstacking of base with slight flipping
Dg	distortion of grooves: minor groove broadens and major groove narrows
D'g	the distorted groove reverts back to the original undistorted state
	after the base first moves out the following events may occur,
B	both backbones may bend further
B'	bent backbones straighten
N3	coupled motion of neighboring base on the 3' side of the same strand, either to maintain stacking with base being flipped out, or to interact with the orphan WC partner
N5	coupled motion of base or base-pair on the 5' side of the base being flipped out to maintain similar interactions as above
A5	the flipped base vertically aligns along its own backbone on 5' side
A3	the flipped base vertically aligns along its own backbone on 3' side
	origin of sequence effects
Ib'	the flipped-out base interacts with the backbone or/and bases of complementary strand that may be bent to further facilitate this interaction
Ib	the flipped base may interact with the backbone or/and bases of its own strand that maybe bent.
	final flipping out
F	the base may flip further out
R	the neighboring bases that moved during coupled motion may move back within the helix to maintain their own base pairing

partner of the base, or the bases on its 5' or 3' sides. Earlier computational studies note that the WC partner base may follow the flipping base by moving toward the same groove. The possibility of the WC partner moving toward the opposite groove was also reported.^{67,84} Subsequent simulations have revealed that when purines flank the base that is being flipped out, they may move with it to retain the stacking interactions.¹⁵⁸ A similar effect has been observed during the present analysis for guanine flipping via the major groove pathway. The adenine on its 3' side breaks hydrogen-bonding interactions with its partner base and aligns its plane to stack with the flipping base. Several other studies have also shown the importance of stacking interactions in the case of purines.⁶⁷ However, it remains to be seen whether the stacking interactions are essential when a purine is flipped out or when purines flank the flipping base.

Interestingly, the base on the 3' side of the flipping base has been found to undergo significant distortion in both guanine and cytosine flipping via the major groove. This observation agrees with a previous study in which a similar distortion of bases on the 3' side of the damaged base was reported.¹⁵⁹ However, the base on the 5' side has also been shown to undergo coupled motion when cytosine flips along the minor groove pathway. Moreover, when cytosine flips either via the major or minor groove, the partner base guanine changes its orientation so as to interact with the neighboring bases, such as thymine, on the opposite strand (Figures 5, parts o and r), either by forming hydrogen bonds, or by interacting with the sugar-phosphate backbone. In some cases when purine flips out, the partner pyrimidine base was found to change its orientation, either to better stack with the pyrimidine on its 5' side (Figure 5c), or to come in close contact with the base on its 5' side (Figure 5k).

Table 3. Observed Mechanisms for Flipping of Adenine, Guanine, Cytosine, and Thymine toward the Major and Minor Grooves^a

base	groove	sequence of events								
		S	C	Bb	U	A3	Ib	Dg	D'g	F
adenine	major	S	C	Bb	U	A3	Ib	Dg	D'g	F
	minor	S	C	Dg	U	B	A5	Ib'	F	
guanine	major	C	Dg	U	N3	A3	Ib'	Ib	F	A5
	minor	C	Dg	U			A5	Ib	F	
cytosine	major	S	C	U	A5	N3	B		F	
	minor	C	Bb	Dg	N5	U	A5	Ib'	F	R
thymine	major	S	Bb	U	Ib	A3	A5	C	F	B'
	minor	S	Bb	Dg	U		A5	Ib	F	

^aSee Table 2 for the codes.

Another remarkable experimental finding is the working of the enzyme thymine DNA glycosylase (TDG), which flips out T from a GT mismatch that has been formed from deamination of methylcytosine. How this enzyme knows the history of the nucleobase it interacts with is unknown.⁴² A recent F-NMR study has shown that there may be a link between the sequence that is more prone to methylation, and a similar sequence that has been observed to interact favorably with TDG.⁴² In any case, the enzyme differentiates between T in GT from the T in AT. This observation suggests an important role for the WC partner of the base being flipped out.

To the best of our knowledge, this is the first report linking major and minor groove pathways for base-flipping with the alignment of the base toward the 3' or 5' side of its backbone during an intermediate step, which shares some similarities with the e-motif structure first reported by Gao et al. and seen during recent computational work using the CHARMM27 force field.^{87,160} It consists of two cytosines in a CC mismatched base-pair aligned toward the 5' side of their respective backbones, with simultaneous flipping toward the minor groove. While this process is beyond the scope of the current investigation, in which only one base is flipped out at a time, it is worth highlighting that when any of the four bases flips toward the minor groove, the base aligns itself along its own backbone on the 5' side. In contrast, when the base flips via the major groove pathway, it usually aligns itself toward the 3' side of its own backbone. However, for cytosine and thymine, both alignments have been observed when the base follows the major groove pathway.

An interesting hypothesis is that when the alteration in sequence on the 5' side of the base being flipped out has a significant impact on the rates, the base is flipping via the minor groove pathway. Experimental confirmation could be achieved using a minor groove ligand and blocking the minor groove pathway. If the alteration in sequence on the 5' side then has a limited impact on rates, that would be consistent with the hypothesis. Similarly, when the sequence on the 3' side of the base is important, the base is probably following the major groove pathway. However, this hypothesis is subject to several caveats, highlighted below.

A computational study using the conformational flooding method suggested that, as the base cytosine flips out via the major groove, it first interacts with the backbone of the complementary strand and then with its own backbone.⁸⁵ In particular, the amino group in the nucleobase interacts with the anionic phosphate groups in the backbone.⁷⁹ The present work suggests similar interactions when purines are flipped out. Figure 6 depicts the polar contacts between the hydrogens of the amino

group and the oxygen of either a nearby base or the sugar or phosphate group in the backbone.

One of the main causes of sequence effects is the interaction of the flipping base with the nearby bases.^{43,67,161} However, these nearby bases are not limited to adjacent bases or next-nearest neighbors, but include more remote bases.⁴³ We observed polar contacts between the flipping base and its third- and fourth-nearest neighbors. This effect is evident in the following pathways: adenine flipping via the minor groove, and guanine flipping via the major and minor groove (Figure 5).

Although most computational studies report that flipping via the major groove is more favorable, several experiments have hypothesized that, since the enzymes interact with the DNA double helix from the major groove side, the base may prefer to flip toward the minor groove.^{4,78,162,163} Several reports have investigated base-flipping from a bent or unwound DNA duplex. A recent study using the CHARMM36 force field suggested that minor groove flipping may be favored by bent DNA.¹⁵⁴ In fact, there is an early report on DNA untwisting and bending, indicating that, as the DNA bends toward the major groove, its minor groove widens, thereby facilitating base opening via the minor groove pathway.^{73,76}

Kinetics of Base-Flipping. Base-flipping rates have been determined using several experimental techniques. NMR imino-H exchange studies indicate that the lifetimes of AT and GC base-pairs are 1 to 5 ms and 10 to 50 ms, respectively.^{37,38} AFM investigations suggest that the AT base-pair lifetime is of the order of seconds, even in the absence of stacking interactions.⁵⁴ Another approach involves the formation of a host-guest complex. The lifetime of purines has been reported to be around 1000 s using these complexation studies.⁵⁷⁻⁵⁹ More recently, a ddFCS study was performed on a mismatched base-pair, and the lifetime was found to be of the order of several seconds,⁵³ differing from NMR results by 4 orders of magnitude. The fundamental question that arises from these observations is, what pathways and open states are sampled by different experimental techniques that lead to the difference in base-pair lifetimes?

The equilibrium constant for base-flipping has been determined in several computational studies.^{2,43,79,164} Umbrella sampling suggested that "NMR imino proton exchange experiments on duplex DNA primarily monitor the opening of purine bases"¹⁵⁵ because the calculated rates for purines were significantly higher than for pyrimidines.¹⁵⁵ In contrast, another umbrella sampling investigation, utilizing a polarizable force field, highlighted the importance of sequence effects and suggested that there are cases when the rates of flipping pyrimidines are higher than for purines.¹⁶⁵

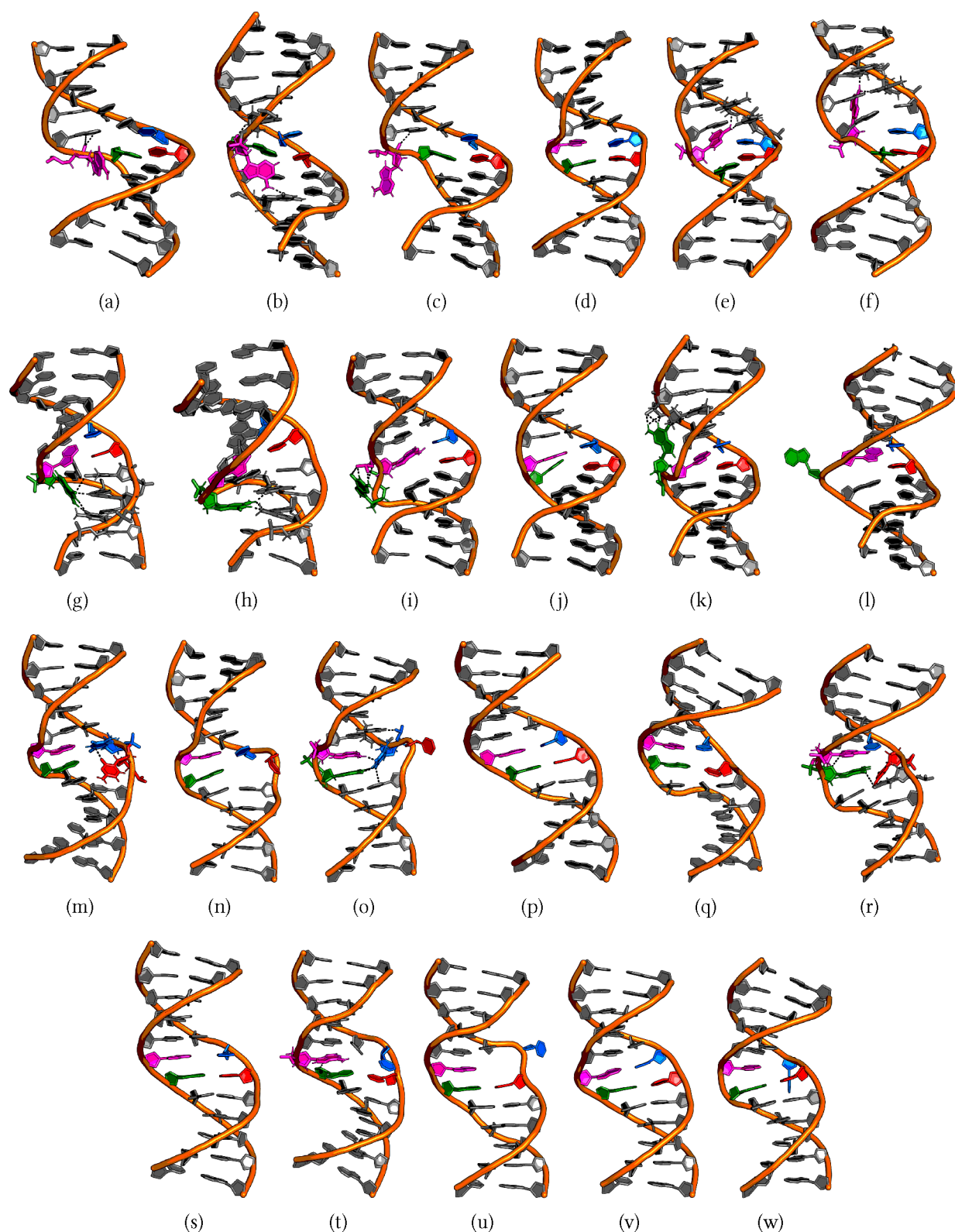


Figure 5. Snapshots of various steps in the base-flipping mechanism for individual bases. Adenine flipping via major (a–c) and minor groove (d–f), guanine flipping via major (g–i) and minor groove (j–l), cytosine flipping via major (m–o) and minor groove (p–r), and thymine flipping via major (s–u) and minor groove (v–w).

The present analysis estimates flipping rates of different bases by different extents along the major and minor grooves. Interestingly, the equilibrium constant of flipping guanine via the minor groove is significantly higher than via the major

groove (Table 4). This difference may be attributed to the increased stabilization possible on the minor groove side, where the flipped base can interact well with neighboring base pairs. The rate constant for opening guanine via the minor groove is

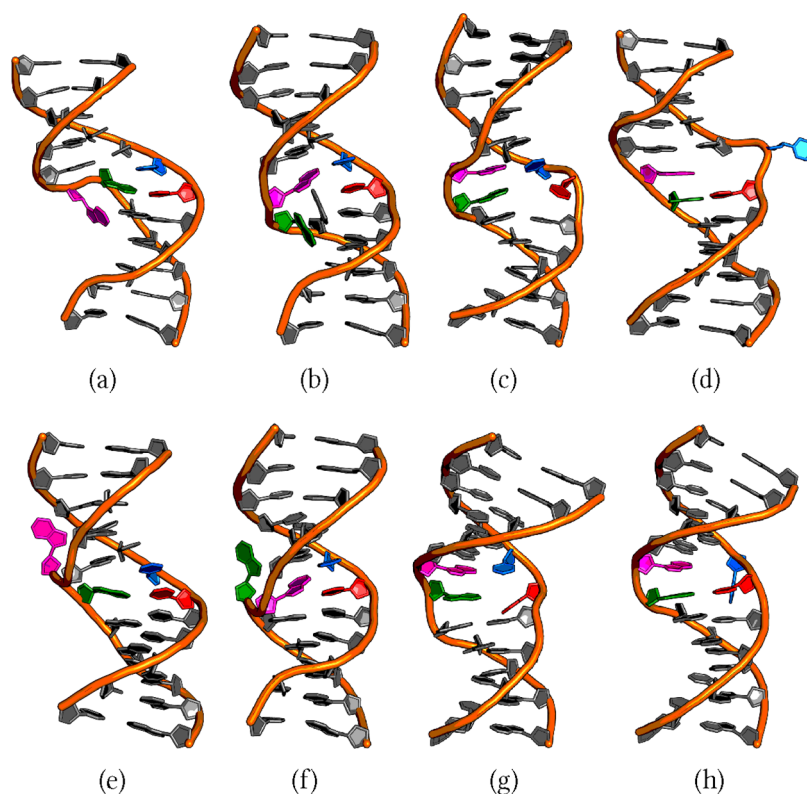


Figure 6. Panels a–d show adenine, guanine, cytosine and thymine flipping via the major groove, respectively, and panels e–h show adenine, guanine, cytosine and thymine flipping via the minor groove, respectively.

Table 4. Kinetic Data for Base-Flipping^a

Base	J or N	CPDb dihedral angle	barrier for opening	barrier for closing	rate of opening (s ⁻¹)	lifetime of closed state (s)	rate of closing (s ⁻¹)	lifetime of open state (s)	equilibrium constant
A	J	-172.11	25.51	10.60	1.09×10^{-8}	9.17×10^7	6.37×10^1	1.57×10^{-2}	1.70×10^{-10}
	N	126.56	15.35	5.04	1.62×10^{-2}	6.17×10^1	4.54×10^4	2.20×10^{-5}	3.57×10^{-7}
G	J	-47.99	15.50	1.47	1.49×10^{-2}	6.71×10^1	4.55×10^4	2.20×10^{-5}	3.27×10^{-7}
	N	-122.02	29.19	3.07	1.26×10^{-12}	7.94×10^{11}	1.62×10^{-1}	6.18	7.78×10^{-12}
		67.00	22.67	5.43	7.67×10^{-8}	1.30×10^7	8.96×10^{-4}	1.12×10^{-3}	8.56×10^{-5}
C	J	132.08	25.99	4.28	4.33×10^{-10}	2.31×10^9	8.96×10^{-4}	1.12×10^{-3}	4.83×10^{-7}
		-45.41	14.39	7.69	7.66×10^{-2}	1.31×10^1	4.97×10^3	2.01×10^{-4}	1.54×10^{-5}
	N	-124.29	22.84	2.46	5.28×10^{-8}	1.89×10^7	1.02×10^4	9.82×10^{-5}	5.19×10^{-12}
T	J	172.92	26.07	5.52	2.37×10^{-10}	4.22×10^9	2.84×10^2	3.52×10^{-3}	8.34×10^{-13}
		47.52	15.08	5.75	3.95×10^{-2}	2.53×10^1	4.55×10^4	2.20×10^{-5}	8.68×10^{-7}
	N	-46.83	11.19	0.80	1.63×10^1	6.12×10^{-2}	4.55×10^4	2.19×10^{-5}	3.59×10^{-4}
	J	177.71	19.29	6.05	3.16×10^{-5}	3.16×10^4	4.82×10^3	2.07×10^{-4}	6.55×10^{-9}
N	76.30	14.39	7.70	7.66×10^{-2}	1.31×10^1	4.97×10^3	2.01×10^{-4}	1.54×10^{-5}	
	121.28	14.79	3.22	2.61×10^{-2}	3.83×10^1	4.97×10^3	2.01×10^{-4}	5.26×10^{-6}	

^aThe letters “J” and “N” in the second column stand for the pathway via major and minor grooves, respectively. The barriers for opening and closing are the free energy barriers at 300 K in kcal/mol.

small, and the equilibrium constant for opening is high, because the corresponding flipped-out state is relatively stable, which decreases the rate of base closing. In a previous investigation on uracil DNA glycosylase (UDG), it was found that the enzyme increases the equilibrium constant for AT bp opening by stabilizing the open state.⁴⁰ Another NMR study revealed a similar trend.¹⁶⁶

It has been suggested that NMR studies analyze the rate of base-pair wobbling instead of full base-pair opening.^{2,43,44} In our calculations, the rates of base opening, and hence the lifetimes of bases in the closed state, for flipping thymine by small and large angles via the major groove differ by 6 orders of magnitude,

whereas for thymine flipping by small angles into the minor groove, and large angles into the major groove, the rates differ by 3 orders of magnitude. Since this difference is close to the difference in rates observed using NMR and ddFCS studies, it is possible that NMR imino-H exchange studies monitor base-flipping through small angles, while ddFCS reports on larger angles.⁵³

The potential energy change as the base flips out via major and minor grooves is plotted against the integrated path length in Figure 7. Different open states have been sampled along the two grooves. For cytosine, the open state considered for the minor

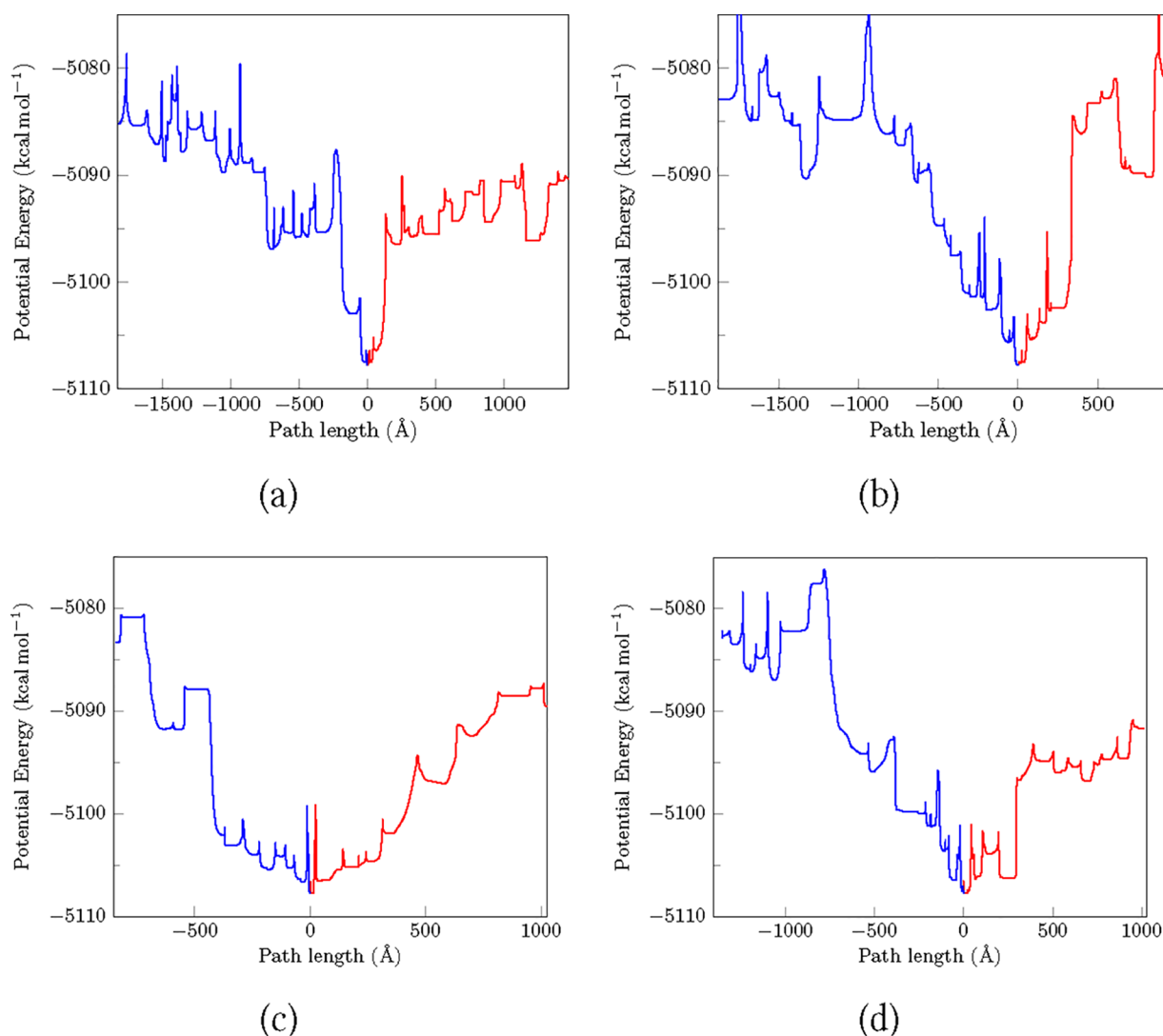


Figure 7. Potential energy as a function of integrated path length for flipping pathways of (a) adenine, (b) guanine, (c) cytosine, and (d) thymine along the major and minor groove. Positive and negative path lengths correspond to flipping along the minor and major groove, respectively. The pathways are between closed state (as shown within Figure 4) and open states labeled as [A] and [B] (Figures 2 and 3) for the base flipped out via major and minor grooves, respectively.

groove has a smaller angle than the open state considered for the major groove.

Giudice et al. have previously reported free energy plots for the base-opening angle. They attributed the initial quadratic increase in energy to the breaking of hydrogen-bonds.⁶⁷ The subsequent linear increase in energy was interpreted in terms of loss of stacking interactions as the base flipped out further. They also reported that purines preferred to flip via the major groove. The effective water bridging that was possible on the major groove side favored base-pair opening.⁶⁷

In contrast, the present study suggests that the purines may prefer the minor groove pathway. This difference might be explained in terms of DNA distortions, which is permitted in the current work, but not in the previous study. Giudice et al. also discussed symmetric base-pair opening for pyrimidines, and found that it was then equally probable for the base to flip out via major or minor grooves.⁶⁷ Indeed, this is the case for thymine in our analysis. However, most pathways that were sampled here involve base-flipping via the major groove for cytosine. Even for some dihedral angles corresponding to the base opened toward

the minor groove, the pathway corresponds to the base moving by more than 180° via the major groove.

Limitations of the Present Study. The present study comes with its own set of limitations. Parametrizing a force field to represent all possible conformations of a large macromolecule like DNA is clearly challenging,¹⁶⁷ and different barriers for flipping the same base in the same sequence have been reported using alternative force fields.^{110,168,169} When the base flips out, its environment changes from hydrophobic within the helix to polyanionic, when it moves close to the backbone. Finally, it ends up in an aqueous environment.¹⁶⁵ During this process, there may be a considerable change in the charge distribution of the nucleobase, which might be better represented by a polarizable force field.¹⁶⁵

A recent X-ray crystallography study revealed the existence of a spine of hydration in the minor groove of DNA,¹⁷⁰ and a computational study revealed that the water channel fills the gap left in the helix at the abasic site.¹⁷¹ In the current investigation, water has been modeled using an implicit solvent model. Comparison with explicit solvent will be considered in future work.

Biomolecules interact with water molecules as well as surrounding ions, and the salt concentration influences the free energies.^{172,173} In an experimental study, magnesium ions were added to an extract of human cells to facilitate DNA repair.¹⁷⁴ Additionally, several investigations have provided evidence for the influence of ions on the width of the minor groove.^{172,173} A recent study has also confirmed that the barrier obtained for base-flipping varies with the salt concentration.¹⁶⁹

The same base in alternative sequences can behave differently.^{44,175} A prominent example of this effect is provided by AT-tracts, which form when four or more AT base-pairs occur together without a 5'-TA-3' step in between. The lifetime of an AT base-pair in AT-tracts is much higher than for an isolated AT base-pair. Conversely, the lifetime of a GC bp in GC-tracts is significantly lower than for an isolated GC bp. Hence, the opening rates of bases in AT-tracts are slower than in GC-tracts.^{68,69,109,111,176}

The sequence also influences the dimensions of grooves in a DNA helix.^{177,178} In particular, sequences rich in AT base-pairs have a relatively narrow minor groove,^{179–181} and NMR studies indicate that the methyl group on the fifth carbon of thymine is the underlying cause.¹⁸² The sequence also determines the stacking of bases within the duplex¹⁸³ and the water-mediated hydrogen-bonding interactions.⁴³ In general, the pathways, barriers, and rates for base-flipping all depend on the sequence.^{184,185} The mechanisms presented in the present work are expected to be generic, however, the details may be specific for the sequence under consideration. In particular, we expect that base opening toward the major or minor groove is likely to be sequence dependent.

CONCLUSIONS

Perhaps the most interesting hypothesis presented here is the relationship between the sequence and the direction of base opening, i.e., toward the major or minor groove. In particular, base-flipping along the minor groove pathway was found to align toward the 5' side of the backbone. The base was found to align toward the 3' side of the backbone when flipping along the major groove pathway. However, in some cases for cytosine and thymine, the base-flipping along the major groove pathway was also found to align toward the 5' side. The sequence effects may be caused by interactions of the flipping base with the neighboring base-pairs on the side toward which it aligns.

Another observation is that purines might prefer to flip via the minor groove pathway. While this suggestion contradicts previous studies,⁶⁷ in which purine flipping toward the minor groove was found to be restricted due to steric reasons, we found that bending and subsequent distortion of the minor groove lower the barrier and promote base-flipping toward the minor groove. A special case occurs when guanine flips along the minor groove pathway because the flipped-out state is relatively stable. The equilibrium constant for base-opening is high and correlated with the reduced rate of base-closing.

Our results may be compared with two experimental observations. First, in an earlier study on the CC mismatched base-pair, both the Cs were found to be aligned toward the 5' direction when the bases were opened toward the minor groove.^{87,160} We find that the alignment toward 5' is associated with flipping toward the minor groove for all four bases: A, G, C, and T. Second, NMR studies show that AT-tracts have a low rate of base-pair opening,⁶⁸ and crystal structure studies reveal that AT-tracts have a narrow minor groove.^{180,181} A computational study reveals that NMR imino proton exchange monitors

flipping of purine bases.¹⁵⁵ These findings may be connected by the preference of flipping via the minor groove pathway for purines, including adenine.

In future work, we will investigate the influence of stretching, bending, and twisting on the energy landscape for base-flipping. The correlation between sequence effects and the direction of opening is another important question that needs to be resolved. One approach would be to study various sequences that are already known to interact with specific enzymes experimentally.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jpcb.2c00340>.

Detailed explanation of the discrete path sampling approach, calculation of free energy and rate constants for multistep pathways, input files for use with the PATHSAMPLE program and OPTIM interfaced with AMBER12, and the CPPTRAJ script used to calculate the CPDb dihedral, along with a figure defining the CPDb dihedral; free energy disconnectivity graphs at 300 K to distinguish flipping via major and minor groove for adenine, guanine, cytosine, and thymine flipped out one at a time (PDF)

AUTHOR INFORMATION

Corresponding Authors

Nicy – Yusuf Hamied Department of Chemistry, University of Cambridge, Cambridge CB2 1EW, U.K.; orcid.org/0000-0003-0245-9977; Email: nn320@cam.ac.uk

Debayan Chakraborty – Department of Chemistry, The University of Texas at Austin, Austin, Texas 78712, United States; orcid.org/0000-0003-4339-5818; Email: debayan.chakraborty@utexas.edu

David J. Wales – Yusuf Hamied Department of Chemistry, University of Cambridge, Cambridge CB2 1EW, U.K.; orcid.org/0000-0002-3555-6645; Email: djw34@cam.ac.uk

Complete contact information is available at: <https://pubs.acs.org/doi/10.1021/acs.jpcb.2c00340>

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

D.J.W. gratefully acknowledges the EPSRC for financial support under EP/N035003/1. Nicy acknowledges the British Council India; the Allen, Meek, and Read Fund, and the Cambridge Commonwealth, European, and International Trust; and the Santander fund, St Edmund's College, University of Cambridge, for funding.

REFERENCES

- (1) Roberts, R. J.; Cheng, X. Base flipping. *Annu. Rev. Biochem.* **1998**, *67*, 181–198.
- (2) Priyakumar, U. D.; MacKerell, A. D. Computational approaches for investigating base flipping in oligonucleotides. *Chem. Rev.* **2006**, *106*, 489–505.
- (3) Klimasauskas, S.; Liutkeviciute, Z. In *DNA and RNA modification enzymes: Structure, mechanism, function and evolution*; Grosjean, H., Ed.; CRC Press: 2009; pp 37–50.

- (4) Klimašauskas, S.; Kumar, S.; Roberts, R. J.; Cheng, X. Hhal methyltransferase flips its target base out of the DNA helix. *Cell* **1994**, *76*, 357–369.
- (5) Klimašauskas, S.; Szyperski, T.; Serva, S.; Wüthrich, K. Dynamic modes of the flipped-out cytosine during Hhal methyltransferase-DNA interactions in solution. *EMBO J.* **1998**, *17*, 317–324.
- (6) Yamagata, Y.; Kato, M.; Odawara, K.; Tokuno, Y.; Nakashima, Y.; Matsushima, N.; Yasumura, K.; Tomita, K. I.; Ihara, K.; Fujii, Y.; Nakabeppu, Y.; Sekiguchi, M.; Fujii, S. Three-dimensional structure of a DNA repair enzyme, 3-methyladenine DNA glycosylase II, from *Escherichia coli*. *Cell* **1996**, *86*, 311–319.
- (7) Slupphaug, G.; Mol, C. D.; Kavli, B.; Arvai, A. S.; Krokan, H. E.; Tainer, J. A. A nucleotide-flipping mechanism from the structure of human uracil-DNA glycosylase bound to DNA. *Nature* **1996**, *384*, 87–92.
- (8) Hollis, T.; Ichikawa, Y.; Ellenberger, T. DNA bending and a flip-out mechanism for base excision by the helix-hairpin-helix DNA glycosylase, *Escherichia coli* AlkA. *EMBO J.* **2000**, *19*, 758–766.
- (9) Bochtler, M.; Szczepanowski, R. H.; Tamulaitis, G.; Grazulis, S.; Czapska, H.; Manakova, E.; Siksnys, V. Nucleotide flips determine the specificity of the *Ec*/18kI restriction endonuclease. *EMBO J.* **2006**, *25*, 2219–2229.
- (10) Christine, K. S.; MacFarlane, A. W., IV; Yang, K.; Stanley, R. J. Cyclobutylpyrimidine dimer base flipping by DNA photolyase. *J. Biol. Chem.* **2002**, *277*, 38339–38344.
- (11) Bird, A. DNA methylation patterns and epigenetic memory. *Genes Dev.* **2002**, *16*, 6–21.
- (12) Goodsell, D. S. Recognition in action: Flipping pyrimidine dimers. *J. Mol. Recognit.* **2005**, *18*, 193–195.
- (13) McCullough, A. K.; Dodson, M. L.; Schärer, O. D.; Lloyd, R. S. The role of base flipping in damage recognition and catalysis by T4 endonuclease V. *J. Biol. Chem.* **1997**, *272*, 27210–27217.
- (14) Banerjee, A.; Yang, W.; Karplus, M.; Verdine, G. L. Structure of a repair enzyme interrogating undamaged DNA elucidates recognition of damaged DNA. *Nature* **2005**, *434*, 612–618.
- (15) Hu, J.; Ma, A.; Dinner, A. R. A two-step nucleotide-flipping mechanism enables kinetic discrimination of DNA lesions by AGT. *Proc. Natl. Acad. Sci. U. S. A.* **2008**, *105*, 4615–4620.
- (16) Schneider, T. D. Strong minor groove base conservation in sequence logos implies DNA distortion or base flipping during replication and transcription initiation. *Nucleic Acids Res.* **2001**, *29*, 4881–4891.
- (17) Yakubovskaya, E.; Mejia, E.; Byrnes, J.; Hambardjiev, E.; Garcia-Diaz, M. Helix unwinding and base flipping enable human MTERF1 to terminate mitochondrial transcription. *Cell* **2010**, *141*, 982–993.
- (18) Cleaver, J. E.; Cortés, F.; Lutze, L. H.; Morgan, W. F.; Player, A. N.; Mitchell, D. L. Unique DNA repair properties of a xeroderma pigmentosum revertant. *Mol. Cell. Biol.* **1987**, *7*, 3353–3357.
- (19) David, S. S.; O'Shea, V. L.; Kundu, S. Base-excision repair of oxidative DNA damage. *Nature* **2007**, *447*, 941–950.
- (20) Reddy Parine, N. R.; Alanazi, I. O.; Shaik, J. P.; Aldhaini, S.; Aljebreen, A. M.; Alharbi, O.; Almadi, M. A.; Azzam, N. A.; Alanazi, M. TDG gene polymorphisms and their possible association with colorectal cancer: A case control study. *J. Oncol.* **2019**, *2019*, 1–9.
- (21) Omeershfudin, U. N. M.; Kumar, S. Bacterial DNA adenine methyltransferase as a novel drug target for antibiotics: Current status and future drug discovery challenges. *Int. J. Curr. Microbiol. Appl. Sci.* **2019**, *8*, 2494–2504.
- (22) Zdorevskiy, O. O.; Volkov, S. N. The possibility of blocking the process of DNA base pairs opening by hydrogen peroxide. *Ukr. J. Phys.* **2019**, *64*, 500–508.
- (23) Mancuso, P.; Tricarico, R.; Bhattacharjee, V.; Cosentino, L.; Kadariya, Y.; Jelinek, J.; Nicolas, E.; Einarson, M.; Beeharry, N.; Devarajan, K.; et al. Thymine DNA glycosylase as a novel target for melanoma. *Oncogene* **2019**, *38*, 3710–3728.
- (24) Jiang, Y. L.; Krosky, D. J.; Seiple, L.; Stivers, J. T. Uracil-directed ligand tethering: An efficient strategy for uracil DNA glycosylase (UNG) inhibitor development. *J. Am. Chem. Soc.* **2005**, *127*, 17412–17420.
- (25) Ianniello, Z.; Paiardini, A.; Fatica, A. N6-methyladenosine (m6A): A promising new molecular target in acute myeloid leukemia. *Front. Oncol.* **2019**, *9*, 1–11.
- (26) Johnson, T. B.; Coghill, R. D. Researches on pyrimidines. C111. The discovery of 5-methyl-cytosine in tuberculinic acid, the nucleic acid of the tubercle bacillus. *J. Am. Chem. Soc.* **1925**, *47*, 2838–2844.
- (27) Wyatt, G. R. Occurrence of 5-methyl-cytosine in nucleic acids. *Nature* **1950**, *166*, 237–238.
- (28) Hotchkiss, R. D. The quantitative separation of purines, and nucleosides by paper chromatography. *J. Biol. Chem.* **1948**, *175*, 315–332.
- (29) Markham, R.; Smith, J. D. Chromatographic studies of nucleic acids. I. A technique for the identification and estimation of purine and pyrimidine bases, nucleosides and related substances. *Biochem* **1949**, *45*, 294–298.
- (30) Meselson, M.; Yuan, R.; Heywood, J. Restriction and modification of DNA. *Annu. Rev. Biochem.* **1972**, *41*, 447–466.
- (31) Mandal, C.; Kallenbach, N. R.; Englander, S. Base-pair opening and closing reactions in the double helix: A stopped-flow hydrogen exchange study in poly (rA) · poly(rU). *J. Mol. Biol.* **1979**, *135*, 391–411.
- (32) Englander, S. W.; Kallenbach, N. R. Hydrogen exchange and structural dynamics of proteins and nucleic acids. *Q. Rev. Biophys.* **1983**, *16*, 521–655.
- (33) Min, J. H.; Pavletich, N. P. Recognition of DNA damage by the Rad4 nucleotide excision repair protein. *Nature* **2007**, *449*, 570–575.
- (34) Hosfield, D. J.; Guan, Y.; Haas, B. J.; Cunningham, R. P.; Tainer, J. A. Structure of the DNA repair enzyme endonuclease IV and its DNA complex: Double-nucleotide flipping at abasic sites and three-metal-ion catalysis. *Cell* **1999**, *98*, 397–408.
- (35) Horton, J. R.; Liebert, K.; Bekes, M.; Jeltsch, A.; Cheng, X. Structure and substrate recognition of the *Escherichia coli* DNA adenine methyltransferase. *J. Mol. Biol.* **2006**, *358*, 559–570.
- (36) Larivière, L.; Moréra, S. A base-flipping mechanism for the T4 phage β -glucosyltransferase and identification of a transition-state analog. *J. Mol. Biol.* **2002**, *324*, 483–490.
- (37) Leroy, J.-l.; Broseta, D.; Guéron, M. Proton exchange and base-pair kinetics of poly (rA) · poly (rU) and poly (rI) · poly (rC). *J. Mol. Biol.* **1985**, *184*, 165–178.
- (38) Guéron, M.; Kochoyan, M.; Leroy, J. L. A single mode of DNA base-pair opening drives imino proton exchange. *Nature* **1987**, *328*, 89–92.
- (39) Russu, I. M. Probing site-specific energetics in proteins and nucleic acids by hydrogen exchange and nuclear magnetic resonance spectroscopy. *Methods Enzymol.* **2004**, *379*, 152–175.
- (40) Cao, C.; Jiang, Y. L.; Stivers, J. T.; Song, F. Dynamic opening of DNA during the enzymatic search for a damaged base. *Nat. Struct. Mol. Biol.* **2004**, *11*, 1230–1236.
- (41) Miller, P. A.; Shajani, Z.; Meints, G. A.; Caplow, D.; Goobes, G.; Varani, G.; Drobny, G. P. Contrasting views of the internal dynamics of the Hhal methyltransferase target DNA reported by solution and solid-state NMR spectroscopy. *J. Am. Chem. Soc.* **2006**, *128*, 15970–15971.
- (42) Dow, B. J.; Malik, S. S.; Drohat, A. C. Defining the role of nucleotide flipping in enzyme specificity using 19F NMR. *J. Am. Chem. Soc.* **2019**, *141*, 4952–4962.
- (43) Banavali, N. K.; MacKerell, A. D. Free energy and structural pathways of base flipping in a DNA GCGC containing sequence. *J. Mol. Biol.* **2002**, *319*, 141–160.
- (44) Giudice, E.; Lavery, R. Nucleic acid base pair dynamics: The impact of sequence and structure using free-energy calculations. *J. Am. Chem. Soc.* **2003**, *125*, 4998–4999.
- (45) Várnai, P.; Canalia, M.; Leroy, J. L. Opening mechanism of G·T/U pairs in DNA and RNA duplexes: A combined study of imino proton exchange and molecular dynamics simulation. *J. Am. Chem. Soc.* **2004**, *126*, 14659–14667.
- (46) Allan, B. W.; Reich, N. O. Targeted base stacking disruption by the EcoRI DNA methyltransferase. *Biochemistry* **1996**, *35*, 14757–14762.

- (47) Holz, B.; Klimášauskas, S.; Serva, S.; Weinhold, E. 2-Aminopurine as a fluorescent probe for DNA base flipping by methyltransferases. *Nucleic Acids Res.* **1998**, *26*, 1076–1083.
- (48) Jean, J. M.; Hall, K. B. 2-Aminopurine fluorescence quenching and lifetimes: Role of base stacking. *Proc. Natl. Acad. Sci. U. S. A.* **2001**, *98*, 37–41.
- (49) Rachofsky, E. L.; Osman, R.; Ross, J. B. Probing structure and dynamics of DNA with 2-aminopurine: Effects of local environment on fluorescence. *Biochemistry* **2001**, *40*, 946–956.
- (50) Neely, R. K.; Daujotyte, D.; Grazulis, S.; Magennis, S. W.; Dryden, D. T.; Klimášauskas, S.; Jones, A. C. Time-resolved fluorescence of 2-aminopurine as a probe of base flipping in M.HhaI-DNA complexes. *Nucleic Acids Res.* **2005**, *33*, 6953–6960.
- (51) Li, X.; Zhu, R.; Yu, A.; Zhao, X. S. Ultrafast photoinduced electron transfer between tetramethylrhodamine and guanosine in aqueous solution. *J. Phys. Chem. B* **2011**, *115*, 6265–6271.
- (52) Yin, Y.; Wang, P.; Yang, X. X.; Li, X.; He, C.; Zhao, X. S. Panorama of DNA hairpin folding observed via diffusion decelerated fluorescence correlation spectroscopy. *Chem. Commun.* **2012**, *48*, 7413–7415.
- (53) Yin, Y.; Yang, L.; Zheng, G.; Gu, C.; Yi, C.; He, C.; Gao, Y. Q.; Zhao, X. S. Dynamics of spontaneous flipping of a mismatched base in DNA duplex. *Proc. Natl. Acad. Sci. U. S. A.* **2014**, *111*, 8043–8048.
- (54) Fuhrmann, A.; Getfert, S.; Fu, Q.; Reimann, P.; Lindsay, S.; Ros, R. Long lifetime of hydrogen-bonded dna basepairs by force spectroscopy. *Biophys. J.* **2012**, *102*, 2381–2390.
- (55) Beckwitt, E. C.; Kong, M.; Van Houten, B. Studying protein-DNA interactions using atomic force microscopy. *Semin. Cell Dev. Biol.* **2018**, *73*, 220–230.
- (56) Janicijevic, A.; Ristic, D.; Wyman, C. The molecular machines of DNA repair: scanning force. *J. Microsc.* **2003**, *212*, 264–272.
- (57) Klimášauskas, S.; Roberts, R. J. M.HhaI binds tightly to substrates containing mismatches at the target base. *Nucleic Acids Res.* **1995**, *23*, 1388–1395.
- (58) Spies, M. A.; Schowen, R. L. The trapping of a spontaneously “flipped-out” base from double helical nucleic acids by host-guest complexation with β -cyclodextrin: The intrinsic base-flipping rate constant for DNA and RNA. *J. Am. Chem. Soc.* **2002**, *124*, 14049–14053.
- (59) Stivers, J. T. Extrahelical damaged base recognition by DNA glycosylase enzymes. *Chem. - A Eur. J.* **2008**, *14*, 786–793.
- (60) Lee, A. J.; Wallace, S. S. Visualizing the search for radiation-damaged DNA bases in real time. *Radiat. Phys. Chem.* **2016**, *128*, 126–133.
- (61) Woźniak, A. K.; Schröder, G. F.; Grubmüller, H.; Seidel, C. A.; Oesterhelt, F. Single-molecule FRET measures bends and kinks in DNA. *Proc. Natl. Acad. Sci. U. S. A.* **2008**, *105*, 18337–18342.
- (62) Holz, B.; Dank, N.; Eickhoff, J. E.; Lipps, G.; Krauss, G.; Weinhold, E. Identification of the binding site for the extrahelical target base in N6-adenine DNA methyltransferases by photo-cross-linking with duplex oligodeoxyribonucleotides containing 5-iodouracil at the target position. *J. Biol. Chem.* **1999**, *274*, 15066–15072.
- (63) Price, M. A.; Tullius, T. D. Using hydroxyl radical to probe DNA structure. *Methods Enzymol.* **1992**, *212*, 194–219.
- (64) Renbaum, P.; Razin, A. Footprint analysis of M.SssI and M.HhaI methyltransferases reveals extensive interactions with the substrate DNA backbone. *J. Mol. Biol.* **1995**, *248*, 19–26.
- (65) Sasse-Dwight, S.; Gralla, J. D. KMnO₄ as a probe for lac promoter DNA melting and mechanism in vivo. *J. Biol. Chem.* **1989**, *264*, 8074–8081.
- (66) Bischerour, J.; Chalmers, R. Base-flipping dynamics in a DNA hairpin processing reaction. *Nucleic Acids Res.* **2007**, *35*, 2584–2595.
- (67) Giudice, E.; Várnai, P.; Lavery, R. Energetic and conformational aspects of A:T base-pair opening within the DNA double helix. *ChemPhysChem* **2001**, *2*, 673–677.
- (68) Leroy, J. L.; Charretier, E.; Kochoyan, M.; Gueron, M. Evidence from base-pair kinetics for two types of adenine tract structures in solution: Their relation to DNA curvature. *Biochemistry* **1988**, *27*, 8894–8898.
- (69) Dornberger, U.; Leijon, M.; Fritzsche, H. High base pair opening rates in tracts of GC base pairs. *J. Biol. Chem.* **1999**, *274*, 6957–6962.
- (70) Dallmann, A.; Dehmel, L.; Peters, T.; Mügge, C.; Griesinger, C.; Tuma, J.; Ernsting, N. P. 2-Aminopurine incorporation perturbs the dynamics and structure of DNA. *Angew. Chemie - Int. Ed.* **2010**, *49*, 5989–5992.
- (71) Stivers, J. T.; Jiang, Y. L. A mechanistic perspective on the chemistry of DNA repair glycosylases. *Chem. Rev.* **2003**, *103*, 2729–2759.
- (72) Endo, M. AFM-based single-molecule observation of the conformational changes of DNA structures. *Methods* **2019**, *169*, 3–10.
- (73) Ramstein, J.; Lavery, R. Energetic coupling between DNA bending and base pair opening. *Proc. Natl. Acad. Sci. U. S. A.* **1988**, *85*, 7231–7235.
- (74) Lavery, R.; Zakrzewska, K.; Sklenar, H. JUMNA (junction minimisation of nucleic acids). *Comput. Phys. Commun.* **1995**, *91*, 135–158.
- (75) Briki, F.; Ramstein, J.; Genest, D.; Lavery, R. Evidence for the stochastic nature of base pair opening in DNA: A brownian dynamics simulation. *J. Am. Chem. Soc.* **1991**, *113*, 2490–2493.
- (76) Bernet, J.; Zakrzewska, K.; Lavery, R. Modelling base pair opening: The role of helical twist. *J. Mol. Struct. THEOCHEM* **1997**, *398–399*, 473–482.
- (77) Torrie, G. M.; Valleau, J. Nonphysical sampling distributions in monte carlo free-energy estimation: Umbrella sampling. *J. Comput. Phys.* **1977**, *23*, 187–199.
- (78) Chen, Y. Z.; Mohan, V.; Griffey, R. H. Spontaneous base flipping in DNA and its possible role in methyltransferase binding. *Phys. Rev. E* **2000**, *62*, 1133–1137.
- (79) Várnai, P.; Lavery, R. Base flipping in DNA: Pathways and energetics studied with molecular dynamic simulations. *J. Am. Chem. Soc.* **2002**, *124*, 7272–7273.
- (80) Gurry, T.; Stultz, C. M. Mechanism of amyloid- β fibril elongation. *Biochemistry* **2014**, *53*, 6981–6991.
- (81) Dellago, C.; Bolhuis, P. G.; Csajka, F. S.; Chandler, D. Transition path sampling and the calculation of rate constants. *J. Chem. Phys.* **1998**, *108*, 1964–1977.
- (82) Röder, K.; Joseph, J. A.; Husic, B. E.; Wales, D. J. Energy Landscapes for Proteins: From Single Funnel to Multifunctional Systems. *Adv. Theory Simulations* **2019**, *2*, 1800175.
- (83) Wales, D. J. Perspective: Insight into reaction coordinates and dynamics from the potential energy landscape. *J. Chem. Phys.* **2015**, *142*, 130901.
- (84) Lindahl, V.; Villa, A.; Hess, B. Sequence dependency of canonical base pair opening in the DNA double helix. *PLoS Comput. Biol.* **2017**, *13*, e1005463.
- (85) Bouvier, B.; Grubmüller, H. A molecular dynamics study of slow base flipping in DNA using conformational flooding. *Biophys. J.* **2007**, *93*, 770–786.
- (86) Laio, A.; Parrinello, M. Escaping free-energy minima. *Proc. Natl. Acad. Sci. U.S.A.* **2002**, *99*, 12562–12576.
- (87) Kingsland, A.; Maibaum, L. DNA base pair mismatches induce structural changes and alter the free-energy landscape of base flip. *J. Phys. Chem. B* **2018**, *122*, 12251–12259.
- (88) Hagan, M. F.; Dinner, A. R.; Chandler, D.; Chakraborty, A. K. Atomistic understanding of kinetic pathways for single base-pair binding and unbinding in DNA. *Proc. Natl. Acad. Sci. U. S. A.* **2003**, *100*, 13922–13927.
- (89) Cao, L.; Lv, C.; Yang, W. Hidden conformation events in DNA base extrusions: A generalized-ensemble path optimization and equilibrium simulation study. *J. Chem. Theory Comput.* **2013**, *9*, 3756–3768.
- (90) Chakraborty, D.; Collepardo-Guevara, R.; Wales, D. J. Energy landscapes, folding mechanisms, and kinetics of RNA tetraloop hairpins. *J. Am. Chem. Soc.* **2014**, *136*, 18052–18061.
- (91) Chakraborty, D.; Wales, D. J. Probing helical transitions in a DNA duplex. *Phys. Chem. Chem. Phys.* **2017**, *19*, 878–892.

- (92) Chakraborty, D.; Wales, D. J. Energy landscape and pathways for transitions between Watson-Crick and Hoogsteen base pairing in DNA. *J. Phys. Chem. Lett.* **2018**, *9*, 229–241.
- (93) Chakraborty, D.; Wales, D. J. Dynamics of an adenine-adenine RNA conformational switch from discrete path sampling. *J. Chem. Phys.* **2019**, *150*, 125101.
- (94) Case, D. A.; Walker, R. C.; Cheatham, T. E.; Simmerling, C.; Roitberg, A.; Merz, K. M.; Luo, R.; Darden, T. *Amber 18*; 2018; <http://ambermd.org/doc12/Amber18.pdf>.
- (95) Weiner, S. J.; Kollman, P. A.; Singh, U. C.; Case, D. A.; Ghio, C.; Alagona, G.; Profeta, S.; Weiner, P. A new force field for molecular mechanical simulation of nucleic acids and proteins. *J. Am. Chem. Soc.* **1984**, *106*, 765–784.
- (96) Weiner, S. J.; Kollman, P. A.; Nguyen, D. T.; Case, D. A. An all atom force field for simulations of proteins and nucleic acids. *J. Comput. Chem.* **1986**, *7*, 230–252.
- (97) Malolepsza, E.; Strodel, B.; Khalili, M.; Trygubenko, S.; Fejer, S.; Wales, D. J. Symmetrisation of the AMBER and CHARMM force fields. *J. Comput. Chem.* **2010**, *31*, 1402–1409.
- (98) Pérez, A.; Marchán, I.; Svozil, D.; Šponer, J.; Cheatham, T. E.; Lughton, C. A.; Orozco, M. Refinement of the AMBER force field for nucleic acids: Improving the description of α/γ conformers. *Biophys. J.* **2007**, *92*, 3817–3829.
- (99) Zgarbová, M.; Otyepka, M.; Šponer, J.; Mládek, A.; Banáš, P.; Cheatham, T. E.; Jurečka, P. Refinement of the Cornell et al. Nucleic acids force field based on reference quantum chemical calculations of glycosidic torsion profiles. *J. Chem. Theory Comput.* **2011**, *7*, 2886–2902.
- (100) Zgarbová, M.; Luque, F. J.; Šponer, J.; Cheatham, T. E.; Otyepka, M.; Jurečka, P. Toward improved description of DNA backbone: Revisiting epsilon and zeta torsion force field parameters. *J. Chem. Theory Comput.* **2013**, *9*, 2339–2354.
- (101) Galindo-Murillo, R.; Robertson, J. C.; Zgarbová, M.; Šponer, J.; Otyepka, M.; Jurečka, P.; Cheatham, T. E. Assessing the current state of Amber force field modifications for DNA. *J. Chem. Theory Comput.* **2016**, *12*, 4114–4127.
- (102) Krepl, M.; Zgarbová, M.; Stadlbauer, P.; Otyepka, M.; Banáš, P.; Koča, J.; Cheatham, T. E.; Jurečka, P.; Šponer, J. Reference simulations of noncanonical nucleic acids with different χ variants of the AMBER force field: Quadruplex DNA, quadruplex RNA, and Z-DNA. *J. Chem. Theory Comput.* **2012**, *8*, 2506–2520.
- (103) Onufriev, A.; Bashford, D.; Case, D. A. Modification of the generalized Born model suitable for macromolecules. *J. Phys. Chem. B* **2000**, *104*, 3712–3720.
- (104) Onufriev, A.; Bashford, D.; Case, D. A. Exploring protein native states and large-scale conformational changes with a modified generalized born model. *Proteins* **2004**, *55*, 383–394.
- (105) Wells, R. D.; Collier, D. A.; Hanvey, J. C.; Shimizu, M.; Wohlrab, F. The chemistry and biology of unusual DNA structures adopted by oligopurine-oligopyrimidine sequences. *FASEB J.* **1988**, *2*, 2939–2949.
- (106) GMIN: A program for basin-hopping global optimization, basin-sampling, and parallel tempering. <http://www-wales.ch.cam.ac.uk/software.html>.
- (107) Mochizuki, K.; Whittleston, C. S.; Somani, S.; Kusumaatmaja, H.; Wales, D. J. A conformational factorisation approach for estimating the binding free energies of macromolecules. *Phys. Chem. Chem. Phys.* **2014**, *16*, 2842–2853.
- (108) Oakley, M. T.; Johnston, R. L. Energy landscapes and global optimization of self-assembling cyclic peptides. *J. Chem. Theory Comput.* **2014**, *10*, 1810–1816.
- (109) Chen, C.; Russu, I. M. Sequence-dependence of the energetics of opening of at basepairs in DNA. *Biophys. J.* **2004**, *87*, 2545–2551.
- (110) Wang, X.; Sun, Z. Determination of base-flipping free-energy landscapes from nonequilibrium stratification. *J. Chem. Inf. Model.* **2019**, *59*, 2980–2994.
- (111) Leijon, M.; Gräslund, A. Effects of sequence and length on imino proton exchange and base pair opening kinetics in DNA oligonucleotide duplexes. *Nucleic Acids Res.* **1992**, *20*, 5339–5343.
- (112) Song, K.; Campbell, A. J.; Bergonzo, C.; de los Santos, C.; Grollman, A. P.; Simmerling, C. An improved reaction coordinate for nucleic acid base flipping studies. *J. Chem. Theory Comput.* **2009**, *5*, 3105–3113.
- (113) Wales, D. J. Discrete path sampling. *Mol. Phys.* **2002**, *100*, 3285–3306.
- (114) Wales, D. J. Some further applications of discrete path sampling to cluster isomerization. *Mol. Phys.* **2004**, *102*, 891–908.
- (115) OPTIM: A program for geometry optimization and pathway calculations. <http://www-wales.ch.cam.ac.uk/software.html>.
- (116) Wales, D. J. *Energy landscapes: Applications to clusters, biomolecules and glasses*; Cambridge University Press: 2004.
- (117) Joseph, J. A.; Röder, K.; Chakraborty, D.; Mantell, R. G.; Wales, D. J. Exploring biomolecular energy landscapes. *Chem. Commun.* **2017**, *53*, 6974–6988.
- (118) Wales, D. J. Exploring energy landscapes. *Annu. Rev. Phys. Chem.* **2018**, *69*, 401–425.
- (119) Trygubenko, S. A.; Wales, D. J. A doubly nudged elastic band method for finding transition states. *J. Chem. Phys.* **2004**, *120*, 2082–2094.
- (120) Sheppard, D.; Terrell, R.; Henkelman, G. Optimization methods for finding minimum energy paths. *J. Chem. Phys.* **2008**, *128*, 134106.
- (121) Mills, G.; Jónsson, H.; Schenter, G. K. Reversible work transition state theory: application to dissociative adsorption of hydrogen. *Surf. Sci.* **1995**, *324*, 305–337.
- (122) Jónsson, H.; Mills, G.; Jacobsen, K. W. In *Classical and Quantum Dynamics in Condensed Phase Simulations*; Berne, B. J., Ciccotti, G., Coker, D. F., Eds.; World Scientific: Singapore, 1998; Chapter 16, pp 385–404.
- (123) Henkelman, G.; Uberuaga, B. P.; Jónsson, H. A climbing image nudged elastic band method for finding saddle points and minimum energy paths. *J. Chem. Phys.* **2000**, *113*, 9901–9904.
- (124) Henkelman, G.; Jónsson, H. Improved tangent estimate in the nudged elastic band method for finding minimum energy paths and saddle points. *J. Chem. Phys.* **2000**, *113*, 9978–9985.
- (125) Munro, L. J.; Wales, D. J. Defect migration in crystalline silicon. *Phys. Rev. B* **1999**, *59*, 3969–3980.
- (126) Nocedal, J. Updating quasi-Newton matrices with limited storage. *Mathematics of Computation* **1980**, *35*, 773–782.
- (127) Liu, D.; Nocedal, J. On the limited memory BFGS method for large scale optimization. *Math. Prog.* **1989**, *45*, 503–528.
- (128) Carr, J. M.; Trygubenko, S. A.; Wales, D. J. Finding pathways between distant local minima. *J. Chem. Phys.* **2005**, *122*, 234903.
- (129) Becker, O. M.; Karplus, M. The topology of multidimensional potential energy surfaces: Theory and application to peptide structure and kinetics. *J. Chem. Phys.* **1997**, *106*, 1495–1517.
- (130) Wales, D. J.; Miller, M. A.; Walsh, T. R. Archetypal energy landscapes. *Nature* **1998**, *394*, 758–760.
- (131) Strodel, B.; Whittleston, C. S.; Wales, D. J. Thermodynamics and kinetics of aggregation for the GNNQQNY peptide. *J. Am. Chem. Soc.* **2007**, *129*, 16005–16014.
- (132) Evans, D. A.; Wales, D. J. Folding of the GB1 hairpin peptide from discrete path sampling. *J. Chem. Phys.* **2004**, *121*, 1080–1090.
- (133) Carr, J. M.; Wales, D. J. Folding pathways and rates for the three-stranded beta-sheet peptide Beta3s using discrete path sampling. *J. Phys. Chem. B* **2008**, *112*, 8760–8769.
- (134) Trygubenko, S. A.; Wales, D. J. Kinetic analysis of discrete path sampling stationary point databases. *Mol. Phys.* **2006**, *104*, 1497–1507.
- (135) Wales, D. J. Calculating rate constants and committer probabilities for transition networks by graph transformation. *J. Chem. Phys.* **2009**, *130*, 204111.
- (136) Swinburne, T. D.; Wales, D. J. Defining, calculating, and converging observables of a kinetic transition network. *J. Chem. Theory Comput.* **2020**, *16*, 2661–2679.
- (137) Keepers, J. W.; Kollman, P. A.; Weiner, P. K.; James, T. L. Molecular mechanical studies of DNA flexibility: Coupled backbone torsion angles and base-pair openings. *Proc. Natl. Acad. Sci. U. S. A.* **1982**, *79*, 5537–5541.

- (138) Keepers, J. O. E.; Kollman, P. A.; James, T. L. Molecular mechanical studies of base-pair opening in d(CGCGC):d(GCGCG), dG₅·dC₅, d(TATAT):d(ATATA), and dA₅·dT₅ in the B and Z forms of DNA. *Biopolymers* **1984**, *23*, 2499–2511.
- (139) Feig, M.; Zacharias, M.; Pettitt, B. M. Conformations of an adenine bulge in a DNA octamer and its influence on DNA structure from molecular dynamics simulations. *Biophys. J.* **2001**, *81*, 352–370.
- (140) Drew, H. R.; Wing, R. M.; Takano, T.; Broka, C.; Tanaka, S.; Itakura, K.; Dickerson, R. E. Structure of a B-DNA dodecamer: conformation and dynamics. *Proc. Natl. Acad. Sci. U. S. A.* **1981**, *78*, 2179–2183.
- (141) Chen, Y. Z.; Mohan, V.; Griffey, R. H. The opening of a single base without perturbations of neighboring nucleotides: A study on crystal b-DNA duplex d(CGCGAATTCGCG)₂. *J. Biomol. Struct. Dyn.* **1998**, *15*, 765–777.
- (142) Lee, B.; Richards, F. M. The interpretation of protein structures: Estimation of static accessibility. *J. Mol. Biol.* **1971**, *55*, 379.
- (143) Weiser, J.; Shenkin, P. S.; Still, W. C. Approximate atomic surfaces from linear combinations of pairwise overlaps (LCPO). *J. Comput. Chem.* **1999**, *20*, 217–230.
- (144) Wodak, S. J.; Janin, J. Analytical approximation to the accessible surface area of proteins. *Proc. Natl. Acad. Sci. U. S. A.* **1980**, *77*, 1736–1740.
- (145) Weiser, J.; Weiser, A. A.; Shenkin, P. S.; Still, W. C. Neighbor-list reduction: Optimization for computation of molecular van der Waals and solvent-accessible surface areas. *J. Comput. Chem.* **1998**, *19*, 797–808.
- (146) Hasel, W.; Hendrickson, T. F.; Still, W. C. A rapid approximation to the solvent accessible surface areas of atoms. *Tetrahedron Comput. Methodol.* **1988**, *1*, 103–116.
- (147) Kleinjung, J.; Fraternali, F. POPSCOMP: An automated interaction analysis of biomolecular complexes. *Nucleic Acids Res.* **2005**, *33*, W342–W346.
- (148) Fraternali, F. Parameter optimized surfaces (POPS): analysis of key interactions and conformational changes in the ribosome. *Nucleic Acids Res.* **2002**, *30*, 2950–2960.
- (149) Cavallo, L.; Kleinjung, J.; Fraternali, F. POPS: A fast algorithm for solvent accessible surface areas at atomic and residue level. *Nucleic Acids Res.* **2003**, *31*, 3364–3366.
- (150) Roe, D. R.; Cheatham, T. E. PTRAJ and CPPTRAJ: Software for processing and analysis of molecular dynamics trajectory data. *J. Chem. Theory Comput.* **2013**, *9*, 3084–3095.
- (151) Imhof, P.; Zahran, M. The effect of a G:T mismatch on the dynamics of DNA. *PLoS One* **2013**, *8*, e53305.
- (152) Krivov, S. V.; Karplus, M. Free energy disconnectivity graphs: Application to peptide models. *J. Chem. Phys.* **2002**, *117*, 10894–10903.
- (153) Evans, D. A.; Wales, D. J. Free energy landscapes of model peptides and proteins. *J. Chem. Phys.* **2003**, *118*, 3891–3897.
- (154) Ma, N.; van der Vaart, A. Free energy coupling between DNA bending and base flipping. *J. Chem. Inf. Model.* **2017**, *57*, 2020–2026.
- (155) Priyakumar, U. D.; MacKerell, A. D. NMR imino proton exchange experiments on duplex DNA primarily monitor the opening of purine bases. *J. Am. Chem. Soc.* **2006**, *128*, 678–679.
- (156) Zheng, H.; Cai, Y.; Ding, S.; Tang, Y.; Kropachev, K.; Zhou, Y.; Wang, L.; Wang, S.; Geacintov, N. E.; Zhang, Y.; Broyde, S. Base flipping free energy profiles for damaged and undamaged DNA. *Chem. Res. Toxicol.* **2010**, *23*, 1868–1870.
- (157) Cheng, X.; Blumenthal, R. M. Finding a basis for flipping bases. *Structure* **1996**, *4*, 639–645.
- (158) O'Neil, L. L.; Wiest, O. Sequence dependence in base flipping: Experimental and computational studies. *Org. Biomol. Chem.* **2008**, *6*, 485–492.
- (159) O'Neil, L. L.; Wiest, O. Structures and energetics of base flipping of the thymine dimer depend on DNA sequence. *J. Phys. Chem. B* **2008**, *112*, 4113–4122.
- (160) Gao, X.; Huang, X.; Smith, G. K.; Zheng, M.; Liu, H. New antiparallel duplex motif of DNA CCG repeats that is stabilized by extrahelical bases symmetrically located in the minor groove. *J. Am. Chem. Soc.* **1995**, *117*, 8883–8884.
- (161) Coman, D.; Russu, I. M. A nuclear magnetic resonance investigation of the energetics of basepair opening pathways in DNA. *Biophys. J.* **2005**, *89*, 3285–3292.
- (162) Kumar, S.; Cheng, X.; Klimašauskas, S.; Sha, M.; Posfai, J.; Roberts, R. J.; Wilson, G. G. The DNA (cytosine-5) methyltransferases. *Nucleic Acids Res.* **1994**, *22*, 1–10.
- (163) O'Gara, M.; Klimašauskas, S.; Roberts, R. J.; Cheng, X. Enzymatic C5-cytosine methylation of DNA: Mechanistic implications of new crystal structures for HhaI methyltransferase-DNA-AdoHcy complexes. *J. Mol. Biol.* **1996**, *261*, 634–645.
- (164) Giudice, E.; Várnai, P.; Lavery, R. Base pair opening within B-DNA: Free energy pathways for GC and AT pairs from umbrella sampling simulations. *Nucleic Acids Res.* **2003**, *31*, 1434–1443.
- (165) Lemkul, J. A.; Savelyev, A.; MacKerell, A. D. Induced polarization influences the fundamental forces in DNA base flipping. *J. Phys. Chem. Lett.* **2014**, *5*, 2077–2083.
- (166) Cao, C.; Jiang, Y. L.; Krosky, D. J.; Stivers, J. T. The catalytic power of uracil DNA glycosylase in the opening of thymine base pairs. *J. Am. Chem. Soc.* **2006**, *128*, 13034–13035.
- (167) Šponer, J.; Banáš, P.; Jurečka, P.; Zgarbová, M.; Kührová, P.; Havrila, M.; Krepl, M.; Stadlbauer, P.; Otyepka, M. Molecular dynamics simulations of nucleic acids. From tetranucleotides to the ribosome. *J. Phys. Chem. Lett.* **2014**, *5*, 1771–1782.
- (168) Priyakumar, U. D.; MacKerell, A. D. Base flipping in a GCGC containing DNA dodecamer: A comparative study of the performance of the nucleic acid force fields, CHARMM, AMBER, and BMS. *J. Chem. Theory Comput.* **2006**, *2*, 187–200.
- (169) Sun, Z.; Zhang, J. Z. Thermodynamic insights of base flipping in TNA duplex: force fields, salt concentrations, and free-energy simulation methods. *CCS Chemistry* **2021**, *3*, 1026–1039.
- (170) McDermott, M. L.; Vanselow, H.; Corcelli, S. A.; Petersen, P. B. DNA's chiral spine of hydration. *ACS Cent. Sci.* **2017**, *3*, 708–714.
- (171) Barsky, D.; Foloppe, N.; Ahmadi, S.; Wilson, D. M.; MacKerell, A. D. New insights into the structure of abasic DNA from molecular dynamics simulations. *Nucleic Acids Res.* **2000**, *28*, 2613–2626.
- (172) Hamelberg, D.; McFail-Isom, L.; Williams, L. D.; Wilson, W. D. Flexible structure of DNA: Ion dependence of minor-groove structure and dynamics. *J. Am. Chem. Soc.* **2000**, *122*, 10513–10520.
- (173) Hamelberg, D.; Williams, L. D.; Wilson, W. D. Influence of the dynamic positions of cations on the structure of the DNA minor groove: Sequence-dependent effects. *J. Am. Chem. Soc.* **2001**, *123*, 7745–7755.
- (174) Thomas, D. C.; Roberts, J. D.; Kunkel, T. A. Heteroduplex repair in extracts of human HeLa cells. *J. Biol. Chem.* **1991**, *266*, 3744–3751.
- (175) Folta-Stogniew, E.; Russu, I. M. Sequence dependence of base-pair opening in a DNA dodecamer containing the CACA/GTGT sequence motif. *Biochemistry* **1994**, *33*, 11016–11024.
- (176) Moe, J. G.; Russu, I. M. Proton exchange and base-pair opening kinetics in 5'-d(CGCGAATTCGCG)-3' and related dodecamers. *Nucleic Acids Res.* **1990**, *18*, 821–827.
- (177) Langley, D. R. Molecular dynamic simulations of environment and sequence dependent dna conformations: The development of the bms nucleic acid force field and comparison with experimental results. *J. Biomol. Struct. Dyn.* **1998**, *16*, 487–509.
- (178) Hancock, S. P.; Ghane, T.; Cascio, D.; Rohs, R.; Di Felice, R.; Johnson, R. C. Control of DNA minor groove width and Fis protein binding by the purine 2-amino group. *Nucleic Acids Res.* **2013**, *41*, 6750–6760.
- (179) Koo, H. S.; Wu, H. M.; Crothers, D. M. DNA bending at adenine·thymine tracts. *Nature* **1986**, *320*, 501–506.
- (180) Nelson, H. C.; Finch, J. T.; Luisi, B. F.; Klug, A. The structure of an oligo(dA)·oligo(dT) tract and its biological implications. *Nature* **1987**, *330*, 221–226.
- (181) Coll, M.; Frederick, C. A.; Wang, A. H.; Rich, A. A bifurcated hydrogen-bonded conformation in the d(A.T) base pairs of the DNA dodecamer d(CGCAAATTTGCG) and its complex with distamycin. *Proc. Natl. Acad. Sci. U. S. A.* **1987**, *84*, 8385–8389.

(182) Wärmländer, S.; Sponer, J. E.; Sponer, J.; Leijon, M. The influence of the thymine C5 methyl group on spontaneous base pair breathing in DNA. *J. Biol. Chem.* **2002**, *277*, 28491–28497.

(183) Hunter, C. A. Sequence-dependent DNA structure. *J. Mol. Biol.* **1993**, *230*, 1025–1054.

(184) Oguey, C.; Foloppe, N.; Hartmann, B. Understanding the sequence-dependence of DNA groove dimensions: Implications for DNA interactions. *PLoS One* **2010**, *5*, e15931.

(185) Krueger, A.; Protozanova, E.; Frank-Kamenetskii, M. D. Sequence-dependent basepair opening in DNA double helix. *Biophys. J.* **2006**, *90*, 3091–3099.